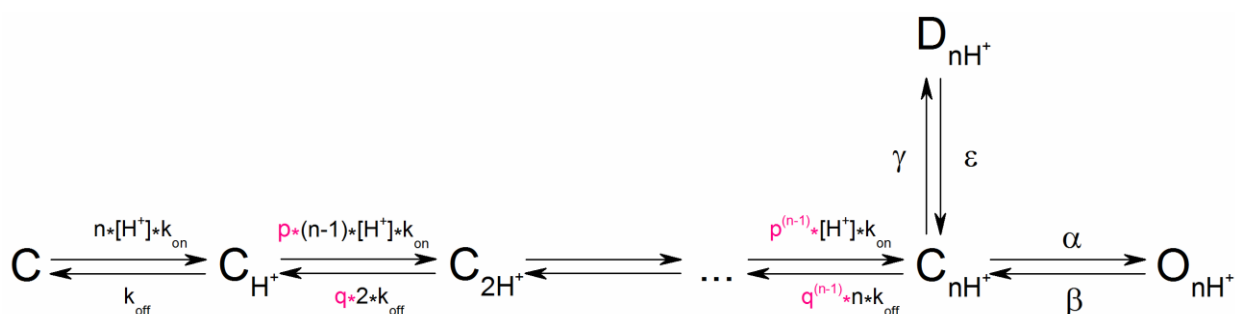


1 Supplementary Materials

2 We simulated the kinetic of proton-activated currents at the ASIC1a channel
 3 (taking into account the rate of change of the solution) using the method of
 4 numerical solution of a system of kinetic equations using an algorithm similar to
 5 that described [1]. The accuracy of this modeling method was verified by
 6 comparing our calculations with analytical solutions for a three-state [2] and
 7 four-state [3] models of receptor activation by the ligand. Comparison of
 8 dose-response curves fitted to the maximum amplitude of the current generated by
 9 our program with these analytical solutions showed complete coincidence of
 10 numerical simulation predictions (the relative amplitude error was less than 10⁻⁷%
 11 over the entire range of ligand concentrations (10⁻⁹–10⁻³ M). Thus, our algorithm can
 12 be successfully used to simulate the currents of ligand-activated receptors,
 13 including rapidly activated, such as ASIC channels.

14 As is known from the literature the activation of ASICs may occur after the
 15 binding of numerous protons (≤16) [4,5]. At first, to simulate the proton dependence
 16 of ASIC1a activation, we developed a theoretical Model 1 describing several
 17 consecutive closed states, which can eventually turn into an open or desensitized
 18 state:



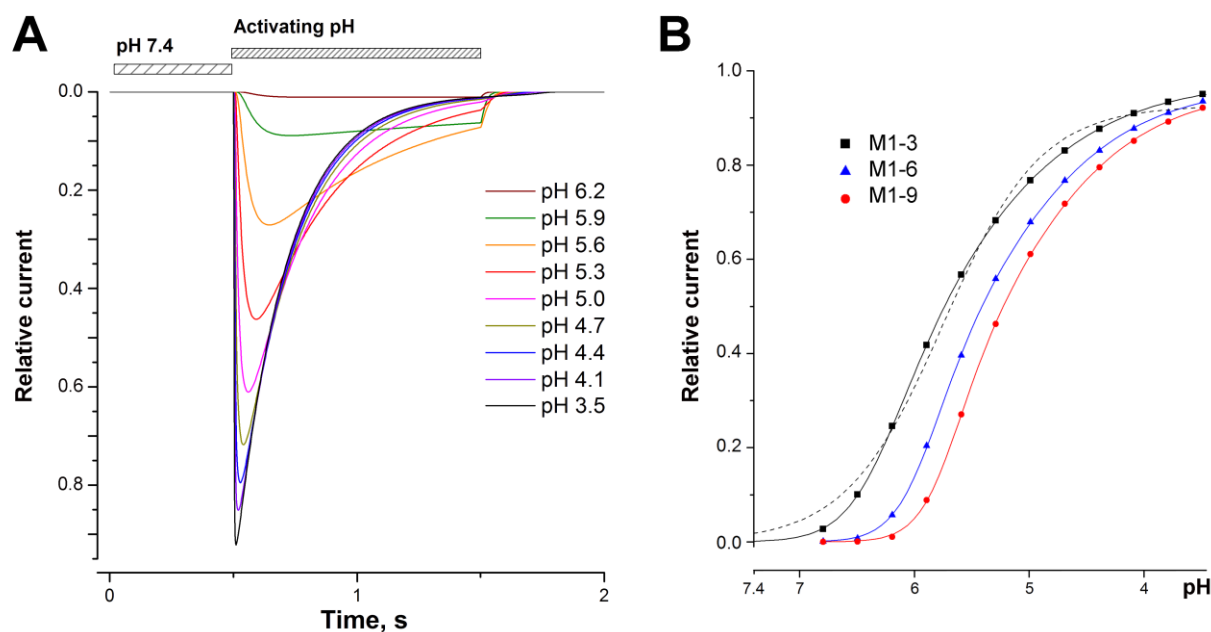
19

20 **Scheme S1.** Model 1/2. Model 2 with introduced cooperativity factors (*p* and *q* < 1).

21 The kinetic constants $k_{on} = 2 \times 10^9 \text{ M}^{-1}\text{s}^{-1}$ and $k_{off} = 10^3 \text{ s}^{-1}$ for protons binding to
 22 ASIC1a were suggested early in literature by [6]. Other constant: The rate constant
 23 of the channel opening $\alpha = 1000 \text{ s}^{-1}$; the rate constant of the channel closing $\beta = 1000$
 24 s^{-1} ; the rate constants of the channel transition to desensitization $\gamma = 10 \text{ s}^{-1}$, and the
 25 rate constants of the channel transition from desensitization $\epsilon = 0.02 \text{ s}^{-1}$, were chosen
 26 on the basis of the speed and magnitude of the desensitization current decrease.
 27 The coefficients "p" and "q", highlighted in red, reflect the cooperativity of the
 28 interaction of protons with the channel and they were introduced into the Model 2
 29 (see description below).

30 We investigated dose-dependence of the probability of the channel opening
 31 for Model 1 with 3, 6, and 9 identical binding sites (M1-3, M1-6, M1-9). This pool of
 32 sites number was chosen in view of homotrimeric ASIC1a organization. All three
 33 models reproduced similar shape of the current generated by various proton

34 concentrations from standard conditioning pH 7.4 (change solution time is 50 ms;
35 time application is 1 s) (Figure S1A).



36
37 **Figure S1.** (A) The set of generated traces predicted for a model M1-9. (B) The dose
38 dependence curves obtained for models M1-3, M1-6, M1-9 by fitting with equation F₂ (solid
39 lines) and by logistic equation F₁ for model M1-3 (dash line).

40 The pH-dependence of the channel activation was analyzed for all three
41 models (Figure S1B). In all cases the dependences of the maximum amplitude of the
42 current versus proton concentration keep an asymmetrical shape of curve like the
43 experimental curve, and poorly fitted by the logistic equation. Otherwise a reliable
44 fitting of we fitted the dose-response with the following equation F₂:

$$45 F(x) = A / (1 + (x/[pH_{501}])^{n_{H1}}) \times (1 + (x/[pH_{502}]^{n_{H2}})).$$

46 The dose-dependence activation of all models is perfectly fitted by this equation.
47 The Hill coefficients obtained by the fittings are shown in Table S1.

48 **Table S1.** The Hill coefficients trend for selected models.

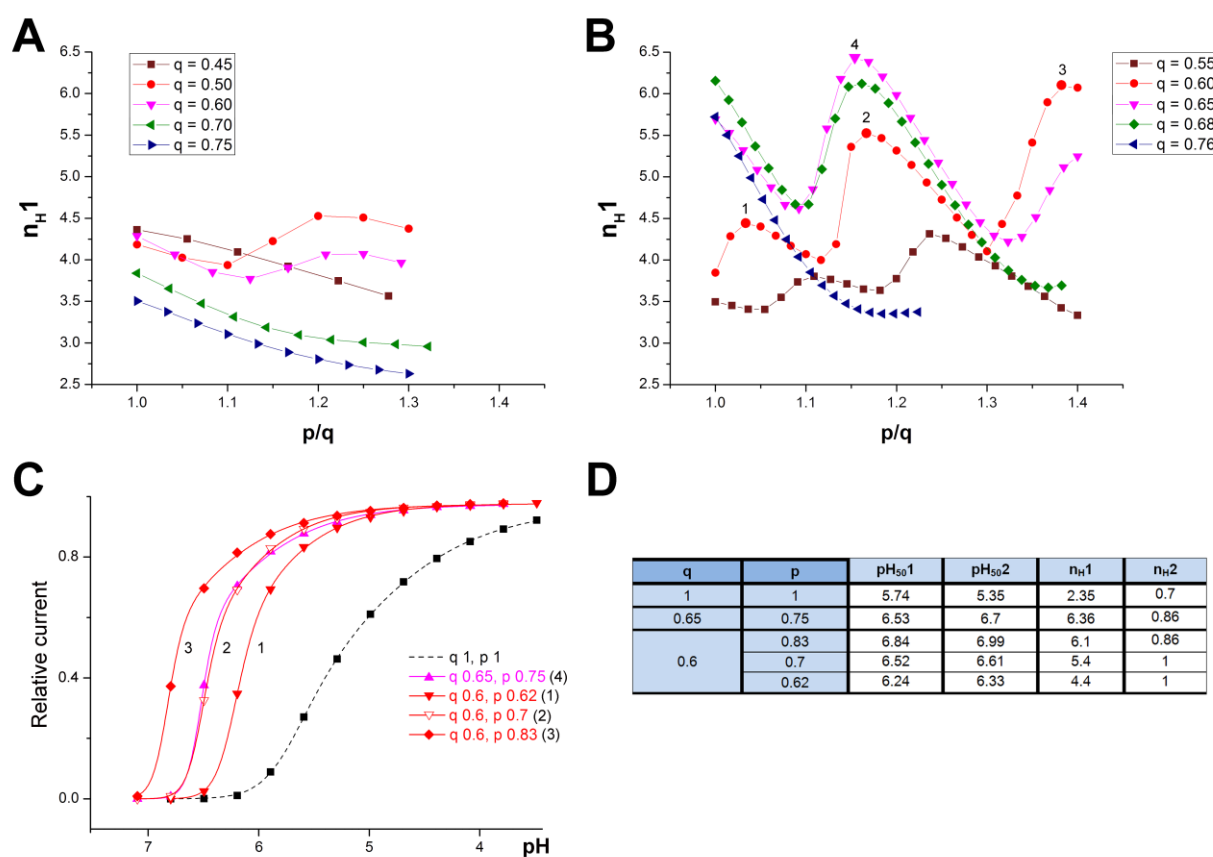
	M1-3	M1-6	M1-9
n _{H1}	1.71	2.17	2.35
n _{H2}	0.61	0.68	0.70

49 The value of the n_{H2} always remains less than one (as in the experimental
50 data). The value n_{H1}, which determines the steepness of the left part of the
51 proton-dependence curve of activation (Figure S1B) has a tendency to grow when a
52 number of proton sites in Model 1 is increased, but this tendency is not enough to
53 achieve higher n_{H1} values. Therefore, the Model 1 without modification failed to
54 obtain high n_{H1} values close to experimental (more than 6) by a simple increasing of
55 proton binding sites number.

56 In order to obtain higher n_{H1}, we designed the Model 2 with two parametric
57 coefficients p and q. We assumed that the binding of protons is a cooperative

58 process, causing the conformational rearrangements in the channel that were
 59 described [7] as a collapse of the pocket where protons bind. As a result, each
 60 successive proton will have different binding/dissociation characteristics than the
 61 previous one. Thus, we believed that the rate constants of both binding and
 62 dissociation decrease with each subsequent process of interaction of the proton
 63 with the channel and introduced the cooperativity coefficient $p < 1$ (for binding
 64 constants) and coefficient q (for dissociation constants) that always less than p .

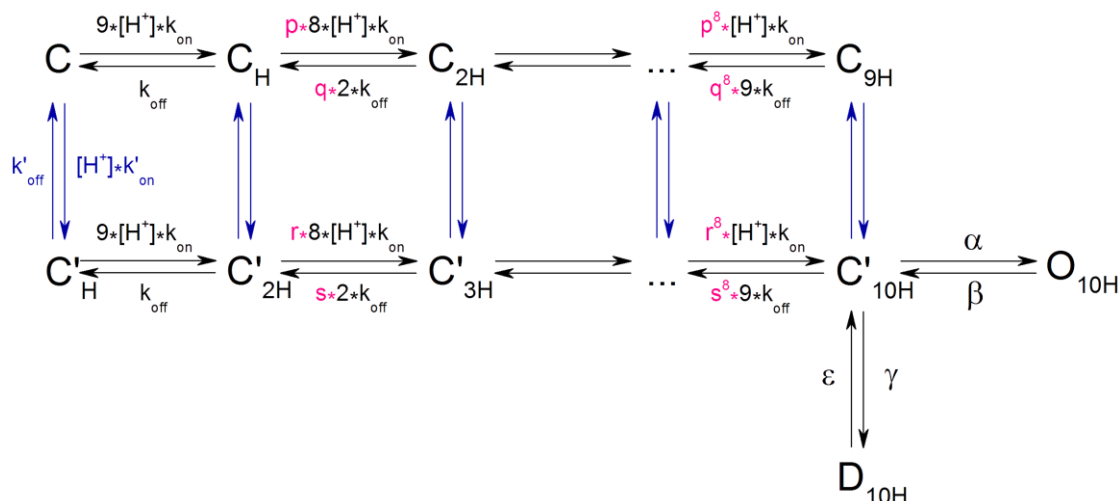
65 As a result, we managed to achieve an increase in the value of n_{H1} (Figure
 66 S2A, B). For the M2-6, n_{H1} does not exceed a value of 4.5 for any values of the “ p ”
 67 and “ q ” coefficients (Figure S2A). For the M2-9 model, the value of n_{H1} , about 6.4,
 68 is achieved only in a narrow range of q and p values. For further modeling, we take
 69 the q p and values of the main maximum (Figure S2B,C). Thus, we assume that the
 70 number of proton-binding sites equal to nine is a necessary and sufficient, and best
 71 calculation is produced for q coefficient of 0.65 and p coefficient of 0.75 (Figure
 72 S2D).



73

74 **Figure S2.** Optimization of the models. (AB) of the Plot of different variants of the q and p
 75 coefficients affecting the n_{H1} coefficient in the model M2-6—panel (A) and model
 76 M2-9—panel (B). Dose-dependence curves (C) and tabular data (D) obtained by fitting
 77 with equation F_2 for M2-9 model with various cooperativity coefficients q and p (points 1–4
 78 from Panel B) vs. control $q = p = 1$.

79 It should be noted, that the M2-9 model with selected p and q values 0.75 and
 80 0.65, respectively, give the satisfactory result for n_{H1} and n_{H2} but false pH_{501} and
 81 pH_{502} values. Experimental pH_{501} is greater than pH_{502} (Figure 6a in main text)
 82 while this model produce more high value for pH_{502} (Figure S2D). To overcome this
 83 inconsistency, we introduced into Model 2 an additional proton binding site with
 84 and noncooperative less affine site (Model 3; see also the scheme and text in the
 85 main results).



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Scheme S2. Model 3

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We introduced the following assumptions into this model:

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1) The tenth site is low affine, $k'_{on} = 2 \times 10^9 M^{-1}s^{-1}$ and $k'_{off} = 1.2 \times 10^3 s^{-1}$ (the last constant was taken based on the difference between pH_{501} and pH_{502} in the experiment);

90

2) Proton binding in the main (highly cooperative) pool does not affect proton binding in the low affinity region;

92

3) Proton binding in the low affinity site also does not affect proton binding in the cooperative pool, i.e., $p = r$, $q = s$.

94

95

Under these assumptions, the entire dose-dependence is shifted to the right in the region of acidic pH. Returning the curve to the left is possible by increasing the cooperativity of the binding by increasing the coefficients of cooperativity p and r. This allowed us to obtain a dose-dependence, which is fitted by equation F₂ with parameter values close to the experimental ones (see Table S2).

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Table S2. Comparison of the parameters of the pH dependence of the model M3-10 activation with non-fixed and fixed per unit coefficient n_{H2} .

101
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	pH_{501}	pH_{502}	n_{H1}	n_{H2}
M3-10 (n_{H2} free)	6.62	6.61	5.6	0.77
M3-10 (n_{H2} fixed)	6.65	6.53	6.4	1
Oocyte experimental data	6.67 ± 0.01	6.59 ± 0.01	6.7 ± 0.5	0.96 ± 0.06

103 It should be noted that, although the parameters of the fitting of the model
104 agree well with the experimental data, the coefficient n_{H1} becomes smaller. This can
105 be compensated for by increasing the number of highly operational sites, or by
106 fixing the second coefficient when fitted by equation F₂. Since the experimental data
107 showed the second coefficient very close to 1, we adjusted the data of the M3-10
108 model with the coefficient n_{H2} fixed to the value 1. These fittings are shown in the
109 main text and in Figure 6.

110

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