

Electronic Supporting Information

Deep eutectic solvent for an expeditious sono-synthesis of novel series of bis-quinazolin-4-one derivatives as potential anticancer agents

Wael Abdelgayed Ahmed Arafa

Chemistry Department, College of Science, Jouf University, P.O. Box 2014, Sakaka, Aljouf, Kingdom of Saudi Arabia

Chemistry Department, Faculty of Science, Fayoum University P.O. Box 63514, Fayoum City, Egypt

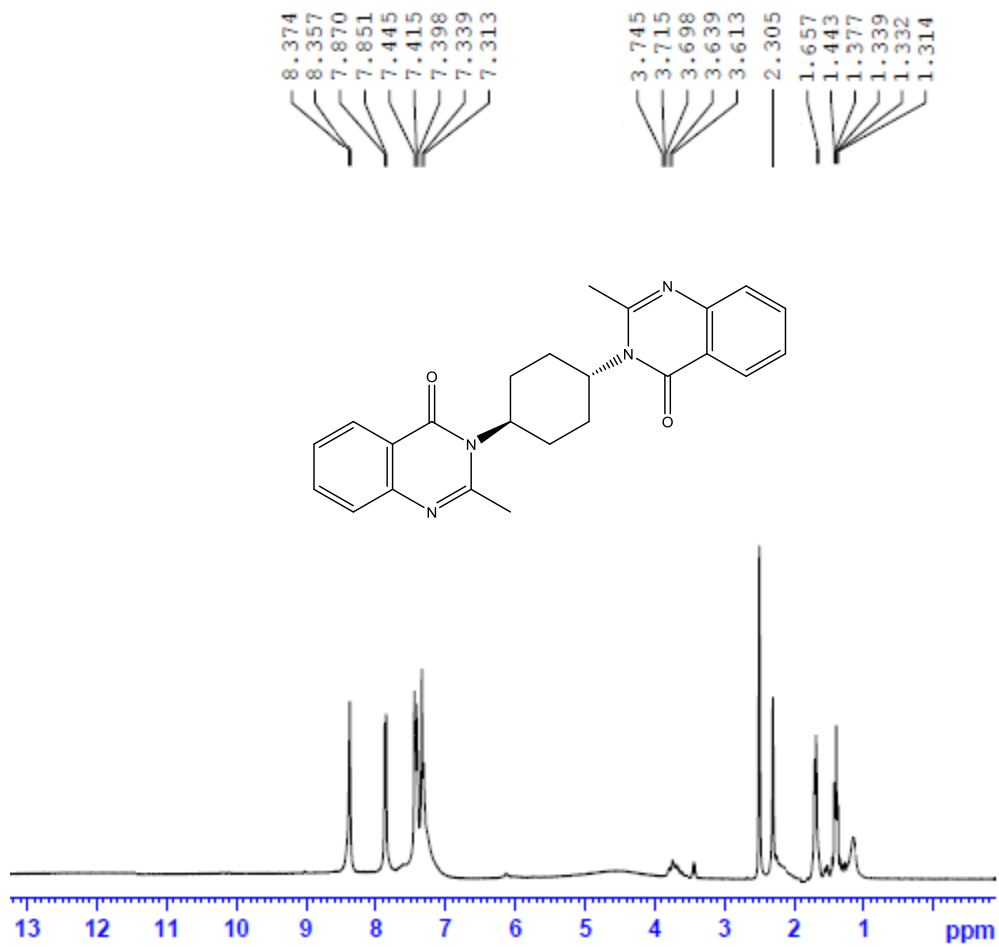
E-mail: waa00@fayoum.edu.eg

Index

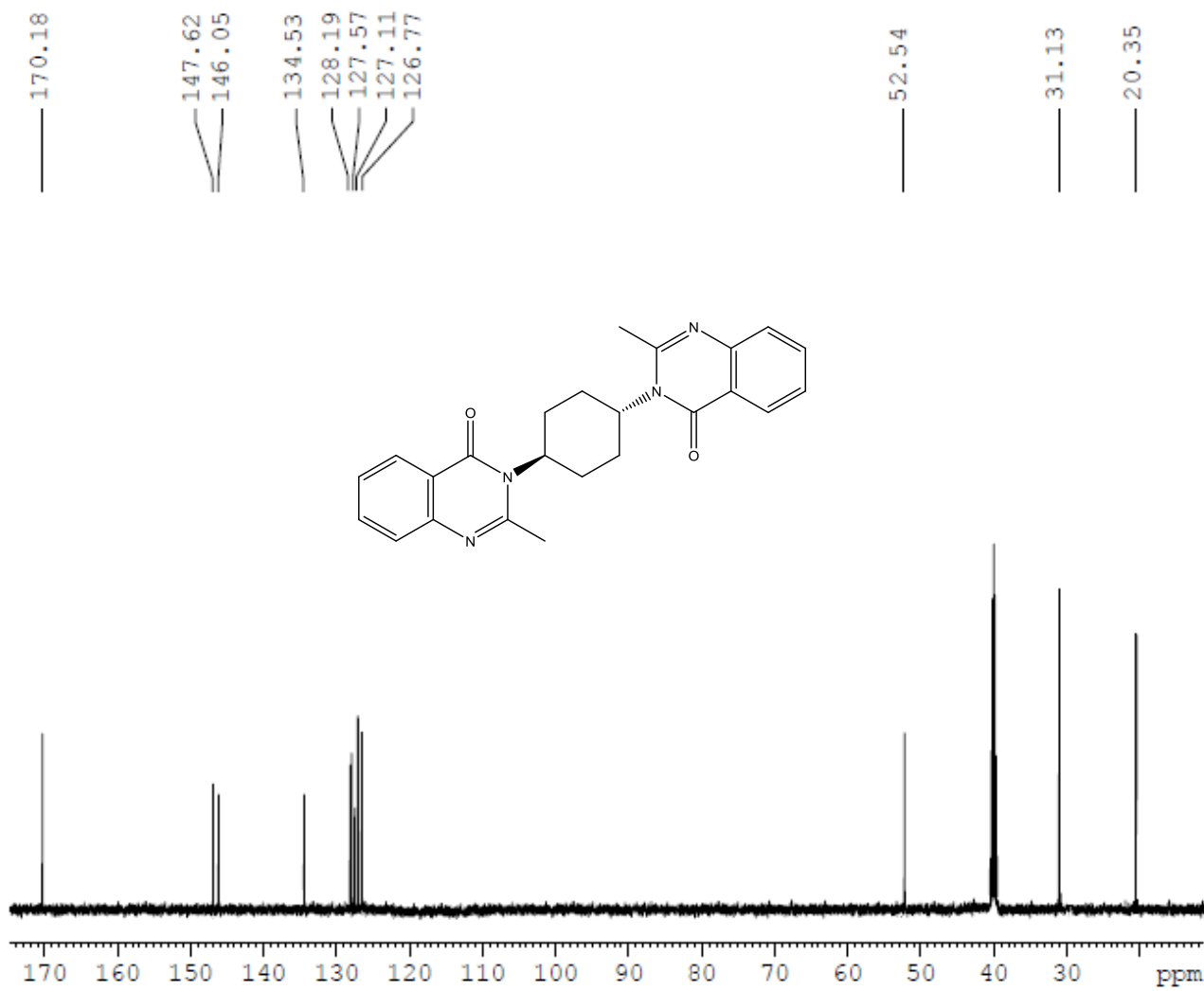
No.	Content	Page No.
1.	Experimental details for the synthesis of 1a-e	S1
2.	Copies of ¹ H NMR and ¹³ C NMR of the synthesized products	S2-S58
3.	Table 3. The anti-proliferative activities of the newly prepared bis-quinazolinone derivatives against human cancer cell lines.	S59
4.	Green metrics calculations	S62
5.	References	S62

Synthesis of **1a-e**

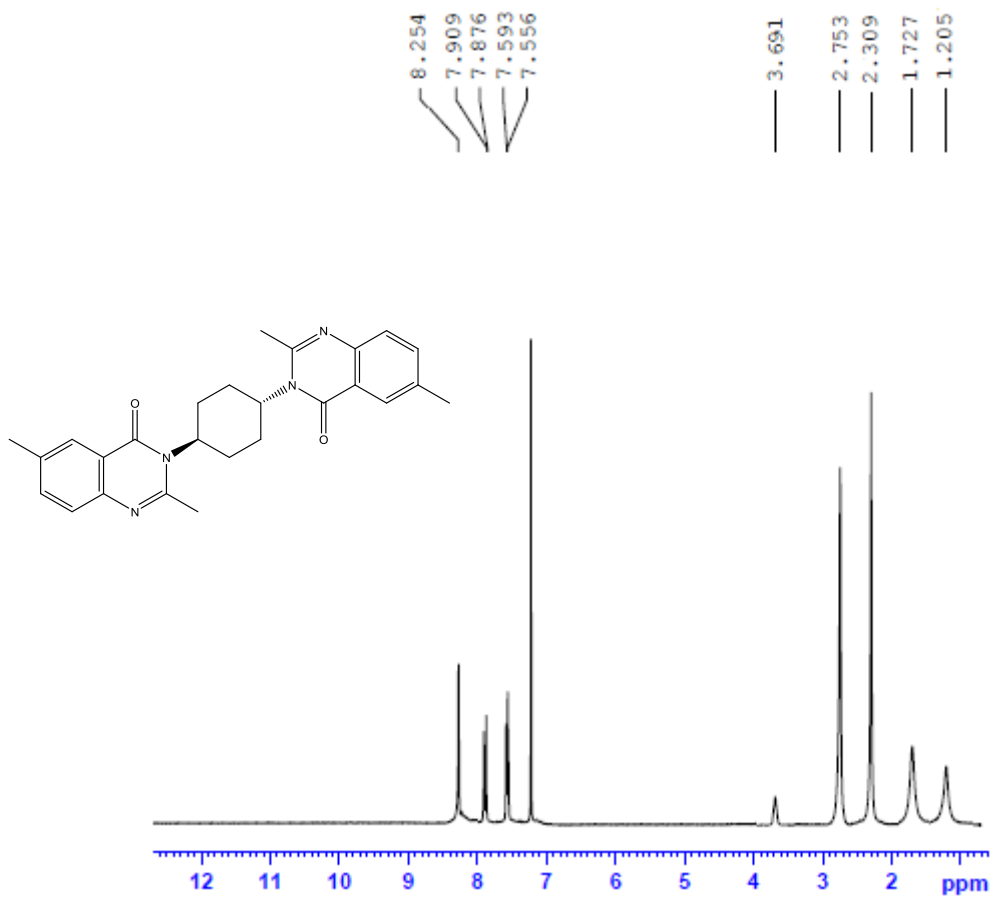
For the synthesis of derivatives **1a-e**, a mixture of anthranilic acid derivatives (1 mmol) and acetic anhydride (\approx 10 mL, 10 mmol) was sonicated at 40 °C for 10 min. The excess of acetic anhydride was distilled off under vacuum and the residue was dissolved in DCM. The obtained crystals filtered, dried and used for the next steps.



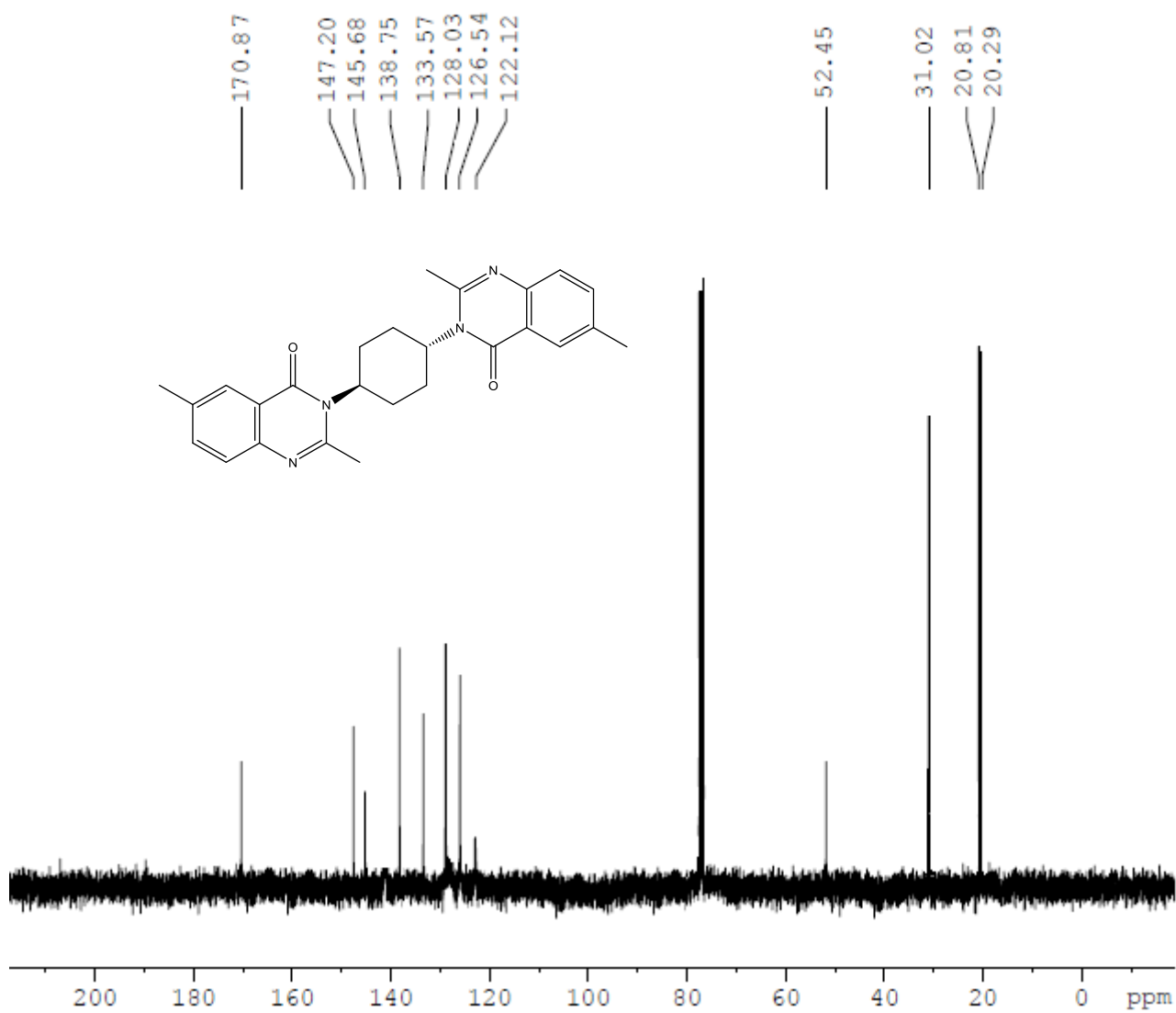
¹H NMR of **3a**



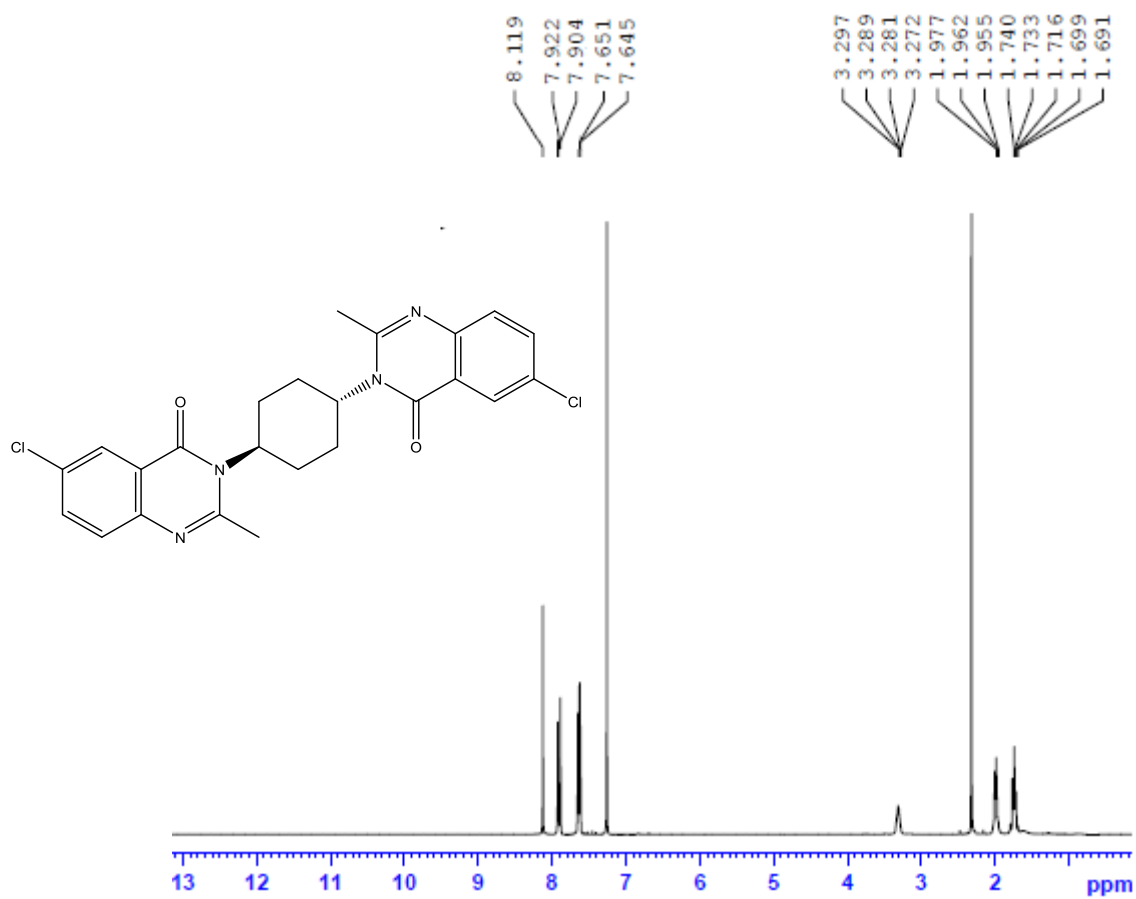
^{13}C NMR of 3a



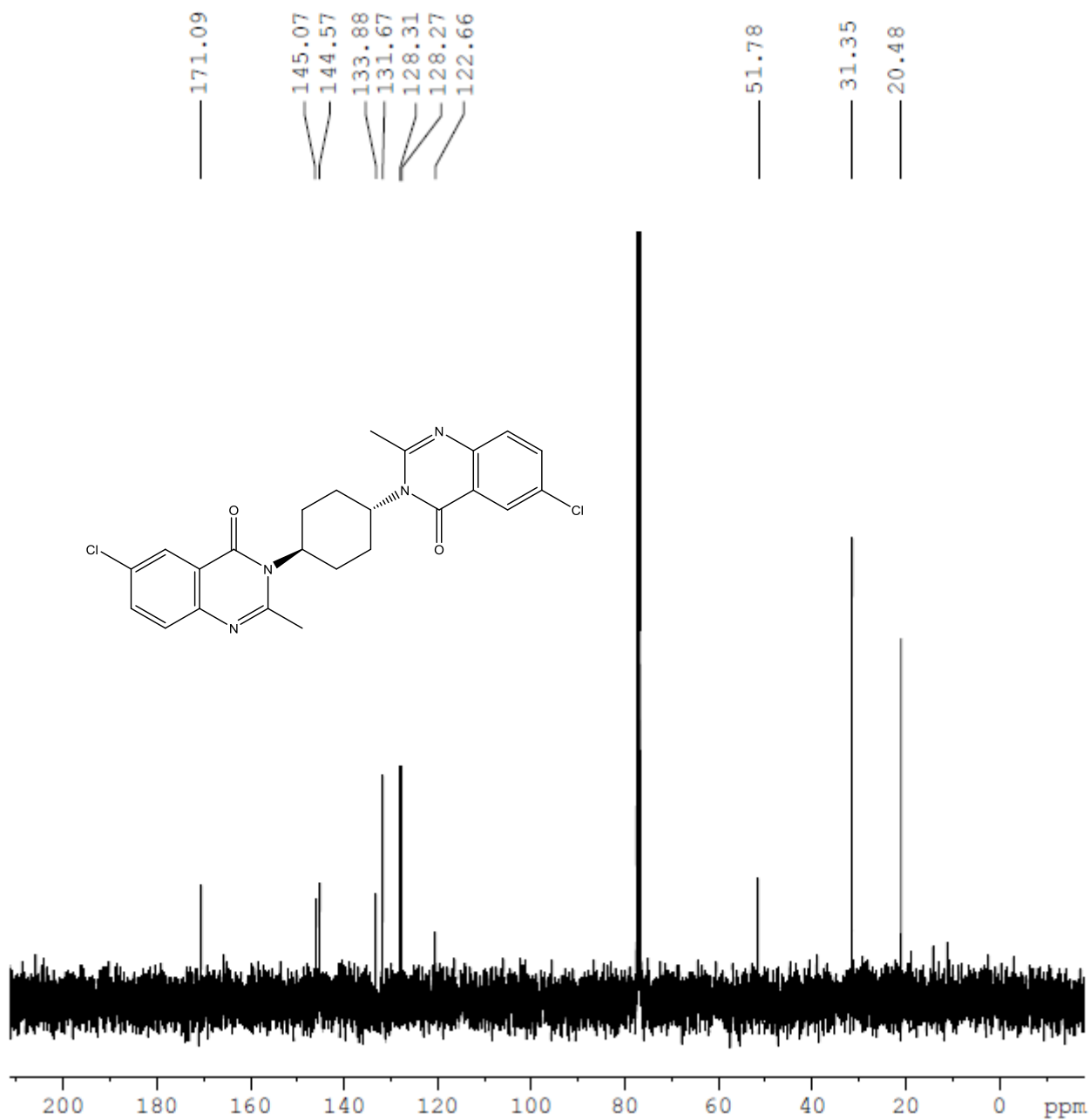
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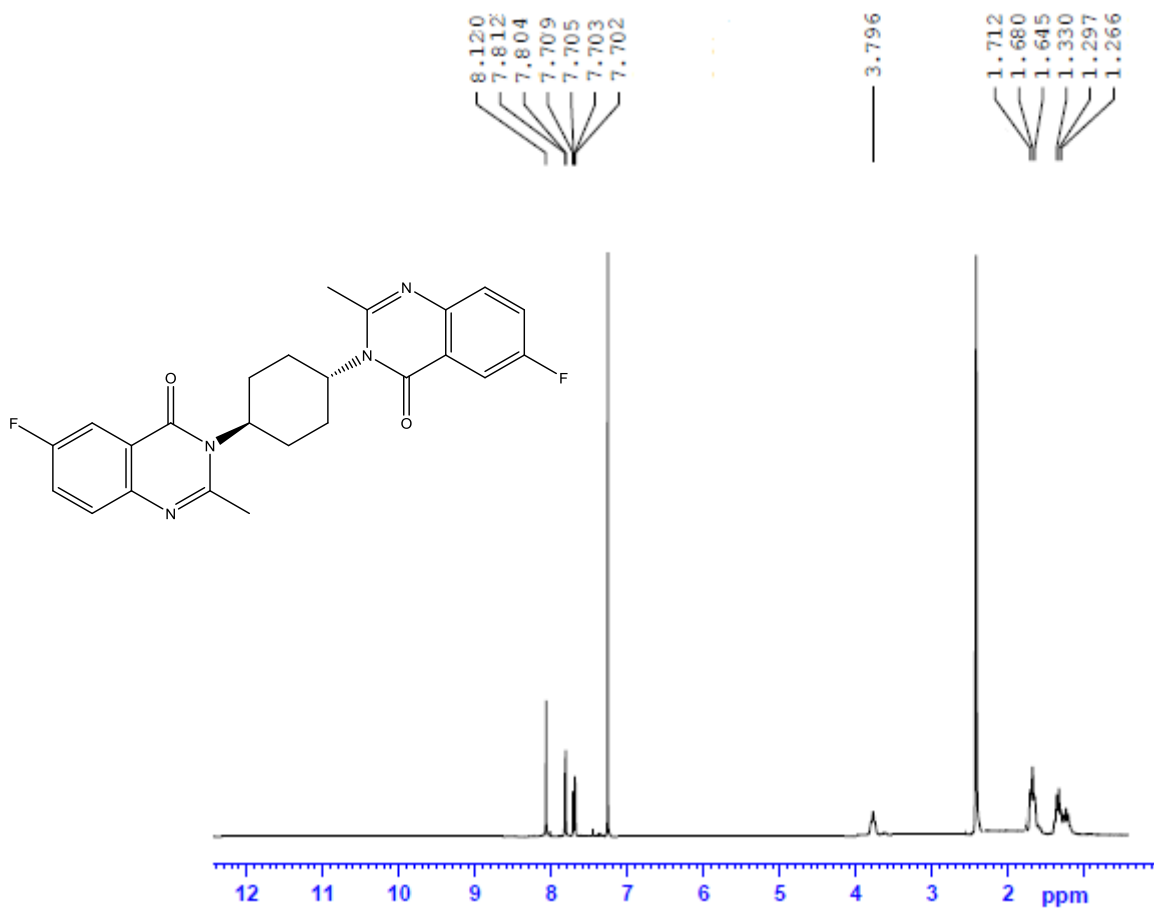
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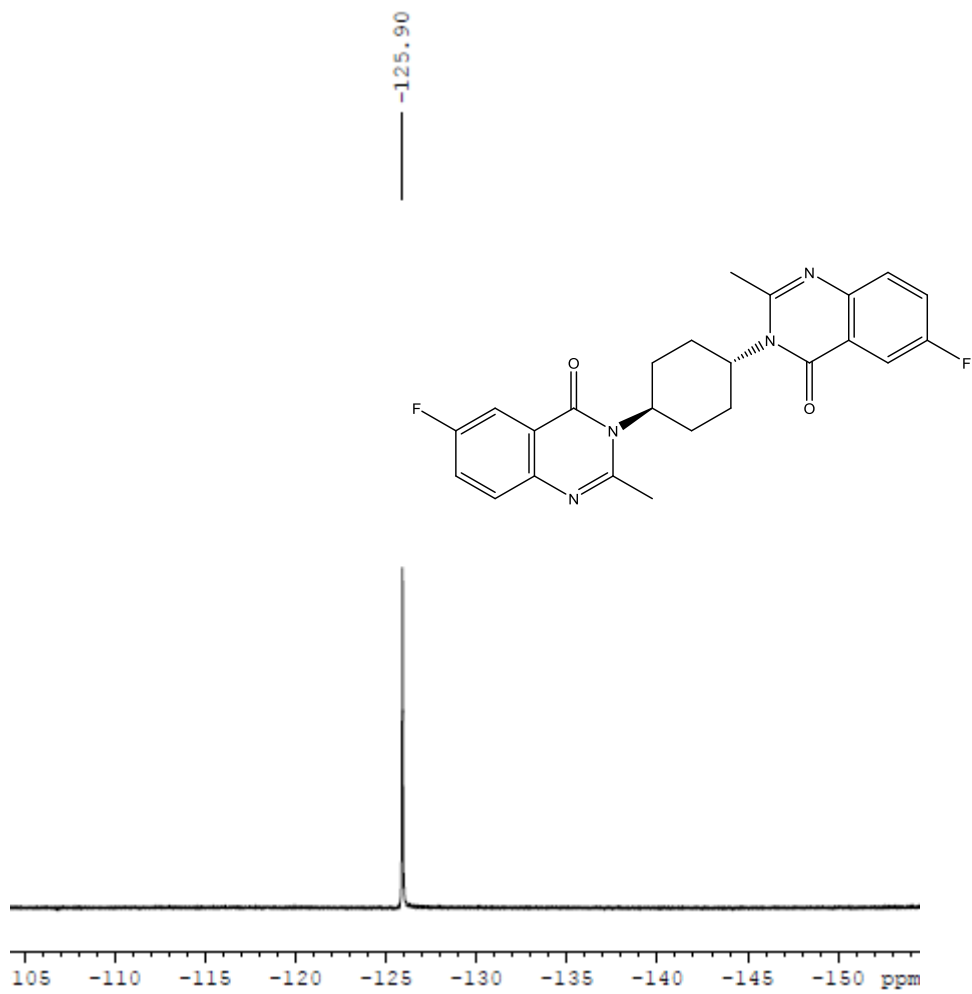
¹H NMR of **3c**



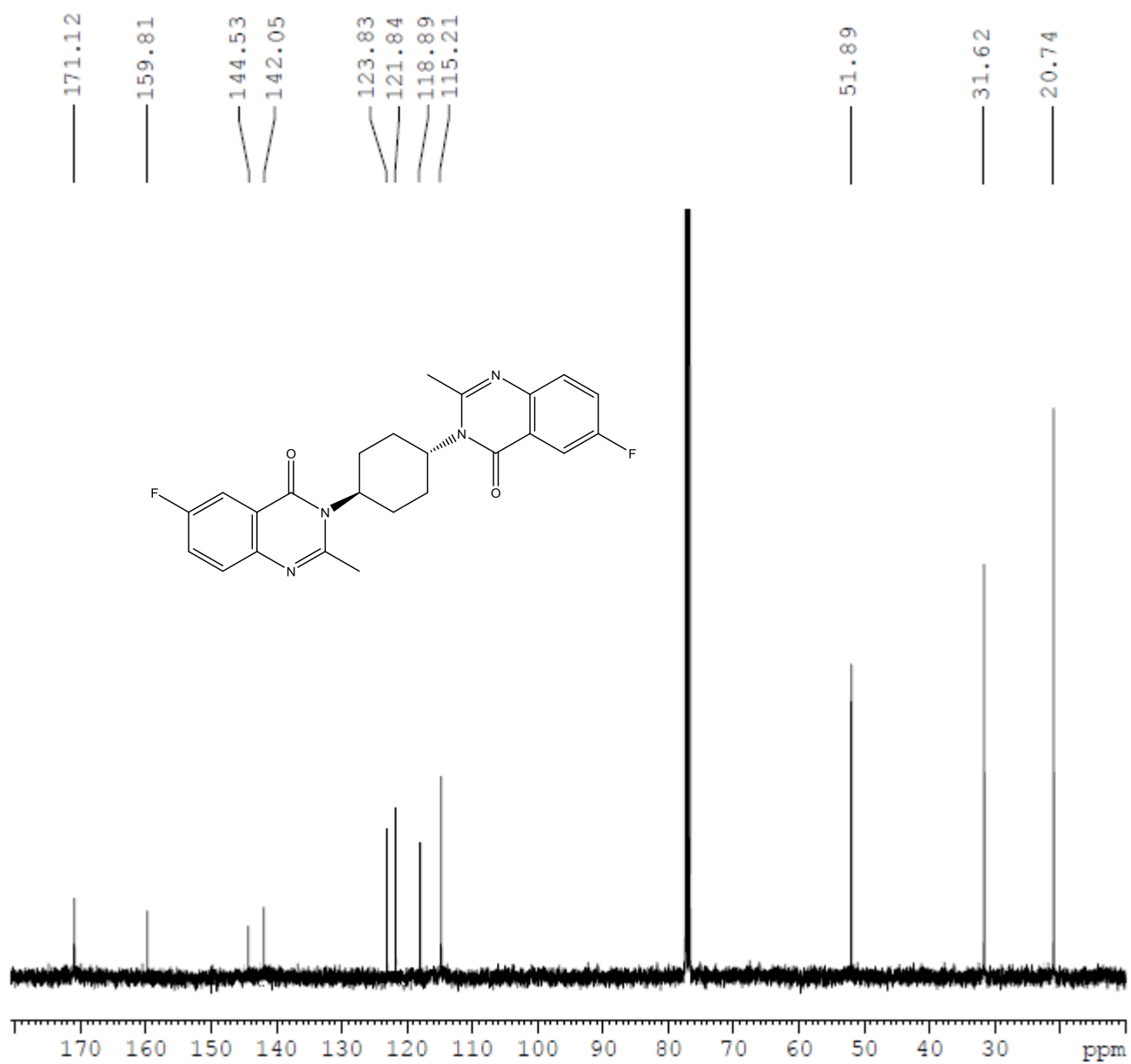
^{13}C NMR of **3c**



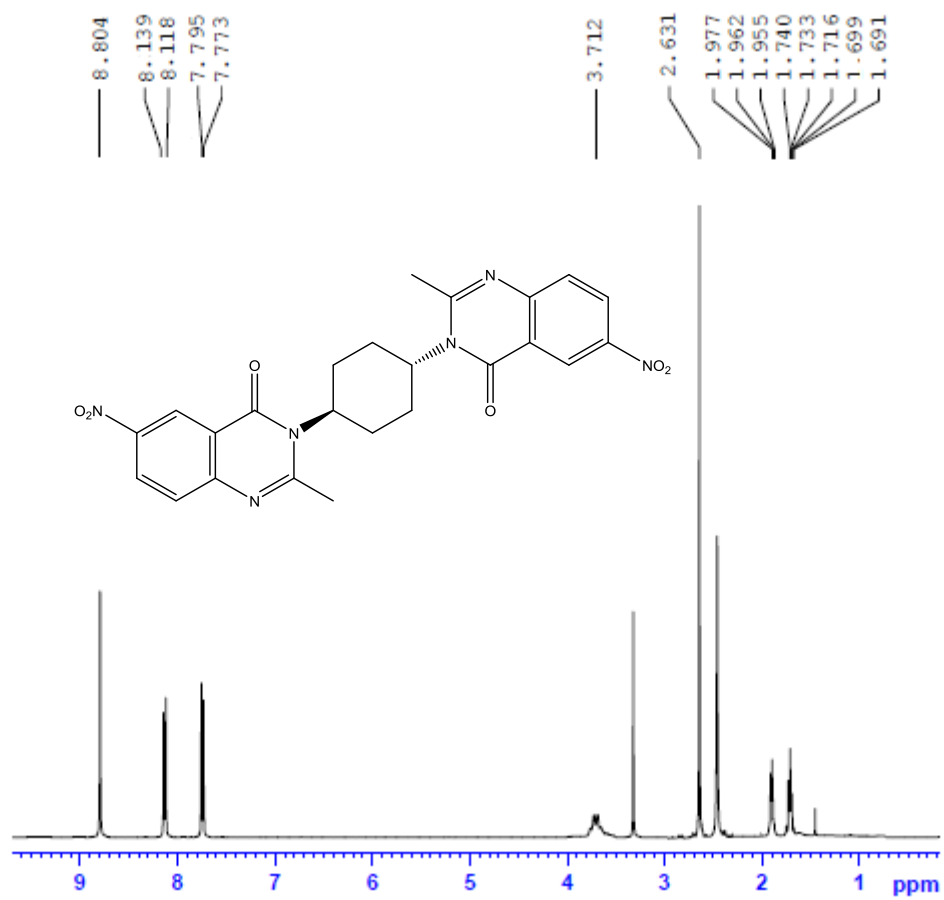
^1H NMR of **3d**



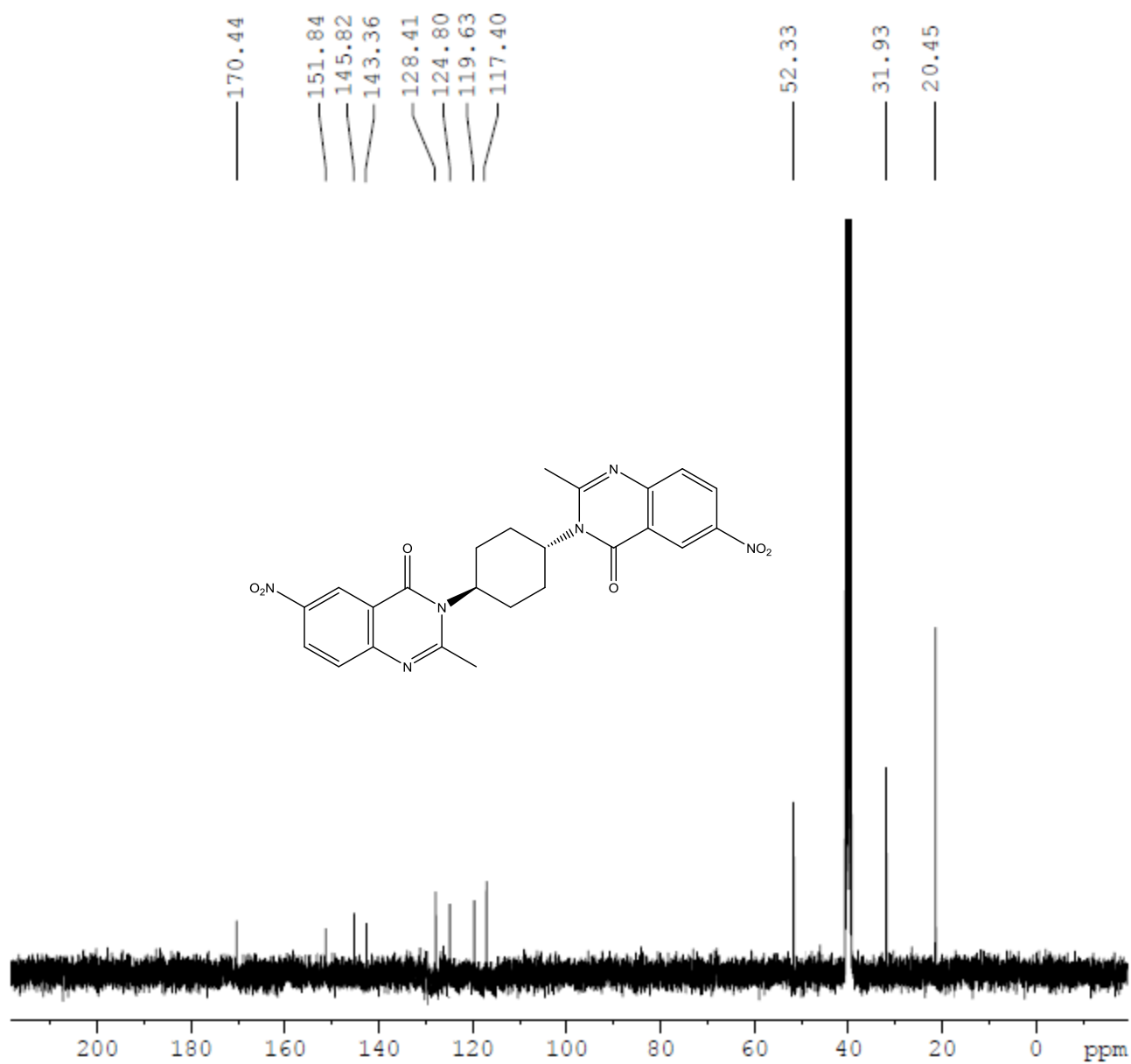
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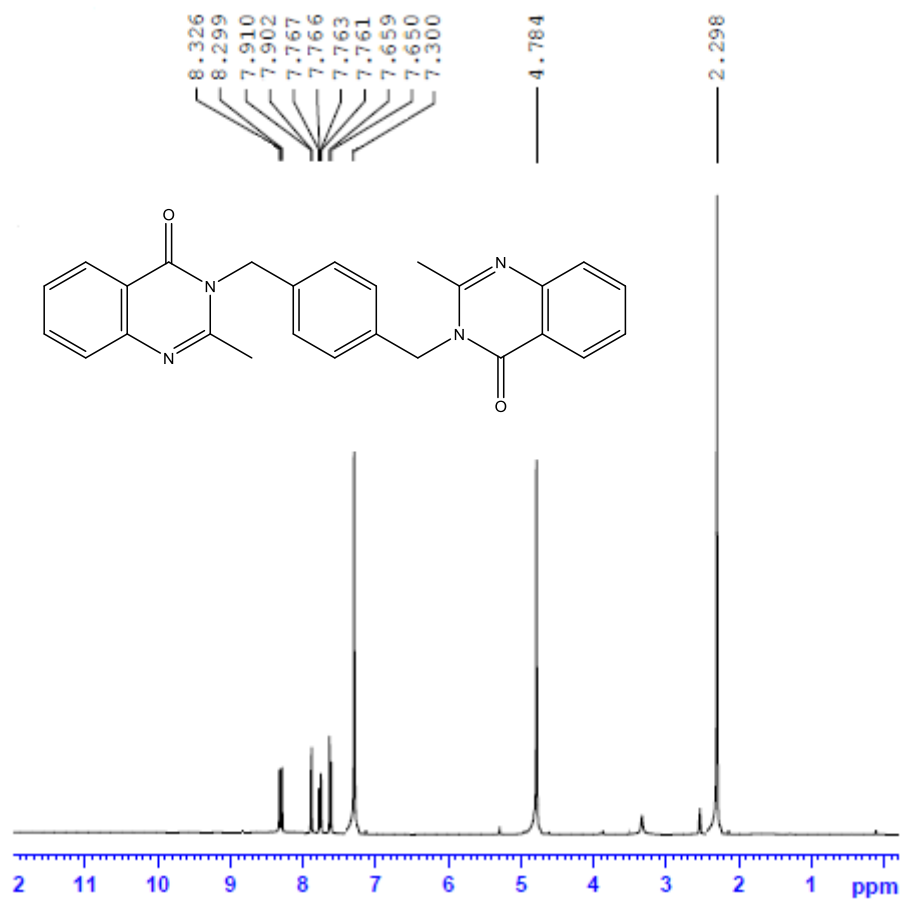
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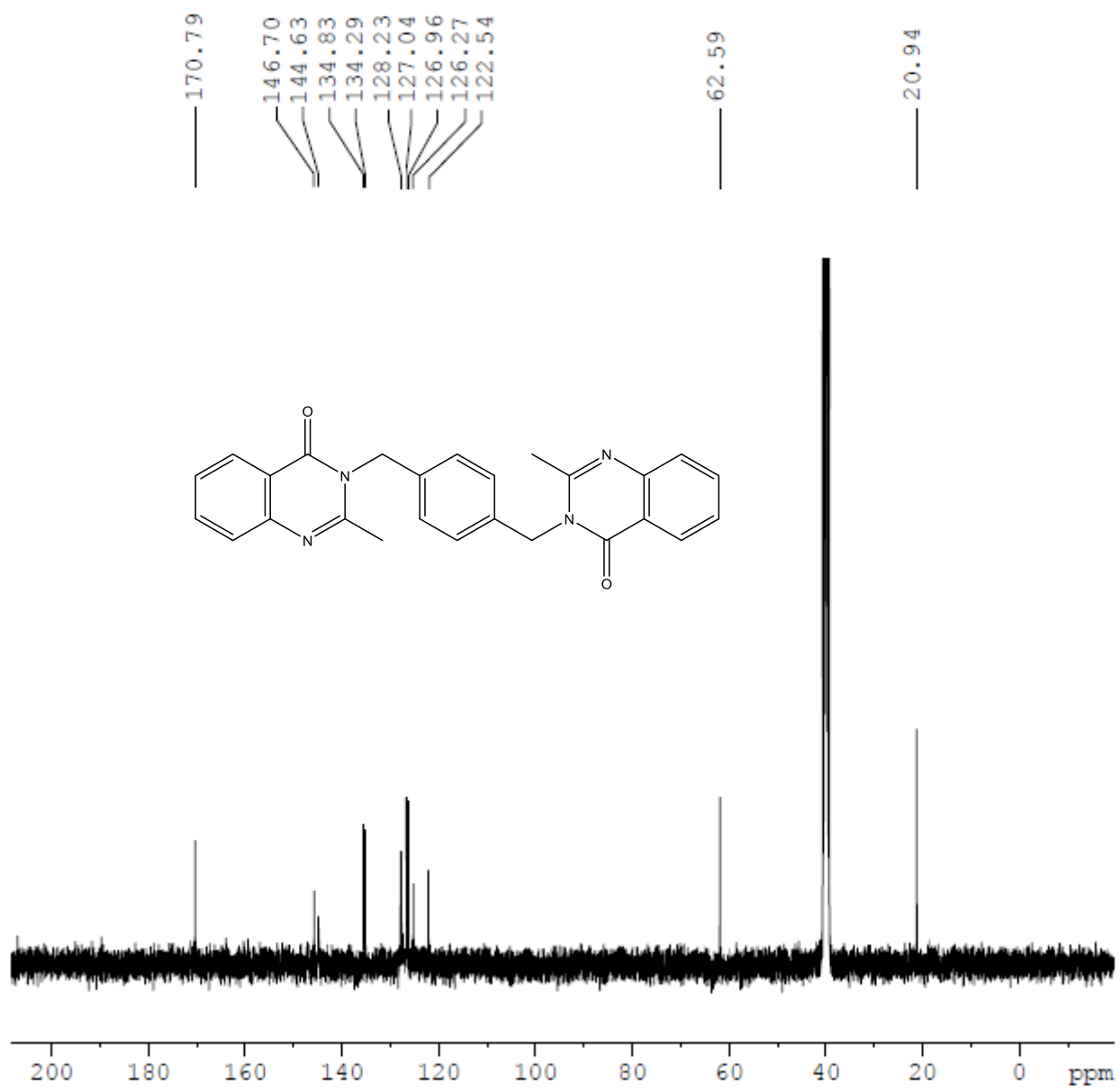
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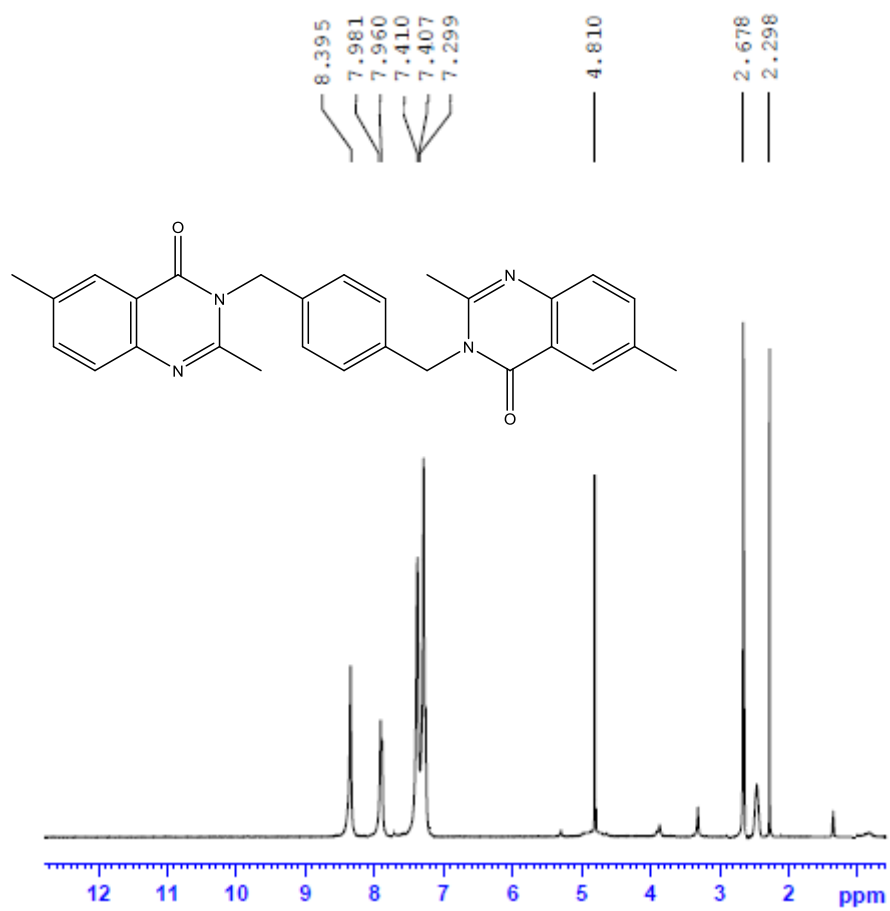
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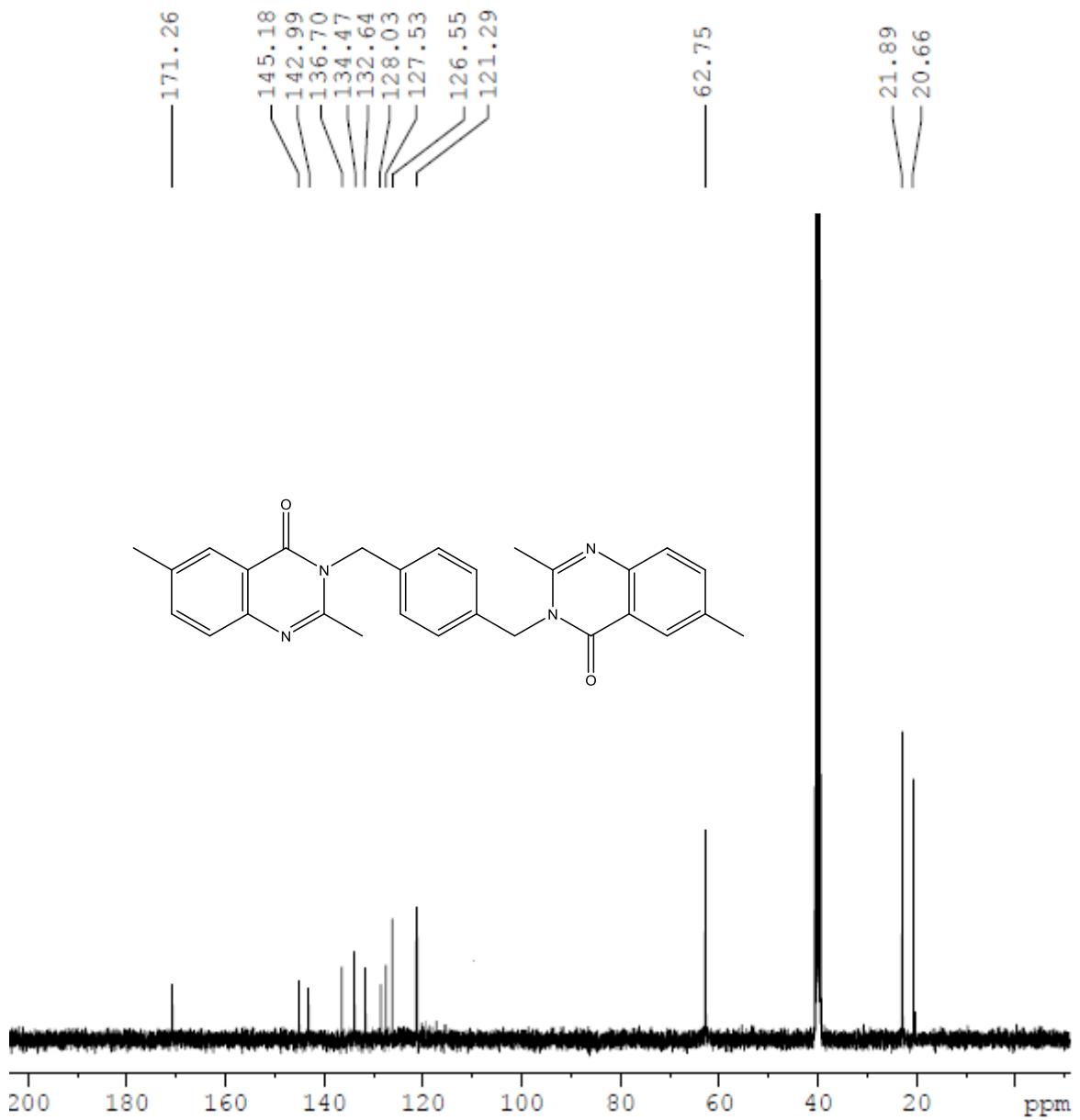
¹H NMR of 4a



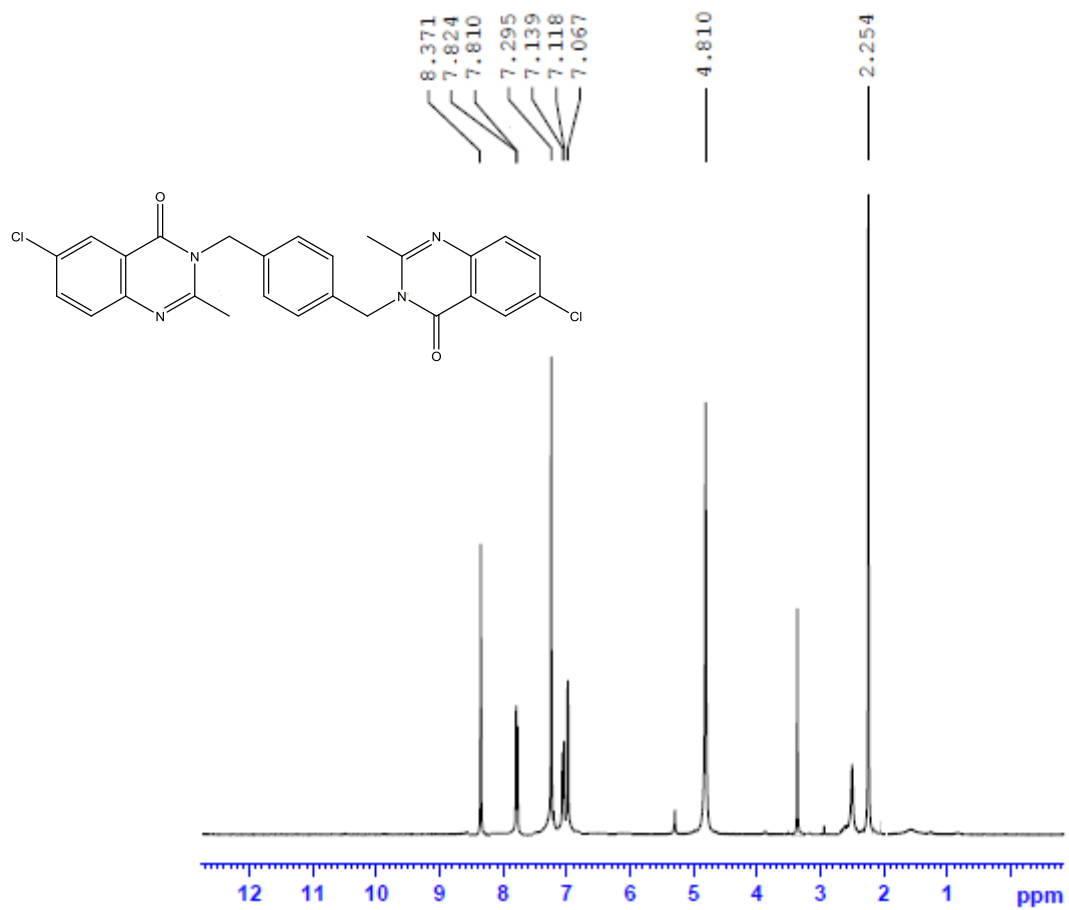
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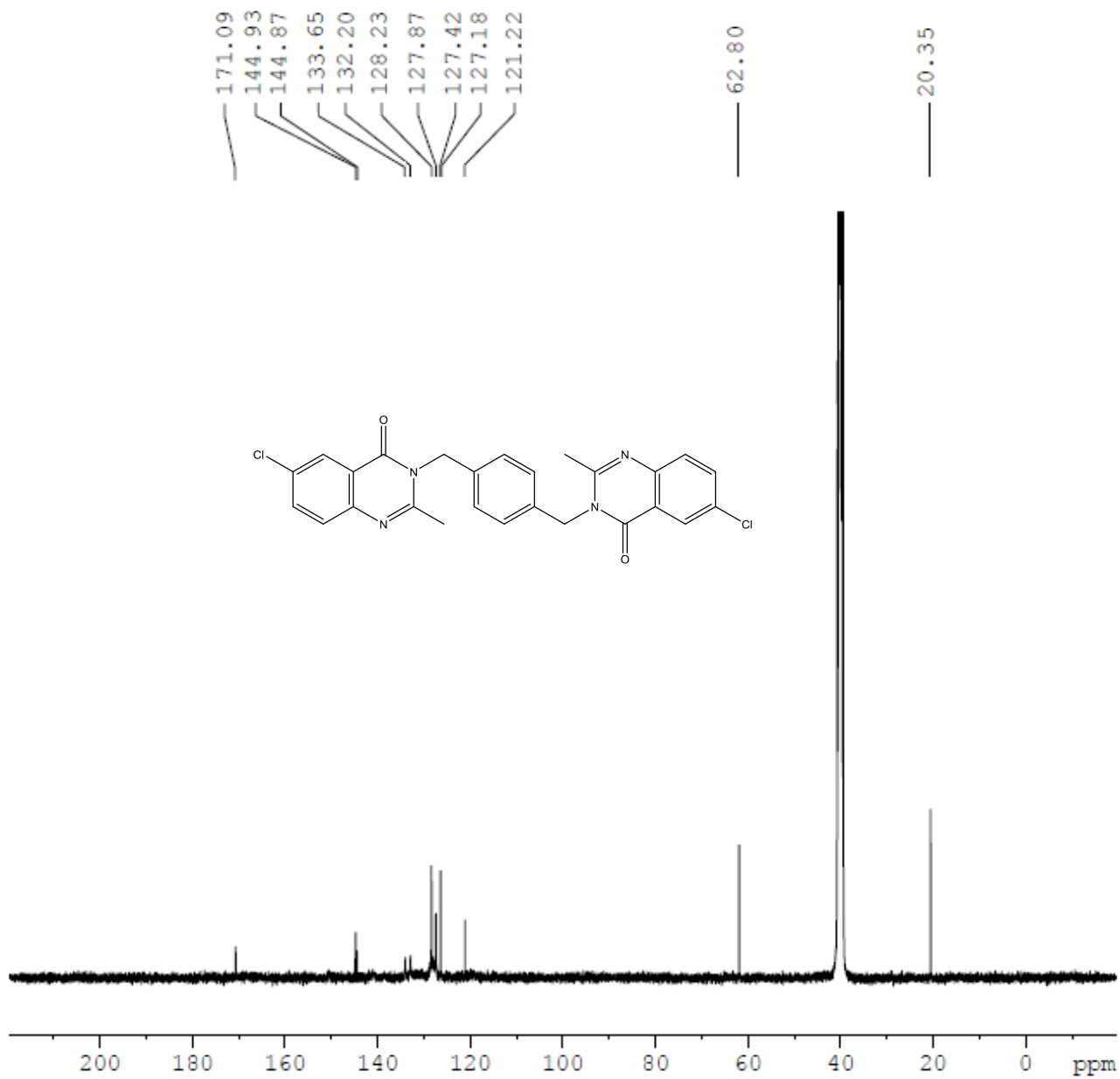
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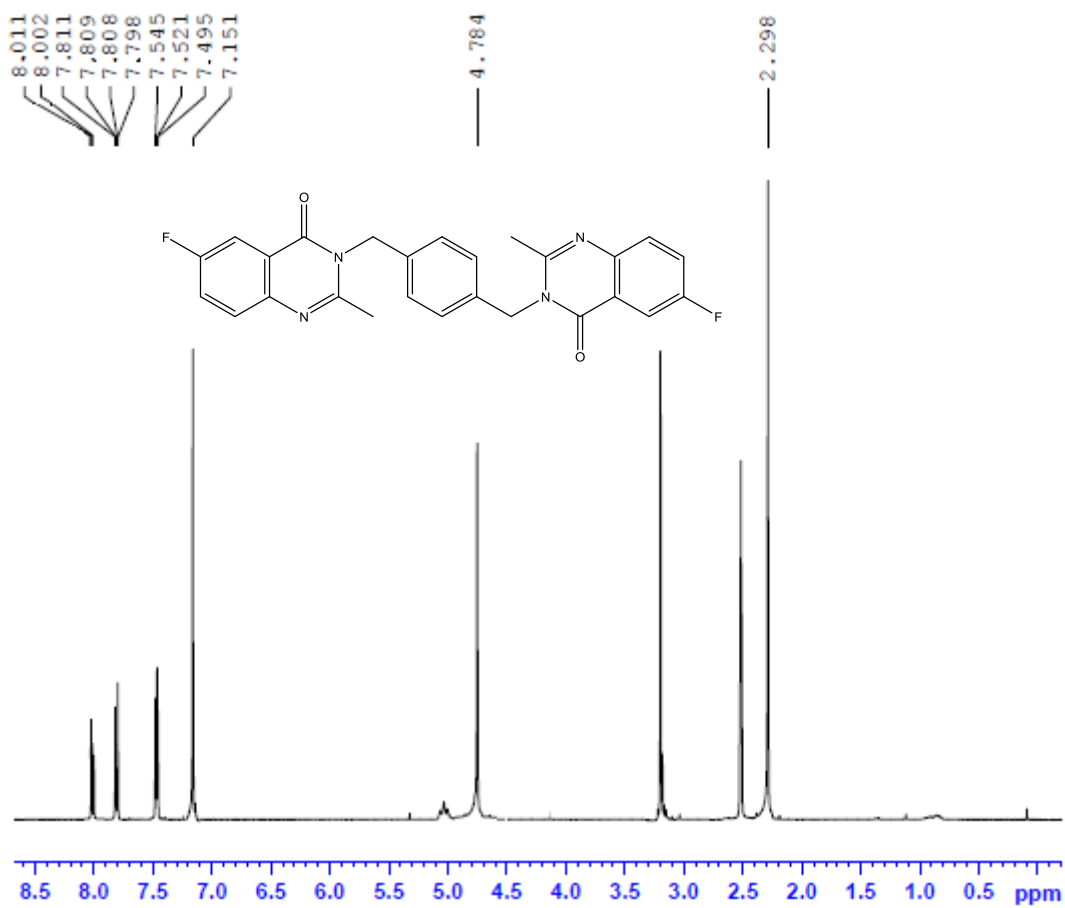
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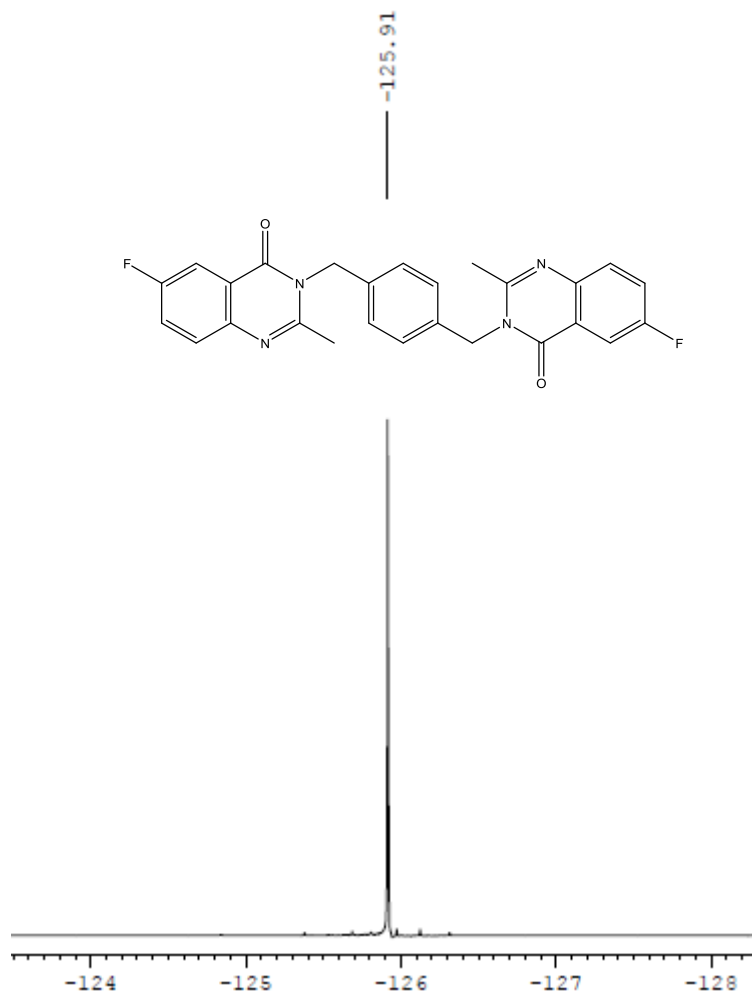
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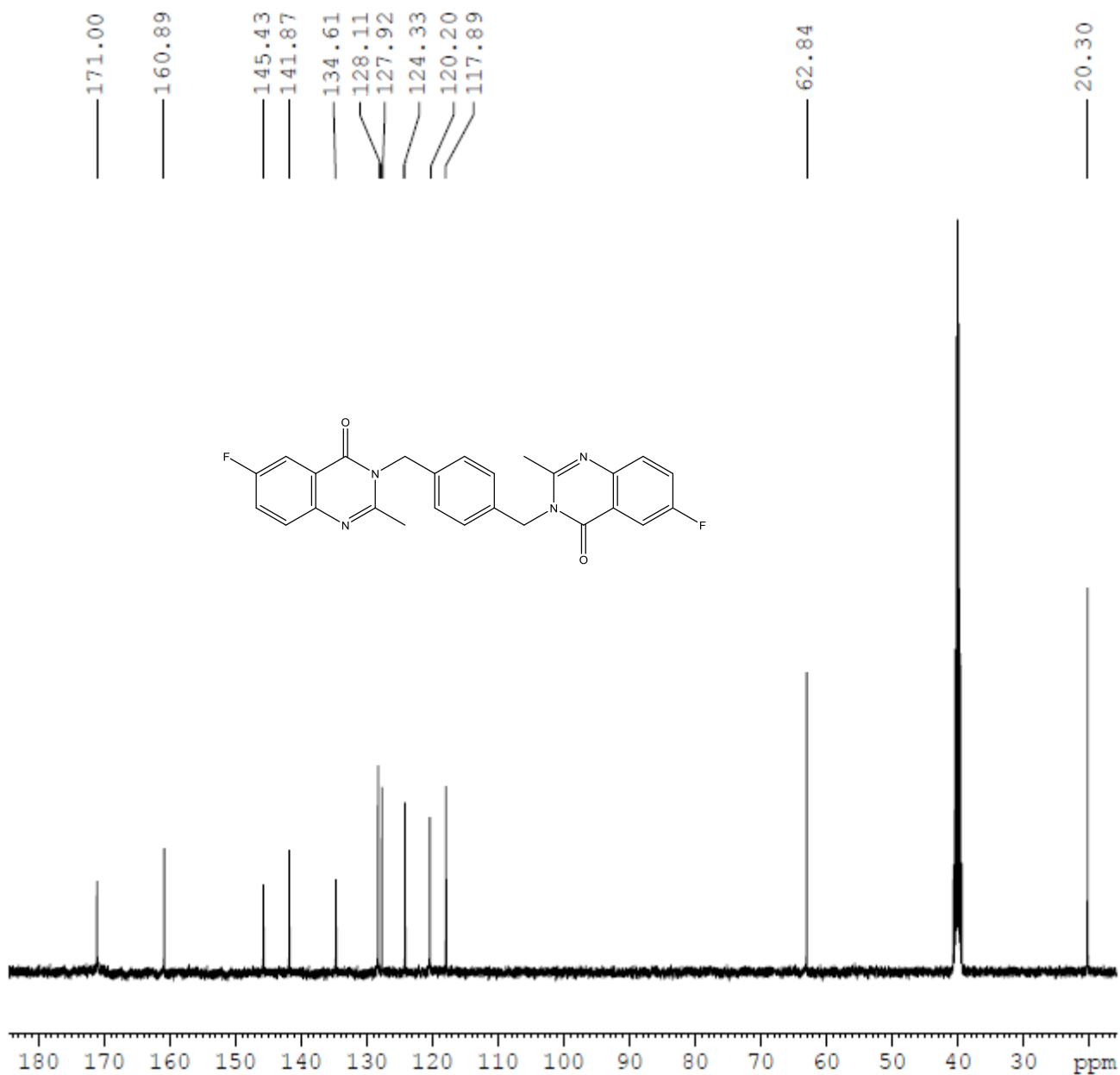
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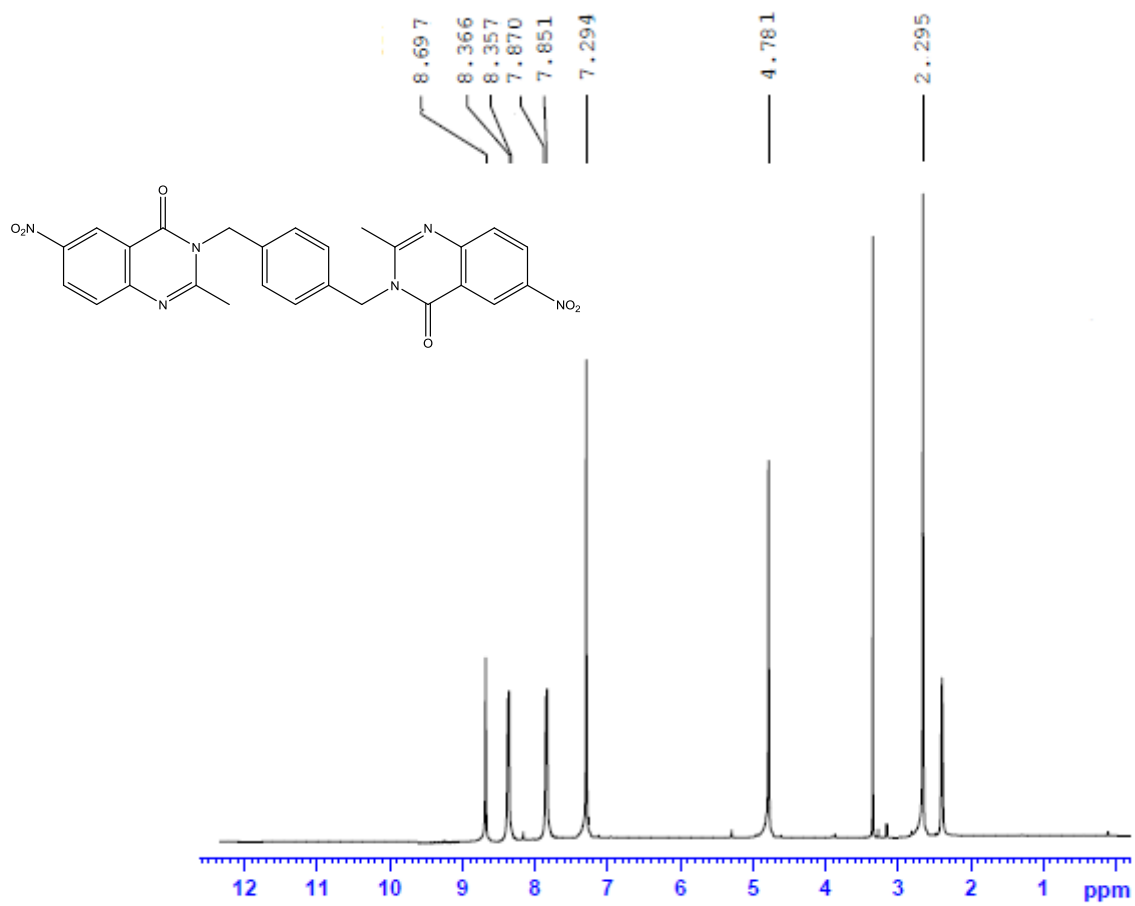
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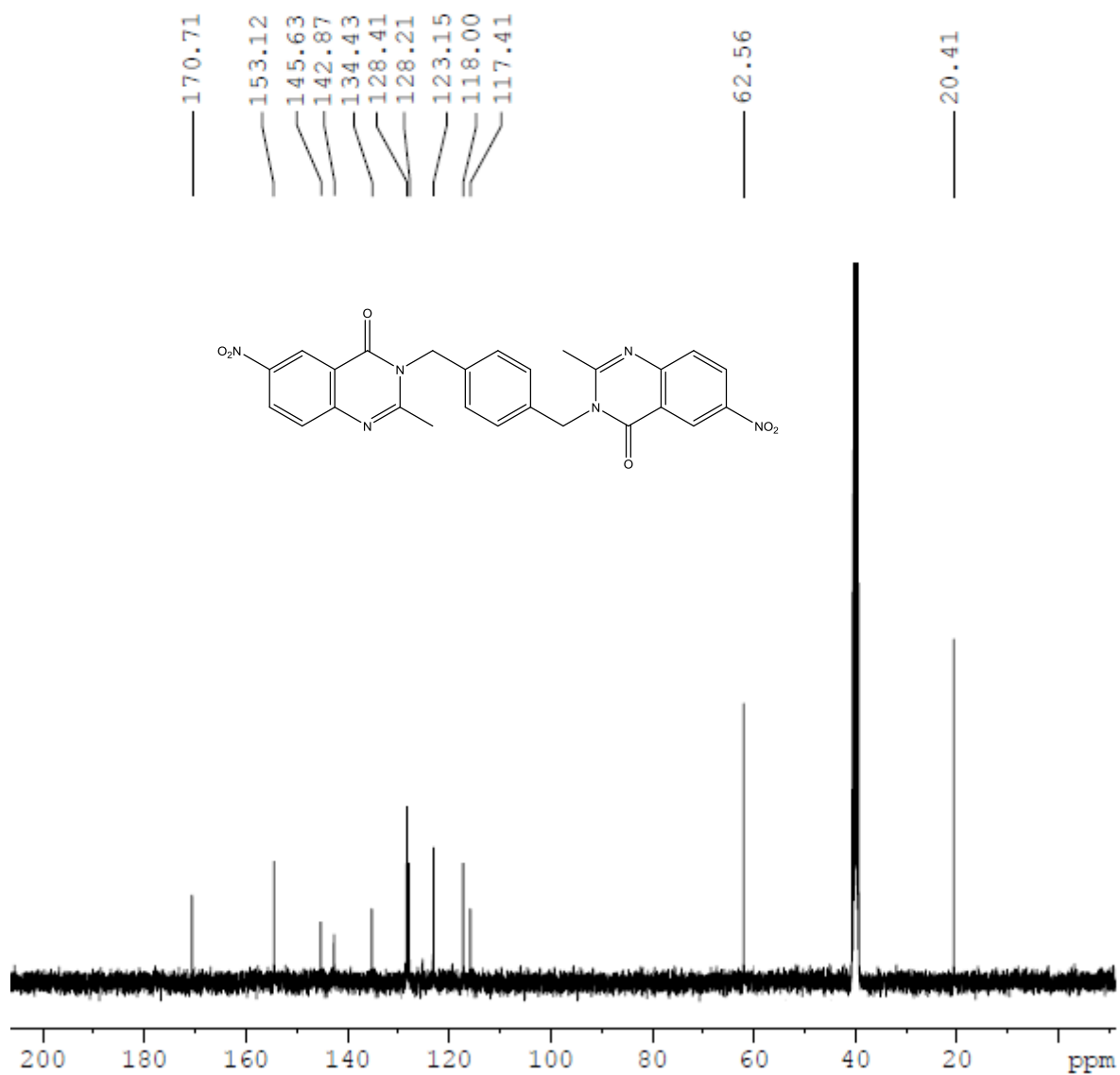
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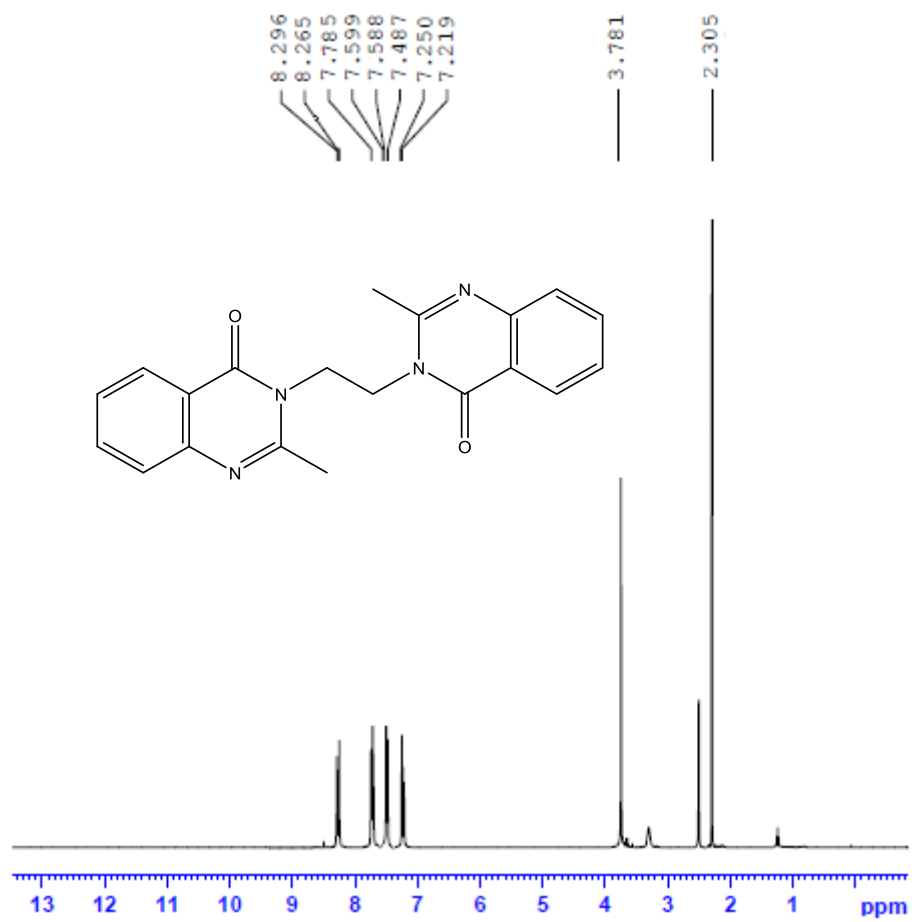
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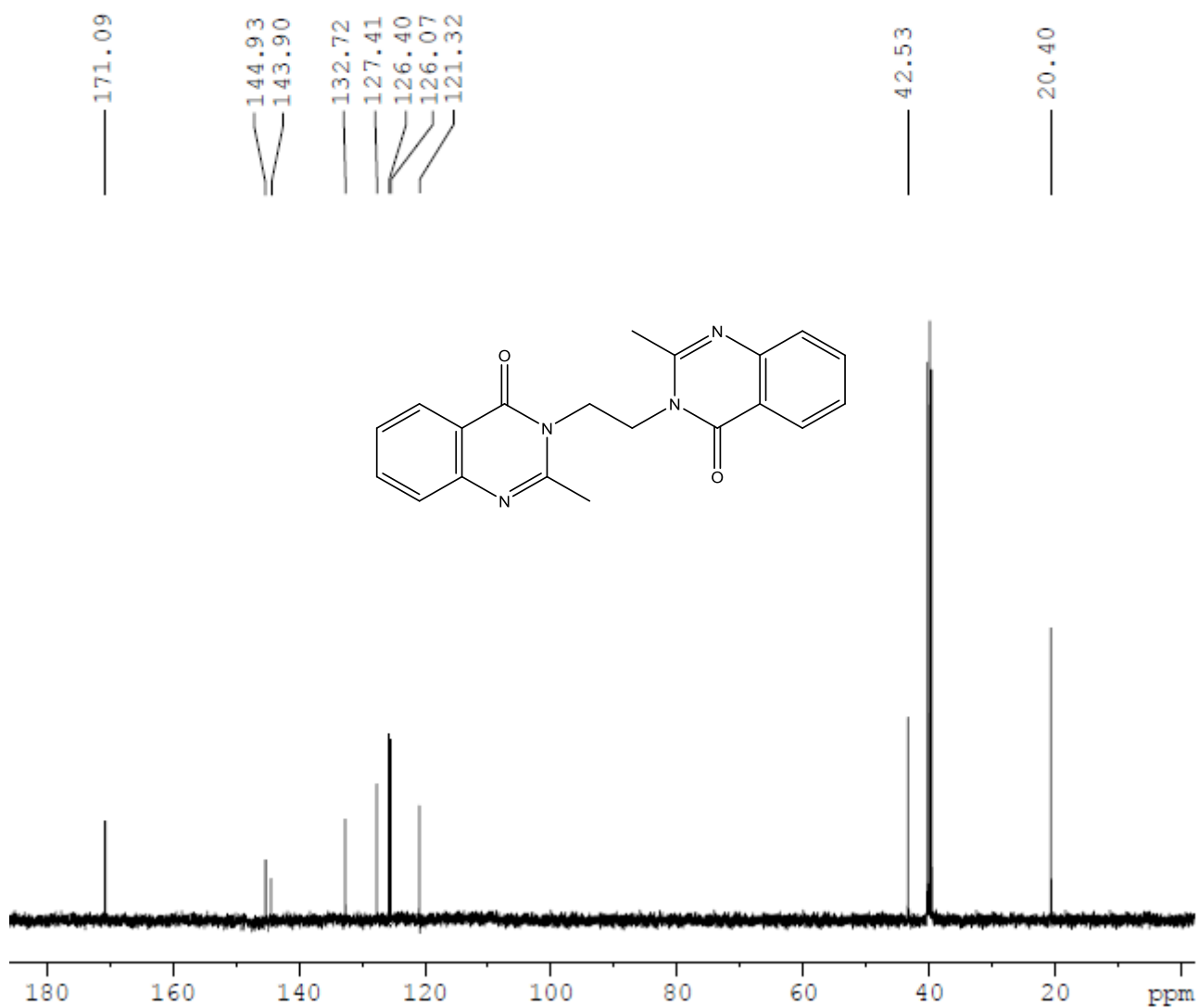
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^{13}C NMR of **4e**



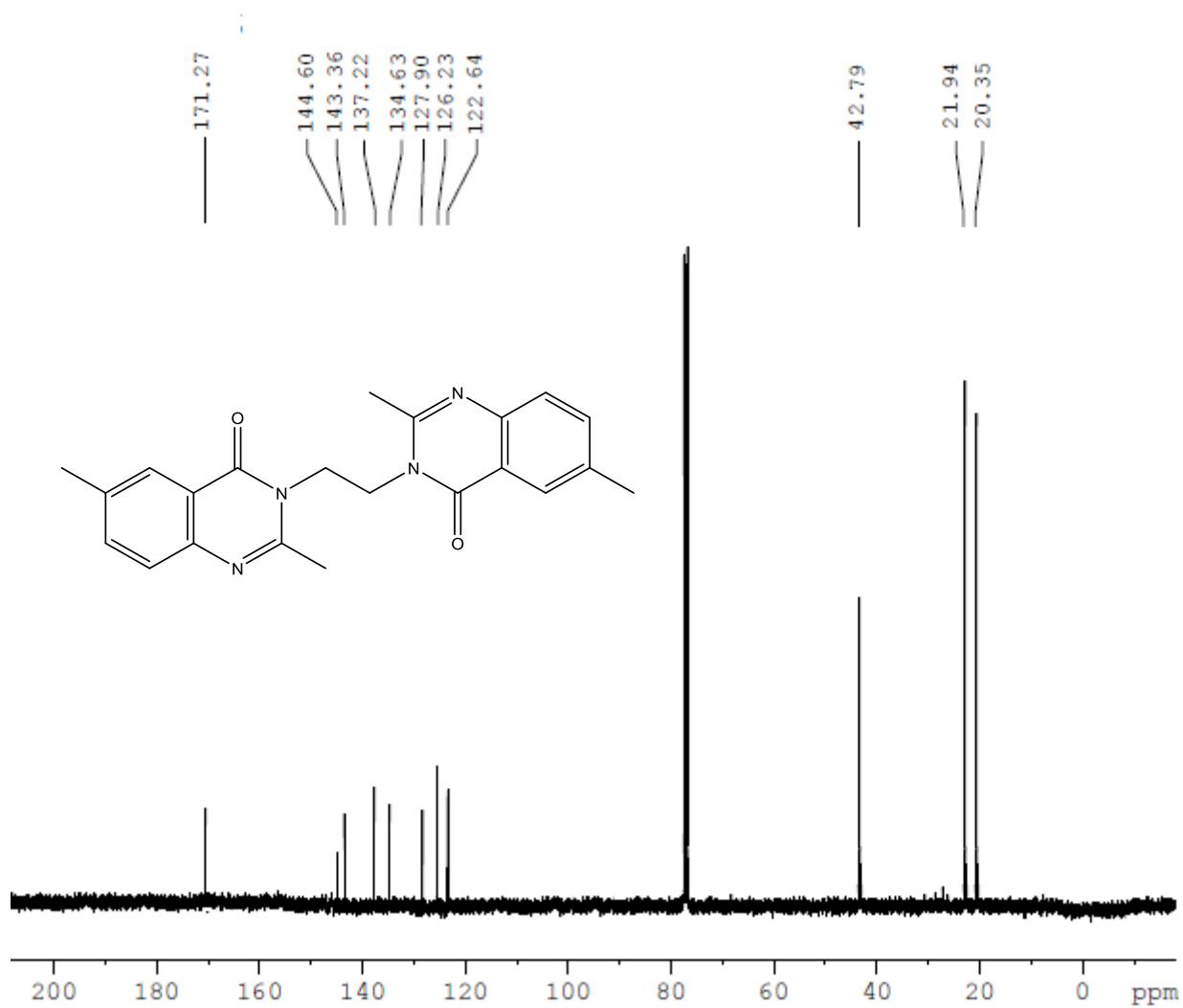
¹H NMR of **5a**



^{13}C NMR of **5a**



^1H NMR of **5b**

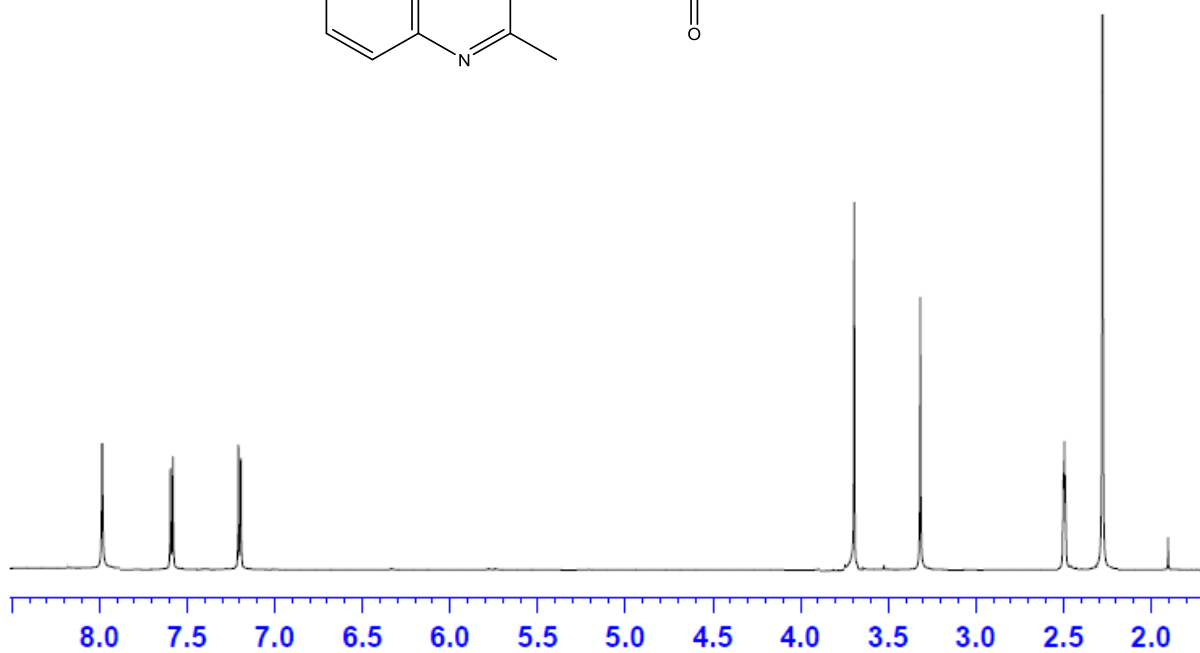
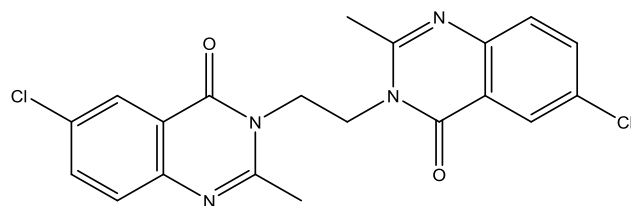


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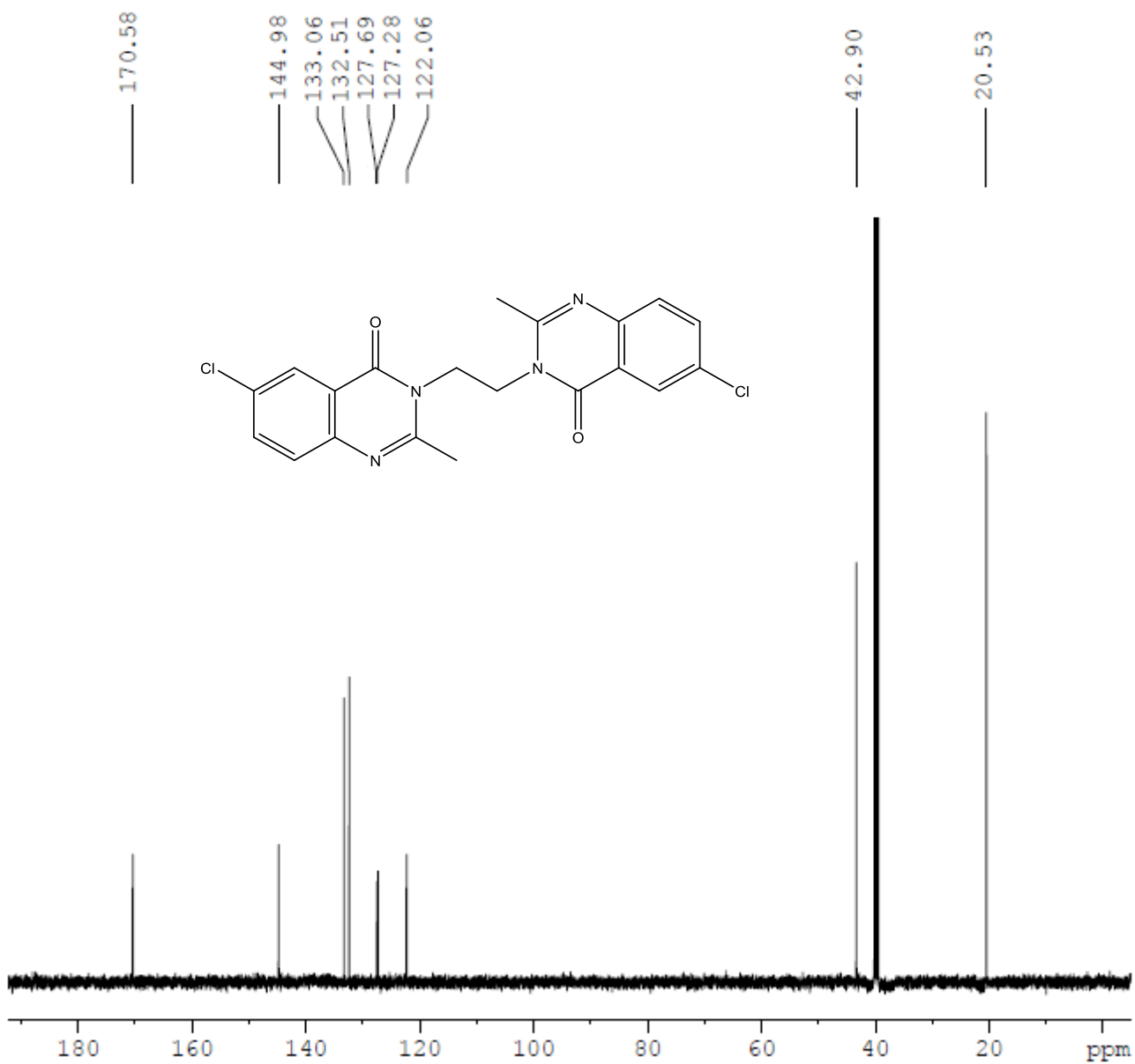
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2.295



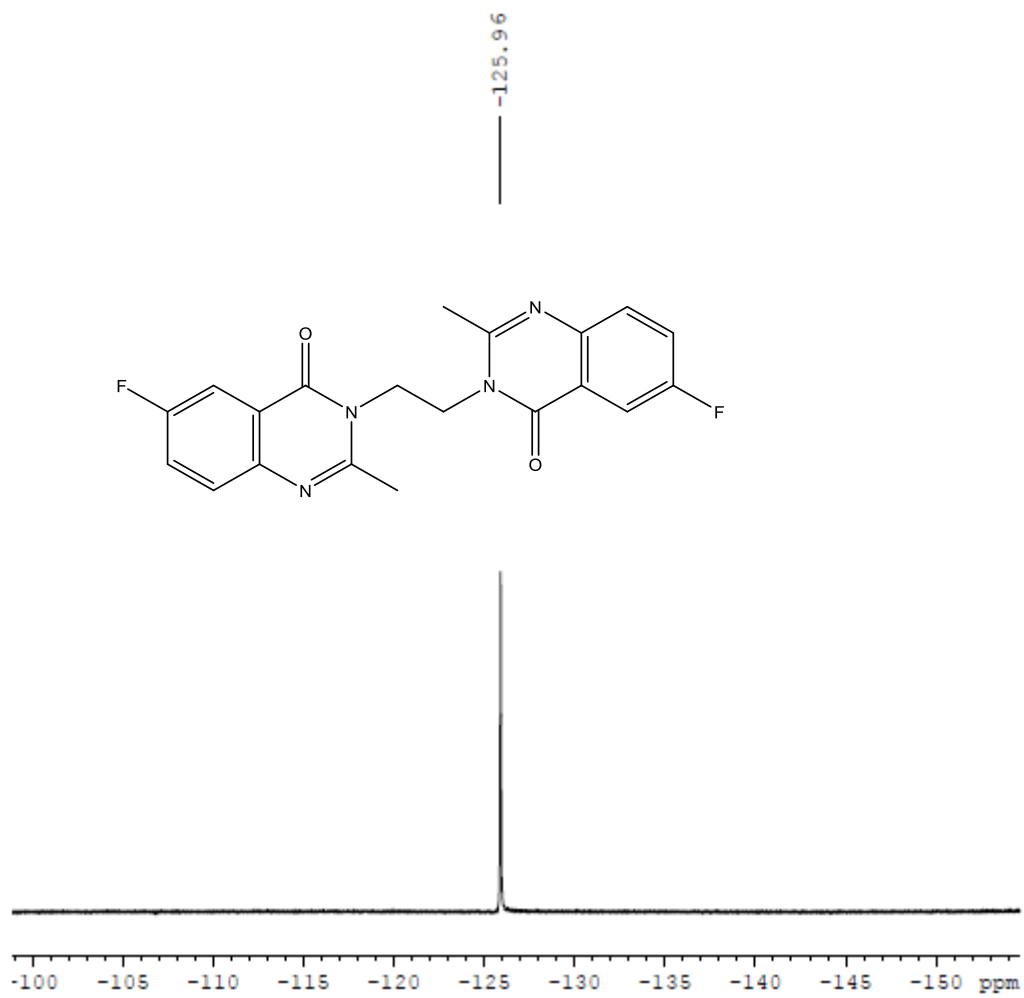
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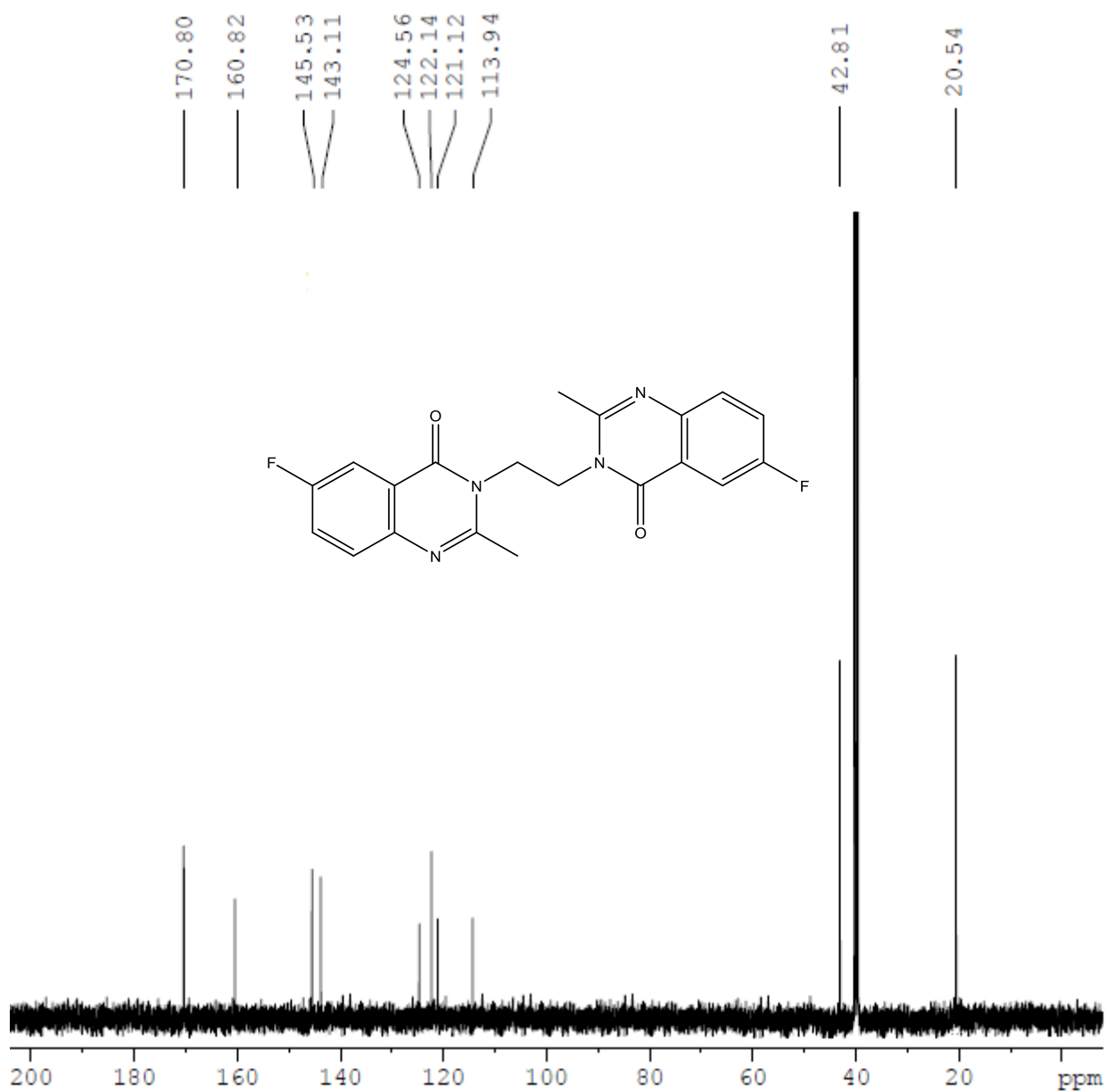
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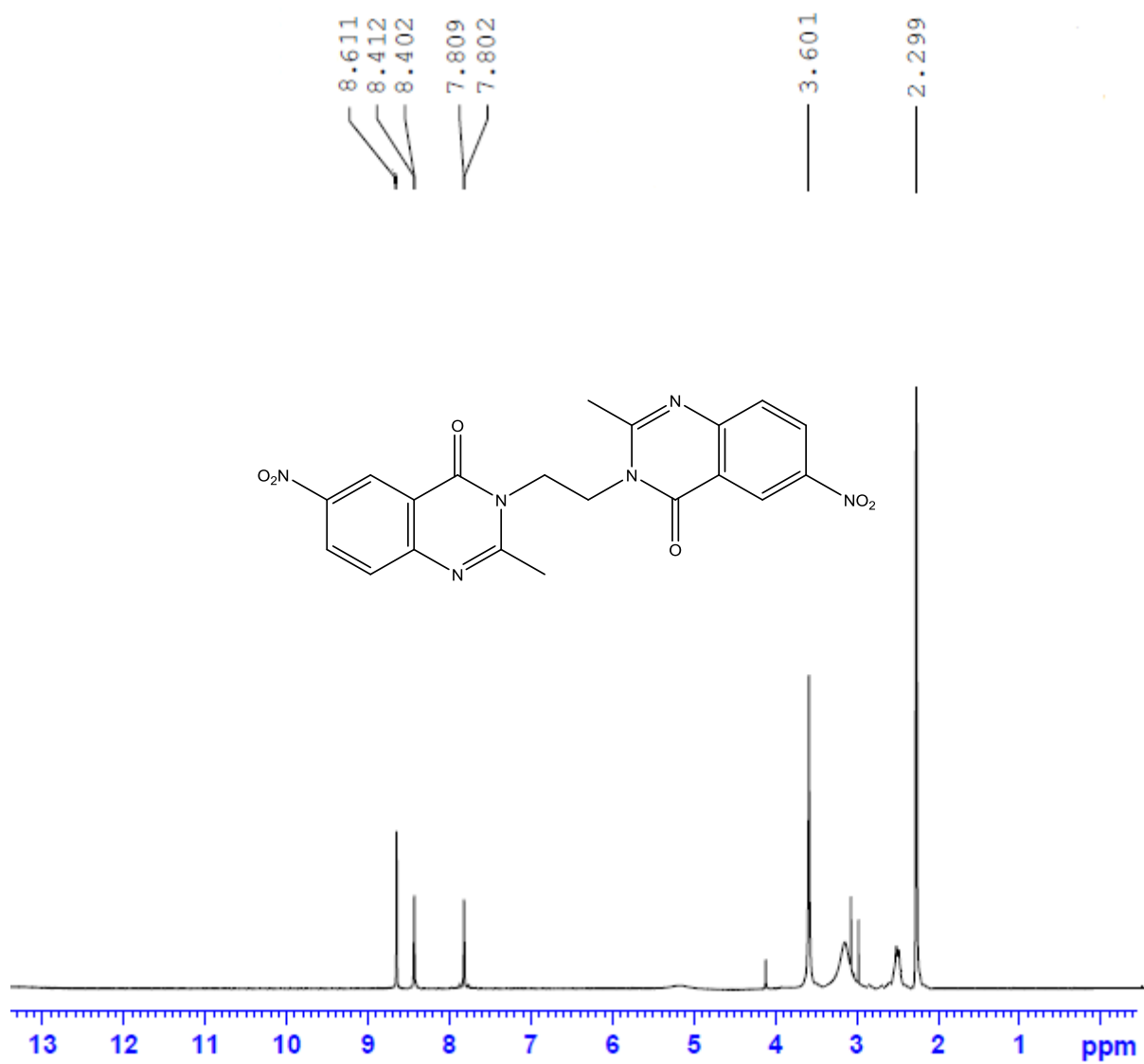
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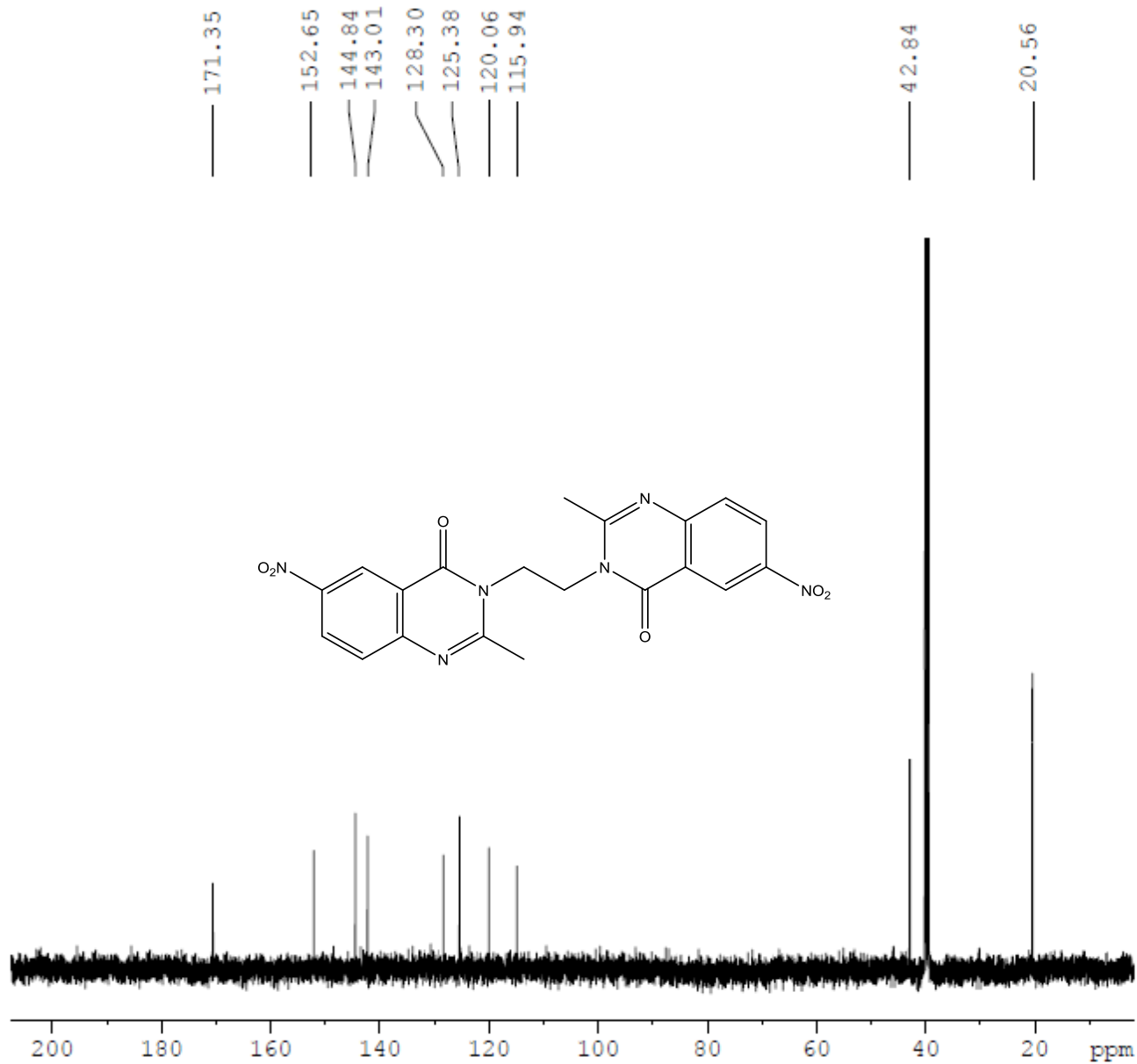
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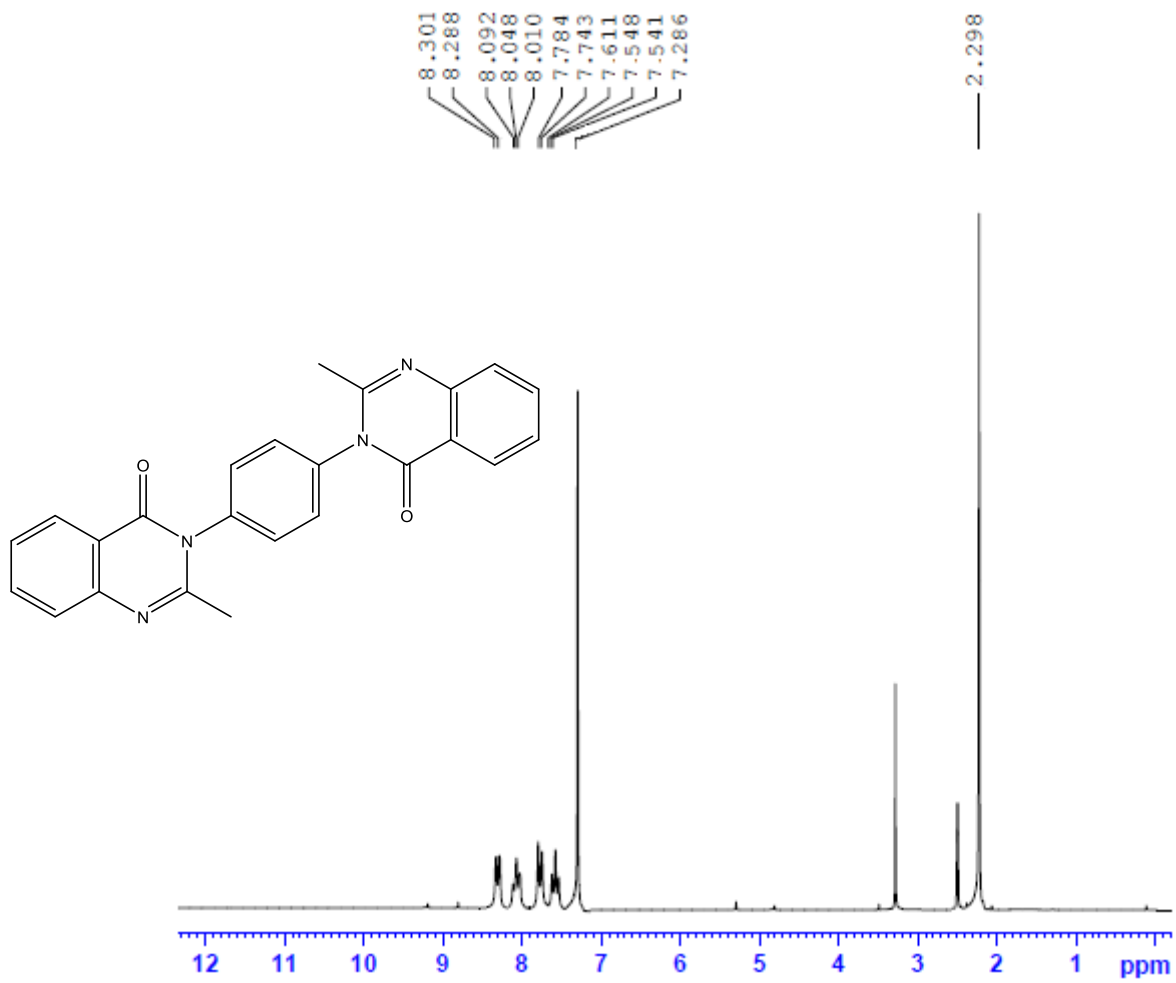
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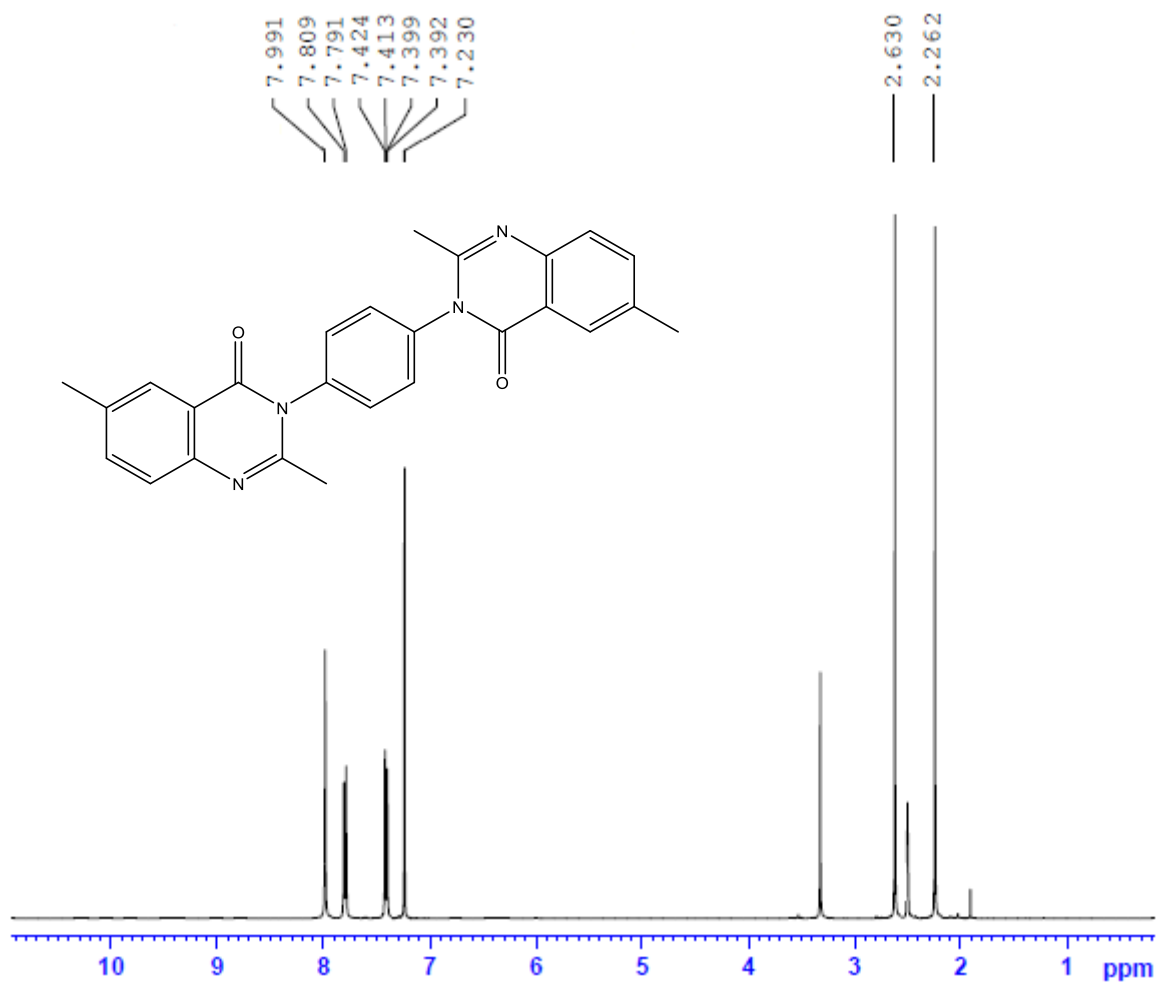
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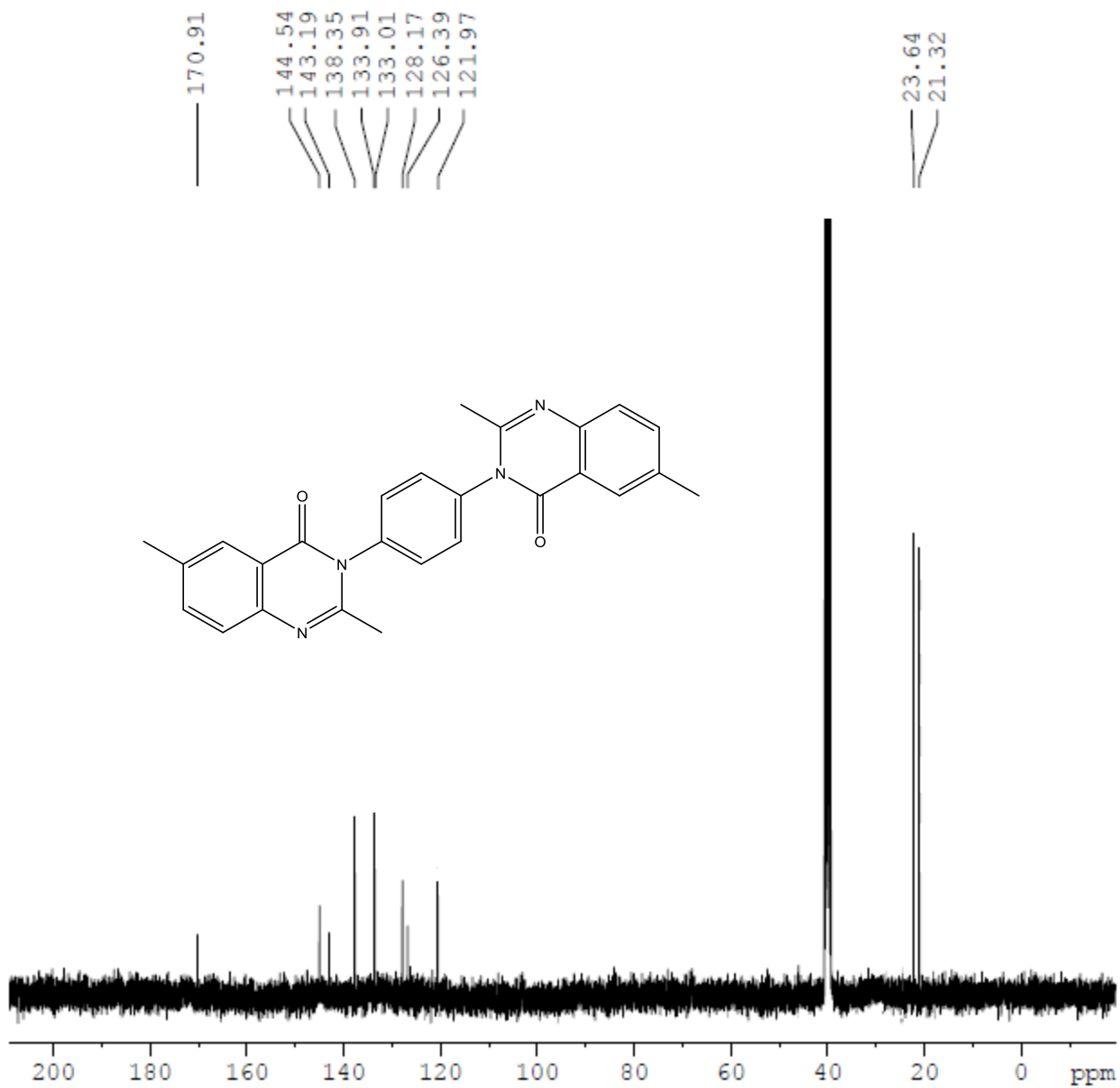
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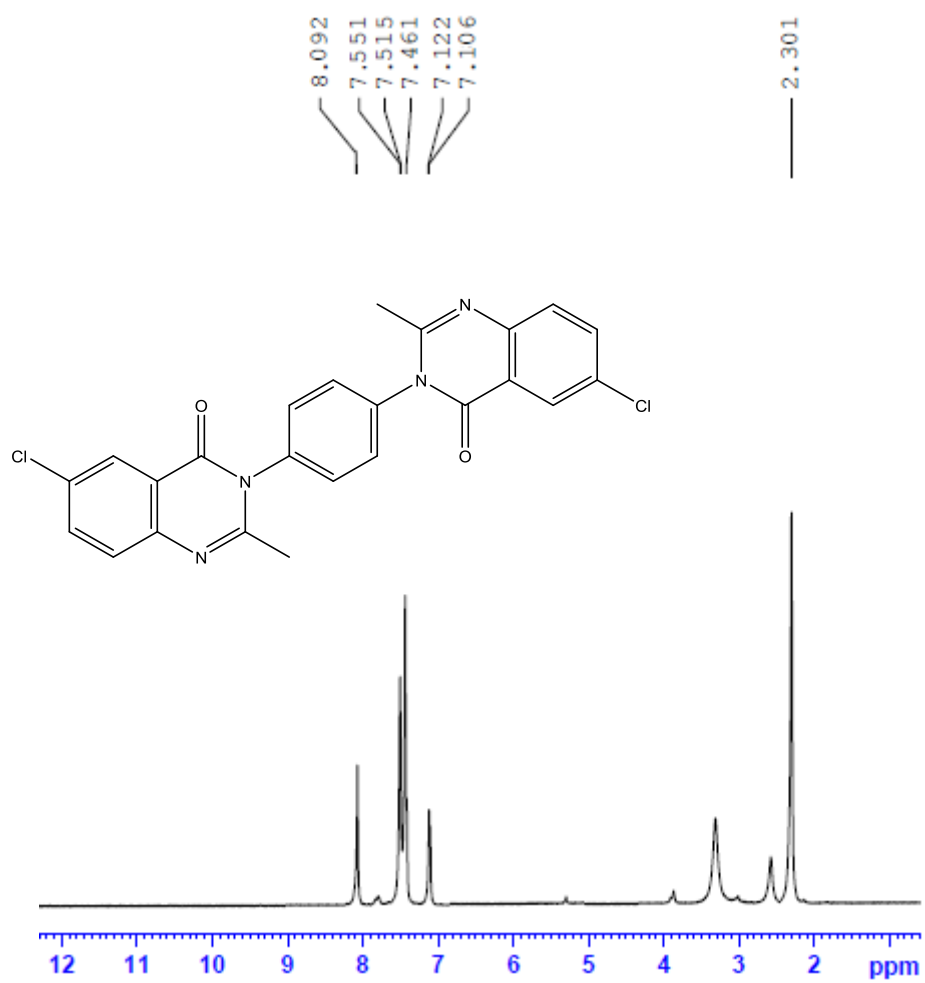
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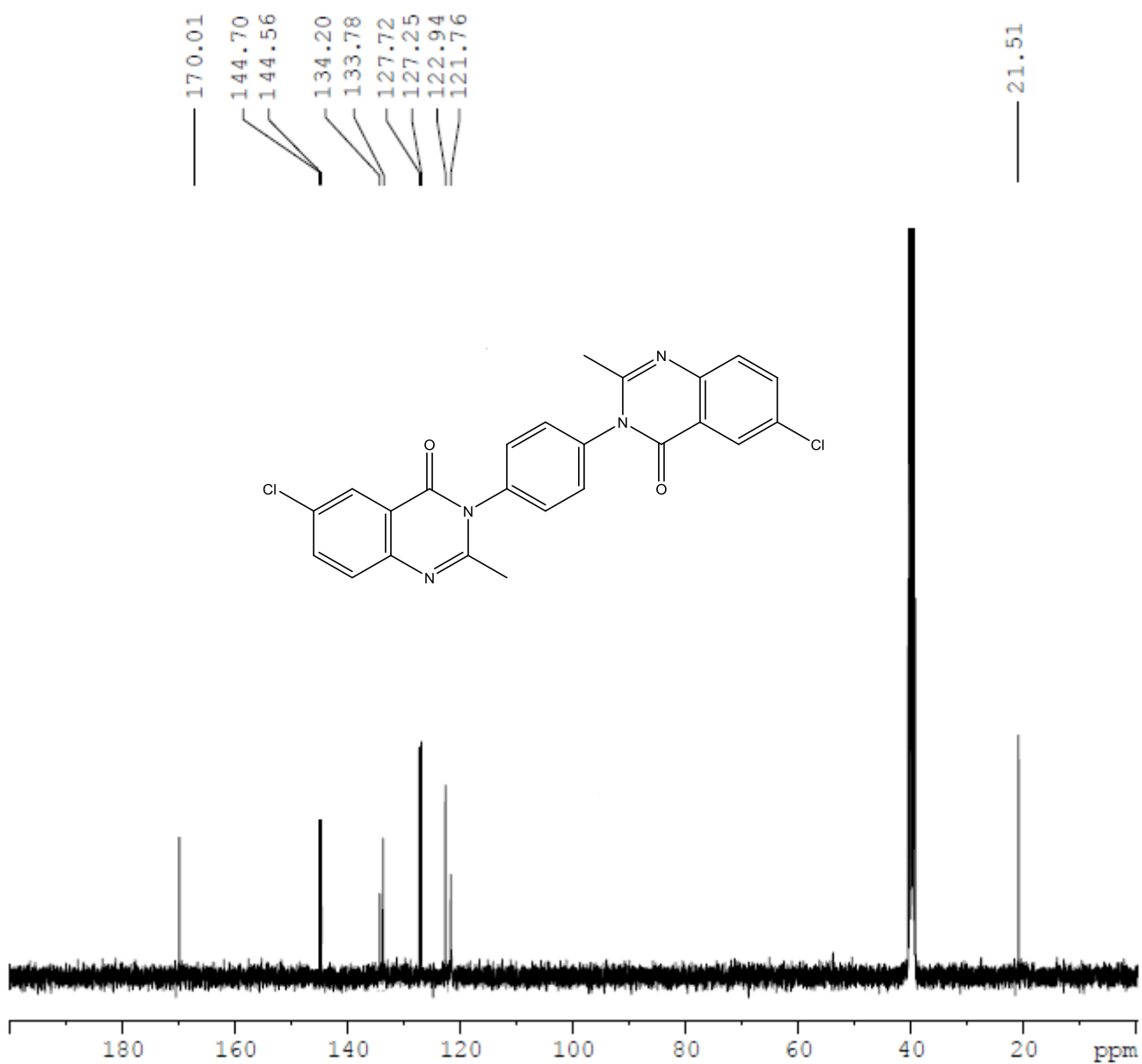
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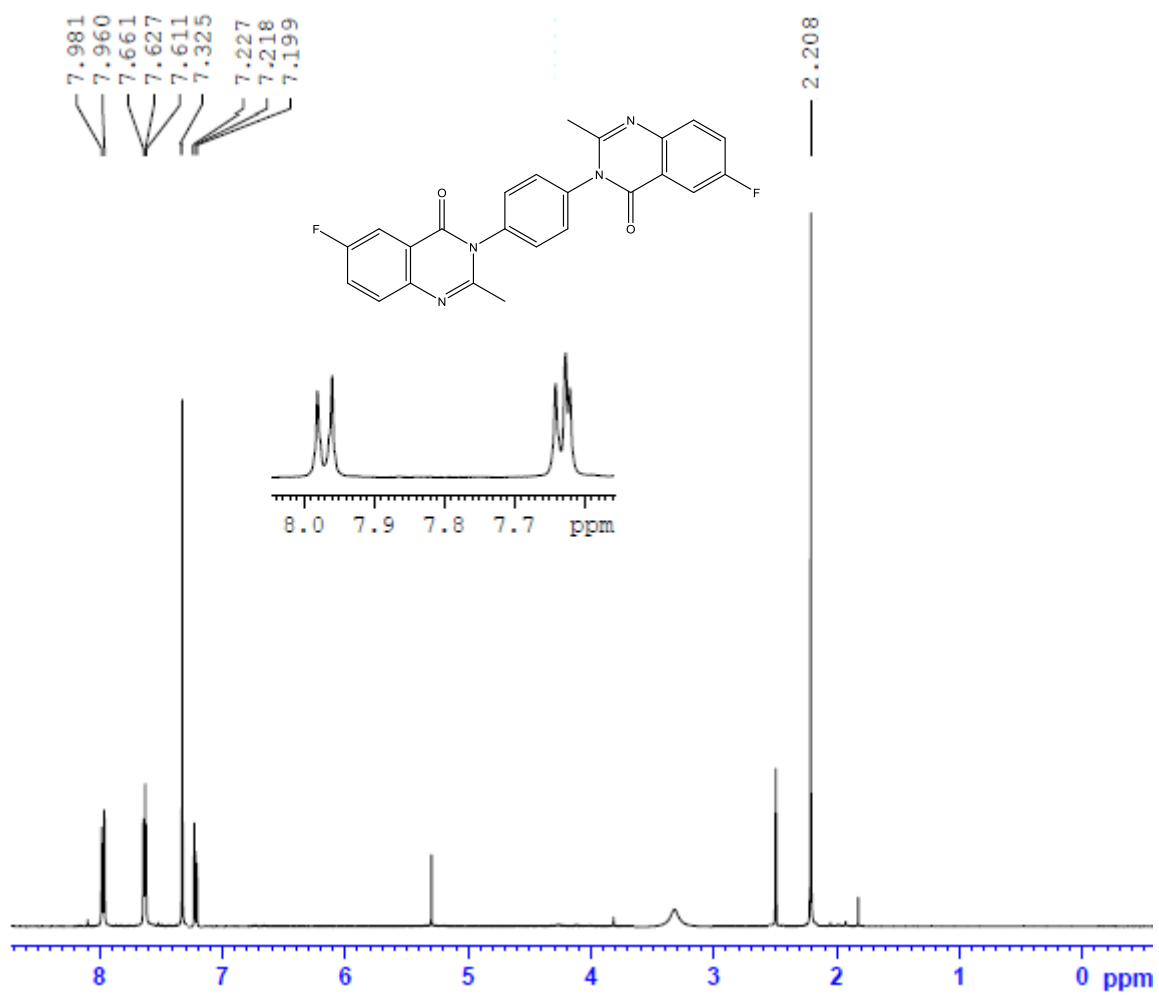
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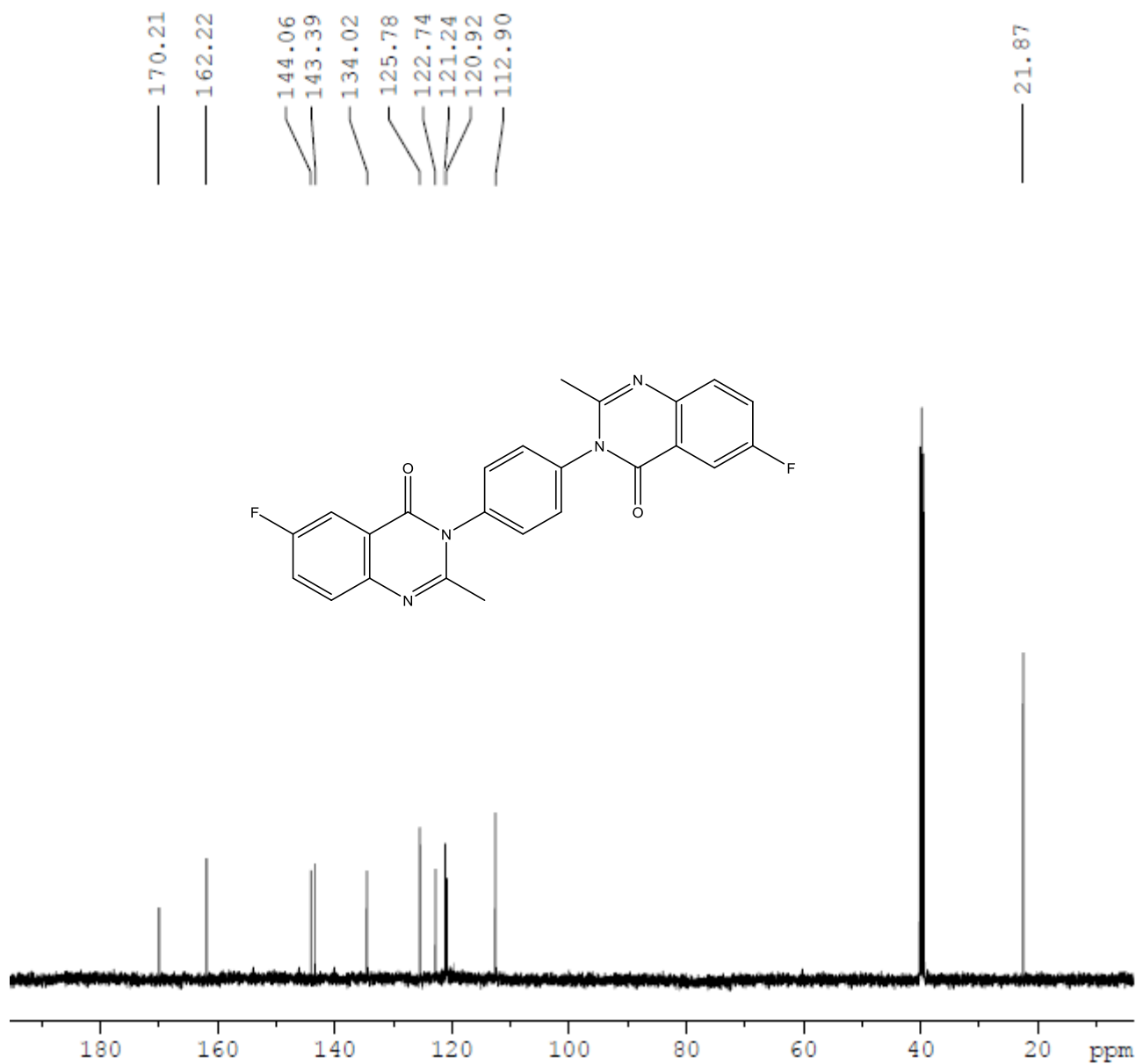
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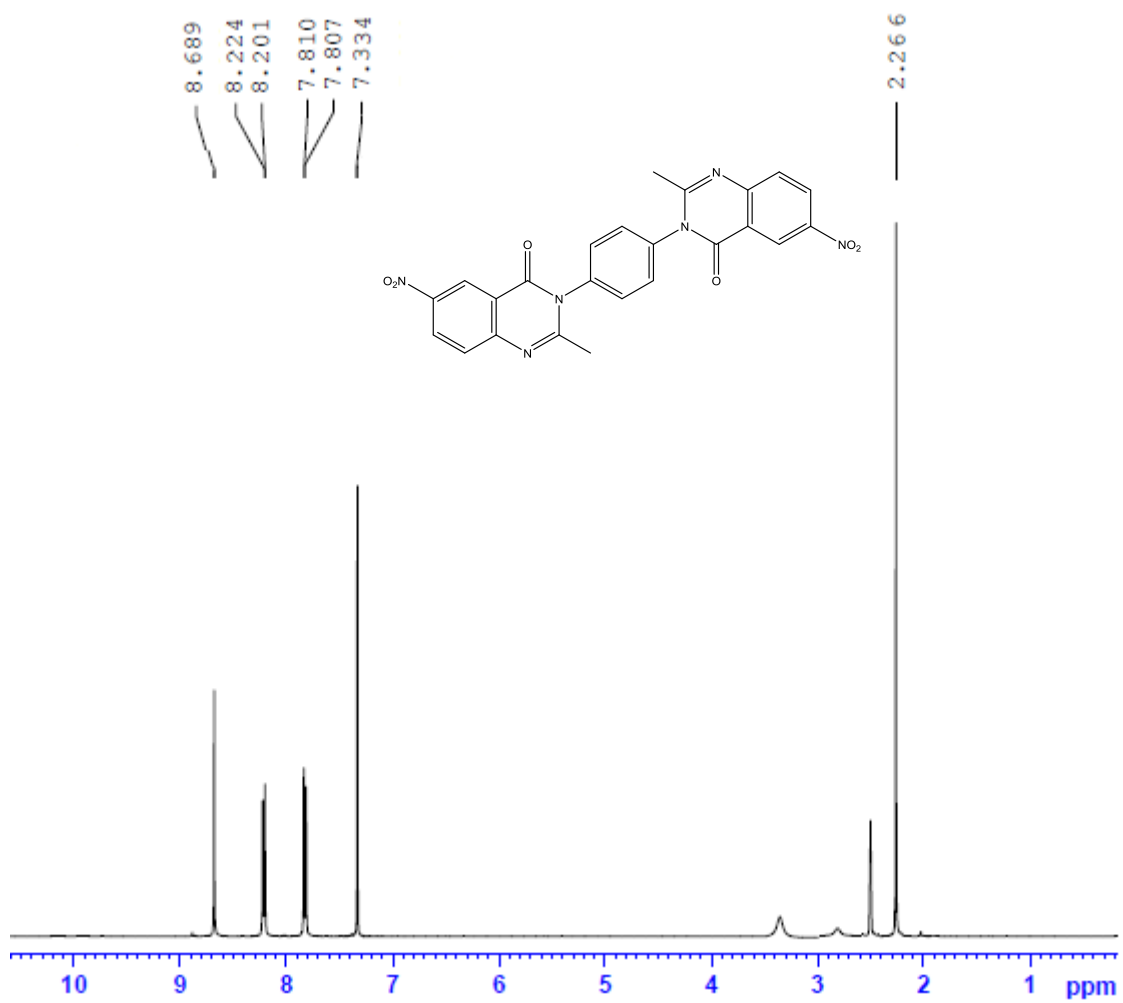
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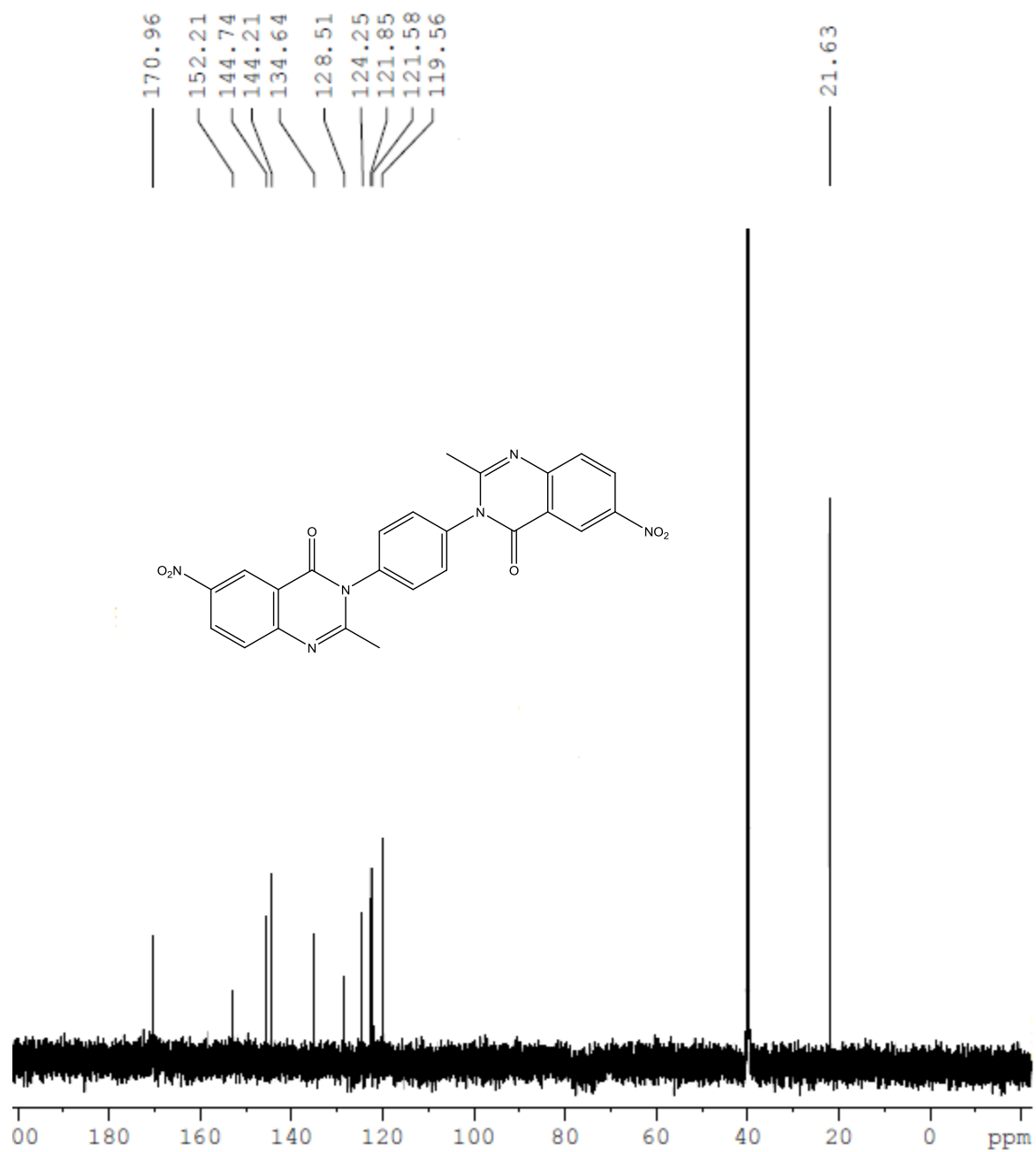
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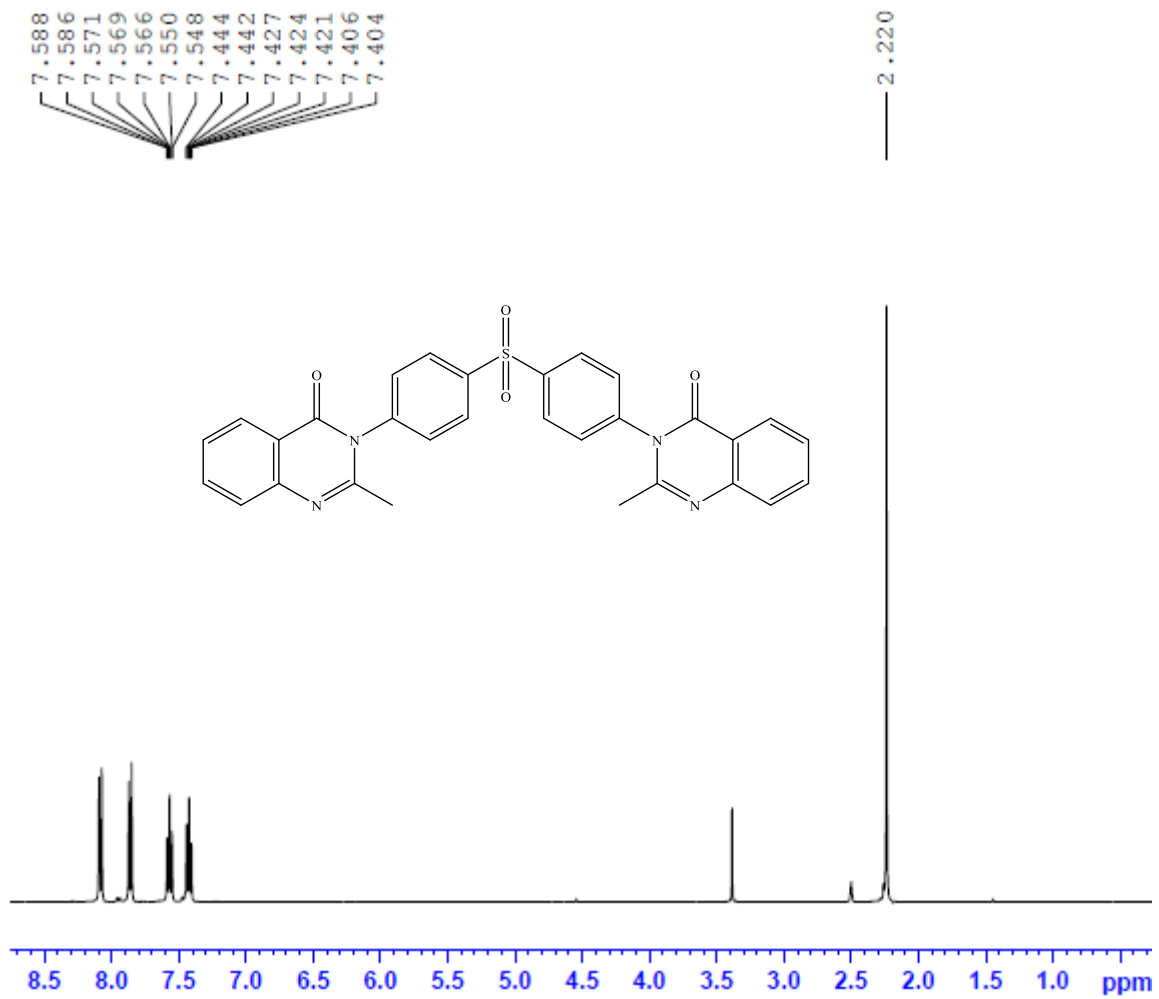
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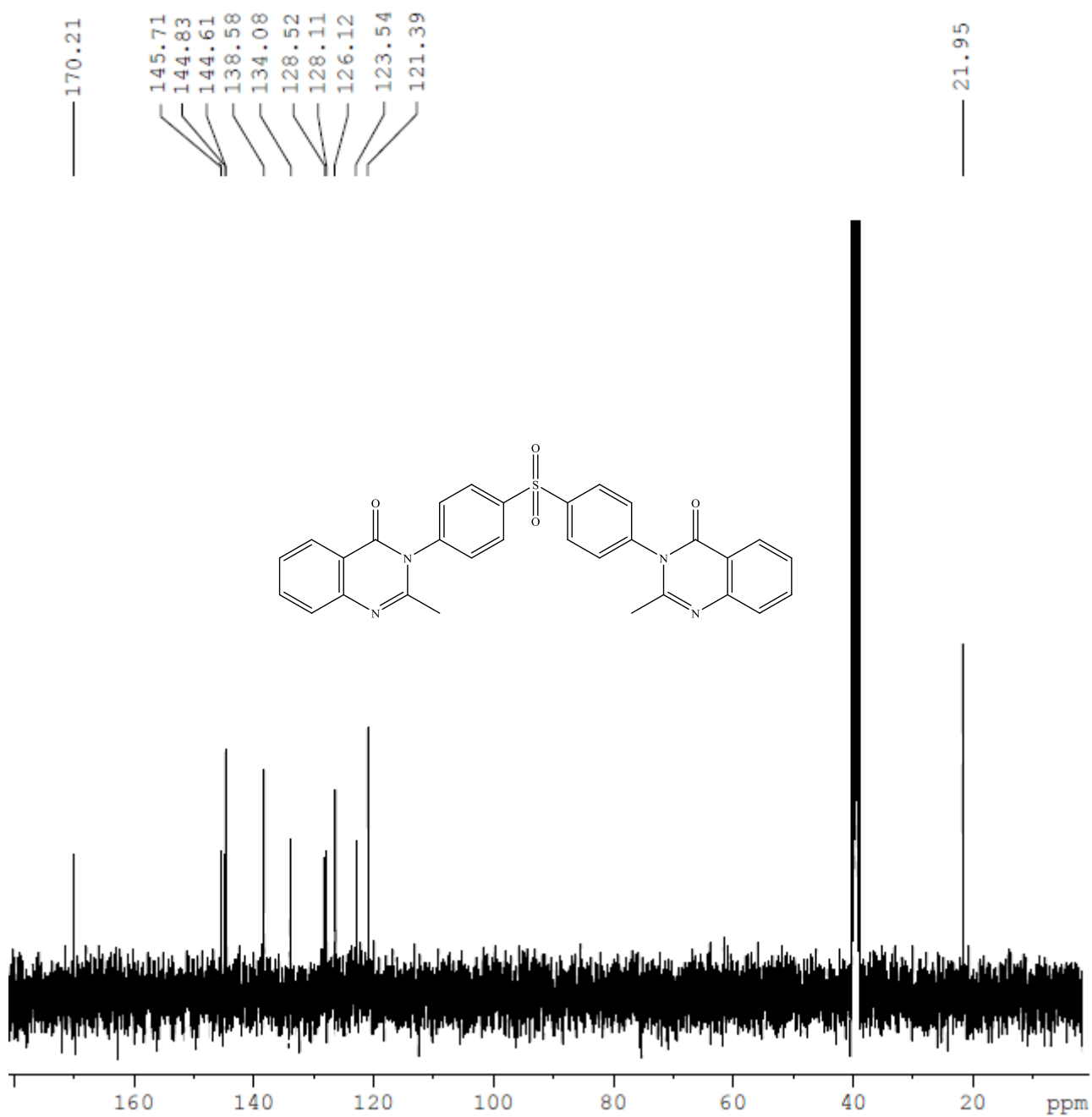
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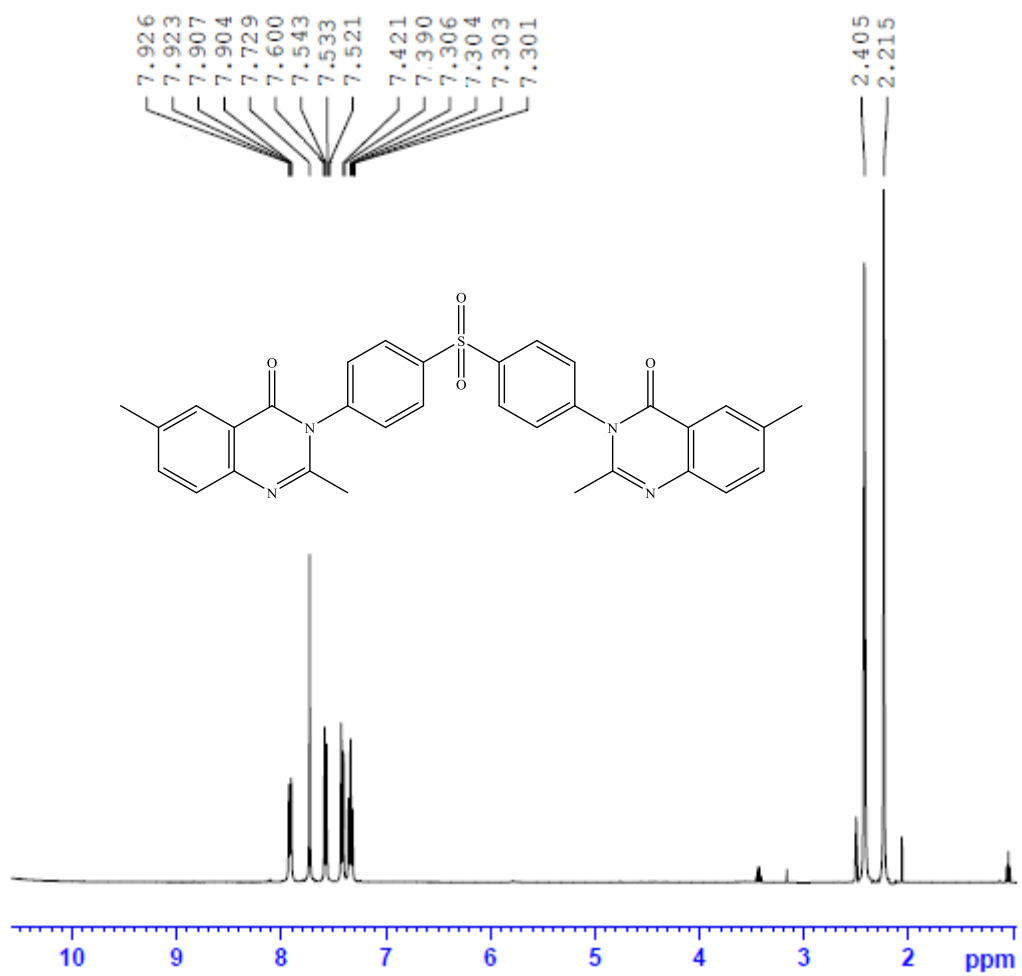
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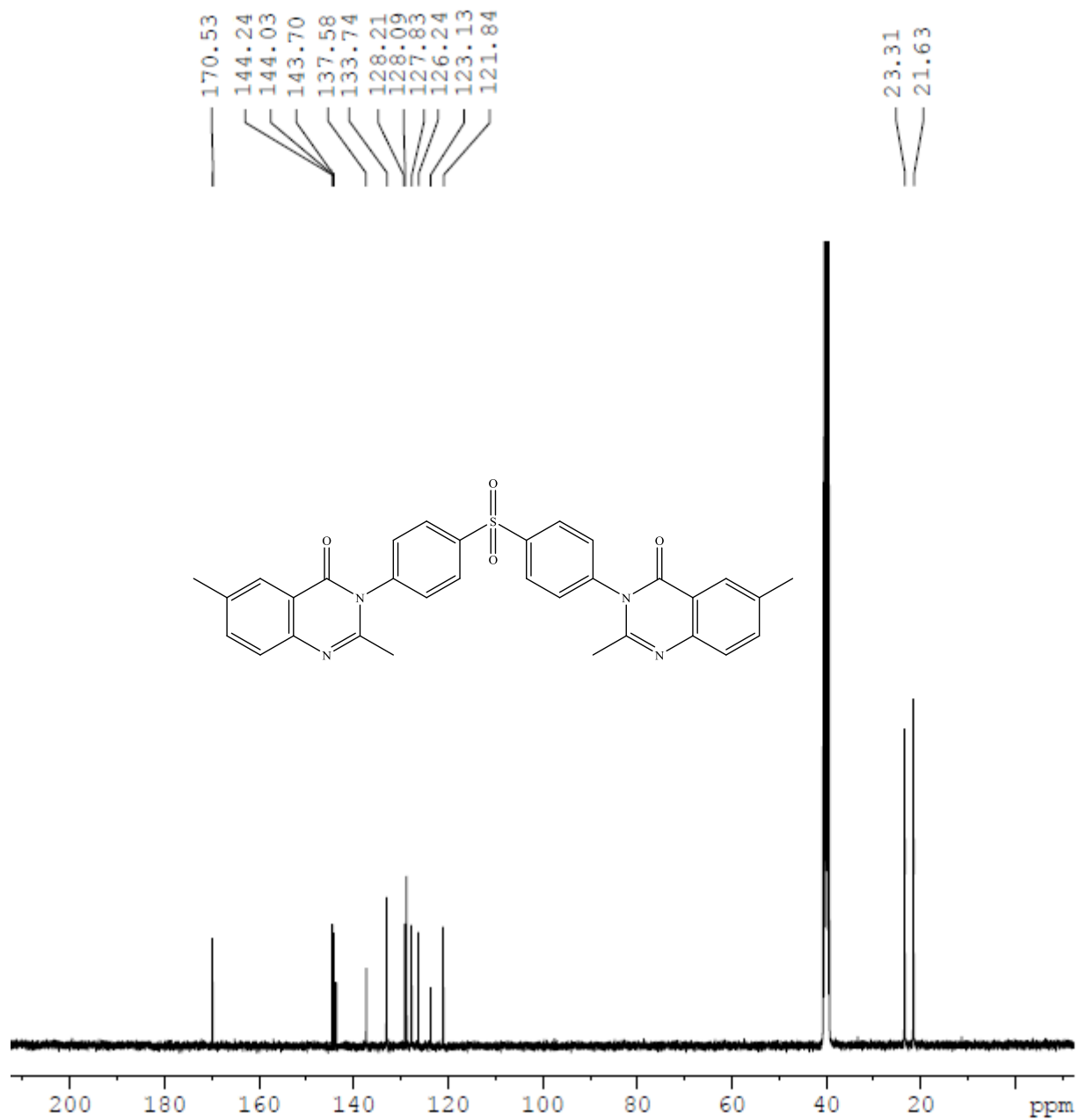
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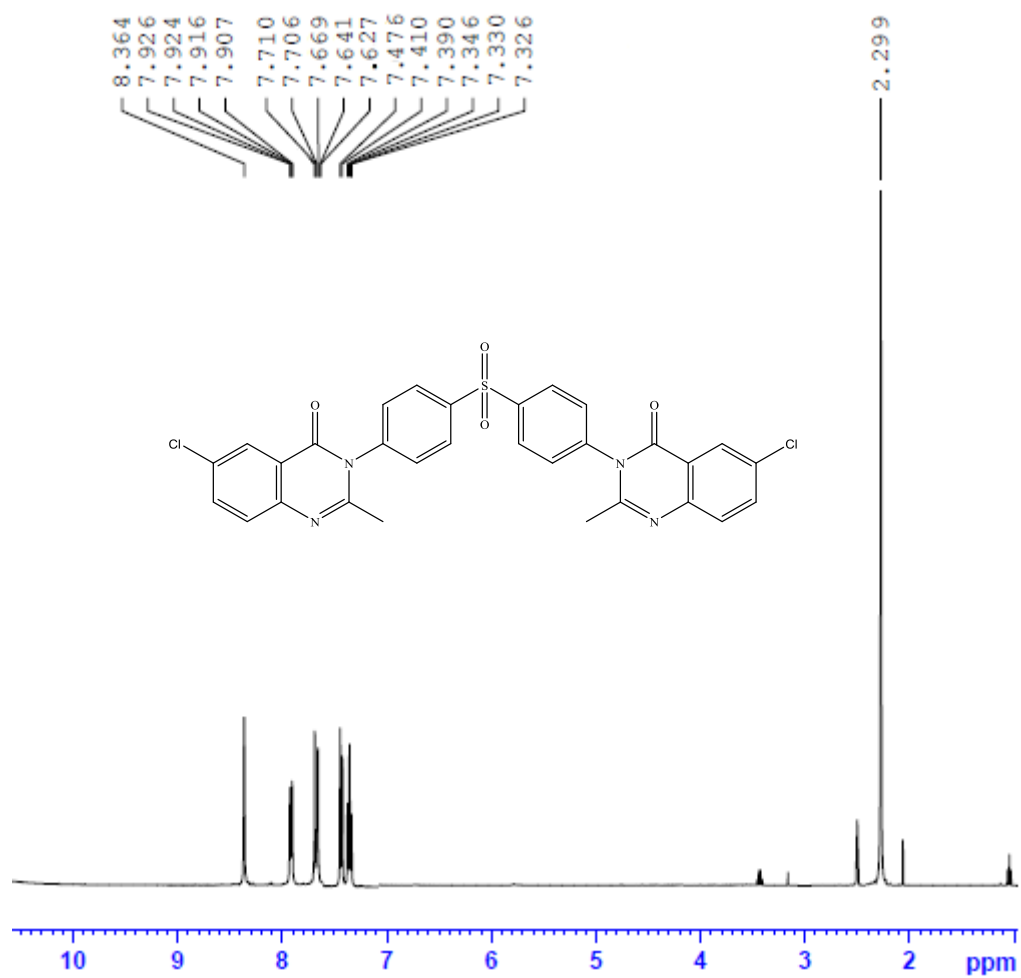
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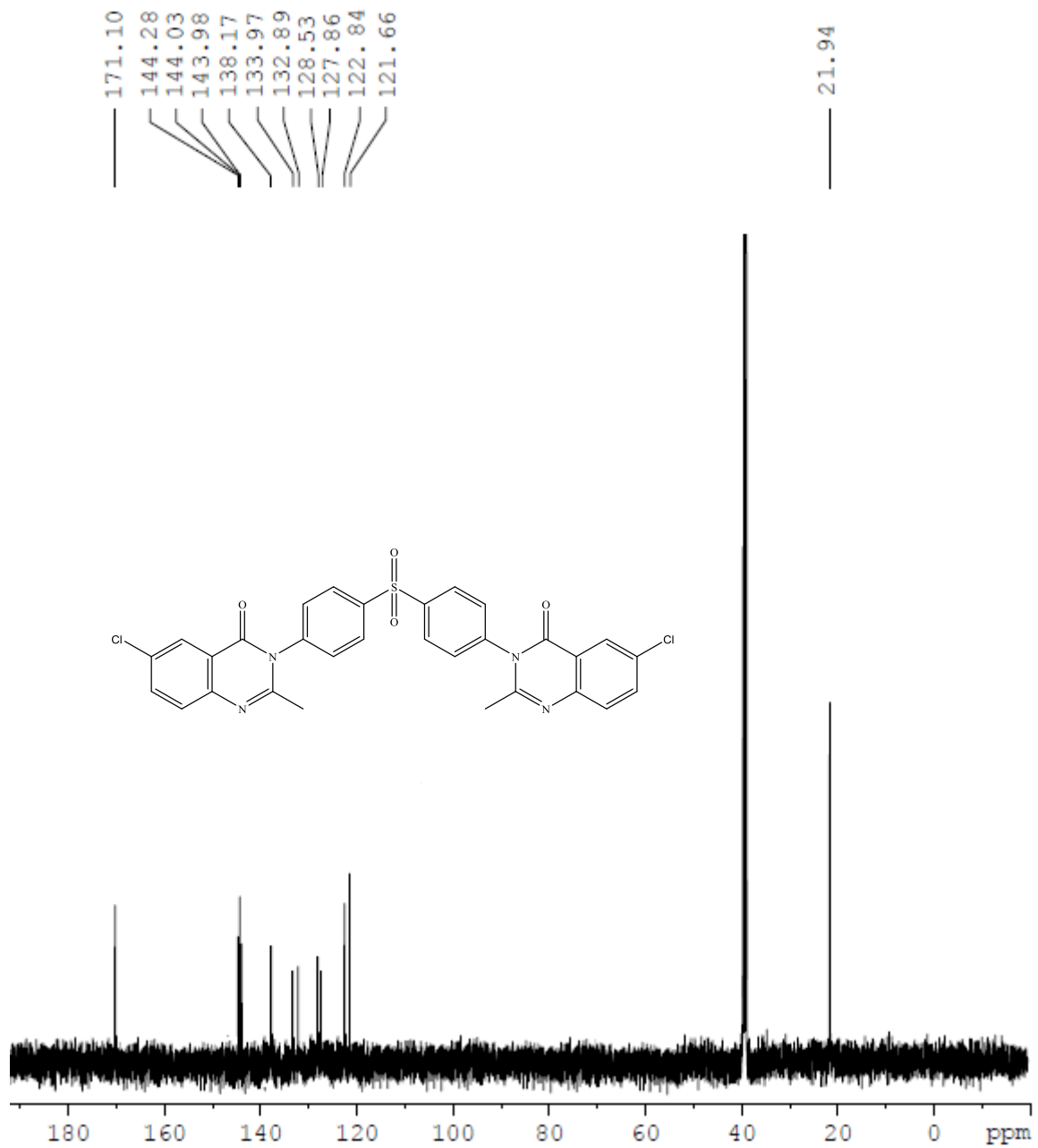
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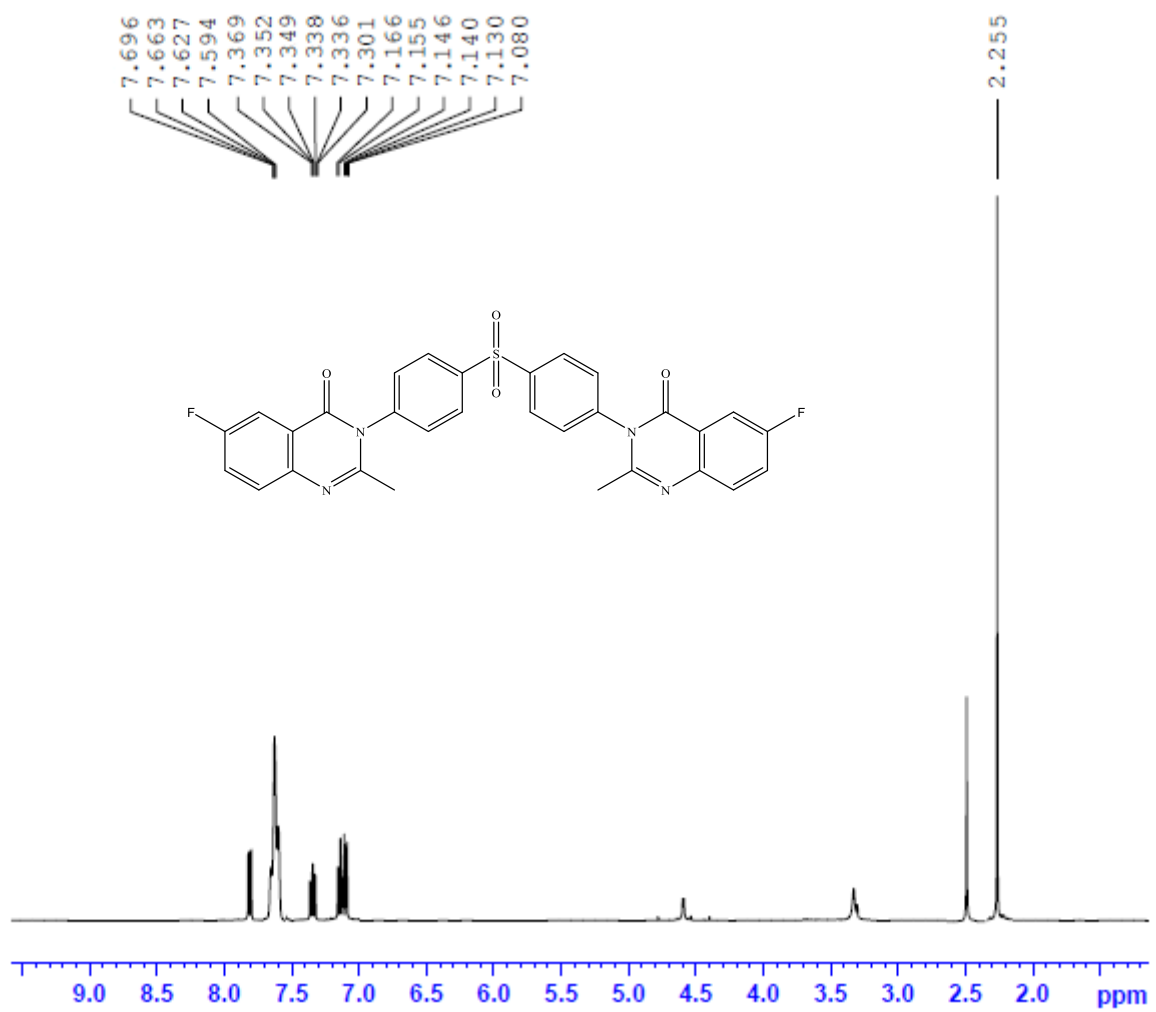
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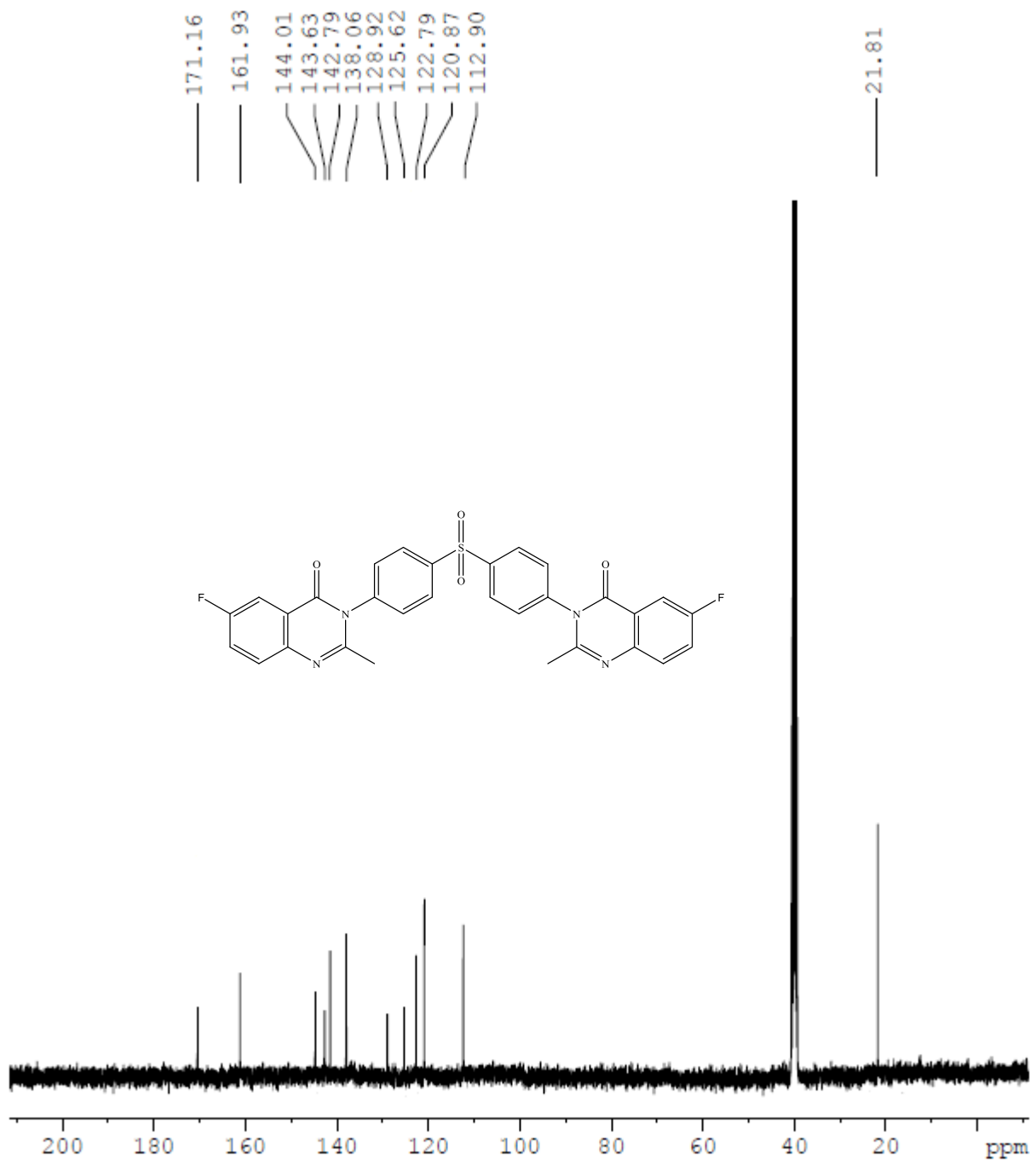
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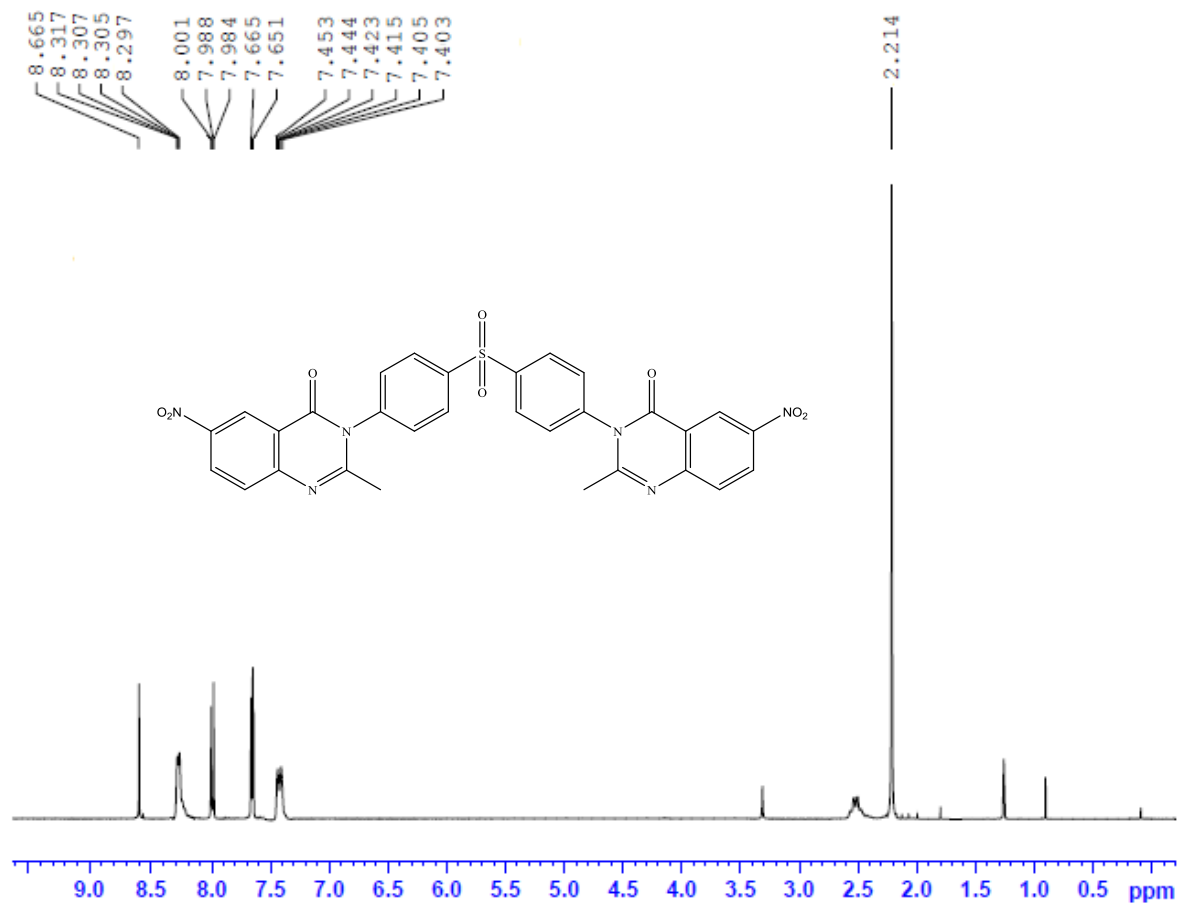
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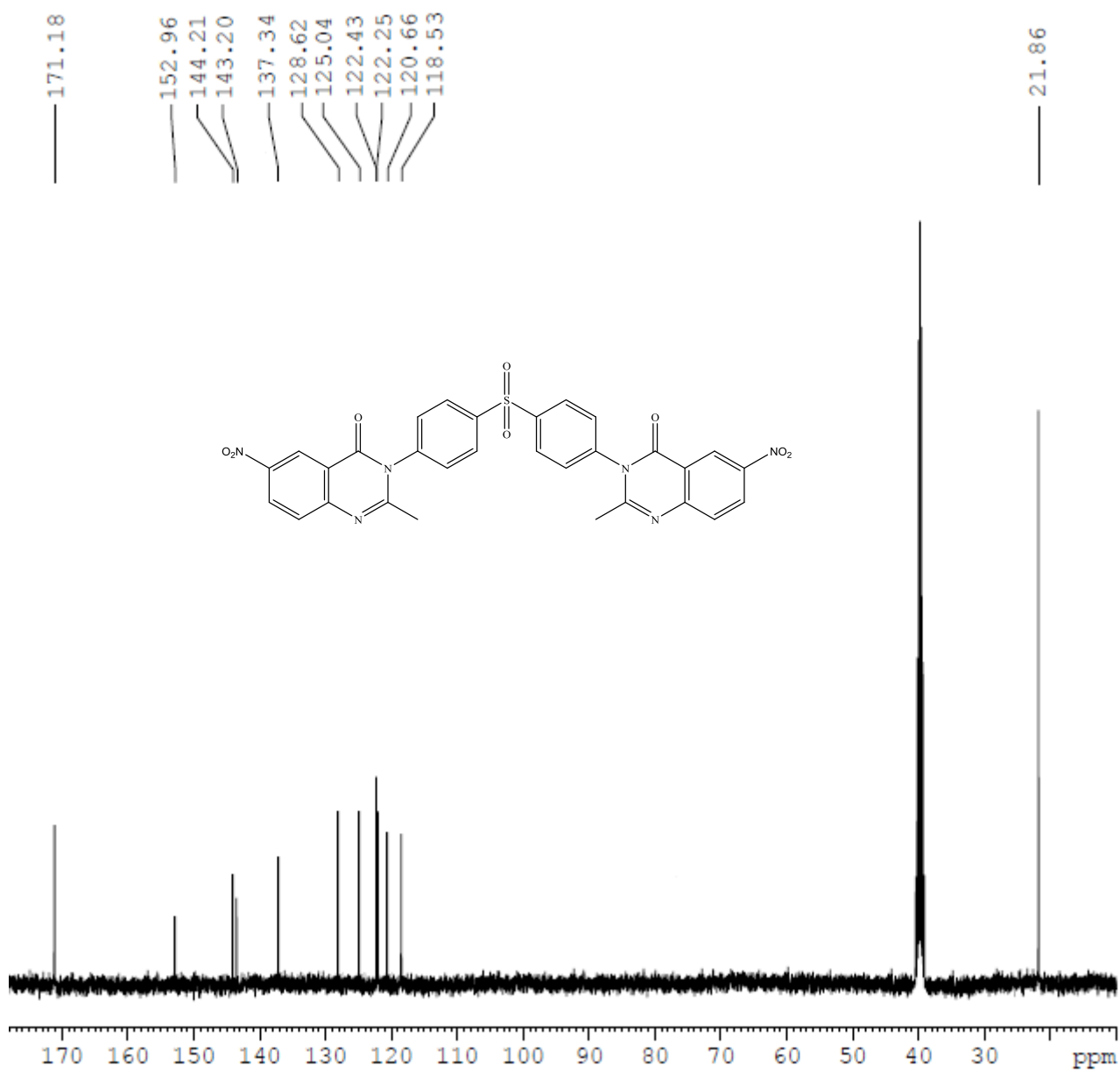
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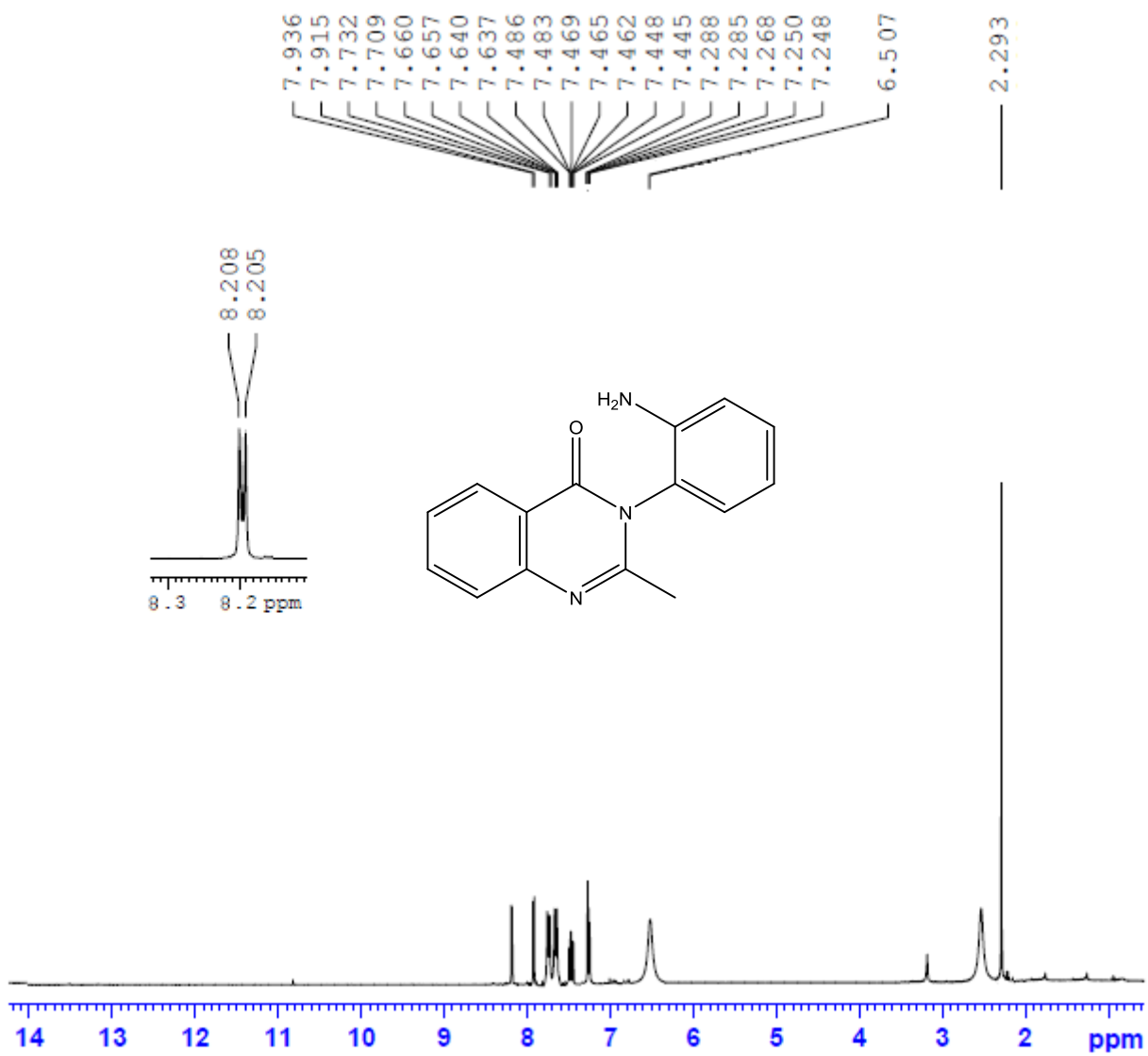
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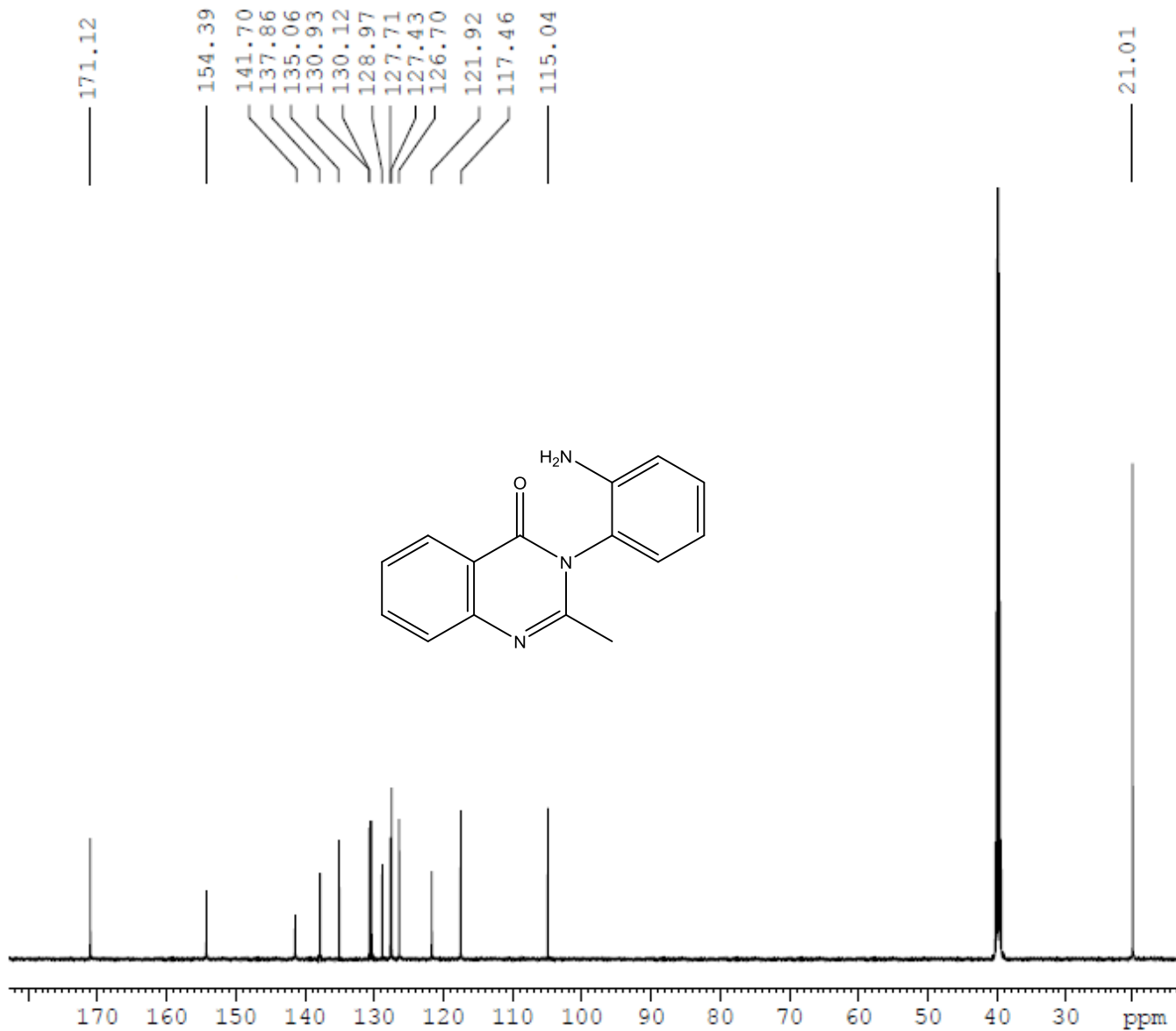
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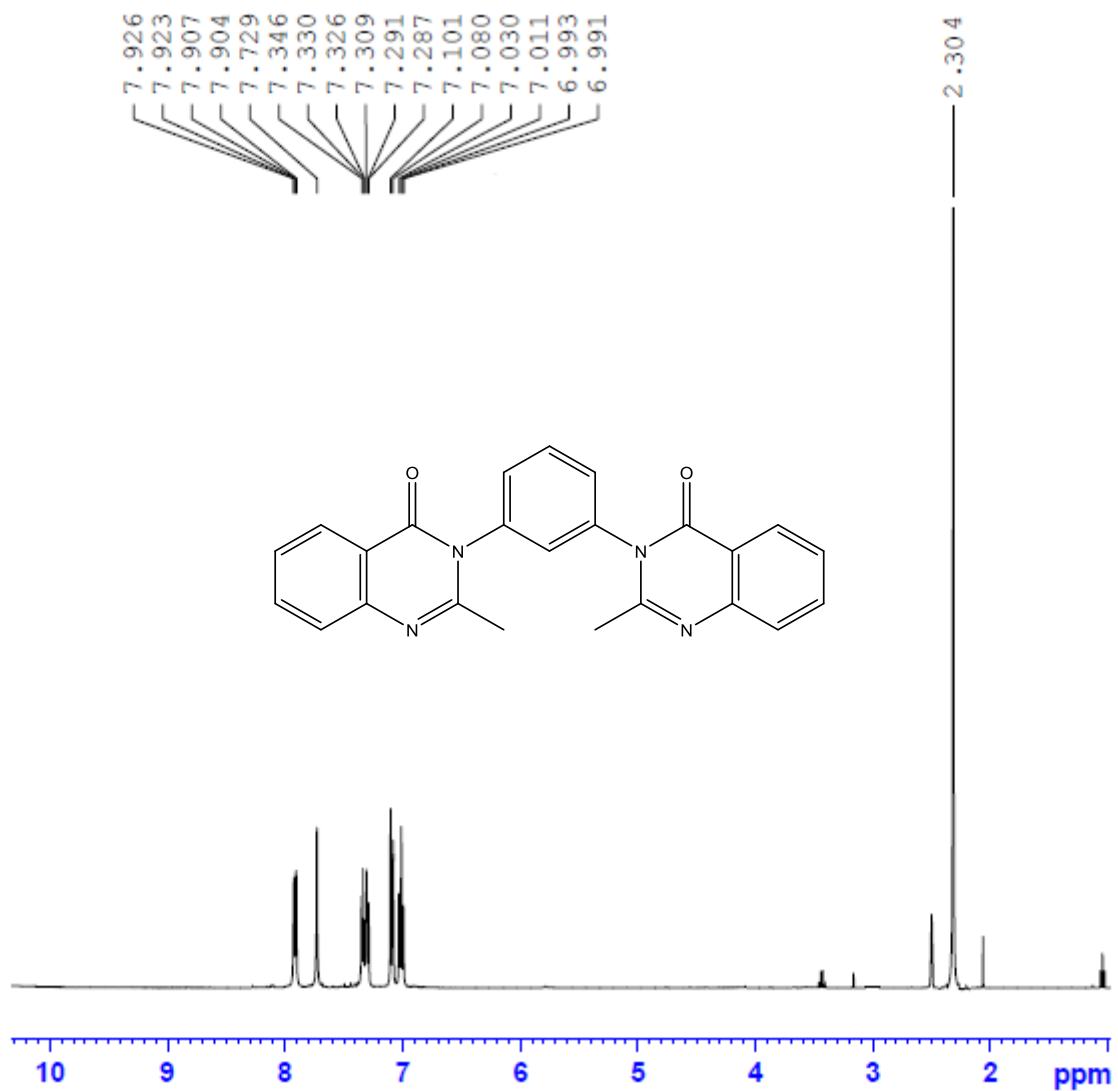
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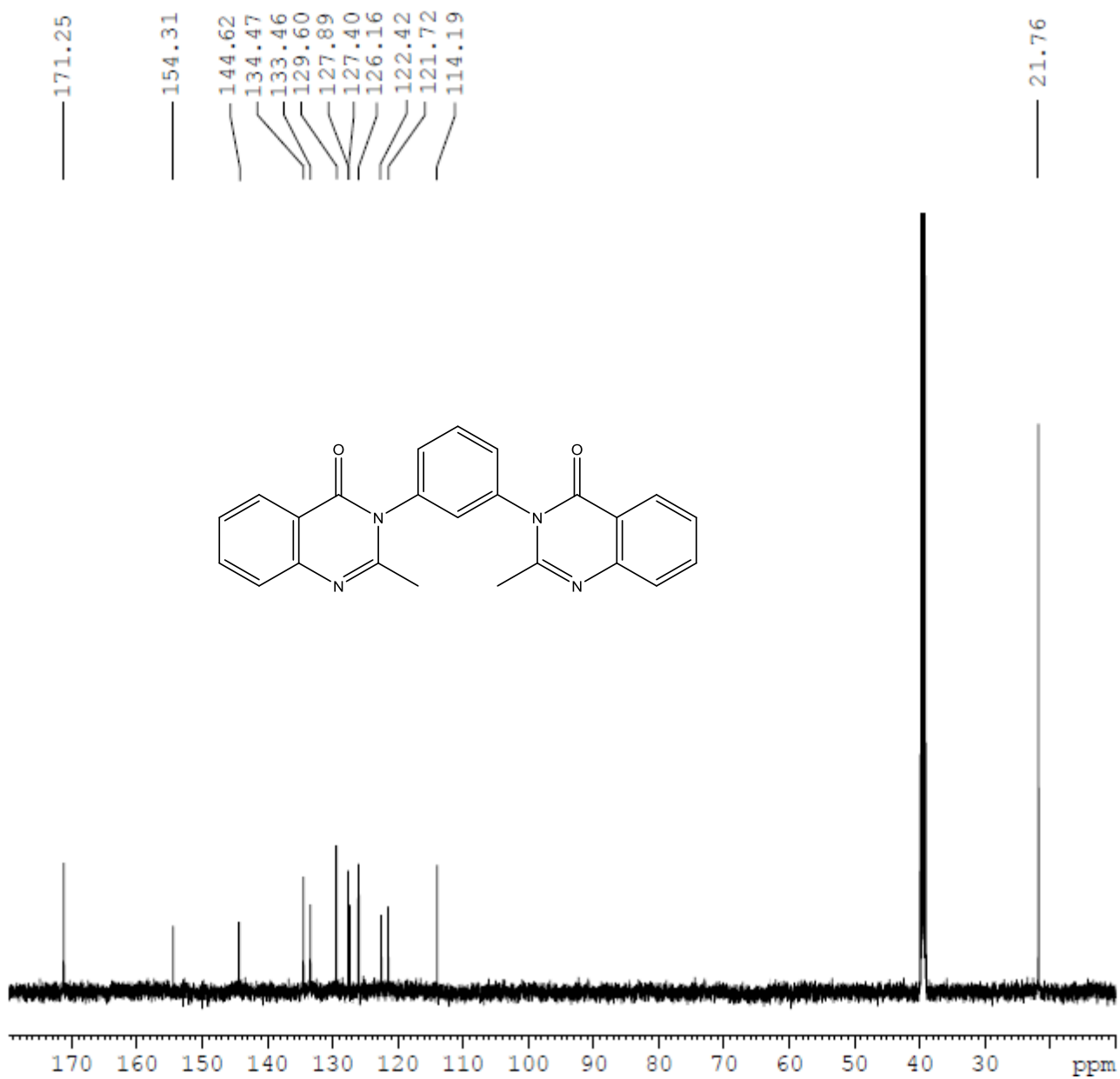
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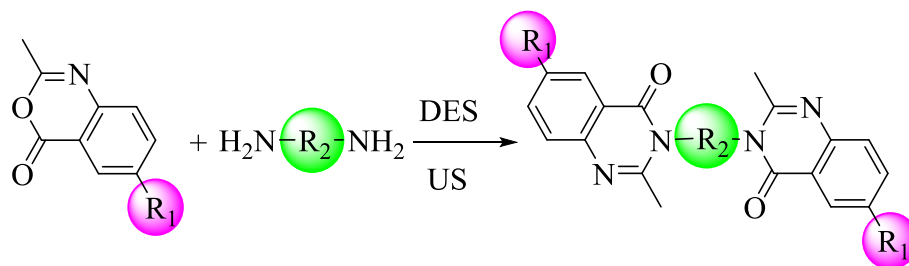


¹H NMR of **11**

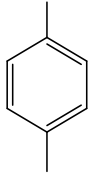
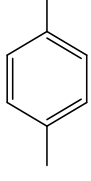
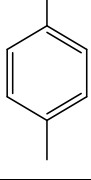
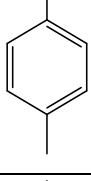
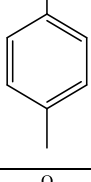
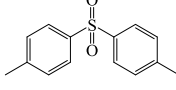


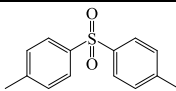
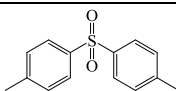
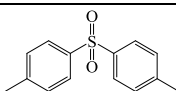
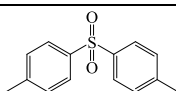
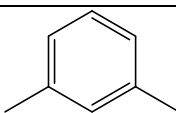
^{13}C NMR of **11**

Table 3. The anti-proliferative activities of the newly prepared bis-quinazolinone derivatives against human cancer cell lines.



Compd.	R ₁	R ₂	Conc.&IC ₅₀	Cells viability (%)	
				MCF-7 (μM)	A549 (μM)
3a	H		10 μM	51.00 ± 2.02	60.16 ± 1.56
			30 μM	57.95 ± 1.45	97.20 ± 0.41
			IC ₅₀ μM	9.78	8.15
3b	CH ₃		10 μM	31.52 ± 1.82	39.33 ± 2.76
			30 μM	94.90 ± 0.46	90.12 ± 0.25
			IC ₅₀ μM	12.70	15.70
3c	Cl		10 μM	56.64 ± 2.23	61.12 ± 1.82
			30 μM	86.85 ± 0.89	96.98 ± 0.40
			IC ₅₀ μM	9.05	8.14
3d	F		10 μM	69.73 ± 2.31	67.49 ± 0.27
			30 μM	89.10 ± 1.93	89.95 ± 0.43
			IC ₅₀ μM	7.11	8.11
3e	NO ₂		10 μM	83.12 ± 1.61	77.06 ± 1.78
			30 μM	81.18 ± 0.55	82.85 ± 1.64
			IC ₅₀ μM	4.55	5.34
4a	H		10 μM	79.06 ± 2.70	60.37 ± 3.31
			30 μM	89.85 ± 1.02	62.90 ± 1.45
			IC ₅₀ μM	5.38	6.11
4b	CH ₃		10 μM	78.02 ± 2.63	75.98 ± 2.21
			30 μM	96.50 ± 0.20	78.71 ± 1.76
			IC ₅₀ μM	6.73	6.79
4c	Cl		10 μM	76.84 ± 1.24	79.50 ± 2.61
			30 μM	81.08 ± 1.72	81.75 ± 0.87
			IC ₅₀ μM	4.65	6.07
4d	F		10 μM	60.25 ± 3.31	94.29 ± 0.39
			30 μM	61.92 ± 1.01	96.09 ± 0.244
			IC ₅₀ μM	6.19	5.55
4e	NO ₂		10 μM	95.58 ± 0.26	80.22 ± 2.10
			30 μM	95.26 ± 0.90	87.03 ± 1.45

			IC ₅₀ μM	2.75	2.98
5a	H	-CH ₂ -CH ₂ -	10 μM	51.03 ± 1.92	32.44 ± 2.91
			30 μM	58.07 ± 1.55	88.68 ± 1.83
			IC ₅₀ μM	9.79	12.32
5b	CH ₃	-CH ₂ -CH ₂ -	10 μM	27.07 ± 1.70	39.99 ± 2.11
			30 μM	87.65 ± 1.24	89.67 ± 1.76
			IC ₅₀ μM	13.55	15.73
5c	Cl	-CH ₂ -CH ₂ -	10 μM	80.22 ± 2.10	61.87 ± 2.34
			30 μM	87.03 ± 1.45	82.10 ± 1.93
			IC ₅₀ μM	6.12	7.98
5d	F	-CH ₂ -CH ₂ -	10 μM	76.84 ± 1.24	54.49 ± 0.27
			30 μM	81.08 ± 1.72	89.95 ± 0.43
			IC ₅₀ μM	5.39	7.95
5e	NO ₂	-CH ₂ -CH ₂ -	10 μM	91.11 ± 1.87	91.54 ± 0.56
			30 μM	92.80 ± 2.52	97.83 ± 0.24
			IC ₅₀ μM	5.21	5.05
6a	H		10 μM	75.50 ± 2.17	54.49 ± 0.27
			30 μM	79.02 ± 0.90	89.95 ± 0.43
			IC ₅₀ μM	6.98	8.41
6b	CH ₃		10 μM	61.13 ± 2.35	59.16 ± 1.53
			30 μM	82.84 ± 0.92	95.12 ± 0.22
			IC ₅₀ μM	7.48	8.13
6c	Cl		10 μM	94.78 ± 0.49	91.98 ± 0.49
			30 μM	96.88 ± 0.21	98.02 ± 0.21
			IC ₅₀ μM	5.53	5.15
6d	F		10 μM	54.22 ± 2.78	82.65 ± 1.67
			30 μM	86.56 ± 1.54	81.78 ± 0.45
			IC ₅₀ μM	5.70	4.54
6e	NO ₂		10 μM	95.26 ± 0.45	95.40 ± 0.24
			30 μM	95.87 ± 0.21	97.60 ± 0.21
			IC ₅₀ μM	2.73	3.43
7a	H		10 μM	72.68 ± 0.79	82.73 ± 1.63
			30 μM	89.08 ± 0.94	82.10 ± 0.50
			IC ₅₀ μM	4.24	4.58

7b	CH ₃		10 μM	53.03 ± 2.29	53.13 ± 2.20
			30 μM	86.90 ± 1.24	86.84 ± 1.19
			IC ₅₀ μM	5.78	5.77
7c	Cl		10 μM	93.82 ± 0.49	72.34 ± 0.78
			30 μM	96.21 ± 0.34	88.43 ± 0.19
			IC ₅₀ μM	3.91	4.21
7d	F		10 μM	61.03 ± 2.82	96.46 ± 0.65
			30 μM	78.09 ± 1.79	96.68 ± 0.17
			IC ₅₀ μM	1.46	3.45
7e	NO ₂		10 μM	81.08 ± 1.77	95.99 ± 0.14
			30 μM	86.97 ± 1.26	96.76 ± 0.87
			IC ₅₀ μM	1.26	2.75
11	H		10 μM	61.80 ± 1.11	79.8 ± 1.05
			30 μM	88.54 ± 1.09	83.20 ± 1.02
			IC ₅₀ μM	5.42	5.94
Sorafenib			10 μM	69.85 ± 1.02	89.8 ± 1.14
			30 μM	90.79 ± 1.43	91.21 ± 1.44
			IC ₅₀ μM	4.03	5.20

Green Metrics Calculations^{1,2}

$$\% \text{ Atomic Efficiency (AE)} = \frac{\text{Mol Wt. of desired product}}{\text{Mol Wt. of all reagents}} \times 100$$

$$\% \text{ Carbon Efficiency (CE)} = \frac{\text{Mass of carbon in product}}{\text{Total mass of carbon in the reactants}} \times 100$$

$$\text{Reaction Mass Efficiency (RME)} = \frac{\text{Mass of the isolated product}}{\text{Total mass of reactants used in the reaction}} \times 100$$

$$\% \text{ Yield Economy (YE)} = \frac{\text{Reaction percent}}{\text{Time in min}} \times 100$$

$$\text{E-Factor (EF)} = \frac{\text{Mass of the total waste}}{\text{Mass of the crude product}}$$

$$\text{Process Mass Intensity (PMI)} = \frac{\text{Total mass used in process}}{\text{Mass of product}}$$

References

1. D. Curzons, D. J. C. Constable, D. N. Mortimer and V. L. Cunningham, *Green Chem.*, 2001, **3**, 1-6.
2. C. Jimenez-Gonzalez, D. J. C. Constable and C. S. Ponder, *Chem. Soc. Rev.*, 2012, **41**, 1485-1498.