

Supporting Information

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Base-Assisted Conversion of Protonated D-Fructose to 5-HMF: Searching for Gas-Phase Green Models

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A.R. Conceptualization:Lead; Investigation:Equal; Methodology:Lead; Writing - Original Draft:Lead; Writing -Review & Editing:Lead

A.T. Data curation:Equal; Investigation:Equal; Writing - Original Draft:Lead; Writing - Review & Editing:Equal G.d. Data curation:Lead; Writing - Original Draft:Lead; Writing - Review & Editing:Equal F.P. Data curation:Equal; Writing - Original Draft:Supporting

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Fig. 1S. (a) CAD spectrum of $[162 \cdot H]^+$ ions at m/z 163 (in red) isolated from the sequence $[180 \cdot H \cdot B]^+ \rightarrow [162 \cdot H \cdot B]^+ \rightarrow [162 \cdot H]^+$. (b) CAD spectrum of isolated m/z = 163 ions (in red) obtained from the acid catalyzed dehydration of D-fructose (D-fructose dissolved in H₂O/methanol, 0.1% formic acid).



Fig. 2S. (a) CAD spectrum of $[96 \cdot H]^+$ ions at m/z 97 (in red) isolated from a solution containing the D-fructose/NH₃ protonated adduct, $[180 \cdot H \cdot NH_3]^+$. (b) CAD spectrum of isolated $[96 \cdot H]^+$ ions at m/z = 97 (in red) obtained from the standard protonated 2-furaldehyde, $[2 \cdot FA]H^+$.



Fig. 3S. Optimized geometries at the B3LYP/6-31G(2df,p) level of theory of the transition states identified on the potential energy surface of the protonated D-fructose dehydration assisted by ammonia. Bond lengths in angstroms.



Fig. 4S. Optimized geometries at the B3LYP/6-31G(2df,p) level of theory of the transition states identified on the potential energy surface of the protonated D-fructose dehydration after ammonia loss (path 3). Bond lengths in angstroms.