

# Catalytic One-Step Deoxytrifluoromethylation of Alcohols

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## Supporting Information – Experimental Details

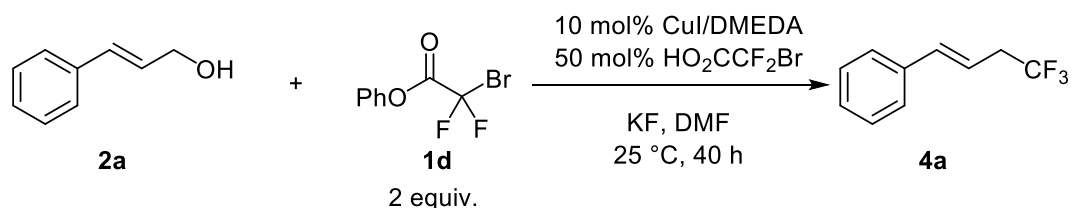
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## Stability of the Phenyl Bromodifluoroacetate (**1d**)

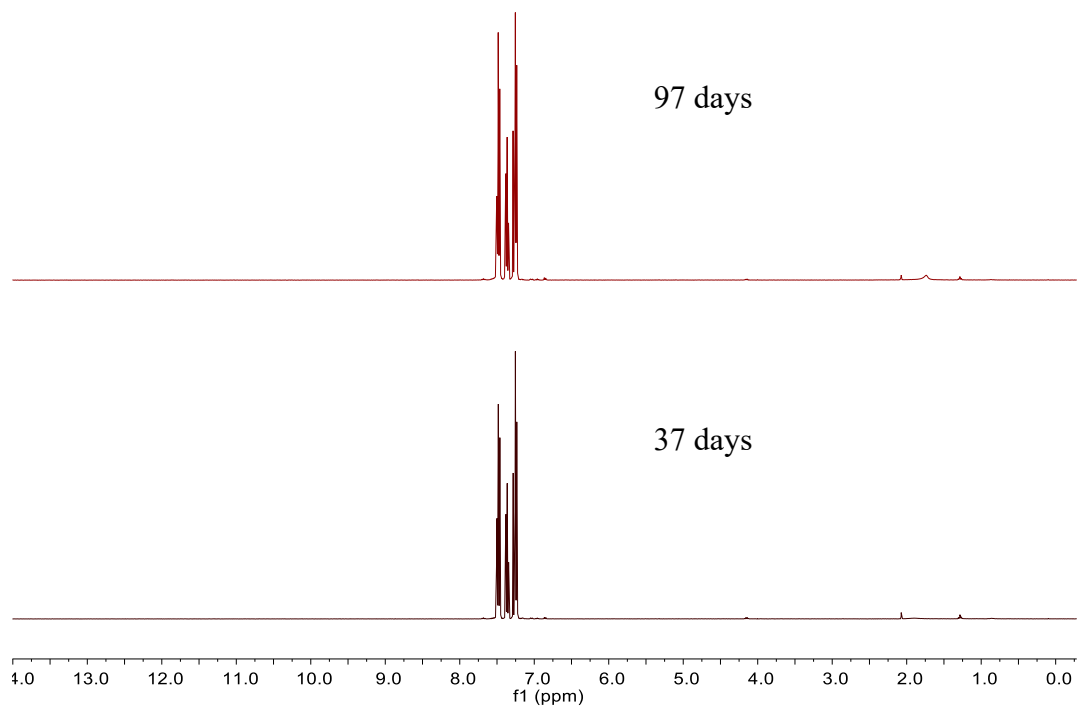
The stability of phenyl bromodifluoroacetate (**1d**) was tested by storing it in 1-dram vials sealed with a PTFE-lined screw cap under three distinct conditions, namely 1) on the bench, 2) in a desiccator, and 3) in the freezer. The stability of the compound was evaluated by <sup>1</sup>H- and <sup>19</sup>F-NMR analysis (Figure S1 and S2), as well as the deoxytrifluoromethylation reaction of cinnamyl alcohol (**2a**) (Scheme S1). The compound was stable in all situations and showed no signs of decomposition by NMR. The reaction yields decreased as the reagent aged. However, the reaction yields were still satisfactory even after 6–8 months. During this work, the only decomposition pathway observed was the hydrolysis of the compound not properly protected against residual moisture. No light sensitivity was observed.



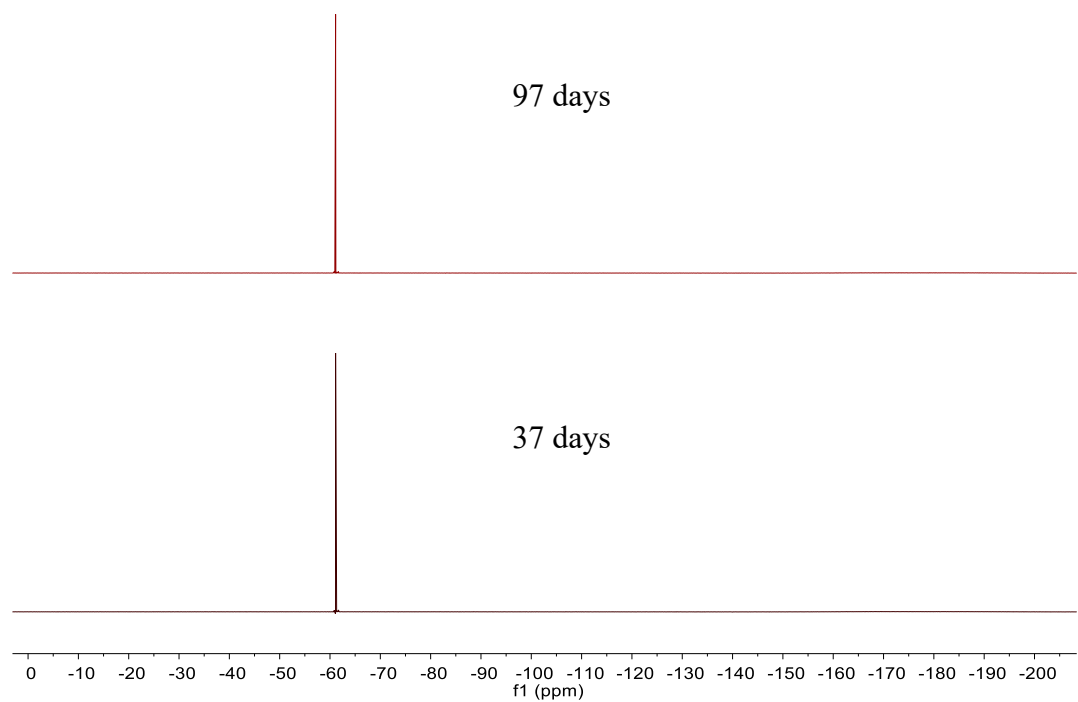
Period	Storage Condition		
	Bench	Desiccator	Freezer
<i>Fresh reagent</i>			86
37 days	86	82	
97 days	75	70	
183 days			73
242 days			67

<sup>a</sup> <sup>19</sup>F-NMR Yields.

**Scheme S1:** Phenyl bromodifluoroacetate (**1d**) provided satisfactory yields of trifluoromethyl compound **4a** after extended storage on the bench, in the desiccator, or in the freezer.



**Figure S1:** Representative  $^1\text{H}$ -NMR spectra of phenyl bromodifluoroacetate (**1d**) obtained during the study of its stability. The spectra showed are from the sample stored on the bench.



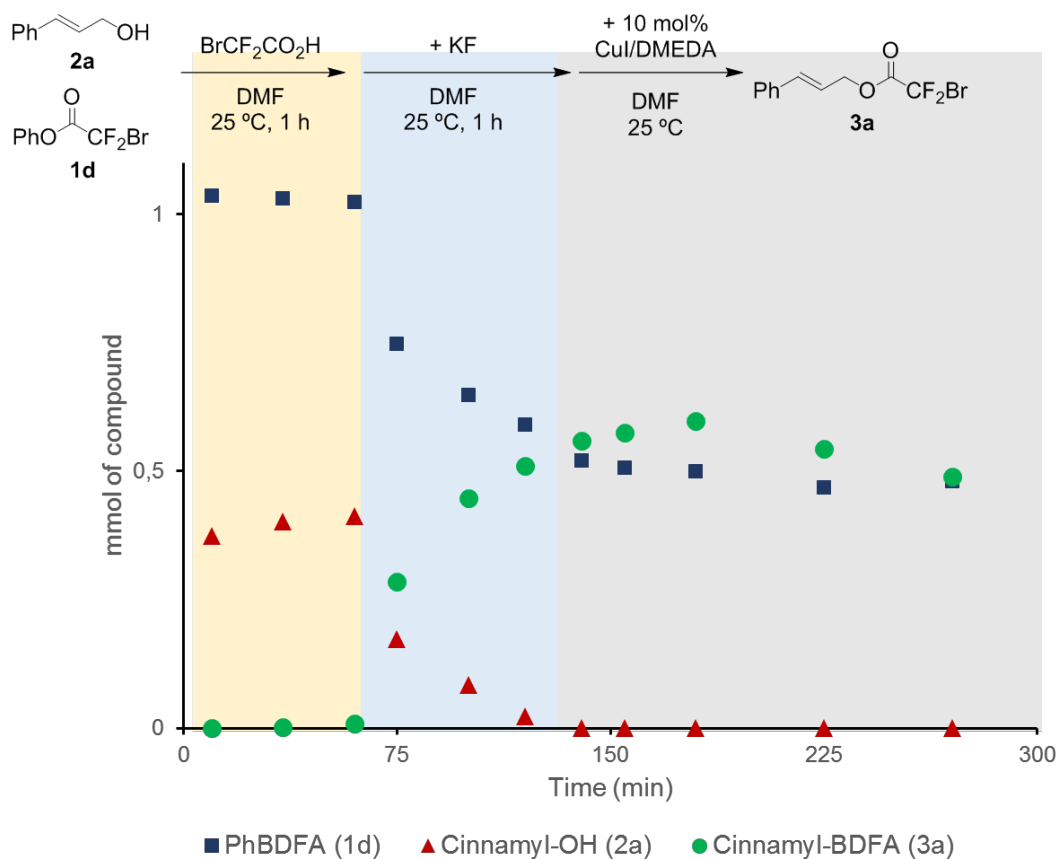
**Figure S2:** Representative  $^{19}\text{F}$ -NMR spectra of phenyl bromodifluoroacetate (**1d**) obtained during the study of its stability. The spectra showed are from the sample stored on the bench.

## Mechanistic Experiments (Scheme 2)

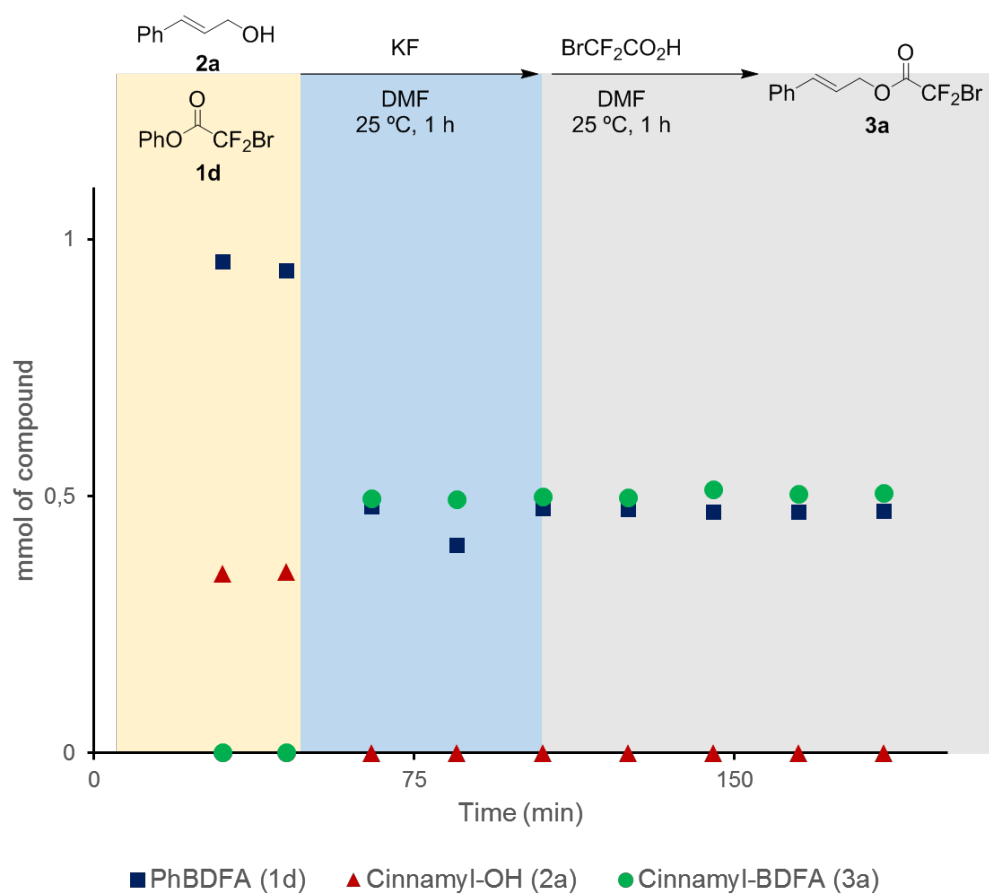
To better understand the reaction pathway, some experiments were done to probe the mechanism of the reaction. Specifically, the stability of phenyl bromodifluoroacetate (**1d**) under reaction conditions, the kinetic profile of the first hours of reaction, and a stepwise addition to study the formation of the intermediate ester.

### Study of the Formation of Putative “Active Ester Intermediate” *In Situ*

KF facilitated the fast formation of the putative bromodifluoroacetate ester intermediate, as shown by stepwise addition of KF and bromodifluoroacetic acid to a DMF solution of cinnamyl alcohol **2a** and phenyl bromodifluoroacetate (**1d**). The key role of KF was both in the presence and absence of bromodifluoroacetic acid.

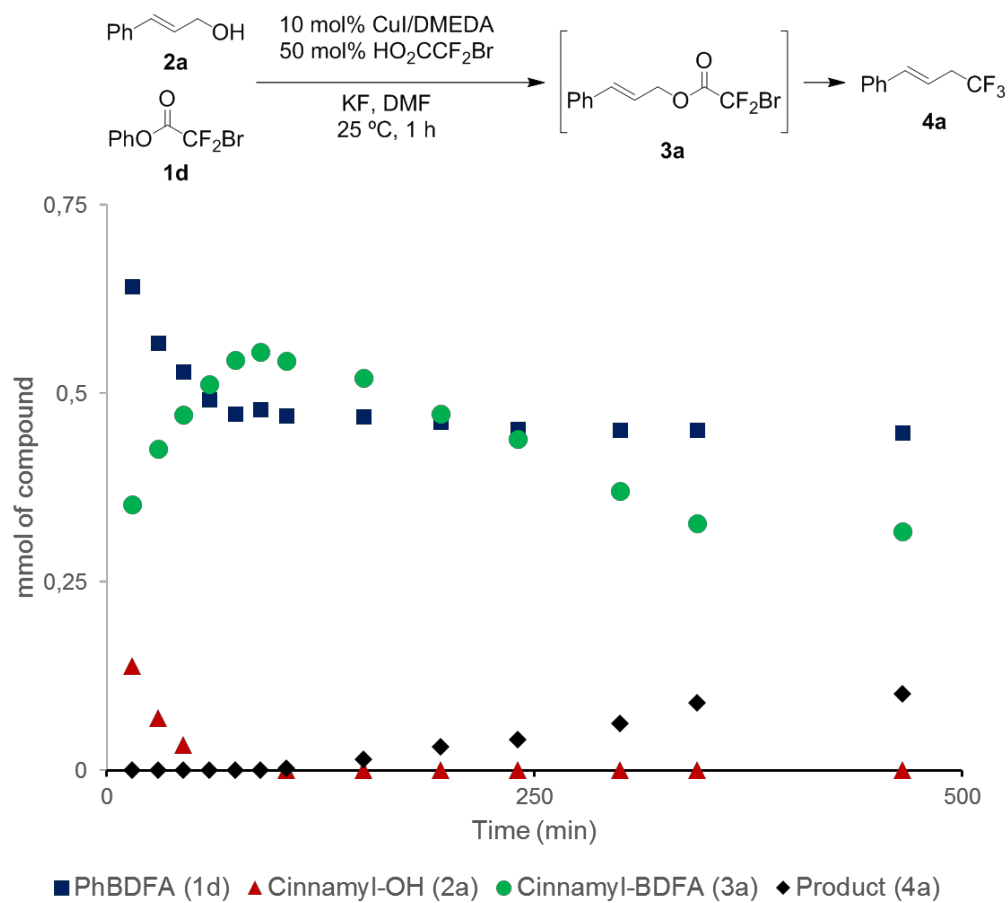


**Figure S3:** Cinnamyl Bromodifluoroacetate (**3a**) Formation Upon Stepwise Addition of KF and Bromodifluoroacetic Acid.



**Figure S4:** Cinnamyl Bromodifluoroacetate (**3a**) Formation Upon Stepwise Addition of KF and Bromodifluoroacetic Acid.

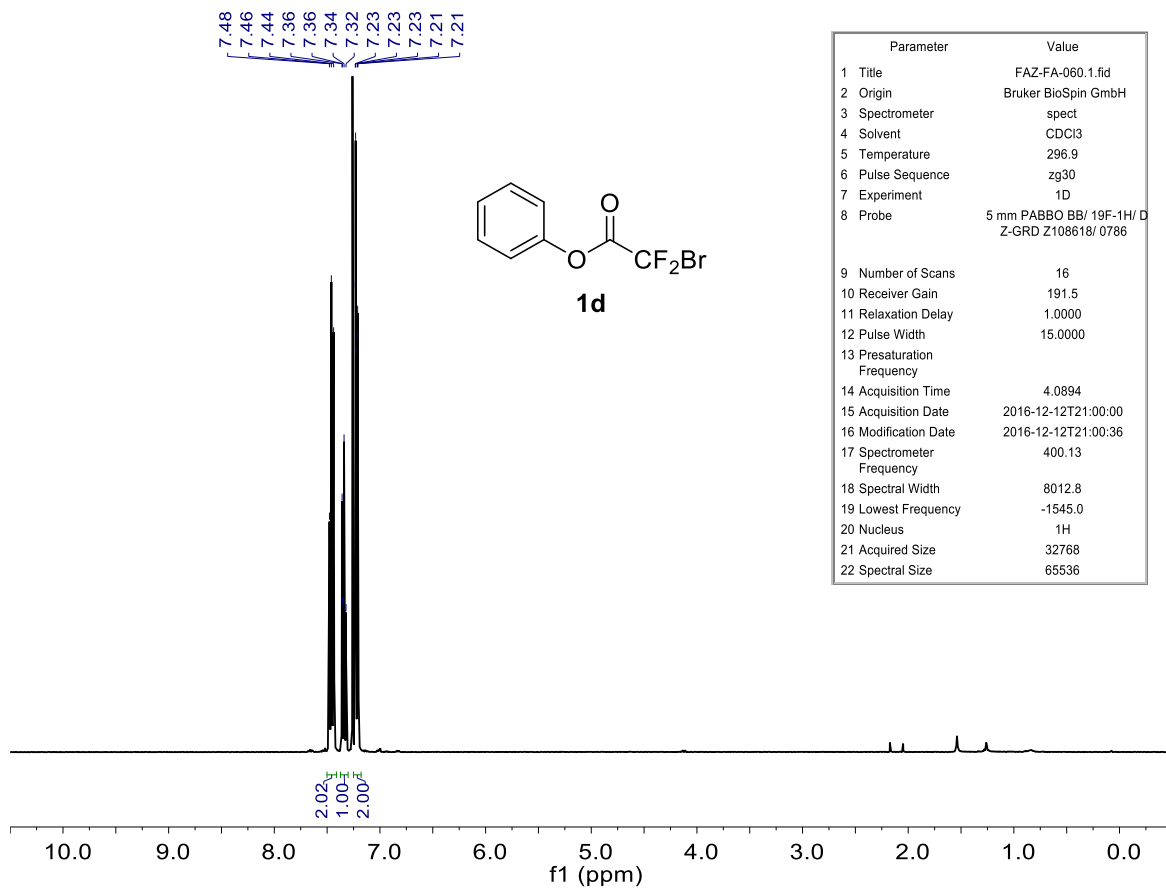
## Time-Course Analysis of the Reaction



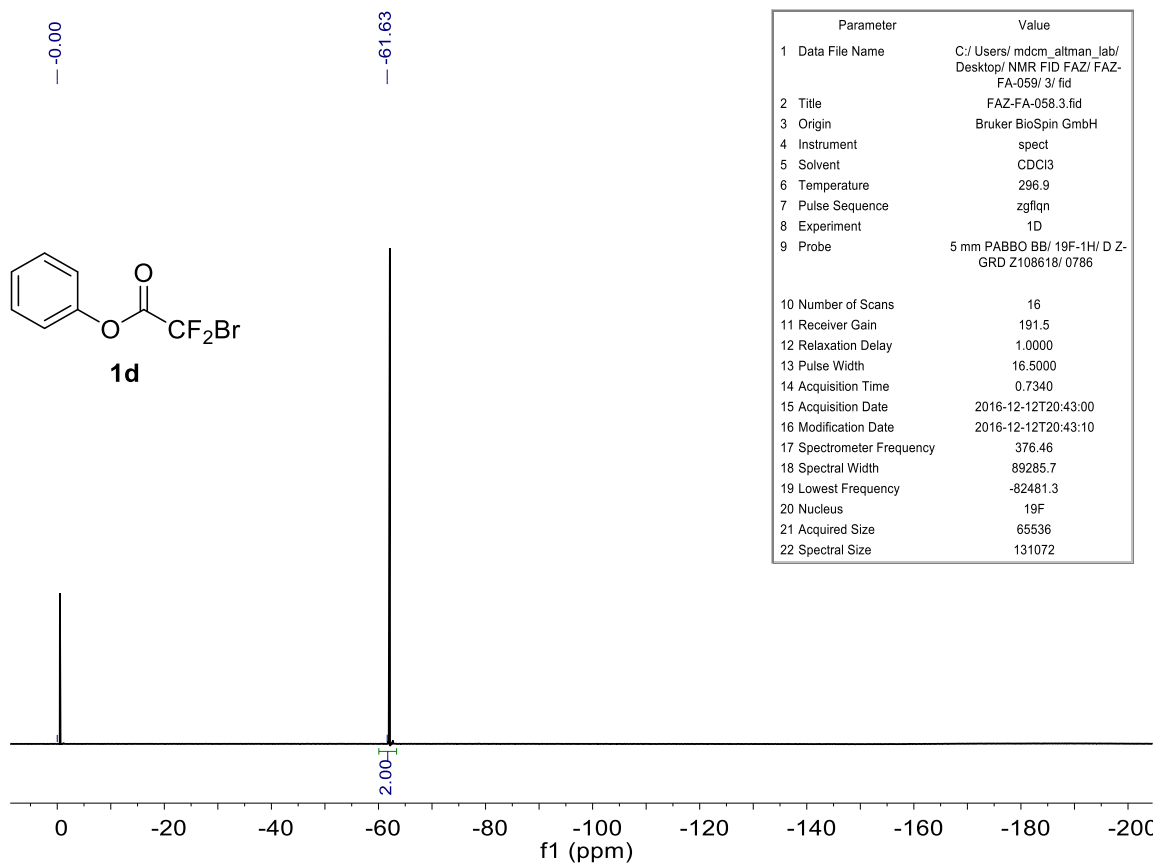
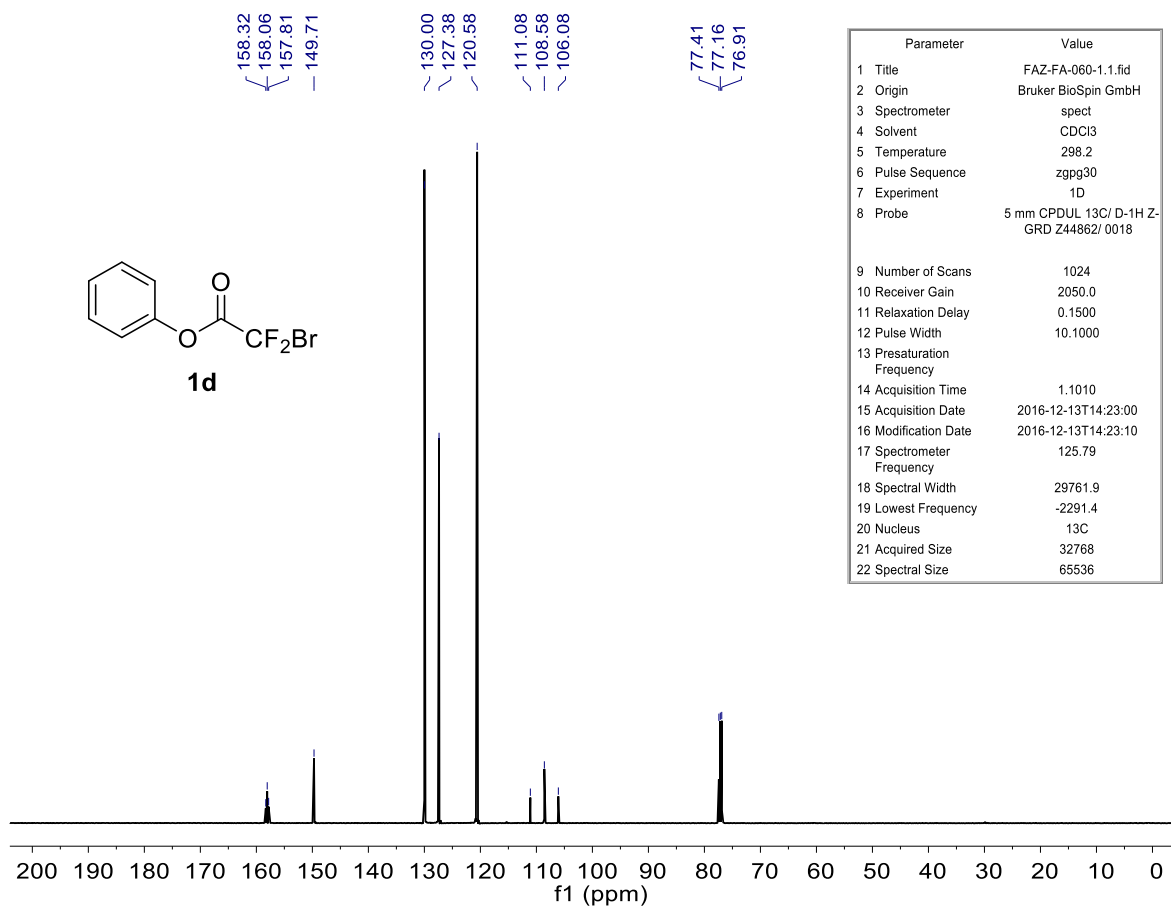
**Figure S5:** Time Course Analysis of the Deoxytrifluorometylation of Alcohol **2a**

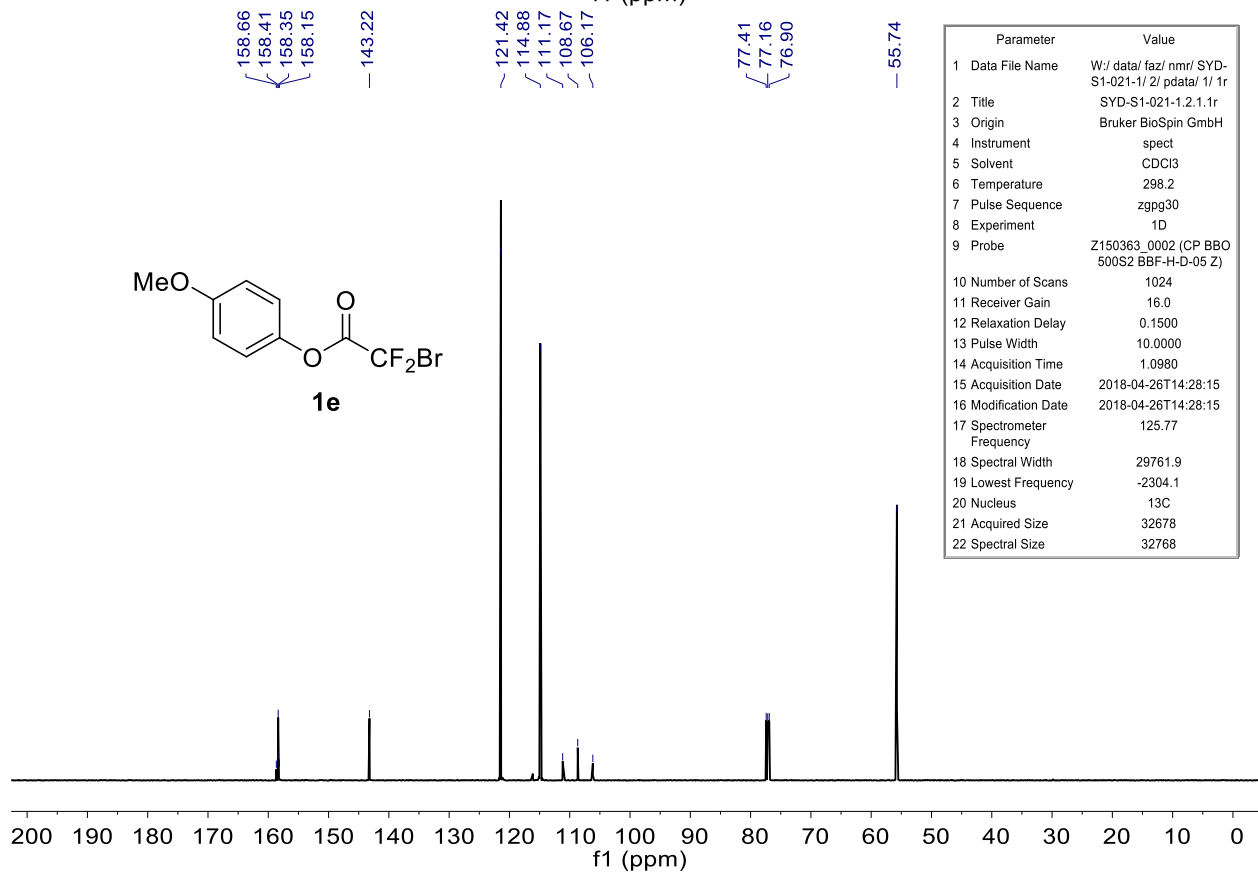
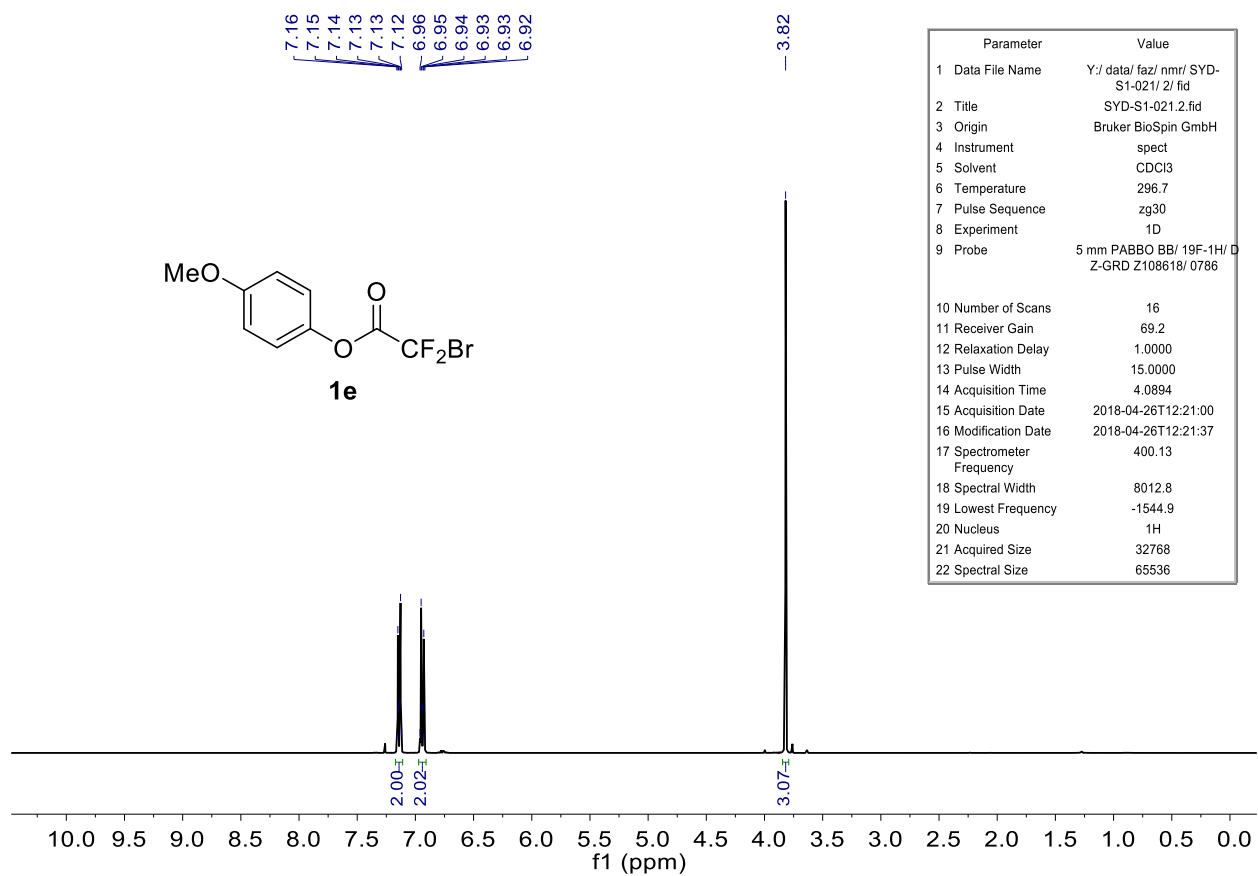
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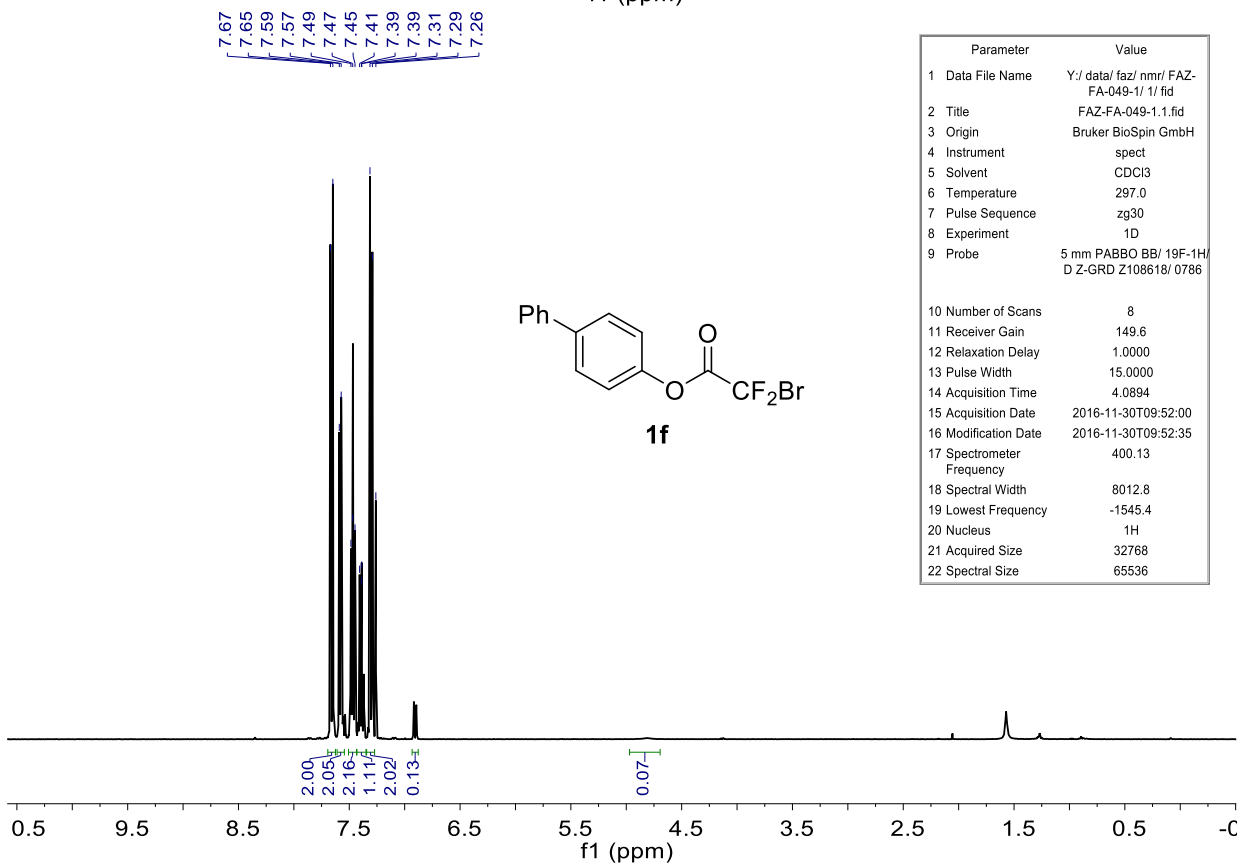
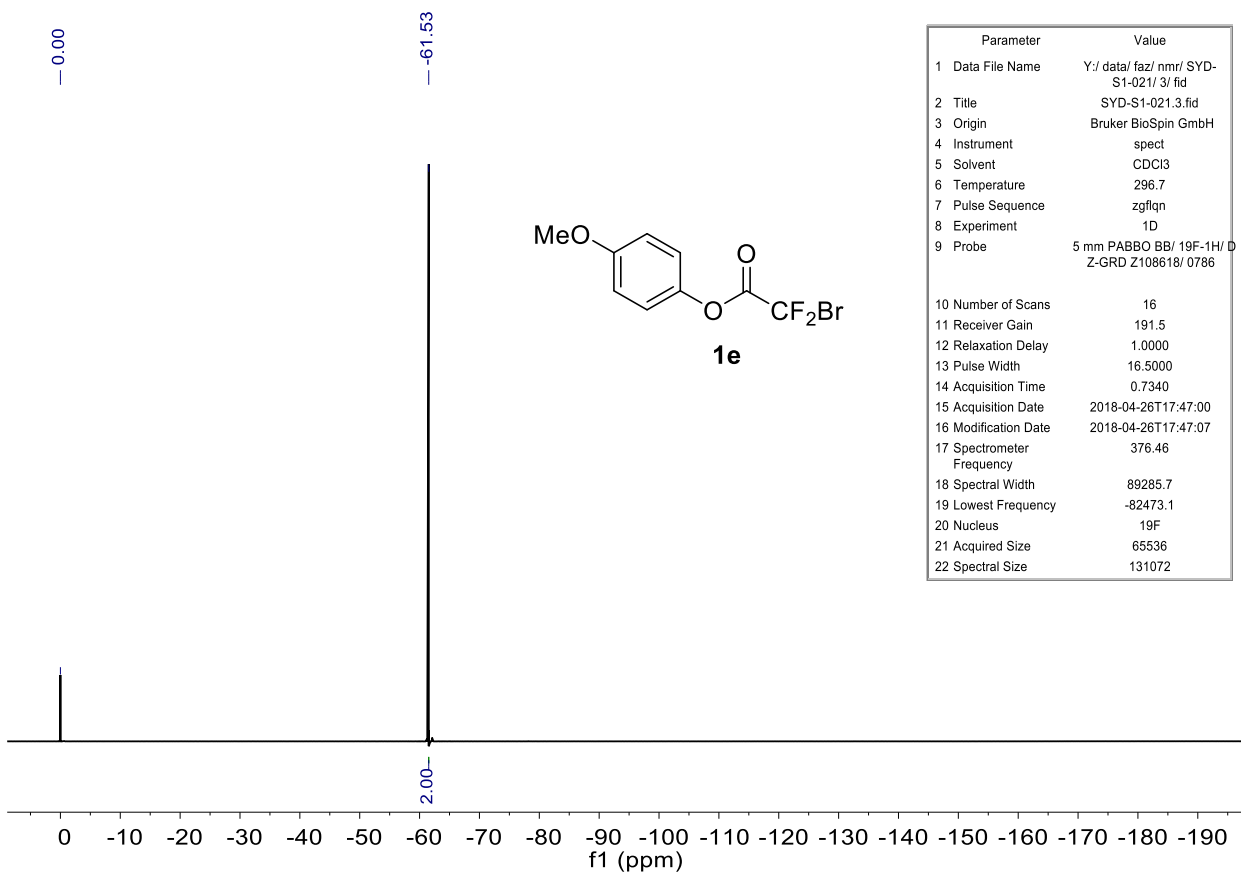
## Bromodifluoroacetate Reagents

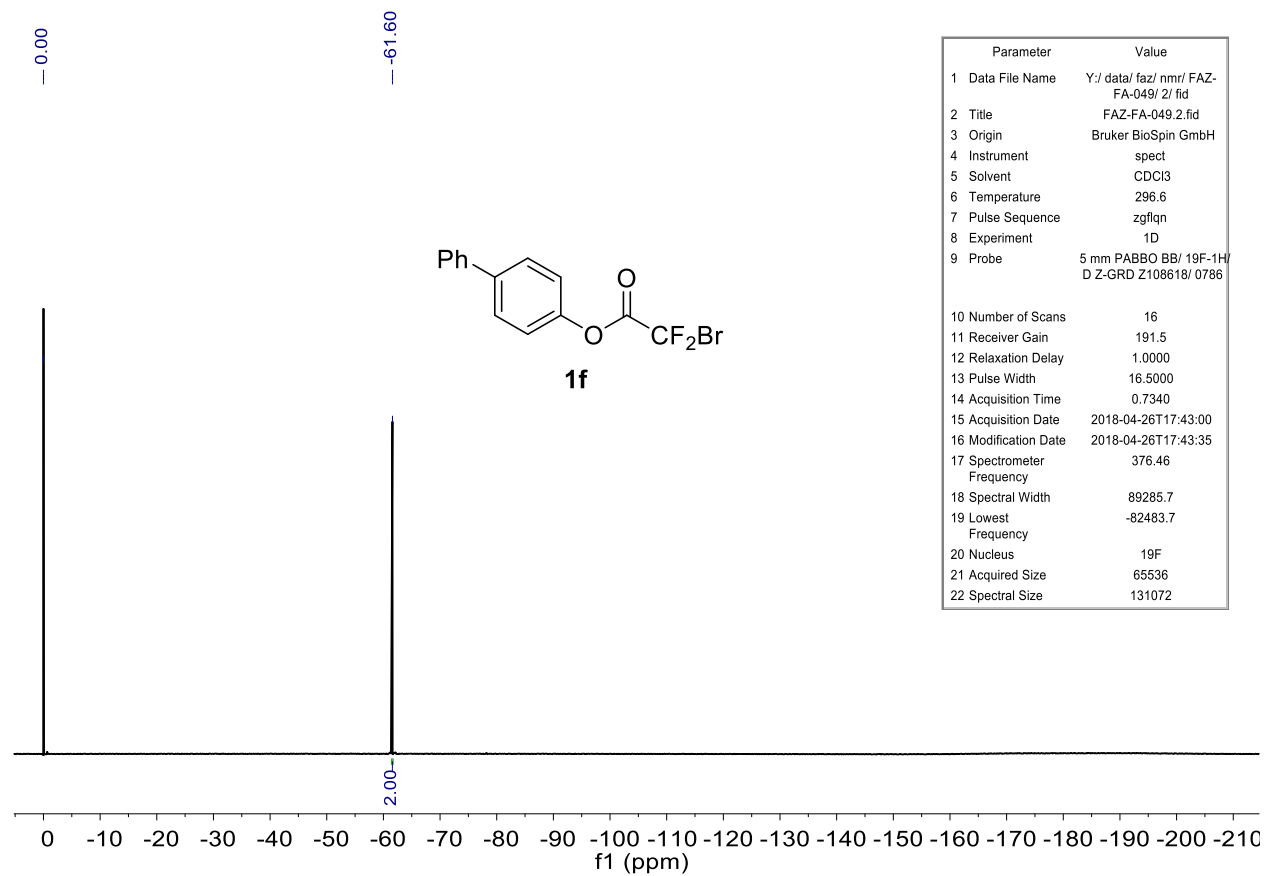
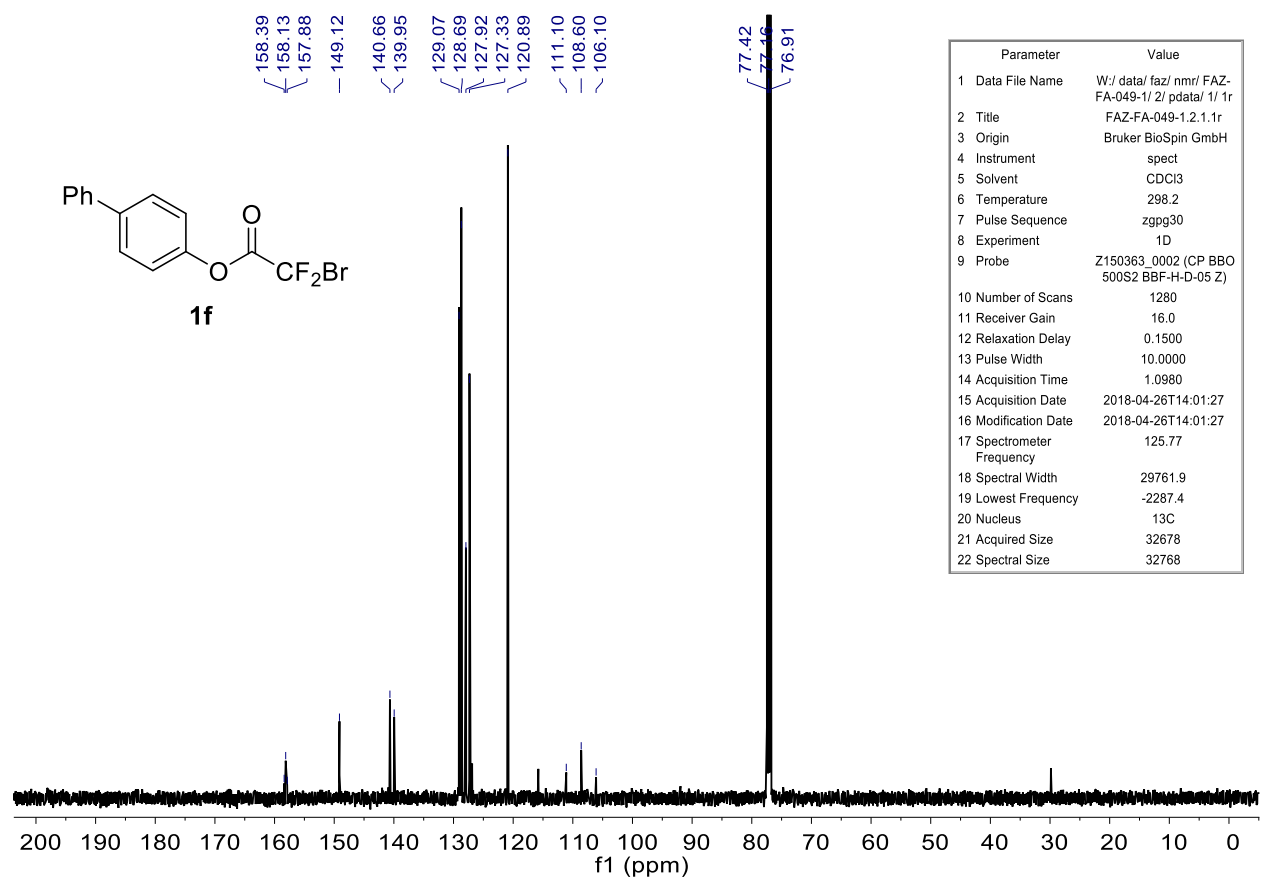


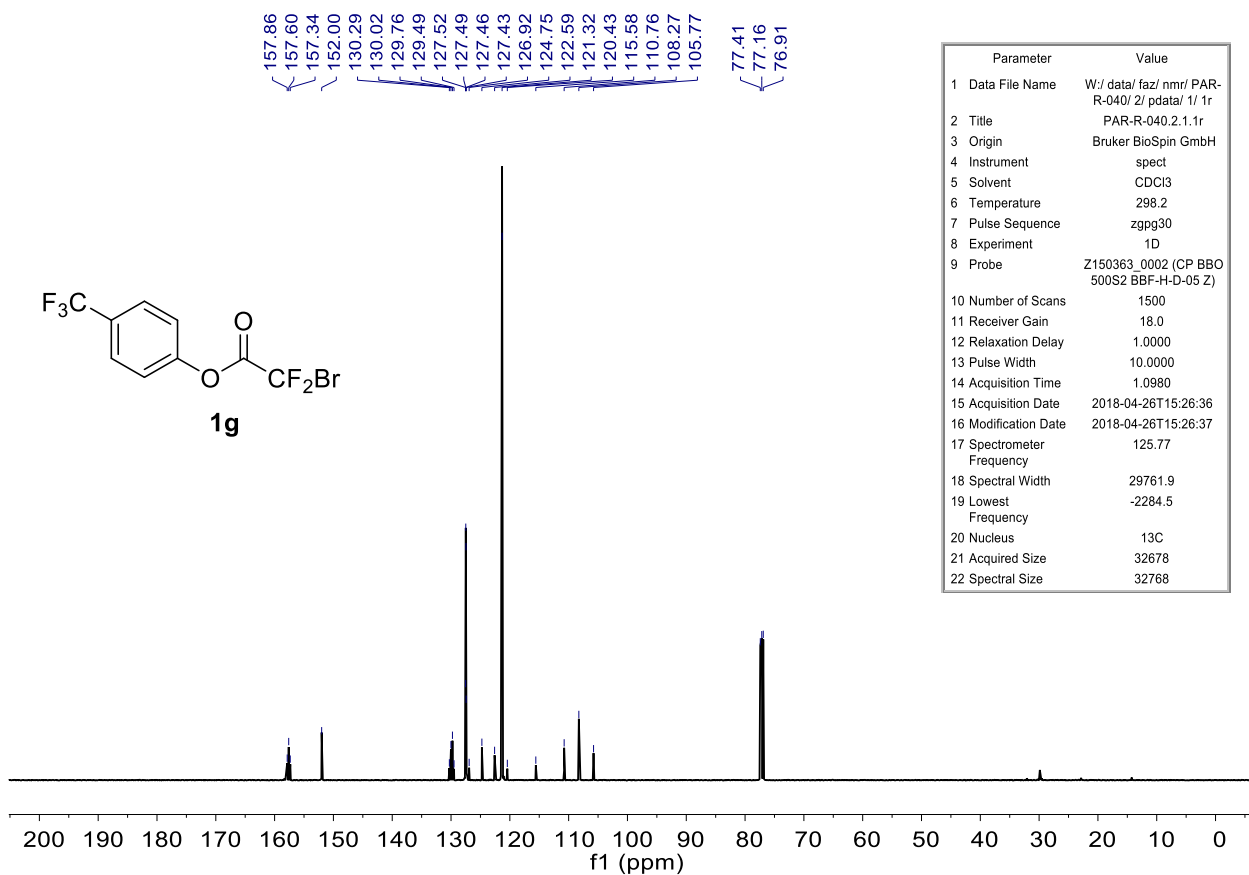
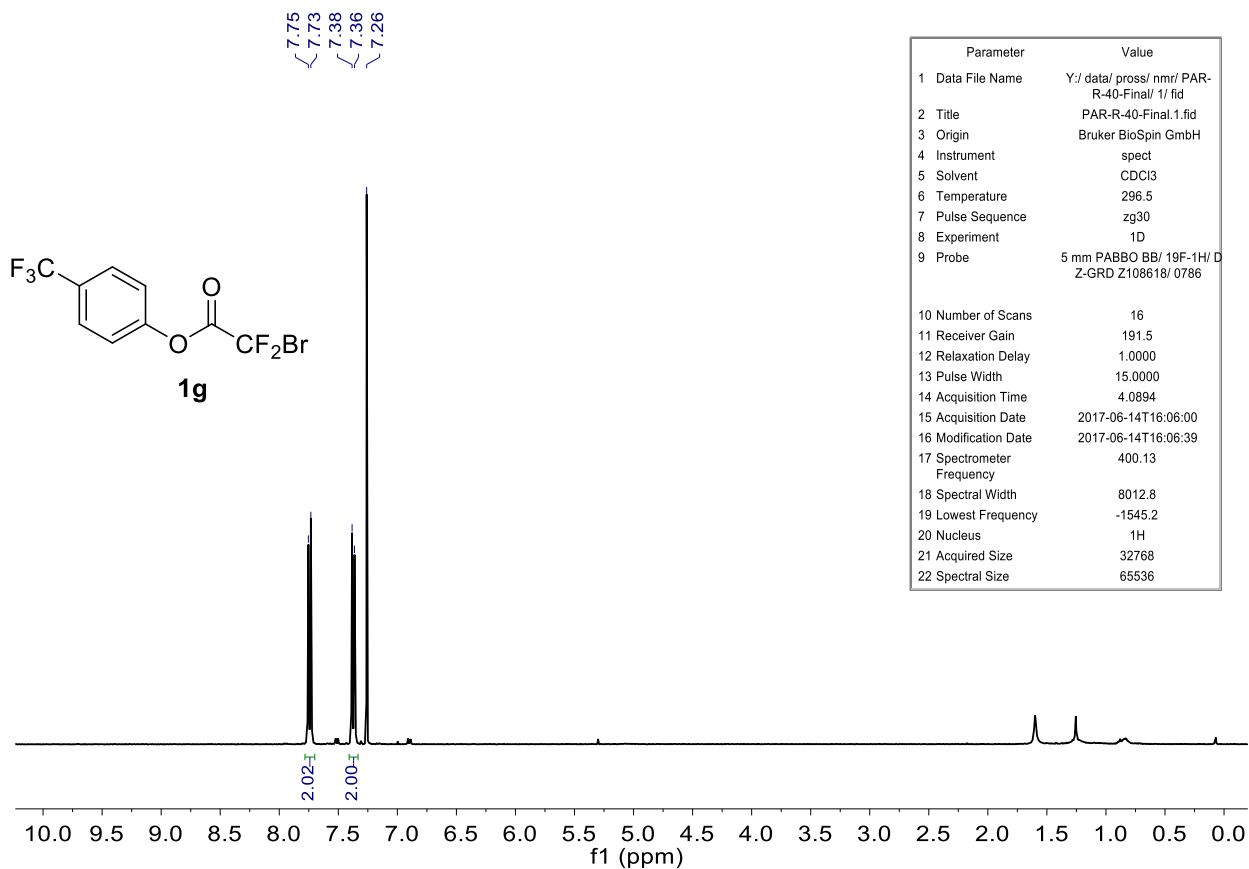


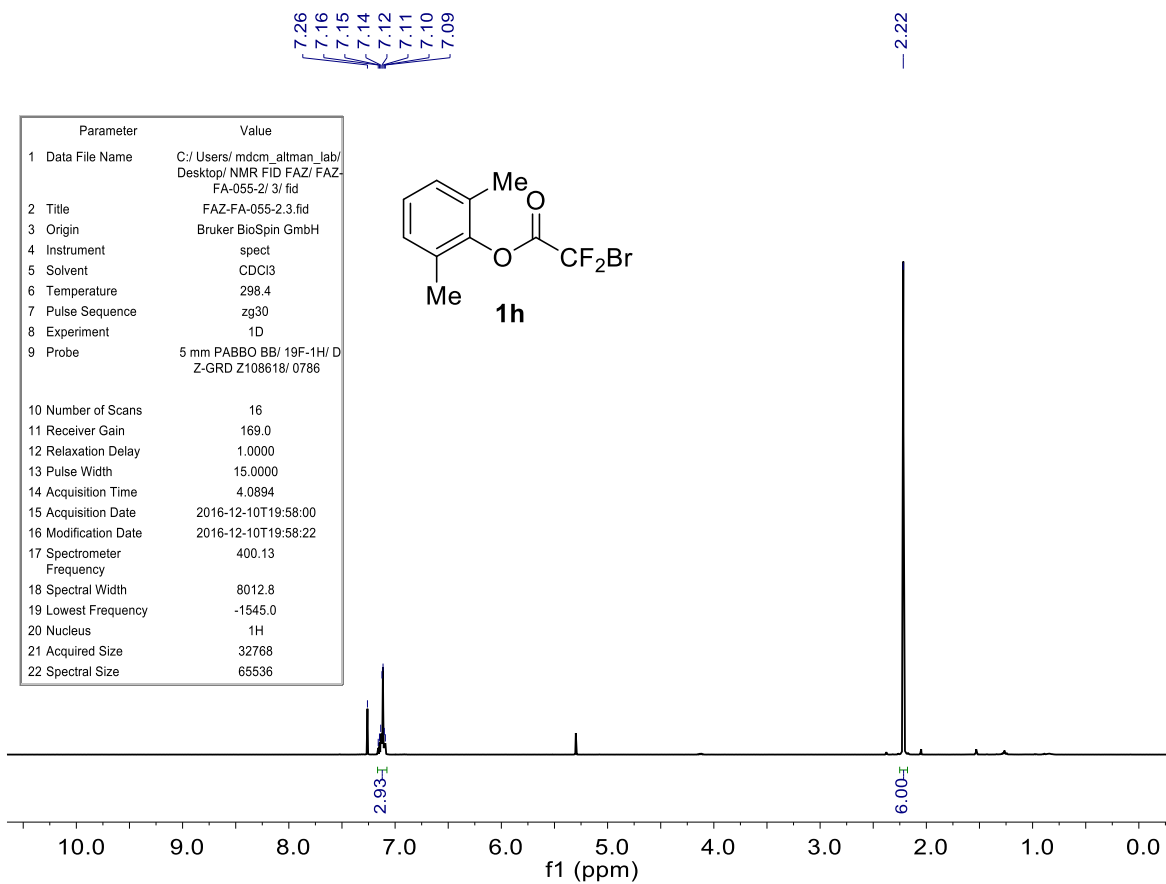
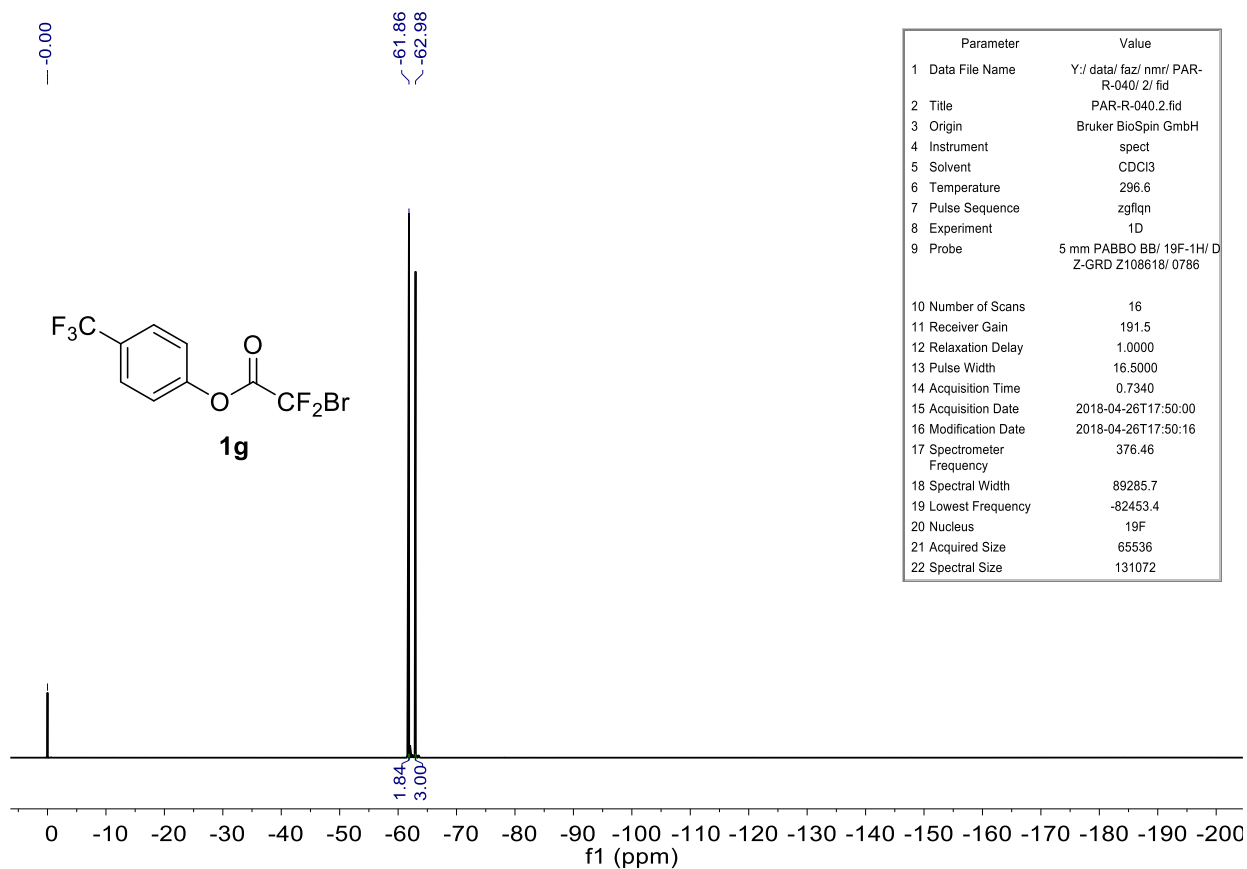


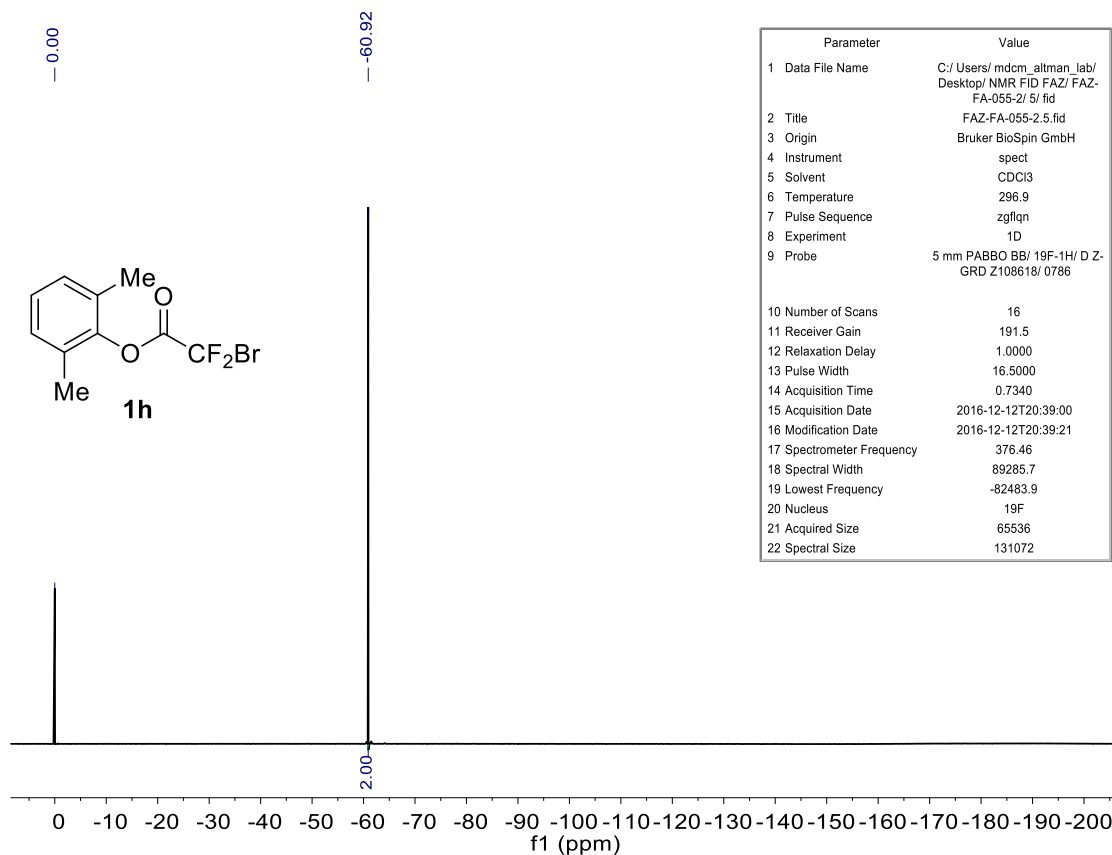
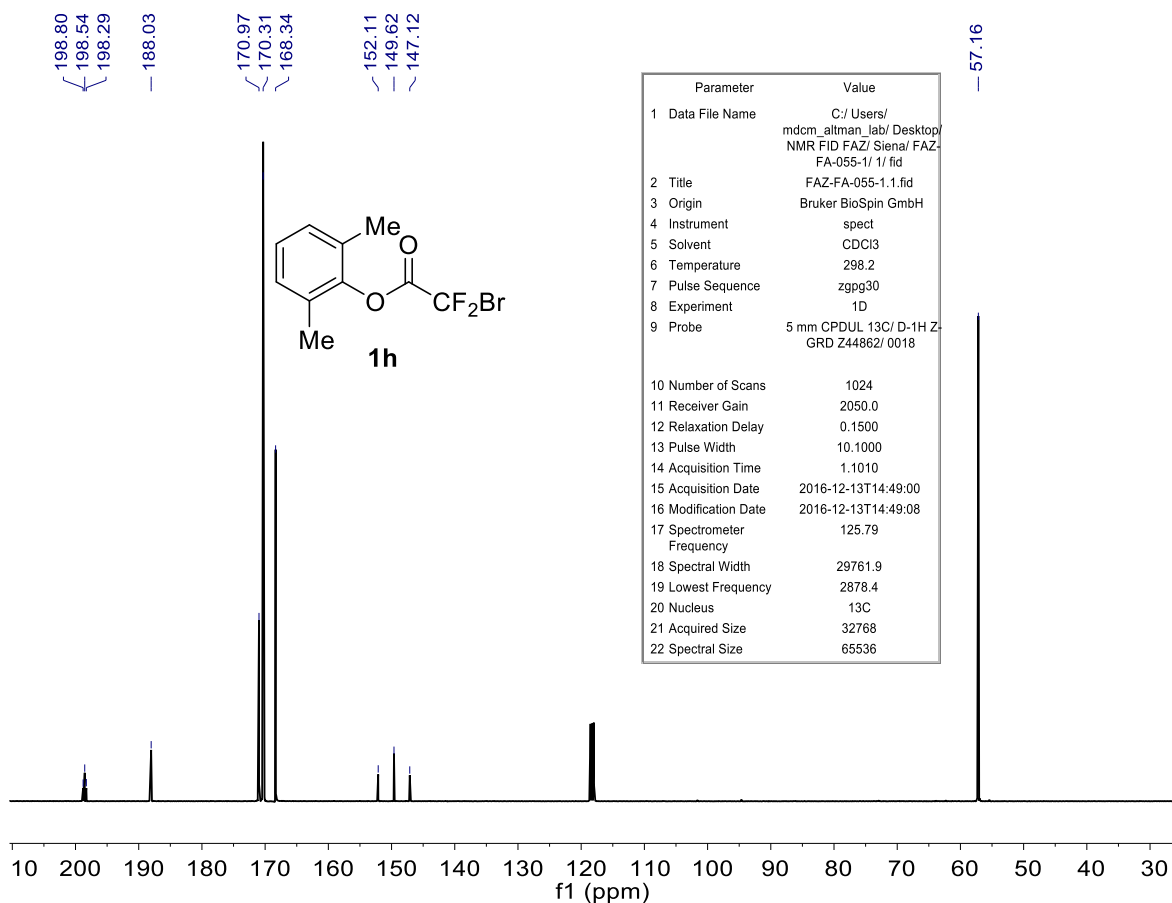


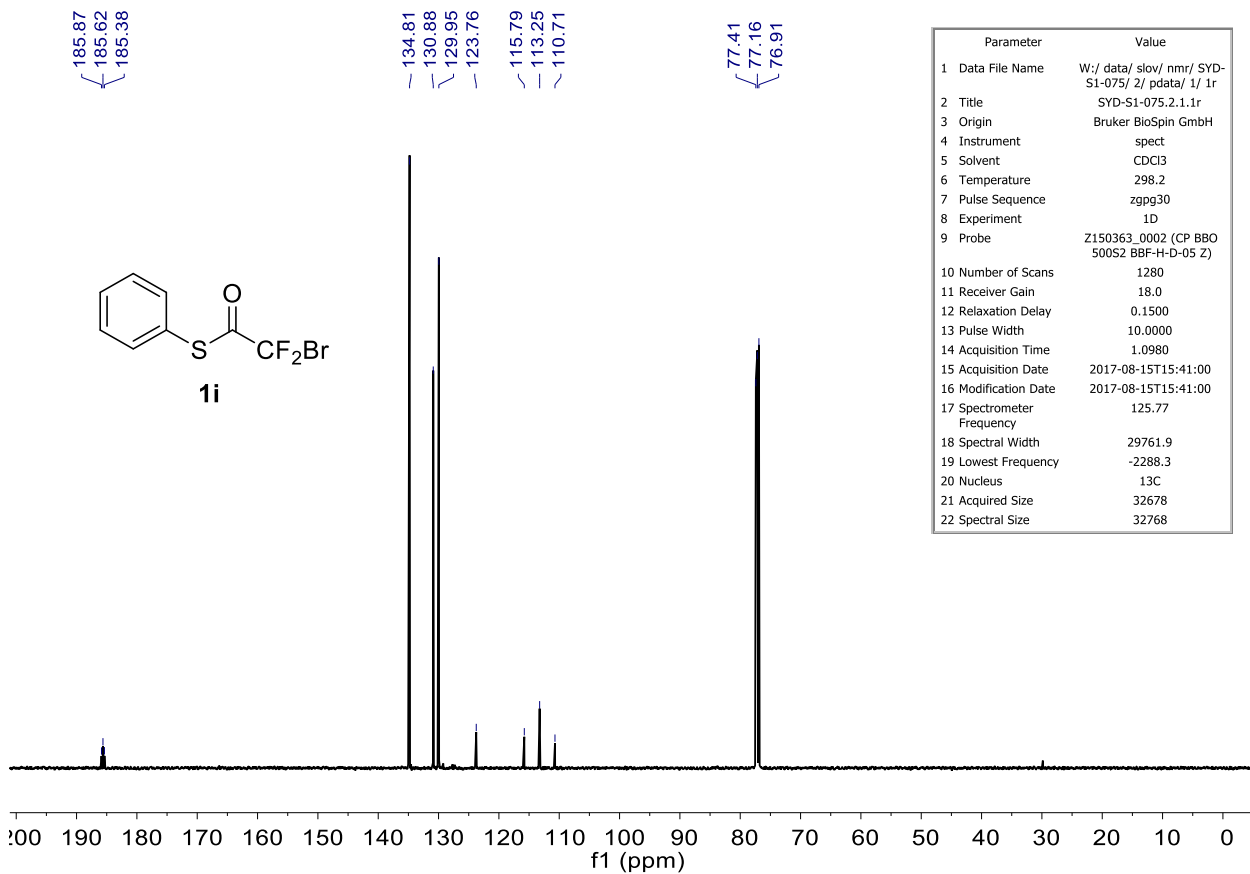
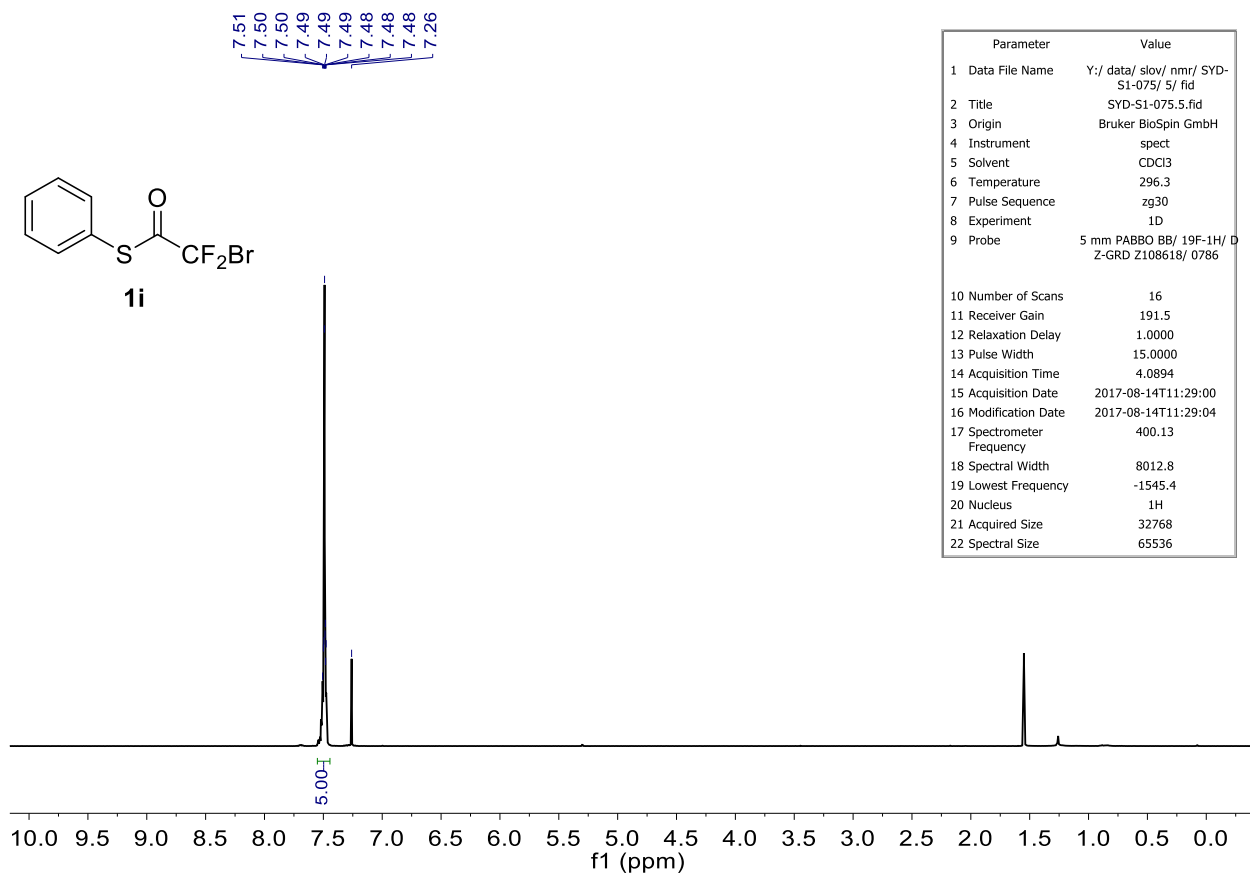




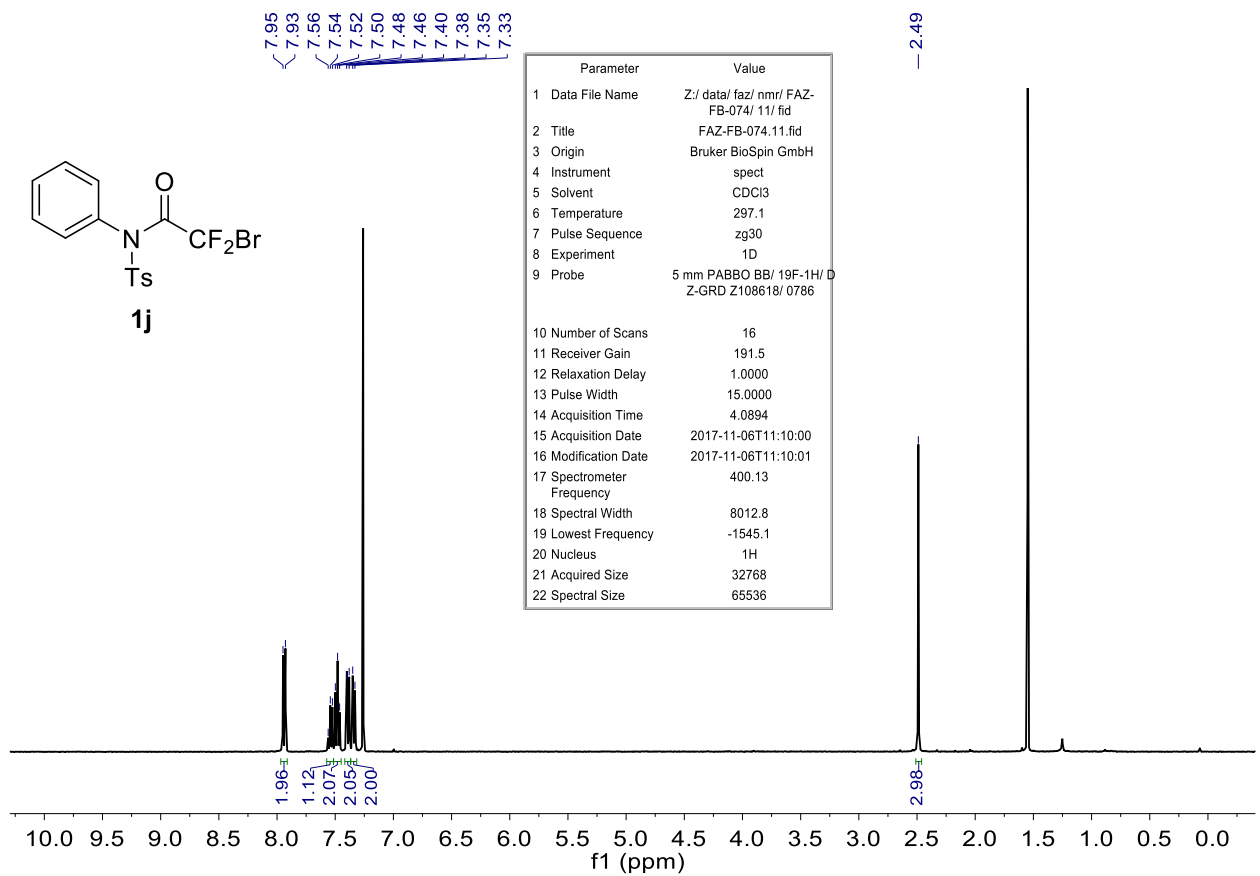
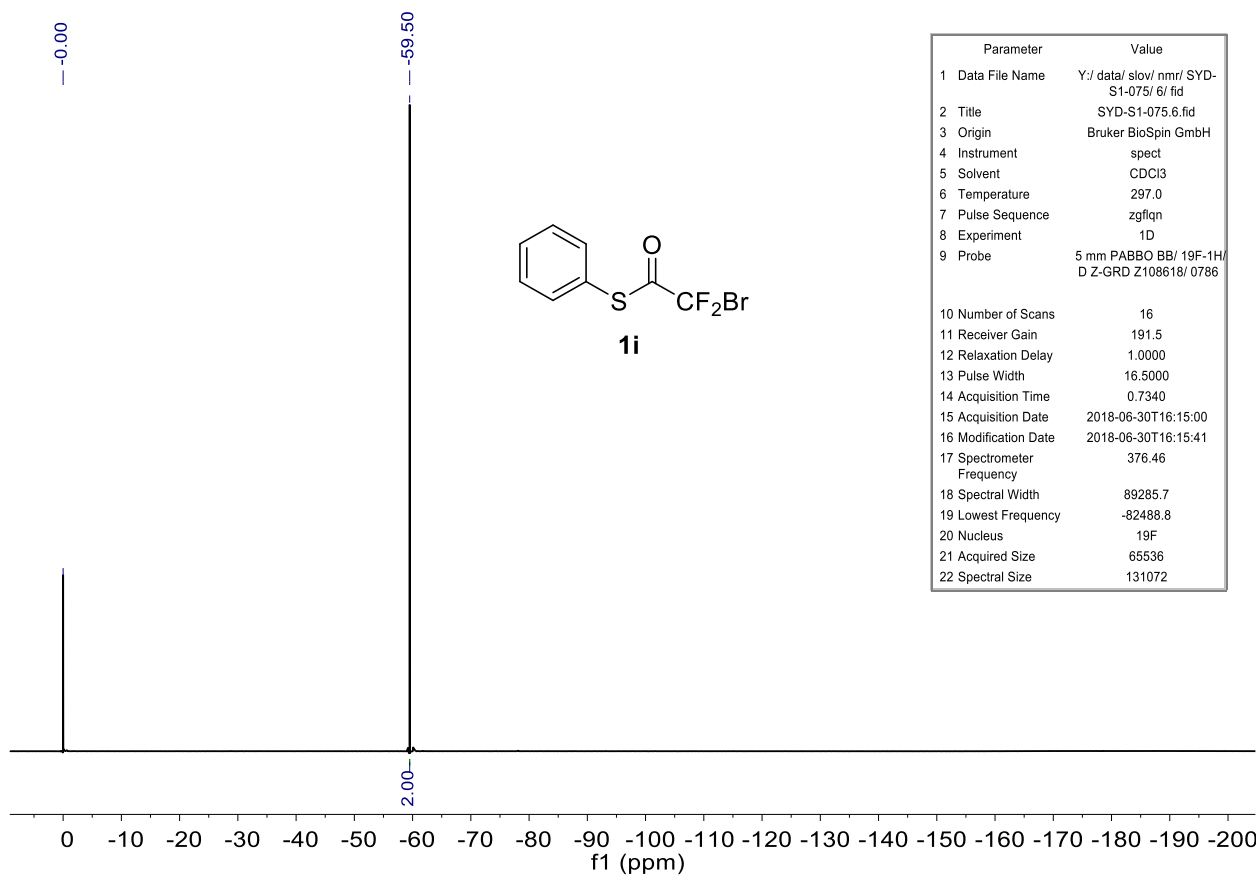


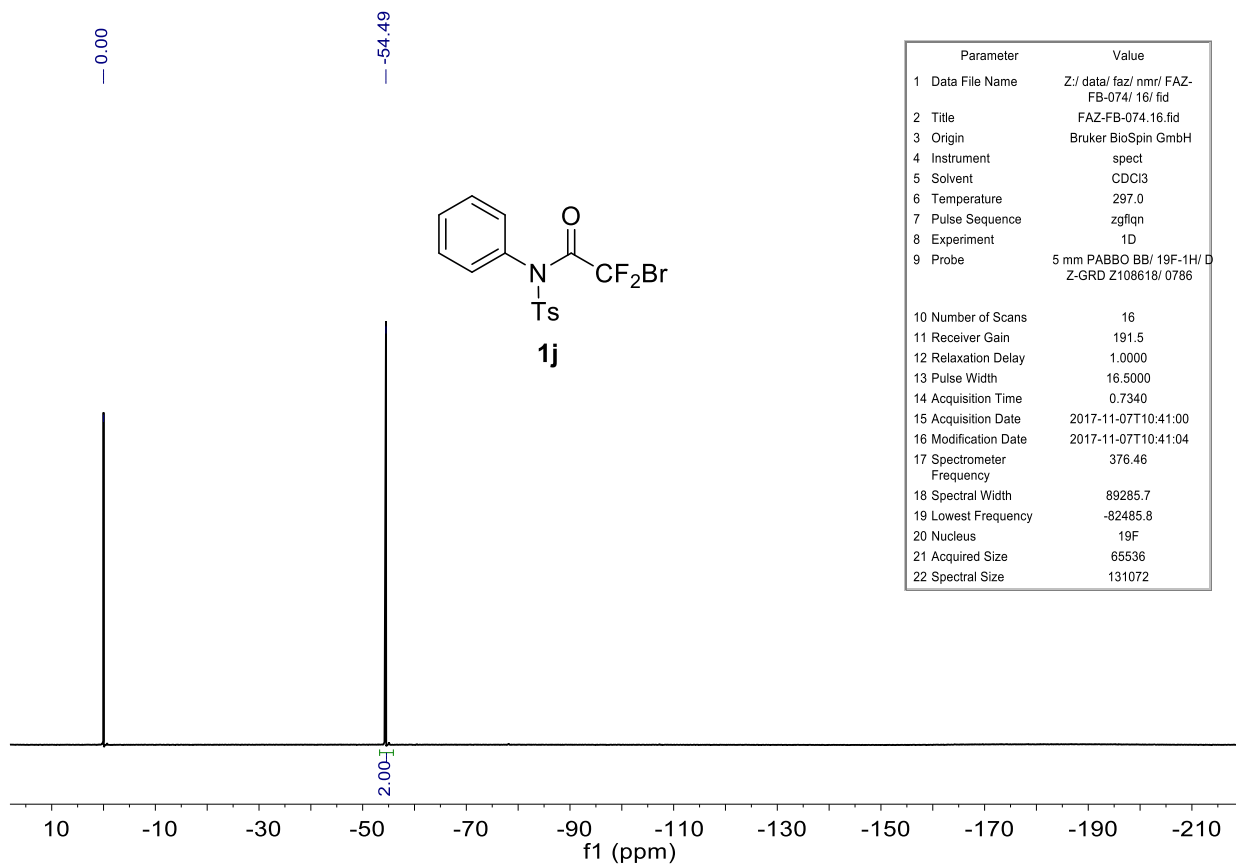
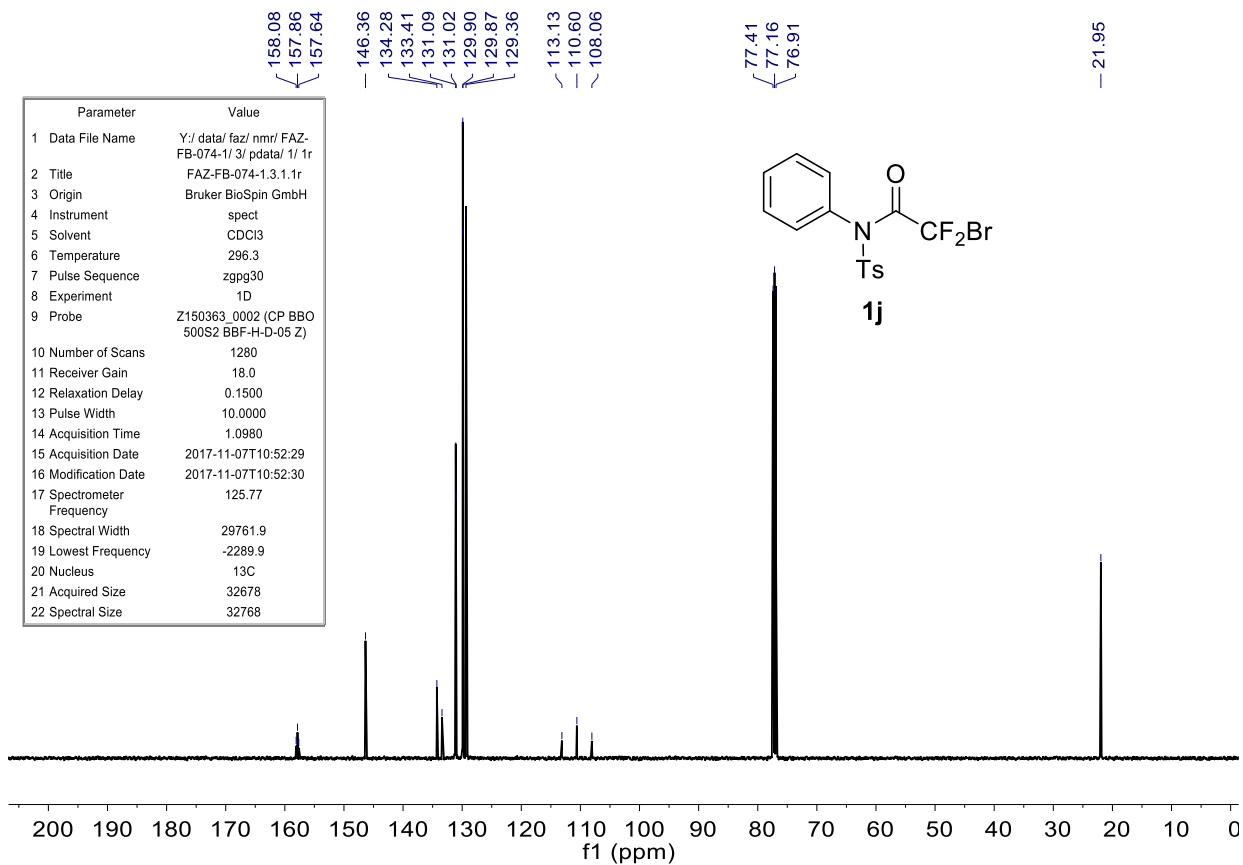






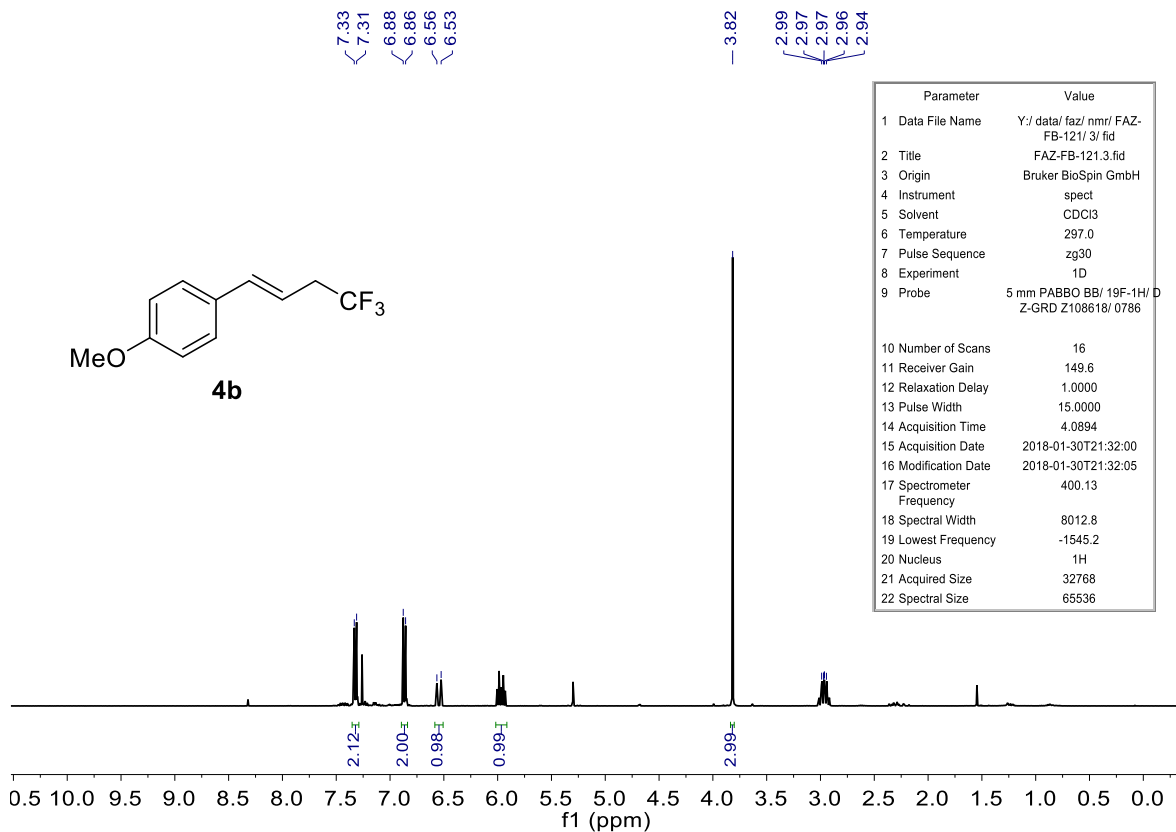
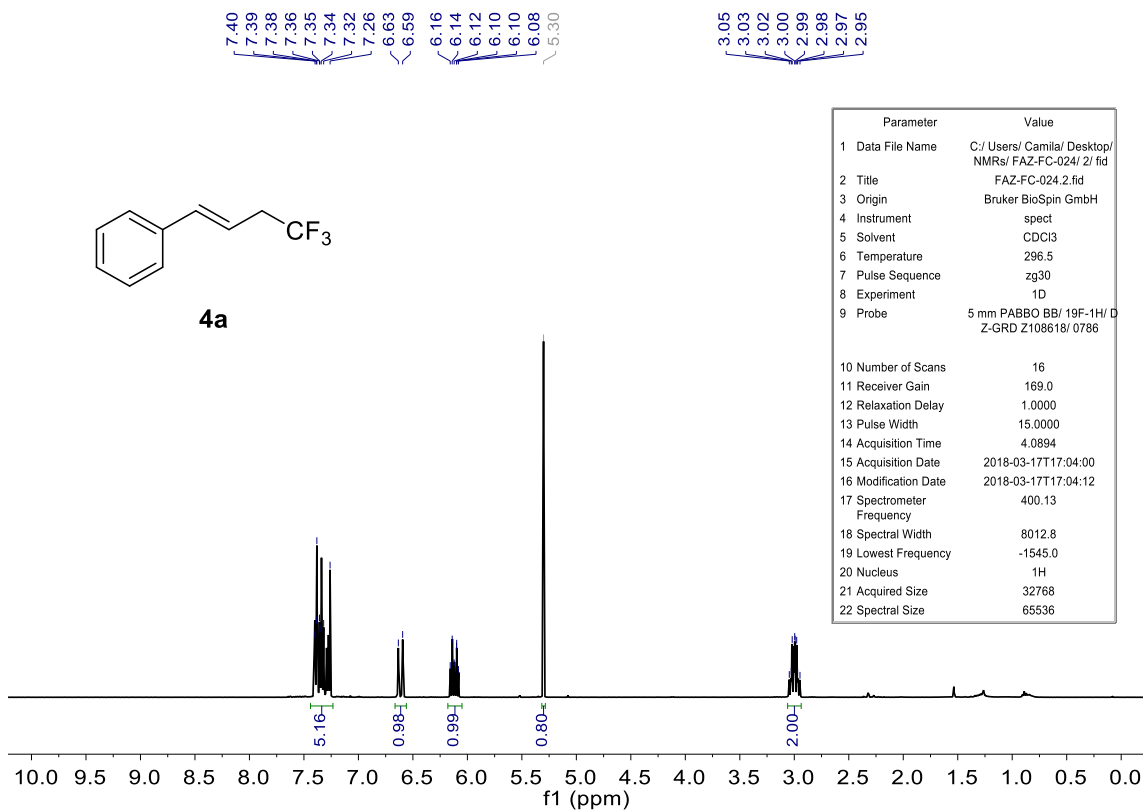




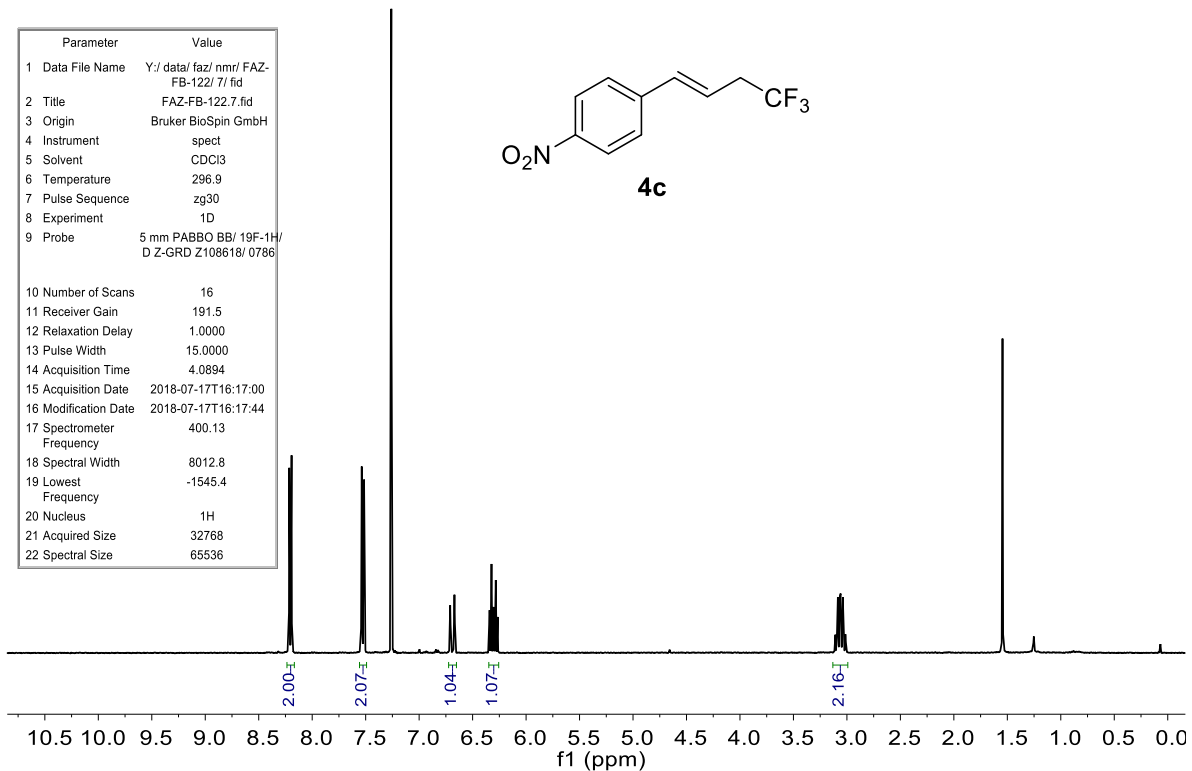
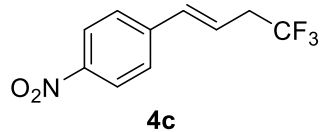


## Reaction Scope

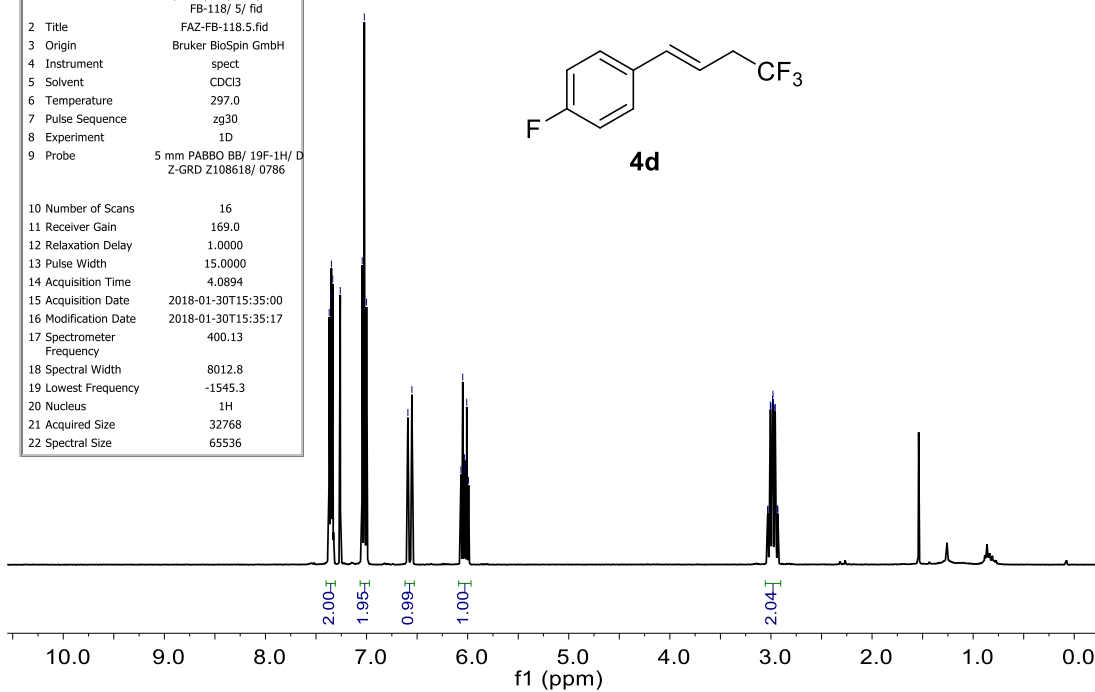
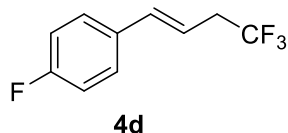
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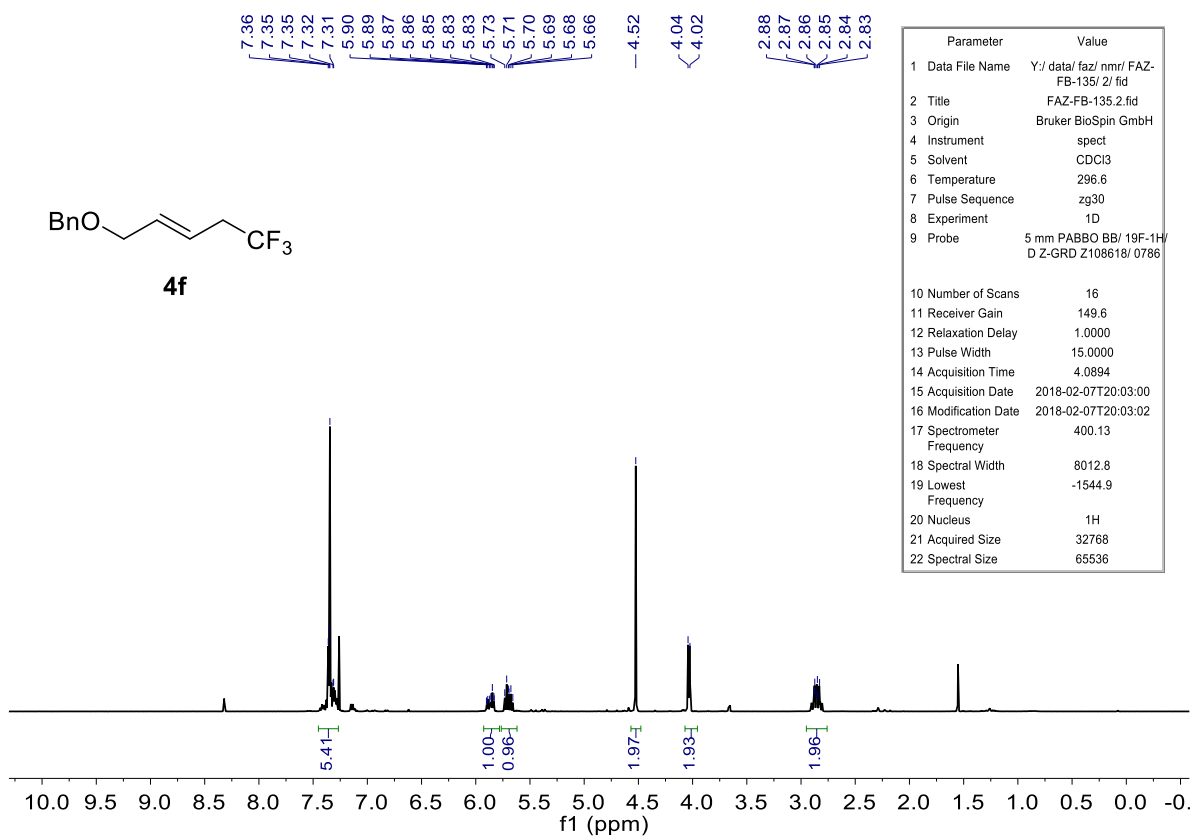
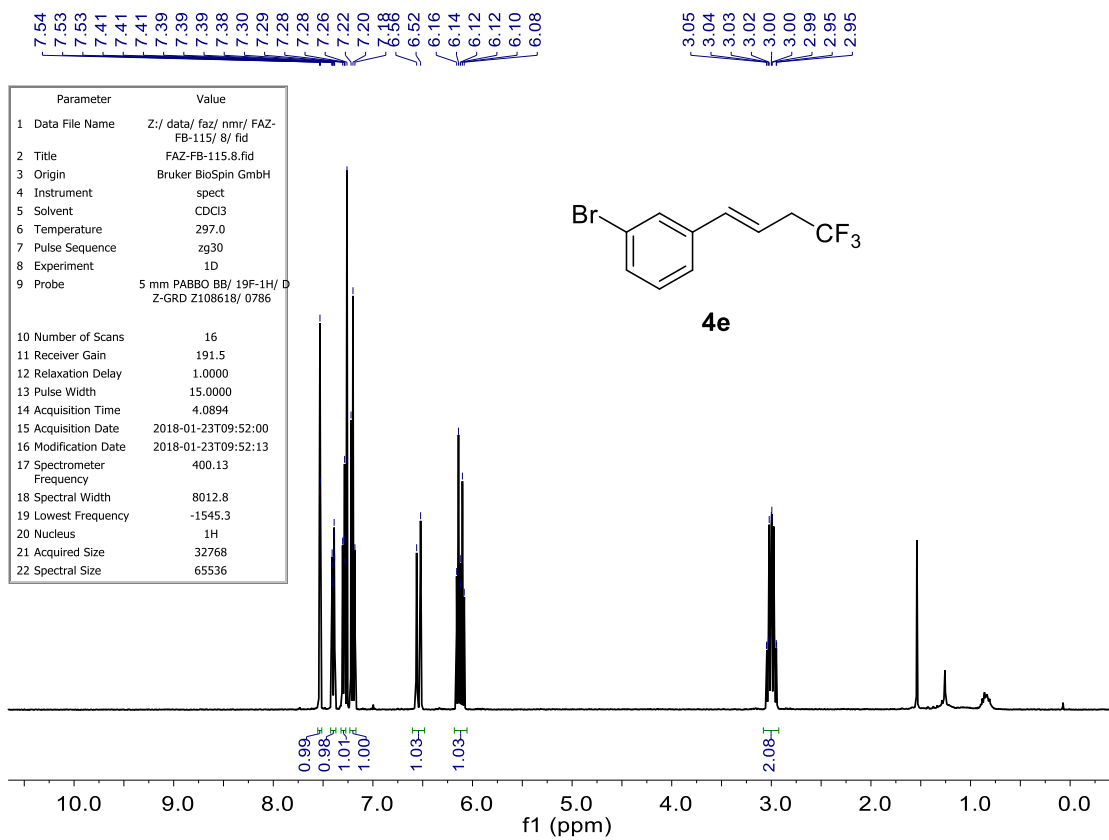


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4 Instrument	spect
5 Solvent	CDCl3
6 Temperature	296.9
7 Pulse Sequence	zg30
8 Experiment	1D
9 Probe	5 mm PABBO BB/ 19F-1H/ D Z-GRD Z108618/ 0786
10 Number of Scans	16
11 Receiver Gain	191.5
12 Relaxation Delay	1.0000
13 Pulse Width	15.0000
14 Acquisition Time	4.0894
15 Acquisition Date	2018-07-17T16:17:00
16 Modification Date	2018-07-17T16:17:44
17 Spectrometer Frequency	400.13
18 Spectral Width	8012.8
19 Lowest Frequency	-1545.4
20 Nucleus	1H
21 Acquired Size	32768
22 Spectral Size	65536

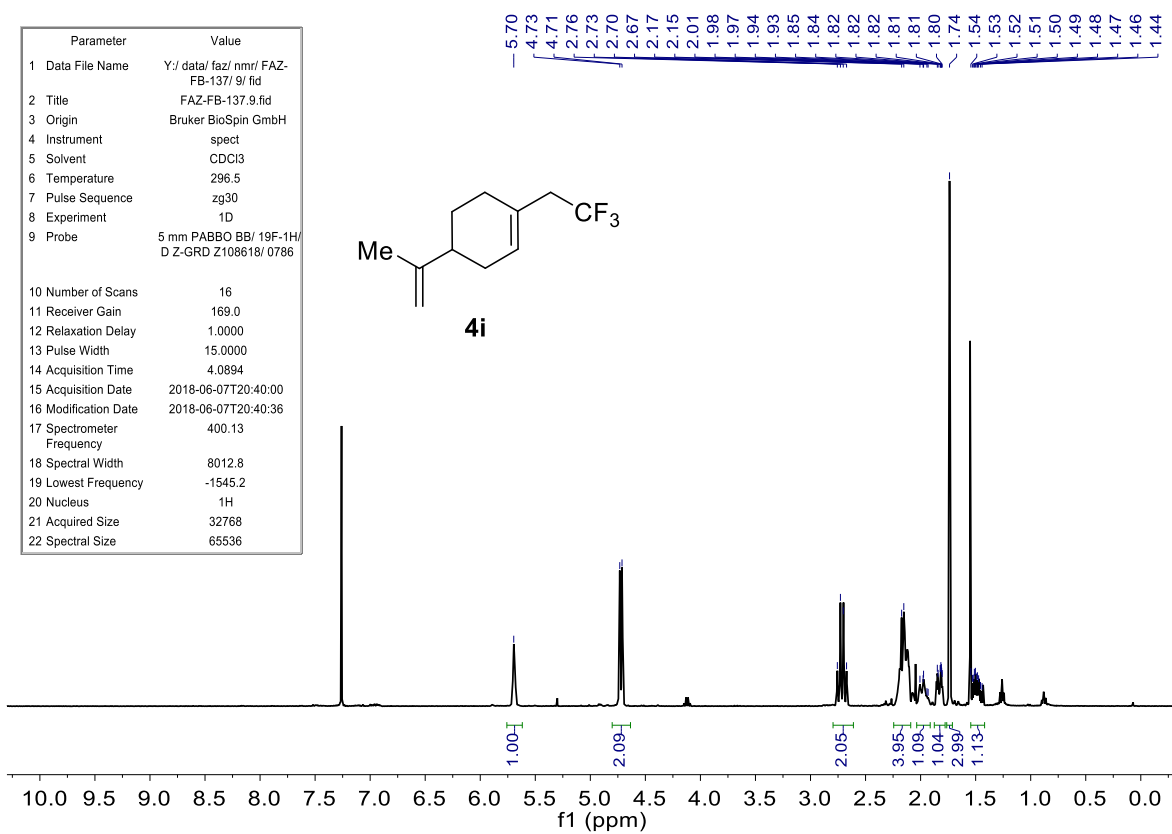
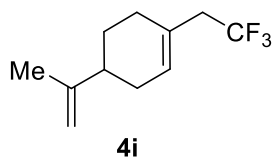


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4 Instrument	spect
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6 Temperature	297.0
7 Pulse Sequence	zg30
8 Experiment	1D
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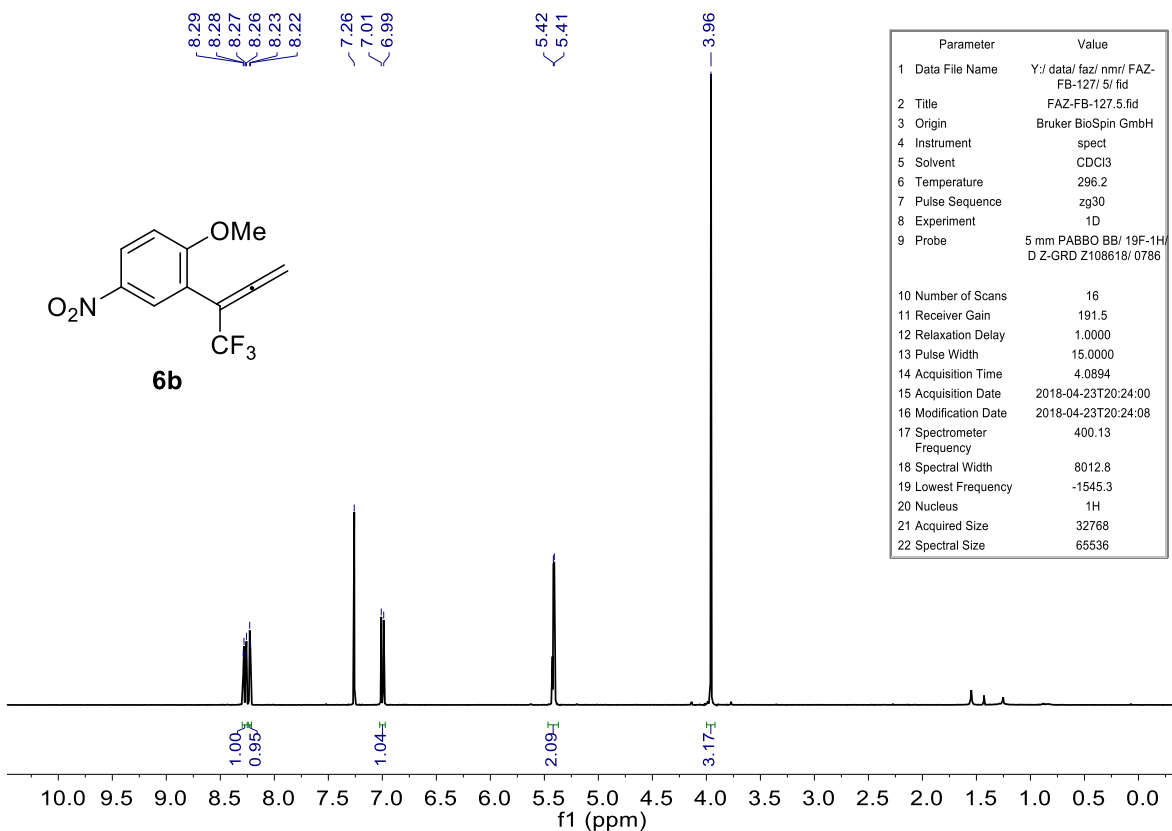
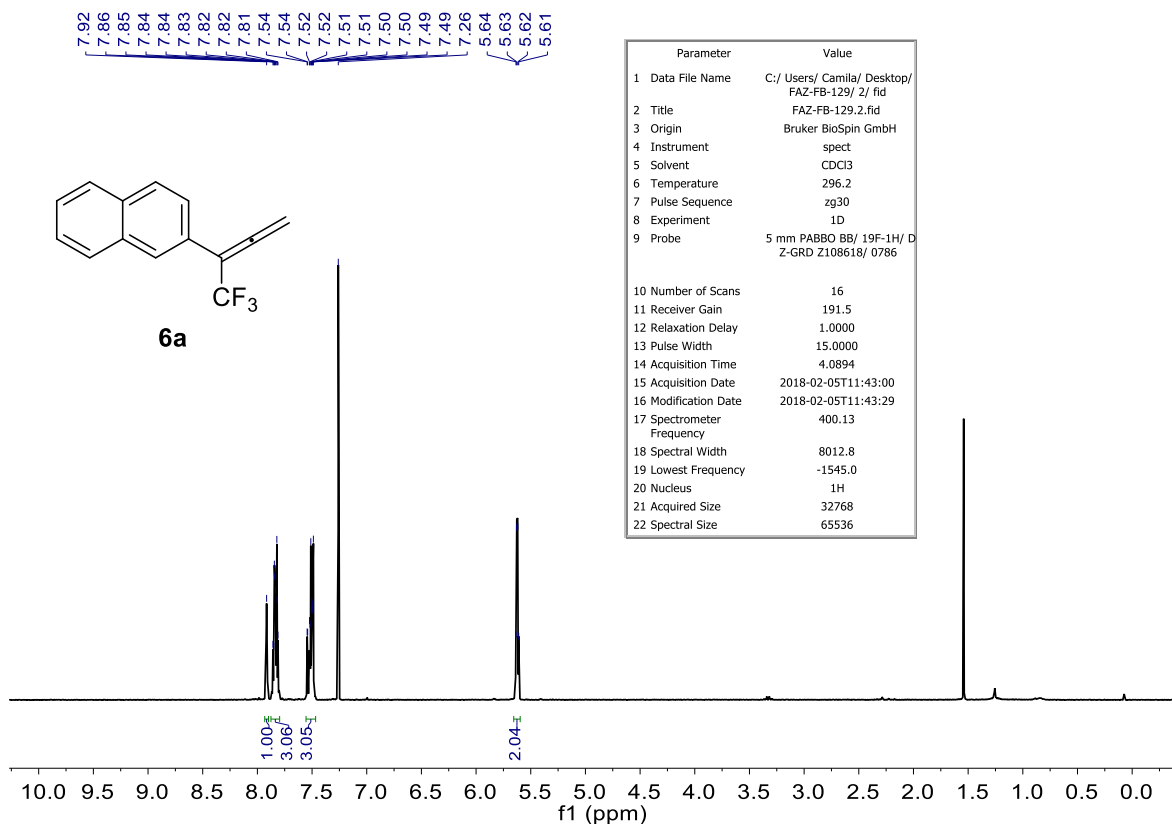


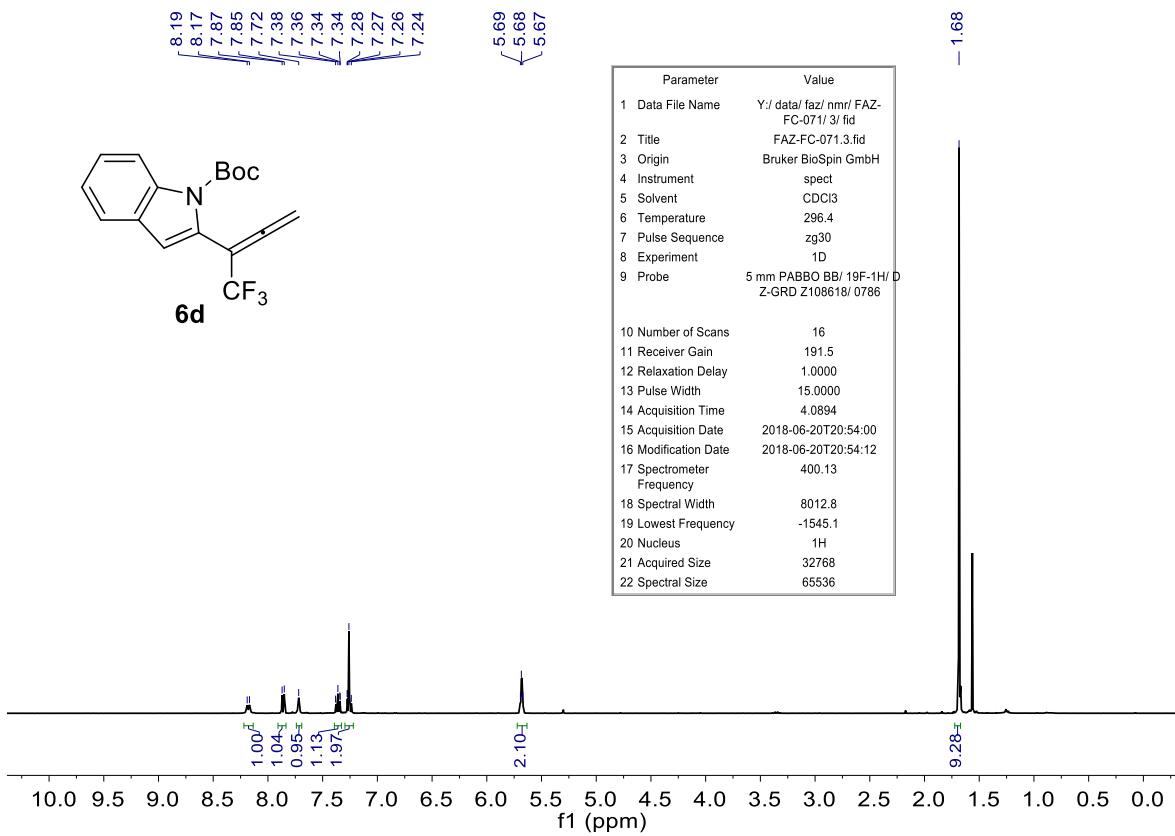
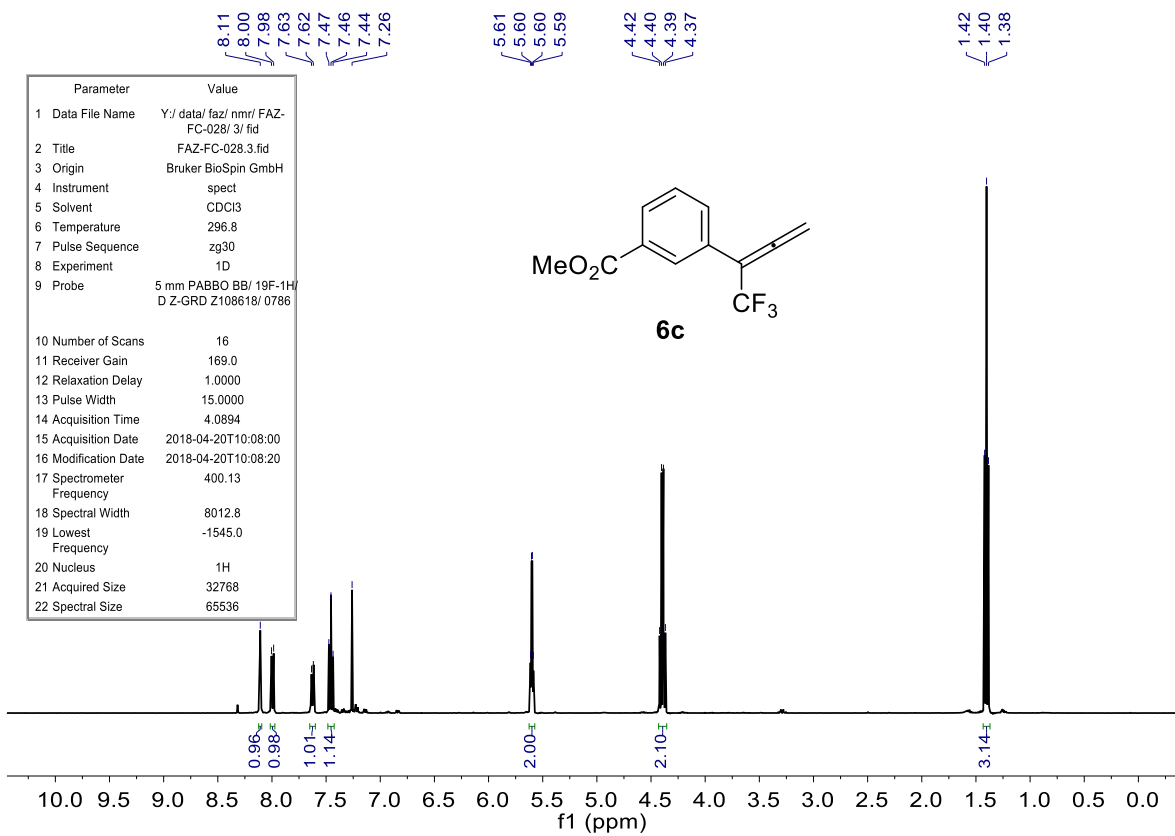


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5 Solvent	CDCl3
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7 Pulse Sequence	zg30
8 Experiment	1D
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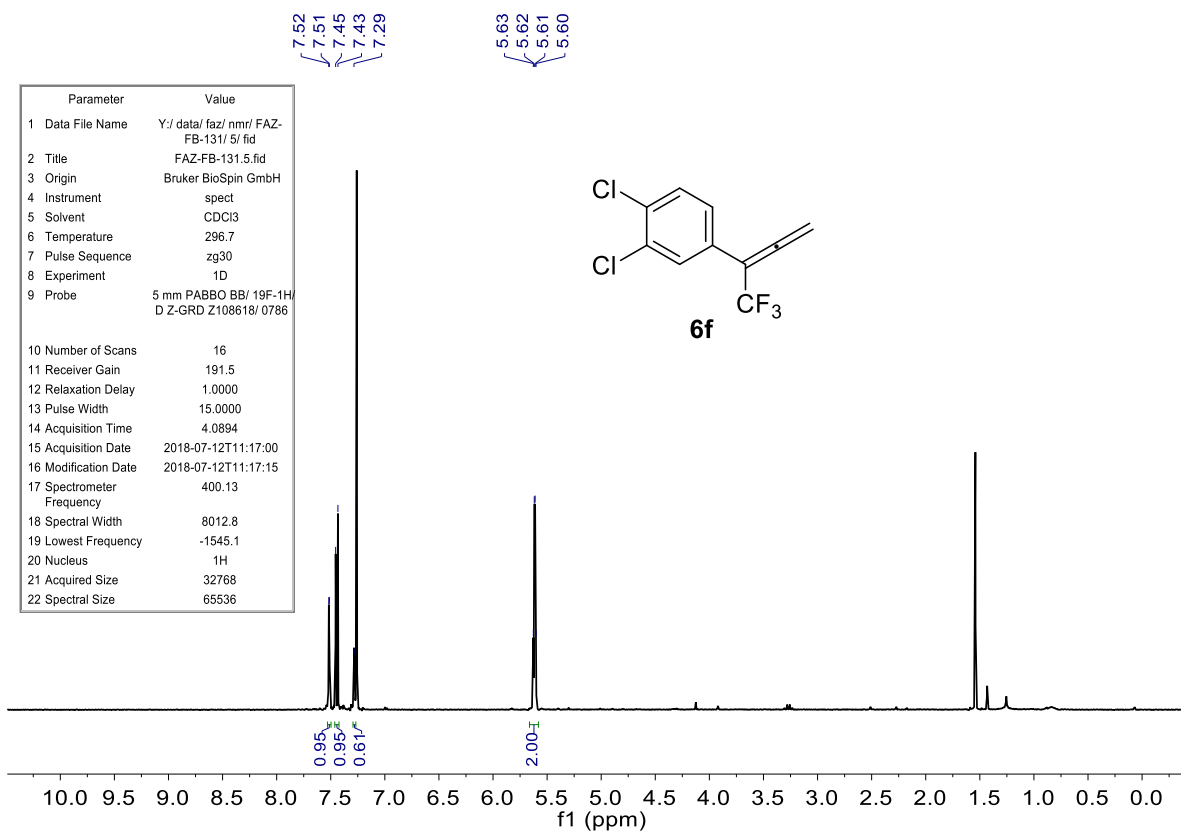
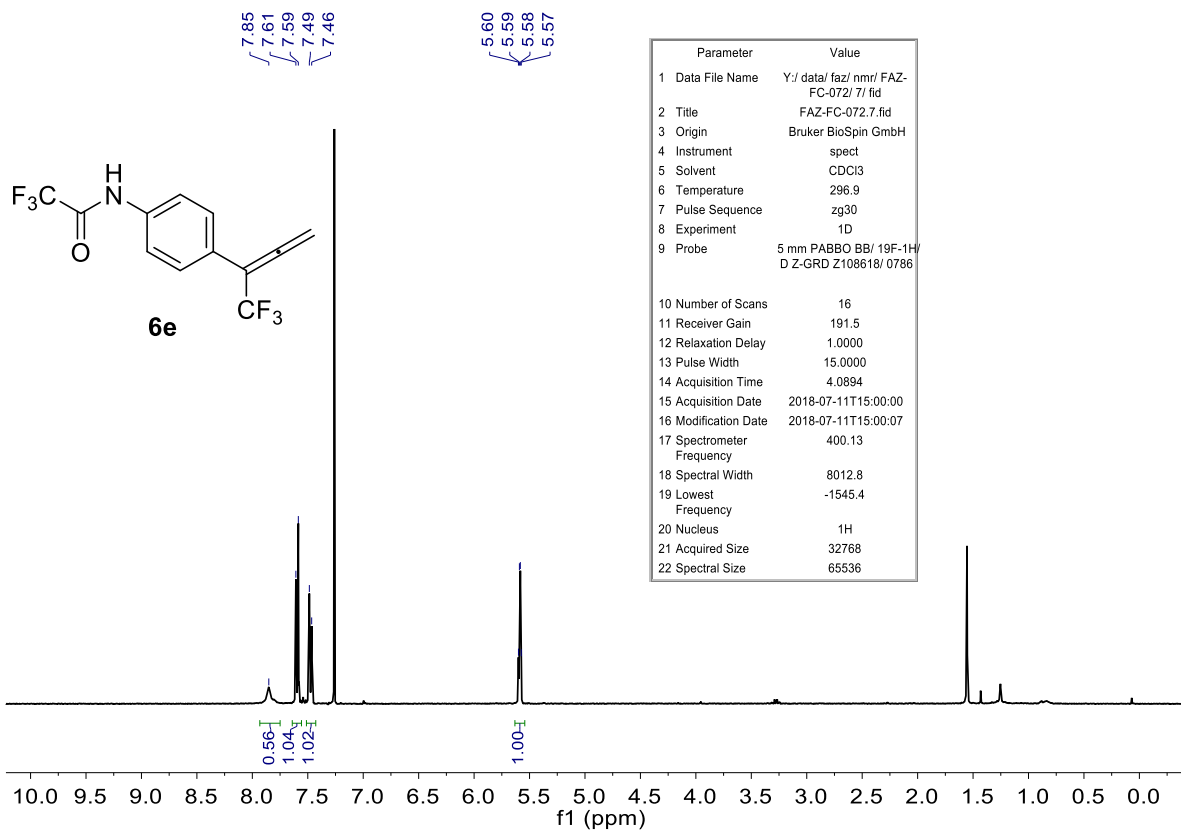


# Allenyl Trifluoromethanes (Table 3)

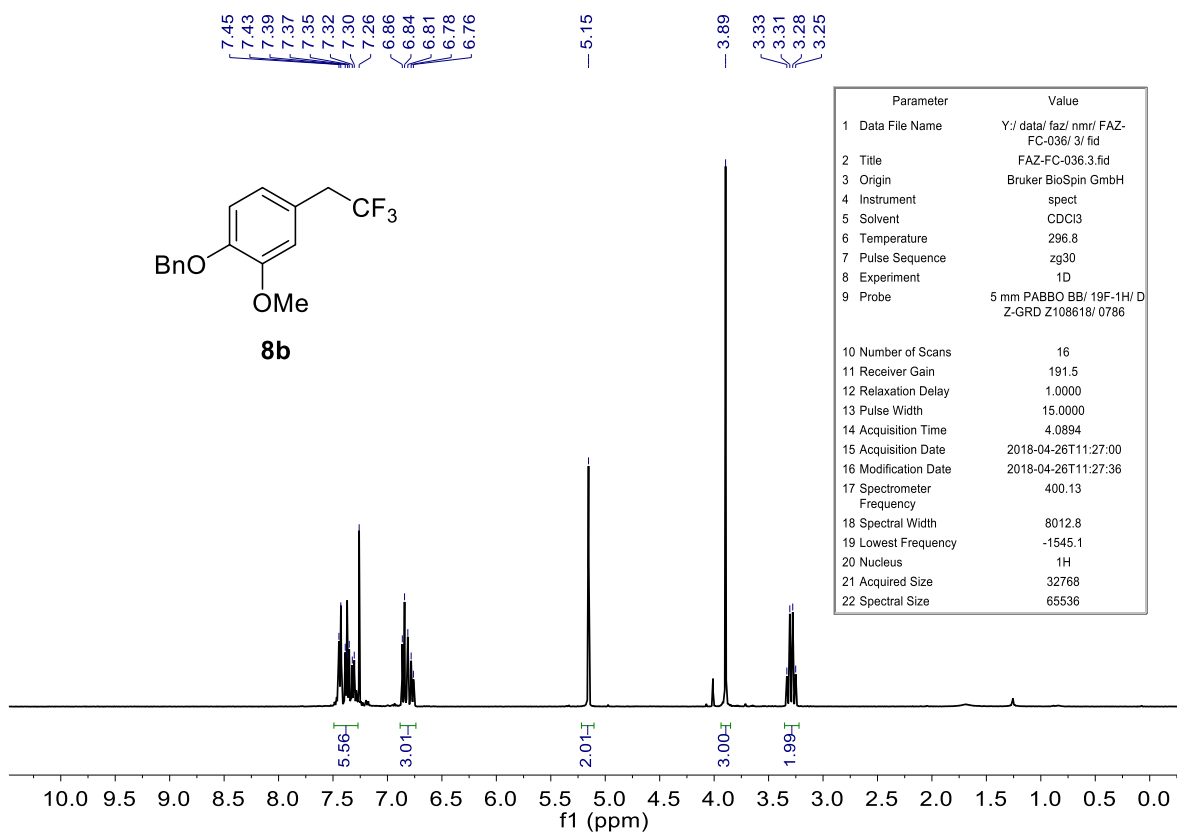
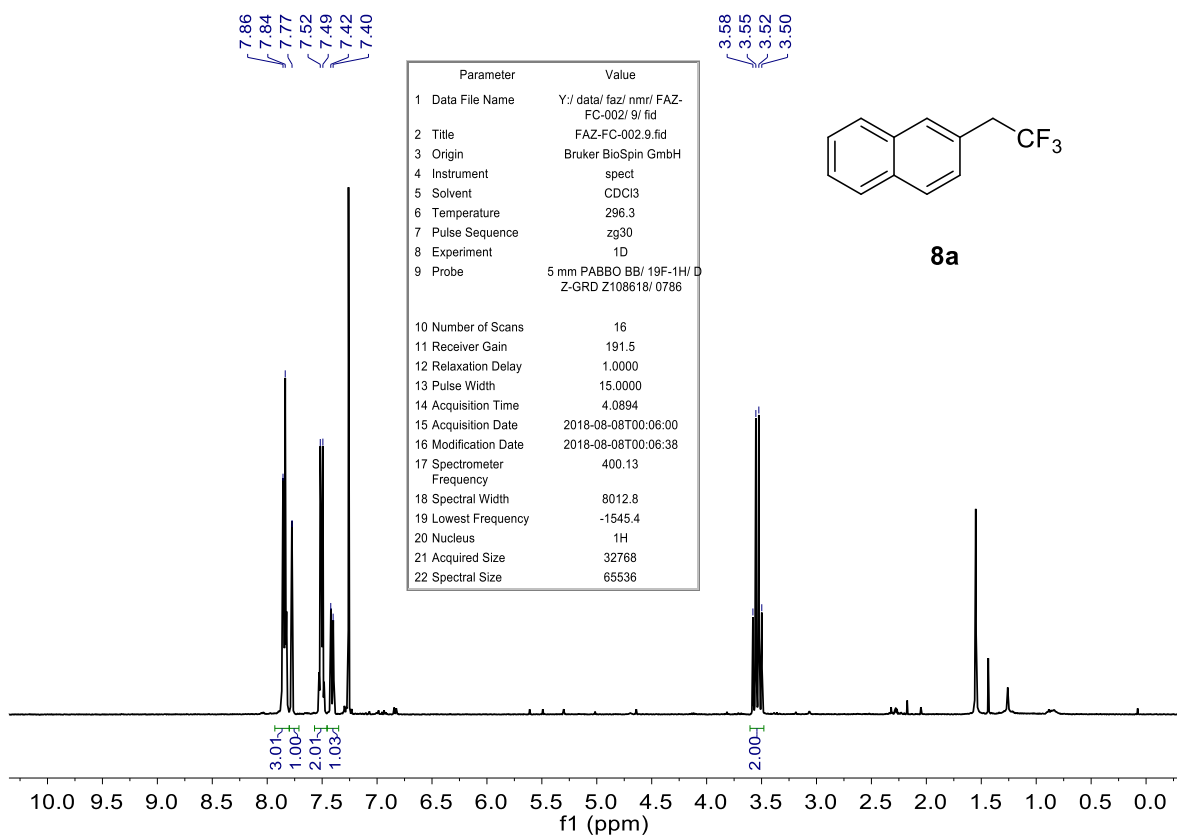


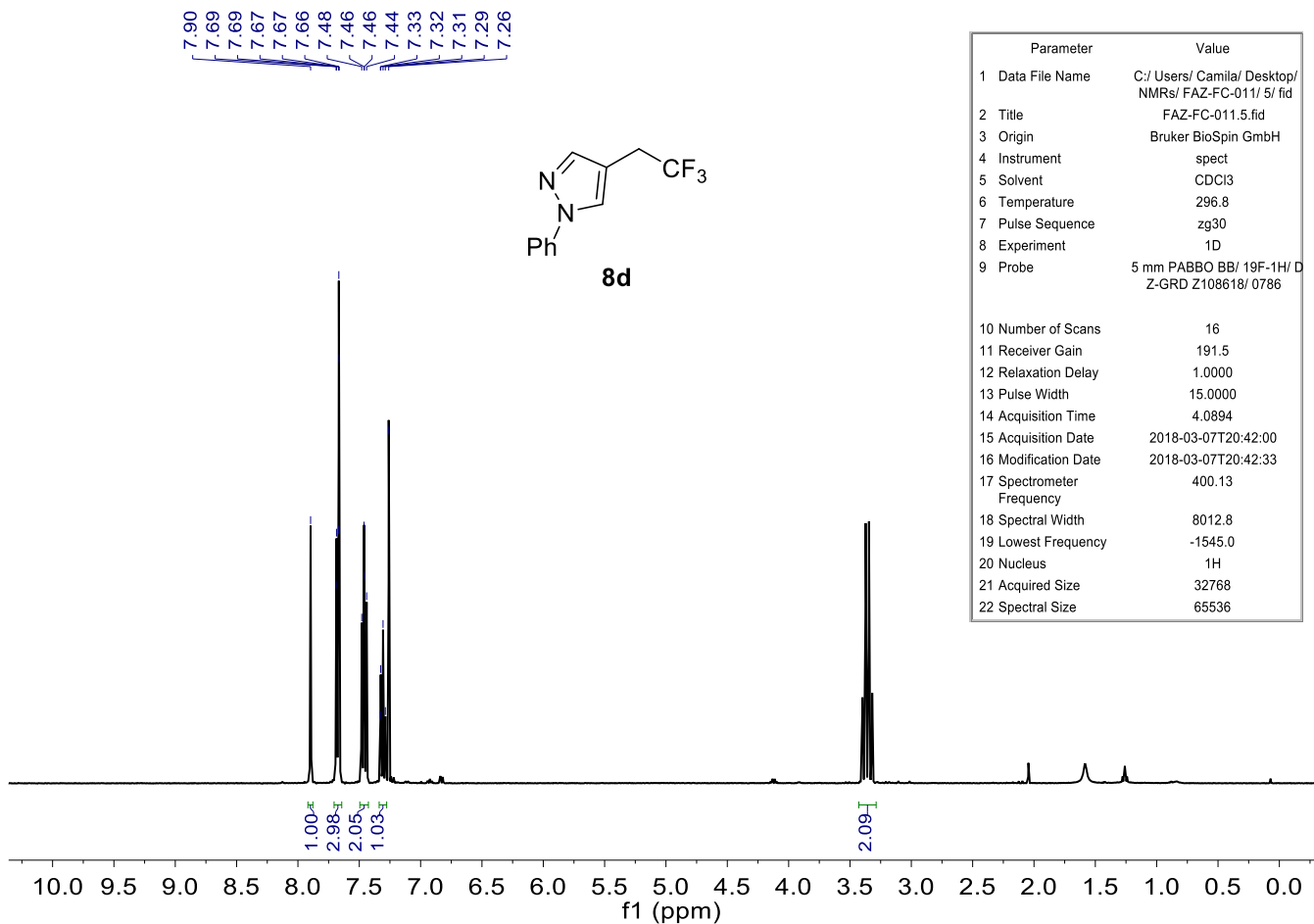
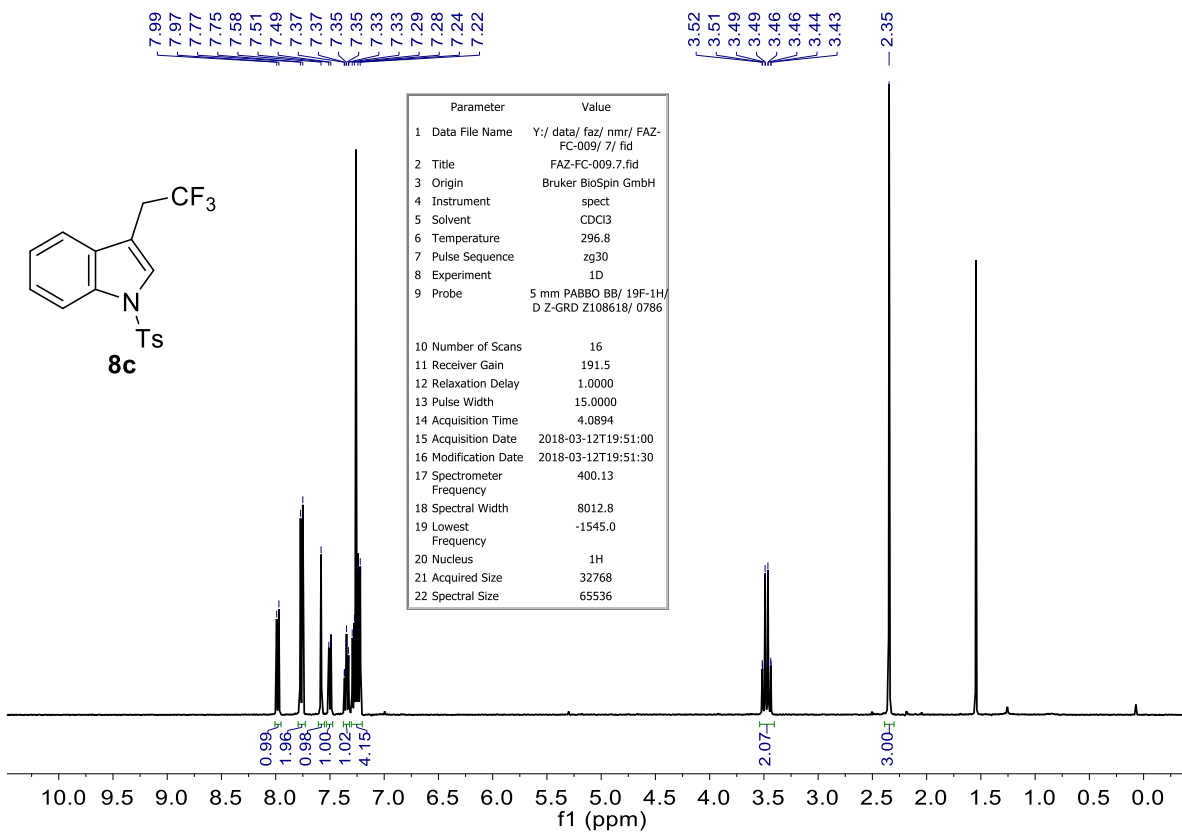


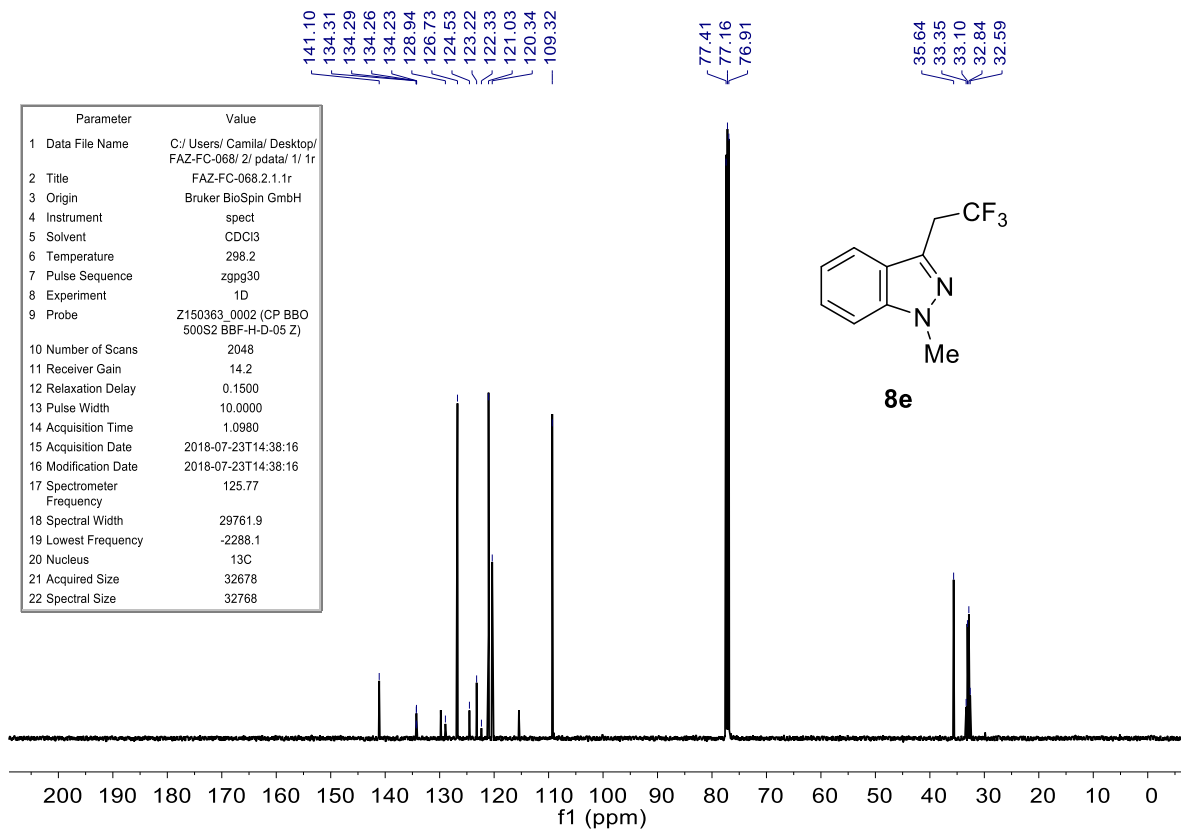
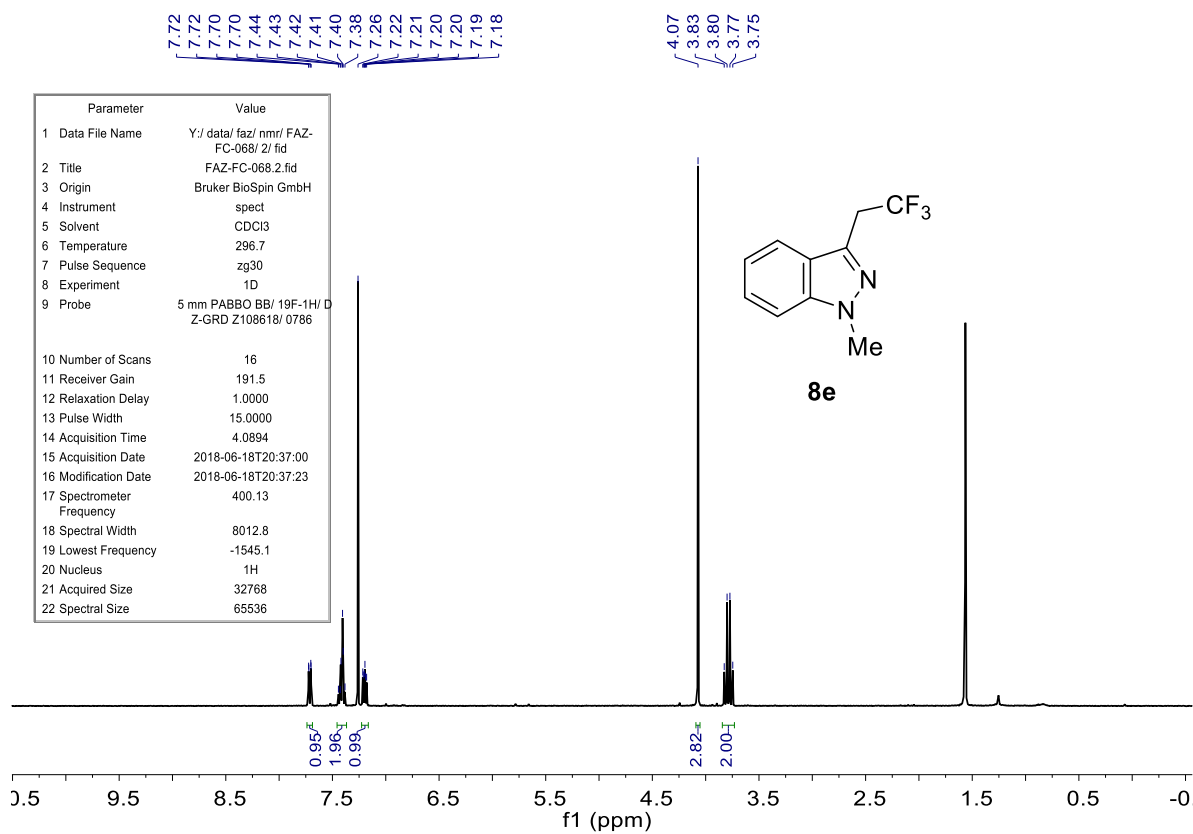


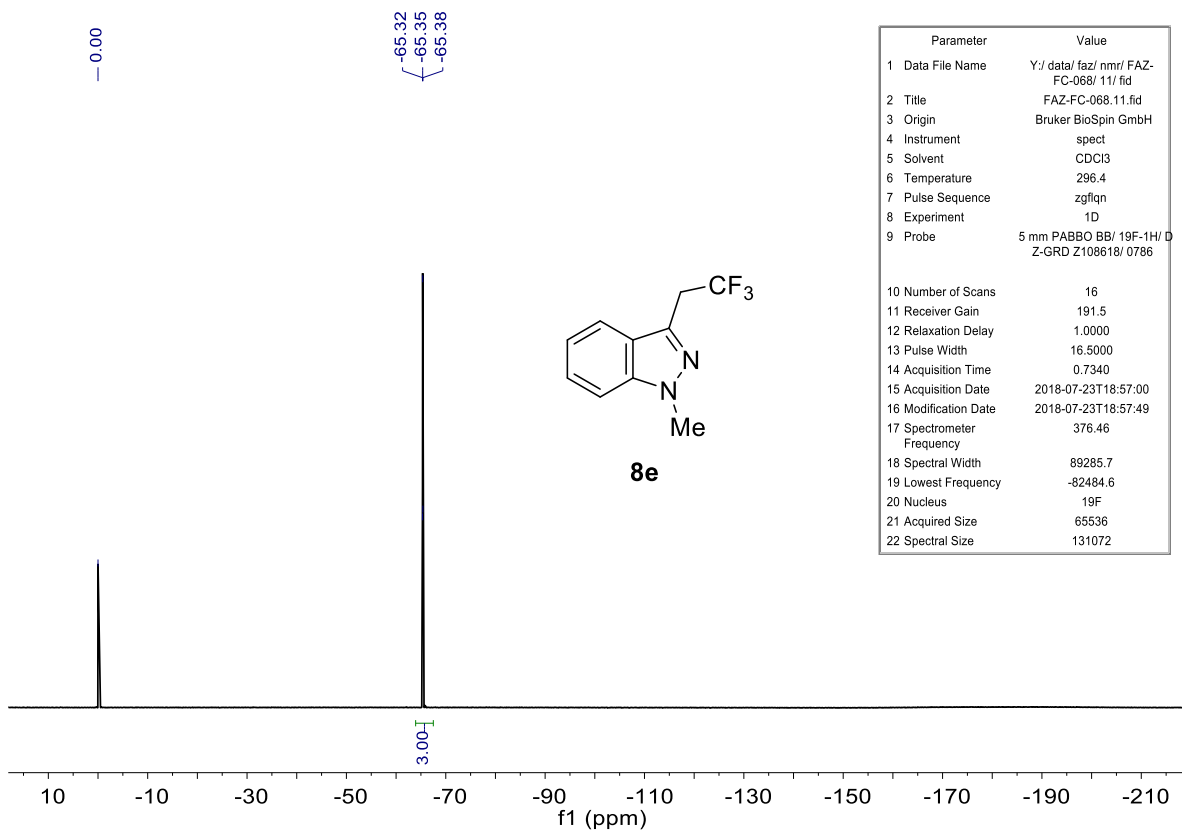


# Benzyl Trifluoromethanes (Table 4)









# Synthetic Applications (Scheme 1)

