Polyethylene glycol (PEG₄₀₀) modified layered double hydroxides: synthesis, characterization and study on adsorption characteristics for removal of acid orange II

from aqueous solution

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Figure S1. Nitrogen adsorption–desorption isotherms taken at 77 K and the BJH desorption plots of pore volume versus pore diameter for the PLDH and LDH/PEG-15.



Figure S2. Pseudo second-order kinetic plots for the PLDH and the LDH/PEG-15 (solid lines for 100 mg/l and dotted lines for 200 mg/l of initial AO-II concentration).



Figure S3. AO-II uptake per gram of the adsorbents as a function of initial AO-II concentration at different reaction temperatures (Initial AO-II concentration: 0 - 800 mg/l, adsorbent dose: 0.5 g/l, contact time: 2 h, error bar represents 5% error value).



Figure S4. Van't Hoff plot showing relationship between 1/T and ln(K_d)



Figure S5. Influence of solution pH on the AO-II adsorption capacity of the PLDH and the LDH/PEG-15, and on the equilibrium pH of the solution (Initial AO-II concentration: 200 mg/l, contact time: 2 h, temp: 30 °C, adsorbent dose: 0.5 g/l).



Figure S6. Particle size distribution of the Pristine LDH (PLDH).



Figure S7. Particle size distribution of the LDH/PEG-15.



Figure S8. Particle size distribution of the LDH/PEG-50.



Figure S9. Influence of LDH/PEG-15 dose on the AO-II adsorption in aqueous solution.