

Supplementary Materials

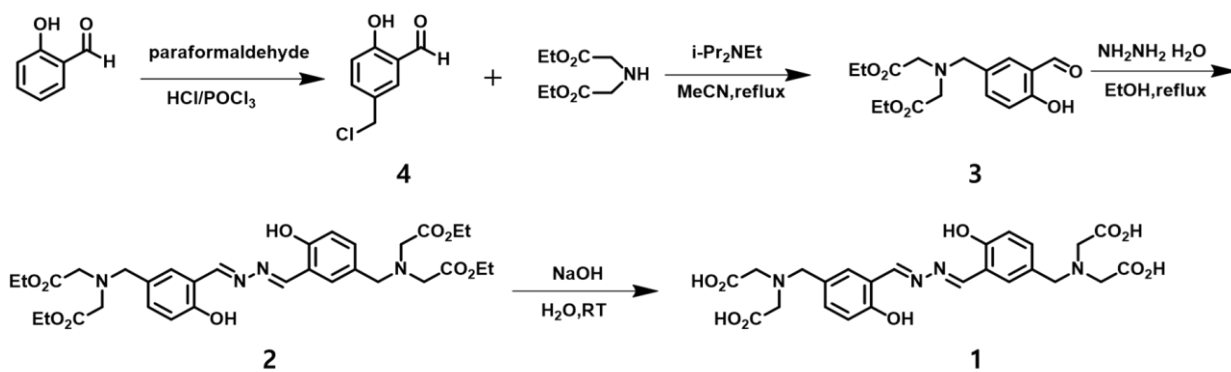
Spectroscopic Study of the Salicyladazine Derivative– UO_2^{2+} Complex and Its Immobilization to Mesoporous Silica

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Scheme S1. Synthesis route of compound 1.

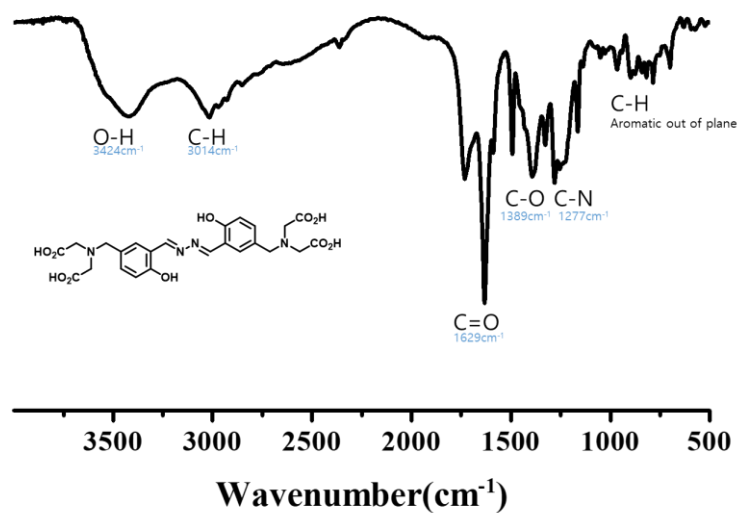


Figure S1. FT-IR spectrum of compound 1.



Figure S2. ¹H NMR spectrum of compound 1 in DMSO-d₆.

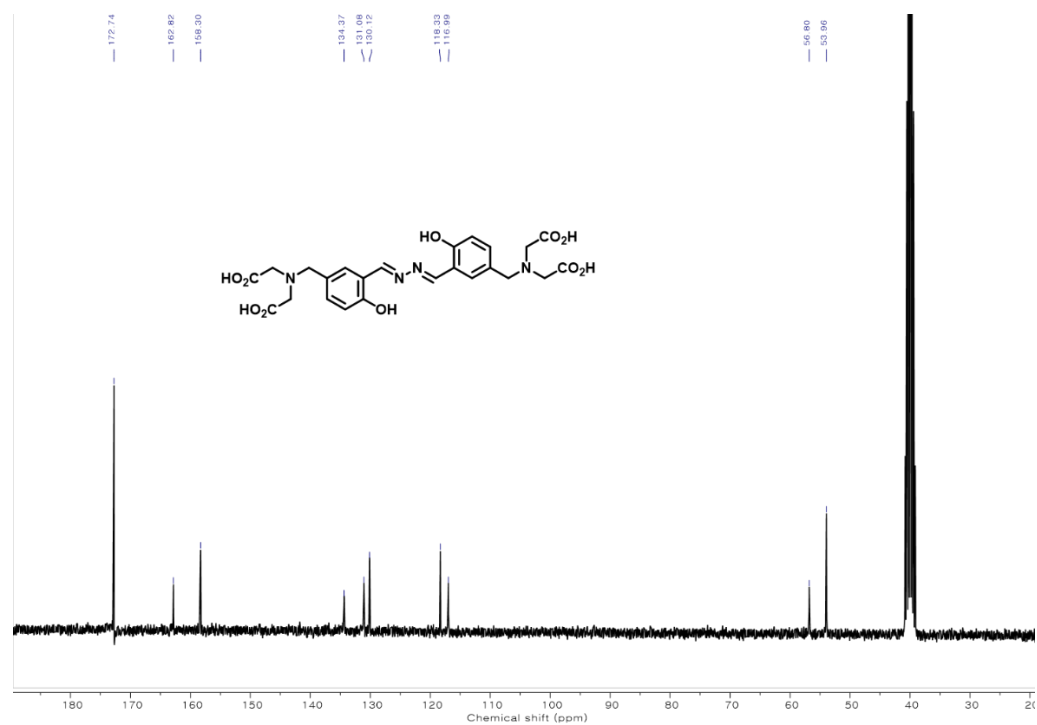


Figure S3. ¹³C NMR spectrum of compound 1 in DMSO-d₆.

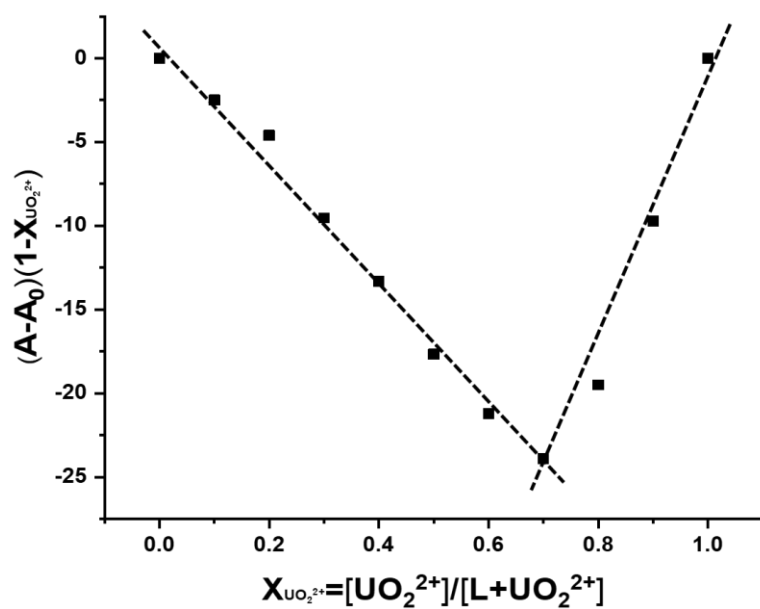


Figure S4. Job's plot for complex formed between compound 1 and UO_2^{2+} .

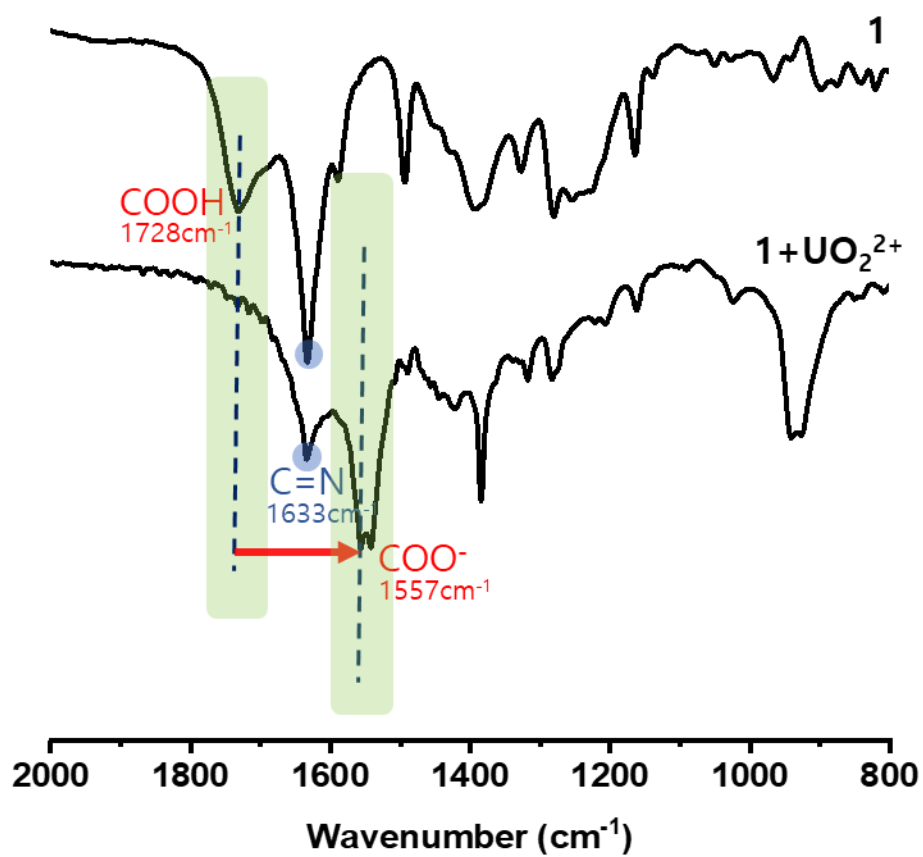


Figure S5. FT-IR spectra of 1 and 1 with UO_2^{2+} .

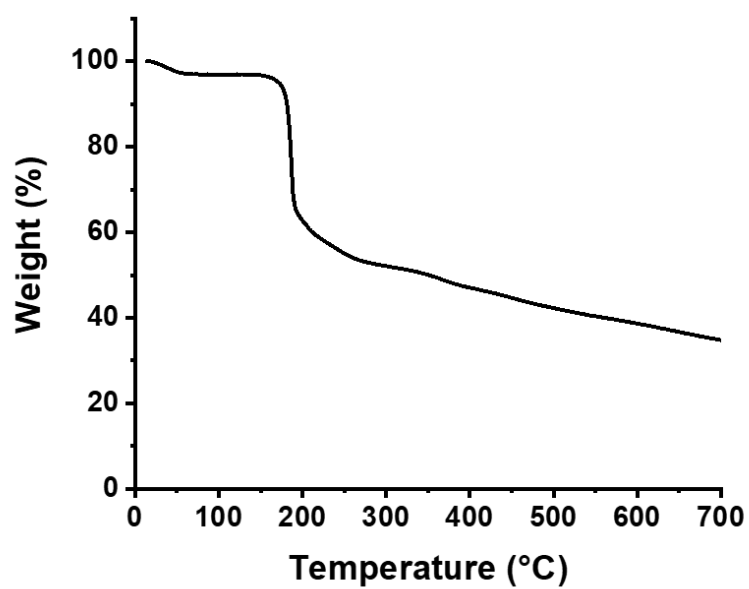


Figure S6. TGA thermogram of compound 1.

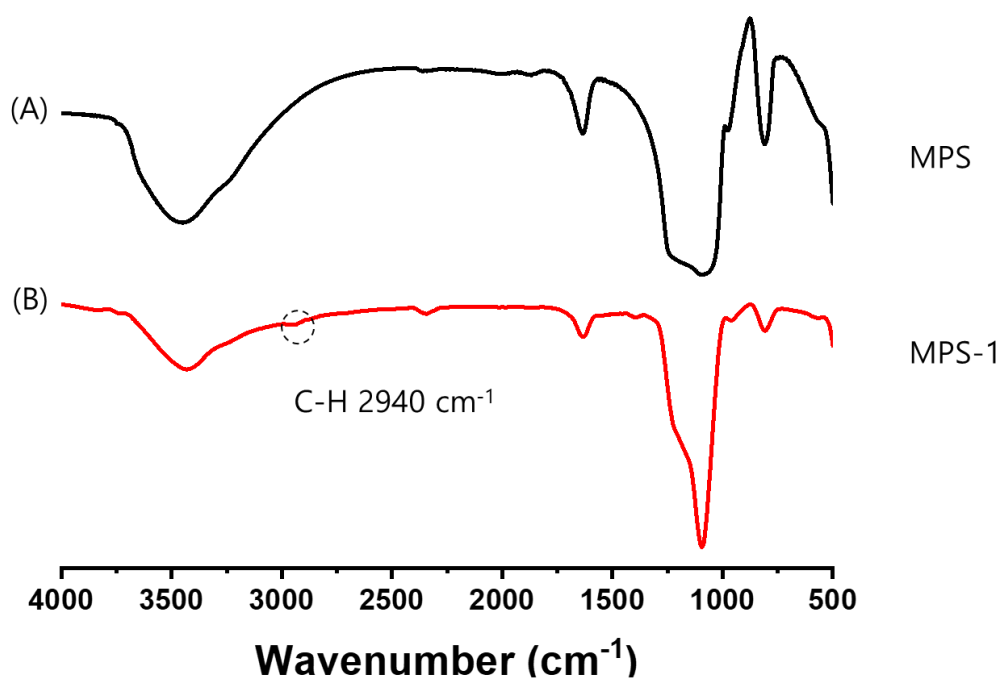


Figure S7. FT-IR spectra of (A) MPS and (B) MPS-1.

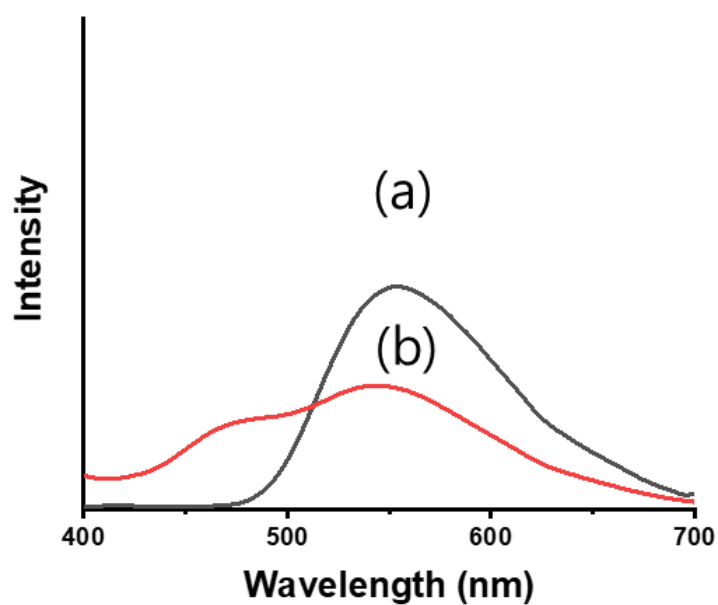


Figure S8. Fluorescence spectra of (a) MPS-1 (2 mg) in 3.5% NaCl solution (2 mL) and (b) MPS-1 (2 mg) with UO_2^{2+} solution (100 ppb) in 3.5% NaCl (2 mL).

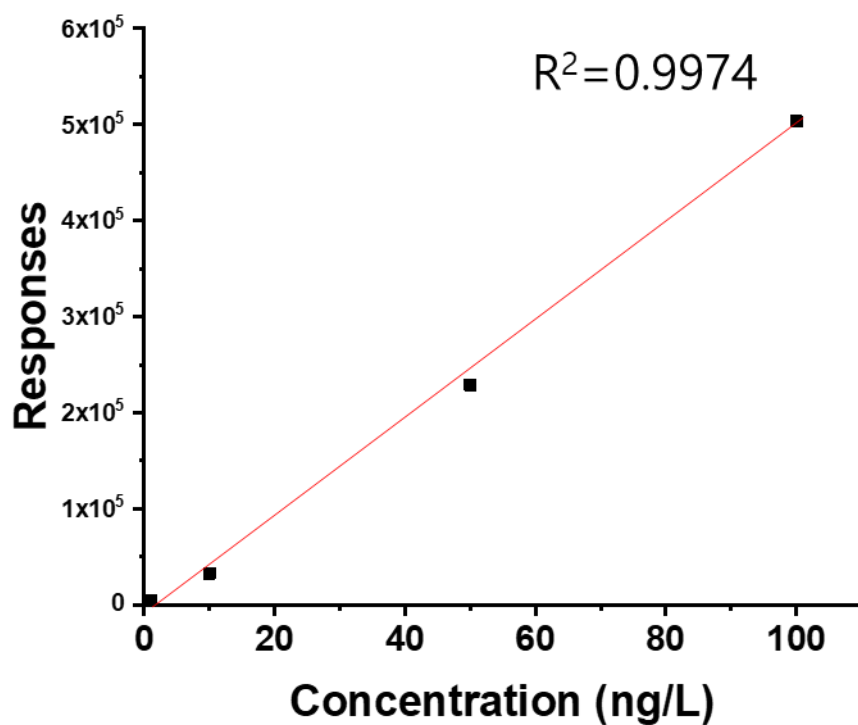
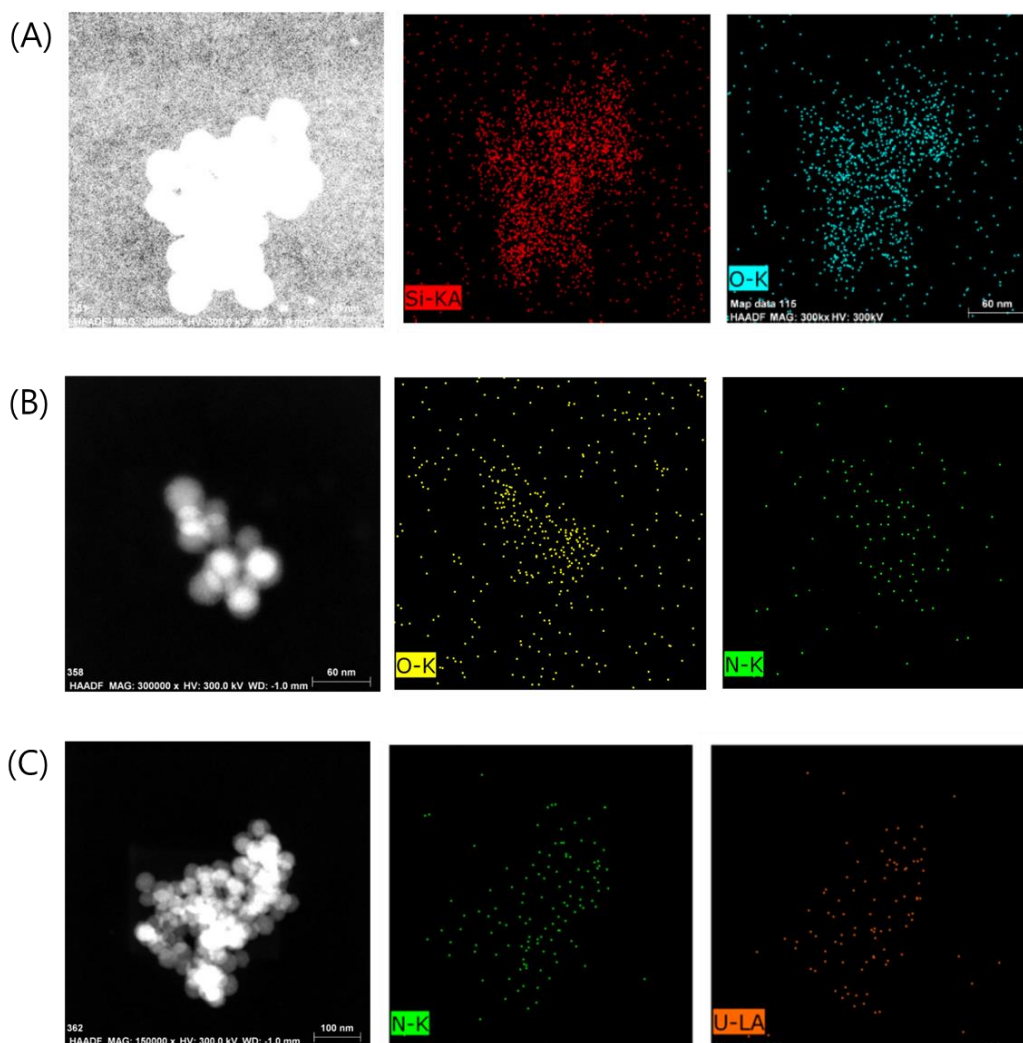


Figure S9. Linear equation of various concentrations of UO_2^{2+} .

Table S1. Adsorption Capacities of **MPS-1** for UO_2^{2+} (100 ppb) solution.^a

MPS-1 (mg)	Removal of UO_2^{2+} (%)	SD	RSD(%)
1	69.65	4.14	13.64
3	96.28	0.44	12.02
5	95.16	0.60	12.47

MPS (mg)	Removal of UO_2^{2+} (%)	SD	RSD(%)
5	1.85	0.18	9.72

^aSD = standard deviation; RSD = relative standard deviation.**Figure S10.** TEM EDX mapping of (A) **MPS** (B) **MPS-1** and (C) **MPS-1** with UO_2^{2+} .**Table S2.** Adsorption Capacities of **MPS-1** (5 mg) with various metal ions (100 ppb) solution.

Removal of metal ion (%)								
UO_2^{2+}	Na^+	Mg^{2+}	Ca^{2+}	Cu^{2+}	Ag^+	Ni^{2+}	Mn^{2+}	Pb^{2+}
95.26	5.21	15.18	42.31	27.54	21.87	23.11	17.59	26.62