

The Role of Biorelevant Dissolution Media in the
Selection of Optimal Salt Forms of Oral Drugs:
Maximising the Gastrointestinal Solubility and *In
Vitro* Activity of the Antimicrobial Molecule,
Clofazimine.

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Biorelevant dissolution media, Supersaturating Drug Delivery Systems, Clofazimine, Drug
Repurposing, Antimicrobial resistance,

TABLES:

Solid form	Bond observed and peak position (1/cm)	
	$N_C^+ - H$	$C=N_C$
CFZ F III	n/a	1626.5
CFZ.Sul	3299.8	1624.3 (br)
CFZ.Pho	3325.2 (br)	1620.9
CFZ.Cit	3323.9 (br)	1616.9
CFZ.Ace	3325.2 (br)	1621.4

Table S1: Summary of the peaks position for the key function groups in various CFZ solid forms from the FTIR spectra

Bond observed	Chemical shift (ppm)	Intermolecular distance** (Å)	Chemical shift (ppm)	Intermolecular distance** (Å)	Chemical shift (ppm)	Intermolecular distance** (Å)
	CFZF I*	49.4	1.465	149 - 151.4 (m)	1.293	149 - 151.4 (m)
	CFZF II	48.7	1.457	149.4 - 151.6 (m)	1.298	149.4 - 151.6 (m)
	CFZF III	49	1.457	149.73	1.296	150.7
	Citrate	48.84 - 46.46 (m), 45.47, 44.71	1.471	143.05, 141.65	1.328	151.08
	Phosphate	48.21, 47.16	1.469	143.84	1.322	152.35 - 150.24 (m), 149.70
	Sulphate	51.21, 48.23	1.462	144.26 - 142.46 (m)	1.322	150.96
Solution NMR:						
CFZ in CDCl ₃	49.4	n/a	150.4	n/a	151.1	n/a
CFZ from Gaussian 09 software (cal)	50.46	n/a	152.12	n/a	152.55	n/a

*taken from FI resolved as DAKXU101 exhibits disorder in its isopropyl group.

**Calculated using Mercury software from single crystal cif files (average taken for solid Z' numbers greater > 1)

Table S2: Summary of bond lengths from crystal structures of CFZ solid forms and ¹³C chemical shifts of the same key functional groups in CPMAS ¹³C spectra for CFZ solid forms. The chemical shifts of the CFZ carbons from solution NMR and molecular modelling are also shown for comparison

FIGURES:

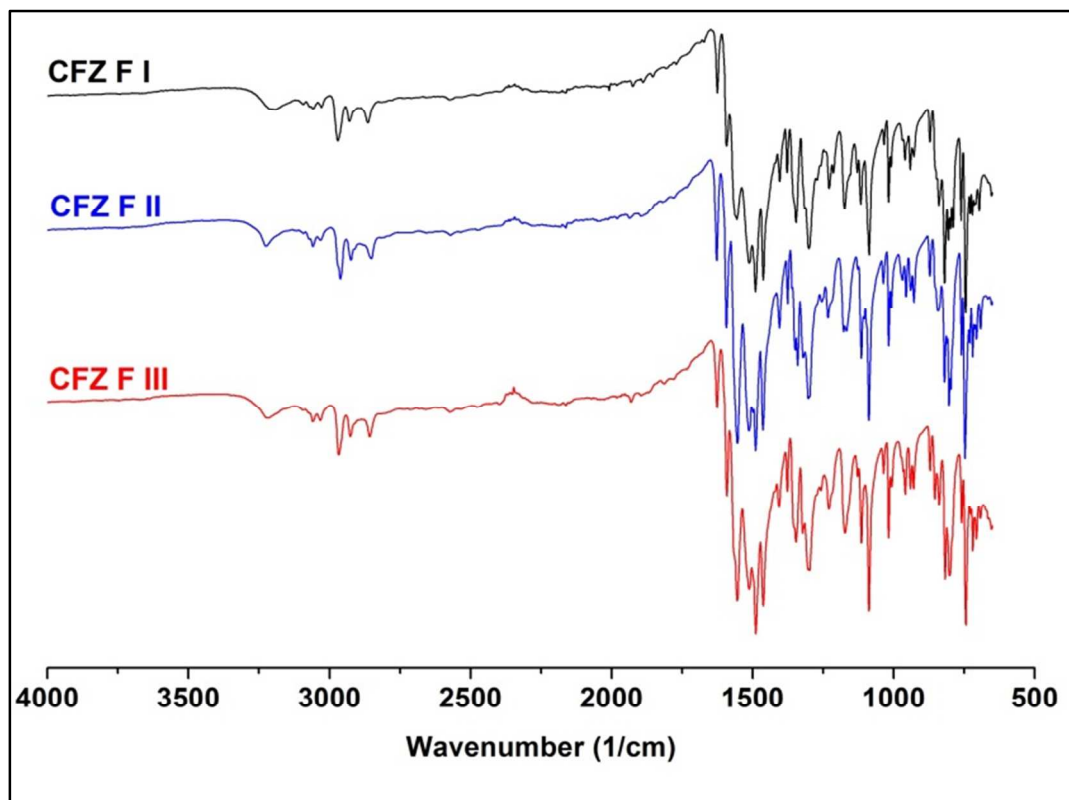


Figure S1: Comparison of the full FTIR spectra for the CFZ polymorphs F I, F II and F III.

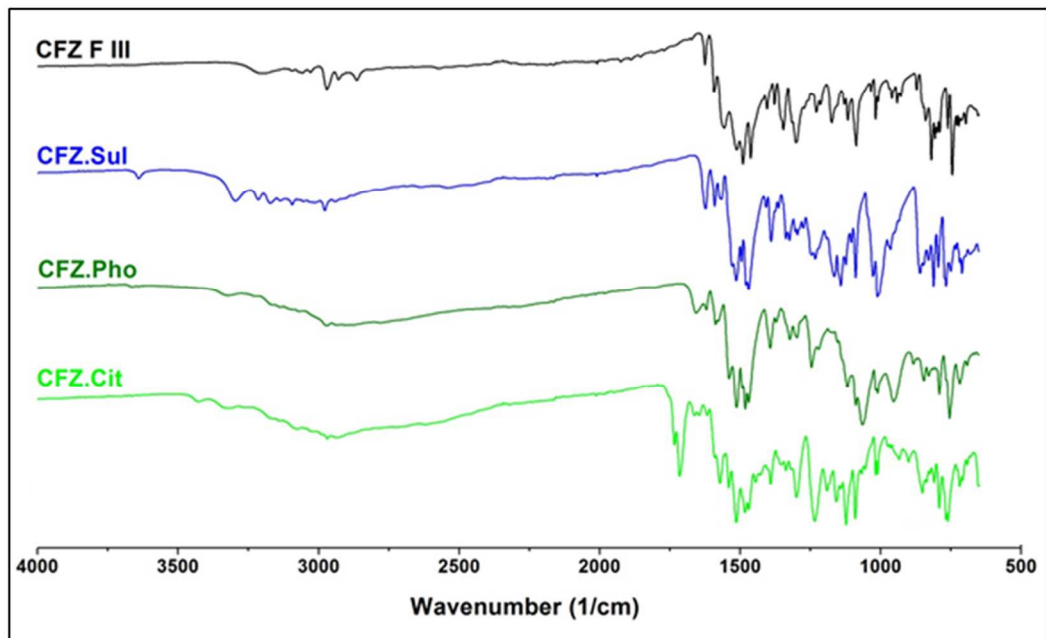


Figure S2: Comparison of the full FTIR spectra of the new CFZ salts with CFZ F III.

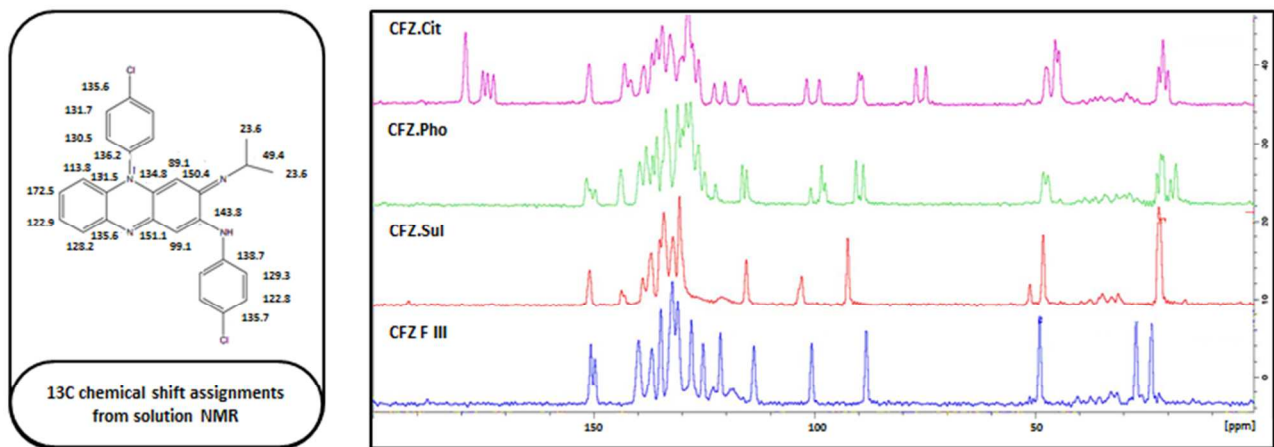


Figure S3: Left; Chemical structure of CFZ with ^{13}C peaks assigned. Right; Comparison of full ^{13}C CPMAS spectra obtained of acetate, citrate, phosphate, sulphate and CFZ F III, peaks corresponding to the secondary ketimine carbon (green) and the isopropyl carbon (blue) are highlighted.