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SUPPORTING INFORMATION

Title: Synthesis and Avidin Binding of Ruthenium Complexes Functionalized with a Light-Cleavable Free Biotin Moiety

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1. Synthesis and Characterization

1.1. NMRs and numbering of the synthesized compounds





Figure S 1: 1H NMR (300 MHz, dMeOH) of compound 1.

1.1.2. Bis(N-biotinyl)-3,6-dioxaoctane-1,8-diamine (2)



Figure S 2: ¹H NMR (300 MHz, dMeOH) ligand **2**.



Figure S 3: ¹H NMR (300 MHz, dMeOH) complex [5]Cl₄.



1.1.4. $[Ru(tpy)(bpy)(1)]Cl_2$ -compound [6]Cl₂ (as mixture with complex [7]₂)

Figure S 4: ¹H-NMR of complex [6]Cl₂. The small dd signal at 9.84 indicates the formation of the regioisomere, complex [7]Cl₂.

1.1.5. [Ru(tpy)(bpy)(2)](PF₆)₂ – (compound [8](PF₆)₂)



Figure S 5: ¹H NMR (400 MHz, dMeOH) of complex [8]Cl₂

1.2. Analytical scale synthesis of $[Ru(tpy)(bpy)(1)](PF_6)_2 - (compound [6](PF_6)_2)$

Table S1. Results of the reaction of ligand **1** with $[Ru(tpy)(DP_2)](PF_6)_2$ in H_2O under different conditions.^a Isomeric ratios were obtained from ¹H NMR measurements (in D_2O or MeOD, 300 MHz) after workup of the reaction by silica column chromatography.

			NMR yield	
	Temperature (K)	Time (h)	[6] ²⁺	[7] ²⁺
Α	333	48	61%	39%
В	353	48	83%	17%
С	Reflux	48	84%	16%
D	Reflux	2	63%	37%

^[a] Conditions: $[RuOH_2] = 5.0 \text{ mM}$, [1] = 25 mM, in H₂O:acetone (3:1) solution, under argon, in the dark.

2. Photochemistry





Figure S 6 Plot of the moles of the ruthenium thioether complex $[8]^{2+}$ vs the moles of photons Q absorbed by the complex.



Figure S 7. Evolution of the UV-Vis spectra of a solution of [8](PF₆)₂ during blue light irradiation (λ = 465 nm, $\Delta\lambda_{1/2}$ = 25 nm). Conditions [Ru]₀ = 9.4×10^s M in PBS solution (pH = 7.03, I = 50 mM), photon flux Φ_{p} = 1.87×10^s (Einstein/s); photon flux Φ_{p} = 1.29×10^s (Einstein/s). T = 298 K, irradiation pathlength: 3.00 cm, UV-vis absorbance pathlength: 1.00 cm.