Supporting Information

A Unified Strategy for the Syntheses of the Isoquinolinium Alkaloids Berberine, Coptisine, and Jatrorrhizine

Luis M. Mori-Quiroz, Sidnee L. Hedrick, Andrew R. De Los Santos, and Michael D. Clift*

Department of Chemistry, The University of Kansas, Lawrence, KS, 66045

*email: mclift@ku.edu

Table of Contents

1.	General Information	.S2
2.	Synthesis of Precursors and Targets	.83
3.	NMR Comparison Tables	.S 7
4.	Copies of ¹ H, ¹³ C, and ¹⁹ F NMR Spectra	.S10

1. General Information

Reactions were run in oven-dried glassware under argon atmosphere and stirred magnetically. Methanol and dichloromethane were purified by passing the solvent through activated alumina using a solvent purification system. Purification of products was carried out by flash chromatography using silica gel (230-400 Mesh, Grade 60). Reactions were monitored by Thin Layer Chromatography (TLC) (silica gel 60 F_{254}) and visualized using UV or KMnO₄, phosphomolybdic acid (PMA) and ceric ammonium molybdate (CAM) stain solutions. ¹H NMR and ¹³C NMR were recorded at room temperature in a Bruker 400 or Bruker 500 instrument. Chemical shifts are reported in ppm and referenced with respect to residual protic solvents: chloroform at 7.26 ppm, DMSO at 2.50 ppm, and methanol at 3.31 ppm (¹H NMR); and the carbon resonances of the solvent: chloroform at 77.16 ppm, DMSO at 39.52 ppm, and methanol at 49.0 ppm (¹³C NMR). The following abbreviations were used to refer to multiplicities: s = singlet, d = doublet, dd = doublet, t = triplet, q = quadruplet, m = multiplet. Infrared spectra were recorded using a Shimadzu FTIR-8400S spectrometer. Mass spectra were obtained on a Micromass LCT Premier quadrupole and time-of-flight tandem mass analyzer.

2. Synthesis of Precursors and Targets



5-(Dimethoxymethyl)-5,6,7,8-tetrahydro-[1,3]dioxolo[4,5-g]isoquinoline (5ab): To a 100 mL round bottom flask was added 3,4-(methylenedioxy)phenethylamine **2ab** (826 mg, 5 mmol) followed by dichloromethane (25 mL). A 2,2-dimethoxyacetaldehyde solution (60 wt.% in water, 2.3 mL, 15 mmol, 3 equiv) was then added, followed by addition of MgSO₄ (13 g MgSO₄ / 1 g amine). The mixture was stirred under air at room temperature for 2 h. MgSO₄ was filtered through celite and the organic collection was transferred into a new 100 mL RBF and concentrated to ~25 mL. Trifluoroacetic acid

(11.1 mL, 145 mmol, 29 equiv) was added and the reaction stirred at room temperature for 2 h. The reaction mixture was extracted with 40 mL water. The organic phase was extracted with water (2 x 40 mL). The combined aqueous extracts were basified with 3 M NaOH and extracted with dichloromethane (3 x 50 mL). The combined organic extracts were dried over MgSO₄, filtered, and concentrated. Column chromatography (3-5% methanol / dichloromethane) afforded **5ab** as a brown oil (865 mg, yield 69%). ¹H NMR (400 MHz, CDCl₃) δ 6.90 (s, 1H), 6.55 (d, *J* = 1.0 Hz, 1H), 5.88 (s, 2H), 4.41 (d, *J* = 6.0 Hz, 1H), 3.91 (d, *J* = 5.9 Hz, 1H), 3.433 (s, 3H), 3.427 (s, 3H), 3.17 (m, 1H), 2.92 (m, 1H), 2.71 (m, 2H), 2.20 (s, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 146.2, 145.6, 129.5, 127.6, 108.8, 108.2, 107.4, 100.7, 57.0, 56.1, 55.2, 40.9, 30.2. FTIR (film) 3349, 2902, 1484, 1205, 1152, 1039 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₁₃H₁₈NO₄]⁺ 252.1236, found 252.1248.



6-(2,3-Dimethoxybenzyl)-5-(dimethoxymethyl)-5,6,7,8-tetrahydro-[1,3]dioxo lo[4,5-g]isoquinoline (6): To a 100 mL round bottom flask was added isoquinoline **5ab** (1.73 g, 6.89 mmol), 2,3-dimethoxybenzaldehyde (1.26 g, 7.58 mmol, 1.1 equiv) methanol (17 mL) and acetic acid (0.80 mL, 13.78 mmol, 2 equiv). Sodium triacetoxyborohydride (1.9 g, 8.96 mmol, 1.3 equiv) was then added slowly. The flask was capped with a needle to relieve gas pressure and the needle was removed after ~30 min. After 23 h a second portion of sodium

triacetoxyborohydride (292 mg, 1.38 mmol, 0.2 equiv) was added and the reaction ran for 2 h more. The reaction was basified (pH~8) with 3 M NaOH and then concentrated to remove most methanol. The mixture was diluted with water (10 mL) extracted with dichloromethane (3 x 20 mol). The combined organic layers were dried with MgSO₄ filtered, and concentrated. Column chromatography (15-30% ethyl acetate in hexanes) afforded product **6** as a yellow oil (2.19 g, yield 79%). ¹H NMR (500 MHz, CDCl₃) δ 7.10 (dd, J = 7.8, 1.7 Hz, 1H), 7.03 (t, J = 7.9 Hz, 1H), 6.82 (m, 1H), 6.80 (s, 1H), 6.57 (s, 1H), 5.89 (dd, J = 7.8, 1.5 Hz, 2H), 4.39 (d, J = 5.0 Hz, 1H), 3.86 (s, 3H), 3.84 (d, overlapped, 1H), 3.77 (s, 3H), 3.75 (d, overlapped, 1 H), 3.70 (d, J = 5.1 Hz, 1H), 3.33 (s, 3H), 3.32 (s, 3H), 3.24 (ddd, J = 12.6, 9.4, 5.0 Hz, 1H), 2.91–2.77 (m, 2H), 2.51 (m, 1H); ¹³C NMR (126 MHz, CDCl₃) δ 152.8, 147.9, 146.3, 145.4, 133.5, 129.2, 126.7, 123.9, 122.5, 111.1, 110.1, 108.3, 107.9, 100.6, 62.7, 60.9, 56.8, 55.9, 53.7, 52.2, 45.0, 25.1; FTIR (film) 2933, 2829, 1481, 1208, 1153, 1072 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₂₂H₂₈NO₆]⁺ 402.1917, found 402,1892.



9,10-Dimethoxy-5,6,8,13-tetrahydro-[1,3]dioxolo[4,5-g]isoquinolino[3,2-*a***]isoquinolin-7-ium triflate (S8):** To a 25 mL round bottom flask was added a magnetic stirbar and amine 6 (200 mg, 0.498 mmol, 1 equiv). The flask was evacuated / backfilled with Ar (3 cycles), then dichloromethane (10 mL) was added. The solution was cooled to 0 °C and triflic acid (88 μ L, 0.1 mmol, 2 equiv) was added dropwise. After 10 minutes the cold bath was removed and the reaction stirred at room temperature for 2 h. The reaction was then cooled back to 0 °C. A yellow solid precipitated overnight. The solid was filtered, washed with

dichloromethane, and dried under vacuum to afford iminium **S8** (133 mg, 55%) as a yellow solid. ¹H NMR (400 MHz, DMSO- d_6) δ 7.77 (s, 1H), 7.15 (d, *overlapped*, 1H), 7.14 (s, 1H), 7.08 (d, J = 8.4 Hz, 1H), 6.25 (s, 2H), 5.10 (t, J = 4.0 Hz, 2H), 4.58 (d, J = 4.1 Hz, 2H), 4.14 – 4.06 (m, 2H), 3.83 (s, 3H), 3.82 (s, 3H), 3.11 (t, J = 7.8 Hz, 2H); ¹³C NMR (126 MHz, DMSO- d_6) δ 170.7, 153.9, 151.1, 147.4, 143.9, 135.4, 122.6, 121.4, 120.7 (q, J = 322.9 Hz, CF₃), 119.5, 119.3, 113.5, 108.6, 108.4, 103.1, 60.3, 56.0, 52.3, 51.1, 31.5, 25.1; ¹⁹F NMR (376 MHz, DMSO- d_6) δ -77.8; HRMS (ESI) *m/z* calcd for [C₂₀H₂₀NO₄]⁺ 338.1392, found 338.1405.



Lambertine (8):¹ To a dry 25 mL round bottom flask was added a magnetic stirbar and amine 6 (219 mg, 0.523 mmol, 1 equiv). The flask was evacuated / backfilled with Ar (3 cycles), then dichloromethane (10.5 mL) was added. The solution was cooled to 0 °C and triflic acid (0.14 mL, 1.569 mmol, 3 equiv) was added dropwise. After 5 minutes, the cold bath was removed and the reaction stirred at room temperature. After 50 minutes the reaction was opened to air, diluted with dichloromethane (40 mL), and transferred to a separatory funnel.

The solution was washed with saturated NaHCO_{3 (aq)} (2 x 10 mL). The organic phase was dried over Na₂SO₄, filtered, and concentrated to give a yellow oil. Due to instability, compound **8** was not further purified. ¹H NMR (500 MHz, CDCl₃) δ 7.17 (s, 1H), 6.73 (d, J = 1.7 Hz, 2H), 6.58 (s, 1H), 5.95 (s, 1H), 5.94 (s, 2H), 4.32 (s, 2H), 3.84 (s, 6H), 3.13 (t, J = 5.9 Hz, 2H), 2.87 (t, J = 5.9 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 150.6, 147.4, 146.8, 144.7, 141.8, 128.9, 128.7, 124.7, 122.3, 119.0, 111.6, 108.0, 103.9, 101.1, 96.5, 60.9, 56.1, 49.5, 49.2, 30.0. HRMS (ESI) *m/z* calcd for [C₂₀H₂₀NO₄]⁺ 338.1392, found 338.1399. The spectral data is in accord with published values.²



Berberine (1a): To a round bottom flask was added a magnetic stirbar and compound **6** (97.3 mg, 0.243 mmol). The flask was evacuated / backfilled with Ar (3 cycles), then dichloromethane (5 mL) was added, followed by triflic acid (65 μ L, 0.729 mmol, 3 equiv). After 1.5 h the reaction was opened to air, diluted with dichloromethane (40 mL), and transferred to a separatory funnel. The solution was washed with saturated NaHCO_{3 (aq)} (2 x 10 mL). The organic phase was dried over Na₂SO₄, filtered, and concentrated to give the crude **lambertine**

(8), which was immediately submitted to oxidation. Potassium acetate (29.74 mg, 0.303 mmol, 1.25 equiv) was added to the flask containing 8, along with ethanol (10 mL). A solution of iodine (61.5 mg, 0.243 mmol, 1 equiv) in ethanol (10 mL) was then added dropwise with vigorous stirring. After 1 h, 10% Na₂S₂O₃ (10 mL) was added and the mixture concentrated to remove ethanol. Water (5 mL) and 20% NaOH (aq) (5 mL) were added to the resulting residue and swirled for ~5 minutes. The mixture was extracted with dichloromethane (4 x 30 mL). The combined organic extracts were extracted with 1 M HCl (9 mL) and water (5 x 20 mL). The combined aqueous extracts were collected and concentrated to afford **1a** as a yellow solid (88.9 mg, yield 99%), mp = 204-206. ¹H NMR (500 MHz, DMSO-*d*₆) δ 9.90 (s, 1H), 8.95 (s, 1H), 8.21 (d, *J* = 9.1 Hz, 1H), 8.00 (d, *J* = 9.1 Hz, 1H), 7.80 (s, 1H), 7.09 (s, 1H), 6.17 (s, 2H), 4.94 (t, *J* = 6.3 Hz, 2H), 4.09 (s, 3H), 4.07 (s, 3H), 3.21 (t, *J* = 6.4 Hz, 2H); ¹³C NMR (126 MHz, DMSO-*d*₆) δ 150.4, 149.8, 147.7, 145.5, 143.7, 137.5, 133.0, 130.7, 126.8, 123.5, 121.4, 120.5, 120.2, 108.4, 105.4, 102.1, 61.9, 57.1, 55.2, 26.3; FTIR (film) 3212, 2983, 2910, 1601, 1506, 1365, 1243, 1205, 1037 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₂₀H₁₈NO₄]⁺ 336.1230, found 336.1249. The spectral data is in accord with published values.³



6-(Benzo[d][1,3]dioxol-4-ylmethyl)-5-(dimethoxymethyl)-5,6,7,8-tetrahydro-[1,3]dioxolo[4,5-g]isoquinoline (9): Following the procedure for the preparation of **6**, to **5ab** (1g, 3.98 mmol), 1,3-benzodioxole-4-carbaldehyde (0.66 g, 4.38 mmol, 1.1 equiv), methanol (10 mL), acetic acid (0.46 mL, 7.96 mmol, 2.0 equiv), sodium triacetoxyborohydride (1.10 g, 5.17 mmol, 1.3 equiv), and a second portion of sodium triacetoxyborohydride after 17 h (0.25 g, 1.2 mmol, 0.3 equiv) afforded, after workup and column chromatography (first column, 20-25% ethyl acetate / hexanes; second

column, 10% acetone / hexanes), compound **9** (869 mg, yield 57%) as a white solid, mp = 90–91 °C; ¹H NMR (400 MHz, CDCl₃) δ 6.91 (dd, J = 7.8, 1.4 Hz, 1H), 6.82 – 6.71 (m, 3H), 6.56 (s, 1H), 5.92 (s, 2H), 5.90 (d, J = 1.4 Hz, 1H), 5.89 (d, J = 1.5 Hz, 1H), 4.37 (d, J = 5.3 Hz, 1H), 3.80 – 3.70 (m, 2H), 3.68 (d, J = 5.2 Hz, 1H), 3.33 (s, 3H), 3.32 (s, 3H), 3.31 – 3.23 (m, 1H), 2.91 – 2.79 (m, 2H), 2.53 (m, 1H); ¹³C NMR (126 MHz, CDCl₃) δ 147.3, 146.4, 146.3, 145.4, 128.9, 126.5, 123.4, 121.5, 120.8, 110.1, 108.3, 107.7, 107.5, 100.8, 100.7, 62.4, 56.8, 53.5, 52.1, 45.1, 24.9. FTIR (film) 2931, 1481, 1154, 1069 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₂₁H₂₄NO₆]⁺ 386.1604, found 386.1609.

¹ Chatterjee, R.; Maiti, P. C. J. Indian Chem. Soc. 1955, 32, 609-610.

² Zhou, S.; Tong, R. Chem. Eur. J. 2016, 22, 7084–7089.

³ Blaskó, G.; Cordell, G. A.; Bhamarapravati, S.; Beecher, C. W. W. Heterocycles 1988, 27, 911–916.



Coptisine (1b): To a round bottom flask was added a magnetic stirbar, compound **9** (100 mg, 0.259 mmol), and dichloromethane (5.2 mL). The flask was then flushed with argon for ~5 min. The solution was cooled to 0 °C and triflic acid (69 μ L, 0.78 mmol, 3.0 equiv) was added dropwise. After 5 min, the ice bath was removed and the reaction stirred at room temperature for 1.5 h. The reaction mixture was then diluted with dichloromethane (~20 mL) and washed with sodium bicarbonate (2 x 5 mL). The organic phase was dried over Na₂SO₄ and

concentrated to give the crude enamine, which was immediately submitted to the oxidation step. A stirbar and potassium acetate (31.8 mg, 0.324 mmol, 1.25 equiv) were added to the flask containing the crude enamine along with ethanol (8.1 mL). A solution of iodine (65.74 mg, 0.259 mmol, 1.0 equiv) in ethanol (0.5 mL) was then added dropwise to the reaction mixture with vigorous stirring. After 1 h, Na₂S₂O_{3 (aq)} (10% w/w, 2 mL) was added to the reaction mixture to reduce excess iodine. The solution was then concentrated to remove ethanol. Water (5 mL) and 20% NaOH_(aq) (5 mL) were added to the resulting residue and swirled for ~5 minutes. The solution was extracted with dichloromethane (6 x 10 mL). The combined organic extracts were extracted with 1 M HCl (9 mL) and water (6 x 20 mL). The aqueous layers were combined and concentrated to afford **1b** as an orange solid (91 mg, yield ~99%). Mp = 240 °C (decomp.); ¹H NMR (500 MHz, CD₃OD) δ 9.73 (s, 1H), 8.75 (s, 1H), 7.89 (d, *J* = 8.7 Hz, 1H), 7.87 (d, *J* = 8.8 Hz, 1H), 7.66 (s, 1H), 6.97 (s, 1H), 6.47 (s, 2H), 6.11 (s, 2H), 4.89 (m, 2H), 3.26 (t, *J* = 6.4 Hz, 2H). 13C NMR ¹³C NMR (126 MHz, CD₃OD) δ 152.2, 150.0, 149.3, 145.8, 145.3, 139.0, 134.5, 131.8, 123.1, 122.5, 122.3, 122.0, 113.7, 109.4, 106.5, 106.2, 103.7, 57.2, 28.1. FTIR (film) 2910, 1606, 1504, 1056 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₁₉H₁₄NO₄]⁺ 320.0923, found 320.0920. The spectral data is in accord with published values.⁴



1-(Dimethoxymethyl)-7-methoxy-1,2,3,4-tetrahydroisoquinolin-6-ol (5c) and 1-(dimethoxymethyl)-7methoxy-1,2,3,4-tetrahydroisoquinolin-8-ol (iso-5c): Phenethylamine 2c (393 mg, 2.35 mmol) was added to a 50 mL pear-shaped flask. A solution of 2,2dimethoxyacetaldehyde (60 wt.% in water, 1.06 mL, 7.05

mmol, 3.0 equiv) was extracted with dichloromethane (2 x 1 mL). The combined organic phases were dried over Na₂SO₄, and then filtered into the flask containing amine **2c**. Dichloromethane (3 mL) was added to the flask and the mixture stirred at room temperature. After 30 min, the reaction was cooled to 0 °C and trifluoroacetic acid (2.70 mL, 35.25 mmol, 15 equiv) was added dropwise. After 1 h, 3 M NaOH (aq) was added until neutral *p*H. The mixture was then extracted with dichloromethane (3 x 20 mL) and the combined organic extracts were dried over Na₂SO₄, filtered, and concentrated. Column chromatography (10% methanol / dichloromethane) afforded a mixture of **5c** and **iso-5c** (6.7:1.0 ratio) as a thick golden brown oil (489 mg, yield 82%). Major isomer: ¹H NMR (500 MHz, CDCl₃) δ 6.86 (s, 1H), 6.66 (s, 1H), 4.51 (d, *J* = 5.6 Hz, 1H), 4.25 (d, *J* = 5.6 Hz, 1H), 3.85 (s, 3H), 3.46 (s, 3H), 3.43 (s, 3H), 3.37 (m, 1H), 3.15 (m, 1H), 2.83 (t, *J* = 6.0 Hz, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 145.4, 145.2, 127.4, 122.2, 114.7, 110.2, 106.3, 56.4, 56.1, 55.9, 55.9, 40.4, 27.2; Minor isomer: ¹H NMR (500 MHz, CDCl₃) δ 6.79 (m, 1H), 6.69 (m, 1H), 4.95 (d, *J* = 3.3 Hz, 1H), 4.73 (d, *J* = 3.3 Hz, 1H), 3.87 (s, 3H), 3.61 – 3.54 (m, 1H), 3.53 (s, 3H), 3.29 (s, 3H), 3.23 (m, 1H), 2.95 (m, 1H), 2.87 (m, 1H); ¹³C NMR (126 MHz, CDCl₃) δ 162.9, 162.7, 142.7, 120.1, 110.7, 104.1, 57.0, 56.6, 56.3, 52.8, 40.3, 26.4 (note: one carbon could not be located due to overlap); FTIR (film) 3384, 1674, 1151 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₁₃H₂₀NO₄]⁺ 254.1392, found 254.1394.



2-(2,3-dimethoxybenzyl)-1-(dimethoxy me thyl)-7-methoxy-1,2,3,4-tetrahydroiso qui nolin-6-ol (10c) 2-(2,3-dimethoxybenzyl)-1-(dimethoxymethyl)-7-methoxy-1,2,3,4tetrahydroisoquinolin-8-ol (iso-10c): To a 50 mL round bottom flask was added 2,3dimethoxybenzaldehyde (618 mg, 3.72 mmol,

1.93 equiv), a mixture of 5c and iso-5c (6.7:1.0 ratio, 489 mg, 1.93 mmol), and 1,2-dichloroethane (8.4 mL).

⁴ Jung, H. A.; Yoon, N. Y.; Bae, H. J.; Min, B.-S.; Choi, J. S. Arch. Pharm. Res. 2008, 31, 1405–1412.

Sodium triacetoxyborohydride (1.187 g, 5.6 mmol, 2.9 equiv) was added and the reaction stirred at room temperature for 3 h. Saturated NaHCO3 (aq) (17 mL) was added to the mixture and then extracted with dichloromethane (3 x 20 mL). The organic collection was dried over MgSO₄ and concentrated. ¹H NMR analysis showed a mixture of 10 and iso-10c (3.4:1.0 ratio). Column chromatography (0-3% methanol / dichloromethane - first column and 5-20% acetonitrile / dichloromethane - second column) afforded a mixture of 10 and iso-10c as a thick yellow oil (465 mg, yield 60%, 3.3:1.0 ratio). Major isomer: ¹H NMR (400 MHz, CDCl₃) δ 7.12 (m, 1H), 7.04 (m, 1H), 6.87 - 6.77 (m, 2H), 6.65 (s, 1H), 5.48 (s, 1H), 4.39 (d, J = 5.5 Hz, 1H), 3.86 (s, 3H), 3.85 (s, 3H), 3.853H), 3.76 (s, 3H), 3.70 (d, J = 5.6 Hz, 1H), 3.33 (s, 3H), 3.31 (s, 3H), 3.22 (m, 1H), 2.95 - 2.79 (m, 2H), 2.59 -2.43 (m, 1H). (Note: benzylic protons and one methoxy group hidden under signals at ~3.86). ¹³C NMR (126 MHz, CDCl₃) δ 152.8, 147.9, 144.5, 144.4, 133.6, 128.7, 125.0, 123.9, 122.5, 114.0, 112.50, 110.9, 107.7, 62.6, 60.9, 56.9, 56.1, 55.8, 53.6, 52.1, 44.9, 24.1. Minor isomer: (400 MHz, CDCl₃) δ 7.73 (s, 1H), 7.00 – 7.06 (m, 2H), 6.87 - 6.77 (m, 2H), 6.70 (m, 1H), 4.65 (d, J = 5.7 Hz, 1H), 4.08 (d, 5.6 Hz, 1H), 3.88 (s, 3H), 3.77 (s, 3H), 3.42 (s, 3H), 3.36 (s, 3H), 3.22 (m, 1H), 2.95 – 2.79 (m, 2H), 2.59 – 2.43 (m, 1H). ¹³C NMR (126 MHz, CDCl₃): 8 152.9, 148.0, 147.0, 144.3, 133.1, 128.6, 123.8, 122.6, 121.7, 120.5, 111.4, 111.1, 107.9, 67.2, 58.6, 56.9, 55.9, 52.9, 51.9, 44.2, 23.5. (Note: one carbon hidden under major isomer signal). FTIR (film) 3374, 1635, 1206, 1150 cm⁻¹. HRMS (ESI) m/z calcd for $[C_{22}H_{30}NO_6]^+404.2073$, found 404.2077.



Jatrorrhizine (1c): To a 50 mL round bottom flask was added compound **10** (375 mg, 0.93 mmol), dichloromethane (18.6 mL), and a magnetic stirbar. The flask was then flushed with argon for ~5 min. The solution was cooled to 0 °C and triflic acid (247 μ L, 2.79 mmol, 3.0 equiv) was added dropwise. After 5 min, the ice bath was removed and the reaction ran for 45 min. The mixture was diluted with dichloromethane (50 mL) and washed with sodium bicarbonate (2x). The organic layers were then dried over Na₂SO₄ and concentrated to a

crude enamine that was immediately submitted to oxidation. Potassium acetate (114 mg, 1.16 mmol, 1.25 equiv) was added to the flask containing crude enamine, along with ethanol (25 mL). A solution of iodine (236 mg, 0.93 mmol, 1.0 equiv) in ethanol (6 mL) was then added dropwise with vigorous stirring. After 1 h, Na₂S₂O_{3 (aq)} (10% w/w, 6 mL) was added. The mixture was then filtered through celite, rinsed with methanol, and the filtrate was concentrated. Column chromatography (2-16% methanol / dichloromethane – first column and 0-10% methanol / dichloromethane – second column) followed by purification through an ion exchange column (Dowex 50W XS H form, 200-400 mesh using methanol/water and flushing with 3M HCl/methanol (1:1), then 6M HCl/methanol (6:1), and 15% methanol / chloroform – third column) afforded **1c** (141 mg, yield 41%) as an orange solid. mp = 201-203 °C; ¹H NMR (500 MHz, CD₃OD) δ 9.74 (s, 1H), 8.77 (s, 1H), 8.11 (d, *J* = 9.1 Hz, 1H), 8.00 (d, *J* = 9.1 Hz, 1H), 7.67 (s, 1H), 6.87 (s, 1H), 4.91 (m, 2H), 4.21 (s, 3H), 4.11 (s, 3H), 4.03 (s, 3H), 3.21 (t, *J* = 6.4 Hz, 2H). ¹³C NMR (126 MHz, MeOD) δ 152.0, 151.8, 149.7, 146.2, 145.7, 140.3, 135.5, 130.3, 128.1, 124.4, 123.2, 120.9, 119.4, 115.9, 110.0, 62.5, 57.7, 57.4, 56.9, 27.7. FTIR (film) 3355, 2943, 1604, 1510, 1361, 1109 cm⁻¹. HRMS (ESI) *m/z* calcd for [C₂₀H₂₀NO₄]⁺ 338.1392, found 338.1393. The spectral data is in accord with published values.⁴

3. NMR Comparison Tables



Atom	¹ H NMR (natural) ³	¹ H NMR (synthetic)	13 C NMR (natural) ³	¹³ C NMR (synthetic)
	DMSO- d_6	DMSO- d_6	DMSO- d_6	DMSO- d_6
	δ multiplicity (J in Hz)	δ multiplicity (J in Hz)	δ	δ
1	7.79 s	7.80 s	105.4	105.4
2	-	-	147.6	147.7
3	-	-	149.8	149.8
4	7.09 s	7.09 s	108.4	108.4
4a	-	-	130.6	130.7
5	3.22 t (6.2)	3.21 t (6.4)	26.4	26.3
6	4.95 t (6.3)	4.93 t (6.3)	55.2	55.2
8	9.91 s	9.90 s	145.4	145.5
8a	-	-	121.4	121.4
9	-	-	143.6	143.7
10	-	-	150.4	150.4
11	8.20 d (9.1)	8.21 d (9.1)	126.7	126.8
12	8.01 d (9.1)	8.00 d (9.1)	123.5	123.5
12a	-	-	132.9	133.0
13	8.96 s	8.95 s	120.2	120.2
13a	-	-	137.4	137.5
13b	-	-	120.4	120.5
OCH ₂ O	6.17 s	6.17 s	102.1	102.1
9-OCH ₃	4.10 s	4.09	62.0	61.9
10-OCH ₃	4.07 s	4.07	57.1	57.1



coptisine 1b

Atom	¹ H NMR (natural) ⁴	¹ H NMR (synthetic)	¹³ C NMR (natural) ⁴	¹³ C NMR (synthetic)
	CD ₃ OD	CD ₃ OD	CD ₃ OD	CD ₃ OD
	δ multiplicity (J in	δ multiplicity (J in Hz)	δ	δ
	Hz)			
1	7.63 s	7.66 s	105.3	106.5
2	-	-	148.1	149.3
3	-	-	148.8	150.0
4	6.84 s	6.97 s	108.2	109.4
4a	-	-	130.6	131.8
5	3.23 m	3.26 t (6.4)	27.0	28.1
6	4.89 m	4.89 m	56.0	57.2
8	9.71 s	9.73 s	144.6	145.8
8a	-	-	112.5	113.7
9	-	-	144.1	145.3
10	-	-	151.0	152.2
11	7.87 d (8.0)	7.89 d (8.7)	120.7	122.0
12	7.83 d (8.0)	7.87 d (8.8)	121.9	123.1
12a	-	-	133.2	134.5
13	8.71 s	8.75 s	120.7	122.7
13a	-	-	137.8	139.0
13b	-	-	121.6	122.5
2,3-OCH ₂ O	6.09 s	6.11 s	105.0	106.2
9,10-OCH ₂ O	6.45 s	6.47 s	102.5	103.7

Note: Although all resonances in the ¹³C NMR spectrum of synthetic coptisine are shifted from that of natural coptisine,⁴ the consistent difference (~1.2 ppm) led us to believe these differences are due to the lack of reference in the reported spectrum of natural coptisine.



jatrorrhizine 1c

Atom	¹ H NMR (natural) ⁴	¹ H NMR (synthetic)	¹³ C NMR (natural) ⁴	¹³ C NMR (synthetic)
	CD ₃ OD	CD ₃ OD	CD ₃ OD	CD ₃ OD
	δ multiplicity (J in	δ multiplicity (J in	δ	δ
	Hz)	Hz)		
1	7.63 s	7.67 s	108.8	110.0
2	-	-	150.5	151.8
3	-	-	148.4	149.7
4	6.84 s	6.87 s	114.7	115.9
4a	-	-	129.1	130.3
5	3.18 m	3.21 t (6.4)	26.4	27.7
6	4.48 m	4.91 m	56.2	56.9
8	9.71 s	9.74 s	145.0	146.2
8a	-	-	118.2	119.4
9	-	-	150.6	152.0
10	-	-	144.5	145.7
11	8.08 d (8.0)	8.11 d (9.1)	123.1	124.4
12	7.97 d (8.0)	8.00 d (9.1)	126.8	128.1
12a	-	_	134.2	135.5
13	8.74 s	8.77 s	119.7	120.9
13a	-	-	139.1	140.3
13b	-	-	122.0	123.2
2-OMe	4.00 s	4.03 s	56.4	57.4
9-OMe	4.18 s	4.21 s	61.3	62.5
10-OMe	4.09 s	4.11 s	55.7	55.7

Note: Although all resonances in the ¹³C NMR spectrum of synthetic jatrorrhizine are shifted from that of natural jatrorrhizine,⁴ the consistent difference (~1.2 ppm) led us to believe these differences are due to the lack of reference in the reported spectrum of natural jatrorrhizine.

4. Copies of ¹H, ¹³C, and ¹⁹F NMR Spectra

























