

Fig. 1S. The IR spectrum of [4-NH₂-2-Me(Q)H][VO(bcma)(H₂O)]2H₂O.



Fig. 2S. The IR spectrum of [4-NH₂-2-Me(Q)H][VO(ceida)(H₂O)]3H₂O.



Fig. 3S. The IR spectrum of 4-NH₂-2-Me(Q).



Fig. 4S. The temperature dependence of experimental $\chi_m T$ and $\chi_m (\chi_m$ per one V(IV) ion) for [4-NH₂-2-Me(Q)H][VO(bcma)(H₂O)]2H₂O.



Fig. 5S. The temperature dependence of experimental $\chi_m T$ and $\chi_m (\chi_m$ per one V(IV) ion) for [4-NH₂-2-Me(Q)H][VO(ceida)(H₂O)]3H₂O.



Fig. 6S. Concentration distribution curves of the species presented in the $[4-NH_2-2-Me(Q)H]-$ [VO(bcma)(H₂O)]2H₂O solution as a function of the pH calculated based on the stability constants listed in Table 1.



Fig. 7S. Concentration distribution curves of the species presented in the $[4-NH_2-2-Me(Q)H]_2[VO(pmida)]3H_2O$ solution as a function of the pH calculated based on the stability constants listed in Table 1.



Fig. 8S. Concentration distribution curves of the species presented in the $[4-NH_2-2-Me(Q)H][VO(ceida)(H_2O)]3H_2O$ solution as a function of the pH calculated based on the stability constants listed in Table 1.