

**Ionic Liquids Based Catanionic Coacervates: The Novel Microreactors for
Membrane Free Sequestration of Dyes and Curcumin**

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Table SI-1: Sodium Salicylate induced Structural Transition in various surfactants.

Sr no	Surfactant	R =[NaSal]/[Surfactant]	Transition*	Ref
1	Cetyltrimethylammonium Chloride	1.0	STT	1
2	Cetyltrimethylammonium Bromide	1.0	STT	2,3
		1.2	STW	4
		0.96	STR	5,6
3	N-cetyl-N-(2-hydroxyethyl) dimethylammonium bromide	0.4 to 1.0	STW	7
4	N-cetyl-N,N-di(2-hydroxyethyl) methylammonium bromide	0.4 to 1.0	STW	7
5	Star-typetrimeric surfactants	0.24 to 0.30	STR	8
		0.45 \leq R \leq 0.66	STW	
		R > 0.79	STV	
6	Ethanediyl- α,ω -bis(dimethyldodecyl ammonium bromide) (12-2-12)	R > 0.96	STV	8
8	N-tetradecyl-N-(2-hydroxyethyl) dimethylammonium bromide	0.70	STW	9
9	1,2-bis[N-ethyl-N-(sodium 2-hydroxyl-3-sulfopropyl)-dodecyl-ammonium] ethane betaine	\leq 0.3	STW	10
		0.8	STLW	
10	Didecyldimethylammoniumformate	0.25	WTV	11
		0.43	STW	

11	N-methyl-N-cetylpyrrolidinium bromide	0.4 (40 mM/100 mM)	STG	12
12	hexanediyl- α,ω -bis(dimethylcetyl ammonium bromide) (16-6-16)	1.8	STW	13

*STT = Spherical micelle to Threadlike Micelle, STR = Spherical micelle to Rodlike Micelle, STW = Spherical micelle to Wormlike Micelle, STLW = Spherical micelle to Long Wormlike Micelle, STV = Spherical micelle to Vesicles, STG = Spherical micelle to Hydrogel, WTV= Wormlike micelle to vesicle.

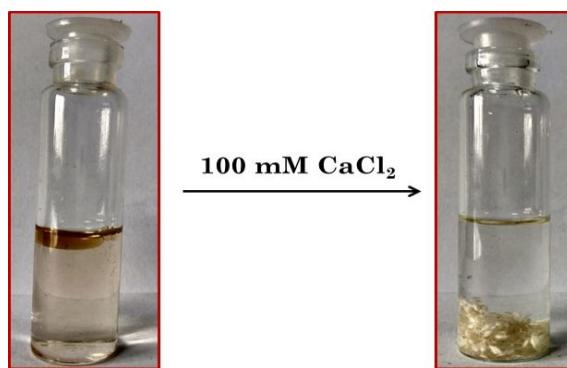


Figure SI-1. Visual Images of Complex coacervates before and after addition of 100 mM of CaCl₂. (Photograph courtesy of 'Ankit Shah')

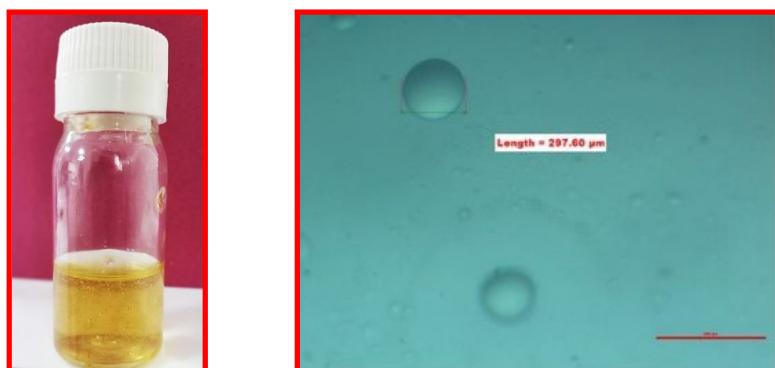


Figure SI-2. Visual and Optical microscopy images of the complex coacervates with C₈EMeImBr (112.5mM) and NaSal (100 mM) in the presence of 0.15 % w/v of NaAlg (Scale bar = 500 µm). (Photograph courtesy of 'Ankit Shah')

References

1. Wang, Z.; Larson, R.G. Molecular Dynamics Simulations of Threadlike Cetyltrimethylammonium Chloride Micelles: Effects of Sodium Chloride and Sodium Salicylate Salts. *J. Phys. Chem. B* **2009**, *113*, 13697–13710.
2. Mohanty, S.; Davis, H. T.; McCormick, A. V. Stratification in Drying Films Containing Bidisperse Mixtures of Nanoparticles. *Langmuir*, **2001**, *17*, 7160–7171.

3. Shikata, T.; Hirata, H.; Kotaka, T. Micelle formation of detergent molecules in aqueous media. 2. Role of free salicylate ions on viscoelastic properties of aqueous cetyltrimethylammonium bromide-sodium salicylate solutions. *Langmuir*, **1988**, *4*, 354–359.
4. Peia, X.; Zhaoa, J.; Wei, X. Comparative study of the viscoelastic wormlike micellar solutions of hexanediy1- α,ω -bis(dimethylcetyl)ammonium bromide and cetyltrimethylammonium bromide in the presence of sodium salicylate, *Colloids and Surfaces A: Physicochem. Eng. Aspects*, **2010**, *366*, 203–207.
5. Aswal, V.K.; Goyal, P.S.; Thiagarajan, P. Small-Angle neutron-scattering and viscosity studies of CTAB/NaSal viscoelastic micellar solutions. *J Phys Chem B*, **1998**, *102*, 2469–2473.
6. Lin, C.; Zhao, J. Nile red probing for sphere-to-rod-to-wormlike micelle transition in aqueous surfactant solution. *Dyes and Pigments*, **2010**, *84*, 223–228.
7. Dai, C.; Wu, X.; Li, W.; You, Q.; Zhao, M.; Du, M.; Liu, Y.; Li, Y. The role of hydroxyethyl groups in the construction of wormlike micelles in the system of quaternary ammonium surfactant and sodium salicylate. *Soft Matter*, **2015**, *11*, 7817-7820.
8. Kusano, T.; Akutsu, K.; Iwase, H.; Yoshimura, T.; Shibayama, M. Structural study on aggregation behavior of star-type trimericsurfactant in the presence of sodium salicylate. *Colloids and Surfaces A: Physicochem. Eng. Aspects*, **2016**, *497*, 109–116.
9. Dai, C.; Lia, W.; Cui, Y.; Sun, Y.; Wu, W.; Xu, Z.; Liu, Y.; Yang, Z.; Wu, X. The effect of functional groups on the sphere-to-wormlike micellar transition in quaternary ammonium surfactant solutions. *Colloids and Surfaces A: Physicochem. Eng. Aspects*, **2016**, *500*, 32–39.
10. Xiang, F.; Geng Q.; Xing Q.; Hu C.; Xue C.; Jia J.L.; Li J. L. Effects of sodium salicylate on the microstructure of a novel zwitterionic gemini surfactant and its rheological responses. *Colloid Polym Sci*, **2014**, *292*, 915–921.
11. Zhao, L.; Zhang, H.; Wang, W.; Wang, G. Effects of sodium salicylate on didecyldimethylammonium formate properties and aggregation behaviors, *J Mol. Liq.* **2017**, *225*, 897–902.

12. Yana, H.; Zhaoa, M.; Zheng, L. A hydrogel formed by cetylpyrrolidinium bromide and sodium salicylate. *Colloids and Surfaces A: Physicochem. Eng. Aspects*, **2011**, *392*, 205– 212.
13. Manohar, C.; Rao, U.R.K.; Valaulikar, B.S.; Lyer, R.M. On the origin of viscoelasticity in micellar solutions of cetyltrimethylammonium bromide and sodium salicylate. *J. Chem. Soc. Chem. Commun.* **1986**, *5*, 379–381.