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Supplementary Materials for

Carbon doping of WS₂ monolayers: Bandgap reduction and p-type doping transport

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Supplementary Materials

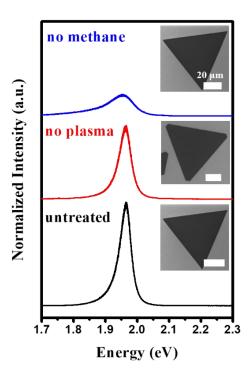


Fig. S1. PL responses of reference samples. Pristine WS₂ (bottom), WS₂ annealed at 400°C in a CH₄ environment (middle), and Ar/H₂ plasma treated WS₂ in the absence of CH₄ (top). As CH₄ cannot be thermally decomposed at 400°C in the absence of plasma, the spectra from heat treated and pristine samples are very similar. Ar/H₂ plasma treatment (without CH₄) causes sulfur atoms to be sputtered away from WS₂, therefore the top panel shows both a quenching and a defect related shoulder in the PL response, as expected. In all panels, insets exhibit the corresponding scanning electron microscopy (SEM) images of the WS₂, showing no morphology changes among these three different samples.

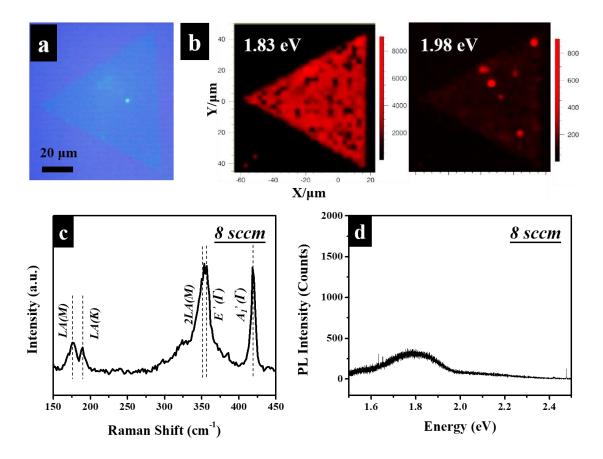


Fig. S2. PL mapping of carbon-doped WS₂, Raman and PL of heavily carbon-doped WS₂. (a) and (b) show the uniformity of the PL of the carbon-doped WS₂ @ 5sccm, measured with a 488 nm laser. (a) Optical image of a WS₂ triangular monolayer. (b) PL mappings showing the intensity of the response at 1.83 eV and 1.98eV, respectively. The mapping at 1.83 exhibits a mostly uniform PL signal, with the exception a few areas with local structural defects (such as holes in the triangles) or undoped parts (that emit at 1.98 eV). (c) Raman and (d) PL of heavily-carbon-doped WS₂ (@ 8sccm). Extra carbon doping may result in severe PL quenching but the characteristic Raman signatures of WS₂ are preserved.

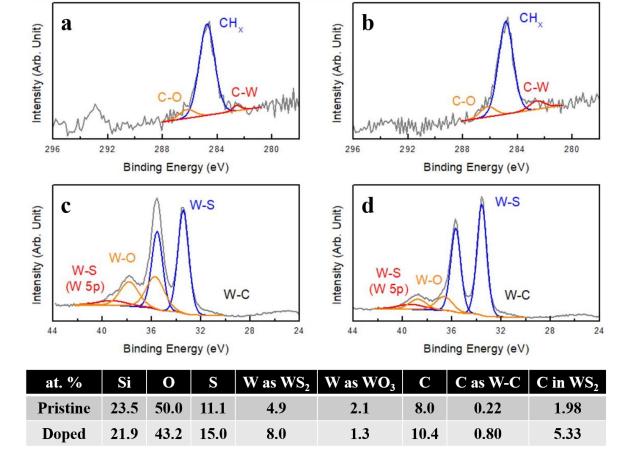


Fig. S3. XPS analyses of the carbon-doped WS₂. X-ray photoelectron spectroscopy (XPS) elemental analyses of pristine WS₂ (a) C 1s, (c) W 4f, and carbon-doped WS₂ (b) C 1s and (d) W 4f core levels. The table list shows elemental compositions in both the pristine and carbon-doped WS₂. The carbon exists on the material surface mostly in the form of CH_X and CO_X (intense peak centered at ~ 284.8 eV). The C 1s line scans show that the carbon in the form of carbide bonding (red curve at around 282.5 eV) increased three times after doping, which indicates the formation of the new W-C bonds that substitute sulfur atoms in the WS₂ lattice. We have also characterized and compared the pristine WS₂ and carbon-doped WS₂ samples by ToF-SIMS (Time of Flight Secondary Ion Mass Spectrometry), and FTIR (Fourier Transform Infrared Spectroscopy). The low signal to noise ratio lead to difficulties in detecting the characteristic features from doped carbon species.

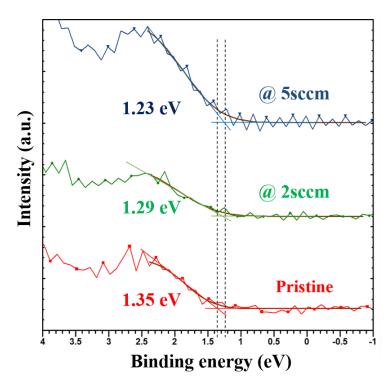


Fig. S4. VBM of the carbon-doped WS₂. The carbon incorporation into WS₂ gradually saturates the sulfur vacancies, thus decreasing the energy of the valence band maximum (VBM) to the Fermi level (FL), as shown in the figure, from 1.35 eV (pristine) to 1.29 eV (doped @2 sccm) and 1.23 eV (doped @5 sccm). All three spectra show similar valence band signatures of WS₂ that indicate that the binding energy shift was induced by carbon doping in WS₂.

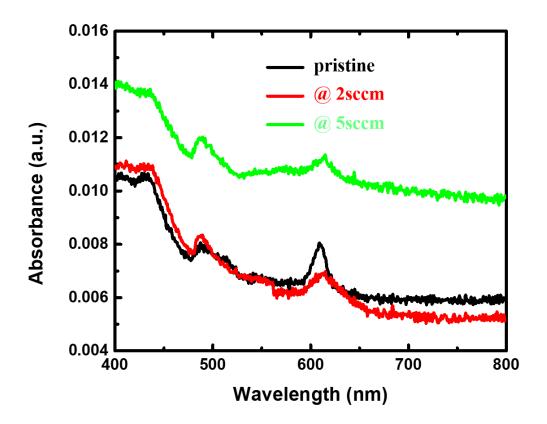


Fig. S5. UV-Vis spectra of the carbon-doped WS₂. The A exciton can be observed at 608 nm (2.039 eV) in the as grown WS₂, which corresponds to the excitonic absorption at the K point of the Brillouin zone. This is one of the characteristic absorption peaks of the monolayered WS₂. As the percentage of C in WS₂ increases, the A excitonic transition gradually shifts from 2.039 eV (608 nm) to 2.020 eV (614 nm) in lightly-doped WS₂ (@2 sccm), and 2.016 eV (615 nm) in medium-doped WS₂(@5 sccm). The red shifts in the A excitonic transitions further support the bandgap reduction via C doping.

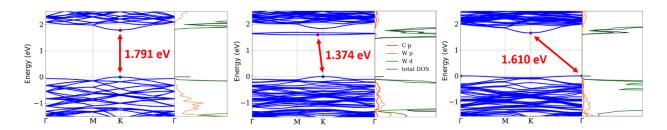
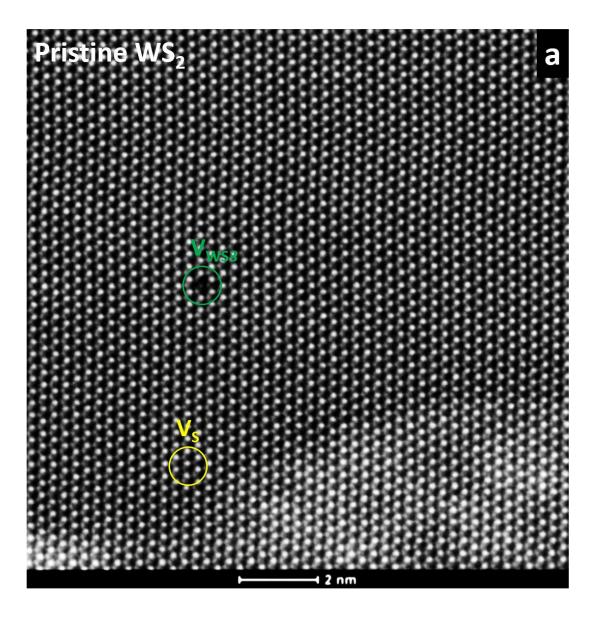
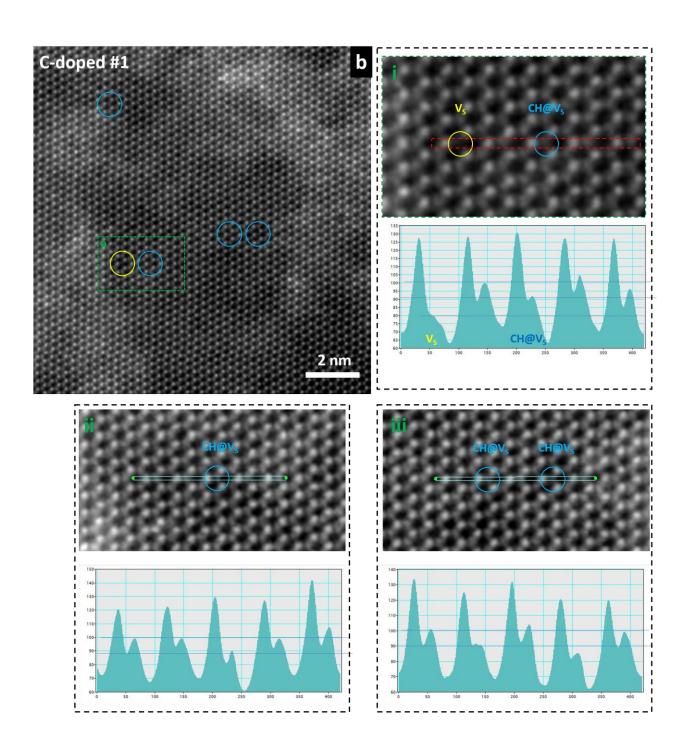
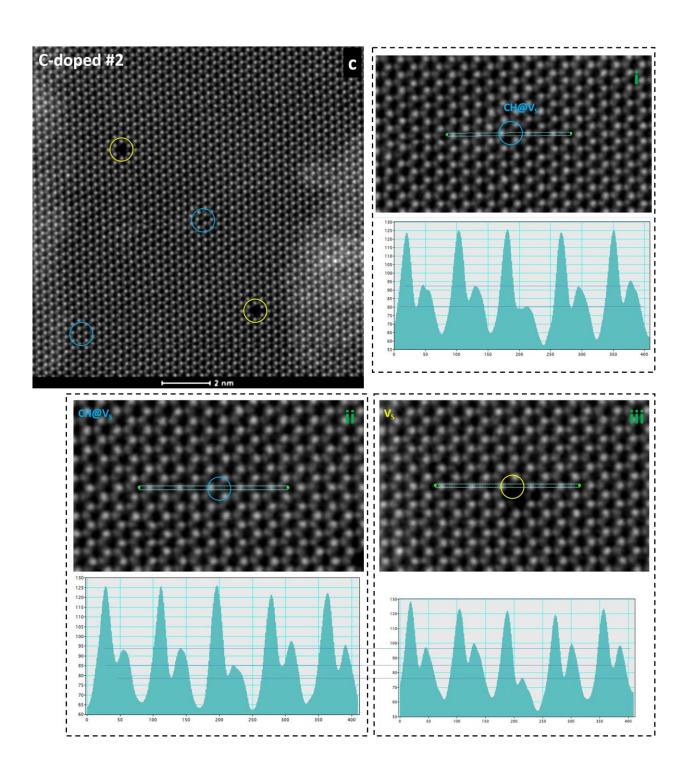


Fig. S6. Complementary band structures and DOS of monolayer pristine WS_2 and carbon-doped WS_2 . Band structure and density of states (DOS) of (a) monolayer pristine WS_2 , (b) carbon-doped WS_2 with CH_2 units bonded at the β position of the single sulfur vacancy, and (c) CH_2 -doped WS_2 with the dopant at the α position. In the DOS, p-orbitals of C and W atoms, d-orbital of W atoms, and total DOS are plotted in different colors.







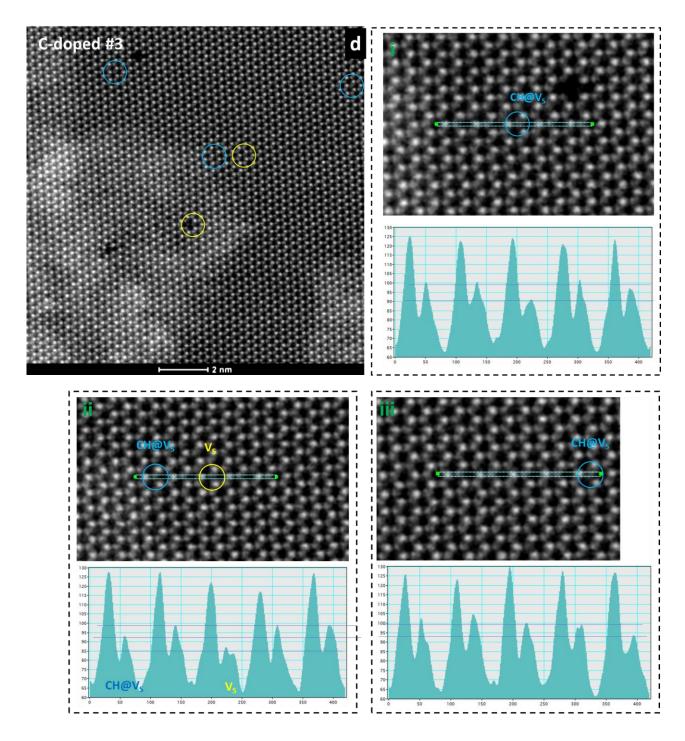


Fig. S7. Experimental STEM examples and identification of sulfur monovacancies and carbon dopant in pristine and carbon-doped WS₂ monolayers. A single sulfur vacancy (Vs) is highlighted with a yellow circle, and tungsten vacancies (V_{WS3}) with green circles. The hexagonal WS₂ lattice can also be identified from the ADF intensity. Examples of carbon-doped WS₂ monolayers (b-d) showing the CH@Vs (blue circles) and Vs defects (yellow circles) identified by the ADF image intensity. The line profiles clearly show differences in the intensities when comparing CH@Vs and Vs. We made sure that this intensity profile line-scan avoid the contaminated polymer residue regions by only analyzing atomically clean areas.

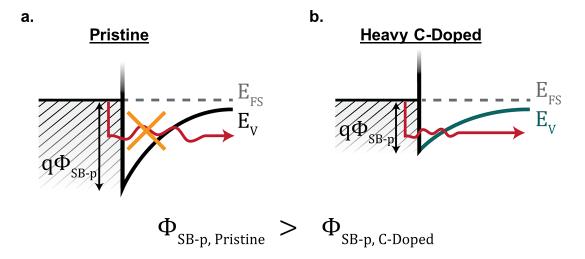


Fig. S8. Band diagram showing the change in Fermi level pinning. Schematic of the valence band position (E_V) near the source for a (a) pristine and (b) carbon-doped FET (@ 8sccm), respectively. For the pristine devices, a tall and wide Schottky tunnel barrier (Φ_{SB-p}) prevents any hole conduction and hence no p-branch is observed. Conversely, after heavy doping (@ 8sccm), the metal Fermi level pins closer to the WS₂ valence band. Thus, heavily-doped samples have a smaller Schottky barrier height for hole injection and, therefore, exhibit significant hole conduction.

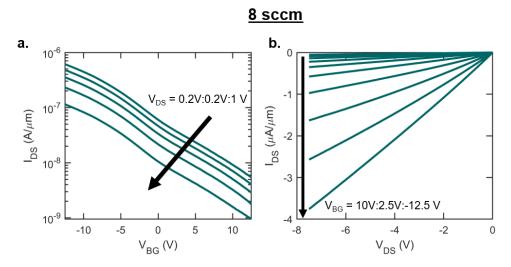


Fig. S9. Transfer and output characteristics of a carbon-doped WS₂-based FET (@8 sccm CH₄). (a) Drain voltage (I_{DS}) versus back gate voltage (V_{BG}) and (b) I_{DS} versus drain voltage (V_{DS}). The device shows heavy p-doping and linear I_{DS} versus V_{DS} characteristics, indicating the metal-WS₂ Fermi level is pinned relatively close to the valence band.

Table S1. Local strain of WS_2 with premade single vacancies and different ligands on C atoms and dopant positions. The local strain is compared to pristine WS_2 .

Premade Vacancy Type	Type of Dopant	Strain Compared to Pristine WS ₂		
		Doped C Position		
		α	β	γ
Single Vacancy	С	-0.21%	-0.24%	0.50%
	СН	-0.43%	-0.10%	0.53%
	CH ₂	-0.32%	-0.11%	0.74%