

Supporting information

Selective Bactericidal Activity of Divalent Metal Salts of Lauric Acid

Yoshiaki Yamamoto[†], Toshiya Morikawa[‡], Takahiro Kawai[†] and Yoshimune Nonomura^{†*}

Table S1. Bactericidal activities of the lauric acid and its divalent metal salts.

		<i>S. aureus</i>			
		0 h	1 h	3 h	24 h
Lauric acid	Average	5.1	3.0	<1	<1
	Standard deviation	0.4	1.0	0.0	0.0
Mg salt	Average	5.4	2.9	0.9	<1
	Standard deviation	0.2	0.8	1.5	0.0
Ca salt	Average	5.5	4.3	2.5	<1
	Standard deviation	0.3	0.9	2.3	0.0
Mn salt	Average	5.6	<1	<1	<1
	Standard deviation	0.0	0.0	0.0	0.0
Co salt	Average	5.6	2.4	<1	<1
	Standard deviation	0.1	0.5	0.0	0.0
Ni salt	Average	5.5	2.2	<1	<1
	Standard deviation	0.2	0.2	0.0	0.0
Cu salt	Average	5.6	<1	<1	<1
	Standard deviation	0.1	0.0	0.0	0.0
Zn salt	Average	6.1	5.7	5.6	4.9
	Standard deviation	0.6	0.1	0.1	0.2
		<i>S. epidermidis</i>			
		0 h	1 h	3 h	24 h
Lauric acid	Average	5.4	5.3	5.3	4.8
	Standard deviation	0.0	0.0	0.0	0.0
Mg salt	Average	5.8	5.8	5.8	4.1
	Standard deviation	0.0	0.0	0.1	0.1
Ca salt	Average	5.7	5.4	5.4	5.1
	Standard deviation	0.1	0.1	0.1	0.3
Mn salt	Average	5.9	5.9	5.3	5.4
	Standard deviation	0.0	0.1	0.5	0.3
Co salt	Average	5.8	5.1	4.0	<1
	Standard deviation	0.0	0.3	0.4	0.0
Ni salt	Average	5.8	5.0	4.0	<1
	Standard deviation	0.0	0.1	0.5	0.0
Cu salt	Average	5.9	<1	<1	<1
	Standard deviation	0.0	0.0	0.0	0.0
Zn salt	Average	5.9	5.8	5.8	5.5
	Standard deviation	0.0	0.1	0.0	0.1
		<i>P. acnes</i>			
		0 h	1 h	3 h	24 h
Lauric acid	Average	5.5	5.6	1.8	<1
	Standard deviation	0.2	0.6	1.6	0.0
Mg salt	Average	5.6	5.5	5.1	<1
	Standard deviation	0.1	0.0	0.2	0.0
Ca salt	Average	5.5	5.2	4.6	0.8
	Standard deviation	0.1	0.2	0.5	1.4
Mn salt	Average	5.8	4.5	<1	<1
	Standard deviation	0.1	0.0	0.0	0.0
Co salt	Average	5.8	3.7	<1	<1
	Standard deviation	0.1	0.0	0.0	0.0
Ni salt	Average	5.7	4.2	<1	<1
	Standard deviation	0.1	0.1	0.0	0.0
Cu salt	Average	5.9	5.3	<1	<1
	Standard deviation	0.1	0.2	0.0	0.0
Zn salt	Average	5.9	5.8	5.5	3.1
	Standard deviation	0.0	0.1	0.0	0.1

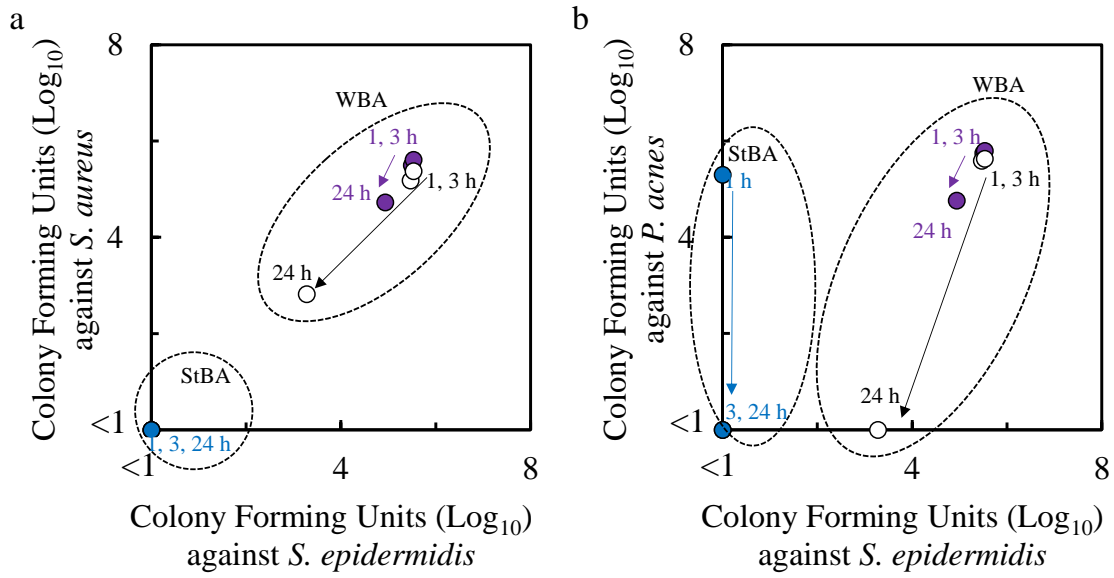


Figure S1. Bactericidal activities of metallic chlorides. Temporal changes of CFU for *S. aureus* and *S. epidermidis* (a) and *P. acnes* and *S. epidermidis* (b). Purple circle: CoCl_2 ; blue circle: CuCl_2 ; white circle: ZnCl_2 .

Table S2. Bactericidal activities of the metallic chlorides.

		<i>S. aureus</i>			
		0 h	1 h	3 h	24 h
Co salt	Average	5.7	5.6	5.5	4.7
	Standard deviation	0.0	0.1	0.1	0.3
Ni salt	Average	5.7	5.5	5.4	4.2
	Standard deviation	0.1	0.1	0.1	0.3
Cu salt	Average	5.7	<1	<1	<1
	Standard deviation	0.1	0.0	0.0	0.0
Zn salt	Average	5.6	5.4	5.2	2.8
	Standard deviation	0.1	0.1	0.0	0.6
		<i>S. epidermidis</i>			
		0 h	1 h	3 h	24 h
Co salt	Average	5.5	5.5	5.5	4.9
	Standard deviation	0.0	0.1	0.0	0.1
Ni salt	Average	5.5	5.3	5.2	4.2
	Standard deviation	0.0	0.0	0.0	0.2
Cu salt	Average	5.5	<1	<1	<1
	Standard deviation	0.0	0.0	0.0	0.0
Zn salt	Average	5.5	5.5	5.5	3.3
	Standard deviation	0.0	0.1	0.0	0.2
		<i>P. acnes</i>			
		0 h	1 h	3 h	24 h
Co salt	Average	5.9	5.8	5.7	4.8
	Standard deviation	0.1	0.1	0.1	0.2
Ni salt	Average	5.9	5.8	5.7	3.8
	Standard deviation	0.0	0.0	0.1	0.2
Cu salt	Average	5.7	5.3	<1	<1
	Standard deviation	0.0	0.1	0.0	0.0
Zn salt	Average	5.7	5.6	5.6	<1
	Standard deviation	0.0	0.0	0.1	0.0

Table S3. The surface tension of the solubility of metal salts of lauric acid in 2wt% aqueous ethanol at 25°C and 50% humidity. The metal salt concentration was 0.1 g L⁻¹.

	Mg salt	Ca salt	Mn salt	Co salt	Ni salt	Cu salt	Zn salt	Water
Surface tension / mN m ⁻¹	70.8±2.0	71.2±1.3	72.7±1.4	70.6±1.2	71.7±1.2	71.4±1.8	72.1±1.7	71.7±3.2