Supporting information for

Metal Ion-Chelated Tannic Acid Coating for Hemostatic Dressing

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Preparation of Fluorescently Labeled Proteins

Labeling of bovine serum albumin (BSA), human Immunoglobulin G (IgG) and fibrinogen (Fgn) with fluorescein isothiocyanate (FITC) was performed according to the instruction of the FITC manufacturer. Briefly, FITC was dissolved in anhydrous DMSO at 1 mg mL⁻¹, and a solution of 2 mg mL⁻¹ of protein in 0.1 M sodium carbonate buffer, pH 9.8 was prepared. For each 1 mL of protein solution, 50 μ L of FITC solution was added slowly, while gently and continuously stirring the protein solution. The reaction was incubated in the dark for 12 h at 4 °C. Afterwards, excess FITC was removed by dialysis membrane (7 kDa).

The ratio of fluorescein to protein (F/P) of the product was estimated by measuring the absorbance at 498 nm and 280 nm via a UV spectrophotometer (Figure S1). The F/P molar ratio is defined as the ratio of moles of FITC to moles of protein in the conjugate. It was calculated from the following formulas:

$$MolarF / P = \frac{MW}{389} \times \frac{A_{498}E^{0.1\%}}{195[A_{280} - (0.35 \times A_{498})]}$$

Where, MW is molecular weight of the protein; $E^{0.1\%}$ is the absorption at 280 nm of a protein at 1.0 mg mL⁻¹; A₂₈₀ and A₄₉₈ is the absorbance of the conjugate sample at 280 nm and 498 nm, respectively.

Table S1. Physicochemical properties of the three model proteins and their coupling efficiency with FITC. MW: molecular weight; pI: isoelectric point; F/P molar ratio: the ratio of mole of FITC to mole of protein in the conjugate.

Protein	BSA	lgG	Fgn
MW(kDa)	66	150	340
pl	4.6	8.6	5.6
F/P molar ratio	3	3	2
Shape	Prism shape	Sphere	Cylinder
Dimensions (nm)	4.0*4.0*14.0	6.6*7.7*10.1	diameter 6.5, length 47.5
Plasma concentration (mg/mL)	40	6.5-16.5	2.0-4.0
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Figure S1. Absorption spectrum of three FITC labeled proteins.



Figure S2. Fluorescence intensity of model protein adsorption on the TA coated-silicon slide.