

## Supplementary information

### A Platinum(II) Molecular Hinge with Motions Visualized by Phosphorescence Changes

Yeye Ai<sup>[a,b]</sup> Michael Ho-Yeung Chan,<sup>[b]</sup> Alan Kwun-Wa Chan,<sup>[b]</sup> Maggie Ng,<sup>[b]</sup> Yongguang Li,\*<sup>[a]</sup> Vivian Wing-Wah Yam\*<sup>[a,b]</sup>

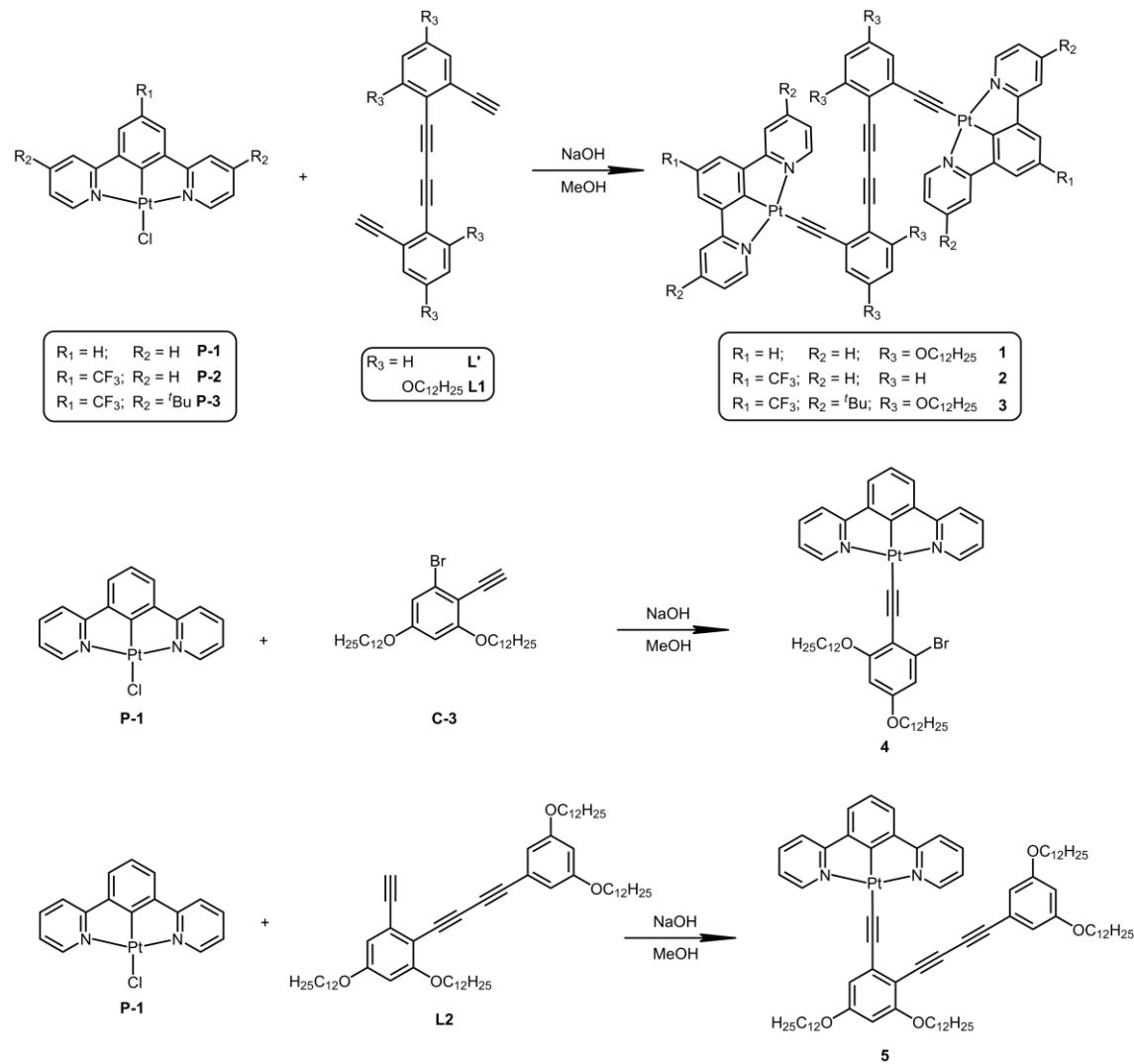
[a] Lehn Institute of Functional Materials, School of Chemistry, Sun Yat-Sen University, Guangzhou 510275, P. R. China

[b] Institute of Molecular Functional Materials [Areas of Excellence Scheme, University Grants Committee (Hong Kong)] and Department of Chemistry, The University of Hong Kong, Pokfulam Road, Hong Kong, P. R. China

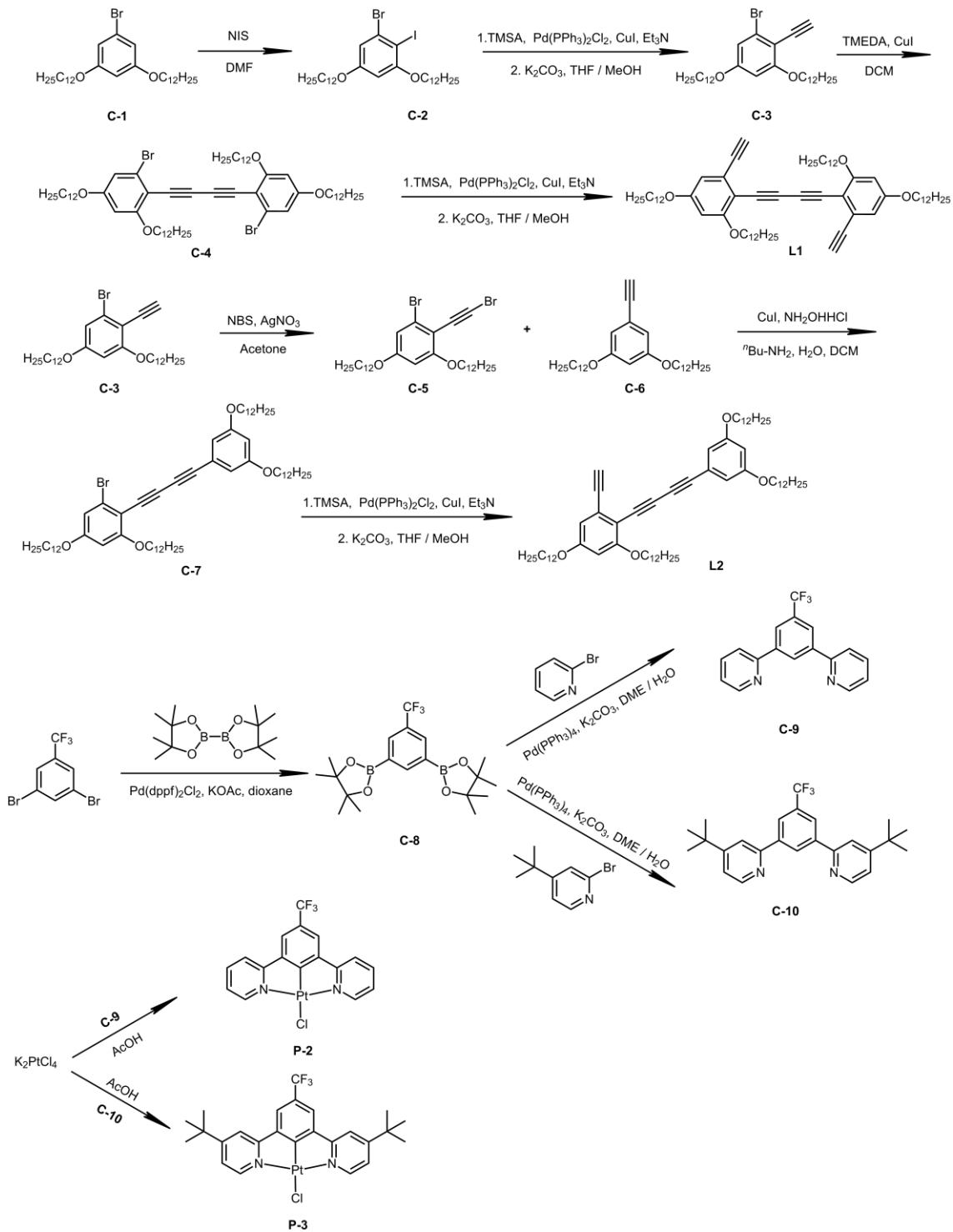
E-mail: liyongguang@mail.sysu.edu.cn; wwyam@hku.hk

## Material and Methods

The synthetic routes of **1–5** were shown in Scheme S1. Complex **6**, chloroplatinum(II) precursors **P-1** and **P-2**, compounds **C-1** and **C-6**, and 1,1'-(1,3-butadiyne-1,4-diyl)bis[2-ethynylbenzene] (**L'**) in Schemes S1 and S2 were synthesized according to the reported literatures (1–4). Compound **L1**, **L2** and **P-3** were synthesized according to Scheme S2.



**Scheme S1.** Synthetic routes of **1–5**.



**Scheme S2.** Synthetic routes of ligands and chloroplatinum(II) precursors.

## Experimental Procedures

### Synthesis of alkynylplatinum(II) complexes

**Synthesis of 1.** A mixture of **L1** (30 mg, 0.030 mmol) and NaOH (6 mg, 0.15 mmol) in a methanol (50 mL) solution was stirred for 20 minutes at room temperature under nitrogen atmosphere. Subsequently,  $[(\text{N}^{\wedge}\text{C}^{\wedge}\text{N})\text{PtCl}]$  (29 mg, 0.064 mmol) was added to the above reaction mixture. The mixture was stirred for 24 hours at room temperature. Slow precipitation with methanol (100 mL) in batches for 10 min afforded the pure product as a yellow solid. Yield: 43 mg, 0.023 mmol, 77 %;  $^1\text{H}$  NMR (400 MHz, 298 K,  $\text{CDCl}_3$ , relative to  $\text{Me}_4\text{Si}$ ,  $\delta$  / ppm): 0.85–0.90 (m, 12H,  $-\text{CH}_3$ ), 1.25–1.52 (m, 72H,  $-\text{CH}_2-$ ), 1.75–1.91 (m, 8H,  $-\text{CH}_2-$ ), 3.95–4.02 (m, 8H,  $-\text{OCH}_2-$ ), 6.28 (d,  $J = 2.1$  Hz, 2H, phenyl), 6.59 (d,  $J = 2.1$  Hz, 2H, phenyl), 6.71 (t,  $J = 7.6$  Hz, 2H, phenyl), 6.96 (d,  $J = 7.6$  Hz, 4H, phenyl), 7.26–7.32 (m, 8H, pyridyl), 7.71 (t,  $J = 8.4$  Hz, 4H, pyridyl), 9.49 (d,  $J = 6.1$  Hz, 4H, pyridyl); MS (MALDI):  $m/z$ : 1837.914 [ $M+\text{H}]^+$ ; elemental analysis calcd (%) for  $\text{C}_{100}\text{H}_{126}\text{N}_4\text{O}_4\text{Pt}_2$ : C 65.33, H 6.91, N 3.05; found: C 65.54, H 6.78, N 2.95; mp/ $^{\circ}\text{C}$ : complex decomposed before melting.

**Synthesis of 2.** A mixture of **L'** (20 mg, 0.080 mmol) and NaOH (8 mg, 0.32 mmol) in a methanol (50 mL) solution was stirred for 20 min at room temperature under nitrogen atmosphere. Subsequently,  $[(\text{N}^{\wedge}\text{C}^{\wedge}\text{N}-\text{CF}_3)\text{PtCl}]$  (89 mg, 0.17 mmol) was added to the above reaction mixture. The mixture was stirred for 24 hours at room temperature. Slow precipitation with methanol (100 mL) in batches for 10 min afforded the pure product as an orange solid. Yield: 78 mg, 0.063 mmol, 79 %;  $^1\text{H}$  NMR (400 MHz, 298 K,  $\text{CDCl}_3$ , relative to  $\text{Me}_4\text{Si}$ ,  $\delta$  / ppm): 7.06 (s, 4H, phenyl), 7.15 (t,  $J = 7.6$  Hz, 2H, phenyl), 7.24–7.32 (m, 10H, pyridyl and phenyl), 7.40 (d,  $J = 7.6$  Hz, 2H, phenyl), 7.55 (d,  $J = 7.7$  Hz, 2H, phenyl), 7.78 (t,  $J = 7.5$  Hz, 4H, pyridyl), 9.46 (d,  $J = 4.9$  Hz, 4H, pyridyl); MS (MALDI):  $m/z$ : 1237.945 [ $M+\text{H}]^+$ ; elemental analysis calcd (%) for  $\text{C}_{54}\text{H}_{28}\text{F}_6\text{N}_4\text{Pt}_2 \bullet 0.5\text{CH}_2\text{Cl}_2$ : C 51.17, H 2.29, N 4.38; found: C 51.07, H 2.50, N 4.32; mp/ $^{\circ}\text{C}$ : complex decomposed before melting.

Synthesis of 3. The procedures were similar to that of **1**, except **P-3** (50 mg, 0.078 mmol) was used in place of  $[(N^{\wedge}C^{\wedge}N)PtCl]$  to give the product as a deep-red solid. Yield: 52 mg, 0.024 mmol, 64 %;  $^1H$  NMR (400 MHz, 298 K,  $CDCl_3$ , relative to  $Me_4Si$ ,  $\delta$  / ppm): 0.86–0.89 (m, 12H,  $-CH_3$ ), 1.27–1.49 (m, 108H,  $-CH_2-$  and  $-CH_3$ ), 1.74–1.81 (m, 4H,  $-CH_2-$ ), 1.88–1.95 (m, 4H,  $-CH_2-$ ), 3.94 (t,  $J$  = 6.5 Hz, 4H,  $-OCH_2-$ ), 4.05 (t,  $J$  = 6.5 Hz, 4H,  $-OCH_2-$ ), 6.29 (d,  $J$  = 1.8 Hz, 2H, phenyl), 6.50 (d,  $J$  = 1.8 Hz, 2H, phenyl), 6.94 (s, 4H, phenyl), 7.26 (s, 4H, pyridyl, mixed with solvent peak), 7.34 (d,  $J$  = 5.9 Hz, 4H, pyridyl), 9.30 (d,  $J$  = 5.9 Hz, 4H, pyridyl); MS (MALDI):  $m/z$ : 2199.628  $[M+H]^+$ ; elemental analysis calcd (%) for  $C_{118}H_{156}F_6N_4O_4Pt_2$ : C 64.48, H 7.10, N 2.55; found: C 64.20, H 6.98, N 2.77; mp/ $^{\circ}C$ : 197–199.

Synthesis of 4. The procedures were similar to that of **1**, except **C-3** (33 mg, 0.060 mmol) was used in place of **L1** to give the product as a yellow solid. Yield: 25 mg, 0.025 mmol, 47 %;  $^1H$  NMR (400 MHz, 298 K,  $CDCl_3$ , relative to  $Me_4Si$ ,  $\delta$  / ppm): 0.85–0.90 (m, 6H,  $-CH_3$ ), 1.20–1.48 (m, 36H,  $-CH_2-$ ), 1.72–1.80 (m, 2H,  $-CH_2-$ ), 1.89–1.96 (m, 2H,  $-CH_2-$ ), 3.93 (t,  $J$  = 6.5 Hz, 2H,  $-OCH_2-$ ), 3.03 (t,  $J$  = 6.8 Hz, 2H,  $-OCH_2-$ ), 6.40 (d,  $J$  = 1.8 Hz, 1H, phenyl), 6.76 (d,  $J$  = 1.8 Hz, 1H, phenyl), 7.17–7.26 (m, 3H, pyridyl and phenyl, mixed with solvent peak), 7.52 (d,  $J$  = 7.7 Hz, 2H, phenyl), 7.67 (d,  $J$  = 8.0 Hz, 2H, pyridyl), 7.92 (t,  $J$  = 7.6 Hz, 2H, pyridyl), 9.79 (d,  $J$  = 5.4 Hz, 2H, pyridyl); MS (MALDI):  $m/z$ : 975.541  $[M+H]^+$ ; elemental analysis calcd (%) for  $C_{48}H_{63}N_2O_2BrPt$ : C 59.20, H 6.47, N 2.88; found: C 59.49, H 6.20, N 2.75; mp/ $^{\circ}C$ : 121–123.

Synthesis of 5. The procedures were similar to that of **1**, except **L2** (50 mg, 0.052 mmol) was used in place of **L1** to give the product as a yellow solid. Yield: 45 mg, 0.032 mmol, 66 %;  $^1H$  NMR (400 MHz, 298 K,  $CDCl_3$ , relative to  $Me_4Si$ ,  $\delta$  / ppm): 0.85–0.89 (m, 6H,  $-CH_3$ ), 1.26–1.50 (m, 36H,  $-CH_2-$ ), 1.69–1.88 (m, 4H,  $-CH_2-$ ), 3.83 (t,  $J$  = 6.5 Hz, 2H,  $-OCH_2-$ ), 3.97–4.03 (m, 2H,  $-OCH_2-$ ), 6.30 (d,  $J$  = 2.0 Hz, 1H, phenyl), 6.41 (s, 1H, phenyl), 6.45 (d,  $J$  = 1.4 Hz, 2H, phenyl), 6.73 (d,  $J$  = 2.0 Hz,

1H, phenyl), 7.20 (t,  $J$  = 7.8 Hz, 1H, phenyl), 7.30 (t,  $J$  = 6.0 Hz, 2H, pyridyl, mixed with solvent peak), 7.50 (d,  $J$  = 7.8 Hz, 2H, phenyl), 7.64 (d,  $J$  = 8.2 Hz, 2H, pyridyl), 7.85 (t,  $J$  = 8.2 Hz, 2H, pyridyl), 9.73 (d,  $J$  = 6.0 Hz, 2H, pyridyl); MS (MALDI):  $m/z$ : 1389.135 [ $M+H$ ]<sup>+</sup>; elemental analysis calcd (%) for C<sub>82</sub>H<sub>116</sub>N<sub>2</sub>O<sub>4</sub>Pt: C 70.94, H 8.36, N 2.02; found: C 70.78, H 8.49, N 2.25; mp/°C: 64–66.

### Synthesis of chloroplatinum(II) precursor complexes

Synthesis of P-3. The compound was synthesized according to the modification of a reported procedure (5). **C-10** (130 mg, 0.31 mmol) and K<sub>2</sub>PtCl<sub>4</sub> (125 mg, 0.30 mmol) were added into glacial acetic acid (15 mL), and the mixture was heated for 110 °C for 3 days. After cooling to room temperature, slow participation with ethanol (30 mL) in batches afforded the pure product as an orange solid. Yield: 110 mg, 0.17 mmol, 57 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si,  $\delta$  / ppm): 1.43 (s, 18H, –CH<sub>3</sub>), 7.35 (d,  $J$  = 6.1 Hz, 2H, pyridyl), 7.69–7.73 (m, 4H, pyridyl and phenyl), 9.24 (d,  $J$  = 6.1 Hz, 2H, pyridyl); MS (MALDI):  $m/z$ : 606.278 [ $M-Cl$ ]<sup>+</sup>; elemental analysis calcd (%) for C<sub>25</sub>H<sub>26</sub>N<sub>2</sub>F<sub>3</sub>ClPt: C 46.80, H 4.06, N 4.37; found: C 46.65, H 4.26, N 4.50; mp/°C: > 300.

### Synthesis of ligand precursors

Synthesis of C-2. The compound was synthesized according to the modification of a reported procedure (6). To a solution of **C-1** (4.4 g, 8.4 mmol) in dry DMF (50 mL) was added *N*-iodosuccinimide (2.0 g, 8.8 mmol). The mixture was heated at 70 °C for overnight under nitrogen atmosphere. Upon completion of the reaction, dichloromethane (200 mL) was added into the reaction mixture, which was washed with deionized water for 2 times. The organic layer was dried over anhydrous MgSO<sub>4</sub>. Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography on silica gel using dichloromethane–petroleum ether (1:20 v / v) as the eluent to give **C-2** as a white solid. Yield: 3.9 g, 6.0 mmol, 71 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si,  $\delta$  / ppm): 0.82 (t,  $J$  = 6.7 Hz, 6H, –CH<sub>3</sub>), 1.21–1.52 (m, 36H, –CH<sub>2</sub>–), 1.65–1.80 (m, 4H, –CH<sub>2</sub>–), 3.81–3.86 (m,

4H, –OCH<sub>2</sub>–), 6.24 (d, *J* = 2.4 Hz, 1H, phenyl), 6.77 (d, *J* = 2.4 Hz, 1H, phenyl); MS (MALDI): *m/z*: 652.35 [M]<sup>+</sup>.

**Synthesis of C-3.** The compound was synthesized by a Sonogashira coupling reaction according to the modification of reported procedure (7). **C-2** (1.2 g, 1.8 mmol), Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (66 mg, 0.09 mmol), CuI (18 mg, 0.09 mmol) and trimethylsilylacetylene (TMSA, 1.0 mL, 7.2 mmol) were added to the flask with degassed triethylamine (60 mL), and the mixture was stirred overnight at room temperature under nitrogen atmosphere. Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography on silica gel using dichloromethane–petroleum ether (1:20 v / v) as the eluent to give **C-3-TMS** (0.9 g, 1.45 mmol) as a light-yellow solid. Subsequently, to the THF–methanol (1:1 v / v, 50 mL) solution mixture of **C-3-TMS** (0.5 g, 0.8 mmol) was added K<sub>2</sub>CO<sub>3</sub> (167 mg, 1.2 mmol). The reaction mixture was stirred for 2 hours at room temperature. Then the reaction mixture was poured into dichloromethane (100 mL) and was washed with deionized water for 2 times. The organic layer was dried over anhydrous MgSO<sub>4</sub>. Removal of the solvent under reduced pressure afforded the crude product, which was used for subsequent reactions without further purification. Yield: 390 mg, 0.71 mmol, 69 %; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, 298 K, relative to Me<sub>4</sub>Si,  $\delta$  / ppm): 0.88 (t, *J* = 6.6 Hz, 6H, –CH<sub>3</sub>), 1.26–1.52 (m, 36H, –CH<sub>2</sub>–), 1.72–1.85 (m, 4H, –CH<sub>2</sub>–), 3.90–3.99 (m, 4H, –OCH<sub>2</sub>–), 6.32 (d, *J* = 1.5 Hz, 1H, phenyl), 6.69 (d, *J* = 1.5 Hz, 1H, phenyl); MS (MALDI): *m/z*: 551.170 [M+H]<sup>+</sup>.

**Synthesis of C-4:** The compound was synthesized according to the modification of a reported procedure (8). CuI (82 mg, 0.44 mmol) was dispersed in a solution of dichloromethane (50 mL), *N,N,N',N'*-tetramethylethylenediamine (0.16 mL, 1.1 mmol) was added and the mixture was stirred for 5 min. The dichloromethane solution of **C-3** (1.2 g, 2.2 mmol) was then added to the mixture and stirred for overnight at room temperature at the oxygen atmosphere. Upon completion of the reaction, the reaction mixture was washed with deionized water for 2 times and the

organic layer was dried over anhydrous MgSO<sub>4</sub>. Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography on silica gel using dichloromethane–petroleum ether (1:5 v / v) as the eluent to give **C-4** as a pale-yellow solid. Yield: 900 mg, 0.82 mmol, 75 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 0.85–0.89 (m, 12H, –CH<sub>3</sub>), 1.23–1.50 (m, 72H, –CH<sub>2</sub>–), 1.72–1.86 (m, 8H, –CH<sub>2</sub>–), 3.91–3.99 (m, 8H, –OCH<sub>2</sub>–), 6.34 (d, J = 2.0 Hz, 2H, phenyl), 6.70 (d, J = 2.0 Hz, 2H, phenyl); MS (MALDI): *m/z*: 1095.655 [M+H]<sup>+</sup>.

**Synthesis of C-5 and C-7.** The compounds were synthesized according to the modification of a reported procedure (9). To a solution of **C-3** (1.32 g, 2.0 mmol) and AgNO<sub>3</sub> (0.41 g, 2.4 mmol) in acetone (30 mL) was added. NBS (0.53 g, 3.0 mmol), and the mixture was stirred in the dark at room temperature for overnight. Removal of the solvent under reduced pressure afforded the crude product **C-5** (1.0g, 1.6 mmol) which was used without further purification. CuI (0.3 g, 1.6 mmol) was dispersed in a mixture of butylamine (6 mL) and water (8 mL), and the mixture was stirred for 5 min. Subsequently, NH<sub>2</sub>OH•HCl was added in portions until the solution turned colorless. **C-6** (0.6 g, 1.3 mmol) was added and the mixture was stirred for 5 min. The dichloromethane solution of **C-5** (0.8 g, 1.3 mmol) was then added dropwise to the mixture and stirred for another 30 min at room temperature. Upon completion of the reaction, the reaction mixture was washed with deionized water for 2 times and the organic layer was dried over anhydrous MgSO<sub>4</sub>. Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography on silica gel using dichloromethane–petroleum ether (1:5 v / v) as the eluent to give **C-7** as a yellow solid. Yield: 630 mg, 0.62 mmol, 38 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 0.85–0.90 (m, 12H, –CH<sub>3</sub>), 1.24–1.50 (m, 72H, –CH<sub>2</sub>–), 1.72–1.87 (m, 8H, –CH<sub>2</sub>–), 3.89–4.00 (m, 8H, –OCH<sub>2</sub>–), 6.35 (d, J = 1.8 Hz, 1H, phenyl), 6.46 (s, 1H, phenyl), 6.65 (d, J = 2.0 Hz, 2H, phenyl), 6.71 (d, J = 1.8 Hz, 1H, phenyl); MS (MALDI): *m/z*: 1019.942 [M+H]<sup>+</sup>.

Synthesis of C-8. The compounds were synthesized according to the modification of a reported procedure (10). 3,5-Dibromobenzotrifluoride (1.0 g, 3.3 mmol), bispinacolatodiboron (1.8 g, 7.2 mmol), Pd(dppf)Cl<sub>2</sub> (178 mg, 0.23 mmol) and KOAc (2.6 g, 26.2 mmol) were added to the flask with degassed dioxane (70 mL), and the mixture was stirred at 110 °C for 24 hours under nitrogen atmosphere. Upon cooling to the room temperature, ethyl acetate (150 mL) was added. The reaction mixture was washed with deionized water for 2 times and the organic layer was dried over anhydrous MgSO<sub>4</sub>. Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography on silica gel using ethyl acetate–petroleum ether (1:15 v / v) as the eluent to give **C-8** as a white solid. Yield: 1.0 g, 2.5 mmol, 78 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 1.35 (s, 2H, –CH<sub>3</sub>), 8.13 (s 2H, phenyl), 8.41 (s 1H, phenyl); MS (MALDI): *m/z*: 399.55 [M+H]<sup>+</sup>.

Synthesis of C-9. The compounds were synthesized according to the modification of a reported procedure (10). 2-Bromopyridine (1.1 g, 6.7 mmol), **C-8** (1.0 g, 3.0 mmol), Pd(PPh<sub>3</sub>)<sub>4</sub> (345 mg, 0.30 mmol) and Na<sub>2</sub>CO<sub>3</sub> (2.1 g, 15.0 mmol) were added to the flask with degassed DME/H<sub>2</sub>O mixture (60 mL, 1:1, v/v), and the mixture was stirred at 100 °C for 24 hours under nitrogen atmosphere. Upon cooling to the room temperature, ethyl acetate (100 mL) was added. The reaction mixture was washed with deionized water for 2 times and the organic layer was dried over anhydrous MgSO<sub>4</sub>. Removal of the solvent under reduced pressure afforded the crude product, which was purified by column chromatography on silica gel using ethyl acetate–petroleum ether (1:8 v / v) as the eluent to give **C-9** as a white solid. Yield: 0.7 g, 2.3 mmol, 77 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 7.32 (t, *J* = 6.2 Hz, 2H, pyridyl), 7.83 (t, *J* = 7.0 Hz, 2H, pyridyl), 7.89 (d, *J* = 7.9 Hz, 2H, pyridyl), 8.34 (s, 2H, phenyl), 8.75 (d, *J* = 4.6 Hz, 2H, pyridyl), 8.83 (s, 1H, phenyl); MS (MALDI): *m/z*: 301.23 [M+H]<sup>+</sup>;

### Synthesis of C-10.

The procedures were similar to that of compound **C-9**, except 2-bromo-4-(tert-butyl)pyridine (296 mg, 1.38 mmol) was used in place of 2-Bromopyridine. The crude product was purified by column chromatography on silica gel using ethyl acetate–petroleum ether (1:10 v / v) as the eluent to give the desired product as a gray-white solid. Yield: 130 mg, 0.31 mmol, 50 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 1.40 (s, 18H, –CH<sub>3</sub>), 7.31 (d, *J* = 5.2 Hz, 2H, pyridyl), 7.80 (s, 2H, pyridyl), 8.28 (s, 2H, phenyl), 8.65 (d, *J* = 5.2 Hz, 2H, pyridyl), 8.74 (s, 1H, phenyl); MS (MALDI): *m/z*: 413.34 [M+H]<sup>+</sup>.

### **Synthesis of alkynyl ligands**

Synthesis of L1. The procedures were similar to that of compound **C-3**, except **C-4** (200 mg, 0.2 mmol) was used in place of **C-2** with heating at 70 °C. The crude product was purified by column chromatography on silica gel using dichloromethane–petroleum ether (1:3 v / v) as the eluent to give the desired product as an off-white solid. Yield: 110 mg, 0.11 mmol, 52 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 0.86–0.89 (m, 12H, –CH<sub>3</sub>), 1.20–1.47 (m, 72H, –CH<sub>2</sub>–), 1.73–1.87 (m, 8H, –CH<sub>2</sub>–), 3.22 (s, 2H, –C≡C–H), 3.93 (t, *J* = 6.5 Hz, 4H, –OCH<sub>2</sub>–), 4.00 (t, *J* = 6.7 Hz, 4H, –OCH<sub>2</sub>–), 6.44 (d, *J* = 1.8 Hz, 2H, phenyl), 6.62 (d, *J* = 1.8 Hz, 2H, phenyl); MS (MALDI): *m/z*: 986.669 [M]<sup>+</sup>; mp/°C: 53–55.

Synthesis of L2. The procedures were similar to that of compound **L1**, except **C-7** (400 mg, 0.39 mmol) was used in place of **C-4**. The crude product was purified by column chromatography on silica gel using dichloromethane–petroleum ether (1:3 v / v) as the eluent to give the desired product as a yellow solid. Yield: 275 mg, 0.29 mmol, 74 %; <sup>1</sup>H NMR (400 MHz, 298 K, CDCl<sub>3</sub>, relative to Me<sub>4</sub>Si, δ / ppm): 0.83–0.90 (m, 12H, –CH<sub>3</sub>), 1.24–1.51 (m, 72H, –CH<sub>2</sub>–), 1.72–1.87 (m, 8H, –CH<sub>2</sub>–), 3.34 (s, 1H, –C≡C–H), 3.89–4.01 (m, 8H, –OCH<sub>2</sub>–), 6.42 (d, *J* = 1.3 Hz, 1H, phenyl), 6.46 (s, 1H, phenyl), 6.60 (d, *J* = 1.3 Hz, 1H, phenyl), 6.65 (d, *J* = 1.8 Hz, 2H, phenyl); MS (MALDI): *m/z*: 962.981 [M]<sup>+</sup>; mp/°C: 45–47.

## **Physical Measurements and Instrumentations**

All  $^1\text{H}$  NMR spectra were recorded on a Bruker DPX 400 FT-NMR spectrometer 400 MHz. Matrix assisted laser desorption ionization time of flight mass were performed on a Bruker ultrafleXtreme MALDI-TOF/TOF mass spectrometer using positive-ion mode (Bruker Daltonics). Elemental analyses of the complexes were performed on a Vario EL elemental analyzer. Melting points (mp) were measured on a Buchi M-560 apparatus with heating rate of 1 °C/min.

Electronic absorption spectra and temperature-dependent electronic absorption spectra were recorded using a Shimadzu UV-3600 spectrophotometer and a Varian Cary 50 UV-vis spectrophotometer, respectively, with the monitoring of temperature using the Varian Cary single-cell Peltier thermostat. The photoluminescence spectra were measured on Edinburgh Instruments FS5 and Edinburgh Instruments FLS980 fluorescence spectrophotometers. Emission lifetime measurements were performed using LP980 transient absorption spectrometer (Edinburgh Instruments Ltd, UK). The excitation source was the 355 nm output (third harmonic) of an Nd: YAG laser (Quanta-Ray Lab130 Pulsed Nd:YAG Laser) and the probe light source was a Xe900 450 W xenon arc lamp. All solutions for photophysical studies were prepared under high vacuum in a 10-cm<sup>3</sup> round-bottomed flask equipped with a sidearm 1-cm fluorescence cuvette and sealed from the atmosphere by a Rotaflo HP6/6 quick-release Teflon stopper. Solutions were rigorously degassed on a high-vacuum line in a two-compartment cell with no less than four successive freeze-pump-thaw cycles. The emission quantum yields were measured on a Hamamatsu C9920-02G absolute PL quantum yield measurement system. Time-resolved emission spectra were obtained using an Edinburgh Instrument LP980KS transient absorption spectrometer equipped with a R928P photomultiplier tube, with capability for time-resolved emission measurements. Dynamic light scattering spectra were obtained using a Malvern Panalytical Zetasizer Nano ZS90 particle size analyzer. Scanning electron microscope experiments were performed by using ZEISS GeminiSEM 500. The atomic force microscopy measurements were performed on Bruker Dimension Scanfast in a tapping mode. The SEM and AFM samples were

prepared by dropcasting solutions onto silicon wafers and allowing them to dry naturally. Transmission electron micrographs (TEM) were recorded on a FEI Tecnai G2 Spirit. The samples for TEM were prepared by dropcasting solutions on the carbon-coated copper grids.

### Time-Dependent Emission and UV-Vis Spectral Changes

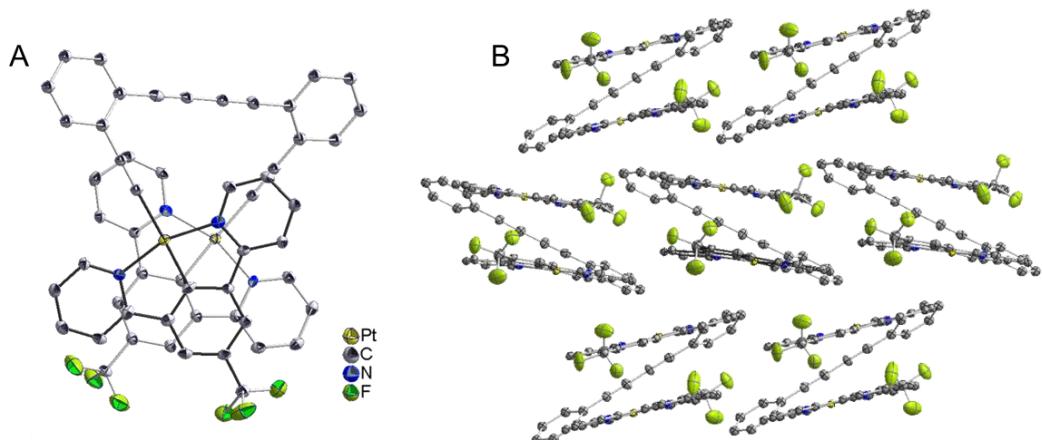
1. Solutions of **1** in dichloromethane (0.3 mL,  $5.4 \times 10^{-4}$  M) were injected into hexane, cyclohexane, decane, or methylcyclohexane (2.7 mL) to yield the respective solution mixtures of **1** in hexane-dichloromethane, cyclohexane-dichloromethane, decane-dichloromethane, methylcyclohexane-dichloromethane (9:1, v/v) ( $5.4 \times 10^{-5}$  M). After rapid mixing, the emission intensity at 516 nm was recorded with time.
2. Solutions of **1** in dichloromethane (0.3 mL) were injected into hexane or cyclohexane (2.7 mL) to yield solution mixtures of **1** in hexane-dichloromethane, cyclohexane-dichloromethane (9:1, v/v) respectively. After rapid mixing, the absorbance intensity at 425 nm was recorded with time under different conditions.

### Crystal Structure Determination

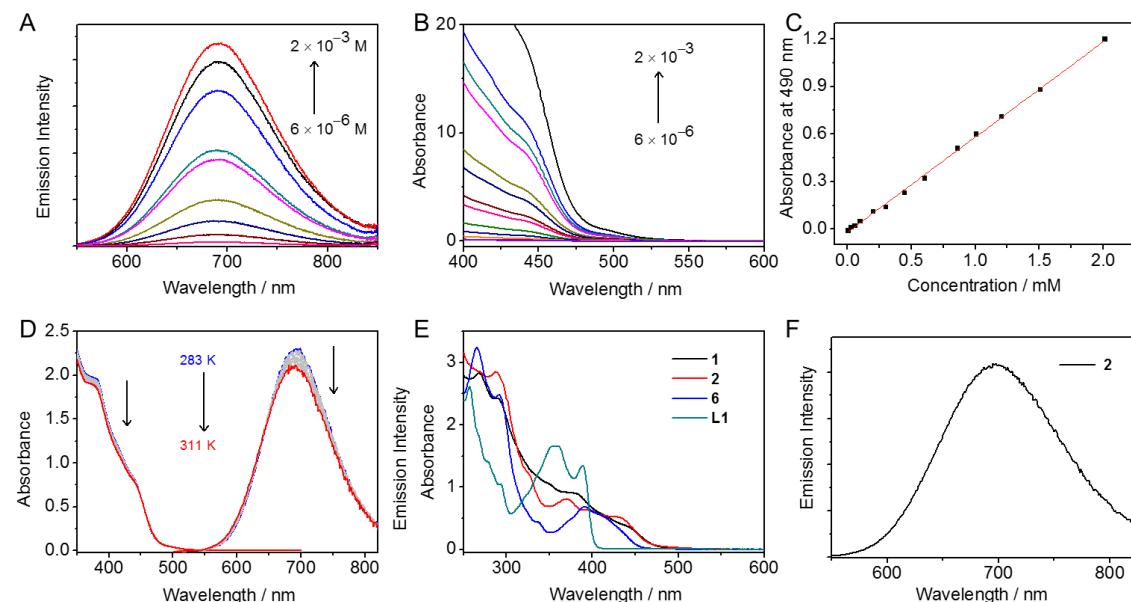
Single crystals of **2** suitable for X-ray diffraction studies were grown by vapor diffusion of diethyl ether into a concentrated dichloromethane solution of **2**. Single-crystal diffraction data of **2** were collected on an Agilent SuperNova X-ray diffractometer using micro-focus dual with X-ray Source of Cu-K $\alpha$  radiation ( $\lambda = 1.54178 \text{ \AA}$ ) at 150 K. Using Olex2 (11), the structure of the crystal was solved with the ShelXS (12) structure solution program using direct methods and refined with the XL (12) refinement package using least squares minimization. CCDC XXX contains the supplementary crystallographic data for **2**. This can be obtained free of charge from the Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

## Computational Details

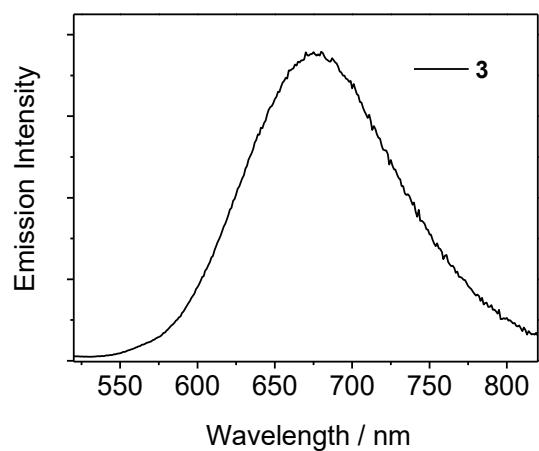
All the density functional theory (DFT) calculations were performed using the Gaussian 09 suite of programs (13). The ground-state geometries of both the open and closed forms of **1** and **3** were fully optimized by DFT with the hybrid meta exchange-correlation M06 functional (14), which is recommended for application in organometallic and metal coordination compounds and for non-covalent interactions (14, 15), in conjunction with the solvation model density (SMD) continuum method (16) using dichloromethane as the solvent. Vibrational frequencies were then computed for all stationary points to verify that each was a minimum (NIMAG = 0) on the potential energy surface (PES). The lowest-lying triplet excited state ( $T_1$ ) of the open form of **1** has also been optimized with the unrestricted method (UM06) in order to calculate its emission energy. All calculations were performed with the Stuttgart effective core potentials (ECPs) and the associated basis set to describe Pt (17), with two f-type polarization functions ( $\zeta = 0.70$  and  $0.14$ ) (18), and the 6-31G(d,p) basis set (19–22) to describe all other atoms. All the DFT calculations were performed with a pruned (99,590) grid for numerical integration. The Cartesian coordinates of the optimized ground-state geometries of the open and closed forms of **1** and **3**, and those of the optimized  $T_1$  geometry of the open form of **1**, are given in Tables S3–S7.



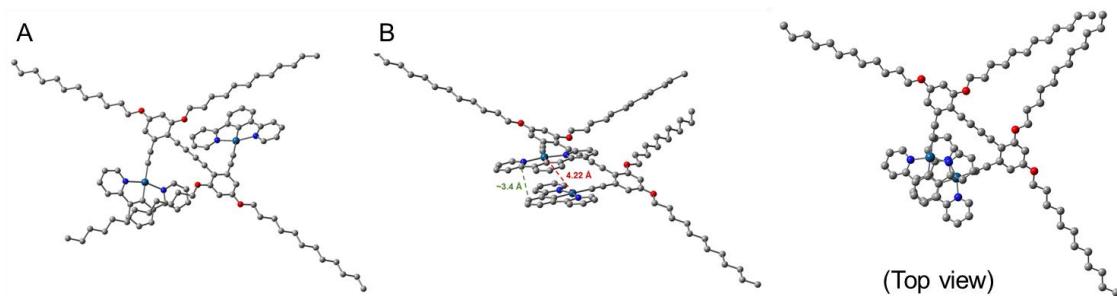
**Figure S1.** (A) Perspective drawing of **2**. Hydrogen atoms and solvent molecules are omitted for clarity. Thermal ellipsoids were shown at the 30 % probability level.(B) Crystal packing of **2** without intermolecular  $\pi-\pi$  interaction.



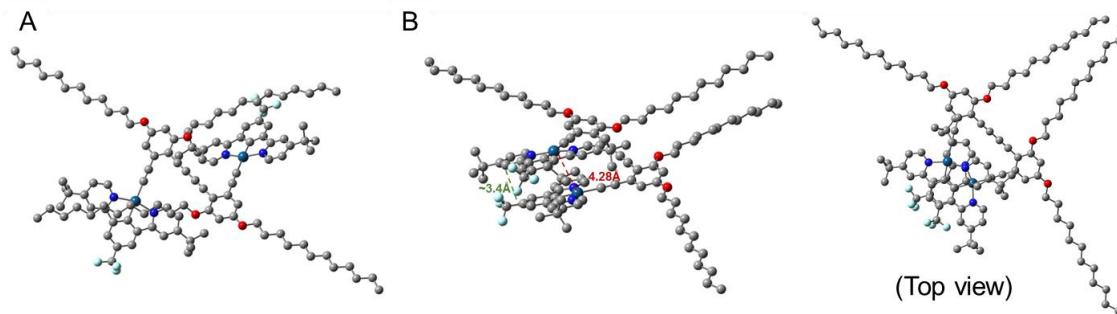
**Figure S2.** Concentration-dependent (A) emission and (B) UV-vis absorption spectra of **1** in dichloromethane from  $6.0 \times 10^{-6}$  to  $2.0 \times 10^{-3}$  M. (C) A plot of apparent absorbance at 490 nm as a function of concentration and the solid line gives the linear relationship between absorbance and concentration. The apparent absorbance is the absorbance corrected to 1-cm pathlength equivalence. (D) Temperature-dependent UV-vis absorption and emission spectra of **1** in dichloromethane at  $8.2 \times 10^{-5}$  M. (E) UV-vis absorption spectra of **1**, **2**, **6** and **L1** in dichloromethane solutions. (F) Emission spectrum of **2** in degassed dichloromethane solution ( $2.5 \times 10^{-5}$  M).



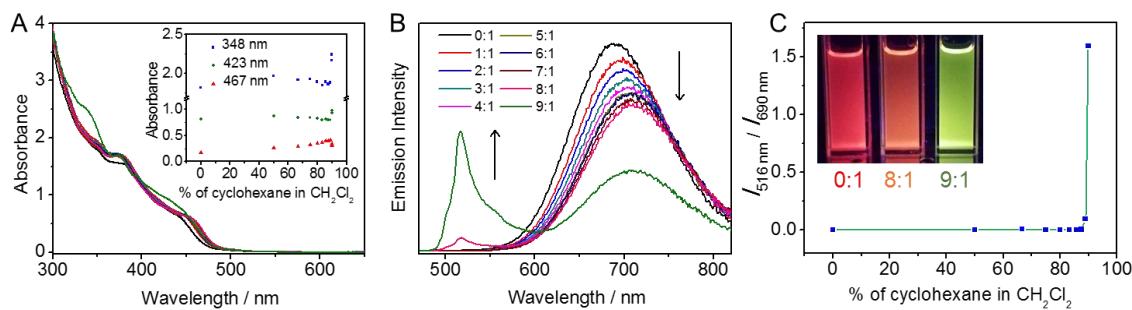
**Figure S3.** Emission spectrum of **3** in degassed dichloromethane solution ( $1.3 \times 10^{-5}$  M).



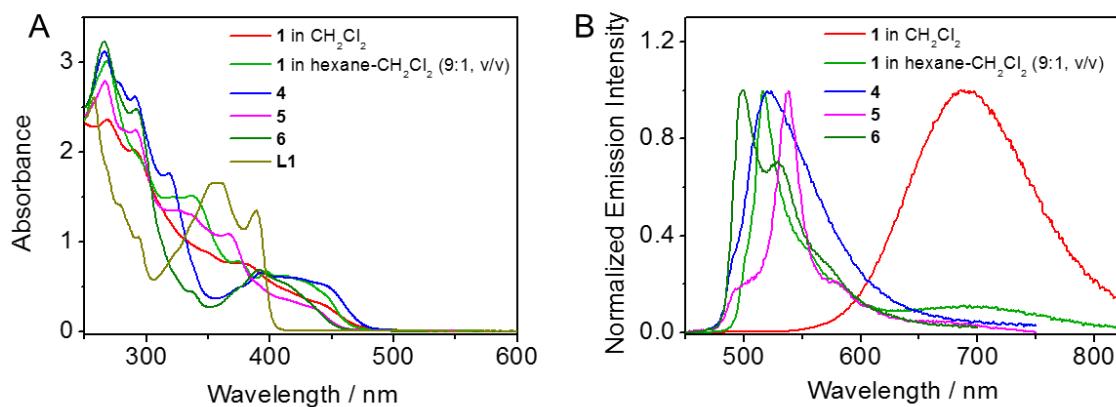
**Figure S4.** Optimized structures of molecular hinge **1** in (A) open form and (B) closed form.



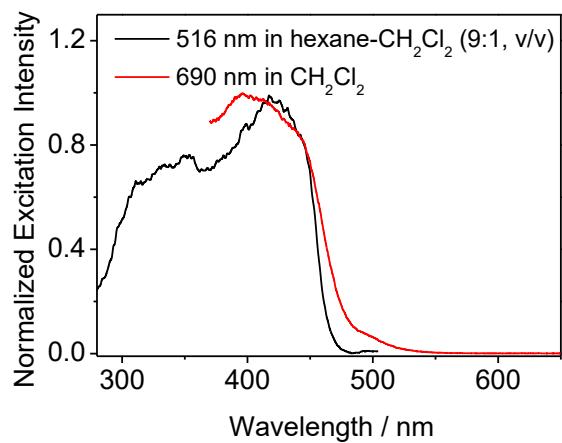
**Figure S5.** Optimized structures of molecular hinge **3** in (A) open form and (B) closed form.



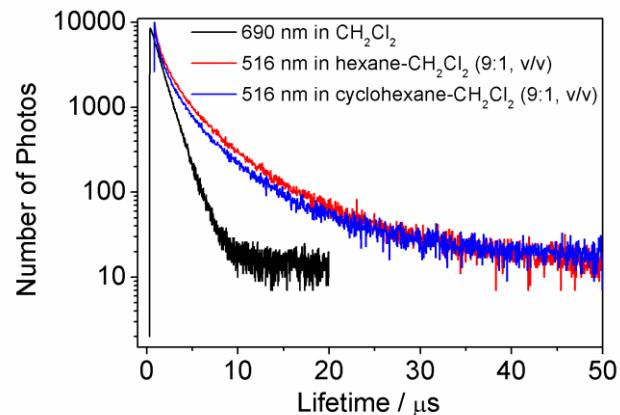
**Figure S6.** Solvent-dependent (A) UV-vis absorption and (B) emission spectra of **1** upon increasing cyclohexane portion in dichloromethane (v/v,  $5.6 \times 10^{-5}$  M). Inset of (A): Plots of the corresponding absorbance at 348, 423 and 467 nm upon increasing cyclohexane content. (C) The phosphorescence intensity ratio between 516 and 690 nm ( $I_{516} / I_{690}$ ) upon increasing cyclohexane portion in dichloromethane (v/v,  $5.6 \times 10^{-5}$  M). Inset: Photographs of different emission colors in cyclohexane-dichloromethane (0:1, 8:1, 9:1, v/v) mixtures.



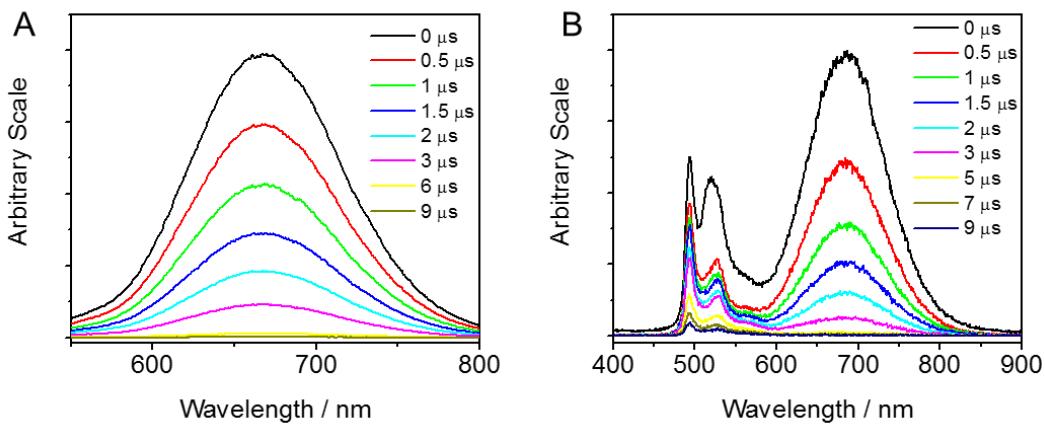
**Figure S7.** (A) UV-vis absorption of **1**, **4–6** and **L1** in dichloromethane solutions and UV-vis absorption of **1** in hexane-dichloromethane (9:1, v/v) mixture. (B) Emission spectra of **1**, **4–6** in dichloromethane solutions and emission spectrum of **1** in hexane-dichloromethane (9:1, v/v) mixture.



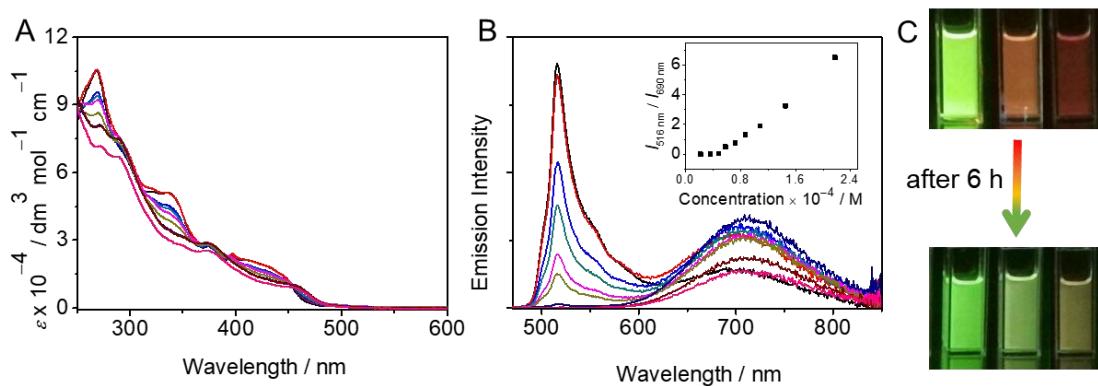
**Figure S8.** Excitation spectra of **1** with emission peaks at 690 nm in dichloromethane and 516 nm in hexane-dichloromethane (9:1, v/v) mixture respectively.



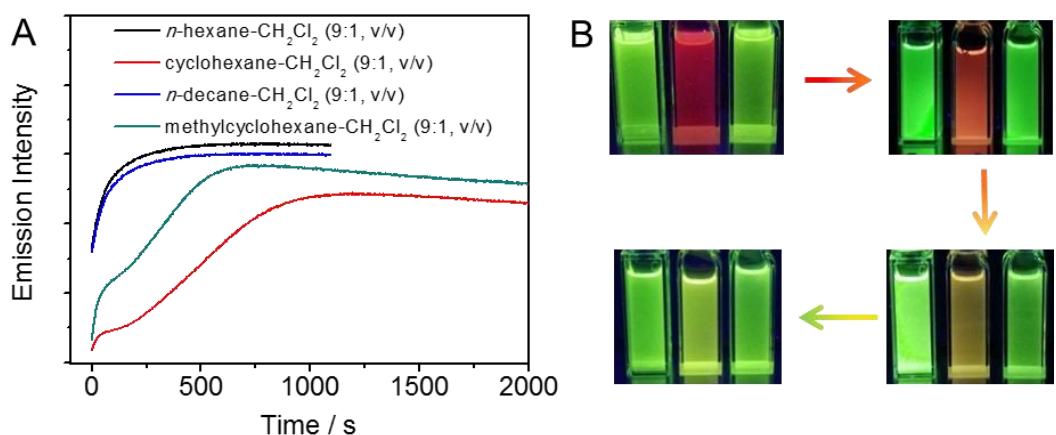
**Figure S9.** Emission decay profiles monitored at 690 nm in dichloromethane solution with closed form ( $\tau = 1.4 \mu\text{s}$ ) and 516 nm in hexane/cyclohexane-dichloromethane (9:1, v/v) mixtures with open form ( $\tau = 3.6 \mu\text{s}$ ) after a laser pulse.



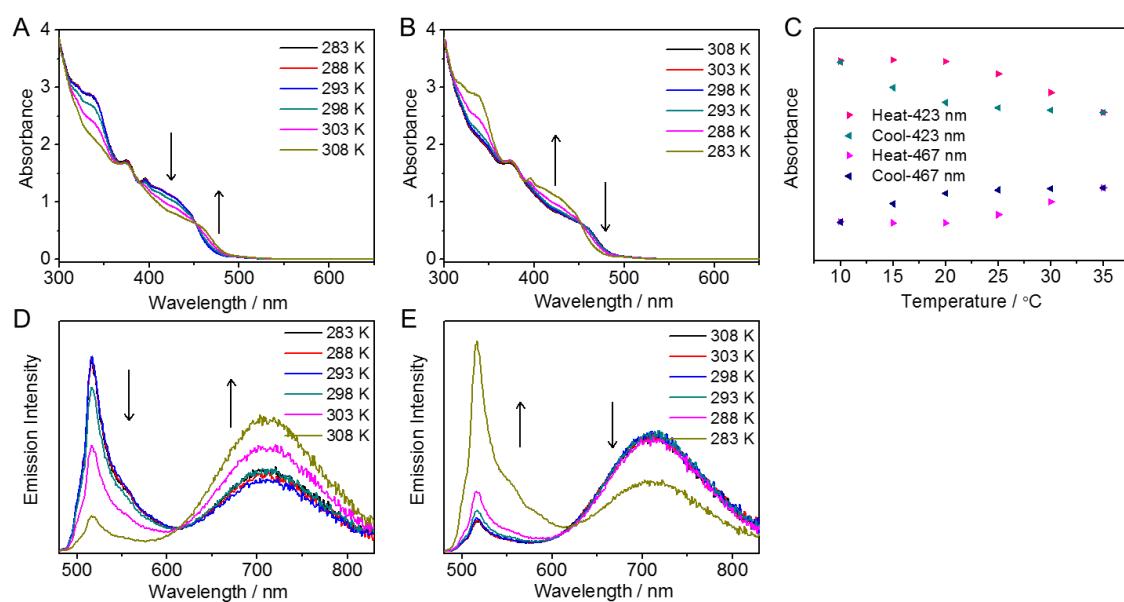
**Figure S10.** Time-resolved emission spectra of **1** in (A) dichloromethane solution ( $5.7 \times 10^{-5}$  M) and (B) hexane-dichloromethane (9:1, v/v) mixture ( $5.7 \times 10^{-5}$  M).



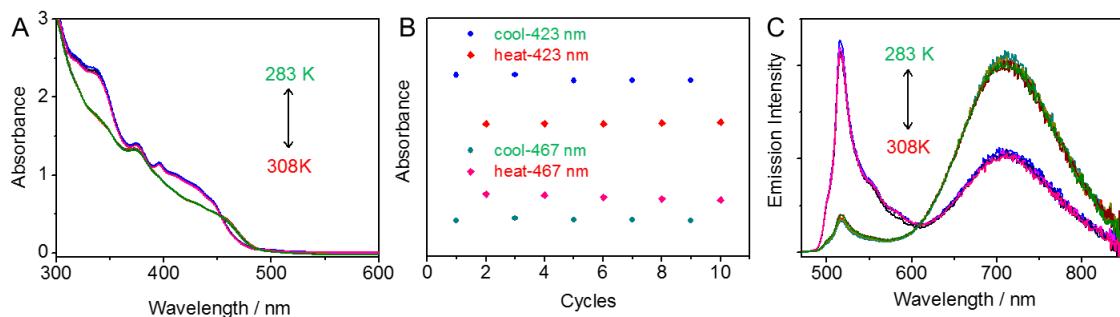
**Figure S11.** Concentration-dependent (A) UV-vis absorption and (B) emission spectra of **1** in cyclohexane-dichloromethane (9:1, v/v) mixtures at different concentrations from  $2.2 \times 10^{-5}$  to  $2.2 \times 10^{-4}$  M at room temperature. Inset: A plot of the corresponding phosphorescence intensity ratio between 516 and 690 nm ( $I_{516} / I_{690}$ ) in cyclohexane-dichloromethane (9:1, v/v) mixtures at different concentrations. (C) Photographs of emission color changes in cyclohexane-dichloromethane (9:1, v/v) mixtures at concentrations of  $7.2 \times 10^{-5}$ ,  $5.8 \times 10^{-5}$ ,  $4.8 \times 10^{-5}$  M (from left to right).



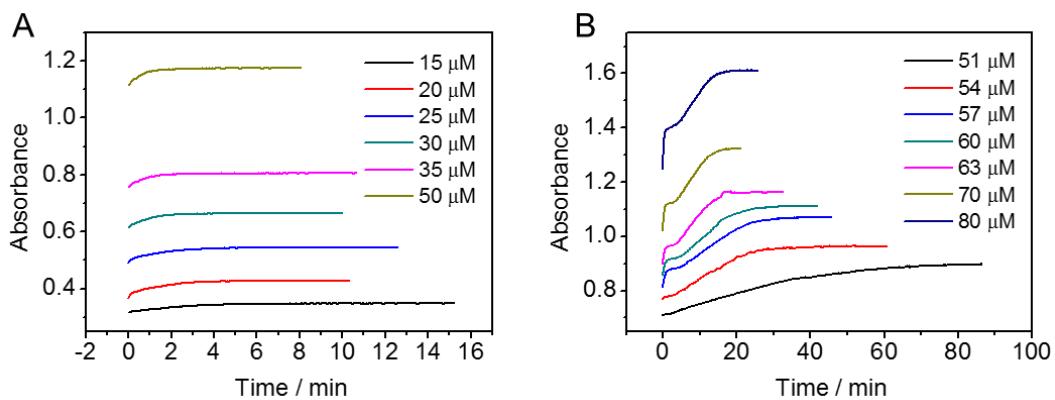
**Figure S12.** (A) Time-dependent emission spectral changes of **1** in *n*-hexane/*n*-decane/cyclohexane/methylcyclohexane-dichloromethane (9:1, v/v) mixtures ( $5.4 \times 10^{-5}$  M). (B) Photographs of hexane-dichloromethane, cyclohexane-dichloromethane and decane-dichloromethane (9:1, v/v, left to right) mixtures with increasing time.



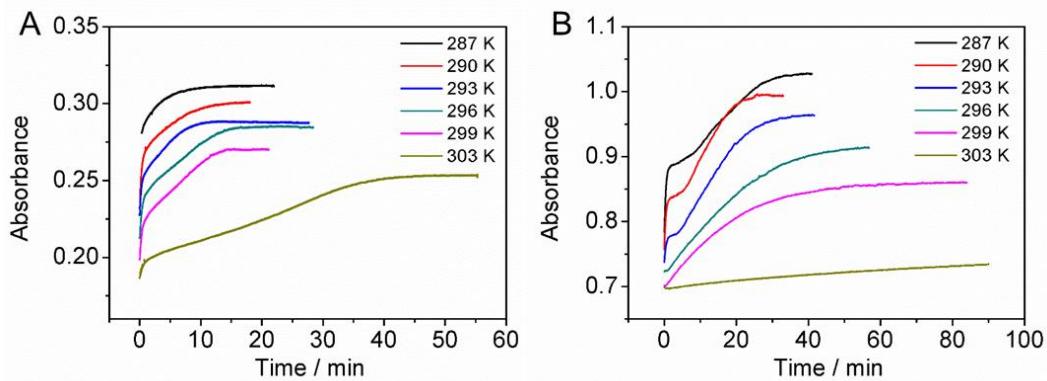
**Figure S13.** Temperature-dependent UV-vis absorption spectra of **1** in cyclohexane-dichloromethane (9:1, v/v) ( $5.6 \times 10^{-5}$  M) mixture with (A) heating and (B) cooling processes. (C) Plots of the corresponding absorbance at 423 and 467 nm at different temperatures with heating and cooling processes. Temperature-dependent emission spectra of **1** in cyclohexane-dichloromethane (9:1, v/v) ( $5.6 \times 10^{-5}$  M) mixture with (D) heating and (E) cooling processes.



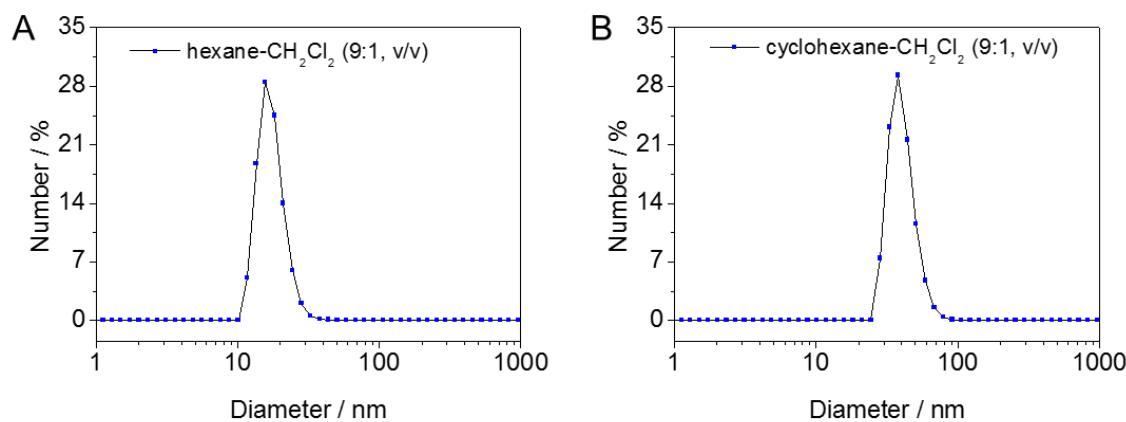
**Figure S14.** (A) Reversible processes of the molecular hinge between open form and closed form with electronic absorption spectra of **1** in cyclohexane-dichloromethane (9:1, v/v) ( $5.6 \times 10^{-5}$  M) mixture in 5 cycles. (B) Plots of the corresponding absorbance at 423 and 467 nm with the number of repeating cycles. (C) Reversible processes of the molecular hinge between open form and closed form with emission spectra of **1** in cyclohexane-dichloromethane (9:1, v/v) ( $5.6 \times 10^{-5}$  M) mixture in 5 cycles.



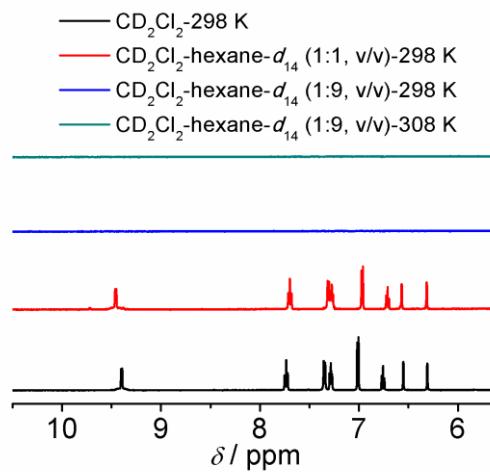
**Figure S15.** (A) Time-dependent UV-vis spectra of complex **1** in hexane-dichloromethane (9:1, v/v) mixtures at different concentrations from 15 to 50  $\mu$ M at 293 K. (B) Time-dependent UV-vis spectra of complex **1** in cyclohexane-dichloromethane (9:1, v/v) mixtures at different concentrations from 51 to 80  $\mu$ M at 293 K.



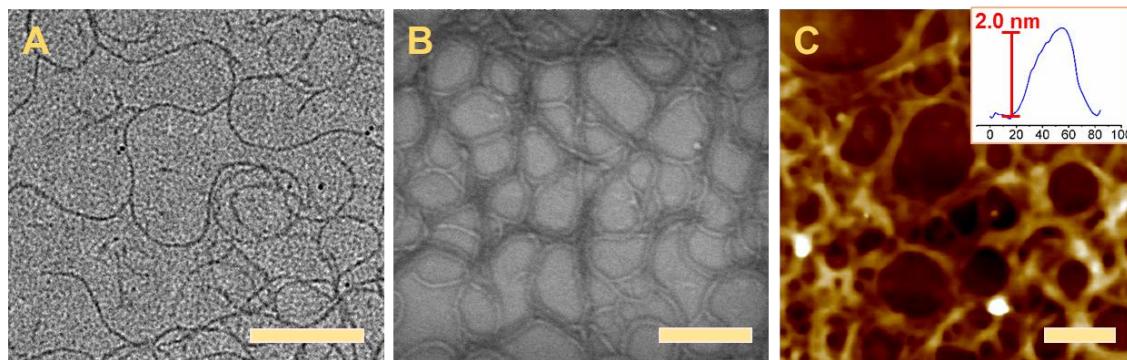
**Figure S16.** Time-dependent UV-vis spectra of complex **1** in (A) hexane-dichloromethane (9:1, v/v) ( $1.5 \times 10^{-5}$  M) and (B) cyclohexane-dichloromethane (9:1, v/v) ( $5.7 \times 10^{-5}$  M) mixtures at different temperatures.



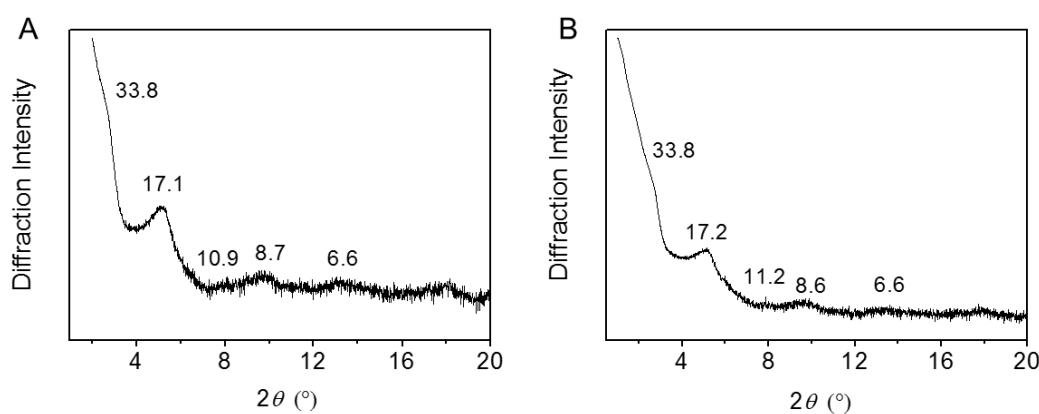
**Figure S17.** DLS studies of **1** in (A) hexane-dichloromethane (9:1, v/v) mixture ( $5.7 \times 10^{-5}$  M) and (B) cyclohexane-dichloromethane (9:1, v/v) mixture ( $5.7 \times 10^{-5}$  M) respectively.



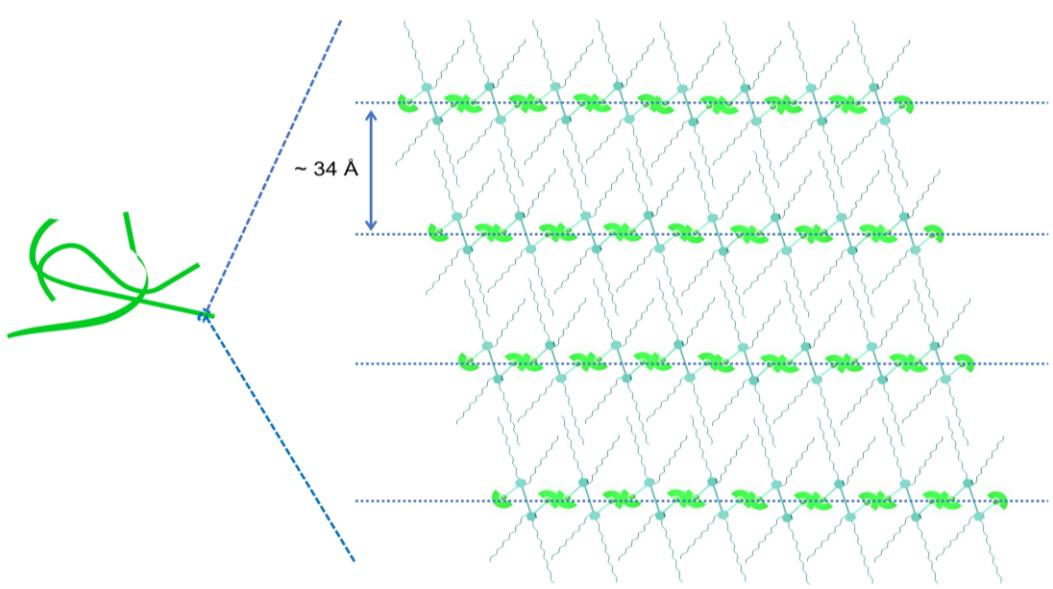
**Figure S18.**  $^1\text{H}$  NMR spectra of **1** upon increasing the hexane- $d_{14}$  portion in  $\text{CD}_2\text{Cl}_2$  ( $4.6 \times 10^{-4} \text{ M}$ ).



**Figure S19.** (A) TEM, (B) SEM and (C) AFM images of **1** prepared from cyclohexane-dichloromethane (9:1, v/v) mixture ( $5.6 \times 10^{-5} \text{ M}$ ). Scale bars: 200 nm.



**Figure S20.** XRD patterns of **1** prepared from (A) hexane-dichloromethane (9:1, v/v) mixture and (B) cyclohexane-dichloromethane (9:1, v/v) mixture with the ratio of about 1:1/2:1/3:1/4:1/5.



**Figure S21.** Schematic representation of one-dimensional nanofiber formed in hexane/cyclohexane-dichloromethane (9:1, v/v) mixtures.

**Video 1.** A video showing the luminescence changes in hexane-dichloromethane mixture of **1**.

**Video 2.** A video showing the luminescence changes in cyclohexane-dichloromethane mixture of **1**.

**Table S1.** Crystallographic and structural refinement data for complex **2**

Identification code	Complex <b>2</b>
Empirical formula	C <sub>54</sub> H <sub>28</sub> F <sub>6</sub> N <sub>4</sub> Pt <sub>2</sub>
Formula weight	1236.98
Temperature/K	150.00
Crystal system	triclinic
Space group	P $\bar{1}$
<i>a</i> /Å	10.136 1(2)
<i>b</i> /Å	14.9235(3)
<i>c</i> /Å	27.9163(6)
$\alpha/^\circ$	97.5834(17)
$\beta/^\circ$	97.4100(18)
$\gamma/^\circ$	90.1700(18)
<i>V</i> /Å <sup>3</sup>	4150.06(16)
<i>Z</i>	4
$\rho_{calc}$ /g · cm <sup>-3</sup>	1.980
$\mu$ /mm <sup>-1</sup>	13.046
<i>F</i> (000)	2360.0
Crystal size/mm <sup>3</sup>	0.696 × 0.513 × 0.233
Radiation	Cu–K $\alpha$ ( $\lambda = 1.54184$ Å )
2 $\theta$ range for data collection/°	7.158 to 148.476
Index ranges	$-12 \leq h \leq 12, -18 \leq k \leq 16, -30 \leq l \leq 34$
Reflections collected	27688
Independent reflections	16187 [ $R_{int} = 0.0849, R_{sigma} = 0.0681$ ]
Data/restraints/parameters	16187/852/776
Goodness-of-fit on <i>F</i> <sup>2</sup>	1.085
Final <i>R</i> indexes [ $I \geq 2\sigma(I)$ ]	$R_1 = 0.0849, wR_2 = 0.2293$
Final <i>R</i> indexes [all data]	$R_1 = 0.0906, wR_2 = 0.2379$
Largest diff. peak/hole / e Å <sup>-3</sup>	5.51/-4.46

**Table S2.** Selected bond distances ( $\text{\AA}$ ) and bond angles ( $^\circ$ ) for **2** with estimated standard deviations (e.s.d.s.) in parentheses

Pt(1)–N(1)	2.033(8)	Pt(1)–C(12)	1.949(9)
Pt(1)–N(2)	2.039(8)	Pt(1)–C(18)	2.049(9)
Pt(2)–N(3)	2.045(8)	Pt(2)–C(49)	1.957(9)
Pt(2)–N(4)	2.035(8)	Pt(2)–C(37)	2.062(9)
C(18)–C(19)	1.207(13)	C(28)–C(29)	1.195(15)
C(26)–C(27)	1.184(15)	C(36)–C(37)	1.212(13)
<hr/>			
N(1)–Pt(1)–N(2)	160.2(3)	N(1)–Pt(1)–C(12)	80.8(4)
N(1)–Pt(1)–C(18)	100.5(3)	N(2)–Pt(1)–C(12)	79.3(4)
N(2)–Pt(1)–C(18)	99.3(3)	Pt(1)–C(18)–C(19)	175.0(8)
C(12)–Pt(1)–C(18)	178.2(4)	C(37)–Pt(2)–C(49)	178.5(4)
N(3)–Pt(2)–N(4)	160.4(3)	N(3)–Pt(2)–C(49)	80.1(4)
N(3)–Pt(2)–C(37)	99.0(3)	N(4)–Pt(2)–C(49)	80.3(4)
N(4)–Pt(2)–C(37)	100.6(3)	Pt(2)–C(37)–C(36)	177.9(8)
C(26)–C(27)–C(28)	178.1(11)	C(27)–C(28)–C(29)	178.6(12)

**Table S3.** Cartesian coordinates of the optimized ground-state geometry of the open form of **1** in dichloromethane

Energy = -4642.03829696 Hartree							
1	Pt	-1.536839	-3.138260	2.683490	54	C	3.208195
2	C	-0.563727	-4.812014	2.953996	55	Pt	2.059432
3	C	-1.097882	-5.764057	3.823743	56	C	0.951137
4	C	0.646398	-5.004361	2.283740	57	C	1.527379
5	C	-0.396847	-6.957233	4.026828	58	C	-0.390127
6	C	1.344730	-6.198167	2.490218	59	C	0.732679
7	C	0.812442	-7.159939	3.355522	60	C	-1.181546
8	H	-0.773661	-7.730331	4.697130	61	C	-0.609096
9	H	2.294093	-6.394517	1.992181	62	H	1.138538
10	H	1.357093	-8.087805	3.515975	63	H	-2.233268
11	N	0.104955	-2.825549	1.457942	64	H	-1.226182
12	C	1.018642	-3.854750	1.452935	65	N	3.432993
13	C	0.348432	-1.715585	0.747697	66	C	2.934311
14	C	2.193602	-3.734910	0.718391	67	C	4.707786
15	C	1.500829	-1.555723	-0.006321	68	C	3.753357
16	H	-0.412637	-0.939682	0.804643	69	C	5.560378
17	C	2.439591	-2.579918	-0.014182	70	H	5.032267
18	H	2.913271	-4.549948	0.730848	71	C	5.072268
19	H	1.654485	-0.632068	-0.557307	72	H	3.348394
20	H	3.362568	-2.477975	-0.580377	73	H	6.584694
21	N	-2.785343	-4.083492	4.039599	74	H	5.713917
22	C	-3.919062	-3.578243	4.545873	75	N	0.250903
23	C	-2.353031	-5.326289	4.439769	76	C	-0.774090
24	C	-4.687655	-4.264853	5.471962	77	C	0.015212
25	H	-4.198461	-2.592061	4.183608	78	C	-2.053372
26	C	-3.090465	-6.051476	5.370048	79	C	-1.239429
27	C	-4.263081	-5.520939	5.889774	80	H	0.869786
28	H	-5.597679	-3.813951	5.854608	81	C	-2.290684
29	H	-2.737538	-7.032109	5.679590	82	H	-2.858989
30	H	-4.840634	-6.084771	6.617729	83	H	-1.377532
31	C	-2.563104	-1.364428	2.356040	84	H	-3.291386
32	C	-3.128774	-0.297477	2.102538	85	O	7.634443
33	C	-3.706306	0.957358	1.754361	86	O	3.477048
34	C	-5.099164	1.104524	1.663351	87	C	8.473735
35	C	-2.861745	2.053701	1.458768	88	H	8.045087
36	C	-5.640848	2.323242	1.261776	89	H	8.535757
37	H	-5.729666	0.252289	1.895823	90	C	9.836532
38	C	-3.439920	3.283986	1.053798	91	H	9.735507
39	C	-4.815337	3.414563	0.949299	92	H	10.221832
40	H	-5.285530	4.342293	0.637617	93	C	10.811972
41	C	-1.456184	1.931612	1.492349	94	H	10.412926
42	C	-0.237663	1.847185	1.449284	95	H	10.893134
43	C	1.110205	1.711092	1.328265	96	C	12.195035
44	C	2.309259	1.549509	1.155701	97	H	12.112408
45	C	3.677013	1.272947	0.944999	98	H	12.593801
46	C	4.441325	1.988898	-0.006456	99	C	13.177519
47	C	4.281826	0.206763	1.659114	100	H	12.778065
48	C	5.787257	1.651174	-0.216910	101	H	13.257755
49	C	5.605376	-0.132546	1.430319	102	C	14.561502
50	C	6.356132	0.598340	0.496388	103	H	14.480398
51	H	6.356933	2.214164	-0.949329	104	H	14.961440
52	H	6.092484	-0.946053	1.959468	105	C	15.544702
53	C	3.821839	3.022864	-0.766157	106	H	15.144397

107	H	15.625980	2.025639	-3.087888	160	H	-21.765911	-0.099560	-0.235488
108	C	16.928489	0.321750	-3.289871	161	H	-22.443418	-1.346999	0.821620
109	H	16.846617	-0.752711	-3.523991	162	H	-21.899406	0.194140	1.502876
110	H	17.329600	0.375330	-2.264133	163	C	-3.026630	5.482882	0.215935
111	C	17.911147	0.972999	-4.250452	164	H	-3.738703	5.971762	0.900062
112	H	17.508775	0.920631	-5.275813	165	H	-3.557885	5.273777	-0.729422
113	H	17.994404	2.047199	-4.015606	166	C	-1.818288	6.355173	-0.022778
114	C	19.294009	0.341451	-4.219499	167	H	-1.345053	6.585950	0.942647
115	H	19.698366	0.394007	-3.194846	168	H	-1.078445	5.780323	-0.602610
116	H	19.211254	-0.733024	-4.453830	169	C	-2.166682	7.637299	-0.760060
117	C	20.276572	0.991319	-5.182191	170	H	-2.593713	7.384973	-1.746570
118	H	19.868594	0.940470	-6.203564	171	H	-2.957079	8.180847	-0.216685
119	H	20.360072	2.063098	-4.944346	172	C	-0.962133	8.544722	-0.955994
120	C	21.650380	0.344776	-5.142969	173	H	-0.548502	8.824349	0.026671
121	H	22.348116	0.817154	-5.844142	174	H	-0.167399	7.978069	-1.471661
122	H	22.091422	0.414069	-4.139776	175	C	-1.279423	9.796922	-1.756967
123	H	21.592713	-0.721056	-5.400726	176	H	-2.058984	10.382653	-1.241895
124	O	-6.961827	2.558563	1.133266	177	H	-1.713914	9.498719	-2.726411
125	O	-2.561798	4.263494	0.779150	178	C	-0.065073	10.676100	-2.010651
126	C	-7.877234	1.498349	1.387348	179	H	0.339605	11.038279	-1.051241
127	H	-7.660890	0.649389	0.718492	180	H	0.734301	10.063060	-2.463925
128	H	-7.764072	1.144359	2.424801	181	C	-0.363388	11.856963	-2.920679
129	C	-9.269278	2.028881	1.147971	182	H	-1.165804	12.470636	-2.478686
130	H	-9.325058	2.429557	0.125142	183	H	-0.766615	11.480135	-3.876494
131	H	-9.452893	2.873739	1.828010	184	C	0.848703	12.731736	-3.198480
132	C	-10.331228	0.958483	1.344016	185	H	1.245396	13.123019	-2.247170
133	H	-10.134675	0.116016	0.661137	186	H	1.655104	12.111277	-3.625231
134	H	-10.258807	0.545725	2.363179	187	C	0.553865	13.889462	-4.139497
135	C	-11.739031	1.482693	1.108091	188	H	-0.246956	14.514363	-3.710693
136	H	-11.801118	1.915181	0.095972	189	H	0.150808	13.494987	-5.087235
137	H	-11.939809	2.314666	1.802905	190	C	1.768559	14.756084	-4.431614
138	C	-12.814417	0.419878	1.267592	191	H	2.169499	15.156402	-3.485573
139	H	-12.615926	-0.408766	0.567741	192	H	2.571583	14.130142	-4.855659
140	H	-12.751249	-0.018152	2.277410	193	C	1.475314	15.907940	-5.381489
141	C	-14.219276	0.952926	1.034036	194	H	0.672917	16.530937	-4.957052
142	H	-14.275884	1.403536	0.029224	195	H	1.075958	15.505249	-6.325286
143	H	-14.420474	1.774145	1.741870	196	C	2.699311	16.762551	-5.662094
144	C	-15.300854	-0.106990	1.171090	197	H	2.475776	17.590839	-6.344328
145	H	-15.103655	-0.924722	0.458071	198	H	3.099703	17.195324	-4.735866
146	H	-15.242703	-0.562813	2.173423	199	H	3.501535	16.166244	-6.116574
147	C	-16.703530	0.434282	0.943088	200	C	3.932741	-1.596300	3.190791
148	H	-16.757577	0.899681	-0.055122	201	H	4.823128	-1.370337	3.799416
149	H	-16.903034	1.245746	1.662596	202	H	4.219931	-2.353092	2.439973
150	C	-17.788152	-0.624502	1.064065	203	C	2.795710	-2.090601	4.051843
151	H	-17.593605	-1.431408	0.338074	204	H	2.567804	-1.331091	4.813939
152	H	-17.729815	-1.096405	2.058975	205	H	1.893944	-2.182435	3.425045
153	C	-19.190325	-0.077815	0.846964	206	C	3.098099	-3.424907	4.713280
154	H	-19.246933	0.403707	-0.143556	207	H	3.279771	-4.186138	3.934352
155	H	-19.388928	0.722052	1.579735	208	H	4.032060	-3.355314	5.294871
156	C	-20.275763	-1.138356	0.953177	209	C	1.965690	-3.892137	5.614558
157	H	-20.208889	-1.626615	1.937745	210	H	1.800587	-3.147813	6.410704
158	H	-20.081824	-1.928763	0.211667	211	H	1.030764	-3.922232	5.028051
159	C	-21.670715	-0.572611	0.750931	212	C	2.208540	-5.259346	6.232853

213	H	2.394156	-5.989229	5.426609
214	H	3.128324	-5.236862	6.840735
215	C	1.045986	-5.745750	7.083279
216	H	0.884639	-5.050071	7.923250
217	H	0.121241	-5.711022	6.480364
218	C	1.237567	-7.156115	7.617244
219	H	2.163966	-7.203029	8.213416
220	H	1.392285	-7.845750	6.769325
221	C	0.068965	-7.645388	8.457981
222	H	-0.080003	-6.961564	9.309961
223	H	-0.858334	-7.585811	7.862333
224	C	0.247055	-9.064545	8.974363
225	H	1.179213	-9.125253	9.560225
226	H	0.384850	-9.749452	8.121040
227	C	-0.915563	-9.547882	9.826865
228	H	-1.849900	-9.483458	9.244304
229	H	-1.049735	-8.866025	10.683090
230	C	-0.739451	-10.969599	10.339193
231	H	0.196332	-11.032317	10.915938
232	H	-0.610953	-11.649349	9.482621
233	C	-1.905353	-11.433227	11.195161
234	H	-2.846117	-11.406993	10.629409
235	H	-1.767159	-12.457964	11.559022
236	H	-2.034070	-10.785109	12.072084

**Table S4.** Cartesian coordinates of the optimized ground-state geometry of the closed form of **1** in dichloromethane

Energy = -4642.04582043 Hartrees								
1	Pt	-3.020472	-3.593160	1.447847	54	C	1.917646	-3.016856
2	C	-2.453559	-5.378089	2.007680	55	Pt	0.458826	-4.470731
3	C	-3.332127	-6.448623	1.835464	56	C	-0.920208	-5.822376
4	C	-1.182860	-5.533728	2.565857	57	C	-0.667048	-7.140912
5	C	-2.917470	-7.727349	2.223513	58	C	-2.124823	-5.430664
6	C	-0.770345	-6.813069	2.946807	59	C	-1.657438	-8.105538
7	C	-1.643963	-7.892821	2.773910	60	C	-3.113602	-6.396678
8	H	-3.563508	-8.596167	2.097256	61	C	-2.865862	-7.721479
9	H	0.218876	-6.986874	3.371795	62	H	-1.507352	-9.146423
10	H	-1.318728	-8.888311	3.068944	63	H	-4.074919	-6.138647
11	N	-1.117916	-3.178341	2.155794	64	H	-3.638956	-8.469849
12	C	-0.441612	-4.271741	2.643253	65	N	1.396392	-6.152849
13	C	-0.525402	-1.976517	2.154507	66	C	0.648755	-7.304436
14	C	0.839659	-4.122211	3.163605	67	C	2.607216	-6.180831
15	C	0.751816	-1.780117	2.657851	68	C	1.154040	-8.494191
16	H	-1.108175	-1.159905	1.731499	69	C	3.148493	-7.338997
17	C	1.440038	-2.870260	3.176510	70	H	3.139404	-5.232724
18	H	1.356934	-4.994497	3.556317	71	C	2.407274	-8.514540
19	H	1.190092	-0.786902	2.629713	72	H	0.550463	-9.396476
20	H	2.441430	-2.748438	3.581825	73	H	4.133005	-7.311871
21	N	-4.691631	-4.704125	0.927654	74	H	2.803058	-9.441232
22	C	-5.794278	-4.227838	0.331570	75	N	-1.029592	-3.305563
23	C	-4.592851	-6.046935	1.207309	76	C	-2.170761	-3.994021
24	C	-6.857764	-5.042323	-0.022354	77	C	-0.968730	-1.980793
25	H	-5.800266	-3.157238	0.140035	78	C	-3.248115	-3.318510
26	C	-5.638584	-6.903745	0.873662	79	C	-2.019514	-1.264437
27	C	-6.775622	-6.402814	0.256871	80	H	-0.041128	-1.499292
28	H	-7.729651	-4.609569	-0.502674	81	C	-3.174372	-1.947153
29	H	-5.546819	-7.963604	1.097497	82	H	-4.141206	-3.875872
30	H	-7.592604	-7.069964	-0.005847	83	H	-1.924358	-0.190622
31	C	-3.615468	-1.677916	0.914652	84	H	-4.016076	-1.413977
32	C	-3.925321	-0.517053	0.633876	85	O	7.338302	-0.487509
33	C	-4.227764	0.836304	0.311186	86	O	3.762256	2.545984
34	C	-5.563377	1.255790	0.199587	87	C	7.847204	-1.805496
35	C	-3.180314	1.765117	0.109319	88	H	7.524809	-2.442574
36	C	-5.851929	2.582650	-0.109654	89	H	7.444810	-2.245347
37	H	-6.350484	0.526287	0.361154	90	C	9.352316	-1.716354
38	C	-3.504336	3.114616	-0.188369	91	H	9.713787	-1.216362
39	C	-4.824123	3.518914	-0.300883	92	H	9.643441	-1.078033
40	H	-5.094783	4.543370	-0.538475	93	C	9.997055	-3.087245
41	C	-1.819590	1.386607	0.155575	94	H	9.700668	-3.715708
42	C	-0.624082	1.130325	0.132063	95	H	9.609029	-3.595035
43	C	0.705235	0.843401	0.073349	96	C	11.514289	-3.018279
44	C	1.897778	0.584161	-0.002512	97	H	11.897803	-2.485942
45	C	3.272061	0.273031	-0.101772	98	H	11.810411	-2.408288
46	C	3.704110	-1.060744	-0.309625	99	C	12.174804	-4.384708
47	C	4.243636	1.303880	-0.010619	100	H	11.882739	-4.990093
48	C	5.074422	-1.346595	-0.415966	101	H	11.786627	-4.922092
49	C	5.593087	1.010238	-0.131192	102	C	13.691625	-4.312713
50	C	6.004659	-0.315701	-0.329472	103	H	14.076412	-3.760275
51	H	5.378357	-2.376119	-0.576735	104	H	13.983716	-3.719504
52	H	6.359558	1.776287	-0.075049	105	C	14.359393	-5.677349
53	C	2.754106	-2.115104	-0.427685	106	H	14.069194	-6.268497

107	H	13.974368	-6.232498	-1.965847	160	H	-22.043568	2.271392	-2.561643
108	C	15.876045	-5.601071	-1.173939	161	H	-22.942558	1.265647	-1.415252
109	H	16.258808	-5.038092	-0.306444	162	H	-22.259630	2.812137	-0.891741
110	H	16.166388	-5.016133	-2.062444	163	C	-2.637160	5.297947	-0.668915
111	C	16.548663	-6.963975	-1.225418	164	H	-3.290911	5.776607	0.077644
112	H	16.257650	-7.548755	-0.336944	165	H	-3.126797	5.387687	-1.652541
113	H	16.167894	-7.527588	-2.093352	166	C	-1.263707	5.923803	-0.669419
114	C	18.065196	-6.884153	-1.301328	167	H	-0.839489	5.828321	0.341118
115	H	18.357298	-6.300833	-2.190505	168	H	-0.612609	5.337687	-1.334187
116	H	18.445582	-6.318587	-0.434327	169	C	-1.256949	7.381497	-1.096914
117	C	18.741769	-8.246021	-1.348802	170	H	-1.569544	7.463127	-2.150187
118	H	18.445602	-8.827766	-0.462127	171	H	-1.997587	7.951207	-0.512682
119	H	18.363663	-8.807398	-2.217209	172	C	0.119129	8.003150	-0.914292
120	C	20.255796	-8.145464	-1.418309	173	H	0.375148	7.988102	0.158196
121	H	20.733306	-9.131384	-1.456355	174	H	0.872869	7.368509	-1.409742
122	H	20.573761	-7.589161	-2.310015	175	C	0.239487	9.424036	-1.439201
123	H	20.657039	-7.616203	-0.543923	176	H	-0.569128	10.044185	-1.018344
124	O	-7.100735	3.072132	-0.248711	177	H	0.086990	9.429027	-2.530904
125	O	-2.446434	3.926061	-0.345946	178	C	1.581760	10.055005	-1.104071
126	C	-8.201380	2.183831	-0.091765	179	H	1.691912	10.095963	-0.006679
127	H	-8.104578	1.341587	-0.795877	180	H	2.396417	9.401267	-1.462088
128	H	-8.203868	1.767601	0.928744	181	C	1.768815	11.450374	-1.677045
129	C	-9.469850	2.954897	-0.361878	182	H	0.919888	12.086176	-1.375397
130	H	-9.420703	3.377376	-1.376321	183	H	1.735012	11.405102	-2.778113
131	H	-9.532183	3.805042	0.332964	184	C	3.066749	12.103685	-1.230080
132	C	-10.698539	2.069673	-0.222706	185	H	3.078968	12.157908	-0.127444
133	H	-10.594664	1.193310	-0.883130	186	H	3.918993	11.457235	-1.504495
134	H	-10.750085	1.670239	0.803072	187	C	3.279738	13.496744	-1.800588
135	C	-11.996199	2.791343	-0.550903	188	H	2.402759	14.123096	-1.566721
136	H	-11.944401	3.183718	-1.579786	189	H	3.325203	13.442148	-2.900953
137	H	-12.103298	3.672310	0.102796	190	C	4.533524	14.173872	-1.270113
138	C	-13.220957	1.901161	-0.410072	191	H	4.475844	14.232224	-0.168623
139	H	-13.096521	1.008767	-1.045822	192	H	5.413117	13.543612	-1.488248
140	H	-13.282344	1.525498	0.624733	193	C	4.760188	15.568759	-1.832927
141	C	-14.523250	2.597534	-0.772822	194	H	3.859617	16.178145	-1.660397
142	H	-14.461497	2.970090	-1.808817	195	H	4.876463	15.505326	-2.925789
143	H	-14.651036	3.491103	-0.139664	196	C	5.968128	16.256517	-1.220214
144	C	-15.741910	1.698533	-0.632205	197	H	6.137424	17.251975	-1.646783
145	H	-15.605127	0.799627	-1.256217	198	H	5.840161	16.379186	-0.134872
146	H	-15.809126	1.334348	0.406433	199	H	6.882922	15.668711	-1.378378
147	C	-17.046784	2.380537	-1.012867	200	C	4.685882	3.629136	0.211378
148	H	-16.978976	2.744578	-2.051542	201	H	5.210457	3.683335	-0.756803
149	H	-17.186589	3.278877	-0.388918	202	H	5.444562	3.458430	0.991898
150	C	-18.260998	1.475059	-0.874604	203	C	3.939464	4.911913	0.489680
151	H	-18.117027	0.574706	-1.494908	204	H	3.115951	5.024665	-0.230529
152	H	-18.330918	1.114183	0.165055	205	H	3.482751	4.854187	1.488534
153	C	-19.567322	2.149874	-1.262574	206	C	4.887524	6.100965	0.407291
154	H	-19.497905	2.512061	-2.301922	207	H	5.795252	5.882202	0.993902
155	H	-19.714277	3.049362	-0.641632	208	H	5.227345	6.223568	-0.633672
156	C	-20.780218	1.241122	-1.127017	209	C	4.293387	7.409437	0.906385
157	H	-20.847510	0.880995	-0.088734	210	H	3.353657	7.624904	0.371224
158	H	-20.630518	0.343610	-1.747082	211	H	4.018092	7.299797	1.967954
159	C	-22.076489	1.929490	-1.518690	212	C	5.256050	8.577812	0.748403

213	H	6.240727	8.288753	1.153034
214	H	5.420967	8.773879	-0.324864
215	C	4.803032	9.858672	1.431774
216	H	3.824843	10.171925	1.027625
217	H	4.637182	9.660946	2.503679
218	C	5.805488	10.992195	1.277098
219	H	5.949038	11.210480	0.204511
220	H	6.789345	10.657112	1.646762
221	C	5.410666	12.270673	1.999009
222	H	4.447568	12.635134	1.601281
223	H	5.231537	12.052299	3.064782
224	C	6.456142	13.368099	1.875966
225	H	6.676681	13.543484	0.808539
226	H	7.402141	13.021316	2.324241
227	C	6.041794	14.681658	2.518775
228	H	5.745479	14.505769	3.566403
229	H	5.137982	15.065493	2.013182
230	C	7.132347	15.741440	2.474354
231	H	7.467185	15.869887	1.432439
232	H	8.011041	15.378140	3.029086
233	C	6.677160	17.076347	3.037496
234	H	6.346516	16.974248	4.079698
235	H	7.475311	17.827364	3.015480
236	H	5.828933	17.477305	2.465979

**Table S5.** Cartesian coordinates of the optimized ground-state geometry of the open form of **3** in dichloromethane

Energy = -5944.48838658 Hartrees							
1	Pt	4.608124	2.604801	-0.470329	54	C	-6.601867
2	C	4.845309	4.473266	-0.986428	55	C	-4.260941
3	C	6.077785	5.083021	-0.747445	56	C	-6.932073
4	C	3.773890	5.148343	-1.580561	57	C	-4.588267
5	C	6.250214	6.420715	-1.106715	58	C	-5.918746
6	C	3.941280	6.485293	-1.930451	59	H	-7.951072
7	C	5.178597	7.100109	-1.689877	60	H	-3.840456
8	H	7.190806	6.943499	-0.938129	61	N	-6.842100
9	H	3.137059	7.069107	-2.378564	62	C	-7.484758
10	N	2.737391	3.019494	-1.253065	63	C	-7.561572
11	C	2.580586	4.303439	-1.723697	64	C	-8.863155
12	C	1.698880	2.178986	-1.313527	65	C	-8.931358
13	C	1.371664	4.704841	-2.268915	66	H	-7.000260
14	C	0.472050	2.542684	-1.848460	67	C	-9.626222
15	H	1.871420	1.180424	-0.916053	68	H	-9.332859
16	C	0.283084	3.827333	-2.357869	69	H	-9.450353
17	H	1.289441	5.723703	-2.643433	70	N	-2.959460
18	H	-0.321649	1.801516	-1.853064	71	C	-2.954013
19	N	6.564709	2.916162	0.135073	72	C	-1.821328
20	C	7.368261	2.026565	0.728910	73	C	-1.786580
21	C	7.038333	4.186738	-0.094116	74	C	-0.634339
22	C	8.659964	2.332859	1.128090	75	H	-1.877746
23	H	6.945329	1.036824	0.883297	76	C	-0.592817
24	C	8.321242	4.535558	0.296417	77	H	-1.821608
25	C	9.170090	3.615485	0.920760	78	H	0.244776
26	H	9.248644	1.554427	1.602584	79	O	-5.189049
27	H	8.654866	5.554665	0.107668	80	O	-0.690639
28	C	4.375279	0.602753	0.013745	81	C	-6.566485
29	C	4.240584	-0.603288	0.234567	82	H	-6.702666
30	C	4.080347	-2.001975	0.455747	83	H	-6.916481
31	C	5.207517	-2.829903	0.433010	84	C	-7.336859
32	C	2.778704	-2.545134	0.672825	85	H	-6.961804
33	C	5.059086	-4.206815	0.602381	86	H	-7.140185
34	H	6.180002	-2.377762	0.265037	87	C	-8.830132
35	C	2.661465	-3.947652	0.855126	88	H	-9.016094
36	C	3.792924	-4.752757	0.813582	89	H	-9.188631
37	H	3.684467	-5.825183	0.954691	90	C	-9.640198
38	C	1.672196	-1.665234	0.633717	91	H	-9.295716
39	C	0.738976	-0.875917	0.618639	92	H	-9.440477
40	C	-0.318751	-0.021121	0.624076	93	C	-11.135810
41	C	-1.268068	0.749508	0.638052	94	H	-11.334753
42	C	-2.305694	1.706540	0.653730	95	H	-11.475624
43	C	-3.651945	1.350643	0.404146	96	C	-11.956059
44	C	-1.986729	3.066835	0.908204	97	H	-11.625940
45	C	-4.647419	2.340339	0.395909	98	H	-11.748768
46	C	-2.976244	4.037587	0.888656	99	C	-13.452391
47	C	-4.305960	3.669667	0.630765	100	H	-13.659660
48	H	-5.672836	2.042573	0.201963	101	H	-13.780715
49	H	-2.761577	5.084942	1.078567	102	C	-14.277199
50	C	-4.001112	-0.006901	0.152860	103	H	-13.955063
51	C	-4.306784	-1.179972	-0.075259	104	H	-14.064288
52	Pt	-4.806891	-3.138076	-0.535602	105	C	-15.773951
53	C	-5.273541	-4.976236	-0.997512	106	H	-15.986615

107	H	-16.096257	7.900400	0.994619	160	C	-0.657890	-5.174427	1.812307
108	C	-16.599179	9.869351	0.280289	161	H	-0.798639	-5.670956	0.837735
109	H	-16.382315	10.387889	1.229082	162	H	-0.264963	-5.935530	2.501963
110	H	-16.281649	10.561295	-0.517536	163	C	-1.992597	-4.632971	2.302547
111	C	-18.097427	9.628557	0.170352	164	H	-1.879933	-4.211468	3.314194
112	H	-18.310788	9.102669	-0.773140	165	H	-2.304088	-3.792312	1.656327
113	H	-18.413890	8.945367	0.973699	166	C	-3.091777	-5.681805	2.297720
114	C	-18.904040	10.913889	0.238316	167	H	-3.175517	-6.105457	1.280839
115	H	-19.981786	10.727977	0.164744	168	H	-2.811145	-6.523517	2.952243
116	H	-18.723875	11.444518	1.182633	169	C	-4.445695	-5.131664	2.713531
117	H	-18.629399	11.596545	-0.576643	170	H	-4.678309	-4.250141	2.087174
118	O	6.075869	-5.089474	0.589042	171	H	-4.398463	-4.766376	3.752841
119	O	1.505706	-4.631871	1.011886	172	C	-5.565710	-6.147916	2.572012
120	C	7.406510	-4.609226	0.428012	173	H	-5.596245	-6.501551	1.526461
121	H	7.505656	-4.092699	-0.540429	174	H	-5.341875	-7.037036	3.184994
122	H	7.640301	-3.881032	1.221324	175	C	-6.934145	-5.606114	2.950902
123	C	8.328117	-5.801334	0.503514	176	H	-7.135065	-4.682503	2.377613
124	H	8.045668	-6.519797	-0.279908	177	H	-6.933291	-5.307460	4.012272
125	H	8.171189	-6.308894	1.466696	178	C	-8.053096	-6.603698	2.700559
126	C	9.790835	-5.415048	0.352516	179	H	-8.087425	-6.839079	1.621459
127	H	9.942563	-4.904058	-0.611977	180	H	-7.818229	-7.554742	3.206592
128	H	10.062690	-4.683227	1.130316	181	C	-9.418599	-6.114565	3.152175
129	C	10.716885	-6.617994	0.442433	182	H	-9.630526	-5.144769	2.667380
130	H	10.439723	-7.348600	-0.335141	183	H	-9.398116	-5.909910	4.235243
131	H	10.554329	-7.129398	1.405438	184	C	-10.539372	-7.092354	2.838285
132	C	12.189509	-6.267137	0.302100	185	H	-10.575097	-7.264557	1.747574
133	H	12.356547	-5.760091	-0.662581	186	H	-10.306994	-8.074935	3.281120
134	H	12.468111	-5.535796	1.078738	187	C	-11.906233	-6.631750	3.323275
135	C	13.099080	-7.481797	0.401147	188	H	-12.096211	-5.609159	2.958459
136	H	12.817657	-8.211611	-0.376061	189	H	-11.895057	-6.560927	4.421531
137	H	12.924593	-7.989256	1.364463	190	C	-13.027270	-7.549762	2.868734
138	C	14.577310	-7.150937	0.268558	191	H	-14.004455	-7.224412	3.243845
139	H	14.755299	-6.645661	-0.695196	192	H	-13.086729	-7.580610	1.771944
140	H	14.861055	-6.422546	1.046219	193	H	-12.864185	-8.578687	3.215690
141	C	15.473828	-8.374956	0.371419	194	C	-0.250821	4.655598	1.327235
142	H	15.188504	-9.101942	-0.407062	195	H	-0.826672	5.151265	2.124701
143	H	15.290771	-8.880860	1.333976	196	H	-0.421331	5.215657	0.391004
144	C	16.955518	-8.057351	0.244246	197	C	1.217554	4.612729	1.671884
145	H	17.140441	-7.551217	-0.717795	198	H	1.342918	4.161708	2.666998
146	H	17.243064	-7.332917	1.024245	199	H	1.729319	3.945740	0.960757
147	C	17.842738	-9.288004	0.346295	200	C	1.861541	5.989061	1.629134
148	H	17.555132	-10.011671	-0.434584	201	H	1.819918	6.373637	0.594439
149	H	17.655198	-9.795678	1.307183	202	H	1.284479	6.700303	2.242591
150	C	19.327380	-8.979649	0.222784	203	C	3.308487	5.959628	2.093616
151	H	19.614433	-8.261486	1.006376	204	H	3.349066	5.662025	3.154153
152	H	19.512570	-8.469641	-0.735273	205	H	3.842890	5.168576	1.537076
153	C	20.193166	-10.223852	0.320815	206	C	4.041156	7.276969	1.895683
154	H	19.944425	-10.941165	-0.472421	207	H	3.956169	7.576883	0.837191
155	H	21.260738	-9.990917	0.233443	208	H	3.551009	8.075505	2.476408
156	H	20.043467	-10.735311	1.280918	209	C	5.513502	7.185745	2.265138
157	C	0.346367	-4.065484	1.625247	210	H	5.620817	7.004919	3.346936
158	H	-0.089628	-3.295083	0.979834	211	H	5.938742	6.295404	1.770080
159	H	0.623779	-3.593359	2.580474	212	C	6.333092	8.400569	1.857016

213	H	6.016556	9.286636	2.430477	266	C	0.666805	-5.694797	-2.623943
214	H	6.122837	8.636513	0.800030	267	C	-11.581475	-2.880770	1.665516
215	C	7.827595	8.174582	2.021751	268	H	-11.297015	-1.839068	1.853365
216	H	8.067266	8.002005	3.084268	269	H	-12.674256	-2.942943	1.749129
217	H	8.095551	7.236111	1.504663	270	H	-11.145489	-3.499238	2.461002
218	C	8.695182	9.295335	1.472291	271	C	-11.721916	-2.420161	-0.788376
219	H	8.493973	10.229405	2.021631	272	H	-11.392683	-1.384393	-0.642514
220	H	8.408250	9.494308	0.425546	273	H	-11.421958	-2.733703	-1.796563
221	C	10.180345	8.972992	1.533105	274	H	-12.818590	-2.434379	-0.740369
222	H	10.369572	8.037716	0.976153	275	C	-11.702427	-4.760792	0.037601
223	H	10.469469	8.759332	2.576184	276	H	-11.295106	-5.493098	0.746701
224	C	11.073060	10.070882	0.975977	277	H	-12.792113	-4.746178	0.165667
225	H	10.905801	10.998652	1.544302	278	H	-11.495446	-5.114784	-0.980314
226	H	10.769885	10.292557	-0.058961	279	C	1.059442	-6.835736	-1.677951
227	C	12.544930	9.697282	1.013905	280	H	0.273894	-7.598113	-1.605102
228	H	12.734647	8.785331	0.431618	281	H	1.968581	-7.326815	-2.048979
229	H	13.182646	10.489499	0.605182	282	H	1.262025	-6.451506	-0.670106
230	H	12.876705	9.501773	2.042176	283	C	1.834552	-4.717557	-2.724518
231	C	5.295338	8.547920	-2.036388	284	H	1.614056	-3.884589	-3.404461
232	C	-6.221251	-8.996385	-1.922846	285	H	2.110283	-4.302202	-1.746490
233	F	6.514151	9.046880	-1.792345	286	H	2.714058	-5.242831	-3.116227
234	F	5.024309	8.780732	-3.330426	287	C	0.386726	-6.260072	-4.021007
235	F	4.421346	9.292652	-1.334051	288	H	1.295980	-6.732093	-4.414842
236	F	-7.509950	-9.309159	-1.749180	289	H	-0.402522	-7.020922	-4.011222
237	F	-5.913678	-9.287245	-3.198599	290	H	0.088129	-5.466748	-4.718152
238	F	-5.503650	-9.844387	-1.166837					
239	C	10.559969	4.041622	1.371163					
240	C	11.391207	2.859785	1.862360					
241	H	10.943098	2.375946	2.739120					
242	H	11.526271	2.102543	1.079295					
243	H	12.386272	3.213864	2.157977					
244	C	10.400224	5.042567	2.522217					
245	H	11.386671	5.394815	2.852142					
246	H	9.812295	5.919259	2.222278					
247	H	9.898586	4.578718	3.381186					
248	C	11.305267	4.708777	0.210336					
249	H	11.416564	4.021384	-0.637852					
250	H	10.799186	5.612459	-0.149443					
251	H	12.309359	5.005474	0.539819					
252	C	-1.008077	4.271984	-3.031900					
253	C	-2.099958	3.208762	-2.943094					
254	H	-2.351488	2.964310	-1.901463					
255	H	-3.010896	3.584490	-3.424916					
256	H	-1.814291	2.282064	-3.456546					
257	C	-0.700001	4.534515	-4.512003					
258	H	0.038671	5.334685	-4.643061					
259	H	-0.312780	3.630988	-5.000264					
260	H	-1.617532	4.835675	-5.034011					
261	C	-1.533948	5.557488	-2.385695					
262	H	-2.417146	5.908280	-2.934801					
263	H	-1.839210	5.377819	-1.346165					
264	H	-0.795666	6.368196	-2.393910					
265	C	-11.144034	-3.360711	0.277372					

**Table S6.** Cartesian coordinates of the optimized ground-state geometry of the closed form of **3** in dichloromethane

Energy = -5944.50391589 Hartrees							
1	Pt	-2.588530	-2.552894	1.164861	54	C	-0.204323
2	C	-1.998857	-4.379903	1.516334	55	C	-1.743026
3	C	-2.873361	-5.431646	1.227446	56	C	-1.191662
4	C	-0.702904	-4.587075	1.995404	57	C	-2.735465
5	C	-2.433385	-6.738427	1.417410	58	C	-2.440993
6	C	-0.259351	-5.896703	2.182974	59	H	-1.025150
7	C	-1.133023	-6.950152	1.892955	60	H	-3.728180
8	H	-3.062637	-7.596984	1.182470	61	N	1.863873
9	H	0.748637	-6.115647	2.533558	62	C	1.144445
10	N	-0.670439	-2.202647	1.848450	63	C	3.103641
11	C	0.036905	-3.331889	2.189375	64	C	1.693883
12	C	-0.077725	-1.009471	1.957001	65	C	3.686627
13	C	1.334054	-3.224127	2.666373	66	H	3.625361
14	C	1.219022	-0.859369	2.424336	67	C	2.983969
15	H	-0.675534	-0.152285	1.652169	68	H	1.084800
16	C	1.960790	-1.979116	2.800588	69	H	4.697602
17	H	1.863042	-4.138308	2.931027	70	N	-0.679630
18	H	1.631906	0.143915	2.469856	71	C	-1.823979
19	N	-4.268910	-3.609809	0.577868	72	C	-0.658307
20	C	-5.401810	-3.091388	0.080538	73	C	-2.923962
21	C	-4.157077	-4.971565	0.686761	74	C	-1.739915
22	C	-6.460673	-3.877301	-0.331601	75	H	0.265959
23	H	-5.435712	-2.005623	0.019968	76	C	-2.910658
24	C	-5.200546	-5.800323	0.286310	77	H	-3.808240
25	C	-6.382293	-5.273202	-0.233086	78	H	-1.646275
26	H	-7.348538	-3.386644	-0.722949	79	O	7.731414
27	H	-5.065202	-6.873847	0.385174	80	O	4.195420
28	C	-3.186123	-0.601922	0.811355	81	C	8.218707
29	C	-3.479028	0.578468	0.606161	82	H	7.766011
30	C	-3.801539	1.942113	0.357286	83	H	7.931504
31	C	-5.146492	2.342611	0.297164	84	C	9.719814
32	C	-2.773001	2.893205	0.158612	85	H	9.970844
33	C	-5.463830	3.673550	0.040249	86	H	10.133792
34	H	-5.917291	1.593984	0.451593	87	C	10.345119
35	C	-3.125813	4.249371	-0.069942	88	H	9.935554
36	C	-4.455493	4.631732	-0.143644	89	H	10.057215
37	H	-4.750330	5.658538	-0.337304	90	C	11.861609
38	C	-1.408066	2.528304	0.148052	91	H	12.147895
39	C	-0.214324	2.271299	0.082161	92	H	12.268867
40	C	1.112324	1.979967	-0.009380	93	C	12.505664
41	C	2.302298	1.712892	-0.097974	94	H	12.111320
42	C	3.674713	1.384626	-0.161586	95	H	12.207653
43	C	4.093934	0.051239	-0.395181	96	C	14.023204
44	C	4.659253	2.387669	0.036462	97	H	14.319627
45	C	5.460258	-0.271009	-0.374138	98	H	14.416004
46	C	6.005365	2.056759	0.056995	99	C	14.677472
47	C	6.401675	0.725601	-0.136451	100	H	14.294716
48	H	5.752878	-1.301529	-0.550605	101	H	14.374234
49	H	6.779173	2.798737	0.225349	102	C	16.195446
50	C	3.135898	-0.970082	-0.648846	103	H	16.497585
51	C	2.300274	-1.849607	-0.871472	104	H	16.577042
52	Pt	0.859226	-3.282903	-1.266032	105	C	16.856856
53	C	-0.499618	-4.626677	-1.662437	106	H	16.482472

107	H	16.549953	-6.678447	-0.896997	160	C	-1.000180	7.180990	-0.075544
108	C	18.374875	-6.064895	0.069023	161	H	-0.644058	6.925878	0.933574
109	H	18.749318	-5.476620	-0.785367	162	H	-0.259232	6.774654	-0.778878
110	H	18.682171	-5.510384	0.971457	163	C	-1.092515	8.690778	-0.226457
111	C	19.041626	-7.432473	0.068451	164	H	-1.276299	8.949102	-1.281481
112	H	18.670864	-8.015757	0.925559	165	H	-1.962172	9.070769	0.334240
113	H	18.729626	-7.986663	-0.830352	166	C	0.162242	9.393190	0.268429
114	C	20.557167	-7.344581	0.120480	167	H	0.267133	9.213866	1.351128
115	H	21.027878	-8.334451	0.122823	168	H	1.049979	8.933921	-0.197069
116	H	20.950447	-6.792741	-0.743500	169	C	0.168450	10.889302	0.000555
117	H	20.891163	-6.816672	1.023517	170	H	-0.790837	11.325863	0.324778
118	O	-6.723108	4.147044	-0.045686	171	H	0.226906	11.067533	-1.085808
119	O	-2.088357	5.092929	-0.199776	172	C	1.306887	11.613551	0.701766
120	C	-7.804481	3.240273	0.137252	173	H	1.166910	11.532580	1.792450
121	H	-7.763084	2.449997	-0.631014	174	H	2.260052	11.101014	0.485075
122	H	-7.721950	2.752671	1.121957	175	C	1.420597	13.078875	0.312389
123	C	-9.091625	4.021457	0.041275	176	H	0.431301	13.558714	0.400214
124	H	-9.142231	4.519566	-0.937968	177	H	1.694959	13.152564	-0.753865
125	H	-9.081232	4.817084	0.800725	178	C	2.426744	13.850057	1.151663
126	C	-10.305224	3.127182	0.241209	179	H	2.118752	13.816103	2.209892
127	H	-10.320758	2.345713	-0.535688	180	H	3.405726	13.341861	1.107465
128	H	-10.213747	2.596184	1.202665	181	C	2.589045	15.298546	0.718717
129	C	-11.618263	3.893864	0.216461	182	H	1.602574	15.791909	0.718900
130	H	-11.716438	4.425000	-0.744410	183	H	2.939722	15.329838	-0.327544
131	H	-11.596199	4.676467	0.992453	184	C	3.548150	16.089515	1.593991
132	C	-12.832151	3.004064	0.432565	185	H	3.205601	16.053396	2.641651
133	H	-12.865010	2.231108	-0.353060	186	H	4.538157	15.599952	1.586913
134	H	-12.718219	2.459518	1.384587	187	C	3.702463	17.540671	1.164082
135	C	-14.146848	3.768278	0.446252	188	H	2.716508	18.030320	1.181589
136	H	-14.267189	4.309407	-0.506851	189	H	4.036489	17.571895	0.113723
137	H	-14.108977	4.544044	1.228939	190	C	4.678784	18.310518	2.036591
138	C	-15.357051	2.877273	0.679252	191	H	4.781361	19.354732	1.718153
139	H	-15.401711	2.106667	-0.108226	192	H	4.353122	18.315144	3.085406
140	H	-15.228137	2.329362	1.627506	193	H	5.678287	17.854633	2.010585
141	C	-16.672358	3.639847	0.714920	194	C	5.140016	4.708677	0.196939
142	H	-16.806155	4.184676	-0.234341	195	H	5.798657	4.613244	-0.681545
143	H	-16.624626	4.412753	1.499966	196	H	5.773197	4.645484	1.096627
144	C	-17.879023	2.747183	0.960058	197	C	4.399014	6.024023	0.149918
145	H	-17.930669	1.977194	0.172352	198	H	3.686701	6.001452	-0.687711
146	H	-17.739948	2.198496	1.906453	199	H	3.807308	6.153424	1.067660
147	C	-19.194840	3.507865	1.008978	200	C	5.378330	7.175775	-0.027078
148	H	-19.338291	4.054880	0.062215	201	H	6.118901	7.159638	0.789118
149	H	-19.142649	4.279424	1.795256	202	H	5.952009	7.014473	-0.954460
150	C	-20.399717	2.613713	1.261641	203	C	4.729672	8.550958	-0.085008
151	H	-20.251262	2.066080	2.205250	204	H	3.937388	8.553231	-0.851775
152	H	-20.452822	1.846630	0.473531	205	H	4.227586	8.768396	0.872326
153	C	-21.704648	3.388873	1.315742	206	C	5.741085	9.642156	-0.403813
154	H	-21.889133	3.918506	0.371758	207	H	6.569952	9.595286	0.321735
155	H	-22.564589	2.735166	1.502004	208	H	6.193965	9.430516	-1.386785
156	H	-21.683211	4.143355	2.113237	209	C	5.168863	11.051059	-0.413756
157	C	-2.325004	6.492801	-0.296702	210	H	4.281353	11.088084	-1.069110
158	H	-3.051944	6.805073	0.469231	211	H	4.808506	11.310940	0.596644
159	H	-2.751989	6.733732	-1.283948	212	C	6.178660	12.089795	-0.879067

213	H	6.483886	11.857105	-1.912749	266	C	-3.936448	1.598943	-3.413692
214	H	7.094991	12.006968	-0.271350	267	C	-3.100162	1.903136	-4.055917
215	C	5.662260	13.517843	-0.815583	268	H	-4.842290	2.080551	-3.802482
216	H	4.709882	13.590677	-1.369720	269	H	-3.759012	1.992842	-2.403877
217	H	5.418844	13.771929	0.230477	270	H	-5.349026	-0.250669	-2.541328
218	C	6.644404	14.541720	-1.363206	271	C	-5.607865	-1.316002	-2.591329
219	H	6.876565	14.301657	-2.414049	272	H	-5.158997	0.007945	-1.491048
220	H	7.599471	14.459239	-0.818115	273	H	-6.221201	0.320925	-2.886049
221	C	6.132065	15.969979	-1.270793	274	H	-4.392189	-0.373865	-4.851635
222	H	5.899227	16.206543	-0.217726	275	C	-3.538152	-0.143913	-5.501622
223	H	5.173441	16.050417	-1.812474	276	H	-4.584460	-1.451974	-4.912088
224	C	7.103607	17.006276	-1.814615	277	H	-5.272158	0.145456	-5.253183
225	H	7.332884	16.771521	-2.865568	278	H	3.633907	-8.788712	0.429194
226	H	8.058559	16.926375	-1.272506	279	C	4.903803	-8.990265	-0.406235
227	C	6.567017	18.423328	-1.707167	280	H	5.633784	-8.187213	-0.248142
228	H	6.365710	18.691229	-0.660785	281	H	5.383842	-9.937174	-0.127660
229	H	7.271574	19.161854	-2.107108	282	H	4.671016	-9.031207	-1.477922
230	H	5.624034	18.532062	-2.259847	283	C	4.008247	-8.727635	1.915087
231	C	-0.696118	-8.368500	2.046381	284	H	4.467194	-9.676998	2.219642
232	C	-3.467548	-7.562480	-2.345544	285	H	4.728559	-7.928928	2.128101
233	F	-4.663635	-7.066279	-2.688868	286	H	3.123033	-8.564051	2.541703
234	F	-3.105316	-8.420073	-3.312840	287	C	2.711753	-9.984003	0.206469
235	F	-3.661257	-8.314662	-1.242673	288	H	1.790724	-9.908631	0.797089
236	F	-1.437886	-9.034835	2.944764	289	H	2.438728	-10.098506	-0.850463
237	F	0.582411	-8.482091	2.429466	290	H	3.228451	-10.901315	0.513812
238	F	-0.816512	-9.055967	0.891335					
239	C	3.382998	-1.892606	3.338117					
240	C	3.899339	-0.456252	3.372021					
241	H	3.292364	0.181871	4.026865					
242	H	4.924296	-0.449441	3.763210					
243	H	3.923300	-0.002095	2.372022					
244	C	3.404927	-2.451024	4.766155					
245	H	2.745689	-1.873010	5.426530					
246	H	3.087986	-3.500301	4.804122					
247	H	4.423681	-2.395305	5.171555					
248	C	4.314671	-2.724687	2.448263					
249	H	4.060218	-3.792230	2.470134					
250	H	4.265227	-2.380781	1.406085					
251	H	5.350797	-2.621986	2.798004					
252	C	-7.556158	-6.137148	-0.670279					
253	C	-8.768477	-5.789073	0.201699					
254	H	-9.056800	-4.735552	0.105844					
255	H	-9.629433	-6.400740	-0.097608					
256	H	-8.562268	-5.989391	1.260938					
257	C	-7.265857	-7.628348	-0.525138					
258	H	-6.417515	-7.943257	-1.145187					
259	H	-7.058810	-7.906383	0.516234					
260	H	-8.143643	-8.200805	-0.848494					
261	C	-7.879972	-5.843366	-2.139635					
262	H	-8.724507	-6.464916	-2.463582					
263	H	-8.158653	-4.795534	-2.301915					
264	H	-7.023915	-6.072658	-2.786163					
265	C	-4.131575	0.084825	-3.411757					

**Table S7.** Cartesian coordinates of the optimized T<sub>1</sub> geometry of the open form of **1** in dichloromethane

Energy = -4641.95790161 Hartrees									
1	Pt	3.671448	3.390323	-0.874539	54	C	-4.024170	-2.138761	-0.588695
2	C	3.427858	5.288124	-1.273955	55	Pt	-3.665803	-4.107758	-1.113475
3	C	4.469987	6.172881	-0.989146	56	C	-3.224394	-5.956099	-1.569441
4	C	2.208970	5.713482	-1.807116	57	C	-4.125409	-6.965853	-1.226783
5	C	4.286676	7.534513	-1.248984	58	C	-1.985790	-6.225543	-2.156203
6	C	2.029227	7.076264	-2.064772	59	C	-3.773613	-8.295719	-1.476859
7	C	3.070316	7.967798	-1.786140	60	C	-1.636787	-7.557248	-2.402190
8	H	5.070689	8.262438	-1.037721	61	C	-2.534291	-8.573501	-2.060356
9	H	1.094320	7.456715	-2.475544	62	H	-4.442619	-9.115819	-1.216987
10	H	2.928178	9.027342	-1.986668	63	H	-0.678338	-7.819020	-2.849553
11	N	1.746825	3.372891	-1.643476	64	H	-2.258566	-9.607980	-2.250530
12	C	1.259823	4.612589	-1.992030	65	N	-5.356591	-5.085353	-0.411015
13	C	0.962305	2.292849	-1.761569	66	C	-5.341941	-6.447687	-0.597968
14	C	-0.047568	4.739961	-2.450412	67	C	-6.419706	-4.506091	0.166207
15	C	-0.342365	2.370128	-2.224050	68	C	-6.428779	-7.218104	-0.195631
16	H	1.409463	1.347175	-1.460600	69	C	-7.525790	-5.231237	0.581175
17	C	-0.855711	3.615682	-2.567203	70	H	-6.361348	-3.427341	0.286631
18	H	-0.424282	5.726854	-2.708660	71	C	-7.528407	-6.610103	0.393749
19	H	-0.938612	1.464005	-2.292792	72	H	-6.401020	-8.293988	-0.353114
20	H	-1.880683	3.712978	-2.917236	73	H	-8.363256	-4.717123	1.042141
21	N	5.491484	4.146926	-0.234087	74	H	-8.381197	-7.209405	0.705381
22	C	6.488191	3.437293	0.314577	75	N	-1.816118	-3.844759	-2.015918
23	C	5.631257	5.505460	-0.396466	76	C	-1.203636	-5.011802	-2.409653
24	C	7.673202	4.022123	0.730638	77	C	-1.200467	-2.670594	-2.214290
25	H	6.307688	2.369829	0.412726	78	C	0.062364	-4.970827	-2.984500
26	C	6.804625	6.136181	0.004599	79	C	0.054263	-2.579873	-2.797272
27	C	7.831533	5.394391	0.571241	80	H	-1.742526	-1.790162	-1.874398
28	H	8.450329	3.405245	1.170500	81	C	0.697941	-3.751722	-3.178414
29	H	6.902384	7.210048	-0.134644	82	H	0.544686	-5.901105	-3.273359
30	H	8.748678	5.884813	0.886884	83	H	0.514117	-1.604274	-2.929536
31	C	3.928504	1.370724	-0.492041	84	H	1.691050	-3.717691	-3.619370
32	C	4.063709	0.151266	-0.334434	85	O	-6.227089	3.424500	0.205994
33	C	4.172438	-1.256449	-0.227528	86	O	-1.549004	2.992943	0.769810
34	C	5.427441	-1.872357	-0.281278	87	C	-7.520452	2.875004	-0.020346
35	C	2.963821	-2.055646	-0.130012	88	H	-7.552122	2.393881	-1.011438
36	C	5.527174	-3.261029	-0.279588	89	H	-7.732683	2.099623	0.733284
37	H	6.308230	-1.241554	-0.353916	90	C	-8.526418	3.995953	0.067240
38	C	3.113607	-3.491222	-0.151205	91	H	-8.268426	4.769912	-0.670415
39	C	4.366310	-4.055946	-0.222296	92	H	-8.452065	4.466615	1.058544
40	H	4.460435	-5.138828	-0.217649	93	C	-9.943704	3.498609	-0.170476
41	C	1.748419	-1.425038	-0.052585	94	H	-10.006564	3.026133	-1.164113
42	C	0.655664	-0.814079	0.077618	95	H	-10.182270	2.705692	0.556801
43	C	-0.500036	-0.208766	0.190394	96	C	-10.983592	4.603262	-0.069577
44	C	-1.613838	0.384600	0.250052	97	H	-10.742973	5.399577	-0.792859
45	C	-2.785706	1.074451	0.253856	98	H	-10.922257	5.072494	0.926004
46	C	-4.069605	0.443074	-0.005832	99	C	-12.402112	4.111995	-0.310750
47	C	-2.768611	2.503210	0.501503	100	H	-12.464045	3.649126	-1.309557
48	C	-5.234571	1.224113	-0.013031	101	H	-12.637429	3.308753	0.407010
49	C	-3.923034	3.238254	0.457148	102	C	-13.447087	5.210638	-0.197422
50	C	-5.165105	2.594575	0.206361	103	H	-13.209827	6.016393	-0.911778
51	H	-6.181625	0.732890	-0.215058	104	H	-13.386874	5.670130	0.803109
52	H	-3.931893	4.310238	0.630477	105	C	-14.865612	4.720814	-0.442672
53	C	-4.099820	-0.937050	-0.299028	106	H	-14.927920	4.268054	-1.446162

107	H	-15.100122	3.909909	0.266808	160	H	21.617317	-6.503875	-0.050747
108	C	-15.911320	5.817564	-0.317684	161	H	22.573500	-5.256096	0.763475
109	H	-15.673802	6.631167	-1.023078	162	H	21.479439	-6.286503	1.698667
110	H	-15.851225	6.266302	0.687772	163	C	0.888243	-4.117177	0.598703
111	C	-17.330236	5.331695	-0.568557	164	H	0.198998	-3.487742	0.018647
112	H	-17.392896	4.890474	-1.577184	165	H	1.133558	-3.581943	1.528451
113	H	-17.565816	4.513030	0.131649	166	C	0.230940	-5.442352	0.897732
114	C	-18.374364	6.428345	-0.431315	167	H	0.065173	-5.979431	-0.050629
115	H	-18.315806	6.864625	0.579838	168	H	0.902321	-6.067245	1.504783
116	H	-18.134923	7.250930	-1.125777	169	C	-1.095856	-5.223852	1.607748
117	C	-19.794766	5.949304	-0.690684	170	H	-0.919429	-4.725575	2.574993
118	H	-19.853210	5.522781	-1.704033	171	H	-1.711800	-4.523550	1.013373
119	H	-20.028954	5.122551	-0.002278	172	C	-1.893569	-6.497966	1.829010
120	C	-20.822824	7.056747	-0.536246	173	H	-2.071730	-6.987031	0.855183
121	H	-21.841728	6.703808	-0.733283	174	H	-1.306239	-7.212363	2.429281
122	H	-20.806660	7.471184	0.480423	175	C	-3.227630	-6.226356	2.505363
123	H	-20.619080	7.884204	-1.228786	176	H	-3.775532	-5.469268	1.915705
124	O	6.686821	-3.945001	-0.325000	177	H	-3.051940	-5.767706	3.492798
125	O	2.079894	-4.361572	-0.153930	178	C	-4.104384	-7.457118	2.667543
126	C	7.909657	-3.215910	-0.334121	179	H	-4.291874	-7.901468	1.673555
127	H	7.974242	-2.606305	-1.250266	180	H	-3.566993	-8.226959	3.245848
128	H	7.940932	-2.528857	0.526699	181	C	-5.431929	-7.144239	3.339253
129	C	9.042038	-4.210332	-0.263381	182	H	-5.926212	-6.323103	2.791053
130	H	8.967917	-4.903117	-1.114357	183	H	-5.246576	-6.757202	4.354843
131	H	8.924439	-4.815538	0.647501	184	C	-6.380856	-8.328977	3.412337
132	C	10.399522	-3.524788	-0.260703	185	H	-6.591755	-8.686146	2.389140
133	H	10.522244	-2.942933	-1.188444	186	H	-5.888422	-9.170013	3.927777
134	H	10.440322	-2.793560	0.562989	187	C	-7.691163	-8.001669	4.111289
135	C	11.553128	-4.505639	-0.118974	188	H	-8.148452	-7.117377	3.634562
136	H	11.511342	-5.241723	-0.938462	189	H	-7.485425	-7.703409	5.152406
137	H	11.424730	-5.083551	0.810957	190	C	-8.687535	-9.149817	4.100445
138	C	12.916991	-3.833270	-0.109570	191	H	-8.912183	-9.428613	3.056651
139	H	13.054646	-3.269383	-1.046980	192	H	-8.225526	-10.042146	4.554850
140	H	12.948607	-3.084055	0.698881	193	C	-9.985468	-8.829286	4.826391
141	C	14.067618	-4.812191	0.066443	194	H	-10.439206	-7.933033	4.375498
142	H	14.037515	-5.561887	-0.741576	195	H	-9.757456	-8.558093	5.868666
143	H	13.925341	-5.375914	1.003406	196	C	-10.976040	-9.980497	4.796585
144	C	15.433334	-4.143185	0.084715	197	H	-11.906013	-9.737212	5.323402
145	H	15.581186	-3.586652	-0.855660	198	H	-11.239723	-10.248386	3.764861
146	H	15.458700	-3.387369	0.887304	199	H	-10.553547	-10.877517	5.268357
147	C	16.581506	-5.121731	0.278626	200	C	-1.392781	4.391967	0.971750
148	H	16.557854	-5.877303	-0.524182	201	H	-2.081041	4.738320	1.759169
149	H	16.430837	-5.678658	1.218376	202	H	-1.650501	4.926156	0.039975
150	C	17.947592	-4.453528	0.303171	203	C	0.043597	4.635083	1.365522
151	H	18.101441	-3.900963	-0.638638	204	H	0.232092	4.164516	2.341528
152	H	17.968178	-3.694159	1.102569	205	H	0.701970	4.127748	0.643122
153	C	19.094834	-5.430761	0.507943	206	C	0.386789	6.114607	1.418786
154	H	19.076436	-6.190890	-0.290885	207	H	0.259769	6.553077	0.413241
155	H	18.941134	-5.983324	1.449905	208	H	-0.323199	6.644269	2.075373
156	C	20.461316	-4.762012	0.534976	209	C	1.809599	6.363442	1.894236
157	H	20.475576	-4.000638	1.330279	210	H	1.933090	5.957445	2.911466
158	H	20.614990	-4.214832	-0.408033	211	H	2.505362	5.794170	1.253222
159	C	21.595181	-5.750175	0.747467	212	C	2.202102	7.831860	1.875949

213	H	2.026531	8.238114	0.865202
214	H	1.543299	8.402343	2.551773
215	C	3.653931	8.068603	2.258361
216	H	3.838654	7.678480	3.272850
217	H	4.301707	7.478775	1.585673
218	C	4.064178	9.530959	2.192405
219	H	3.426801	10.122790	2.870175
220	H	3.864211	9.918630	1.178423
221	C	5.525643	9.762435	2.542021
222	H	5.731618	9.348418	3.542951
223	H	6.161539	9.189246	1.845671
224	C	5.931948	11.227360	2.509649
225	H	5.298046	11.795888	3.210153
226	H	5.722090	11.644400	1.510574
227	C	7.394706	11.457286	2.855510
228	H	8.031363	10.900476	2.147714
229	H	7.607509	11.029341	3.849315
230	C	7.793542	12.925450	2.844532
231	H	7.157749	13.476651	3.554473
232	H	7.576499	13.351809	1.852906
233	C	9.257115	13.139870	3.188973
234	H	9.911278	12.622536	2.474779
235	H	9.530892	14.201257	3.179292
236	H	9.489776	12.747203	4.187667

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