## **Supplementary Data**







**Fig. S2. CO binding to BvHbs by stopped-flow rapid mixing.** (*A*) Time courses for CO binding to BvHb1.2 following stopped-flow rapid mixing at different CO concentrations (13, 25, 50, 125, 250, and 500 mM CO, from right to left) monitored at 430 nm. (*B*) Time courses for CO binding to BvHb1.1 following stopped-flow rapid mixing at different CO concentrations (13, 25, 50, 125, 250 mM CO, from right to left). Time courses were monitored at 430 nm. *Insets* in (*A*) and (*B*) show the absorbance spectra associated with the binding of CO to the deoxy-ferrous BvHbs as indicated by the color gradient bar (from green to red).



**Fig. S3. Oxygen binding and dissociation.** Time courses for rebinding of oxygen to BvHb1.1 (*A*) and BvHb1.2 (*B*) after flash photolysis at different oxygen concentrations. From right to left: 5, 31, 62, 125, and 250  $\mu$ M for BvHb1.1 and 5, 32, 63, 125, 250  $\mu$ M for BvHb1.2. The time courses were monitored at 410 nm. *Insets*: Absorbance spectra associated with the dissociation of oxygen from BvHb1.1 (*A*) and BvHb1.2 (*B*). The direction of the spectra change is indicated by the arrow, from oxyHb to deoxyHb.



**Fig. S4. BvHbs Nitrite reductase activity.** (*A*-*C*) Static spectra of ferric BvHbs (black line) and after binding NO (Fe<sup>3+</sup>-NO, red line). BvHb1.1, BvHb1.2, and BvHb2 are, respectively, represented in A, B, and C. (*D*) Stopped flow experiment, carried out at a very low dithionite concentration, showing the initial ferrous form of BvHb2 (green) converted to the ferric NO form (yellow) proving a nitrite reductase activity and then to the ferrous NO form (red). The sequence of the reaction goes from dark green to yellow and finally red.