Author contributions: C.M.W. conceived the project, wrote the first draft of the paper with subsequent assistance from all authors, and was the principle supervisor of H.X. and S.D.H. C.M.W., H.X., S.D.H. and T.F-J. chose the COT/cubane targets. H.X. S.D.H. and T.F-J undertook the synthetic preparation of all COT/cubane analogues, performed the log P measurements, and obtained the respective characterisation data. J.-K.T., D.-Y.J. and X.C designed the warfarin study and analyzed results after evaluation of warfarin and related compounds (6-8) against the VKOR assay and all 27 human VKOR mutation studies. S.G. and M.T.S designed the metabolism study of warfarin and related compounds (6-8) and analysed the results. T.H.J.B., K.-A.C. and K.N.J. designed and analyzed results after evaluation of moclobemide and related compounds (9, 14 and 21). A.K. and M.T.S. designed the pravadoline and related compounds (10, 15 and 22) study and analysed results. G.M.B. designed and performed the cancer cell assay experiments and analysed results for SAHA and related compounds (11, 16 and 23). C.P. and J.Mc. designed the scabies study and analysed results for benzyl benzoate and related compounds (12, 17-19 and 24-26). C.-E.M. and G.H.W. designed the diflubenzuron and related compounds (13, 20 and 27) study and analysed results. G.P.S. and J.T. provided cubane expertise, and G.P.S. co-supervised H.X, S.D.H and T F-J. J.D.V. provided expertise on P450 metabolism in relation to the metabolism study. P.V.B. performed the X-ray crystallographic structure determination. J.M.B. performed all in silico calculations and geometry optimisations, assisted with graphical layouts and manuscript preparation.

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Supplementary Information Part 1: Synthesis and Physical Property Measurements

Abbreviations

°C	degrees Celsius	DMSO	dimethyl sulfoxide
Δ	heat/reflux	d	doublet
Δ	chemical shift	EI	electron ionisation
μL	microliter	et al	et alii / et aliae (and
μm	micrometer	EtOAc	ethyl acetate
Acetone-	deuterated acetone	ESI	electrospray ionisation
a ₆ ANOVA	analysis of variance	eV	electron volts
Boc	tert-Butyl carbonate	GCMS	gas chromatography
Br.	broad	g	gram
BPU	benzoylphenyl ureas	HCI	hydrochloric acid
calcd	calculated	h	hour(s)
CDCI ₃	deuterated chloroform	HRMS	high resolution mass
СОТ	1,3,5,7-	hu	photoirradiation
CSIRO	Commonwealth Sciontific and Industrial	Hz	Hertz
D₂O	Research Organisation	IC ₅₀	50% maximal inhibitory
D20		IGR	insect growth regulator
dec	decomposition	in vacuo	in a vacuum
	dichloromothano	in vitro	within the glass
		in vivo	within the living
		J	coupling constant
DMSO- <i>d</i> ₆	aminopyridine deuterated dimethyl sulfoxide		

LC- MS/MS	liquid chromatography-	THF	tetrahydrofuran
	spectrometry/mass	TLC	thin layer
Methanol-	deuterated methanol	t	tertiary
MHz	mega-hertz	t	triplet
min	minute(s)	UV	ultraviolet
mL	millilitre(s)	VKOR	vitamin K epoxide
mmol	millimole(s)	wt	reductase weight
mm	millimetre		
m.p.	melting point		
m/z	mass to charge ratio		
m	multiplet		
NADPH	nicotinamide adenine dinucleotide phosphate, reduced form		
nmol	nanomole(s)		
NMR	nuclear magnetic		
PBS	phosphate buffered		
post hoc	after this		
ppm	parts per million		
Prep.	preparative		
q	quartet		
rh	relative humidity		
rpm	revolutions per minute		
SE/SEM	standard error/standard		
S	singlet		
TEA	triethylamine		

General Experimental

Synthetic procedures pertaining to warfarin (6), moclobemide (9), pravadoline (10), SAHA (11), benzyl benzoate (12), diflubenzuron (13), and analogues (performed by the group led by Prof. Craig Williams at the School of Chemistry and Molecular Biosciences):

Reactions employing microwave conditions were carried out in a CEM Discover LabMate. Argon was dried by passing through a drying tube containing 4Å molecular sieves and Drierite[™]. Glassware was oven dried (160 °C) before use with anhydrous solvents and reagents. THF and diethylether were freshly distilled to dryness over elemental sodium/benzophenone under an argon atmosphere. DCM was freshly distilled to dryness over calcium hydride under an argon atmosphere. Unless stated otherwise commercially available chemicals were used without further purification. 2-Mercaptopyridine N-oxide sodium salt was concentrated to dryness then washed with ethyl acetate. Diphenylphosphoryl azide (DPPA) was distilled to dryness following known procedures.^[41] Acetonitrile, methanol, ethanol, *t*-butanol, N, N, N', N'-tetramethylethylenediamine (TMEDA), N,N-dimethylformamide (DMF), triethylamine (TEA), N,N-diisopropylamine and 2,2,6,6-tetramethylpiperidine were freshly distilled to dryness over calcium hydride under an argon atmosphere or reduced pressure.^[42] Chloroform was washed with water, dried over magnesium sulfate then distilled to dryness over P₂O₅ under an argon atmosphere. NMR spectra were recorded under standard conditions (unless stated otherwise) using Bruker AV 500, 400 and 300 MHz or Bruker AS 500 MHz spectrometers and were referenced with residual monoprotic solvent peaks (e.g. CDCl₃, C₆D₆ etc.).^[43] Samples run in D₂O were referenced using a dioxane standard (¹H = δ 3.75 ppm, ¹³C = 67.2 ppm). Coupling constants (*J*) are quoted to the nearest 0.1 Hz. The following abbreviations are used to report multiplicities: s = singlet, d = doublet, t = triplet, m = multiplet, quin = quintet, sext = sextet, sep = septet, br. = broad. High resolution ESI mass spectra were recorded using a Bruker MicroTOF-Q (quadrupole-Time of Flight) with a Bruker ESI source. Optical rotations were performed on a JASCO P-2000 polarimeter. Melting points were determined using a Digimelt MPA 160 melting point apparatus and are reported uncorrected. Crystallographic data were collected at 190K on an Oxford Diffraction Gemini Ultra S CCD diffractometer employing graphite monochromated Mo-Ka radiation (0.71073 Å) in the range $2 < 2\theta < 50^{\circ}$. Data reduction and empirical absorption corrections were performed with CrysAlisPro (version 1.171.38.43). The structure was solved *via* dual space methods with SHELXT. Refinement was performed by full-matrix least-squares analysis against F² using SHELXL-2014 within the WinGX package. Carbon-bound hydrogen atoms were included at estimated positions using a riding model. The $U_{iso}(H)$ of the methyl group were constrained to 1.5 times U_{eq} of the parent carbon atom, while $U_{iso}(H)$ of the remaining carbon-bound positions were constrained to 1.2 times U_{eq} of the respective parent carbon atoms. The nitrogen-bound hydrogen atom was located from a difference map; its coordinates were allowed to refine freely, and $U_{iso}(H)$ was constrained to 1.5 times U_{eq} of the parent nitrogen atom. Drawings of the molecule were produced with ORTEP-3 with all thermal displacement elipsoids depicted at the 50% probability level. Flash column chromatography was run using Merck silica gel 60 (230–400 mesh). Fractions were initially visualised using UV irradiation and subsequently by heating TLC plates exposed to either ceric ammonium molybdate (Goofy's stain) or 10 % aqueous potassium permanganate. TLC was performed with Merck precoated silica gel plates (silica gel 60 F₂₅₄) 0.2 mm). Preparative Chiral HPLC was performed at the Analytical and Preparative Enantioselective Chromatography facility at the School of Chemistry and Molecular Biosciences, University of Queensland.

Warfarin



Scheme S1: synthesis of cubyl warfarin (7), cyclooctatetraenyl warfarin (8) and determination of the absolute stereochemistry of enantiomer 8a.



Scheme S2: separation of commercially available racemic warfarin (6) and cubyl warfarin (7).

4-(Methoxycarbonyl)cubane-1-carboxylic acid (S2)



Following the procedure of Eaton *et al*:^[44] dimethyl cubane-1,4-dicarboxylate (**S1**) (6.079 g, 27.60 mmol) was suspended in THF (250 mL). A solution of sodium hydroxide (1.220 g, 30.50 mmol) in methanol (14 mL) was added dropwise and the solution was left to stir for 16 h. The THF was removed *in vacuo* and the residual solid was suspended in water (200 mL), and washed with DCM (3 x 100 mL). The aqueous phase was acidified to pH 2 with hydrochloric acid (10 M) and washed again with DCM (3 x 100 mL). The combined organic phases were dried over magnesium sulfate and concentrated to give the title compound (5.385 g, 95%) as a white solid. ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 4.28 (s, 6H), 3.73 (s, 3H).

Methyl cubane-1-carboxylate (S4)



Following the procedure of Ko *et al*:^[45] to a solution of 4methoxycarbonylcubane-1-carboxylic acid (**S2**) (9.654 g, 46.82 mmol) in anhydrous DCM (500 mL) was added oxalyl chloride (4.81 mL, 56.08 mmol) and anhydrous DMF (0.3 mL) under an argon atmosphere. After 1 h, the mixture was concentrated *in vacuo* and further dried under high vacuum (1 h). Separately, freshly ground 2-mercaptopyridine *N*-oxide sodium salt (**S3**) (10.639 g, 71.33 mmol) and DMAP (59 mg, 0.48 mmol) were suspended in anhydrous chloroform (500 mL) and heated to reflux whilst under irradiation from a 500-W tungsten lamp. [Note: Chloroform (1000 mL) was washed with water (3 x 1000 mL) and dried over molecular sieves prior to use in order to remove the ethanol stabilisier.] The newly formed acid chloride was suspended in anhydrous chloroform (500 mL) and added slowly over 1 h to the refluxing mixture under an argon atmosphere. After reflux (4 h) the suspension was washed with water (3 x 500 mL), dried over magnesium sulfate and concentrated to give a brown oil. Purification by column chromatography (10% ethyl acetate/petroleum ether v/v) gave the title compound (6.820 g, 90%) as a white sweet smelling solid. ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 4.26–4.24 (m, 3H), 4.04–3.99 (m, 4H), 3.70 (s, 3H).

Cubane-1-carboxylic acid (S5)



Methyl cubane-1-carboxylate (**S4**) (6.600 g, 0.041 mmol) was suspended in THF (200 mL). A solution of sodium hydroxide (2.004 g, 50.10 mmol) in methanol (12 mL) was added dropwise and the solution was left to stir for 16 h. The THF was removed *in vacuo* and the residual solid was suspended in water (200 mL) and washed with DCM (3 x 100 mL). The aqueous phase was acidified to pH 2 with hydrochloric acid (10 M) and washed again with DCM (3 x 100 mL). The combined organic phases were dried over magnesium sulfate and concentrated to give the title compound (5.324 g, 88%) as a yellow solid. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 4.33–4.29 (m, 3H), 4.07–4.00 (m, 4H).



Following the procedure of Priefer *et al*:^[46] to a solution of cubane-1-carboxylic acid (**S5**) (100 mg, 0.67 mmol) in anhydrous THF (10 mL) was slowly added borane dimethylsulfide complex (5 M in diethyl ether, 0.40 mL, 2.02 mmol) under an argon atmosphere. The solution was left to stir for 1 h then water (5 mL) was cautiously added. The THF was removed *in vacuo* and the residue was

washed with DCM (3 x 10 mL). The combined organic phases were dried over magnesium sulfate, concentrated and purified by column chromatography (50% ethyl acetate/petroleum ether v/v) to give the title compound (85 mg, 95%) as a white solid. ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 4.08–4.01 (m, 1H), 3.96–3.88 (m, 6H), 3.76 (d, *J* = 5.5 Hz, 1H).



To a solution of cubylmethanol (**S6**) (1.60 g, 11.9 mmol) and *N*-methylmorpholine-*N*-oxide (2.70 g, 23.80 mmol) in anhydrous DCM (20 mL) under an argon atmosphere, was added tetrapropylammonium perruthenate (0.21 g, 0.59 mmol) and 4Å molecular sieves (200 mg). The reaction mixture was stirred at room temperature for 1 hour. The resulting solution was filtered through Celite and eluted with additional anhydrous DCM (20 mL). 1-(Triphenylphosphoranylidene)-2-propanone (**S7**) (7 g, 23.80 mmol) was then added to the eluent and the resulting mixture was heated to reflux in DCM for 12 h. After complete conversion of starting material the solvent was removed and the residue was purified by column chromatography on silica gel (5% ethyl acetate/hexane v/v) to give the title compound (1.35 g, 68% yield, over two steps) as a colorless oil. ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 7.00 (d, *J* = 16.0 Hz, 1H), 5.92 (d, *J* = 16.0 Hz, 1H), 4.09-4.05 (m, 4H), 3.99- 3.95 (m, 3H), 2.27 (s, 3H); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 198.7, 146.5, 127.7, 58.5, 50.1, 48.5, 44.4, 27.0; HRMS-ESI calcd for C₁₂H₁₃O+ ([M+H]⁺): 173.0961; found: 173.0960.



To a solution of 4-(cubanyl) but-3-en-2-one **S8** (45 mg, 0.26 mmol) and 4hydroxycoumarin (**S9**) (45 mg, 0.27 mmol) in dimethyl sulfoxide (1.5 mL) was added (*L*)-Proline (15 mg, 0.13 mmol) and the mixture was stirred at room temperature for 24 h. Water was added to the mixture and the aqueous layer was extracted with ethyl acetate (3 x 8 mL). The combined organic layers were dried (Na₂SO₄), and the solvent removed *in vacuo* to give the crude product which was purified by column chromatography on silica gel (20 % ethyl acetate/hexane v/v) the title compound (60 mg, 70%) as a white solid. The cubane analogue of warfarin (**7**) existed as an equilibrium in solution with the corresponding cyclic hemiketal (**S10**).



m.p. 170–172°C (dec.); ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 9.77 (s, 1H, -OH, keto), 7.92-7.91 (m, 1H, ArH, keto), 7.79-7.74 (m, 0.48H, ArH, ketal), 7.51-7.46 (m, 1.5H, ArH), 7.30-7.21 (m, 3H, ArH), 3.98-3.81 (m, 10H, cubane), 3.38 (d, J = 10 Hz, 1H, CH₂, keto), 3.31-3.22 (m, 1.5H, CH, keto, CH₂, CH, ketal), 3.15 (s, 0.2H, -OH, ketal), 3.01 (s, 0.26H, ketal), 2.76 (d, J = 20.0 Hz, 1H, CH₂, keto), 2.28-2.20 (m, 0.3 H, CH₂, ketal), 2.24 (s, 3H, CH₃, keto), 2.00-1.95 (m, 0.26H, CH₂, ketal), 1.77 (s, 0.6H, CH₃, ketal), 1.76-1.70 (m, 0.2H, CH₂, ketal), 1.71 (s, 0.7H, CH₃, ketal); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 124.3, 162.6, 161.8, 158.0, 152.6, 152.5, 131.5, 131.3, 123.7, 123.6, 123.5, 122.6, 116.6, 116.4, 116.1, 115.7, 106.1, 102.5, 99.1, 60.9, 59.6, 49.2, 49.0, 48.9, 48.0, 47.4, 44.2, 44.1, 44.0, 43.3, 34.1, 33.4, 31.5, 29.8, 28.0; HRMS-ESI calcd for C₂₁H₁₈O₄Na⁺ ([M+Na]⁺): 357.1097; found: 357.1110.

Enantiomers 7a and 7b and cyclic hemiketal S10a



Racemic cubyl warfarin (**7**) (200 mg) was separated by preparative chiral HPLC to provide two enantiomers **7a** (90 mg) and **7b** (90 mg). **7a**: $[\alpha]^{24}$ _D 29.6 (c 0.06, CHCl₃). **7b**: $[\alpha]^{24}$ _D -23.4 (c 0.08, CHCl₃). X-ray quality crystals of **S10a** were grown from ethyl acetate and petroleum ether.

<u>CHROMATO</u>	GRAM REPORT
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168:49 mV 158:5 148:5 138:5 128:5 128:5 18:5 18:5 98:5	1.2 V1.3 V.1.4 V.1.1 10.9 12.1 14.9 16.9 18
Operation Mode :Standard mode Method name :premix_20IPA_80Hex_5 Col Name :AD Column Size CSP Density :0.6000 Packed Length :25.0000 Inner Diameter :2.1000 Inj Vol µL :2000	UV Wavelengths (nm) :254,-,-,- Flow mL/min :10.00 Press bar :150 Run Time :19.00 min Meth. Vol mL :220 Sample ID :HX_1-24 Data Path :C:\Program Files\PDR\&lice\ Hui Xing_HX-1-24
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1 HX 1-24 1 (+) 9.90 12.17 11.22 88.735 2 HX_1-24_2 (+) 13.13 15.28 14.03 98.400	6471.409 49.31 6471.381 477.805 0.0000 6652.765 50.69 6652.756 861.424 0.0000

Total Peak Area :13124.173 mVs





PEAK LIST REPORT

No	Name	Type	Start (min)	End (min)	RT (min)	Height (mV)	Area (mVs)	<pre>% Area (%)</pre>	Col. Area (mVs)	Plates (#)	Conc (mg/mL)
1	HX_1-24_2	(+)	13.75	15.35	14.75	20.128	1302.169	100.00	1302.157	958.435	0.0000

Total Peak Area :1302.169 mVs

(E)-3-Cyclooctatetraenyl-1-methylprop-2-en-1-one (S11)



To a solution of cubylmethanol (**S6**) (52 mg, 0.39 mmol), NMO (141 mg, 1.20 mmol) and 4Å molecular sieves (30 mg) in anhydrous DCM (5 mL) was added TPAP (1 mg, 0.004 mmol) under an argon atmosphere. The mixture was stirred at rt for 20 min until the total consumption of the alcohol. The solvent was removed under reduced pressure to dryness. To the intermediate aldehyde was added toluene (5 mL) and 1-(triphenylphosphoranylidene)-2-propanone (**S7**) (191 mg, 0.60 mmol) and the resulting mixture was stirred at 60 °C for 6 h. After cooling, the solvent was removed, and the residue was purified by column chromatography (5% ethyl acetate/hexane v/v) to give the title compound (35 mg, 52%) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 7.17 (d, *J* = 16.0 Hz, 1H), 6.22-6.21 (m, 1H), 5.98-5.79 (m, 7H), 2.28 (s, 3H); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 198.3, 144.7, 140.4, 139.8, 133.7, 131.7, 131.3, 130.9, 129.1, 127.0, 27.5; HRMS-ESI calcd for C₁₂H₁₂ONa⁺ ([M+Na]⁺): 195.0780; found: 195.0788.



To a solution of (*E*)-3-cyclooctatetraenyl-1-methylprop-2-en-1-one (**S11**) (60 mg, 0.35 mmol) and 4-hydroxycoumarin (**S9**) (62 mg, 0.38 mmol) in dimethyl sulfoxide (1.5 mL) was added (*D*)-Proline (20 mg, 0.18 mmol) and the mixture was stirred at rt for 24 h. Water was added to the mixture and the aqueous layer was extracted with ethyl acetate (3 x 8 mL). The combined organic layers were dried (Na₂SO₄), and the solvent removed *in vacuo* to give the crude product which was purified by column chromatography (15% ethyl acetate/hexane v/v) to give title

compound (51 mg, 42% yield) as a yellow solid. m.p. 98.2–99.8 °C; ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 7.91-7.90 (m, 1H), 7.57-7.53 (m, 1H), 7.34-7.28 (m, 2H), 6.13-5.68 (m, 8H), 4.97 (s, 1H), 3.49 (d, *J* = 5.5 Hz, 1H), 2.46 (dd, *J* = 14.5, 1.5 Hz, 1H), 2.03-1.99 (m, 1H), 1.69 (m, 1H); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 162.0, 159.5, 153.0, 143.4, 133.9, 132.9, 132.7, 132.0, 131.8, 131.3, 131.2, 128.4, 123.9, 123.2, 116.6, 115.4, 100.8, 99.6, 34.7, 34.0, 28.6; HRMS-ESI calcd for C₂₁H₁₈O₄Na⁺ ([M+Na]⁺): 357.1097; found: 357.1094.

Cyclooctatetraenyl warfarin (8) (alternative procedure)



Cubyl warfarin (**7**) (200 mg, 0.60 mmol) and $[Rh(nbd)Cl]_2$ (27 mg, 0.06 mmol) were suspended in toluene (15 mL) and stirred at 60 °C for 48 h. The solvent was removed and the residue was purified by column chromatography (15% ethyl acetate/hexane v/v) to give the title compound (100 mg, 50%).

Enantiomers 8a and 8b



Racemic cyclooctatetraenyl warfarin (**8**) (150 mg) was separated by chiral preparative HPLC to provide two enantiomers **8a** (70 mg) and **8b** (70 mg). **8a**: $[\alpha]^{24}$ 289.3 (c 0.06, CHCl₃); **8b**: $[\alpha]^{24}$ -307.7 (c 0.08, CHCl₃).

File name :20161012_103204.ccf File folder :C:\Program Files\PDR\Alice\Hui X Run started :10/12/2016 10:12:04 AM	ing_HX-5-37\
10000 €3 mV 900.0 800.0 500.0 500.0 500.0 300.0 200.0 200.0 10	3
Operation Mode :Standard mode Method name :30IPA_HEX70 Col Name :AD Column Size CSP Density :0.6000 Packed Length :25.0000 Inner Diameter :2.1000 Inj Vol μL :1500	UV Wavelengths (nm) :254,-,-,-,- Flow mL/min :10.00 Press bar :150 Run Time :20.00 min Meth. Vol mL :200 Sample ID :HX-5-37 Data Path :C:\Program Files\PDR\&lice\ Hui Xing_HX-5-37
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Total Peak Area :58426.197 mVs	
<u>C H R O M A T O</u>	GRAM REPORT
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193.14 mV 173.1 153.1 133.1 113.1 93.1 73.1	
1.9 3.9 5.9 7.9 9.9	9 11.9 13.9 15.9 17.9 199
Operation Mode :Standard mode Method name :301PA_HEX70 Col Name :AD Column Size CSP Density :0.6000 Packed Length :25.0000 Inner Diameter :2.1000 Inj Vol μL :1000	UV Wavelengths (nm) :254,-,-,-,- Flow mL/min :10.00 Press bar :150 Run Time :20.00 min Meth. Vol mL :300 Sample ID :HX-5-37 1 Data Path :C:\Program Files\PDR\Alice\ Hui Xing_HX-5-37
Eluent and Gradient Time Solvent A Solvent B (min) (\$) (\$) 	
PEAK LIST REPORT	
No Name Type Start End RT Height (min) (min) (min) (mV) 	Area % Area Col. Area Plates Conc (mVs) (%) (mVs) (#) (mg/mL)
1 HX-5-37_1 (+) 8.01 8.67 8.25 116.166	2054.890 100.00 2054.873 4807.575 0.0000
LOCAL LOAN MICH .2001.090 MVS	



0.00 100.0

PEAK LIST REPORT

No Name	Type	Start (min)	End (min)	RT (min)	Height (mV)	Area (mVs)	\$ Area (\$)	Col. Area (mVs)	Plates (#)	Conc (mg/mL)
1 HX-5-37_2	(+)	13.23	14.52	13.57	19.917	755.578	100.00	755.574	2800.494	0.0000

Total Peak Area :755.578 mVs



Determination of the absolute stereochemistry of 8a: PTAD adducts S13 and S14

(*R*)-COT warfarin (**8a**) (15 mg, 0.05 mmol) and PTAD (**S12**) (8 mg, 0.05 mmol) were stirred in ethyl acetate (3 mL) at 60 °C for 1 h. The solvent was removed *in vacuo* then purification by column chromatography (20% ethyl acetate/hexane v/v) gave the title compounds (20 mg, 87% yield) as a slight yellow solid. X-ray quality crystals of **S13** (10 mg) were grown from MeOH and MeCN. Isolation of a pure sample of **S14** was not possible due to degradation during crystallization. **S13** and **S14** existed in equilibrium in solution.



It was not possible to obtain a pure sample of **S14** due to degradation during recrystallisation. Full characterisation of a pure sample of either epimer was not possible due to the rapid equilibrium that exists between **S13** and **S14**.

S13: m.p 154 °C (dec.) ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 8.06-7.97 (m, 1H), 7.68 (t, *J* = 7.5 Hz, 1H), 7.46-7.26 (m, 7H), 5.84-5.22 (m, 7H), 3.77-3.61 (m, 1H), 3.44 (s, 1H), 2.66 (br. s, 1H), 2.10 (s, 3H), 1.92 (dd, *J* = 5.0, 15.0 Hz, 1H).

S13 and **S14** mixture: ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 8.06-7.97 (m, 1.7H), 7.71-7.66 (m, 1.8H), 7.47-7.26 (m, 12H), 5.84-5.23 (m, 12.7H), 4.86-4.73 (br m, 1H), 4.23 (br. s, 1H,), 3.74-3.58 (m, 2H, OH and CH, **S13**), 3.31-3.19 (m, 0.84H, CH, **S14**), 2.89 (t, J = 15.0 Hz, 0.85H, CH₂, **S14**), 2.66 (br. s, 1H, CH₂, **S13**), 2.10 (s, 3H, CH₃, **S13**), 2.02-1.94 (m, 0.83H, CH₂, **S14**), 2.00 (s, 2.5H, CH₃, **S14**), 1.92 (dd, J = 5.0, 15.0 Hz, 1H, CH₂, **S13**); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 190.1, 187.8, 164.9, 164.5, 164.1, 155.0, 154.7, 151.5, 151.1, 150.9, 149.7, 137.6, 137.3, 135.9, 135.5, 135.2, 134.4, 134.2, 134.0, 132.4, 132.0, 131.6, 131.4, 130.5, 130.4, 130.2, 129.1, 128.3, 127.5, 127.1, 125.4, 125.4, 125.2, 120.1, 119.6, 118.3, 118.2, 87.5, 84.5, 70.2, 69.5, 50.5, 48.4, 47.7, 45.5, 39.1, 38.8, 37.6, 27.0, 26.2; HRMS-ESI calcd for C₂₉H₂₃N₃O₆Na⁺ ([M+Na]⁺): 532.1479; found: 532.1456.

Warfarin (6)



Purchased from Sigma-Aldrich.

Warfarin enantiomers 6a and 6b



Racemic warfarin **6** (200 mg) was separated by chiral preparative HPLC to provide two enantiomers **6a** (90 mg) and **6b** (90 mg). **6a**: $[\alpha]^{24}$ _D 17.70 (c 0.14, CHCl₂); **6b**: $[\alpha]^{24}$ _D -19.46 (c 0.15, CHCl₂).

CHROMATOGRAM REPORT File name :20160804_171116.ccf File folder :C:\Program Files\PDR\Alice\Hui Xing_warfarin\ Run started :8/4/2016 4:51:15 PM File name File folder -- ALP 900.0 700.0 600.0 500.0 400.0 300.0 φρ V:1-3 1-1 UV Wavelengths (nm) :254, Flow mL/min :10.00 Press bar :150 :Standard mode :premix_30IPA_70Hex_1 :AD Operation Mode Method name Run Time :20.00 min Meth. Vol mL :200 Sample ID :HX Warfarin Data Path :C:\Program Files\PDR\Alice\ Col Name Column Size Cor Density :0.6000 Packed Length :25.0000 Inner Diameter :2.1000 Inj Vol µL :1000 Hui Xing_warfarin Eluent and Gradient Time Solvent A Solvent B (min) (%) (%) (min) 0.00 50.0 50.0 PEAK LIST REPORT Type Start No Name End RT Height (mV) Area % Area Col. Area
(%) (mVs) Plates Conc (mg/mL) (min) (min) (min) (mVs) (%) (#) 1 HX WF 1 (+) 7.75 8.71 7.95 896.504 15702.359 2 HX_WF_2 (+) 16.73 18.15 17.28 458.640 14610.948 51.80 15702.338 5310.566 48.20 14610.933 1.856 0.0000 Total Peak Area :30313.308 mVs

S23

CHROMATOGRAM REPORT





Scheme S3: synthesis of moclobemide (9) and cubane (14) and COT (21) analogues.

Methyl 4-chlorocubane-1-carboxylate (S15)



To a solution of 4-methoxycarbonylcubane-1-carboxylic acid (S2) (2.964 g, 14.37 mmol) in anhydrous DCM (120 mL) was added oxalyl chloride (1.73 mL, 20.12 mmol) and anhydrous DMF (5 drops) under an argon atmosphere. After 1 h, the mixture was concentrated in vacuo and further dried under high vacuum (1 h). Separately, freshly ground 2-mercaptopyridine N-oxide sodium salt (S3) (3.237 g, 21.70 mmol) and DMAP (17 mg, 0.136 mmol) were suspended in carbon tetrachloride (20 mL) and heated to reflux under an argon atmosphere whilst under irradiation from a 500-W tungsten lamp. The newly formed acid chloride was suspended in carbon tetrachloride (20 mL) and added slowly to the refluxing mixture under an argon atmosphere. After reflux (2 h) the carbon tetrachloride was removed by distillation at atmospheric pressure. The resulting residue was suspended in diethyl ether (100 mL) then washed with water (2 x 40 mL), brine (2 x 40 mL), dried over magnesium sulfate, and concentrated in vacuo to give an orange oil. Purification by column chromatography (5% ethyl acetate/petroleum ether v/v) gave the title compound (2.104 g, 74%) as a white solid.^{[47] 1}H-NMR (300 MHz, CDCl₃): δ (ppm) 4.26–4.12 (m, 3H), 4.20–4.15 (m, 3H), 3.71 (s, 3H).

4-Chlorocubane-1-carboxylic acid (S16)



Methyl 4-chlorocubane-1-carboxylate (**S16**) (2.091 g, 10.63 mmol) was suspended in THF (50 mL). A solution of sodium hydroxide (444 mg, 11.10 mmol) in methanol (5 mL) was added dropwise and the solution was left to stir for 17 h.

The THF was removed *in vacuo* and the residual solid was suspended in water (50 mL), and washed with chloroform (3 x 25 mL). The aqueous phase was acidified to pH 2 with hydrochloric acid (10 M) and washed again with chloroform (3 x 25 mL). The combined organic phases were dried over magnesium sulfate and concentrated to give the title compound (1.446 g, 75%) as a white solid.^{[48] 1}H-NMR (300 MHz, CDCl₃): δ (ppm) 4.31–4.26 (m, 3H), 4.23–4.18 (m, 3H).



To a solution of 4-chlorocubane-1-carboxylic acid (**S16**) (262 mg, 1.43 mmol) in anhydrous DCM (25 mL) was added oxalyl chloride (0.15 mL, 1.72 mmol) and anhydrous DMF (2 drops) under an argon atmosphere. After 1 h, the mixture was concentrated in vacuo and further dried under high vacuum (1 h). The newly formed acid chloride was taken up in anhydrous DCM (25 mL) and cooled to 0 °C under an argon atmosphere. 4-(2-Aminoethyl)morpholine (S17) (0.21 mL, 1.72 mmol) was slowly added followed by anhydrous triethylamine (1.00 mL, 7.15 mmol). The solution was allowed to warm to rt, and then left to stir for 17 h before being washed with water (2 x 25 mL) then brine (25 mL). The remaining organic phase was dried over magnesium sulfate, concentrated, and then purified by column chromatography (5% methanol/DCM v/v) to give the title compound (384 mg, 91%) as a white solid. m.p 174.2–175.8 °C; ¹H-NMR (400 MHz, C₆D₆): δ (ppm) 5.53 (br. s, 1H), 3.82-3.73 (m, 6H), 3.46 (t, J = 4.5 Hz, 4H), 3.18 (q, J = 6.0 Hz, 2H), 1.98–1.93 (m, 6H); ¹³C-NMR (100 MHz, C₆D₆): δ (ppm) 170.1, 72.5, 67.0, 58.7, 57.3, 53.9, 53.5, 46.1, 35.5; HRMS-ESI calcd for C₁₅H₂₀N₂O₂³⁷Cl⁺ ([M+H]⁺): 297.1178; found: 297.1193.

COT analogue of moclobemide (21)



The cubane analogue of moclobemide (**14**) (12 mg, 0.041 mmol) and $[Rh(nbd)Cl]_2$ (2 mg, 0.004 mmol) were suspended in toluene (5 mL) and heated to reflux for 7 h. The solvent was removed and the residue was purified by column chromatography (5% methanol/DCM v/v) to give the title compound (9 mg, 75%) as an orange oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 6.95–6.79 (m, 1H), 6.36–6.26 (m, 1H), 6.17–6.14 (m, 1H), 6.06–6.04 (m, 2H), 6.01–5.95 (m, 1H), 5.83–5.79 (m, 1H), 3.71–3.67 (m, 4H), 3.44–3.36 (m, 2H), 2.54–2.42 (m, 6H); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 165.3, 165.3, 138.0, 136.5, 136.2, 135.9, 134.5, 134.2, 132.6, 132.4, 131.8, 131.2, 131.1, 131.0, 129.8, 129.7, 129.3, 128.3, 67.2, 56.7, 56.5, 53.4, 53.3, 36.2; HRMS-ESI calcd for C₁₅H₂₀N₂O₂³⁵Cl⁺ ([M+H]⁺): 295.1208; found: 295.1199.

Moclobemide (9)



Adapted from the procedure of More *et al*:^[49] to a solution of 4-chlorobenzoic acid (**S18**) (307 mg, 1.96 mmol) in anhydrous DCM (60 mL) was added oxalyl chloride (0.25 mL, 2.87 mmol) and anhydrous DMF (5 drops) under an argon atmosphere. The solution was heated to reflux for 1 h. After cooling, the DCM was removed *in vacuo* and the residual brown oil was further dried under high vacuum (1 h). The newly formed acid chloride was taken up in anhydrous DCM (25 mL) and cooled to 0 °C under an argon atmosphere. 4-(2-Aminoethyl)morpholine (**S17**) (0.33 mL, 2.30 mmol) and anhydrous triethylamine (0.81 mL, 5.76 mmol) were slowly added. The solution was allowed to warm to rt, and then left to stir for 17 h.

After extensive concentration *in vacuo*, the residual material was taken up in DCM (30 mL) then washed with water (3 x 30 mL). The remaining organic phase dried over magnesium sulfate, concentrated and crystalised from a minimal volume of boiling ethanol to give the title compound (217 mg, 41%) as a white crystaline solid. Data reported are consistent with Allen *et al.*^[50] ¹H-NMR (400 MHz, Acetone-*d*₆): δ (ppm) 7.89–7.85 (m, 2H), 7.69 (br. s, 1H), 7.50–7.47 (m, 2H), 3.60 (t, *J* = 4.5 Hz, 4H), 3.51 (q, *J* = 6.6 Hz, 2H), 2.55 (t, *J* = 6.6 Hz, 2H), 2.45 (t, *J* = 4.5 Hz, 4H).



Scheme S4: synthesis of pravadoline (10) and cubane (15) and COT (22) analogues.

Methyl 4-iodocubane-1-carboxylate (S19)



Following the procedure of Priefer et al: [51] to a solution of 4methoxycarbonylcubane-1-carboxylic acid (S2) (1.000 g, 4.84 mmol) in anhydrous DCM (50 mL) was added oxalyl chloride (0.50 mL, 5.82 mmol) and anhydrous DMF (1 drop) under an argon atmosphere. After 1 h, the mixture was concentrated in vacuo and further dried under high vacuum (1 h). Separately, freshly ground 2mercaptopyridine N-oxide sodium salt (S3) (952 mg, 6.38 mmol), DMAP (6 mg, 0.049 mmol) and 2,2,2-trifluoroiodoethane (2.39 mL, 24.20 mmol) were suspended in anhydrous DCM (50 mL) and heated to reflux under an argon atmosphere whilst under irradiation from a 500-W tungsten lamp. The newly formed acid chloride was suspended in anhydrous DCM (50 mL) and added slowly to the refluxing mixture. After reflux (3 h) the suspension was washed with water (3 x 50 mL) then dried purification magnesium sulfate. Concentration and column over by chromatography (50% ethyl acetate/petroleum ether v/v) gave the title compound (1.210 g, 87%) as an off-yellow solid. ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 4.40– 4.36 (m, 3H), 4.30–4.26 (m, 3H), 3.70 (s, 3H).

Methyl 4-methoxycubane-1-carboxylate (S20)



Methyl 4-iodocubane-1-carboxylate (**S19**) (1080 mg, 3.75 mmol) was suspended in anhydrous methanol (80 mL) in a quartz tube under an argon atmosphere. The tube was placed in a Rayonet (Srinivasen-Griffin) reactor equipped with 2537 Å lamps and irradiated for 17 h. Sodium bicarbonate (5 drops)

was added and the methanol was removed *in vacuo* Purification by column chromatography (DCM) gave the title compound (417 mg, 58%) as a brown oil.^[52] ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 4.18–4.15 (m, 3H), 4.01–3.97 (m, 3H), 3.70 (s, 3H), 3.32 (s, 3H).

4-Methoxycarbonylcubane-1-carboxylic acid (S21)



Methyl 4-methoxycubane-1-carboxylate (**S20**) (407 mg, 2.12 mmol) was suspended in THF (30 mL). A solution of sodium hydroxide (95 mg, 3.96 mmol) in methanol (5 mL) was added dropwise and the solution was left to stir for 21 h. The THF was removed *in vacuo* and the residual solid was suspended in water (20 mL), and washed with DCM (3 x 20 mL). The aqueous phase was acidified to pH 2 with hydrochloric acid (3 M) and washed again with DCM (3 x 20 mL). The combined organic phases were dried over magnesium sulfate and concentrated to give the title compound (315 mg, 83%) as a white solid.^[48] m.p 148.7–149.9 °C; ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 4.21–4.18 (m, 3H), 4.05–4.01 (m, 3H), 3.33 (s, 3H); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 177.7, 91.5, 56.4, 51.7, 50.8, 42.7; HRMS-ESI calcd for C₁₀H₉O₃⁻ ([M-H]⁻): 177.0557; found: 177.0577.



To a solution of 4-methoxycubane-1-carboxylic acid (**S21**) 28 mg, 0.16 mmol) in anhydrous DCM (5 mL) was added oxalyl chloride (0.02 mL, 0.19 mmol) and anhydrous DMF (1 drop) under an argon atmosphere. After 1 h, the mixture was concentrated *in vacuo* and further dried under high vacuum (1 h). Separately, a

solution of 2-methylindole (**S22**) (22 mg, 0.17 mmol) in anhydrous DCM (5 mL) was added to a solution of methylmagnesium bromide (3 M in diethyl ether, 63 μ L, 0.19 mmol) at 0 °C under an argon atmosphere. The solution was warmed to rt, then the newly formed acid chloride was dissolved in anhydrous DCM (5 mL) and slowly added. The mixture was heated to reflux for 2 h. After cooling, saturated ammonium chloride (10 mL) was added and the phases were separated. The aqueous phase was washed with DCM (3 x 10 mL) and the combined organic phases were dried over magnesium sulfate, concentrated *in vacuo* and purified by column chromatography (30% ethyl acetate/petroleum ether v/v) to give the title compound as an orange solid (35 mg, 75%).^[53] m.p 161.2–163.8 °C; ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 8.54 (br. s, 1H), 7.82–7.80 (m, 1H), 7.33–7.31 (m, 1H), 7.25–7.18 (m, 2H), 4.30–4.23 (m, 6H), 3.40 (s, 3H), 2.71 (s, 3H); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 196.0, 143.6, 134.4, 126.8, 122.5, 122.0, 120.2, 114.3, 110.8, 90.3, 64.5, 51.7, 50.2, 43.4, 15.0; HRMS-ESI calcd for C₁₉H₁₇O₂NNa⁺ ([M+Na]⁺): 314.1151; found: 314.1144.





Sodium hydride (50% dispersion in mineral oil, 40 mg, 0.83 mmol) was suspended in anhydrous DMF (2 mL) under an argon atmosphere. To this was added 3-(4-methoxycubanone)-2-methylindole (**S23**) (20 mg, 0.069 mmol) as a solution in anhydrous DMF (2 mL). The mixture was left to stir for 10 min, then *N*-(2-chloroethyl)-morpholine hydrochloride (**S24**) (21 mg, 0.11 mmol) was slowly added as a solution in anhydrous DMF (2 mL). The mixture was heated to 100 °C. After 17 h the DMF was removed *in vacuo* and the residual material was taken up in water (5 mL) then washed with DCM (3 x 5mL). The combined organic phases were dried over magnesium sulfate, concentrated and purified by column

chromatography (60% ethyl acetate/petroleum ether v/v) to give the title compound (17 mg, 61%) as a yellow solid.^[53] m.p 121.5–124.0 °C; ¹H-NMR (400 MHz, CDCl₃): δ (ppm) 7.79–7.74 (m, 1H), 7.34–7.31 (m, 1H), 7.25–7.21 (m, 2H), 4.29–4.20 (m, 8H), 3.71 (t, J = 4.5 Hz, 4H), 3.39 (s, 3H), 2.73 (s, 3H), 2.67 (t, J = 7.3 Hz, 2H), 2.52 (t, J = 4.5 Hz, 4H); ¹³C-NMR (100 MHz, CDCl₃): δ (ppm) 196.1, 144.3, 135.7, 126.4, 122.2, 121.9, 120.2, 114.3, 109.4, 90.2, 67.0, 64.7, 57.5, 54.2, 51.7, 50.2, 43.6, 41.1, 12.5; HRMS-ESI calcd for C₂₅H₂₉O₃N₂⁺ ([M+H]⁺): 405.2173; found: 405.2176.





The cubane analogue of pravadoline (**15**) (49 mg, 0.12 mmol) and [Rh(nbd)Cl]₂ (6 mg, 0.004 mmol) were suspended in toluene (5 mL) and heated to 60 °C for 4 h. The solvent was removed and the residue was purified by column chromatography (ethyl acetate) to give the title compound (40 mg, 82%) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 8.11–8.05 (m, 1H), 7.31–7.30 (m, 1H), 7.23–7.16 (m, 2H), 6.54–6.48 (m, 2H), 6.17–5.81 (m, 3H), 5.02–4.97 (m, 1H), 4.24 (t, 2H, J = 7.2 Hz), 3.70 (t, 4H, J = 4.6 Hz), 3.66–3.63 (m, 3H), 2.72–2.66 (m, 5H), 2.51 (t, 4H, J = 4.6 Hz); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 191.9, 191.2, 157.5, 156.7, 145.2, 144.0, 143.3, 143.0, 141.2, 140.4, 135.9, 135.9, 134.9, 133.0, 132.3, 131.9, 129.8, 129.1, 128.3, 127.4, 127.4, 122.2, 122.1, 121.6, 121.5, 121.5, 113.6, 113.3, 109.3, 109.2, 99.9, 99.6, 67.0, 57.6, 57.6, 55.6, 55.4, 54.2, 41.2, 12.8, 12.7; HRMS-ESI calcd for C₂₅H₂₉O₃N₂+ ([M+H]⁺): 05.2173; found: 405.2172.

3-(4-Methoxyphenone)-2-methyl indoline (S26)



MeMgBr, DCM, 0 °C \rightarrow Δ , 42%

To a solution of p-anisic acid (S25) (520 mg, 3.42 mmol) in anhydrous DCM (15 mL) was added oxalyl chloride (0.34 mL, 3.95 mmol) and anhydrous DMF (2 drops) under an argon atmosphere. The solution was heated to reflux. After 2 h, the DCM was removed in vacuo and the residual brown solid was further dried under high vacuum (1 h). Separately, a solution of 2-methylindole (S22) (455 mg, 3.42 mmol) in anhydrous DCM (15 mL) was added to a solution of methylmagnesium bromide (3M in diethyl ether, 1.32 mL, 3.95 mmol) at 0 °C under an argon atmosphere. The solution was warmed to rt, then the newly acid chloride was dissolved in anhydrous DCM (15 mL) and slowly added. The mixture was heated to reflux for 2 h. After cooling, saturated ammonium chloride (10 mL) was added and the phases were separated. The aqueous phase was washed with DCM (3 x 10 mL) and the combined organic phases were dried over magnesium sulfate, concentrated *in vacuo* and purified by column chromatography (30% ethyl acetate/petroleum ether) to give the title compound as a yellow solid (380 mg, 42%). Data reported are consistent with Gong et al.[53] ¹H-NMR (300 MHz, DMSO d_6): δ (ppm) 7.64 (d, J = 8.1 Hz, 2H), 7.39–7.32 (m, 2H), 7.13–6.98 (m, 4H), 3.85 (s, 3H), 2.42 (s, 3H).



Sodium hydride (50% dispersion in mineral oil, 99 mg, 2.06 mmol) was suspended in anhydrous DMF (1 mL) under an argon atmosphere. To this was added (4-methoxyphenyl)(2-methyl-1H-indol-3-yl)methanone (**S26**) (104 mg, 0.39 mmol) as a solution in anhydrous DMF (3 mL). The mixture was left to stir for 30 min, then *N*-(2-chloroethyl)-morpholine hydrochloride (**S24**) (100 mg, 0.54 mmol) was slowly added as a solution in anhydrous DMF (3 mL) and the mixture was heated to 100 °C. After 16 h saturated ammonium chloride (10 mL) was added and the mixture was extracted with DCM (2 x 10 mL). The combined organic phases were washed with water (2 x 10 mL), brine (10 mL) then dried over magnesium sulfate and concentrated. Purification by column chromatography (ethyl acetate) gave the title compound (89 mg, 60%) as an orange solid. Data are consistent with Gong *et al.*^[53] ¹H-NMR (500 MHz, Acetone-*d*₆): δ (ppm) 7.73 (d, *J* = 9.0 Hz, 2H), 7.50 (d, *J* = 8.1 Hz, 1H), 7.38 (dt, *J* = 8.1, 0.9 Hz), 7.19–7.16 (m, 1H), 7.06–7.02 (m, 3H), 4.40 (t, *J* = 6.8 Hz, 2H), 3.90 (s, 3H), 3.61–3.59 (m, 4H), 2.75 (t, *J* = 6.8 Hz, 2H), 2.58 (s, 3H), 2.51–2.49 (m, 4H).



Scheme S5: synthesis of cubane (16) and COT (23) analogues of SAHA.

t-Butyl cubanylcarbamate (S27)



Following the procedure of Chalmers *et al*:^[54] a solution of cubanecarboxylic acid (**S5**) (548 mg, 3.70 mmol), anhydrous triethylamine (0.52 mL, 3.73 mmol) and diphenylphosphoryl azide (0.80 mL, 3.72 mmol) in anhydrous tert-butyl alcohol (12 mL) was heated at reflux under an argon atmosphere for 24 h. After cooling to rt, the mixture was concentrated *in vacuo* and purified by column chromatography (40% ethyl acetate / petroleum ether v/v) to give the title compound (453 mg, 56%) as a white solid. ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 5.02 (br. s, 1H), 4.06 (m, 3H), 3.93 (m, 4H), 1.45 (s, 9H).





Adapted from the procedure of Eaton *et al*:^[55] acetyl chloride (1.0 mL, 14 mmol) was added slowly dropwise to a vigorously stirring solution of anhydrous methanol (2 mL) at -30 °C under a nitrogen atmosphere. After 15 minutes of stirring, solid tert-butyl cubanylcarbamate (**S27**) (79 mg, 0.36 mmol) was added and the solution continued to stir for 1 h at -30 °C. After warming to rt, the mixture was concentrated *in vacuo* to give a brown solid, which was then filtered and washed (25% acetone / diethyl ether: acetone v/v) to give the title compound (40 mg, 72%) as a white solid. ¹H-NMR (300 MHz, D₂O): δ (ppm) 4.22 (m, 3H), 4.02 (m, 4H).



Following the procedure of Chalmers *et al*:^[54] dry ethyl hydrogen suberate^[56] (**S29**) (286 mg, 1.41 mmol) was dissolved in anhydrous DCM (5 mL) under an
argon atmosphere at 0 °C. Oxalyl chloride (0.15 mL, 1.70 mmol) and anhydrous *N*,*N*-dimethylformamide (1 drop) were added and the solution was left to stir for 2 h. The DCM was removed *in vacuo* and the residual solid was exposed to high vacuum for 1 h to give ethyl 8-chloro-8-oxooctanoate. The nascent acid chloride was dissolved in anhydrous DCM (26 mL, 55 mM) under an argon atmosphere. Separately, aminocubane hydrochloride (**S28**) (30 mg, 0.2 mmol) was suspended in anhydrous DCM (20 mL) and cooled to -30 °C. A portion of the previously made acid chloride solution (3.69 mL, 0.20 mmol) was added and the solution was stirred for 5 min. Anhydrous triethylamine (0.11 mL, 0.79 mmol) was added dropwise and the solution stirred for 15 min, allowed to warm to rt then concentrated *in vacuo*. The residue was purified by column chromatography (40% ethyl acetate / petroleum ether v/v) to give the title compound (47 mg, 82%) as a white solid. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 6.00 (br. s, 1H), 4.13 (m, 5H), 3.93 (m, 4H), 2.27 (t, *J* = 7.4 Hz, 2H), 2.17 (t, *J* = 7.4 Hz, 2H), 1.61 (m, 4H), 1.32 (m, 4H), 1.24 (t, *J* = 7.2 Hz, 3H).

Cubane analogue of SAHA (16)



Adapted from the procedure of Chalmers *et al*:^[54] methanolic potassium hydroxide (2M, 0.89 mL, 1.79 mmol) was slowly added to methanolic hydroxylamine hydrochloride (1M, 0.89 mL, 0.89 mmol) at 0 °C. The solution was then left to stir for 30 min. Ethyl 8-(cubylamino)-8-oxooctanoate (**S30**) (25 mg, 83 µmol) was dissolved in anhydrous methanol (1.24 mL) and added slowly dropwise under an argon atmosphere. The solution was allowed to stir overnight. The solvent was removed *in vacuo* and water (5 mL) was added followed by glacial acetic acid until pH 6-7 was reached. The mixture was extracted with ethyl acetate (5 x 3 mL) and the combined organic phases were washed with water (3 x 5 mL), concentrated *in vacuo* then crystalised from methanol to give the title compound (11 mg, 46%) as a white solid. ¹H-NMR (500 MHz, DMSO-*d*₆): δ (ppm) 10.32 (s,

1H), 8.65 (s, 1H), 8.41 (s, 1H), 4.01 (m, 3H), 3.90 (m, 1H), 3.85 (m, 3H), 2.05 (t, J = 7.4 Hz, 2H), 1.92 (t, J = 7.4 Hz, 2H), 1.46 (m, 4H), 1.22 (m, 4H).



Ethyl 8-(cubylamino)-8-oxooctanoate (**S30**) (95 mg, 0.31 mmol) and [Rh(nbd)Cl]₂ (14 mg, 30 μmol) were suspended in toluene (5.5 mL) and stirred at 60 °C for 3 h. The solvent was removed and the residue was purified by column chromatography (40% ethyl acetate/petroleum ether v/v) to give the title compound (67 mg, 71%) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 6.38-6.34 (m, 1H), 5.98-5.74 (m, 6H), 4.12 (q, *J* =7.2, 2H), 2.28 (t, *J* = 7.4 Hz, 2H), 2.20 (t, *J* = 7.4 Hz, 2H), 1.68-1.59 (m, 4H), 1.35-1.32 (m, 4H), 1.25 (t, *J* = 7.2 Hz, 3H); ¹³C-NMR (125 MHz, CDCl₃) δ (ppm) 173.7, 171.3, 134.5, 134.3, 133.0, 131.1, 130.9, 130.5, 129.3, 116.4, 60.2, 37.5, 34.2, 28.8, 28.7, 25.2, 24.7, 14.2; HRMS-ESI calcd for C₁₈H₂₅NO₃Na⁺ ([M+Na]⁺): 326.1732, found: 326.1727.

COT analogue of SAHA (23)



Adapted from the procedure of Chalmers *et al*:^[54] methanolic potassium hydroxide (2M, 0.75 mL, 1.45 mmol) was slowly added to methanolic hydroxylamine hydrochloride (1M, 0.75 mL, 0.75 mmol) at 0 °C. The solution was then left to stir for 30 min. Ethyl 8-((cyclooctatetraenyl)amino)-8-oxooctanoate (**S31**) (22 mg, 73 µmol) was dissolved in anhydrous methanol (1.10 mL) and added slowly dropwise under an argon atmosphere. The solution was allowed to stir overnight. The solvent was removed *in vacuo* and water (4 mL) was added followed by glacial acetic acid until pH 6-7 was reached. The mixture was extracted with ethyl acetate (5 x 3 mL) and the combined organic phases were washed with water (3 x 5 mL), concentrated *in vacuo* then crystalised from methanol/ethyl

acetate to give the title compound (16 mg, 74%) as a white solid. m.p. 122.0–124.0 °C (dec.); ¹H-NMR (500 MHz, Methanol- d_4): δ (ppm) 6.19 (br. s, 1H), 5.89-5.76 (m, 6H), 2.21 (t, J = 7.4, 2H), 2.08 (t, J = 7.4, 2H), 1.65-1.58 (m, 4H), 1.36-1.33 (m, 4H); ¹³C-NMR (125 MHz, Methanol- d_4): δ (ppm) 174.8, 172.9, 136.9, 134.3, 133.4, 132.3, 131.8, 131.7, 130.9, 118.1, 37.7, 33.7, 29.8, 29.8, 26.7, 26.6; HRMS-ESI calcd for C₁₆H₂₂N₂O₃Na⁺ ([M+Na]⁺): 313.1528; found: 313.1523.



Purchased from Cayman Chemical.

Benzyl Benzoate



Scheme S6: synthesis of cubyl (17-19) and cyclooctatetraenyl (24-26) analogues of benzyl benzoate.

Cubylmethyl cubanecarboxylate (17)



To a solution of cubane-1-carboxylic acid (**S5**) (148 mg, 1.00 mmol) in anhydrous DCM (10 mL) was added oxalyl chloride (0.10 mL, 1.20 mmol) and anhydrous DMF (1 drop) under an argon atmosphere. After 1 h, the mixture was concentrated *in vacuo* and further dried under high vacuum (1 h). The newly formed acid chloride was taken up in anhydrous DCM (6 mL) and a solution of cubylmethanol (**S6**) (134 mg, 1.00 mmol) and DMAP (134 mg, 1.1 mmol) in anhydrous DCM (6 mL) was slowly added under an argon atmosphere. After stirring for 2 h, saturated sodium bicarbonate (10 mL) was added and the phases were separated. The aqueous phase was washed with DCM (2 x 10 mL) then the combined organic phases were dried over magnesium sulfate, concentrated and purified by column chromatography (15% ethyl acetate/petroleum ether v/v) to give the title compound (225 mg, 85%) as a white solid.^[54] ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 4.24 (s, 2H), 4.21 (m, 3H), 3.99 (m, 5H), 3.88 (m, 6H).

Cyclooctatetraenylmethyl cyclooctatetraenecarboxylate (24)



Cubylmethyl cubanecarboxylate (**17**) (100 mg, 0.38 mmol) and [Rh(nbd)Cl]² (17 mg, 0.04 mmol) were suspended in toluene (5 mL) and heated to reflux for 4 h. The solvent was removed and the residue was purified by column chromatography (10% ethyl acetate/petroleum ether v/v) to give the title compound (80 mg, 80%) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 7.04 (br. s, 1H), 6.03–5.78 (m, 13H), 4.56 (s, 2H); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 165.4, 142.9, 138.6, 134.2, 133.6, 133.3, 133.2, 132.4, 132.2, 132.0, 131.9, 131.6, 131.3, 131.2, 130.3, 129.8, 129.7, 67.5; HRMS-ESI calcd for C₁₈H₁₆O₂Na⁺ ([M+Na]⁺): 287.1043; found: 287.1054.

Cubylmethyl benzoate (18)



To a solution of benzoyl chloride (0.31 mL, 2.68 mmol) in anhydrous DCM (3 mL) was slowly added a solution of cubylmethanol (**S6**) (300 mg, 2.24 mmol) and DMAP (390 mg, 3.19 mmol) in anhydrous DCM (3 mL) under an argon atmosphere. After stirring for 2 h, saturated sodium bicarbonate (10 mL) was added and the phases were separated. The aqueous phase was washed with DCM (2 x 10 mL) then the combined organic phases were dried over magnesium sulfate, concentrated and purified by column chromatography (10% ethyl acetate/petroleum ether v/v) to give the title compound (410 mg, 76%) as a clear oil.^[54] ¹H-NMR (300 MHz, CDCl₃): δ (ppm) 8.06–8.03 (m, 2H), 7.59–7.53 (m, 1H), 7.47–7.41 (m, 2H), 4.48 (s, 2H), 4.08–4.02 (m, 1H), 3.98–3.96 (m, 6H).





Cubylmethyl benzoate (**18**) (126 mg, 0.53 mmol) and [Rh(nbd)Cl]₂ (24 mg, 0.05 mmol) were suspended in toluene (5 mL) and heated to reflux for 4 h. The solvent was removed and the residue was purified by column chromatography (10% ethyl acetate/petroleum ether v/v) to give the title compound (91 mg, 72%) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 8.06–8.04 (m, 2H), 7.58–7.55 (m, 1H), 7.46–7.42 (m, 2H), 5.95–5.79 (m, 7H), 4.75 (s, 2H); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 166.3, 138.7, 133.4, 133.1, 132.3, 132.1, 131.9, 131.4, 131.2, 130.3, 130.1, 129.8, 128.5, 67.6; HRMS-ESI calcd for C₁₆H₁₄O₂Na⁺ ([M+Na]⁺): 261.0886; found: 261.0898.

Benzyl cubanecarboxylate (19)

To a solution of cubane-1-carboxylic acid (**S5**) (206 mg, 1.39 mmol) in anhydrous DCM (10 mL) was added oxalyl chloride (0.15 mL, 1.62 mmol) and anhydrous DMF (1 drop) under an argon atmosphere. After 1 h, the mixture was concentrated *in vacuo* and further dried under high vacuum (1 h). The newly formed acid chloride was taken up in anhydrous DCM (6 mL) and a solution of benzyl alcohol (158 mg, 1.46 mmol) and DMAP (202 mg, 1.65 mmol) in anhydrous DCM (6 mL) was slowly added under an argon atmosphere. After stirring for 2 h, saturated sodium bicarbonate (10 mL) was added and the phases were separated. The aqueous phase was washed with DCM (2 x 10 mL) then the combined organic phases were dried over magnesium sulfate, concentrated and purified by column chromatography (10% ethyl acetate/petroleum ether v/v) to give the title compound (312 mg, 94%) as a clear oil.^{[54] 1}H-NMR (500 MHz, CDCl₃): δ (ppm) 7.40–7.30 (m, 5H), 5.15 (s, 2H), 4.28–4.26 (m, 3H), 4.03–3.97 (m, 4H).

Benzyl cyclooctatetraenecarboxylate (26)



Benzyl cubanecarboxylate (**19**) (97 mg, 0.41 mmol) and [Rh(nbd)Cl]₂ (19 mg, 0.04 mmol) were suspended in toluene (5 mL) and heated to 60 °C 16 h. The solvent was removed and the residue was purified by column chromatography (10% ethyl acetate/petroleum ether v/v) to give the title compound (69 mg, 71%) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 7.37–7.31 (m, 5H), 7.08 (br. s, 1H), 6.07–5.79 (m, 6H), 5.19 (s, 2H); ¹³C-NMR (125 MHz, CDCl₃): δ (ppm) 165.7, 143.0, 136.1, 134.3, 133.6, 133.3, 132.4, 131.6, 130.3, 129.7, 128.7, 128.3, 128.3, 66.7; HRMS-ESI calcd for C₁₆H₁₄O₂Na⁺ ([M+Na]⁺): 261.0886; found: 261.0896.

Benzyl benzoate (12)



Purchased from Sigma-Aldrich.

Diflubenzuron



Scheme S7: synthesis of cubane (20) and COT (27) analogues of diflubenzuron.

Cubane analogue of diflubenzuron (20)



To a solution of 4-chloro-cubane carboxylic acid (**S16**) (1 g, 5.50 mmol) in anhydrous toluene (10 mL), was added triethylamine (0.9 mL, 6.6 mmol) and diphenylphosphorazide (1.50 g, 5.50 mmol) under an argon atmosphere. The mixture was stirred at room temperature for 30 min, and then heated at 90 °C for 2 h to generate the isocyanate. A separate solution of 2,6-difluorobenzamide (**S32**) (0.95 g, 6.05 mmol) in anhydrous toluene (20 mL) under an argon atmosphere was prepared and heated to reflux in toluene. To this solution was added dropwise (10 mins.) the above hot isocyanate solution while maintaining reflux. After addition the reaction mixture was stirred under reflux condition for 12 h. On cooling a precipitate formed, which was filtered and washed with toluene and acetone to afford the title compound as a white solid (924 mg, 50%). Data reported are consistent with Chalmers *et al.*^[54] ¹H NMR (400 MHz, CDCl₃) δ : 9.65 (br. s, 1H), 8.83 (br. s, 1H), 7.53-7.45 (m, 1H), 7.03-6.99 (m, 2H), 4.11-4.07 (m, 3H), 4.01-3.98 (m, 3H).



The cubane analogue of diflubenzuron (**20**) (200 mg, 0.6 mmol) and $[Rh(nbd)Cl]_2$ (27 mg, 0.06 mmol) were suspended in toluene (30 mL) and stirred at 60 °C for 24 h. The solvent was removed and the residue was purified by column chromatography (20% ethyl acetate/petroleum ether v/v) to give the title compound (150 mg, 75%) as a yellow solid. m.p. 146 °C (dec.). ¹H-NMR (500 MHz, CDCl₃): δ (ppm) 9.74-9.68 (m, 1H), 9.27-9.24 (m, 1H), 7.51-7.45 (m, 1H), 7.03-6.99 (m, 2H),

6.23-5.68 (m, 6H); ¹³C-NMR (125 MHz, CDCl₃) δ (ppm) 161.8, 161.03, 161.0, 160.9, 159.0, 159.0, 158.9, 150.6, 134.0, 133.6, 133.5, 133.3, 133.1, 132.9, 132.2, 131.6, 130.7, 130.4, 130.3, 130.0, 129.6, 128.9, 128.0, 117.7, 115.7, 115.4, 112.5, 112.3; HRMS-ESI calcd for C₁₆H₁₁ClF₂N₂O₂Na⁺([M+Na]⁺): 359.0369; found: 359.0362.

Diflubenzuron (13)



13

Purchased from AK Scientific.

logP Measurements

logP measurements pertaining to warfarin (6), moclobemide (9), pravadoline (10), SAHA (11), benzyl benzoate (12), diflubenzuron (13), and analogues (performed by the group led by Prof. Craig Williams at the School of Chemistry and Molecular Biosciences):

A shake-flask procedure was adapted from several sources.^[57] All experiments were conducted at room temperature (*ca* 25 °C). Solvents were saturated before taking measurements: Milli-q water (500 mL) and *n*-octanol (100 mL) were stirred vigorously (24 h) then poured into a separatory funnel and left to partition (3 h). Most of each phase was separated, discarding the phase boundary. The separated phases were used in all subsequent procedures. Measurements of volume were made using a micropipette.

Prior to taking measurements it was useful to estimate the mass expected in the aqueous phase. Estimations can be made using ClogP (calculated using <u>www.molinspiration.com</u>) of the analyte and the following derivation of the standard logP equation:

$$y = \frac{B\Sigma_m}{AP + B}$$

where:

 $A = volume_{octanol}$ $B = volume_{water}$ $y = mass_{water}$ $\Sigma_m = total\ mass$ $P = partition\ coefficient$

A general procedure is as follows:

Stock solution: the desired analyte was accurately weighed and *fully dissolved* in a known volume of *n*-octanol.

Standard solution: a known volume of stock solution was dissolved in a known volume of suitable HPLC solvent.

Aqueous 1: a known volume of water and known volume of stock solution were added to a 10 mL centrifuge tube. Transfer of the stock solution was aided by washing with a known volume of *n*-octanol. The tube was inverted 100 times then centrifuged (20 min). The *n*-octanol and a small volume around the phase boundary were removed. The bottom of the centrifuge tube was pierced with a needle and a known volume of water ($\leq 80\%$ of the total volume) was drained into another 10 mL plastic centrifuge tube. The water was extracted with organic solvent (3 x). The combined organic phases were concentrated then dissolved in a known volume of HPLC solvent.

Aqueous 2-5: the procedure was repeated four more times, varying the phase ratios (e.g. 1:10, 1:20, 1:40 etc.) for each repeat.

Quantification: the standard solution and aqueous 1-5 were each subjected to HPLC. The area under curve (AUC) from the UV (or PDA) trace for each was measured. The concentration of analyte in aqueous 1-5 was calculated using the known concentration of the standard solution.

logP: the analyte concentrations were used to calculate a logP for each repeat. The mean of a minimum of three repeats that lay within a range of ± 0.3 was calculated to give the final logP of the analyte.

An example procedure is as follows:

Warfarin stock solution: Warfarin (21.2 mg, 0.069 mmol) was dissolved in *n*-octanol (4.000 mL) to give a stock solution of 5.3 mg mL⁻¹.

Warfarin standard solution: Warfarin stock solution (94 μ L, 0.500 mg) was suspended in MeOH (906 μ L) to give a standard solution of 0.5 mg mL⁻¹. HPLC AUC: 4612006.

Warfarin aqueous 1: stock solution (450 μ L, 2.385 mg) was washed with *n*-octanol (50 μ L) into water (5.000 mL) in a 10 mL plastic centrifuge tube. The tube was inverted 100 times then subjected to centrifugation for (20 min). The *n*-octanol and a small volume around the phase boundary were removed. The bottom of the tube was pierced with a needle and the water layer (4.000 mL) was transferred to another 10 mL plastic centrifuge tube. The aqueous phase was extracted with ethyl acetate (3 x 2 mL). The combined organic phases were concentrated then dissolved in MeOH (300 μ L) for HPLC analysis. HPLC AUC: 1272563.

Quantification:

$$HPLC_{conc.} = \frac{1272563}{4612006} \times 0.5 \ mg \ mL^{-1} = 0.14 \ mg \ mL^{-1}$$
$$HPLC_{Mass} = 0.14 \ mg \ mL^{-1} \times 0.300 \ mL = 0.041 \ mg$$
$$5.000 \ mL \ H_2O_{Mass} = \frac{0.041 \ mg}{0.8} = 0.051 \ mg$$
$$500 \ \muL \ n - octanol_{Mass} = 2.39 \ mg \ - 0.051 \ mg = 2.34 \ mg$$
$$logP = log \frac{\left[\frac{2.34 \ mg}{0.500 \ mL}\right]}{\left[\frac{0.051 \ mg}{5.000 \ mL}\right]} = 2.66$$

logP: the mean of the remaining measurements (excluding outliers i.e. values greater than one standard deviation from the mean) was calculated as 2.73. The logP of Warfarin has previously been measured as 2.70.^[58]

The derivation of the standard logP equation is as follows:

$logP = log \frac{[octanol]}{[water]}$
$P = \frac{[octanol]}{[water]}$
$P = \frac{\left(\frac{x}{A}\right)}{\left(\frac{y}{B}\right)}$
$P = \frac{x}{A} \times \frac{B}{y}$
$x + y = \Sigma_m$
$x = \Sigma_m - y$
$P = \frac{\Sigma_m - y}{A} \times \frac{B}{y}$
$P = \frac{B(\Sigma_m - y)}{Ay}$
$yAP = B\Sigma_m - yB$
$yAP + yB = B\Sigma_m$
$y(AP+B) = B\Sigma_m$
$y = \frac{B\Sigma_m}{AP + B}$



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Figure S1: summary of logP values. Values in [brackets] denote logP value. 'a' denotes a value measured using the shake-flask procedure. 'b' denotes a value measured using a HPLC calibration curve.^[54]

Computational Methods

Molecular geometry calculations pertaining to cubane (**1**) and cyclooctatetraene (**2**) (performed by Dr Jed Burns at the School of Chemistry and Molecular Biosciences):

Geometry optimisations were initially conducted using the well-known B3LYP functional with the 6-311+G(d,p) basis set.^[59] This level of theory has been used previously to compute reliable structures, especially in the context of NMR prediction of organic compounds.^[60] To ensure the validity of the results, computed structures were reoptimised using the modern M06 and M06-2X functionals (6-311+G(d,p) basis set),^[61] which are parameterised to include medium range interactions that mimic dispersion. All structures obtained were confirmed as minima by frequency calculations (no imaginary frequencies). Calculations were conducted using the Gaussian 09 (rev. A.02) and 16 (rev. A.03) software packages,^[62] chemical figures were generated with CYL view.^[63]

In order to confirm the accuracy of the DFT calculations, computed structures were compared with those determined by X-ray crystallography.

Crystal structures for cubane^[64] yielded distances of 4.70 ("para"), 3.95 ("meta") and 2.76 ("ortho") angstroms (Å), 0.05 - 0.2 shorter than the calculated distances. We observed similar distances in our own crystallographic data for **S10a** (4.706 ("para"), 3.842 ("meta") and 2.710 ("ortho") Å).

No unsubstituted or monosubstituted cyclooctatetraene structures could be found in the Cambridge Crystallographic Data Centre (CCDC) database. Equivalent distances from benzocyclooctatetraene^[65] were found to be 4.98 ("para") and 4.33 ("meta") Å. Our synthesised warfarin COT-analogue **S13** possess "para" and "meta" distances of 4.997 and 4.286 Å. These values are 0.3 – 0.4 Å shorter than the calculated distances.

Discrepancies between the values are likely due to the differences in physical state (calculations were conducted in the gas phase, crystallographic data are solid

state measurements). Nevertheless, the distances are similar and enable meaningful comparisons to be made.



Figure S2: Space filled model comparisons of cubane (1) and cyclooctatetraene (COT) (2). Bond distances are in angstroms [plain text - M06-2X/6-311+G(d,p), *italics* -M06/6-311+G(d,p), <u>underlined</u> -B3LYP/6-311+G(d,p)], angles are in degrees.



Figure S3: A) Energy minimized geometries [M06-2X/6-311+G(d,p)] with selected distances and angles comparing 1,4-disubstituted COT with that of 1,3- and 1,4- disubstituted cubane. Bond distances are in angstroms, angles are in degrees; **B)** Face centered view of 1,4-COT; **C)** Angle between C1-C2-C3; **D)** Dihedral angle of C1 and C4 along the C2-C3 axis. Methyl hydrogen atoms have been omitted for clarity.

Computational data are listed as follows:

Energy (E) Enthalpy (H) Gibbs Free Energy (G)

xyz coordinates

Separate data are listed for B3LYP, M06 and M06-2X calculations. All energies are in Hartrees.

Cubane (1)

B3LYP

E = -309.5325529 H = -309.394348

G = -309.428088

С	-0.78674	-0.97407	-0.53303
С	-0.78522	-0.53335	0.97513
С	-0.78642	0.53409	-0.97375
С	-0.78489	0.97481	0.53441
С	0.78664	0.97406	0.53321
Н	-1.41293	1.75502	0.96208
Н	-1.41569	0.96151	-1.75312
С	0.78511	0.53353	-0.97512
С	0.78631	-0.53426	0.97374
Н	-1.41352	-0.96018	1.75560
С	0.78479	-0.97480	-0.53458
Н	1.41692	1.75311	0.95962
Н	-1.41627	-1.75369	-0.95961

Н	1.41633	-0.96152	1.75254
Н	1.41417	0.96019	-1.75502
Н	1.41359	-1.75444	-0.96210

M06

E = -309.3022645 H = -309.163510 G = -309.197197

С	0.77942	0.59637	-0.92689
С	0.77844	0.92787	0.59613
С	0.78045	-0.92665	-0.59540
С	0.77947	-0.59516	0.92763
С	-0.77934	-0.59624	0.92698
Н	1.40899	-1.07569	1.67684
Н	1.41075	-1.67508	-1.07611
С	-0.77836	-0.92796	-0.59600
С	-0.78037	0.92674	0.59526
Н	1.40715	1.67726	1.07744
С	-0.77940	0.59502	-0.92772
Н	-1.40936	-1.07738	1.67534
Н	1.40890	1.07786	-1.67551
Н	-1.41121	1.67492	1.07563
Н	-1.40760	-1.67710	-1.07695
Н	-1.40945	1.07521	-1.67667

M06-2X

E = -309.4067648 H = -309.266762 G = -309.300406

С	0.65241	-0.68866	0.96537
С	0.47981	0.86415	0.92456
С	-0.85499	-0.86600	0.59249
С	-1.02756	0.68683	0.55169
С	-0.65251	0.68862	-0.96541
Н	-1.85276	1.23890	0.99493
Н	-1.54207	-1.56136	1.06851
С	-0.47978	-0.86425	-0.92457
С	0.85496	0.86610	-0.59252
Н	0.86543	1.55818	1.66743
С	1.02765	-0.68680	-0.55169
Н	-1.17651	1.24195	-1.74098
Н	1.17626	-1.24199	1.74102
Н	1.54202	1.56159	-1.06842
Н	-0.86542	-1.55841	-1.66732
Н	1.85296	-1.23887	-0.99478

COT (2)

B3LYP

E = -309.6690978

H = -309.529376

G = -309.567551

С 0.67035 1.56508 -0.37512 С 0.37541 -0.59575 1.59477 -0.77933 2.51417 0.92923 Н 0.37513 С -1.56509 0.67035 -2.47492 0.89728 0.92869 Н С -1.59479 -0.59576 -0.37540 Η -2.51420 -0.77934 -0.92922

С	-0.67035	-1.56507	-0.37513
Н	-0.89726	-2.47491	-0.92871
С	0.59576	-1.59477	0.37541
Н	0.77931	-2.51418	0.92923
С	1.56509	-0.67035	0.37513
Н	2.47493	-0.89727	0.92869
Н	0.89728	2.47491	-0.92869
С	1.59478	0.59576	-0.37541
Н	2.51419	0.77932	-0.92922

M06

E = -309.4093751 H = -309.269995 G = -309.308114

С	-0.64143	-1.55571	-0.38870
С	0.59955	-1.57198	0.38900
Н	0.77766	-2.48020	0.96575
С	1.55574	-0.64144	0.38870
Н	2.45908	-0.84425	0.96500
С	1.57203	0.59955	-0.38900
Н	2.48025	0.77768	-0.96571
С	0.64144	1.55570	-0.38870
Н	0.84422	2.45906	-0.96499
С	-0.59956	1.57199	0.38901
Н	-0.77764	2.48023	0.96571
С	-1.55574	0.64145	0.38870
Н	-2.45911	0.84424	0.96496
Н	-0.84425	-2.45906	-0.96500
С	-1.57201	-0.59956	-0.38900
Н	-2.48025	-0.77765	-0.96570

M06-2X

E = -309.5219292

H = -309.381326

G = -309.419433

С	0.63394	-1.56035	0.39594
С	-0.60874	-1.56993	-0.39658
Н	-0.78311	-2.47200	-0.97905
С	-1.56035	-0.63394	-0.39594
Н	-2.45958	-0.82330	-0.97810
С	-1.56994	0.60874	0.39658
Н	-2.47201	0.78312	0.97904
С	-0.63395	1.56034	0.39594
Н	-0.82330	2.45958	0.97809
С	0.60874	1.56994	-0.39658
Н	0.78311	2.47201	-0.97903
С	1.56035	0.63395	-0.39594
Н	2.45959	0.82330	-0.97808
Н	0.82330	-2.45958	0.97810
С	1.56994	-0.60874	0.39658
Н	2.47201	-0.78312	0.97903

Cubane, meta-substituted

B3LYP E = -388.1958208 H = -387.999573 G = -388.040253

C -1.10916 1.18010 -0.00011

С	-1.11586	-0.39525	-0.00004
С	0.00007	-0.39027	-1.11148
С	0.00007	1.17936	-1.11120
С	1.10925	1.18013	0.00006
С	-0.00010	1.17942	1.11114
С	-0.00007	-0.39021	1.11154
С	1.11582	-0.39529	0.00009
С	2.34276	-1.26107	0.00001
Н	-0.00005	-1.02385	2.00005
Н	0.00009	-1.02397	-1.99994
Н	2.00201	1.80581	0.00009
С	-2.34279	-1.26103	-0.00001
Н	0.00008	1.80847	-1.99988
Н	-0.00014	1.80859	1.99979
Н	-2.00192	1.80575	-0.00013
Н	2.96083	-1.07196	0.88519
Н	2.07772	-2.32453	-0.00015
Н	2.96082	-1.07168	-0.88511
Н	-2.96091	-1.07165	0.88507
Н	-2.96081	-1.07194	-0.88523
Н	-2.07769	-2.32447	0.00018

M06 E = -387.8965152 H = -387.700524 G = -387.741516

С	-1.10044	1.17063	-0.00002
С	-1.10466	-0.39089	-0.00002
С	0.00005	-0.38778	-1.10345
С	-0.00001	1.17016	-1.10308

С	1.10043	1.17064	-0.00001
С	-0.00002	1.17020	1.10305
С	-0.00002	-0.38774	1.10349
С	1.10467	-0.39088	0.00005
С	2.32348	-1.25042	0.00001
Н	-0.00002	-1.02682	1.98984
Н	0.00012	-1.02690	-1.98977
Н	1.99736	1.79265	-0.00001
С	-2.32348	-1.25044	-0.00001
Н	-0.00003	1.80094	-1.99199
Н	-0.00003	1.80101	1.99194
Н	-1.99734	1.79260	-0.00003
Н	2.94207	-1.06131	0.88587
Н	2.05809	-2.31466	-0.00010
Н	2.94207	-1.06113	-0.88581
Н	-2.94211	-1.06116	0.88578
Н	-2.94203	-1.06133	-0.88589
Н	-2.05807	-2.31468	0.00011

M06-2X

E = -388.0274736

H = -387.828797

G = -387.869546

С	-1.10334	1.17362	-0.00001
С	-1.10648	-0.39131	-0.00001
С	0.00003	-0.38897	-1.10630
С	-0.00002	1.17338	-1.10630
С	1.10333	1.17361	-0.00002
С	-0.00000	1.17341	1.10627
С	-0.00002	-0.38894	1.10633

С	1.10648	-0.39130	0.00004
С	2.33227	-1.25537	0.00001
Н	-0.00002	-1.02626	1.99033
Н	0.00009	-1.02633	-1.99027
Н	1.99783	1.79395	-0.00003
С	-2.33227	-1.25539	-0.00001
Н	-0.00003	1.80260	-1.99275
Н	0.00000	1.80266	1.99270
Н	-1.99782	1.79395	-0.00000
Н	2.94453	-1.06172	0.88577
Н	2.06087	-2.31521	-0.00009
Н	2.94455	-1.06156	-0.88571
Н	-2.94459	-1.06159	0.88569
Н	-2.94451	-1.06174	-0.88579
Н	-2.06086	-2.31522	0.00010

COT, syn-isomer

B3LYP
E = -388.3263027
H = -388.128323
G = -388.173558

С	1.56833	0.95231	-0.33804
С	0.67073	1.99209	0.18365
Н	1.16789	2.89759	0.52900
С	-0.67078	1.99209	0.18364
Н	-1.16794	2.89759	0.52899
С	-1.56837	0.95230	-0.33803
Н	-2.35419	1.31879	-0.99844
С	-1.57935	-0.35512	-0.03291

С	-2.62163	-1.28662	-0.60439
Н	-3.20518	-1.75688	0.19581
Н	-3.31058	-0.76047	-1.26797
Н	-2.15159	-2.09908	-1.16921
С	-0.66847	-0.97671	0.95500
Н	-1.16948	-1.55589	1.73100
С	0.66849	-0.97671	0.95499
Н	1.16950	-1.55590	1.73098
Н	2.35415	1.31883	-0.99844
С	1.57936	-0.35511	-0.03293
С	2.62168	-1.28657	-0.60438
Н	3.20529	-1.75674	0.19583
Н	2.15169	-2.09908	-1.16914
Н	3.31057	-0.76040	-1.26801

M06

E = -388.0013364 H = -387.803677

G = -387.848687

С	1.55039	0.94731	-0.35157
С	0.66796	1.98421	0.18121
Н	1.16854	2.88482	0.53945
С	-0.66819	1.98419	0.18122
Н	-1.16879	2.88479	0.53947
С	-1.55055	0.94723	-0.35155
Н	-2.32732	1.30196	-1.03250
С	-1.56034	-0.35188	-0.03361
С	-2.57520	-1.29421	-0.60278
Н	-3.17148	-1.75330	0.19631
Н	-3.25566	-0.78955	-1.29368

Н	-2.08720	-2.11791	-1.13780
С	-0.66589	-0.95303	0.96939
Н	-1.17353	-1.52148	1.75221
С	0.66599	-0.95303	0.96938
Н	1.17363	-1.52151	1.75218
Н	2.32709	1.30213	-1.03256
С	1.56040	-0.35179	-0.03358
С	2.57538	-1.29399	-0.60277
Н	3.17167	-1.75305	0.19633
Н	2.08749	-2.11771	-1.13785
Н	3.25582	-0.78921	-1.29361

M06-2X

E = -388.1386489

H = -387.938792

G = -387.984015

С	1.54655	0.94332	-0.36796
С	0.66770	1.99192	0.17561
Н	1.17348	2.88365	0.54013
С	-0.66839	1.99187	0.17569
Н	-1.17420	2.88355	0.54024
С	-1.54706	0.94308	-0.36779
Н	-2.30511	1.28951	-1.06890
С	-1.55563	-0.35092	-0.03121
С	-2.55716	-1.31535	-0.61093
Н	-3.15666	-1.76965	0.18428
Н	-3.22863	-0.81995	-1.31292
Н	-2.04625	-2.13085	-1.13079
С	-0.66624	-0.93295	0.99967
Н	-1.17771	-1.47254	1.79593

С	0.66651	-0.93294	0.99961
Н	1.17804	-1.47257	1.79580
Н	2.30445	1.28999	-1.06911
С	1.55570	-0.35065	-0.03125
С	2.55780	-1.31461	-0.61076
Н	3.15752	-1.76841	0.18457
Н	2.04741	-2.13048	-1.13052
Н	3.22906	-0.81893	-1.31276

Cubane, para-substituted

B3LYP

E = -388.1960872

H = -387.999831

G = -388.040503

С	1.36551	-0.00007	0.00009
С	0.45179	-0.77466	-1.02327
С	-0.45184	-1.27350	0.15919
С	0.45165	-0.49889	1.18256
С	-0.45179	0.77470	1.02326
С	0.45186	1.27352	-0.15920
С	-0.45167	0.49891	-1.18254
С	-1.36550	0.00007	-0.00007
С	-2.86673	-0.00001	-0.00002
Н	-0.82010	0.89773	-2.12778
Н	-0.82032	-2.29151	0.28640
Н	-0.82026	1.39392	1.84124
Н	0.82025	-1.39386	-1.84126
Н	0.82006	-0.89771	2.12781
Н	0.82035	2.29151	-0.28642

С	2.86673	-0.00005	0.00002
Н	-3.26265	1.01413	-0.12686
Н	-3.26259	-0.61699	-0.81485
Н	-3.26247	-0.39725	0.94173
Н	3.26265	0.61665	0.81504
Н	3.26261	-1.01424	0.12651
н	3.26247	0.39749	-0.94161

M06

E = -387.8967981 H = -387.700817 G = -387.741842

С	1.35106	-0.00001	-0.00001
С	0.44821	0.84242	-0.95590
С	-0.44827	-0.40665	-1.20747
С	0.44827	-1.24904	-0.25162
С	-0.44821	-0.84246	0.95587
С	0.44824	0.40662	1.20748
С	-0.44825	1.24904	0.25160
С	-1.35106	0.00002	0.00003
С	-2.84214	0.00002	0.00002
Н	-0.82392	2.25352	0.45395
Н	-0.82402	-0.73367	-2.17852
Н	-0.82390	-1.51994	1.72459
Н	0.82391	1.51988	-1.72463
Н	0.82396	-2.25351	-0.45398
Н	0.82397	0.73363	2.17854
С	2.84214	0.00003	0.00000
Н	-3.23843	0.32650	0.96930
Н	-3.23843	0.67618	-0.76738

Н	-3.23844	-1.00264	-0.20187
Н	3.23845	-0.67632	0.76723
Н	3.23844	-0.32620	-0.96936
н	3.23841	1.00265	0.20215

M06-2X

- E = -388.0277336
- H = -387.829088
- G = -387.869785

С	1.35303	0.00007	0.00007
С	0.44959	-1.20461	-0.42575
С	-0.44953	-0.97110	0.83030
С	0.44947	0.23356	1.25617
С	-0.44958	1.20460	0.42580
С	0.44950	0.97111	-0.83027
С	-0.44944	-0.23358	-1.25615
С	-1.35303	-0.00006	-0.00008
С	-2.85245	-0.00000	-0.00005
Н	-0.82415	-0.42041	-2.26084
Н	-0.82436	-1.74779	1.49434
Н	-0.82447	2.16804	0.76635
Н	0.82448	-2.16806	-0.76627
Н	0.82420	0.42038	2.26085
Н	0.82431	1.74783	-1.49431
С	2.85245	0.00001	-0.00001
Н	-3.24087	0.77740	-0.66467
Н	-3.24085	-0.96432	-0.34084
Н	-3.24063	0.18696	1.00557
Н	3.24087	0.96425	0.34100
Н	3.24088	-0.77755	0.66441

COT, anti-isomer

B3LYP

E = -388.3260108

H = -388.128031

G = -388.173222

С	0.70683	-1.36461	0.20267
С	1.64539	-0.43735	-0.04722
С	1.40257	0.80917	-0.80585
Н	2.11220	1.00026	-1.61063
С	0.49404	1.75717	-0.54497
Н	0.51917	2.66185	-1.15106
С	-0.49394	1.75718	0.54494
С	-1.40253	0.80922	0.80580
С	-1.64539	-0.43729	0.04719
С	-3.08311	-0.64813	-0.36742
Н	-3.21981	-1.60882	-0.86775
Н	-3.41485	0.14454	-1.04672
Н	-3.74893	-0.61589	0.50301
Н	-2.11219	1.00038	1.61053
Н	-0.51907	2.66187	1.15102
С	3.08305	-0.64816	0.36758
Н	3.74909	-0.61516	-0.50265
Н	3.21985	-1.60916	0.86729
Н	3.41443	0.14412	1.04751
С	-0.70688	-1.36457	-0.20279
Н	-1.03439	-2.27987	-0.69635
Н	1.03429	-2.27995	0.69620

M06

E = -388.00117

H = -387.803694

G = -387.848806

С	-0.70157	-1.36683	-0.20640
С	-1.62886	-0.43724	0.04972
С	-1.37628	0.78933	0.82206
Н	-2.07486	0.96757	1.64252
С	-0.47813	1.73786	0.55394
Н	-0.48960	2.63829	1.16941
С	0.47808	1.73787	-0.55393
С	1.37627	0.78935	-0.82204
С	1.62886	-0.43722	-0.04971
С	3.05592	-0.61801	0.36884
Н	3.20845	-1.56663	0.89052
Н	3.37581	0.19565	1.03099
Н	3.72444	-0.59153	-0.50136
Н	2.07486	0.96763	-1.64247
Н	0.48956	2.63830	-1.16939
С	-3.05590	-0.61802	-0.36891
Н	-3.72448	-0.59135	0.50123
Н	-3.20844	-1.56672	-0.89044
Н	-3.37567	0.19554	-1.03123
С	0.70159	-1.36681	0.20645
Н	1.03571	-2.27421	0.71485
Н	-1.03566	-2.27425	-0.71479

M06-2X

E = -388.1384734

H = -387.938565

G = -387.983471

С	0.70517	-1.36843	0.21207
С	1.62974	-0.43936	-0.05307
С	1.36757	0.78478	-0.84227
Н	2.04889	0.95125	-1.67547
С	0.47040	1.73317	-0.56685
Н	0.46238	2.62660	-1.18775
С	-0.47034	1.73317	0.56684
С	-1.36753	0.78480	0.84226
С	-1.62975	-0.43932	0.05306
С	-3.06553	-0.60807	-0.37269
Н	-3.21653	-1.55056	-0.89992
Н	-3.37036	0.21311	-1.02740
Н	-3.72849	-0.58644	0.49806
Н	-2.04885	0.95132	1.67545
Н	-0.46233	2.62660	1.18773
С	3.06549	-0.60810	0.37276
Н	3.72849	-0.58655	-0.49796
Н	3.21645	-1.55055	0.90009
Н	3.37031	0.21313	1.02742
С	-0.70520	-1.36841	-0.21213
Н	-1.03133	-2.26668	-0.73509
Н	1.03126	-2.26670	0.73505

Supplementary Information Part 2: Biological Assay Methods and Results

Warfarin Study: Inhibition of VKOR Activity

Inhibition of VKOR activity (performed by the group led by Dr Jack Tie at the University of North Carolina at Chapel Hill):

Evaluation of the half-maximal inhibition concentration (IC_{50}) of warfarin (**6**) and the synthetic compounds (**7** and **8**) to VKOR: FIXgla-PC/HEK293 reporter cells were plated in a 24-well plate in 1 ml of complete growth medium so that the cells were ~70% confluent at the time for drug treatment, as described previously.^[66] Next day, the cell culture medium was replaced with a complete growth medium containing 5 µM vitamin K epoxide (KO) with increasing concentrations of warfarin (**6**) or the synthetic compounds (**7** and **8**). The cells were cultured for forty-eight hours, and the cell culture medium was collected to determine the efficiency of carboxylation of the reporter protein using ELISA. The IC₅₀ of warfarin and the synthetic compounds were determined using GraphPadTM Prism software (version 7.03; GraphPad Software, San Diego, CA, USA).

Evaluation of the resistance of VKOR and its naturally occurring mutants to warfarin (6) and racemic COT warfarin (8): FIXgla-PC/HEK293 reporter cells with their endogenous VKOR/VKORC1L1 genes knocked out were plated in a 24-well plate in 1 ml of complete growth medium so that the cells will be ~70% confluent at the time for transfection. Wild-type VKOR or its naturally occurring mutants was transiently expressed in these reporter cells using transfection reagent Xfect (Takara Bio USA, Inc., Mountain View, CA, USA). Transfected cells were cultured with 5 μM KO and increasing concentration of warfarin (6)/ *racemic COT warfarin* (8). Cell culture medium was collected after a 48-hour incubation and directly used for the determination of reporter protein carboxylation by ELISA. The resistance of VKOR and its naturally occurring mutants to warfarin (6) and *racemic COT warfarin* (8) was expressed as IC₅₀ (Figure S6) and normalized resistance (resistance for wild-type VKOR is normalized as 1, Figure 1, Main Text).
ELISA: A 96-well high binding ELISA microplate was coated with 100 µL/well anti-carboxylated FIXgla mAb (Green Mountain Antibodies, Burlington, VT, USA) overnight at 4°C. The concentration of the coating antibodies was 2 µg/ml in 50 mM carbonate buffer (pH 9.6). After being washed 5 times with TBS-T wash buffer (20 mM Tris-HCl, pH 7.6, 150 mM NaCl, and 0.1% Tween 20), the plate was blocked with 0.2% BSA in TBS-T wash buffer for 2 hours at room temperature. Cell culture medium samples and carboxylated reporter protein standards (0.12-250 ng/mL) containing 5 mM CaCl₂ were added at 100 µL/well and incubated for 2 hours at room temperature. After being washed with TBS-T wash buffer containing 5 mM CaCl₂, horseradish peroxidase conjugated sheep anti-human protein C IgG (100 µL/well at 1:2500 in TBS-T wash buffer with 5 mM CaCl₂) (Affinity Biologicals Inc., Ancaster, Canada) was added to each well and incubated for 45 minutes at room temperature. After the unbound detecting antibody was washed off, 100 µL of ABTS solution was added to each well and the absorbance was determined at 405 nm with a THERMOmax microplate reader (Molecular Devices, Sunnyvale, CA, USA). The concentration of carboxylated FIXgla-PC was determined using the logit (OD)-log (FIXgla-PC) plot.



Figure S4: (Cubyl and COT Warfarin Combined results). Inhibition of VKOR activity by warfarin enantiomers (**6a** and **6b**), cubyl warfarin enantiomers (**7a** and **7b**) and racemic COT warfarin (**8**).



Figure S5: Inhibition of VKOR activity by the COT warfarin enantiomers (8b and 8a).



Figure S6: Comparison of the resistances of racemic warfarin (6) and racemic COT warfarin (8) with the 27 naturally occurring VKOR mutants.

Warfarin Study: Human Plasma Protein Binding

Human plasma protein binding (performed by the group led by Prof. Maree Smith at the Centre for Integrated Preclinical Drug Development):

1. INTRODUCTION

Protein binding was assessed by an ultrafiltration method using Centrifree[®] Ultrafiltration Devices with a 30000 dalton molecular weight cut-off point. Briefly, known concentrations of analytes were added to human plasmas samples and incubated for 20 min at 37 °C. The samples were then subject to ultrafiltration to separate protein bound and unbound drug. The extent of protein binding was then defined as the percentage difference between the total and unbound concentration of the drug.

Item	Description
(<i>R</i>)-warfarin (6a)	Racemic warfarin (6) was purchased from Sigma- Aldrich and resolved using chiral HPLC (see Supplementary Information Part 1 for details). Expiry date: 16 Oct 2021, Purity:100%
(S)-warfarin (6b)	Racemic warfarin (6) was purchased from Sigma- Aldrich and resolved using chiral HPLC (see Supplementary Information Part 1 for details). Expiry date: 16 Oct 2021, Purity: 100%
(<i>R</i>)-COT warfarin (8a)	Expiry date: 16 Oct 2021, Purity: 100%
(S)-COT warfarin (8b)	Expiry date: 16 Oct 2021, Purity: 100%
Human plasma	Pooled blank human (Li-Hep as the anticoagulant), pH adjusted to 7.40 using NaH ₂ PO ₄
Centrifree® Ultrafiltration Devices	Millipore Item no. 4104 – 30 000 NMWL

2. MATERIALS & REAGENTS

Table S1: materials and reagents for warfarin human plasma protein binding study.

3. METHOD

3.1. LC-MS/MS: Analytical method

Analyte	Q1 Mass	Q3 Mass	Dwell time	DP	CE	СХР	
(R)-warfarin (6a)	307.07	160.982	150	-205	-26	-13	
(S)-warfarin (6b)	307.07	160.982	150	-205	-26	-13	
(R)-COT warfarin (8a)	333.157	255.030	150	-30	-22	-17	
(S)-COT warfarin (8b)	333.157	255.030	150	-30	-22	-17	
Parameter Table	R	acemic war	farin (6)	Rac (8)	' warfarin		
CUR	40	0.00		35.0	0		
CAD	Lo	OW		Med	lium		
IS	-4	500		-200	0		
ТЕМ	60)0		450			
GS1	30	0.00		60.0	0		
GS2	40	0.00	60.00				
EP	-1	0.0	-10.				
HPLC							
Autosampler Temperature	10°C						
Column	Symmetry® C18 3.5 µm 2.1 x 100 mm (Waters [®])						
Mistral Temperature	40°C						
(Column oven)							
Mobile phase A	0.1% formi	c acid in wat	ter				
Mobile phase B	0.1% formid	c acid in ace	tonitrile				
Needle wash solvent 1 (S1)	water						
Needle wash solvent 2 (S2)	0.1 % formi	c acid in 50	% methano	ol in water			
Needle wash solvent 3 (S3)	50% aceton	itrile in wate	er				
Gradient	Time (min	n) Pum	p flow	Pump f	raction B	%	
	1:00	0.4		10			
	2:00	0.4		30			
	3:50	0.4		80			
	4:10	0.4		10			
	5:10	0.4	0.4 10				

Table S2: MS/MS Parameters: Q1 – first mass filter; Q3 – mass analyser; Dwell time – amount of time instrument spends at each transition; DP: declustering potential; EP: entrance potential; CE: collision energy, CXP: collision cell exit potential for product ions, GS1: Nebuliser gas pressure (psi), GS2: Heater gas pressure (psi), CUR: Curtain gas pressure, CAD: Collision gas pressure (psi), IS: Spray voltage (V), TEM: Turbo gas temperature (°C).

3.2. Sample Preparation

Analyte stock solutions (40 μ L) were spiked into solutions of each plasma (1960 μ L) as listed below to have a final concentration of 200 ng/mL & 2000 ng/mL.

Matrix	Analytes concentration in the stock solutions spiked into the matrix (µg/mL)	Analytes final concentration (ng/mL)			
Plasma (human)	10	200			
Plasma (human)	100	2000			

 Table S3:
 sample preparation.

Spiked samples were then incubated at 37°C for 20 minutes in a water bath. Aliquots of the samples from each test group (500 μ L) were transferred to the ultrafiltration devices (n=3) and centrifuged for 20 minutes at 1000 xg. After centrifugation, aliquots of the ultrafiltrates (50 μ L) were transferred to separate test tubes for extraction, alongside triplicate aliquots of the samples prior to ultrafiltration from each test group.

3.3. Calibration Curve Preparation

Standard solution aliquots at 100 μ g/mL were diluted as shown below.

Standard ID / Conc.	Preparation Details
St A 100 µg/mL	
St B 75 µg/mL	mix 150μ L of A + 50 μ L 50% methanol in water
St C 50 µg/mL	mix 120μ L of B + 60 μ L 50% methanol in water
St D 25 µg/mL	mix 100μ L of C + 100μ L 50% methanol in water
St E 10 μg/mL	mix 80μ L of A + 120 μ L 50% methanol in water
St F 5 µg/mL	mix 100μ L of E + 100μ L 50% methanol in water
St G 2.5 μg/mL	mix 100μ L of F + 100 μ L 50% methanol in water
St H 1 µg/mL	mix 80μ L of F + 120 μ L 50% methanol in water
St I 0.5 μg/mL	mix 100 μ L of F + 100 μ L 50% methanol in water

Table S4: calibration curve preparation.

Aliquots of the resulting standard solutions (10 μ L) were then spiked into 490 μ L of the human plasma and mixed well.

3.4. Extraction

Aliquots of the spiked standards and samples (50 μ L) (before and after incubation and ultrafiltration) were mixed with pure acetonitrile (200 μ L) and the samples vortexed immediately. All tubes, were then centrifuged at 14000 xg for 5 minutes. Supernatants were diluted 20 times using 10% acetonitrile in water and 5 μ L of diluted solutions were injected into the LC-MS/MS system.

4. **RESULTS**

4.1. Plasma Protein Binding

Plasma protein binding was calculated by comparison of the concentration of each analyte spiked into human plasma, before and after ultrafiltration. The extent of binding was similar for all four analytes in the range of 92.6-99.98%. (see **Table S5–Table S8** below).

Matrix: Human Plasma									
	Concentrations of (R)-warfarin (6a)								
Replicate	200 ng/mL Pre-filter	200 ng/mL Post-filter	200 ng/mL2000 ng/mL2000Post-filterPre-filterPost						
1	200	1.3	20	60	11.8				
2	243	0.94	2020		2020		2020		9.8
3	259	0.11	2210		9.1				
Mean	234	0.78	2097		10.22				
	Calculations: Plasma protein binding								
Plasma protein binding (%): 100 – [Concentration post-filter Concentration pre-filter x 100]									
200 ng/mL $100 - \left[\frac{0.78}{234} \times 100\right]$ 99.67					99.67				
200	00 ng/mL	$100 - \left[\frac{10.22}{2097} \ge 100\right]$		99.51					

 Table S5: (R)-warfarin (6a) binding to human plasma.

Matrix: Human Plasma								
	Concentrations of (S)-warfarin (6b)							
Replicate	200 ng/mL Pre-filter	200 ng/mL Post-filter	2000 ng/mL Post-filter					
1	217	0.12	24	70	11.4			
2	142	0.12	1300		1300		7.24	
3	125	0.06	1560		8.41			
Mean	161.3	0.1	1776.7		9.0			
Calculations: Plasma protein binding								
Plasma protein binding (%): 100 – [Concentration post-filter Concentration pre-filter x 100]								
200 ng/mL $100 - \left[\frac{0.1}{161.3} \times 100\right]$ 99.94					99.94			
20	00 ng/mL	$100 - \left[\frac{9.0}{1776.7} \ge 100\right] \qquad 99.49$			99.49			

 Table S6: (S)-warfarin (6b) binding to human plasma.

Matrix: Human Plasma									
	Concentrations of (R)-COT warfarin (8a)								
Replicate	200 ng/mL Pre-filter	200 ng/mL Post-filter	200 ng/mL2000 ng/mL2000Post-filterPre-filterPost						
1	177	14.1	18	10	29.1				
2	206	13.7	2360		2360		2360		24.5
3	190	14.5	2360		24.6				
Mean	191	14.1	2176.7		26.1				
	Calculations: Plasma protein binding								
F	Plasma protein binding (%): 100 – [Concentration post-filter Concentration pre-filter x 100]								
200 ng/mL $100 - \left[\frac{14.1}{191} \times 100\right]$ 92.6%					92.6%				
2000 ng/mL $100 - [\frac{26.1}{2176.7} \times 100]$ 98.8%				98.8%					

 Table S7: (R)-COT warfarin (8a) binding to human plasma.

Matrix: Human Plasma							
Concentrations of (S)-COT warfarin (8b)							
Replicate	200 ng/mL Pre-filter	200 ng/mL Post-filter	2000 ng/mL Post-filter				
1	237	0.05	23	60	0.48		
2	255	0.22	16	20	0.39		
3	238	0.04	2030		0.41		
Mean	243.3	0.1	2003.3		0.43		
Calculations: Plasma protein binding							
Plasma protein binding (%): 100 – [Concentration post-filter Concentration pre-filter x 100]							
200 ng/mL $100 - \left[\frac{0.1}{243.3} \times 100\right]$ 99.96					99.96		
2000 ng/mL $100 - \left[\frac{0.43}{2003.3} \ge 100\right]$				99.98			

 Table S8: (S)-COT warfarin (8b) binding to human plasma.

Warfarin Study: Liver Microsome Metabolic Stability

Human liver microsome metabolic stability (performed by the group led by Prof. Maree Smith at the Centre for Integrated Preclinical Drug Development):

1. Methods

1.1 Compounds

Item	Description
(<i>R</i>)-warfarin (6a)	Racemic warfarin (6) was purchased from Sigma- Aldrich and resolved using chiral HPLC (see Supplementary Information Part 1 for details). Expiry date: 16 Oct 2021, Purity:100%
(S)-warfarin (6b)	Racemic warfarin (6) was purchased from Sigma- Aldrich and resolved using chiral HPLC (see Supplementary Information Part 1 for details). Expiry date: 16 Oct 2021, Purity: 100%
(<i>R</i>)-COT warfarin (8a)	Expiry date: 16 Oct 2021, Purity: 100%
(S)-COT warfarin (8b)	Expiry date: 16 Oct 2021, Purity: 100%
Human liver microsomes	Gibco, 20 mg/mL, Catalogue # HMMCPL, Lot # PL050 B
β-Nicotinamide adenine dinucleotide 2'- phosphate reduced tetrasodium salt hydrate (NADPH)	Sigma, Catalogue # N-1630-250 mg, Lot # SLBH5704V

 Table S9: compounds tested.

1.2 Experimental Plan

The reaction conditions for the incubations with substrate and liver microsomes are summarised in **Table S10** and detailed further in sections 1.2.1 to 1.2.4 below.

1.2.1 Metabolic stability of warfarin enantiomers (6a and 6b) and COT warfarin enantiomers (15a and 15b) in human liver microsomes

Aliquots (10 µL) of a stock concentration of 250 µM of each of the warfarin enantiomers (6a and 6b) and the COT warfarin enantiomers (8a and 8b) dissolved in 50% methanol in water were added to the reaction tubes. The reaction mixture (final volume of 250 µL) comprised the following: 0.1 M phosphate buffer (pH 7.4), β-Nicotinamide adenine dinucleotide 2'-phosphate (NADPH) (1mM), and pooled human liver microsomes (0.5 mg/mL). The final concentration of warfarin enantiomers (6a and 6b) and COT warfarin enantiomers (8a and 8b) in the incubation medium was 10 µM. Metabolism was started by adding NADPH after a 5 min pre-incubation period to each of the reaction tubes and incubated at 37°C in a shaking water bath prior to stopping the reaction with 500 µL aliquots of ice cold acetonitrile added to samples collected at 0, 15, 30, 45 and 60 after metabolism initiation. Two negative controls (no NADPH, and no microsomes) were used in parallel with the study samples. A positive control; midazolam, 10 µM was incubated under the same conditions for 1h. Samples were then vortex-mixed and centrifuged for 5 min at 14,000 rpm. The supernatants were diluted 10 times with 10% acetonitrile in water. Then, aliquots of 50 µL of the diluted samples were mixed with 50 µL of internal standard (mefenamic acid 10 ng/mL in 10% acetonitrile in water) and 400 µL of 10% acetonitrile in water. Samples were transferred to a 96-well plate from which 5 μ L aliquots were injected to the HPLC-MS/MS system.

	Metabolic stability incubation condition
Substrate concentrations	10 µM
Incubation volume	250 μL
Incubation medium	Phosphate buffer 100 mM pH 7.4
Incubation total time	1 h
Liver microsome protein concentrations	0.5 mg/mL
Cofactor concentrations	1 mM (NADPH)
Stop reaction solvent	500 µL acetonitrile

 Table S10:
 summary of human liver microsome incubation conditions.

1.2.2 Bioanalysis: Calibration Curves for warfarin enantiomers (6a and 6b) and COT warfarin enantiomers (8a and 8b)

A mix of stock solutions of warfarin enantiomers (**6a** and **6b**)/COT warfarin enantiomers (**8a** and **8b**) (100 µg/mL) and midazolam (100 µg/mL) in 50% methanol in water were diluted to 75, 50, 25, 12.5, 0.25 µg/mL using 50% methanol in water, and aliquots of the working standard solutions (10 µL) were added to plastic tubes. Next, aliquots (190 µL) of phosphate buffer (100 mM, pH 7.4) and microsome solutions in phosphate buffer (100 mM, pH 7.4) (50 µL) were added to the tubes, followed by 500 µL aliquots of acetonitrile. Tubes were vortex-mixed and centrifuged for 5 min at 14,000 rpm. Aliquots of supernatant (50 µL) were diluted 10 times with 10% acetonitrile in water. Then, aliquots of 50 µL of the diluted standards were mixed with 50 µL of internal standard (mefenamic acid 10 ng/mL in 10% acetonitrile in water) and 400 µL of 10% acetonitrile in water. Samples were transferred to a 96-well plate from which 5 µL aliquots were injected to the HPLC-MS/MS system.

1.2.4 Mass spectrometry conditions

The LC-MS/MS parameters used for the analysis of warfarin enantiomers (**6a** and **6b**), COT warfarin enantiomers (**8a** and **8b**) and midazolam are summarized in **Table S11**.

HPLC									
Autosampler Tempera	ture	10°C							
Column			Symmetry® C18 3.5 µm 2.1 x 100 mm						
Mistral Temperature			40°C						
(Column oven)									
Mobile phase A		0.1% Formic acid in water							
Mobile phase B		0.1	l% Formic a	acid in acet	onitrile				
Needle wash solvent 1		W	ater						
Needle wash solvent 2		0.1	1% formic a	cid in 50%	methar	nol ir	n water		
		50	% acetonitri	le in water	•				
Gradient		Т	ime (min)	Pump flo)W	Pun	np fraction	B%	
		0.	.00	0.4		10			
		1:02		0.4	0.4		10		
		2:30		0.4		95			
		3:20		0.4		95	10		
		5:50		0.4		10			
		5.	.10	0.4		10			
MS/MS									
Analyte	Q1 Ma	ass	Q3	Dwell	DP		CE	CXP	
			Mass	time					
Warfarin (6)	307.0	70	160.980	150	-205	.0	-26.0	-13.0	
COT warfarin (8)	333.1	57	255.030	150	-30.0		-22.0	-17.0	
Midazolam 325.80		00	0 291.100 150		136.0		39.0	22.0	
1-hydroxymidazolam	342.300		203.046	150	111.	0	37.0	20.0	
4-hydroxymidazolam	342.0	00	325.000	150	106.	0	31.0	28.0	
Mefenamic acid	240.0	53	196.085	150	-135	.0	-24.0	-11.0	
able S11: mass spectrometry conditions for the analytes of interest									

warfarin (6) / COT warfarin (8)	midazolam
40.0	40.0
LOW	LOW
-4500	4500
600	600
30.0	30.0
40.0	40.0
-10.0	10.0
	warfarin (6) / COT warfarin (8) 40.0 LOW -4500 600 30.0 40.0 -10.0

MS/MS Parameters: Q1 – first mass filter; Q3 – mass analyser; Dwell time – amount of time instrument spends at each transition; DP- declustering potential; EP – entrance potential; CE – collision energy; CXP: collision cell exit potential for product ions, GS1: Nebuliser gas pressure (psi), GS2: Heater gas pressure (psi), CUR: Curtain gas pressure, CAD: Collision gas pressure (psi), IS: Spray voltage (V), TEM: Turbo gas temperature (°C).

2. Results:

2.1 (S)-warfarin (6b)

Time	(<i>S</i>)-warfarin (6b) (µM)			Average
(min)				
0	8.5	10.7	11.5	10.2
15	9.0	10.7	11.0	10.2
30	8.9	10.8	12.7	10.8
45	9.3	8.2	8.5	8.7
60	8.5	8.6		8.6

Table S12: concentration of (*S*)-warfarin (**6b**) in incubation medium (µM).

	(<i>S</i>)-warfarin (6b) (μM)		
Negative Control			
(No NADPH)	10.9	13.0	8.5
Negative Control			
(No Microsome)	12.0	10.8	10.0

 Table S13: concentration of (S)-warfarin (6b) in negative controls.

Positive Control	1.2 x 10 ⁵	1.4 x 10 ⁵	1.4 x 10 ⁵
(The area of formed			
metabolite; 1-			
hydroxymidazolam)			
Positive Control	$1.2 \ge 10^5$	$1.2 \ge 10^5$	$1.2 \ge 10^5$
(The area of formed			
metabolite; 4-			
hydroxymidazolam)			

Table S14: formation of midazolam metabolites after 1 hour incubation (positive control).

Time	LN (<i>S</i>)-warfarin (6b) (µM)		
(min)			
0	2.1	2.4	2.4
15	2.2	2.4	2.4
30	2.2	2.4	2.5
45	2.2	2.1	2.1
60	2.1	2.2	

Table S15: concentration of (S)-warfarin (6b) in incubation medium (Logarithmic scale).

(S)-warfarin (6b) metabolic stability



T1/2 = -0.693 / -0.0033 = 210 min ± 120

Figure S7: (S)-warfarin (6b) metabolic stability.

2.2 (*R*)-warfarin (6a)

Time (min)	(R)-warfai	(R)-warfarin (6a) (uM)		
0	11.4	11.0	10.9	11.1
15	11.2	12.1		11.7
30	10.7	14.2	13.4	12.8
45	10.3	9.1	10.3	9.9
60	12.5	12.2	9.9	11.5

Table S16: concentration of (*R*)-warfarin (**6a**) in incubation medium (μ M).

	(R)-warfarin (6a) (μ M)			
Negative Control				
(No NADPH)	7.8	10.4	11.7	
Negative Control				
(No Microsome)	10.5	8.6	11.4	

Table S17: concentration of (*R*)-warfarin (6a) in negative controls.

Positive Control	2.6 x 10 ⁵	2.9 x 10 ⁵	
(The area of formed			
metabolite; 1-			
hydroxymidazolam)			
Positive Control	7.7×10^3	$1.2 \text{ x } 10^4$	
(The area of formed			
metabolite; 4-			
hydroxymidazolam)			

Table S18: formation of midazolam metabolites after 1 hour incubation (positive control).

Time	LN (<i>R</i>)-warfarin (6a) (µM)		
(min)			
0	2.4	2.4	2.4
15	2.4	2.5	
30	2.4	2.7	2.6
45	2.3	2.2	2.3
60	2.5	2.5	2.3

Table S19: concentration of (*R*)-warfarin (6a) in incubation medium (Logarithmic scale).



(R)-warfarin (6a) metabolic stability

Figure S8: (R)-warfarin (6a) metabolic stability.

2.3 (R)-COT warfarin (8a)

Time	(<i>R</i>)-COT warfarin (8a) (µM)			Average
(min)				
0	8.6	8.8	8.8	8.7
15	7.1	7.8	8.1	7.7
30	8.3	7.8	6.9	7.7
45	6.2	7.7	7.2	7.0
60	6.3	5.5	6.7	6.2

Table S20: concentration of (*R*)-COT warfarin (8a) in incubation medium (µM).

	(<i>R</i>)-COT warfarin (8a)			
Negative Control				
(No NADPH)	9.2	8.9	8.7	
Negative Control				
(No Microsome)	8.0	9.6	8.0	

 Table S21: concentration of (R)-COT warfarin (8a) in negative controls.

Positive Control	7.7 x 10 ⁵	6.2 x 10 ⁵	7.0 x 10 ⁵
(The area of formed			
metabolite; 1-			
hydroxymidazolam)			
Positive Control	2.1×10^5	$1.8 \ge 10^5$	$2.0 \ge 10^5$
(The area of formed			
metabolite; 4-			
hydroxymidazolam)			

 Table S22: formation of midazolam metabolites after 1 hour incubation (positive control).

Time	LN (R)-COT warfarin (8a) (μ M)		
(min)			
0	2.2	2.2	2.2
15	2.0	2.1	2.1
30	2.1	2.1	1.9
45	1.8	2.0	2.0
60	1.8	1.7	1.9

Table S23: concentration of (R)-COT warfarin (8a) in incubation medium (Logarithmic scale).



Figure S9: (*R*)-COT warfarin (8a) metabolic stability.

2.4	(S)-	СОТ	warfarin	(8b)
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Time	(<i>S</i>)-COT warfarin (8b) (μM)			Average
(min)				
0	10.2	10.0	10.3	10.2
15	9.6	10.1	10.1	9.9
30	8.0	7.6	7.0	7.5
45	7.4	7.2		7.3
60	7.6	7.6	7.2	7.5

Table S24: concentration of (*S*)-COT warfarin (**8b**) in incubation medium (µM).

	(S)-COT warfarin (8b) (μM)		
Negative Control (No NADPH)	12.0	12.1	11.0
Negative Control (No Microsome)	11.0	11.2	10.0

Table S25: concentration of (S)-COT warfarin (8b) in negative controls.

Positive Control	2.8×10^5	$2.0 \ge 10^5$	2.6×10^5
(The area of formed			
metabolite; 1-			
hydroxymidazolam)			
Positive Control	$2.2 \text{ x } 10^4$	$1.4 \text{ x } 10^4$	2.5×10^4
(The area of formed			
metabolite; 4-			
hydroxymidazolam)			

Table S26: formation of midazolam metabolites after 1 hour incubation (positive control).

0	2.3	2.3	2.3
15	2.3	2.3	2.3
30	2.1	2.0	2.0
45	2.0	2.0	
60	2.0	2.0	2.0

Table S27: concentration of (S)-COT warfarin (8b) in incubation medium (Logarithmic scale).



(S)-COT warfarin (8b) metabolic stability

Figure S10: (S)-COT warfarin (8b) metabolic stability.

2.5 Sample Chromatograms



Figure S11: representative chromatograms of one of the standards.



Figure S12: representative chromatograms of one of midazolam metabolite formation in positive control samples.



Figure S13: representative chromatograms of (S)-COT warfarin (8b) after 1 h incubation.

3. MS Investigation:

The below MS investigation was conducted on (*S*)-warfarin (**6b**), (*S*)-COT warfarin (**8b**) (for [M+16] and [M+34] metabolites) and (*R*)-COT warfarin (**8a**) (for [M+2] and [M+18] metabolites) according to the incubation procedures and MS parameters outlined in subsections **1.2.1** and **1.4.1** respectively.

Using Precursor Ion Scan, the m/z [M+16] and [M+34] were identified and using the Multiple Reaction Monitoring (MRM) method, the relative quantity of the metabolites at time 0, and 60 min of the incubation were compared (see below).



Figure S14: (S)-warfarin (6b) potential hydroxy metabolite (323.1 = [M + 16]) fragmentation pattern.



Figure S15: (S)-COT warfarin (8b) fragmentation pattern.



Figure S16: Potential hydroxy metabolite of (S)-COT warfarin (8b) [M + 16] fragmentation pattern.



Figure S17: Potential hydrated epoxy metabolite of (*S*)-COT warfarin (**8b**) [M + 34] fragmentation pattern.

MRM peaks



Figure S18: Comparison of [M+34] and [M+16] metabolites for (*S*)-COT warfarin (**8b**) after 0 h. **Top:** Potential hydrated epoxy metabolite of (*S*)-COT warfarin (**8b**) [M+34] chromatogram after 0 h incubation (intensity: 6.4e³). **Bottom:** Potential hydroxy metabolite of (*S*)-COT warfarin (**8b**) [M+16] chromatogram after 0 h incubation (intensity: 5.2e⁴).



Figure S19: Comparison of [M+34] and [M+16] metabolites for (*S*)-COT warfarin (**8b**) after 1 h. **Top:** Potential hydrated epoxy metabolite of (*S*)-COT warfarin (**8b**) [M+34] chromatogram after 1 h incubation (intensity: 2.1e⁴). **Bottom:** Potential hydroxy metabolite of (*S*)-COT warfarin (**8b**) [M+16] chromatogram after 1 h incubation (intensity: 5.5e⁶).

Using Precursor Ion Scan, the m/z [M+2] and [M+18] were identified and using Multiple Reaction Monitoring (MRM) method, the relative quantity of the metabolites at time 0, and 60 min of the incubation were compared.

No M+2 or M+18 with similar fragmentation pattern were found in the Precursor Ion Scan from (R)- and (S)-warfarin (**6a** and **6b**). However, [M+2] and [M+18] for COT-warfarin (**15**) were detected. Data for (R)-COT warfarin (**8a**) is shown below.



Figure S21: Potential carbonyl reduction metabolite [M + 2] of (R)-COT warfarin (**8a**) fragmentation pattern.



Figure S22: Potential carbonyl reduction and hydroxy/epoxy metabolites [M + 18] of (R)-COT warfarin (**8a**) fragmentation pattern.

$\begin{array}{c} \text{WRM peaks} \\ \hline & \text{SC = AMM(1) transfer 50 100 101 000 b tr} [NN-10] from Sample 10 [NC 101 MM] of 2011 04 55 strlf [Lines Sam$

Figure S23: Potential carbonyl reduction and hydroxy/epoxy metabolites [M+18] of (R)-COT warfarin (**8a**) chromatogram. Time 0 h; intensity: $2e^3$.



Figure S24: Potential carbonyl reduction and hydroxy/epoxy metabolites [M+18] of (*R*)-COT warfarin (**8a**) chromatogram. Time 1 h; intensity: $3.5e^{5}$.



Figure S25: Potential carbonyl reduction metabolite of (*R*)-COT warfarin (**8a**) [M+2] chromatogram (left hand peak), Time 0 h (intensity: 2.5e4).



Figure S26: Potential carbonyl reduction metabolite of (*R*)-COT warfarin (**8a**) [M+2] chromatogram (left hand peak). Time 1 h; intensity: $2.8e^5$ after 1 h incubation.

Moclobemide Study

Behavioural evaluation in mice (performed by the group led by A/ Prof. Tom Burne at the Queensland Brain Institute):

Experimental Animals

All animal care and experimental procedures complied with the Australia Code of Practice for the Care and Use of Animals for Scientific Purposes (8th edition, 2013). Ethics approval was obtained from the Animal Ethics Committee of The University of Queensland (Brisbane, Australia).

Sixty-four adult male C57BL6/J mice were purchased from the Animal Resources Centre (Perth, WA, Australia). Mice weighed 34 g (\pm 4g) upon arrival and they were housed in a purpose-built Physical Containment Level 2 (PC2) animal holding facility in groups of four in individually ventilated cages (OptiMice cages, Animal Care Systems, CO, USA) at 21°C, 40–60% humidity and 12/12 h light/dark cycle (lights on 0700 h) with ad libitum access to pelleted food (Specialty Feeds, WA) and water. The mice were habituated to the QBI animal house for 4–5 days prior to testing. One mouse became unwell and was euthanised prior to the experiment.

Test Compounds Preparation and Administration

Working solutions of each of moclobemide (9), the cubane (14) and COT (21) analogues were prepared at concentrations of 3.15mg/ml – made up in 50% EtOH and Saline, 25mg/ml – made up in 100% EtOH and 8mg/ml – made up in 100% EtOH , respectively. Working solutions were then diluted in sterile water for injection (Pfizer, West Ryde, NSW, Australia) to the final concentration of 5 mg/ml, and stored refrigerated and protected from light the day before the experiment.

Open Field activity monitoring

Each mouse was injected via the intraperitoneal route with 5 mg/ml (10ml/kg) and returned to its home cage 30 minutes prior to open field activity monitoring. The mouse was then placed in a clear open field (27.5 x 27.5 x 30cm, Med

Associates Inc, USA) within a sound attenuated chamber and activity levels were recorded for 30 minutes. The light level was set at 18 lux. As a measure of spontaneous activity, distance travelled was calculated using activity monitor tracking software based on beam breaks from three 16 beam infrared arrays and was sampled in 1 min time bins.

Data analysis

Results were analysed for statistical significance using the SPSS statistics software package (ver. 24, SPSS Inc., Chicago, Illinois). Data were pooled into 5 minute time bins and analysed using ANOVA to assess the difference between three groups, with a main effect of test compound and repeated measure on time bin. A p value of <0.05 was considered to be statistically significant.



Figure S27: Average distance travelled values for moclobemide (9) vs. the cubane analogue (14). Single bolus i.p. injections of water (vehicle), 9 or 14 were administered to adult male C57BL6/J mice at doses of 5 mg / kg. After a wait period of thirty minutes, the mice were individually placed in the center of an enclosure ("open field") surrounded by walls that prevented escape. The total distance travelled by each mouse in thirty minutes was measured. There was a significant main effect of Time (F_{5,145}=27.67, p<0.001) and test compound (F_{1,29}=3.35, p<0.05).



Figure S28: Average distance travelled values for moclobemide (**9**) vs. the COT analogue (**21**). Single bolus i.p. injections of water (vehicle), **9** or **21** were administered to adult male C57BL6/J mice at doses of 5 mg / kg. After a wait period of thirty minutes, the mice were individually placed in the center of an enclosure ("open field") surrounded by walls that prevented escape. The total distance travelled by each mouse in thirty minutes was measured. There was a significant main effect of Time (F_{5,140}=22.65, p<0.001) and test compound (F_{1,28}=4.77, p<0.01).

Pravadoline Study

Antinociception evaluation (performed by the group led by Prof. Maree Smith at the Centre for Integrated Preclinical Drug Development):

Experimental Animals

All animal care and experimental procedures complied with the Australia Code of Practice for the Care and Use of Animals for Scientific Purposes (8th edition, 2013). Ethics approval was obtained from the Animal Ethics Committee of The University of Queensland (Brisbane, Australia).

Male Sprague-Dawley (SD) rats were purchased from the Animal Resources Centre (Perth, WA, Australia). Rats weighed 180-200 g upon arrival and they were housed in a purpose-built Physical Containment Level 2 (PC2) animal holding facility in groups of three to four in individually ventilated cages (BioZone, Thorne Hill, Ramsgate, Kent, UK). Rat chow (Specialty Feeds, Glen Forrest, WA, Australia) and tap water were available *ad libitum* throughout the housing period. Rats were maintained in cages that contained recycled paper bedding material (FibreCycle Pty Ltd, Yatala, QLD, Australia) and environmental enrichment comprising Kimwipes (Kimberly-Clark Professional, Milsons Point, NSW, Australia), a rodent hutch (red Perspex hutch) and Rat Chewsticks (Able Scientific, Welshpool, WA, Australia) in each cage. The animal holding facility had a 12 h/12 h light/dark cycle and a mean (±SEM) room temperature of 23 (±3) °C. Animals were acclimatised for at least four days in the animal holding facility prior to initiation of any experimentation.

Rat model of Freund's Complete Adjuvant (FCA) induced Inflammatory Pain

Rats, whilst anaesthetised with 3% isoflurane (Abbott Australasia Pty Ltd, Botany, NSW, Australia) delivered in oxygen, received an intraplantar (i.pl.) injection of 150 μ L of Freund's Complete Adjuvant (FCA; Sigma-Aldrich, MO, USA) into the plantar aspect of the left hindpaws. Clinical observations were performed post-FCA injection and at least once weekly until study completion (≤ day 14).

Test Compounds Preparation and Administration

Working solutions of each of pravadoline (**10**), cubane (**15**) and COT (**22**) analogues were prepared at concentrations of 300 ug/100-150 μ L, 321 ug/150 μ L and 321 ug/150 μ L respectively. Working solutions were prepared using pure DMSO and then diluted with 40% Captisol (CyDex Pharmaceuticals Inc, KS, USA) in sterile water for injection (Pfizer, West Ryde, NSW, Australia) so that the final vehicle concentrations were 10% DMSO/40% Captisol. Working solutions were stored refrigerated and protected from light with a one week expiry date from the date of preparation.

Rats were anaesthetised briefly with 3% isoflurane (Abbott Australasia Pty Ltd, Botany, NSW, Australia) delivered in oxygen to facilitate intraplantar (i.pl.) drug administration. The i.pl. injection volumes were fixed at 150 μ L and injections were made using a Hamilton syringe (SGE Analytical Science Pty Ltd, Ringwood, VIC, Australia) with a 25G x 5/8 needle (Terumo Corporation, Shibuya-ku, Tokyo, Japan). Each rat received a maximum of three doses according to a 'washout' protocol in the inflamed hindpaw with at least 3 days between each successive dose.

Rat model of acute pain: Noxious mechanical stimuli to the hindpaws

The Randall Sellito apparatus was used to apply acute noxious mechanical stimuli (Ugo Basile, Comerio, Italy) to the hindpaws of rats. As previously described,^[67] a noxious mechanical stimulus with increasing force was applied to the medial portion of the hindpaw until a withdrawal response was evoked. A cut-off force of 250 g was used to prevent tissue damage. Baseline Paw Pressure Thresholds (PPTs) were determined in both hindpaws prior to i.pl. FCA. Baseline PPTs were the mean of three readings for the corresponding hindpaw, with a 5-min interval between consecutive measurements. Following dose administration, inflamed hindpaw PPTs were measured at the following post-dosing times 30, 45, 60, 75, 90, 120 and 180 min.
Data Analysis

For individual rats, delta (Δ) PPTs were calculated by subtracting the predosing PPT from the corresponding post-dosing PPTs and any negative Δ PPTs were arbitrarily assigned a value of 0. For each rat, the extent and duration of antinociception (pain relief) was determined by using trapezoidal integration to estimate the area under the Δ PPT versus time curve (Δ PPT AUC values) using GraphPadTM Prism software (version 7.03; GraphPad Software, San Diego, CA, USA).

Analysis of variance (ANOVA) with a post-hoc Dunn's Multiple Comparison test was performed on the mean (±SEM) Δ PPT AUC values for groups of SD-rats administered single i.pl. bolus doses of the cubane (**15**) and COT (**22**) analogues or the positive control item, pravadoline (**10**) relative to animals administered vehicle (10%DMSO/40% Captisol). Microsoft Excel (version 16.0.4549.1000; Microsoft Corporation, Redmond, WA, USA) and GraphPadTM Prism software (version 7.03; GraphPad Software, San Diego, CA, USA).were used for all data and statistical analysis. The statistical significance criterion was p ≤ 0.05.

Only responder-rats were included for the data plotting and analysis. Responders were defined as FCA-rats that received a test or positive control item that evoked a Δ PPT AUC response that was > 2 standard deviations above that of vehicle (i.e. mean vehicle + 2 SD).



PPT

Figure S29: Mean (±SEM) paw pressure threshold (PPT) versus time curves for the ipsilateral hindpaws of FCA-rats that responded¹ following administration of single intraplantar bolus doses of vehicle (10%DMSO/40% Captisol; n=6) at 150 μ L, pravadoline (**10**) at 300 μ g/100-150 μ L (n=11), pravadocube (**15**) at 321 μ g/150 μ L (n=9) and pravadocot (**22**) at 321 μ g/150 μ L (n=5) at time 0 (pre-dosing) and at 30, 45, 60, 75, 90, 120 and 180 min post-dosing.

¹ Responders were defined as FCA-rats that received a test item (pravadoline (**10**), cubane (**15**) or COT (**22**) analogues) that evoked a Δ PPT AUC response that was > 2 standard deviations above that of vehicle (i.e. mean vehicle + 2 SD; refer to **Figure S30**). N=6, n=2 and n=4 were excluded from pravadoline (**10**), cubane (**15**) and COT (**22**) analogues respectively using the above mentioned exclusion criteria.



Figure S30: Mean (±SEM) extent and duration of action quantified as the mean (±SEM) areas under the Δ PPT versus time curves (Δ PPT AUC values) for the ipsilateral (FCA injected) hindpaws of FCA-rats that responded¹ following administration of single intraplantar bolus doses of vehicle (10%DMSO/40% Captisol; n=6) at 150 µL, pravadoline (**10**) at 300 µg/100-150 µL (n=11), cubane analogue (**15**) at 321 µg/150 µL (n=9) and COT analogue (**22**) at 321 µg/150 µL (n=5) at time 0 (pre-dosing) and at 30, 45, 60, 75, 90, 120 and 180 min post-dosing.

¹ Responders were defined as FCA-rats that received a test item (pravadoline (**10**), cubane (**15**) or COT (**22**) analogues) that evoked a \triangle PPT AUC response that was > 2 standard deviations above that of vehicle (i.e. mean vehicle + 2 SD).

* *p*≤0.05, ** *p*<0.01, *** *p*<0.001; one-way ANOVA (post-hoc Dunn's multiple comparisons test).

	10%DMSO/40%Captisol - 150uL (n=6)	Pravadoline - 300µg/100-150µL (n=11)	Cubane-Pravadoline - 321ug/150uL (n=9)	COT-Pravadoline - 321ug/150uL (n=5)
Number of values	6	11	9	5
Minimum	313	1600	1275	1763
25% Percentile	341	1725	1494	2544
Median	431.3	1913	3088	3774
75% Percentile	712.2	4875	3901	6019
Maximum	1312	6188	5151	6238
Mean	558.3	2875	2827	4180
Std. Deviation	377.4	1721	1382	1844
Std. Error of Mean	154.1	518.9	460.7	824.5
Lower 95% Cl of mean	162.3	1719	1764	1890
Upper 95% Cl of mean	954.4	4031	3889	6469
Sum	3350	31628	25441	20898
D'Agostino & Pearson normality test				
K2	N too small	3.436	1.014	N too small
P value		0.1794	0.6022	
Passed normality test (alpha=0.05)?		Yes	Yes	
P value summary		ns	ns	
Shapiro-Wilk normality test				
W	0.6835	0.7298	0.9106	0.9383
P value	0.0041	0.0011	0.3199	0.6536
Passed normality test (alpha=0.05)?	No	No	Yes	Yes
P value summary	**	**	ns	ns
KS normality test				
KS distance	0.382	0.3246	0.2157	0.2101
P value	0.0065	0.0019	>0.1000	>0.1000
Passed normality test (alpha=0.05)?	No	No	Yes	Yes
P value summary	**	**	ns	ns

 Table S28:
 summary table of normality test results for the vehicle and three test items.

SAHA Study

Inhibition of cancer cell growth in culture (performed by the group led by Dr Glen Boyle at the QIMR Berghofer Medical Research Institute):

Materials and Methods

SAHA (11) was purchased from Cayman Chemical (Ann Arbor, Michigan 48108 USA; Catalogue #10009929). The cubane (16) and COT (23) analogues were synthesized as detailed **Supplementary Information Part 1**. Both compounds were dissolved in DMSO. MM96L and MCF7 are human tumor cell lines derived from melanoma and breast cancer respectively. NFF are early passage neonatal foreskin fibroblasts.

Sulforhodamine B assay for cell survival assessment of cultured cells.

Cells were seeded at 2,500 per microtitre well (96-well plate) in 10% FCS-RPMI 1640 culture medium, treated, and allowed to grow until the controls were nearly confluent (6 days). Culture media was removed; the wells were then washed twice with phosphate buffered solution (PBS), fixed with ethanol for a minimum of 5 min. and washed with water. Sulforhodamine B (SRB) solution (50 μ L of 0.4% in 1% acetic acid) was added and the mixture left at room temperature for a minimum of 15 min. The plate was washed rapidly with tap water and then twice with 0.1-1% acetic acid, the liquid being removed by tapping each time. After addition of 100 μ L/well of 10 mM Tris base (unbuffered, pH > 9), plates were left for a minimum of 5 min, then the absorbance was read at 564 nm on an Biotek Synergy H4 Multi Mode Plate Reader, with a 3 second prior shaking operation. Data were exported to an Excel spreadsheet. After subtraction of a blank (i.e. wells with no cells, A564 typically ~0.04), growth inhibition was calculated as a percentage of the untreated control and plotted against dose.^[68] Non-linear regression was performed using GraphPad[™] Prism version 7.02 for Windows (GraphPad Software, La Jolla California USA).

Compound	NFF	MCF-7	MM96L
SAHA (11)	32 ± 5	17 ± 1	26 ± 5
Cubane analogue (16)	142 ± 21	53 ± 4	137 ± 5
COT analogue (23)	77 ± 11	37 ± 3	73 ± 8

Table S29: IC₅₀ Values (ng/ml).



Figure S31: Cell survival of NFF, MCF-7 and MM96L cells following treatment with SAHA (**11**), and cubane (**16**) or COT (**23**) analogues for 6 days.



Figure S32: Cell survival of NFF, MCF-7 and MM96L cells following treatment with SAHA (**11**) for 6 days.



Figure S33: Cell survival of NFF, MCF-7 and MM96L cells following treatment with cubane analogue (16) for 6 days.





Figure S34: Cell survival of NFF, MCF-7 and MM96L cells following treatment with COT analogue (23) for 6 days.

Benzyl Benzoate Study

Acaricidal activity of scabies mites (performed by the group led by Professor James McCarthy at the QIMR Berghofer Medical Research Institute):

Method

Three concentrations (100mM, 50mM, 25mM) of each compound for testing were prepared using mineral oil as diluent. Each compound concentration was spread thinly in each duplicate dish and live mites (at least 10) were placed to allow contact with compounds. Mite status was observed under the microscope within one hour of contact and hourly thereafter, for up to 7 hours. Mite status was again observed after 24 hours. (Mortality was described as absence of leg movement or gut peristalsis when touched with needle). Benzyl Benzoate (**12**) (25mM) was used as the positive control acaricide and mineral oil as negative control. The bioassay was performed twice.

Data was analysed using Survival Analysis in Graph Pad Prism[™] (v7) and statistical significance between survival curves compared by the Log Rank Test. Results are expressed as median survival time.

Results

No solubility issues were identified; all three compounds dissolved well in mineral oil at room temperature. Of the three concentrations tested, the highest concentration (100mM) showed discriminatory activity against scabies mites within 7 hours of observation with benzyl cyclooctatetraenecarboxylate (26) giving the fastest killing effect (Median survival of mites= 4 hrs). At lower concentrations (50 mM and 25mM) of benzyl cyclooctatetraenecarboxylate (26), median survival of mites (7 hours) was the same. On the other hand, mites remained alive within 7 hours of exposure to cyclooctatetraenemethyl benzoate (25) of the same lower concentrations. Median survival of mites in cyclooctatetraenylmethyl cyclooctatetraenecarboxylate (24) was the same (7 hrs) in all concentrations (100mM, 50mM, 25mM) tested. Structure -activity relationship was evident in the mites bioassay performed.

	26	25	24	Benzyl	Mineral Oil
				Benzoate	
				(12)	
Concentration(mM)	(hour)	(hour)	(hour)	(hour)	(hour)
100	4	6	7	-	*
50	7	>7	7	-	*
25	7	>7	7	1	*

Table S30: median survival time of scabies mites in cycloctatetraene derivatives 24-26 in contact

 bioassays. *mites remained alive >24 hrs

When mites were checked after 24 hrs of exposure to test compounds, % mortality was observed to be dose-dependent with all three compounds demonstrating acaricidal activity with longer exposure time. This is in comparison to mites in contact with mineral oil (negative control) that remained alive (39/42=93%), after 24 hours. Data as shown in **Table S31** below:

	26	25	24
Concentration	*	*	*
(mM)			
100	38/40(95%)	39/40(98%)	30/40(75%)
50	36/40(90%)	31/40(78%)	27/40(68%)
25	26/40(65%)	26/40(65%)	23/40(58%)

Table S31: proportion of dead mites after 24 hours of contact with compounds. *no. of dead mites/total no. of mites exposed (percentage).

Cyclooctatetraene compounds (100mM)



Figure S35: survival of scabies mites in 100mM. Cyclooctatetraene compounds (**24-26**) compared to survival in mineral oil (negative control) and benzyl benzoate (**12**) (positive acaricidal control).



Figure S36: survival of scabies mites in 50mM. Cyclooctatetraene compounds (24-26) compared to survival in mineral oil (negative control) and benzyl benzoate (12) (positive acaricidal control).

Cyclooctatetraene compounds (50mM)



Figure S37: survival of scabies mites in 25mM. Cyclooctatetraene compounds (**24-26**) compared to survival in mineral oil (negative control) and benzyl benzoate (**12**) (positive acaricidal control).

Diflubenzuron Study

Tribolium castaneum (rust-red flour beetle) evaluation (performed by the group led by Prof. Gimme Walter at the School of Biological Sciences, The University of Queensland):

Introduction

Tribolium has often been used in investigations of the mode of action of insecticides, using insect growth regulators (IGRs) such as benzoylphenyl ureas (BPUs), including diflubenzuron (**13**), an inhibitor of chitin formation.^[69] Diflubenzuron (**13**) is considered most effective on the larval stages of arthropods, inducing abortive molting.^[70] We tested the mortality imposed on field strain *T. castaneum* larvae by diflubenzuron (**13**) at different doses relative to that caused by the COT analogue (**27**). In general, methods are similar to those we used previously.^[54]

Materials and Methods - Rearing *T. castaneum*

Three hundred *T. castaneum* adults (collected from the field at Dalby, Queensland [27° 11' S 151° 16' E]) were placed in a 400 g whole-wheat flour enriched with 5% (20 g) torula yeast mixture at 25 °C and ~70% rh overnight. The following day the adults were sieved from the medium, which was retained so as to culture the larvae that hatched from the eggs deposited in it. After three weeks these larvae were ~6 mm long, so were close to pupation, and were used in experiments.

Experimental methods

The treatment doses and application methods for testing chitin inhibition in insects using IGR's vary considerably in the literature (e.g. direct application to larval and adult insects, and direct or acetone dissolved application to food sources).^[71] We used the following method, which is fairly representative.^[71] Controls comprised a Petri dish containing 10 late instar larvae in a 10 g organic

whole wheat flour/yeast medium prepared as described above. The diflubenzuron (13) and COT analogue (27) treatments were conducted in the same manner as the controls, with the appropriate weight of the relevant compound (53, 35, 18, 9 and 4 μ mol) added to the flour medium and mixed thoroughly, 30 seconds clockwise then 30 seconds anticlockwise, before adding the larvae. All treatments were held at 25 °C and 70% rh for 10 days. The larvae were sieved from the medium to record mortality. Data were analysed by means of ANOVA's. No significance was determined from the ANOVA's, so post hoc tests were unnecessary.

Comparison with equimolar doses of (13) and (27)

The efficacy of the COT analogue (**27**) and diflubenzuron (**13**) at equimolar doses were tested on a field strain of *T. castaneum* (see above). Three replicates of each dose level were run, plus three blank controls.

Results

None of the treatments imposed mortality that was significantly higher than that recorded in the control tests. The culture medium could have imposed stresses on insects adapted to field conditions and led to an increase in the variance in recorded mortality. Similar levels of variance were also seen in a previous study^[54] that tested the cubane analogue of diflubenzuron (i.e. **20**) against field strain beetle larvae. However, the controls for laboratory strain tests returned lower levels of mortality.



Figure S38: mean mortality for the cubane analogue (20) and diflubenzuron (13) at different concentrations.



Figure S39: mean mortality for COT analogue(27) and diflubenzuron (20) at different concentrations.

Comparing the efficacy of equimolar doses of the cubane (20) and COT (27) analogues against that of diflubenzuron (13)

Tests comparing diflubenzuron (13) against the cubane (20) and COT (27) analogues were conducted separately, so the data could not be combined for statistical analysis. However, comparison of the efficiency of compounds 20 and 27 against diflubenzuron (13) is valid because field strain beetles from the same locality (Dalby) were used in both experiments.

In neither study were the mortalities associated with the cubane (**20**) and COT (**27**) analogues significantly different from the mortality levels recorded in the untreated controls. Nevertheless, in one experiment using a mid-range concentration of the cubane analogue (**20**) significantly higher mortality was recorded than with diflubenzuron (**13**),^[54] although this could be attributed to chance. Results from the COT analogue (**27**) experiment were generally similar in that no significant differences from the controls were recorded.

Supplementary Information Part 3: Crystallographic Data

Crystallographic information for compounds **S10a**, **S13** and **23** (collection and analysis by Prof. Paul Bernhardt at the School of Chemistry and Molecular Biosciences):

CCDC 1847176 (for compound **S10a**), 1847177 (for compound **S13**) and 1847178 (for compound **23**) contain the supplementary crystallographic data for this paper. These data are provided free of charge by The Cambridge Crystallographic Data Centre.







Table S32: ci	rystal data	and structure	refinement for	[.] S10a
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CCDC	1847176	
Identification code	S10a	
Empirical formula	$C_{21} H_{18} O_4$	
Formula weight	334.35	
Temperature	190(2) K	
Wavelength	1.54184 Å	
Crystal system	Orthorhombic	
Space group	<i>P</i> 2 ₁ 2 ₁ 2 ₁	
Unit cell dimensions	$a = 5.4085(2) \text{ Å}$ $\alpha = 90^{\circ}.$	
	$b = 14.7083(4) \text{ Å} \qquad \beta = 90^{\circ}.$	
	$c = 19.7970(7) \text{ Å}$ $\gamma = 90^{\circ}.$	
Volume	1574.85(9) Å ³	
Z	4	
Density (calculated)	1.410 Mg/m^3	
Absorption coefficient	0.791 mm ⁻¹	
F(000)	704	
Crystal size	0.6 x 0.15 x 0.15 mm ³	
Theta range for data collection	3.74 to 62.45°.	
Index ranges	-6<=h<=5, -16<=k<=12, -21<=l<=22	
Reflections collected	5483	
ndependent reflections 2481 [R(int) = 0.0169]		
Completeness to theta = 62.45°	99.7 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	1 and 0.73628	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2481 / 0 / 228	
Goodness-of-fit on F ²	1.068	
Final R indices [I>2sigma(I)]	(I)] $R1 = 0.0287, wR2 = 0.0753$	
R indices (all data) $R1 = 0.0293$, wR2 = 0.0760		
Absolute structure parameter	-0.01(15)	
Largest diff. peak and hole	0.147 and -0.250 e.Å ⁻³	





Table S33: crystal data and structure refinement for S13.

CCDC	1847177	
Identification code	S13	
Empirical formula	$C_{29}H_{23}N_3O_6$	
Formula weight	509.50	
Temperature	190(2) K	
Wavelength	1.54184 Å	
Crystal system	Orthorhombic	
Crystal system Space group	Orthorhombic P 2 ₁ 2 ₁ 2 ₁	
Crystal system Space group Unit cell dimensions	Orthorhombic P $2_1 2_1 2_1$ a = 9.4892(1) Å	α= 90°.
Crystal system Space group Unit cell dimensions	Orthorhombic P $2_1 2_1 2_1$ a = 9.4892(1) Å b = 10.7010(1) Å	$\alpha = 90^{\circ}.$ $\beta = 90^{\circ}.$
Crystal system Space group Unit cell dimensions	Orthorhombic P $2_1 2_1 2_1$ a = 9.4892(1) Å b = 10.7010(1) Å c = 23.5032(3) Å	$\alpha = 90^{\circ}.$ $\beta = 90^{\circ}.$ $\gamma = 90^{\circ}.$

Ζ Density (calculated) Absorption coefficient F(000) Crystal size Theta range for data collection Index ranges Reflections collected Independent reflections Completeness to theta = 62.42° Absorption correction Max. and min. transmission Refinement method Data / restraints / parameters Goodness-of-fit on F² Final R indices [I>2sigma(I)] R indices (all data) Absolute structure parameter

Largest diff. peak and hole

4 1.418 Mg/m^3 0.832 mm⁻¹ 1064 0.4 x 0.1 x 0.1 mm³ 3.76 to 62.42°. -9<=h<=10, -12<=k<=11, -26<=l<=27 10005 3772 [R(int) = 0.0200] 99.9 % Semi-empirical from equivalents 1 and 0.89725 Full-matrix least-squares on F² 3772 / 0 / 345 1.057 R1 = 0.0284, wR2 = 0.0711R1 = 0.0291, wR2 = 0.07180.01(15) 0.229 and -0.198 e.Å-3



Table S34: crystal data and structure refinement for 23

CCDC	1847178			
Identification code	23			
Empirical formula	$C_{16}H_{22}N_2O_3$	$C_{16}H_{22}N_2O_3$		
Formula weight	290.36			
Temperature	190(2) K	190(2) K		
Wavelength	1.54184 Å			
Crystal system	Monoclinic			
Space group	$P2_{1}/c$			
Unit cell dimensions	a = 18.325(5) Å	$\alpha = 90^{\circ}$.		
	b = 9.2693(8) Å	$\beta = 90.000(9)^{\circ}.$		
	c = 9.2733(8) Å	$\gamma = 90^{\circ}.$		
Volume	1575.2(5) Å ³			
Z	4			
Density (calculated)	1.224 Mg/m^3			
Absorption coefficient	0.688 mm ⁻¹			
F(000)	624	624		
Crystal size	0.2 x 0.08 x 0.02 mm ³			
eta range for data collection 4.83 to 62.40° .				
Index ranges -21<=h<=19, -10<=k<=10, -10<		10, -10<=l<=10		
Reflections collected	4181			
Independent reflections	4181 [R(int) = 0.0000]			

98.3 % Completeness to theta = 62.40° Absorption correction Semi-empirical from equivalents Max. and min. transmission 1 and 0.7873 Full-matrix least-squares on F² Refinement method 4181 / 0 / 192 Data / restraints / parameters Goodness-of-fit on F² 0.986 Final R indices [I>2sigma(I)] R1 = 0.0511, wR2 = 0.1466R1 = 0.0771, wR2 = 0.1631R indices (all data) 0.213 and -0.245 e.Å⁻³ Largest diff. peak and hole

Supplementary Information Part 4: ¹H and ¹³C Spectra

Spectroscopic data pertaining to warfarin (6), moclobernide (9), pravadoline (10), SAHA (11), benzyl benzoate (12), diflubenzuron (13), and analogues (performed by the group led by Prof. Craig Williams at the School of Chemistry and Molecular Biosciences):






































S153









































Supplementary Information Part 5: Bibliography

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