

1 Supporting information

## 2 Heterocycle Effects on the Liquid Crystallinity of 3 Terthiophene Analogues

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### 7 1. Synthesis and Identification

8 All solvents used were reagent grade. 1-Bromooctane was purchased from TCI America.  
9 Triethylamine was purchased from Anachemia. All other reagents were purchased from Sigma-  
10 Aldrich. All reagents were used as received without further purification. Column chromatography  
11 was performed on silica gel 60 (230-400 mesh) purchased from Silicycle Inc. CDCl<sub>3</sub> was obtained from  
12 Cambridge Isotope Laboratories Inc.

13 400 MHz <sup>1</sup>H NMR spectra were run on a Bruker AMX-400 400 MHz NMR spectrometer. 500  
14 MHz <sup>1</sup>H NMR spectra were run on a Varian AS500 Unity Inova 500 MHz spectrometer. High  
15 resolution mass spectrometry was carried out on a Bruker micrOTOF II LC/MS (ESI<sup>+</sup>) by Nonka  
16 Sevova at Notre Dame Mass Spectrometry and Proteomics facility.

17 The final target compounds were prepared according to the procedures described in the  
18 experimental section of the paper. The synthetic procedures for any precursors and literature  
19 compounds are outlined below. The percent yield and detailed <sup>1</sup>H and <sup>13</sup>C NMR and mass spec data  
20 for all compounds are given below.

#### 21 1.1. Precursors and literature compounds

22 *2-Decylthiophene*: In an oven-dried 100 mL 3-neck round bottom flask, thiophene (1.0 g, 11.9  
23 mmol) in dry THF was cooled to -78 °C under a nitrogen atmosphere. n-Butyllithium (0.95 eq., 2.5 M  
24 solution in hexanes) was added drop-wise. The mixture was stirred at 0 °C for 60 minutes. Once  
25 cooled back to -78 °C, 1-bromodecane (0.90 eq.) was added slowly. The reaction was allowed to warm  
26 to room temperature and stirred overnight. It was then quenched with water, extracted with diethyl  
27 ether (3x50 mL), and the combined organic fractions were dried over magnesium sulfate. The crude  
28 was purified by vacuum distillation to give the product as a yellow oil (81 % yield). <sup>1</sup>H NMR (500  
29 MHz, Chloroform-*d*) δ 7.18 (dd, *J* = 5.1, 1.1 Hz, 1H), 7.00 (dd, *J* = 5.1, 3.4 Hz, 1H), 6.87 (d, *J* = 3.2 Hz,  
30 1H), 2.93 (t, *J* = 7.7 Hz, 2H), 1.79 (p, *J* = 7.4 Hz, 2H), 1.53 – 1.33 (m, 16H), 1.02 (t, *J* = 6.8 Hz, 3H) ppm.

31 *5-Decyl-2-thiophenecarboxylic acid*: In an oven-dried 100 mL 3-neck round bottom flask, 2-  
32 decylthiophene (1.0 g, 4.46 mmol) in dry diethyl ether was cooled to -78 °C under a nitrogen  
33 atmosphere. n-Butyllithium (1.2 eq., 2.5 M solution in hexanes) was added drop-wise. The mixture  
34 was stirred at 0 °C for 60 minutes. Once cooled back to -78 °C, carbon dioxide (excess) in the form of  
35 dry ice chunks was added. The reaction was allowed to warm to room temperature and stirred  
36 overnight. The precipitate was filtered and washed with diethyl ether then 10% hydrochloric acid  
37 and dried under high vacuum overnight to give the product as a white solid (69 % yield). <sup>1</sup>H NMR  
38 (500 MHz, Chloroform-*d*) δ 7.70 (d, *J* = 3.6 Hz, 1H), 6.80 (d, *J* = 3.5 Hz, 1H), 4.65 (s, 1H), 2.83 (t, *J* = 7.6  
39 Hz, 2H), 1.69 (p, *J* = 7.5 Hz, 2H), 1.43 – 1.18 (m, 15H), 0.88 (t, *J* = 6.8 Hz, 3H) ppm.

40 *Hydrazide derivative 1*: In an oven-dried 25 mL 3-neck round bottom flask, 5-decyl-2-  
41 thiophenecarboxylic acid (1.0 g, 3.52 mmol) was refluxed in thionyl chloride (5 mL, excess) under a  
42 nitrogen atmosphere for 2 hours. After cooling to room temperature, remaining thionyl chloride was  
43 removed via high vacuum overnight to generate 5-decyl-2-thiophenecarbonyl chloride in situ. This –  
44 oyl chloride was added to an oven-dried 50 mL 3-neck round bottom flask containing dry NMP.  
45 Triethylamine (3 mL, excess) was added followed by the drop-wise addition of hydrazine hydrate  
46 (0.7 eq.) at 0 °C. The mixture was warmed to room temperature and stirred overnight. The resulting

47 precipitate was collected by filtration. This crude product was purified by column chromatography  
48 on silica treated with 2% trimethylamine using a gradient from 0 to 50 % of ethyl acetate in hexanes  
49 as the eluent. The pure product was isolated as a white powder (37 % yield). <sup>1</sup>H NMR (500 MHz,  
50 Chloroform-*d*) δ 9.04 (s, 2H), 7.52 (d, *J* = 3.7 Hz, 2H), 6.76 (d, *J* = 3.7 Hz, 2H), 2.81 (t, *J* = 7.6 Hz, 4H),  
51 1.67 (p, *J* = 7.5 Hz, 4H), 1.26 (s, 28H), 0.88 (t, *J* = 6.9 Hz, 6H) ppm.

52 α-Terthienyl, **Th**<sub>3</sub>: Magnesium turnings (0.90 g, 37.0 mmol) were added to a 100 mL 3-neck round  
53 bottom flask, which was then sealed, purged with N<sub>2</sub>, and flame dried. Added dry diethyl ether (50  
54 mL) and cooled to 0 °C, followed by slow addition of 2-bromothiophene (3.26 mL, 33.7 mmol).  
55 Solution was warmed to room temperature and then refluxed for 2 hours to generate the Grignard  
56 solution. An oven-dried 250 mL 3-neck round bottom flask was charged with [1,3-  
57 bis(diphenylphosphino)propane] dichloronickel(II) (catalytic, ~2%) and cycled between N<sub>2</sub> and  
58 vacuum three times. Dry diethyl ether (150 mL) and then 2,5-dibromothiophene (1.71 mL, 15.3 mmol)  
59 were added and the solution was cooled to 0 °C. The Grignard solution prepared in situ was then  
60 added dropwise, the reaction was warmed to room temperature and then refluxed for 2 hours. After  
61 cooling to room temperature, the reaction was quenched by slow addition of 10% hydrochloric acid,  
62 the organic was separated out, and the aqueous was further extracted with diethyl ether (3x75 mL).  
63 The combined organic fractions were dried over magnesium sulfate. The crude was purified by  
64 column chromatography on silica using 5% ethyl acetate in hexanes as the eluent to give the final  
65 product (57 % yield). <sup>1</sup>H NMR (500 MHz, Chloroform-*d*) δ 7.23 (dd, *J* = 5.1, 1.2 Hz, 2H), 7.18 (dd, *J* =  
66 3.6, 1.2 Hz, 2H), 7.09 (s, 2H), 7.03 (dd, *J* = 5.1, 3.6 Hz, 2H) ppm.

67 **Th**<sub>3</sub>(10): A flame-dried 50 mL three-neck round bottom flask was charged with **Th**<sub>3</sub> (0.15 g, 0.60  
68 mmol) and cycled between high vacuum and N<sub>2</sub> atmosphere three times. Dry THF (25 mL) was added  
69 by cannula and the solution was cooled to around -78 °C prior to the drop-wise addition of *n*-  
70 butyllithium solution (2.5 M, 2.5 equivalents). After stirring for 30 minutes, potassium *t*-butoxide (1.0  
71 M solution, 2.5 equivalents) was added and the solution was stirred for another 30 minutes. Then 1-  
72 decylbromide (2.2 equivalents) was added drop-wise. This mixture was warmed to room  
73 temperature and stirred overnight. The reaction was quenched with water (40 mL) and extracted with  
74 hexanes (3x30 mL). The combined organic fractions were dried over magnesium sulfate. Further  
75 purification was done by column chromatography on silica using hexanes as the eluent.  
76 Recrystallization from ethanol afforded the product as a light yellow solid (49 % yield). <sup>1</sup>H NMR (500  
77 MHz, Chloroform-*d*) δ 6.96 (t, *J* = 3.3 Hz, 4H), 6.67 (d, *J* = 3.4 Hz, 2H), 2.78 (t, *J* = 7.6 Hz, 4H), 1.67 (p, *J*  
78 = 7.4 Hz, 4H), 1.42 – 1.21 (m, 28H), 0.92 – 0.84 (m, 6H) ppm. <sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ  
79 145.56, 136.29, 134.79, 124.92, 123.60, 123.31, 32.06, 31.76, 30.35, 29.76, 29.71, 29.52, 29.48, 29.24, 22.84,  
80 14.27 ppm.

81 **Th-Thd-Th core**: In a 150 mL high pressure vessel, 2-thiophenecarboxaldehyde (1.5 g, 13.4  
82 mmol), sulfur (1.5 equivalents), and hydrazine hydrate (4.0 equivalents) were combined in propanol  
83 (10 mL). The mixture was allowed to react for 4 hours at 150 °C under high pressure. Once cooled to  
84 room temperature, the precipitate was collected by vacuum filtration. The crude was then dissolved  
85 in dichloromethane (150 mL), washed with saturated sodium sulfide (2x120 mL), water (1x150 mL),  
86 and then dried over magnesium sulfate. Recrystallization from ethanol afforded the final product as  
87 a light orange solid (88 % yield). <sup>1</sup>H NMR (500 MHz, Chloroform-*d*) δ 7.56 (dd, *J* = 3.7, 1.1 Hz, 2H),  
88 7.50 (dd, *J* = 5.1, 1.1 Hz, 2H), 7.13 (dd, *J* = 5.0, 3.7 Hz, 2H) ppm. <sup>13</sup>C NMR (126 MHz, Chloroform-*d*) δ  
89 161.18, 132.35, 129.65, 129.55, 128.14 ppm.

## 90 1.2. Oxadiazole derivative

91 **Th-Oxd-Th(10)**: 64% yield. <sup>1</sup>H NMR (500 MHz, Chloroform-*d*) δ 7.60 (d, *J* = 3.7 Hz, 2H), 6.84 (dd,  
92 *J* = 3.7, 1.0 Hz, 2H), 2.87 (t, *J* = 7.6 Hz, 4H), 1.72 (p, *J* = 7.6 Hz, 4H), 1.42 – 1.23 (m, 30H), 0.88 (t, *J* = 6.9  
93 Hz, 6H) ppm. <sup>13</sup>C NMR (126 MHz, Chloroform-*d*) δ 160.25, 151.93, 129.83, 125.48, 122.39, 32.04, 31.61,  
94 30.40, 29.72, 29.67, 29.46, 29.17, 22.83, 14.27 ppm. HRMS (ESI<sup>+</sup> of *m* + H<sup>+</sup>): *m/z* calcd for C<sub>30</sub>H<sub>47</sub>N<sub>2</sub>O<sub>5</sub>:  
95 515.3085, found: 515.3124.

## 96 1.3. Thiadiazole derivatives

97 **Th-Thd-Th(4)**: 62% yield.  $^1\text{H}$  NMR (500 MHz, Chloroform-*d*)  $\delta$  7.35 (d,  $J$  = 3.7 Hz, 2H), 6.79 (dd,  
98  $J$  = 3.7, 1.0 Hz, 2H), 2.89 – 2.81 (m, 4H), 1.75 – 1.66 (m, 4H), 1.48 – 1.36 (m, 4H), 0.95 (t,  $J$  = 7.4 Hz, 6H)  
99 ppm.  $^{13}\text{C}$  NMR (126 MHz, Chloroform-*d*)  $\delta$  160.96, 150.97, 129.74, 129.53, 125.30, 33.68, 30.15,  
100 22.28, 13.96 ppm. HRMS (ESI<sup>+</sup> of  $m + \text{H}^+$ ):  $m/z$  calcd for  $\text{C}_{18}\text{H}_{23}\text{N}_2\text{S}_3$ : 363.0979, found: 363.1018.

101 **Th-Thd-Th(6)**: 29% yield.  $^1\text{H}$  NMR (500 MHz, Chloroform-*d*)  $\delta$  7.35 (d,  $J$  = 3.7 Hz, 2H), 6.79 (dd,  
102  $J$  = 3.7, 1.0 Hz, 2H), 2.85 (t,  $J$  = 7.6 Hz, 4H), 1.71 (p,  $J$  = 7.6 Hz, 4H), 1.43 – 1.29 (m, 12H), 0.92 – 0.87 (m,  
103 6H) ppm.  $^{13}\text{C}$  NMR (126 MHz, Chloroform-*d*)  $\delta$  160.97, 151.04, 129.73, 129.54, 125.29, 31.57, 30.47,  
104 28.86, 22.70, 14.24, 14.19 ppm. HRMS (ESI<sup>+</sup> of  $m + \text{H}^+$ ):  $m/z$  calcd for  $\text{C}_{22}\text{H}_{31}\text{N}_2\text{S}_3$ : 419.1605, found:  
105 419.1644.

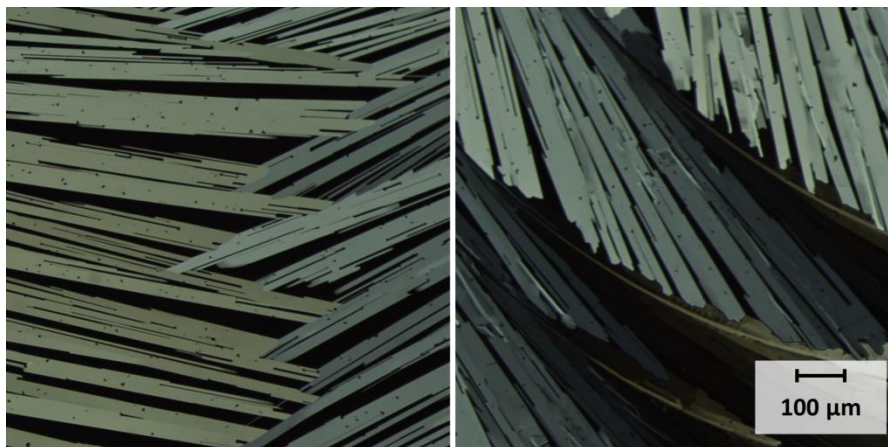
106 **Th-Thd-Th(8)**: 30% yield.  $^1\text{H}$  NMR (500 MHz, Chloroform-*d*)  $\delta$  7.35 (d,  $J$  = 3.7 Hz, 2H), 6.79 (d,  $J$   
107 = 3.7 Hz, 2H), 2.84 (t,  $J$  = 7.6 Hz, 4H), 1.71 (p,  $J$  = 7.6 Hz, 4H), 1.43 – 1.21 (m, 20H), 0.88 (t,  $J$  = 6.9 Hz,  
108 6H) ppm.  $^{13}\text{C}$  NMR (126 MHz, Chloroform-*d*)  $\delta$  160.97, 151.03, 129.73, 129.52, 125.28, 31.98, 31.61,  
109 30.46, 29.37, 29.20, 22.80, 14.27, 14.23 ppm. HRMS (ESI<sup>+</sup> of  $m + \text{H}^+$ ):  $m/z$  calcd for  $\text{C}_{26}\text{H}_{39}\text{N}_2\text{S}_3$ : 475.2231,  
110 found: 475.2270.

111 **Th-Thd-Th(10)**: 51% yield.  $^1\text{H}$  NMR (500 MHz, Chloroform-*d*)  $\delta$  7.35 (d,  $J$  = 3.7 Hz, 2H), 6.79 (d,  
112  $J$  = 3.7 Hz, 2H), 2.84 (t,  $J$  = 7.6 Hz, 4H), 1.71 (p,  $J$  = 7.5 Hz, 4H), 1.43 – 1.22 (m, 28H), 0.88 (t,  $J$  = 6.9 Hz,  
113 6H) ppm.  $^{13}\text{C}$  NMR (126 MHz, Chloroform-*d*)  $\delta$  160.97, 151.04, 129.74, 129.52, 125.29, 32.05, 31.61,  
114 30.48, 29.73, 29.68, 29.47, 29.20, 22.84, 14.27 ppm. HRMS (ESI<sup>+</sup> of  $m + \text{H}^+$ ):  $m/z$  calcd for  $\text{C}_{30}\text{H}_{47}\text{N}_2\text{S}_3$ :  
115 531.2857, found: 531.2896.

116 **Th-Thd-Th(12)**: 69% yield.  $^1\text{H}$  NMR (500 MHz, Chloroform-*d*)  $\delta$  7.35 (d,  $J$  = 3.6 Hz, 2H), 6.79 (d,  
117  $J$  = 3.7 Hz, 2H), 2.84 (t,  $J$  = 7.6 Hz, 4H), 1.71 (p,  $J$  = 7.6 Hz, 4H), 1.41 – 1.22 (m, 36H), 0.88 (t,  $J$  = 6.9 Hz,  
118 6H) ppm.  $^{13}\text{C}$  NMR (126 MHz, Chloroform-*d*)  $\delta$  160.96, 151.03, 129.74, 129.51, 125.29, 32.07, 31.61,  
119 30.47, 29.78, 29.68, 29.50, 29.46, 29.48, 29.19, 22.85, 14.27 ppm. HRMS (ESI<sup>+</sup> of  $m + \text{H}^+$ ):  $m/z$  calcd for  
120  $\text{C}_{34}\text{H}_{55}\text{N}_2\text{S}_3$ : 587.3483, found: 587.3522.

## 121 2. Polarized Optical Microscopy (POM)

122 Polarized optical microscopy was carried out using an Olympus BX50 microscope equipped  
123 with a Linkam LTS350 heating stage. All images shown are of a size of ca. 920 x 1400  $\mu\text{m}$ .

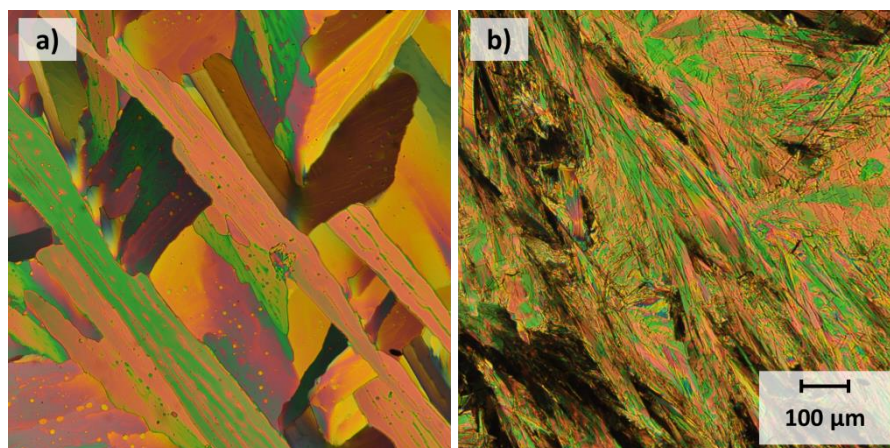


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**Figure S1.** POM of Th-Oxd-Th(10) showing needle-like textures of the crystalline state at 72.0 °C.



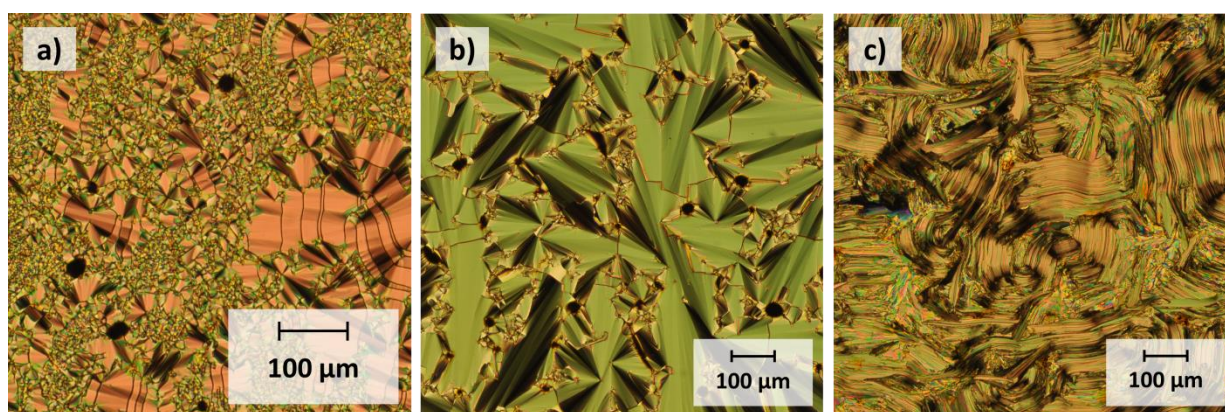


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**Figure S2.** POM of Th-Thd-Th(4) showing needle-like textures of the crystalline state at 94.0 °C (a) and at 25.0 °C (b).



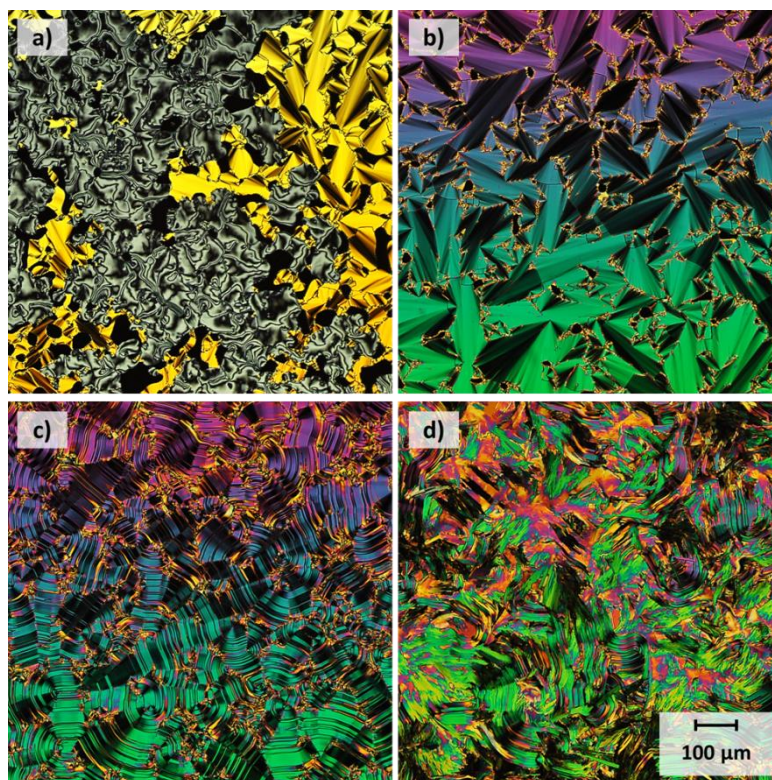
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**Figure S3.** POM of Th-Thd-Th(6) showing schlieren and focal-conic textures of the SmC phase at 111.0 °C (a and b) and striations across focal-conic textures of the CrJ phase at 25.0 °C (c).





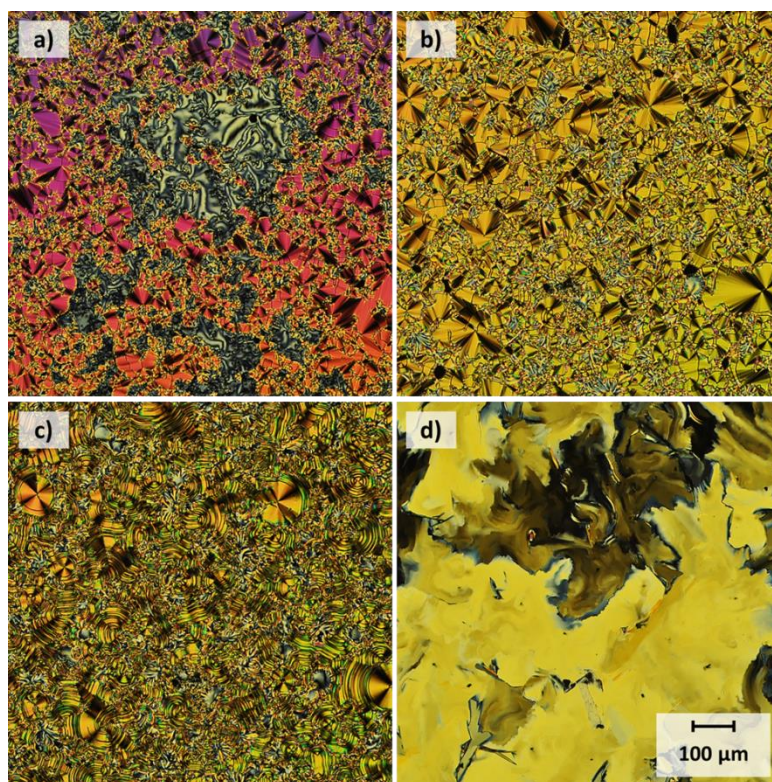
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**Figure S4.** POM of Th-Thd-Th(8) showing schlieren and focal-conic textures of the SmC phase at 125.0 °C (a and b), striations across focal-conic textures of the CrJ phase at 88.0 °C (c), and changes upon cooling to the crystalline phase at 74.0 °C (d).



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**Figure S5.** POM of Th-Thd-Th(12) showing schlieren and focal-conic textures of the SmC phase at 131.0 °C (a and b respectively), striations across focal-conic textures of the CrJ phase at 96.0 °C (c), and changes upon cooling to the crystalline phase at 87.0 °C (d).

140 **3. Variable Temperature Powder X-ray Diffraction (VT-XRD)**

141 X-ray scattering experiments were conducted using a Rigaku R-Axis Rapid diffractometer  
 142 equipped with an in-house built temperature controller. [1]

143 **Table 1.** XRD parameters of the phases of **Th<sub>3</sub>(10)** and **Th-Thd-Th(n)** (n = 6,8,10,12).

Compound (molecular length <sup>a</sup> , nm)	Temperature (°C)	Phase	Layer spacing <sup>b</sup> (nm)	Correlation length <sup>c</sup> (nm)	# of layers	Tilt angle, $\theta$ (°) <sup>d</sup>
<b>Th<sub>3</sub>(10)</b> (3.91)	91	SmF	3.60	26	7	23
	25	Cr	3.85	27	8	10
<b>Th-Thd-Th(6)</b> (2.95)	110	SmC	2.26	31	13	40
	76	CrJ	2.28	36	16	39
	25	CrJ	2.17	38	17	43
<b>Th-Thd-Th(8)</b> (3.47)	124	SmC	2.65	25	9	40
	83	CrJ	2.93	31	11	32
	25	Cr	2.49	38	15	44
<b>Th-Thd-Th(10)</b> (4.00)	126	SmC	3.01	31	10	41
	88	CrJ	3.33	32	10	34
	25	Cr	3.10	34	11	39
<b>Th-Thd-Th(12)</b> (4.49)	127	SmC	3.38	29	9	41
	101	SmI	3.66	29	8	35
	95	CrJ	3.69	25	7	35
	25	Cr	3.38	29	9	41

144 <sup>a</sup>calculated by DFT using B3LYP/6-31G\*; <sup>b</sup>determined based on  $d_{001}$  peak; <sup>c</sup>calculated using Debye-Scherrer  
 145 equation applied to  $d_{001}$  peak; <sup>d</sup>calculated from molecular length and layer spacing

146 **Table S2.** XRD data of **Th<sub>3</sub>(10)** and **Th-Thd-Th(n)** (n=6,8,10,12).

Compound (molecular length <sup>a</sup> , Å)	Temperature (°C)	Phase	d-spacings (Å)	Miller indices (hkl)
<b>Th<sub>3</sub>(10)</b> (39.1)	91	SmF	36.0	001
			4.37	110
			38.5	001
	25	Cr	23.3	---
			19.6	002
			4.37	110
			3.78	---
	110	SmC	3.06	---
			22.7	001
			4.37	alkyl halo
<b>Th-Thd-Th(6)</b> (29.5)	76	CrJ	22.7	001
			11.4	002
			9.08	---

			7.22	003
			6.65	---
			6.24	---
			5.55	004
			4.82	---
			4.43	110
			3.62	---
			3.54	---
			3.27	---
			3.20	---
			21.7	001
			10.9	002
			7.12	003
			6.45	---
			5.78	---
	25	CrJ'	4.78	---
			4.34	110
			4.03	---
			3.55	---
			3.16	---
			3.02	---
	124	SmC	26.5	001
			4.37	alkyl halo
			29.3	001
			14.7	002
			9.83	003
			7.06	004
			6.73	---
			5.18	---
	83	CrJ	4.87	---
			4.45	110
			4.29	---
			4.11	---
			3.81	---
			3.58	---
			3.30	---
			24.9	001
			12.5	002
			8.79	003
	25	Cr	5.78	---
			4.75	---
			4.32	110
			4.15	---

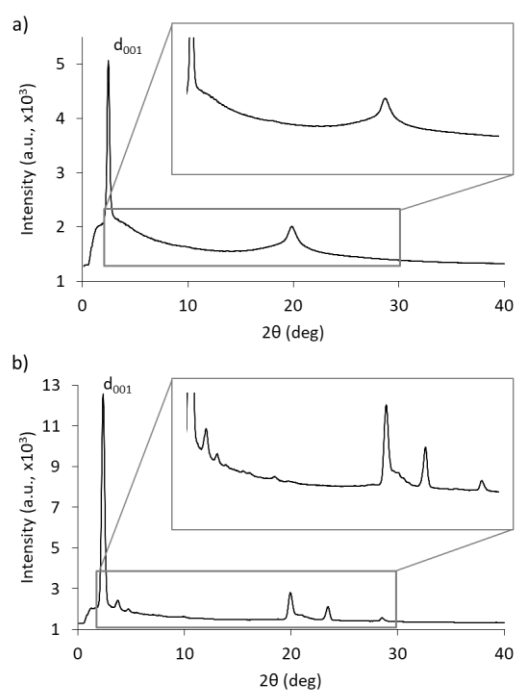
Th-Thd-Th(8) (34.7)

			3.54	---
			3.29	---
			3.14	---
			30.1	001
	126	SmC	15.1	002
			4.37	alkyl halo
			33.3	001
			16.7	002
			11.1	003
			9.94	---
			8.84	---
	88	CrJ	7.59	---
			5.16	---
			4.29	110
			4.27	---
			4.07	---
			3.98	---
			3.80	---
			31.0	001
			15.4	002
			9.78	003
			7.25	---
			6.36	---
			5.04	---
	25	Cr	4.31	---
			4.22	110
			4.11	---
			3.94	---
			3.79	---
			3.65	---
			3.54	---
			33.8	001
	127	SmC	17.1	002
			4.37	alkyl halo
			36.6	001
	101	SmI	18.5	002
			4.37	110
			36.9	001
			18.8	002
	95	CrJ	12.4	003
			9.98	---
			8.95	004



	5.13	---	
	4.52	---	
	4.33	110	
	4.24	---	
	4.02	---	
	3.88	---	
	3.74	---	
	3.52	---	
	33.8	001	
	17.1	002	
	11.8	003	
	9.83	---	
	8.61	004	
	7.65	---	
25	Cr	5.07	---
		4.26	110
		4.17	---
		3.80	---
		3.67	---
		3.55	---
		3.44	---

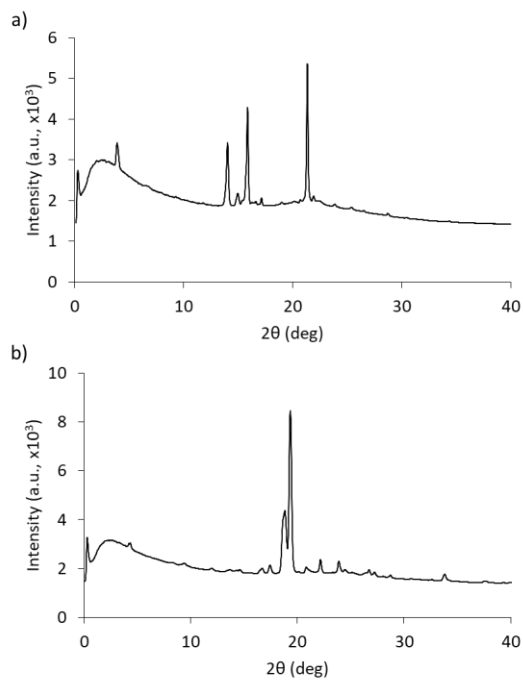
147 <sup>a</sup>calculated by DFT using B3LYP/6-31G\*



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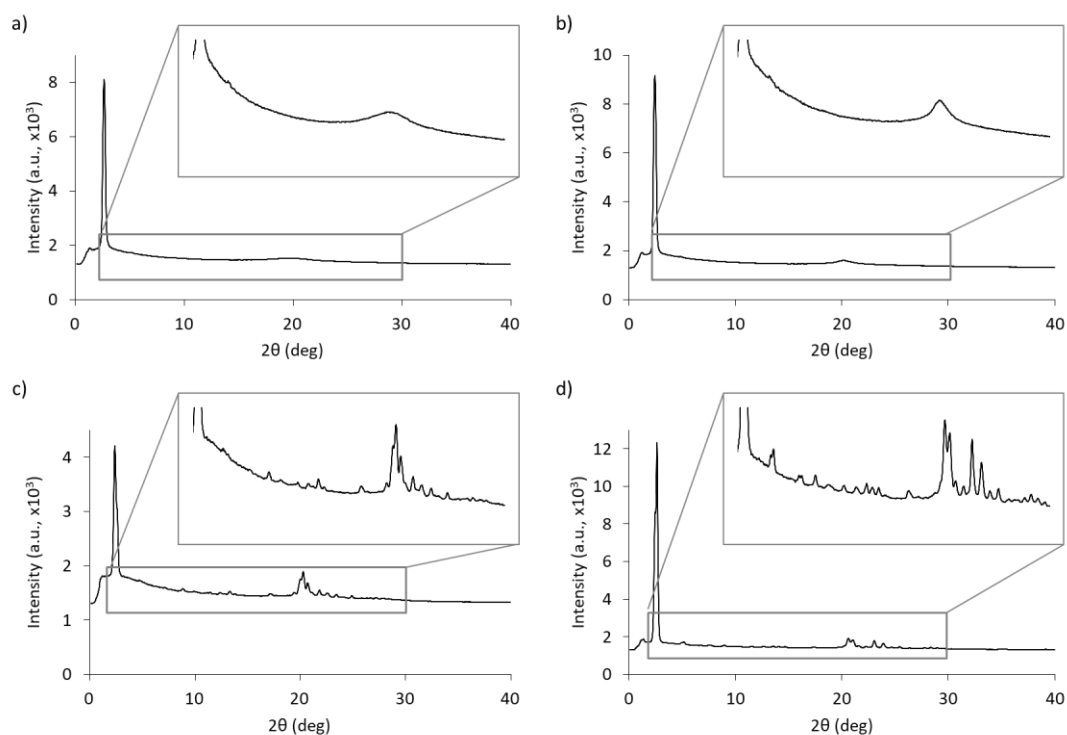
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**Figure S6.** XRD of  $\text{Th}_3(10)$  at 91 °C, in the SmF phase (a) and at 25 °C, in the crystalline phase (b).



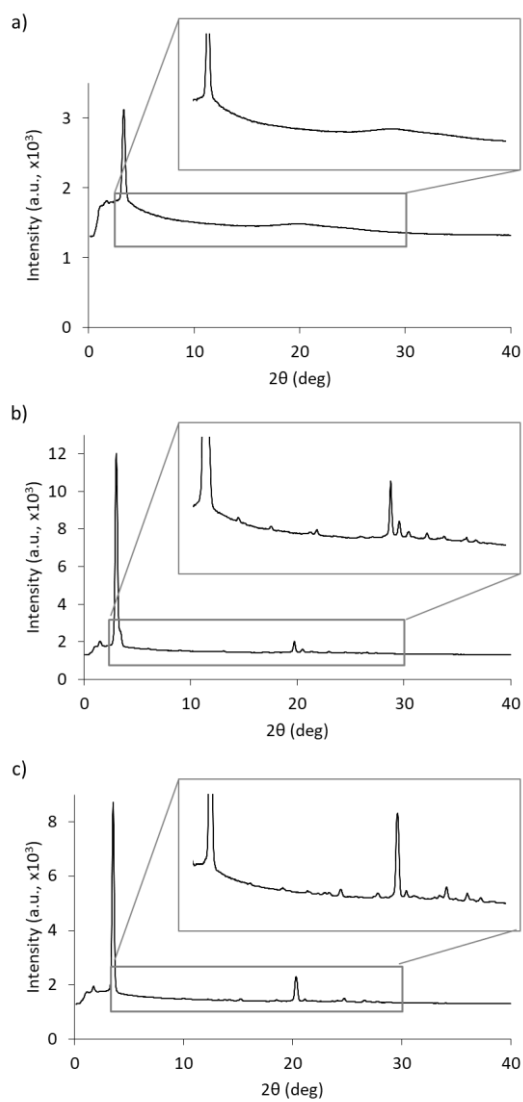
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151 **Figure S7.** XRD of Th-Oxd-Th(10) at 70 °C, in the Cr phase (a) and at 25 °C, in the Cr'' phase (b).



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153 **Figure S8.** XRD of Th-Thd-Th(12) at 127 °C, in the SmC phase (a); at 101 °C, in the SmI phase (b); at  
154 95 °C, in the CrI phase (c); and at 25 °C, in the crystalline phase (d).

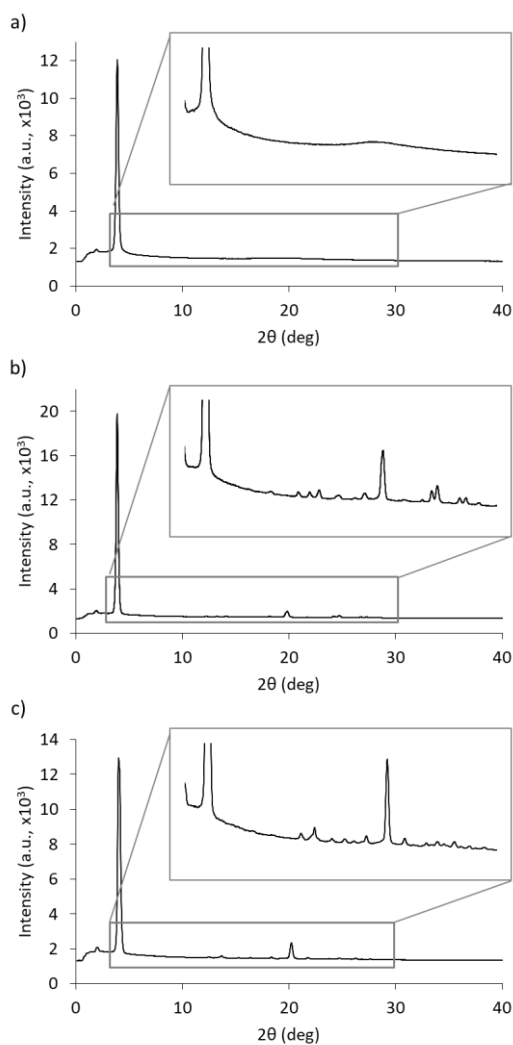


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**Figure S9.** XRD of Th-Thd-Th(8) at 124 °C, in the SmC phase (a); at 83 °C, in the CrI phase (b); and at 25 °C, in the crystalline phase (c).

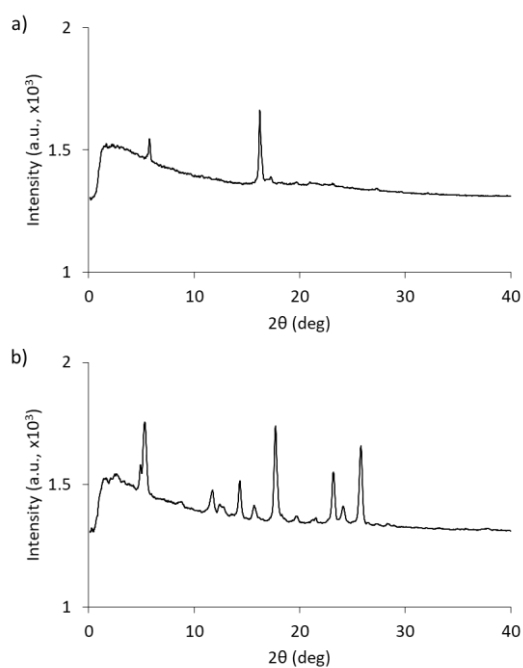


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**Figure S10.** XRD of Th-Thd-Th(6) at 110 °C, in the SmC phase (a); at 76 °C, in the CrI phase (b); and at 25 °C, in the crystalline phase (c).



161



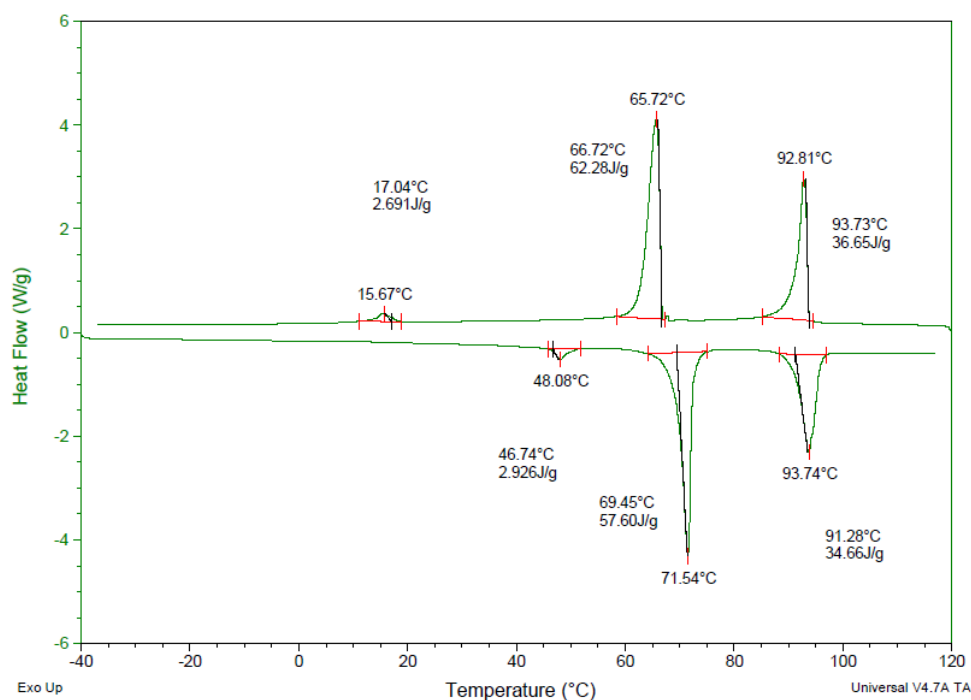
162 **Figure S11.** XRD of Th-Thd-Th(4) at 94 °C, in the Cr phase (a) and at 25 °C, in the Cr'' phase (b).

#### 163 4. Differential Scanning Calorimetry (DSC)

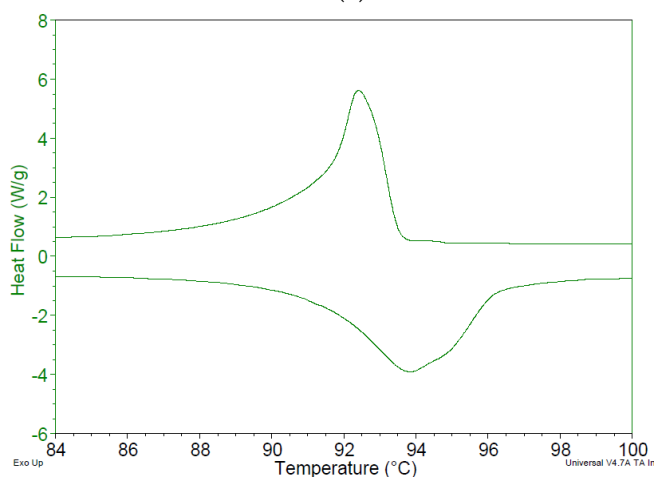
164 Phase transition temperatures and enthalpies were determined using differential scanning  
 165 calorimetry (DSC) on a TA Instruments DSC Q2000 equipped with a TA Instruments Refrigerated  
 166 Cooling System 90, heating and cooling at a rate of 10 °C min<sup>-1</sup>.

167 **Table S3.** Phase transition temperatures and enthalpies of Th<sub>3</sub>(n) reported by Boucher et al. [2].

n	Transition Temperature/ °C [enthalpy/ kJ mol <sup>-1</sup> ]
5	Cr 46 [15.2] CrG 76 [13.6] I
6	Cr 51 [16.7] CrG 81 [16.4] I
7	Cr 52 [17.8] CrG 75 [3.8] SmF 82 [0.7] SmC 86 [7] I
8	Cr 64 [24.4] CrG 71 [3.3] SmF 85 [1.6] SmC 90 [7.3] I
9	Cr 66 [30.8] SmF 91 [1.8] SmC 94 [10.4] I
10	Cr1 43 [2.3] Cr2 71 [32.1] SmF 92 [2.7] SmC 95 [10.5] I

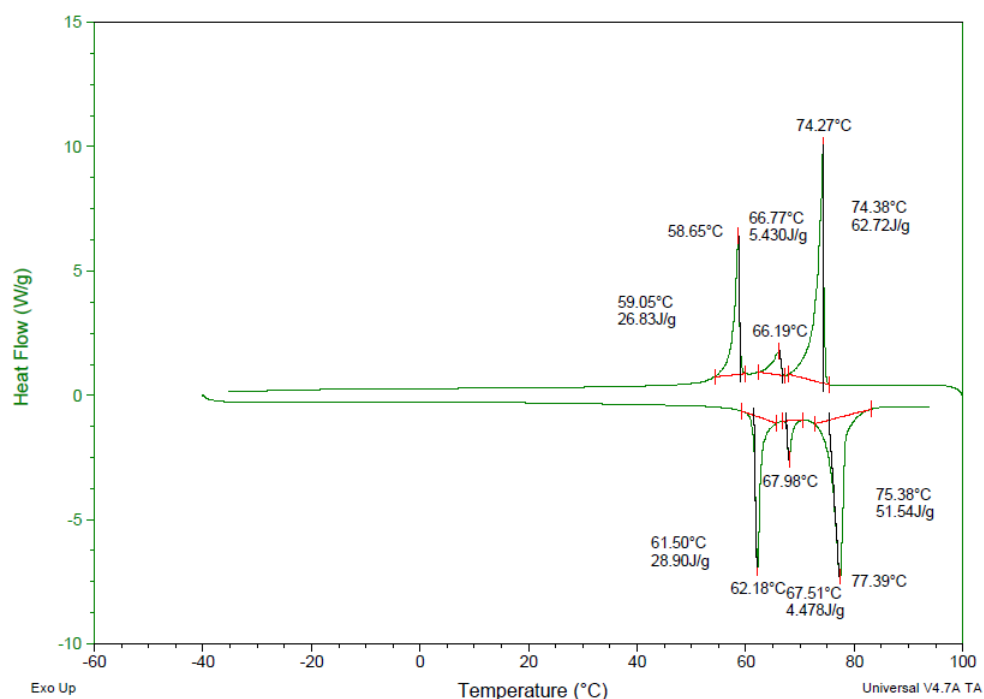


(a)



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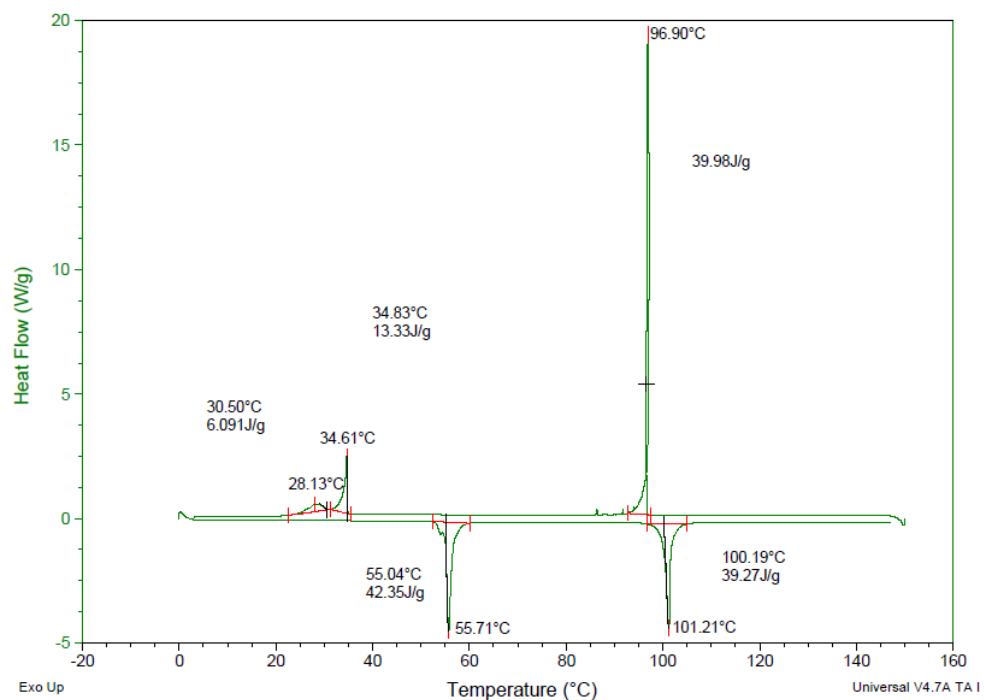
(b)  
**Figure S12.** DSC thermogram of Th<sub>3</sub>(10) (a), including zoomed in high temperature peak (b).



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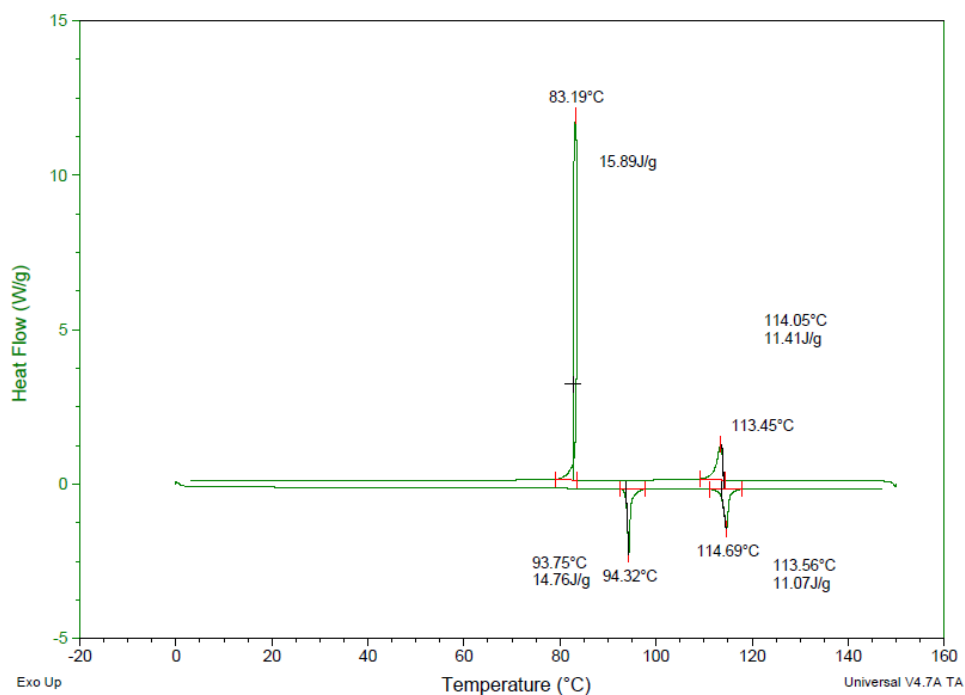
**Figure S13.** DSC thermogram of Th-Oxd-Th(10).



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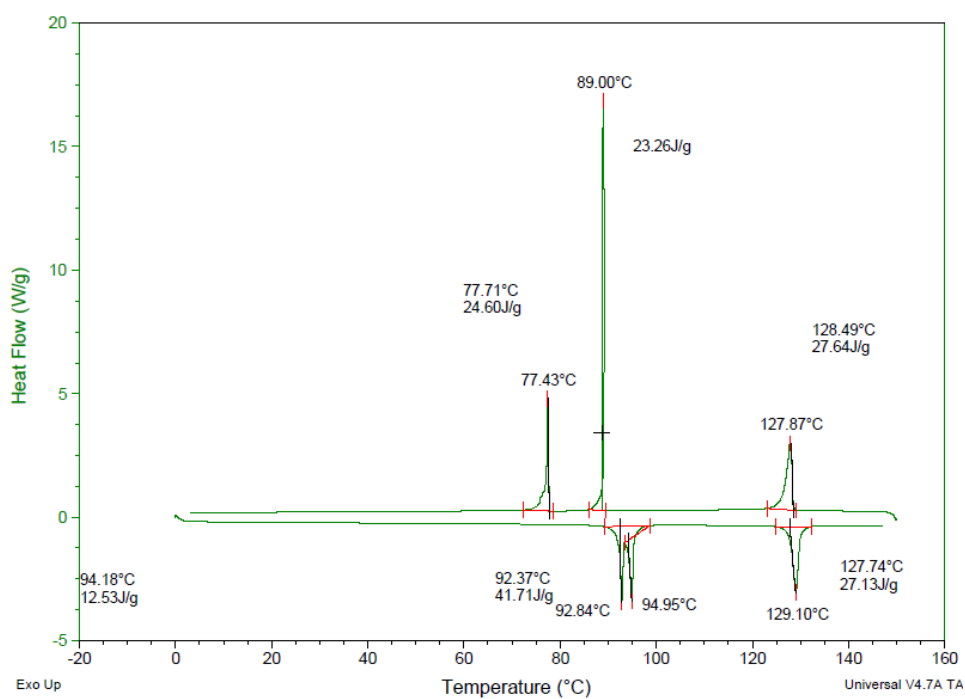
**Figure S14.** DSC thermogram of Th-Thd-Th(4).



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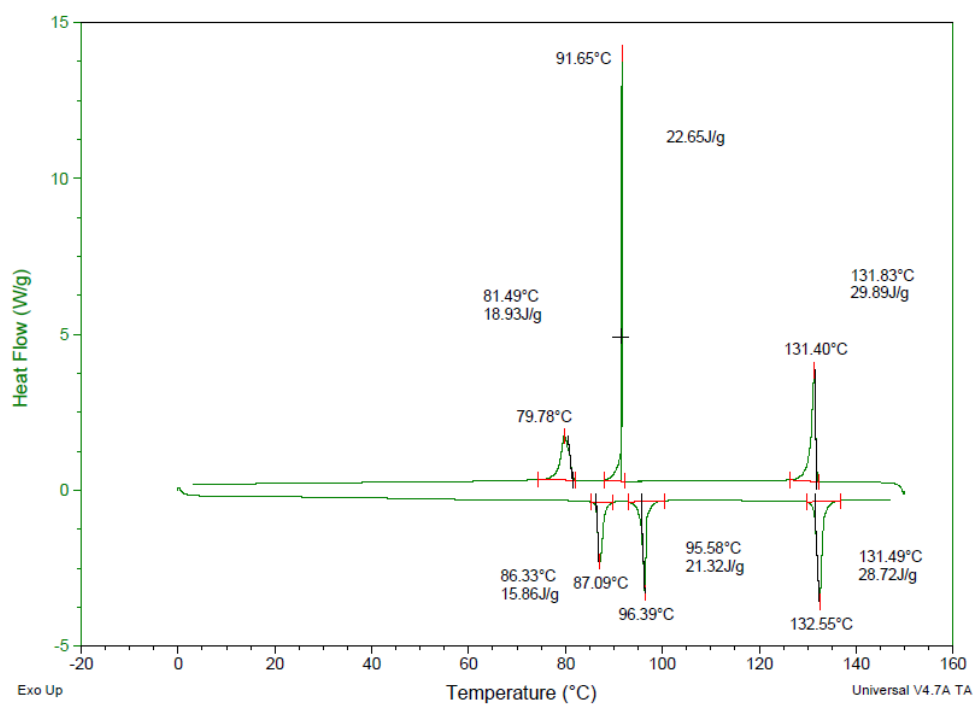
Figure S15. DSC thermogram of Th-Thd-Th(6).



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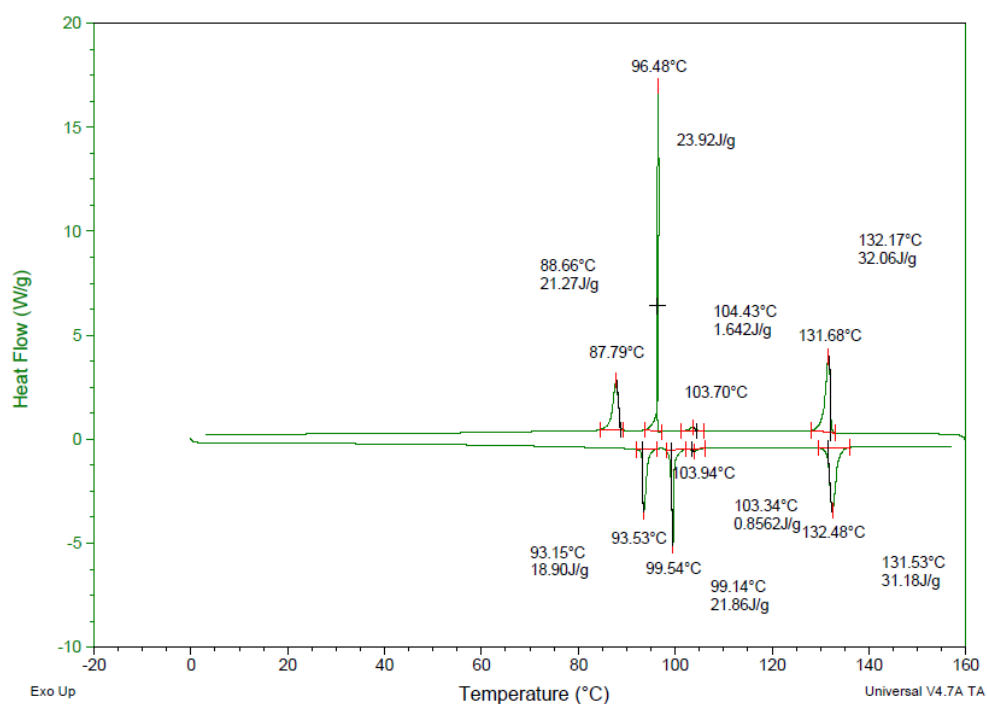
Figure S16. DSC thermogram of Th-Thd-Th(8).



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Figure S17. DSC thermogram of Th-Thd-Th(10).



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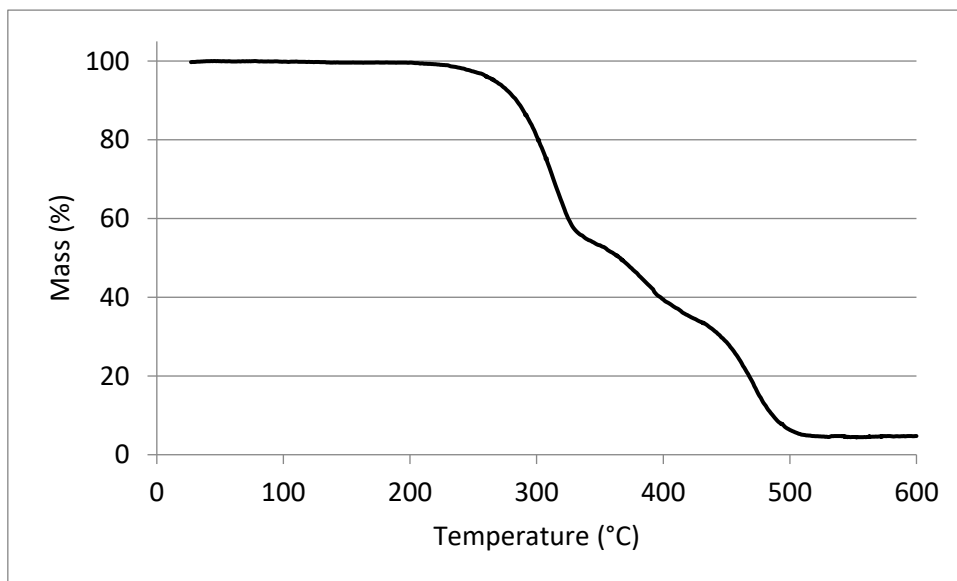
Figure S18. DSC thermogram of Th-Thd-Th(12).

## 181 5. Thermogravimetric Analysis (TGA)

182 Thermogravimetric analysis was carried out on a Shimadzu TGA-50 with a heating rate of 2 °C  
 183 per minute.

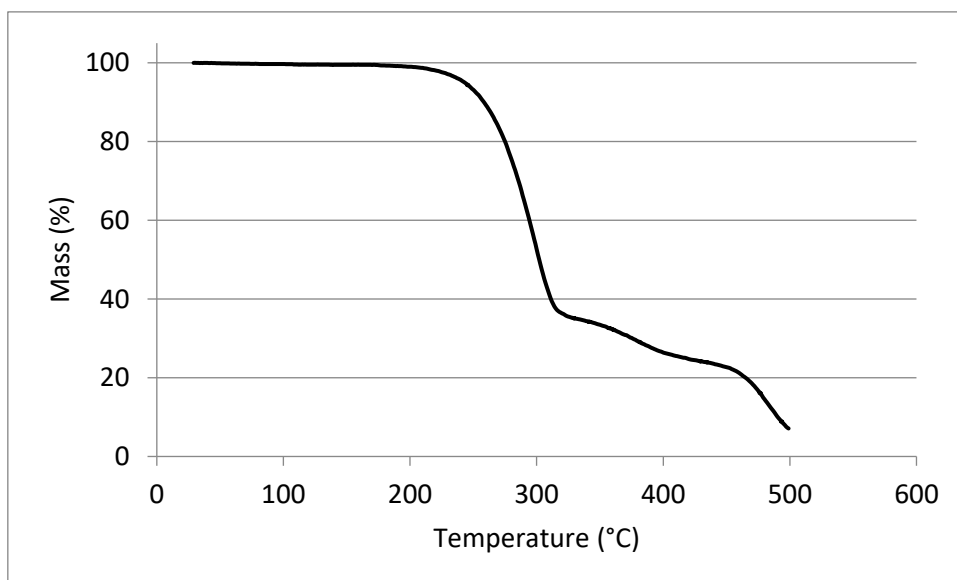
184 TGA of Th-Oxd-Th(10):





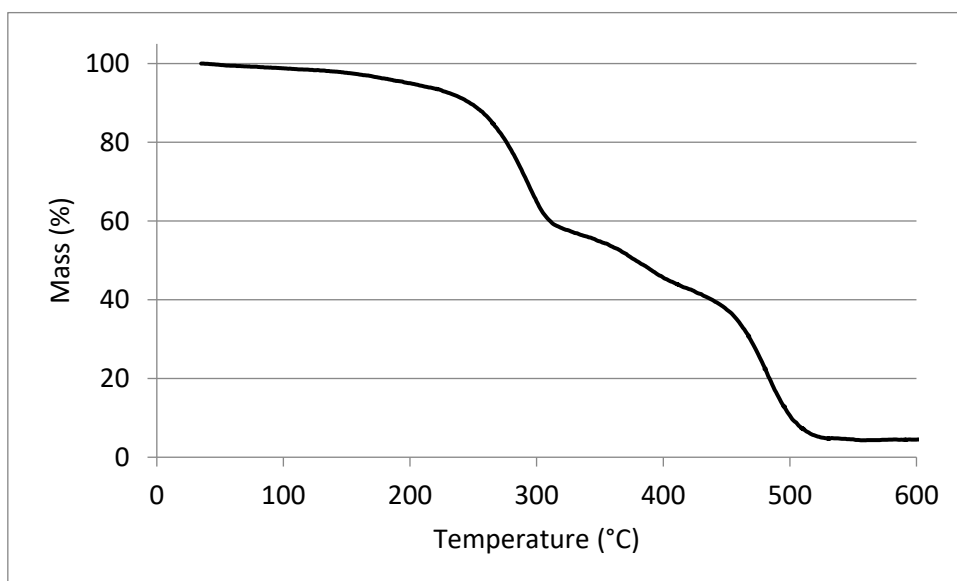
185  
186

TGA of Th-Thd-Th(4):



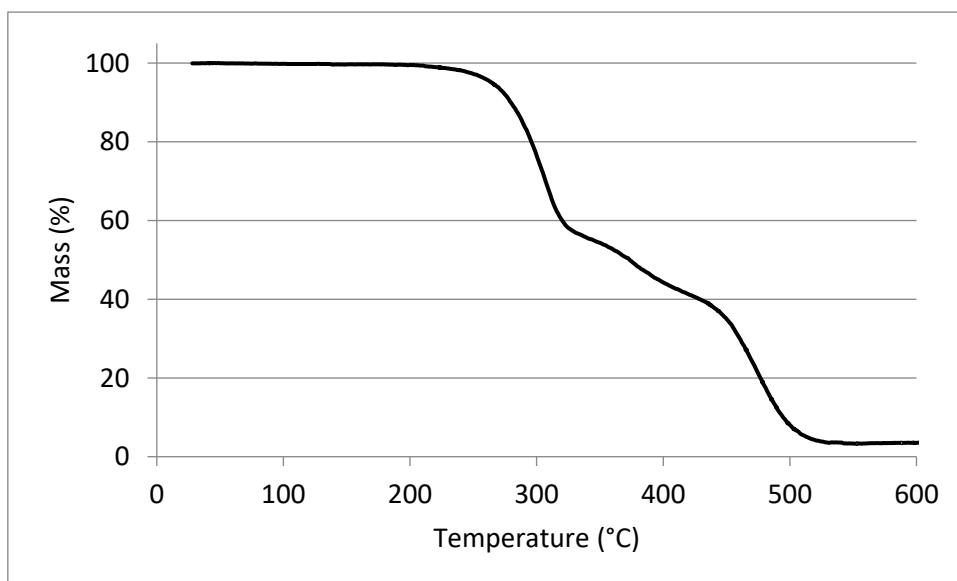
187  
188

TGA of Th-Thd-Th(6):



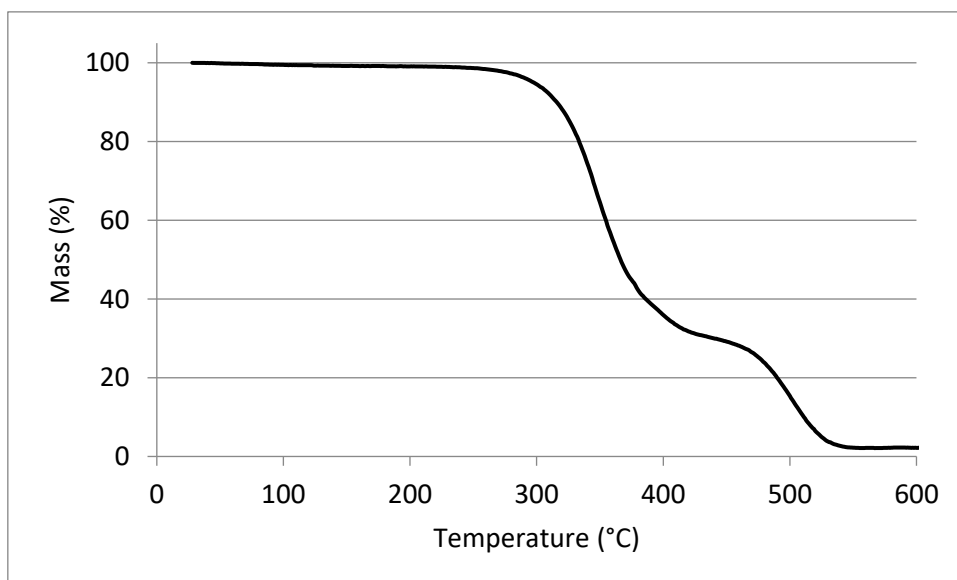
189

190 TGA of Th-Thd-Th(8):



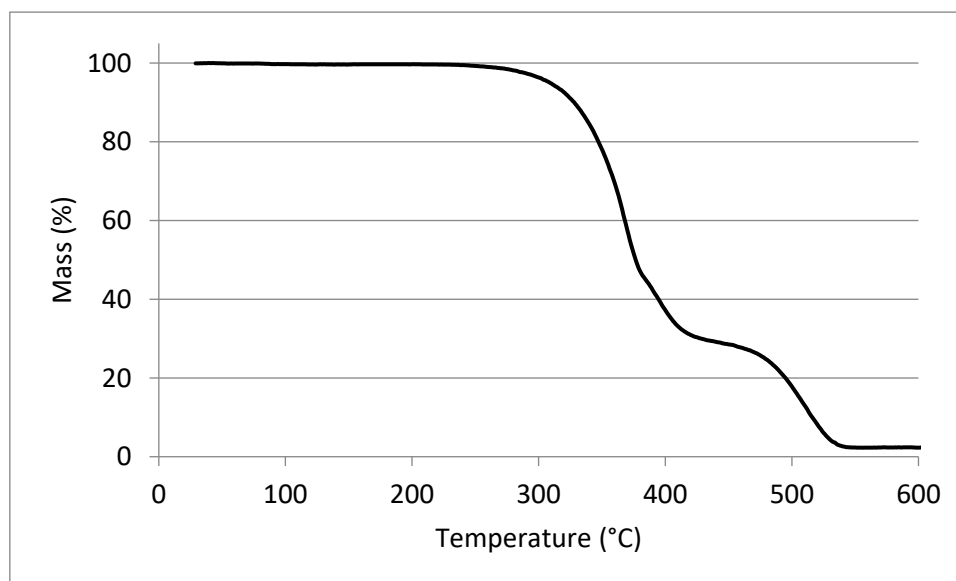
191

192 TGA of Th-Thd-Th(10):



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194 TGA of Th-Thd-Th(12):



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## 196 6. Single Crystal X-ray Diffraction

197 Suitable crystals (**Th-Thd-Th**, **Th-Thd-Th(4)** and **Th-Thd-Th(10)**) were coated in paratone oil,  
 198 mounted on a MiTeGen Micro Mount, and transferred to the the X-ray diffractometer. Data was  
 199 collected on a Bruker Smart instrument equipped with an APEX II CCD area detector fixed at a  
 200 distance of 5.0 cm (**Th-Thd-Th**, **Th-Thd-Th(4)**) or 4.0 cm (**Th-Thd-Th(10)**), from the crystal and a Cu  
 201  $K\alpha$  fine focus sealed tube ( $\lambda = 1.54178 \text{ \AA}$ ) operated at 1.5 kW (45 kV, 0.65 mA), filtered with a graphite  
 202 monochromator. Data was collected at 150K (**Th-Thd-Th(4)**, **Th-Thd-Th(10)**) or 296K (**Th-Thd-Th**);  
 203 the temperature was regulated using an Oxford Cryosystems Cryostream. Additional  
 204 crystallographic information can be found in Table S4. All diffraction data were processed using the  
 205 Bruker SAINT software package and were corrected for absorption effects using the multi-scan  
 206 technique (SADABS [3] or TWINABS [4]). The structures were solved with direct methods (SIR92)  
 207 and subsequent refinements were performed using SHELXL [5] and ShelXle. [6] Hydrogen atoms on  
 208 carbon atoms were included at geometrically idealized positions (C–H bond distance 0.95Å) and  
 209 were not refined. The isotropic thermal parameters of the hydrogen atoms were fixed at 1.2 times  
 210 that of the preceding carbon atom. Diagrams were prepared using ORTEP-3 [7] and POV-RAY. [8]  
 211 Thermal ellipsoids are shown at the 50% probability level.

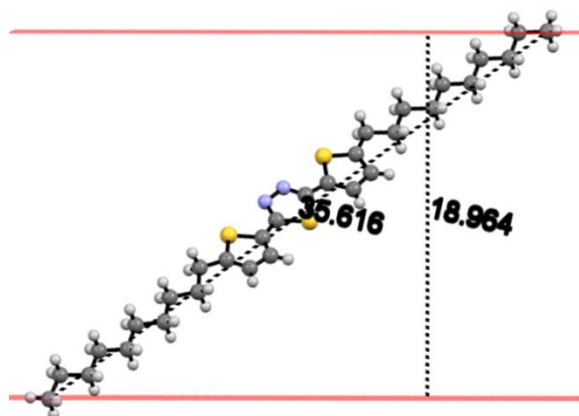
212

**Table S4.** Crystallographic information table for **Th-Thd-Th**, **Th-Thd-Th(4)**, and **Th-Thd-Th(10)**.

Compound	<b>Th-Thd-Th</b>	<b>Th-Thd-Th(4)</b>	<b>Th-Thd-Th(10)</b>
Chemical Formula	$C_{10}H_6N_2S_3$	$C_{18}H_{22}N_2S_3$	$C_{30}H_{46}N_2S_3$
Formula Mass	250.35	362.55	530.87
$a/\text{\AA}$	6.34810(10)	7.3146(4)	7.5511(11)
$b/\text{\AA}$	9.63630(10)	7.6299(3)	9.6532(16)
$c/\text{\AA}$	17.8284(2)	18.0456(9)	23.523(4)
$\alpha/^\circ$	90	85.306(4)	91.278(8)
$\beta/^\circ$	94.9750(10)	84.800(4)	94.604(7)
$\gamma/^\circ$	90	67.452(4)	100.468(9)
Unit cell volume/ $\text{\AA}^3$	1086.49(2)	925.09(8)	1679.5(5)
Temperature/K	159(2)	150(2)	150(2)
Space group	P21/c	P-1	P-1

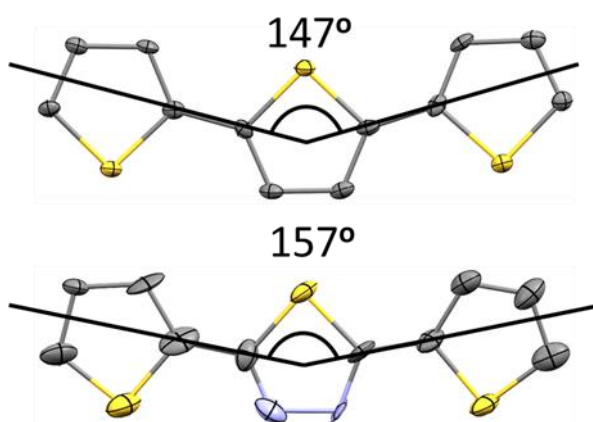
Number of formula unit per cell/Z	4	2	2
Radiation type	Cu K $\alpha$	Cu K $\alpha$	Cu K $\alpha$
Absorption coefficient, $\mu/\text{mm}^{-1}$	5.949	3.651	2.142
No. of reflections collected	5857	9288	17667
No. unique reflections	1922	3161	5526
$R_{\text{int}}$	0.0288	0.0693	0.0630
Final $R_1$ values ( $I > 2\sigma(I)$ )	0.0449	0.0867	0.2077
Final $wR(F^2)$ values ( $I > 2\sigma(I)$ )	0.1299	0.2800	0.5868
Final $R_1$ values (all data)	0.0475	0.1088	0.2178
Final $wR(F^2)$ (all data)	0.1331	0.2965	0.6060
Goodness of fit	1.080	1.161	3.232

213 All crystal structure figures below follow the legend: sulphur in yellow, nitrogen in light blue,  
214 carbon in grey, hydrogen in white.



215

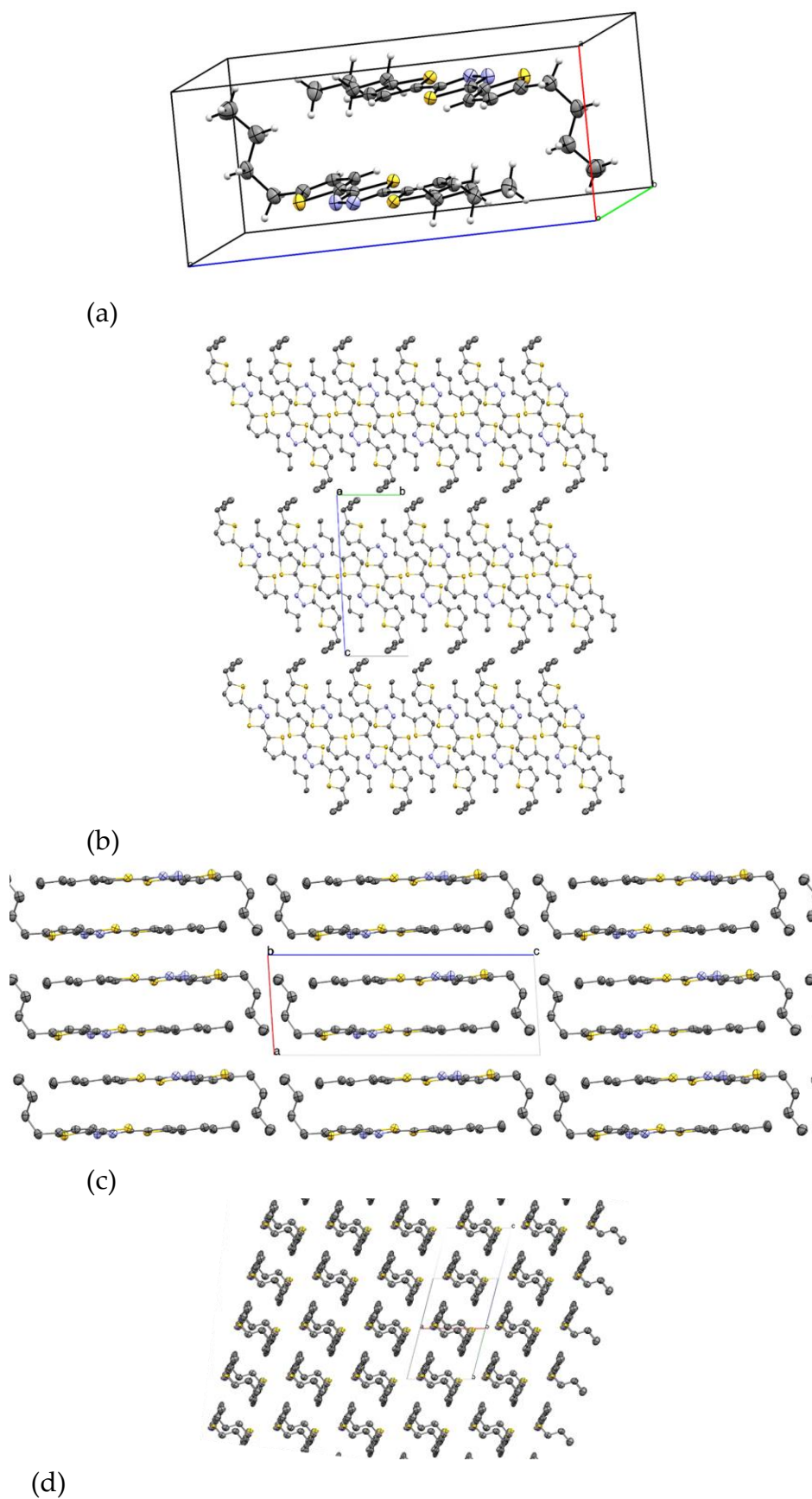
216 **Figure S19.** Estimation of the tilt angle in the room temperature smectic phase of **Th-Thd-Th(10)**  
217 based on the single crystal structure.



218

219 **Figure S20.** Bond angles at the 2 and 5 positions of the central heterocycle for **Th<sub>3</sub>(8)[9]** and **Th-Thd-**  
220 **Th(10)**. Alkyl chains and hydrogens omitted for clarity.

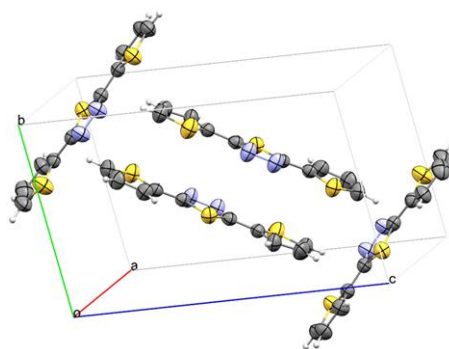




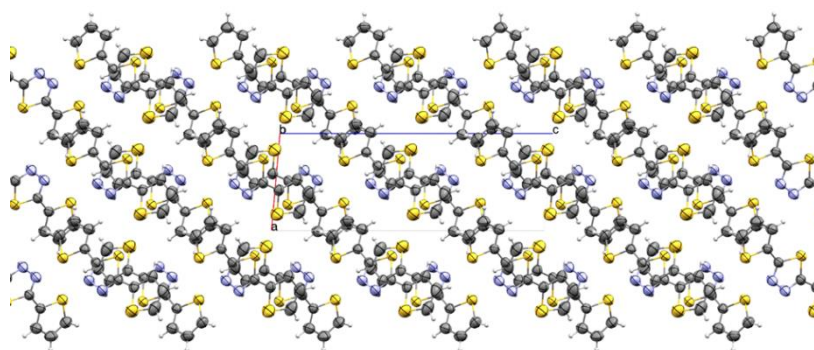
221  
222

**Figure S21.** Single crystal structure of Th-Thd-Th(4) showing the unit cell (a), lamellar order looking down the a- or b- axis (b and c respectively), and intra-layer packing looking down the c-axis (d).

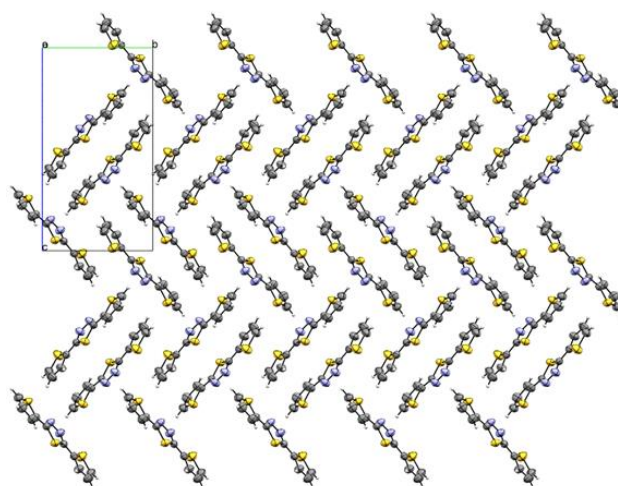
a)



b)

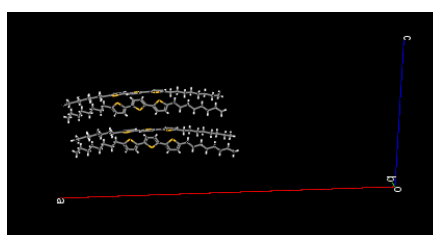


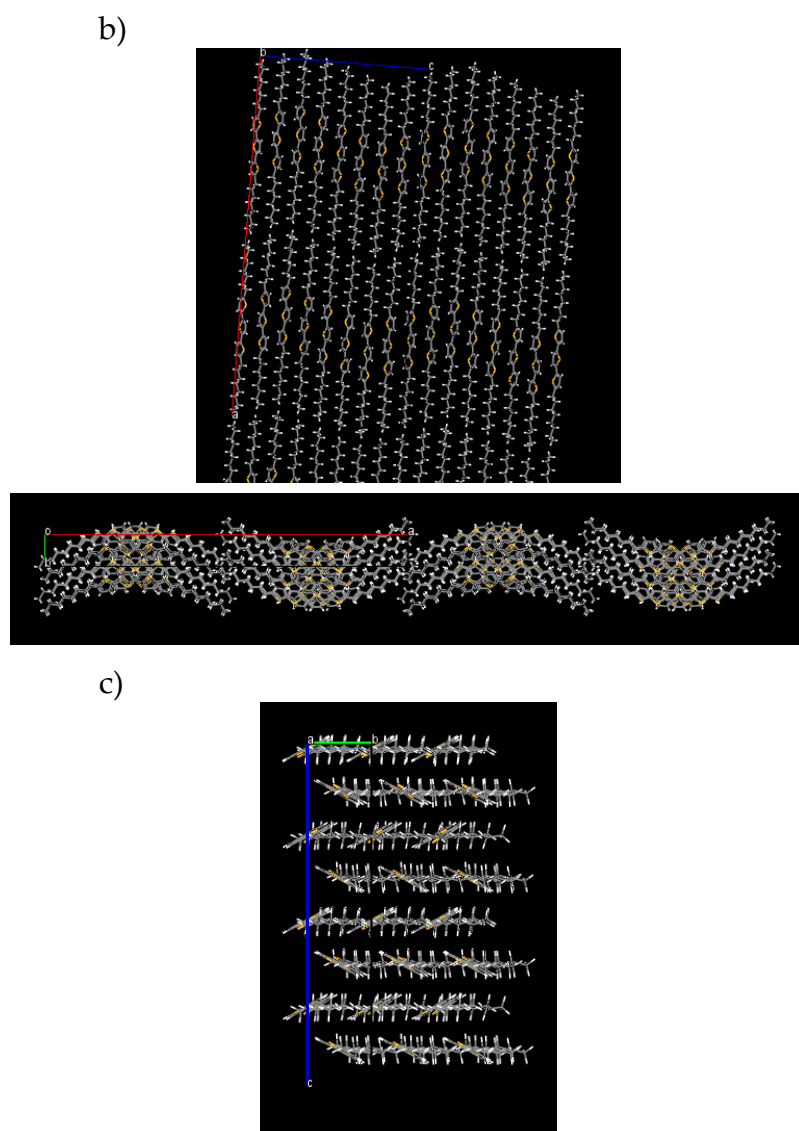
c)

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**Figure S22.** Single crystal structures of **Th-Thd-Th** showing the unit cell (a), order looking down the b-axis (b), and herringbone packing looking down the a-axis (c).

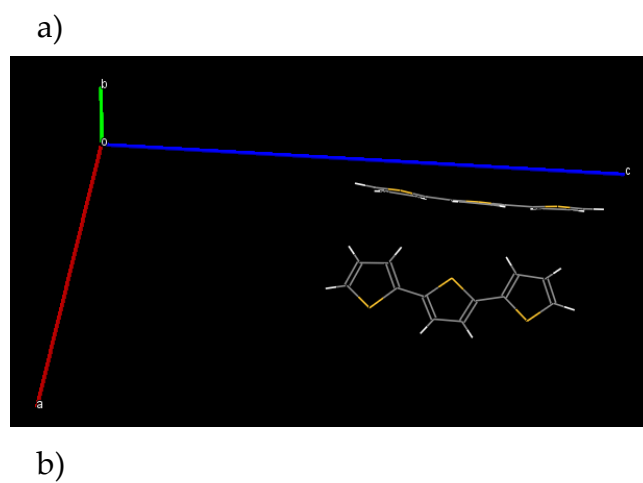
a)

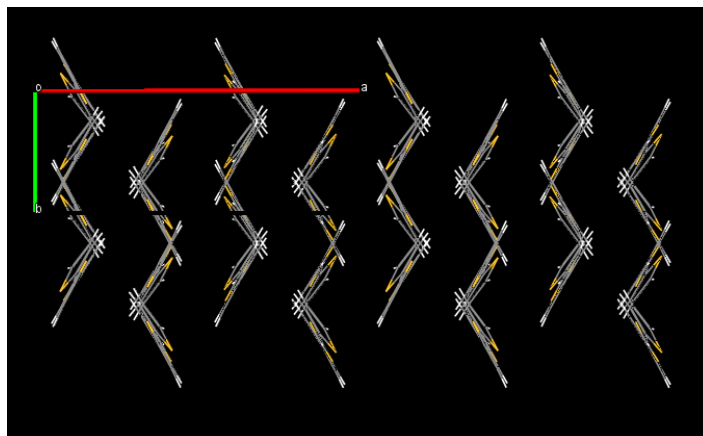




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226

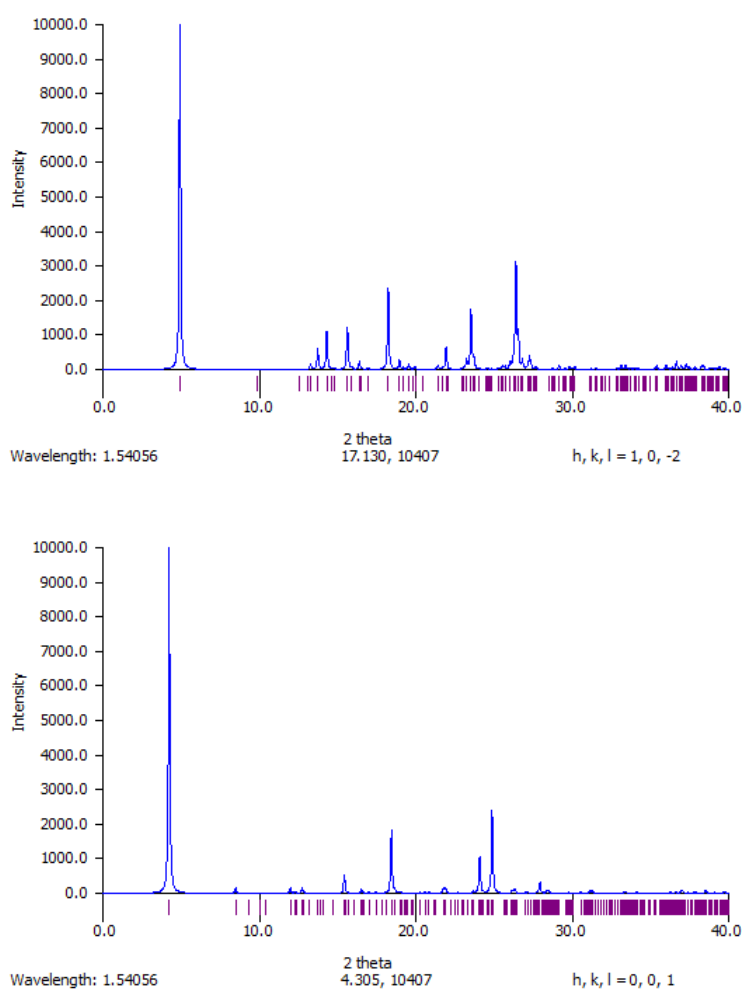
**Figure S23.** Literature single crystal structures of **Th<sub>3</sub>(8)[9]** showing the unit cell (a), lamellar order (b), and intra-layer packing (c).





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**Figure S24.** Literature single crystal structures of Th<sub>3</sub>[10] showing the unit cell (a) and herringbone packing (b).



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**Figure S25.** Simulated powder XRD patterns of Th-Thd-Th(4) (a) and Th-Thd-Th(10) (b) as determined based on the single crystal structure.

## 231 7. UV/vis Absorption Spectroscopy

232

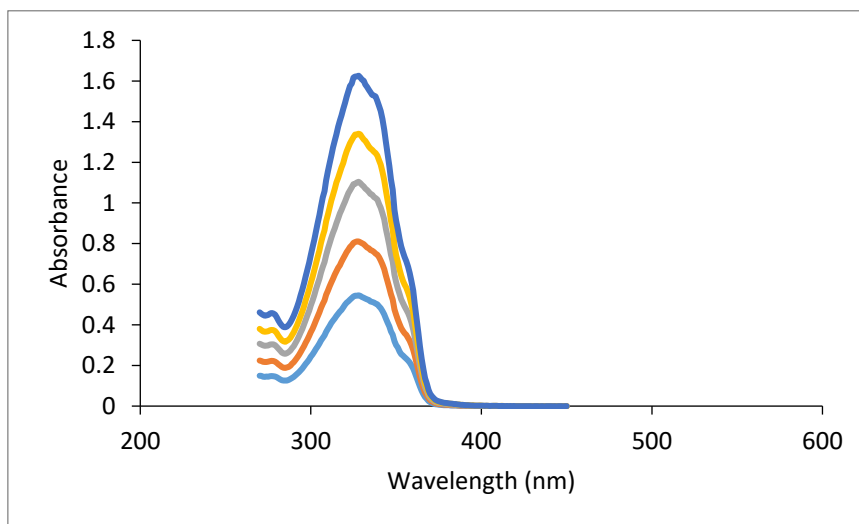
**Table S5.** Summary of UV-Vis spectral properties.

Compound	$\lambda_{abs}^{max}$ (nm)	$\epsilon$ ( $M^{-1}cm^{-1}$ )	$\lambda_{abs}^{onset}$ (nm)	$E_{g,opt}$ (eV) <sup>a</sup>
----------	-------------------------------	-----------------------------------	---------------------------------	----------------------------------

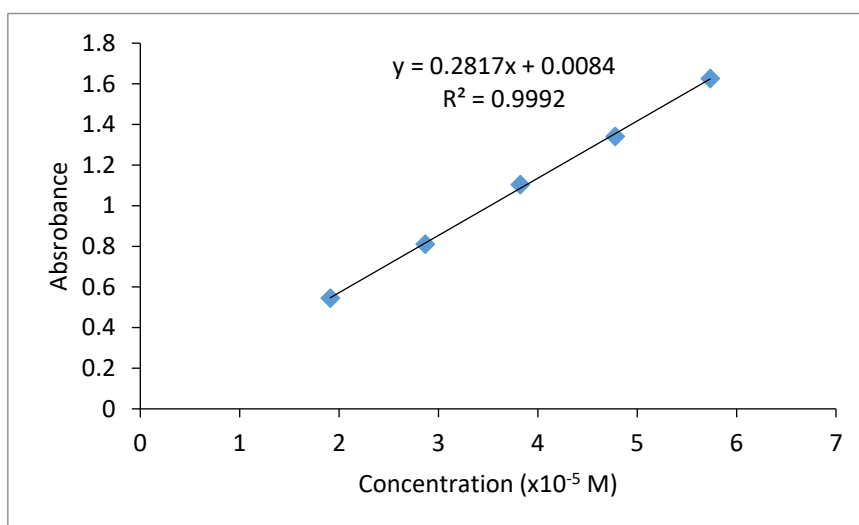
<b>Th-Oxd-Th(10)</b>	328	28200	371	3.35
<b>Th-Thd-Th(10)</b>	359	26800	409	3.04
<b>Th<sub>3</sub>(10)</b>	367	25900	423	2.93

233 <sup>a</sup>Optical bandgap, estimated by  $E_g = hc/\lambda_{abs}^{onset}$

a)



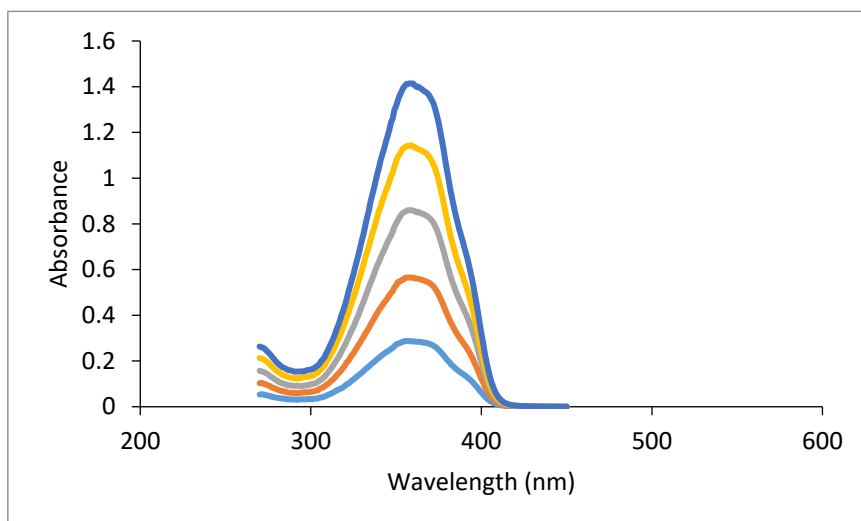
b)



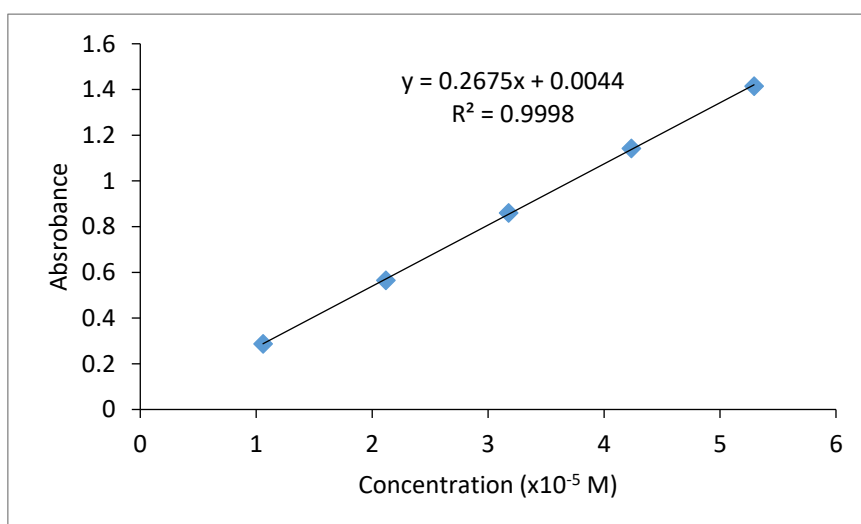
234  
235

**Figure S26.** Absorption spectra of **Th-Oxd-Th(10)** in chloroform at various concentrations ranging from around  $6 \times 10^{-5}$  to  $2 \times 10^{-5}$  M (top to bottom) (a) and the resulting Beer-Lambert plot (b).

a)



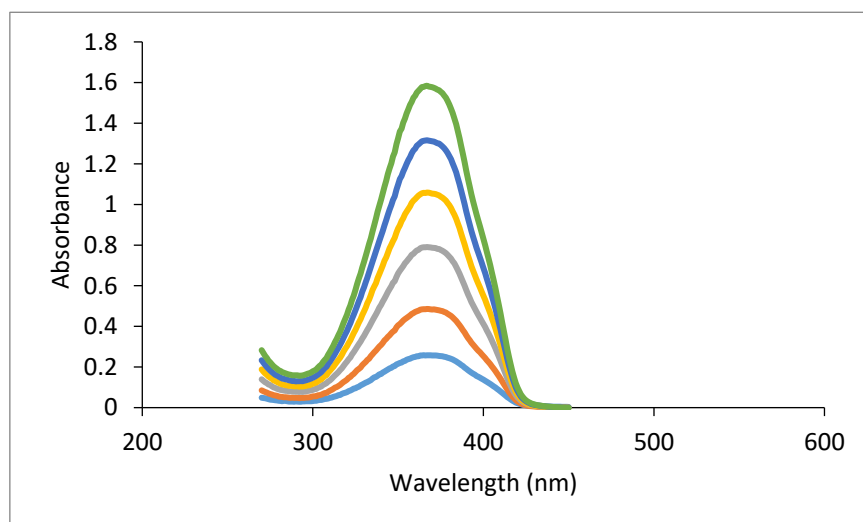
b)



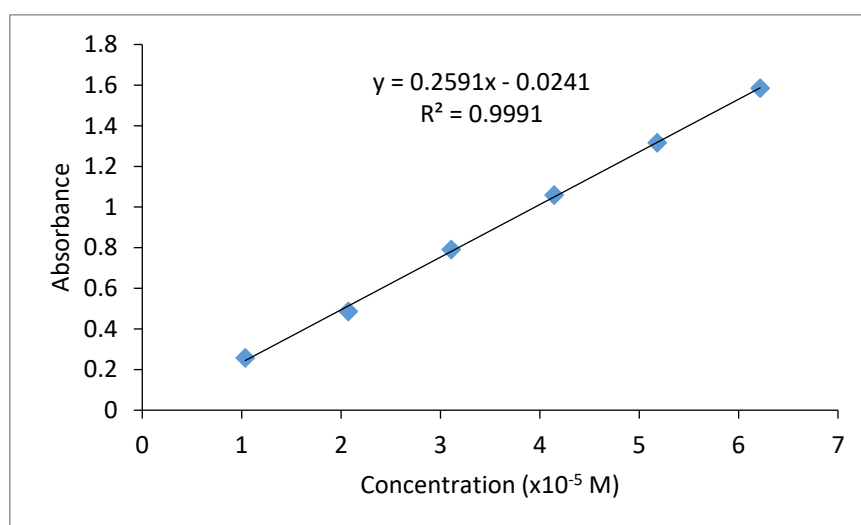
236  
237

**Figure S27.** Absorption spectra of **Th-Thd-Th(10)** in chloroform at various concentrations ranging from around  $5 \times 10^{-5}$  to  $1 \times 10^{-5}$  M (top to bottom) (a) and the resulting Beer-Lambert plot (b).

a)



b)



238 **Figure S28.** Absorption spectra of Th<sub>3</sub>(10) in chloroform at various concentrations ranging from  
 239 around  $6 \times 10^{-5}$  to  $1 \times 10^{-5}$  M (top to bottom) (a) and the resulting Beer-Lambert plot (b).

## 240 8. Fluorescence Excitation/Emission Spectroscopy

241

**Table S6.** Summary of fluorescence spectral properties.

Compound	$\lambda_{ex}^{max}$ (nm) <sup>a</sup>	$\lambda_{em}^{max}$ (nm) <sup>b</sup>	Stokes shift (nm)	$\Phi_f^c$
Th-Oxd-Th(10)	340	387	47	0.48
Th-Thd-Th(10)	369	430	61	0.49
Th <sub>3</sub> (10)	380	443	63	0.56

242 <sup>a</sup> Observed at the corresponding  $\lambda_{em}^{max}$ ; <sup>b</sup> Measured at the respective  $\lambda_{ex}^{max}$ ; <sup>c</sup> Relative to quinine  
 243 sulfate standard ( $\Phi_f = 0.54$ ) in 0.1 M H<sub>2</sub>SO<sub>4</sub>

244

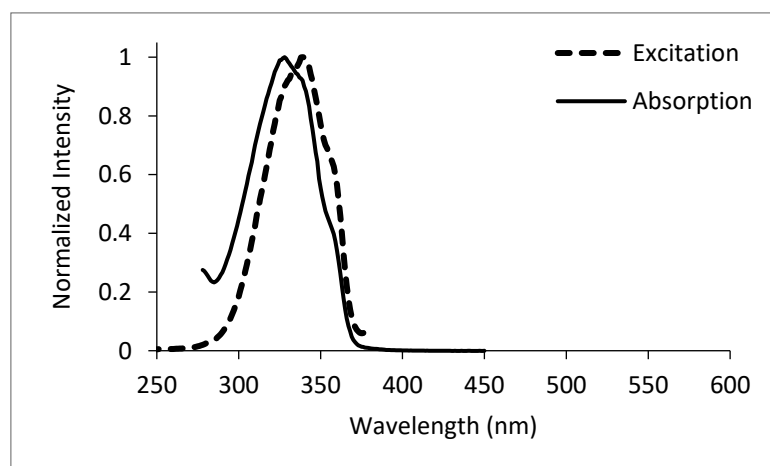
**Table S7.** Detailed fluorescence data used for quantum yield determination.

Compound	Concentration (x10 <sup>-6</sup> M)	Maximum absorbance, A <sub>max</sub>	Integrated emission intensity, I <sub>em</sub>	I <sub>em</sub> /A <sub>max</sub> gradient <sup>a</sup>	$\Phi_f^b$
Th-Oxd-Th(10)	0.96	552038	43751369	74.9	0.48
	1.91	898329	69736325		
	2.87	1235771	94299072		
	3.82	1567427	119646050		
	4.78	1915589	145914659		
Th-Thd-Th(10)	2.12	275332	21930717	76.4	0.49
	6.35	559616	43049313		

	8.47	696109	53070730		
	10.59	828772	63336304		
	12.70	983655	75462409		
	16.94	1258332	97024851		
<b>Th<sub>3</sub>(10)</b>	0.52	376360	38399486		
	0.62	423623	42422719		
	0.73	526075	51875681	87.1	0.56
	0.83	599034	58557104		
	0.93	685502	64905104		
Quinine sulfate	6.65	622805	59563561		
	8.87	792582	76626907		
	11.09	985531	93600171		
	13.30	1151426	110217780	98.2	0.54
	15.52	1305979	126343550		
	17.74	1459479	142261748		

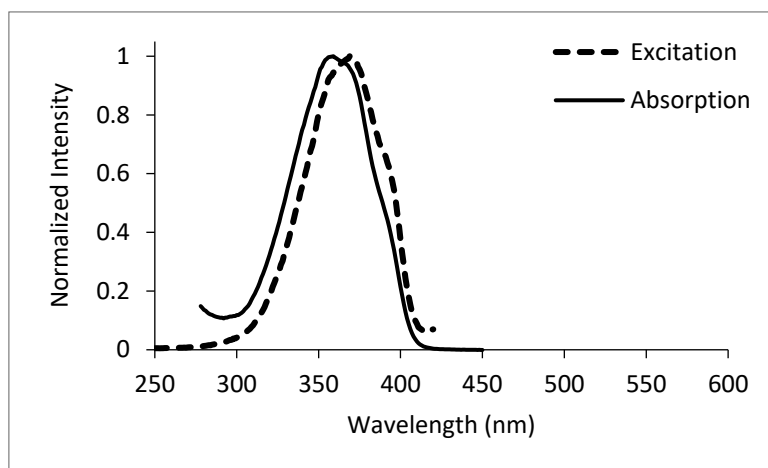
245 <sup>a</sup> Determined from slope of  $I_{em}$  vs.  $A_{max}$  at various concentrations, see plots below; <sup>b</sup>  
 246 Calculated according to:  $\Phi_{f,x} = \Phi_{f,ref} (I_{em,x}/A_{max,x}) (A_{max,ref}/I_{em,ref}) (\eta_x/\eta_{ref})^2$ , [11] using the  
 247 quinine sulfate standard ( $\Phi_f = 0.54$ ) [12] in 0.1 M  $H_2SO_4$ , the gradients above determined  
 248 from experiment, and  $\eta_{chloroform} = 1.444$  and  $\eta_{water} = 1.333$ .

a)

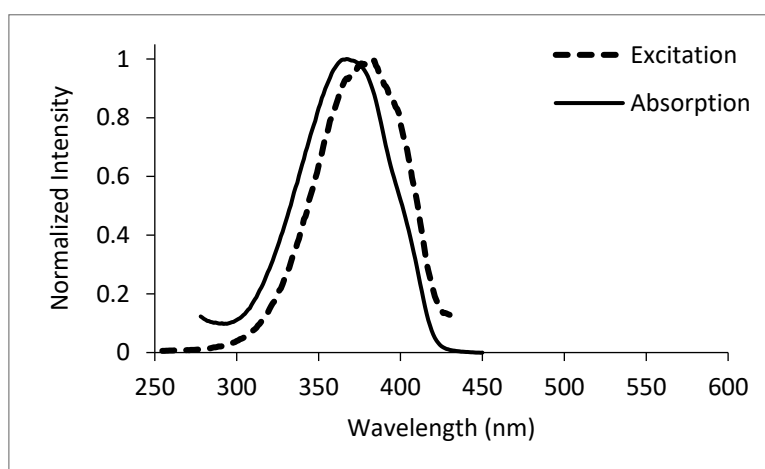


b)





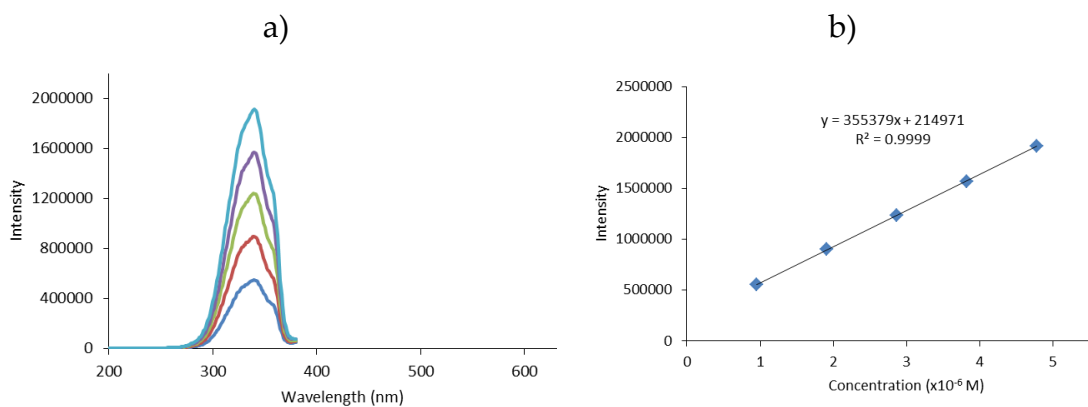
c)

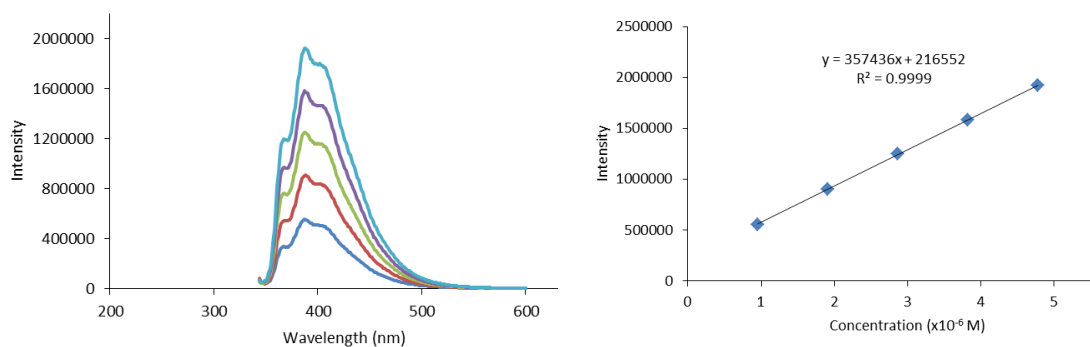


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251

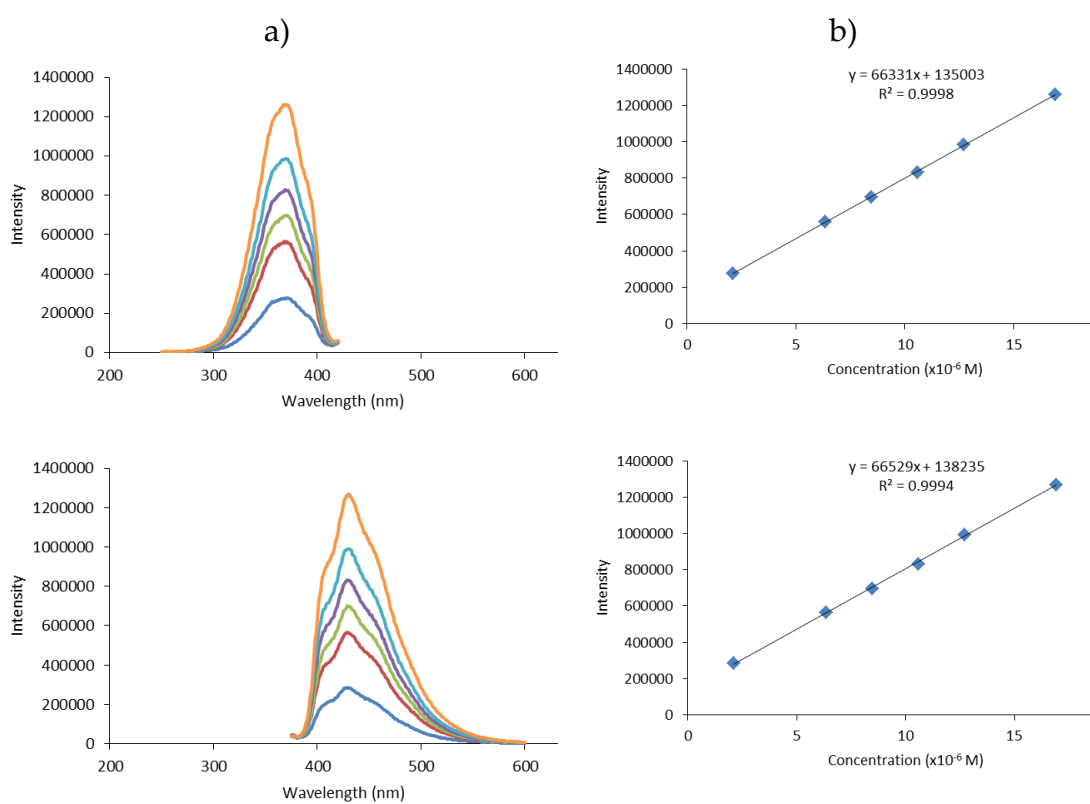
**Figure S29.** Overlay of UV/vis absorption (solid line) and fluorescence excitation (dashed line) spectra of Th-Oxd-Th(10) (a), Th-Thd-Th(10) (b), and Th<sub>3</sub>(10) (c) in CHCl<sub>3</sub>.





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**Figure S30.** Fluorescence excitation (top) and emission (bottom) spectra of **Th-Oxd-Th(10)** in chloroform at various concentrations ranging from around  $5 \times 10^{-6}$  to  $1 \times 10^{-6}$  M (shown in descending order) (a) and the respective Beer-Lambert plots demonstrating the linear range (b).

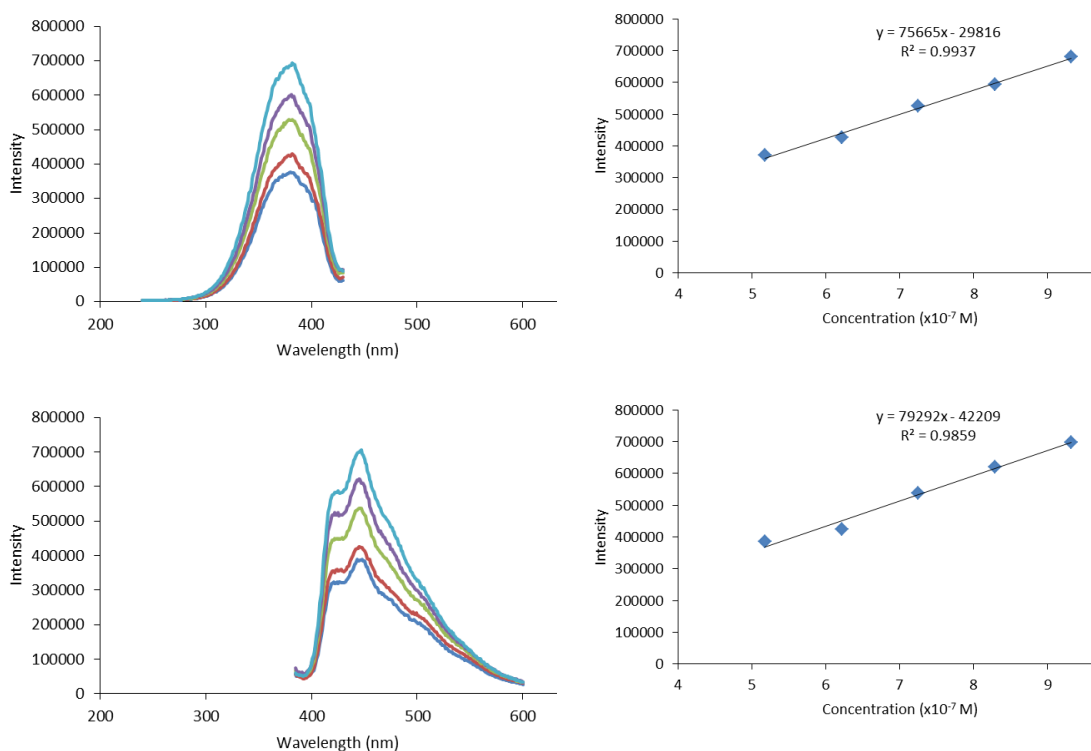


255  
256  
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**Figure S31.** Fluorescence excitation (top) and emission (bottom) spectra of **Th-Thd-Th(10)** in chloroform at various concentrations ranging from around  $16 \times 10^{-6}$  to  $2 \times 10^{-6}$  M (shown in descending order) (a) and the respective Beer-Lambert plots demonstrating the linear range (b).

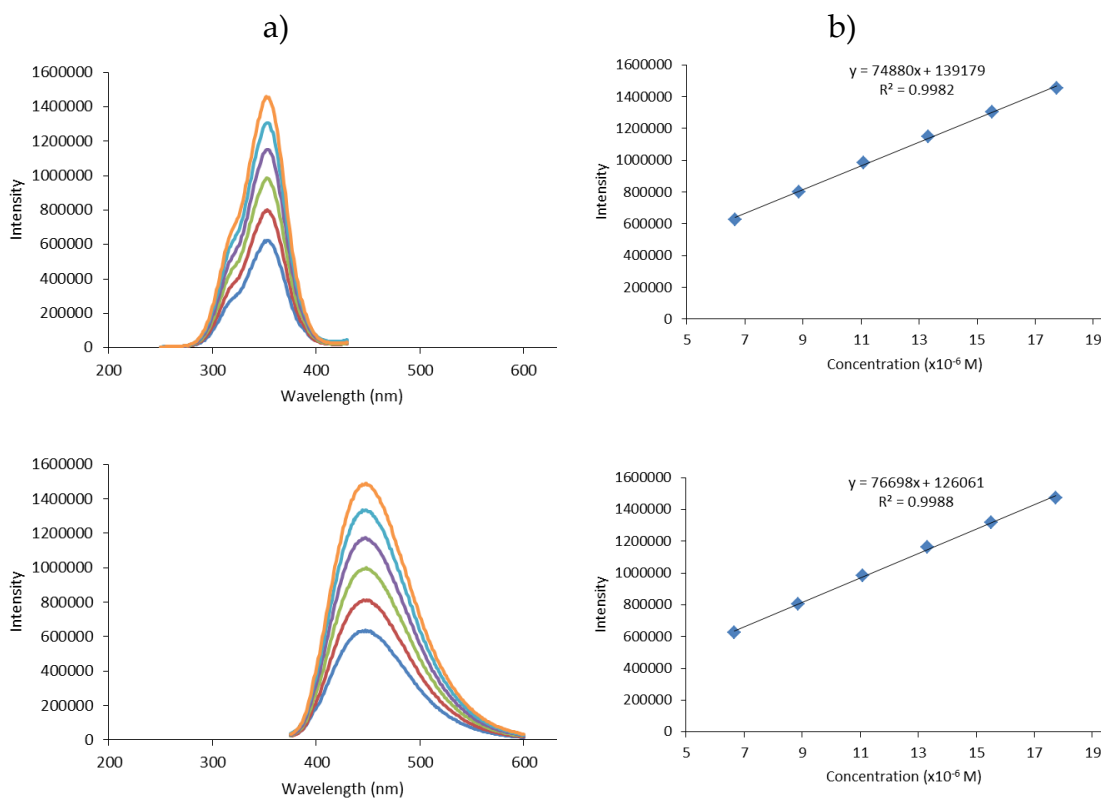
a)

b)



258  
259  
260

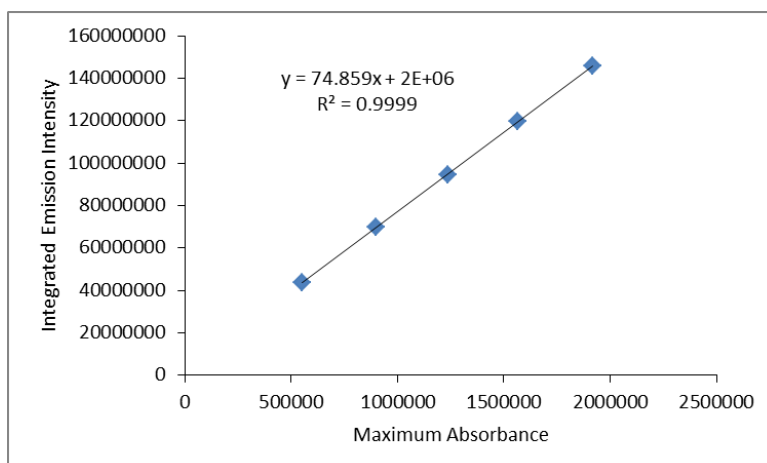
**Figure S32.** Fluorescence excitation (top) and emission (bottom) spectra of Th<sub>3</sub>(10) in chloroform at various concentrations ranging from around  $9 \times 10^{-7}$  to  $5 \times 10^{-7}$  M (shown in descending order) (a) and the respective Beer-Lambert plots demonstrating the linear range (b).



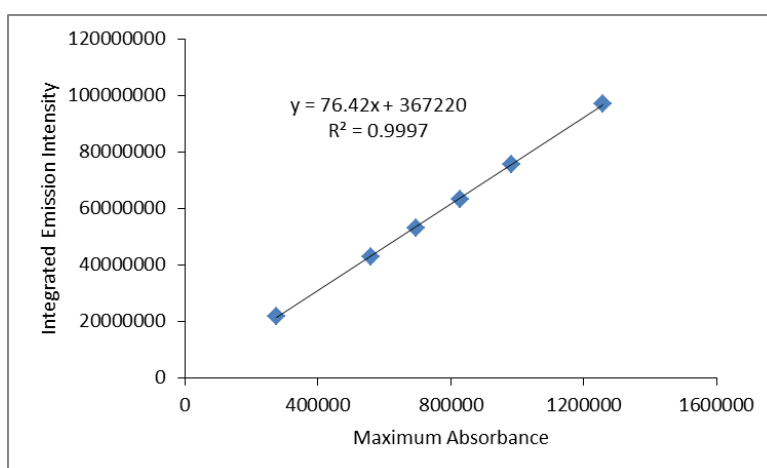
261  
262  
263

**Figure S33.** Fluorescence excitation (top) and emission (bottom) spectra of quinine sulfate in 0.1 M sulfuric acid at various concentrations ranging from around  $17 \times 10^{-6}$  to  $6 \times 10^{-6}$  M (shown in descending order) (a) and the respective Beer-Lambert plots demonstrating the linear range (b).

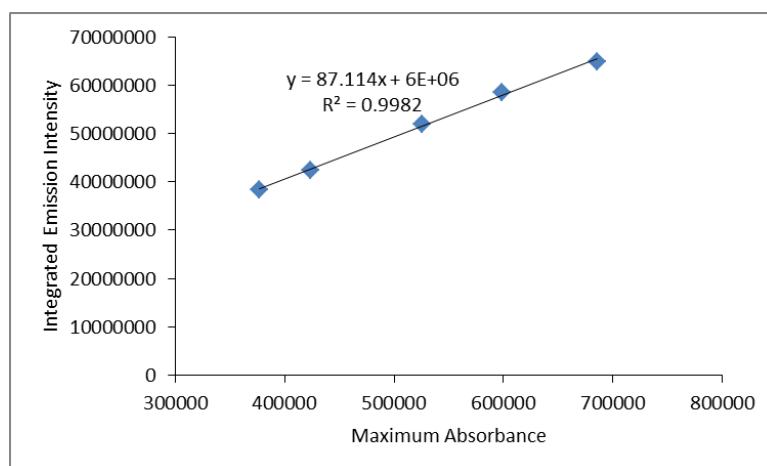
a)



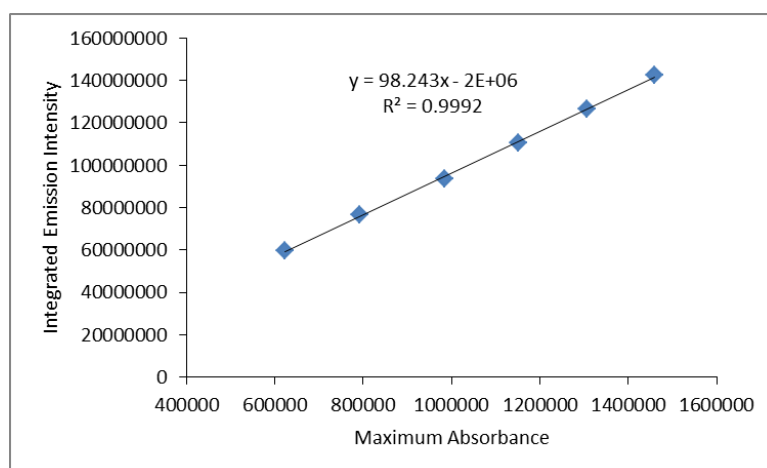
b)



c)



d)



264 **Figure S34.** Integrated emission intensity vs. maximum absorbance plots created based on the  
 265 fluorescence spectra at various concentrations shown above for **Th-Oxd-Th(10)** (a), **Th-Thd-Th(10)**  
 266 (b), **Th<sub>3</sub>(10)** (c), and quinine sulfate (d).

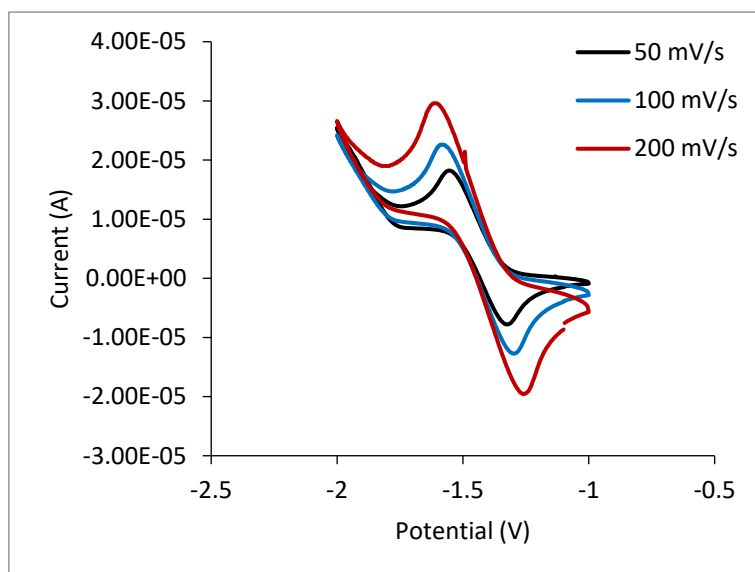
### 267 9. Cyclic Voltammetry (CV)

268 Cyclic voltammetry scans were performed on a PAR-263A potentiometer. All analytes were  
 269 scanned as 1.0 mM solutions in THF (**Th-Ox-Th(10)** and **Th-Thd-Th(6)** reduction) or MeCN (**Th-**  
 270 **Thd-Th(6)** oxidation) with a 0.1 M tetrabutylammonium perchlorate supporting electrolyte. The  
 271 apparatus was a 20 ml glass cell equipped with a glassy carbon working electrode, platinum counter  
 272 electrode and silver wire reference electrode. All scans were reference internally to a Fc/Fc<sup>+</sup> reversible  
 273 redox couple.

274 **Table S8.** Summary of the raw CV data for select compounds.

Compound	Process/ solvent	Analyte (compound)			Ferrocene reference		
		E <sub>a</sub> (V)	E <sub>c</sub> (V)	E <sup>1/2</sup> (V) <sup>a</sup>	E <sub>a</sub> (V)	E <sub>c</sub> (V)	E <sup>1/2</sup> (V)
<b>Th-Oxd-Th(10)</b>	Reduction/ THF	n/a, irreversible			1.29	0.53	0.91
<b>Th-Thd-Th(6)</b>	Reduction/ THF	-1.32	-1.64	-1.48	1.35	0.50	0.93
	Oxidation/ MeCN	n/a, irreversible			0.92	0.56	0.74

275 <sup>a</sup> Irreversible processes estimated based on middle of onset slope

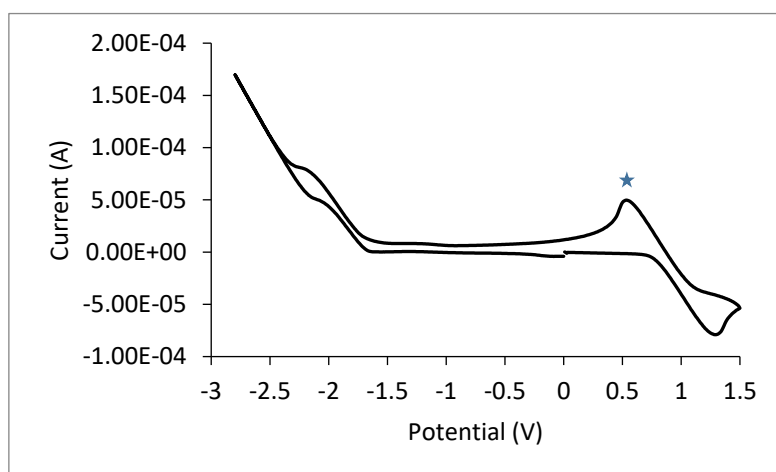


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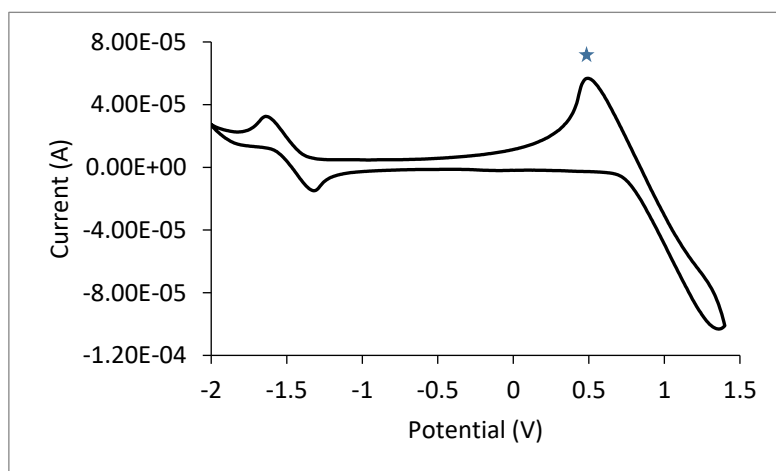
277  
278

**Figure S35.** Unprocessed cyclic voltammograms showing the reversible reduction of **Th-Thd-Th(6)** at different scan rates.

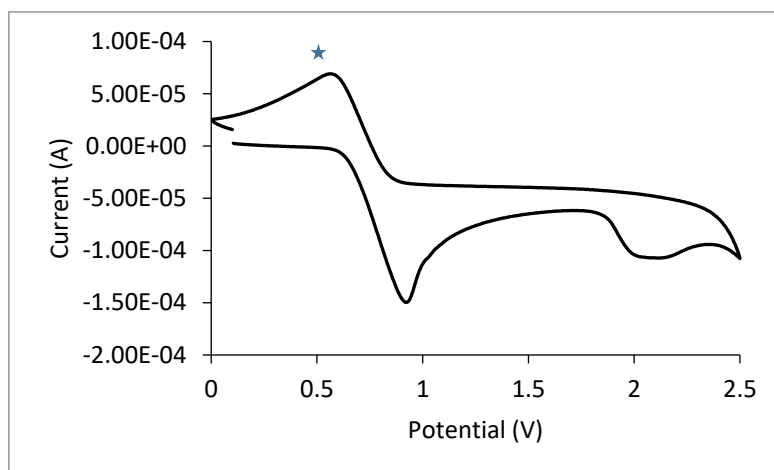
a)



b)



c)



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**Figure S36.** Unprocessed cyclic voltammograms showing the ferrocene internal standard signal (blue star) used for reference for **Th-Oxd-Th(10)** reduction in THF (a), **Th-Thd-Th(6)** reduction in THF (b), and **Th-Thd-Th(6)** oxidation in MeCN (c).

## 283 10. DFT and TD-DFT Calculations

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*Density functional theory (DFT)* – DFT calculations were carried out using B3LYP functional[13,14] and 6-31G\* basis set (C, H, N, O, S)[15–18] in Gaussian 16[19] to determine the optimized molecular structures (Figure 6) as well as the frontier molecular orbitals and respective HOMO/LUMO energy levels (Figure S36). All of these results were consistent with those obtained by different basis sets used during TD-DFT calculations (see following).

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*Time-dependent density functional theory (TD-DFT)* – Geometry optimization calculations were completed using the Gaussian 16 program (Revision B.01)[19], the restricted B3LYP functional[13,14], the 6-31G\* basis set (C, H, N, O, S), with a polarized continuum model (PCM) for CHCl<sub>3</sub> (dielectric  $\epsilon = 4.9$ ).[15–18] Frequency calculations at the same level of theory confirmed that the optimized structures were located at a minimum on the potential energy surface. Single-point calculations were done using restricted B3LYP functional, the TZVP basis set (C, H, N, O, S)[20,21], with a PCM for CHCl<sub>3</sub>. The intensities of 30 lowest energy transitions using TD-DFT[22,23] calculations were performed using the CAM-B3LYP functional[24–26], the TZVP basis set (C, H, N, O, S), with a PCM for CHCl<sub>3</sub>.

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**Table S9.** Summary of predicted properties determined by DFT (B3LYP/6-31G\*).

Compound	Gas phase			Solvated in THF		
	HOMO (eV)	LUMO (eV)	$E_g^{\text{calc}}$ (eV)	HOMO (eV)	LUMO (eV)	$E_g^{\text{calc}}$ (eV)
<b>Th<sub>3</sub>(10)</b>	-4.88	-1.51	3.37	-5.05	-1.69	3.36
<b>Th-Oxd-Th(10)</b>	-5.59	-1.62	3.97	-5.77	-1.80	3.97
<b>Th-Thd-Th(10)</b>	-5.54	-1.90	3.64	-5.71	-2.08	3.62
<b>Th-Thd-Th(6)</b>	-5.54	-1.90	3.64	-5.71	-2.09	3.62

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**Table S10.** Summary of predicted properties determined by TD-DFT (CAM-B3LYP/TZVP).

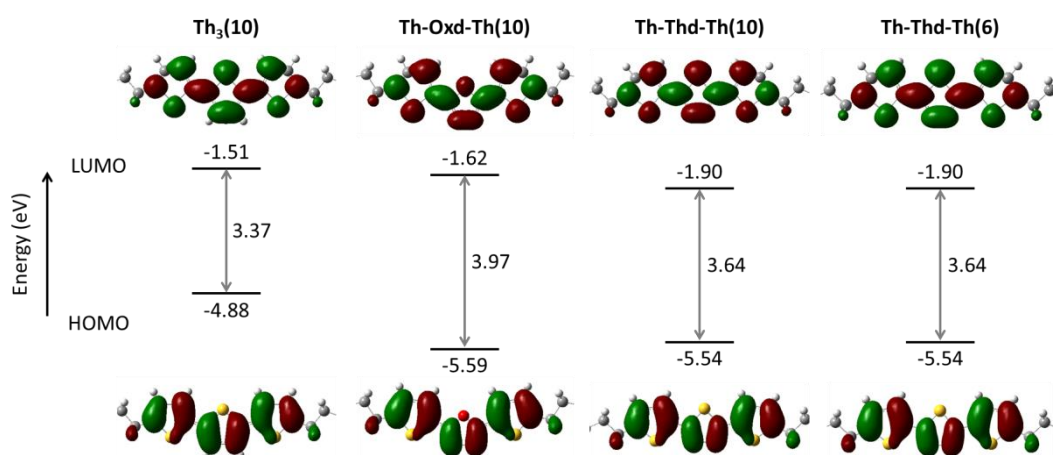
Compound	TD-DFT
----------	--------

	$\lambda_{abs}^{max}$ (nm)	$f^a$	$E_g^{opt}$ (eV) <sup>b</sup>
<b>Th<sub>3</sub>(10)</b>	383	1.15	2.67
<b>Th-Oxd-Th(10)</b>	323	1.16	3.26
<b>Th-Thd-Th(10)</b>	354	1.18	2.94

<sup>a</sup>Oscillator strength

<sup>b</sup>Optical bandgap, estimated by  $E_g = hc/\lambda_{abs}^{onset}$

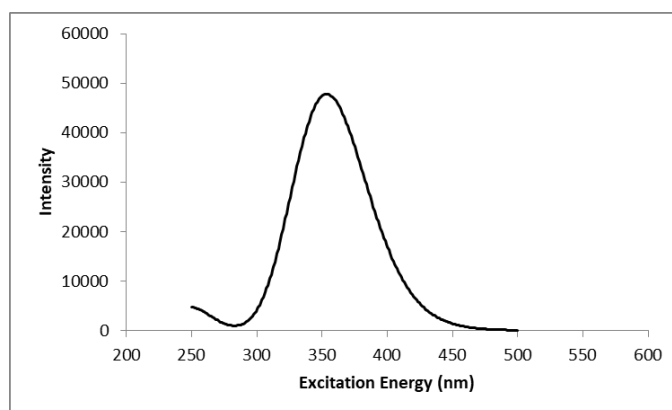
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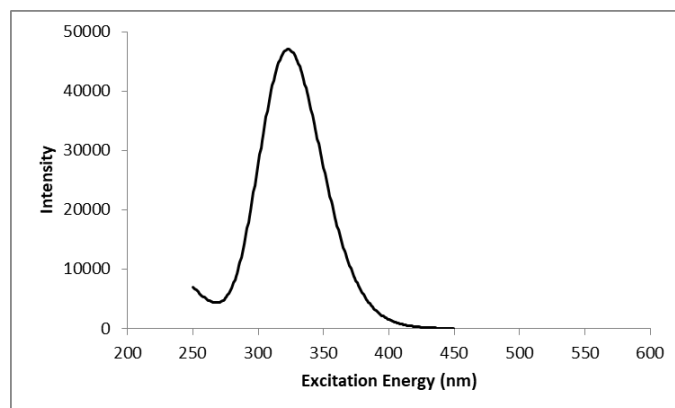
**Figure S37.** Frontier molecular orbitals of select compounds calculated by DFT at B3LYP/6-31G\* level.

a)

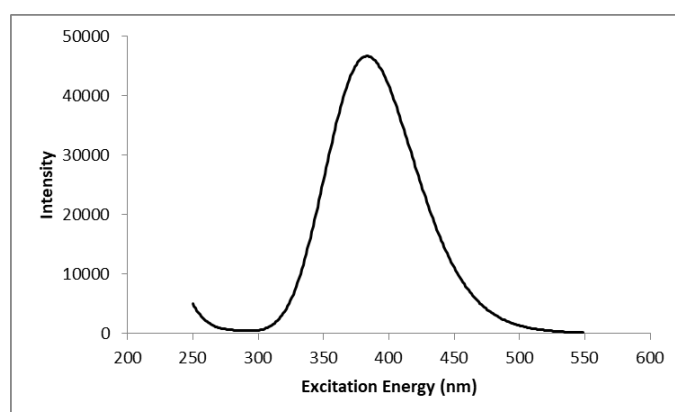


b)





c)



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307 **Figure S38.** TD-DFT predicted UV/vis spectra in chloroform of **Th<sub>3</sub>(10)** (a), **Th-Oxd-Th(10)** (b), and  
 308 **Th-Thd-Th(10)** (c).

### 309 11. Complete List of Computational Data

#### 310 Optimized coordinates (Å) for Th-Oxd-Th(10):

311	C	1.06861315	-2.08694106	0.00077733
312	C	-1.06861352	-2.08694115	0.00071126
313	C	2.39766853	-1.54132287	-0.00139990
314	C	2.77436144	-0.21785222	-0.00711548
315	S	3.79545595	-2.59381429	0.00380009
316	C	4.18572883	-0.04510711	-0.00774990
317	H	2.06185720	0.59918123	-0.01079215
318	C	4.88823187	-1.22612957	-0.00253064
319	H	4.66347744	0.92751470	-0.01198087
320	C	-2.39766880	-1.54132337	-0.00161983
321	C	-2.77436145	-0.21785256	-0.00731636
322	S	-3.79545651	-2.59381625	0.00318546
323	C	-4.18572881	-0.04510757	-0.00804950
324	H	-2.06185703	0.59918136	-0.01084878
325	C	-4.88823212	-1.22613025	-0.00291490
326	H	-4.66347720	0.92751460	-0.01222002
327	N	0.69495173	-3.33705542	0.00614971
328	N	-0.69495230	-3.33705639	0.00589109
329	C	6.37730125	-1.45485264	-0.00210701
330	H	6.64913616	-2.06181825	0.87321688

331	H	6.64960845	-2.06146371	-0.87757292
332	C	7.21222653	-0.16644298	-0.00169072
333	H	6.95424034	0.43684545	-0.88254785
334	H	6.95256581	0.43731071	0.87835320
335	C	8.72046191	-0.44679165	-0.00034091
336	H	8.97778842	-1.05622778	-0.87899553
337	H	8.97639213	-1.05525476	0.87939382
338	C	9.57263471	0.82886404	-0.00039303
339	H	9.31489655	1.43723484	-0.87997259
340	H	9.31352133	1.43825006	0.87807735
341	C	11.08204180	0.55434118	0.00093129
342	H	11.34045757	-0.05583227	-0.87730798
343	H	11.33914876	-0.05461481	0.88039787
344	C	11.93592221	1.82897310	0.00067000
345	H	11.67705214	2.43912377	0.87889034
346	H	11.67831709	2.43789854	-0.87877272
347	C	13.44553753	1.55547064	0.00193890
348	H	13.70423043	0.94509834	-0.87622063
349	H	13.70300327	0.94644163	0.88139028
350	C	14.30000861	2.82952366	0.00155407
351	H	14.04226918	3.44048647	0.87974964
352	H	14.04348258	3.43913280	-0.87793698
353	C	15.80965123	2.55653223	0.00280353
354	H	16.06751629	1.94662021	-0.87482891
355	H	16.06631717	1.94801871	0.88175698
356	C	-6.37730151	-1.45485311	-0.00253797
357	H	-6.64960136	-2.06129019	-0.87812652
358	H	-6.64914399	-2.06199258	0.87266316
359	C	-7.21222663	-0.16644339	-0.00187290
360	H	-6.95422436	0.43702490	-0.88260214
361	H	-6.95258163	0.43713048	0.87829899
362	C	-8.72046207	-0.44679200	-0.00060999
363	H	-8.97641050	-1.05543125	0.87899758
364	H	-8.97777064	-1.05605206	-0.87939196
365	C	-9.57263461	0.82886387	-0.00042410
366	H	-9.31487576	1.43741292	-0.87987429
367	H	-9.31354154	1.43807172	0.87817585
368	C	-11.08204175	0.55434112	0.00080914
369	H	-11.33916919	-0.05479501	0.88014500
370	H	-11.34043741	-0.05565229	-0.87756110
371	C	-11.93592184	1.82897329	0.00078963
372	H	-11.67829497	2.43808021	-0.87852102
373	H	-11.67707320	2.43894256	0.87914228
374	C	-13.44553726	1.55547096	0.00196491
375	H	-13.70302454	0.94625918	0.88128339
376	H	-13.70420900	0.94528133	-0.87632779
377	C	-14.30000797	2.82952426	0.00182434
378	H	-14.04345988	3.43931684	-0.87753308
379	H	-14.04229034	3.44030374	0.88015380
380	C	-15.80965070	2.55653298	0.00297909
381	H	-16.06633871	1.94783575	0.88179890
382	H	-16.06749411	1.94680457	-0.87478728

383	C	-16.65554604	3.83407879	0.00278355
384	H	-16.44637672	4.44739458	-0.88259487
385	H	-17.72753839	3.60505452	0.00362778
386	H	-16.44520612	4.44844191	0.88715775
387	C	16.65554688	3.83407775	0.00236187
388	H	17.72753913	3.60505337	0.00328081
389	H	16.44639984	4.44720832	-0.88315009
390	H	16.44518506	4.44862597	0.88660225
391	O	-0.00000010	-1.23211328	-0.00293686

392 **Optimized coordinates (Å) for Th-Thd-Th(10):**

393	C	-1.20806377	0.50845006	-0.08189367
394	C	1.20806231	0.50844896	-0.08186841
395	C	-2.62897218	0.26196466	-0.10378363
396	C	-3.30724199	-0.93120312	-0.21603580
397	S	-3.75383772	1.60215727	0.02841576
398	C	-4.72047485	-0.77714338	-0.19652581
399	H	-2.81147972	-1.89206474	-0.30921571
400	C	-5.13166814	0.52795398	-0.07024494
401	H	-5.40815518	-1.61115399	-0.27339070
402	C	2.62897093	0.26196222	-0.10372872
403	C	3.30724201	-0.93120506	-0.21597812
404	S	3.75383480	1.60215297	0.02850384
405	C	4.72047463	-0.77714681	-0.19643585
406	H	2.81148083	-1.89206528	-0.30917861
407	C	5.13166637	0.52794882	-0.07013243
408	H	5.40815598	-1.61115720	-0.27329419
409	N	-0.67872948	1.70492656	0.03120256
410	N	0.67872675	1.70492594	0.03121690
411	C	-6.52843051	1.08964312	-0.01520250
412	H	-6.63703425	1.69821823	0.89371850
413	H	-6.67075511	1.78587795	-0.85425675
414	C	-7.63742108	0.02837637	-0.04680507
415	H	-7.54715630	-0.56880960	-0.96422752
416	H	-7.50229530	-0.66611570	0.79336948
417	C	-9.03990065	0.64625750	0.02226512
418	H	-9.17222518	1.34284177	-0.81852838
419	H	-9.12561737	1.25118318	0.93680907
420	C	-10.16370184	-0.39780305	-0.00177323
421	H	-10.07753765	-1.00321688	-0.91622769
422	H	-10.02923954	-1.09455842	0.83884452
423	C	-11.56787885	0.21628248	0.06913067
424	H	-11.70294970	0.91089331	-0.77327290
425	H	-11.65144896	0.82523051	0.98159146
426	C	-12.69382056	-0.82574116	0.05176856
427	H	-12.55817735	-1.52007882	0.89440866
428	H	-12.61038804	-1.43514402	-0.86048329
429	C	-14.09796667	-0.21162192	0.12344441
430	H	-14.23434321	0.48104843	-0.72048688
431	H	-14.18006524	0.39991695	1.03443403
432	C	-15.22472807	-1.25259941	0.11041363
433	H	-15.08953979	-1.94519598	0.95476991

434	H	-15.14373892	-1.86493410	-0.80029356
435	C	-16.62884142	-0.63855282	0.18189774
436	H	-16.76479738	0.05219514	-0.66254269
437	H	-16.70983608	-0.02632273	1.09146857
438	C	6.52842797	1.08963542	-0.01504594
439	H	6.67075337	1.78593260	-0.85404804
440	H	6.63702916	1.69814343	0.89392059
441	C	7.63741992	0.02837253	-0.04672512
442	H	7.54715185	-0.56875040	-0.96418831
443	H	7.50229921	-0.66617767	0.79340221
444	C	9.03989870	0.64625129	0.02238311
445	H	9.12562012	1.25110739	0.93697263
446	H	9.17221608	1.34290006	-0.81835809
447	C	10.16370221	-0.39780474	-0.00174333
448	H	10.07753458	-1.00314607	-0.91624547
449	H	10.02924615	-1.09462729	0.83881974
450	C	11.56787801	0.21627877	0.06920174
451	H	11.65145398	0.82514851	0.98171422
452	H	11.70293960	0.91096257	-0.77314316
453	C	12.69382278	-0.82573984	0.05173967
454	H	12.61038518	-1.43506245	-0.86056533
455	H	12.55818835	-1.52015233	0.89431953
456	C	14.09796741	-0.21162234	0.12345875
457	H	14.18007275	0.39983248	1.03450416
458	H	14.23433321	0.48112666	-0.72040975
459	C	15.22473246	-1.25259453	0.11032012
460	H	15.14373686	-1.86484426	-0.80044361
461	H	15.08955482	-1.94527067	0.95461285
462	C	16.62884428	-0.63854945	0.18184824
463	H	16.70984597	-0.02640562	1.09147650
464	H	16.76478881	0.05227942	-0.66252784
465	S	-0.00000002	-0.77125293	-0.20301612
466	C	17.74810507	-1.68492171	0.17003676
467	H	17.71426080	-2.29059304	-0.74418072
468	H	18.73699733	-1.21476990	0.22174887
469	H	17.65926346	-2.36929533	1.02298639
470	C	-17.74809840	-1.68493049	0.17019756
471	H	-18.73699188	-1.21477756	0.22187505
472	H	-17.71426122	-2.29068893	-0.74396246
473	H	-17.65924526	-2.36922236	1.02321157

474 **Optimized coordinates (Å) for Th<sub>3</sub>(10):**

475	C	1.26837879	-1.21007656	0.00008198
476	C	0.70852443	-2.47104433	0.00033753
477	C	-0.70852454	-2.47104417	0.00033264
478	C	-1.26837862	-1.21007627	0.00007325
479	S	0.00000022	0.01108963	-0.00024612
480	H	1.30156813	-3.37954852	0.00055460
481	H	-1.30156847	-3.37954821	0.00053860
482	C	2.66280229	-0.82805285	0.00003919
483	C	3.22398045	0.42972758	-0.00007935
484	S	3.93527594	-2.04662194	0.00014213

485	C	4.64875931	0.42350348	-0.00008727
486	H	2.63410326	1.34072849	-0.00015208
487	C	5.20090459	-0.83106076	0.00002478
488	H	5.24307275	1.33020577	-0.00017238
489	C	-2.66280200	-0.82805223	0.00002622
490	C	-3.22397990	0.42972830	-0.00029443
491	S	-3.93527604	-2.04662098	0.00043851
492	C	-4.64875882	0.42350458	-0.00026313
493	H	-2.63410247	1.34072902	-0.00056956
494	C	-5.20090450	-0.83105947	0.00007857
495	H	-5.24307197	1.33020702	-0.00051052
496	C	6.64764594	-1.25032333	0.00005105
497	H	6.84287393	-1.88653100	0.87558288
498	H	6.84287901	-1.88662508	-0.87541167
499	C	-6.64764594	-1.25032175	0.00018201
500	H	-6.84289086	-1.88674621	-0.87518910
501	H	-6.84286218	-1.88640668	0.87580539
502	C	7.63992573	-0.07865094	-0.00001001
503	H	7.45912084	0.55298895	-0.88027122
504	H	7.45910740	0.55309150	0.88017475
505	C	-7.63992573	-0.07864935	-0.00002881
506	H	-7.45909269	0.55321695	0.88006408
507	H	-7.45913566	0.55286667	-0.88038190
508	C	-9.10082769	-0.54674119	0.00009762
509	H	-9.27876687	-1.18346403	-0.87896767
510	H	-9.27872242	-1.18312624	0.87941647
511	C	9.10082774	-0.54674268	0.00002772
512	H	9.27873888	-1.18324276	0.87925998
513	H	9.27875073	-1.18335047	-0.87912420
514	C	10.10788798	0.61059592	-0.00003658
515	H	9.92872375	1.24731056	-0.87907160
516	H	9.92870528	1.24742444	0.87891218
517	C	-10.10788803	0.61059728	-0.00009860
518	H	-9.92873544	1.24720246	-0.87921525
519	H	-9.92869396	1.24753527	0.87876850
520	C	-11.57042912	0.14715030	0.00002368
521	H	-11.74885435	-0.49003806	0.87900680
522	H	-11.74889954	-0.49036015	-0.87871669
523	C	11.57042906	0.14714916	0.00000916
524	H	11.74886742	-0.49014292	0.87891440
525	H	11.74888663	-0.49025751	-0.87880911
526	C	12.57923310	1.30306212	-0.00005519
527	H	12.40049099	1.94038510	-0.87892945
528	H	12.40046707	1.94050387	0.87872807
529	C	-12.57923331	1.30306307	-0.00016181
530	H	-12.40050068	1.94028629	-0.87911035
531	H	-12.40045802	1.94060463	0.87854715
532	C	-14.04195172	0.84012335	-0.00004244
533	H	-14.22049793	0.20274951	0.87889690
534	H	-14.22054301	0.20243838	-0.87874691
535	C	-15.05135891	1.99530071	-0.00022078
536	H	-14.87386773	2.63313839	-0.87918136

537	H	-14.87382402	2.63344742	0.87850664
538	C	-16.51415231	1.53281315	-0.00010308
539	H	-16.69196820	0.89604880	0.87832949
540	H	-16.69201296	0.89574338	-0.87830512
541	C	14.04195156	0.84012266	-0.00000369
542	H	14.22053285	0.20253421	-0.87878026
543	H	14.22050803	0.20265236	0.87886359
544	C	15.05135854	1.99530022	-0.00006701
545	H	14.87385921	2.63323209	-0.87895759
546	H	14.87383159	2.63335269	0.87873044
547	C	16.51415205	1.53281294	-0.00001206
548	H	16.69200450	0.89583600	-0.87828311
549	H	16.69197640	0.89595580	0.87835153
550	C	17.51529582	2.69276663	-0.00007501
551	H	17.38491046	3.32826236	0.88475111
552	H	18.54962514	2.32973201	-0.00003320
553	H	17.38493935	3.32814112	-0.88499243
554	C	-17.51529629	2.69276667	-0.00027913
555	H	-17.38494750	3.32804847	-0.88526422
556	H	-18.54962551	2.32973186	-0.00019049
557	H	-17.38490347	3.32835509	0.88447929

#### 558 TD-DFT excitation energies and oscillator strengths for Th-Oxd-Th(10):

559	Excited State 1: Singlet-A 3.8397 eV 322.90 nm f = 1.1612 <S**2> = 0.000			
560	139 ->142	0.16639		
561	140 ->141	0.67844		
562	This state for optimization and/or second-order correction.			
563	Total Energy, E(TD-HF/TD-DFT) = -2151.57241331			
564	Copying the excited state density for this state as the 1-particle RhoCI density.			
565	Excited State 2:	Singlet-A	4.7066 eV 263.42 nm f = 0.0585 <S**2> = 0.000	
566	138 ->141	0.10602		
567	139 ->141	0.39208		
568	140 ->142	0.56073		
569	Excited State 3:	Singlet-A	5.1694 eV 239.84 nm f = 0.0007 <S**2> = 0.000	
570	137 ->141	0.58692		
571	138 ->142	-0.35150		
572	Excited State 4:	Singlet-A	5.1769 eV 239.50 nm f = 0.1757 <S**2> = 0.000	
573	137 ->142	-0.34830		
574	138 ->141	0.57678		
575	139 ->141	-0.12485		
576	Excited State 5:	Singlet-A	5.3774 eV 230.57 nm f = 0.0001 <S**2>=0.000	
577	133 ->142	0.12674		
578	139 ->141	0.55159		
579	140 ->142	-0.40040		
580	Excited State 6:	Singlet-A	5.6984 eV 217.58 nm f=0.0000 <S**2> = 0.000	
581	134 ->141	0.64152		
582	134 ->145	-0.14015		
583	136 ->141	0.20251		
584	Excited State 7:	Singlet-A	5.9211 eV 209.39 nm f = 0.0347 <S**2> = 0.000	
585	133 ->141	0.21742		
586	139 ->142	0.61505		
587	140 ->141	-0.13869		

588	140 ->145	0.17856	
589	Excited State 8:	Singlet-A	5.9270 eV 209.19 nm f = 0.0000 <S**2> = 0.000
590	139 ->144	0.33420	
591	140 ->143	0.59840	
592	Excited State 9:	Singlet-A	5.9946 eV 206.83 nm f = 0.0008 <S**2> = 0.000
593	139 ->143	0.37585	
594	140 ->144	0.56656	
595	Excited State 10:	Singlet-A	6.2904 eV 197.10 nm f = 0.0067 <S**2>=0.000
596	116 ->141	-0.10331	
597	118 ->141	-0.21384	
598	122 ->141	-0.17124	
599	134 ->142	0.54317	
600	134 ->150	0.16147	
601	134 ->167	-0.12540	
602	136 ->142	0.17894	
603	Excited State 11:	Singlet-A	6.3596 eV 194.96 nm f = 0.0000 <S**2> = 0.000
604	137 ->143	0.48569	
605	138 ->144	-0.45662	
606	Excited State 12:	Singlet-A	6.3618 eV 194.89 nm f = 0.0000 <S**2> = 0.000
607	134 ->142	-0.10948	
608	137 ->144	-0.45141	
609	138 ->143	0.47057	
610	Excited State 13:	Singlet-A	6.5740 eV 188.60 nm = 0.0144 <S**2> = 0.000
611	133 ->141	-0.16914	
612	137 ->141	-0.30996	
613	137 ->145	0.15244	
614	138 ->142	-0.34737	
615	139 ->150	-0.18597	
616	140 ->145	0.42003	
617	Excited State 14:	Singlet-A	6.6269 eV 187.09 nm f=0.0002 <S**2>=0.000
618	137 ->142	0.50133	
619	138 ->141	0.37664	
620	138 ->145	-0.15985	
621	139 ->145	0.15232	
622	140 ->150	-0.19883	
623	Excited State 15:	Singlet-A	6.7394 eV 183.97 nm f=0.0251 <S**2>=0.000
624	137 ->141	0.20498	
625	137 ->145	-0.10627	
626	138 ->142	0.43762	
627	139 ->142	-0.13396	
628	139 ->150	-0.14540	
629	140 ->145	0.44440	
630	Excited State 16:	Singlet-A	6.8924 eV 179.88 nm f=0.0019 <S**2>=0.000
631	115 ->142	0.11594	
632	116 ->141	0.18119	
633	118 ->141	0.40486	
634	122 ->141	0.36915	
635	134 ->142	0.28832	
636	135 ->141	-0.10094	
637	Excited State 17:	Singlet-A	6.9960 eV 177.22 nm f = 0.3755 S**2>=0.000
638	113 ->141	-0.14081	
639	137 ->142	0.28490	

640	139 ->145	-0.28752					
641	140 ->150	0.52557					
642	Excited State 18:	Singlet-A	7.0016 eV	177.08 nm	f=0.0091	<S**2>=0.000	
643	133 ->141	0.60542					
644	138 ->142	-0.11100					
645	139 ->142	-0.23165					
646	140 ->141	0.10649					
647	Excited State 19:	Singlet-A	7.0156 eV	176.73 nm	f=0.0000	<S**2>=0.000	
648	110 ->142	0.10494					
649	112 ->141	-0.10370					
650	114 ->142	0.20899					
651	115 ->141	0.36301					
652	117 ->141	0.10405					
653	121 ->141	0.31625					
654	122 ->142	0.28720					
655	135 ->142	-0.14145					
656	136 ->141	-0.21074					
657	Excited State 20:	Singlet-A	7.0690 eV	175.39 nm	f=0.0085	<S**2>=0.000	
658	110 ->141	0.17679					
659	114 ->141	0.37381					
660	115 ->142	0.24180					
661	116 ->141	-0.19952					
662	118 ->141	-0.19438					
663	121 ->142	0.21296					
664	122 ->141	0.18842					
665	122 ->145	0.10014					
666	135 ->141	-0.21212					
667	136 ->142	-0.14511					
668	Excited State 21:	Singlet-A	7.0864 eV	174.96 nm	f=0.0000	<S**2>=0.000	
669	139 ->144	-0.16326					
670	140 ->147	-0.13963					
671	140 ->149	0.13149					
672	140 ->152	-0.21062					
673	140 ->153	0.29669					
674	140 ->154	0.32012					
675	140 ->159	-0.27770					
676	140 ->162	-0.23557					
677	Excited State 22:	Singlet-A	7.3232 eV	169.30 nm	f=0.0000	<S**2>=0.000	
678	139 ->147	-0.18742					
679	139 ->149	0.16438					
680	139 ->152	-0.14061					
681	140 ->144	-0.12821					
682	140 ->146	0.34304					
683	140 ->148	-0.28060					
684	140 ->151	0.21270					
685	140 ->161	-0.16377					
686	140 ->163	0.13354					
687	140 ->169	-0.11490					
688	140 ->170	0.15360					
689	Excited State 23:	Singlet-A	7.4056 eV	167.42 nm	f=0.0000	<S**2>=0.000	
690	139 ->144	0.14018					
691	139 ->158	0.11313					



692	139 ->163	-0.19155					
693	140 ->153	0.50296					
694	140 ->159	0.18105					
695	140 ->162	0.22337					
696	140 ->168	0.12670					
697	Excited State 24:	Singlet-A	7.4385 eV	166.68 nm	f=0.0000	<S**2>=0.000	
698	136 ->141	0.11977					
699	139 ->146	-0.19282					
700	139 ->148	0.16348					
701	139 ->151	-0.13568					
702	140 ->147	0.33618					
703	140 ->149	-0.26475					
704	140 ->152	0.16583					
705	140 ->153	0.11341					
706	140 ->154	0.10884					
707	140 ->159	-0.11322					
708	140 ->160	-0.15120					
709	140 ->162	-0.14973					
710	140 ->168	-0.11390					
711	Excited State 25:	Singlet-A	7.4553 eV	166.30 nm	f=0.0009	<S**2>=0.000	
712	139 ->143	0.16129					
713	139 ->153	-0.11429					
714	139 ->154	0.10306					
715	139 ->159	-0.15927					
716	139 ->162	-0.17987					
717	139 ->168	-0.11816					
718	140 ->146	-0.15137					
719	140 ->148	0.12997					
720	140 ->151	-0.13059					
721	140 ->155	0.15112					
722	140 ->158	-0.26395					
723	140 ->163	0.37400					
724	140 ->170	0.15430					
725	Excited State 26:	Singlet-A	7.5037 eV	165.23 nm	f=0.0000	<S**2>=0.000	
726	104 ->142	-0.13497					
727	105 ->141	0.13491					
728	107 ->141	0.17020					
729	110 ->142	0.14865					
730	112 ->141	-0.20770					
731	114 ->142	0.13178					
732	115 ->141	0.29385					
733	121 ->141	-0.12739					
734	134 ->141	-0.11645					
735	135 ->142	0.15633					
736	136 ->141	0.33751					
737	140 ->147	-0.12163					
738	140 ->153	-0.11847					
739	Excited State 27:	Singlet-A	7.5451 eV	164.32 nm	f=0.0224	<S**2>=0.000	
740	108 ->141	-0.13687					
741	111 ->141	0.18370					
742	113 ->141	0.16512					
743	133 ->142	0.44961					

744	137 ->150	0.21846					
745	138 ->145	0.23926					
746	139 ->141	-0.10379					
747	140 ->167	0.18617					
748	Excited State 28:	Singlet-A	7.5572 eV	164.06 nm	f=0.0000	<S**2>=0.000	
749	114 ->142	-0.21459					
750	116 ->142	0.25632					
751	118 ->142	0.45963					
752	118 ->150	0.13839					
753	118 ->167	-0.10476					
754	122 ->142	0.25208					
755	Excited State 29:	Singlet-A	7.5636 eV	163.92 nm	f=0.0011	<S**2>=0.000	
756	104 ->141	-0.20068					
757	105 ->142	0.10349					
758	107 ->142	0.12348					
759	110 ->141	0.24180					
760	112 ->142	-0.15071					
761	114 ->141	0.23230					
762	115 ->142	0.21505					
763	118 ->141	0.11659					
764	135 ->141	0.37713					
765	136 ->142	0.17981					
766	Excited State 30:	Singlet-A	7.6332 eV	162.43 nm	f=0.2315	<S**2>=0.000	
767	137 ->145	0.46672					
768	138 ->142	0.18205					
769	138 ->150	0.40275					
770	139 ->150	-0.12803					
771	140 ->166	-0.13622					

## 772 TD-DFT excitation energies and oscillator strengths for Th-Thd-Th(10):

773	Excited State 1:	Singlet-A	3.5070 eV	353.53 nm	f=1.1804	<S**2>=0.000	
774	143 ->146	0.13462					
775	144 ->145	0.68560					
776	This state for optimization and/or second-order correction.						
777	Total Energy, E(TD-HF/TD-DFT) = -2474.55713381						
778	Copying the excited state density for this state as the 1-particle RhoCI density.						
779	Excited State 2:	Singlet-A	4.4743 eV	277.10 nm	f=0.0001	<S**2>=0.000	
780	143 ->145	0.54962					
781	144 ->146	0.41215					
782	Excited State 3:	Singlet-A	4.7187 eV	262.75 nm	f=0.0000	<S**2>=0.000	
783	140 ->145	0.68229					
784	140 ->148	0.13663					
785	Excited State 4:	Singlet-A	4.9413 eV	250.92 nm	f=0.0964	<S**2>=0.000	
786	141 ->146	-0.27999					
787	142 ->145	0.60751					
788	143 ->145	0.16448					
789	Excited State 5:	Singlet-A	4.9658 eV	249.67 nm	f=0.0061	<S**2>=0.000	
790	141 ->145	0.62700					
791	142 ->146	-0.28821					
792	Excited State 6:	Singlet-A	5.1295 eV	241.71 nm	f=0.0139	<S**2>=0.000	
793	142 ->145	0.14946					
794	143 ->145	-0.38413					

795	143 ->148	-0.12713					
796	144 ->146	0.53809					
797	Excited State	7:	Singlet-A	5.3658 eV	231.06 nm	f=0.0006	<S**2>=0.000
798	136 ->147	-0.13272					
799	144 ->147	0.67553					
800	Excited State	8:	Singlet-A	5.6407 eV	219.80 nm	f=0.0013	<S**2>=0.000
801	133 ->145	0.52447					
802	140 ->146	-0.40254					
803	140 ->155	-0.12979					
804	144 ->147	-0.10696					
805	Excited State	9:	Singlet-A	5.7387 eV	216.05 nm	f=0.1112	<S**2>=0.000
806	136 ->145	0.21706					
807	143 ->146	0.58147					
808	144 ->148	-0.28253					
809	Excited State	10:	Singlet-A	5.8126 eV	213.30 nm	f=0.0303	<S**2>=0.000
810	136 ->146	-0.10062					
811	137 ->145	0.64223					
812	140 ->147	0.16350					
813	Excited State	11:	Singlet-A	5.8558 eV	211.73 nm	f=0.0002	<S**2>=0.000
814	143 ->147	-0.18070					
815	143 ->150	0.26175					
816	144 ->149	0.59726					
817	Excited State	12:	Singlet-A	6.0214 eV	205.91 nm	f=0.0011	<S**2>=0.000
818	133 ->145	-0.14580					
819	140 ->146	-0.11898					
820	143 ->149	0.36392					
821	144 ->150	0.50896					
822	144 ->152	-0.15647					
823	Excited State	13:	Singlet-A	6.1099 eV	202.92 nm	f=0.0038	<S**2>=0.000
824	133 ->145	0.39510					
825	140 ->146	0.49361					
826	140 ->155	0.14768					
827	143 ->149	0.10918					
828	144 ->150	0.14781					
829	Excited State	14:	Singlet-A	6.1768 eV	200.73 nm	f=0.0000	<S**2>=0.000
830	137 ->147	0.38238					
831	142 ->147	0.10202					
832	143 ->147	0.53709					
833	143 ->150	0.10905					
834	144 ->149	0.13514					
835	Excited State	15:	Singlet-A	6.3335 eV	195.76 nm	f=0.0067	<S**2>=0.000
836	136 ->145	0.30318					
837	141 ->145	-0.20038					
838	141 ->148	-0.16839					
839	142 ->146	-0.28521					
840	143 ->155	0.18149					
841	144 ->148	0.43790					
842	Excited State	16:	Singlet-A	6.3823 eV	194.26 nm	f=0.0002	<S**2>=0.000
843	141 ->147	0.15331					
844	141 ->150	-0.40636					
845	141 ->152	0.12986					
846	142 ->149	0.47299					

847	Excited State	17:	Singlet-A	6.3871 eV	194.12 nm	f=0.0000	$\langle S^{*2} \rangle = 0.000$
848	141 ->149		0.47975				
849	142 ->147		0.13512				
850	142 ->150		-0.39778				
851	142 ->152		0.12746				
852	143 ->150		-0.10594				
853	Excited State	18:	Singlet-A	6.4570 eV	192.01 nm	f=0.0031	$\langle S^{*2} \rangle = 0.000$
854	137 ->145		-0.10085				
855	141 ->146		0.50008				
856	142 ->145		0.30150				
857	142 ->148		0.21920				
858	143 ->148		0.17744				
859	144 ->155		0.19138				
860	Excited State	19:	Singlet-A	6.5242 eV	190.04 nm	f=0.0757	$\langle S^{*2} \rangle = 0.000$
861	141 ->145		0.20347				
862	141 ->148		0.17401				
863	142 ->146		0.44590				
864	143 ->146		0.20349				
865	144 ->148		0.37967				
866	Excited State	20:	Singlet-A	6.6944 eV	185.21 nm	f=0.0010	$\langle S^{*2} \rangle = 0.000$
867	136 ->145		0.55620				
868	137 ->146		-0.11522				
869	141 ->145		0.10055				
870	142 ->146		0.20532				
871	143 ->146		-0.24397				
872	144 ->145		0.11098				
873	144 ->148		-0.10053				
874	Excited State	21:	Singlet-A	6.8314 eV	181.49 nm	f=0.2001	$\langle S^{*2} \rangle = 0.000$
875	140 ->147		-0.27549				
876	141 ->146		-0.26409				
877	143 ->148		0.29197				
878	144 ->153		-0.12434				
879	144 ->155		0.43394				
880	Excited State	22:	Singlet-A	6.8374 eV	181.33 nm	f=0.0000	$\langle S^{*2} \rangle = 0.000$
881	117 ->146		-0.15989				
882	118 ->145		0.37991				
883	123 ->146		0.13395				
884	124 ->145		0.27591				
885	133 ->146		0.36296				
886	139 ->145		-0.17930				
887	Excited State	23:	Singlet-A	6.9028 eV	179.61 nm	f=0.0064	$\langle S^{*2} \rangle = 0.000$
888	114 ->145		-0.14222				
889	117 ->145		-0.33787				
890	118 ->146		0.23649				
891	119 ->145		0.12307				
892	123 ->145		0.34030				
893	124 ->146		0.21139				
894	138 ->145		-0.26525				
895	139 ->146		-0.12211				
896	Excited State	24:	Singlet-A	6.9601 eV	178.14 nm	f=0.0000	$\langle S^{*2} \rangle = 0.000$
897	137 ->147		-0.12830				
898	143 ->150		0.22079				

899	143 ->178	-0.11140					
900	144 ->156	0.14121					
901	144 ->158	0.47414					
902	144 ->163	-0.18195					
903	144 ->169	0.20726					
904	Excited State 25:	Singlet-A	7.0702 eV	175.36 nm	f=0.0000	<S**2>=0.000	
905	111 ->145	-0.13921					
906	117 ->146	0.10414					
907	123 ->146	-0.17663					
908	124 ->145	-0.21380					
909	133 ->146	0.44611					
910	133 ->155	0.15483					
911	139 ->145	0.27099					
912	144 ->158	-0.12091					
913	Excited State 26:	Singlet-A	7.1966 eV	172.28 nm	f=0.0000	<S**2>=0.000	
914	118 ->145	-0.12226					
915	137 ->147	-0.48892					
916	139 ->145	-0.11810					
917	143 ->147	0.32931					
918	144 ->158	-0.19148					
919	144 ->159	-0.10344					
920	Excited State 27:	Singlet-A	7.2334 eV	171.41 nm	f=0.0000	<S**2>=0.000	
921	110 ->146	-0.10281					
922	111 ->145	-0.20097					
923	114 ->146	-0.13037					
924	115 ->145	-0.16591					
925	117 ->146	-0.17092					
926	118 ->145	0.35564					
927	133 ->146	-0.26648					
928	133 ->155	-0.10352					
929	137 ->147	-0.15400					
930	139 ->145	0.26259					
931	143 ->147	0.11741					
932	Excited State 28:	Singlet-A	7.2689 eV	170.57 nm	f=0.0132	<S**2>=0.000	
933	133 ->147	-0.12287					
934	137 ->146	0.62523					
935	137 ->155	0.12744					
936	143 ->146	-0.11783					
937	144 ->148	-0.10865					
938	Excited State 29:	Singlet-A	7.2892 eV	170.09 nm	f=0.0107	<S**2>=0.000	
939	140 ->147	-0.12446					
940	143 ->151	0.18465					
941	143 ->154	-0.16801					
942	143 ->156	0.14777					
943	143 ->165	0.10297					
944	144 ->150	0.20034					
945	144 ->152	0.30387					
946	144 ->153	0.28802					
947	144 ->157	-0.21871					
948	144 ->166	0.15341					
949	Excited State 30:	Singlet-A	7.3187 eV	169.41 nm	f=0.0002	<S**2>=0.000	
950	143 ->152	0.18274					

951	143 ->153	0.16843
952	143 ->157	-0.14315
953	144 ->151	0.34681
954	144 ->154	-0.29297
955	144 ->156	0.21631
956	144 ->163	0.14280
957	144 ->165	0.15538
958	144 ->169	-0.18547

### 959 TD-DFT excitation energies and oscillator strengths for Th<sub>3</sub>(10):

960	Excited State 1:	Singlet-A	3.2352 eV	383.23 nm	f=1.1501	<S**2>=0.000
961	143 ->146	0.10496				
962	144 ->145	0.69331				
963	This state for optimization and/or second-order correction.					
964	Total Energy, E(TD-HF/TD-DFT) = -2442.48397494					
965	Copying the excited state density for this state as the 1-particle RhoCI density.					
966	Excited State 2:	Singlet-A	4.4727 eV	277.20 nm	f=0.0114	<S**2>=0.000
967	140 ->145	-0.13771				
968	143 ->145	0.41305				
969	144 ->146	0.53540				
970	Excited State 3:	Singlet-A	4.9600 eV	249.97 nm	f=0.0009	<S**2>=0.000
971	139 ->146	-0.12084				
972	140 ->145	0.12421				
973	142 ->145	0.24007				
974	143 ->145	0.52412				
975	144 ->146	-0.33903				
976	Excited State 4:	Singlet-A	5.0025 eV	247.84 nm	f=0.0001	<S**2>=0.000
977	140 ->145	0.18829				
978	141 ->146	-0.15922				
979	142 ->145	0.57887				
980	143 ->145	-0.17322				
981	144 ->146	0.24500				
982	Excited State 5:	Singlet-A	5.1535 eV	240.58 nm	f=0.0025	<S**2>=0.000
983	140 ->146	0.11974				
984	141 ->145	0.61946				
985	142 ->146	-0.24148				
986	144 ->149	0.12542				
987	Excited State 6:	Singlet-A	5.2003 eV	238.42 nm	f=0.0002	<S**2>=0.000
988	139 ->148	0.11993				
989	144 ->148	0.65392				
990	144 ->150	0.20644				
991	Excited State 7:	Singlet-A	5.2659 eV	235.45 nm	f=0.2023	<S**2>=0.000
992	140 ->145	0.59885				
993	141 ->146	0.21978				
994	142 ->145	-0.17621				
995	143 ->149	0.10311				
996	144 ->146	0.11921				
997	144 ->157	-0.12382				
998	Excited State 8:	Singlet-A	5.3190 eV	233.10 nm	f=0.0000	<S**2>=0.000
999	143 ->150	-0.22838				
1000	144 ->147	0.63700				
1001	Excited State 9:	Singlet-A	5.4861 eV	226.00 nm	f=0.0009	<S**2>=0.000

1002	143 ->147	-0.30436					
1003	144 ->148	-0.19118					
1004	144 ->150	0.56937					
1005	Excited State 10:	Singlet-A	5.7371 eV	216.11 nm	f=0.0898	<S**2>=0.000	
1006	139 ->145	-0.19445					
1007	141 ->145	-0.14198					
1008	143 ->146	0.38087					
1009	144 ->149	0.51601					
1010	Excited State 11:	Singlet-A	6.0924 eV	203.51 nm	f=0.0080	<S**2>=0.000	
1011	139 ->145	0.51640					
1012	140 ->146	0.11207					
1013	143 ->146	-0.24081					
1014	143 ->157	-0.10360					
1015	144 ->149	0.35163					
1016	Excited State 12:	Singlet-A	6.2170 eV	199.43 nm	f=0.0000	<S**2>=0.000	
1017	140 ->148	0.43473					
1018	140 ->150	0.17204					
1019	142 ->148	0.35622					
1020	143 ->148	-0.32545					
1021	Excited State 13:	Singlet-A	6.2258 eV	199.15 nm	f=0.0001	<S**2>=0.000	
1022	140 ->147	-0.21337					
1023	141 ->148	-0.15016					
1024	141 ->150	0.42272					
1025	142 ->147	0.41946					
1026	143 ->147	-0.14830					
1027	Excited State 14:	Singlet-A	6.2301 eV	199.01 nm	f=0.0000	<S**2>=0.000	
1028	140 ->148	0.16562					
1029	140 ->150	-0.16123					
1030	141 ->147	0.47248					
1031	142 ->150	0.38420					
1032	143 ->150	-0.12741					
1033	Excited State 15:	Singlet-A	6.3848 eV	194.19 nm	f=0.0000	<S**2>=0.000	
1034	139 ->147	0.10077					
1035	143 ->150	0.20691					
1036	144 ->158	0.12446					
1037	144 ->160	-0.29750					
1038	144 ->162	0.52143					
1039	144 ->166	-0.10013					
1040	144 ->173	-0.10046					
1041	Excited State 16:	Singlet-A	6.5583 eV	189.05 nm	f=0.0785	<S**2>=0.000	
1042	139 ->145	-0.21790					
1043	140 ->146	0.22034					
1044	141 ->145	0.17733					
1045	141 ->149	-0.15636					
1046	142 ->146	0.47097					
1047	143 ->146	-0.28815					
1048	144 ->149	0.13947					
1049	Excited State 17:	Singlet-A	6.5981 eV	187.91 nm	f=0.0572	<S**2>=0.000	
1050	140 ->145	0.22905					
1051	141 ->146	-0.35047					
1052	142 ->145	-0.20032					
1053	142 ->149	0.18471					

1054	143 ->149	-0.25489					
1055	144 ->157	0.35732					
1056	144 ->163	0.12281					
1057	Excited State 18:	Singlet-A	6.7410 eV	183.93 nm	f=0.0047	<S**2>=0.000	
1058	121 ->146	-0.15110					
1059	124 ->145	-0.43713					
1060	126 ->145	0.48565					
1061	Excited State 19:	Singlet-A	6.7490 eV	183.71 nm	f=0.0404	<S**2>=0.000	
1062	139 ->145	0.32421					
1063	140 ->146	0.17189					
1064	142 ->146	0.32710					
1065	143 ->146	0.41507					
1066	144 ->145	-0.11702					
1067	144 ->149	-0.18198					
1068	Excited State 20:	Singlet-A	6.7642 eV	183.29 nm	f=0.0000	<S**2>=0.000	
1069	140 ->148	0.10227					
1070	143 ->148	0.12138					
1071	143 ->168	-0.10086					
1072	144 ->151	0.14359					
1073	144 ->153	-0.14284					
1074	144 ->156	0.16845					
1075	144 ->160	0.50528					
1076	144 ->162	0.26350					
1077	Excited State 21:	Singlet-A	6.8325 eV	181.46 nm	f=0.0029	<S**2>=0.000	
1078	139 ->145	-0.12983					
1079	140 ->146	0.55172					
1080	141 ->145	-0.19479					
1081	141 ->149	0.24496					
1082	142 ->146	-0.10988					
1083	144 ->169	0.10752					
1084	144 ->172	0.12705					
1085	Excited State 22:	Singlet-A	6.8371 eV	181.34 nm	f=0.0005	<S**2>=0.000	
1086	143 ->151	0.15153					
1087	143 ->153	-0.14615					
1088	143 ->156	0.12087					
1089	144 ->150	-0.12148					
1090	144 ->152	0.39264					
1091	144 ->154	-0.35542					
1092	144 ->155	0.26005					
1093	144 ->164	0.14586					
1094	Excited State 23:	Singlet-A	6.8658 eV	180.58 nm	f=0.0000	<S**2>=0.000	
1095	143 ->152	0.13304					
1096	143 ->154	-0.13296					
1097	143 ->155	0.10352					
1098	144 ->151	0.37744					
1099	144 ->153	-0.31825					
1100	144 ->156	0.19876					
1101	144 ->160	-0.23391					
1102	144 ->166	0.17936					
1103	144 ->175	0.12662					
1104	Excited State 24:	Singlet-A	6.9527 eV	178.33 nm	f=0.5434	<S**2>=0.000	
1105	140 ->149	0.13155					



1106	141 ->146	0.40916					
1107	142 ->145	0.13324					
1108	142 ->149	-0.18033					
1109	143 ->149	-0.18570					
1110	144 ->157	0.40079					
1111	144 ->163	0.11667					
1112	144 ->170	-0.13318					
1113	144 ->171	0.11015					
1114	Excited State 25:	Singlet-A	6.9796 eV	177.64 nm	f=0.0000	<S**2>=0.000	
1115	121 ->145	0.51090					
1116	125 ->145	0.32714					
1117	126 ->146	-0.25877					
1118	Excited State 26:	Singlet-A	7.0175 eV	176.68 nm	f=0.0009	<S**2>=0.000	
1119	139 ->148	0.10075					
1120	143 ->147	-0.17571					
1121	143 ->158	-0.12106					
1122	143 ->162	-0.22564					
1123	144 ->150	-0.13309					
1124	144 ->159	-0.10324					
1125	144 ->164	-0.28123					
1126	144 ->168	0.44776					
1127	Excited State 27:	Singlet-A	7.1144 eV	174.27 nm	f=0.0000	<S**2>=0.000	
1128	140 ->148	0.24262					
1129	142 ->148	0.10358					
1130	143 ->148	0.45693					
1131	143 ->150	0.21545					
1132	144 ->162	-0.23352					
1133	144 ->165	0.16471					
1134	144 ->173	-0.14445					
1135	Excited State 28:	Singlet-A	7.1596 eV	173.17 nm	f=0.0020	<S**2>=0.000	
1136	114 ->145	-0.17717					
1137	118 ->145	-0.28277					
1138	119 ->145	0.10392					
1139	121 ->146	0.11144					
1140	124 ->145	-0.26405					
1141	125 ->146	0.23664					
1142	126 ->145	-0.26854					
1143	126 ->149	-0.11796					
1144	137 ->145	0.28692					
1145	138 ->146	-0.12358					
1146	Excited State 29:	Singlet-A	7.2077 eV	172.02 nm	f=0.0083	<S**2>=0.000	
1147	139 ->146	-0.13113					
1148	141 ->146	0.12497					
1149	143 ->169	-0.11253					
1150	143 ->172	-0.13142					
1151	144 ->157	0.16827					
1152	144 ->170	0.57395					
1153	144 ->171	-0.19441					
1154	Excited State 30:	Singlet-A	7.2624 eV	170.72 nm	f=0.0000	<S**2>=0.000	
1155	112 ->145	0.25008					
1156	116 ->145	0.13405					
1157	120 ->145	-0.10514					

1158	121 ->145	0.23442
1159	124 ->146	0.14355
1160	125 ->145	-0.25429
1161	137 ->146	-0.13388
1162	138 ->145	0.35980
1163	143 ->148	-0.12958

## 1164 References

1. Lavigueur, C.; Foster, E.J.; Williams, V.E. A simple and inexpensive capillary furnace for variable-temperature X-ray diffraction. *J. Appl. Crystallogr.* **2008**, *41*, 214–216.
2. Boucher, N.; Leroy, J.; Sergeev, S.; Pouzet, E.; Lemaure, V.; Lazzaroni, R.; Cornil, J.; Geerts, Y.H.; Sferrazza, M. Mesomorphism of dialkylterthiophene homologues. *Synth. Met.* **2009**, *159*, 1319–1324.
3. Bruker, APEX3, SAINT and SADABS, Bruker AXS Inc., Madison, Wisconsin, USA, 2016.
4. G. M. Sheldrick, TWINABS (Version 2012/1), University of Göttingen, Germany, 1996.
5. Sheldrick, G.M. Crystal structure refinement with SHELXL. *Acta Crystallogr. Sect. C Struct. Chem.* **2015**, *71*, 3–8.
6. Hübschle, C.B.; Sheldrick, G.M.; Dittrich, B. ShelXle: a Qt graphical user interface for SHELXL. *J. Appl. Crystallogr.* **2011**, *44*, 1281–1284.
7. Farrugia, L.J. WinGX and ORTEP for Windows: an update. *J. Appl. Crystallogr.* **2012**, *45*, 849–854.
8. Fenn, T.D.; Ringe, D.; Petsko, G.A. POVScript+: a program for model and data visualization using persistence of vision ray-tracing. *J. Appl. Crystallogr.* **2003**, *36*, 944–947.
9. Lercher, C.; Röthel, C.; Roscioni, O.M.; Geerts, Y.H.; Shen, Q.; Teichert, C.; Fischer, R.; Leising, G.; Sferrazza, M.; Gbabode, G.; et al. Polymorphism of dioctyl-terthiophene within thin films: The role of the first monolayer. *Chem. Phys. Lett.* **2015**, *630*, 12–17.
10. Van Bolhuis, F.; Wynberg, H.; Havinga, E.E.; Meijer, E.W.; Staring, E.G.J. The X-ray structure and MNDO calculations of  $\alpha$ -terthienyl: A model for polythiophenes. *Synth. Met.* **1989**, *30*, 381–389.
11. Resch-Genger, U.; Rurack, K. Determination of the photoluminescence quantum yield of dilute dye solutions (IUPAC Technical Report). *Pure Appl. Chem.* **2013**, *85*, 2005–2013.
12. Brouwer, A.M. Standards for photoluminescence quantum yield measurements in solution (IUPAC Technical Report). *Pure Appl. Chem.* **2011**, *83*, 2213–2228.
13. Stephens, P.J.; Devlin, F.J.; Chabalowski, C.F.; Frisch, M.J. Ab Initio Calculation of Vibrational Absorption and Circular Dichroism Spectra Using Density Functional Force Fields. *J. Phys. Chem.* **1994**, *98*, 11623–11627.
14. Becke, A.D. Density-functional thermochemistry. III. The role of exact exchange. *J. Chem. Phys.* **1993**, *98*, 5648–5652.
15. Tomasi, J.; Mennucci, B.; Cancès, E. The IEF version of the PCM solvation method: an overview of a new method addressed to study molecular solutes at the QM ab initio level. *J. Mol. Struct. THEOCHEM* **1999**, *464*, 211–226.
16. Barone, V.; Cossi, M.; Tomasi, J. Geometry optimization of molecular structures in solution by the polarizable continuum model. *J. Comput. Chem.* **1998**, *19*, 404–417.
17. Barone, V.; Cossi, M.; Tomasi, J. A new definition of cavities for the computation of solvation free energies by the polarizable continuum model. *J. Chem. Phys.* **1997**, *107*, 3210–3221.
18. Miertuš, S.; Scrocco, E.; Tomasi, J. Electrostatic interaction of a solute with a continuum. A direct utilization of AB initio molecular potentials for the prevision of solvent effects. *Chem. Phys.* **1981**, *55*, 117–129.
19. Gaussian 16, Revision B.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, G. A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A. V. Marenich, J. Bloino, B. G. Janesko, R. Gomperts, B. Mennucci, H. P. Hratchian, J. V. Ortiz, A. F. Izmaylov, J. L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V. G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. J. Bearpark, J. J. Heyd, E. N. Brothers, K. N. Kudin, V. N. Staroverov, T. A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. P. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, J. M. Millam, M. Klene, C. Adamo, R. Cammi, J. W. Ochterski, R. L. Martin, K. Morokuma, O. Farkas, J. B. Foresman, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2016.

- 1210 20. Schäfer, A.; Huber, C.; Ahlrichs, R. Fully optimized contracted Gaussian basis sets of triple zeta valence  
1211 quality for atoms Li to Kr. *J. Chem. Phys.* **1994**, *100*, 5829–5835.
- 1212 21. Schäfer, A.; Horn, H.; Ahlrichs, R. Fully optimized contracted Gaussian basis sets for atoms Li to Kr. *J.*  
1213 *Chem. Phys.* **1992**, *97*, 2571–2577.
- 1214 22. Stratmann, R.E.; Scuseria, G.E.; Frisch, M.J. An efficient implementation of time-dependent density-  
1215 functional theory for the calculation of excitation energies of large molecules. *J. Chem. Phys.* **1998**, *109*, 8218–  
1216 8224.
- 1217 23. Casida, M.E. Time-Dependent Density Functional Response Theory for Molecules. In *Recent Advances in*  
1218 *Density Functional Methods; Recent Advances in Computational Chemistry; WORLD SCIENTIFIC, 1995;*  
1219 *Vol. Volume 1*, pp. 155–192 ISBN 978-981-02-2442-4.
- 1220 24. Huff, G.S.; Gallaher, J.K.; Hodgkiss, J.M.; Gordon, K.C. No single DFT method can predict Raman cross-  
1221 sections, frequencies and electronic absorption maxima of oligothiophenes. *Synth. Met.* **2017**, *231*, 1–6.
- 1222 25. Jacquemin, D.; Michaux, C.; Perpète, E.A.; Maurel, F.; Perrier, A. Photochromic molecular wires: Insights  
1223 from theory. *Chem. Phys. Lett.* **2010**, *488*, 193–197.
- 1224 26. Yanai, T.; Tew, D.P.; Handy, N.C. A new hybrid exchange–correlation functional using the Coulomb-  
1225 attenuating method (CAM-B3LYP). *Chem. Phys. Lett.* **2004**, *393*, 51–57.