# **Supplementary Information**

# Single-Molecule Detection of Deoxyribonucleoside Triphosphates in Microdroplets

Boris Breiner, Kerr Johnson, Magdalena Stolarek, Ana-Luisa Silva, Aurel Negrea, Neil M. Bell, Tom H. Isaac, Mark Dethlefsen, Jasmin Chana, Lindsey A. Ibbotson, Rebecca N. Palmer, James Bush, Alexander J. Dunning, David M. Love, Olympia Pachoumi, Douglas J. Kelly, Aya Shibahara, Mei Wu, Maciej Sosna, Paul H. Dear, Fabian Tolle, Edoardo Petrini, Michele Amasio, Leigh R. Shelford, Monica S. Saavedra, Eoin Sheridan, Jekaterina Kuleshova, Gareth J. Podd, Barnaby W. Balmforth, and Cameron A. Frayling

Base 4 Innovation Ltd, Broers Building, J.J Thomson Avenue, Cambridge CB3 0FA, UK.

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# **Materials and Equipment**

Dye-labelled DNA oligos were purchased from ATDbio (Southampton, UK), All other oligos were purchased from Integrated DNA Technologies (Coralville, IA, USA). We use PAGE purified Probe oligos and double HPLC purified Capture and Ligation oligos.

Enzymes were purchased from New England Biolabs (Ipswich, MA, USA), Qiagen Beverly (Enzymatics) (Beverly, MA, USA) or Agilent (Santa Clara, CA, USA).

Spermine was purchased from Abcam (Cambridge, UK), Tetraethylenepentamine (TEPA; Acros, technical. grade) and Ambion nuclease-free water was purchased from Fisher Scientific (Waltham, MA, USA), and the latter was used throughout. All other reagents were purchased from Sigma (St. Louis, MO, USA) at the highest purity available, unless otherwise noted.

384-well PS Microplate were purchased from Greiner Bio-one (Frickenhausen, Germany; REF 784076).

# **Typical Detection Reaction and Time-Course – Bulk Solution**

In a typical, four-color 70 µL (1x) or 175 µL (2.5x) detection reaction, the following reagents were combined in a 1.5 mL tube.

**Supplementary Table 1:** Composition of a typical 70 µL (1x) or 175 µL (2.5x) four-color detection reaction.

Component	Stock Concentration	Final Concentration	Volume (µL) 1x	Volume (µL) 2.5x
Water	-	-	39.695	99.238
Buffer <sup>1</sup>	5x	1x	14	35
TIPP	2,000 U/mL	66.6 U/mL	2.33	5.825
ТЕРА	0.3%	0.015%	3.5	8.75
Spermine	100 mM	1 mM	0.7	1.75
Spermidine	100 mM	1 mM	0.7	1.75
Oligo mix <sup>2</sup>	100x	1x	0.7	1.75
<i>Pfu</i> Ultra II Fusion HS	100 Reactions/mL	1.43 Reaction/mL	1	2.5
Bst L.F. DNA pol.	8,000 U/mL	28.57 U/mL	0.25	0.625
<i>E. coli</i> ligase	10,000 U/mL	17.85 U/mL	0.125	0.3125
Total	-		63	157.5

<sup>1</sup> see Supplementary Table 8 for further information on the buffer composition <sup>2</sup> see Supplementary Table 7 for further information on the Oligo Mix

From the 2.5x volume mix, two samples of 63 µL were taken and added to either 7 µL of 250 pM dNTP or 7 µL of water, to give the 25 pM and 0 pM samples, respectfully. Each sample (25 pM or 0 pM) was split into five reactions by adding 14 µL of each sample into 5 separate PCR tubes. The five tubes (for each sample), were incubated in a PCR thermocycler for the required amount of time (Supplementary Table 2).

Supplementary Table 2: Description of the time-course used in typical, four-color detection reaction. The thermocycler program is 10 min @ 37 °C, followed by 90 min @ 69.5 °C.

Tube	Sample	Time Point (min)	Temperature (°C)
1	25 pM & 0 pM	0	RT
2	25 pM & 0 pM	10	37
3	25 pM & 0 pM	40	69.5
4	25 pM & 0 pM	70	69.5
5	25 pM & 0 pM	100	69.5

# Fluorescence Measurements and Data Analysis – Bulk solution

#### a. Fluorescent Intensity Measurements of the four color detection reaction

Fluorescence measurements (Fluorescence Intensity = FI) were performed on a Clariostar Plate Reader (BMG LABTECH GmbH, Ortenberg, Germany). The four colors (532 nm, 594 nm, 655 nm, 700 nm) are defined below (Supplementary Table 3).

**Supplementary Table 3**: Definition of the four colors measured by the Clariostar fluorescent plate reader.

Color	Excitation (nm)	Dichroic (nm)	Emission (nm)	Gain
700 nm	705-10	722.5	785-100	2450
655 nm	643-20	665.2	690-25	2030
594 nm	580-15	603.0	647-57	1707
532 nm	530-12	548.5	570-18	1663

Prior to reading, samples of  $10.5 \,\mu$ L were plated onto a 384-well PS Microplate (Greiner Bioone, Frickenhausen, Germany; 784076). Water was used on the 384 well plate as a blank reference for background correction.

#### b. Calculation of the Fluorescent Intensity Signal

$$S = FI_{i(25 \text{ pM})} - FI_{i(0 \text{ pM})}$$

Where:-

S	= Signal
FI <sub>i(25 pM)</sub>	= Fluorescent Intensity for the 25 pM sample at time i
FI <sub>i(0 pM)</sub>	= Fluorescent Intensity for the 0 pM sample at time ${\rm i}$

#### c. Ratio of the Fluorescent Intensity Signal over background

$$R = \frac{FI_{i(25 \text{ pM})} - FI_{i(0 \text{ pM})}}{FI_{i(0 \text{ pM})} - \frac{(FI_{0(25 \text{ pM})} + FI_{0(0 \text{ pM})})}{2}}$$

Where:-

R	= Ratio
$FI_{i(25 \text{ pM})}$	= Fluorescent Intensity for the 25 pM sample at time i
$FI_{i(0 \ pM)}$	= Fluorescent Intensity for the 0 pM sample at time i
FI <sub>0(25 pM)</sub>	= Fluorescent Intensity for the 25 pM sample at 0 min
FI <sub>0(0 pM)</sub>	= Fluorescent Intensity for the 0 pM sample at 0 min

#### d. Digestion of the Oligo Mix with Exonuclease T and Exonuclease I

To ascertain the maximum amount of signal that can be generated from the labelled probes within a standard reaction mix, the Oligo Mix was digested using Exonuclease T and Exonuclease I (Supplementary Table 4 & 5).

**Supplementary Table 4:** Reaction mixture for the digestion of the Oligo Mix with Exonuclease T and Exonuclease I, reactions were incubated at 37 °C for 30 mins.

Component	Stock Concentration	Volume (µL)
Water	-	53.3
Buffer <sup>1</sup>	5x	14
Oligo mix <sup>2</sup>	100x	0.7
Exonuclease T	5,000 U/mL	1
Exonuclease I	20,000 U/mL	1

<sup>1</sup> see Supplementary Table 8 for further information on the buffer composition <sup>2</sup> see Supplementary Table 7 for further information on the Oligo Mix

**Supplementary Table 5:** The maximum fluorescent intensity that can be obtained from the four labelled Probe oligos (700 nm, 655 nm, 594 nm and 532 nm) after digestion of the Oligo Mix with Exonuclease T and Exonuclease I.

Digestion	700 nm	655 nm	594 nm	532 nm
Exo T/Exo I	166960	165572	111612	191448
PFu UF	70808	52271	38205	41887
% Digested	42.4	31.6	34.2	21.9

# **DNA Oligos - Design and Sequences**

Nomenclature for non-standard bases and modifications: -

F = dT with corresponding fluorophore attached

Q = dT with corresponding fluorescence quencher attached

/3C6/ = 3'-terminal hexanediol spacer

\* = phosphorothioate linker

/5Phos/ = 5'-terminal phosphate group

/3InvdT/ = 3'-terminal 3'-5'-inverted deoxythymidine base

Black Hole Quencher 1 (BHQ1) was incorporated into the oligos using BHQ-1-dT-CE Phosphoramidite. Black Hole Quencher 2 (BHQ2) was incorporated into the oligos using BHQ-2-dT-CE Phosphoramidite and ATTO dyes were incorporated onto an amino modifier C6 dT-phosphoramidite post-synthesis (Supplementary Table 6).

**Supplementary Table 6:** Description of the Probe, Ligation, Capture, Captured and Target oligos used.

Oligo	Color	Sequence	F/Q
	700	5'- CAAGTGACTCGGTCCAGGGCCCAGFFQATTTCAGCACA TA-3'	F = ATTO 700 Q = BHQ2
Brobo	655	5'- TCGTGCCTCATCGAAACTGACGAGGFFQFFGGTTTGTG GT/3C6/-3'	F = ATTO 655 Q = BHQ2
FIODE	594	5'- CTCCAACGGACCCTAACGCACTGGAFFQCTTATGTATT GT-3'	F = ATTO 594 Q = BHQ2
	532	5'- TCACTCATGGAACATCACTCCTCCGFFQTTCCTTTCCA GA/3C6/-3'	F = ATTO 532 Q = BHQ1
	700	5'- /5Phos/GGACCGAGTCAGAACTAGATGTAGG/3Invd T/-3	
LIG	655	5'- /5Phos/GTTTCGATGAGGCACGTCTTAGATGTACG/3 InvdT/-3'	
	594	5'- /5Phos/GTTAGGGTCCGTTGGAGCTTAGATGTACG/3 InvdT/-3'	
	532	5'- /5Phos/TGATGTTCCATGAGTGTCTTAGATGTACG/3 InvdT/-3'	
	700	5'-TCCGTGAGTAATGTGGTGAAATAAACTGGGCC*C- 3'	
САР	655	5'-TCGCTGAGCCTCCACAAACCAATAACCTCGT*C- 3'	
	594	5'-TCGCTGAGCCTCAATACATAAGAAATCCAGT*G- 3'	

532	5' -TCGCTGAGCCACTGGAAAGGAAAAACGGAGG*A-	
700		
	TUCGTGAGTAATGTGGTGAAATAAAUTGGGUU*UT-3'	
655	5'-TCGCTGAGCCTCCACAAACCAATAACCTCGT*CA-	
000	3'	
504	5'-TCGCTGAGCCTCAATACATAAGAAATCCAGT*GC-	
594	3'	
500	5'-TCGCTGAGCCACTGGAAAGGAAAACGGAGG*AG-	
532	3'	
	5'-	
700	TCCGTGAGTAATGTGGTGAAATAAACTGGGCC*CTGGA	
	CCGAGTCTTAACTAGATGTAGG/3InvdT/-3'	
	5 <b>' -</b>	
655	TCGCTGAGCCTCCACAAACCAATAACCTCT*CAGTTTC	
	GATGAGGCACGTCTTAGATGTACG/3InvdT/-3'	
	5'-	
594	TCGCTGAGCCTCAATACATAAGAAATCCAGT*GCGTTA	
	GGGTCCGTTGGAGCTTAGATGTACG/3InvdT/-3'	
	5'-	
532	TCGCTGAGCCACTGGAAAGGAAAAACGGAGG*AGTGAT	
-	GTTCCATGAGTGTCTTAGATGTACG/3InvdT/-3'	
	532 700 655 594 532 700 655 594 594	5325' - TCGCTGAGCCACTGGAAAGGAAAAACGGAGG*A- 3'7005' - TCCGTGAGTAATGTGGTGAAATAAACTGGGCC*CT-3'6555' - TCGCTGAGCCTCCACAAACCAATAACCTCGT*CA- 3'5945' - TCGCTGAGCCTCAATACATAAGAAATCCAGT*GC- 3'5325' - TCGCTGAGCCACTGGAAAGGAAAAACGGAGG*AG- 3'7005' - TCGCTGAGCCACTGGAAATAAACTGGGCC*CTGGA CCGAGTCTTAACTAGATGTAGG/31nvdT/-3'6555' - TCGCTGAGCCTCCACAAACCAATAACTTCGGCC*CTGGA CCGAGTCTTAACTAGATGTAGG/31nvdT/-3'6555' - TCGCTGAGCCTCCACAAACCAATAACCTCT*CAGTTCC GATGAGGCACGTCTTAGATGTACG/31nvdT/-3'5945' - TCGCTGAGCCTCCACATACATAAGAAATCCAGT*GCGTTA GGGTCCGTTGGAGCCTCAATACATAAGAAATCCAGT*GCGTTA GGGTCCGTTGGAGCCTCAATACATAAGAAATCCAGT*GCGTTA GCGTCCGTTGGAGCCTCAATACATAAGAAATCCAGT*GCGTTA GCGTCCGTTGGAGCCTCAATACATAAGAAATCCAGT*GCGTTA GCGTCCGTTGGAGCCTCAATACATAAGAAAACCGAAGG*AGTGAT GTTCCATGAGTGTCTTAGATGTACG/31nvdT/-3'

## **Choice of Dyes**

Dyes were chosen for their photostability, solubility in aqueous media, and, most importantly, compatibility with our enzymes. Multiplexing the fluorophores is achieved by matching the dye excitation wavelengths to laser lines at 532 nm (ATTO 532), 594 nm (ATTO 594), 640 nm (ATTO 655), and 701 nm (ATTO 700). We found ATTO 700 to be superior, with respect to enzymatic compatibility, to the ATTO 488 and other dyes of similar spectral properties that are traditionally used in multiplexing applications.(1, 2) Use of the shorter wavelength dyes resulted in very high false signal due to unusually rapid single-strand digestion by the exonuclease. The dyes were attached to the Probe Oligo using standard postsynthetic protocols (-NH2 – NHS-ester amide chemistry(3)). As quenchers, we found that the quenchers BHQ-1 (for ATTO 532) and BHQ-2 (for ATTO 594, 655 and 700) provided >99% quenching efficiency, thereby guaranteeing a large difference between quenched and unquenched states(4). The quenchers were attached during the solid-state DNA oligo synthesis by using the corresponding quencher-labelled phosphoramidites.

# Preparation of the 100x Oligo Mix

For ease of use, all the Probe, Ligation and Capture oligos were pre-mixed, in nuclease-free water, at a 100 times the required final concentration; before addition (of the oligo mix) to the detection reaction (Supplementary Table 7).

**Supplementary Table 7:** Description of the Probe, Ligation and Capture oligos used for the 100x Oligo Mix.

Oligo	100x Oligo Mix (nM)	Final Oligo Concentration (nM)
Probe, 700 nm	600	6
Probe, 655 nm	600	6
Probe, 594 nm	600	6
Probe, 532 nm	1000	10
LIG, 700 nm	4000	40
LIG, 655 nm	100	10
LIG, 594 nm	1000	10
LIG, 532 nm	2000	20
CAP, 700 nm	50	0.5
CAP, 655 nm	50	0.5
CAP, 594 nm	50	0.5
CAP, 532 nm	50	0.5

# **Optimisation of the Buffer used in the Detection Reaction**

The reaction buffer had to be chosen in a way that would allow all enzymes to work in the same environment. We found that the limiting factor in buffer optimization was *Pfu*Ultra II Fusion HS DNA polymerase, with respect to the specificity of its dsDNA exonuclease-activity. Both *Bst* L.F. DNA polymerase and *E. coli* DNA ligase tolerated a much wider range of buffer conditions. While these two enzymes did not work in the proprietary buffer supplied with *Pfu*Ultra II Fusion HS DNA polymerase, it was found that using a modified version of this buffer made it possible to have all enzymes work in the same buffer (Supplementary Table 8). Optimisation of the Tris-HCI, KCI and MgCl<sub>2</sub> components of the buffer are described below (Supplementary Figures 1 and 2).

**Supplementary Table 8:** Final Concentration of the components that make up the 1x buffer used in a typical, four-color detection reaction.

Component	Concentration in 1x buffer	Concentration in 5x buffer stock
Tris-HCI, pH 7.6	12 mM	60 mM
MgCl <sub>2</sub>	5 mM	25 mM
KCI	20 mM	100 mM
Triton X-100	0.1%	0.5%
DTT	0.5 mM	2.5 mM
NAD	400 nM	2 µM

#### a. Titration of Tris-HCl, pH 7.6 vs. KCl at 5 mM MgCl<sub>2</sub>

To optimise the Tris-HCl, pH 7.6 and KCl concentrations a cross-titration experiment was preformed, leaving the MgCl<sub>2</sub> concentration at 5 mM. The cross-titration showed that concentrations of 12 mM Tris-HCl and 20 mM KCl provided a good operating window for the reaction, which could accommodate slight increases and decreases in salt concentration without adversely affecting the performance of the detection reaction (Supplementary Figure 1). Accommodation of additional salts is desirable as it allows for some tolerance to water loss from microdroplets.



**Supplementary Figure 1:** Cross-Titration of Tris-HCl, pH 7.6 vs. KCl, using 5 mM MgCl<sub>2</sub>. Tris-HCl, pH 7.6 was titrated from 3 to 48 mM, while KCl was titrated from 5 to 40 mM using the four color standard reaction with 0 and 25 pM dNTPs. The fluorescent signal (25 pM dNTPs– 0 pM dNTPs) for the four colors is shown in the tables and plotted; 700 nm (top left), 655nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

#### b. Titration of Tris-HCI, pH 7.6 vs. MgCl<sub>2</sub> at 20 mM KCI

Following on from the KCl optimisation (Supplementary Figure 1), a cross-titration of Tris-HCl, pH 7.6 and MgCl<sub>2</sub> was performed with a fixed KCl concentration of 20 mM. 12 mM Tris-HCl, pH 7.6 and 5 mM MgCl<sub>2</sub> provided a good operating window for the reaction. Therefore,

12 mM Tris-HCl, pH 7.6, 5 mM MgCl<sub>2</sub> and 20 mM KCl were used as the final concentrations for the 1x buffer.



Olemat (FOA Observat)							Olemat (FOO Observal)								
Signal (594	Channel)	48	36	24	12	6	3	Signal (532	Channel)	48	36	24	12	6	3
MgCl <sup>2</sup> (mM)	20	197	331	5251	7784	234	137	(Mi	20	160	302	-67155	13192	525	156
	15	175	2745	26640	56488	6053	217		15	-119	1122	24037	108424	13203	418
	10	1882	5847	61663	73105	44271	612		10	660	3047	84597	117514	95877	1211
	8	3596	7041	71330	79922	60190	633	E)	8	1156	4017	109785	166429	130208	1684
	5	1687	4857	72998	85563	63440	4576	MgCI 2	5	359	1920	107652	175157	138494	17174
	3	-51	788	22822	81431	76417	12832		3	-145	-34	7248	160655	155851	47503
	1.25	-54	24	1409	41167	62137	5217		1.25	-43	-141	433	47890	115409	29229
	0.6125	-522	213	441	3400	45844	469		0.6125	-537	226	217	1656	71684	3587

**Supplementary Figure 2:** Cross-Titration of Tris-HCl, pH 7.6 vs. MgCl<sub>2</sub>, using 20 mM KCl. Tris-HCl, pH 7.6 was titrated from 3 to 48 mM, while MgCl<sub>2</sub> was titrated from 0.6125 to 20 mM using the four color standard reaction with 0 and 25 pM dNTPs. The fluorescent signal (25 pM dNTPs– 0 pM dNTPs) for the four colors is shown in the tables and plotted; 700 nm (top left), 655nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

# **Optimisation of Spermine, Spermidine and TEPA Additives**

Reaction additives spermine, spermidine and TEPA were found to enhance the detection reaction and in an identical way as to how the buffer was optimised, cross-titrations were used to optimise the concentrations of spermine, spermidine and TEPA in the detection reaction



#### a. Titration of Spermine vs. TEPA at 1.0 mM Spermidine

**Supplementary Figure 3:** Cross-Titration of Spermine vs. TEPA, using 1 mM Spermidine. Spermine was titrated from 0 to 1.25 mM, while TEPA was titrated from 0 – 0.015 % using the four color standard reaction with 0 and 25 pM dNTPs. The fluorescent signal (25 pM dNTPs– 0 pM dNTPs) for the four colors is shown in the tables and plotted; 700 nm (top left), 655nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

#### b. Titration of Spermine vs. TEPA at 1.5 mM Spermidine



**Supplementary Figure 4:** Cross-Titration of Spermine vs. TEPA, using 1.5 mM Spermidine. Spermine was titrated from 0 to 1.25 mM, while TEPA was titrated from 0 – 0.015 % using the four color standard reaction with 0 and 25 pM dNTPs. The fluorescent signal (25 pM dNTPs– 0 pM dNTPs) for the four colors is shown in the tables and plotted; 700 nm (top left), 655nm (top right), 594 nm (bottom left) and 532 nm (bottom right).





**Supplementary Figure 5:** Cross-Titration of Spermine vs. TEPA, using 2 mM Spermidine. Spermine was titrated from 0 to 1.25 mM, while TEPA was titrated from 0 – 0.015 % using the four color standard reaction with 0 and 25 pM dNTPs. The fluorescent signal (25 pM dNTPs– 0 pM dNTPs) for the four colors is shown within the tables and plotted; 700 nm (top left), 655 nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

Cross-titrations of Spermine vs. TEPA were performed at fixed concentrations of spermidine (Supplementary Figures 3, 4 & 5). From this data 1 mM spermine, 1 mM spermidine and 0.015% TEPA were used as the final additive concentration in the four color detection reaction.

## **Screening of Capture Polymerase and Ligation Enzyme**

The choice of capture polymerase was based on initial screenings using a variety of commercially available DNA polymerases. In these reactions, the other enzymes were the same as described earlier, i.e., *E. coli* DNA ligase and *Pfu*Ultra II Fusion HS DNA polymerase. The tables and graphs below show the fluorescence intensity after 100 minutes of incubation. Note that conditions were not yet fully optimized for capture and ligation, giving results that differ from those in reactions performed under finalized conditions.

The choice of ligase was based on initial screenings using a variety of commercially available DNA ligases. In these reactions, the other enzymes were the same as described earlier, i.e., *Bst* L.F. DNA polymerase and *Pfu*Ultra II Fusion HS DNA polymerase. The tables and graphs below show the fluorescence intensity after 100 minutes of incubation. Note that conditions were not yet fully optimized for capture and ligation, giving results that differ from those in reactions performed under finalized conditions.

As before, "captured" refers to a construct of CAP+N (simulating that capture has taken place), and "target" refers to a construct of CAP+N+LIG (simulating that both capture and ligation have taken place).





**Supplementary Figure 6:** Screening of capture polymerase – Part 1. Different polymerases (from different suppliers) were used to test their ability to capture dNTPs at 25 pM, using the four color standard reaction, compared to 25 pM Captured and 25 pM Target oligos. The four colors are shown in the tables and plotted; 700 nm (top left), 655 nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

#### b. Screening of Capture Polymerase – Part 2



**Supplementary Figure 7:** Screening of capture polymerase – Part 2. Different polymerases (from different suppliers) were used to test their ability to capture dNTPs at 25 pM, using the four color standard reaction, compared to 25 pM Captured and 25 pM Target oligos. The four colors are shown in the tables and plotted; 700 nm (top left), 655 nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

#### c. Screening of Ligation Enzyme



**Supplementary Figure 8:** Screening of Ligation Enzyme. Different Ligases (from different suppliers) were used to test their ability to capture dNTPs at 25 pM, using the four color standard reaction, compared to 25 pM Captured and 25 pM Target oligos. For the four colors the data from 0 pM dNTPs, 25 pM dNTPs, 25 pM Captured and 25 pM Targets are shown in the tables and plotted; 700 nm (top left), 655 nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

# Digestion of oligo mix using the exonuclease activity of different

#### polymerases

Finding a suitable exonuclease proved the most difficult among the enzymes needed in our reaction cascade. "Classical" exonucleases (Exonuclease T, Exonuclease I, Lambda Exonuclease) all lacked sufficient dsDNA selectivity for our intended purpose, resulting in unwanted digestion of Probe Oligo even in absence of dNTPs. Most polymerases with 3' -> 5' exonuclease activity were either insufficiently selective as well, or their exonuclease activity was too slow to result in sufficient signal being generated. The three polymerases that exhibited suitable levels of selectivity and reactivity were Phusion II DNA polymerase, Phire Hot Start II DNA Polymerase and *Pfu*Ultra II Fusion HS DNA polymerase. All three polymerases required different buffer systems and reaction temperatures for optimal reactivity (Supplementary Table 9):

#### a. Polymerase - Required Buffers and Temperatures

1x Buffer Component							
Polymerase	<i>Pfu</i> Ultra II Fusion HS	Phusion II	Phire Hot Start II				
рН	7.6	8.0	8.0				
Tris-HCI	12 mM	40mM	5 mM				
MgCl₂	5 mM	15mM	15 mM				
KCI	20 mM	100mM	60 mM				
Triton X-100	0.10 %	0.10%	0.10%				
DTT	0.5 mM	0.5 mM	0.5 mM				
NAD	400 nM	400 nM	400 nM				
Temperature	69.5 ⁰C	68.5 °C	69.0 °C				

Supplementary Table 9: Description of the three Polymerase buffer conditions used.

#### b. Screening of different Polymerases



**Supplementary Figure 9:** Screening of different Polymerases. Different Polymerases were tested for their ability to digest the probe DNA, using the four color standard reaction, at 0 pM and 25 pM dNTPs concentration (data from the 655 nm channel is shown).

#### c. Digestion temperature for *Pfu*Ultra II Fusion HS DNA polymerase



**Supplementary Figure 10:** Screening of digestion temperature for *Pfu*Ultra II Fusion HS DNA polymerase. *Pfu*Ultra II Fusion HS DNA polymerase was screened from 60 to 74 °C for its ability to digest the probe DNA, using the four color standard reaction, at 0 and 25 pM dNTPs concentration. For the four colors the data from 0 pM and 25 pM dNTPs as well as the signal (0 pM – 25 pM) is shown within the tables and plotted; 700 nm (top left), 655nm (top right), 594 nm (bottom left) and 532 nm (bottom right).

# **Choice of Protecting Group on LIG**

In order to protect the oligos from unwanted exonuclease digestion, a protecting group was needed at the 3'-end of LIG. Several commercially available end modifications were tested for their suitability by subjecting the corresponding 3'-modified LIG oligos to an excess of *Pfu*Ultra II Fusion HS DNA polymerase during a 2-hour incubation at 70°C. The figure below shows a picture of the resulting 8% denaturing PAGE analysis of the reactions.



Lane number	End modification		
1	(none), unreacted reference sample		
2	(none)		
3	RNA base (Uracil)		
4	Phosphorothioate		
5	Hexanediol		
6	5'-3' inverted dT		
7	Spacer 9		
8	C3 spacer		
9	dspacer (abasic furan)		

**Supplementary Figure 11**: Image (detail) of 8% denaturing (formamide) polyacrylamide gel showing the digestion of LIG by PfuUltra II Fusion HS bearing a variety of 3' end modifications.

It is apparent that unprotected oligos are digested almost completely under these conditions, and that the RNA base offers no protection. The most efficient protecting group is the 5'-3' inverted dT, which was consequently used for our experiments.

## **Specificity of Detection**



**Supplementary Figure 12**: Testing for chemical cross-talk by adding either intended or "incorrect" dNTP to the reaction. The individual capture channels were tested separately (A – Probe, LIG and CAP for dGTP detection only, B – Probe, LIG and CAP for dCTP detection only, C – Probe, LIG and CAP for dATP detection only, D – Probe, LIG and CAP for dTTP detection only). Note that in each case the intended dNTP is present at 1000-fold lower concentration than the three others.

# **Fluorescence Measurements of Droplets**

Fluorescent images of microdroplets were acquired using a custom fluorescent microscope consisting of an Andor Zyla 5.5 sCMOS camera, four CW laser excitation lines (Vortran Stradus 532 nm, Cobalt Mambo 594 nm, Vortran Stradus 640 nm, Vortran Stradus 701 nm), a filter cube changer to select the appropriate excitation, dichroic and emission filters (see Supplementary Table 10), and a galvanometer scanning mirror system to raster the laser over the area of interest. The exposure time per laser for each 300  $\mu$ m x 300  $\mu$ m field of view was 30 s.

Laser Wavelength	Excitation filter	Dichroic	Emission filter		
701 nm	Semrock	Chroma	Semrock		
	FF01-692/40	725dcxxr	FF01-795/150		
640 nm	Semrock	Semrock	Semrock		
	FF01-640/14	Di02-R635	FF01-679/41		
594 nm	Semrock	Semrock	Semrock		
	FF01-591/6	Di02-R594	FF01-647/57		
532 nm	Semrock FF01-532/3	Semrock Di02-R532	Semrock FF01-582/75 + FF01-562/40		

Supplementary Table 10: Filter sets used for fluorescence measurements of droplets





**Supplementary Figure 13**: Example images of a field of view of droplets in: A-D) the four color channels used for dNTP detection, and E) brightfield used for droplet position and size detection. Scale bars are 50 µm.

#### B. Fluorescence Intensity vs. Droplet Diameter – All Color Channels



**Supplementary Figure 14**: Droplet intensity vs. droplet diameter extracted from each of the four fluorescence channels: A) 532 nm, B) 594 nm, C) 655 nm, D) 700 nm.





**Supplementary Figure 15**: Histograms showing frequency of occurrence vs droplet intensity for all four color channels (A 532 nm, B 594 nm, C 655 nm, D 700 nm) in the 3-7 µm diameter range. In each case, two peaks are visible, one for droplets which contain no nucleotide (blue) and one for droplets which contain one (or more) nucleotides (green).





**Supplementary Figure 16**: Droplet intensity in a given color channel plotted against droplet intensity in another channel. All six of the possible pairs are shown: A) 532 nm vs. 594 nm, B) 594 nm vs. 655 nm, C) 532 nm vs. 655 nm, D) 594 nm vs. 700 nm, E) 532 nm vs. 700 nm, F) 655 nm vs. 700 nm. These plots allow for a visual examination of the correlation between color channels. The different diameter subsets are colored from dark green at small droplets to light green at larger droplets.

For higher dNTP concentration or larger droplets a fraction of the droplets will begin to appear on the diagonal as the probability of a droplet containing more than one type of dNTP becomes larger. For instance, 12.5 pM of each dNTP in 5  $\mu$ m droplets gives  $P(\geq 1)_{532}P(\geq 1)_{594} \approx 15\%$ .

#### E. Fluorescence Intensity – Channel Correlations (3 - 4 micrometer diameters)



**Supplementary Figure 17**: Droplet intensity in a given color channel plotted against droplet intensity in another channel. All six of the possible pairs are shown: A) 532 nm vs. 594 nm, B) 594 nm vs. 655 nm, C) 532 nm vs. 655 nm, D) 594 nm vs. 700 nm, E) 532 nm vs. 700 nm, F) 655 nm vs. 700 nm. These plots allow for a visual examination of the correlation between color channels. Only the 3-4  $\mu$ m subset of the droplets is shown in each case. The Pearson correlation coefficient (and r<sup>2</sup> value) for sub-figures A - F are 0.072 (0.005), 0.149 (0.022), 0.135 (0.018), 0.124 (0.015), 0.090 (0.008), and 0.157 (0.025) respectively.

Supplementary Figure 17 shows the 3 - 4 µm diameter droplet range. As an example, consider the following. From Table 1 in the manuscript, the expected fraction of droplets with

signal for 12.5 pM at 3  $\mu$ m and 4  $\mu$ m diameter is 10% and 22% respectively. For two colours to be positive, this implies between 1% (if all droplets are 3  $\mu$ m) and 4.8% (if all droplets are 4  $\mu$ m). We see between 3.4% and 5.0% for the combinations of colours (532 nm, 594 nm and 700 nm) which have 12.5 pM dNTP concentration, which is close to the expected range. The 655 nm channel has 37.5 pM dNTP concentration so a higher fraction would be expected. At 3  $\mu$ m diameter we expect 28% positive, at 4  $\mu$ m 53% positive, which then gives a 2.8% to 11.7% range for double positive between the 12.5pM and 37.5pM channels. We observe 9.5% to 10.6%, again within expected bounds.

### **Droplet Finding Algorithm**

First the gradient of the brightfield droplet image is taken to get the directional change in the intensity of the image. The circle finding algorithm is based on performing a matched filter(5). This involves correlating a known template image with the gradient image of interest. In this case the template is a mathematically-defined ring,  $exp^{\frac{-(r-r_0)^2}{w^2}}$ , with direction pointing radially outwards, where *r* is distance from the droplet's centre,  $r_0$  is the template's size parameter, and *w* is the template's width parameter. This is performed multiple times, with a range of template image sizes,  $r_0$ , to allow a range of droplet sizes to be found. Suitable thresholds are chosen such that droplets are identified and spurious matches are minimised.

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