

**SUPPLEMENTARY MATERIAL TO ACCOMPANY:**

**ION MOBILITY SPECTROMETRY AND TANDEM MASS SPECTROMETRY ANALYSIS OF  
ESTRADIOL GLUCURONIDE ISOMERS**

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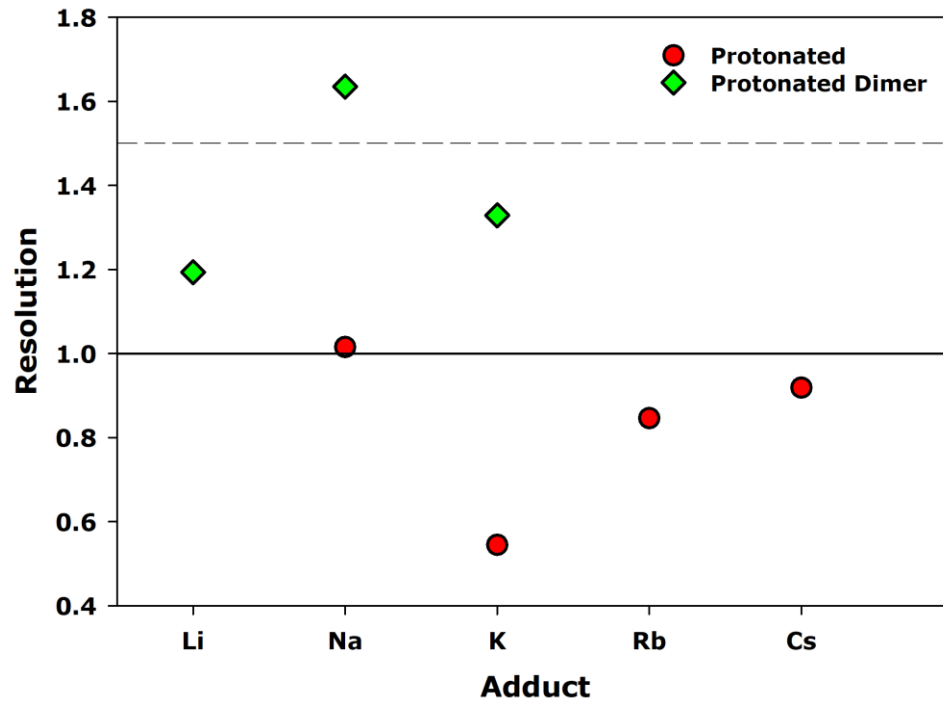
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**Figure S1.** Scatter plot of the resolution values of standards of the neutral estradiol glucuronide with alkali adducts (red circle) and the dimeric neutral estradiol glucuronide with alkali adducts (green diamond).

**Table S1.** Material sources for chemicals used in this study.

<b>Material</b>	<b>Source</b>	<b>Location</b>
<b>Polyalanine</b>	Sigma-Aldrich	St. Louis, MO
<b>Estradiol-3-Glucuronide</b>	Cayman Chemical	Ann Arbor, MI
<b>Estradiol-17-Glucuronide</b>	Cayman Chemical	Ann Arbor, MI
<b>HPLC Grade Water</b>	Sigma-Aldrich	St. Louis, MO
<b>Methanol</b>	Fisher Scientific	Pittsburg, PA
<b>Acetonitrile</b>	Fisher Scientific	Pittsburg, PA

**Table S2.** Resolution values for standards of the neutral estradiol glucuronide with alkali adducts ( $[M+H+X]^+$ ) and the dimeric neutral estradiol glucuronide with alkali adducts ( $[2M+2H+X]^+$ ). N.D. represents not detected.

<b>X</b>	<b>[M+H+X]<sup>+</sup></b>	<b>[2M+2H+X]<sup>+</sup></b>
<b>Li</b>	N.D.	1.19
<b>Na</b>	1.02	1.64
<b>K</b>	0.54	1.33
<b>Rb</b>	0.85	N.D.
<b>Cs</b>	0.92	N.D.

**Table S3.** Collision cross sections, standard deviation ( $n = 4$ ), error, and resolution of the sodiated dimer of the estradiol glucuronide isomers as standards and in a mixture.

<b>Species</b>	<b>Standard CCS (<math>\text{\AA}^2</math>)</b>	<b>Mixture CCS (<math>\text{\AA}^2</math>)</b>	<b>% Error</b>	<b>Resolution</b>
<b>E3G</b>	$302.5 \pm 3.4$	$306.7 \pm 2.6$	1.38	1.23
<b>E17G</b>	$310.5 \pm 1.9$	$315.6 \pm 2.8$	1.64	

**Table S4.** Collision cross sections ( $\Omega$ ) and standard deviation ( $n = 4$ ) of metal adducts of estradiol-3-glucuronide (E3G) and estradiol-17-glucuronide (E17G).

Adduct	E3G		E17G	
	$\Omega$ ( $\text{\AA}^2$ )	Standard Deviation	$\Omega$ ( $\text{\AA}^2$ )	Standard Deviation
[M+Li] <sup>+</sup>	212.3	0.89	214.0	1.06
[M+Na] <sup>+</sup>	216.8	0.92	223.2	1.02
[M+K] <sup>+</sup>	220.5	1.18	224.9	0.93
[M+Rb] <sup>+</sup>	221.0	1.03	225.0	1.03
[M+Cs] <sup>+</sup>	225.7	1.98	227.0	1.03
[M-H+2Na] <sup>+</sup>	230.4	1.04	234.8	1.11
[M-H+2K] <sup>+</sup>	233.8	2.48	231.4	2.04
[M-H+2Rb] <sup>+</sup>	254.1	1.21	254.8	1.26
[M-H+2Cs] <sup>+</sup>	281.5	1.57	281.6	1.51
[2M+H+Li] <sup>+</sup>	305.8	1.54	311.3	1.37
[2M+H+Na] <sup>+</sup>	302.5	1.72	310.5	0.93
[2M+H+K] <sup>+</sup>	303.1	1.84	312.6	0.41

**Table S5.** Fraction of  $m/z$  271 and standard deviation ( $n = 4$ ) of metal adducts of E3G andE17G, where  $\text{Abundance}_{271} = \sum \frac{PA_{271}}{PA_{271} + PA_{447}}$ .

Collision Energy (V)	E3G		E17G	
	Abundance <sub>271</sub> (%)	Standard Deviation	Abundance <sub>271</sub> (%)	Standard Deviation
<b>0</b>	0.02	0.007	0.01	0.002
<b>5</b>	0.02	0.003	0.00	0.001
<b>10</b>	0.03	0.004	0.01	0.003
<b>15</b>	0.17	0.016	0.06	0.008
<b>20</b>	0.81	0.050	0.26	0.024
<b>25</b>	2.64	0.038	0.68	0.033
<b>30</b>	11.42	0.177	2.21	0.064
<b>35</b>	54.42	0.448	9.89	0.355
<b>40</b>	92.68	0.191	33.56	0.559
<b>45</b>	98.78	0.107	65.77	0.609