



Supplementary Material for

Particulate methane monooxygenase contains only mononuclear copper centers

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Materials and Methods

Production of $^{63}\text{CuSO}_4$

5 50 mg aliquots of ^{63}CuO (Cambridge Isotope Laboratories) were dissolved in 1.5 mL of 3 M trace metal grade H_2SO_4 while heating at 60-80°C and shaking at 300 rpm. The reaction was judged to be complete when no black CuO particles were visible in the blue CuSO_4 solution (after several days).

Growth of *M. capsulatus* (Bath) and preparation of Vivo-pMMO EPR sample

10 Isotopically enriched ^{63}Cu , ^{15}N -*M. capsulatus* (Bath) was grown in a 12 L bioreactor as described previously with minor modifications (13, 23). Briefly, cells were grown in a solution of 3.9 mM phosphate buffer, pH 6.8, 40 μM NaFe(III)EDTA , trace element solution, 50 μM $^{63}\text{CuSO}_4$ and 1X ^{15}N -nitrate mineral salts (NMS) (Cambridge Isotopes Laboratories, potassium ^{15}N -nitrate) with a constant flow of 1:4 methane-to-air gas mix. NaOH and H_2SO_4 were used to maintain the growth medium pH between 6.7 and 7.4. Cells were harvested by centrifugation (6,000 x g for 20 min at 4°C) at OD_{600} of 11.3. Of this yield, 0.75 g of cell pellet was set aside on ice, while the rest was frozen in liquid nitrogen and stored at -80°C. To prepare whole cell EPR samples, the 0.75 g of cell pellet was resuspended in 50 mL of 12.2 mM dibasic sodium phosphate, 7.8 mM monobasic sodium phosphate, 5 mM magnesium chloride, pH 7.0 and centrifuged at 4°C, 10,000 x g, for 10 min, as done previously (49). The supernatant was poured off, the wash and centrifugation procedure was repeated, and the supernatant was again poured off. The cells were resuspended in 200 μL of phosphate buffer, and this suspension was then aliquoted into a Wilmad quartz X-band EPR tube (Sigma Aldrich) and custom quartz Q-band EPR tube (~180 μL and ~80 μL , respectively, as with all EPR samples in this study). EPR samples were frozen in liquid nitrogen, where they were stored until analysis. Natural abundance *M. capsulatus* (Bath) was grown and harvested under similar conditions except natural abundance CuSO_4 and NMS were used for the growth medium instead of $^{63}\text{CuSO}_4$ and ^{15}N -NMS.

pMMO purification and X-band Continuous Wave (CW) and Q-band EPR/ENDOR sample preparation

30 *M. capsulatus* (Bath) cells were resuspended in 25 mM PIPES, 250 mM NaCl , pH 7.2-7.3 and lysed by sonication (either using 90% power setting, 1 s on, 1 s off sonication for 10 min, or a 90% power setting, 1 s on, 2 s off sonication for 8 min) and membranes containing pMMO were isolated, solubilized, and purified by size exclusion chromatography (using a 120 mL Superdex 200 equilibrated in 25 mM PIPES, 250 mM NaCl , 0.02 % DDM, pH 7.2-7.3, 1 mL/min flow rate) as described previously (13, 50), although for this study the solubilized membranes were concentrated to 2 mL (rather than 1 mL) before running size exclusion chromatography and the PIPES buffer used throughout (i.e. with and without DDM) was kept at pH 7.2-7.3 (rather than pH 7.0). After size exclusion chromatography, the sample was concentrated via centrifugation at 5,000 x g, at 4°C using a 100 kDa cutoff Amicon spin concentrator; samples were subsequently either frozen in liquid nitrogen and stored at -80°C until use, or, to make Purified-pMMO EPR samples, aliquoted into X- or Q-band EPR tubes and frozen in liquid nitrogen. To generate Reduced/Purified-pMMO, Purified-pMMO was degassed and made anaerobic on a Schlenk line, and then brought into a Coy anaerobic glovebox. 10-12 protein equivalents of sodium ascorbate in anaerobic 25 mM PIPES, pH 7.5 or 25 mM PIPES, 250 mM NaCl , 0.02 % DDM, pH 7.5 were added to the protein solution. Reduced/Purified-pMMO samples were allowed to equilibrate for at least 1 hr at room temperature in the glovebox before being aliquoted into X- or Q-band EPR tubes

which were frozen in liquid nitrogen in the glovebox. While the Cu_C(II) EPR signal was always significantly more diminished by ascorbate reduction than the Cu_B(II) signal, occasionally some Cu_C(II) was apparent even after this procedure. When this occurred, the sample was thawed anaerobically, and ascorbate was added until the Cu_C(II) signal was no longer evident. For Purified-pMMO in D₂O samples, the protein was buffer exchanged into 25 mM PIPES, 250 mM NaCl, 0.02% DDM buffer in D₂O (99.9%-D, Sigma-Aldrich), pH 7.3. To prepare Reduced/Purified-pMMO in D₂O, Purified-pMMO in D₂O was anaerobically reduced as above except the ascorbate was prepared in the deuterated buffer mentioned above (made anaerobic). Unless otherwise noted, the concentrations of all EPR/ENDOR samples of Purified- or Reduced/Purified-pMMO were comparable, only varying between 300 and 500 μM.

For Purified-pMMO incubated with nitrite, NaNO₂ was dissolved in 25 mM PIPES, 250 mM NaCl, 0.02% DDM, pH 7.2 to generate a 4 M NaNO₂ stock solution. An equal volume of this stock and natural abundance Purified-pMMO were mixed, and the sample was incubated at room temperature for 2 hr and 15 min. The sample was then transferred to either an X-band or Q-band EPR tube and flash frozen in liquid nitrogen.

For Purified- and Reduced/Purified-pMMO samples incubated with H₂¹⁷O, the enzyme was desalted into 2X wash buffer (50 mM PIPES, 500 mM NaCl, 0.04% DDM, pH 7.2) using a PD-10 desalting column at 4°C. The protein was concentrated to 90 μL, ~630 μM protein using a 100 kDa cutoff Amicon filter. The protein was then diluted with 90 μL of 90% ¹⁷O-enriched H₂O (Sigma Aldrich). To generate the Purified-pMMO in H₂¹⁷O, 80 μL of this sample was put in a Q-band EPR tube inside of a cryotube, parafilm to make airtight, and then left in the dark at room temperature overnight. To generate the Reduced/Purified-pMMO, with the remaining 100 μL of pMMO, the protein was made anaerobic using a Schlenk line and brought into the glovebox, where it was treated with 12 protein equivalents of ascorbate (in anaerobic 25 mM PIPES, 250 mM NaCl, 0.02% DDM, pH 7.2). The protein was incubated overnight in the dark at room temperature in the glovebox. In the morning this sample was transferred to a Q-band EPR tube. Both Q-band samples were flash frozen in liquid nitrogen in the morning (the anaerobic sample was flash frozen in the glovebox).

Design, expression, and purification of spmoB constructs and NeMCO

spmoB was recombinantly expressed in *E. coli* into inclusion bodies and refolded in 20 mM Tris, 250 mM NaCl, pH 8.0 as described previously (27), except Tris base was used instead of Tris-Cl.

To optimize spmoB solubility and create a solubly expressing spmoB construct, we redesigned the artificial flexible linker that connects PmoB subdomain 1 (PmoB D1) and subdomain 2 (PmoB D2) in spmoB, and tested various combinations of N- and C-terminal soluble tag attachments to spmoB. To reduce flexibility and aggregation propensity, the original linker in spmoB was replaced with the following sequence: PmoB(residues 173-182)-GEPS-Protein GB1-GE-PmoA(residues 170-203)-S, which is followed by the D2 domain starting with PmoB residue 267 (Fig. S9), in which Protein GB1 is the B1 domain of Protein G. Solubility improvements were screened by the split-GFP assay and confirmed by SDS-PAGE (51). Out of the screened variants, constructs Con7 with a Set12 tag (Set12 is the T7B9 peptide (52)) at both C- and N-termini, and SUMO7 (Con7 with an additional N-terminal SUMO tag) were identified to be solubly expressed at ≥10 mg/L in *E. coli* (Fig. S9). In order to test whether the foldedness of the copper-binding PmoB D1 affects the activity, error-prone PCR was performed on the PmoB D1 subdomain alone, and the resulting variants were selected by a twin-arginine translocation (TAT) pathway-based

ampicillin survival assay, which has been shown to isolate variants with improved folding (53, 54). From this assay, a single amino acid mutation variant (PmoB D1_E75D) was identified to grant host expression *E. coli* cells with improved ampicillin survival rate than PmoB D1_WT, and assembled back to the SUMO7 construct to create SUMO7_E75D. To validate that methanol produced in the methane oxidation activity assays was a result of enzymatic turnover at the previously reported pMMO active site (Cu_B), we produced two negative control constructs which should have exhibited no methane oxidation activity. For SUMO7_3A, the three Cu_B residues (H33, H137, and H139) were mutated to alanine to abrogate copper binding at this site. The second negative control was a multicopper oxidase (NeMCO, derived from *Nitrosomonas europaea*)(55) that does not oxidize methane.

SUMO7, SUMO7_E75D, SUMO7_3A, and NeMCO were expressed in chemically-competent *E. coli* BL-21(DE3)Gold cells by either auto induction or IPTG induction. For auto induction, 5 mL of overnight culture in LB with 100 $\mu\text{g}/\text{mL}$ of ampicillin (Amp100) was inoculated into 1 L of ZYM-5052 media with Amp100 at 37°C until OD_{600} hit 0.5, and then at 20°C for 24 hours to express protein (56). For IPTG induction, 1 L of LB/Amp100 was inoculated by 5 mL of overnight culture at 37°C, and induced by 1 mM IPTG at OD_{600} between 0.6 and 1.0. Expression was then conducted at 27°C for 4 hours. Post-expression culture was transferred to a capped bottle, supplemented with water to replace the headspace gas, and treated with 1x ZYM-5052 sugar solution together with 1 mM of CuSO_4 for 2 hours to allow *in-vivo* copper loading (56). Cells were then harvested, lysed by a cell disruptor, and centrifuged at 15,000 x g for 40 min. Soluble protein was then purified by Strep-Tactin beads (IBA) and stored in either 1x PBS (137 mM NaCl, 2.7 mM KCl, 10 mM Na_2HPO_4 , 1.8 mM KH_2PO_4 , pH 7.4) or 1x TBS (50 mM Tris, 150 mM NaCl, pH 7.6). Protein was concentrated by 30kDa MWCO spin-filters (Amicon) to ≤ 1 mL volume before further analysis.

For the refolded SUMO7 sample, soluble protein was expressed in C41(DE3) chemically competent *E. coli* cells in ZYM-5052 autoinduction media with 50-100 $\mu\text{g}/\text{mL}$ ampicillin at 37°C (56). After the OD_{600} reached ~ 0.5 , the temperature was lowered to 18°C for overnight expression. Cells were harvested by centrifugation, frozen in liquid nitrogen and stored at -80°C until use. Cells were then thawed and resuspended in 50 mM Tris, 150 mM NaCl, pH 8.0 and lysed by sonication; cell debris was pelleted by centrifugation, and the protein was purified via Strep-Tactin column (IBA-GMBH). After purification, the protein was unfolded in 8 M urea at a protein concentration of 0.7 mg/mL (protein concentration determined using the Pierce™ 660 nm assay (VWR International) and a molecular weight of 69.292 kDa) and was subsequently refolded in the presence of CuSO_4 following a previously described protocol (24). After refolding, the protein was ultracentrifuged for 40 min at 125,000 x g, 4°C, and concentrated using a 10 kDa cutoff Amicon spin concentrator. Aliquots of 50 μM SUMO7 were then immediately tested for methane oxidation activity, and the rest of the protein was frozen in liquid nitrogen and stored at -80°C until later use. To prepare the refolded SUMO7 X-band EPR sample, protein was thawed, concentrated using an Amicon spin concentrator to 250 μM , transferred to a Wilmad quartz X-band EPR tube and frozen in liquid nitrogen.

Optical spectroscopy

All optical spectroscopy measurements were performed on an Agilent 8453 spectrophotometer with an Agilent 89090A Peltier temperature controller.

Methane oxidation activity assays

All Purified- and Reduced/Purified-pMMO preparations used to produce paramagnetic resonance spectroscopy samples were verified as active (capable of catalyzing methane oxidation) using a very slightly adjusted bicelle-reconstitution ^{13}C methane oxidation activity assay (13). Duroquinol was generated from duroquinone (VWR International) as described previously (25). For all methane oxidation activity assays, 100 μL of 2.6-8.0 mg/mL purified pMMO in 3-6% DMPC:CHAPSO 2.8:1 bicelle solution (Molecular Dimensions or Anatrace), or 100 μL of spmoB/spmoB construct (concentrations and length of reaction provided in figures), were mixed with a small scoop of duroquinol placed into sealed septa top vials (Agilent). A 2:1 ratio of ^{13}C -methane (Sigma-Aldrich) and air, totaling 3 mL, was injected into the top. Reactions were incubated in a water bath shaking at 200 rpm (at least 5 min at 30°C for pMMO, otherwise temperature and time of reaction are provided in figures) after which point the reactions were either immediately stopped using 500 μL of chloroform spiked with 1 mM dichloromethane or were frozen and later thawed; upon thawing 500 μL of chloroform spiked with 1 mM dichloromethane was added to the vial. The amount of $^{12/13}\text{C}$ -methanol after reaction was determined using a PoraBOND Q column (Agilent) on an Agilent 7890B/5977A MSD GC/MS instrument monitoring the 33 m/z ion (^{13}C -methanol) and 31 m/z ion (^{12}C -methanol) relative to the 49 m/z ion (dichloromethane internal standard) as described previously (13). The GC oven temperature holding times were varied to adjust for changing retention times over the life of the GC column. Absolute ^{13}C -methanol concentrations were determined by comparison of ^{13}C -methanol peak intensities to those of known concentration ^{13}C -methanol (Sigma-Aldrich) standards in buffer. A signal-to-noise cutoff of 1.0 for the ^{13}C -methanol integrated peak area was used to distinguish real signal from noise; samples with signal-to-noise less than 1.0 were treated as 0 μM ^{13}C -methanol.

X-band CW-EPR and Q-band ENDOR measurements

All CW X-band EPR measurements (except those of SUMO7) were performed using a Bruker ESP-300 spectrometer with a liquid helium flow Oxford Instruments ESR-900 cryostat. SUMO7 EPR was measured on a modified Varian E-4 with a liquid nitrogen finger-dewar at 77 K. All wide scans spanning ≥ 1000 G were background subtracted. Collection conditions (Fig. 1, wide and inset scans): 9.36-9.37 GHz microwave frequency, 90 s per scan, 320 ms time constant, 12.5 G modulation amplitude, temperature at 20 K, at least 5 scans. SI spectra collection conditions are provided below the figures. All EPR simulations were performed using EasySpin (57).

Pulsed ENDOR measurements were collected at ~ 2 K in a liquid helium immersion dewar on a spectrometer described elsewhere, with SpinCore PulseBlaster ESR_PRO 400MHz digital word generator, and Agilent Technologies Acquiris DP235 500 MS/sec digitizer using SpecMan software (58, 59). For all ^1H and strongly-coupled ^{15}N ENDOR measurements, a Davies [π - T_{RF} - $\pi/2$ - τ - π - τ -echo] pulse sequence was employed, in which T_{RF} denotes the interval during which the RF was applied. For weakly-coupled remote ^{15}N ENDOR measurements, a Doan/ReMims [$\pi/2$ - τ_1 - $\pi/2$ - T_{RF} - $\pi/2$ - τ_2 - π - $(\tau_1 + \tau_2)$ -echo] or a Mims [$\pi/2$ - τ - $\pi/2$ - T_{RF} - $\pi/2$ - τ -echo] pulse sequence was employed (60). Hyperfine coupling (A) of $I = 1/2$ nuclear spin (^{15}N and ^1H) and $S = 1/2$ electron spin (Cu(II)) produce an ENDOR spectrum given by the equation $\nu_{\pm} = |\nu_n \pm \frac{A}{2}|$, where ν_+ and ν_- are the high and low frequency peaks of the doublet, respectively, while ν_n is the Larmor frequency for the nucleus (n). For Fig. 2, collection conditions were as follows: (strongly-coupled ^{15}N) 34.592 GHz microwave frequency, 100 ms repetition time, $\pi = 80$ ns, $\tau = 600$ ns, $T_{\text{RF}} = 15$ μs , and RF tail = 5 μs ; (weakly-coupled ^{15}N) 34.608 GHz microwave frequency, 20 ms repetition time, $\pi = 60$ ns, $\tau_1 = 200$ ns, $\tau_2 = 400$ ns, $T_{\text{RF}} = 60$ μs , and RF tail = 10 μs , ^{15}N -Larmor

corrected. The sign of the ^{15}N hyperfine coupling was determined through the Pulsed ENDOR Saturation Recovery (PESTRE) experiment (Fig. S14), which provides the absolute sign of A/g_n (61).

5 X-band pulsed DEER sample preparation and measurements

75 μL of 360 μM natural abundance Purified-pMMO in 2X wash buffer (50 mM PIPES, 500 mM NaCl, 0.04% DDM, pH 7.2) was aliquoted into a Wilmad quartz X-band EPR tube. The altered buffer conditions do not affect the Cu(II) ligation (Fig. S15). X-band pulsed EPR spectra were recorded on a Bruker E680 spectrometer (Bruker, Rheinstetten, Germany) using a Bruker MD5-W1 EPR probehead equipped with a self-modified cryogen-free cryostat (Advanced Research Systems Inc, Macungie, PA, USA). The microwave pulses were amplified using a 1kW-TWT (Applied Systems Engineering Inc., Fort Worth, TX, USA). All EPR experiments were carried out at 17K. The field-swept spectrum was obtained by integrating the Hahn echo signal as a function of the magnetic field after a two-pulse sequence.

15 For the 4-pulse PELDOR experiments recorded at 17 K, pulse lengths were 20 nsec for $\pi/2$ and 20 nsec for π . The pump pulse length was 16 nsec and $\Delta\nu$ ($\nu_{\text{obs}} - \nu_{\text{pump}}$) was 80 MHz, with the detection field set at a g value of 2.07. The pulse separation τ_1 was 200 nsec while τ_2 varied between 2500-3500 ns, and the echo signals were integrated using a video amplifier bandwidth of 20 MHz. The pump pulse was stepped out by 16 nsec for a total number of points in T that depended upon the τ_2 used. Nuclear modulation artifacts were suppressed using a systematic variation of the interpulse delay time τ_1 and an appropriate phase cycling routine.

Supplementary TextDiscrepancy between the paramagnetic resonance analysis and previous data supporting a dicopper site

5 Previous EXAFS scattering was fit with a short Cu-Cu distance (~ 2.5 - 2.6 Å) indicative of a dicopper center. However, all three possible oxidation states of a dicopper center, Cu(II)-Cu(II), Cu(II)-Cu(I), or Cu(I)-Cu(I), are not consistent with the current analysis. Two Cu(II) ions at ~ 2.5 - 2.6 Å separation, with no magnetic interaction between them, would produce a large dipolar splitting in the EPR spectrum that is not observed. A (anti)ferromagnetically coupled Cu(II)-Cu(II) center would not contribute to the paramagnetic resonance spectroscopy, but we do not observe the characteristic charge transfer optical absorbance features associated with the superexchange coupling pathway requisite to forming such centers (62, 63). To the best of our knowledge, there is no dicopper Cu(I)-Cu(I) metal site in biology that does not oxidize in air, so if a Cu(I)-Cu(I) site were present, it would oxidize to either Cu(II)-Cu(I) or Cu(II)-Cu(II) during the aerobic purification procedure. In addition, in previous studies from our laboratory (13, 15), Cu(II) EPR spin quantitation of purified pMMO indicated that nearly all Cu in the protein is EPR active Cu(II), inconsistent with a dicopper Cu(I)-Cu(I) site. Finally, we consider a Cu(II)-Cu(I) center. As noted previously, we do not observe any of the characteristic features associated with such valence-delocalized center (24). As described in the main text, the 4 N ligands in the Cu_B site all are bound to the single Cu(II) giving rise to the Cu_B EPR signal; the terminal NH_2 and the side chains of His33, His137, and His139 are coordinated in a square plane of equatorial coordination. Thus, there are no ligands available to coordinate a second copper ion, excluding the possible presence of a valence-localized Cu(II)-Cu(I) site. Viewed from a different perspective, the Cu(II) EPR signal from spmoB differs from that of Cu_B in pMMO, yet the EXAFS for both proteins was fit with the same Cu-Cu scattering interaction (within ~ 0.1 Å) (27). It is implausible that the two samples with distinct copper centers coincidentally both contain dicopper centers with the same Cu-Cu distance. The optical absorbance feature proposed to be assignable to a dicopper site in pMMO is discussed in the caption to Fig. S5, and spmoB methane oxidation activity proposed to occur at a dicopper Cu_B site is discussed in the caption to Fig. S11.

Table S1. Summary of the paramagnetic spectroscopic parameters of the Cu_B and Cu_C sites.*

	g₁, g₂, g₃	Cu A₁, A₂, A₃ (MHz)	Directly Coord. ¹⁵N A₁, A₂[§] (MHz)[†]	Remote ¹⁵N A₁ (MHz)		¹H A₁, A₂ (MHz)[‡]
Cu_B	2.242,	570,	-48, -54	1.7	His33-NH ₂	~10, ~10
	2.068,	30,	-53, -54	2.3	H _{ε-1}	- , 4.5
	2.035	14	-53, -54	2.3	H _{ε-2}	- , 2.5
			-53, -66	N/A	H _β	- , ~1
					H _δ	- , ~1
Cu_C	2.30,	440,	-48, -54			
	2.07,	40,	-53, -66	~2	H _x O	~8, ~10
	2.05	20				

* For simulations featuring the full Vivo-, Purified-, and Reduced/Purified-pMMO EPR spectra in Fig. 1 using the above parameters, anisotropic *g*-strain (for both Cu_B and Cu_C, EasySpin *g*-strain parameters 0, 0, 0.03 for *g*₃, *g*₂, *g*₁, respectively) was employed, with additional isotropic 14 G and 25 G fwhm Gaussian line width broadening for Cu_B and Cu_C, respectively. For simulations depicted in insets, the Cu_B *g*₁ value was changed very slightly to 2.252, and isotropic 13 G fwhm Gaussian line width broadening was employed, with no anisotropic broadening. Simulation of Purified-pMMO represents the sum of 1 Cu_B equivalent and 0.32 Cu_C equivalents. For comparison to ¹⁴N ENDOR measurements, see Table S3.

§ Strongly-coupled ¹⁵N A₂ ENDOR measurements are shown in Fig. S16. The four features seen in these spectra are plausibly, but not uniquely assigned as listed.

† PESTRE experiments determined that all directly-coordinated ¹⁵N possess positive spin density $\rho > 0$ (Fig. S14); the ¹⁵N hyperfine couplings are therefore negative because the ¹⁵N nuclear *g_n* < 0.

‡ The ¹H ENDOR assignment to individual C-H/N-H bonds of the histidine imidazole ring and amine are made by analogy to Cu(II)(histidine)₂ complex (34).

Table S2. Vivo-pMMO ¹⁵N A₁ Gaussian fitting parameters used in Fig. 2.*

Resonance		Gaussian Function Parameters			R ²
Remote ¹⁵ N	Red	$k = 30.32$ $a_1 = 0.08824$ $b_1 = 0.8631$			0.997
	Blue [§]	$a_2 = 0.1194$ $b_2 = 1.149$			
Directly-coord. ¹⁵ N	Red	$k = 6.57$ $a_1 = 1.341$ $b_1 = 27.5$			0.9938
	Blue [†]	$a_2 = 1.329$ $b_2 = 32.28$	$a_3 = 1.75$ $b_3 = 31.13$	$a_4 = 0.9132$ $b_4 = 30.77$	

* The colors red or blue define the color of the Gaussian curves shown in Fig. 2, which were used to fit the ¹⁵N ν₊ resonance. R² values correspond to the fit of the full normalized Gaussian function to the experimental spectrum over a range of $\nu - \nu_{^{15}\text{N}} = 0.627\text{-}1.604$ MHz for remote ¹⁵N, and over a range of 25.651-28.043 MHz for directly-coordinated ¹⁵N resonance. The MatLab Curve Fitting tool was used to fit the Gaussian function parameters, using the following equations: normalized Gaussian function,

remote ¹⁵N:
$$f(x) = k \left(\frac{1}{a_1\sqrt{2\pi}} e^{-\frac{(x-b_1)^2}{2(a_1)^2}} + 2 \frac{1}{a_2\sqrt{2\pi}} e^{-\frac{(x-b_2)^2}{2(a_2)^2}} \right),$$

directly-coordinated ¹⁵N:

$$f(x) = k \left(\frac{1}{a_1\sqrt{2\pi}} e^{-\frac{(x-b_1)^2}{2(a_1)^2}} + \frac{1}{a_2\sqrt{2\pi}} e^{-\frac{(x-b_2)^2}{2(a_2)^2}} + \frac{1}{a_3\sqrt{2\pi}} e^{-\frac{(x-b_3)^2}{2(a_3)^2}} + \frac{1}{a_4\sqrt{2\pi}} e^{-\frac{(x-b_4)^2}{2(a_4)^2}} \right),$$

where k is a factor that scales calculated and experimental spectra, a_n is related to the width of the peak (in MHz), and b_n defines the center of the Gaussian peak (in MHz).

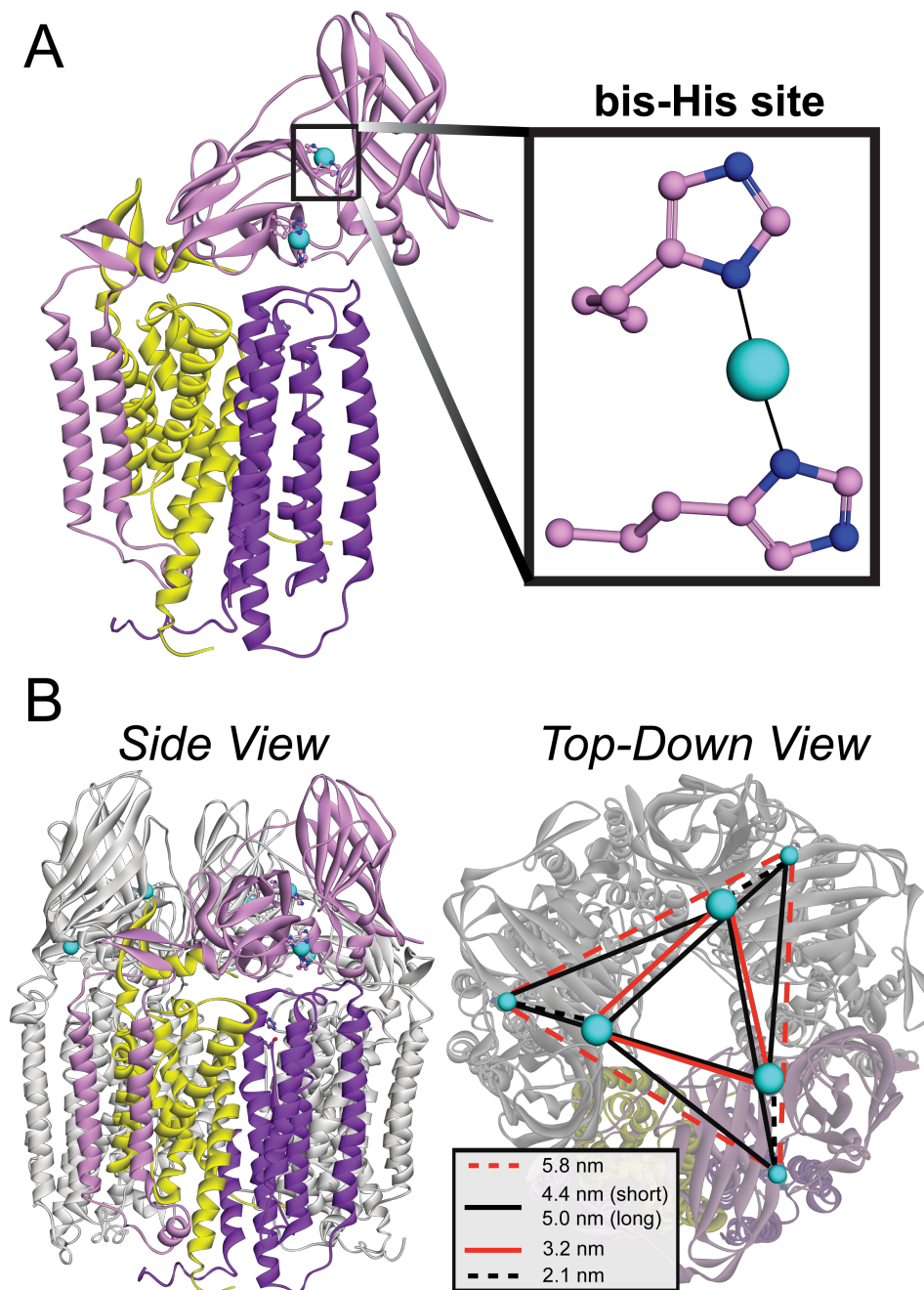
[§] In Fig. 2 the blue peak intensity is twice that produced by simply using the parameters to account for the fact that the function used for total fitting also doubles the intensity of this peak relative to the red peak

[†] Blue, directly-coordinated ¹⁵N resonance function is defined by the sum of three equal area Gaussian functions

Table S3. Conversion of ^{15}N hyperfine coupling to equivalent ^{14}N hyperfine coupling for comparison with past characterization of Purified pMMO.

	$^{15}\text{N } A_1 \text{ (MHz)}$		$^{14}\text{N } A_1 \text{ (MHz)}$	Source
^{15}N-labelled	$A_1\text{-}^{15}\text{N}_1$	48	$[34]^\dagger$	This study
	$A_1\text{-}^{15}\text{N}_2$	53	$[38]^\dagger$	
Natural abundance	N/A		35.5	(24)

5 † Values in brackets correspond to hyperfine coupling predicted based on conversion between the
 two nuclear gyromagnetic ratios: $A_{15\text{N}} = 1.403(A_{14\text{N}})$, (64). The two distinguishable ^{15}N
 hyperfine couplings observed with ^{15}N -labelled Purified-pMMO are not resolved from one another
 in the natural abundance Purified-pMMO ^{14}N ENDOR spectrum from (24), presumably due to the
 10 substantial quadrupolar broadening of the ^{14}N ENDOR response.



5 **Fig. S1.** (A) One protomer of the *M. capsulatus* (Bath) pMMO modeled with an unoccupied PmoC metal site and a Cu-bound bis-His site. ***Inset:*** A closer view of the bis-His site. (B) Full pMMO trimer with Cu_B to bis-His Cu distances (***inset***).

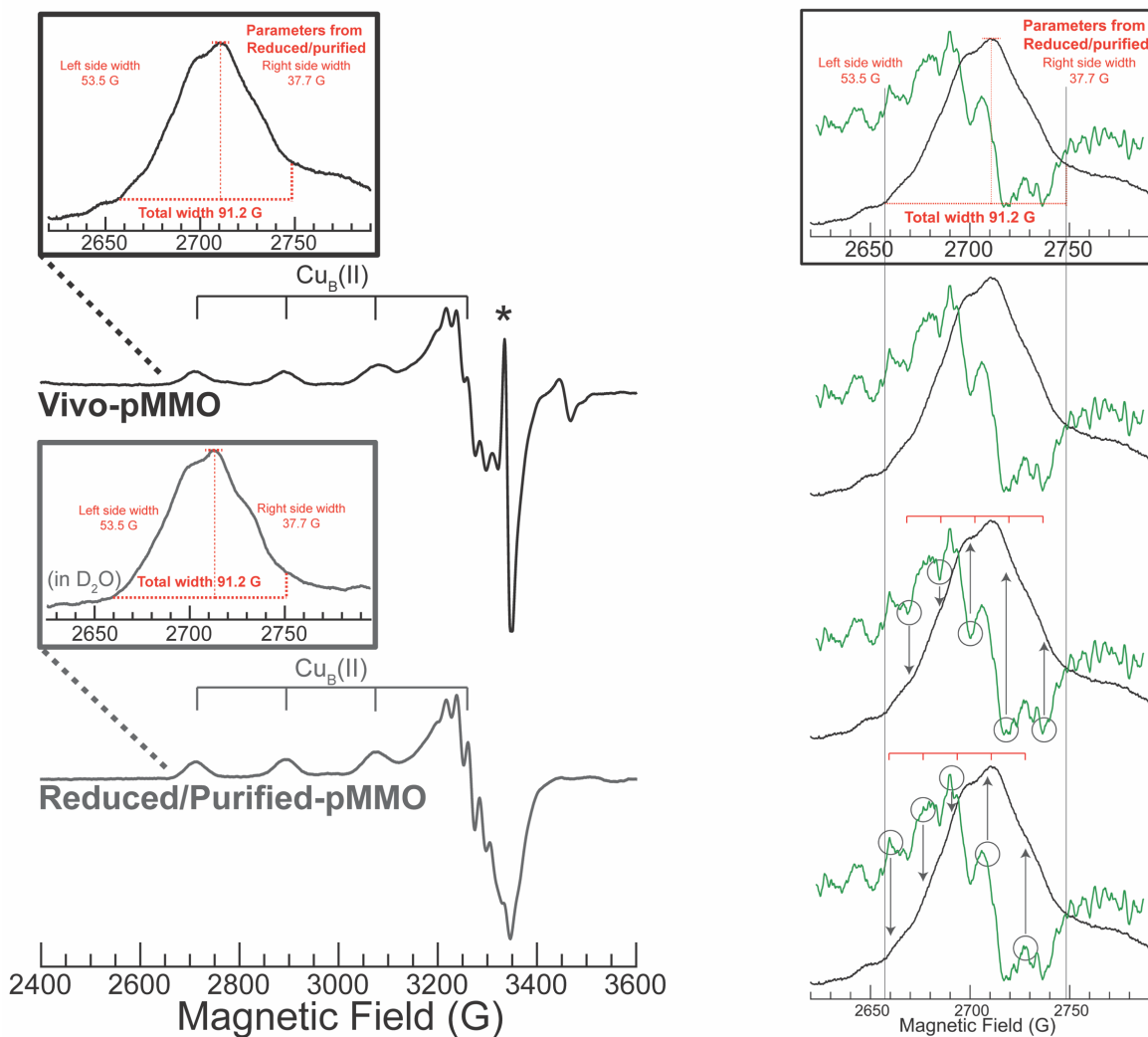


Fig. S2. Support for the assignment of 4 directly coordinated N equatorial ligands on Cu_B from the EPR shown in Fig. 1. We first distinguished the region corresponding to $\text{Cu}_B(\text{II})$ resonance from resonance attributable to the contaminant paramagnet(s) clearly present at the high and low field edge of the Vivo-pMMO single Cu(II) hyperfine line in this region. Toward this end, we measured the total width of the low field $\text{Cu}_B(\text{II})$ hyperfine line in Reduced/Purified-pMMO (left, bottom), since this spectrum does not contain these contaminants. Specifically, we measured the width of the total resonance to the left and to the right of the hyperfine transition maximum height peak. We then used these bounds to define the $\text{Cu}_B(\text{II})$ resonance in Vivo-pMMO (left, top and right). There are 5 evenly spaced local minima in the second derivative (green) of the experimental spectrum, as well as the 5 corresponding evenly spaced local maxima, as shown by the circles in the figure below, indicative of 5 hyperfine lines. Additionally, these peak and valley features clearly correspond to the expected splittings on the absorption-like experimental spectrum. 5 hyperfine lines can only be accounted for by 4 approximately equivalent $I = 1/2$ nuclei, such as that expected from 4 ^{15}N equatorial ligands.

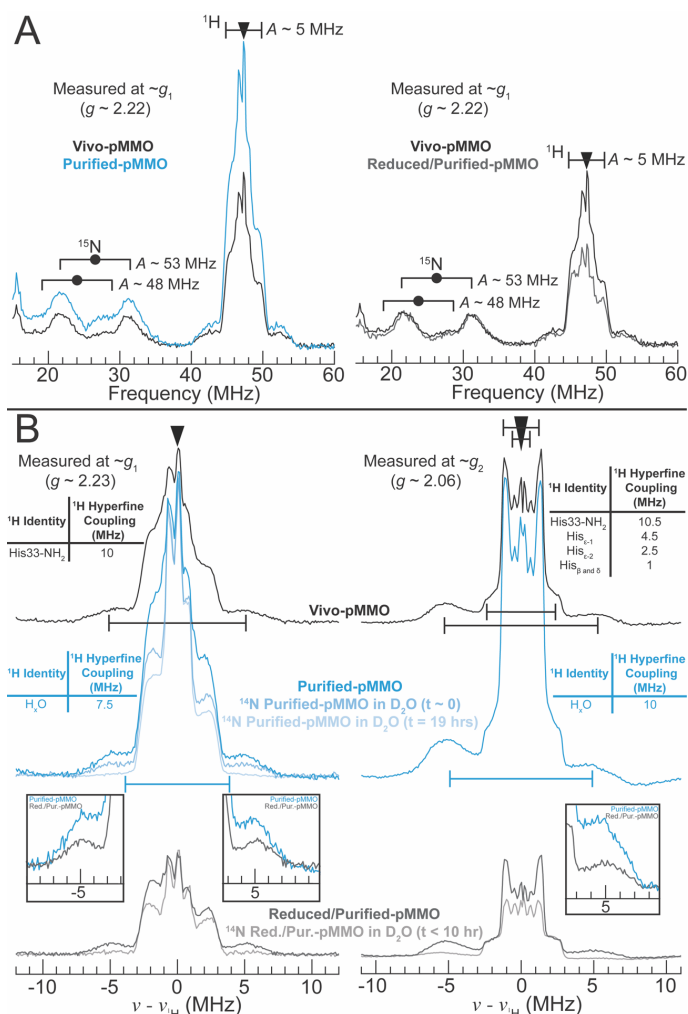
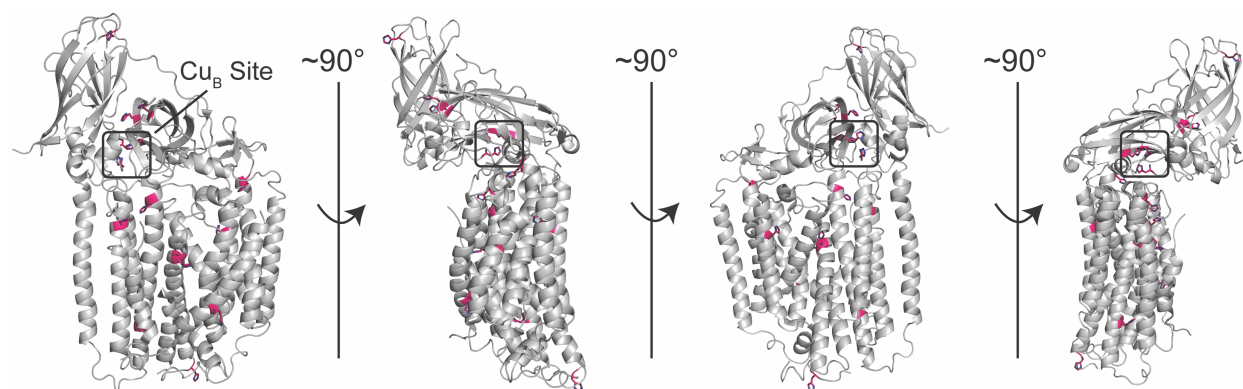


Fig. S3. ^1H and ^{15}N Davies ENDOR for Vivo-, Purified-, and Reduced/Purified-pMMO. Times incubated in D₂O buffer are listed. **(A)** ENDOR measurements simultaneously probing ^{15}N and ^1H , with heights normalized to echo intensity. Filled circle defines half the ^{15}N hyperfine coupling magnitude ($|A|/2$) and associated goalpost width indicates twice the ^{15}N Larmor frequency ($\nu_{^{15}\text{N}}$); triangle defines the ^1H Larmor frequency ($\nu_{^1\text{H}}$) and associated goalpost width defines hyperfine coupling magnitude ($|A|$). The weakly-coupled ^1H resonance ($A < \sim 5$ MHz) in Vivo-pMMO that is less intense in Reduced/Purified-pMMO is attributed to solvent H_xO near the Cu_B site in vivo that is not present in purified enzyme. Collection conditions: 34.5-34.6 GHz microwave frequency, 100 ms repetition time, $\pi = 200$ ns, $\tau = 600$ ns, $T_{\text{RF}} = 60$ μs , and RF tail = 10 μs . **(B)** ^1H Larmor-centered ^1H ENDOR measured at (*left*) g_1 and (*right*) g_2 , where triangle defines the ^1H Larmor frequency ($\nu_{^1\text{H}}$) and goalpost width defines the hyperfine coupling magnitude ($|A|$). Vivo-, Purified-, and Reduced/Purified-pMMO ENDOR intensities were normalized to ENDOR intensities in **(A)**. *Insets* highlight the increased intensity and altered lineshape of the Purified-pMMO large coupling ^1H resonance relative to Reduced/Purified-pMMO, evidence for a coordinated H_xO (with hyperfine coupling, $|A_1| \sim 8$ MHz, $|A_2| \sim 10$ MHz) (65) on Cu_C. Collection conditions: 34.5-34.8 GHz microwave frequency, 100 ms repetition time, $\pi = 200$ ns, $\tau = 600$ ns, $T_{\text{RF}} = 60$ μs , and RF tail = 10 μs .



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Fig. S4. One protomer of *M. capsulatus* (Bath) pMMO from the crystal structure in Fig. 1, with all protomers colored gray, shown from several angles, with His residues highlighted in pink. The Cu_B site is the only site where three His side chains come together.

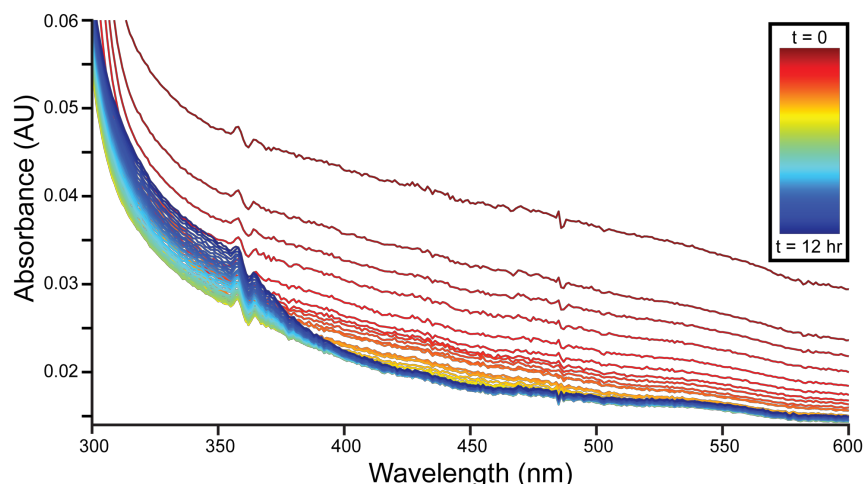


Fig. S5. UV-Vis spectra monitoring formation of the 345 nm optical feature. A sample of 25 mM PIPES, 250 mM NaCl, 0.02% DDM, pH 7.2 (*M. capsulatus* (Bath) Purified-pMMO wash buffer) with 120 μ M CuSO₄ (to represent the Cu equivalents present in the pMMO sample used previously (37)) and 4 μ M methanol (to represent background-level contamination present to varying degrees in most pMMO preparations) was made anaerobic in a sealed cuvette and incubated with 5 Cu(II) equivalents of ascorbate for 30 min. 2.5 Cu(II) equivalents of H₂O₂ were then added using a gas-tight syringe, and optical spectra were measured every 15 min for 12 hr at room temperature. Formation of the 345 nm feature is observed on the timescale defined previously (37) and can be attributed to ketal derivatives of dehydroascorbic acid generated by the reaction of ascorbate and H₂O₂ (66). The drifting baseline in early timepoints is an instrument artifact. The variable intensity of the 345 nm feature previously observed upon mutation of the Cu_B ligands is likely due to the strong dependence of dehydroascorbic acid ketal derivatives on trace metals for stability, since mutation of the copper binding ligands leads to lowered levels of copper in the sample. Consistent with this conclusion, close inspection of the spmoB_H137,139A (Cu_B ligating His residues mutated) optical spectrum reveals the presence of a small amount of 345 nm absorbance despite mutation of two of the three Cu_B His ligands (37). Conversely, the spmoB_H48N_137,139A (bis-His and Cu_B residues mutated) optical spectrum shows no evidence of 345 nm absorbance. The only difference between spmoB_H137,139A and spmoB_H48N_137,139A is mutation of the nonconserved bis-His residue His48. This mutation decreases the total amount of Cu bound by the protein by approximately 0.5 Cu equivalents (24). The reason this was not noted previously was likely the low intensity of the 345 nm feature in the spmoB_H137,139A optical spectrum.

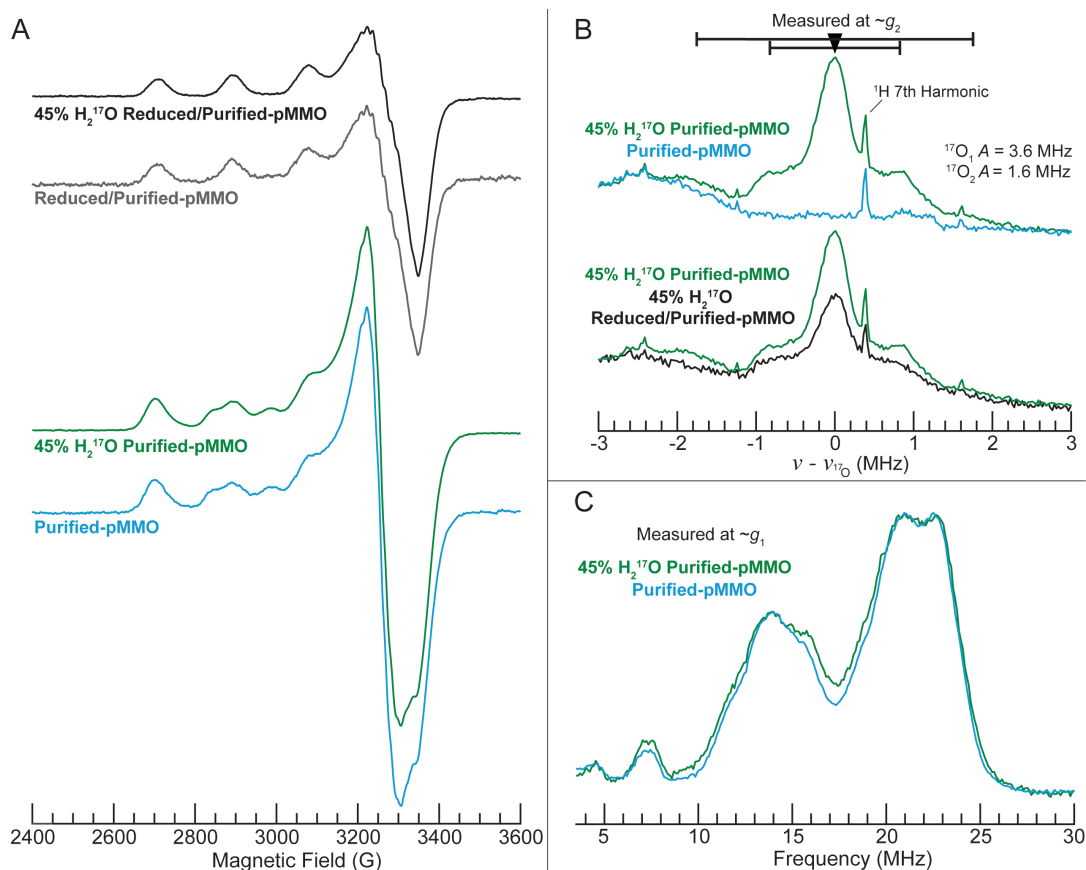


Fig. S6. EPR and ENDOR characterization of (all samples were natural abundance N and Cu) Purified- and Reduced/Purified-pMMO with H₂¹⁷O. (A) X-band EPR measurements assessing Purified- and Reduced/Purified-pMMO, both in 45% H₂¹⁷O compared to natural abundance H₂O (~0% ¹⁷O). If Cu_B(II) featured an equatorial H_xO ligand, both H₂¹⁷O Purified- and H₂¹⁷O Reduced/Purified-pMMO would exhibit broadening of the ¹⁴N hyperfine lines relative to the natural abundance H₂O samples, due to the large hyperfine coupling to the ¹⁷O (approximately equivalent to that of an equatorial N ligand) (67). However, neither sample exhibits such broadening. **Conditions:** 9.36-9.37 GHz microwave frequency, 90 s scan rate, 320 ms time constant, 12.5 G modulation amplitude, temperature 20 K. Protein concentrations of natural abundance Purified- and Reduced/Purified-pMMO were 167 μ M. (B) Mims ¹⁷O-Larmor centered ENDOR testing for weakly-coupled ¹⁷O on Cu_B and Cu_C. The ¹⁷O hyperfine coupling $A = 1.6$ MHz observed in both H₂¹⁷O Reduced/Purified-pMMO and Purified-pMMO is of the magnitude expected for an axial H_xO ligand (68). Thus, there exists an axially coordinated H₂O on Cu_B. The larger coupling ¹⁷O $A = 3.6$ is attributed to a H_xO coordinated to Cu_C. **Conditions:** 34.6-34.7 GHz microwave frequency, 20 ms repetition time (H₂¹⁷O Purified-pMMO and natural abundance Purified-pMMO), 100 ms repetition time (H₂¹⁷O Reduced/Purified-pMMO), $\pi = 100$ ns, $T_{RF} = 60$ μ s, $\tau = 1400$ ns and RF tail = 10 μ s. (C) The equivalence of the Davies ENDOR responses between natural abundance and H₂¹⁷O Purified-pMMO samples in this region indicates that Cu_B does not feature an equatorial H_x¹⁷O ligand whose coupling would be ~30-33 MHz (¹⁷O Larmor frequency $\nu_{17O} = 6.4$ MHz) as shown previously (67). **Conditions:** 34.5-34.7 GHz microwave frequency, 20 ms repetition time, $\pi = 80$ ns, $T_{RF} = 15$ μ s, $\tau = 475$ ns, RF tail = 5 μ s.

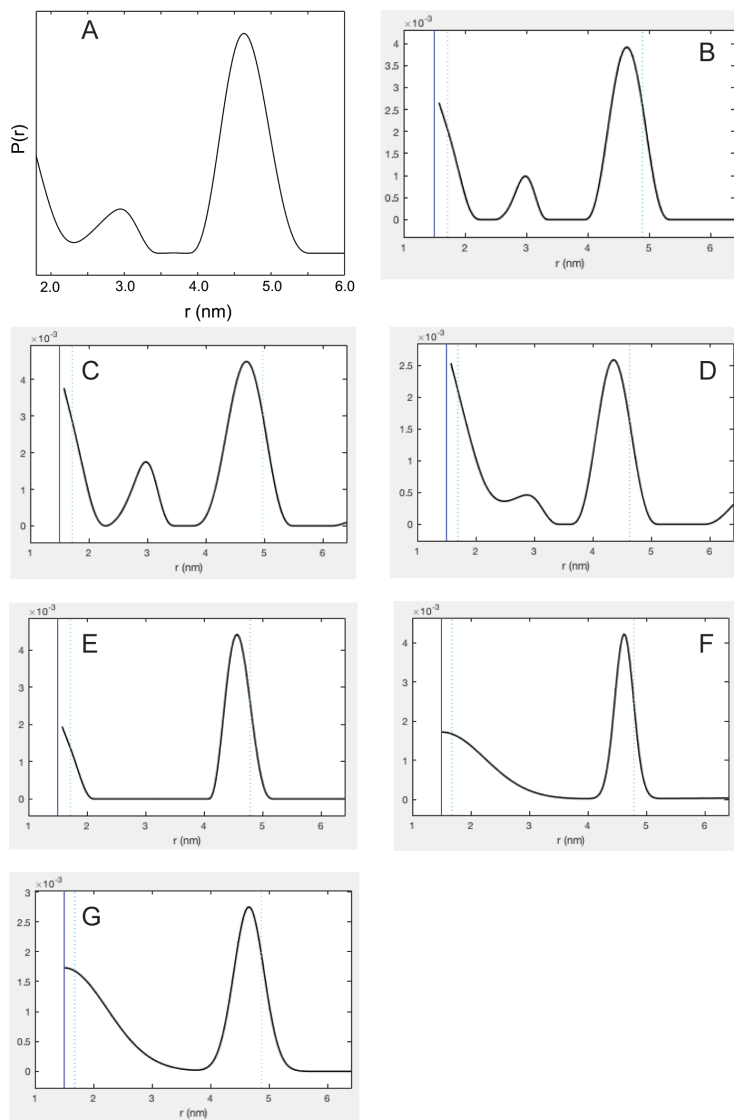


Fig. S7. Alternative analyses of DEER distance distributions showing that the feature at 4.5 nm is robust to analysis procedure, whereas the 2.8 nm feature is highly variable. The distribution shown in (A) is the same as that presented in Fig. 3. *Analysis procedures:* (A) manually optimized phase correction, zero time, using a 3rd order homogenous background function, followed by Tikhonov regularization. (B) Automatic phase correction, zero time, using a 2nd order homogenous background function, followed by Tikhonov regularization. (C) Automatic phase correction, zero time, using a 1st order polynomial background function, followed by Tikhonov regularization. (D) Automatic phase correction, zero time, using a 2nd order polynomial background function, followed by Tikhonov regularization. (E) Manually optimized phase correction, zero time, using a 2nd order homogenous background function, followed by Tikhonov regularization. (F) Automatic phase correction, zero time, using a 1st order polynomial background function, followed by 2 Gaussian fit function. (G) Automatic phase correction, zero time, using a 1st order polynomial background function, followed by 2 Gaussian homogeneous fit function.

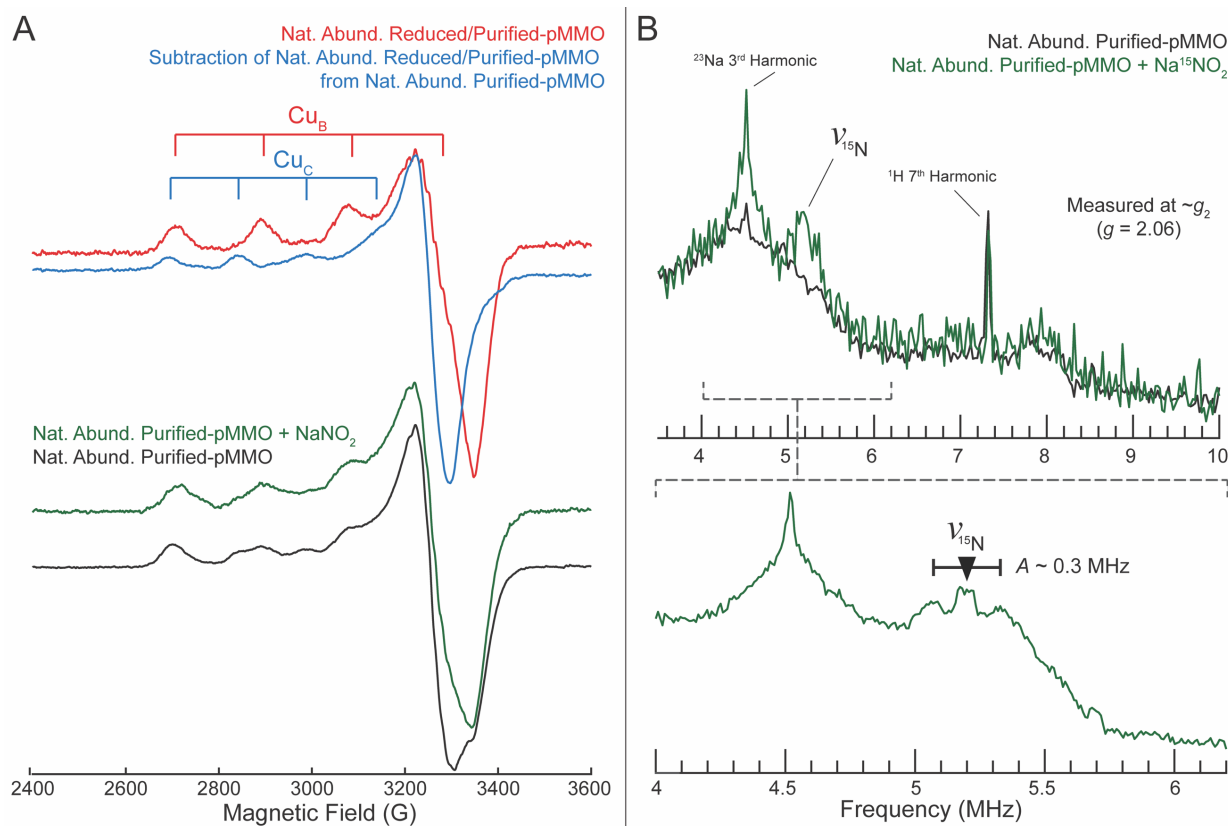
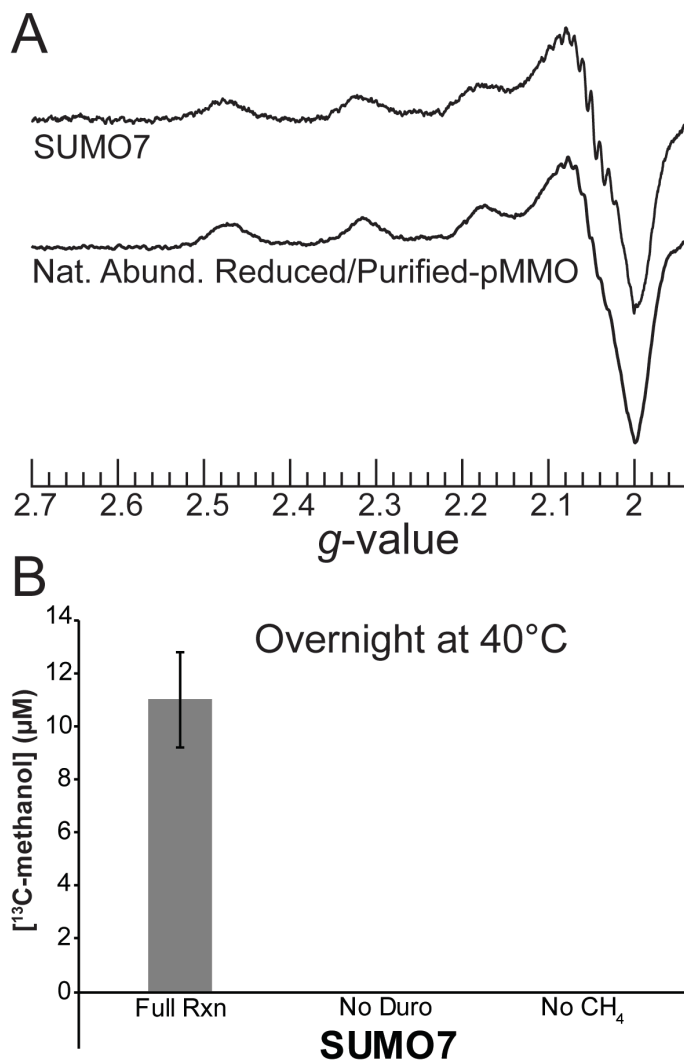
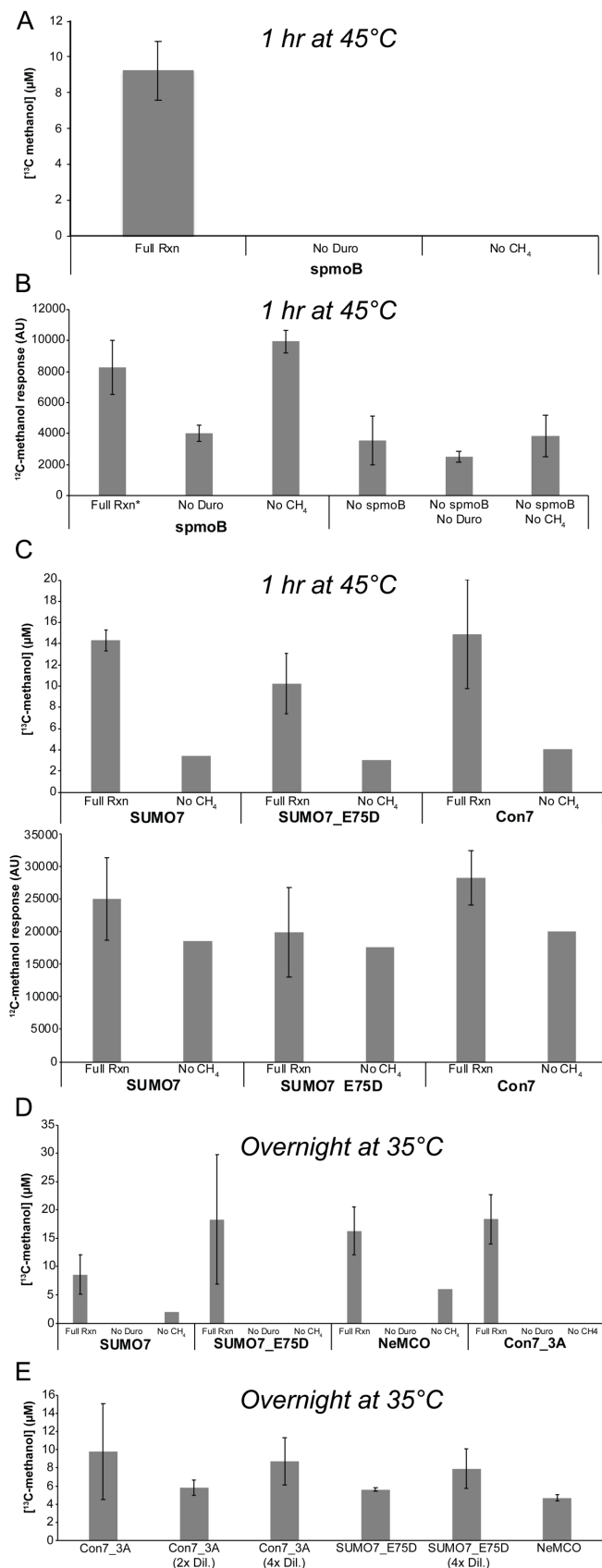


Fig. S8. Natural abundance Purified-pMMO before and after incubation with 2 M NaNO₂. At this concentration of NaNO₂, pMMO in bicelles has <25% methane oxidation activity relative to samples without NaNO₂. **(A)** X-band CW-EPR showing nitrite-induced changes to the Cu_C(II) EPR signal, which are most evident in the g_{\perp} region (~ 3300 G), a collapse of the otherwise resolved features. **Conditions:** 9.365 GHz microwave frequency, 90 s scan rate, 320 ms time constant, 12.5 G modulation amplitude, temperature 20 K. Purified- and Reduced/Purified-pMMO protein concentrations were 167 μ M, while the Purified-pMMO sample after nitrite incubation was 83 μ M. **(B)** Q-band Mims ENDOR comparing natural abundance Purified-pMMO to natural abundance Purified-pMMO incubated with 2 M Na¹⁵NO₂. The bottom expanded view shows the ¹⁵N Larmor frequency ($\nu_{^{15}\text{N}}$) region at higher signal/noise ratio, with the resolved hyperfine coupling of the magnitude expected for ¹⁵NO₂⁻ coordinated to the Cu(II) via oxygen(s) highlighted ($A \sim 0.3$ MHz) (42). Triangle defines the $\nu_{^{15}\text{N}}$ and associated goalpost width defines hyperfine coupling magnitude ($|A|$). **Conditions:** (top) 34.7 GHz microwave frequency, 20 ms repetition time, $\pi = 100$ ns, $T_{\text{RF}} = 60$ μ s, $\tau = 1400$ ns, RF tail = 10 μ s, Purified-pMMO scan time = 17 min, Purified-pMMO + Na¹⁵NO₂ scan time = 40 min; (bottom) 34.7 GHz microwave frequency, 10 ms repetition time, $\pi = 100$ ns, $T_{\text{RF}} = 60$ μ s, $\tau = 1000$ ns, RF tail = 10 μ s, scan time = 3.5 hrs.



5 **Fig. S10.** Refolded SUMO7 and natural abundance Reduced/Purified-pMMO X-band EPR
 10 spectra, and SUMO7 ¹³C-methane oxidation activity. **(A)** The inactive SUMO7 exhibits a Cu_B(II)
 EPR signal similar to Reduced/Purified-pMMO although the Cu_B(II) site formed in SUMO7
 cannot be identical to that of Reduced/Purified-pMMO since the N-terminal residue of SUMO7 is
 not His33 (and thus His33 cannot coordinate via both side chain and amine). **Conditions:** The
 SUMO7 and Reduced/Purified-pMMO concentrations are 250 and 167 µM, respectively. Spectra
 15 normalized to unity. SUMO7, 9.125 GHz microwave frequency, 1 min scan rate, 30 ms time
 constant, 4 G modulation amplitude, 77 K temperature; natural abundance Reduced/Purified-
 pMMO, 9.367 GHz microwave frequency, 90 s scan rate, 320 ms time constant, 12.5 G modulation
 amplitude, 20 K temperature. **(B)** ¹³C-methane oxidation activity assay of the same batch of
 refolded SUMO7 as probed by EPR in **(A)**. The SUMO7 protein concentration was 50 µM.
 SUMO7 does not exhibit methane oxidation activity above that observed for control reactions
 which do not feature Cu_B (e.g. NeMCO and Con7_3A in Fig. S11). Error bars indicate n = 3
 replicates of reaction assay samples.



* For 1 of the 4 "(+) spmoB, Full Rxn" replicates in (B), the entire MS-detected elution profile off the GC column was anomalous, so the replicate was omitted.

Fig. S11. Results of ^{13}C -methane oxidation activity assays performed on spmoB, its variants, as well as NeMCO. Representative assays in which appreciable ^{13}C -methanol was observed are shown. There were other assays run under the same conditions in which no samples showed any ^{13}C -methanol. Error bars indicate $n \geq 2$ replicates of reaction assay samples. (A) Standard spmoB reaction time/temperature (24) generates ^{13}C -methanol from ^{13}C -methane, although the spmoB protein concentration ($\sim 30 \mu\text{M}$) was lower than traditionally assayed for activity ($\sim 100 \mu\text{M}$). (B) The ^{12}C -methanol absolute MS responses are consistently higher in samples containing both spmoB and duroquinol than in conditions with either individually or neither. While the ^{12}C -methanol levels are higher on average in the "(+) protein, No CH_4 " than in the "(+) protein, Full Rxn" for this assay, this trend was often reversed, as shown in (C). In these such situations, where ^{12}C -methanol was produced over the course of the assay and the "(+) protein, Full Rxn" sample accumulated more ^{12}C -methanol than the "(+) protein, No CH_4 " control, the spmoB construct would have appeared active using the GC-FID method of total methanol quantitation, as performed previously (24, 27). The differences in ^{12}C -methanol levels for samples containing both spmoB and duroquinol can be attributed to minor differences in the amount of duroquinol added since standard protocol calls for addition of a small scoop of this insoluble compound. Moreover, ^{12}C -methanol is a common contaminant in many buffer reagents. As ^{13}C represents $\sim 1\%$ of natural abundance carbon, we monitored ^{12}C -methanol levels to ensure that any ^{13}C -methanol observed after reaction was a product of methane oxidation, and not merely observation of the natural abundance ^{13}C -methanol from ^{12}C -methanol contamination. The spmoB protein concentration was $100 \mu\text{M}$. (C) **Top:** ^{13}C -methanol produced via spmoB construct reaction with ^{13}C -methane, with **bottom:** ^{12}C -methanol responses. Con7, SUMO7, and SUMO7_E75D protein concentrations were $22 \mu\text{M}$, $6 \mu\text{M}$, and $17 \mu\text{M}$, respectively. (D) ^{13}C -methane oxidation activity assays run overnight on several spmoB constructs as well as NeMCO. The activity detected for spmoB variant constructs is comparable to that observed for an unrelated enzyme (NeMCO) or a variant in which all three Cu_B His ligands are replaced with Ala (Con7_3A). SUMO7, SUMO7_E75D, NeMCO, and Con7_3A protein concentrations were $114 \mu\text{M}$, $118 \mu\text{M}$, $60 \mu\text{M}$, and $131 \mu\text{M}$, respectively. (E) ^{13}C -methane oxidation activity of diluted samples. There is no dependence on protein concentration. Initial Con7_3A, SUMO7_E75D, and NeMCO protein concentrations (prior to dilution) are same as in (D).

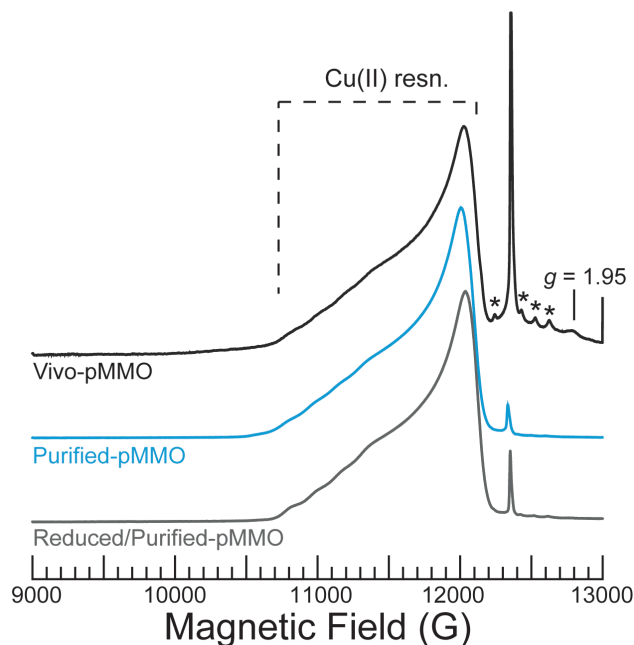
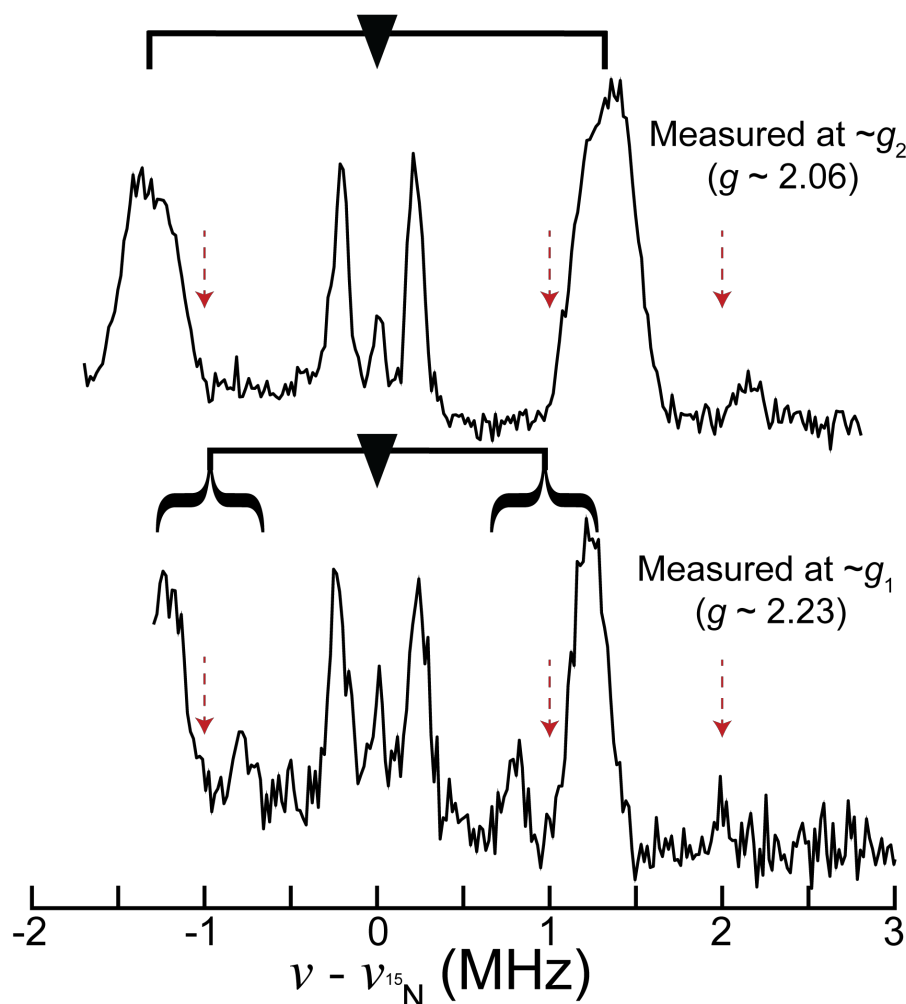


Fig. S12. Rapid-passage Q-band absorption-display CW EPR spectra of the 3 EPR samples shown in Fig. 1 and 2. The Cu(II) resonance is by far the dominant paramagnetic species. We note the extremely weak $g \sim 1.95$ feature in Vivo-pMMO, the intensity of which is exaggerated in the derivative-display of the X-band spectrum in Fig. 1 (~ 3470 G); this feature contributes negligible integrated intensity. Asterisks denote the hyperfine lines for the small amount of Mn(II) ($I = 5/2$). **Conditions:** (Purified- and Reduced/Purified-pMMO) ~ 34.8 GHz microwave frequency, 2 min and 45 second scan rate, 64 ms time constant, 1 G modulation amplitude, temperature ~ 2 K. (Vivo-pMMO) is the same as the others except scan rate was twice as fast.



5 **Fig. S13.** Vivo-pMMO weakly-coupled ^{15}N Mims ENDOR showing both ν_+ and ν_- resonance, measured at g_1 and g_2 . Triangle defines $\nu_{^{15}\text{N}}$ and goalpost splittings define ν_+ and ν_- partners; red arrows denote Mims "holes". For the Mims pulse sequence, the ENDOR response (R) is defined by the equation $R \sim 1 - \cos(2\pi A\tau)$. Consequently, for all integer values of $A\tau$ the ENDOR response/intensity is 0, leading to the so-called Mims "holes". For both ^{15}N -Larmor corrected spectra above, τ was chosen such that Mims holes appear every $|A| = Z(2 \text{ MHz})$, where Z is an integer. The g_1 ^{15}N ν_+ and ν_- responses both show such Mims holes (at $\nu_{^{15}\text{N}} \pm 1 \text{ MHz}$), confirming their assignment to weakly-coupled ^{15}N resonance. We note these must be Mims holes, as Fig. 2 indicates that for appropriate values of τ , ^{15}N resonance is non-zero at this frequency. The microwave amplifier employed for these measurements cannot measure frequencies low enough to observe the full ν_- peak at g_1 . Collection conditions: 34.64 GHz microwave frequency, 50 ms repetition time, $\pi = 100 \text{ ns}$, $\tau = 500 \text{ ns}$, $T_{\text{RF}} = 60 \mu\text{s}$, and RF tail = 10 μs .

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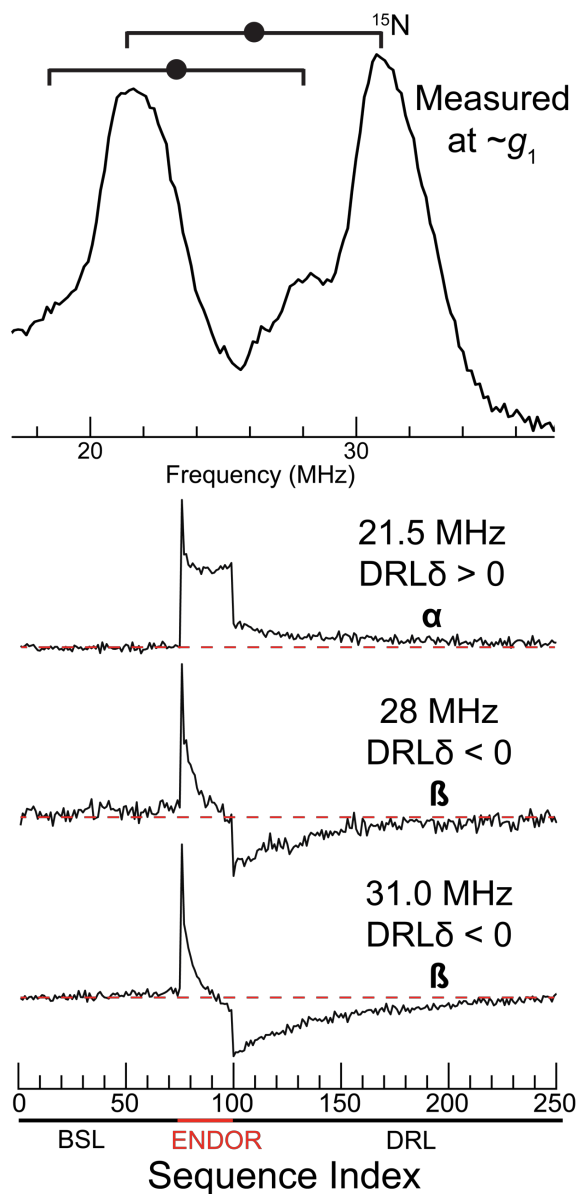
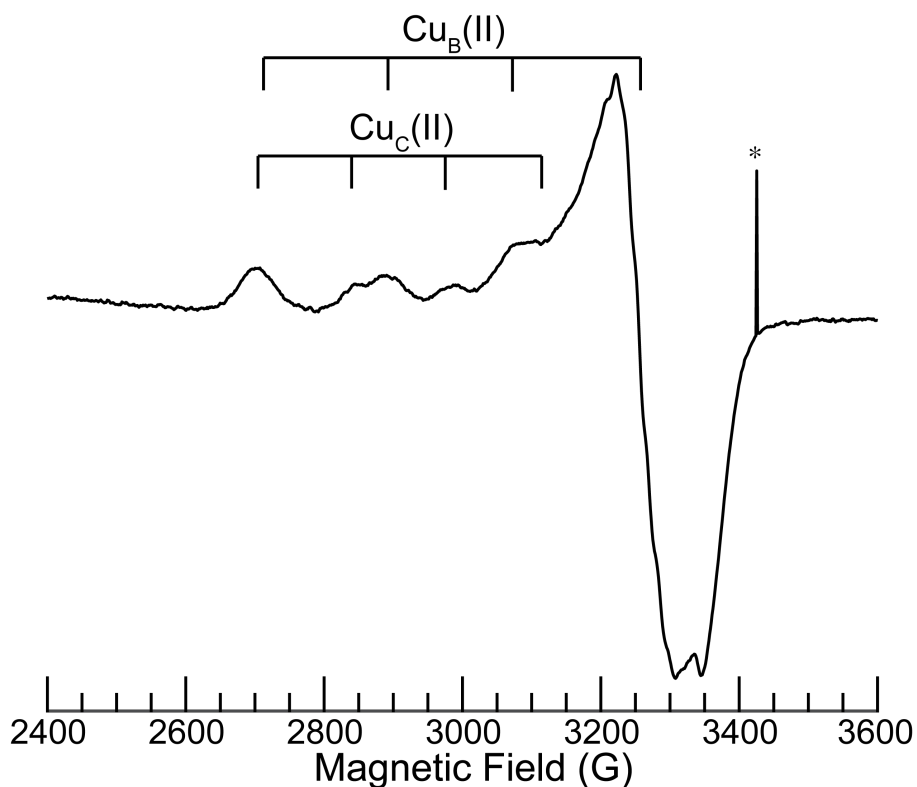
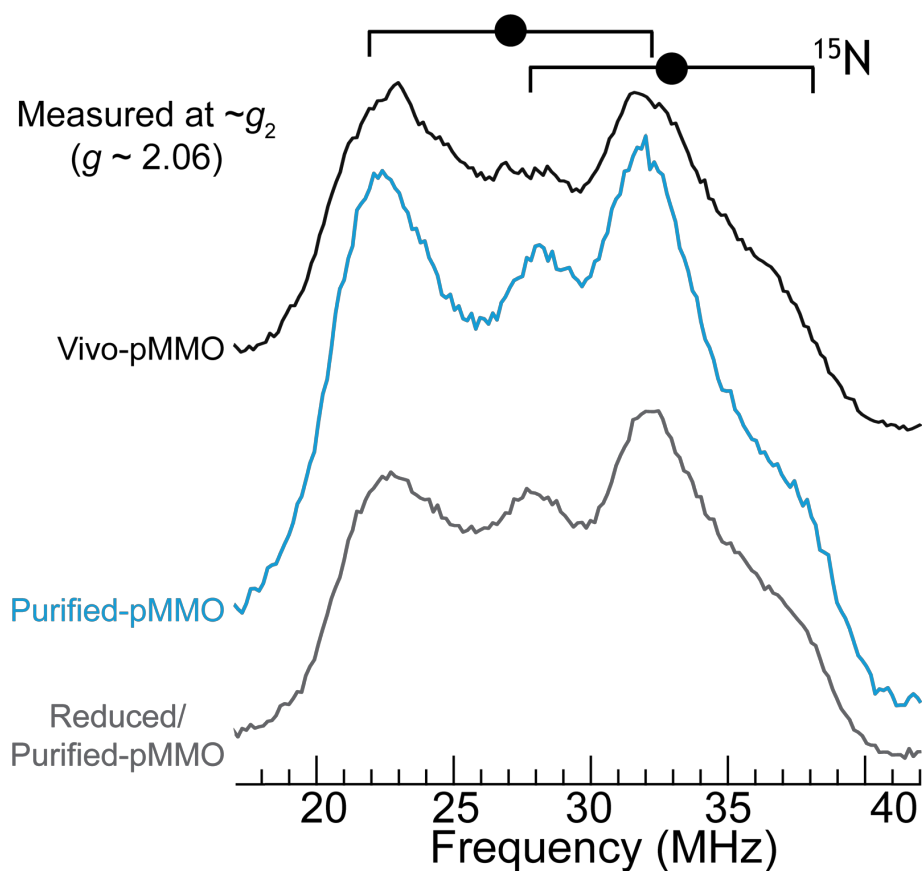


Fig. S14. PESTRE hyperfine sign assignments for strongly-coupled ^{15}N resonance in Vivo-pMMO. **Top:** Davies ENDOR measurements defining the ^{15}N resonance at g_1 . **Bottom:** PESTRE responses measured at measured at g_1 . The PESTRE experiment utilizes three discrete Davies ENDOR sequences. First, a baseline signal is established (BSL) during which time the microwave is applied with no RF, saturating the EPR transition. Next, a single frequency RF is applied to saturate the ENDOR transition (ENDOR). Finally, the RF is turned off (DRL) and the direction of echo relaxation relative to the BSL ($\text{DRL}\delta \equiv \text{DRL} - \text{BSL}$) defines the manifold producing the observed resonance (labeled as α or β to indicate the higher or lower energy electron spin manifolds, respectively). For ν_+ $\text{DRL}\delta > 0$, $A/g_n < 0$, while ν_+ $\text{DRL}\delta < 0$ defines $A/g_n > 0$; likewise, for ν_- $\text{DRL}\delta < 0$, $A/g_n < 0$, while ν_- $\text{DRL}\delta > 0$ defines $A/g_n > 0$. Consequently, as expected, the PESTRE measurements are consistent with exclusively strongly-coupled ^{15}N $A/g_n > 0$, indicating a positive spin density $\rho > 0$ since $\rho \propto A/g_n$. Collection conditions: 34.5-34.6 GHz microwave frequency, 100 ms repetition time, $\pi = 80$ ns, $\tau = 600$ ns, $T_{\text{RF}} = 15$ μs , and RF tail = 5 μs .



5 **Fig. S15.** X-band CW EPR of the Purified-pMMO sample used for DEER measurements (buffer
 conditions: 50 mM PIPES, 500 mM NaCl, 0.04% DDM, pH 7.2). Bracket width denotes A_1 Cu
 hyperfine splitting, and the center of the brackets indicate g_1 using the values from Table S1.
 Asterisk denotes an instrument artifact sometimes observed on the spectrometer. The altered buffer
 conditions do not alter the $\text{Cu}_B(\text{II})$ or $\text{Cu}_C(\text{II})$ signals from those observed under the standard
 Purified-pMMO buffer conditions (as in Fig. 1). Collection conditions: 9.366 GHz microwave
 10 frequency, 90 s scan rate, 320 ms time constant, 12.5 G modulation amplitude, 20 K temperature,
 5 scans.



5 **Fig. S16.** ^{15}N Davies ENDOR measured at g_2 for Vivo-, Purified-, and Reduced/Purified-pMMO. Intensities normalized to match those of ^{15}N in Fig. S3A. Goalpost width indicates twice the ^{15}N Larmor frequency ($\nu_{^{15}\text{N}}$), filled circle defines half the hyperfine coupling magnitude ($|A/2|$). Collection conditions: 34.5-34.7 GHz microwave frequency, 50 ms repetition time, $\pi = 80$ ns, $\tau = 400$ ns, $T_{\text{RF}} = 15$ μs , and RF tail = 5 μs .

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