Supplementary Information for "Selenium isotopes as tracers of a late volatile contribution to Earth from the outer Solar System"

5

6 Supplementary Notes

7 <u>S1 Sample description</u>

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9 Detailed sample information is summarized in Supplementary Table S1. Nine spinel-bearing 10 mantle peridotites from different post-Archean settings and localities including the Pyrenean and 11 Ronda orogenic massifs and the Lanzo and External Ligurides transitional massifs were analyzed 12 in this study. We have also included two peridotite xenoliths hosted by Cenozoic alkali basalts 13 from the French Massif Central (FMC) to cover a wider range of tectonic settings and ages. The 14 incorporation of Ronda, Pyrenees and External Ligurides massifs to the subcontinental lithospheric mantle (SCLM) has been dated as Mid-Proterozoic on the basis of whole-rock Os model ages^{48,49}. 15 16 According to Nd model ages, the northern and central part of the Lanzo massif (sampled by peridotite L212) were isolated from the convective mantle in the late Proterozoic/early 17 Palaeozoic⁵⁰. Selected xenoliths come from the southern part of the FMC, interpreted as SCLM 18 accreted or rejuvenated during the Variscan orogeny⁵¹. 19

Most of the selected peridotites have been extensively studied for their major and trace element compositions, including platinum-group element^{23,24,47,52-57} (Supplementary Table S1), and Se abundances ^{5,16,22,44,47} (Supplementary Table S2). Exception is sample 15/EDL-007, for which we provide the whole-rock major and trace element data in Supplementary Table S4. These samples cover a wide range of rock texture, mineralogy and geochemical features usually

25 reported for mantle rocks. Samples mostly from the Pyrenees comprise LREE-depleted, 26 amphibole-free, coarse-granular to mylonitic fertile orogenic lherzolites (TUR 7, FONB 93, 84-1, 27 15 EDL-007, LiX) along with a highly refractory harzburgite (64-3). All these rocks are devoid 28 of modal metasomatism features apart from minor pargasite in 84-1. Three other orogenic 29 peridotites display clear evidence of melt-rock interactions: L212 (Lanzo), a cpx-poor lherzolite, 30 displays selective enrichment in LREE over HREE interpreted as produced by chromatographic fractionation of a percolating melt ⁵⁰. The two Ronda lherzolites are from the fine-granular 31 32 domain of the Ronda peridotite massif that was affected by pervasive porous melt flow⁵⁸. DR-29 33 is an olivine-rich "secondary" lherzolite resulting from melt/rock reactions at increasing melt mass. Moreover, it contains abundant disseminated Ti pargasite. DR 33 is an amphibole-free cpx-34 35 poor lherzolite showing abundant euhedral Cr spinel and secondary cpx, which were precipitated 36 by melt-rock reaction at decreasing melt volume. The two French Massif Central xenoliths are a 37 coarse-granular LILE and LREE enriched lherzolite with secondary phases (5SC30) and a depleted lherzolite (EG3)^{23,24}. 38 39 Selected peridotites show no correlation with petrogenetic indicators of melt

depletion/metasomatism such as Al₂O₃ contents (Fig. 1b), or Pd/Ir ratios, considered sensitive
 tracers of refertilization and metasomatic processes affecting the HSEs and Se-Te systematics^{16,59}
 (Supplementary Figure S1), further supporting that partial melting and metasomatism do not
 produce a Se isotopic variation in mantle peridotites.

44 <u>S2 Reproducibility and accuracy of Se measurements</u>

45 Accuracy of Se concentration analyses was checked against the USGS reference material BHVO-

46 2 (Hawaiian basalt). We report an average value of $170 \pm 4 \text{ ng g}^{-1}$ Se (n = 4) in agreement with

47 previously published values^{20,44,60} (Supplementary Table S2). Reproducibilities of Se

48 concentration measurements estimated from at least 4 replicate analyses of peridotites were better
49 than ~3% r.s.d. (Supplementary Table S2), similar to the long-term analytical reproducibility of
50 3% r.s.d. reported by Yierpan, et al. ²⁰ in different reference materials.

51 The long-term analytical reproducibility of Se isotope measurements is 0.07‰ based on 52 multiple analyses of the inter-laboratory standard solution MH-495 (Supplementary Table S3). In 53 addition, the analytical reproducibility obtained from peridotite samples analyzed multiple times 54 $(n \ge 5)$ is 0.10‰ (see Methods and Supplementary Table S2). An exception is harzburgite 64-3, 55 with a relatively poor reproducibility of $\pm 0.20\%$ (2 s.d., n=3) that may be attributed to sample heterogeneity or to measurement-related uncertainties such as low signals on ⁸²Se. For this 56 57 particular sample, it has been shown that heterogeneity plays an important role in its Se 58 concentration reproducibility, where the more material was digested, the better the reproducibility⁴⁴. Although in our case the Se concentration reproducibility is 3% r.s.d. 59 (Supplementary Table S2), the depleted Se content (Se= 5.79 ± 0.17 ng g⁻¹, Supplementary Table 60 61 S2) may contribute to the poor reproducibility of its Se isotopic composition. On the other hand, 62 internal errors (s.e.) of Se isotope measurements are inversely correlated with the measured signals on ⁸²Se and with the amount of Se analyzed (Supplementary Table S2, Supplementary 63 Figure S2). These correlations show that analyses of low Se contents (~7 ng of Se) have the 64 highest internal errors (up to 0.11‰, 2 s.e.; Supplementary Table S2), as also described by 65 Kurzawa, et al. ²¹ and which may result in poor reproducibility of the $\delta^{82/76}$ Se measurements. 66 However, no apparent correlation is observed between the measured signals on ⁸²Se (as low as 67 0.2 V) or the amount of Se analyzed and the $\delta^{82/76}$ Se of the samples (Supplementary Table S2, 68 69 Supplementary Figure S2).

No high-precision Se isotopic data have been published for peridotites following the
 pioneering study of Rouxel, et al. ²⁷, who provided the first Se isotopic value for the USGS

72	reference peridotite (PCC-1) and for the BSE (Fig. 2). Thus, we performed an accuracy test for
73	the type of peridotite-matrix analyzed here, already tested in standard solutions ²¹ and previously
74	used for testing other isotope systems ^{61,62} . Increasing amounts of the fertile lherzolite TUR 7
75	(from ~5 to 50 ng of Se) were admixed to a constant amount of MH-495 reference material (~10
76	or 200 ng of Se) (Supplementary Table S5). Each mixture was then spiked accordingly to its Se
77	concentration and processed as a normal sample. The measured $\delta^{82/76}Se$ of the mixtures are
78	positively correlated with the fraction of Se from TUR 7, forming a mixing line between the
79	$\delta^{82/76}$ Se of the end members TUR 7 and MH-495 (r ² = 0.99) and remaining within maximum
80	uncertainty of 0.14‰ (Supplementary Figure S3). These results demonstrate that we can discard
81	potential bias induced by peridotite-matrix or interferences during analysis.
82	S3 Potential mixtures between chondrite-like material
83	a) Mixtures between CM chondrites Murchison and Mighei

A potential mixture between the CM chondrites Murchison and Mighei could result in a Se isotope signature that fits with that of the BSE. To calculate the possible range of mixtures we use the following mass balance equation:

87
$$\frac{f \times C_1 \times \delta^{82/76} Se_1 + (1-f) \times C_2 \times \delta^{82/76} Se_2}{f \times C_1 + (1-f) \times C_2} = \delta^{82/76} Se_{BSE}$$
(Eq.1)

- f : Fraction of Murchison in the mixture
- 1 f: Fraction of Mighei in the mixture
- C_1 : Se concentration of Murchison (µg/g)
- C_2 : Se concentration of Mighei ($\mu g/g$)
- $\delta^{82/76}Se_1$: Se isotope signature of Murchison (‰)
- $\delta^{82/76}Se_2$: Se isotope signature of Mighei (‰)
- $\delta^{82/76}Se_{BSE}$: Se isotope signature of BSE from this study (‰)

97 Se concentrations of Murchison $(12.7 \pm 0.8 \ \mu g \ g^{-1})$ and Mighei $(24.3 \pm 1.4 \ \mu g \ g^{-1})$ and 98 their respective $\delta^{82/76}$ Se values $(0.20 \pm 0.13\%$ and $-0.30 \pm 0.10\%)$ used in the equation are from 99 Labidi, et al. ¹⁹. Our calculations result in a *f* value, fraction of Murchison in the mixture, that 100 ranges between 0.49 and 0.89 (*f*= 0.69 ± 0.20) and thus allows for proportional contributions 101 from Mighei that range between 0.11 and 0.51. The uncertainty was calculated by propagating 102 the errors associated with the concentration of Se in both chondrites (Supplementary Table S6) 103 and the uncertainties of the $\delta^{82/76}$ Se values¹⁹.

104 b) Potential mixtures of CI and different types of chondrite-like material

105 Potential mixtures between CI and different types of chondrite-type material might also account 106 for the Se isotope composition of the BSE. Following Eq. 1, we calculate mixtures between CI 107 and CO, CV, each sub-type of CM (Mighei and Murchison), ordinary and enstatite chondrites 108 that would result in a Se isotope signature that fits with that of the BSE (Supplementary Table 109 S7). Our calculations show that for all potential mixtures, the average contribution from a CI 110 chondrite-like material is always higher than 85%, and in most cases higher than 90% (CI-111 Enstatite; CI-Murchison and CI-Mighei), clearly indicating that carbonaceous chondrites 112 dominated the late-accreted material. The uncertainty for each mixture was calculated by 113 propagating the errors associated with the concentration of Se in both components (Supplementary Tables S6 and S7) and the uncertainties of the $\delta^{82/76}$ Se values¹⁹. 114

115 <u>S4 Mass of the late veneer and its contribution to the volatile budget of the BSE</u>

116 To calculate the mass of the late veneer we considered that $\sim 40\%$ of S is in the pre-late veneer

117 mantle³⁰ and use a BSE S content of $200 \pm 40 \ \mu g/g^{31}$. To obtain the percentage of the Earth's

118 mass added to the BSE by a CI or CM chondrite-like late veneer ('a' term) we use the same mass

119 balance approach than in the mixture of chondrite material (see above):

120	
121 122	$C_{BSE}^{A} \times M_{BSE} = C_{CC}^{A} \times a \times M_{E} + C_{PLV}^{A} \times (M_{BSE} - a \times M_{E}) $ Eq.2
123 123 124 125 126 127 128 129	C_{BSE}^X : Concentration of element X in the BSE C_{CC}^X : Concentration of element X in the chondrite C_{PLV}^X : Concentration of element X in the pre-late veneer mantle <i>a</i> : Percentage of the Earth's mass added to the BSE by the late veneer M_{BSE} , M_E : mass of the BSE and the Earth from ^{31, and references therein}
130	Abundances (\pm s.d.) of CI and CM chondrite elements used in Eq. 2 are reported in
131	Supplementary Table S6 ^{5,18-19,60,63} . Our results show that $0.15 \pm 0.03\%$ of the Earth's mass would
132	be added by a CI chondrite-like late veneer, which is lower than the 0.26 \pm 0.05 % resulting if we
133	consider CM chondrite-like late veneer. The associated uncertainties in both calculations were
134	calculated by propagating the errors associated with the concentration of S, Se and $\delta^{82/76}$ Se, both
135	in the BSE and chondrites (CI or Murchison and Mighei mixture; Supplementary Table S6).
136	More complex scenarios involving mixtures of CI and different types of chondrite-type material
137	(c.f. S3b) would result in a late veneer of 0.15-0.16% the Earth's mass. Abundances (\pm s.d.),
138	fractions and estimated late veneer mass of potential mixtures are reported in Supplementary
139	Table S7.
140	Since the late veneer delivered 100% of Se and Te based on their partition coefficients ⁹ , a
141	CI chondrite-like late veneer would result in Se and Te BSE abundances of 47 ± 9 and 5 ± 1 ng g ⁻
142	¹ , respectively (Supplementary Table S6). These BSE estimates are within error of those obtained
143	assuming a CM chondrite-like late veneer (Se = 65 ± 17 ng g ⁻¹ ; Te= 6 ± 1 ng g ⁻¹ ; Supplementary
144	Table S6). Our results imply Se and Te BSE abundances lower than previous estimates based on
145	Se and Te contents and relative proportions in mantle fertile lherzolites (~80 ng g ⁻¹ of Se and ~11

146 ng g^{-1} of Te $)^{5,6,31}$. It has been shown however that Se-Te systematics in peridotites might not be

representative of the BSE and instead represent the result of refertilization and metasomatic
processes^{16,17}.

149	Regarding other elements such as water, carbon and nitrogen, the contribution of the late
150	veneer would have been heterogeneous and it relies on how well constrained are the estimate
151	BSE abundances of these elements (Supplementary Table S6). Following Palme and O'Neill ³¹
152	water content in the BSE, a CI chondrite-like late veneer would have provided either ~20-40% of
153	water if the Earth has chondritic refractory lithophile element ratios, i.e. a water abundance of
154	$1100 \pm 220 \ \mu g \ g^{-1}$, or ~30-60% of water if Earth is non-chondritic (BSE water of 675 $\pm 135 \ \mu g \ g^{-1}$
155	¹). Similarly, a CM chondrite-like late veneer would have delivered either ~25-45% or ~40-70%
156	of water assuming a chondritic or a non-chondritic model, respectively. Hirschmann ⁶⁴
157	constrained the carbon content in the BSE, providing two different estimates of 90 \pm 27 $\mu g~g^{\text{-1}}$
158	and 108 \pm 34 μg g^-1. According to those estimates, a CI chondrite-like late veneer would have
159	delivered ~57-120% or ~46 – 100% of the BSE carbon, respectively, whereas a CM chondrite-
160	like late veneer would have delivered ~60-130% or ~50 – 110%. These estimates imply that a CI
161	or CM chondrite-like late veneer could have delivered more than 100% of the BSE carbon.
162	Recent studies, however, have suggested that ~10-30% of the present day carbon budget
163	corresponds to pre-late veneer mantle abundances ³⁹ , which supports that the late veneer delivered
164	~80% of the BSE carbon budget, in agreement with the average values provided by our
165	calculations. Nitrogen abundance in the BSE (1.9 \pm 0.37 μg g $^{-1},$ $^{65})$ is much lower than the
166	calculated amounts delivered by the late veneer (CI chondrite contribution: 4.3 \pm 1.1 μg g^-1; CM
167	chondrite contribution: $3.8 \pm 1.8 \ \mu g \ g^{-1}$; Supplementary Table S6). However, the excess nitrogen
168	abundance (CI chondrite contribution: between 120 to 320%; CM chondrite contribution:
169	between 80 to 320%) can be reconciled with considerable atmospheric loss 41 after the late
170	veneer. This does not preclude that nitrogen was already in the mantle previous to the late veneer.

- 171 Supplementary Table Captions
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- 173 Supplementary Table S1.
- 174 Details on rock types, localities, Al₂O₃ content and Pd_N/Ir_N ratios of selected peridotites.
- 175 Supplementary Table S2.
- 176 Detailed information about the Se data obtained per sample measurement together with published

177 data.

178 Supplementary Table S3.

- 179 The $\delta^{82/76}$ Se value of the inter-laboratory standard MH 495 (30 ng mL⁻¹ Se) measured together
- 180 with the samples in this study and previously reported

181 Supplementary Table S4.

182 New major and trace element data for peridotite 15/EDL-007.

183 Supplementary Table S5.

184 Se isotope data of mixtures between MH-495 standard and peridotite TUR-7.

185 Supplementary Table S6.

- 186 Recent estimates of the volatile content of carbonaceous chondrites and the BSE, together with
- 187 the estimated contribution of a CI and CM chondrite-like late veneer.

188 Supplementary Table S7.

- 189 Summary of potential mixtures between CI chondrite-like material and different types of
- 190 chondrites that could result in a Se isotope signature that would overlap that of the BSE.

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Supplementary Figures

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198 Supplementary Figure S1. Pd_N/Ir_N vs. Se isotope data of mantle peridotites.

199 Circles represent orogenic and transitional peridotites and triangles mantle xenoliths. N = CI-

200 chondrite normalized ³¹. Error bars indicate 2 s.d. uncertainties of more than two combined

201 measurements. If not available, the analytical uncertainty obtained for repeated analyses of

202 peridotites is reported ($\pm 0.10\%$, 2s.d.) (see Methods). The shaded field represents the 2 s.d. of

203 the mean of peridotites as discussed in the text. Pd and Ir concentrations are taken from published

204 data using the same pulverized powder batch (Supplementary Table S1).





208 Supplementary Figure S2. Data obtained during Se isotope measurements of peridotites.

209 Internal errors (s.e.) vs. (a) measured signals on ⁸²Se and (b) amount of Se analyzed. (c) $\delta^{82/76}$ Se

- 210 (‰) vs. the amount of Se analyzed (Supplementary Table S2).
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216 **Supplementary Figure S3.** Accuracy test. $\delta^{82/76}$ Se (‰) values of the mixture between a

constant amount of the standard solution MH-495 (~10 or 200 ng of Se) and different amounts of

218 the fertile lherzolite TUR 7 (from ~5 to 50 ng of Se) vs. the fraction of Se from TUR 7. Error bars

219 represent the internal precision of sample measurement reported as 2 s.e. (Supplementary Table

220 S5). The doped samples fall on the range defined by the Se isotope compositions of the end

221 members MH-495 ($\delta^{82/76}$ Se = -3.25 ± 0.07 ‰, Supplementary Table S3) and TUR 7 ($\delta^{82/76}$ Se = -

222 0.06 ± 0.10 ‰; Supplementary Table S2) and remain within maximum uncertainty of 0.14‰ (2)

- 223 s.d.).
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225 Supplementary References

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