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## Supporting Information

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White-Light-Emitting Melamine-Formaldehyde Microspheres through Polymer-Mediated Aggregation and Encapsulation of Graphene Quantum Dots

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#### **Supporting Information**

# White-light-emitting melamine-formaldehyde microspheres through polymer mediated aggregation and encapsulation of graphene quantum dots

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Figure S1. AFM image and topographic heights of the GQDs.



**Figure S2.** XPS spectra (full survey, C1s, N1s, O1s) of the GQDs (A). Deconvoluted XPS spectra for C1s (B) and N1s (C).



**Figure S3.** UV-Vis absorption and fluorescence spectra of the GQDs solution, excited at different wavelengths (A). UV-Vis absorptions of the series of GQDs-microspheres (B), the GQDs solution (0.1% wt.) and of the GQDs-MF prepolymer mixture (C1 and C2).



Figure S4. pH-zeta potential curves of GQDs and MF prepolymer in aqueous solutions

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Figure S5. SEM images of GQDs-MF microsphere samples prepared with varied GQDs doping concentrations.



Figure S6. FTIR spectra of the GQDs, GQDs-MF microspheres and blank MF microspheres.



**Figure S7.** Emissions spectra of the GQDs solution (A) and aqueous suspension of GQDs-MF microspheres at varied temperatures, excited by 360 nm.



**Figure S8.** Emission spectra and maximum emission intensity of the GQDs solution and aqueous suspension of GQDs-MF microspheres in solutions of different pH, excited at 360 nm.



**Figure S9.** Emission spectra of the GQDs and GQDs extracted from the decomposed GQDs-MF microspheres, the spectra are normalized by maximum intensity, excited at 360 nm.



**Figure S10.** Time-resolved luminescence decay curves of the GQDs solution and series of GQDs-MF microspheres. The emissions were collected at 450 nm, excited at 340 nm.



**Figure S11.** A comparison of the photo-stability of the GQDs solution, GQDs-MF microspheres, Rhodamine 110, and Fluorescein, under UV irradiation by a 300W high pressure mercury lamp

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**Figure S12.** White-light-emitting block and film prepared with GQDs-MF-PMMA and GQDs-MF-PVA materials.

Doping Concentration	QY	CCT (K)	CRI	CIE
0.1% wt solution	0.55	NA	NA	(0.16, 0.14)
2.0% wt.	0.83	10810	0.75	(0.28, 0.28)
2.5% wt.	0.74	7814	0.81	(0.30, 0.29)
3.0% wt.	0.62	6653	0.84	(0.31, 0.31)
3.5% wt.	0.51	5969	0.87	(0.32, 0.32)
4.0% wt.	0.43	5638	0.88	(0.33, 0.32)
<ul><li>3.0% wt.</li><li>3.5% wt.</li><li>4.0% wt.</li></ul>	0.62 0.51 0.43	6653 5969 5638	0.84 0.87 0.88	(0.31, 0.31) (0.32, 0.32) (0.33, 0.32)

**Table S1.** QY, CCT, CRI and CIE coordinates for the GQDs solution and GQDs-MF microspheres prepared with different GQDs doping concentrations.

Author and Date	CIE	CRI	QY	EQE	Ref.
This work	(0.28, 0.28); (0.30, 0.29); (0.31, 0.31); (0.32, 0.32); (0.33, 0.32);	0.75-0.88	43-83%	NA	This work
P. Dong et. al. 2017	(0.33, 0.36)	83.9	3.62%	NA	23
Z. Luo et. al. 2016	(0.24, 0.25); (0.25, 0.27); (0.26, 0.28); (0.27, 0.29)	NA	NA	0.24 - 0.19%	46
T. Ghosh et. al. 2015	(0.29, 0.34)	NA	1-2%	NA	25
R. Sekiya et. al. 2014	(0.24, 0.27); (0.34, 0.40); (0.34, 0.39); (0.32, 0.38)	NA	1-2%	NA	21
C.M. Luk et. al. 2012	(0.33. 0.38)	72.0	12%	NA	24

**Table S2.** A comparison of some recently developed GQDs based white-light-emitting materials.