Supporting Information

Structural Properties of Gas Phase Molybdenum Oxide Clusters $[Mo_4O_{13}]^{2-}$, $[HMo_4O_{13}]^-$, and $[CH_3Mo_4O_{13}]^-$ Studied by Collision-Induced Dissociation

Manuel Plattner, Aristeidis Baloglou, Milan Ončák, Christian van der Linde, and Martin K. Beyer[*]

Institut für Ionenphysik und Angewandte Physik, Universität Innsbruck, Technikerstraße 25, 6020 Innsbruck, Austria

[*] E-mail: martin.beyer@uibk.ac.at

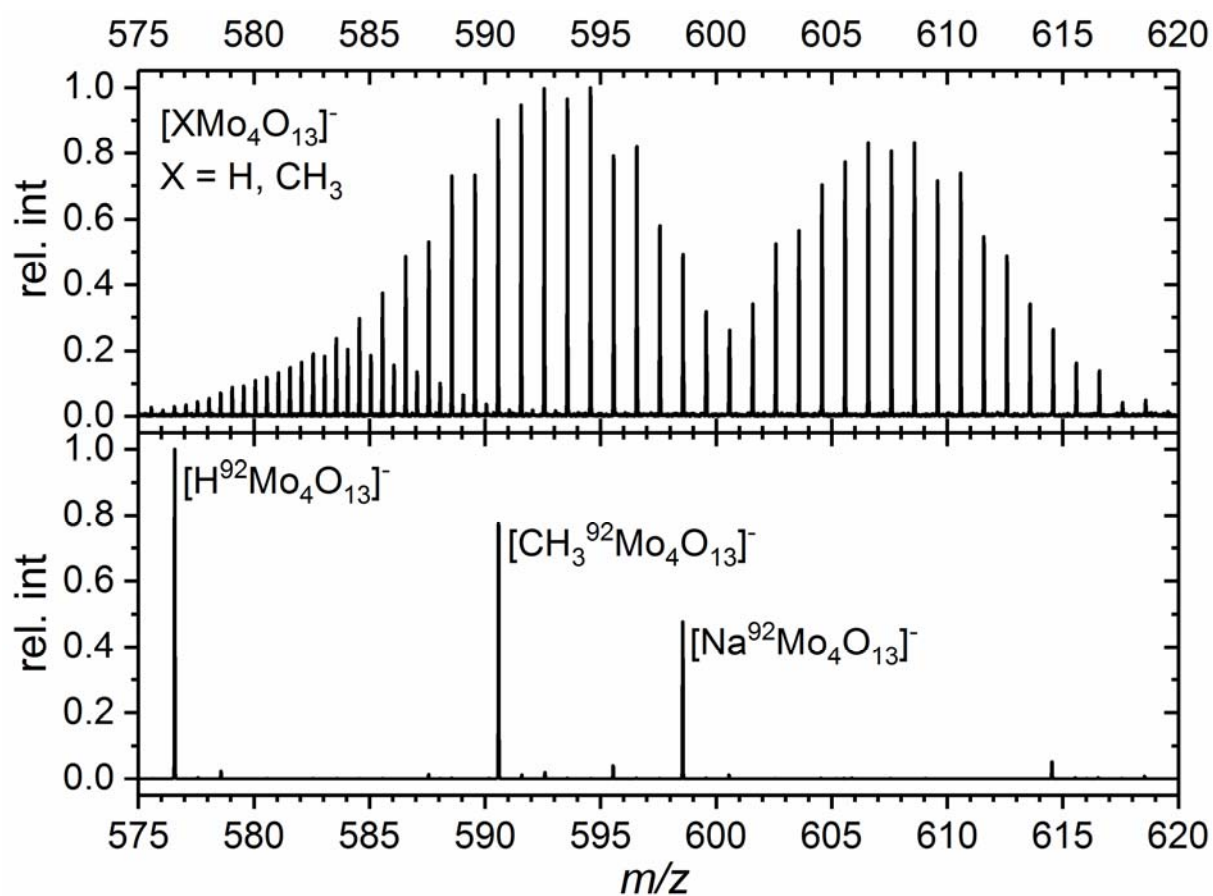## Mass spectra of natural and isotopically enriched molybdate clusters

Figure S1: Isotopologues of natural tetramolybdate clusters (top) and the corresponding mass spectrum of the tetramolybdate clusters containing only isotopically enriched ^{92}Mo (bottom). Unambiguous identification of the mass peaks is only possible for the isotopically enriched case. The doubly charged isotope distribution in the lower mass range is $[Mo_8O_{25}]^{2-}$. The mass peak $[Na^{92}Mo_4O_{13}]^-$ is irrelevant for the present study and probably originates from Na^+ impurities inside the ESI inlet system.
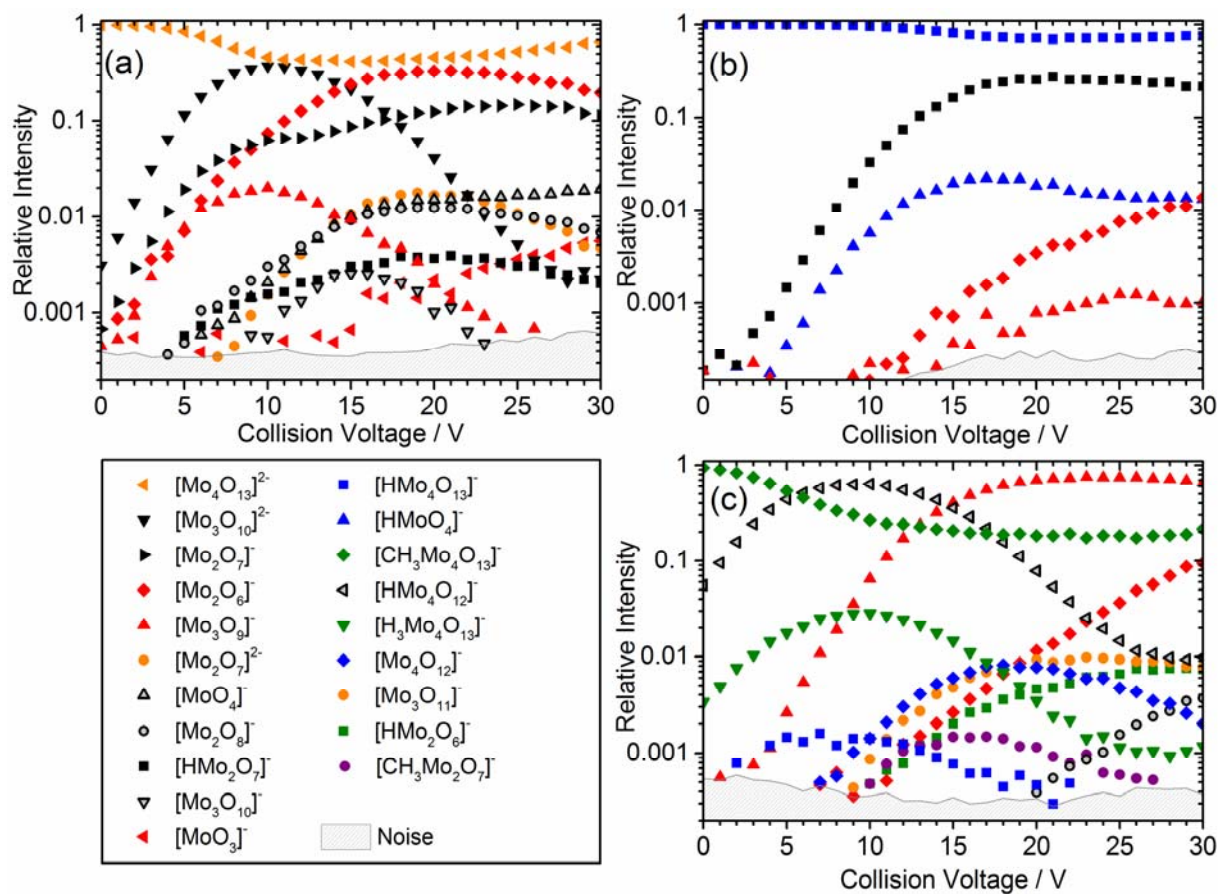
Figure S2: CID breakdown curves of the $[Mo_4O_{13}]^{2-}$ (a), $[HMo_4O_{13}]^-$ (b) and $[CH_3Mo_4O_{13}]^-$ (c) clusters.

Table S1: Overview of all measured fragment ions, sorted by parent ion.

$[Mo_4O_{13}]^{2-}$ Fragment	Measured m/z	Exact m/z	$[HMo_4O_{13}]^{-}$ Fragment	Measured m/z	Exact m/z	$[CH_3Mo_4O_{13}]^{-}$ Fragment	Measured m/z	Exact m/z
$Mo_4O_{13}^{2-}$	287.780	287.781	$HMo_4O_{13}^{-}$	576.568	576.570	$CH_3Mo_4O_{13}^{-}$	590.583	590.585
$Mo_3O_{10}^{2-}$	217.834	217.835	$HMo_2O_7^{-}$	296.786	296.786	$HMo_4O_{12}^{-}$	560.574	560.575
$Mo_2O_7^{-}$	295.778	295.779	$HMoO_4^{-}$	156.892	156.894	$H_3Mo_4O_{13}^{-}$	578.584	578.585
$Mo_2O_6^{-}$	279.783	279.784	$Mo_2O_6^{-}$	279.782	279.784	$Mo_4O_{12}^{-}$	559.566	559.567
$Mo_3O_9^{-}$	419.675	419.675	$Mo_3O_9^{-}$	419.675	419.675	$Mo_3O_{11}^{-}$	451.665	451.665
$Mo_2O_7^{2-}$	147.887	147.890				$Mo_2O_6^{-}$	279.783	279.784
MoO_4^{-}	155.885	155.887				$HMo_2O_6^{-}$	280.791	280.791
$Mo_2O_8^{-}$	311.773	311.773				$CH_3Mo_2O_7^{-}$	310.802	310.802
$HMo_2O_7^{-}$	296.786	296.786				$Mo_3O_9^{-}$	419.675	419.675
$Mo_3O_{10}^{-}$	435.670	435.670				$HMo_4O_{13}^{-}$	576.569	576.570
MoO_3^{-}	139.889	139.892				$Mo_2O_8^{-}$	311.773	311.773

Table S2: Energies of calculated structures on the wB97XD level of theory. See Figures for the respective structures.

Ion	Isomer	Singlet/Douplet	Triplet/Quartet	Pentet/Hextet
[HMo$_2$O$_6$]$^-$	I	0.00	-	-
	II	0.25	0.64	-
	III	0.67	0.71	-
	IV	2.25	0.71	-
[CH$_3$OH]	I	0.00	-	-
[Mo$_2$O$_6$]$^-$	I	0.00	3.41	-
[Mo$_2$O$_8$]$^-$	I	0.00	2.89	-
	II	2.62	2.60	-
	III	2.71	2.72	-
[CO$_2$]	I	0.00	4.49	-
[Mo$_4$O$_{13}$]$^{2-}$	I	0.00	2.47	-
	II	0.20	3.32	-
	III	0.47	3.25	-
	IV	2.75	2.75	-
	V	2.85	2.85	-
	VI	3.26	3.21	-
	VII	4.08	4.08	-
	VIII	4.21	4.31	-
[CH$_2$O$_2$]	I	0.00	1.51	-
[Mo$_2$O$_8$]$^{2-}$	I	0.00	2.01	-
	II	2.12	2.12	-
	III	2.53	-	-
[Mo$_2$O$_6$]$^{2-}$	I	0.00	0.25	-
[HMo$_3$O$_{10}$]$^-$	I	0.00	2.95	-
	II	0.17	3.02	-
	III	0.50	3.07	-
	IV	0.77	2.91	-
	V	0.87	-	-
	VI	1.18	3.61	-
	VII	1.20	3.92	-

$[Mo_2O_7]^{2-}$	I	0.00	3.39	-
$[CH_3O]$	I	0.00	-	-
$[MoO_4]^-$	I	0.00	3.49	-
	II	3.49	3.89	7.33
$[HMoO_4]^-$	I	0.00	3.19	-
	II	5.02	5.41	-
	III	5.39	5.43	-
	IV	6.05	-	-
$[Mo_2O_5]^-$	I	0.00	0.02	3.13
$[HMo_4O_{12}]^-$	I	0.30	0.00	-
	II	1.22	0.10	-
	III	0.23	0.33	-
	IV	0.36	0.25	-
	V	0.35	0.35	-
	VI	0.39	2.28	-
	VII	1.75	1.25	-
$[MoO_3]^-$	I	0.00	2.96	6.49
$[Mo_3O_{10}]^-$	I	0.00	2.76	-
	II	0.32	2.29	-
	III	0.45	3.05	-
$[CH_3Mo_2O_7]$	I	0.00	3.26	-
$[CH_2O]$	I	0.00	2.92	-
$[HMo_4O_{13}]^-$	I	0.00	3.14	-
	II	0.21	2.70	-
	III	0.35	3.51	-
	IV	0.42	3.73	-
	V	0.48	2.61	-
	VI	0.61	3.77	-
	VII	0.66	2.72	-
	VIII	1.17	3.46	-
	IX	1.17	3.41	-
	X	1.31	2.95	-
	XI	1.40	3.74	-
	XII	1.73	4.20	-
	XIII	1.40	3.74	-
$[HMoO_3]$	I	0.00	0.37	-
$[HMo_2O_7]$	I	0.00	5.44	-

	II	0.69	2.91	-
[CH$_3$Mo$_2$O$_5$]	I	0.00	2.23	-
	II	1.19	1.28	-
	III	1.52	1.32	-
	IV	1.63	1.87	-
	V	2.33	2.35	-
[Mo$_3$O$_9$]$^-$	I	0.00	3.30	-
	II	0.27	3.45	-
[Mo$_2$O$_6$]	I	0.00	2.71	5.57
[HMo$_2$O$_7$]$^-$	I	0.00	2.73	-
	II	0.92	-	-
[HMoO$_4$]	I	0.00	3.41	-
	II	1.49	3.12	-
	III	2.59	4.97	-
	IV	3.25	4.64	-
[Mo$_2$O$_7$]$^-$	I	0.00	3.12	-
[O$_2$]	I	0.50	0.00	-
[CH$_3$Mo$_4$O$_{13}$]$^-$	I	0.00	2.66	-
	II	0.18	2.55	-
	III	0.23	2.48	-
	IV	0.26	3.55	-
	V	0.37	2.66	-
	VI	0.45	2.66	-
	VII	0.52	2.68	-
	VIII	0.92	-	-
	IX	0.97	1.71	-
	X	1.10	-	-
	XI	1.10	-	-
	XII	1.21	-	-
	XIII	1.22	-	-
	XIV	1.22	-	-
	XV	1.22	-	-
	XVI	1.45	-	-
	XVII	1.52	-	-
	XVIII	1.52	-	-
	XIX	1.54	3.12	-

	XX	1.54	-	-
	XXI	1.65	-	-
	XXII	1.65	-	-
	XXIII	1.65	-	-
	XXIV	1.65	-	-
	XXV	1.66	-	-
	XXVI	1.67	-	-
	XXVII	1.72	-	-
	XXVIII	1.74	-	-
	XXIX	1.84	-	-
	XXX	1.95	-	-
	XXXI	2.05	-	-
	XXXII	2.19	-	-
	XXXIII	2.25	-	-
	XXXIV	2.31	-	-
	XXXV	2.35	-	-
	XXXVI	2.49	-	-
	XXXVII	2.57	-	-
	XXXVIII	2.66	-	-
	XXXIX	2.66	-	-
	XL	2.76	-	-
	XLI	2.93	-	-
$[H_2O]$	I	0.00	-	-
$[CH_3MoO_2]$	I	0.00	2.47	-
$[CH_3MoO_4]$	I	0.00	3.23	-
	II	1.49	3.17	-
	III	1.62	-	-
	IV	2.45	4.76	-
$[Mo_3O_{10}]^{2-}$	I	0.00	3.54	-
	II	1.14	3.58	-
	III	4.24	4.14	-
	IV	4.98	4.81	-
$[Mo_3O_9]$	I	0.00	3.08	-
	II	1.38	3.61	-
$[Mo_4O_{12}]^-$	I	0.00	3.33	-

	II	0.36	3.47	-
	III	0.80	3.44	-
	IV	1.11	-	-
	V	1.53	3.71	-
[CH$_2$Mo$_2$O$_7$]	I	0.00	3.23	-
[H$_3$Mo$_4$O$_{13}$]$^-$	I	0.01	0.00	-
	II	0.44	0.61	-
	III	0.55	0.54	-
	IV	0.59	0.59	-
	V	1.05	1.01	-
	VI	1.22	1.24	-
	VII	1.26	1.24	-
	VIII	1.58	1.61	-
	IX	2.00	2.00	-
	X	2.04	2.03	-
	XI	2.32	2.32	-
	XII	2.59	2.57	-
[CH$_3$Mo$_2$O$_7$]$^-$	I	0.00	3.16	-
	II	0.94	3.81	-
[H$_2$]	I	0.00	-	-
[MoO$_3$]	I	0.00	1.70	4.41
[Mo$_3$O$_{11}$]$^-$	I	0.00	-	-
	II	2.38	2.67	-
[Mo$_2$O$_7$]	I	0.00	0.00	-
	II	0.71	0.67	-
[MoO$_4$]	I	0.02	0.00	-
[Mo$_2$O$_5$]	I	0.00	0.53	3.12

Table S3: Benchmarking various properties of small molybdenum oxide and sulfide clusters using different DFT functionals along with the def2TZVP basis set. Experimental values taken from Ref. [1–3].

quantity	ωB97XD	B3LYP	M06	exp.
D_0(Mo$^+$-O)	4.76	5.07	5.10	5.06 ± 0.02
D_0(OMo$^+$-O)	5.32	5.25	5.24	5.57 ± 0.14
IE(MoO)	7.38	7.57	7.48	7.79 ± 0.22
D_0(Mo$^+$–S)	3.43	3.73	3.75	3.68 ± 0.06
D_0(SMo$^+$–S)	3.89	3.87	4.35	4.10 ± 0.10
D_0(Mo$^+$–S2)	2.81	3.17	3.53	3.42 ± 0.11
D_0(S$_2$Mo$^+$–S)	3.45	3.48	3.71	3.37 ± 0.09
D_0(SMo$^+$–S$_2$)	2.83	2.92	3.50	3.11 ± 0.13

References

1. Sievers, M.R., Chen, Y.-M., Armentrout, P.B.: Metal oxide and carbide thermochemistry of Y^+, Zr^+, Nb^+, and Mo^+. J. Chem. Phys. **105**, 6322–6333 (1996)

2. Sievers, M.R., Armentrout, P.B.: Reactions of CO and CO_2 with Gas-Phase Mo^+, MoO^+, and MoO_2^+. J. Phys. Chem. A **102**, 10754–10762 (1998)

3. Kretzschmar, I., Schröder, D., Schwarz, H., Armentrout, P.B.: Structure, Thermochemistry, and Reactivity of MS_n^+ Cations (M=V, Mo; n=1–3) in the Gas Phase. Int. J. Mass Spectrom. **228**, 439–456 (2003)

4. Zhai, H.-J., Huang, X., Waters, T., Wang, X.-B., O'Hair, R.A.J., Wedd, A.G., Wang, L.-S.: Photoelectron spectroscopy of doubly and singly charged group VIB dimetalate anions: $M_2O_7^{2-}$, $MM'O_7^{2-}$, and $M_2O_7^-$ (M, M' = Cr, Mo, W). J. Phys. Chem. A **109**, 10512–10520 (2005)

5. Zhai, H.-J., Kiran, B., Cui, L.-F., Li, X., Dixon, D.A., Wang, L.-S.: Electronic structure and chemical bonding in MO(n)- and MO(n) clusters (M = Mo, W; n = 3-5): a photoelectron spectroscopy and ab initio study. J. Am. Chem. Soc. **126**, 16134–16141 (2004)
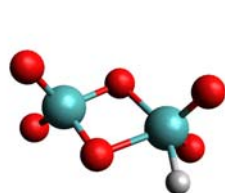
Table S4: Benchmarking vertical detachment energies (in eV) for various molybdenum oxides using different DFT functionals along with the def2TZVP basis set. Experimental values are taken from [4, 5].

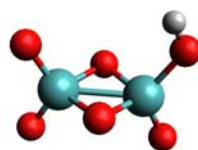| Ion	ωB97XD	B3LYP	M06	exp.
$[MoO_3]^-$	3.29	3.41	3.26	3.33(5)
$[MoO_4]^-$	5.49	5.37	5.44	5.45(5)
$[MoO_5]^-$	5.48	5.39	5.41	5.40(5)
$[Mo_2O_7]^-$	6.90	6.60	6.76	~ 6.5
$[Mo_2O_7]^{2-}$	2.30	1.80	1.98	~ 2.2

Calculated structures on the wB97XD/def2TZVP level of theory

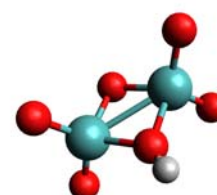Zero-point corrected energies are given in eV relative to the most stable structure in Figures S3.1-S3.43.
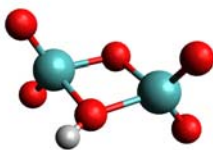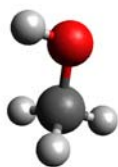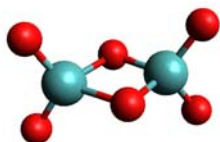

^{1}I E=0.00   II E=0.25   III E=0.67

IV* E=0.71

Figure S3.1: Calculated Molecules of [HMo$_2$O$_6$]$^-$ in lowest multiplicity unless marked with *.
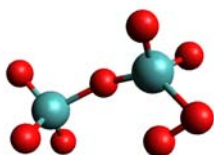

I E=0.00

Figure S3.2: Calculated Molecules of [CH$_3$OH] in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.S3.3: Calculated Molecules of [Mo$_2$O$_6$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00   II* E=2.60   III E=2.71

Figure S3.4: Calculated Molecules of [Mo$_2$O$_8$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.5: Calculated Molecules of [CO_2] in lowest multiplicity unless marked with *.

I E=0.00

II E=0.20

III E=0.47

IV E=2.75

V E=2.85

VI* E=3.21

VII* E=4.08

VIII E=4.21

Figure S3.6: Calculated Molecules of [Mo_4O_{13}]$^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.7: Calculated Molecules of [CH_2O_2] in lowest multiplicity unless marked with *.

I E=0.00

II E=0.17

III E=0.50

IV E=0.77

V E=0.87

VI E=1.18

VII E=1.20

Figure S3.10: Calculated Molecules of [HMo_3O_{10}]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.11: Calculated Molecules of $[Mo_2O_7]^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.12: Calculated Molecules of $[CH_3O]$ in lowest multiplicity unless marked with *.

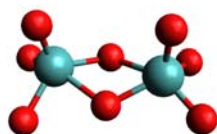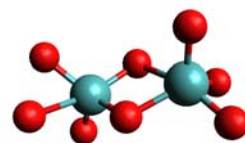I E=0.00   II E=3.49

Figure S3.13: Calculated Molecules of $[MoO_4]^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=5.02   III E=5.39

IV E=6.05

Figure S3.14: Calculated Molecules of [HMoO$_4$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.16: Calculated Molecules of [Mo$_2$O$_5$]$^-$ in lowest multiplicity unless marked with *.
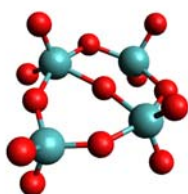

I* E=0.00

II* E=0.10

III E=0.23

IV* E=0.25

V E=0.35

VI E=0.39

VII* E=1.25

Figure S3.17: Calculated Molecules of [HMo$_4$O$_{12}$]$^-$ in lowest multiplicity unless marked with *.
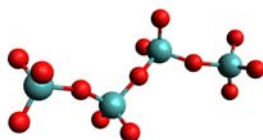

I E=0.00

Figure S3.18: Calculated Molecules of [MoO$_3$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=0.32   III E=0.45

Figure S3.19: Calculated Molecules of [Mo$_3$O$_{10}$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.20: Calculated Molecules of [CH$_3$Mo$_2$O$_7$] in lowest multiplicity unless marked with *.

I E=0.00

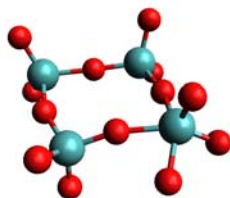Figure S3.21: Calculated Molecules of [CH$_2$O] in lowest multiplicity unless marked with *.

I E=0.00

II E=0.21

III E=0.35

IV E=0.42

V E=0.48

VI E=0.61

VII E=0.66

VIII E=1.17

IX E=1.17

X E=1.31

XI E=1.40

XII E=1.73

XIII E=1.40

Figure S3.21: Calculated Molecules of [HMo$_4$O$_{13}$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

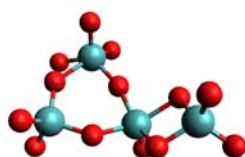Figure S3.22: Calculated Molecules of [HMoO$_3$] in lowest multiplicity unless marked with *.
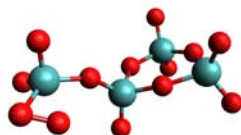

I E=0.00

II E=0.69

Figure S3.23: Calculated Molecules of [HMo$_2$O$_7$] in lowest multiplicity unless marked with *.

I E=0.00   II E=1.19   III* E=1.32

IV E=1.63   V E=2.33

Figure S3.24: Calculated Molecules of [$CH_3Mo_2O_5$] in lowest multiplicity unless marked with *.

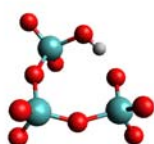I E=0.00   II E=0.27

Figure S3.25: Calculated Molecules of [Mo_3O_9]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.26: Calculated Molecules of [Mo_2O_6] in lowest multiplicity unless marked with *.

I E=0.00   II E=0.92
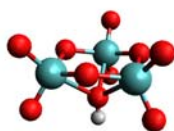
Figure S3.27: Calculated Molecules of $[HMo_2O_7]^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=1.49   III E=2.59

IV E=3.25

Figure S3.28: Calculated Molecules of $[HMoO_4]$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.29: Calculated Molecules of $[Mo_2O_7]^-$ in lowest multiplicity unless marked with *.

I* E=0.00

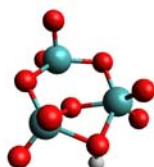Figure S3.30: Calculated Molecules of [O_2] in lowest multiplicity unless marked with *.

I E=0.00

II E=0.18

III E=0.23

IV E=0.26

V E=0.37

VI E=0.45

VII E=0.52

VIII E=0.92

IX E=0.97

X E=1.10

XI E=1.10

XII E=1.21

XIII E=1.22

XIV E=1.22

XV E=1.22

XVI E=1.45

XVII E=1.52

XVIII E=1.52

XIX E=1.54

XX E=1.54

XXI E=1.65

XXII E=1.65

XXIII E=1.65

XXIV E=1.65

XXV E=1.66

XXVI E=1.67

XXVII E=1.72

XXVIII E=1.74

XXIX E=1.84

XXX E=1.95

XXXI E=2.05   XXXII E=2.19   XXXIII E=2.25

XXXIV E=2.31   XXXV E=2.35   XXXVI E=2.49

XXXVII E=2.57   XXXVIII E=2.66   XXXIX E=2.66

XL E=2.76   XLI E=2.93

Figure S3.28: Calculated Molecules of [CH$_3$Mo$_4$O$_{13}$]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.29: Calculated Molecules of [H$_2$O] in lowest multiplicity unless marked with *.
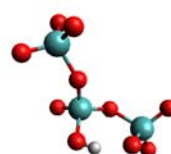

I E=0.00

Figure S3.30: Calculated Molecules of [CH$_3$MoO$_2$] in lowest multiplicity unless marked with *.

I E=0.00   II E=1.49   III E=1.62
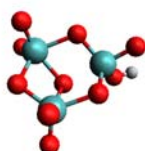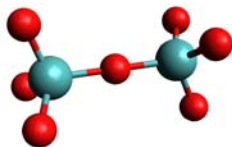

IV E=2.45

Figure S3.31: Calculated Molecules of [CH$_3$MoO$_4$] in lowest multiplicity unless marked with *.

I E=0.00   II E=1.14   III* E=4.14

IV* E=4.81

Figure S3.32: Calculated Molecules of [Mo$_3$O$_{10}$]$^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00   II E=1.38

Figure S3.33: Calculated Molecules of [Mo$_3$O$_9$] in lowest multiplicity unless marked with *.

I E=0.00   II E=0.36   III E=0.80

IV E=1.11   V E=1.53

Figure S3.34: Calculated Molecules of $[Mo_4O_{12}]^-$ in lowest multiplicity unless marked with *.

I E=0.00

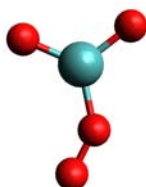Figure S3.35: Calculated Molecules of $[CH_2Mo_2O_7]$ in lowest multiplicity unless marked with *.

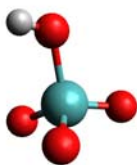I* E=0.00   II E=0.44   III* E=0.54

IV* E=0.59   V* E=1.01   VI E=1.22

VII* E=1.24   VIII E=1.58   IX* E=2.00

X* E=2.03   XI E=2.32   XII* E=2.57

Figure S3.36: Calculated Molecules of $[H_3Mo_4O_{13}]^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=0.94

Figure S3.37: Calculated Molecules of $[CH_3Mo_2O_7]^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.38: Calculated Molecules of [H_2] in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.39: Calculated Molecules of [MoO_3] in lowest multiplicity unless marked with *.
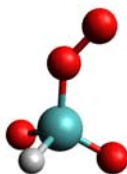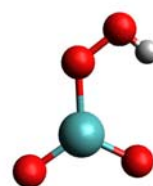

I E=0.00   II E=2.38

Figure S3.40: Calculated Molecules of [Mo_3O_{11}]$^-$ in lowest multiplicity unless marked with *.

I E=0.00   II* E=0.67

Figure S3.41: Calculated Molecules of [Mo_2O_7] in lowest multiplicity unless marked with *.
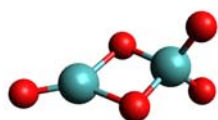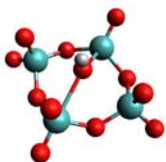
I* E=0.00

Figure S3.42: Calculated Molecules of [MoO$_4$] in lowest multiplicity unless marked with *.

I E=0.00

Figure S3.43: Calculated Molecules of [Mo$_2$O$_5$] in lowest multiplicity unless marked with *.

Calculated transition states at the wB97XD/def2TZVP level of theory

I → II E=1.44   I→V E=0.94   II→V E=1.30

Figure S3.44: Calculated transition states of $[HMo_4O_{13}]^-$ in lowest multiplicity. The given energies are the barriers in eV.

I → II E=3.22   I→III E=3.17   II→III E=2.94

Figure S3.45: Calculated transition states of $[CH_3Mo_4O_{13}]^-$ in lowest multiplicity. The given energies are the barriers in eV.

Cartesian coordinates (in Ångstrom) of transition states optimized at the wB97XD/def2TZVP level of theory along the respective ZPE-corrected energy (in eV)

[HMo$_4$O$_{13}$]$^-$ I → II
E = -34075.44

Mo	0.05649	1.91937	0.16136
O	3.08540	-0.12063	-1.80126
O	0.04093	3.16713	-0.95099
O	-0.09795	2.57053	1.69645
O	-1.89258	1.38214	0.07042
Mo	2.85571	0.00605	-0.14047
Mo	-3.01481	-0.02602	-0.15868
O	-3.74932	-0.53663	1.26517
O	-4.20224	0.28928	-1.31135
O	1.98349	1.56754	0.32256
O	0.22923	-0.02030	-0.60424
O	4.38060	0.05915	0.55800
Mo	0.17281	-1.90016	0.17668
O	1.99015	-1.53733	0.55696
O	-0.53363	-2.19807	1.66559
O	-1.69751	-1.24718	-0.81416
O	0.20581	-3.31151	-0.72524
H	-0.88717	-0.48095	-1.05622

[HMo$_4$O$_{13}$]$^-$ I → V
E = -34075.94

Mo	2.53305	-0.00005	-0.48350
O	0.12887	-3.39184	-0.30941
O	1.70077	-0.00003	-2.05801
O	4.19800	-0.00002	-0.66424
O	1.98669	1.47142	0.40667
Mo	-0.00054	-1.95088	0.53090
Mo	-0.00061	1.95099	0.53082
O	0.03360	2.27275	2.17218
O	0.12915	3.39180	-0.30970
O	1.98669	-1.47166	0.40652
O	-0.00240	-0.00008	-0.38908
O	0.03361	-2.27228	2.17233
Mo	-2.53126	-0.00006	-0.48627
O	-1.87108	-1.56810	0.30651
O	-4.18129	0.00011	-0.18013
O	-1.87091	1.56814	0.30630
O	-2.35553	-0.00023	-2.16001
H	0.64385	-0.00008	-1.46121

[HMo$_4$O$_{13}$]$^-$ II → V
E = -34075.59

Mo	2.71602	-0.22221	-0.40588
O	0.05383	2.22969	2.02049
O	3.97030	-0.13217	0.70728
O	3.38741	-0.37535	-1.93795
O	1.61895	-1.69162	-0.02832
Mo	0.09812	2.22516	0.34434
Mo	-0.18921	-2.00227	0.47769
O	-0.65576	-3.36977	-0.58542
O	-0.18231	-2.53784	2.05505
O	1.64467	1.27440	-0.32431
O	-0.57868	-0.18544	0.46808
O	0.44012	3.80311	-0.10523
Mo	-2.44651	0.13281	-0.42625
O	-1.56319	1.89710	-0.39577
O	-2.85948	0.10413	-2.05469
O	-2.17612	-1.79386	-0.16101
O	-3.82445	0.42541	0.48397
H	-1.69579	-2.78857	-0.71331

[CH$_3$Mo$_4$O$_{13}$]$^-$ I → II
E = -35142.15

O	0.00933	-3.24832	-1.34011
O	-0.06774	-2.70725	1.30551
Mo	0.10367	-2.00949	-0.21280
O	1.97051	-1.62734	-0.46587
O	-1.66642	-1.26018	-0.64446
O	4.25350	-0.02644	-0.58102
O	2.89085	0.16430	1.74076
Mo	2.70393	0.03169	0.06740
O	0.37054	0.06772	0.41798
Mo	-2.82640	0.01281	0.00435
O	-3.88431	0.57554	-1.17935
O	-3.70391	-0.58746	1.31873
O	1.98270	1.61055	-0.68773
O	-1.67843	1.32692	0.65220
Mo	0.13270	1.91350	-0.26995
O	-0.56442	2.25702	-1.76184
O	0.21922	3.28457	0.70492
C	-0.65284	0.40768	2.32081
H	0.39157	0.61302	2.50668
H	-1.37882	1.10249	2.70928
H	-0.93096	-0.63600	2.28322

[CH$_3$Mo$_4$O$_{13}$]$^-$ I → III
E = -35142.20

O	3.89823	-0.39649	1.07355
O	3.57749	-1.20229	-1.51849
Mo	2.77380	-0.60157	-0.16397
O	1.45062	-1.73936	0.36042
O	1.94098	1.01429	-0.60184
O	-0.71071	-2.97013	1.65887
O	-0.56817	-2.94832	-1.09623
Mo	-0.50765	-1.99730	0.30631
O	-0.44999	0.04482	0.43255
Mo	0.56985	2.20939	-0.03593
O	0.70693	2.52917	1.60017
O	1.06540	3.61081	-0.81005
O	-2.32589	-1.32122	0.06263
O	-1.26750	2.17579	-0.55022
Mo	-2.25907	0.68655	0.08320
O	-3.02340	1.16106	1.49276
O	-3.45011	0.37870	-1.16053
C	-3.41622	-1.63514	-1.68305
H	-2.55324	-2.27771	-1.80841
H	-4.20756	-1.99516	-1.04664
H	-3.72399	-1.08780	-2.55935

[CH$_3$Mo$_4$O$_{13}$]$^-$ II → III
E = -35142.43

O	-4.31806	0.03598	-0.33167
O	-2.69118	-0.09687	1.86591
Mo	-2.71256	0.01694	0.16945
O	-1.86209	-1.44320	-0.57559
O	-1.89067	1.56560	-0.38273
O	-0.02625	-3.01584	-1.71687
O	0.01103	-2.93927	1.00258
Mo	0.00551	-1.99652	-0.38419
O	0.13562	0.03590	0.30010
Mo	0.03501	1.98076	-0.39600
O	-0.01290	2.76428	-1.88009
O	0.13952	3.10782	0.84708
O	1.90969	-1.53572	-0.64025
O	1.97868	1.52888	-0.60887
Mo	2.60633	-0.01610	0.13209
O	4.25890	-0.05319	-0.16045
O	2.38871	-0.02137	1.85213
C	0.31107	0.12007	2.57104
H	0.88278	-0.56034	3.17808
H	0.52597	1.17500	2.65481
H	-0.68325	-0.17247	2.27405

Cartesian coordinates (in Ångstrom) of species optimized at the wB97XD/def2TZVP level of theory along the respective ZPE-corrected energy (in eV)

[HMo$_2$O$_6$]$^-$ I 1/2

E = -16022.04

Mo	1.495589	0.000346	-0.095315
Mo	-1.460429	0.000224	0.018274
O	-0.054987	0.006121	1.217293
O	-0.253002	-0.004390	-1.338410
O	2.366369	-1.388003	0.294968
O	2.373747	1.383161	0.299087
O	-2.433005	1.396627	0.068489
O	-2.430946	-1.397108	0.080205
H	1.977878	0.004796	-1.737337

[HMo$_2$O$_6$]$^-$ II 1/2

E = -16021.78

Mo	1.257240	-0.097016	0.000235
Mo	-1.289490	0.020129	-0.000024
O	0.007134	0.092026	1.444152
O	0.007495	0.092778	-1.443787
O	1.938898	-1.634238	-0.000627
O	2.349119	1.480105	-0.000325
O	-2.167427	1.496727	0.000077
O	-2.210493	-1.418180	-0.000511
H	1.956694	2.355479	-0.000709

[HMo$_2$O$_6$]$^-$ II 3/4

E = -16021.40

Mo	1.467921	-0.083624	-0.000125
Mo	-1.526250	0.012285	-0.000005
O	0.055343	-0.099425	1.230436
O	0.055299	-0.099058	-1.230653
O	2.360379	-1.516965	0.000102
O	2.412831	1.582952	0.000432
O	-2.263823	1.558752	-0.000038
O	-2.555825	-1.353552	0.000283
H	1.936188	2.414593	0.000952

[HMo$_2$O$_6$]$^-$ III 1/2

E = -16021.37

Mo	1.315171	0.000050	-0.023936
Mo	-1.315124	0.000078	-0.024098
O	0.000005	0.000216	-1.433584
O	-0.000194	0.000100	1.530038
O	2.183434	1.475835	-0.038560
O	2.182072	-1.476479	-0.038962
O	-2.182533	-1.476202	-0.038666
O	-2.182968	1.476097	-0.038561
H	-0.000528	-0.001946	2.483829

[HMo$_2$O$_6$]$^-$ III 3/4

E = -16021.33

Mo	1.601707	-0.000004	-0.009434
Mo	-1.601702	-0.000006	-0.009430
O	-0.000006	-0.000067	-1.091775
O	0.000008	-0.000461	1.339080
O	2.502784	1.445914	-0.108557
O	2.503357	-1.445558	-0.108809
O	-2.503312	-1.445575	-0.108807
O	-2.502877	1.445857	-0.108544
H	0.000159	-0.000456	2.291588

[HMo$_2$O$_6$]$^-$ IV 1/2

E = -16019.79

Mo	-1.349844	-0.047641	-0.000010
Mo	1.349748	-0.047598	0.000025
O	0.000008	0.297387	1.346471
O	0.000228	0.299022	-1.345884
O	-2.270897	1.388689	-0.000051
O	-2.440215	-1.342460	-0.000256

O	2.439637	-1.342873	-0.000713
O	2.271808	1.388017	0.000460
H	-0.000489	-1.502217	-0.000847

[HMo$_2$O$_6$]$^-$ IV 3/4

E = -16021.33

Mo	-1.601689	0.000187	0.009501
Mo	1.601700	0.000094	0.009396
O	0.000111	0.003639	1.091808
O	-0.000176	0.006676	-1.339090
O	-2.508607	1.442256	0.111017
O	-2.497615	-1.449087	0.106157
O	2.499229	-1.448006	0.109178
O	2.507051	1.443320	0.108173
H	-0.000384	-0.002199	-2.291610

[CH$_3$OH] I 1/2

E = -3147.90

C	0.659632	-0.020077	0.000019
H	1.024029	-0.539354	0.893766
H	1.023104	-0.549797	-0.887931
H	1.082029	0.983939	-0.006169
O	-0.743325	0.121832	0.000016
H	-1.140359	-0.748982	0.000092

[Mo$_2$O$_6$]$^-$ I 1/2

E = -16006.13

Mo	1.403437	0.000044	0.000054
Mo	-1.446626	0.000056	-0.000169
O	0.094879	0.000873	-1.314181
O	0.094674	-0.000593	1.314302
O	2.373825	1.396229	0.001130
O	2.373153	-1.396494	-0.001190
O	-2.354370	-1.438436	0.000728
O	-2.355414	1.437896	-0.000184

[Mo$_2$O$_6$]$^-$ I 3/4

E = -16002.72

Mo	1.484339	-0.044199	0.000165
Mo	-1.518875	0.013273	-0.000210
O	0.076084	-0.033485	-1.227412
O	-2.346481	1.509351	-0.000129
O	2.495595	1.618392	-0.000184
O	2.352218	-1.494367	-0.000053
O	0.075406	-0.032261	1.227275
O	-2.471509	-1.405271	0.000741

[Mo$_2$O$_8$]$^-$ I 1/2

E = -20097.82

Mo	1.995961	0.011890	-0.005209
Mo	-1.707964	-0.256045	-0.015308
O	-1.622347	1.835191	-0.054689
O	0.075407	-0.469774	-0.219335
O	-2.467261	-1.018309	-1.311297
O	-2.165620	-0.989854	1.428957
O	2.568889	0.636216	-1.491856
O	2.873163	-1.388239	0.437358
O	-2.872569	1.473771	0.097896
O	2.098357	1.202812	1.220681

[Mo$_2$O$_8$]$^-$ I 3/4

E = -20094.93

Mo	1.626513	0.036895	-0.024119
Mo	-1.312880	-0.191671	-0.114347
O	-1.642221	1.929641	-0.158447
O	0.254872	-0.222288	-1.292729
O	-2.630507	-0.906287	-0.864071
O	-1.074098	-1.785012	0.976397
O	2.578754	1.412450	-0.342652
O	2.624579	-1.317418	0.234108

O	-2.158579	1.398184	0.913969
O	0.400625	0.303306	1.260375

[Mo$_2$O$_8$]$^-$ II 1/2

E = -20095.20

Mo	1.682747	0.061810	-0.017638
Mo	-1.682723	-0.062230	-0.017382
O	-0.042654	-1.036184	0.413789
O	-2.497989	1.553699	0.584515
O	0.042323	1.036893	0.410790
O	1.915047	0.173717	-1.679398
O	2.500669	-1.549717	0.591546
O	-2.833044	-1.037234	0.773470
O	2.832037	1.042745	0.767624
O	-1.916519	-0.181713	-1.678481

[Mo$_2$O$_8$]$^-$ II 3/4

E = -20095.22

Mo	1.700893	-0.053153	0.022893
Mo	-1.700870	0.053308	0.022644
O	-0.106217	1.046504	-0.372338
O	-2.390622	-1.514504	-0.809642
O	0.106116	-1.046699	-0.370927
O	1.927898	0.041753	1.686592
O	2.392493	1.510937	-0.814361
O	-2.905244	1.067973	-0.624760
O	2.905588	-1.070498	-0.619976
O	-1.930133	-0.036279	1.686342

[Mo$_2$O$_8$]$^-$ III 1/2

E = -20095.11

Mo	1.573720	-0.129478	-0.010758
Mo	-1.441329	0.193983	0.013583
O	-0.153335	1.024499	0.900307
O	-2.588482	-1.553854	-0.579878
O	-2.216702	1.286524	-1.016576
O	1.987165	-1.024694	1.352999
O	2.141038	1.682473	0.119750
O	-2.575252	-0.468941	1.209101
O	2.740911	-0.464856	-1.196843
O	-0.030394	-0.819802	-0.803690

[Mo$_2$O$_8$]$^-$ III 3/4

E = -20095.11

Mo	1.572846	-0.129474	-0.010311
Mo	-1.442220	0.194960	0.014133
O	-0.151757	1.025334	0.897615
O	-2.586394	-1.554423	-0.581395
O	-2.219472	1.287747	-1.014301
O	1.985675	-1.027736	1.351595
O	2.154452	1.678109	0.120825
O	-2.575121	-0.470150	1.209125
O	2.738517	-0.464706	-1.198662
O	-0.031687	-0.817980	-0.804873

[CO$_2$] I 1/2

E = -5131.94

C	-0.000000	0.000356	0.000000
O	1.155926	-0.000294	0.000000
O	-1.155926	0.000027	0.000000

[CO$_2$] I 3/4

E = -5127.45

C	-0.000000	0.460418	0.000000
O	1.059676	-0.172654	0.000000
O	-1.059676	-0.172659	0.000000

[Mo$_4$O$_{13}$]$^{2-}$ I 1/2

E = -34060.70

Mo 0.120352   2.061413 -0.145228
O -3.490710   0.001552   1.708918
O -0.116543   3.180903   1.098048
O   0.048301   2.871607 -1.624947
O   1.997105   1.828387   0.035802
Mo -2.862425 -0.000004   0.125238
Mo 2.580880   0.000036   0.157782
O   3.754513   0.001054 -1.066054
O   3.352821 -0.001365   1.666091
O -1.780299   1.429650 -0.144483
O   0.460807   0.000081 -0.018858
O -4.159798 -0.000873 -0.980908
Mo 0.120305 -2.061280 -0.145286
O -1.780782 -1.430766 -0.140885
O   0.046312 -2.872108 -1.624569
O   1.997283 -1.828303   0.032763
O -0.114354 -3.180687   1.098426

[Mo$_4$O$_{13}$]$^{2-}$ I 3/4
E = -34058.23
Mo -0.138892 -2.060152 -0.249408
O   3.376377 -0.219651   1.830828
O   0.194693 -3.217133   0.936670
O -0.153638 -2.830050 -1.751722
O -1.979105 -1.793031   0.115489
Mo 2.820180   0.027007   0.241666
Mo -2.550040   0.062303   0.291041
O -3.828338   0.083323 -0.821174
O -3.180829   0.027671   1.862989
O   1.729810 -1.334759 -0.285882
O -0.486857   0.001913 -0.188693
O   4.168068   0.137575 -0.794081
Mo -0.087858   1.941805 -0.353207
O   1.820574   1.550410   0.158216
O -0.023110   2.536777 -1.915928
O -1.941001   1.885385   0.179008
O   0.075562   3.324008   1.041298

[Mo$_4$O$_{13}$]$^{2-}$ II 1/2
E = -34060.51
Mo 4.937626 -0.328274   0.358191
O   5.137318 -0.838112   1.980385
O   5.910439   1.047503   0.051492
O   5.362890 -1.596223 -0.710732
Mo 1.531446   1.075551 -0.354368
Mo -1.529628 -1.067495 -0.358781
O -1.392852 -2.439083   0.608376
O -1.600834 -1.579827 -1.961378
O -3.032182 -0.148866   0.067872
O   0.000700   0.003863 -0.114121
O   1.601400   1.598061 -1.953850
O   1.396743   2.440684   0.621808
O   3.033779   0.153188   0.064386
Mo -4.939281   0.320929   0.359597
O -5.188568   0.631420   2.024866
O -5.314422   1.716319 -0.558613
O -5.915270 -0.992859 -0.144847

[Mo$_4$O$_{13}$]$^{2-}$ II 3/4
E = -34057.39
Mo 4.952040 -0.260041   0.339038
O   5.169474 -0.735631   1.970104
O   5.874581   1.146068   0.011967
O   5.429798 -1.528785 -0.707824
Mo 1.495044   1.010357 -0.374163
Mo -1.560729 -1.156999 -0.289182
O -1.474930 -2.437806   0.807311
O -1.653837 -1.789491 -1.851833
O -2.997911 -0.127486   0.055004
O   0.016873 -0.112406 -0.130487
O   1.538420   1.540548 -1.973679
O   1.319661   2.368819   0.608196
O   3.036966   0.141080   0.034394
Mo -4.880815   0.490912   0.345474
O -5.235675   0.854075   1.978220
O -5.445360   1.553255 -0.869463
O -5.607149 -1.314443 -0.043037

[Mo$_4$O$_{13}$]$^{2-}$ III 1/2
E = -34060.23
Mo 0.557904 -1.118736   0.040469
Mo 4.119201   0.105042 -0.020887
Mo -1.543666   1.997985   0.009299
Mo -3.337439 -0.906776 -0.028312
O -3.135690   1.001430 -0.042073
O -1.507885   2.955564   1.403812
O   2.220291 -0.361301   0.044158
O   5.046733 -1.255907   0.462110
O -4.214690 -1.410630   1.340407
O -1.669596 -1.447699   0.027211
O -0.233494   0.758940   0.012495
O   4.543615   0.551831 -1.622340
O   4.420136   1.415913   1.044282
O -1.450596   3.009113 -1.344268
O -4.134531 -1.437823 -1.435419
O   0.583222 -2.075655   1.423257
O   0.603484 -2.110733 -1.316614

[Mo$_4$O$_{13}$]$^{2-}$ III 3/4
E = -34057.45
Mo -0.427421 -1.170923 -0.025775
Mo -4.086446   0.043677   0.031601
Mo 1.498567   2.019405   0.002846
Mo 3.321614 -0.896743   0.009138
O   3.094780   1.001412   0.017810
O   1.452864   2.998007 -1.385559
O -2.390818 -0.411099 -0.023577
O -5.057551 -0.921274 -0.994156
O   4.195448 -1.374393 -1.372461
O   1.675255 -1.528699 -0.012373
O   0.181646   0.816296 -0.002991
O -4.690658   0.025138   1.624221
O -4.354243   1.810295 -0.659134
O   1.430672   3.007745   1.383654
O   4.166838 -1.390115   1.403091
O -0.644852 -1.988196 -1.507212
O -0.667525 -2.021049   1.435185

[Mo$_4$O$_{13}$]$^{2-}$ IV 1/2
E = -34057.95
Mo 3.030845   0.019416   0.113343
O   4.021029   0.034236 -1.277468
O   3.854842 -0.140485   1.600365
O   1.749307   1.513653   0.150498
Mo 0.094188 -2.371253 -0.127550
Mo 0.127155   2.335863 -0.090166
O   0.009110   2.887785 -1.691517
O   0.028530   3.675369   0.954986
O -1.205940   1.159062   0.278030
O   0.182239 -3.310923 -1.539259
O -1.488631 -1.431444 -0.123843
O   0.138012 -3.411941   1.212858
O   1.500368 -1.231039 -0.068192
Mo -2.854257 -0.060307   0.138616
O -4.041012 -0.766203 -0.864385
O -3.407598 -0.212160   1.722222
O -3.429396   1.634567 -0.534026

[Mo$_4$O$_{13}$]$^{2-}$ IV 3/4
E = -34057.95
Mo -3.030287 -0.017944   0.118233
O -4.025732 -0.032514 -1.268896
O -3.848167   0.142723   1.608562
O -1.750667 -1.514522   0.150715
Mo -0.093733   2.371398 -0.132201
Mo -0.128849 -2.336293 -0.094679
O -0.014196 -2.884481 -1.697612
O -0.029975 -3.678900   0.946481
O   1.208571 -1.164943   0.273498
O -0.177981   3.302949 -1.549490
O   1.487155   1.428345 -0.118674
O -0.139580   3.419526   1.202400
O -1.499529   1.231183 -0.069474
Mo 2.854030   0.059266   0.143655
O   4.044865   0.774341 -0.848955
O   3.400116   0.201183   1.730570
O   3.439025 -1.626130 -0.542919

[Mo$_4$O$_{13}$]$^{2-}$ V 1/2
E = -34057.86
Mo -2.017909   1.894809 -0.223339
O   2.250821 -1.285055   0.128885
O -2.152822 -2.330153 -1.756042
O -2.803622   2.561882   1.135542
O -2.665077   0.276665 -0.610070
Mo 3.581899 -0.138905 -0.380931
Mo -2.148704 -1.779780 -0.159798
O -3.918891 -1.473182   0.503778
O -2.141881 -3.181996   0.815529
O -2.210806   2.951697 -1.549009
O   2.486914   1.286485 -0.035234
O   4.036058 -0.283264 -2.019188
Mo 0.930542   0.178506   0.750985
O   4.969072 -0.200316   0.613651
O -0.214526   1.639882   0.153549
O -0.476129 -1.035349   0.282971
O   1.025292   0.260895   2.404327

[Mo$_4$O$_{13}$]$^{2-}$ V 3/4
E = -34057.85
Mo 2.295237   1.761852   0.148378
O -2.021094 -1.234273 -0.010337
O   3.177187   0.096081   0.363200
O   2.442777   2.689739   1.562502
O   2.981535   2.615738 -1.149337
Mo -3.578552   0.082534   0.388680
Mo 2.459612 -1.663431   0.235743
O   3.278835 -2.543017 -0.966305
O   2.621656 -2.468233   1.723295
O -3.582533   0.340658   2.064845
O -2.178722   1.157559 -0.373939
O -4.444564 -1.604110   0.021552
Mo -0.783862 -0.102163 -0.797034
O -4.952060   0.865339 -0.279302
O   0.599234   1.324585 -0.177610
O   0.756978 -1.402699 -0.189245
O -0.739516 -0.251025 -2.462104

[Mo$_4$O$_{13}$]$^{2-}$ VI 1/2
E = -34057.44
Mo -2.404935 -1.340211 -0.606620
Mo -1.654031   1.924639   0.135727
Mo 0.837260 -0.682734   1.069419
O -2.719765   0.530002 -0.601133
O -0.850283 -1.532110   0.235809
O -0.142684   1.135621   0.631495
O -1.394477   3.138724 -1.020686
O -2.453688   2.607190   1.471971
O -3.650675 -2.139503   0.229453
O -2.359643 -1.925592 -2.198525
O   2.551038   0.364228   1.133033
O   1.605359 -0.656631 -1.065912
Mo 3.099539   0.229083 -0.635806
O   3.260584   1.739336 -1.421809
O   4.538521 -0.674009 -0.856913
O   0.472688 -0.869046   2.686740
O   1.784404 -2.404789   0.972196

[Mo$_4$O$_{13}$]$^{2-}$ VI 3/4
E = -34057.49
Mo -2.418710 -1.343779 -0.596909
Mo -1.667242   1.923070   0.131172
Mo 0.839437 -0.671801   1.054745
O -2.745349   0.525229 -0.580129
O -0.848126 -1.525276   0.214732
O -0.145910   1.136798   0.600975
O -1.429643   3.131505 -1.036177
O -2.442111   2.611163   1.479199
O -3.644190 -2.154036   0.259205
O -2.395672 -1.923924 -2.191734
O   2.542599   0.407559   1.123200
O   1.639303 -0.685753 -1.059842
Mo 3.121644   0.224095 -0.627595
O   3.288066   1.712253 -1.454187
O   4.570428 -0.675486 -0.794817

O   0.470067   -0.867415   2.668383
O   1.796110   -2.383441   0.973775

[Mo₄O₁₃]²⁻ VII 1/2
E = -34056.62
Mo   3.750691   0.867301   -0.074099
Mo   1.396384   -0.999473   -0.100117
Mo   -1.718869   -0.915823   0.145773
Mo   -3.933920   1.016109   -0.045287
O   2.124988   0.288907   1.024896
O   2.814468   -0.482460   -1.188270
O   -3.074600   -0.299403   1.229858
O   -2.375356   0.344558   -1.080967
O   -0.000545   -0.897978   -1.198853
O   -0.293304   -0.836977   1.221760
O   1.749688   -2.591882   0.323967
O   -2.132792   -2.509141   -0.208237
O   -5.447659   0.523874   -0.672211
O   4.306072   1.241552   1.728562
O   3.456779   2.352316   -0.819722
O   -3.828537   2.648390   0.447776
O   5.355795   0.385644   -0.421474

[Mo₄O₁₃]²⁻ VII 3/4
E = -34056.62
Mo   -3.750691   -0.867300   -0.074099
Mo   -1.396381   0.999471   -0.100119
Mo   1.718869   0.915821   0.145771
Mo   3.933919   -1.016108   -0.045289
O   -2.124991   -0.288909   1.024901
O   -2.814471   0.482460   -1.188269
O   3.074599   0.299401   1.229861
O   2.375359   -0.344559   -1.080969
O   0.000549   0.897981   -1.198849
O   0.293299   0.836981   1.221761
O   -1.749692   2.591881   0.323971
O   2.132788   2.509141   -0.208239
O   5.447659   -0.523868   -0.672209
O   -4.306071   -1.241550   1.728561
O   -3.456781   -2.352320   -0.819719
O   3.828539   -2.648388   0.447781
O   -5.355791   -0.385640   -0.421469

[Mo₄O₁₃]²⁻ VIII 1/2
E = -34056.49
Mo   -2.521087   -1.866509   -0.230795
O   -3.168092   0.024069   -0.478030
O   -0.796458   -1.152729   0.298721
O   -3.328587   -2.653031   1.054624
O   -2.466974   -2.707408   -1.717376
Mo   -2.532976   1.708289   -0.228995
O   -2.704391   2.618855   -1.653326
O   -3.397071   2.475788   1.017457
O   -0.741656   1.583927   0.231249
Mo   0.432708   0.139833   0.829896
O   0.510115   0.203599   2.505753
O   1.973008   0.072233   -0.165094
O   3.987593   -1.428406   0.804675
O   4.111041   -0.086949   -2.093522
Mo   3.932086   0.095519   -0.413053
O   4.349826   1.666252   0.088223
O   5.290311   -1.021146   0.332118

[Mo₄O₁₃]²⁻ VIII 3/4
E = -34056.39
Mo   2.506033   1.873890   -0.209688
O   3.160250   -0.016702   -0.164097
O   0.700709   1.299483   0.349383
O   3.227210   2.856352   0.988911
O   2.522794   2.497363   -1.799023
Mo   2.547438   -1.724817   -0.202310
O   2.796201   -2.435413   -1.725761
O   3.317673   -2.665677   0.985844
O   0.739872   -1.553736   0.148163
Mo   -0.430122   -0.128721   0.780119
O   -0.538413   -0.320114   2.444453
O   -1.953998   -0.055553   -0.253852
O   -3.969076   1.097371   1.125012

O   -4.158782   0.521679   -2.010789
Mo   -3.915212   -0.072396   -0.436356
O   -4.284781   -1.729705   -0.331660
O   -5.277380   0.777884   0.601648

[CH₂O₂] I 1/2
E = -5159.39
C   0.728185   0.000002   -0.000001
O   -0.435662   -0.734371   -0.000002
O   -0.435667   0.734369   -0.000001
H   1.300785   0.000004   -0.927022
H   1.300738   0.000003   0.927051

[CH₂O₂] I 3/4
E = -5157.88
C   0.000000   -0.625618   0.000002
O   -0.940771   0.390179   0.000002
O   0.940770   0.390181   0.000002
H   0.000004   -1.244549   -0.907547
H   0.000004   -1.244620   0.907501

[Mo₂O₈]²⁻ I 1/2
E = -20098.31
Mo   2.018499   0.020559   0.000010
Mo   -1.772003   -0.191211   0.000022
O   -1.737771   1.762541   0.000007
O   0.130168   -0.201327   0.000370
O   -2.199326   -1.063617   -1.407016
O   -2.199870   -1.063276   1.407105
O   2.517966   0.884525   -1.407374
O   2.764806   -1.536531   0.000204
O   -3.088426   1.228461   -0.000363
O   2.518346   0.885149   1.406897

[Mo₂O₈]²⁻ I 3/4
E = -20096.30
Mo   -1.996431   0.001045   0.000034
Mo   1.791174   -0.279718   -0.000223
O   2.124155   1.792060   0.659253
O   -0.109458   0.058002   -0.001240
O   2.338283   -1.067666   1.428127
O   2.341667   -1.064680   -1.428942
O   -2.597565   0.661537   1.480358
O   -2.539161   -1.633473   -0.151493
O   2.126038   1.792660   -0.656672
O   -2.606360   0.924595   -1.328403

[Mo₂O₈]²⁻ II 1/2
E = -20096.19
Mo   1.562489   0.099518   -0.011757
Mo   -1.562433   -0.099530   0.011786
O   -0.153340   -0.854592   -0.880670
O   -2.844504   -0.156489   -1.144965
O   -2.048046   -1.086198   1.320497
O   2.047870   1.086233   -1.320425
O   2.050168   -1.764284   -0.288464
O   -2.050513   1.764123   0.288210
O   2.844591   0.156456   1.144954
O   0.153480   0.854710   0.880710

[Mo₂O₈]²⁻ II 3/4
E = -20096.19
Mo   1.562491   0.099682   -0.011534
Mo   -1.562470   -0.099700   0.011530
O   -0.154955   -0.849159   -0.886347
O   -2.847316   -0.152637   -1.142290
O   -2.045408   -1.091747   1.317177
O   2.045494   1.091646   -1.317240
O   2.050343   -1.763846   -0.293086
O   -2.050655   1.763795   0.293146
O   2.847338   0.152502   1.142333
O   0.155046   0.849383   0.886330

[Mo₂O₈]²⁻ III 1/2
E = -20095.78

Mo   -1.461567   0.001695   0.005065
Mo   1.619393   -0.000716   0.001295
O   0.390326   -0.120015   1.326307
O   2.636573   -1.392921   -0.116816
O   -0.977524   -1.847801   0.316029
O   -0.953662   1.843701   -0.330425
O   -2.526447   0.443920   1.263169
O   2.632286   1.390771   0.159096
O   -2.484448   -0.441600   -1.289243
O   0.454307   0.118808   -1.361507

[Mo₂O₆]²⁻ I 1/2
E = -16003.98
Mo   -1.293334   -0.000009   -0.000023
Mo   1.293292   -0.000007   0.000025
O   0.000139   0.000003   -1.459863
O   0.000016   0.000005   1.459867
O   -2.194556   -1.486643   0.000024
O   -2.194436   1.486692   0.000004
O   2.194483   1.486624   -0.000011
O   2.194578   -1.486597   -0.000029

[Mo₂O₆]²⁻ I 3/4
E = -16003.73
Mo   -1.522558   -0.000056   -0.000003
Mo   1.522557   -0.000045   0.000003
O   0.000010   -0.000367   -1.226491
O   -0.000017   -0.000746   1.226491
O   -2.438950   -1.475411   -0.000074
O   -2.438196   1.475770   0.000078
O   2.437404   1.476228   0.000080
O   2.439754   -1.474939   -0.000083

[HMo₃O₁₀]⁻ I 1/2
E = -26073.63
Mo   2.324310   -0.878157   0.023527
O   1.477853   0.886876   0.152700
O   -1.351960   -2.080932   -0.982659
O   1.017623   -2.030321   -0.040274
O   3.309870   -1.155511   1.378401
Mo   0.001232   1.946197   -0.012236
O   -0.150188   2.953160   1.324862
O   0.119269   2.917941   -1.378808
O   -1.506619   0.831502   -0.163398
Mo   -2.308404   -0.869182   0.008909
O   -3.890788   -0.835321   -0.556609
O   -2.328048   -1.299866   1.632798
O   3.263530   -0.951407   -1.389725
H   -0.404108   -2.240978   -0.666697

[HMo₃O₁₀]⁻ I 3/4
E = -26070.68
Mo   2.101320   -0.698653   -0.034054
O   1.405719   1.036128   0.259581
O   -0.120887   -1.109414   -0.699322
O   2.502627   -1.001699   -1.641625
O   1.413124   -2.213099   0.854703
Mo   -0.243535   1.934088   0.070816
O   -0.249184   3.317144   1.035810
O   -0.478523   2.399616   -1.532151
O   -1.551338   0.793990   0.614446
Mo   -2.040818   -1.063653   0.033701
O   -3.169224   -1.111346   -1.235198
O   -2.341608   -2.115619   1.328017
O   3.552922   -0.688532   0.829655
H   -0.021627   -1.672213   -1.470757

[HMo₃O₁₀]⁻ II 1/2
E = -26073.46
Mo   -0.854565   -1.695387   0.035991
O   1.008523   -1.537973   0.432785
O   -1.836747   -0.104682   0.432464
O   -1.261625   -2.503350   -1.382409
O   -1.367970   -2.712432   1.273708
Mo   1.895895   0.107882   0.035790
O   2.801760   0.158787   -1.380805
O   0.000685   0.000639   -1.266272

Mo -1.041115   1.587695   0.035801
O   0.827648   1.642674   0.432685
O   -1.665156   2.538003   1.275436
O   3.030311   0.171370   1.276214
O   -1.538416   2.346291 -1.381088
H   -0.001119 -0.002575 -2.220164

[HMo₃O₁₀]⁻ II 3/4
E = -26070.61
Mo -2.128973   0.008002   0.166279
O   -0.722686 -1.381929   0.400533
O   -0.713126   1.388871   0.393960
O   -2.726216   0.005588 -1.442776
O   -3.273877   0.014589   1.413064
Mo   1.044052 -1.712059   0.103727
O   1.287883 -2.578369 -1.320948
O   -0.130536 -0.007859 -2.567943
Mo   1.056906   1.705602   0.101446
O   1.866002 -0.006285   0.050621
O   1.701555   2.599945   1.375257
O   1.684843 -2.610833   1.376419
O   1.311780   2.568926 -1.323057
H   -1.108368 -0.006045 -2.442022

[HMo₃O₁₀]⁻ III 1/2
E = -26073.14
Mo -1.637386   0.867104 -0.043176
O   -1.743709 -0.985752   0.338771
O   0.019149   2.224073   0.013915
O   -2.750400   1.323559 -1.215263
O   -2.099899   1.567964   1.428112
Mo -0.016101 -1.840321   0.027725
O   -0.022986 -2.810684 -1.340717
O   0.001257   0.124908 -0.917863
Mo   1.652245   0.838944 -0.042945
O   1.724836 -1.014807   0.342105
O   2.128393   1.534295   1.426608
O   -0.026106 -2.882878   1.350580
O   2.772946   1.272555 -1.216260
H   0.024274   2.813582   0.772750

[HMo₃O₁₀]⁻ III 3/4
E = -26070.57
Mo -1.695541   0.809344 -0.052136
O   -1.810609 -1.012827   0.337742
O   0.000547   2.093405   0.091232
O   -2.729666   1.236740 -1.304828
O   -2.261909   1.569870   1.349299
Mo -0.000744 -1.761674 -0.053712
O   -0.000703 -3.094816 -1.055012
O   0.000168   0.031349 -0.942406
Mo   1.696459   0.807907 -0.051818
O   1.810244 -1.014256   0.337179
O   2.262447   1.567618   1.350344
O   -0.002918 -2.192869   1.864584
O   2.731443   1.235444 -1.303966
H   0.000306   2.708512   0.828643

[HMo₃O₁₀]⁻ IV 1/2
E = -26072.87
Mo   0.108018   1.725603 -0.092436
O   -1.694200   0.930278   0.475817
O   1.697176   0.871369   0.543430
O   0.338834   3.059915 -1.072024
O   -0.247867   2.617342   1.563673
Mo -1.720631 -0.868052 -0.012573
O   -2.488481 -1.191631 -1.473935
O   -0.081801   0.380105 -1.231069
Mo   1.665580 -0.971394   0.052610
O   -0.025776 -1.761495   0.277716
O   2.643418 -1.726760   1.190512
O   -2.685496 -1.580688   1.169701
O   2.376801 -1.276054 -1.437223
H   -0.885473   2.202349   2.147936

[HMo₃O₁₀]⁻ IV 3/4
E = -26070.72

Mo -0.087345 -1.738494 -0.093600
O   1.709101 -1.045899   0.399246
O   -1.816348 -0.863247   0.375125
O   -0.134970 -2.889688 -1.293502
O   -0.198924 -2.656061   1.593376
Mo   1.771128   0.772702   0.001235
O   2.878110   1.104471 -1.218689
O   0.016138   0.121685 -0.994010
Mo -1.662684   0.956620 -0.022670
O   0.111326   1.982184 -0.067825
O   -2.175339   1.719164   1.394370
O   2.352608   1.487834   1.415745
O   -2.717669   1.414921 -1.247170
H   -1.078412 -2.617680   1.978134

[HMo₃O₁₀]⁻ V 1/2
E = -26072.76
Mo -0.834900 -1.636925 -0.015247
O   1.039604 -1.748679   0.239357
O   -2.311918 -0.027466 -0.140309
O   -1.390343 -2.807271 -1.094072
O   -1.410685 -1.962362   1.532283
Mo   1.849923   0.022794 -0.011366
O   2.885163   0.037296 -1.332753
O   -0.092598 -0.001392 -0.905619
Mo -0.876322   1.615410 -0.015089
O   0.995423   1.772213   0.241692
O   -1.460720   1.930537   1.531418
O   2.811594   0.033663   1.371568
O   -1.457560   2.770724 -1.096482
H   -2.689130 -0.031770 -1.025179

[HMo₃O₁₀]⁻ VI 1/2
E = -26072.45
Mo -3.180419 -0.528077 -0.017382
O   -1.197392 -0.024343 -0.096538
O   4.508209   0.144059   0.445948
O   -3.610621 -0.704230   1.624291
O   -3.416523 -1.984480 -0.871053
Mo   0.140071   1.106279   0.103277
O   0.765535   1.685323 -1.525055
O   -0.380946   2.436451   0.951181
O   1.501130   0.321098   0.966048
Mo   3.041525 -0.611782   0.045743
O   2.588135 -0.311306 -1.602530
O   3.107502 -2.268754   0.396796
O   -4.049151   0.750585 -0.741068
H   1.423556   1.055160 -1.912955

[HMo₃O₁₀]⁻ VI 3/4
E = -26070.03
Mo   3.207117 -0.555484 -0.059078
O   1.501570   0.080397 -0.860392
O   -4.412418   0.835708 -0.138061
O   4.535831   0.034316 -0.949882
O   3.236046 -2.259461 -0.005449
Mo   0.144797   1.052591 -0.113102
O   0.426866   0.886285   1.737963
O   0.090342   2.683558 -0.475632
O   -1.603376   0.160145 -0.083700
Mo -3.315703 -0.443904 -0.058303
O   -3.710115 -1.393457   1.531832
O   -3.588167 -1.528841 -1.323357
O   3.162011   0.125266   1.530244
H   1.370418   0.774145   1.971733

[HMo₃O₁₀]⁻ VII 1/2
E = -26072.44
Mo -1.833280   0.000012   0.092130
O   -1.019748 -1.544564   0.566879
O   -1.020148   1.544566   0.567217
O   -2.649429   0.000222 -1.614293
O   -3.235104 -0.001276   1.028248
Mo   0.983099 -1.515606 -0.030294
O   1.386933 -2.605632 -1.243519
O   -0.001258 -0.000035 -0.988749
Mo   0.982799   1.515719 -0.030218
O   2.160303   0.000192   0.071183

O   1.373321   2.240757   1.445942
O   1.373654 -2.240614   1.445904
O   1.386298   2.605857 -1.243410
H   -3.608496 -0.001060 -1.611150

[HMo₃O₁₀]⁻ VII 3/4
E = -26069.72
Mo   0.002877 -1.714224 -0.034149
O   1.732601 -1.008189   0.367831
O   -1.728672 -1.012862   0.369151
O   0.004580 -3.332312 -0.974845
O   0.004111 -1.833501   1.873810
Mo   1.636938   0.898293 -0.022921
O   2.433744   1.367100 -1.426268
O   0.000481 -0.193575 -1.101727
Mo -1.640040   0.893362 -0.022605
O   -0.003019   1.835111   0.313331
O   -2.593316   1.465031   1.244971
O   2.587258   1.471739   1.246068
O   -2.437395   1.357961 -1.427034
H   0.006470 -4.184125 -0.535948

[Mo₂O₇]²⁻ I 1/2
E = -18053.67
O   -2.488477 -0.307114 -1.593904
O   2.485810   0.411516   1.570806
Mo -1.902201   0.000289 -0.000071
O   2.485500 -1.567187 -0.427237
O   -2.487032   1.533541   0.533112
Mo   1.902367   0.000112 -0.000162
O   -2.485041 -1.228550   1.062310
O   2.488339   1.153697 -1.142353
O   0.000027   0.001990 -0.001513

[Mo₂O₇]²⁻ I 3/4
E = -18050.27
O   -2.493636 -0.601296   1.480538
O   2.687517   0.508413 -1.493601
Mo -1.880757   0.034147 -0.004430
O   2.638412   0.761117   1.406762
O   -2.535002 -0.865206 -1.326995
Mo   1.940934   0.059057 -0.005476
O   -2.364631   1.687975 -0.152013
O   1.742110 -1.917107   0.162609
O   0.009299 -0.063218 -0.025295

[CH₃O] I 1/2
E = -3129.96
C   0.570291   0.000067   0.013477
H   1.002236   0.908987   0.453855
H   1.002052 -0.907169   0.457555
H   0.866253 -0.002399 -1.052729
O   -0.786536   0.000022   0.007557

[MoO₄]⁻ I 1/2
E = -10049.16
Mo -0.042668 -0.000040   0.006635
O   0.263474 -0.010610   1.709038
O   1.733952   0.004292 -0.761301
O   -0.887110 -1.409309 -0.499995
O   -0.886308   1.415838 -0.482574

[MoO₄]⁻ I 3/4
E = -10045.67
Mo   0.112685   0.000008 -0.000018
O   -1.232515 -0.000205   1.490858
O   -1.231839   0.000007 -1.491287
O   0.936250   1.512011   0.000366
O   0.936509 -1.511857   0.000157

[MoO₄]⁻ II 1/2
E = -10038.65
Mo   0.152469 -0.137317   0.117060
O   1.489058   1.240992   0.013125
O   2.104472 -0.647811 -0.422013

O   -1.915822   -0.103397   0.471541
O   -2.478171   0.231131   -0.677216

$[MoO_4]^-$ II 3/4
E = -10045.67
Mo   0.329285   0.005681   -0.000130
O   0.933778   1.613309   0.000226
O   1.442836   -1.302538   0.000327
O   -1.438857   -0.444055   -0.000095
O   -2.666505   0.103460   0.000223

$[MoO_4]^-$ II 5/6
E = -10045.27
Mo   0.340625   -0.000023   -0.053931
O   1.114477   1.526981   0.169023
O   1.116128   -1.526280   0.168470
O   -1.661518   0.000589   -0.586469
O   -2.357370   -0.001170   0.532115

$[MoO_4]^-$ II 7/8
E = -10041.83
Mo   0.385024   0.001963   -0.189183
O   1.476405   1.136464   0.470620
O   0.135981   -1.901966   0.326671
O   -1.777550   -0.272209   0.086844
O   -1.856214   1.027405   0.109075

$[HMoO_4]^-$ I 1/2
E = -10067.57
Mo   -0.067334   -0.002623   0.001043
O   -0.571282   0.744804   -1.464183
O   1.857973   -0.303286   -0.066310
O   -0.406073   1.041589   1.326167
O   -0.825464   -1.531019   0.202034
H   2.386774   0.493462   -0.025469

$[HMoO_4]^-$ I 3/4
E = -10064.38
Mo   -0.168443   -0.000090   0.006594
O   -1.056575   -1.468158   0.066179
O   1.431888   0.070612   1.146966
O   1.333590   -0.028461   -1.319406
O   -1.104816   1.437972   -0.009668
H   2.241896   -0.091916   0.650492

$[HMoO_4]^-$ II 1/2
E = -10062.54
Mo   0.341016   0.004939   0.140862
O   0.754202   1.588031   -0.332487
O   1.384508   -1.209217   -0.455328
O   -1.393633   -0.475110   0.011709
O   -2.609285   0.073769   -0.194881
H   0.591015   -0.027233   1.851676

$[HMoO_4]^-$ II 3/4
E = -10062.15
Mo   0.352165   0.000005   0.142570
O   0.948019   1.491264   -0.456600
O   0.948472   -1.491051   -0.456636
O   -1.659649   -0.000078   0.572828
O   -2.176117   -0.000156   -0.637454
H   0.723279   -0.000068   1.834948

$[HMoO_4]^-$ III 1/2
E = -10062.18
Mo   0.375612   -0.010562   0.000014
O   0.577459   1.698238   0.000025
O   1.773125   -0.998970   -0.000105
O   -1.412673   -0.753956   0.000145
O   -2.620770   -0.003595   -0.000096
H   -2.312851   0.909876   -0.000333

$[HMoO_4]^-$ III 3/4

E = -10062.14
Mo   0.379845   -0.001007   -0.070155
O   1.319382   1.408408   0.282991
O   0.936919   -1.624537   0.172176
O   -1.508573   0.199360   -0.653001
O   -2.415647   0.124253   0.503084
H   -2.610128   -0.817561   0.504530

$[HMoO_4]^-$ IV 1/2
E = -10061.52
Mo   0.289444   -0.062587   -0.000069
O   1.502903   1.432749   0.000098
O   0.903968   -1.634930   0.000146
O   -1.474125   -0.010254   -0.000147
O   -2.759310   0.368929   0.000225
H   2.455854   1.376692   0.000310

$[Mo_2O_5]^-$ I 1/2
E = -13949.63
Mo   -1.579842   0.012252   -0.003939
Mo   1.315043   -0.054988   -0.009239
O   -0.015743   -0.984544   0.881524
O   0.031582   1.019231   -0.812920
O   -3.645126   0.018604   -0.004303
O   2.479596   -1.109342   -0.706293
O   2.539886   1.280411   0.711177

$[Mo_2O_5]^-$ I 3/4
E = -13956.34
Mo   -1.545550   -0.000278   -0.000003
Mo   1.303367   -0.000175   -0.000003
O   -0.020036   -0.002567   1.300922
O   -0.020025   -0.000120   -1.300926
O   -3.233568   0.000358   0.000011
O   2.276344   -1.400366   -0.001122
O   2.268748   1.405074   0.001147

$[Mo_2O_5]^-$ I 5/6
E = -13956.32
Mo   -1.577479   -0.186827   -0.000007
Mo   1.290836   0.042174   -0.000001
O   0.016209   -0.162384   1.314761
O   0.016234   -0.162523   -1.314769
O   -3.029737   0.691501   0.000030
O   2.473903   -1.188205   0.000086
O   2.028269   1.581038   -0.000068

$[Mo_2O_5]^-$ I 7/8
E = -13953.21
Mo   1.538307   -0.000131   0.000000
Mo   -1.327113   0.000005   0.000000
O   -0.052920   -0.000311   1.324252
O   -0.052918   -0.000316   -1.324251
O   3.598077   0.000356   -0.000000
O   -2.299820   1.405287   -0.000003
O   -2.301190   -1.404353   0.000001

$[HMo_4O_{12}]^-$ I 1/2
E = -32027.89
Mo   -0.000040   2.194452   0.085745
O   -3.027868   -0.199835   1.453291
O   -0.000118   2.397773   1.752115
O   -0.000005   3.740971   -0.558505
O   1.611445   1.460040   -0.556148
Mo   -2.604009   -0.113789   -0.175318
Mo   2.604080   -0.113736   -0.175313
O   4.032352   -0.002556   -1.055344
O   3.027825   -0.199880   1.453318
O   -1.611443   1.460038   -0.556291
O   -0.000187   -0.295927   0.840286
O   -4.032236   -0.002618   -1.055419
Mo   0.000017   -2.223720   0.022779
O   -1.744432   -1.640286   -0.734453
O   1.744386   -1.640145   -0.734503
O   0.000064   -3.701189   0.797990
H   -0.000240   -0.225757   1.797782

$[HMo_4O_{12}]^-$ I 3/4
E = -32028.19
Mo   0.000037   2.193167   0.075369
O   -3.019745   -0.190136   1.465772
O   -0.000010   2.402213   1.740889
O   0.000031   3.737098   -0.574963
O   1.612010   1.455404   -0.561360
Mo   -2.607854   -0.113058   -0.166403
Mo   2.607857   -0.113105   -0.166163
O   4.042376   -0.001939   -1.036066
O   3.019811   -0.190283   1.465757
O   -1.611895   1.455361   -0.561438
O   -0.000072   -0.299315   0.837904
O   -4.042389   -0.001829   -1.036036
Mo   -0.000041   -2.221125   0.015268
O   -1.755075   -1.642596   -0.724869
O   1.754968   -1.642567   -0.724923
O   -0.000016   -3.718979   0.756292
H   0.000090   -0.226400   1.795174

$[HMo_4O_{12}]^-$ II 1/2
E = -32026.97
Mo   -2.300929   -1.993791   0.076662
O   -2.183424   -2.309488   1.721820
O   -0.847659   -2.610870   -0.818604
O   -3.681266   -2.743149   -0.522918
Mo   -2.096661   1.718925   -0.110011
Mo   1.631236   2.099974   0.118561
O   2.373411   3.586456   0.205856
O   2.682466   0.610206   0.053479
O   -0.271846   2.026249   0.051111
O   -2.905205   2.401333   1.191323
O   -2.676522   2.445814   -1.506281
O   -2.413597   -0.108094   -0.176044
Mo   2.942955   -1.327521   -0.060329
O   3.593733   -1.699942   -1.581789
O   4.013254   -1.826623   1.157055
H   0.110392   -2.481035   -0.442616
O   1.375702   -2.074092   0.149683

$[HMo_4O_{12}]^-$ II 3/4
E = -32028.09
Mo   -0.747626   1.883361   -0.622063
O   0.051920   0.403698   -1.239367
O   0.257215   2.673760   0.508218
O   -1.137685   2.944258   -1.869640
Mo   -2.763214   -0.530938   0.789532
Mo   0.261104   -1.628442   -0.937285
O   0.374635   -2.750876   -2.164282
O   1.886810   -1.353881   0.140920
O   -1.492896   -1.587790   0.036163
O   -4.304211   -0.952973   0.254855
O   -2.721361   -0.640776   2.470609
O   -2.299813   1.222670   0.229821
Mo   3.059321   -0.009441   0.598378
O   4.401321   -0.664934   1.380074
O   3.572935   0.798838   -0.788611
H   1.594240   1.848519   1.416411
O   2.211528   1.175609   1.764242

$[HMo_4O_{12}]^-$ III 1/2
E = -32027.96
Mo   0.103968   -2.358885   0.437780
O   3.726635   -0.025163   0.312782
O   0.116109   -2.145161   2.102063
O   -0.125771   -3.997411   0.153927
O   -1.417696   -1.487425   -0.338439
Mo   2.258568   -0.016279   -0.505900
Mo   -2.502052   0.017570   -0.436689
O   -3.348597   0.022351   -1.886261
O   -3.618463   0.023879   0.815889
O   1.728795   -1.912182   -0.293776
O   2.430782   -0.017510   -2.181219
Mo   0.137538   2.357610   0.437479
O   1.755286   1.887349   -0.294069
O   -0.067824   3.999383   0.152897
O   -1.398055   1.508030   -0.336713

O   0.147853   2.144328   2.102406
H   0.650603 -0.004427   0.347142

[HMo₄O₁₂]⁻ III 3/4

$[HMo_4O_{12}]^-$ III 3/4
E = -32027.86
Mo   0.053405 -2.036671 -0.487116
O   -4.118858 -0.047758 -0.624832
O   0.767331 -0.578352 -1.527226
O   0.345058 -3.520868 -1.208145
O   1.433470 -1.514301   0.777249
Mo -2.857581 -0.155925   0.504844
Mo   2.603679 -0.071929   0.587588
O   3.415719   0.087298   2.047200
O   3.770931 -0.350105 -0.587423
O   -1.628575 -1.713726   0.058127
O   -3.304872 -0.111942   2.139919
Mo   0.024411   2.130745 -0.448779
O   -1.314082   1.049897   0.108594
O   -0.232756   3.684289   0.142729
O   1.669389   1.526682   0.285879
O   0.064213   2.232562 -2.133756
H   0.459832 -0.330690 -2.401109

$[HMo_4O_{12}]^-$ IV 1/2
E = -32027.83
Mo   0.078595   2.454291   0.275643
Mo -2.735657   0.042837 -0.377068
Mo -0.014370 -2.447430   0.272574
Mo   2.660253 -0.045774 -0.240451
O   1.677495 -1.527020   0.282564
O   -0.378165 -2.924919   1.852739
O   -1.201338   1.328164 -0.250935
O   -3.763456   0.016065   1.213921
O   4.173273 -0.072994   0.491359
O   1.742493   1.484347   0.258593
O   -1.245982 -1.299362 -0.317366
O   -3.438136   0.081864 -1.883280
O   0.078404 -3.799880 -0.733665
O   2.888689 -0.063275 -1.904982
O   -0.222474   2.979266   1.853613
O   0.180351   3.780337 -0.763994
H   -3.459732 -0.025533   2.122128

$[HMo_4O_{12}]^-$ IV 3/4
E = -32027.94
Mo   0.376034   2.527600   0.081698
Mo -2.698724   0.401293 -0.164636
Mo -0.356119 -2.431627   0.015686
Mo   2.653440 -0.397941 -0.053847
O   1.414260 -1.738458   0.276449
O   -1.005453 -2.926623   1.499494
O   -1.155917   1.691097 -0.338635
O   -3.016370 -0.411394   1.553438
O   3.887148 -0.467225   1.082090
O   1.788828   1.239724   0.025945
O   -1.349942 -1.111613 -0.647390
O   -4.079442   0.762626 -1.045752
O   -0.326613 -3.741104 -1.047082
O   3.341788 -0.605394 -1.570601
O   0.290172   3.171128   1.636387
O   0.681314   3.778626 -1.004065
H   -2.692665 -1.302801   1.723926

$[HMo_4O_{12}]^-$ V 1/2
E = -32027.84
Mo   0.550231 -2.357670 -0.147041
O   2.575060 -0.140520   1.744019
O   0.967301 -2.635230   1.477479
O   0.766161 -3.770070 -1.042291
O   -1.144809 -1.763910 -0.276291
Mo   2.553800   0.521590 -0.019811
Mo -2.716640 -0.569890   0.064249
O   -3.808170 -0.582080 -1.233051
O   -3.398540 -0.786330   1.600879
O   1.698960 -1.008130 -0.774991
O   4.110210   0.644481 -0.610651
Mo -0.465870   2.482930 -0.062411
O   1.379970   1.969660 -0.147541

O   -0.852060   3.378880 -1.435401
O   -1.441120   0.974120   0.020479
O   -0.711580   3.439960   1.301639
H   2.165010 -0.999010   1.936449

$[HMo_4O_{12}]^-$ V 3/4
E = -32027.84
Mo   0.534087 -2.363722 -0.127620
O   2.582143 -0.140908   1.727570
O   0.965995 -2.629199   1.494993
O   0.747617 -3.783052 -1.012198
O   -1.165225 -1.780883 -0.247364
Mo   2.553667   0.510182 -0.040095
Mo -2.720693 -0.550930   0.048774
O   -3.767895 -0.534553 -1.284647
O   -3.458340 -0.759760   1.560427
O   1.674749 -1.017163 -0.774707
O   4.106949   0.613701 -0.642016
Mo -0.445879   2.480699 -0.045647
O   1.401784   1.976176 -0.163622
O   -0.856554   3.378799 -1.409986
O   -1.415616   0.968101   0.050041
O   -0.673192   3.433193   1.324623
H   2.171008 -0.997211   1.927774

$[HMo_4O_{12}]^-$ VI 1/2
E = -32027.80
Mo   0.284953   2.009839 -0.592472
O   0.554553   2.331439 -2.214886
O   0.266048   3.448978   0.269761
O   -1.602551   1.564805 -0.554629
Mo   2.266879 -0.156521   0.563460
Mo -2.293691   0.214974   0.526633
O   -1.973082   0.509445   2.150608
O   -3.960900   0.265512   0.342331
O   2.211185   1.541264 -0.033802
O   0.323859   0.097627   0.161925
O   2.959318 -0.116240   2.080293
Mo -0.059095 -2.155196 -0.505803
O   1.628468 -1.930678   0.372897
O   -0.258268 -3.814513 -0.368339
O   -1.767373 -1.529735   0.043712
O   0.136669 -1.837557 -2.139876
H   3.496660 -0.592848 -0.536314

$[HMo_4O_{12}]^-$ VI 3/4
E = -32025.91
Mo -2.331886   0.167142   0.482488
O   -2.347660 -0.064795   2.147373
O   -3.875084   0.556640 -0.050986
O   -1.376084   1.860237   0.353528
Mo -0.389233 -2.342588 -0.413042
Mo   0.221637   2.346831 -0.436412
O   0.091747   2.377061 -2.109046
O   0.582944   3.909465   0.056321
O   -2.253423 -1.765299 -0.041574
O   -0.818869 -0.415835 -0.751659
O   -0.311007 -3.445757 -1.647078
Mo   2.559277 -0.387479   0.418561
O   1.440433 -1.779468   0.084541
O   3.921641 -0.485216 -0.563701
O   1.627421   1.220945   0.100232
O   3.066459 -0.426475   2.023534
H   -0.492857 -3.256056   1.021127

$[HMo_4O_{12}]^-$ VII 1/2
E = -32026.44
Mo -3.489760 -1.501506 -0.340532
O   -5.023331 -2.022046 -0.695006
O   -2.296101 -2.773582   0.249886
Mo -2.021696   1.854947 -0.218897
Mo   0.826516 -0.043269   1.033620
O   1.037993   0.267379   2.666421
O   2.390223 -0.052162   0.188639
O   -0.327185   1.273460   0.293707
O   -2.699409   2.814171   0.975449
O   -1.931268   2.766993 -1.619369
O   -3.085488   0.345756 -0.496099

Mo   4.265877 -0.269305 -0.535014
O   4.703648   1.196231 -1.295397
O   5.312317 -0.612057   0.770181
H   -1.378062 -2.574922   0.528315
O   4.261613 -1.555877 -1.657413
O   0.029326 -1.540955   0.872282

$[HMo_4O_{12}]^-$ VII 3/4
E = -32026.94
Mo -2.491610 -1.836439 -0.306961
O   -2.957018 -2.790378 -1.594132
O   -2.623210 -2.664422   1.422486
Mo -2.335592   1.830148 -0.207904
Mo   0.491986   0.100186   0.726688
O   0.605269   0.013717   2.391106
O   2.097813   0.091143 -0.011570
O   -0.481231   1.608989   0.265690
O   -3.187184   2.515263   1.071039
O   -2.522784   2.793646 -1.571560
O   -2.866525   0.129235 -0.511806
Mo   4.103021 -0.058349 -0.330479
O   4.669799   1.470629 -0.834399
O   4.837833 -0.526164   1.137065
H   -1.814004 -2.839247   1.909016
O   4.361638 -1.234687 -1.538938
O   -0.488622 -1.238687   0.159334

$[MoO_3]^-$ I 1/2
E = -7992.37
Mo   0.163696 -0.091196 -0.000117
O   -1.293047   1.185225   0.000090
O   -1.814632 -0.787418   0.000236
O   2.248275   0.080974   0.000287

$[MoO_3]^-$ I 3/4
E = -8002.08
Mo -0.000092 -0.000033 -0.116213
O   1.092680   1.303483   0.203317
O   0.582998 -1.597556   0.203372
O   -1.675196   0.294245   0.203428

$[MoO_3]^-$ I 5/6
E = -7999.13
Mo -0.060565   0.000218   0.000033
O   -0.819135   1.559214 -0.000062
O   1.976997 -0.011809 -0.000047
O   -0.839897 -1.548550 -0.000062

$[MoO_3]^-$ I 7/8
E = -7995.59
Mo -0.000024   0.005276   0.000240
O   0.000488 -1.730501 -0.000413
O   -1.855852   0.851090 -0.000423
O   1.855488   0.851713 -0.000423

$[Mo_3O_{10}]^-$ I 1/2
E = -26054.87
Mo   1.931572   0.000551   0.038577
O   0.962546   1.584429   0.394374
O   0.963594 -1.584003   0.394159
O   2.622070   0.000897 -1.492986
O   3.211971   0.000919   1.125220
Mo -0.926925   1.611618 -0.013891
O   -1.325455   2.845439 -1.078174
O   -0.317142 -0.000133 -1.125851
Mo -0.925922 -1.612161 -0.013907
O   -2.128473 -0.000622 -0.201199
O   -1.538545 -2.023327   1.502989
O   -1.539970   2.022679   1.502878
O   -1.323898 -2.846316 -1.077994

$[Mo_3O_{10}]^-$ I 3/4
E = -26052.11
Mo -0.000432   1.728459 -0.068871
O   -1.787919   0.990562   0.366928

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O   1.787376   0.991233   0.367183
O   -0.000290   3.024802   -1.113264
O   -0.001648   2.288609   1.814726
Mo -1.744215   -0.835701   -0.020211
O   -2.770432   -1.232133   -1.288184
O   0.000027   -0.083379   -1.018324
Mo   1.744717   -0.834926   -0.020054
O   0.000480   -1.891195   0.012344
O   2.362858   -1.580207   1.359832
O   -2.361982   -1.581180   1.359750
O   2.771159   -1.230724   -1.288027

[Mo₃O₁₀]⁻ II 1/2
E = -26054.55
Mo   2.049985   -0.000629   -0.198273
O   0.588564   1.386482   -0.102415
O   0.588027   -1.386987   -0.099366
O   2.719101   -0.002283   -1.721573
O   3.164397   0.721992   1.163198
Mo -1.184160   1.722315   0.026686
O   -1.550164   2.555631   1.446945
O   -1.734260   2.622423   -1.289789
O   -1.975991   0.000505   0.042110
Mo -1.185110   -1.721740   0.026989
O   -1.733964   -2.620733   -1.290755
O   -1.553617   -2.555724   1.446216
O   3.164150   -0.721020   1.164567

[Mo₃O₁₀]⁻ II 3/4
E = -26052.58
Mo   1.962570   -0.590684   -0.240782
O   0.889060   1.178214   -0.319091
O   0.220959   -1.591149   -0.140859
O   2.966393   -1.212077   -1.429942
O   3.481276   0.402173   0.929512
Mo -0.735295   1.910319   -0.094718
O   -0.842438   2.729144   1.379655
O   -1.101509   2.984500   -1.347251
O   -1.913381   0.413324   -0.102178
Mo -1.570546   -1.432199   0.091153
O   -2.418327   -2.333570   -1.056516
O   -2.022333   -1.953680   1.632387
O   2.542468   -0.025917   1.737103

[Mo₃O₁₀]⁻ III 1/2
E = -26054.41
Mo   3.209210   -0.659796   -0.009221
O   1.491835   0.378248   -0.098984
O   -3.420623   -1.346093   -1.525739
O   2.922724   -2.185126   -0.726360
O   3.647563   -0.842555   1.632162
Mo   0.009000   1.371142   0.011842
O   0.008422   2.301504   1.411515
O   -0.117137   2.410852   -1.302833
O   -1.534986   0.217385   0.037816
Mo -3.162735   -0.628718   -0.027181
O   -4.416460   0.442499   0.306935
O   -3.291367   -2.004433   1.257053
O   4.418782   0.193922   -0.862626

[Mo₃O₁₀]⁻ III 3/4
E = -26051.82
Mo -2.605127   -0.855159   -0.000008
O   -1.416408   0.498195   -0.000193
O   2.474029   -1.804393   1.393853
O   -2.472043   -1.805454   1.392897
O   -2.474111   -1.804180   -1.394007
Mo -0.000009   1.895138   -0.000003
O   0.000126   2.799821   -1.435371
O   -0.000049   2.799832   1.435360
O   1.416241   0.498039   0.000183
Mo   2.605130   -0.855170   -0.000006
O   4.382721   -0.173306   -0.000965
O   2.472311   -1.805305   -1.393052
O   -4.382785   -0.173494   0.001387

[CH₃Mo₂O₇]⁻ I 1/2

E = -19133.78
O   2.736860   -1.417600   0.105661
O   2.570092   0.819029   -1.465066
Mo   2.078345   0.116338   -0.025430
O   0.214549   0.059581   0.076284
O   -2.604161   1.058444   -0.561204
O   -2.146512   -0.833847   1.541145
Mo -1.623307   -0.391579   0.015702
O   2.806130   1.154934   1.362662
O   -1.866610   -1.681588   -1.019390
C   -3.561981   1.953765   -0.032741
H   -4.480925   1.868052   -0.613348
H   -3.173571   2.969251   -0.118256
H   -3.767997   1.728583   1.015887

[CH₃Mo₂O₇]⁻ I 3/4
E = -19130.52
O   2.448960   -0.843308   1.451360
O   2.731466   -0.842692   -1.297657
Mo   1.886093   -0.200853   0.002865
O   0.078973   -0.394318   -0.188686
O   -1.392575   1.673531   -0.730904
O   -2.443352   -0.253858   1.591044
Mo -1.881350   -0.332566   0.004150
O   2.212227   1.679601   0.076281
O   -2.748112   -1.335765   -1.030891
C   -0.880836   2.683605   0.022628
H   -1.020628   3.643308   -0.485098
H   0.228549   2.521359   0.029281
H   -1.222802   2.671794   1.061070

[CH₂O] I 1/2
E = -3115.29
C   0.000000   -0.524279   0.000000
O   -0.000000   0.670794   -0.000000
H   0.939280   -1.110338   0.000000
H   -0.939280   -1.110338   0.000000

[CH₂O] I 3/4
E = -3112.38
C   -0.048886   0.593453   -0.000000
O   -0.048886   -0.704649   0.000000
H   0.342205   1.038239   0.924402
H   0.342205   1.038239   -0.924402

[HMo₄O₁₃]⁻ I 1/2
E = -34076.89
Mo   2.393277   -0.050816   -0.059357
O   0.058685   2.740744   1.580854
O   2.741096   -0.058185   1.580889
O   3.874633   -0.082220   -0.844883
O   1.605460   -1.674830   -0.631400
Mo   0.050886   2.392434   -0.059256
Mo -0.050761   -2.392481   -0.059254
O   -0.082282   -3.873659   -0.845129
O   -0.058127   -2.740722   1.580881
O   1.675113   1.604992   -0.631636
O   0.000085   0.000244   0.628561
O   0.082211   3.873624   -0.845118
Mo -2.393444   0.050821   -0.059307
O   -1.605436   1.674780   -0.631145
O   -3.874592   0.082281   -0.845224
O   -1.675048   -1.604998   -0.631288
O   -2.741676   0.058170   1.580839
H   0.000842   0.000029   1.591701

[HMo₄O₁₃]⁻ I 3/4
E = -34073.75
Mo -0.002605   -2.318594   -0.075196
O   2.861126   -0.132291   1.451632
O   -0.004159   -2.452246   1.827898
O   -0.004963   -4.003340   -0.133730
O   -1.616036   -1.586265   -0.804480
Mo   2.432053   -0.021036   -0.166093
Mo -2.432244   -0.015535   -0.165998
O   -3.871190   0.050481   -1.025371
O   -2.860326   -0.126003   1.452044

O   1.612621   -1.590468   -0.804202
O   0.000416   -0.119417   0.694903
O   3.870570   0.041547   -1.026438
Mo   0.002854   2.381195   0.060824
O   1.612770   1.639961   -0.580163
O   0.004376   3.926897   -0.587109
O   -1.608614   1.643177   -0.580107
O   0.003121   2.580434   1.727008
H   -0.000126   -0.072988   1.656411

[HMo₄O₁₃]⁻ II 1/2
E = -34076.67
Mo -0.037860   2.163649   -0.212015
O   -3.593292   -0.402035   1.701816
O   -0.326319   3.757456   0.216122
O   -0.031254   2.141288   -1.886486
O   1.701732   1.905031   0.465151
Mo -2.782047   -0.105836   0.258067
Mo   2.508463   0.173294   0.283666
O   3.698217   0.416553   -0.874020
O   3.225645   -0.082425   1.786286
O   -1.633823   1.336366   0.464427
O   0.582591   -0.194370   -0.182192
O   -3.926935   0.190585   -0.937503
Mo   0.123812   -2.088071   -0.285002
O   -1.728349   -1.549913   -0.207917
O   0.230925   -2.891115   -1.752135
O   2.247584   -1.864858   -0.068841
O   0.201063   -3.202360   0.977917
H   2.698297   -2.489088   0.500926

[HMo₄O₁₃]⁻ II 3/4
E = -34074.19
Mo -0.074656   2.187801   0.314069
O   3.357101   -0.207543   -1.905621
O   0.163290   3.775849   -0.163481
O   -0.137463   2.236540   1.985996
O   -1.789583   1.854133   -0.435842
Mo   2.723318   -0.047621   -0.354329
Mo -2.437547   0.065531   -0.406674
O   -3.802153   0.099347   0.569521
O   -2.885875   -0.227994   -2.004687
O   1.556543   1.402778   -0.298238
O   -0.611932   -0.156469   0.370954
O   4.002092   0.173541   0.716825
Mo -0.088313   -2.028045   0.501397
O   1.765646   -1.549349   0.109660
O   -0.106578   -2.703870   2.017021
O   -2.045215   -2.034808   -0.143618
O   0.181105   -3.266698   -0.999211
H   -2.333491   -2.625647   -0.841664

[HMo₄O₁₃]⁻ III 1/2
E = -34076.53
Mo -0.211296   2.005208   0.492970
Mo -2.839264   -0.099590   -0.461850
Mo   0.457619   -2.028819   0.536200
Mo   2.839300   0.223229   -0.559841
O   2.181401   -1.454415   -0.102310
O   0.586031   -2.690748   2.083299
O   -1.833663   1.527054   -0.296447
O   -4.217142   0.447910   -1.243138
O   4.162495   0.620004   0.394721
O   1.433341   1.400509   -0.283258
O   -0.513962   -0.563909   0.512907
O   -2.239764   -1.434271   -1.624383
O   -3.356897   -0.709062   1.009228
O   -0.271645   -3.152140   -0.524924
O   3.328963   0.250139   -2.166112
O   -0.216122   1.850018   2.157378
O   -0.130652   3.652595   0.210173
H   -1.646126   -2.150669   -1.331176

[HMo₄O₁₃]⁻ III 3/4
E = -34073.38
Mo   0.212367   -2.174901   0.387160
Mo   2.826462   0.105693   -0.343425
Mo -0.480877   2.347258   0.384630

Mo -2.748099 -0.303630 -0.461531
O   -1.957528   1.349236 -0.480930
O   -0.875439   3.185190   1.796858
O   1.829454 -1.497606 -0.310330
O   4.409941 -0.322149 -0.696726
O   -3.759680 -0.455895   0.871427
O   -1.412712 -1.630366 -0.410101
O   0.088862   0.423133   0.618398
O   2.270096   1.263130 -1.675225
O   2.823177   0.861614   1.156359
O   0.588895   3.129419 -0.713852
O   -3.677318 -0.496787 -1.850379
O   0.171316 -2.129203   2.057326
O   0.288483 -3.799368 -0.006033
H   1.685797   2.031626 -1.461383

[HMo₄O₁₃]⁻ IV 1/2
E = -34076.46
Mo   0.343039 -1.955347 -0.857073
O   1.378818   0.414192   2.013420
O   0.356320 -3.614903 -0.639029
O   0.219875 -1.741466 -2.511600
O   -1.252460 -1.453243   0.076563
Mo   2.462385   0.187362   0.750920
Mo -2.478118 -0.304181   0.868230
O   -4.026209 -0.631726   0.307530
O   -2.496386 -0.473271   2.536524
O   2.005788 -1.434296 -0.171262
O   0.546971   0.573656 -0.765416
O   3.922820 -0.259630   1.440203
Mo -0.536726   1.968471 -0.778215
O   2.926242   1.837282 -0.016925
O   -1.127726   2.273792 -2.329215
O   -1.972281   1.424074   0.378133
O   0.339334   3.317389 -0.218049
H   2.226850   2.500339 -0.129207

[HMo₄O₁₃]⁻ IV 3/4
E = -34073.16
Mo -1.175779   1.503740 -1.151773
O   -0.318298 -0.318389   1.727182
O   -1.940526   2.989866 -1.002636
O   -1.046461   1.271003 -2.805697
O   0.442283   1.750551 -0.256196
Mo -1.738434 -0.938168   0.990673
Mo   1.721211   1.167608   1.056552
O   2.991107   2.224517   0.844379
O   1.260176   1.099906   2.901301
O   -2.390925   0.363680 -0.189129
O   -0.317319 -0.849972 -1.035864
O   -2.929893 -0.978524   2.161852
Mo   1.296437 -1.574128 -0.968799
O   -1.591825 -2.752264   0.588557
O   2.048751 -1.634424 -2.481502
O   2.251166 -0.464598   0.236778
O   1.092515 -3.151544 -0.330486
H   -0.750294 -3.078633   0.212296

[HMo₄O₁₃]⁻ V 1/2
E = -34076.40
Mo   2.219446 -0.093526   0.602196
O   0.041104   3.734637 -0.133527
O   2.352143 -0.266969   2.250045
O   3.971213   0.134333 -0.050772
O   1.918827 -1.761818 -0.078493
Mo   0.089415   2.096128 -0.494766
Mo -0.012210 -2.024976 -0.571419
O   0.165680 -2.410483 -2.192860
O   -0.217557 -3.438189   0.308977
O   1.823198   1.723856   0.268388
O   0.281217 -0.075637   0.115177
O   0.263628   1.974818 -2.158189
Mo -2.510470   0.041545   0.472852
O   -1.726339   1.645482 -0.078398
O   -4.169536   0.144507   0.220053
O   -1.845597 -1.368737 -0.524417
O   -2.254882 -0.208766   2.117461
H   4.155636   0.578576 -0.879805

[HMo₄O₁₃]⁻ V 3/4
E = -34074.28
Mo   0.251533 -2.223223 -0.550280
O   -4.058940 -0.114218 -0.432910
O   0.213426 -3.161792 -1.919050
O   0.120065 -3.263202   1.050590
O   2.169017 -1.745238 -0.147710
Mo -2.585731 -0.194300   0.376470
Mo   2.360012   0.137960   0.458810
O   2.445400 -0.136870   2.112700
O   3.892655   0.504248 -0.115310
O   -1.617082 -1.607398 -0.224310
O   0.767028 -0.331837 -0.765000
O   -2.895962 -0.381298   2.022450
Mo -0.110860   2.445960 -0.367320
O   -1.586658   1.380712   0.111690
O   -0.403537   4.008102   0.170310
O   1.453206   1.864238   0.425970
O   0.043091   2.523229 -2.036170
H   -0.761795 -3.298105   1.431340

[HMo₄O₁₃]⁻ VI 1/2
E = -34076.28
Mo   2.581126 -1.771554 -0.170354
O   1.027775 -2.482724   0.194118
O   3.779096 -2.388493   0.857971
O   3.011066 -2.065502 -1.783256
Mo   1.864831   1.888661   0.254832
Mo -1.851335   1.875519 -0.270977
O   -2.270018   2.581990 -1.731690
O   -2.630652   2.728297   0.942009
O   -2.386644   0.101954 -0.257004
O   -0.007577   1.962604 -0.024378
O   2.641136   2.864892 -0.867743
O   2.232079   2.477418   1.782466
O   2.408767   0.164519   0.089627
Mo -2.581622 -1.745699   0.181885
O   -2.447509 -1.914888   1.846563
O   -4.093126 -2.300382 -0.299927
O   -1.293245 -2.688314 -0.681940
H   -0.315204 -2.701882 -0.340732

[HMo₄O₁₃]⁻ VI 3/4
E = -34073.12
Mo   2.664635 -1.572677 -0.402361
O   1.182327 -2.416199 -0.074675
O   3.870281 -1.593548   1.147315
O   3.606993 -2.019233 -1.735408
Mo   1.665204   1.880933   0.444320
Mo -1.934232   1.764449 -0.493596
O   -2.350972   2.261147 -2.040209
O   -2.834488   2.671456   0.591481
O   -2.323122 -0.047250 -0.290558
O   -0.128283   2.037234 -0.213437
O   2.567079   3.232988   0.006891
O   1.617670   1.768889   2.121727
O   2.357657   0.383672 -0.288569
Mo -2.434150 -1.835902   0.330202
O   -2.076469 -1.861247   1.968464
O   -3.988605 -2.432046   0.109051
O   -1.263320 -2.876642 -0.614190
H   -0.275222 -2.825629 -0.402765

[HMo₄O₁₃]⁻ VII 1/2
E = -34076.22
Mo   0.165102   1.864782 -0.178049
O   -3.910022   0.000185   1.626477
O   0.119220   3.110054   0.946402
O   -0.275774   2.428186 -1.699070
O   2.059685   1.594836 -0.441684
Mo -2.978644 -0.000224   0.227522
Mo   0.000626   0.000867   0.145137
O   4.423544   0.000684 -0.467729
O   2.994983   0.000867   1.821487
O   -1.945741   1.405653   0.187594
O   0.218652 -0.000324   0.358568
O   -4.126901 -0.000134 -1.267019
Mo   0.166829 -1.865303 -0.178113

O   -1.946401 -1.406080   0.188629
O   -0.275873 -2.428377 -1.698733
O   2.060256 -1.594643 -0.441624
O   0.119749 -3.110433   0.946461
H   -5.079441   0.001217 -1.170948

[HMo₄O₁₃]⁻ VII 3/4
E = -34074.16
Mo -0.806949   1.973852 -0.426327
O   2.778415 -0.240014   2.139144
O   -1.254770   3.135151 -1.557307
O   0.318852   2.646530   0.659712
O   -2.305621   1.274805   0.483315
Mo   3.186805   0.012618   0.528241
Mo -2.873497 -0.529297   0.693856
O   -2.909039 -0.942624   2.323310
O   -4.412489 -0.756951   0.053054
O   4.800065 -0.403847   0.286684
O   -0.092424   0.527628 -1.219934
O   2.975567   1.825119   0.115571
Mo   0.232840 -1.456952 -0.942999
O   2.086932 -0.995743 -0.558903
O   0.532305 -2.581261   0.598904
O   -1.633474 -1.488544 -0.229803
O   0.212222 -2.287296 -2.375317
H   2.181294   2.287085   0.436225

[HMo₄O₁₃]⁻ VIII 1/2
E = -34075.72
Mo   3.309397 -1.415227   0.285991
O   3.650486 -1.510800 -1.351194
O   1.878204 -2.501294   0.726612
O   4.655861 -1.917127   1.145598
Mo   1.891195   1.917234   0.285705
Mo -0.878447 -0.001375 -1.082086
O   -0.047676 -1.476396 -0.879763
O   -1.081429   0.270948 -2.721707
O   -2.449374 -0.029320 -0.254013
O   0.234568   1.359239 -0.347180
O   2.691053   2.815846 -0.877245
O   1.718120   2.879881   1.642154
O   2.874623   0.377749   0.706386
Mo -4.302583 -0.290195   0.515142
O   -4.539581   0.910346   1.706654
O   -5.444550 -0.129560 -0.743594
O   -4.378353 -1.851399   1.202597
H   1.082802 -2.423208   0.157961

[HMo₄O₁₃]⁻ VIII 3/4
E = -34073.43
Mo -3.027686   0.298415 -0.789873
O   -3.987703 -0.754431   0.099163
O   -2.016453 -0.626316 -2.013450
O   -4.021810   1.380817 -1.599994
Mo -0.386878   0.949334   1.524400
Mo   0.679907 -2.158360   0.310341
O   -0.473257 -2.461663 -0.939263
O   1.383018 -3.497387   1.058350
O   1.971228 -0.797806 -0.209717
O   -0.096540 -0.824900   1.614183
O   -0.665368   1.579551   3.054204
O   0.981080   1.704309   0.835928
O   -1.879353   1.203474   0.407163
Mo   2.678467   0.790107 -0.833857
O   3.489238   1.985098   0.396519
O   4.117536   0.387451 -1.629196
O   1.677889   1.531575 -1.965154
H   -1.476047 -1.417042 -1.732376

[HMo₄O₁₃]⁻ IX 1/2
E = -34075.72
Mo -0.899618 -0.039177   0.923653
Mo -4.385072 -0.294129 -0.472736
Mo   1.923425   1.937383 -0.209446
Mo   3.381104 -1.410765 -0.285988
O   3.064488   0.453055 -0.258733
O   1.944633   2.752949 -1.669286
O   -2.422724 -0.184089   0.021416

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```
O   -5.110641  -1.499366   0.495011
O    2.579594  -2.175798   1.196374
O    5.023057  -1.735259  -0.237350
O    0.214814   1.308959   0.166557
O   -5.090696   1.231569  -0.170895
O   -4.498380  -0.694477  -2.128222
O    2.436802   3.001473   0.975881
O    2.752178  -2.043221  -1.702744
O   -0.008450  -1.492535   0.865584
O   -1.188572   0.336336   2.529640
H    1.597926  -2.195913   1.203829
```

$[HMo_4O_{13}]^-$ IX 3/4
E = -34073.48
```
Mo   0.858316   1.763327  -1.275840
Mo   2.122110  -0.937136   0.727502
Mo  -1.520258   1.106125   1.353479
Mo  -1.470991  -1.709153  -0.783919
O   -1.964577  -0.621531   0.628485
O   -1.348291   1.139907   3.020837
O    2.037619   0.350804  -0.572262
O    2.505558  -2.709420   0.165516
O   -1.900279  -0.873350  -2.363895
O   -2.263537  -3.184296  -0.717867
O   -0.071673   1.774037   0.481772
O    3.546002  -0.679899   1.597090
O    0.823595  -0.811695   1.809004
O   -2.840229   2.080637   0.989311
O    0.214471  -1.940490  -0.711586
O   -0.224165   1.139068  -2.474482
O    1.704995   3.177322  -1.640266
H   -1.301364  -0.101621  -2.584595
```

$[HMo_4O_{13}]^-$ X 1/2
E = -34075.57
```
Mo  -0.600486  -0.629116  -0.815190
Mo  -3.859443   0.142422   0.385989
Mo   1.895992   1.959328  -0.012117
Mo   2.837690  -1.301114   0.421806
O    3.098258   0.580444   0.440727
O    2.448247   2.826407  -1.339842
O   -2.153870   0.268000  -0.612773
O   -5.207768   0.384303  -0.620011
O    3.960149  -2.069145  -0.566601
O    1.151392  -1.434271  -0.238785
O    0.350123   1.042669  -0.397245
O   -3.718684  -1.511471   0.905341
O   -3.897729   1.211652   1.706464
O    1.689973   3.026661   1.265886
O    2.962192  -1.941201   1.971058
O   -0.508220  -0.908671  -2.444764
O   -1.332162  -2.115435  -0.019154
H   -2.232828  -2.083428   0.417098
```

$[HMo_4O_{13}]^-$ X 3/4
E = -34073.93
```
Mo   0.315671  -1.363882   0.522331
Mo   3.595581   0.113399  -0.282709
Mo  -1.068810   2.004218   0.180901
Mo  -3.160699  -0.641402  -0.362189
O   -2.743810   1.201238  -0.178159
O   -1.114130   2.809758   1.657691
O    2.216231  -1.029112  -0.027559
O    3.738460   1.234869   0.970961
O   -4.504619  -1.062512   0.563101
O   -1.661289  -1.464542   0.228441
O    0.098461   0.602528   0.215481
O    5.232751  -0.848181  -0.332719
O    3.487071   0.898839  -1.775089
O   -0.666760   3.108358  -1.022839
O   -3.473249  -1.005552  -1.977899
O    0.457521  -1.769832   2.117371
O    0.437210  -2.883132  -0.629709
H    1.335231  -3.059782  -0.922699
```

$[HMo_4O_{13}]^-$ XI 1/2
E = -34075.48
```
Mo  -4.136154   0.088309  -0.149872
```

```
O   -4.360973  -0.080849  -1.830244
O   -4.833978   1.541755   0.417641
O   -4.742144  -1.263979   0.690625
Mo  -0.465553  -0.945209   0.273987
Mo   3.162963  -1.123153  -0.184414
O    3.882852  -1.711569  -1.581298
O    3.962707  -1.781157   1.135770
O    3.208090   0.761045  -0.127720
O    1.363539  -1.491794  -0.144959
O   -1.264009  -1.917514  -0.822799
O   -0.676483  -1.602575   1.788457
O   -2.125927   0.323601   0.274705
Mo   1.787417   2.008023   0.035694
O    0.318704   0.945632   0.244030
O    2.005074   3.015757   1.361744
O    1.665585   2.962448  -1.340811
H   -1.868550   1.218839   0.512282
```

$[HMo_4O_{13}]^-$ XI 3/4
E = -34073.14
```
Mo   4.150066   0.051253   0.276840
O    4.471157  -0.315728   1.908812
O    4.972939   1.471989  -0.194713
O    4.516948  -1.263242  -0.743885
Mo   0.491568  -0.538399  -0.657174
Mo  -2.911803  -1.388219   0.276511
O   -3.047821  -2.050583   1.812690
O   -4.002901  -2.165168  -0.735562
O   -3.230075   0.478332   0.329014
O   -1.200047  -1.496005  -0.369206
O    1.118088  -1.893139   0.576816
O    0.833308  -0.780575  -2.252308
O    2.139615   0.492821   0.035909
Mo  -2.070101   1.944356   0.050808
O   -0.428287   1.183058  -0.251199
O   -2.570161   2.856001  -1.267778
O   -2.024072   2.944800   1.399103
H    1.901809   1.401884   0.245113
```

$[HMo_4O_{13}]^-$ XII 1/2
E = -34075.16
```
Mo   3.994251  -1.048712  -0.116624
O    2.159675  -1.051991  -0.046070
O    4.751908  -1.880701   1.143489
O    4.643384  -1.525152  -1.601794
Mo   1.912268   1.001330   0.243292
Mo  -1.912267   1.001128  -0.243405
O   -1.718032   1.614067  -1.785690
O   -1.659161   2.224939   0.870450
O   -3.863055   0.798089  -0.124695
O    0.000046   0.231559  -0.000540
O    1.658385   2.225726  -0.869716
O    1.717931   1.613373   1.785888
O    3.863160   0.798175   0.124682
Mo  -3.994174  -1.048758   0.116555
O   -4.641742  -1.524234   1.602694
O   -4.753353  -1.881234  -1.142332
O   -2.159556  -1.052190   0.044607
H   -0.000032  -0.732910  -0.000198
```

$[HMo_4O_{13}]^-$ XII 3/4
E = -34072.69
```
Mo   4.118526  -0.906076  -0.153591
O    2.298824  -1.091987  -0.011267
O    5.009463  -1.712742   1.033296
O    4.740018  -1.254461  -1.685120
Mo   1.874005   0.919853   0.387545
Mo  -1.830938   0.688764  -0.527181
O   -1.703371   1.104275  -2.123508
O   -1.531840   2.151914   0.725114
O   -3.702685   0.979092   0.075237
O    0.043107   0.016683   0.005156
O    1.457320   2.228550  -0.573710
O    1.665936   1.332199   1.991557
O    3.820899   0.911089   0.177102
Mo  -4.137384  -0.826178   0.234638
O   -4.609317  -1.306152   1.783719
O   -5.259897  -1.369438  -0.905131
O   -2.372453  -1.222050  -0.187459
```

```
H    0.136102  -0.943023   0.004078
```

$[HMo_4O_{13}]^-$ XIII 1/2
E = -34075.48
```
Mo  -0.465557  -0.945351  -0.274223
Mo  -4.136009   0.088370   0.150013
Mo   1.787366   2.008025  -0.035741
Mo   3.162839  -1.123105   0.184609
O    3.208061   0.761065   0.128066
O    2.005228   3.015644  -1.361842
O   -2.125775   0.323529  -0.274842
O   -4.742502  -1.263423  -0.690905
O    3.962616  -1.781025  -1.135609
O    1.363410  -1.491930   0.145055
O    0.318787   0.945552  -0.244235
O   -4.360602  -0.081367   1.830360
O   -4.833519   1.542215  -0.416849
O    1.665291   2.962555   1.340674
O    3.882743  -1.711561   1.581463
O   -0.676217  -1.602373  -1.788870
O   -1.264331  -1.917913   0.822120
H   -1.868369   1.218777  -0.512332
```

$[HMo_4O_{13}]^-$ XIII 3/4
E = -34073.14
```
Mo  -0.491568  -0.538285  -0.657225
Mo  -4.150091   0.051226   0.276834
Mo   2.070204   1.944309   0.050849
Mo   2.911742  -1.388299   0.276491
O    3.230125   0.478241   0.329018
O    2.570308   2.855979  -1.267702
O   -2.139664   0.492892   0.035788
O   -4.517320  -1.262463  -0.744802
O    4.002914  -2.165297  -0.735469
O    1.200038  -1.495962  -0.369347
O    0.428361   1.183078  -0.251199
O   -4.470851  -0.316986   1.908593
O   -4.972906   1.472431  -0.193458
O    2.024192   2.944710   1.399175
O    3.047562  -2.050684   1.812677
O   -0.833255  -0.780441  -2.252372
O   -1.118270  -1.892740   0.576976
H   -1.901950   1.401952   0.245109
```

$[HMoO_3]^-$ I 1/2
E = -8017.96
```
Mo   0.053131  -0.003588   0.000218
O    1.103621  -1.358560  -0.000397
O    0.788059   1.547935  -0.000389
O   -1.866186  -0.236913  -0.000335
H   -2.435453   0.530999  -0.000191
```

$[HMoO_3]^-$ I 3/4
E = -8017.59
```
Mo   0.082986  -0.000008  -0.006623
O    0.892491  -1.533738   0.020024
O    0.893337   1.533326   0.020003
O   -1.922978   0.000356  -0.097604
H   -2.388202   0.000807   0.738783
```

$[HMo_2O_7]^-$ I 1/2
E = -18065.11
```
Mo   1.883743   0.017323   0.027410
Mo  -1.854208  -0.054323  -0.017964
O    0.020551   0.132479  -0.106838
O   -2.666456   1.618731  -0.181078
O   -2.378940  -1.038071  -1.259346
O    2.350223  -1.480018   0.611580
O    2.712703   0.246344  -1.642995
O   -2.279865  -0.768952   1.428729
O    2.472274   1.219678   1.034496
H   -3.084407   2.112459   0.526879
```

$[HMo_2O_7]^-$ I 3/4
E = -18059.67
```
Mo   1.869633  -0.003020   0.015650
```

Mo -1.882085   0.006398 -0.016879
O   0.001962   0.000006 -0.169915
O   -2.194297   1.593420   0.812353
O   -2.659576   0.068678 -1.480166
O   2.392301 -1.455747   0.661214
O   2.729173   0.214132 -1.640586
O   -2.180256 -1.693529   0.668540
O   2.365282   1.256886   0.999354
H   -3.113718 -0.012636   1.245279

[HMo₂O₇] II 1/2
E = -18064.41
O   -2.622508   1.271592 -0.046005
O   -2.455019 -1.454785   0.151476
Mo -1.580167 -0.033286   0.033146
O   -0.213000 -0.038662 -1.234754
O   2.645625 -0.638024   1.080990
O   1.874073   1.641425 -0.285849
Mo   1.325678   0.112548   0.010267
O   -0.118997   0.132401   1.245688
O   1.873197 -1.134588 -1.297861
H   2.821549 -1.563861   1.267199

[HMo₂O₇] II 3/4
E = -18062.20
O   -2.612343   1.249508 -0.066784
O   -2.445661 -1.493659   0.034569
Mo -1.571475 -0.062516 -0.006760
O   -0.162927 -0.030278 -1.268352
O   2.415322 -1.105653   0.940159
O   1.900524   1.752526   0.047278
Mo   1.270700   0.232181   0.017962
O   -0.207831   0.050122   1.284556
O   2.433045 -1.072410 -1.194049
H   2.071524 -1.927215   1.310478

[CH₃Mo₂O₅] I 1/2
E = -15039.61
Mo -1.392235 -0.142449 -0.086777
Mo   1.446283   0.035129   0.012022
O   -0.002489   0.032665   1.233198
O   0.098449 -0.045636 -1.314991
O   -2.290134 -1.498619   0.229831
O   2.445587 -1.307306   0.075033
O   2.363001   1.437749   0.036647
C   -2.507562   1.623718   0.113423
H   -2.987529   1.820977 -0.856081
H   -1.865352   2.469335   0.367442
H   -3.287078   1.523996   0.870050

[CH₃Mo₂O₅] I 3/4
E = -15037.38
Mo -1.361999 -0.269047 -0.159520
Mo   1.497584   0.058638   0.033388
O   -0.016209   1.080935   0.440955
O   0.274497 -1.185119 -0.681304
O   -2.649097 -1.174144   0.390896
O   2.356234 -0.474058   1.376033
O   2.536779   0.746220 -1.094669
C   -2.833295   1.845084   0.197198
H   -3.686403   1.443994 -0.338041
H   -2.193551   2.575009 -0.281398
H   -2.832484   1.796996   1.278529

[CH₃Mo₂O₅] II 1/2
E = -15038.43
Mo -1.764616   0.212023 -0.000044
Mo   1.015895 -0.347757   0.000025
O   -0.355137   0.014345 -1.294095
O   -0.354967   0.014931   1.293995
O   -3.436027   0.279510   0.000127
O   2.333571   0.959257   0.000069
O   1.428582 -1.959379   0.000243
C   3.715406   1.221616 -0.000188
H   3.965998   1.802464   0.888969
H   3.965339   1.803959 -0.888551
H   4.294339   0.295413 -0.001170

[CH₃Mo₂O₅] II 3/4
E = -15038.34
Mo -1.785360   0.167592   0.000100
Mo   1.037678 -0.348119 -0.000023
O   -0.367975 -0.066079 -1.286892
O   -0.367885 -0.066737   1.287034
O   -3.428239   0.483123 -0.000332
O   2.295291   1.014163   0.000107
O   1.533062 -1.935751 -0.000369
C   3.668709   1.320072   0.000054
H   3.899324   1.910981   0.887897
H   3.899904   1.907920 -0.889676
H   4.277101   0.413027   0.001808

[CH₃Mo₂O₅] III 1/2
E = -15038.10
Mo -1.186598 -0.001021   0.026762
Mo   1.647317   0.000377 -0.005627
O   0.237418 -0.000918   1.265413
O   0.223442 -0.000759 -1.246963
O   -2.998620 -0.000014 -0.117694
O   2.601692 -1.381296 -0.015947
O   2.599387   1.383637 -0.015685
C   -4.396984   0.002340   0.008508
H   -4.796232 -0.908920 -0.437942
H   -4.800584   0.869590 -0.514998
H   -4.678494   0.047112   1.061241

[CH₃Mo₂O₅] III 3/4
E = -15038.30
Mo   1.201920 -0.490994   0.000006
Mo -1.566584   0.154430 -0.000042
O   -0.258377 -0.265368 -1.283906
O   -0.258112 -0.263451   1.284042
O   2.975379   0.016269 -0.000337
O   -2.908192 -0.859171   0.001073
O   -2.049448   1.766126 -0.001035
C   3.844952   1.125685   0.000220
H   4.872129   0.759525   0.017274
H   3.670457   1.742249   0.884127
H   3.693577   1.724581 -0.899898

[CH₃Mo₂O₅] IV 1/2
E = -15037.98
Mo -1.606223 -0.150366 -0.000087
Mo   1.393812 -0.238836   0.000038
O   -0.258342 -1.406683 -0.000278
O   -0.102747   1.238908 -0.000106
O   -3.260648 -0.334038   0.000384
O   2.284083 -0.281365 -1.433138
O   2.283791 -0.281422   1.433393
C   0.130652   2.643830 -0.000006
H   -0.823644   3.170675 -0.000211
H   0.696172   2.914856 -0.891744
H   0.695741   2.914806   0.892022

[CH₃Mo₂O₅] IV 3/4
E = -15037.75
Mo -1.665772 -0.145524 -0.168566
Mo   1.346939 -0.233656   0.030214
O   -0.021281 -1.396920 -0.238701
O   0.050172   1.239964 -0.087321
O   -3.118425 -0.377317   0.620132
O   2.512909 -0.233616 -1.178501
O   2.083054 -0.362242   1.533826
C   0.142117   2.655732   0.057419
H   -0.822794   3.090926 -0.199517
H   0.907047   3.038686 -0.618167
H   0.395113   2.902750   1.088471

[CH₃Mo₂O₅] V 1/2
E = -15037.28
Mo -1.730367   0.088241   0.001161
Mo   1.139962 -0.300574 -0.000345
O   -0.137427   1.160114   0.000672

O   -0.438145 -1.354301   0.001198
O   -3.383023   0.323778 -0.004310
O   2.037994 -0.608395   1.387086
O   2.036229 -0.608430 -1.388937
C   2.628246   1.922038   0.000020
H   3.580956   1.410599   0.000557
H   2.260146   2.336681   0.928065
H   2.261407   2.336383 -0.928680

[CH₃Mo₂O₅] V 3/4
E = -15037.26
Mo -1.734299   0.094279   0.001010
Mo   1.135070 -0.305157 -0.000191
O   -0.128752   1.156967   0.002291
O   -0.444849 -1.353846 -0.000285
O   -3.386127   0.330845 -0.003651
O   2.034543 -0.611847   1.386408
O   2.031943 -0.608791 -1.389086
C   2.678089   1.916564   0.000025
H   3.623194   1.391792   0.000224
H   2.310407   2.329349   0.929073
H   2.311406   2.329747 -0.929264

[Mo₃O₉]⁻ I 1/2
E = -24009.48
Mo -2.179199   0.001154   0.000106
O   -0.717302   1.377152   0.004057
O   -0.718913 -1.376335 -0.001414
O   -3.082802 -0.000447   1.436014
O   -3.080231   0.003212 -1.437405
Mo   1.060068   1.718796   0.000169
O   1.513895   2.586696 -1.373804
O   1.519432   2.588214   1.371327
O   1.858637 -0.001008 -0.000605
Mo   1.058134 -1.719859 -0.000153
O   1.512993 -2.588169   1.373166
O   1.514523 -2.589791 -1.371974

[Mo₃O₉]⁻ I 3/4
E = -24006.19
Mo -2.178403   0.098668 -0.012319
O   -0.648447   1.381078   0.068099
O   -0.841236 -1.371085   0.302241
O   -3.266061   0.226304   1.284495
O   -2.871143   0.049375 -1.561498
Mo   1.150971   1.635231   0.037898
O   1.625516   2.521121 -1.319046
O   1.674296   2.441799   1.426260
O   1.862856 -0.108614 -0.017920
Mo   0.909997 -1.778790   0.077224
O   1.601462 -2.951894   1.061022
O   1.479288 -1.952409 -1.783368

[Mo₃O₉]⁻ II 1/2
E = -24009.21
Mo -0.000168 -0.751039 -0.000061
Mo -2.649572   0.369656   0.000058
O   -1.399374 -0.085931   1.285494
O   -1.396494 -0.080840 -1.285052
O   -0.000795 -2.403348   0.000034
O   -3.088540   2.005632   0.003067
O   -4.035726 -0.605727 -0.003480
O   1.400323 -0.087165 -1.285715
O   1.396213 -0.080840   1.284777
Mo   2.649734   0.369463 -0.000014
O   4.036614 -0.604858   0.004620
O   3.087813   2.005658 -0.003657

[Mo₃O₉]⁻ II 3/4
E = -24006.03
Mo   0.035421 -0.815986   0.000256
Mo -2.702498   0.372412 -0.000160
O   -1.381999 -0.151260   1.220414
O   -1.378939 -0.147457 -1.219874
O   0.049958 -2.475047   0.001198
O   -3.026677   2.281267   0.003672
O   -4.037811 -0.651688 -0.004304

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O    1.418149  -0.091876  -1.291758
O    1.414751  -0.084892   1.291281
Mo   2.640664   0.402902  -0.000137
O    4.062922  -0.523323   0.003878
O    3.018312   2.057806  -0.004292


[Mo2O6] I 1/2
E = -16002.82
Mo  -1.429658  -0.000072  -0.000012
Mo   1.429639  -0.000174   0.000013
O    0.000010  -0.000603  -1.260991
O   -0.000079  -0.000769   1.260992
O   -2.387399  -1.366901  -0.000030
O   -2.387346   1.366814   0.000043
O    2.385231   1.368138   0.000031
O    2.389682  -1.365387  -0.000049


[Mo2O6] I 3/4
E = -16000.11
Mo  -1.413871  -0.019281  -0.000036
Mo   1.428347   0.038670   0.000067
O   -0.008760  -0.028436  -1.278451
O   -0.008937  -0.028603   1.278478
O   -2.376705  -1.388262  -0.000040
O   -2.371160   1.352802   0.000060
O    2.178807   1.512412  -0.000085
O    2.510759  -1.521706  -0.000124


[Mo2O6] I 5/6
E = -15997.25
Mo  -1.476259  -0.058000   0.000002
Mo   1.476094   0.057827  -0.000003
O   -0.000103  -0.000391  -1.213424
O   -0.000096  -0.000405   1.213424
O   -2.209598  -1.548233  -0.000005
O   -2.548301   1.518547   0.000000
O    2.208467   1.548654   0.000007
O    2.550493  -1.517113   0.000001


[Mo2O6] I 7/8
E = -15993.80
Mo  -1.458904  -0.016113   0.011076
Mo   1.488062   0.039559  -0.002480
O   -0.009715   0.020242  -1.205655
O    0.033466  -0.048617   1.213571
O   -2.473434  -1.666610  -0.012039
O   -2.510624   1.604261  -0.026644
O    2.257286   1.512552   0.027916
O    2.549940  -1.544920  -0.042275


[HMo2O7]- I 1/2
E = -18070.63
O   -2.542783   1.354936   0.641266
O   -2.631537  -0.703190  -1.255734
Mo  -1.786759   0.038986  -0.036575
O    0.049669   0.601108  -0.768004
O    1.487709  -1.712421   0.512431
O    2.070741   1.066614   1.242893
Mo   1.660317   0.069891  -0.054544
O   -1.220176  -1.101947   1.177958
O    2.896842   0.154223  -1.196077
H    0.646833  -1.847399   0.989145


[HMo2O7]- I 3/4
E = -18067.90
O   -2.542668   1.333490   0.191199
O   -2.506221  -1.435896  -0.111803
Mo  -1.550497  -0.035197   0.012884
O   -0.288181   0.126016  -1.308373
O    2.449037  -0.898810   1.152539
O    1.948398   1.693256  -0.052764
Mo   1.320670   0.148973  -0.026814
O   -0.212170  -0.161437   1.284991
O    2.114824  -1.082255  -1.310232
H    1.948564  -1.373490   1.820603


[HMo2O7]- II 1/2
E = -18069.71
Mo  -1.935109   0.003030  -0.002025
Mo   1.935096   0.003026   0.002022
O   -0.000055  -0.739338   0.000023
O    2.023279   1.353883   1.034873
O    2.864942  -1.305957   0.590705
O   -2.864973  -1.305122  -0.592483
O   -2.404674   0.410235   1.583670
O    2.404229   0.411977  -1.583362
O   -2.022705   1.355066  -1.033389
H    0.000200  -1.700306  -0.000160


[HMoO4] I 1/2
E = -10062.80
Mo  -0.025872  -0.031946  -0.000700
O   -1.179583   1.455852   0.022887
O    1.752767   0.487395  -0.242059
O   -0.230648  -0.841227   1.452311
O   -0.511761  -0.997479  -1.282628
H    2.440427   0.505393   0.425332


[HMoO4] I 3/4
E = -10059.39
Mo  -0.220671   0.000260  -0.001970
O   -1.014472   1.484624  -0.085200
O    1.558578  -0.218119  -0.952970
O    1.530636   0.048055   1.151503
O   -1.183555  -1.380743   0.043857
H    2.138689   0.518550  -1.174778


[HMoO4] II 1/2
E = -10061.31
Mo   0.140216   0.077101  -0.000068
O    1.565594  -1.136671   0.000058
O    0.444959   1.708654   0.000111
O   -1.530913  -0.422600  -0.711110
O   -1.530613  -0.422624   0.711303
H    2.518700  -1.052324  -0.000027


[HMoO4] II 3/4
E = -10059.68
Mo  -0.216240   0.113164  -0.058773
O   -0.952567  -1.641276   0.026899
O   -1.261279   1.367008   0.247363
O    1.793089   0.312436  -0.501268
O    1.763766  -0.385854   0.601273
H   -1.661973  -1.971400  -0.525669


[HMoO4] III 1/2
E = -10060.21
Mo  -0.372142   0.005024  -0.118998
O   -1.573096   1.098120   0.303126
O   -0.786143  -1.590360   0.193785
O    1.413621   0.651052  -0.032501
O    2.504350  -0.228594   0.097614
H    3.160123   0.351269   0.501712


[HMoO4] III 3/4
E = -10057.83
Mo   0.300144  -0.057289   0.029803
O    1.723435  -0.865443  -0.241570
O    0.350397   1.889895   0.129644
O   -1.425748  -0.858865   0.484195
O   -1.915285   0.051980  -0.525851
H   -2.468441   0.665604  -0.023095


[HMoO4] IV 1/2
E = -10059.55
Mo   0.363606  -0.006121   0.166743
O    0.830951   1.492263  -0.388342
O    1.269883  -1.190670  -0.575675
O   -1.512084  -0.368415   0.167256
O   -2.609385   0.103284  -0.302200


H    0.893633  -0.034625   1.788494


[HMoO4] IV 3/4
E = -10058.16
Mo   0.430598  -0.000004   0.092952
O    1.227986   1.419409  -0.324302
O    1.228129  -1.419325  -0.324344
O   -1.982133  -0.000056   0.413586
O   -2.777752  -0.000000  -0.478258
H    0.345039  -0.000034   1.802558


[Mo2O7]- I 1/2
E = -18052.01
O   -2.483888   1.381718   0.435895
O    1.910615  -1.545680  -0.569578
Mo  -1.563510   0.012608   0.022076
O    1.882281   1.363623  -1.109239
O   -2.582997  -1.317537  -0.267323
Mo   1.410980  -0.066280   0.052342
O   -0.368674   0.313631  -1.274062
O    2.579931   0.418623   1.184950
O   -0.136487  -0.332604   1.208661


[Mo2O7]- I 3/4
E = -18048.89
O    2.429924  -0.738316   1.450865
O   -2.569404  -0.654718  -1.475438
Mo   1.863234  -0.056592   0.004738
O   -2.581025  -0.659082   1.462556
O    2.651582   1.677647  -0.110506
Mo  -1.932789  -0.070616  -0.003383
O    2.446825  -0.906567  -1.343740
O   -2.091640   1.905623   0.019385
O    0.078901   0.043258  -0.010232


[O2] I 1/2
E = -4090.36
O   -0.000000   0.000000   0.597111
O    0.000000   0.000000  -0.597111


[O2] I 3/4
E = -4090.86
O   -0.000000   0.000000   0.597272
O    0.000000   0.000000  -0.597272


[CH3Mo4O13]- I 1/2
E = -35145.37
Mo  -2.329625   0.699653  -0.302497
O   -2.980039   0.534548  -1.844595
O   -1.241161   2.272515  -0.406541
O   -3.553592   1.139798   0.765568
Mo  -0.391557  -1.943358   0.294456
Mo   2.828042  -0.552036  -0.266553
O    3.521541  -1.001737  -1.732305
O    4.045549  -0.480182   0.892244
O    1.973089   1.080798  -0.448357
O    1.529934  -1.765577   0.245042
O   -0.681541  -2.987448  -0.993002
O   -0.662928  -2.757398   1.738910
O   -2.416709  -1.299395   0.199490
Mo   0.534270   2.164733   0.232743
O   -0.478973   0.002591   0.134246
O    0.543389   2.136582   1.907409
O    1.109364   3.679137  -0.196251
C   -3.525918  -2.141883  -0.020067
H   -4.441552  -1.608258   0.242725
H   -3.428490  -3.027676   0.610453
H   -3.565337  -2.444266  -1.069861


[CH3Mo4O13]- I 3/4
E = -35142.71
Mo  -3.469324  -0.253156  -0.454859
O   -3.209758   0.382116  -1.996168
O   -4.696155   0.592861   0.388645
O   -4.539443  -1.768144  -0.707687
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Mo   -0.605449   0.006915   1.027931
Mo   2.359000   -1.782999   -0.190745
O   2.164752   -2.625284   -1.635945
O   3.316672   -2.697996   0.849264
O   3.139130   -0.081038   -0.516913
O   0.763233   -1.370475   0.561318
O   -1.987378   -1.139292   0.353514
O   -0.626678   0.068706   2.679520
O   -2.021235   1.254498   0.379996
Mo   2.390749   1.641399   -0.315716
O   0.707700   1.328127   0.303572
O   3.282291   2.557025   0.780796
O   2.349433   2.457570   -1.787872
C   -1.946852   2.617051   0.029139
H   -2.938382   3.057841   0.154439
H   -1.636305   2.721797   -1.012939
H   -1.233690   3.137986   0.669679

[CH3Mo4O13]- II 1/2
E = -35145.20
Mo   1.631932   -1.764571   -0.127700
O   2.245596   2.046441   1.420555
O   2.045931   -2.234022   1.425976
O   2.494087   -2.791059   -1.137720
O   -0.143572   -2.287570   -0.617320
Mo   1.783172   1.612703   -0.129798
Mo   -1.826086   -1.663240   -0.118484
O   -2.913432   -2.620541   -0.962122
O   -2.148246   -1.920909   1.507140
O   2.389804   -0.107423   -0.638358
O   0.240962   -0.006613   0.801411
O   2.730220   2.553547   -1.147410
Mo   -1.669383   1.817003   -0.117103
O   0.061312   2.295108   -0.613890
O   -2.667259   2.865382   -0.962725
O   -2.231359   0.098459   -0.649280
O   -1.970598   2.103047   1.507650
C   0.255943   -0.003629   2.233301
H   -0.308196   -0.854670   2.616532
H   -0.197334   0.913021   2.612267
H   1.277646   -0.066978   2.609704

[CH3Mo4O13]- II 3/4
E = -35142.83
Mo   0.025143   -2.436350   -0.170222
O   3.128994   -0.000169   1.458969
O   -0.020188   -3.000031   1.418335
O   -0.016284   -3.776848   -1.183356
O   -1.397684   -1.389969   -0.555977
Mo   2.650401   0.000060   -0.152920
Mo   -2.769322   0.000021   -0.161395
O   -4.055416   0.000334   -1.264043
O   -3.179166   -0.000077   1.495535
O   1.669811   -1.528113   -0.555531
O   0.220430   -0.000794   1.417961
O   4.044245   0.000071   -1.086725
Mo   0.025169   2.436381   -0.170185
O   1.670072   1.528569   -0.555057
O   -0.016519   3.777014   -1.183136
O   -1.397262   1.389569   -0.556070
O   -0.020458   2.999688   1.418506
C   -0.178318   0.000112   2.707883
H   -1.295092   0.000182   2.643550
H   0.101343   0.923750   3.225563
H   0.100569   -0.923224   3.226606

[CH3Mo4O13]- III 1/2
E = -35145.14
Mo   1.930641   -0.369711   0.812822
O   0.408182   3.774309   -0.367146
O   1.986925   -0.659538   2.450400
O   3.692742   -0.189179   0.233265
O   1.597324   -1.964233   -0.028815
Mo   0.238222   2.106293   -0.448694
Mo   -0.323026   -2.028229   -0.622872
O   -0.096046   -2.346044   -2.253992
O   -0.723823   -3.452373   0.170138
O   1.731926   1.531864   0.568738
O   0.064841   -0.178828   0.236725

O   0.499307   1.681709   -2.054272
Mo   -2.536908   0.296609   0.404403
O   -1.594470   1.879663   -0.016623
O   -4.155223   0.619207   0.092297
O   -2.075473   -1.165125   -0.638993
O   -2.415926   -0.052773   2.044671
C   4.144370   0.388660   -0.963717
H   5.235640   0.373318   -0.971876
H   3.792376   1.423717   -1.026095
H   3.768439   -0.166682   -1.828589

[CH3Mo4O13]- III 3/4
E = -35142.89
Mo   1.660233   -1.281482   0.744671
O   1.986872   3.516954   0.476375
O   2.407103   -1.699711   2.168432
O   2.723982   -1.731583   -0.763169
O   0.315221   -2.716148   0.272266
Mo   1.350657   2.155443   -0.281565
Mo   -1.358181   -1.952552   -0.473521
O   -1.032280   -2.167067   -2.106995
O   -2.476173   -3.101472   0.021719
O   2.048815   0.649846   0.449008
O   -0.253048   -0.787350   0.815883
O   1.747586   2.220077   -1.919658
Mo   -2.206437   1.341694   0.266175
O   -0.524581   2.100597   -0.105769
O   -3.383223   2.359667   -0.362148
O   -2.415046   -0.312332   -0.528679
O   -2.462276   1.260089   1.922652
C   3.668411   -0.933197   -1.432435
H   4.189811   -1.561352   -2.159911
H   4.407079   -0.514464   -0.741103
H   3.187485   -0.107823   -1.965639

[CH3Mo4O13]- IV 1/2
E = -35145.11
Mo   -0.323853   -2.328552   -0.362836
O   -1.270499   -0.855335   -0.006980
O   -0.718791   -3.555319   0.725900
O   -0.630085   -2.878627   -1.924722
Mo   2.843153   -0.664149   0.292382
Mo   0.897909   2.407948   -0.202268
O   1.020721   3.183877   -1.692710
O   1.145182   3.545574   1.015160
O   -0.687020   1.611687   -0.024663
O   2.242735   1.042072   -0.094736
O   4.266282   -0.982150   -0.539140
O   3.164392   -0.805332   1.933844
O   1.527264   -1.881066   -0.201992
Mo   -2.558713   0.762972   0.098671
O   -2.919130   1.401584   1.609529
O   -3.180862   1.750368   -1.105385
O   -3.860791   -0.610006   0.018222
C   -4.073273   -1.601282   0.981748
H   -3.219813   -2.284735   1.048427
H   -4.958049   -2.179136   0.702800
H   -4.234538   -1.152231   1.969819

[CH3Mo4O13]- IV 3/4
E = -35141.82
Mo   -0.522910   2.361024   -0.053990
O   0.770906   1.340964   0.367559
O   -0.555099   3.801099   1.181324
O   -0.233256   3.098357   -1.550305
Mo   -3.170148   -0.206830   0.124478
Mo   -0.085616   -2.342022   -0.227651
O   -0.091512   -3.280931   -1.626093
O   -0.174521   -3.368837   1.103289
O   1.523294   -1.357905   -0.155119
O   -1.497288   -1.208036   -0.244298
O   -4.332079   -0.353551   -1.101599
O   -3.732804   -0.439098   1.706757
O   -2.081477   1.485087   -0.056771
Mo   3.020357   -0.307608   0.247538
O   3.047598   0.235559   1.834302
O   4.332732   -1.339453   0.083372
O   3.373476   1.070264   -0.953703
C   3.238324   2.470512   -0.943251

H   4.153049   2.910478   -1.348108
H   3.075388   2.846249   0.069718
H   2.391212   2.760348   -1.567549

[CH3Mo4O13]- V 1/2
E = -35145.00
Mo   -0.123726   1.817301   -0.387217
Mo   -3.043870   -0.012357   0.142054
Mo   0.581959   -1.818451   -0.600927
Mo   3.008676   0.186280   0.441965
O   2.396873   -1.508445   -0.145701
O   -0.015944   -3.165470   0.216332
O   -2.033610   1.450895   -0.383143
O   -2.683585   -0.485230   1.906133
O   2.962598   0.252373   2.125079
O   1.839823   1.364594   -0.295818
O   -0.199383   -0.266076   -0.045963
O   -4.660844   0.444050   0.089259
O   -2.844658   -1.337057   -0.880216
O   0.456765   -2.064738   -2.259503
O   4.585273   0.481389   -0.069075
O   -0.088958   2.953474   0.841012
O   -0.030491   2.614144   -1.853981
C   -1.772410   -1.381938   2.502753
H   -0.795390   -0.904923   2.605370
H   -1.658838   -2.287087   1.902712
H   -2.149775   -1.644067   3.493324

[CH3Mo4O13]- V 3/4
E = -35142.72
Mo   -0.168754   1.476613   0.811387
Mo   -3.211832   0.284302   -0.509586
Mo   0.870438   -1.981909   0.216687
Mo   3.111383   0.515076   -0.513943
O   2.531157   -1.285282   -0.353939
O   1.057360   -2.881691   1.628607
O   -2.087564   1.412375   0.414410
O   -3.559758   -1.252858   0.489733
O   4.536609   0.771378   0.345041
O   1.732571   1.456766   0.196582
O   -0.174490   -0.523016   0.519674
O   -4.681295   1.057093   -0.797115
O   -2.521648   -0.162278   -1.978093
O   0.194356   -2.980579   -0.958917
O   3.377836   0.909775   -2.128150
O   -0.073608   1.916363   2.404666
O   -0.281517   2.955868   -0.433084
C   -2.905869   -2.500871   0.548736
H   -2.190360   -2.501985   1.374893
H   -2.373008   -2.718397   -0.380335
H   -3.653358   -3.277133   0.726808

[CH3Mo4O13]- VI 1/2
E = -35144.92
Mo   0.160035   -1.642081   -0.173438
O   4.225325   -0.686849   1.449211
O   0.315731   -2.641728   1.157859
O   0.071256   -2.596421   -1.543581
O   -1.840206   -1.411934   -0.023795
Mo   3.530152   -0.143811   0.012587
Mo   -3.329863   -0.422939   0.252830
O   -4.555763   -0.830286   -0.828158
O   -3.915803   -0.603103   1.822046
O   2.034717   -1.129945   -0.366933
O   0.021532   0.417998   -0.006197
O   4.657408   -0.271121   -1.232738
Mo   -0.935662   1.920322   -0.316474
O   3.083111   1.651962   0.233960
O   -0.736544   2.473383   -1.895073
O   -2.718624   1.347811   -0.033971
O   -0.505891   3.150941   0.756024
C   2.606508   2.291955   1.402430
H   2.321403   3.311380   1.143241
H   3.389353   2.306375   2.165638
H   1.724404   1.782247   1.796074

[CH3Mo4O13]- VI 3/4
E = -35142.72

Mo -0.168718 -1.476027   0.813663
O   -2.518143   0.162765 -1.978266
O   -0.279375 -2.955801 -0.430367
O   -0.075226 -1.914992   2.407280
O   1.733354 -1.455956   0.201104
Mo -3.210299 -0.284537 -0.510900
Mo   3.109813 -0.515391 -0.515532
O   4.538005 -0.770911   0.338749
O   3.370479 -0.912039 -2.130189
O   -2.087063 -1.412560   0.414473
O   -0.174878   0.523430   0.520496
O   -4.679047 -1.057783 -0.800839
Mo   0.870467   1.982036   0.217349
O   -3.560337   1.252107   0.488470
O   1.059565   2.880527   1.629788
O   2.530197   1.285216 -0.355895
O   0.193320   2.982021 -0.956526
C   -2.906727   2.500176   0.549409
H   -3.654486   3.276035   0.728094
H   -2.373403   2.718937 -0.379095
H   -2.191636   2.500406   1.375942

[CH3Mo4O13]- VII 1/2
E = -35144.85
Mo -0.276823 -1.861382 -0.121279
O   3.728892   0.086721   1.798063
O   -0.281258 -3.101209   1.010387
O   0.263602 -2.425209 -1.610597
O   -2.156311 -1.627770 -0.499698
Mo   2.791793   0.068879   0.400938
Mo -3.014697 -0.045660   0.037174
O   -4.542076 -0.082191 -0.660901
O   -3.241551 -0.042702   1.703718
O   1.788268 -1.363662   0.373934
O   -0.405359   0.009903   0.393435
O   3.941305   0.076221 -1.063444
Mo -0.351794   1.873629 -0.152827
O   1.741758   1.464379   0.332387
O   0.157120   2.431009 -1.655202
O   -2.224035   1.566275 -0.514567
O   -0.391420   3.130742   0.959028
C   5.295479 -0.246500 -1.249537
H   5.720863   0.436749 -1.987673
H   5.383276 -1.270246 -1.621859
H   5.855352 -0.157174 -0.313865

[CH3Mo4O13]- VII 3/4
E = -35142.70
Mo   0.427813 -0.181324   1.533818
O   4.431683 -0.956604   0.059532
O   0.370177 -0.096438   3.186166
O   1.427480 -1.679810   0.824872
O   -1.136872 -1.324768   1.013098
Mo   3.321017   0.222259 -0.399836
Mo -2.316078 -1.543388 -0.348450
O   -1.556698 -2.200656 -1.705450
O   -3.604964 -2.530255   0.098787
O   2.126746   0.613455   0.970384
O   -0.546205   1.313056   0.625685
O   2.400804 -0.333871 -1.914624
Mo -1.983108   1.844364 -0.349188
O   4.172837   1.621348 -0.784880
O   -1.509920   2.608807 -1.772645
O   -2.886785   0.219737 -0.733745
O   -2.976283   2.888321   0.522329
C   1.973722 -1.612715 -2.343337
H   2.108175 -1.672326 -3.425255
H   0.918424 -1.748069 -2.101536
H   2.550022 -2.402156 -1.855677

[CH3Mo4O13]- VIII 1/2
E = -35144.46
2.423620   0.553668 -0.660872
-0.417821   2.317207   2.176143
3.556712   0.499584   0.578055
3.236879   0.946438 -2.077481
1.643046 -1.095848 -0.895605
-0.480071   2.348394   0.500465
0.658442 -2.186683   0.412205

-0.180917 -3.754126 -0.215357
1.939765 -2.803170   1.264573
1.150941   1.903442 -0.318150
0.344757 -0.396766   1.217280
-0.789895   3.940091   0.075105
-2.579390 -0.403338 -0.572046
-1.882005   1.337235 -0.166488
-3.513086 -0.359694 -1.982883
-1.201557 -1.490058 -0.772254
-3.540511 -0.934654   0.721885
-0.266377 -1.282899   2.044592
-1.357529 -1.268484   2.061932
0.213067 -1.444274   3.009833
-0.976988 -3.573022 -0.727483

[CH3Mo4O13]- IX 1/2
E = -35144.40
Mo   0.710893 -0.734329   1.429511
Mo   2.372178   0.479272 -0.843198
Mo -1.251119   1.954570   0.228033
Mo -2.066739 -1.154685 -0.742552
O   -2.216072   0.728794 -0.838505
O   -2.356052   2.873186   1.095560
O   2.308088   0.157551   1.078578
O   3.407267 -0.618073 -1.622251
O   -3.588848 -1.800492 -0.444789
O   -1.023788 -1.448168   0.741939
O   -0.254990   0.964244   1.441565
O   0.668642   0.042722 -1.039278
O   2.763523   2.067475 -1.275885
O   -0.337034   3.033577 -0.668967
O   -1.499301 -1.880503 -2.149217
O   0.732335 -1.032873   3.058461
O   1.353689 -2.348918   0.774556
C   1.151695 -3.128484 -0.372713
H   1.920930 -3.901954 -0.398318
H   1.205126 -2.521215 -1.277869
H   0.165157 -3.596918 -0.317033

[CH3Mo4O13]- IX 3/4
E = -35143.66
Mo   0.592423 -0.156511   0.923488
Mo   3.618006   0.345689 -0.446760
Mo -2.309200   1.847845 -0.110613
Mo -2.540330 -1.564458 -0.347487
O   -3.198179   0.201884 -0.450108
O   -3.129534   2.746774   1.053625
O   2.039138   0.990173   0.299247
O   4.628568 -0.785350   0.686560
O   -3.432125 -2.471496   0.755550
O   -0.814065 -1.367666   0.204062
O   -0.679873   1.320561   0.477784
O   3.353810 -0.355147 -1.956270
O   4.616411   1.673641 -0.680579
O   -2.195094   2.760436 -1.521214
O   -2.614219 -2.322947 -1.848346
O   0.609614 -0.235081   2.574665
O   1.908203 -1.529689   0.327191
C   1.772534 -2.796067 -0.055823
H   0.872602 -3.332271   0.220707
H   4.209339 -1.636021   0.849017
H   2.383863 -3.111771 -0.894523

[CH3Mo4O13]- X 1/2
E = -35144.27
-0.549181   2.482755   0.022289
-0.503929   3.926171 -0.831851
1.281344   1.795853 -0.521477
-0.198553   2.756748   1.643004
-2.462773 -0.863080 -0.153312
0.799105 -2.564809   0.075759
1.306752 -3.903830 -0.807775
0.849058 -2.949617   1.710961
1.902660 -1.075913 -0.282033
-0.877561 -2.063372 -0.428807
-3.501900 -0.979852 -1.463779
-3.165310 -1.595104   1.193304
-2.423473   2.210207   0.106184
2.632552   0.663634 -0.089891

-0.888655   0.481141 -0.620960
3.166277   0.907503   1.484476
3.933610   0.872723 -1.137388
-2.970745   1.088123   0.720462
-2.651611   1.019274   1.769841
-4.061719   1.153185   0.689119
-0.132305   0.060500 -1.036065

[CH3Mo4O13]- XI 1/2
E = -35144.27
0.036909 -1.610495   0.709142
3.800232 -0.124392   0.991515
0.204009 -2.410310   2.153253
-0.220965 -3.469965 -0.518686
-1.756745 -1.346366 -0.011868
2.834034 -0.026741 -0.389100
-2.863703   0.153176 -0.272850
-3.287586   0.174608 -1.913795
-4.271323   0.043069   0.656786
1.612676 -1.457485 -0.425262
0.185702   0.204372   1.114188
3.834910 -0.134734 -1.746484
0.088829   1.981845   0.311960
1.916960   1.552977 -0.439193
-0.163723   2.725776 -1.445562
-1.890278   1.602358   0.190430
0.285282   3.154833   1.465773
-0.773883 -3.474347 -1.595331
-0.832167 -4.410463 -2.164736
-1.229291 -2.566719 -2.009699
0.676675   2.798341 -1.906721

[CH3Mo4O13]- XII 1/2
E = -35144.16
2.013241 -0.243218   0.856773
-0.062129   4.052384   0.031231
2.947155 -0.559034   2.200501
3.152425   0.726200 -0.757052
1.991429 -1.648849 -0.396384
0.041019   2.389934 -0.229446
0.079379 -2.159097 -0.417708
0.167960 -1.768626 -2.321714
-0.029805 -3.773119 -0.019129
1.265239   1.652450   0.865635
0.213961 -0.831097   1.040714
0.588805   2.088364 -1.818322
-2.602620 -0.047262   0.284074
-1.610700   1.567377   0.032989
-4.114957   0.194090 -0.417111
-1.823272 -1.442402 -0.573911
-2.859164 -0.357863   1.917532
2.761319   0.625088 -1.909187
3.216813   1.263336 -2.672960
2.001944 -0.109941 -2.201670
-0.705012 -1.597894 -2.685193

[CH3Mo4O13]- XIII 1/2
E = -35144.16
1.851025   0.531309   0.917384
-1.049583   3.486197   0.111568
2.205668   0.938564   2.480365
3.502832   0.374881 -0.071855
1.667940 -1.462260   0.766282
-0.642314   2.030975 -0.621855
0.550302 -2.202517 -0.459284
0.823365 -1.449942 -1.954215
0.898308 -3.845072 -0.596946
1.451977   2.221686 -0.168618
-0.063860   0.469631   0.444731
-0.310826   2.290988 -2.253602
-2.392703 -0.550079   0.407999
-2.269378   1.025273 -0.622891
-3.913846 -1.133132   0.011840
-1.267632 -2.023622   0.014851
-2.449994 -0.234058   2.053789
3.809008   0.055176 -1.272918
4.857948   0.086285 -1.549773
3.040208 -0.265521 -1.972053
2.062446   2.568164 -0.821341

[CH₃Mo₄O₁₃]⁻ XIV 1/2

[CH$_3$Mo$_4$O$_{13}$]⁻ XIV 1/2
E = -35144.16
```
1.851937   -0.529994   0.917394
0.895058    3.845198   -0.596817
2.207403   -0.936881   2.480289
3.503083   -0.371607   -0.072511
1.454237   -2.221487   -0.167321
0.548330    2.202375   -0.459124
-0.640549  -2.030942   -0.621985
-0.307593  -2.289728   -2.253624
-1.046706  -3.486939    0.110517
1.666667    1.463419    0.766624
-0.063083  -0.469546    0.444532
0.822427    1.450025   -1.953992
-2.393729   0.548193    0.408009
-1.269803   2.022353    0.014225
-3.915546   1.129848    0.012326
-2.268901  -1.027163   -0.622613
-2.450101   0.232297    2.053853
3.808509   -0.050681   -1.273425
4.857397   -0.080629   -1.550597
3.039069    0.269690   -1.972007
2.063838   -2.567818   -0.821076
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XV 1/2
E = -35144.15
```
0.687466   -1.868426    0.770704
3.776412    1.814895    0.071132
0.585220   -2.822689    2.118886
1.855456   -2.908842   -0.462493
-0.678448  -1.988228   -0.673696
2.323326    1.042694   -0.305092
-2.208186  -0.946422   -0.544835
-3.055837  -1.326594   -1.947148
-3.202730  -1.397169    0.736470
2.118713   -0.452579    0.676278
-0.564965  -0.222201    1.114889
2.289930    0.631070   -1.951842
-1.035272   2.202073    0.266786
0.841422    2.150168    0.050110
-1.461607   3.675485   -0.408896
-2.023914   0.966623   -0.720263
-1.490520   2.304955    1.884504
2.283937   -2.694161   -1.601578
2.905499   -3.456314   -2.073620
2.054517   -1.766437   -2.128946
-0.804730  -0.067787    2.030971
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XVI 1/2
E = -35143.92
```
-0.043822  -1.422234    1.096206
4.215085    0.208024   -0.396348
0.033368   -2.221121    2.537863
-0.239107  -3.002686   -0.161140
-2.008902  -1.008179    0.660910
2.561654    0.004253   -0.643298
-2.493625   0.191234   -0.604969
-2.471099  -0.424314   -2.192154
-4.096460   0.587955   -0.279087
1.951223   -1.142203    0.592607
-0.074355  -0.130792   -0.816064
2.342298   -0.560642   -2.234638
0.088107    1.860076    0.479851
1.732683    1.758899   -0.390503
0.166781    3.308731    1.306437
-1.564690   1.917529   -0.363593
0.041579    0.586240    1.688517
-0.589869  -3.153143   -1.325725
-0.670808  -4.170290   -1.716806
-0.832809  -2.301434   -1.964679
0.098391   -0.228804   -1.753815
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XVII 1/2
E = -35143.85
```
1.829583   -0.042886    0.968109
-0.118746   3.811505   -0.566826
1.939525   -0.220017    2.621365
```

3.875573   0.176716   0.309143
1.919880   -1.635125   -0.040265
-0.050932   2.131464   -0.552616
0.048580   -1.850032   -0.756053
0.251054   -1.825709   -2.407073
-0.109277 -3.634662   -0.073993
1.572762   1.797378   0.365362
0.006433   -0.141074   0.317361
0.182072   1.640999   -2.147105
-2.481023 -0.077050   0.481131
-1.801542   1.641313   -0.022806
-4.126441   0.032683   0.148047
-1.912857 -1.588487   -0.428589
-2.350197 -0.301126   2.146055
4.193666   -0.085517   -0.835796
5.235625   0.058433   -1.141458
3.463903   -0.470681   -1.556282
-1.028176 -3.892664   0.043114

[CH$_3$Mo$_4$O$_{13}$]⁻ XVIII 1/2
E = -35143.85
```
1.829594   -0.043596    0.968327
-0.117726   3.811423   -0.567447
1.939986   -0.221489    2.621465
3.875449    0.176462    0.308403
1.919581   -1.635517   -0.040362
-0.050173   2.131374   -0.552296
0.048181   -1.849936   -0.756168
0.250734   -1.826255   -2.407173
-0.110310  -3.634069   -0.073054
1.572868    1.797182    0.366463
0.006483   -0.141336    0.317673
0.183312    1.640008   -2.146473
-2.481268  -0.076391    0.480827
-1.800938   1.641857   -0.022344
-4.126584   0.034070    0.147464
-1.913207  -1.587700   -0.429074
-2.350835  -0.300993    2.145700
4.192662   -0.084776   -0.837008
5.234394    0.059319   -1.143389
3.462412   -0.469075   -1.557474
-1.029249  -3.891654    0.044038
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XIX 1/2
E = -35143.83
```
Mo    2.927214    0.388279    0.324293
O     1.910130   -0.993116    0.964210
O     2.782382    1.858840    1.131728
O     4.561279   -0.003942    0.174040
Mo    0.729148   -1.184084   -0.710032
Mo   -2.309775   -0.828975    0.183254
O    -2.765177   -0.762662    1.796630
O    -3.371216   -1.910012   -0.522654
O    -2.750666    0.885196   -0.467768
O    -0.624458   -1.968542    0.677788
O     1.365843   -2.573497   -1.367658
O    -0.798056   -0.945012   -1.512967
O     1.939872    0.166457   -1.281912
Mo   -1.247116    2.041667   -0.157306
O    -0.639129    2.734754   -1.574522
O    -1.575478    3.249013    0.980100
O    -0.239200    0.710302    0.492896
C    -0.308916   -2.313083    2.020774
H     0.553835   -2.978023    2.018405
H    -1.172015   -2.817451    2.453702
H    -0.075147   -1.417562    2.595227
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XIX 3/4
E = -35142.25
```
Mo    3.412824    0.586386    0.402732
O     2.673900   -1.090735    0.070775
O     3.716339    0.901478    2.034737
O     4.787053    0.930954   -0.519917
Mo    0.997591   -0.406568   -0.786864
Mo   -1.899347   -1.476692    0.370842
O    -1.853452   -2.180692    1.898927
O    -2.589467   -2.564108   -0.710435
O    -3.119127   -0.049986    0.506700
O     0.075195   -2.079757   -0.083690
```

O   1.067589   -0.525756   -2.433286
O   -0.768727   0.133746   -0.176094
O   1.839741   1.318408   -0.251413
Mo   -2.420032   1.666011   -0.027236
O   -2.540345   1.903014   -1.685148
O   -3.519093   2.729331   0.703266
O   -0.946876   2.573560   0.730816
C   0.579831   -3.389407   0.096974
H   1.666606   -3.368668   0.013376
H   0.153311   -4.047405   -0.663334
H   0.295759   -3.746931   1.088288

[CH$_3$Mo$_4$O$_{13}$]⁻ XX 1/2
E = -35143.83
```
2.026067    0.089487    0.847974
-0.125173   3.675820   -0.048050
2.260796    0.015665    2.500525
4.042157    0.092356    0.084074
1.898185   -1.530854   -0.148365
-0.043908   1.840057   -0.594908
0.017140   -2.000274   -0.469477
0.160508   -2.468677   -2.075954
-0.097162  -3.383984    0.471991
1.833426    1.808777    0.084146
0.113649    0.027395    0.387971
0.079016    1.705850   -2.248393
-2.715975   0.012818    0.377294
-1.962156   1.607887   -0.156136
-4.333256  -0.011769   -0.099136
-1.825080  -1.401910   -0.459414
-2.668658  -0.141508    2.057227
4.264433   -0.162532   -1.085053
3.454637   -0.402370   -1.782114
5.300585   -0.163685   -1.440187
0.747852    4.013194    0.171649
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XXI 1/2
E = -35143.72
```
-1.171604   1.592417   -0.713663
-1.638841   1.885299   -2.280706
0.683817    2.269086   -0.563645
-1.784298   2.942110    0.490746
-1.574551  -1.446133    0.538322
1.511548   -1.847118   -0.448481
1.202144   -1.924438   -2.094031
2.672932   -3.016689   -0.149821
2.412725   -0.240954   -0.024731
0.021800   -2.482906    0.539995
-2.683395  -2.344250   -0.350272
-2.072898  -1.376083    2.142682
-2.673066   0.397593    0.074437
1.963416    1.516281    0.507939
-0.349932  -0.131462   -0.261903
1.412376    1.468338    2.099168
3.352691    2.464491    0.451420
-3.926446   0.715060    0.417231
-4.430091   1.456850   -0.192058
-4.471831  -0.011771    1.003509
-2.597802   2.734689    0.965562
```

[CH$_3$Mo$_4$O$_{13}$]⁻ XXII 1/2
E = -35143.72
```
-1.573971   1.446450   -0.538570
-2.071543   1.376243   -2.143149
0.022756    2.482780   -0.539784
-2.682820   2.345381    0.349216
-1.171954  -1.591931    0.713896
1.962692   -1.516760   -0.508208
1.411181   -1.468266   -2.099261
3.351644   -2.465478   -0.452352
2.412920    0.240115    0.024768
0.683103   -2.269344    0.563608
-1.785773  -2.941553   -0.490010
-1.639147  -1.884343    2.281033
-2.673032  -0.396459   -0.074291
1.512108    1.846495    0.448741
-0.349669   0.131521    0.261909
1.202473    1.923589    2.094261
2.673933    3.015742    0.150496
```

-3.926521 -0.713686 -0.417034
-4.471404   0.012699 -1.004358
-4.430761 -1.454380   0.193117
-2.599665 -2.734267 -0.964210

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXIII 1/2
E = -35143.72
-1.175037   1.589365 -0.714474
-1.641846   1.879032 -2.282220
0.677899   2.271520 -0.563159
-1.793678   2.938698   0.487697
-1.570272 -1.449410   0.539497
1.516377 -1.842614 -0.449810
1.206896 -1.919245 -2.095398
2.680391 -3.009882 -0.152262
2.414395 -0.234827 -0.024945
0.028374 -2.482540   0.538470
-2.678592 -2.351326 -0.346051
-2.065473 -1.379685   2.144759
-2.672991   0.390049   0.074183
1.958102   1.520184   0.509151
-0.348166 -0.131627 -0.261271
1.405044   1.468675   2.099564
3.344574   2.472603   0.455847
-3.928017   0.703990   0.414731
-4.432698   1.444090 -0.195775
-4.472477 -0.024504   0.999832
-2.606475   2.728850   0.962592

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXIV 1/2
E = -35143.72
-1.571455   1.448668 -0.538764
-2.069155   1.378867 -2.143335
0.026948   2.482630 -0.540110
-2.678922   2.349362   0.349004
-1.174454 -1.590273   0.714136
1.960060 -1.519552 -0.508329
1.408573 -1.470458 -2.099370
3.347806 -2.469924 -0.452304
2.412989   0.236731   0.024241
0.679526 -2.270422   0.563614
-1.790208 -2.939079 -0.489885
-1.642463 -1.881814   2.281192
-2.673524 -0.392578 -0.074435
1.515056   1.844414   0.448874
-0.349482   0.131852   0.261577
1.204872   1.921385   2.094297
2.678914   3.011911   0.151554
-3.927589 -0.707719 -0.417007
-4.471650   0.019920 -1.003570
-4.432832 -1.447987   0.192878
-2.603702 -2.730072 -0.964056

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXV 1/2
E = -35143.72
-2.523386   0.481122 -0.266887
-3.159909   0.362702 -1.818698
-1.701060   2.148180 -0.175949
-3.801436   0.561112   0.826685
-0.115891 -1.946809   0.238592
2.978503 -0.221661 -0.246014
3.643685 -0.535595 -1.763720
4.222663 -0.127520   0.888539
1.976055   1.288821 -0.273397
1.733719 -1.542898   0.237781
-0.250412 -3.056740 -1.013896
-0.343187 -2.756670   1.687174
-2.230280 -1.570570   0.042889
0.203744   1.963800   0.323303
-0.496701   0.026377   0.084839
0.343078   2.332980   1.931491
0.522062   3.519332 -0.740719
-3.184106 -2.513615 -0.017483
-4.209499 -2.180460 -0.079527
-2.864859 -3.526057 -0.217174
-0.291971   3.981167 -0.960375

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXVI 1/2

E = -35143.71
-2.525520   0.530325 -0.266242
-3.144960   0.427500 -1.825929
-1.670894   2.172546 -0.145943
-3.814044   0.619624   0.814984
-0.165024 -1.943434   0.225968
2.967975 -0.271104 -0.231330
3.686759 -0.603168 -1.720580
4.174245 -0.178457   0.943343
2.000853   1.266056 -0.317232
1.696892 -1.569510   0.220836
-0.317189 -3.045417 -1.031330
-0.392094 -2.758271   1.671737
-2.265919 -1.529628   0.051966
0.228770   1.959250   0.316078
-0.499949   0.033998   0.072071
0.395213   2.316661   1.924348
0.629692   3.500553 -0.741507
-3.238866 -2.453993   0.007673
-4.258643 -2.100572 -0.025695
-2.943798 -3.473560 -0.192480
1.546352   3.522976 -1.029878

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXVII 1/2
E = -35143.65
2.215695   0.844102 -0.018296
3.034587   0.895034   1.705823
0.913543   2.251048   0.539294
3.164637   1.522903 -1.203169
0.597194 -1.997407 -0.201529
-2.691331 -0.832748   0.140373
-3.576493 -1.400915   1.450941
-3.735795 -0.704604 -1.168399
-1.952463   0.817844   0.566103
-1.321309 -2.022621 -0.262278
0.925209 -2.664938   1.309853
1.097708 -2.351469 -1.413460
2.698804 -1.159960 -0.051557
-0.828981   2.258935 -0.026885
0.574033 -0.115187 -0.500916
-0.899155   2.382981 -1.700552
-1.528539   3.654015   0.593269
3.772317 -1.776585   0.444707
4.734006 -1.300351   0.286741
3.684104 -2.836955   0.639326
3.501609   0.085888   1.952254

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXVIII 1/2
E = -35143.64
-2.426587   0.614209 -0.167814
-3.336841   0.460908 -1.573238
-1.491236   2.234026 -0.329512
-3.461584   0.810179   1.145497
-0.286059 -1.921232 -0.008346
2.879190 -0.409694 -0.089677
3.765709 -0.889883 -1.440994
3.943635 -0.159704   1.193022
1.913167   1.135655 -0.484930
1.631038 -1.675550   0.356285
-0.301274 -2.755816 -1.444732
-0.482351 -3.036047   1.529422
-2.331247 -1.515997   0.168915
0.392738   2.150754   0.099851
-0.498081   0.028466 -0.052975
0.506962   2.275472   1.763985
0.837564   3.643628 -0.518233
-3.351219 -2.360489   0.360125
-4.325424 -1.922680   0.529409
-3.228676 -3.371526 -0.011210
-1.392144 -3.215138   1.792323

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXIX 1/2
E = -35143.54
1.438387 -1.116264   0.965781
2.206168   3.396030 -0.418329
1.789338 -1.739945   2.468675
3.079349 -1.170968 -0.484584
0.703912 -2.416505 -0.213368
1.346339   1.943267 -0.395470

-1.137796 -1.867660 -0.574068
-1.014277 -1.908427 -2.257464
-2.043085 -3.177428 -0.031877
1.997602   0.818727   0.884439
-0.344694 -0.506121   0.800942
1.577639   1.092777 -1.869640
-2.103358   1.260257   0.253103
-0.429713   2.243127 -0.088176
-3.198966   2.408689 -0.276132
-2.461372 -0.392266 -0.479663
-2.093279   1.236780   2.153679
2.811208 -1.146200 -1.675768
3.607853 -0.914099 -2.388971
1.814855 -1.396295 -2.048208
-1.268930   0.888685   2.511196

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXX 1/2
E = -35143.42
1.760154 -0.149357   1.060810
-0.027855   3.950266 -0.375569
2.155865 -0.371874   2.669170
3.794294   0.273443   0.261411
1.884568 -1.600310 -0.162804
-0.068138   2.284526 -0.566433
0.038553 -1.934019 -0.768000
0.223282 -1.287827 -2.340479
0.064804 -3.617186 -0.780727
1.397962   1.630811   0.432523
-0.074216 -0.580994   0.769881
0.171747   1.542634 -2.275052
-2.416833 -0.137667   0.531692
-1.683032   1.570206   0.100555
-4.002445   0.051355   0.003907
-1.854732 -1.618634 -0.488639
-2.544441 -0.429955   2.181773
4.037918   0.102931 -0.913905
5.043624   0.327969 -1.289852
3.278258 -0.284235 -1.603748
0.227330   0.576918 -2.429461

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXXI 1/2
E = -35143.32
1.534549 -0.128558   1.183095
0.326501   3.719839 -1.233240
1.792062 -0.455383   2.791132
3.525851   0.433789   0.452914
1.832695 -1.554077 -0.066909
0.146141   2.107095 -0.760373
0.103953 -2.085133 -0.730615
0.438493 -2.319443 -2.365978
-0.191975 -3.594006 -0.059671
1.162565   1.751591   0.757197
-0.257089 -0.551635   0.677834
0.799889   1.079146 -1.960756
-2.402244   0.176702   0.565954
-1.595505   1.725173 -0.451995
-3.975035 -0.121798 -0.005066
-1.720758 -1.309091 -1.016365
-2.392149   0.549567   2.217177
3.754650   0.471204 -0.740614
4.760419   0.752586 -1.074530
2.982444   0.240521 -1.484729
-2.295895 -1.594191 -1.725819

[CH$_3$Mo$_4$O$_{13}$]$^-$ XXXII 1/2
E = -35143.19
0.010679   2.371991 -0.216420
-2.992339 -0.000169   1.568297
0.157521   2.912199   1.369738
-0.007586   3.752515 -1.169597
1.530454   1.462268 -0.784452
-2.525037   0.000127 -0.042701
2.546380 -0.000051 -0.067157
4.003604 -0.001364 -0.879670
2.679276   0.000920   1.772355
-1.680841   1.592734 -0.578289
-0.090841   0.001112   0.943322
-3.952408 -0.000205 -0.922764
0.010574 -2.372574 -0.216240

-1.680095 -1.592158 -0.578517
-0.007201 -3.753707 -1.168551
1.529158   -1.461428 -0.784246
0.157103   -2.911463 1.370401
-0.045717 0.000718   2.277401
-0.253030 -0.938821 2.784337
-0.247550 0.941202   2.784888
1.819393   0.000254   2.287884

[CH₃Mo₄O₁₃]⁻ XXXIII 1/2
E = -35143.13
-1.840915 1.432541 -0.434488
-1.569936 0.907023 -2.041330
0.004032   2.335955 -0.216629
-2.817835 2.779519 -0.107953
-1.135903 -1.994769 0.831949
2.213872   -1.263777 -0.760812
2.266779   -1.634839 -2.400364
3.469547   -2.079693 0.008215
2.426698   0.601587   -0.506556
0.614689   -1.725092 -0.032646
-2.359107 -2.510494 -0.251165
-1.047516 -2.793731 2.315742
-3.672395 0.146257   -0.508857
1.545322   1.891202   0.555957
-1.235375 0.049200   1.072811
1.152872   1.153073   2.048055
2.516678   3.244942   0.809937
-3.779222 -0.763924 -1.307942
-3.065630 -0.874672 -2.130471
-4.627382 -1.451036 -1.231391
-0.484469 0.381325   1.606344

[CH₃Mo₄O₁₃]⁻ XXXIV 1/2
E = -35143.06
-2.294375 0.484569 -0.404395
-2.840426 0.253048 -1.978301
-1.356304 2.159524 -0.490048
-3.611685 0.809919 0.588761
-0.139736 -1.930828 0.518552
2.693452   -0.342358 -0.343218
2.203267   -1.169022 -1.976778
4.321109   -0.564988 -0.035545
1.936118   1.355851   -0.210711
1.685558   -1.515643 0.764882
-0.080444 -2.986376 -0.806194
-0.577441 -2.771118 1.907268
-2.246621 -1.548001 0.059095
0.339648   2.291745   0.316472
-0.454640 -0.025536 0.230041
0.204685   2.278210   1.986675
0.774563   3.858686   -0.089268
-3.220641 -2.447802 -0.119206
-4.140019 -2.111836 -0.577220
-2.950376 -3.489406 -0.022009
1.594787   -1.919737 -1.955932

[CH₃Mo₄O₁₃]⁻ XXXV 1/2
E = -35143.02
-2.408435 0.697427 -0.226966
-3.279905 0.745255 -1.664337
-1.310043 2.295937 -0.290346
-3.445197 1.029354 1.057098
-0.387829 -1.961705 0.181259
2.837224   -0.521118 -0.216578
3.782811   -1.017167 -1.520789
3.847716   -0.331660 1.117699
1.971412   1.032094   -0.656272
1.505910   -1.782949 0.144991
-0.698950 -3.088980 -1.028109
-0.709222 -2.658190 1.675879
-2.504494 -1.355401 0.037559
0.517289   2.148773   0.077171
-0.556166 -0.047289 -0.115326
0.833215   1.556371   1.857975
1.237885   3.647142   -0.062906
-3.591293 -2.129963 0.118824
-4.554866 -1.667406 -0.040438
-3.432168 -3.198468 0.111984

0.688570   2.187685   2.565424

[CH₃Mo₄O₁₃]⁻ XXXVI 1/2
E = -35142.88
-2.429526 0.567149 -0.289438
-3.163837 0.475576 -1.799972
-1.451473 2.215852 -0.337779
-3.611490 0.845780 0.874848
-0.250635 -1.957843 0.292376
2.827648   -0.366870 -0.361630
3.683097   -0.959104 -1.677578
3.977749   -0.279949 1.159932
1.850971   1.190912   -0.555555
1.613806   -1.650693 0.301353
-0.445997 -3.097383 -0.931662
-0.563222 -2.687414 1.774582
-2.374350 -1.479910 0.049212
0.351053   2.151338   0.223013
-0.522237 -0.023796 0.027336
0.436924   2.065569   1.893171
0.849935   3.703223   -0.177507
-3.398390 -2.341327 0.009088
-4.375303 -1.949214 -0.235247
-3.151284 -3.392486 -0.027663
4.739243   0.301886   1.103851

[CH₃Mo₄O₁₃]⁻ XXXVII 1/2
E = -35142.80
0.038492   2.103041   -0.172210
-3.482366 0.347091   1.562706
0.071866   2.641492   1.422860
0.152035   3.452184   -1.168993
1.449376   1.006259   -0.492688
-2.950213 0.038109   -0.002716
3.076844   -0.051564 -0.039182
4.164974   -0.164046 -1.334588
3.764526   0.336349   1.461076
-1.640208 1.268614   -0.531911
-0.141387 -0.579802 1.250572
-4.283759 0.130602   -1.022371
-0.236027 -1.927409 -0.178618
-2.183964 -1.637765 -0.105776
-0.031134 -1.559451 -1.804580
1.906897   -1.732464 0.131919
-0.075203 -3.579206 0.112634
0.353440   -0.472689 2.452228
0.673008   -1.362518 2.981801
0.451107   0.523823   2.863554
2.359985   -2.575491 0.088861

[CH₃Mo₄O₁₃]⁻ XXXVIII 1/2
E = -35142.72
-2.672389 0.310816 -0.110292
-3.534802 0.370399 -1.549235
-1.955604 2.067847 0.100317
-3.770466 0.171717 1.154127
0.017805   -1.865886 0.118700
3.232137   -0.158586 -0.089980
4.025962   -0.694087 -1.490852
4.241832   0.288788   1.194600
1.802406   1.267294   -0.568257
1.714480   -1.795200 0.447127
0.105464   -2.860824 -1.205301
-0.344677 -2.800460 1.499468
-2.190565 -1.672996 -0.156533
-0.061914 2.050762   0.101301
-0.594902 0.076202   -0.065560
0.395157   2.523412   1.638098
0.264618   3.310147   -0.971859
-3.058532 -2.695731 -0.173229
-4.107741 -2.470387 -0.044300
-2.642090 -3.684377 -0.048719
2.172952   1.967228   -1.110751

[CH₃Mo₄O₁₃]⁻ XXXIX 1/2
E = -35142.71
-2.196337 1.124585 -0.140146
-3.246294 1.696877 -1.308271

-0.679852 2.266471 -0.136277
-2.886235 1.459982 1.353448
-0.904657 -2.105703 -0.034070
2.649392   -1.170902 -0.023612
3.736877   -1.536385 -1.271744
3.246427   -1.337041 1.552772
1.745134   0.623570   -0.304318
0.899261   -2.096244 -0.278776
-1.563400 -3.357440 -0.945855
-1.126478 -2.430265 1.601970
-2.758683 -0.873013 0.053818
1.199637   2.327700   0.077400
-0.798563 -0.225662 -0.918869
1.609280   2.777531   1.644655
1.862736   3.438492   -1.000134
-3.898405 -1.293306 0.632922
-4.460504 -0.573462 1.212297
-4.006902 -2.361362 0.755110
0.118691   0.061136   -1.046315

[CH₃Mo₄O₁₃]⁻ XL 1/2
E = -35142.62
-2.576385 0.408419 0.049289
-3.390484 0.628488 -1.460246
-1.746717 2.334362 0.019105
-3.497603 0.395038 1.468878
-0.196287 -1.904438 0.141613
2.982598   -0.304616 -0.069913
3.845119   -0.821481 -1.420464
4.064180   -0.029014 1.190347
1.967690   1.188890   -0.466635
1.675615   -1.503690 0.421717
-0.064862 -2.957027 -1.156513
-0.508704 -2.816245 1.510096
-2.230673 -1.653159 -0.289415
0.315771   2.064482   -0.009018
-0.489844 0.134756   0.067956
0.658499   2.721717   1.485284
0.395416   3.293717   -1.149486
-3.105168 -2.592303 -0.669943
-4.012029 -2.239318 -1.140632
-2.752186 -3.612172 -0.728941
-2.105128 2.992883   -0.578545

[CH₃Mo₄O₁₃]⁻ XLI 1/2
E = -35142.44
-2.310449 0.837654 -0.279938
-2.985983 0.683754 -1.812453
-1.121681 2.288749 -0.420554
-3.511095 1.297492 0.800649
-0.674663 -1.952468 0.329895
2.935544   -0.671439 -0.305587
3.387126   -1.314775 -1.805992
4.180270   -0.480679 0.832531
1.804521   0.932237   -0.496252
1.376068   -1.790304 0.570615
-0.698436 -2.890545 -1.066150
-0.950440 -2.919783 1.684943
-2.639332 -1.303872 0.064612
0.686714   2.196898   0.221043
-0.648313 -0.070474 0.340835
0.719192   2.141447   1.900717
1.328554   3.688004   -0.214308
-3.712316 -2.030816 -0.269209
-4.535079 -1.508934 -0.735927
-3.602951 -3.105939 -0.289850
1.728146   -2.537374 1.060108

[H₂O] I 1/2
E = -2079.39
O 0.000000   0.116100   -0.000000
H 0.761524   -0.464400 -0.000000
H -0.761524 -0.464400 0.000000

[CH₃MoO₂] I 1/2
E = -7036.70
Mo 0.114670 0.000004 -0.191288
O   0.865529 1.388120 0.371786

O   0.865352   -1.388219   0.371769
C   -1.946682   0.000071   0.228091
H   -2.292941   0.891620   0.749621
H   -2.292909   -0.891210   0.750095
H   -2.397230   -0.000205   -0.782614

[CH₃MoO₂] I 3/4
E = -7034.23
Mo   0.104566   0.058669   0.000012
O   0.342613   1.699398   -0.000000
O   1.420425   -1.350855   -0.000028
C   -1.932608   -0.545112   -0.000034
H   -2.603707   0.315734   0.000459
H   -2.148043   -1.149042   0.888780
H   -2.148676   -1.148481   -0.889054

[CH₃MoO₄] I 1/2
E = -11131.45
Mo   -0.338994   0.034928   0.001669
O   -1.235035   -1.626108   0.039389
O   1.412928   -0.137154   0.558326
O   -0.385086   0.610758   -1.574343
O   -1.182489   1.060728   1.032179
C   2.704944   -0.085026   -0.010928
H   3.305332   0.633337   0.548163
H   2.657660   0.216561   -1.059769
H   3.162542   -1.072501   0.062674

[CH₃MoO₄] I 3/4
E = -11128.22
Mo   -0.489187   -0.027365   -0.011399
O   0.629232   0.928529   1.422622
O   1.430245   0.431897   -0.542828
O   -0.821029   -1.640467   0.363729
O   -1.702897   0.872262   -0.761786
C   2.593310   -0.332693   -0.378680
H   3.408398   0.310276   -0.045539
H   2.841433   -0.734687   -1.368083
H   2.451740   -1.167891   0.310554

[CH₃MoO₄] II 1/2
E = -11129.96
Mo   -0.220933   0.152357   -0.000018
O   1.500126   -0.521396   -0.000071
O   -0.447499   1.800581   0.000031
O   -1.641720   -0.864540   0.712461
O   -1.641816   -0.864539   -0.712382
C   2.892573   -0.324903   0.000034
H   3.317005   -0.793806   -0.889001
H   3.317115   -0.795684   0.888039
H   3.136912   0.739069   0.001207

[CH₃MoO₄] II 3/4
E = -11128.29
Mo   -0.191291   0.281511   0.031942
O   1.388432   -0.695599   -0.311685
O   -0.083665   1.929757   -0.165454
O   -1.871635   -0.776267   0.620328
O   -1.599435   -1.226001   -0.574818
C   2.699675   -0.634095   0.191223
H   3.157269   -1.622279   0.106093
H   2.716566   -0.331890   1.243754
H   3.292774   0.080171   -0.385716

[CH₃MoO₄] III 1/2
E = -11129.84
Mo   0.146201   0.166766   0.000005
O   0.767812   0.866078   1.387599
O   0.767406   0.865242   -1.388195
O   -1.880574   0.379158   0.000141
O   -1.679872   -0.903741   0.000125
C   1.064736   -1.742924   0.000286
H   2.147321   -1.612504   0.000236
H   0.762696   -2.293861   -0.891606
H   0.762941   -2.294149   0.892085

[CH₃MoO₄] IV 1/2
E = -11129.01
Mo   0.662740   0.003312   0.119288
O   0.738737   1.662168   -0.156129
O   2.039619   -0.830826   -0.356975
O   -0.963353   -0.939773   0.106367
O   -2.309634   -0.744217   -0.156454
C   -2.618754   0.629437   -0.062461
H   -3.689419   0.683265   -0.253141
H   -2.076683   1.209736   -0.816157
H   -2.399417   1.012471   0.939506

[CH₃MoO₄] IV 3/4
E = -11126.69
Mo   0.570585   0.071622   0.066797
O   1.037723   -1.809151   -0.149047
O   1.641847   1.184994   -0.542206
O   -1.067459   0.519003   1.042805
O   -1.620167   -0.475432   0.160366
C   -2.576741   0.138174   -0.695656
H   -2.887150   -0.647584   -1.383163
H   -2.132178   0.971589   -1.243933
H   -3.420341   0.483516   -0.099777

[Mo₃O₁₀]²⁻ I 1/2
E = -26057.41
Mo   3.387013   -0.535255   -0.000690
O   1.508284   0.025864   -0.014343
O   -3.699666   -1.492226   -1.389804
O   3.700059   -1.491562   -1.390178
O   3.691480   -1.456646   1.414021
Mo   -0.000062   1.070719   -0.001099
O   -0.000005   2.052696   1.376691
O   -0.000504   2.086525   -1.354481
O   -1.508001   0.025822   -0.013821
Mo   -3.386918   -0.535324   -0.000672
O   -4.395687   0.852670   -0.015373
O   -3.692276   -1.455822   1.414439
O   4.396159   0.852484   -0.014226

[Mo₃O₁₀]²⁻ I 3/4
E = -26053.87
Mo   3.465146   -0.531650   0.000443
O   1.587444   0.040925   -0.020980
O   -3.695189   -1.723658   -1.212587
O   3.692205   -1.723774   -1.212871
O   3.827076   -1.193134   1.542036
Mo   -0.000158   0.952009   -0.006968
O   -0.001210   2.341313   1.140093
O   -0.000706   2.380189   -1.105838
O   -1.586609   0.038984   -0.024133
Mo   -3.464873   -0.531849   0.000422
O   -4.488787   0.809179   -0.307154
O   -3.824441   -1.193188   1.542620
O   4.489613   0.808482   -0.309142

[Mo₃O₁₀]²⁻ II 1/2
E = -26056.27
Mo   -1.531156   -0.995233   -0.014438
O   0.106600   -2.003292   0.275105
O   -1.792441   0.910599   0.266159
O   -2.461216   -1.605683   -1.299562
O   -2.297695   -1.490151   1.429354
Mo   1.628447   -0.826856   -0.014286
O   2.617732   -1.333401   -1.299918
O   0.000206   0.000925   -0.979326
Mo   -0.097202   1.822372   -0.014943
O   1.684691   1.094931   0.269357
O   -0.146592   2.725256   1.434028
O   2.443089   -1.240629   1.428886
O   -0.154840   2.938958   -1.294828

[Mo₃O₁₀]²⁻ II 3/4
E = -26053.83
Mo   -1.623458   -0.871481   -0.012839
O   -0.000165   -1.872782   0.322405

O   -1.773906   0.989490   0.349387
O   -2.485534   -1.367891   -1.390029
O   -2.512130   -1.472977   1.313384
Mo   1.623284   -0.871763   -0.012878
O   2.485254   -1.368372   -1.390057
O   -0.000019   0.104397   -1.070398
Mo   0.000154   1.764172   -0.086911
O   1.774103   0.989198   0.349273
O   0.000470   2.220878   1.864477
O   2.511855   -1.473376   1.313352
O   0.000180   3.141564   -1.070501

[Mo₃O₁₀]²⁻ III 1/2
E = -26053.17
Mo   2.106183   0.002672   0.210913
O   0.522007   -1.300484   0.121698
O   0.519502   1.302372   0.108864
O   2.724343   0.008755   1.780126
O   3.220901   -0.723383   -1.207666
Mo   -1.200585   -1.703642   -0.026284
O   -1.535545   -2.492216   -1.508083
O   -1.734038   -2.643640   1.301346
O   -2.079331   -0.002323   -0.010050
Mo   -1.205158   1.700997   -0.027404
O   -1.732876   2.636249   1.305773
O   -1.551597   2.492858   -1.504912
O   3.219326   0.721665   -1.212526

[Mo₃O₁₀]²⁻ III 3/4
E = -26053.27
Mo   1.980531   -0.345190   -0.208296
O   0.903149   1.370625   -0.238580
O   0.391195   -1.409884   0.041945
O   2.625311   -0.608406   -1.742510
O   3.232857   0.268903   1.128995
Mo   -0.802196   1.929420   0.024967
O   -0.930313   2.860468   1.452904
O   -1.333345   2.872753   -1.299179
O   -1.790627   0.410073   0.171473
Mo   -1.542557   -1.583515   0.045366
O   -2.127162   -2.195645   -1.449334
O   -2.073162   -2.409240   1.456149
O   3.014264   -1.163405   1.202446

[Mo₃O₁₀]²⁻ IV 1/2
E = -26052.43
Mo   1.777999   -0.734689   0.018974
O   1.763207   1.125734   0.179475
O   2.872162   -0.999622   -1.260028
O   2.398623   -1.278646   1.506280
Mo   -0.031383   1.795801   -0.001000
O   -0.183718   2.708741   1.436802
O   0.098387   2.708189   -1.441436
O   -1.805539   1.078798   -0.177176
Mo   -1.753078   -0.782627   -0.017094
O   -2.376805   -1.341710   -1.498048
O   -2.811923   -1.088133   1.283388
O   -0.475269   -2.195680   0.477783
O   0.554798   -2.179718   -0.511665

[Mo₃O₁₀]²⁻ IV 3/4
E = -26052.60
Mo   2.226156   -0.895729   0.048084
O   1.364897   0.847536   0.549295
O   3.240430   -0.626109   -1.312415
O   2.916056   -1.824137   1.315820
Mo   0.001373   1.923553   0.000194
O   -0.487265   2.921653   1.294992
O   0.493126   2.915958   -1.297727
O   -1.365423   0.850118   -0.545744
Mo   -2.227542   -0.893603   -0.047904
O   -2.919281   -1.818584   -1.317158
O   -3.240578   -0.626322   1.314001
O   -0.519164   -1.672879   0.497708
O   0.517271   -1.671897   -0.500734

[Mo₃O₉] I 1/2

E = -24006.19
Mo -1.998169 -0.360992   0.000305
O   -1.101423   1.301513   0.000534
O   -0.576471 -1.604561   0.003963
O   -2.968385 -0.534153   1.342009
O   -2.961997 -0.536619 -1.345711
Mo   0.686484   1.910761 -0.000162
O   1.018240   2.836254 -1.343407
O   1.019146   2.834244   1.344171
O   1.678074   0.303098 -0.001730
Mo   1.311662 -1.549874   0.000161
O   1.949011 -2.298992   1.343129
O   1.943926 -2.300228 -1.344553

[Mo3O9] I 3/4
E = -24003.12
Mo   1.036525   1.731750 -0.011463
O   1.794131   0.001739 -0.000357
O   -0.808377   1.386286 -0.102273
O   1.426958   2.577008   1.372854
O   1.541424   2.621543 -1.329435
Mo   1.040039 -1.729771 -0.011014
O   1.543589 -2.616799 -1.331391
O   1.435761 -2.576124   1.371193
O   -0.805886 -1.388263 -0.096365
Mo -2.099184 -0.001891   0.035116
O   -2.781945 -0.000162   1.541573
O   -3.226899 -0.005691 -1.492153

[Mo3O9] II 1/2
E = -24004.82
Mo   0.000017 -0.730093 -0.000206
Mo -2.678512   0.372835   0.000103
O   -1.349004 -0.156403   1.235484
O   -1.350572 -0.158581 -1.235651
O   0.000220 -2.376990   0.000321
O   -3.056383   1.999787 -0.001634
O   -4.078628 -0.537361   0.001642
O   1.349614 -0.157052 -1.235661
O   1.349960 -0.157420   1.235623
Mo   2.678444   0.372944   0.000065
O   4.078048 -0.538099 -0.000540
O   3.057010   1.999767   0.000618

[Mo3O9] II 3/4
E = -24002.58
Mo -0.044427 -0.785595   0.000017
Mo -2.651118   0.395334 -0.000013
O   -1.351778 -0.116292   1.266677
O   -1.352806 -0.117957 -1.266845
O   -0.039316 -2.425251   0.000512
O   -3.023805   2.030178 -0.001098
O   -4.056004 -0.518038   0.001048
O   1.488434 -0.070847 -1.291775
O   1.489118 -0.071919   1.291828
Mo   2.625970   0.342952 -0.000121
O   4.018184 -0.586171 -0.001113
O   3.193241   2.124668   0.001381

[Mo4O12]⁻ I 1/2
E = -32012.34
Mo   0.014304 -2.531924 -0.034201
O   -0.001774 -3.506942   1.338150
O   -0.056102 -3.515687 -1.398605
O   -1.355820 -1.371311 -0.008972
Mo   2.707953 -0.000041   0.024308
Mo -2.810409   0.000051   0.032277
O   -3.755585   0.000016 -1.374151
O   1.626120 -1.502607 -0.071531
O   3.816128 -0.000216 -1.236120
Mo   0.014405   2.531914 -0.034225
O   1.626184   1.502547 -0.071940
O   -0.056094   3.515979 -1.398407
O   -1.355754   1.371341 -0.009137
O   -0.001513   3.506633   1.338342
O   -3.670361   0.000134   1.491877
O   3.571743   0.000108   1.462658

[Mo4O12]⁻ I 3/4
E = -32009.01
Mo -0.784838 -2.410286 -0.236105
O   -0.755936 -3.240638 -1.701880
O   -1.246388 -3.481591   0.977448
O   0.823079 -1.684781   0.122573
Mo -2.483893   0.782806   0.118303
Mo   2.595433 -0.772631   0.206893
O   3.373634 -0.973104   1.698625
O   -2.039953 -0.976155 -0.400873
O   -2.275227 -0.029916   1.840325
Mo   0.697331   2.488203 -0.187232
O   -1.177199   2.057647 -0.235497
O   0.984109   3.579883   1.063012
O   1.593831   0.956799   0.073929
O   1.128377   3.201465 -1.650734
O   3.548979 -1.074274 -1.161552
O   -4.083478   1.202184 -0.110134

[Mo4O12]⁻ II 1/2
E = -32011.98
Mo   2.615069 -0.209861   0.474086
O   0.200229   3.685640   0.074364
O   3.206278 -0.132037   2.045990
O   3.926130 -0.240953 -0.578565
O   1.589429 -1.761832   0.256627
Mo -0.092555   2.126663 -0.469881
Mo -0.138570 -2.002346 -0.576056
O   -0.274370 -2.838510 -2.018094
O   1.521405   1.229068   0.114464
O   -0.675646 -0.231620 -0.775589
O   -0.207767   2.264144 -2.136599
Mo -2.309881 -0.163839   0.577642
O   -1.698651   1.690139   0.354421
O   -3.809570 -0.240029 -0.169924
O   -1.628470 -2.026035   0.559435
O   -2.537831 -0.088716   2.243064

[Mo4O12]⁻ II 3/4
E = -32008.87
Mo   0.075366 -2.120026   0.187222
O   -2.846689 -0.000611 -1.844829
O   -0.109429 -3.136718 -1.133416
O   0.197189 -3.065545   1.567429
O   1.913609 -1.582426 -0.051219
Mo -2.716523 -0.000329 -0.165187
Mo   2.948841   0.000379 -0.197276
O   4.604796   0.000305 -0.418946
O   -1.790800 -1.513558   0.378208
O   0.084707   0.000329   0.186329
O   -4.266924 -0.000663   0.493168
Mo   0.074662   2.120056   0.187155
O   -1.791888   1.513240   0.377966
O   0.194893   3.066995   1.566546
O   1.913188   1.583095 -0.049019
O   -0.109970   3.135138 -1.134764

[Mo4O12]⁻ III 1/2
E = -32011.54
Mo -0.541462 -0.012147 -1.008767
Mo -4.057987 -0.000337   0.399893
Mo   2.357311   1.724408   0.272656
Mo   2.372960 -1.714280   0.281039
O   3.276104   2.513167 -0.889744
O   -2.114634 -0.013400 -0.177793
O   3.296443 -2.498227 -0.880937
O   0.660693   1.367915 -0.347542
O   -4.127906 -0.244551   2.087450
O   -4.712929   1.519515 -0.021596
O   0.671057 -1.378296 -0.336195
O   -4.860340 -1.260115 -0.428405
O   -0.520360 -0.017482 -2.674380
O   2.322792   2.669628   1.657721
O   2.351268 -2.654890   1.669495
O   3.070986   0.009125   0.631618

[Mo4O12]⁻ III 3/4

E = -32008.90
Mo -0.357291 -0.277984   1.337797
Mo -3.620383 -0.126140 -0.492788
Mo   1.750730   1.929551 -0.265740
Mo   2.525542 -1.621028 -0.442209
O   1.397767   2.654015 -1.748081
O   -2.013076 -0.425257   0.313638
O   2.086795 -2.258469 -1.951116
O   0.161634   1.415933   0.604796
O   -4.439448   1.169561   0.210320
O   -3.459279   0.106732 -2.154209
O   -0.631064 -0.128994   2.979724
O   -4.747343 -1.638177 -0.283887
O   0.957909 -1.386973   0.790691
O   2.636276   3.013420   0.676062
O   3.826654 -2.383309   0.333305
O   2.655536   0.363424 -0.490809

[Mo4O12]⁻ IV 1/2
E = -32011.23
Mo   0.490117   0.047199   0.812420
Mo   4.029005 -0.082038 -0.344599
Mo -2.259525   1.826382 -0.246474
Mo -2.484757 -1.840855 -0.215991
O   -3.163665   2.714589   0.856432
O   1.997540 -0.066545 -0.095339
O   -3.391229 -2.583880   1.000609
O   -2.845978   0.121661 -0.370063
O   -0.444912   1.580637   0.379059
O   4.370114   0.083749 -2.006590
O   4.658187   1.233249   0.542033
O   -0.617735 -1.272295   0.402793
O   4.621590 -1.564199   0.257925
O   0.784193   0.027445   2.451174
O   -2.288370   2.606601 -1.733070
O   -2.497650 -2.622122 -1.713083

[Mo4O12]⁻ V 1/2
E = -32010.81
Mo -0.405729 -1.508618 -0.030258
Mo -3.880338 -0.014998 -0.023627
Mo   0.869982   2.009412   0.017311
Mo   3.176090 -0.513609   0.011415
O   2.626663   1.294993   0.040600
O   0.622818   2.948134 -1.358413
O   -2.286578 -0.905968 -0.089943
O   4.113405 -0.840796 -1.349239
O   1.600337 -1.408690 -0.033458
O   -0.164091   0.521564   0.010269
O   -4.542406   0.421288   1.470885
O   -4.326289   1.016663 -1.287427
O   0.587103   2.957949   1.379566
O   4.072639 -0.897793   1.384463
O   -0.510236 -2.500774 -1.374370
O   -0.533387 -2.460553   1.339154

[Mo4O12]⁻ V 3/4
E = -32008.63
Mo -0.530270 -1.023159   0.846965
Mo -3.885763 -0.007486 -0.452214
Mo   1.359404   2.031438   0.162829
Mo   2.861332 -0.980243 -0.526215
O   2.782126   0.906957 -0.384033
O   0.933786   3.099337 -1.066325
O   -2.382855 -0.534940   0.468406
O   2.981745 -1.444655 -2.139754
O   1.295810 -1.527917   0.216448
O   0.003226   0.876589   0.533923
O   -5.184706   0.684249   0.385040
O   -3.825046   0.394623 -2.093616
O   1.797433   2.922786   1.522266
O   4.186298 -1.595254   0.311034
O   -1.040507 -2.449443 -0.366982
O   -0.522001 -1.440158   2.448929

[CH2Mo2O7] I 1/2
E = -19118.90
O   2.631367   0.320831   1.374228

```
O     2.631771    0.317697   -1.374380
Mo    1.692611    0.117307   -0.000007
O     0.556915   -1.388614    0.001503
O    -2.582871    1.235659   -0.000096
O    -1.936715   -1.035348    1.376676
Mo   -1.148667   -0.506704    0.000073
O     0.085329    1.077477   -0.001032
O    -1.936208   -1.036439   -1.376453
C    -2.331364    2.419307   -0.000706
H    -3.160811    3.135575   -0.000063
H    -1.293343    2.773151   -0.002042
```

$[CH_2Mo_2O_7]$ I 3/4
E = -19115.67
```
O     2.535254    0.920050    1.207404
O     2.732214   -0.223620   -1.280159
Mo    1.698939    0.134088   -0.012548
O     0.565681   -1.277052    0.550536
O    -2.537352    1.145270   -0.688409
O    -2.042690   -0.396463    1.474825
Mo   -1.120788   -0.515769    0.077259
O     0.082573    0.950836   -0.505809
O    -1.847388   -1.537955   -1.023995
C    -2.585237    2.343415   -0.178249
H    -2.994795    2.381733    0.839978
H    -1.682435    2.940708   -0.363495
```

$[H_3Mo_4O_{13}]^-$ I 1/2
E = -34108.96
```
Mo   -2.756326    0.129384    0.375994
O    -0.319790   -3.392618    0.780053
O    -2.828348    0.164474    2.065706
O    -4.321789    0.195198   -0.256501
O    -1.721197    1.491802   -0.256228
Mo   -0.054710   -1.979996   -0.497702
Mo    0.185896    1.949158   -0.487216
O     0.379646    2.475285   -2.046611
O    -0.092205    3.438239    0.682013
O    -1.888233   -1.422728   -0.160264
O     0.559771   -0.075399   -0.324682
O    -0.010010   -2.554452   -2.050068
Mo    2.504018    0.010125    0.441042
O     1.872671   -2.011505    0.211071
O     3.862237   -0.116967   -0.540071
O     1.986332    1.788070    0.353204
O     2.956947   -0.249903    2.044270
H     0.728793    3.739523    1.081186
H    -1.227925   -3.469839    1.083589
H     2.097971   -2.669809    0.871094
```

$[H_3Mo_4O_{13}]^-$ I 3/4
E = -34108.97
```
Mo   -2.760560    0.128578    0.373564
O    -0.316733   -3.386294    0.789733
O    -2.837585    0.163798    2.063108
O    -4.323754    0.193731   -0.264755
O    -1.720947    1.488053   -0.257301
Mo   -0.053995   -1.979813   -0.495542
Mo    0.184890    1.948751   -0.484912
O     0.380739    2.481437   -2.041784
O    -0.093392    3.432976    0.690356
O    -1.886888   -1.420466   -0.161522
O     0.564299   -0.075350   -0.330573
O    -0.008369   -2.561483   -2.045177
Mo    2.506822    0.010980    0.438112
O     1.874616   -2.010261    0.211425
O     3.863281   -0.112898   -0.545820
O     1.985918    1.788142    0.355832
O     2.963594   -0.250675    2.040092
H     0.727161    3.729726    1.093849
H    -1.224973   -3.465425    1.092340
H     2.098954   -2.666789    0.873574
```

$[H_3Mo_4O_{13}]^-$ II 1/2
E = -34108.41
```
Mo   -2.568140    0.307934   -0.006858
O    -0.600732   -3.301309    1.233773
O    -2.925607    0.379349    1.637274
```

$[H_3Mo_4O_{13}]^-$ II 1/2
E = -34108.52
```
Mo    2.91798   -0.00413     0.25265
O     0.03658    3.29584     1.18311
O     3.90132   -0.06852     1.62724
O     3.90597    0.21098    -1.10003
O     1.93068   -1.51686     0.09929
Mo   -0.00851    1.72940    -0.41839
Mo   -0.00527   -1.88754    -0.29914
O    -0.05106   -2.95988    -1.56066
O     0.03835   -2.78344     1.40468
O     1.73497    1.45623     0.43309
O    -0.04396   -0.05164    -0.97292
O    -0.03007    2.65783    -1.79099
Mo   -2.89158    0.01401     0.25846
O    -1.72454    1.47256     0.50581
O    -3.74593    0.19314    -1.18636
O    -1.92307   -1.52887     0.26315
O    -4.00066   -0.00906     1.53439
H    -0.85308   -2.91112     1.74130
H     0.83469    3.08780     1.68577
H    -0.73987    3.10374     1.72388
```

$[H_3Mo_4O_{13}]^-$ III 3/4
E = -34108.43
```
Mo   -2.568521    0.305440   -0.004861
O    -0.593409   -3.304064    1.230854
O    -2.924547    0.376914    1.639592
O    -4.012999    0.591713   -0.813823
O    -1.430617    1.727456   -0.569862
Mo   -0.319024   -2.171026    0.053773
Mo    0.324416    2.258881   -0.152017
O     0.522747    3.754189   -0.886448
O     0.558142    2.484773    1.511509
O    -2.044430   -1.383275   -0.549755
O    -0.186906   -0.259855    0.815755
O    -0.284374   -2.985926   -1.682147
Mo    2.589994   -0.374539   -0.068820
O     1.653416   -1.937172   -0.167449
O     4.137741   -0.408745   -0.701669
O     1.658509    1.171855   -0.818811
O     2.680754    0.321846    1.701313
H    -0.039317   -0.174325    1.761153
H     2.183597    1.127792    1.897569
H    -1.143255   -2.938707   -2.111786
```

$[H_3Mo_4O_{13}]^-$ IV 1/2
E = -34108.37
```
Mo   -0.315092    2.146340   -0.032304
O    -2.900356   -0.250761    1.713569
O    -0.570830    3.273653    1.153989
O    -0.245720    2.997633   -1.748409
O     1.648714    1.879607   -0.298031
Mo   -2.595312   -0.280950    0.057236
Mo    2.551400    0.312528    0.039433
O     4.114014    0.482152   -0.561647
O     2.653697   -0.027699    1.706083
O    -2.041565    1.353424   -0.590533
O    -0.176894    0.236131    0.766600
O    -4.069756   -0.579528   -0.690883
Mo    0.283064   -2.275810   -0.245645
O    -1.511397   -1.784140   -0.446133
```

```
O     0.630857   -3.729874   -0.977551
O     1.736460   -1.120214   -0.817332
O     0.861602   -2.415757    1.548175
H    -0.036398    0.186197    1.716220
H     1.539594   -1.790879    1.849324
H     0.655665    3.199124   -2.014954
```

$[H_3Mo_4O_{13}]^-$ IV 3/4
E = -34108.37
```
Mo    0.313740   -2.146103   -0.032267
O     2.901694    0.248845    1.713040
O     0.568702   -3.272655    1.154935
O     0.244199   -2.998016   -1.748155
O    -1.649547   -1.878297   -0.297808
Mo    2.596123    0.279654    0.056826
Mo   -2.552296   -0.311082    0.039068
O    -4.114704   -0.480126   -0.562825
O    -2.655014    0.029272    1.705613
O     2.040445   -1.353915   -0.591053
O     0.176663   -0.235756    0.766725
O     4.070445    0.577206   -0.691988
Mo   -0.281577    2.275146   -0.245031
O     1.512941    1.783972   -0.445056
O    -0.628513    3.729457   -0.976849
O    -1.736141    1.121269   -0.816981
O    -0.860138    2.415467    1.548827
H     0.036508   -0.185652    1.716375
H    -1.539131    1.791580    1.849682
H    -0.657254   -3.199566   -2.014495
```

$[H_3Mo_4O_{13}]^-$ V 1/2
E = -34107.92
```
Mo    2.272940    0.194994    0.570614
O    -0.483991    3.373980    0.415084
O     1.342138   -0.576361    2.131536
O     3.852822    0.302002    1.102210
O     2.073568   -1.373554   -0.460330
Mo   -0.199845    2.052229   -0.583590
Mo    0.220121   -1.794205   -0.879370
O     0.196636   -2.185167   -2.495182
O     0.313352   -3.309485    0.314289
O     1.683566    1.815767   -0.208602
O    -0.039099    0.048998   -0.215534
O    -0.266189    2.544366   -2.184081
Mo   -2.218827   -0.247864    0.587730
O    -2.116565    1.331489   -0.283643
O    -3.834900   -0.410435    0.980108
O    -1.808253   -1.732204   -0.329939
O    -1.359635   -0.220746    2.224107
H    -0.558790   -3.667955    0.502278
H     1.374280   -1.540410    2.132281
H    -0.367417   -0.277301    2.239095
```

$[H_3Mo_4O_{13}]^-$ V 3/4
E = -34107.95
```
Mo    2.297218   -0.195595    0.566875
O     0.050269    3.381041    0.195872
O     1.305610    0.182212    2.234790
O     3.910663   -0.105657    1.022178
O     1.806784   -1.761795   -0.261925
Mo    0.177136    1.962131   -0.698079
Mo   -0.124854   -1.875442   -0.774429
O    -0.153879   -2.167124   -2.408058
O    -0.295922   -3.467513    0.284293
O     1.967627    1.472079   -0.377562
O    -0.016513   -0.015839   -0.121902
O     0.110640    2.341638   -2.332211
Mo   -2.245982    0.128514    0.603261
O    -1.883345    1.567712   -0.382588
O    -3.873544    0.230866    0.971606
O    -2.066552   -1.474913   -0.223357
O    -1.418349    0.162776    2.258034
H    -1.208513   -3.666488    0.510857
H     1.739781   -0.098189    3.041658
H    -0.426918    0.177278    2.273694
```

$[H_3Mo_4O_{13}]^-$ VI 1/2
E = -34107.74

Mo   0.787264 -2.158598 -0.226212
O   2.586863   0.548277   1.693269
O   1.476442 -2.295430   1.530033
O   1.408970 -3.444131 -1.086390
O   -1.096560 -2.211820 -0.329999
Mo   2.338270   0.813102   0.034558
Mo -2.338284 -0.813210   0.034505
O   -3.779379 -1.411447 -0.587219
O   -2.586826 -0.548355   1.693224
O   2.064683 -0.784779 -0.835940
O   0.000174 -0.000041   0.584686
O   3.779339   1.411463 -0.587103
Mo -0.787278   2.158736 -0.226229
O   1.096473   2.211597 -0.330475
O   -1.409055   3.444567 -1.085897
O   -2.064607   0.784769 -0.835778
O   -1.476378   2.295199   1.530107
H   -0.000256   0.000146   1.546510
H   -1.964832   1.530043   1.867598
H   1.965157 -1.530433   1.867616

[H₃Mo₄O₁₃]⁻ VI 3/4
E = -34107.73
Mo   0.801133 -2.155661 -0.221348
O   2.575973   0.572259   1.690960
O   1.479682 -2.275128   1.540493
O   1.437879 -3.443998 -1.066799
O   -1.081690 -2.207225 -0.355613
Mo   2.334584   0.824538   0.029244
Mo -2.334563 -0.824509   0.029203
O   -3.774409 -1.430770 -0.587461
O   -2.576303 -0.572275   1.690861
O   2.075328 -0.778321 -0.835389
O   -0.000214   0.000016   0.581273
O   3.774554   1.430767 -0.587148
Mo -0.801146   2.155652 -0.221234
O   1.081711   2.207150 -0.355802
O   -1.437993   3.443980 -1.066640
O   -2.075258   0.778350 -0.835478
O   -1.479444   2.275116   1.540729
H   0.000279   0.000136   1.543063
H   -1.955998   1.502929   1.879579
H   1.956882 -1.503277   1.879186

[H₃Mo₄O₁₃]⁻ VII 1/2
E = -34107.71
Mo -2.384186   0.408112 -0.093192
O   0.021048 -2.579517   1.784731
O   -3.006549   0.579192   1.454139
O   -3.669178   0.609465 -1.157926
O   -1.376796   1.965712 -0.560993
Mo -0.459169 -2.376642 -0.046342
Mo   0.391879   2.353291 -0.025781
O   0.648875   3.897925 -0.631718
O   0.602606   2.449289   1.654703
O   -2.046796 -1.463268 -0.440391
O   -0.237697 -0.025576   0.581712
O   -0.531978 -3.913549 -0.691385
Mo   2.397593 -0.420910 -0.238298
O   1.348609 -1.878199 -0.629356
O   3.965681 -0.518459 -0.804754
O   1.741877   1.325613 -0.779558
O   2.483887 -0.087105   1.643202
H   -0.208015   0.005224   1.542329
H   2.107437   0.747260   1.953982
H   0.814934 -2.114431   2.076221

[H₃Mo₄O₁₃]⁻ VII 3/4
E = -34107.73
Mo   2.337590 -0.612062 -0.089658
O   0.226953   2.585544   1.783342
O   2.935450 -0.824780   1.462171
O   3.605766 -0.932026 -1.145668
O   1.201523 -2.079072 -0.556794
Mo   0.664690   2.331927 -0.051199
Mo -0.595240 -2.314753 -0.029909
O   -0.984629 -3.830165 -0.638771
O   -0.819262 -2.392683   1.649969
O   2.160759   1.276580 -0.453684

O   0.234490   0.009091   0.581328
O   0.870323   3.851253 -0.708582
Mo -2.349566   0.624419 -0.234966
O   -1.186071   2.000131 -0.599754
O   -3.903179   0.865488 -0.797765
O   -1.845413 -1.170329 -0.786236
O   -2.474737   0.286678   1.643979
H   0.196014 -0.023154   1.541572
H   -2.180648 -0.582936   1.947041
H   -0.605047   2.200085   2.083838

[H₃Mo₄O₁₃]⁻ VIII 1/2
E = -34107.39
Mo -0.034791   2.187648 -0.143174
O   -2.726065   0.461750   1.722045
O   -0.089552   2.384364   1.542054
O   -0.071170   3.713340 -0.846879
O   1.535231   1.367470 -0.675734
Mo -2.704884 -0.126597 -0.077125
Mo   2.702184 -0.061222 -0.069646
O   4.195546   0.215400 -0.774195
O   2.749738   0.450896   1.759116
O   -1.537257   1.261679 -0.732563
O   0.075088 -0.519062   0.932722
O   -4.199899   0.113298 -0.784071
Mo -0.005801 -2.018568   0.077375
O   -1.935051 -1.802479 -0.244904
O   0.039146 -2.442912 -1.803018
O   1.979914 -1.721974 -0.342341
O   0.098097 -3.321224   1.124087
H   1.973228   0.920409   2.088234
H   0.918482 -2.424857 -2.185657
H   -1.982853   1.006474   2.014815

[H₃Mo₄O₁₃]⁻ VIII 3/4
E = -34107.36
Mo -0.037591   2.191786 -0.153181
O   -2.699241   0.451378   1.734244
O   -0.093404   2.397057   1.530553
O   -0.073874   3.714828 -0.862471
O   1.534600   1.372334 -0.677029
Mo -2.700147 -0.129967 -0.067345
Mo   2.697224 -0.061220 -0.060325
O   4.198265   0.221906 -0.744855
O   2.727579   0.450235   1.769051
O   -1.540431   1.264835 -0.735925
O   0.073203 -0.517903   0.921975
O   -4.204550   0.113840 -0.752137
Mo -0.002483 -2.020371   0.068806
O   -1.929522 -1.802311 -0.265768
O   0.043152 -2.452851 -1.810693
O   1.979338 -1.722290 -0.352057
O   0.095473 -3.321676   1.117809
H   1.948213   0.918866   2.092365
H   0.923592 -2.434421 -2.190887
H   -1.950652   0.990926   2.022844

[H₃Mo₄O₁₃]⁻ XI 1/2
E = -34106.97
Mo -3.212776 -1.553131 -0.201889
O   -3.960799 -0.640899   1.272245
O   -1.815548 -2.636222   0.330623
O   -4.410788 -2.237559 -1.137088
Mo -1.943783   1.717493 -0.509902
Mo   0.949625   0.416764   1.078755
O   0.076739 -1.032519   1.352741
O   1.288966   1.162013   2.545854
O   2.448834   0.102960   0.179319
O   -0.166224   1.554337   0.008939
O   -2.885064   2.188369   0.813838
O   -2.121336   2.847933 -1.736424
O   -2.536487   0.054132 -1.050425
Mo   4.239053 -0.388167 -0.604927
O   4.109539 -1.256220 -2.062687
O   5.342777   0.906442 -0.623906
O   4.583568 -1.579029   0.890466
H   -1.039349 -2.240760   0.792795
H   3.907223 -1.648872   1.567774
H   -3.850261   0.315458   1.365889

[H₃Mo₄O₁₃]⁻ IX 3/4
E = -34106.97
Mo -3.210210 -1.555398 -0.199111
O   -3.961951 -0.640865   1.271569
O   -1.812738 -2.635541   0.339006
O   -4.405697 -2.243708 -1.134726
Mo -1.944724   1.715774 -0.512637
Mo   0.949507   0.423216   1.076934
O   0.074207 -1.023025   1.359301
O   1.292220   1.175147   2.539819
O   2.446815   0.102435   0.176693
O   -0.165588   1.557170   0.002441
O   -2.884348   2.185277   0.812842
O   -2.127118   2.845194 -1.739362
O   -2.534986   0.050766 -1.050758
Mo   4.237831 -0.391142 -0.604086
O   4.109904 -1.263694 -2.059307
O   5.341280   0.903604 -0.625949
O   4.581012 -1.577248   0.895408
H   -1.038231 -2.237371   0.801500
H   3.903858 -1.645118   1.572114
H   -3.850680   0.315522   1.364354

[H₃Mo₄O₁₃]⁻ X 1/2
E = -34106.93
Mo -3.379752 -1.334980 -0.277147
O   -3.895938 -0.945952   1.287943
O   -2.085022 -2.662160 -0.249439
O   -4.687879 -1.873556 -1.178456
Mo -1.850195   1.828245 -0.303936
Mo   0.961383   0.014444   0.854779
O   -0.077186 -1.339115   0.994208
O   1.528104   0.463318   2.375319
O   2.326601 -0.343820 -0.219213
O   -0.070801   1.450685   0.105484
O   -2.861394   1.876448   1.282624
O   -2.240113   3.101473 -1.303513
O   -2.659046   0.222272 -0.988109
Mo   4.341423 -0.280424 -0.544709
O   4.890336 -1.557111 -1.526934
O   4.895473   1.273478 -0.967715
O   4.645993 -0.633338   1.339984
H   -1.278377 -2.441076   0.265797
H   3.931396 -0.468578   1.961405
H   -3.386081   1.102747   1.537843

[H₃Mo₄O₁₃]⁻ X 3/4
E = -34106.94
Mo -3.361746 -1.352885 -0.265721
O   -3.889957 -0.952441   1.292443
O   -2.051904 -2.664704 -0.215826
O   -4.659029 -1.917111 -1.166912
Mo -1.861154   1.823661 -0.321650
Mo   0.961082   0.051882   0.856200
O   -0.079301 -1.295440   1.038272
O   1.541398   0.538337   2.360054
O   2.317573 -0.335701 -0.219531
O   -0.073946   1.470005   0.076702
O   -2.862711   1.867881   1.271029
O   -2.271570   3.087285 -1.325518
O   -2.655154   0.203946 -0.991596
Mo   4.331954 -0.296552 -0.543837
O   4.869794 -1.596271 -1.501510
O   4.900370   1.243644 -0.996400
O   4.634679 -0.616357   1.346837
H   -1.253919 -2.428353   0.305868
H   3.926182 -0.423881   1.967322
H   -3.379898   1.091194   1.532733

[H₃Mo₄O₁₃]⁻ XI 1/2
E = -34106.65
Mo -0.443980 -0.955249 -0.294181
Mo -4.158335 -0.045885   0.197542
Mo   1.705280   2.000421 -0.051583
Mo   3.232646 -1.067374   0.246096
O   3.178009   0.815702 -0.032440
O   1.924862   3.181641 -1.234504

O   -2.152528   0.195519   -0.342652
O   -5.046563   -1.188820   -0.695284
O   4.122594   -1.858703   -1.227362
O   1.412326   -1.382879   0.102824
O   0.258572   0.960604   -0.410655
O   -4.519953   0.011322   1.857667
O   -4.185243   1.733006   -0.619628
O   1.557446   2.751029   1.449892
O   3.749661   -1.344676   1.807571
O   -0.629133   -1.733310   -1.762617
O   -1.158636   -1.885509   0.904359
H   -2.029300   1.077761   -0.714168
H   4.850938   -2.476099   -1.151780
H   -5.008601   2.218593   -0.622107

[H₃Mo₄O₁₃]⁻ XI 3/4
E = -34106.65
Mo   -0.443608   -0.956443   -0.286808
Mo   -4.159762   -0.046084   0.193530
Mo   1.704017   2.000800   -0.050527
Mo   3.234941   -1.066365   0.241752
O   3.177793   0.817324   -0.031833
O   1.922368   3.182628   -1.233076
O   -2.151488   0.196071   -0.335230
O   -5.042377   -1.189926   -0.703784
O   4.118621   -1.853429   -1.237567
O   1.413621   -1.381837   0.105899
O   0.258461   0.959477   -0.409702
O   -4.529943   0.011762   1.851839
O   -4.182624   1.732868   -0.623909
O   1.555495   2.750749   1.451192
O   3.757407   -1.349082   1.800432
O   -0.632729   -1.738888   -1.752460
O   -1.153681   -1.883593   0.916613
H   -2.026214   1.078788   -0.704944
H   4.847452   -2.470845   -1.167016
H   -5.005677   2.218938   -0.629115

[H₃Mo₄O₁₃]⁻ XII 1/2
E = -34106.38
Mo   0.484311   -0.891226   0.082325
Mo   4.232621   0.066925   -0.039492
Mo   -1.742659   2.004383   -0.063474
Mo   -3.195571   -1.163524   -0.146548
O   -3.149452   0.710965   -0.077110
O   -2.153459   2.699371   1.666454
O   2.211565   0.346825   0.152692
O   4.704566   -1.321559   0.830019
O   -3.872926   -1.785501   1.516010
O   -1.346279   -1.403466   -0.191276
O   -0.189304   1.026283   -0.069269
O   4.646155   -0.076878   -1.687985
O   4.917821   1.484905   0.629382
O   -2.014141   3.099174   -1.299407
O   -3.908125   -1.619323   -1.585649
O   0.768068   -1.560422   1.588644
O   1.229855   -1.821036   -1.094457
H   1.930529   1.262963   0.223015
H   -4.542099   -2.465516   1.600451
H   -2.848700   2.272417   2.174044

[H₃Mo₄O₁₃]⁻ XII 3/4
E = -34106.40
Mo   0.491551   -0.917917   0.149769
Mo   4.224829   0.085271   -0.074139
Mo   -1.752619   1.983629   -0.051165
Mo   -3.180114   -1.133311   -0.178968
O   -3.199271   0.730508   0.094105
O   -2.046531   2.765534   1.663279
O   2.198964   0.340881   0.119012
O   4.739579   -1.206149   0.912702
O   -4.016622   -1.908108   1.341722
O   -1.343772   -1.433304   -0.072487
O   -0.220106   0.974628   -0.106426
O   4.624746   -0.200374   -1.706856
O   4.881179   1.574577   0.453835
O   -2.086285   3.008518   -1.331095
O   -3.726996   -1.452111   -1.723763
O   0.794565   -1.499738   1.688467

O   1.219464   -1.916071   -0.982325
H   1.906324   1.256041   0.133907
H   -4.701902   -2.575255   1.290440
H   -2.668912   2.346691   2.263431

[CH₃Mo₂O₇]⁻ I 1/2
E = -19139.21
O   2.763950   -0.953613   1.128119
O   2.696801   0.012370   -1.506167
Mo   1.868551   -0.049637   -0.012042
O   0.093346   -0.881198   -0.291814
O   -1.907917   1.372699   -0.609798
O   -2.022408   -0.445298   1.624373
Mo   -1.642065   -0.379965   -0.019426
O   1.557426   1.539110   0.573720
O   -2.688676   -1.392354   -0.873184
C   -1.525405   2.565620   0.051446
H   -1.787054   3.406928   -0.593858
H   -0.448749   2.571299   0.245418
H   -2.064327   2.657639   0.999429

[CH₃Mo₂O₇]⁻ I 3/4
E = -19136.05
O   2.544417   -0.767354   1.457286
O   2.550142   -0.979543   -1.335847
Mo   1.923131   -0.148878   0.004787
O   0.136602   -0.148186   0.002640
O   -2.040244   1.624380   -0.029331
O   -2.498326   -0.948332   1.455802
Mo   -1.875356   -0.313647   -0.001300
O   2.586601   1.635678   -0.138664
O   -2.485529   -0.992782   -1.443774
C   -1.022535   2.576626   0.008337
H   -1.467832   3.578924   -0.009095
H   -0.342102   2.497212   -0.849691
H   -0.410692   2.499264   0.917395

[CH₃Mo₂O₇]⁻ II 1/2
E = -19138.27
Mo   1.847627   -0.148262   0.000113
Mo   -1.857442   -0.122558   0.000276
O   0.007725   0.777786   0.009727
O   -2.055682   -1.071969   1.399526
O   -2.935110   1.208553   0.007815
O   2.657808   0.403164   -1.399693
O   2.661956   0.370372   1.410181
O   -2.054714   -1.054323   -1.410895
O   1.621120   -1.835742   -0.019343
C   0.121477   2.215056   0.001183
H   -0.874298   2.651116   0.049465
H   0.623082   2.523559   -0.917440
H   0.709756   2.526669   0.866008

[CH₃Mo₂O₇]⁻ II 3/4
E = -19135.40
Mo   1.807279   -0.161815   0.000289
Mo   -1.793987   -0.204680   -0.000595
O   0.041217   0.851795   0.000206
O   -2.184700   -0.977895   1.463854
O   -2.736942   1.534795   0.009421
O   2.659905   0.310816   -1.404309
O   2.659539   0.307208   1.406358
O   -2.186283   -0.967302   -1.469989
O   1.448811   -1.831779   -0.001935
C   0.190118   2.279440   -0.001642
H   -0.798730   2.732999   0.013615
H   0.730264   2.581199   -0.901584
H   0.757114   2.580852   0.881842

[H₂] I 1/2
E = -31.73
H   0.000000   0.000000   0.372080
H   -0.000000   -0.000000   -0.372080

[MoO₃] I 1/2
E = -7999.00

Mo   0.000086   -0.000071   -0.184904
O   1.605651   -0.148590   0.323684
O   -0.674239   1.464910   0.323470
O   -0.931864   -1.315945   0.323591

[MoO₃] I 3/4
E = -7997.30
Mo   0.023016   0.000076   -0.133289
O   0.861911   1.403792   0.241510
O   -1.842379   0.001059   0.216870
O   0.859634   -1.405252   0.241385

[MoO₃] I 5/6
E = -7994.59
Mo   0.000129   0.061056   0.000016
O   1.611032   -1.020115   -0.000025
O   0.004314   1.711938   -0.000032
O   -1.616022   -1.012369   -0.000025

[MoO₃] I 7/8
E = -7991.01
Mo   -0.000141   -0.006658   -0.023175
O   1.702387   -0.960741   0.040935
O   0.014687   1.932091   0.039784
O   -1.716336   -0.936397   0.040953

[Mo₃O₁₁]⁻ I 1/2
E = -28100.94
Mo   1.267652   1.715875   0.065650
O   2.071783   -0.027806   -0.131065
O   -2.112963   0.943745   -1.709244
O   -0.431386   1.313609   0.353625
O   1.468105   2.625452   -1.347221
Mo   1.277416   -1.711395   -0.096724
O   1.683790   -2.586194   -1.476302
O   1.793500   -2.584964   1.246143
O   -0.537934   -1.404714   -0.023424
Mo   -2.048170   -0.198193   0.213075
O   -2.362831   -0.018626   1.849191
O   -3.257820   -1.220613   -0.334379
O   1.938343   2.559556   1.370400
O   -2.861305   1.417552   -0.753227

[Mo₃O₁₁]⁻ II 1/2
E = -28098.56
Mo   3.348089   -0.515628   -0.007260
O   1.801885   0.426080   -0.268516
O   -2.945501   -1.238324   -1.536055
O   3.364999   -2.135650   -0.976066
O   3.565048   -0.919683   1.611451
Mo   0.033137   1.181738   -0.007871
O   0.281496   2.525971   0.967395
O   -0.461230   1.647040   -1.541581
O   -1.665126   0.419208   0.759551
Mo   -3.328903   -0.563584   -0.018365
O   -4.517858   0.651206   -0.170098
O   -3.823136   -1.767598   1.083157
O   4.676047   0.365065   -0.555366
O   -0.551320   -0.511577   0.801977

[Mo₃O₁₁]⁻ II 3/4
E = -28098.27
Mo   2.075950   -0.662453   0.019541
O   1.309739   1.118600   -0.243506
O   -2.348105   -1.601968   -1.541833
O   1.620867   -2.354946   -0.693230
O   2.733234   -0.803010   1.554623
Mo   -0.255498   2.070913   -0.022424
O   -0.352783   2.700912   1.535041
O   -0.317555   3.354805   -1.109509
O   -1.670508   0.900593   -0.336701
Mo   -1.867426   -0.999540   -0.042491
O   -3.106424   -1.145110   1.092182
O   -0.982258   -2.796573   0.503378
O   3.381385   -0.537846   -1.053266
O   -0.020977   -0.982284   0.531037

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[Mo₂O₇] I 1/2
E = -18046.39
O   -2.329493   -1.504192   0.569812
O   -2.457411   0.176935   -1.581086
Mo -1.869994   -0.010080   -0.025200
O   -2.703361   1.294258   1.037573
O   2.329612   -1.505488   -0.566309
O   2.457084   0.180634   1.580782
Mo 1.870002   -0.010048   0.025200
O   0.000003   0.171613   -0.000385
O   2.703525   1.291908   -1.040385

[Mo₂O₇] I 3/4
E = -18046.39
O   -2.328898   -1.503865   0.571068
O   -2.458340   0.175695   -1.580913
Mo -1.870015   -0.010103   -0.025238
O   -2.702946   1.294980   1.036995
O   2.330080   -1.505442   -0.566037
O   2.456224   0.180661   1.581035
Mo 1.870004   -0.010031   0.025150
O   0.000010   0.171785   -0.001468
O   2.703928   1.291890   -1.040218

[Mo₂O₇] II 1/2
E = -18045.68
O   -2.510212   1.372408   0.079852
O   -0.010276   0.000078   1.262220
Mo -1.554357   0.000073   0.038759
O   1.669735   1.577744   -0.844830
O   -2.509323   -1.372848   0.079248
Mo 1.391497   0.000014   0.071126
O   2.815562   0.000380   0.954491
O   1.669011   -1.578633   -0.843878
O   -0.269479   0.000415   -1.263996

[Mo₂O₇] II 3/4
E = -18045.72
O   -2.511893   1.372486   0.082675
O   -0.006256   0.000407   1.259538
Mo -1.555773   0.000037   0.039326
O   2.824141   -0.000951   0.951993
O   -2.511614   -1.372711   0.082930
Mo 1.392500   0.000022   0.067819
O   1.670401   -1.577160   -0.836213
O   1.671720   1.577520   -0.835469
O   -0.279315   0.000100   -1.267965

[MoO₄] I 1/2
E = -10044.06
Mo -0.000361   0.023537   0.000105
O   -1.572212   -1.017343   0.001190
O   -0.003590   0.972678   -1.382438
O   0.015455   0.952088   1.396489
O   1.562241   -1.030992   -0.015794

[MoO₄] I 3/4
E = -10044.07
Mo -0.000060   -0.020699   0.000151
O   1.566968   1.020028   -0.025617
O   -1.571819   1.014048   -0.000340
O   -0.011646   -0.980953   -1.376799
O   0.016814   -0.944451   1.401962

[Mo₂O₅] I 1/2
E = -13953.61
Mo -1.523799   0.176138   0.000024
Mo 1.278647   -0.043491   -0.000007
O   -0.105546   0.195053   -1.286286
O   -0.105466   0.195831   1.286260
O   -2.940131   -0.685200   -0.000090
O   2.447385   1.156817   -0.000385
O   1.990803   -1.558901   0.000408

[Mo₂O₅] I 3/4
E = -13953.08
Mo -1.526531   0.224354   -0.000453
Mo 1.333380   -0.049948   0.000094
O   -0.330429   0.097448   -1.312670
O   -0.330509   0.102064   1.312155
O   -2.784541   -0.873168   0.001875
O   2.343519   1.293372   -0.001063
O   2.116003   -1.535346   0.001587

[Mo₂O₅] I 5/6
E = -13950.49
Mo -1.518325   -0.195464   -0.000067
Mo 1.304662   0.050934   0.000031
O   -0.070013   -0.183439   -1.278738
O   -0.070289   -0.184570   1.278448
O   -3.239133   0.698886   0.000239
O   2.017418   1.569245   0.000526
O   2.483752   -1.141341   -0.000284

[Mo₂O₅] I 7/8
E = -13947.72
Mo -1.572970   -0.015525   0.000132
Mo 1.302515   0.027752   -0.000131
O   0.123761   -0.011119   -1.318175
O   0.123296   -0.017016   1.317327
O   -3.567008   -0.020058   -0.000231
O   2.247693   1.416077   0.003273
O   2.492149   -1.432081   -0.002198

Table S5: Energies of calculated structures on the M06/def2TZVP level of theory. See Figures for the respective structures.

Ion	Isomer	Singlet/Douplet	Triplet/Quartet	Pentet/Hextet
$[MoO_4]^-$	I	0.00	3.20	-
	II	3.41	3.99	-
$[Mo_2O_8]^{2-}$	I	0.00	2.09	-
	II	2.20	1.56	-
	III	1.87	1.87	-
$[Mo_3O_9]^-$	I	0.00	3.09	-
	II	0.16	3.38	-
$[Mo_2O_6]$	I	0.00	2.68	-
$[Mo_2O_5]^-$	I	0.00	0.22	3.46
$[Mo_2O_6]^-$	I	0.00	3.52	-
$[Mo_2O_7]^-$	I	0.00	2.91	5.64
	II	0.17	3.21	-
$[Mo_2O_8]^-$	I	0.00	2.90	-
	II	2.45	2.38	-
	III	2.50	2.50	-
$[Mo_4O_{13}]^{2-}$	I	0.00	2.52	-
	II	0.41	-	-
	III	0.60	3.30	-
	IV	0.60	4.47	-
	V	0.78	3.19	-
	VI	0.95	2.90	-
	VII	2.87	3.80	-
	VIII	3.25	3.25	-
$[Mo_2O_5]$	I	0.02	0.00	-
$[MoO_3]$	I	0.00	1.80	-
$[Mo_3O_{10}]^-$	I	0.00	2.79	-
	II	0.56	2.76	-
	III	0.68	3.21	-
	IV	1.40	3.86	-
$[Mo_2O_7]$	I	0.00	-	-
	II	1.05	1.05	-
	III	1.60	1.56	-
$[Mo_3O_{10}]^{2-}$	I	0.00	3.18	-
	II	0.85	3.34	-
	III	0.85	3.56	-

	IV	4.40	2.94	-
	V	3.90	4.16	-
[Mo$_3$O$_9$]	I	0.00	3.12	-
	II	1.22	3.46	-
[Mo$_2$O$_7$]$^{2-}$	I	0.00	3.37	6.44
[MoO$_3$]$^-$	I	0.00	2.90	-
[Mo$_2$O$_6$]$^{2-}$	I	0.00	0.45	-

Calculated structures on the M06/def2TZVP level of theory

I E=0.00   II E=3.41

Figure S4.1: Calculated Molecules of [MoO$_4$]$^-$ in lowest multiplicity unless marked with *.

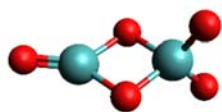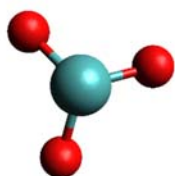I E=0.00   II* E=1.56   III E=1.87

Figure S4.2: Calculated Molecules of [Mo$_2$O$_8$]$^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00   II E=0.16

Figure S4.3: Calculated Molecules of [Mo$_3$O$_9$]$^-$ in lowest multiplicity unless marked with *.

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I E=0.00

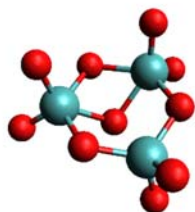Figure S4.4: Calculated Molecules of [Mo_2O_6] in lowest multiplicity unless marked with *.

I E=0.00

Figure S4.5: Calculated Molecules of [Mo_2O_5]$^-$ in lowest multiplicity unless marked with *.
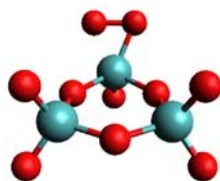

I E=0.00

Figure S4.6: Calculated Molecules of [Mo_2O_6]$^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=0.17

Figure S4.7: Calculated Molecules of [Mo_2O_7]$^-$ in lowest multiplicity unless marked with *.
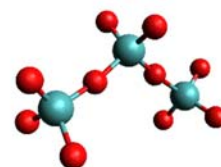
| I E=0.00 | II* E=2.38 | III E=2.50 |

Figure S4.8: Calculated Molecules of $[Mo_2O_8]^-$ in lowest multiplicity unless marked with *.

| I E=0.00 | II E=0.41 | III E=0.60 |

| IV E=0.60 | V E=0.78 | VI E=0.95 |

| VII E=2.87 | VIII E=3.25 |

Figure S4.9: Calculated Molecules of $[Mo_4O_{13}]^{2-}$ in lowest multiplicity unless marked with *.

I* E=0.00

Figure S4.10: Calculated Molecules of [Mo_2O_5] in lowest multiplicity unless marked with *.

I E=0.00

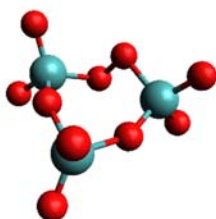Figure S4.11: Calculated Molecules of [MoO_3] in lowest multiplicity unless marked with *.
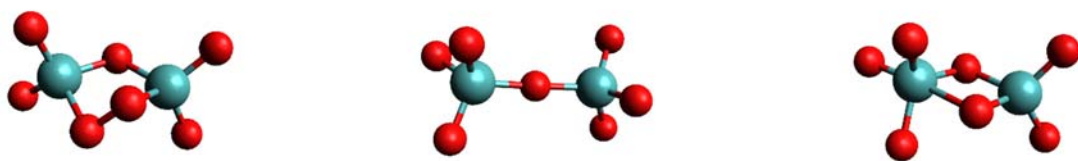

I E=0.00   II E=0.56   III E=0.68

IV E=1.40

Figure S4.12: Calculated Molecules of [Mo_3O_{10}]$^-$ in lowest multiplicity unless marked with *.
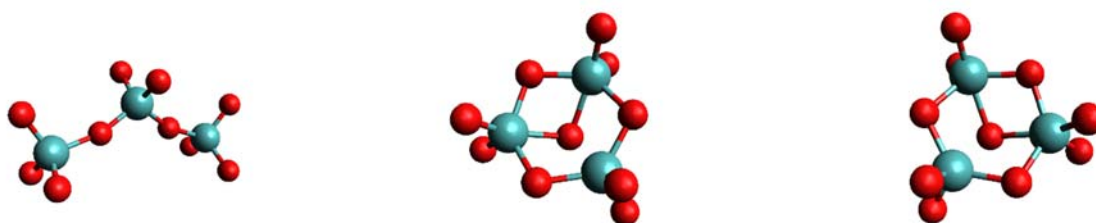
I E=0.00   II E=1.05   III* E=1.56

Figure S4.13: Calculated Molecules of $[Mo_2O_7]$ in lowest multiplicity unless marked with *.

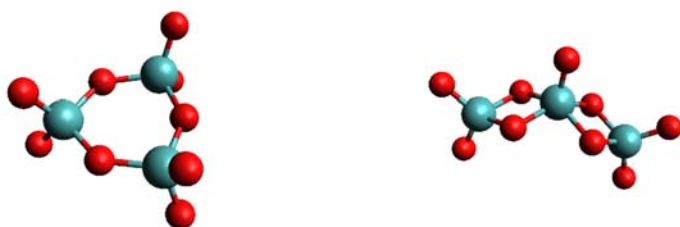I E=0.00   II E=0.85   III E=0.85

IV* E=2.94   V E=3.90

Figure S4.14: Calculated Molecules of $[Mo_3O_{10}]^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00   II E=1.22

Figure S4.15: Calculated Molecules of $[Mo_3O_9]$ in lowest multiplicity unless marked with *.
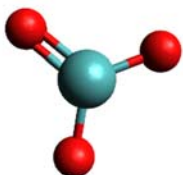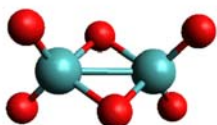
I E=0.00

Figure S4.16: Calculated Molecules of [Mo_2O_7]$^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S4.17: Calculated Molecules of [MoO_3]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S4.18: Calculated Molecules of [Mo_2O_6]$^{2-}$ in lowest multiplicity unless marked with *.

Cartesian coordinates (in Ångstrom) of species optimized at the M06/def2TZVP level of theory along the respective ZPE-corrected energy (in eV)

[MoO₄]⁻ I 1/2

E = -10046.92
```
Mo  -0.039945  -0.004273   0.001240
O   -0.898411   0.842776   1.230894
O   -0.898999   0.077302  -1.489276
O    1.723282   0.750670  -0.211360
O    0.283840  -1.648315   0.463233
```

[MoO₄]⁻ I 3/4

E = -10043.72
```
Mo  -0.158060   0.000009   0.010131
O    1.402610   0.002414  -1.162521
O    1.536938  -0.002044   1.083867
O   -1.055098   1.469004   0.014445
O   -1.054636  -1.469423   0.011024
```

[MoO₄]⁻ II 1/2

E = -10043.51
```
Mo   0.331928   0.006218  -0.000047
O    0.997349   1.592422   0.000084
O    1.403968  -1.340948   0.000122
O   -1.456601  -0.389933  -0.000056
O   -2.687337   0.105816   0.000095
```

[MoO₄]⁻ II 3/4

E = -10042.93
```
Mo   0.335024   0.000002  -0.036200
O    1.117660   1.532566   0.131898
O    1.121515  -1.530630   0.132281
O   -1.660909  -0.001203  -0.599433
O   -2.337141  -0.000743   0.525302
```

[Mo₂O₈]²⁻ I 1/2

E = -20093.70
```
Mo   2.021613   0.021579   0.000068
Mo  -1.774121  -0.190522   0.000111
O   -1.763173   1.769293   0.000887
O    0.128502  -0.187157   0.002767
O   -2.194577  -1.070386  -1.408493
O   -2.198181  -1.071310   1.407083
O    2.532863   0.883732  -1.407764
O    2.763896  -1.540766  -0.000613
O   -3.105061   1.219401  -0.001196
O    2.536396   0.884140   1.406386
```

[Mo₂O₈]²⁻ I 3/4

E = -20091.61
```
Mo  -1.995393  -0.006758   0.000421
Mo   1.801325  -0.275639   0.000075
O    2.064878   1.812880   0.656087
O   -0.104018   0.032365   0.006199
O    2.381492  -1.058255   1.420060
O    2.374397  -1.056331  -1.423764
O   -2.604639   0.707786   1.455788
O   -2.556604  -1.642852  -0.097334
O    2.061861   1.813128  -0.655906
O   -2.598514   0.873863  -1.363731
```

[Mo₂O₈]²⁻ II 1/2

E = -20091.50
```
Mo  -1.463771  -0.000008  -0.000107
Mo   1.610638  -0.000054  -0.000025
O    0.422939   0.123455   1.358707
O    2.630666  -1.390534   0.144281
O   -0.927276  -1.831124  -0.322263
O   -0.925735   1.830974   0.322061
O   -2.510305  -0.433436   1.284248
O    2.629147   1.391401  -0.145435
O   -2.511743   0.434169  -1.283063
O    0.421259  -0.124583  -1.357845
```

[Mo₂O₈]²⁻ II 3/4

E = -20092.14
```
Mo   1.797370  -0.223149  -0.006935
Mo  -1.998299  -0.008186   0.003135
O   -2.608036  -0.319105   1.589807
O   -2.532905   1.552930  -0.506056
O    1.662760   2.318715  -0.053219
O    2.141938  -1.839957   0.454575
O    2.647493   0.788768   1.180065
O   -2.624658  -1.196906  -1.083040
O    2.458375  -0.012259  -1.581314
O   -0.090090  -0.077676   0.019133
```

[Mo₂O₈]²⁻ III 1/2

E = -20091.83
```
Mo   1.565641   0.097191  -0.010741
Mo  -1.565748  -0.097323   0.010783
O   -0.170699  -0.845588  -0.900609
O   -2.875427  -0.153855  -1.121043
O   -2.031977  -1.084481   1.330829
O    2.028655   1.084422  -1.331848
O    2.020643  -1.777710  -0.288782
O   -2.018604   1.777956   0.289762
O    2.876917   0.155542   1.119214
O    0.171056   0.844406   0.902258
```

[Mo₂O₈]²⁻ III 3/4

E = -20091.83
```
Mo   1.565852   0.097365  -0.010500
Mo  -1.565838  -0.097418   0.010469
O   -0.172674  -0.839754  -0.906902
O   -2.879008  -0.149704  -1.117500
O   -2.028434  -1.090514   1.327354
O    2.027048   1.090109  -1.328142
O    2.019611  -1.777504  -0.293986
O   -2.018899   1.777564   0.295150
O    2.879538   0.150939   1.116854
O    0.172747   0.839143   0.907337
```

[Mo₃O₉]⁻ I 1/2

E = -24003.41
```
Mo  -2.177189  -0.000242  -0.000007
O   -0.723082   1.355504   0.005406
O   -0.722705  -1.355600  -0.005678
O   -3.095329  -0.005154   1.427850
O   -3.095541   0.004489  -1.427727
Mo   1.061471   1.710283   0.000439
O    1.504184   2.584546  -1.377058
O    1.511203   2.587565   1.373729
O    1.887210   0.000220  -0.000015
Mo   1.061899  -1.710061  -0.000448
O    1.504617  -2.584104   1.377187
O    1.511995  -2.587360  -1.373610
```

[Mo₃O₉]⁻ I 3/4

E = -24000.32
```
Mo  -1.008335  -1.735187  -0.070090
O   -1.871596  -0.043261  -0.158423
O    0.752832  -1.338984   0.029132
O   -1.525209  -2.602161   1.287412
O   -1.345898  -2.649533  -1.453790
```

[Mo₃O₉]⁻ II 1/2

```
Mo  -1.109209   1.681029   0.035140
O   -1.768799   2.775804  -1.071251
O   -1.381294   2.255857   1.603524
O    0.671924   1.450014  -0.260030
Mo   2.168486   0.139148  -0.056547
O    2.772424  -0.718022   1.626210
O    3.428168   0.424088  -1.122428
```

[Mo₃O₉]⁻ II 1/2

E = -24003.25
```
Mo  -0.000190  -0.717041  -0.000097
Mo  -2.653342   0.356195   0.000068
O   -1.392994  -0.064910   1.295087
O   -1.391328  -0.061744  -1.294799
O   -0.000705  -2.372463   0.000063
O   -3.141550   1.981060   0.001886
O   -4.016150  -0.655270  -0.002132
O    1.394069  -0.066357  -1.295368
O    1.390884  -0.061513   1.294473
Mo   2.653467   0.355998  -0.000019
O    4.017076  -0.654381   0.003689
O    3.141040   1.981032  -0.002645
```

[Mo₃O₉]⁻ II 3/4

E = -24000.03
```
Mo   0.046892  -0.791064   0.000013
Mo  -2.712513   0.360721  -0.000032
O   -1.376895  -0.135663   1.219697
O   -1.377020  -0.135924  -1.219746
O    0.060728  -2.453560  -0.000063
O   -3.106124   2.254058  -0.000085
O   -4.025043  -0.697423   0.000226
O    1.410838  -0.067115  -1.303071
O    1.411248  -0.068128   1.303186
Mo   2.644500   0.396103   0.000015
O    4.049649  -0.560083  -0.000584
O    3.063507   2.043592   0.000467
```

[Mo₂O₆] I 1/2

E = -15998.84
```
Mo  -1.427378  -0.000078  -0.000019
Mo   1.427358  -0.000179   0.000020
O    0.000024  -0.000667  -1.267467
O   -0.000101  -0.000882   1.267468
O   -2.390709  -1.365709  -0.000036
O   -2.390572   1.365672   0.000061
O    2.388339   1.367054   0.000036
O    2.393125  -1.364120  -0.000067
```

[Mo₂O₆] I 3/4

E = -15996.15
```
Mo  -1.405826  -0.020153   0.000008
Mo   1.409213   0.035813  -0.000026
O    0.008624  -0.038699  -1.298717
O    0.008697  -0.038881   1.298689
O   -2.379630  -1.384636  -0.000070
O   -2.357270   1.358676   0.000100
O    2.165887   1.511399   0.000051
O    2.535911  -1.490074   0.000046
```

[Mo₂O₅] I 1/2

E = -13945.94
```
Mo  -1.481406  -0.010140  -0.017307
Mo   1.320206  -0.106121   0.019351
O   -0.170400  -1.052032   0.909355
O   -0.139065   1.126857  -0.701747
```

O   -3.528312   0.046479 -0.081230
O   2.472850 -1.082943 -0.744500
O   2.211224   1.572007   0.607391

[Mo₂O₅]⁻ I 3/4

$[Mo_2O_5]^-$ I 3/4
E = -13952.89
Mo -1.533644 -0.000268   0.000000
Mo   1.299864 -0.000032   0.000000
O   -0.040226 -0.000531   1.308465
O   -0.040226 -0.000525 -1.308465
O   -3.224182   0.000618   0.000000
O   2.267603 -1.408802 -0.000003
O   2.264376   1.410813   0.000003

$[Mo_2O_5]^-$ I 5/6
E = -13952.67
Mo -1.553835 -0.133373 -0.000001
Mo   1.297460   0.033551   0.000005
O   -0.014826 -0.140871   1.322215
O   -0.014859 -0.140964 -1.322240
O   -3.139311   0.486240   0.000022
O   2.442254 -1.236202   0.000040
O   2.072708   1.555861 -0.000057

$[Mo_2O_5]^-$ I 7/8
E = -13949.43
Mo   1.514076   0.000100 -0.000008
Mo -1.313341   0.000199   0.000003
O   -0.023253   0.001627   1.338634
O   -0.023522   0.001806 -1.338702
O   3.565464   0.000177 -0.000010
O   -2.290468   1.404675   0.000138
O   -2.282077 -1.409853 -0.000032

$[Mo_2O_6]^-$ I 1/2
E = -16002.22
Mo   1.415536 -0.000083   0.000102
Mo -1.394983 -0.000031 -0.000090
O   -0.046627   0.000253 -1.329771
O   -0.047423   0.000473   1.330185
O   2.349784   1.422867 -0.000138
O   2.349771 -1.423025 -0.000172
O   -2.357010 -1.404815   0.000039
O   -2.356401   1.404848 -0.000204

$[Mo_2O_6]^-$ I 3/4
E = -15998.70
Mo   1.488918   0.041515 -0.000181
Mo -1.522994 -0.014010   0.000121
O   0.072655   0.034324 -1.224141
O   0.072959   0.038548   1.224066
O   2.357801   1.495719 -0.001714
O   2.523320 -1.603427   0.002263
O   -2.359156 -1.506861 -0.000447
O   -2.488682   1.397290   0.000286

$[Mo_2O_7]^-$ I 1/2
E = -18039.05
O   -2.849866 -1.564737   0.207394
O   -2.636218   1.467254   0.185232
Mo -1.758923   0.064906 -0.131662
O   -0.163211 -1.033315   0.636029
O   2.900366   0.305545 -1.247106
O   2.679260 -1.395446   0.163832
Mo   1.456188   0.068753   0.118678
O   -0.058256   0.249024 -1.141798
O   1.717284   1.269966   1.264586

$[Mo_2O_7]^-$ I 3/4
E = -18047.84
O   -2.586668 -1.314154   0.275152

O   -2.483715   1.378045 -0.449483
Mo -1.560082   0.011084 -0.024954
O   -0.375475   0.322632   1.282649
O   2.595335   0.396424 -1.184553
O   1.905155 -1.537568   0.598108
Mo   1.411052 -0.065032 -0.052945
O   -0.130661 -0.346266 -1.211315
O   1.858431   1.384114   1.098415

$[Mo_2O_7]^-$ I 5/6
E = -18044.93
O   -2.767205 -0.202869 -1.220421
O   -1.638346   1.643890   0.469316
Mo -1.536756   0.047280 -0.066791
O   -2.278480 -1.145405   1.220020
O   2.283434   1.566030   0.327915
O   2.993379 -1.127759 -0.369954
Mo   1.752794 -0.015008 -0.017969
O   0.064946 -0.206110 -1.068255
O   0.208075 -0.697210   1.086372

$[Mo_2O_7]^-$ I 7/8
E = -18042.20
O   -3.063686 -1.075195 -0.112347
O   -2.212691   1.664285   0.096341
Mo -1.765461   0.024003 -0.009655
O   -0.170098 -0.549853   1.070716
O   2.367009   1.212803 -1.126181
O   2.587127 -1.451353   0.098754
Mo   1.503545 -0.172587   0.017271
O   -0.133353 -0.376927 -1.110566
O   2.000752   1.356305   1.043298

$[Mo_2O_7]^-$ II 1/2
E = -18047.66
O   -2.473475   0.803870   1.388414
O   2.474027 -0.878990 -1.368101
Mo -1.875609   0.034186   0.000511
O   2.459647   1.614867 -0.075147
O   -2.653693 -1.697115 -0.009774
Mo   1.921500 -0.009388   0.000422
O   -2.458582   0.817554 -1.385701
O   2.489104 -0.752956   1.435266
O   -0.077953 -0.037419   0.010148

$[Mo_2O_7]^-$ II 3/4
E = -18044.63
O   -2.558328 -0.680101 -1.458881
O   2.416493 -0.825963   1.412704
Mo -1.916005 -0.066788 -0.000065
O   2.416530 -0.831437 -1.409370
O   -2.160067   1.895472 -0.000654
Mo   1.860185 -0.055958   0.000180
O   -2.558947 -0.678531   1.459165
O   2.677641   1.677854 -0.003485
O   0.059731   0.087121 -0.000084

$[Mo_2O_8]^-$ I 1/2
E = -20093.23
Mo   1.656100   0.009832 -0.026165
Mo -1.344173 -0.274381 -0.073738
O   -1.396780   1.965672 -0.047149
O   0.205559 -0.603525 -1.130462
O   -2.599795 -0.371417 -1.192704
O   -1.566202 -1.575358   0.974759
O   2.488056   1.352713 -0.671558
O   2.763153 -1.208303   0.422579
O   -2.064628   1.380819   0.905964
O   0.533022   0.448283   1.263060

$[Mo_2O_8]^-$ I 3/4
E = -20090.33

Mo   1.793739   0.020275 -0.038384
Mo -1.406116 -0.275372 -0.096039
O   -1.529167   1.938200 -0.033815
O   0.151765 -0.416954 -1.085587
O   -2.661554 -0.521591 -1.197367
O   -1.447617 -1.566330   1.004135
O   2.631192   1.391879 -0.610621
O   2.740324 -1.323221   0.416362
O   -2.265305   1.349499   0.868031
O   0.345339   0.487776   1.344586

$[Mo_2O_8]^-$ II 1/2
E = -20090.78
Mo   1.671030 -0.060891   0.015161
Mo -1.670853   0.060492   0.015234
O   -0.054618   1.054829 -0.446896
O   -2.489916 -1.545994 -0.612982
O   0.054871 -1.051578 -0.449971
O   1.849453 -0.135066   1.688550
O   2.490253   1.546070 -0.612276
O   -2.857655   1.053151 -0.707935
O   2.858664 -1.053482 -0.706495
O   -1.851979   0.134167   1.688432

$[Mo_2O_8]^-$ II 3/4
E = -20090.85
Mo   1.664069   0.067704 -0.000109
Mo -1.668383 -0.057475   0.036424
O   -0.150342 -1.261374 -0.238596
O   -2.660648 -0.682040   1.247281
O   0.130551   0.519353   0.990555
O   1.625139   0.807016 -1.517243
O   2.383257 -1.693676 -0.100995
O   -2.453227 -0.308075 -1.458701
O   2.967948   0.724878   0.880483
O   -1.820031   1.840213   0.006562

$[Mo_2O_8]^-$ III 1/2
E = -20090.74
Mo   1.440593   0.191922 -0.010751
Mo -1.574471 -0.128961   0.009233
O   0.152981   1.066946 -0.863651
O   -2.751912 -0.518817   1.173833
O   -1.984849 -0.962399 -1.397694
O   2.229920   1.242960   1.056123
O   2.567646 -0.420265 -1.244291
O   -2.128702   1.694251 -0.050384
O   2.589098 -1.575899   0.562464
O   0.028675 -0.857322   0.771571

$[Mo_2O_8]^-$ III 3/4
E = -20090.74
Mo   1.572536 -0.129100 -0.008609
Mo -1.441065   0.193639   0.011565
O   -0.147496   1.069408   0.855630
O   -2.589403 -1.576456 -0.562447
O   -2.233903   1.243807 -1.053699
O   1.982139 -0.969569   1.394296
O   2.149083   1.687565   0.055367
O   -2.565029 -0.419179   1.246466
O   2.746327 -0.517942 -1.178169
O   -0.031943 -0.856464 -0.772961

$[Mo_4O_{13}]^{2-}$ I 1/2
E = -34052.47
Mo   0.548335   2.187139 -0.454668
O   -2.028936   0.097373   2.088828
O   0.798249   3.858655 -0.292065
O   0.795269   1.835846 -2.092998
O   1.889223   1.469566   0.581981
Mo -2.250362   0.485736   0.454110
Mo   2.092451 -0.529577   0.541300
O   3.495069 -0.508327 -0.416358

```
O     2.524987   -0.782301    2.161098
O    -1.301433    2.062572   -0.005226
O     0.138732   -0.186853    0.003112
O    -3.866889    0.982335    0.298696
Mo   -0.431676   -2.081895   -0.538564
O    -2.141035   -1.011310   -0.600315
O    -0.390859   -2.580430   -2.158193
O     1.397646   -2.235646    0.007984
O    -1.093448   -3.323842    0.412019
```

$[Mo_4O_{13}]^{2-}$ I 3/4
E = -34049.95
```
Mo   -0.117967   -1.955884   -0.445680
O     2.709972    0.195996    1.992704
O     0.087517   -3.390517    0.891641
O    -0.180818   -2.503213   -2.029101
O    -1.931462   -1.851462    0.212789
Mo    2.688698   -0.069143    0.311534
Mo   -2.468356   -0.007710    0.382491
O    -3.817775    0.010324   -0.648741
O    -3.008213    0.037913    1.990618
O     1.830442   -1.645712   -0.058279
O    -0.419754   -0.004779   -0.188354
O     4.297390   -0.141474   -0.250929
Mo   -0.051402    2.057573   -0.340416
O     1.796269    1.293085   -0.527663
O    -0.161539    2.804655   -1.854967
O    -1.867908    1.834834    0.157572
O     0.398270    3.229961    0.796085
```

$[Mo_4O_{13}]^{2-}$ II 1/2
E = -34052.05
```
Mo   -4.964745    0.329593   -0.228081
O    -5.358886    1.568268    0.890450
O    -5.938591   -1.050771    0.070730
O    -5.231825    0.889728   -1.826649
Mo   -1.520183   -1.104186    0.229189
Mo    1.520230    1.104256    0.229102
O     1.534250    1.831261    1.751425
O     1.411468    2.339537   -0.914232
O     3.050149    0.159477   -0.018013
O    -0.000092    0.000194    0.079438
O    -1.412290   -2.338230   -0.915559
O    -1.532697   -1.832803    1.750753
O    -3.050492   -0.159336   -0.015245
Mo    4.964659   -0.329642   -0.228163
O     5.357398   -1.568348    0.890841
O     5.233542   -0.889735   -1.826444
O     5.938270    1.050649    0.071751
```

$[Mo_4O_{13}]^{2-}$ III 1/2
E = -34051.87
```
Mo    0.560268   -1.108316    0.006417
Mo    4.133464    0.106166   -0.003518
Mo   -1.564098    1.990786    0.001376
Mo   -3.337880   -0.910378   -0.004674
O    -3.163487    1.001933   -0.007520
O    -1.506671    2.977520    1.377826
O     2.229956   -0.359795    0.006021
O     5.071456   -1.321510    0.188734
O    -4.177594   -1.441248    1.380537
O    -1.662343   -1.444828    0.001976
O    -0.229988    0.770578    0.001699
O     4.538648    0.856414   -1.495613
O     4.470454    1.186907    1.289338
O    -1.497197    2.987060   -1.367831
O    -4.168349   -1.445793   -1.393699
O     0.592004   -2.083625    1.380579
O     0.596403   -2.094466   -1.359953
```

$[Mo_4O_{13}]^{2-}$ III 3/4
E = -34049.17
```
Mo    0.424002   -1.166346    0.013551
Mo    4.078826    0.049915   -0.017330
Mo   -1.510288    2.011773   -0.003750
Mo   -3.305321   -0.894526   -0.002915
O    -3.126723    1.012357   -0.010392
O    -1.450376    2.994141    1.386023
O     2.369594   -0.395090    0.014462
O     4.958259   -0.641721    1.282753
O    -4.162021   -1.393659    1.385486
O    -1.630231   -1.503376    0.008452
O    -0.193984    0.787443    0.001357
O     4.802330   -0.347565   -1.511140
O     4.317197    1.932285    0.250074
O    -1.437706    2.994607   -1.392810
O    -4.148178   -1.405444   -1.395475
O     0.667635   -2.012460    1.474472
O     0.676995   -2.025801   -1.438426
```

$[Mo_4O_{13}]^{2-}$ IV 1/2
E = -34051.87
```
Mo   -3.336056   -0.916169   -0.016794
O    -3.167534    0.996565   -0.038200
O    -1.659015   -1.444318    0.024851
O    -4.196901   -1.436083    1.359625
O    -4.141261   -1.468553   -1.413978
Mo   -1.571827    1.990487    0.003871
O    -1.500066    3.018846   -1.341310
O    -1.526299    2.944753    1.403412
O    -0.233587    0.775060   -0.016119
Mo    0.562554   -1.101183    0.022311
O     0.597828   -2.053920    1.411996
O     0.601727   -2.109355   -1.327803
O     5.054331   -1.268034    0.461189
O     4.601197    0.581073   -1.597619
Mo    4.136337    0.105813   -0.012733
O     2.229795   -0.347012    0.010343
O     4.436989    1.396499    1.081175
```

$[Mo_4O_{13}]^{2-}$ IV 3/4
E = -34047.99
```
Mo    2.468502    1.852851   -0.185734
O     3.137558    0.026943   -0.050297
O     0.695857    1.317110    0.366159
O     3.174038    2.911680    0.957236
O     2.494473    2.426717   -1.794204
Mo    2.573574   -1.718305   -0.208275
O     2.745616   -2.288085   -1.806050
O     3.391595   -2.736826    0.888690
O     0.782934   -1.526156    0.243460
Mo   -0.451822   -0.129870    0.778368
O    -0.637590   -0.264320    2.440189
O    -1.925729   -0.093042   -0.340595
O    -3.903400    0.518658    1.408854
O    -4.147802    1.068528   -1.744046
Mo   -3.886786   -0.034436   -0.471090
O    -4.281962   -1.625819   -0.940475
O    -5.218790    0.420846    0.826416
```

$[Mo_4O_{13}]^{2-}$ V 1/2
E = -34051.69
```
Mo   -2.513686   -1.205854   -0.553027
Mo   -1.524366    1.991910    0.168574
Mo    0.772332   -0.948988    0.831245
O    -2.804934    0.627339   -0.317312
O    -0.731846   -1.476227   -0.426861
O     0.020075    1.160339    0.135518
O    -1.568615    3.272900   -0.957932
O    -1.881761    2.612703    1.717349
O    -3.362792   -2.083144    0.627632
O    -3.082908   -1.701382   -2.075462
O     2.363110    0.076220    1.205857
O     1.906141   -0.650270   -1.293263
Mo    3.215814    0.210633   -0.488257
O     3.464042    1.803678   -1.064498
O     4.701270   -0.647494   -0.548576
O    -0.145334   -0.736226    2.229562
O     1.385557   -2.508866    0.985680
```

$[Mo_4O_{13}]^{2-}$ V 3/4
E = -34049.27
```
Mo   -2.229732   -1.452858   -0.593294
Mo   -1.777182    1.848619    0.058799
Mo    0.850722   -0.509080    1.185286
O    -2.702440    0.385605   -0.734022
O    -0.838563   -1.470551    0.532199
O    -0.204005    1.209347    0.610402
O    -1.588479    3.102212   -1.074107
O    -2.679197    2.446101    1.373744
O    -3.531074   -2.342033    0.051948
O    -1.856181   -2.078726   -2.126837
O     2.511982    0.625575    1.051296
O     1.528648   -0.731188   -0.963271
Mo    3.019103    0.238479   -0.691200
O     3.132514    1.624777   -1.690250
O     4.475606   -0.659269   -0.820979
O     0.586609   -0.591730    2.834076
O     1.884294   -2.177214    1.167949
```

$[Mo_4O_{13}]^{2-}$ VI 1/2
E = -34051.51
```
Mo    2.874829   -0.404634    0.065982
O     3.937898   -0.645084   -1.229805
O     3.754236   -0.651690    1.491220
O     2.230202    1.300205    0.027325
Mo   -0.268904   -2.407236   -0.065920
Mo    0.629944    2.451845   -0.057993
O     0.694667    3.367788   -1.499076
O     0.641707    3.508659    1.284427
O    -0.753169    1.413230   -0.035002
O    -0.427816   -3.321866   -1.486413
O    -1.493113   -1.133960    0.005743
O    -0.412002   -3.457987    1.258843
O     1.480466   -1.618980   -0.035293
Mo   -3.016246    0.296447    0.054978
O    -4.249595   -0.897237    0.095374
O    -3.225157    1.228979    1.477161
O    -3.331344    1.241722   -1.339007
```

$[Mo_4O_{13}]^{2-}$ VI 3/4
E = -34049.57
```
Mo   -3.029595   -0.030545    0.069637
O    -4.006738   -0.005457   -1.331070
O    -3.899513    0.099034    1.533641
O    -1.767443   -1.531503    0.094245
Mo   -0.100783    2.373864   -0.085921
Mo   -0.114806   -2.330861   -0.052116
O     0.081233   -2.929139   -1.631625
O    -0.042355   -3.641639    1.035590
O     1.186522   -1.120903    0.343412
O    -0.191282    3.310587   -1.502910
O     1.510144    1.472365   -0.055914
O    -0.187347    3.416515    1.254861
O    -1.488167    1.195312   -0.046101
Mo    2.855623    0.068544    0.108131
O     4.011757    0.763789   -0.945746
O     3.499488    0.179931    1.664720
O     3.338896   -1.634153   -0.621692
```

$[Mo_4O_{13}]^{2-}$ VII 1/2
E = -34049.60
```
Mo   -2.032391    1.797305   -0.261494
O     2.199844   -1.256872    0.228303
O    -1.956551   -2.497647   -1.617157
O    -2.726694    2.116944    1.264251
O    -2.604320    0.218676   -0.890256
Mo    3.572823   -0.188387   -0.360137
Mo   -2.126620   -1.757593   -0.105229
O    -3.961505   -1.385425    0.309626
O    -2.217031   -3.034094    1.032032
O    -2.452104    3.054362   -1.340182
O     2.515806    1.300874   -0.190411
```

O   4.064662   -0.505175   -1.966384
Mo   0.937727   0.302701   0.696595
O   4.938691   -0.185909   0.668997
O   -0.173862   1.701292   -0.085544
O   -0.516748   -0.920690   0.419340
O   1.044230   0.585032   2.329435

[Mo₄O₁₃]²⁻ VII 3/4
E = -34048.66
Mo   -0.293022   1.960442   -0.265660
O   -0.703328   -1.861756   0.332730
O   1.094702   0.663559   1.135770
O   -1.493145   2.758342   -1.156994
O   0.719124   3.238795   0.168800
Mo   -2.072134   -0.535069   0.798009
Mo   2.479608   -0.321338   0.625312
O   3.750171   0.645172   -0.001085
O   3.104229   -1.242141   1.929899
O   -1.520741   1.202355   1.107306
O   -1.880068   -0.368179   -1.050959
O   -1.723058   -1.420721   2.480731
Mo   -0.047693   -1.134567   -1.340373
O   -3.748822   -0.740115   0.963702
O   0.458879   0.800927   -1.510667
O   1.783193   -1.439813   -0.718561
O   -0.191621   -2.076135   -2.721436

[Mo₄O₁₃]²⁻ VIII 1/2
E = -34049.22
Mo   3.753904   0.875291   -0.123721
Mo   1.301513   -0.980090   0.172026
Mo   -1.672013   -0.901432   -0.160516
Mo   -3.844993   0.955599   0.069588
O   2.363345   -0.153804   1.332606
O   2.456537   0.018977   -1.140967
O   -2.860578   -0.025083   1.276700
O   -2.742734   0.443173   -1.306799
O   -0.085271   -0.482465   -1.130016
O   -0.261235   -1.128681   1.199000
O   1.828517   -2.568372   -0.048546
O   -2.221784   -2.426189   -0.560491
O   -5.456465   0.406287   -0.103030
O   4.850902   0.082378   1.243622
O   3.286465   2.460271   0.256778
O   -3.840477   2.637528   0.373185
O   5.106121   1.001796   -1.168270

[Mo₄O₁₃]²⁻ VIII 3/4
E = -34049.21
Mo   -3.751813   -0.875806   -0.122758
Mo   -1.301544   0.981743   0.170323
Mo   1.670816   0.900676   -0.158953
Mo   3.844559   -0.955709   0.069089
O   -2.373306   0.164419   1.329882
O   -2.453350   -0.020131   -1.140989
O   2.858394   0.020669   1.277825
O   2.744010   -0.439347   -1.307483
O   0.085777   0.487693   -1.133222
O   0.258356   1.118926   1.202376
O   -1.819805   2.572275   -0.053073
O   2.221042   2.427007   -0.551867
O   5.456150   -0.405374   -0.099533
O   -4.858833   -0.090941   1.242128
O   -3.281858   -2.459739   0.258453
O   3.839712   -2.638596   0.367520
O   -5.101883   -1.004108   -1.169953

[Mo₂O₅] I 1/2
E = -13950.02
Mo   -1.472057   0.230405   0.000014
Mo   1.211748   -0.060468   0.000103
O   -0.154693   0.329096   -1.330291
O   -0.154759   0.329448   1.330431
O   -2.586237   -0.995293   -0.000297

O   2.456020   1.071826   -0.000573
O   1.806289   -1.627249   0.000111

[Mo₂O₅] I 3/4
E = -13950.04
Mo   -1.547814   0.058950   0.000098
Mo   1.318210   -0.014422   -0.000086
O   -0.080465   0.054261   -1.278437
O   -0.079886   0.053626   1.278517
O   -3.189747   -0.214065   -0.000171
O   2.348765   1.311433   0.000283
O   2.206754   -1.439024   -0.000252

[MoO₃] I 1/2
E = -7997.08
Mo   -0.000014   -0.000030   -0.200076
O   1.209138   -1.049738   0.350144
O   0.304732   1.571934   0.350095
O   -1.513795   -0.522039   0.350162

[MoO₃] I 3/4
E = -7995.28
Mo   -0.021221   0.000089   -0.133236
O   -0.864834   -1.404503   0.240605
O   1.845913   0.002356   0.218118
O   -0.869670   1.401681   0.240767

[Mo₃O₁₀]⁻ I 1/2
E = -26048.74
Mo   -0.914745   -1.600972   -0.016999
O   0.968439   -1.614004   0.398588
O   -2.146488   0.000736   -0.261436
O   -1.323628   -2.854354   -1.060113
O   -1.557554   -1.958090   1.505659
Mo   1.907188   -0.000695   0.036237
O   2.640898   -0.001027   -1.477801
O   -0.258585   0.000167   -1.109186
Mo   -0.913484   1.601653   -0.016998
O   0.969708   1.613396   0.398500
O   -1.556015   1.959013   1.505713
O   3.170220   -0.001160   1.148304
O   -1.321530   2.855400   -1.059983

[Mo₃O₁₀]⁻ I 3/4
E = -26045.95
Mo   -1.727279   -0.774370   -0.016234
O   -0.079490   -2.022395   -0.183402
O   -1.742970   1.055360   0.365649
O   -2.886302   -1.152340   -1.176384
O   -2.203086   -1.503794   1.433982
Mo   1.652777   -0.897668   -0.037623
O   2.767533   -1.369046   -1.205995
O   -0.011963   -0.061417   -1.018323
Mo   0.064821   1.750292   -0.072371
O   1.823173   0.933818   0.335534
O   0.237994   2.303553   1.805705
O   2.083310   -1.652692   1.412987
O   0.062628   3.058117   -1.107058

[Mo₃O₁₀]⁻ II 1/2
E = -26048.17
Mo   2.052023   -0.001153   -0.200152
O   0.589182   1.373524   -0.144872
O   0.587852   -1.374254   -0.140930
O   2.759468   -0.003435   -1.708668
O   3.152182   0.718482   1.179307
Mo   -1.186087   1.713126   0.022693
O   -1.521933   2.502796   1.478520
O   -1.747722   2.669554   -1.252231
O   -2.003292   0.001030   -0.005926
Mo   -1.188087   -1.712002   0.023090
O   -1.748497   -2.666796   -1.253595

O   -1.527443   -2.502452   1.477688
O   3.151492   -0.718299   1.181149

[Mo₃O₁₀]⁻ II 3/4
E = -26045.98
Mo   1.908925   -0.284265   -0.271788
O   0.997978   1.395678   -0.085826
O   0.475638   -1.453482   -0.061363
O   2.541293   -0.388801   -1.808082
O   3.184949   0.062696   1.352627
Mo   -0.809227   1.912320   0.025746
O   -1.170975   2.411321   1.606679
O   -1.180034   3.141992   -1.081404
O   -1.638863   0.340120   -0.399124
Mo   -1.498506   -1.577885   0.015800
O   -2.136322   -2.573476   -1.207211
O   -1.996557   -1.972611   1.594634
O   3.016637   -1.226833   1.297844

[Mo₃O₁₀]⁻ III 1/2
E = -26048.06
Mo   3.237843   -0.648233   -0.008339
O   1.488145   0.340618   -0.063158
O   -3.482111   -1.338730   -1.524275
O   3.180505   -1.897472   -1.177474
O   3.441956   -1.315116   1.554645
Mo   0.006304   1.344876   0.012793
O   -0.010480   2.281557   1.411362
O   -0.073461   2.392417   -1.302652
O   -1.558075   0.214306   0.005999
Mo   -3.194248   -0.618816   -0.027643
O   -4.438739   0.466419   0.315439
O   -3.322328   -1.991115   1.261416
O   4.512618   0.438527   -0.359563

[Mo₃O₁₀]⁻ III 3/4
E = -26045.52
Mo   -2.766632   -0.813238   0.000022
O   -1.435276   0.420100   0.000107
O   2.735146   -1.771528   1.399079
O   -2.736219   -1.770723   1.399672
O   -2.735292   -1.771639   -1.398990
Mo   0.000022   1.766528   -0.000008
O   -0.000083   2.691343   -1.422379
O   0.000255   2.690775   1.422718
O   1.435181   0.419987   -0.000386
Mo   2.766603   -0.813290   -0.000019
O   4.461056   0.063869   0.000690
O   2.736423   -1.770927   -1.399577
O   -4.461157   0.063749   -0.000908

[Mo₃O₁₀]⁻ IV 1/2
E = -26047.34
Mo   1.901973   -0.962442   0.025924
O   1.311155   0.726551   0.301238
O   3.037695   -1.003383   -1.222062
O   2.600516   -1.604890   1.422249
Mo   0.004039   2.212635   -0.000053
O   -0.282495   3.130400   1.398496
O   0.293487   3.130486   -1.397972
O   -1.307475   0.730567   -0.301600
Mo   -1.905489   -0.955817   -0.025992
O   -2.606506   -1.595942   -1.422160
O   -3.041459   -0.991722   1.221925
O   -0.405233   -2.033206   0.597663
O   0.397568   -2.034338   -0.597141

[Mo₃O₁₀]⁻ IV 3/4
E = -26044.87
Mo   2.307896   -0.808357   0.031902
O   1.230287   0.650838   0.751333
O   2.903593   -0.511724   -1.528100
O   3.453771   -1.360289   1.155339

Mo 0.005491   1.818136 -0.000107
O -0.683364   2.791819   1.199820
O   0.691269   2.791614 -1.202097
O -1.215041   0.644464 -0.747774
Mo -2.312011 -0.802899 -0.034555
O -3.426271 -1.378136 -1.177783
O -2.951258 -0.474840   1.501791
O -0.642889 -2.116682   0.144040
O   0.632679 -2.123183 -0.082080

[Mo$_2$O$_7$] I 1/2
E = -18043.17
O -2.669010 -1.124350 -0.480787
O -2.223596   1.490348   0.049079
Mo -1.497762 -0.006780 -0.084398
O -0.499411 -0.518666   1.505928
O   2.666688   1.128272 -0.476530
O   2.227328 -1.488962   0.044451
Mo 1.497855   0.006801 -0.084348
O   0.000140   0.000821 -1.263259
O   0.497374   0.512426   1.507031

[Mo$_2$O$_7$] II 1/2
E = -18042.11
O -2.456228   0.135405 -1.588796
O   2.456238   0.139573   1.588390
Mo -1.872707 -0.010343 -0.023138
O   2.354590 -1.487545 -0.605689
O -2.703468   1.331184   0.996598
Mo 1.872642 -0.010289   0.023137
O -2.354403 -1.486022   0.609537
O   2.703629   1.328549 -0.999897
O -0.000015   0.147178 -0.000135

[Mo$_2$O$_7$] II 3/4
E = -18042.11
O -2.456501   0.136994 -1.588455
O   2.456502   0.137026   1.588452
Mo -1.872634 -0.010344 -0.023082
O   2.353858 -1.486877 -0.607822
O -2.703494   1.330144   0.998034
Mo 1.872634 -0.010343   0.023083
O -2.353857 -1.486866   0.607850
O   2.703491   1.330123 -0.998066
O -0.000000   0.148058   0.000001

[Mo$_2$O$_7$] III 1/2
E = -18041.56
O -2.511225   1.370360   0.083325
O -0.004150 -0.000267   1.268111
Mo -1.548249   0.000002   0.045581
O   2.835543   0.000493   0.946782
O -2.508770 -1.372072   0.081869
Mo 1.400242   0.000149   0.074749
O   1.626016 -1.568844 -0.863903
O   1.624218   1.568639 -0.864580
O -0.284599   0.000901 -1.283336

[Mo$_2$O$_7$] III 3/4
E = -18041.61
O -2.508512   1.368437   0.100947
O   0.002763 -0.000467   1.262577
Mo -1.543940 -0.000060   0.050474
O   2.855435   0.003046   0.944545
O -2.503224 -1.372241   0.099759
Mo 1.416369   0.000501   0.076182
O   1.569744 -1.550798 -0.879543
O   1.561462   1.549359 -0.882468
O -0.307921   0.000345 -1.310757

[Mo$_3$O$_{10}$]$^{2-}$ I 1/2
E = -26051.01

Mo 3.395824 -0.532156   0.000214
O   1.513076   0.020822   0.006220
O -3.673299 -1.624529 -1.297496
O   3.762258 -1.308406 -1.488570
O   3.674008 -1.625208   1.296871
Mo 0.000130   1.063730   0.000029
O -0.004802   2.068670   1.364597
O   0.005719   2.067638 -1.365290
O -1.513541   0.021894 -0.005291
Mo -3.395987 -0.532046 -0.000179
O -4.406758   0.845291 -0.187910
O -3.762419 -1.309289   1.488092
O   4.405933   0.845595   0.188442

[Mo$_3$O$_{10}$]$^{2-}$ I 3/4
E = -26047.84
Mo 3.388404 -0.523353 -0.000022
O   1.527669   0.064720   0.000058
O -4.011890 -1.307750 -1.436127
O   3.689803 -1.471997 -1.402822
O   3.690086 -1.471531   1.403029
Mo -0.047496   1.028795   0.000025
O -0.101416   2.023478   1.372476
O -0.101314   2.023586 -1.372348
O -1.460926 -0.151488 -0.000090
Mo -3.377627 -0.621348 -0.000015
O -3.462683   1.365154 -0.000081
O -4.011241 -1.307798   1.436317
O   4.434691   0.842130 -0.000347

[Mo$_3$O$_{10}$]$^{2-}$ II 1/2
E = -26050.16
Mo -1.708699 -0.630766 -0.014416
O -0.340524 -1.983693   0.276352
O -1.547922   1.286755   0.276291
O -2.756083 -1.018309 -1.299809
O -2.569239 -0.947960   1.431335
Mo 1.400618 -1.164485 -0.014457
O   2.259595 -1.878894 -1.299493
O   0.000123   0.000195 -0.995898
Mo 0.308009   1.795200 -0.014589
O   1.888341   0.697155   0.275807
O   0.463508   2.698334   1.431549
O   2.105897 -1.749615   1.431713
O   0.496680   2.896299 -1.299677

[Mo$_3$O$_{10}$]$^{2-}$ II 3/4
E = -26047.67
Mo -1.621892 -0.866191 -0.014533
O -0.000050 -1.870619   0.329068
O -1.785764   0.999286   0.333181
O -2.483127 -1.377031 -1.391101
O -2.516262 -1.459607   1.316500
Mo 1.621837 -0.866280 -0.014556
O   2.483026 -1.377153 -1.391140
O -0.000008   0.109649 -1.088024
Mo 0.000042   1.759474 -0.084345
O   1.785809   0.999194   0.333170
O   0.000213   2.138510   1.882515
O   2.516200 -1.459747   1.316458
O   0.000037   3.155749 -1.045100

[Mo$_3$O$_{10}$]$^{2-}$ III 1/2
E = -26050.16
Mo 1.761235 -0.463985   0.014433
O   1.417886   1.429271 -0.275578
O   0.529038 -1.942085 -0.275230
O   2.646905 -0.698191 -1.432023
O   2.841707 -0.749404   1.299100
Mo -0.478414   1.757152   0.014544
O -0.771707   2.836181   1.298623
O -0.720411   2.639672 -1.432542
O -1.947046   0.513158 -0.273679
Mo -1.282684 -1.293017   0.014546

O -1.927646 -1.942304 -1.432319
O -2.069832 -2.087260   1.298648
O   0.000390   0.000178   0.996504

[Mo$_3$O$_{10}$]$^{2-}$ III 3/4
E = -26047.46
Mo 2.143812   0.331899 -0.024720
O   0.483635   1.376240 -0.445638
O   1.179568 -1.399602   0.068381
O   3.219032   0.362708 -1.362101
O   2.865262   0.741696   1.474606
Mo -1.321425   1.373016 -0.025497
O -1.738470   2.436041   1.239658
O -2.050140   2.045792 -1.416874
O -2.191058 -0.347596 -0.044993
Mo -0.669317 -1.685720   0.016706
O -0.874465 -2.028644 -1.647431
O -1.076473 -3.097993   0.879210
O -0.620508 -0.189414   1.431112

[Mo$_3$O$_{10}$]$^{2-}$ IV 1/2
E = -26046.61
Mo 1.591213 -0.787720 -0.040468
O   1.757880   1.090271 -0.040228
O   2.133470 -1.231698 -1.597424
O   2.750908 -1.285503   1.125757
Mo -0.014387   1.827045   0.023398
O -0.021292   2.619877   1.543158
O -0.020747   2.880462 -1.329100
O -1.775010   1.063216 -0.041956
Mo -1.578676 -0.811854 -0.040737
O -2.113792 -1.265307 -1.597466
O -2.730678 -1.325958   1.126099
O   0.012349 -1.617717   1.240125
O   0.016623 -2.121866 -0.125482

[Mo$_3$O$_{10}$]$^{2-}$ IV 3/4
E = -26048.07
Mo 2.072367   0.333788   0.066194
O   0.434985   1.237147   0.044534
O   3.093802   1.501185 -0.069750
O   2.648583   0.195475   1.655905
Mo -1.460064   1.640076 -0.025826
O -2.020880   2.456059   1.374422
O -1.937751   2.363735 -1.505029
O -1.867675 -0.297317   0.018578
Mo -0.898787 -1.877368   0.004649
O -1.232000 -2.741985 -1.437800
O -1.355448 -2.851906   1.339532
O   0.836550 -1.453703   0.104132
O   2.903875 -0.915296 -1.133059

[Mo$_3$O$_{10}$]$^{2-}$ V 1/2
E = -26047.11
Mo 2.091424   0.000254 -0.207764
O   0.542238   1.271766 -0.141522
O   0.542735 -1.272472 -0.142798
O   2.743495   0.000753 -1.763221
O   3.215005 -0.720143   1.205868
Mo -1.202736   1.688543   0.025647
O -1.520624   2.459362   1.525051
O -1.726836   2.648894 -1.295972
O -2.118886 -0.000275 -0.001712
Mo -1.202183 -1.688739   0.025587
O -1.727045 -2.649682 -1.295286
O -1.519007 -2.459186   1.525389
O   3.214775   0.720680   1.205986

[Mo$_3$O$_{10}$]$^{2-}$ V 3/4
E = -26046.86
Mo 1.972963 -0.369997 -0.212652
O   0.918368   1.344340 -0.290759
O   0.364580 -1.414880   0.050496

```
O     2.616320  -0.707940  -1.735022
O     2.980371  -1.168479   1.233378
Mo   -0.772337   1.941575   0.025685
O    -0.839917   2.907085   1.439490
O    -1.341460   2.856207  -1.307867
O    -1.760554   0.424426   0.247063
Mo   -1.567661  -1.556045   0.052214
O    -2.169069  -2.123798  -1.454163
O    -2.101545  -2.442461   1.426236
O     3.259842   0.243953   1.098606


[Mo3O9] I 1/2
E = -24000.12
Mo   -1.113825  -1.689854   0.000004
O    -1.714792   0.101605  -0.001069
O     0.769473  -1.535979   0.001294
O    -1.658933  -2.514967   1.342475
O    -1.657058  -2.515962  -1.342617
Mo   -0.906518   1.809498  -0.000106
O    -1.349760   2.694426  -1.342047
O    -1.349654   2.692609   1.343060
O     0.945533   1.434478  -0.000307
Mo    2.020342  -0.119654   0.000077
O     3.008210  -0.177510   1.342192
O     3.006991  -0.178649  -1.342847


[Mo3O9] I 3/4
E = -23997.01
Mo    1.041147   1.720791  -0.010858
O     1.829097   0.003095   0.006433
O    -0.806002   1.368063  -0.092887
O     1.430083   2.582643   1.367524
O     1.531786   2.612318  -1.337113
Mo    1.047196  -1.717430  -0.010444
O     1.537703  -2.605550  -1.339079
O     1.442775  -2.579641   1.365836
O    -0.801301  -1.370952  -0.087576
Mo   -2.109226  -0.003436   0.033398
O    -2.804774  -0.002394   1.537697
O    -3.249733  -0.007188  -1.484337


[Mo3O9] II 1/2
E = -23998.91
Mo    0.000005  -0.717407  -0.000210
Mo   -2.683236   0.367487   0.000124
O    -1.352141  -0.151558   1.240934
O    -1.353719  -0.153591  -1.241072
O     0.000269  -2.367843   0.000345
O    -3.079803   1.992396  -0.001336
O    -4.081418  -0.549596   0.001399
O     1.352633  -0.152100  -1.241112
O     1.353115  -0.152311   1.240993
Mo    2.683168   0.367607   0.000010
O     4.080651  -0.550572  -0.000159
O     3.080744   1.992318   0.000407


[Mo3O9] II 3/4
E = -23996.66
Mo   -0.039353  -0.744872   0.000047
Mo   -2.660449   0.377842  -0.000036
O    -1.348306  -0.097259   1.276111
O    -1.348850  -0.098436  -1.276203
O    -0.027885  -2.387662   0.000443
O    -3.088843   2.001092  -0.000658
O    -4.041931  -0.573941   0.000636
O     1.480604  -0.041200  -1.303902
O     1.480954  -0.041625   1.303823
Mo    2.629998   0.330255  -0.000107
O     3.988480  -0.655144  -0.000617
O     3.272248   2.087246   0.000871


[Mo2O7]2- I 1/2
E = -18049.32
```

```
Mo   -1.903884   0.000003  -0.000008
Mo    1.903885   0.000007   0.000018
O    -0.000001   0.000111   0.000053
O     2.494708  -1.324121   0.941347
O     2.495081  -0.153305  -1.617264
O    -2.495213   0.152965   1.617259
O    -2.494770   1.324228  -0.941155
O     2.495184   1.477153   0.675984
O    -2.494991  -1.477084  -0.676272


[Mo2O7]2- I 3/4
E = -18045.95
Mo    1.952371  -0.053781  -0.002603
Mo   -1.883212  -0.039147  -0.002319
O     0.014182  -0.000534  -0.001959
O    -2.507513   0.682703   1.441148
O    -2.508079   0.803580  -1.378515
O     2.709999  -0.529047  -1.478443
O     1.631418   1.904667   0.098547
O    -2.407001  -1.689654  -0.072184
O     2.703908  -0.683840   1.417249


[Mo2O7]2- I 5/6
E = -18042.89
Mo   -1.893703   0.190368   0.021237
Mo    1.923888  -0.016657   0.022001
O     0.036148  -0.054269  -0.013578
O     2.538394  -1.571898   0.474496
O     2.540968   0.410100  -1.541286
O    -2.487003   1.754266  -0.342662
O    -2.103369  -1.631262  -0.941960
O     2.460357   1.161941   1.175802
O    -3.143966  -0.980865   0.962186


[Mo2O7]2- I 7/8
E = -18039.50
Mo   -1.857915   0.151102   0.037120
Mo    1.973188  -0.055161   0.044782
O     0.058922  -0.208369   0.038466
O     2.707935   0.150490   1.599085
O     2.833771  -1.120327  -1.014446
O    -2.441795   1.719116  -0.330343
O    -2.090745  -1.684894  -0.910347
O     1.431902   1.650416  -0.812641
O    -3.105172  -1.010124   1.000242


[MoO3]- I 1/2
E = -8000.05
Mo    0.000043  -0.000034  -0.089409
O     0.055667  -1.716803   0.156484
O     1.459291   0.906752   0.156476
O    -1.515185   0.810228   0.156439


[MoO3]- I 3/4
E = -7997.15
Mo    0.060233   0.000079   0.000011
O     0.826655   1.557831  -0.000021
O    -1.977425  -0.004488  -0.000016
O     0.834549  -1.553758  -0.000021


[Mo2O6]2- I 1/2
E = -16000.16
Mo   -1.320098  -0.000034   0.000005
Mo    1.320138  -0.000009  -0.000006
O    -0.000015  -0.000100  -1.444001
O     0.000010  -0.000042   1.444000
O    -2.232915  -1.481317   0.000010
O    -2.233051   1.481169  -0.000018
O     2.232391   1.481602  -0.000008
O     2.233370  -1.481086   0.000019
```

```
[Mo2O6]2- I 3/4
E = -15999.71
Mo   -1.526597  -0.000072  -0.000004
Mo    1.526595  -0.000061   0.000004
O     0.000010  -0.000496  -1.224561
O    -0.000018  -0.001195   1.224560
O    -2.452483  -1.470747  -0.000139
O    -2.451195   1.471415   0.000146
O     2.450297   1.471930   0.000146
O     2.453400  -1.470211  -0.000151
```

Table S6: Energies of calculated structures on the M06/def2TZVP level of theory. See Figures for the respective structures.

Ion	Isomer	Singlet/Douplet	Triplet/Quartet	Pentet/Hextet
$[MoO_4]^-$	I	0.00	3.13	-
	II	3.39	3.93	-
$[Mo_2O_8]^{2-}$	I	0.00	2.01	-
	II	1.94	1.47	-
	III	1.94	1.95	-
$[Mo_3O_9]^-$	I	0.00	3.08	-
	II	0.28	3.36	-
$[Mo_2O_6]$	I	0.00	2.59	-
$[Mo_2O_5]^-$	I	0.09	0.00	3.08
$[Mo_2O_6]^-$	I	0.00	3.15	6.47
$[Mo_2O_7]^-$	I	0.00	2.76	5.34
$[Mo_2O_8]^-$	I	0.00	-	-
	II	1.06	2.45	-
	III	2.30	2.29	-
	IV	2.41	2.41	-
$[Mo_4O_{13}]^{2-}$	I	0.00	2.58	-
	II	0.06	3.40	-
	III	0.41	2.99	-
	IV	0.49	3.15	-
	V	2.74	2.69	-
	VI	3.21	3.21	-
$[Mo_2O_5]$	I	0.17	0.00	-
$[MoO_3]$	I	0.00	1.59	-
$[Mo_3O_{10}]^-$	I	0.00	2.71	-
	II	0.07	2.72	-
	III	0.42	2.49	-
	IV	1.24	3.52	-
$[Mo_2O_7]$	I	0.00	0.00	-
$[Mo_3O_{10}]^{2-}$	I	0.00	3.14	-
	II	1.26	3.68	-
	III	4.99	3.04	-
	IV	4.13	4.26	-
$[Mo_3O_9]$	I	0.00	2.96	-
	II	1.27	3.43	-

[Mo$_2$O$_7$]$^{2-}$	I	0.00	3.33	6.31
[MoO$_3$]	I	0.00	2.92	-
[Mo$_2$O$_6$]$^{2-}$	I	0.00	0.31	-

Calculated structures on the B3LYP/def2TZVP level of theory

I E=0.00   II E=3.39

Figure S5.1: Calculated Molecules of $[MoO_4]^-$ in lowest multiplicity unless marked with *.

I E=0.00   II* E=1.47   III E=1.94

Figure S5.2: Calculated Molecules of $[Mo_2O_8]^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00   II E=0.28

Figure S5.3: Calculated Molecules of $[Mo_3O_9]^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.4: Calculated Molecules of [Mo_2O_6] in lowest multiplicity unless marked with *.

I* E=0.00

Figure S5.5: Calculated Molecules of [Mo_2O_5]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.6: Calculated Molecules of [Mo_2O_6]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.7: Calculated Molecules of [Mo_2O_7]$^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=1.06   III* E=2.29

IV* E=2.41

Figure S5.8: Calculated Molecules of $[Mo_2O_8]^-$ in lowest multiplicity unless marked with *.

I E=0.00   II E=0.06   III E=0.41

IV E=0.49   V* E=2.69   VI E=3.21

Figure S5.9: Calculated Molecules of $[Mo_4O_{13}]^{2-}$ in lowest multiplicity unless marked with *.

I* E=0.00

Figure S5.10: Calculated Molecules of [Mo_2O_5] in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.11: Calculated Molecules of [MoO_3] in lowest multiplicity unless marked with *.

I E=0.00   II E=0.07   III E=0.42
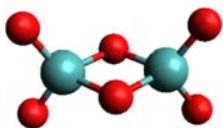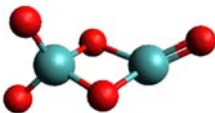

IV E=1.24

Figure S5.12: Calculated Molecules of [Mo_3O_{10}]$^-$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.13: Calculated Molecules of [Mo_2O_7] in lowest multiplicity unless marked with *.

I E=0.00   II E=1.26   III* E=3.04

IV E=4.13

Figure S5.14: Calculated Molecules of [Mo_3O_{10}]$^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00   II E=1.27

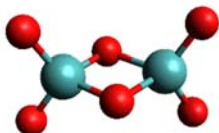Figure S5.15: Calculated Molecules of [Mo_3O_9] in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.16: Calculated Molecules of $[Mo_2O_7]^{2-}$ in lowest multiplicity unless marked with *.

I E=0.00

Figure S5.17: Calculated Molecules of $[MoO_3]^-$ in lowest multiplicity unless marked with *.

I E=0.00

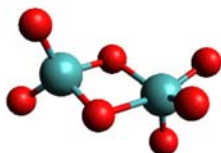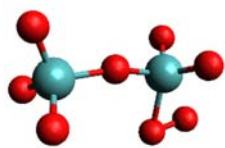Figure S5.18: Calculated Molecules of $[Mo_2O_6]^{2-}$ in lowest multiplicity unless marked with *.

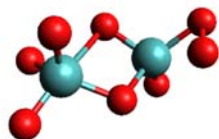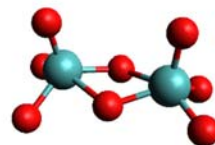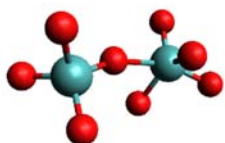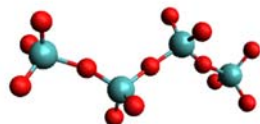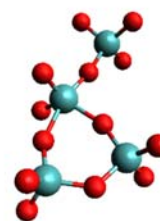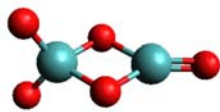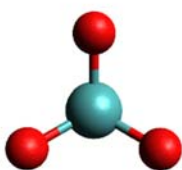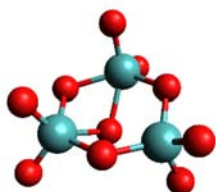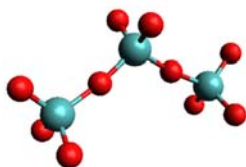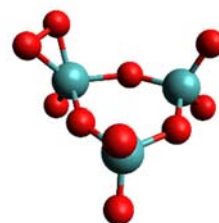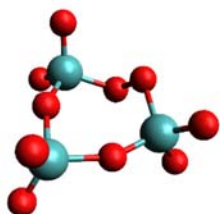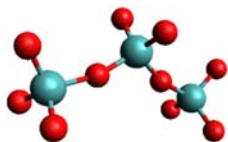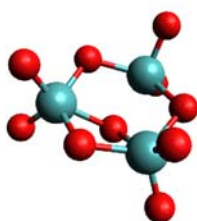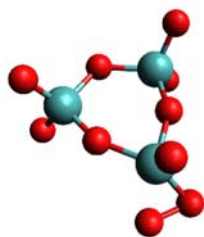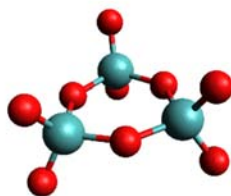# Cartesian coordinates (in Ångstrom) of species optimized at the B3LYP/def2TZVP level of theory along the respective ZPE-corrected energy (in eV)

[MoO₄]⁻ I 1/2
E = -10052.75
```
Mo  -0.040749  -0.000232   0.005774
O    1.478256   0.043220  -1.123659
O   -0.998610  -1.411311  -0.312938
O    0.722564  -0.070159   1.601309
O   -0.988277   1.439470  -0.195027
```

[MoO₄]⁻ I 3/4
E = -10049.62
```
Mo  -0.170746  -0.000002   0.000289
O    1.509296   0.000031  -1.014215
O    1.510971   0.000801   1.013028
O   -1.062639   1.491113  -0.000427
O   -1.061212  -1.491932   0.000096
```

[MoO₄]⁻ II 1/2
E = -10049.37
```
Mo   0.331311   0.006078   0.000635
O    0.927628   1.632169  -0.001112
O    1.463206  -1.305610  -0.001499
O   -1.450052  -0.467431   0.000144
O   -2.680167   0.108963  -0.000868
```

[MoO₄]⁻ II 3/4
E = -10048.83
```
Mo   0.346587   0.000017  -0.049118
O    1.131424   1.537351   0.162892
O    1.135313  -1.535580   0.162992
O   -1.659016  -0.000752  -0.579304
O   -2.427303  -0.001108   0.511287
```

[Mo₂O₈]²⁻ I 1/2
E = -20105.11
```
Mo   2.031274   0.022829  -0.000013
Mo  -1.780135  -0.189791  -0.000025
O   -1.780150   1.781855  -0.000165
O    0.129740  -0.167417  -0.000347
O   -2.196963  -1.080731  -1.417659
O   -2.196472  -1.080601   1.417836
O    2.553554   0.886886  -1.417565
O    2.762861  -1.557596   0.000014
O   -3.144003   1.207469   0.000203
O    2.552950   0.886686   1.417879
```

[Mo₂O₈]²⁻ I 3/4
E = -20103.10
```
Mo  -2.004226  -0.004246  -0.000004
Mo   1.806024  -0.277400   0.000001
O    2.091696   1.813541   0.666652
O   -0.106623   0.026172   0.000196
O    2.377850  -1.065336   1.436676
O    2.378115  -1.063581  -1.437538
O   -2.609252   0.803633   1.422416
O   -2.574122  -1.653052   0.000111
O    2.091702   1.814089  -0.665627
O   -2.608809   0.803176  -1.422866
```

[Mo₂O₈]²⁻ II 1/2
E = -20103.17
```
Mo  -1.571973  -0.096751   0.011829
Mo   1.571616   0.096555  -0.011884
```

```
O    0.151504   0.818905   0.921742
O    2.065120  -1.747336  -0.341740
O   -0.151606  -0.819802  -0.920865
O   -2.061709   1.748452   0.339980
O   -2.062810  -1.140610   1.296038
O    2.869933   0.115042   1.161436
O   -2.870268  -0.115702  -1.161300
O    2.061709   1.142079  -1.294999
```

[Mo₂O₈]²⁻ II 3/4
E = -20103.65
```
Mo   1.794501  -0.187667  -0.020158
Mo  -2.013515   0.004492   0.006444
O   -2.593850   0.122680   1.642239
O   -2.616350   1.355071  -0.908206
O    1.763639   2.126347  -0.069065
O    2.039940  -1.802528   0.545035
O    2.792548   0.755153   1.179163
O   -2.610235  -1.466503  -0.704153
O    2.469855  -0.133695  -1.611232
O   -0.095721   0.005142  -0.001783
```

[Mo₂O₈]²⁻ III 1/2
E = -20103.17
```
Mo   1.571856   0.096801  -0.011912
Mo  -1.571886  -0.096755   0.011794
O   -0.151552  -0.819821  -0.920917
O   -2.869933  -0.115554  -1.161634
O   -2.062300  -1.141565   1.295368
O    2.062972   1.140614  -1.295990
O    2.064560  -1.747709  -0.339685
O   -2.064912   1.747659   0.341375
O    2.869885   0.115792   1.161758
O    0.151436   0.820343   0.920344
```

[Mo₂O₈]²⁻ III 3/4
E = -20103.17
```
Mo   1.571337   0.097187  -0.011551
Mo  -1.571362  -0.097305   0.011559
O   -0.151724  -0.814339  -0.925869
O   -2.870857  -0.109124  -1.160614
O   -2.061534  -1.149294   1.289378
O    2.059166   1.148767  -1.290505
O    2.068347  -1.745651  -0.346635
O   -2.066994   1.745764   0.348195
O    2.872031   0.111284   1.159383
O    0.151699   0.813214   0.926629
```

[Mo₃O₉]⁻ I 1/2
E = -24017.29
```
Mo  -2.179894   0.000601   0.000022
O   -0.727768   1.363180   0.002474
O   -0.728225  -1.361949  -0.001280
O   -3.097569  -0.001480   1.444832
O   -3.096566   0.001776  -1.445430
Mo   1.063465   1.726685   0.000167
O    1.514755   2.602114  -1.390578
O    1.517661   2.603971   1.388780
O    1.870243  -0.000515   0.000425
Mo   1.062639  -1.727197  -0.000117
O    1.514112  -2.603731   1.389891
O    1.515753  -2.603840  -1.389493
```

[Mo₃O₉]⁻ I 3/4
E = -24014.21
```
Mo  -1.297109  -1.580502   0.039374
```

```
O   -1.835768   0.235508  -0.135120
O    0.520075  -1.514991  -0.144298
O   -1.730250  -2.174359   1.579314
O   -2.002942  -2.563078  -1.161075
Mo  -0.814188   1.845717  -0.072620
O   -1.126900   2.821299  -1.436721
O   -1.178080   2.732391   1.338066
O    0.906628   1.257370  -0.071076
Mo   2.149902  -0.361342  -0.043321
O    2.884374   0.503543   1.580685
O    3.360191  -0.793017  -1.147797
```

[Mo₃O₉]⁻ II 1/2
E = -24017.00
```
Mo  -0.000085  -0.741549  -0.000117
Mo  -2.668270   0.366414   0.000054
O   -1.403917  -0.078797   1.293732
O   -1.401244  -0.074175  -1.293266
O   -0.000203  -2.412620   0.000222
O   -3.129711   2.012666   0.002771
O   -4.058096  -0.630215  -0.003091
O    1.404290  -0.079798  -1.293927
O    1.400926  -0.073363   1.293110
Mo   2.668328   0.366362  -0.000003
O    4.058191  -0.630215   0.004357
O    3.129909   2.012580  -0.003562
```

[Mo₃O₉]⁻ II 3/4
E = -24013.93
```
Mo   0.042003  -0.815185   0.000046
Mo  -2.722926   0.374346  -0.000044
O   -1.385188  -0.145597   1.222166
O   -1.384920  -0.145181  -1.222158
O    0.050217  -2.493355   0.000035
O   -3.072653   2.270942   0.000327
O   -4.075934  -0.662498  -0.000268
O    1.423191  -0.085794  -1.300911
O    1.422940  -0.086180   1.300712
Mo   2.658723   0.404143  -0.000008
O    4.092792  -0.532384  -0.000071
O    3.046107   2.072703   0.000204
```

[Mo₂O₆] I 1/2
E = -16008.03
```
Mo  -1.437950  -0.000078  -0.000017
Mo   1.437927  -0.000175   0.000018
O    0.000014  -0.000591  -1.268554
O   -0.000099  -0.000811   1.268556
O   -2.406212  -1.379387  -0.000038
O   -2.406183   1.379271   0.000060
O    2.403969   1.380641   0.000039
O    2.408633  -1.377796  -0.000066
```

[Mo₂O₆] I 3/4
E = -16005.44
```
Mo  -1.419521   0.017385  -0.000020
Mo   1.427781  -0.033487   0.000014
O   -0.001192   0.030483  -1.294340
O   -0.001294   0.030955   1.294318
O   -2.378981  -1.372754   0.000296
O   -2.396845   1.396183  -0.000242
O    2.520153   1.513212   0.000044
O    2.214795  -1.513547  -0.000047
```

[Mo₂O₅]⁻ I 1/2
E = -13954.56

Mo   -1.599492   0.008290   0.001116
Mo   1.316033 -0.023243   0.002923
O   0.004471   1.078742   0.741178
O   -0.025374 -1.065917 -0.770321
O   -3.664216   0.009003 -0.000665
O   2.547572 -1.056039   0.751539
O   2.625707   1.112712 -0.742932

[Mo$_2$O$_5$]$^-$ I 3/4
E = -13961.01
Mo   1.506724 -0.097453   0.035629
Mo   -1.329636   0.030114 -0.008135
O   0.209381 -0.540779 -1.225938
O   0.202791   0.340543   1.295167
O   3.154478   0.361491 -0.137242
O   -2.361555 -1.302883   0.329075
O   -2.134808   1.495157 -0.405408

[Mo$_2$O$_5$]$^-$ I 5/6
E = -13961.10
Mo   -1.576361 -0.168250   0.000000
Mo   1.302046   0.039209   0.000000
O   -0.000283 -0.154932   1.320971
O   -0.000291 -0.154959 -1.320967
O   -3.099595   0.615481   0.000000
O   2.483308 -1.214503   0.000009
O   2.057014   1.586378 -0.000015

[Mo$_2$O$_5$]$^-$ I 7/8
E = -13958.01
Mo   1.543353 -0.000858   0.000003
Mo   -1.332205   0.000379 -0.000001
O   -0.039207   0.001877   1.330533
O   -0.039218   0.001852 -1.330539
O   3.600258   0.002960 -0.000010
O   -2.318910   1.414640 -0.000009
O   -2.311448 -1.418815   0.000014

[Mo$_2$O$_6$]$^-$ I 1/2
E = -16002.18
Mo   -1.619785 -0.024493 -0.145261
Mo   1.650232 -0.012430 -0.069327
O   0.033223 -0.605271 -1.102253
O   -0.008842 -0.412622   1.079664
O   -2.727679 -0.956032   1.223714
O   -2.459645   1.321126 -0.773186
O   1.849347   1.619493   1.039343
O   3.153748 -0.772853 -0.340692

[Mo$_2$O$_6$]$^-$ I 3/4
E = -16011.49
Mo   1.407107 -0.000065 -0.000066
Mo   -1.437733 -0.000049   0.000264
O   0.068613 -0.000947   1.326711
O   0.067022 -0.000016 -1.326175
O   2.381190 -1.412280 -0.000935
O   2.379479   1.413295   0.000220
O   -2.367249   1.441727 -0.000567
O   -2.368265 -1.441182 -0.000293

[Mo$_2$O$_6$]$^-$ I 5/6
E = -16008.34
Mo   1.423978 -0.005283 -0.013685
Mo   -1.446678 -0.076787   0.080261
O   0.138666 -0.507038   1.253058
O   0.089444   0.667138 -1.107510
O   2.270510 -1.327715 -0.711221
O   2.531172   1.177420   0.561113
O   -2.275569   1.699092 -0.214856
O   -2.635044 -1.278032 -0.130109

[Mo$_2$O$_6$]$^-$ I 7/8
E = -16005.02
Mo   -1.492657 -0.047773 -0.028688
Mo   1.523186 -0.057136   0.095294
O   -0.057745   0.445136 -1.122285
O   -0.101144 -0.497137   1.172142
O   -2.379233 -1.412635 -0.570278
O   -2.539770   1.481525   0.557863
O   2.177658   1.783978 -0.333661
O   2.739955 -1.250092 -0.053464

[Mo$_2$O$_7$]$^-$ I 1/2
E = -18049.62
O   -2.797090 -1.633017   0.333135
O   -2.824804   1.382170   0.109184
Mo   -1.814799   0.032882 -0.130282
O   -0.154262 -0.797586   0.756532
O   2.821586 -0.208291 -1.311261
O   2.737277 -1.345203   0.361995
Mo   1.505562   0.120636   0.101706
O   -0.079347   0.302670 -1.106219
O   1.920132   1.493287   1.006661

[Mo$_2$O$_7$]$^-$ I 3/4
E = -18058.10
O   -2.611024 -1.332505   0.214312
O   -2.495757   1.410827 -0.393463
Mo   -1.572446   0.012115 -0.024052
O   -0.375506   0.259456   1.301154
O   2.610705   0.454164 -1.180407
O   1.928265 -1.569528   0.548920
Mo   1.421323 -0.064078 -0.051846
O   -0.133538 -0.296310 -1.222620
O   1.870252   1.346704   1.130571

[Mo$_2$O$_7$]$^-$ I 5/6
E = -18055.34
O   -2.710255 -0.548518 -1.198507
O   -1.815074   1.708195   0.168128
Mo   -1.559874   0.042049 -0.051567
O   -2.380958 -0.904946   1.366025
O   2.425598   1.563829   0.220391
O   2.971527 -1.253564 -0.206304
Mo   1.796023 -0.016975 -0.011754
O   0.097610 -0.203429 -1.069392
O   0.171771 -0.493205   1.052092

[Mo$_2$O$_7$]$^-$ I 7/8
E = -18052.76
O   -2.932789 -1.353341   0.000710
O   -2.530320   1.520596 -0.001336
Mo   -1.821321 -0.044204 -0.000216
O   -0.144661 -0.279251   1.081314
O   2.747041 -0.947380 -1.011984
O   2.745395 -0.951043   1.009687
Mo   1.498613   0.142782   0.000360
O   -0.143856 -0.281013 -1.080558
O   1.953407   1.773897   0.001409

[Mo$_2$O$_8$]$^-$ I 1/2
E = -20104.52
Mo   2.043094   0.023632   0.000002
Mo   -1.733963 -0.251843 -0.000005
O   -1.806173   1.854152 -0.000058
O   0.078970 -0.338460 -0.000126
O   -2.299297 -1.048905 -1.391376
O   -2.299140   1.048981   1.391388
O   2.432059   0.933213 -1.414465
O   2.889774 -1.479729 -0.000501
O   -3.051285   1.394501   0.000132
O   2.432157   0.932316   1.415022

[Mo$_2$O$_8$]$^-$ II 1/2
E = -20103.47
Mo   1.605833   0.111874 -0.042446
Mo   -1.465339 -0.201839 -0.007455
O   0.059206   0.295074 -1.121936
O   -2.967645   0.791380   0.600291
O   -2.080167 -1.627674 -0.708521
O   1.891355   1.509742   0.883115
O   2.909827 -0.038219 -1.149836
O   -2.249813   1.539501 -0.433083
O   2.029262 -1.479415   0.889502
O   -0.329623 -0.518072   1.302447

[Mo$_2$O$_8$]$^-$ II 3/4
E = -20102.07
Mo   1.581987   0.123021 -0.008777
Mo   -1.460779 -0.186306   0.009057
O   -0.025701   0.788672 -0.845652
O   -2.536263   1.551036 -0.611857
O   -2.239112 -1.324935 -0.994156
O   1.998267   1.081221   1.332452
O   2.770064   0.404977 -1.221640
O   -2.612212   0.490316   1.220584
O   2.165869 -1.671382   0.184472
O   -0.157258 -0.987661   0.934325

[Mo$_2$O$_8$]$^-$ III 1/2
E = -20102.22
Mo   1.692592   0.048084   0.030758
Mo   -1.692369   0.045695 -0.033851
O   -2.657351   1.441813 -0.120545
O   -2.359736 -1.164927 -1.091316
O   2.657141   1.446839   0.006872
O   0.024501   0.470253   1.028685
O   2.359223 -1.073935   1.181006
O   -2.201411 -0.940502   1.476071
O   2.200907 -1.055105 -1.397602
O   -0.024446   0.383225 -1.066931

[Mo$_2$O$_8$]$^-$ III 3/4
E = -20102.23
Mo   1.721345 -0.043117   0.027285
Mo   -1.721480   0.043315   0.027272
O   -2.011217 -0.131366   1.692090
O   -2.932007   1.094986 -0.621351
O   2.012834   0.131416   1.691812
O   0.120990 -1.061652 -0.292705
O   2.930559 -1.095467 -0.622617
O   -2.383115 -1.453773 -0.919520
O   2.383797   1.453854 -0.919164
O   -0.121130   1.060965 -0.294968

[Mo$_2$O$_8$]$^-$ IV 1/2
E = -20102.11
Mo   1.971052   0.030347 -0.020782
Mo   -1.893098 -0.011255   0.064268
O   -2.370553 -0.164820   1.690092
O   -1.479880   1.782301 -0.420522
O   0.217163   0.041068   0.372421
O   2.813218 -1.537207   0.590069
O   2.825262   1.262055   0.843860
O   -3.346289   0.005275 -0.903872
O   2.254264   0.173232 -1.701566
O   -1.322441 -1.662135 -0.698781

[Mo$_2$O$_8$]$^-$ IV 3/4
E = -20102.12
Mo   1.965728 -0.031662 -0.020069
Mo   -1.890301   0.011733   0.063835
O   -1.281245   1.651626 -0.694637
O   -1.487061 -1.787635 -0.407857
O   0.217083 -0.059696   0.397661
O   2.799039   1.545291   0.575914

```
O     2.228548   -0.180823   -1.703435
O    -3.330910    0.009968   -0.921082
O     2.844085   -1.250091    0.840255
O    -2.385533    0.175984    1.683409


[Mo4O13]2- I 1/2
E = -34071.74
Mo   -1.531114   -1.092352    0.367516
O    -1.373760   -2.453099   -0.640145
O    -1.611999   -1.648076    1.972884
O     0.000222    0.000055    0.160460
Mo   -4.952395    0.328257   -0.366341
Mo    1.531218    1.092986    0.367121
O     1.612611    1.648481    1.972545
O     1.373060    2.453866   -0.640249
O     3.052583    0.180217   -0.050771
O    -5.968625   -1.013463    0.022867
O    -5.155884    0.754040   -2.027468
O    -5.352539    1.676281    0.636150
O    -3.052418   -0.179072   -0.049531
Mo    4.952311   -0.328781   -0.366290
O     5.351849   -1.674594    0.639402
O     5.155606   -0.758606   -2.026403
O     5.969165    1.013394    0.019727


[Mo4O13]2- I 3/4
E = -34069.16
Mo   -3.027604   -0.001611    0.035629
O    -3.958498    0.005748   -1.412833
O    -3.947617    0.041652    1.490528
O    -1.684905   -1.431564    0.053035
Mo   -0.112151    2.424608   -0.051347
Mo   -0.125006   -2.413846   -0.028804
O    -0.058827   -3.251299   -1.524834
O    -0.103208   -3.561635    1.248836
O     1.292946   -1.272931    0.133641
O    -0.123803    3.331605   -1.506556
O     1.407465    1.375895    0.022077
O    -0.132661    3.506812    1.279004
O    -1.586746    1.339889    0.008685
Mo    2.863218    0.055490    0.086791
O     3.985366    0.989511   -0.851266
O     3.460969    0.035755    1.679965
O     3.557622   -1.448802   -0.842187


[Mo4O13]2- II 1/2
E = -34071.68
Mo    0.123321    2.075398   -0.000181
O    -3.907042    0.001132    1.397758
O    -0.037407    3.058179    1.385851
O    -0.035537    3.059095   -1.385784
O     2.018939    1.838804    0.001013
Mo   -2.904578   -0.000023    0.000123
Mo    2.615942   -0.000027    0.000209
O     3.609186    0.000302   -1.394455
O     3.610431    0.000302    1.394010
O    -1.777141    1.433773   -0.001906
O     0.477979   -0.000068   -0.000228
O    -3.908520   -0.001320   -1.396453
Mo    0.123301   -2.075380   -0.000113
O    -1.776892   -1.433677    0.000943
O    -0.036253   -3.057757   -1.386557
O     2.019129   -1.838726    0.000523
O    -0.036300   -3.059545    1.385086


[Mo4O13]2- II 3/4
E = -34068.34
Mo   -0.010115   -2.087011    0.269107
O     4.452664    0.141565    0.387037
O    -0.014886   -2.522571    1.918902
O     0.243926   -3.517495   -0.630950
O    -1.857548   -1.885920   -0.164993
Mo    2.825103    0.144471   -0.159540
Mo   -2.527355   -0.073055   -0.263804
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O    -3.241930   -0.081080   -1.819445
O    -3.759893   -0.161392    0.922332
O     1.841718   -1.444533    0.061710
O    -0.429960    0.036148    0.043285
O     2.736787   -0.437541   -2.006846
Mo   -0.203047    2.092898    0.217840
O     1.763396    1.540698    0.299200
O     0.098071    3.258184   -0.996770
O    -2.058064    1.803710   -0.088462
O    -0.218360    2.864390    1.741085


[Mo4O13]2- III 1/2
E = -34071.32
Mo    0.558365   -1.123741   -0.001511
Mo    4.141858    0.104254    0.000823
Mo   -1.552346    2.008775   -0.000407
Mo   -3.357261   -0.910425    0.001044
O    -3.149655    1.005269    0.000587
O    -1.487446    3.004305    1.386315
O     2.236219   -0.373382   -0.001168
O     5.093111   -1.342047   -0.000990
O    -4.200637   -1.435529    1.402332
O    -1.673651   -1.451504   -0.000681
O    -0.231076    0.762437   -0.000235
O     4.507657    1.036096   -1.410917
O     4.505892    1.031452    1.416065
O    -1.488401    3.003021   -1.388102
O    -4.203662   -1.436078   -1.398225
O     0.595846   -2.110203    1.381729
O     0.595070   -2.107864   -1.386446


[Mo4O13]2- III 3/4
E = -34068.75
Mo   -0.419519   -1.174126   -0.029976
Mo   -4.088849    0.054151    0.040499
Mo    1.505617    2.027272    0.002155
Mo    3.324055   -0.897845    0.010392
O     3.121995    1.014209    0.019893
O     1.453751    3.012946   -1.401822
O    -2.373323   -0.400780   -0.029642
O    -5.092405   -0.888119   -1.031027
O     4.201387   -1.390414   -1.383740
O     1.639823   -1.506108   -0.019321
O     0.196872    0.780743   -0.011915
O    -4.696221   -0.000509    1.647349
O    -4.375756    1.782885   -0.663066
O     1.421495    3.014042    1.404037
O     4.159579   -1.407946    1.423621
O    -0.663224   -2.018095   -1.507795
O    -0.680815   -2.042474    1.432273


[Mo4O13]2- IV 1/2
E = -34071.25
Mo   -2.863716   -1.220833   -0.552409
Mo   -1.909780    2.012432    0.129621
Mo    0.616624   -0.994554    0.806838
O    -3.154569    0.621174   -0.423884
O    -1.107072   -1.452820   -0.120959
O    -0.424165    1.098004    0.370486
O    -1.745290    3.200490   -1.100403
O    -2.453682    2.758729    1.579354
O    -3.891464   -2.056490    0.528514
O    -3.189168   -1.769041   -2.138537
O     2.075840   -0.124299    0.116377
O     4.125542   -0.192925   -2.104621
Mo    3.941070    0.216888   -0.433488
O     4.326963    1.885996   -0.200804
O     5.008765   -0.753836    0.521078
O     0.429024   -0.674096    2.462277
O     1.132238   -2.614031    0.770671


[Mo4O13]2- IV 3/4
E = -34068.59
Mo   -2.478002   -1.365306   -0.558723
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Mo   -1.732311    1.931350    0.111269
Mo    0.840283   -0.644210    0.985948
O    -2.854233    0.503990   -0.493107
O    -0.821485   -1.493914    0.118437
O    -0.177334    1.139115    0.521171
O    -1.552467    3.105008   -1.124726
O    -2.434968    2.676523    1.488541
O    -3.612571   -2.226740    0.398451
O    -2.566894   -1.933797   -2.173888
O     2.534492    0.477098    1.118530
O     1.754663   -0.712415   -1.077635
Mo    3.234685    0.210728   -0.582048
O     3.472287    1.672640   -1.466200
O     4.698999   -0.705463   -0.619583
O     0.446977   -0.862797    2.612215
O     1.823097   -2.335203    0.926452


[Mo4O13]2- V 1/2
E = -34069.00
Mo   -2.008116    1.930871   -0.205344
O     2.259474   -1.309030    0.070559
O    -2.255506   -2.267801   -1.810214
O    -2.812686    2.699089    1.108681
O    -2.702612    0.303741   -0.533443
Mo    3.603614   -0.132349   -0.382849
Mo   -2.187142   -1.775035   -0.178251
O    -3.920920   -1.507967    0.572302
O    -2.146087   -3.234950    0.753134
O    -2.133291    2.935416   -1.598222
O     2.495858    1.286278    0.015899
O     4.077254   -0.208418   -2.036039
Mo    0.936092    0.134021    0.754949
O     4.996992   -0.235956    0.625389
O    -0.218036    1.626559    0.229309
O    -0.480165   -1.057060    0.233830
O     1.031370    0.143175    2.429162


[Mo4O13]2- V 3/4
E = -34069.04
Mo    2.297500    1.773702    0.144680
O    -2.015965   -1.247326    0.003486
O     3.197850    0.106113    0.368984
O     2.370057    2.681488    1.596883
O     3.028692    2.670440   -1.121466
Mo   -3.586509    0.082890    0.399388
Mo    2.472782   -1.662210    0.246557
O     3.322304   -2.569061   -0.936897
O     2.595209   -2.453006    1.763603
O    -3.572851    0.329547    2.093655
O    -2.179026    1.152993   -0.388657
O    -4.476442   -1.574527    0.006446
Mo   -0.782506   -0.120283   -0.817138
O    -4.977625    0.896158   -0.253416
O     0.609301    1.301869   -0.265468
O     0.764210   -1.389974   -0.229348
O    -0.772362   -0.293736   -2.498614


[Mo4O13]2- VI 1/2
E = -34068.53
Mo    3.735656    0.904513   -0.119601
Mo    1.317942   -1.018647    0.163967
Mo   -1.687659   -0.945620   -0.151975
Mo   -3.828620    0.996941    0.066307
O     2.366454   -0.147338    1.328949
O     2.456143   -0.005991   -1.145617
O    -2.853364   -0.004202    1.276294
O    -2.757143    0.408598   -1.312965
O    -0.090380   -0.561693   -1.141277
O    -0.257133   -1.170945    1.194936
O     1.872053   -2.616359   -0.029732
O    -2.274896   -2.480820   -0.522675
O    -5.473990    0.501929   -0.073684
O     4.858363    0.180704    1.249473
O     3.232276    2.500873    0.225224
O    -3.755906    2.695224    0.337811
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O   5.106595   1.029787   -1.169903

[Mo₃O₁₀]⁻ I 3/4
E = -26060.69
Mo   -1.769644   -0.844277   -0.015423
O   -0.000179   -1.782776   0.150295
O   -1.799621   0.993226   0.361369
O   -2.721712   -1.249687   -1.358875
O   -2.501661   -1.585854   1.330899
Mo   1.769457   -0.844555   -0.015473
O   2.721513   -1.250261   -1.358843
O   0.000012   -0.059860   -1.078762
Mo   0.000163   1.733511   -0.065379
O   1.799865   0.992986   0.361229
O   0.000526   2.214537   1.843192
O   2.501232   -1.586195   1.330946
O   0.000151   3.079317   -1.076004

[Mo₃O₁₀]⁻ II 1/2
E = -26063.34
Mo   3.244053   -0.618424   -0.000347
O   1.524390   0.240052   -0.004548
O   -3.459011   -1.551717   -1.456622
O   3.375754   -1.729455   -1.341380
O   3.461116   -1.545524   1.459817
Mo   0.000001   1.305769   -0.000080
O   0.003950   2.306602   1.368512
O   -0.003965   2.307433   -1.368068
O   -1.524390   0.240040   0.003649
Mo   -3.244048   -0.618418   0.000434
O   -3.377744   -1.723792   1.345368
O   4.496003   0.546993   -0.114993

[Mo₃O₁₀]⁻ II 3/4
E = -26060.68
Mo   -2.820302   -0.800465   -0.000190
O   -1.453296   0.406371   -0.000356
O   2.821539   -1.763224   1.417389
O   -2.825113   -1.760570   1.419024
O   -2.821470   -1.764478   -1.416566
Mo   -0.000007   1.735657   -0.000004
O   0.000077   2.659222   -1.439857
O   -0.000548   2.658516   1.440297
O   1.453538   0.406669   -0.000350
Mo   2.820345   -0.800408   0.000236
O   4.484592   0.105701   0.000792
O   2.824945   -1.761649   -1.418267
O   -4.484450   0.105826   -0.002327

[Mo₃O₁₀]⁻ III 1/2
E = -26062.98
Mo   2.056069   -0.002180   -0.199917
O   0.594358   1.376011   -0.129241
O   0.592183   -1.377561   -0.119893
O   2.746820   -0.007013   -1.733117
O   3.168428   0.735260   1.179124
Mo   -1.187990   1.729785   0.024622
O   -1.536399   2.544286   1.479733
O   -1.748941   2.666859   -1.282573
O   -1.988342   0.001860   0.017637
Mo   -1.191598   -1.727693   0.025489
O   -1.748920   -2.661174   -1.285815
O   -1.548040   -2.544204   1.477528
O   3.167330   -0.733859   1.183100

[Mo₃O₁₀]⁻ III 3/4
E = -26060.91
Mo   1.900555   -0.356502   -0.276535
O   1.036165   1.353880   -0.096901
O   0.427157   -1.481204   -0.056674
O   2.522689   -0.498187   -1.831331
O   3.211224   -0.013423   1.332803
Mo   -0.743414   1.978659   0.037047
O   -1.002027   2.717238   1.557795

[Mo₄O₁₃]²⁻ VI 3/4
E = -34068.53
Mo   -3.732456   -0.904855   -0.118350
Mo   -1.317517   1.020352   0.162161
Mo   1.686622   0.945376   -0.149877
Mo   3.827410   -0.997477   0.065474
O   -2.379475   0.158353   1.324627
O   -2.450866   0.003713   -1.146017
O   2.851804   -0.000411   1.277791
O   2.756269   -0.405042   -1.312974
O   0.090789   0.572134   -1.145806
O   0.253678   1.156222   1.199855
O   -1.862547   2.620703   -0.033414
O   2.275024   2.482356   -0.510635
O   5.472673   -0.501459   -0.072339
O   -4.869478   -0.192930   1.246604
O   -3.227998   -2.500915   0.225630
O   3.755101   -2.696624   0.331920
O   -5.101285   -1.028927   -1.172136

[Mo₂O₅] I 1/2
E = -13958.16
Mo   -1.477639   0.229540   -0.000015
Mo   1.219695   -0.062158   0.000080
O   -0.155346   0.333847   -1.334245
O   -0.155559   0.333723   1.334362
O   -2.623093   -0.988195   -0.000139
O   2.469699   1.083101   -0.000179
O   1.818506   -1.641228   -0.000138

[Mo₂O₅] I 3/4
E = -13958.33
Mo   -1.557725   0.000076   0.000053
Mo   1.330858   0.000096   -0.000040
O   -0.081852   -0.002438   -1.277893
O   -0.081824   0.003314   1.277821
O   -3.238368   0.000016   -0.000123
O   2.298060   1.389095   -0.002582
O   2.295038   -1.390891   0.002706

[MoO₃] I 1/2
E = -8001.71
Mo   -0.000072   -0.000001   -0.196361
O   0.813210   -1.399526   0.343591
O   0.805996   1.403693   0.343590
O   -1.618829   -0.004162   0.343716

[MoO₃] I 3/4
E = -8000.13
Mo   0.021510   -0.000423   -0.138481
O   0.848240   1.428497   0.249889
O   -1.836330   -0.013483   0.226409
O   0.875161   -1.412793   0.250729

[Mo₃O₁₀]⁻ I 1/2
E = -26063.40
Mo   0.931609   1.619691   -0.014010
O   -0.967018   1.594001   0.396982
O   2.132760   -0.000229   -0.171401
O   1.333204   2.852127   -1.104981
O   1.547813   2.062781   1.511642
Mo   -1.938259   0.000234   0.039934
O   -2.638441   -0.000049   -1.505672
O   0.306425   -0.000292   -1.133608
Mo   0.931081   -1.619968   -0.013977
O   -0.967605   -1.593582   0.397715
O   1.547266   -2.062960   1.511704
O   -3.230079   0.001028   1.139697
O   1.332412   -2.852599   -1.104799

O   -1.119131   3.066147   -1.226879
O   -1.653989   0.398835   -0.119721
Mo   -1.560913   -1.564771   0.030678
O   -2.221767   -2.376757   -1.327812
O   -2.062392   -2.152993   1.562841
O   2.981872   -1.314815   1.302134

[Mo₃O₁₀]⁻ IV 1/2
E = -26062.16
Mo   1.964624   -0.954518   0.017225
O   1.338008   0.745483   0.156067
O   3.136809   -1.086098   -1.210099
O   2.619787   -1.494441   1.492681
Mo   -0.007905   2.221197   -0.000082
O   -0.176157   3.135163   1.436819
O   0.154965   3.138746   -1.435317
O   -1.345558   0.737854   -0.159124
Mo   -1.957805   -0.967120   -0.017294
O   -2.610094   -1.513850   -1.491504
O   -3.127743   -1.106548   1.211304
O   -0.456215   -2.064678   0.562201
O   0.471898   -2.064319   -0.562231

[Mo₃O₁₀]⁻ IV 3/4
E = -26059.89
Mo   2.374248   -0.860143   0.017093
O   1.385225   0.775827   0.368806
O   3.131344   -0.875397   -1.516282
O   3.364665   -1.302634   1.339408
Mo   0.000441   1.947039   -0.000190
O   -0.333038   2.925374   1.357653
O   0.332115   2.928944   -1.355949
O   -1.382242   0.774131   -0.371130
Mo   -2.374441   -0.859639   -0.017465
O   -3.361409   -1.305831   -1.341146
O   -3.135405   -0.869560   1.514081
O   -0.661457   -2.122410   0.058058
O   0.658895   -2.121545   -0.050550

[Mo₂O₇] I 1/2
E = -18052.53
O   -2.386962   -1.513927   0.572054
O   -2.478623   0.215259   -1.592872
Mo   -1.880993   -0.010541   -0.022484
O   -2.694204   1.289842   1.041488
O   2.386537   -1.512697   -0.575795
O   2.480246   0.211814   1.592775
Mo   1.881025   -0.010705   0.022497
O   0.000008   0.128439   0.001043
O   2.692834   1.292812   -1.038756

[Mo₂O₇] I 3/4
E = -18052.53
O   -2.385401   -1.511142   0.580577
O   -2.480050   0.206395   -1.593448
Mo   -1.880957   -0.010732   -0.022349
O   -2.693672   1.295967   1.034400
O   2.385504   -1.514998   -0.570423
O   2.479462   0.216919   1.592219
Mo   1.880918   -0.010581   0.022361
O   -0.000025   0.129848   -0.000776
O   2.694386   1.288903   -1.042609

[Mo₃O₁₀]²⁻ I 1/2
E = -26066.09
Mo   3.390934   -0.549755   0.000032
O   1.519921   0.059854   0.000758
O   -3.684467   -1.503116   -1.414071
O   3.684339   -1.502902   -1.414220
O   3.684223   -1.506292   1.412024
Mo   -0.000051   1.104950   0.000049
O   -0.000251   2.114982   1.377924
O   -0.000145   2.113400   -1.378975

O   -1.519696   0.059368   0.000646
Mo   -3.390865   -0.549803   0.000028
O   -4.447189   0.821326   0.001461
O   -3.684429   -1.506037   1.412171
O   4.447603   0.821112   0.001707

[Mo₃O₁₀]²⁻ I 3/4
E = -26062.95
Mo   3.336342   -0.676454   -0.000042
O   1.514242   0.100227   0.000652
O   -3.587839   -1.557711   -1.412646
O   3.851042   -1.454961   -1.452963
O   3.851251   -1.457647   1.451415
Mo   0.012629   1.169710   -0.000084
O   0.001075   2.170497   1.387239
O   0.002754   2.170937   -1.387108
O   -1.499668   0.097204   -0.001243
Mo   -3.334375   -0.588256   -0.000015
O   -4.452013   0.735165   -0.001520
O   -3.587437   -1.554198   1.415081
O   3.829964   1.249233   0.001832

[Mo₃O₁₀]²⁻ II 1/2
E = -26064.83
Mo   -1.699092   -0.695099   -0.014163
O   -0.271951   -1.999205   0.280546
O   -1.595561   1.235190   0.280654
O   -2.733689   -1.118345   -1.316436
O   -2.560850   -1.047988   1.437702
Mo   1.451587   -1.123736   -0.014163
O   2.335430   -1.808010   -1.316482
O   0.000024   0.000158   -0.982978
Mo   0.247501   1.818840   -0.014182
O   1.867553   0.764126   0.280738
O   0.372839   2.741176   1.437963
O   2.187936   -1.693884   1.437665
O   0.398294   2.926755   -1.316201

[Mo₃O₁₀]²⁻ II 3/4
E = -26062.40
Mo   -1.626361   -0.878518   -0.015153
O   -0.000111   -1.897810   0.316871
O   -1.781593   1.002041   0.339433
O   -2.509416   -1.389281   -1.394804
O   -2.516055   -1.466524   1.338453
Mo   1.626222   -0.878747   -0.015186
O   2.509153   -1.389655   -1.394861
O   -0.000013   0.094308   -1.068581
Mo   0.000117   1.780014   -0.085262
O   1.781706   1.001841   0.339388
O   0.000350   2.217377   1.866911
O   2.515932   -1.466782   1.338396
O   0.000164   3.175052   -1.074301

[Mo₃O₁₀]²⁻ III 1/2
E = -26061.09
Mo   1.806403   -0.765869   0.017742
O   1.786234   1.105743   0.166118
O   2.886789   -1.064889   -1.290016
O   2.436853   -1.307058   1.522262
Mo   -0.002891   1.817408   -0.000079
O   -0.135724   2.737835   1.454080
O   0.127739   2.740950   -1.452646
O   -1.790012   1.101503   -0.167292
Mo   -1.804100   -0.770246   -0.017820
O   -2.433946   -1.313838   -1.521752
O   -2.881906   -1.072155   1.291378
O   -0.539448   -2.202942   0.476408
O   0.546508   -2.201943   -0.477716

[Mo₃O₁₀]²⁻ III 3/4
E = -26063.05
Mo   2.099150   0.205630   0.069601

O   0.513068   1.220592   0.068264
O   3.195818   1.355219   -0.680424
O   2.673516   -0.009046   1.666591
Mo   -1.355163   1.726215   -0.022987
O   -1.880570   2.572807   1.389937
O   -1.776969   2.477075   -1.521758
O   -1.894682   -0.184386   0.009048
Mo   -1.028249   -1.831815   -0.000430
O   -1.438884   -2.699713   -1.436732
O   -1.515677   -2.765426   1.368945
O   0.734968   -1.486026   0.058858
O   2.881792   -1.006252   -1.165195

[Mo₃O₁₀]²⁻ IV 1/2
E = -26061.96
Mo   2.085181   -0.000090   -0.207604
O   0.554939   1.275809   -0.136631
O   0.555222   -1.275925   -0.134556
O   2.735123   -0.000915   -1.780280
O   3.215007   0.737589   1.208761
Mo   -1.201191   1.705644   0.026531
O   -1.523728   2.487760   1.534275
O   -1.728604   2.656636   -1.317349
O   -2.106415   0.000285   0.010526
Mo   -1.201086   -1.705760   0.026734
O   -1.727382   -2.656404   -1.317794
O   -1.524781   -2.487946   1.534159
O   3.215374   -0.735804   1.209173

[Mo₃O₁₀]²⁻ IV 3/4
E = -26061.82
Mo   1.990355   -0.314602   -0.210072
O   0.887414   1.372465   -0.217577
O   0.413263   -1.425956   0.016482
O   2.638660   -0.566142   -1.763768
O   3.025200   -1.155371   1.211814
Mo   -0.835640   1.936629   0.027798
O   -0.995130   2.871062   1.471105
O   -1.371459   2.868027   -1.324761
O   -1.776272   0.370363   0.165829
Mo   -1.522579   -1.609976   0.045082
O   -2.121366   -2.248109   -1.450395
O   -2.026813   -2.451812   1.474336
O   3.257792   0.302209   1.137192

[Mo₃O₉] I 1/2
E = -24013.84
Mo   1.684567   1.148680   -0.000070
O   1.541583   -0.742259   -0.000862
O   -0.128028   1.706053   -0.000200
O   2.503727   1.707069   1.358646
O   2.504609   1.708147   -1.357800
Mo   0.152529   -2.033172   -0.000079
O   0.226751   -3.023434   -1.357556
O   0.226798   -3.021338   1.358921
O   -1.413627   -0.963967   -0.000775
Mo   -1.837091   0.884495   -0.000028
O   -2.730607   1.314575   1.358503
O   -2.731237   1.315142   -1.357945

[Mo₃O₉] I 3/4
E = -24010.88
Mo   1.040168   1.740671   -0.010076
O   1.808582   0.006476   -0.006118
O   -0.812261   1.378561   -0.082024
O   1.441942   2.596068   1.386035
O   1.528897   2.639625   -1.350398
Mo   1.053231   -1.733500   -0.009677
O   1.545357   -2.626971   -1.352400
O   1.464268   -2.587844   1.384385
O   -0.801405   -1.383027   -0.078332
Mo   -2.113340   -0.007306   0.030981
O   -2.843363   -0.008926   1.540681
O   -3.227329   -0.013252   -1.500778

[Mo₃O₉] II 1/2
E = -24012.57
Mo   -0.000005   -0.731127   -0.000335
Mo   -2.699817   0.373422   0.000067
O   -1.361674   -0.155410   1.242263
O   -1.363530   -0.159304   -1.242516
O   0.001003   -2.396658   0.000754
O   -3.083203   2.015946   -0.002127
O   -4.114333   -0.544548   0.002599
O   1.361561   -0.155830   -1.242536
O   1.362570   -0.156607   1.242348
Mo   2.699637   0.373698   0.000022
O   4.112320   -0.547057   -0.000424
O   3.086251   2.015504   0.000930

[Mo₃O₉] II 3/4
E = -24010.41
Mo   -0.045083   -0.775083   -0.000040
Mo   -2.674879   0.391602   -0.000008
O   -1.362011   -0.109779   1.275766
O   -1.362305   -0.110524   -1.275883
O   -0.036202   -2.433499   0.000357
O   -3.069287   2.037758   -0.000428
O   -4.085011   -0.542799   0.000479
O   1.490873   -0.063303   -1.301827
O   1.490856   -0.062601   1.301860
Mo   2.645478   0.342606   -0.000046
O   4.070578   -0.592459   0.000105
O   3.253553   2.091797   0.000060

[Mo₂O₇]²⁻ I 1/2
E = -18059.79
Mo   -1.910556   -0.000009   -0.000005
Mo   1.910556   0.000004   -0.000010
O   0.000002   -0.000067   0.000022
O   2.503478   1.633053   -0.108827
O   2.503535   -0.722241   1.468673
O   -2.503570   -1.632901   0.110634
O   -2.503516   0.912287   1.358811
O   2.503674   -0.910753   -1.359797
O   -2.503602   0.720650   -1.469442

[Mo₂O₇]²⁻ I 3/4
E = -18056.45
Mo   -1.871176   -0.107663   0.000003
Mo   1.953175   0.019484   -0.000002
O   0.039674   0.026878   0.000047
O   2.556578   1.652215   -0.000866
O   2.536606   -0.801810   1.420483
O   -2.567831   -1.692327   0.000043
O   -2.765955   1.040671   1.152663
O   2.536513   -0.803264   -1.419685
O   -2.766078   1.040577   -1.152690

[Mo₂O₇]²⁻ I 5/6
E = -18053.48
Mo   -1.904326   0.199832   0.046189
Mo   1.935768   -0.014640   0.016410
O   0.047037   0.090715   0.032232
O   2.455192   -1.569871   0.612056
O   2.536210   0.212793   -1.608003
O   -2.588907   1.743904   -0.338186
O   -2.121125   -1.596733   -0.948048
O   2.590943   1.240129   1.041108
O   -3.084418   -1.093191   0.880197

[Mo₂O₇]²⁻ I 7/8
E = -18050.13
Mo   -1.889394   0.160408   0.053593
Mo   1.989463   -0.060644   0.039226
O   0.058598   -0.006964   -0.027136

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O      2.695718   -0.010565    1.635145
O      2.775824   -1.144058   -1.081241
O     -2.614716    1.701129   -0.273592
O     -2.115072   -1.636711   -0.951244
O      1.675846    1.739522   -0.722140
O     -3.001562   -1.166116    0.932907


[MoO₃]⁻ I 1/2
E = -8004.93
Mo    0.000055   -0.000010   -0.123755
O    -1.054359   -1.346055    0.216553
O     1.692946   -0.239827    0.216611
O    -0.638875    1.585932    0.216551


[MoO₃]⁻ I 3/4
E = -8002.00
Mo    0.060361    0.000113    0.000178
O     0.816070    1.577822   -0.000338
O    -1.973516   -0.013891   -0.000255
O     0.840550   -1.564526   -0.000339


[Mo₂O₆]²⁻ I 1/2
E = -16009.46
Mo   -1.318334    0.000025   -0.000007
Mo    1.318373    0.000001    0.000008
O    -0.000025    0.000057   -1.455472
O    -0.000042   -0.000015    1.455474
O    -2.234159   -1.494388   -0.000007
O    -2.233870    1.494602    0.000023
O     2.234309    1.494374    0.000010
O     2.233584   -1.494765   -0.000030


[Mo₂O₆]²⁻ I 3/4
E = -16009.15
Mo   -1.534406   -0.000067   -0.000005
Mo    1.534402   -0.000054    0.000005
O     0.000010   -0.000352   -1.228029
O    -0.000023   -0.000977    1.228029
O    -2.461322   -1.486055   -0.000122
O    -2.460399    1.486499    0.000130
O     2.459494    1.487015    0.000129
O     2.462265   -1.485498   -0.000136
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