

Supporting Information

Small Naked Pt Nanoparticles Confined in Mesoporous Shell of Hollow Carbon Spheres for High-Performance Nonenzymatic Sensing of H₂O₂ and Glucose

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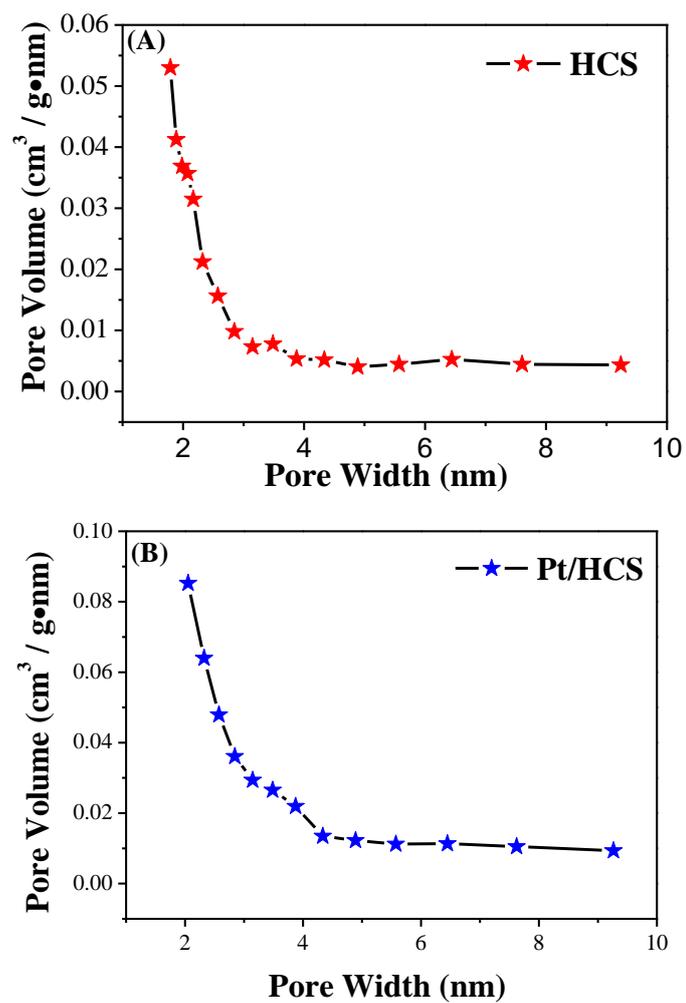


Figure S1 Pore size distributions of HCS (A) and Pt/HCS (B).

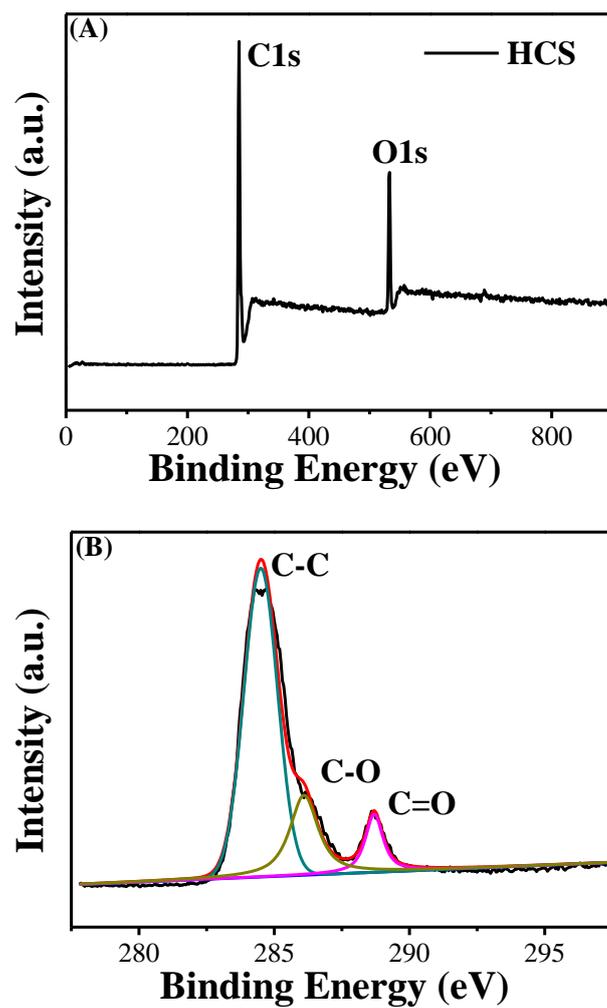


Figure S2 (A) XPS survey spectrum and (B) high-resolution C 1s XPS spectrum of HCS.

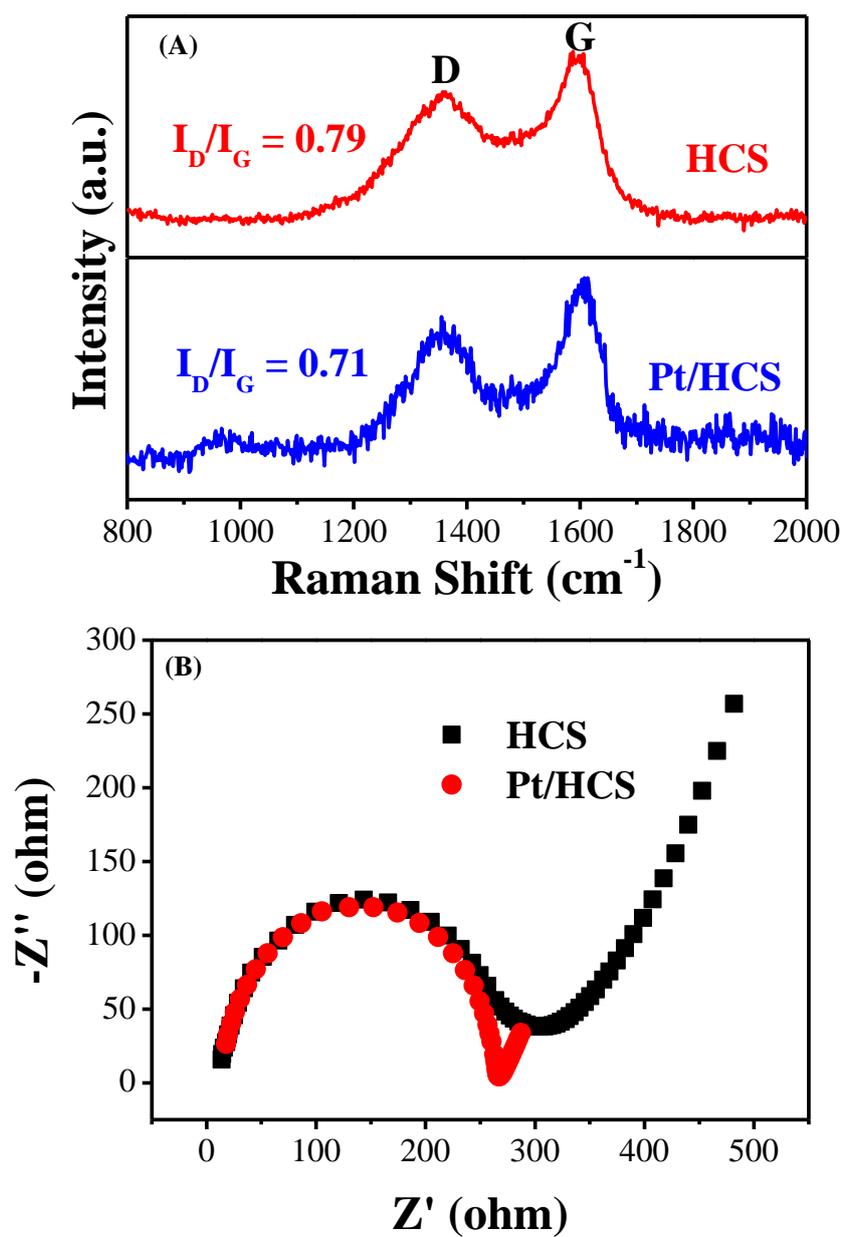


Figure S3 (A) Raman spectra of HCS (red line) and Pt/HCS (blue line). (B) Electrochemical impedance plots (Nyquist plots) of HCS and Pt/HCS in 0.1 M PBS (pH =7.4) buffer solution.

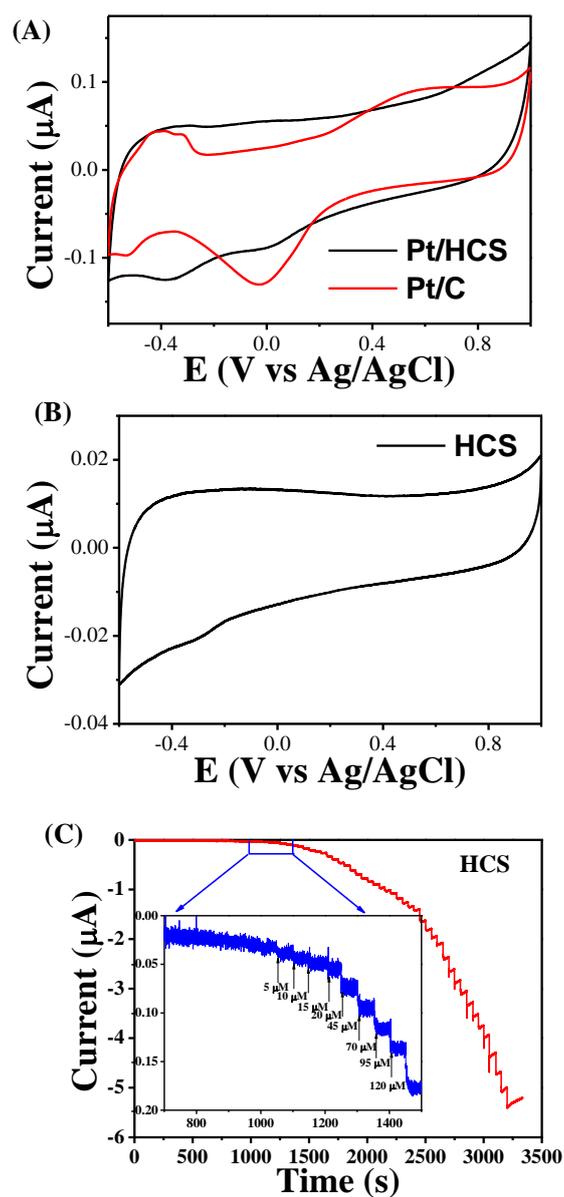


Figure S4 CV curves of Pt/HCS and Pt/C (A), HCS (B) in 0.1 M N₂-saturated PBS (pH = 7.4) solution with a scan rate of 50 mV s⁻¹. (C) Amperometric response of HCS-modified GCE to successive addition of H₂O₂ at the potential of -0.1 V in 0.1 M PBS (pH = 7.4) solution. The inset shows the amplified current signal at low concentrations of H₂O₂.

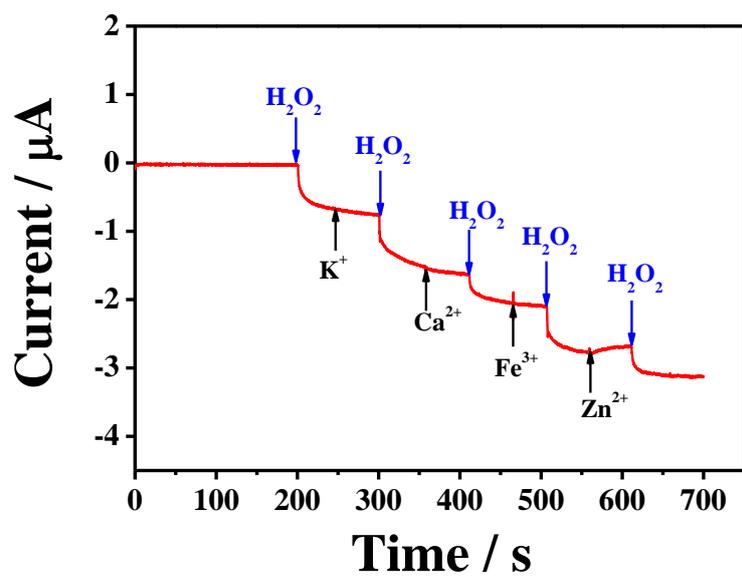


Figure S5 Current responses of Pt/HCS to the addition of 0.1 M H_2O_2 and 0.1 M K^+ , Ca^{2+} , Fe^{3+} and Zn^{2+} in 0.1 M N_2 -saturated PBS solution.

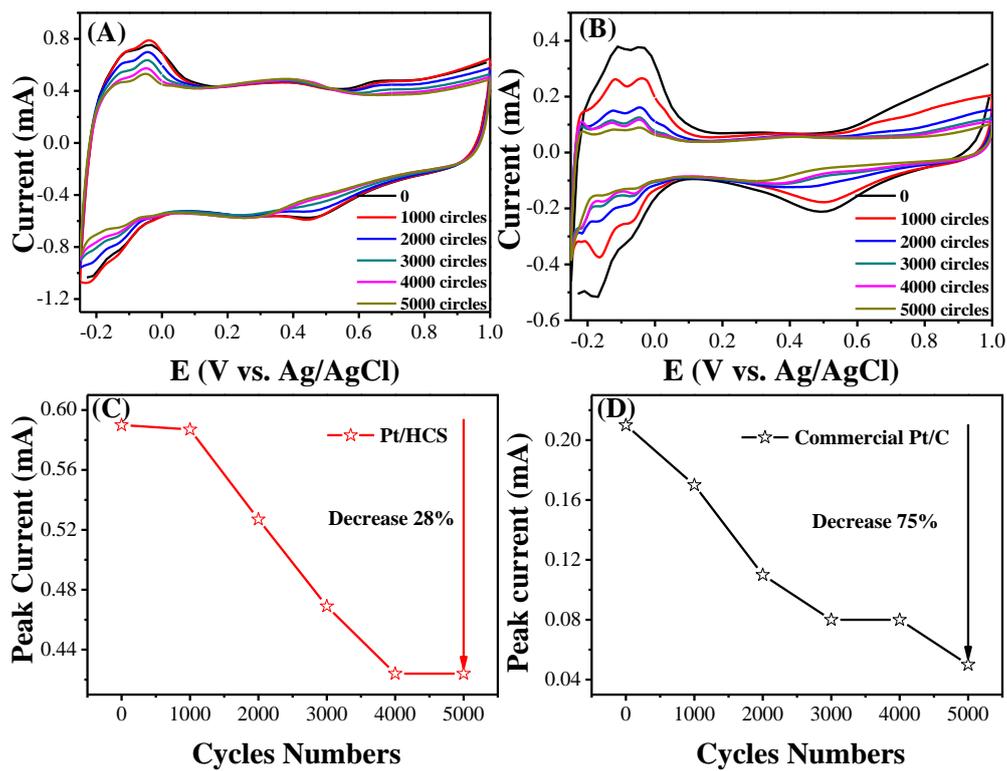


Figure S6 CV curves after 0, 1000, 2000, 3000, 4000 and 5000 cycles of (A) Pt/HCS and (B) commercial Pt/C in 0.1 M HClO₄ with the scan rate of 100 mV s⁻¹. The variations of peak currents after different potential scanning cycles on (C) Pt/HCS and (D) commercial Pt/C.