

# Immobilization of Polyiodide Redox Species in Porous Carbon for Battery-Like Electrodes in Eco-Friendly Hybrid Electrochemical Capacitors

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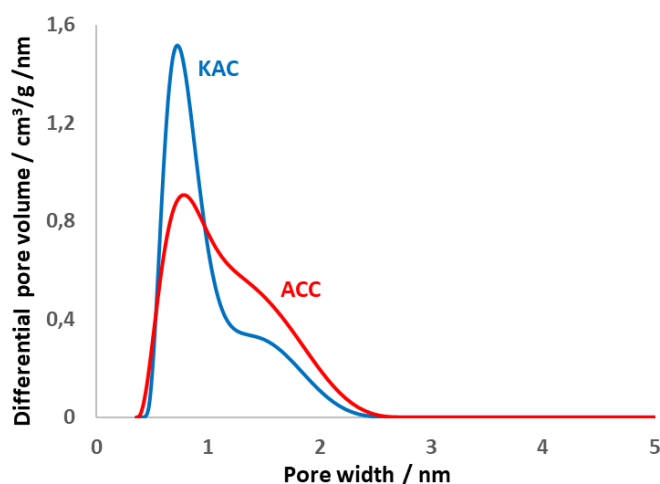
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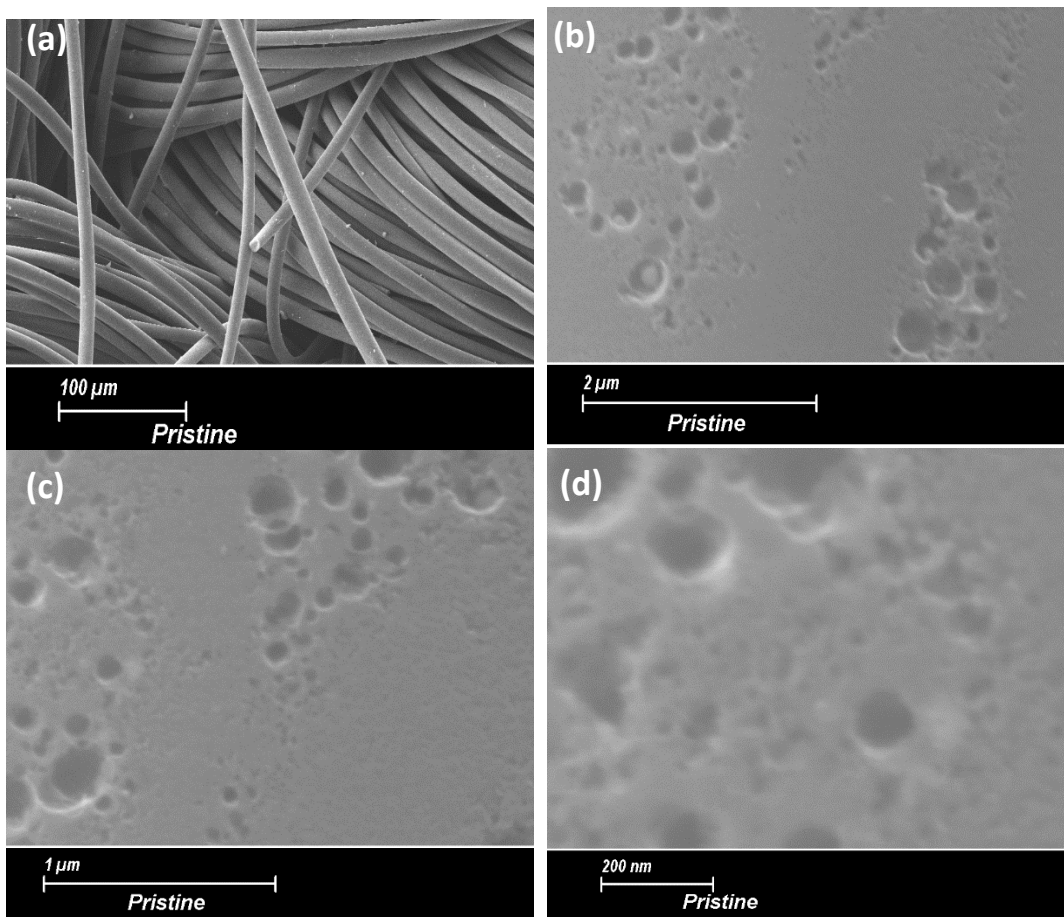
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**Table 1.** Porous data of KOH activated carbon (KAC) and activated carbon cloth (ACC).

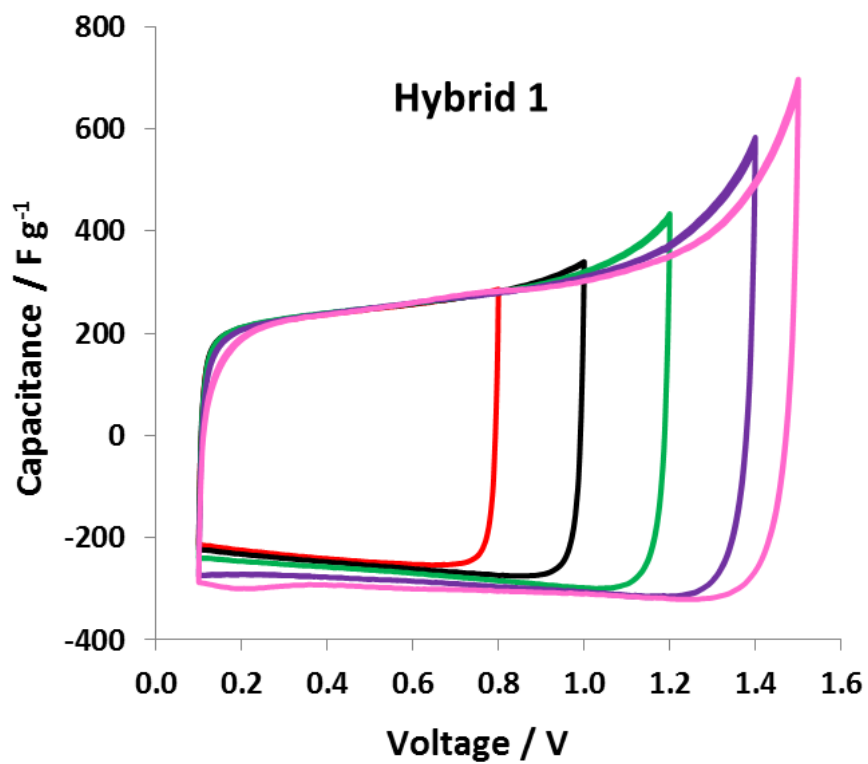
Activated carbon	Surface area (m <sup>2</sup> g <sup>-1</sup> )	Micropore volume (cm <sup>3</sup> g <sup>-1</sup> )	Mesopore volume (cm <sup>3</sup> g <sup>-1</sup> )
ACC	2320	0.84	0.06
KAC	2300	0.88	0.02



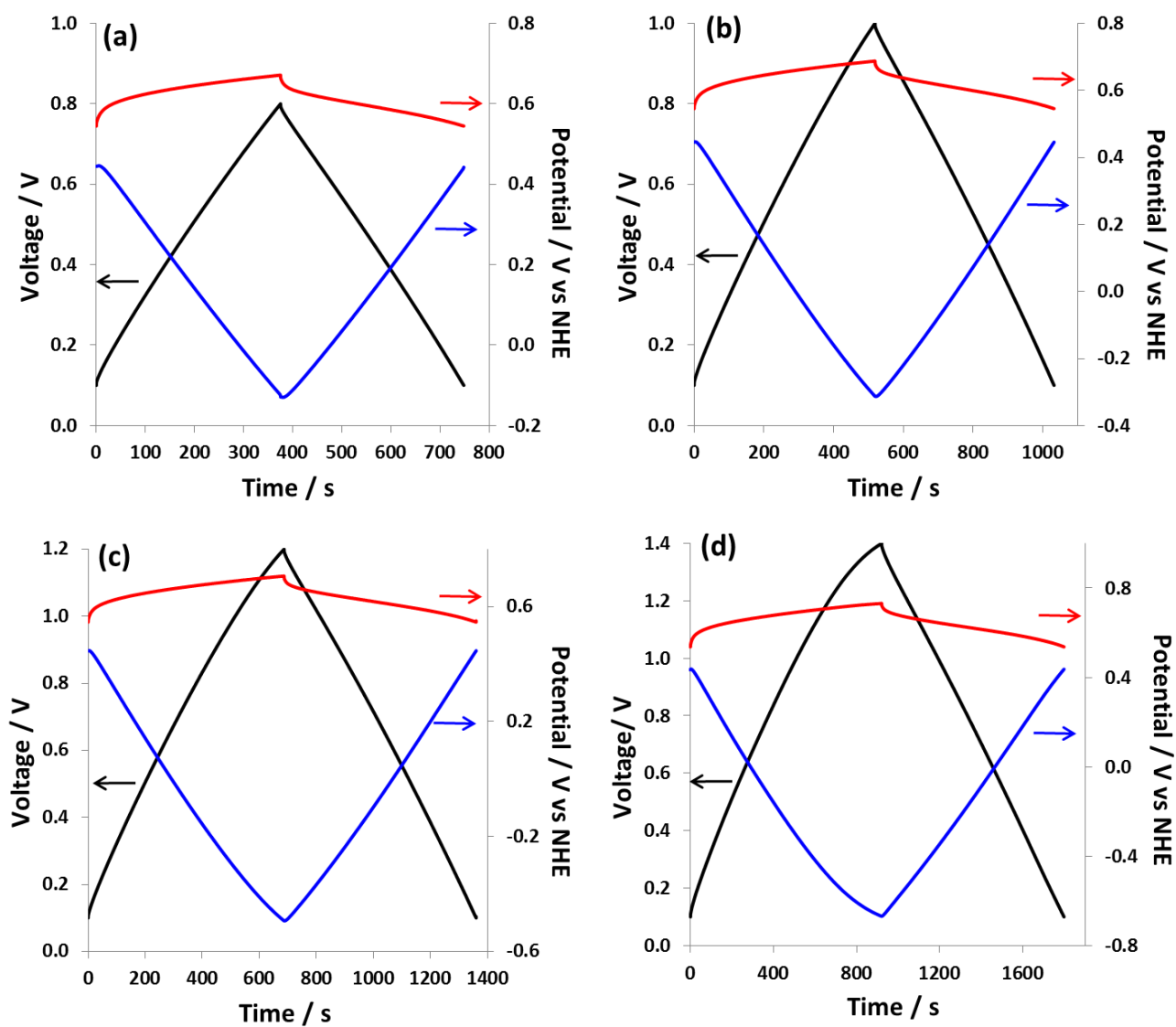
**Figure 1.** Pore size distribution of KAC and activated carbon cloth (ACC).



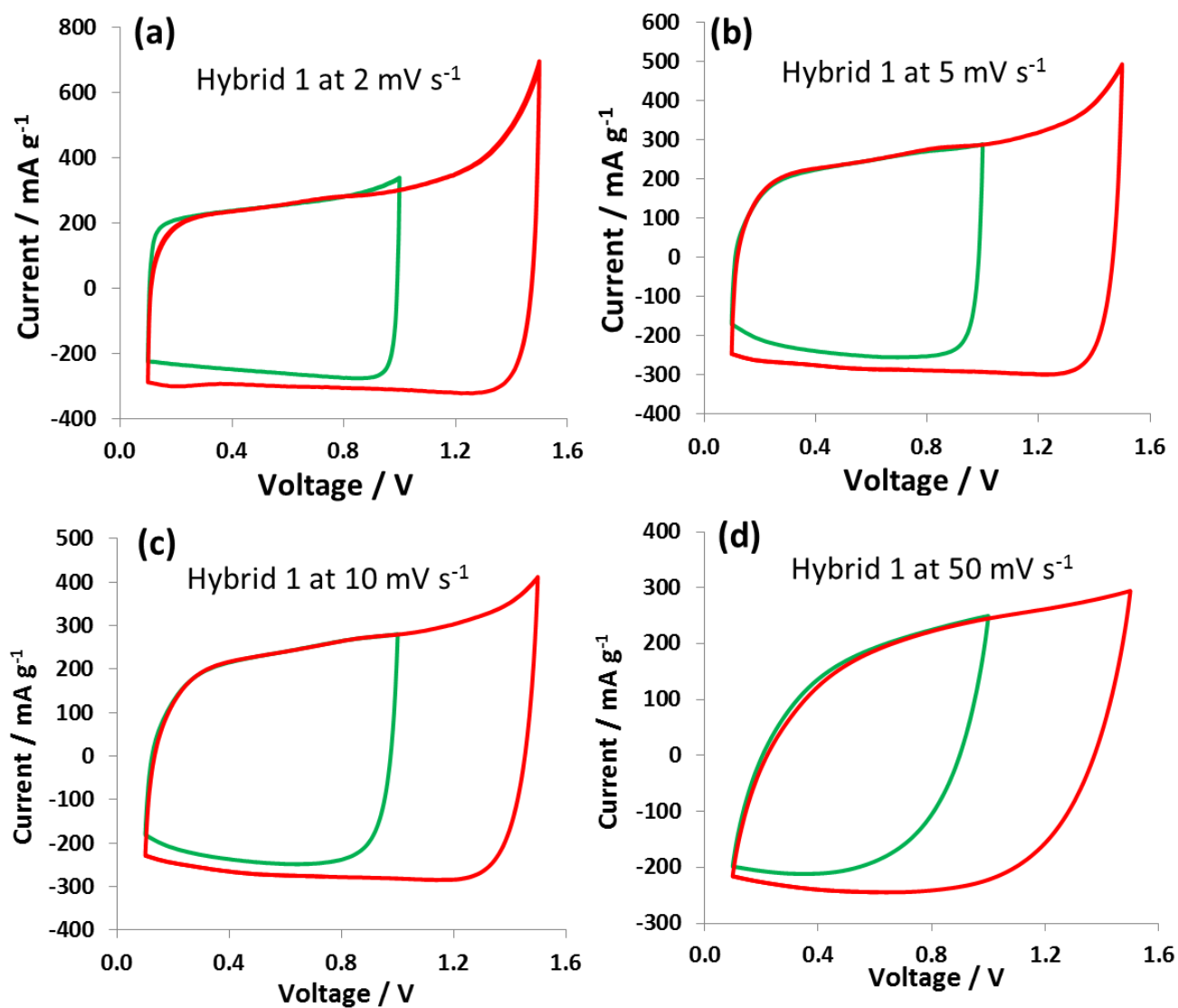
**Figure 2.** Scanning electron micrographs of pristine activated carbon cloth (ACC 507-20).



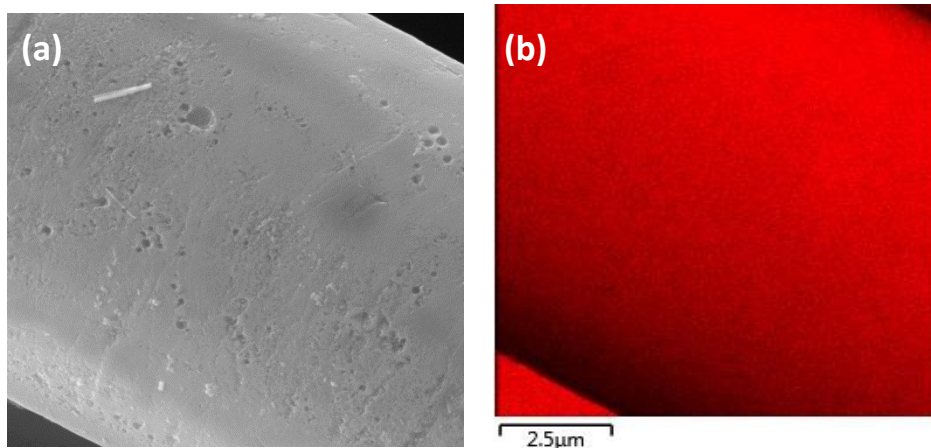
**Figure 3.** Cyclic voltammograms of hybrid cell 1 in 1 mol L<sup>-1</sup> NaI at 2 mV s<sup>-1</sup> with gradual voltage increase from 0.1 V up to 1.5 V.

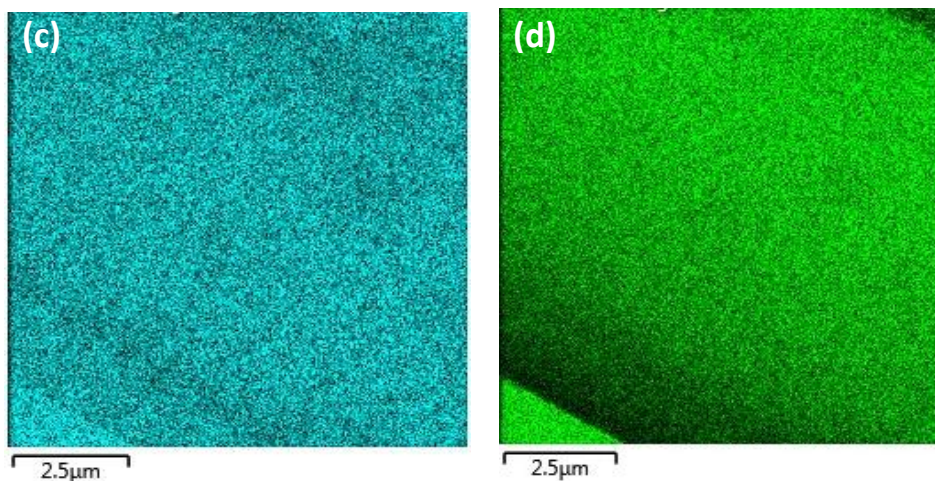


**Figure 4.** Galvanostatic charge/discharge curves of hybrid cell 1 in two electrode cell equipped with reference electrode in 1 mol L<sup>-1</sup> NaI as electrolyte at specific current of 0.1 A g<sup>-1</sup> up to cell voltage (a) 0.8 V, (b) 1.0 V, (c) 1.2 V and (d) 1.4 V.

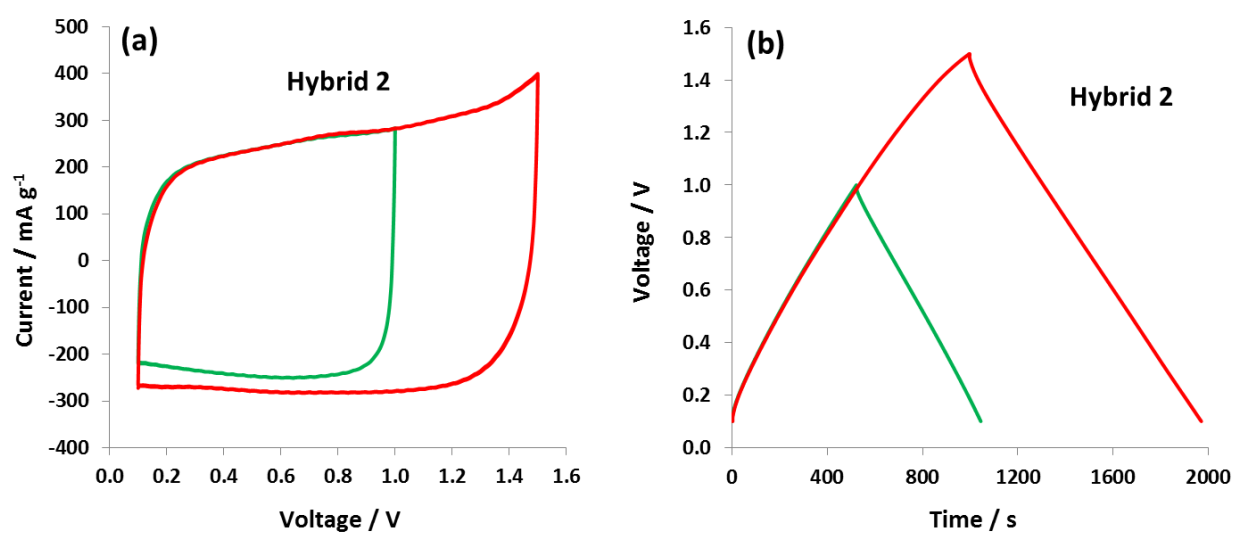


**Figure 5.** Two electrode cell performance of hybrid cell 1 in  $1 \text{ mol L}^{-1} \text{ NaI}$  with CVs up to  $1.0 \text{ V}$  and  $1.5 \text{ V}$  using scan rate of (a)  $2 \text{ mV s}^{-1}$ , (b)  $5 \text{ mV s}^{-1}$ , (c)  $10 \text{ mV s}^{-1}$  and (d)  $50 \text{ mV s}^{-1}$ .

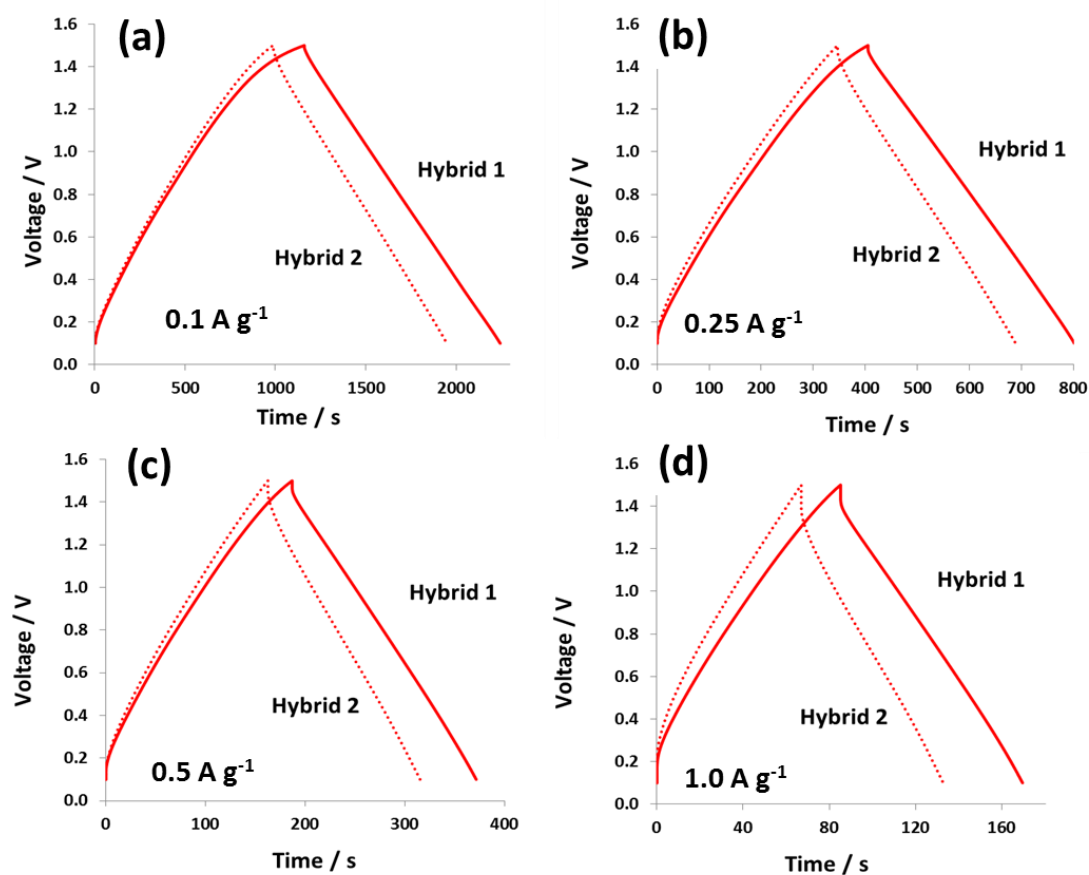




**Figure 6.** (a) Scanning electron micrograph of positive ACC electrode with corresponding surface mapping of (b) carbon, (c) iodine and (d) sodium.



**Figure 7.** Two electrode cell performance of hybrid cell 2 in 1 mol L<sup>-1</sup> NaNO<sub>3</sub> with (a) CVs and (b) galvanostatic charge/discharge up to 1.0 V and 1.5 V using scan rate of 2 mV s<sup>-1</sup> and specific current of 0.1 A g<sup>-1</sup> respectively.



**Figure 8.** Comparison of two electrode cell performance of hybrid cell 1 in 1 mol L<sup>-1</sup> NaI and hybrid cell 2 in 1 mol L<sup>-1</sup> NaNO<sub>3</sub> with galvanostatic charge/discharge up to 1.5 V using specific current (a) 0.1 A g<sup>-1</sup>, (b) 0.25 A g<sup>-1</sup>, (c) 0.5 A g<sup>-1</sup> and (d) 1.0 A g<sup>-1</sup>

**Table 2.** Performance comparison during 5000 galvanostatic charge/discharge cycles for various hybrid cells using iodides as redox active specie in aqueous electrolyte.

Electrolyte.	Cell voltage	Number of galvanostatic charge/discharge cycles	Percentage of capacitance loss	Ref.
KI+Na <sub>2</sub> MoO <sub>4</sub>	1.0 V	5000	35.4%	1
LiTFSI+KI	1.8 V	10000	21%	2
Li <sub>2</sub> SO <sub>4</sub> +KI	1.6 V	10000	9%	3
MnSO <sub>4</sub> +KI	1.5 V	10000	8%	4
ChNO <sub>3</sub> +ChI	1.5 V	20000	8%	5
NaNO <sub>3</sub>	1.5 V	5000	No capacitance decay	This work

#### References:

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