

Synthesis of MgAC-Fe₃O₄/TiO₂ hybrid nanocomposites via sol-gel chemistry for water treatment by photo-Fenton and photocatalytic reactions

Supporting Information

Vu Khac Hoang Bui¹, Duckshin Park², Tuyet Nhung Pham¹, Yejin An³, Jin Seok Choi⁴, Hyun Uk Lee⁵, Oh-Hyeok Kwon⁶, Ju-Young Moon⁶, Ki-Tae Kim^{3*}, and Young-Chul Lee^{1*}

¹ Department of BioNano Technology, Gachon University, 1342 Seongnamdaero, Sujeong-gu, Seongnam-si, Gyeonggi-do 13120, Republic of Korea; hoangvu210190@gmail.com; nhungpham240694@gmail.com

² Korea Railroad Research Institute (KRRRI), 176 Cheoldobakmulkwon-ro, Uiwang-si 16150, Gyeonggi-do, Republic of Korea; dspark@krri.re.kr

³ Department of Environmental Engineering, Seoul National University of Science and Technology, 232 Gongneung-ro, Nowon-gu, Seoul 01811, Republic of Korea; yejin33@seoultech.ac.kr

⁴ Analysis Center for Research Advancement, Korea Advanced Institute of Science and Technology (KAIST), Yuseong-gu, Daejeon 34141, Republic of Korea; ffband@kaist.ac.kr

⁵ Advanced Nano-surface Research Group, Korea Basic Science Institute (KBSI), Daejeon 34133, Republic of Korea; leehe@kbsi.re.kr

⁶ Department Beauty Design Management, Hansung University, 116 Samseongyoro-16gil, Seoul 02876, Republic of Korea; beauty67@hansung.ac.kr; bora7033@naver.com

*Correspondence:

Prof. Ki-Tae Kim; Email: ktkim@seoultech.ac.kr; Tel: +82-2-970-6642 (K.-T.K.); Fax: +82-2-971-5776 (K.-T.K.)

Prof. Young-Chul Lee; Email: dreamdb@gachon.ac.kr; Tel: +82-31-750-8751 (Y.-C. L.); Fax: +82-31-750-4748 (Y.-C. L.)

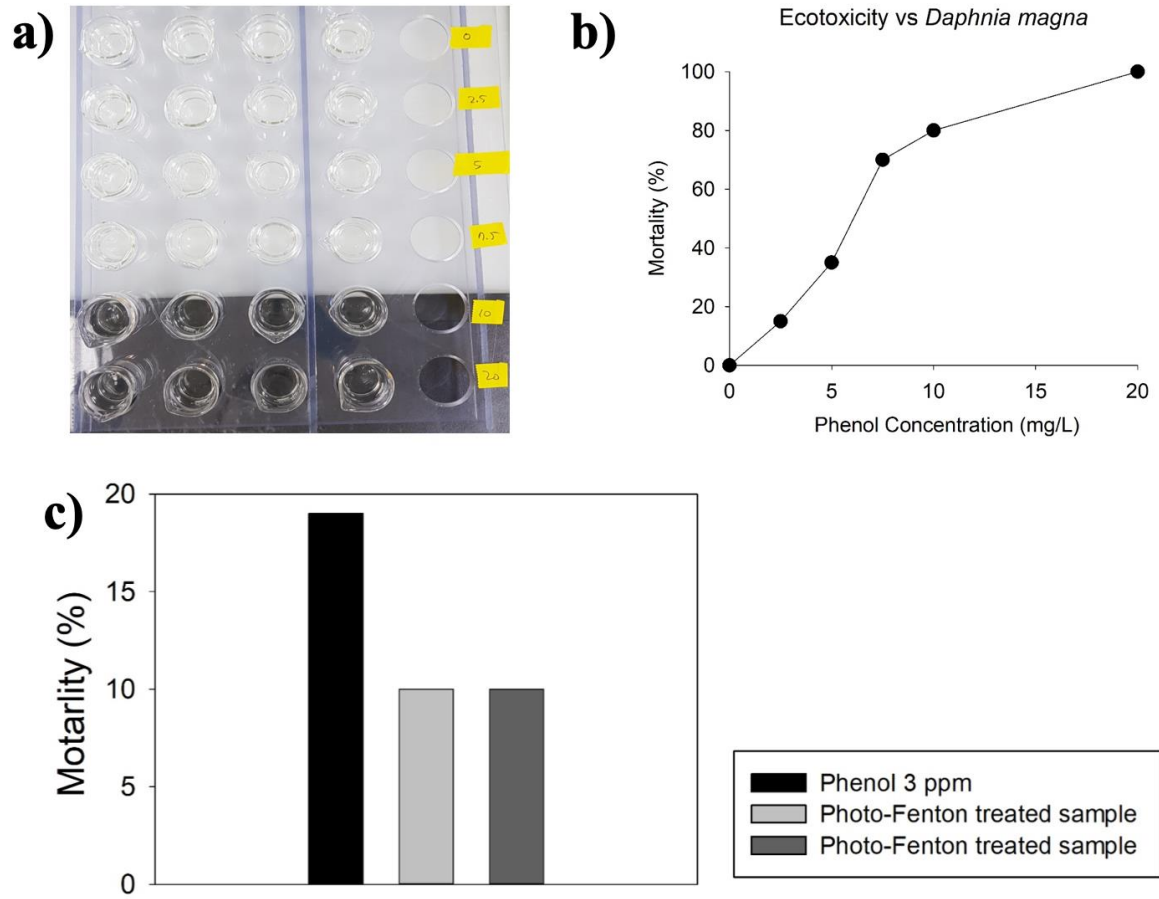


Figure S1. Ecotoxicity against *Daphnia magna*: (a) exposure of *D. magna* to different concentrations of phenol, (b) ecotoxicity of phenol against *D. magna*, and (c) ecotoxicity of phenol 3 ppm (initial sample), photo-Fenton-, and photocatalysis-treated samples

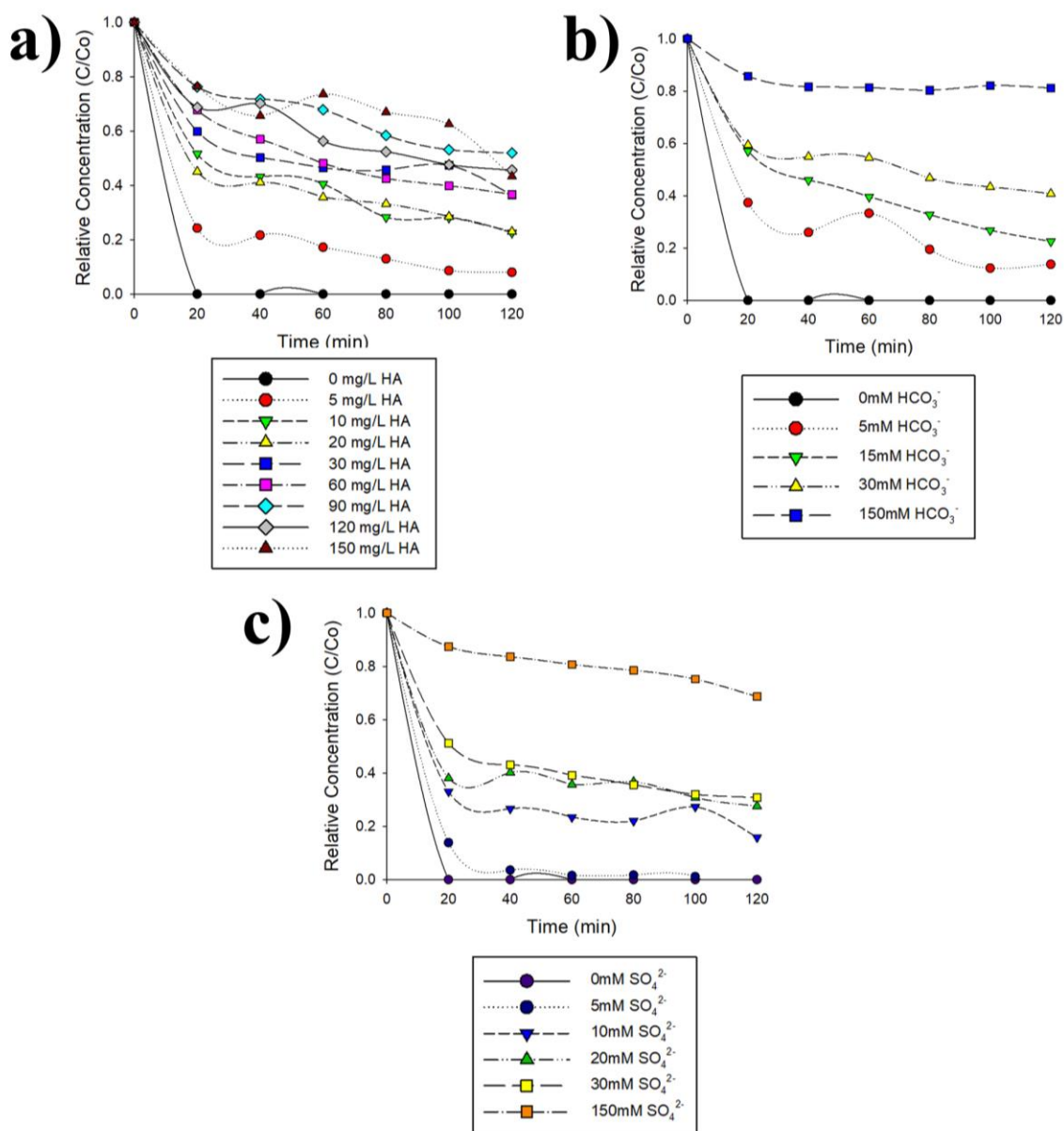


Figure S2. Effects of inhibition agents: a) HA, b) HCO₃⁻, and c) SO₄²⁻ on photo-Fenton performance of MgAC-Fe₃O₄ [0.05 g]/TiO₂ hybrid nanocomposites

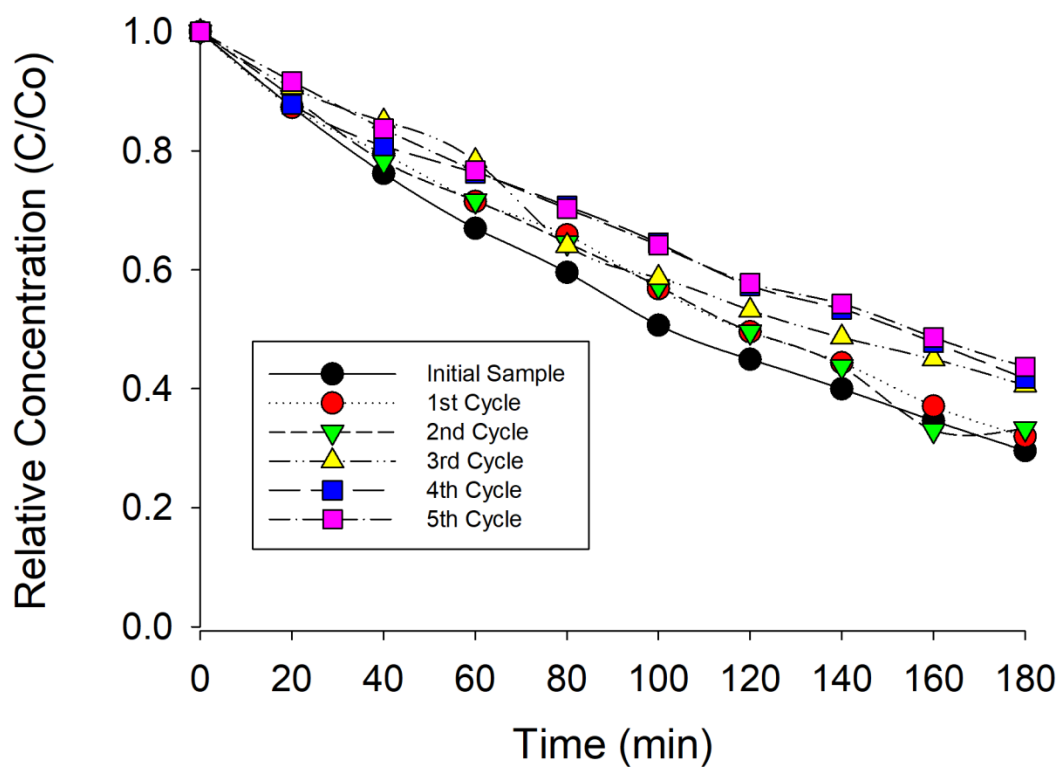


Figure S3. Recycling performances of MgAC-Fe₃O₄ [0.05 g]/TiO₂ on batch mode scale

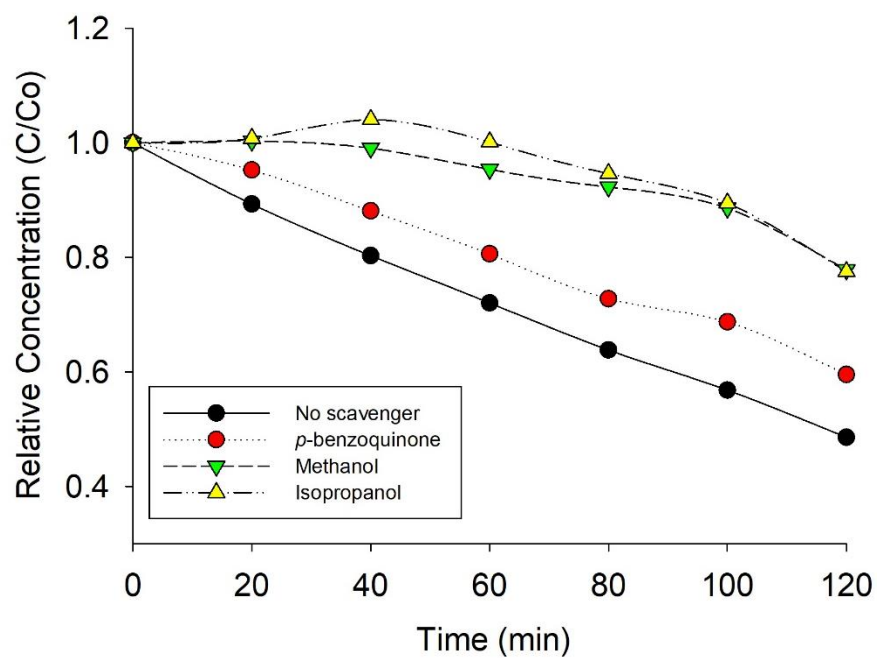


Figure S4. Photocatalytic activities of MgAC-Fe₃O₄ [0.05 g]/TiO₂ against 20 ppm MB in the presence of different reactive species: 0.1 M isopropanol, 0.1 M ethanol, and 5mM *p*-benzoquinone

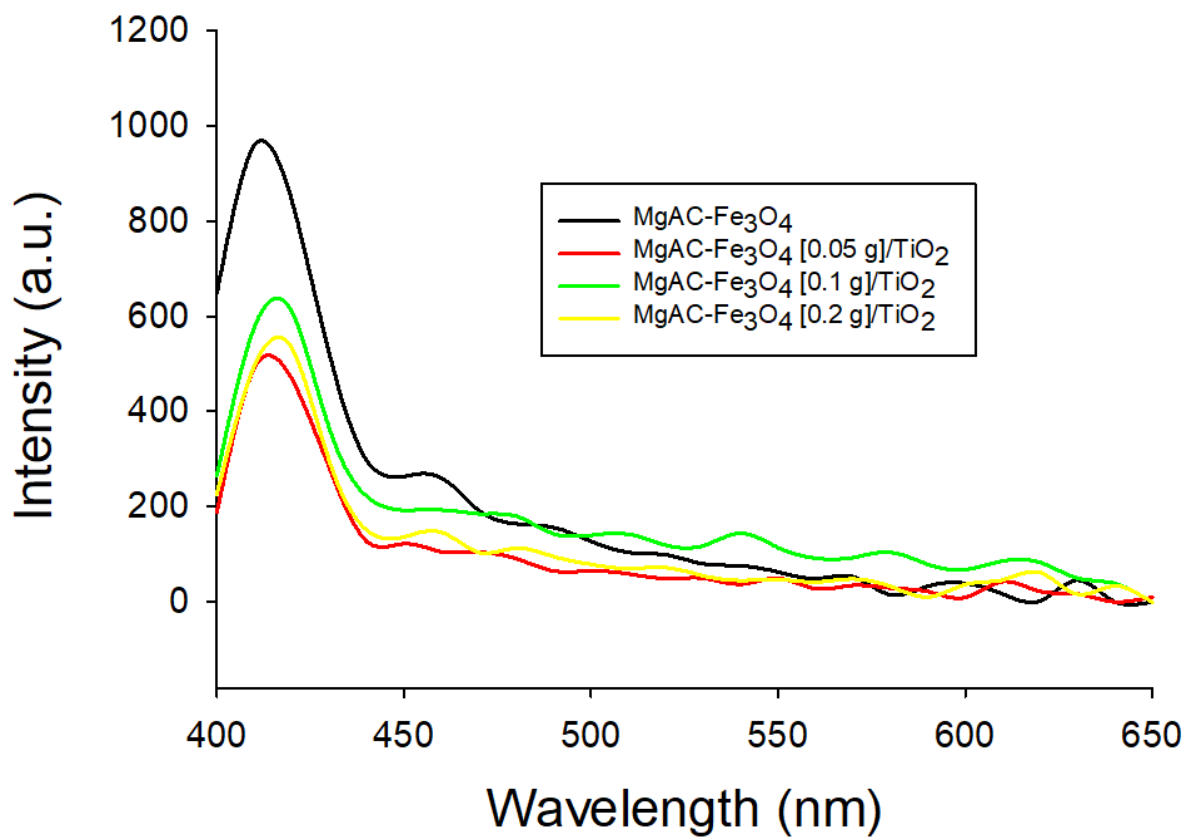


Figure S5. Photoluminescence (PL) spectra of MgAC-Fe₃O₄ and MgAC-Fe₃O₄/TiO₂ hybrid nanocomposites ($\lambda_{\text{ex}} = 365 \text{ nm}$).

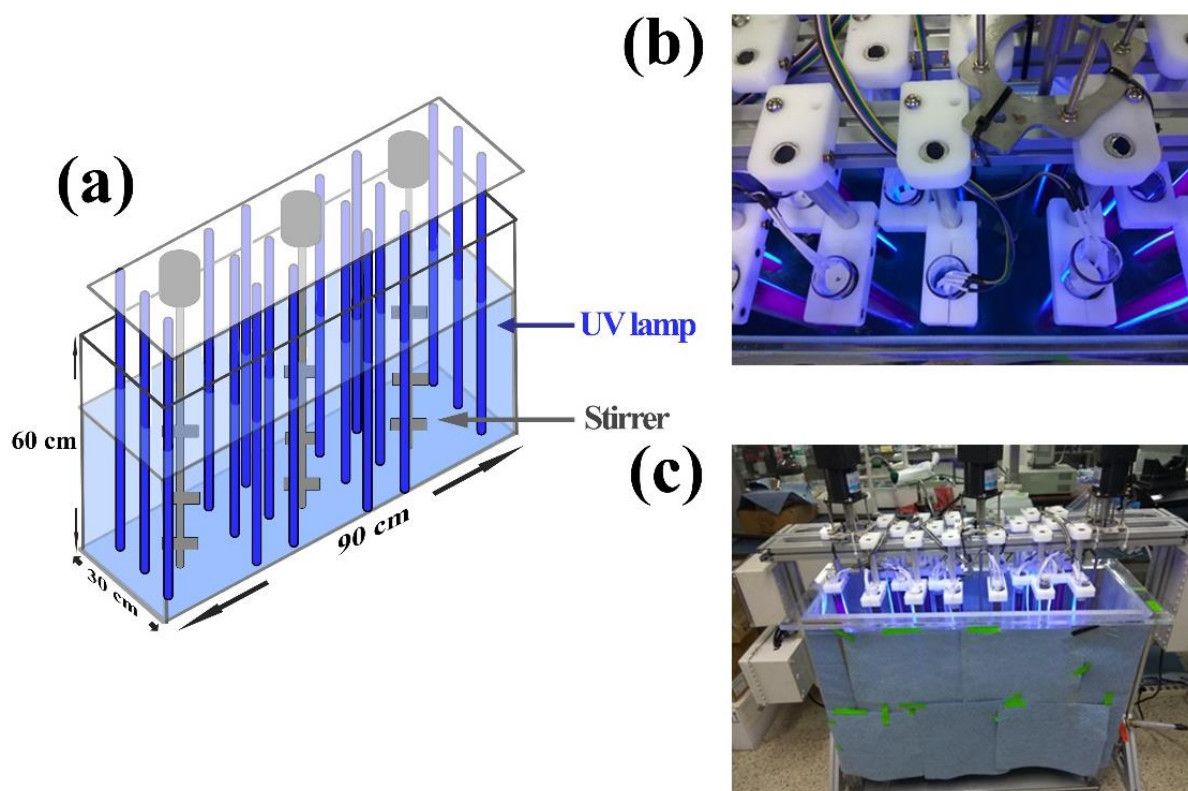


Figure S6. (a) Design of reactor for pilot-scale experiment and photos of reactor from (b) above and (c) front

(a)



(b)



Figure S7. Photos of reactor (a) when finishing reaction and (b) after self-precipitation for 24 hours

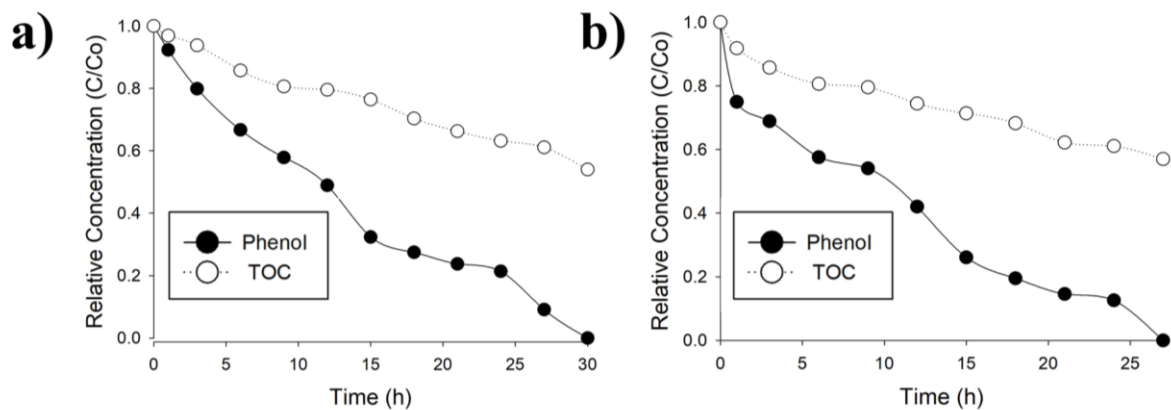


Figure S8. Phenol concentration vs. TOC concentration during (a) photo-Fenton and (b) photocatalytic reactions by MgAC-Fe₃O₄ [0.05 g]/TiO₂ on pilot scale

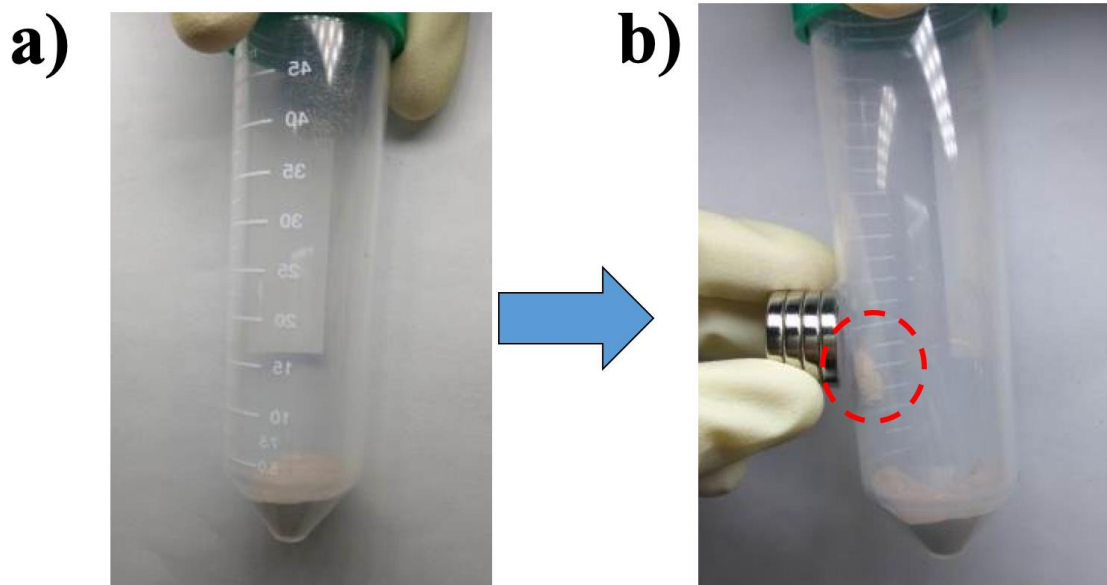


Figure S9. MgAC-Fe₃O₄ [0.05 g]/TiO₂ hybrid nanocomposites (a) before and (b) after applied magnetic field

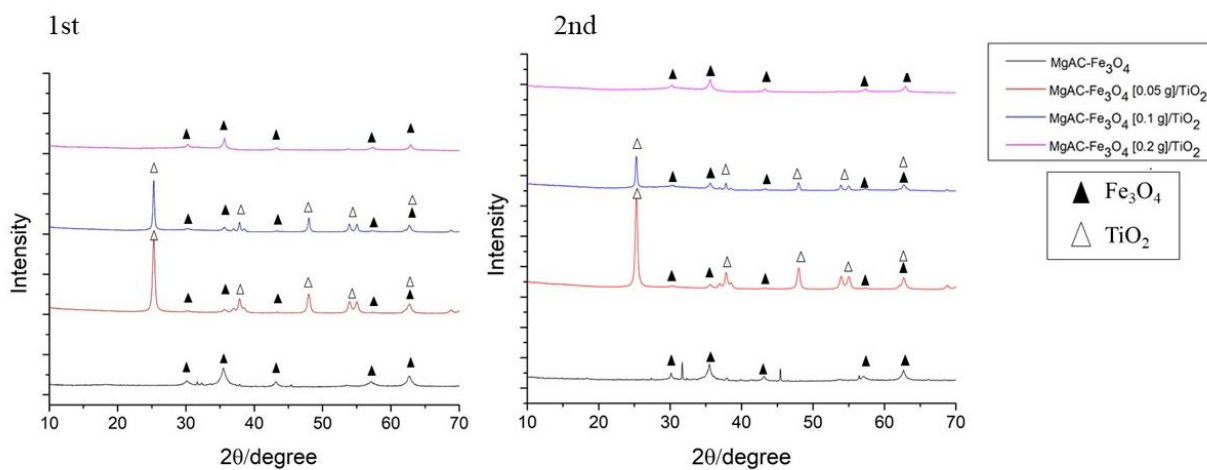


Figure S10. XRD was conducted two times to confirm the absence of TiO₂ peaks due to the excess MgAC-Fe₃O₄ loading.

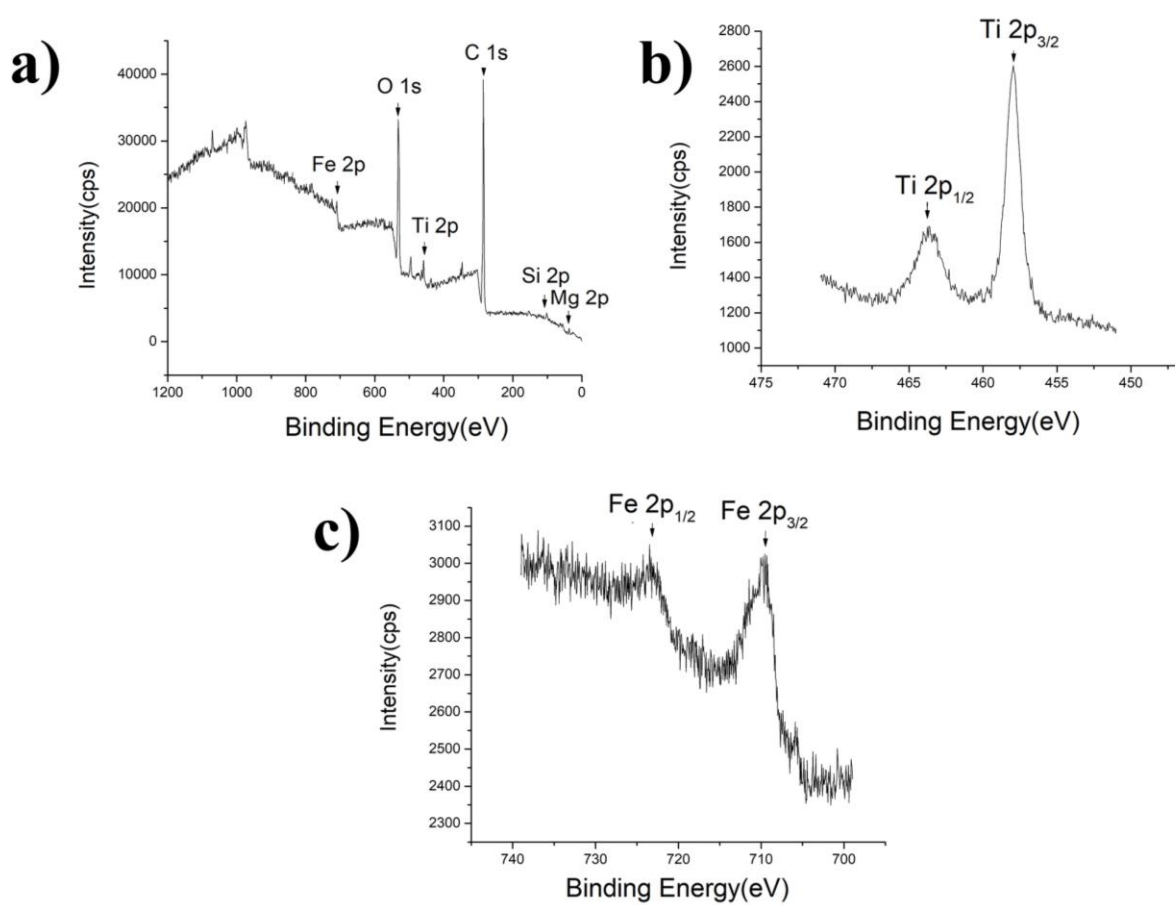


Figure S11. XPS spectra of MgAC-Fe₃O₄ [0.05 g]/TiO₂: (a) full spectra, (b) Ti 2p region, (c) Fe 2p region

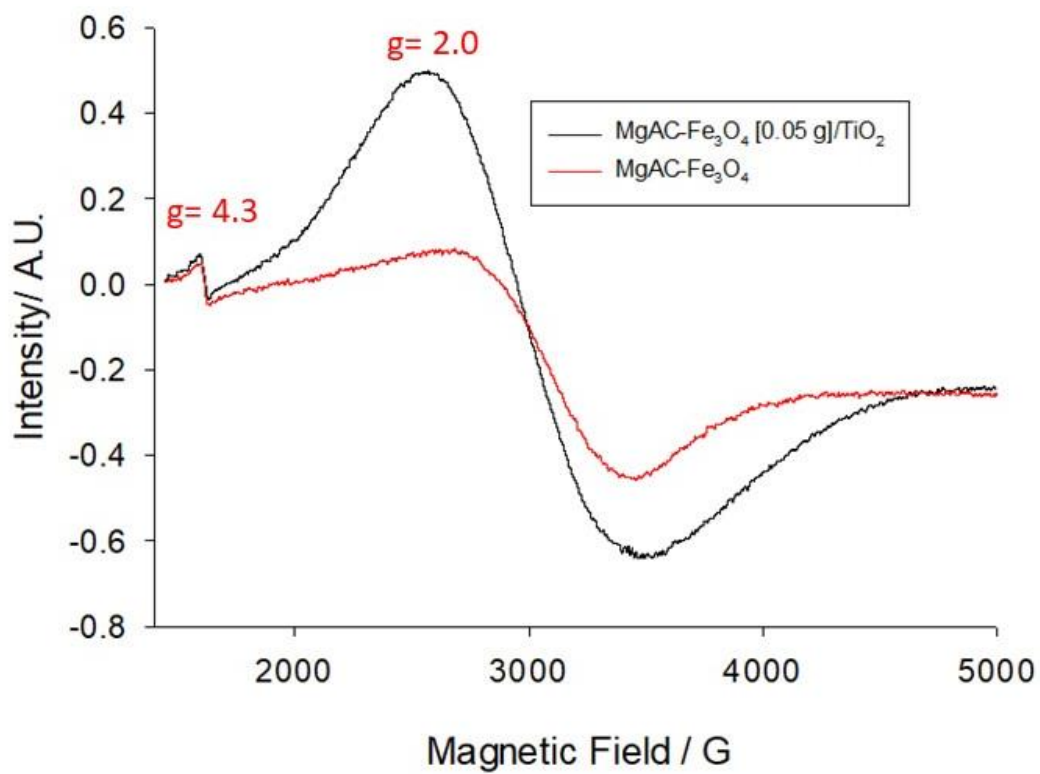


Figure S12. ESR spectra of MgAC-Fe₃O₄ and MgAC-Fe₃O₄ [0.05 g]/TiO₂

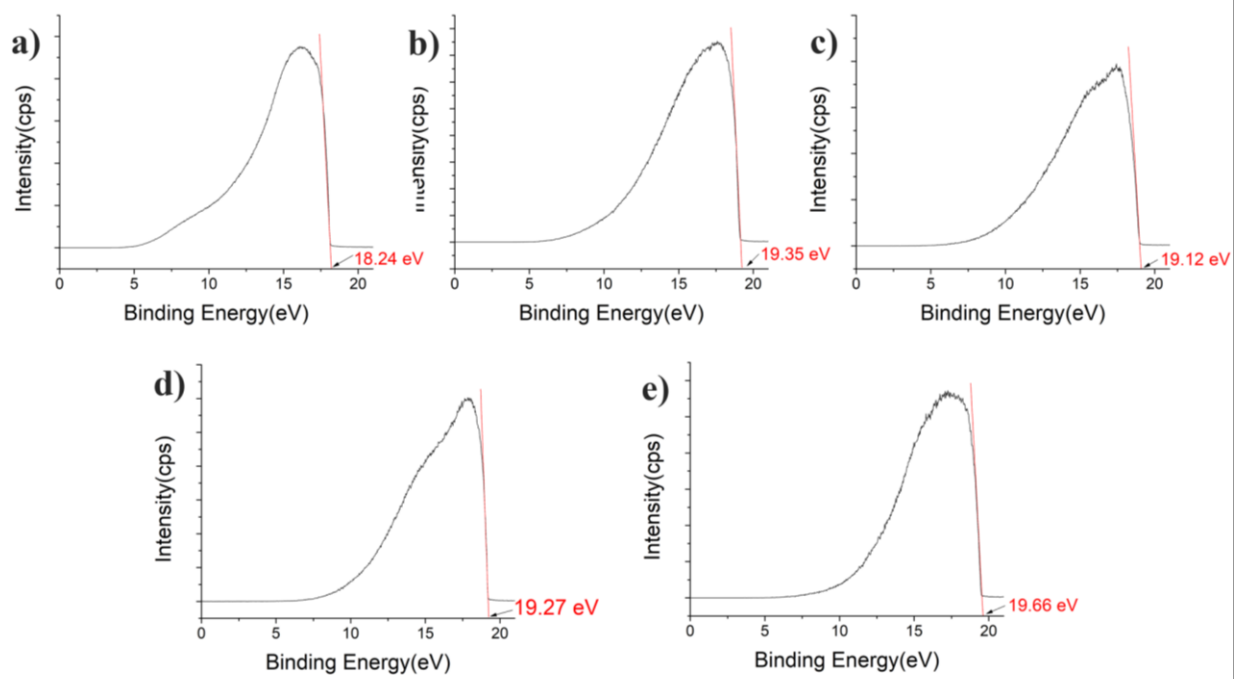


Figure S13. UPS spectra and secondary electron cutoff energies (E_{cutoff}) of (a) MgAC-Fe₃O₄, (b) MgAC-Fe₃O₄ [0.05 g]/TiO₂, (c) MgAC-Fe₃O₄ [0.1 g]/TiO₂, (d) MgAC-Fe₃O₄ [0.2 g]/TiO₂, and (e) MgAC-TiO₂

Table S1. Elemental compositions (wt. %) of as-prepared samples by X-ray fluorescence (XRF) spectrometry

Sample	MgAC-Fe₃O₄ [0.05 g]/TiO₂	MgAC-Fe₃O₄ [0.1 g]/TiO₂	MgAC-Fe₃O₄ [0.2 g]/TiO₂	MgAC-Fe₃O₄
MgO (%)	3.5	3.4	3.5	6.8
SiO ₂ (%)	0.91	0.73	1.1	2.4
TiO ₂ (%)	88.1	81.5	74.3	0.01
Fe ₃ O ₄ (%)	7.48	14.4	21.2	90.8

Table S2. Work functions of materials in this study

Sample	Work function (eV)
MgAC-Fe ₃ O ₄	2.98
MgAC-Fe ₃ O ₄ [0.05 g]/TiO ₂	1.87
MgAC-Fe ₃ O ₄ [0.1 g]/TiO ₂	2.10
MgAC-Fe ₃ O ₄ [0.2 g]/TiO ₂	1.98
MgAC-TiO ₂	1.56

Table S3. Cost estimations

Product	Estimated Cost (USD/kg)*
MgAC	389.80
MgAC-Fe ₃ O ₄	583.45
MgAC-TiO ₂	1476.24
MgAC-Fe₃O₄ [0.05 g]/TiO₂	1479.80

*Estimated cost includes Republic of Korea industrial-electrical fees.

*Estimated cost does not include the cost of 4% H₂/Ar.