SUPPORTING INFORMATION

Reductive cleavage of organic peroxides by iron salts and thiols

(10 pages)

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Figure S-1. Visible monitoring of Fe/RSH promoted reduction (time lapse)

Figure S-2. ¹H NMR monitoring of the disproportionation of diacetone diperoxide (**6**) and the reductive decomposition of dialkyl peroxide (**3**).

Figure S-3. ¹H and ¹³C NMR spectra related to synthesis of alkoxy radical probe 8.

Figure S-4. ¹H NMR of reaction mixture from reaction of probe **8**.

Figure S-5. ¹H NMR spectra of purified reaction mixtures from reduction of peroxides 1-4.

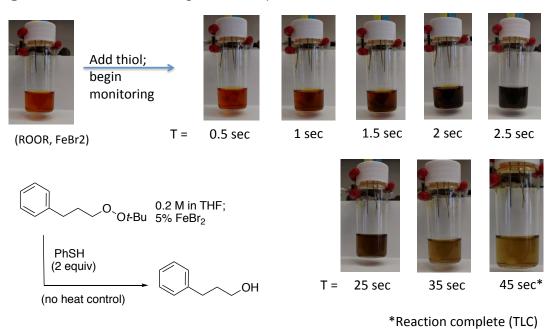
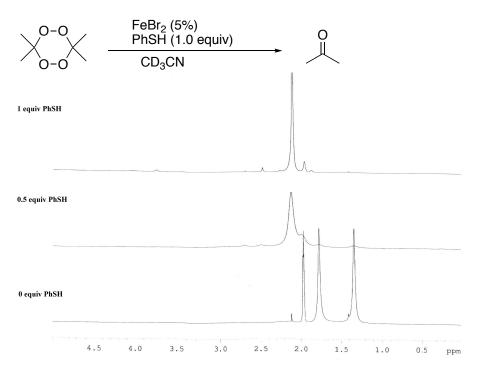


Figure S-1. Visual monitoring of Fe/thiol promoted reduction

Figure S-2

a. ¹H NMR monitoring of the FeBr₂/PhSH-promoted decomposition of diacetone diperoxide (6)



b. ¹H NMR monitoring of reduction of dialkyl peroxide **3** by Fe(SPh)_n/stoichiometric PhSH.

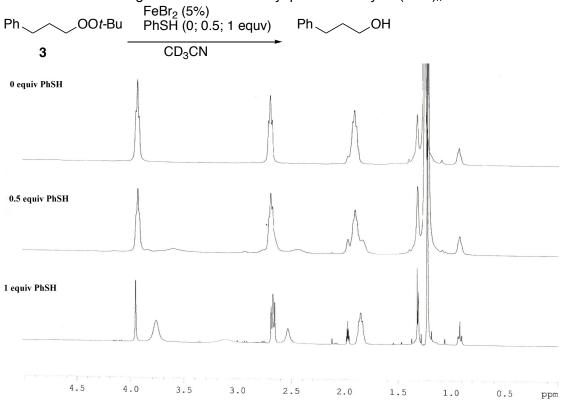
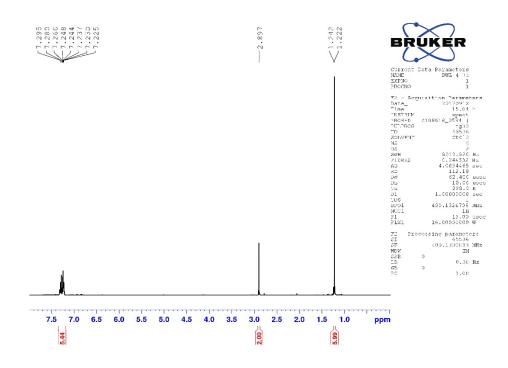
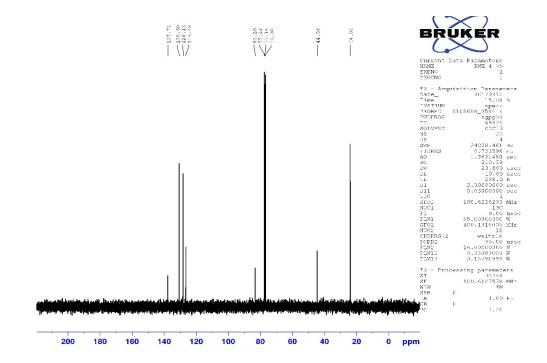


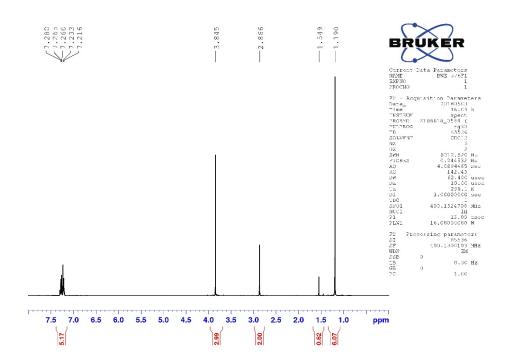
Figure S-3. NMR spectra related to synthesis of probe **8 a.** ¹H NMR of of 1-phenyl-2-methyl-2-propyl hydroperoxide



b. ¹³C NMR of 1-phenyl-2-methyl-2-propyl hydroperoxide



c. ¹H NMR of Methyl 1-phenyl-2-methyl-2-propyl peroxide (8)



d. ¹³C NMR of Methyl 1-phenyl-2-methyl-2-propyl peroxide (8)

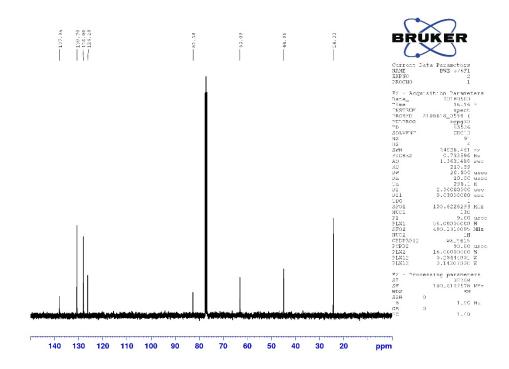


Figure S-4. ¹H NMR analysis of reductive fragmentation of radical probe **8**. (Relative yields shown; isolated yield of 2-methyl-1-phenyl-2-propanol in parentheses).

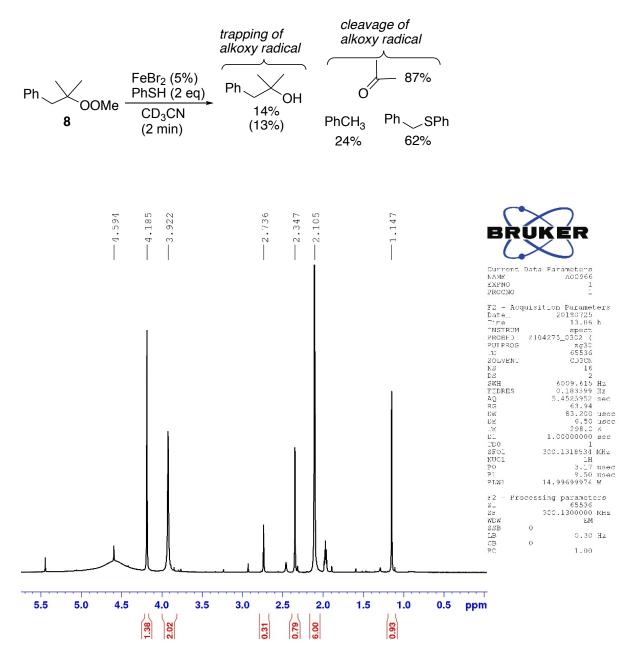


Figure S-5. 1H NMR spectra of purified reaction mixtures from reduction (5% FeBr2, 1.0 equiv PhSH) of peroxides: a) **1**; b. **2**; c. **3**; d. **4**.

