

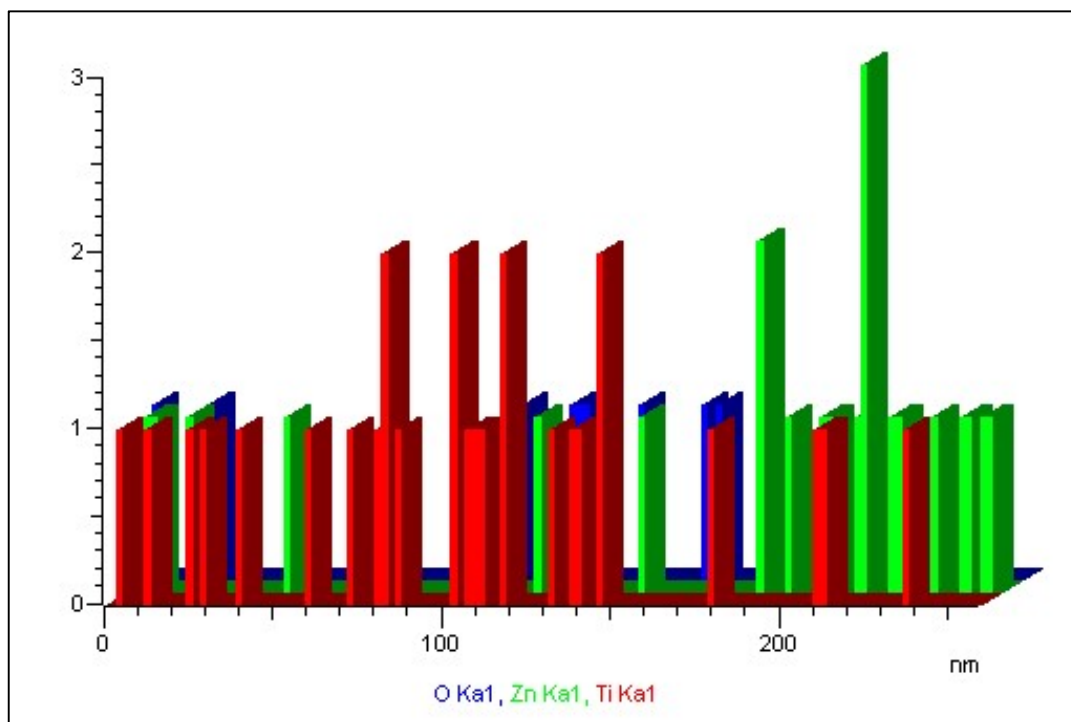
## **Supporting Information**

# **Amperometric Glucose Sensing at Nanomolar Level Using MOF Encapsulated TiO<sub>2</sub> Platform**

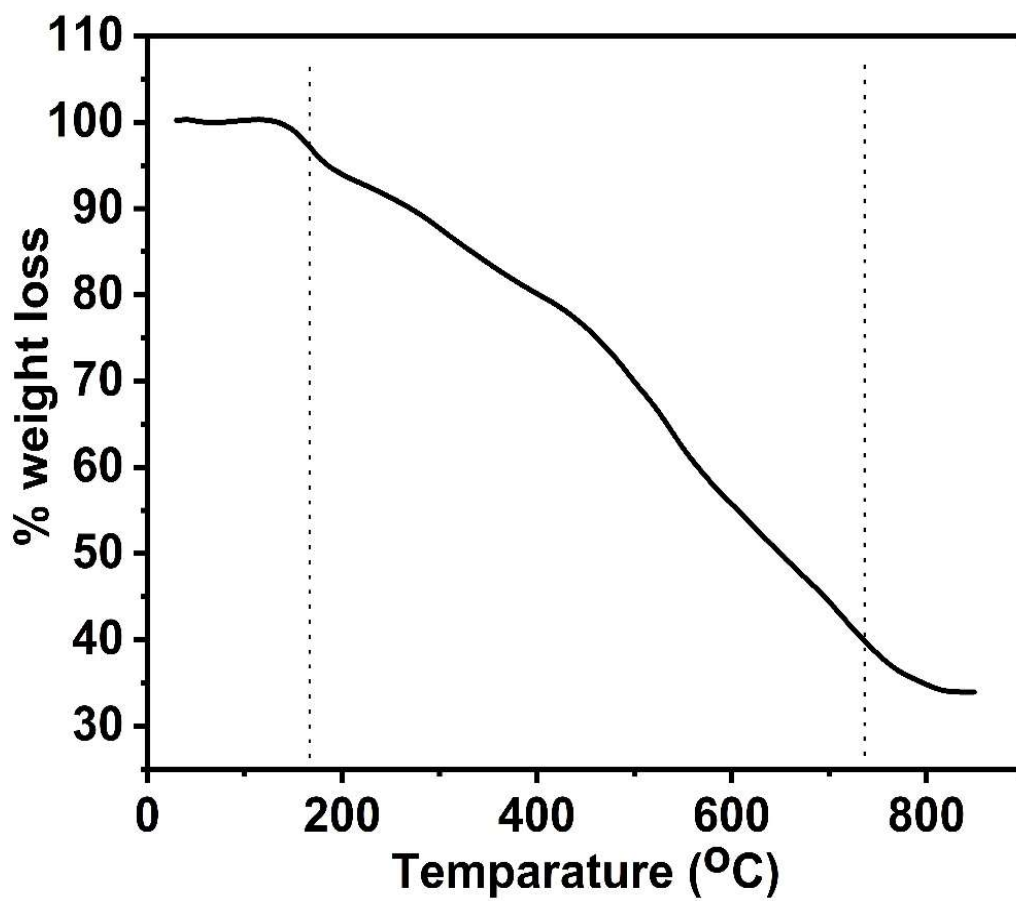
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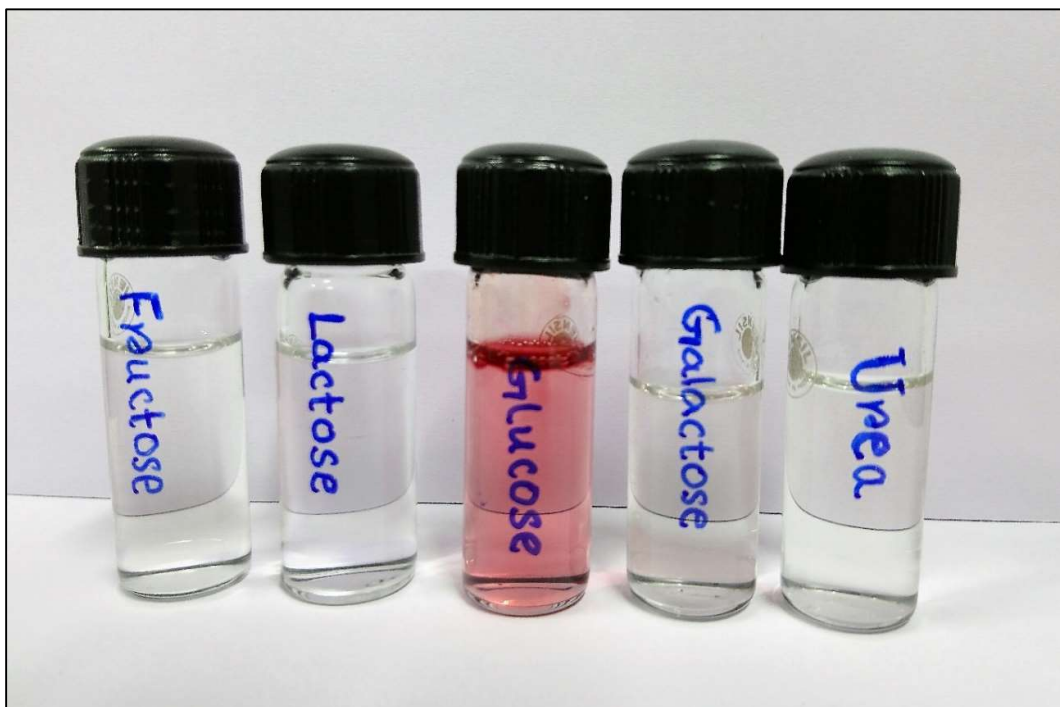
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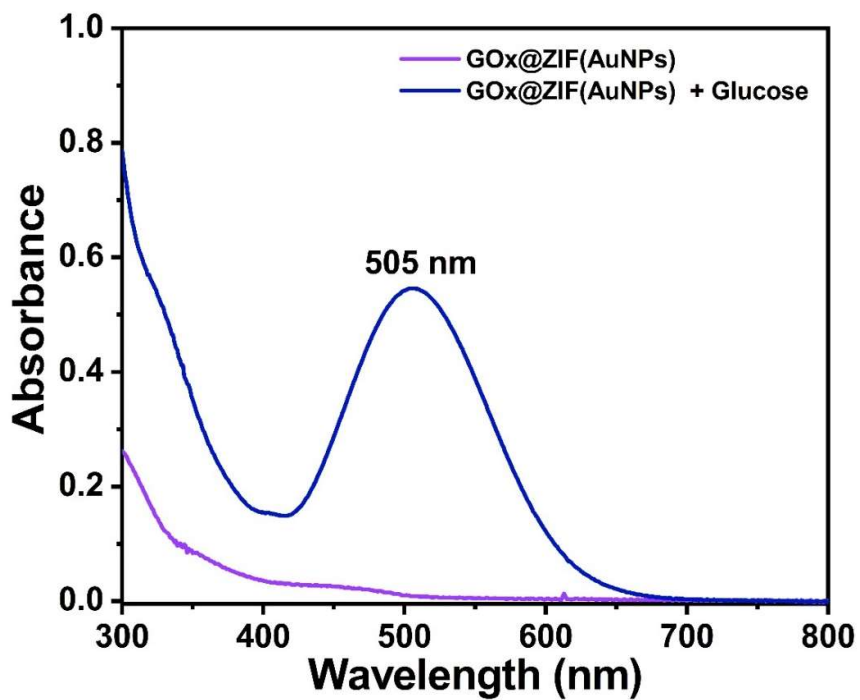
**Figure S1.** STEM-EDX mapping of TiO<sub>2</sub> (red colour building block) over ZIF-8 layers (green colour building block) showing equal spatial distribution of TiO<sub>2</sub>.



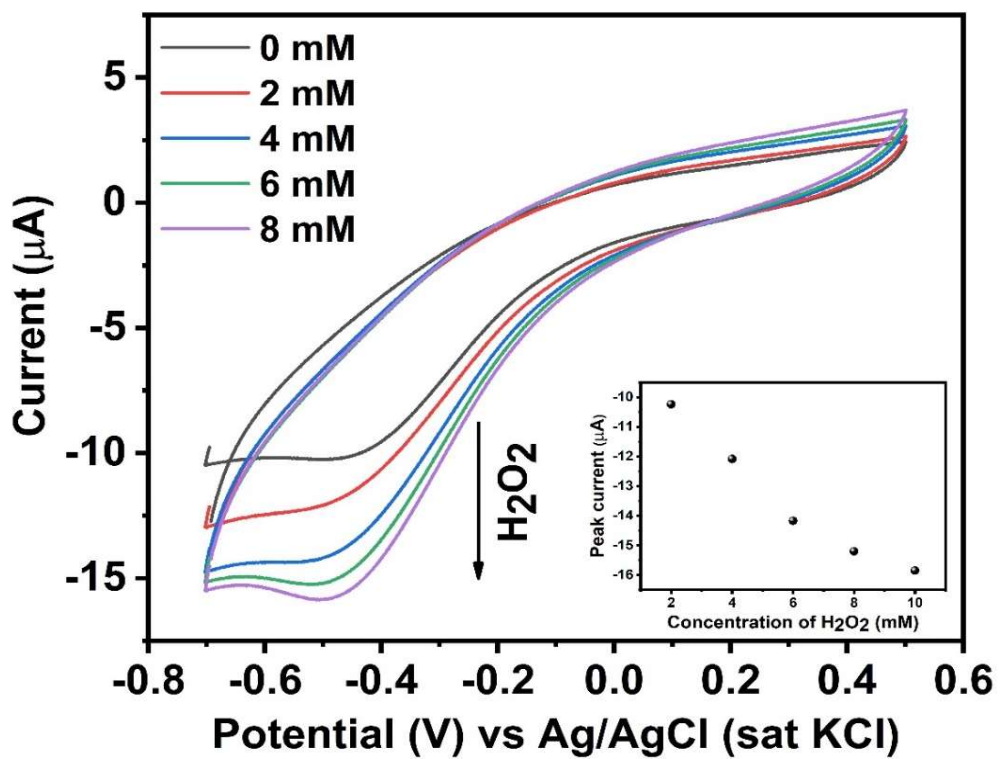
**Figure S2.** Thermogravimetric analysis of GO<sub>x</sub>@ZIF-8(TiO<sub>2</sub>) showing distinct degradation region at different temperature.



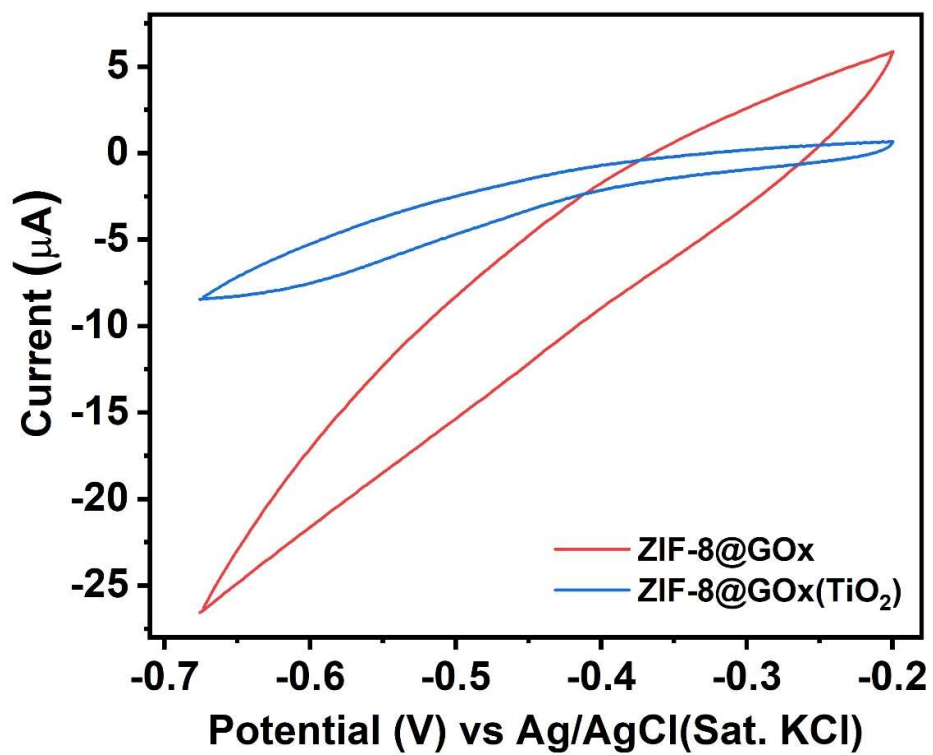
**Figure S3.** Optical image of the GOD-POD assay test output of the synthesized probe towards different interferent analyte.



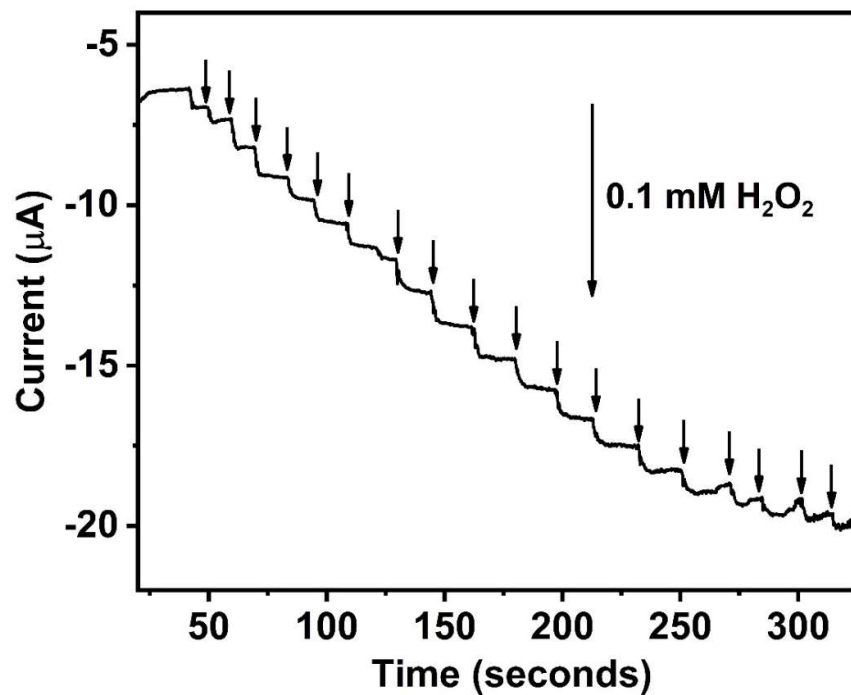
**Figure S4.** UV-VIS result of the probe  $\text{GO}_x@ZIF-8(\text{TiO}_2)$  upon GOD-POD assay test showing characteristic 505 nm peak for presence of  $\text{GO}_x$ .



**Figure S5.** Cyclic voltammetry of synthesized probe with different concentration of H<sub>2</sub>O<sub>2</sub> in pH 7.4 PBS using three electrode cell setups. Calibration of cathodic peak current versus H<sub>2</sub>O<sub>2</sub> concentration is depicted inset.



**Figure S6.** Comparative CV of ZIF-8@GOx and ZIF-8@GOx(TiO<sub>2</sub>) showing almost linear I-V type result for ZIF-8@GOx due to absence of capacitance provided by TiO<sub>2</sub> in as synthesized catalyst.



**Figure S7.** Amperometric response of synthesized probe at -0.45V vs Ag/AgCl (Sat. KCl) by successive addition of 0.1 mM H<sub>2</sub>O<sub>2</sub> in air saturated pH 7.4 PBS resulting a dynamic stair with quick steady state of current.