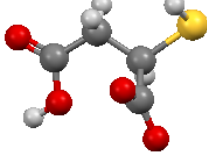
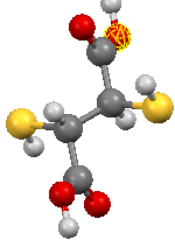
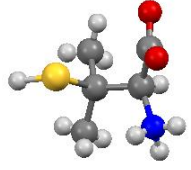
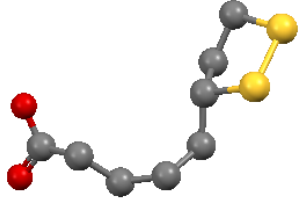
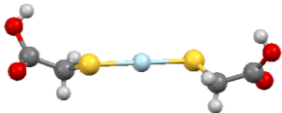
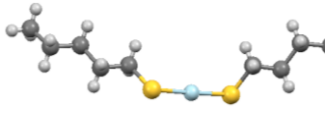
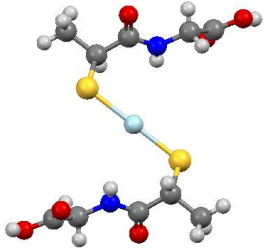
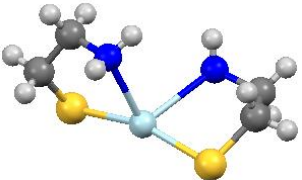
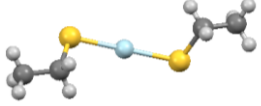
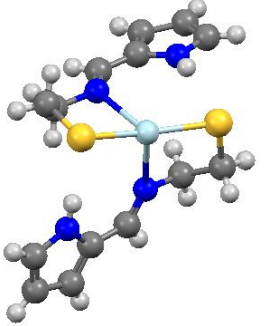
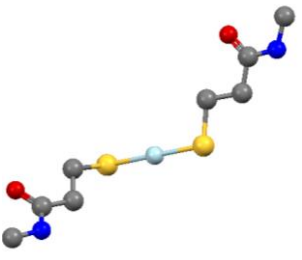
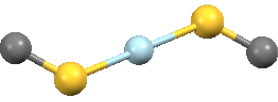
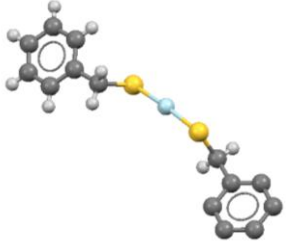
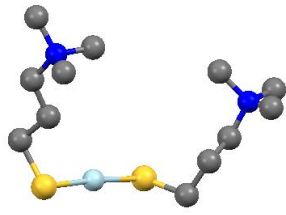
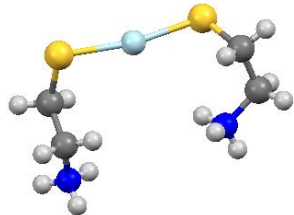
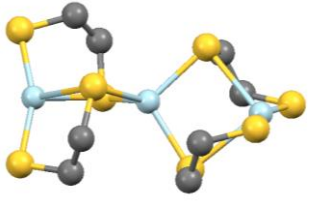
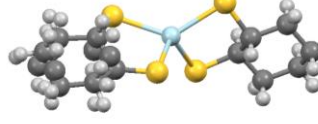


**Table 1S.** Structures of the ligands. Hydrogen in light grey, carbon in grey, nitrogen in blue, sulfur in yellow and oxygen in red. Coordinates obtained from the Cambridge Structural Database, image created with Mercury3.5.

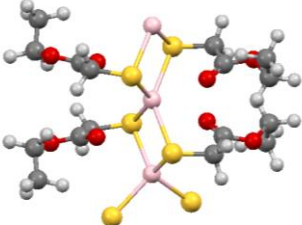
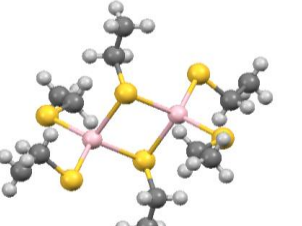
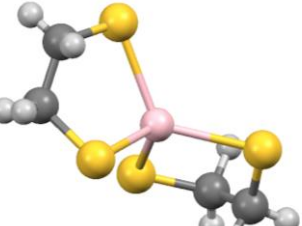
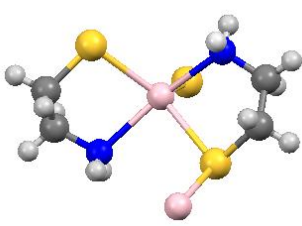
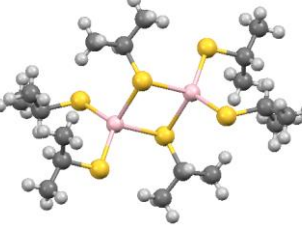
CCDC-Refcode Reference	Structures	Ligand	
<p style="text-align: center;"><b>BECVUC</b></p> <p>K. Igawa, N. Yoshinari, T. Konno, Acta Crystallogr., Sect. C: Cryst. Struct. Commun. 68 (2012) 6332</p>		TMA	
<p style="text-align: center;"><b>JUBFER</b></p> <p>G.J.Pyrka, N.Scott, Q.Fernando, Acta Crystallogr., Sect. C: Cryst. Struct. Commun. 48 (1992) 2007</p>		DMSA	
<p style="text-align: center;"><b>CEDFAS</b></p> <p>H.E. Howard-Lock, C.J.L. Lock, P.S. Smalley, J. Crystallogr. Spectrosc. Res., 13 (1983) 333</p>		DPEN	
<p style="text-align: center;"><b>HAXJIB</b></p> <p>C.-P.Racz, G.Borodi, M.M.Pop, I.Kacso, S.Santa, M.Tomoaia-Cotisel, Acta Crystallogr., Sect. B: Struct. Sci., 68 (2012) 164</p>		LA	

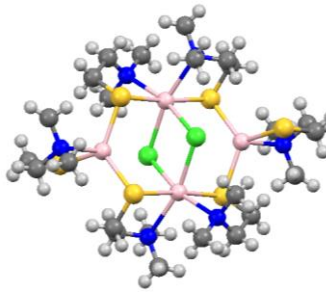
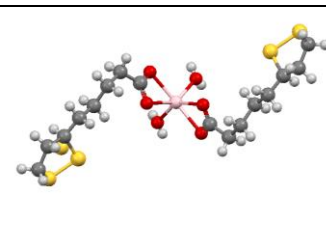
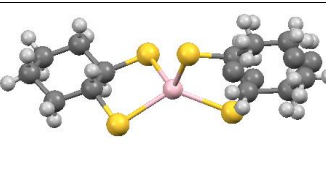
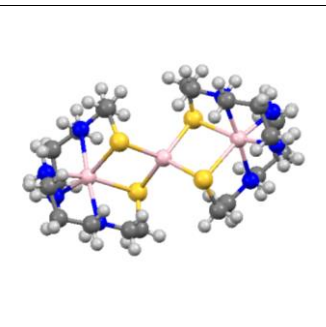
**Table 2S.** Structures of  $\text{Hg}^{2+}$  with different ligands bearing mercapto groups. Hydrogen in light grey, carbon in grey, nitrogen in blue, sulfur in yellow, oxygen in red and mercury in light blue. Coordinates obtained from the Cambridge Structural Database, image created with Mercury3.5.

CCDC-Refcode Reference	$\text{Hg}^{2+}$ Structures	Complex	Ligand
AQON0J J.M. Bramlett, Hee-Jung Im, Xiang-Hua Yu, Tianniu Chen, Hu Cai, L.E. Roecker, C.E. Barnes, Sheng Dai, Zi-Ling Xue, <i>Inorg. Chim. Acta</i> , 317 (2004) 243		$\text{HgL}_2$	TGA  $\text{COOHCH}_2\text{-SH}$
BEPQAO G.G. Hoffmann, I. Steinfatt, W. Brockner, V. Kaiser, <i>Z. Naturforsch. B:Chem.Sci.</i> , 54 (1999) 887		$\text{HgL}_2$	$\text{CH}_3(\text{CH}_2)_3\text{CH}_2\text{SH}$
CAYFUD Z. Popovic, D. Matkovic-Calogovic, J. Hasic, M. Sikirica, D. Vikić-Topić, <i>Croat. Chem. Acta</i> , 72 (1999) 279		$\text{HgL}_2$	$\text{HCOO-CH}_2\text{-NH-CO-CH-}$ $(\text{CH}_3)\text{-SH}$
HAYQEE H.Fleischer, Y.Dienes, B.Mathiasch, V.Schmitt, D.Schollmeyer, <i>Inorg. Chem.</i> , 44, (2005) 8087		$\text{HgL}_2$	$\text{NH}_2\text{-CH}_2\text{-CH}_2\text{-SH}$
MERSET01 K.A. Fraser, W. Clegg, D.C. Craig, M.L. Scudder, I.G. Dance, <i>Acta Crystallogr. Sect.C:Cryst. Struct. Commun.</i> , 51 (1995) 406		$\text{HgL}_2$	$\text{CH}_3\text{-CH}_2\text{-SH}$
AHOQOE Kuan-Yi Wu, Chang-Chih Hsieh, Yih-Chern Horng, <i>J.Organomet.Chem.</i> , 694 (2009) 2085		$\text{HgL}_2$	$\text{R-CH}_2\text{-NH-CH}_3\text{-CH}_2\text{-SH}$

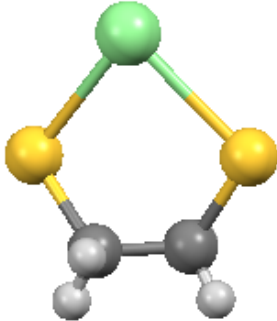
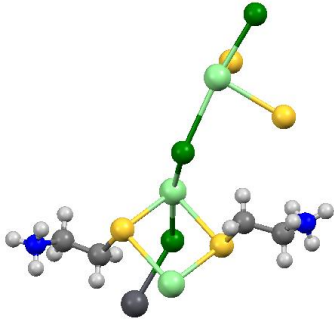
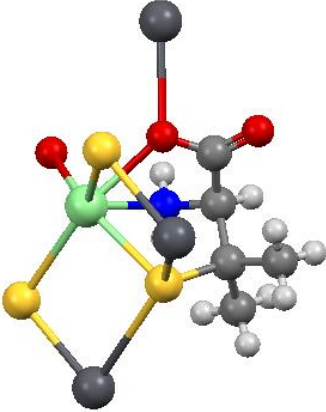
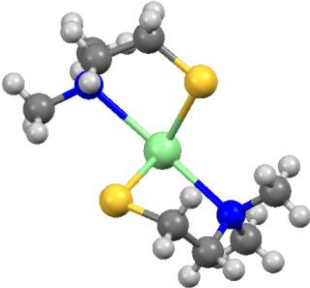
<p><b>MECBHG</b></p> <p>C. Perchard, G. Zuppiroli, P. Gouzerh, Y. Jeannin, F. Robert, J. Mol. Struct., 72 (1981) 119</p>		HgL <sub>2</sub>	CH <sub>3</sub> -NH-CO-CH <sub>2</sub> -CH <sub>2</sub> -SH
<p><b>MERMES</b></p> <p>D.C. Bradley, N.R. Kunchur, J. Chem. Phys., 40 (1964) 2258</p>		HgL <sub>2</sub>	CH <sub>3</sub> -SH
<p><b>TEVQOA01</b></p> <p>R.D.Schluter, G.Krauter, W.S.Rees Junior, J.Cluster Sci., 8 (1997) 123</p>		HgL <sub>2</sub>	Φ-CH <sub>2</sub> -SH
<p><b>VOLFOR</b></p> <p>I. Casals, P. Gonzalez-Duarte, W. Clegg, Inorg. Chim. Acta, 184 (1991) 167</p>		HgL <sub>2</sub>	N(CH <sub>3</sub> ) <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>2</sub> S
<p><b>XIJKOQ</b></p> <p>Chong-Hyeak Kim, S. Parkin, M. Bharara, D. Atwood, Polyhedron, 21 (2002) 225</p>		HgL <sub>2</sub>	NH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> S
<p><b>DAXPUN</b></p> <p>G. Henkel, P. Betz, B. Krebs, Chem. Comm.,(1985) 1498</p>		Hg <sub>3</sub> L <sub>4</sub>	HS-CH <sub>2</sub> -CH <sub>2</sub> -SH
<p><b>WACXAZ</b></p> <p>N. Govindaswamy, J. Moy, M. Millar, S.A. Koch, Inorg. Chem., 91 (1992) 5343</p>		HgL <sub>2</sub>	C <sub>6</sub> H <sub>12</sub> S <sub>2</sub>

**Table 3S.** Structures of Cd<sup>2+</sup> with different ligands bearing mercapto groups. Hydrogen in light grey, carbon in grey, nitrogen in blue, sulfur in yellow, oxygen in red, cadmium in pink and chloride in green. Coordinates obtained from the Cambridge Structural Database, image created with Mercury3.5.

CCDC-Refcode Reference	Cd <sup>2+</sup> Structures	Complex	Ligand
CAHGUN  I.G.Dance, M.L.Scudder, R.Secomb, <i>Inorg. Chem.</i> , 22 (1983) 1794		Cd <sub>3</sub> L <sub>6</sub>	CH <sub>3</sub> -CH <sub>2</sub> -O-CO-CH <sub>2</sub> SH
DETHOZ  A.D. Watson, C.P. Rao, J.R. Dorfman, R.H. Holm, <i>Inorg. Chem.</i> , 24 (1985) 2820		Cd <sub>2</sub> L <sub>6</sub>	CH <sub>3</sub> -CH <sub>2</sub> -SH
DUFZAF  C.P. Rao, J.R. Dorfman, R.H. Holm. <i>Inorg. Chem.</i> , 25 (1986) 428		CdL <sub>2</sub>	SH-CH <sub>2</sub> .CH <sub>2</sub> -SH
HAZGAR  H. Fleischer, Y. Dienes, B. Mathiasch, V. Schmitt, D. Schollmeyer, <i>Inorg. Chem.</i> , 44 (2005) 8087		CdL <sub>2</sub>	SH-CH <sub>2</sub> .CH <sub>2</sub> -NH <sub>2</sub>
HEQMAT  U. Florke, CSD Communication (Private Communication), 2013		Cd <sub>2</sub> L <sub>6</sub>	(CH <sub>3</sub> ) <sub>2</sub> -CH-SH

<p style="text-align: center;"><b>JIZWIY</b></p> <p>I. Casals, P. Gonzalez-Duarte, W. Clegg, C. Foces-Foces, F.H. Cano, M. Martinez-Ripoll, M. Gomez, X. Solans, J. Chem. Soc., Dalton Trans., (1991) 2511</p>		$\text{Cd}_4\text{L}_6\text{Cl}_2$	$(\text{CH}_3)_2\text{-N-CH-CH}_2\text{-SH}$
<p style="text-align: center;"><b>TUQREC</b></p> <p>H. Strasdeit, A. von Dollen, A.-K. Duhme, Z. Naturforsch., B:Chem.Sci., 52 (1997) 17</p>		$\text{CdL}_2 \cdot 2\text{H}_2\text{O}$	Lipoic acid
<p style="text-align: center;"><b>WACXED</b></p> <p>N. Govindaswamy, J. Moy, M. Millar, S.A. Koch, Inorg. Chem., 31 (1992) 5343</p>		$\text{CdL}_2$	Cicloesene-(SH) <sub>2</sub>
<p style="text-align: center;"><b>XICZEO</b></p> <p>M. Mikuriya, Xiao Jian, S. Ikemi, T. Kawahashi, H. Tsutsumi, A. Nakasone, Jong-Wan Lim, Inorg. Chim. Acta, 312 (2001) 183</p>		$\text{Cd}_3\text{L}_4$	$\text{NH}_2\text{-CH}_2\text{CH}_2\text{CH}_2\text{NHCH-CH}_2\text{-SH}$

**Table 4S.** Structures of  $\text{Pb}^{2+}$  with different ligands bearing mercapto groups. Hydrogen in light grey, carbon in grey, nitrogen in blue, sulfur in yellow, oxygen in red, lead in light green and chloride in green. Coordinates obtained from the Cambridge Structural Database, image created with Mercury3.5.

CCDC-Refcode Reference	$\text{Pb}^{2+}$ Structures	Complex	Ligand
DISWIL P.A.W.Dean, J.J.Vittal, N.C.Payne, Inorg. Chem., 24 (1985) 3594		PbL	HS-CH <sub>2</sub> -CH <sub>2</sub> -SH
KEJCOS M.S. Bharara, S. Parkin, D.A. Atwood, Inorg. Chim. Acta, 359 (2006) 3375		Pb <sub>2</sub> L <sub>2</sub>	CH <sub>2</sub> SH-CH <sub>2</sub> NH <sub>2</sub>
DPENPB01 A.C.Schell, M.Parvez, F.Jalilehvand, Acta Crystallogr., Sect. E: Struct. Rep. Online, 68 (2012) m489		PbL	DPEN
NUFQEL01 V.N. Khrustalev, R.R. Aysin, I.V. Borisova, A.S. Peregudov, L.A. Leites, N.N. Zemlyansky, Dalton Trans., 39 (2010) 9480		PbL <sub>2</sub>	HS-CH <sub>2</sub> -CH <sub>2</sub> N(CH <sub>3</sub> ) <sub>2</sub>

**Fig. 1S.** Speciation plots of the metal complexes whose stability constants are reported in Table 3, calculated with Hyss program at metal concentration 0.001 M and ligand concentration 0.002 M.

