Supplementary Information

for

All Small Molecule Organic Solar Cells with Over 14% Efficiency by Optimizing Hierarchical Morphologies

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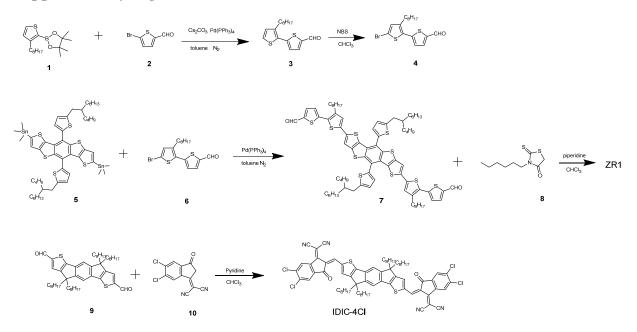
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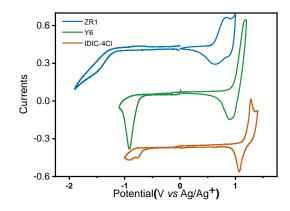
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Supplementary Figures



Supplementary Figure 1. Synthetic routes of ZR1 and IDIC-4Cl (The details of above molecules are given in the Supplementary Methods).



Supplementary Figure 2. Cyclic voltammogram (CV) plots of ZR1 (blue), Y6 (green) and IDIC-4Cl (red) in films.

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报 Report 测 告 试 Test

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1. 测试条件 Test Conditions: 标准上图电池,单晶硅(81+);			有效面积	短路电流 开	路电压	最大功率
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Supplementary Figure 3. Certification of the photovoltaic efficiency of the ZR1: Y6 devices by

National Institute of Metrology, China (NIM).

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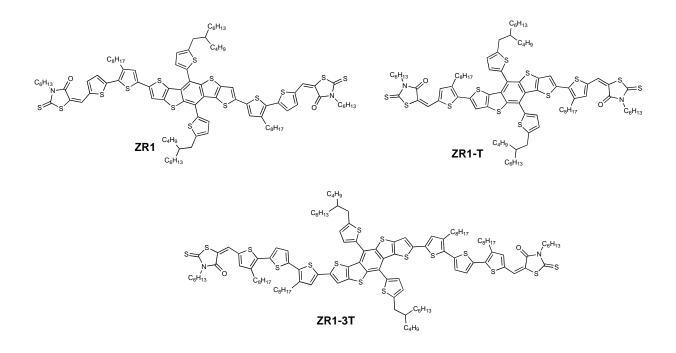
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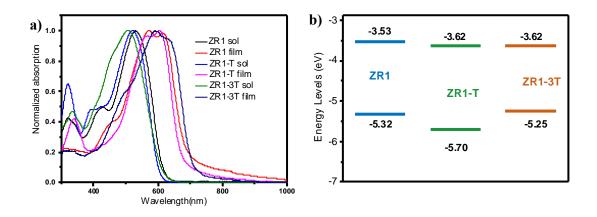
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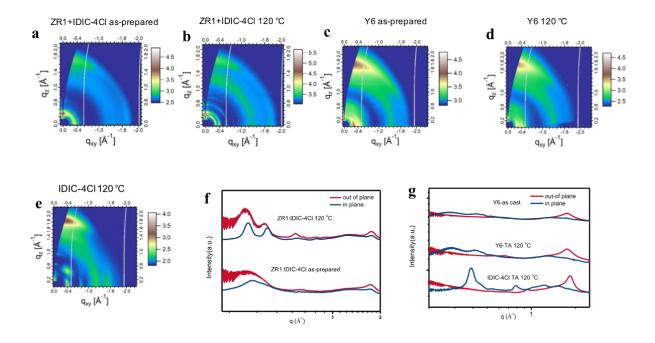
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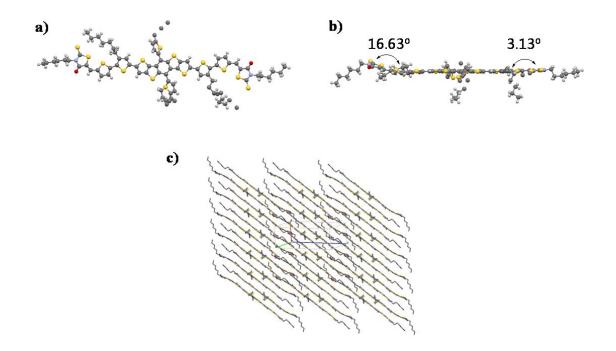
Supplementary Figure 4. ZR1, ZR1-T, ZR1-3T molecular structures.



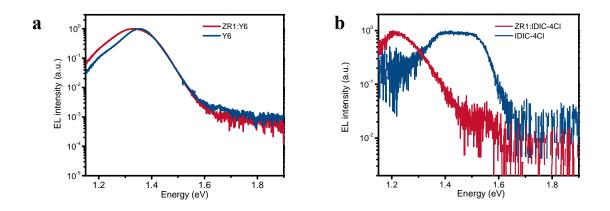
Supplementary Figure 5. a) Normalized UV–vis absorption spectra of ZR1, ZR1-T, ZR1-3T in solution and thin films. **b)** Energy diagrams of ZR1, ZR1-T, ZR1-3T



Supplementary Figure 6. (a–e) 2D GIWAXS patterns in pristine films and in blend films; (f-g) corresponding out-of-plane curves and in-plane curves.



Supplementary Figure 7. Molecular conformation of ZR1 in the single crystal, top view (a) and side view(b), (c) crystal packing of ZR1 backbones in the (010) plane.



Supplementary Figure 8. (a–b) Electroluminescence spectrum of blend film and pure film.

Supplementary Tables

Compound	λ ^{solution} (nm)	λ_{\max}^{film} (nm)	E ^{opt} g,film (eV)	HOMO ^{CV} (eV)	LUMO ^{CV} (eV)
ZR1	534	620	1.84	-5.32	-3.53
IDIC-4Cl	689	756	1.53	-5.91	-4.10
Y6	730	820	1.33	-5.72	-4.10

Supplementary Table 1. Optical properties and electrochemical of ZR1, Y6 and IDIC-4Cl.

Supplementary Table 2. Device optimization of D:A ratios for ZR1: IDIC-4Cl at TA 120°C

Donor/Acceptor	D:A	$V_{\rm OC}$ [V]	$J_{ m SC}$ [mA cm ⁻²]	FF [%]	PCE [%]
ZR1:IDIC-4Cl	1:0.6	0.789	17.14	66.37	9.08
ZR1:IDIC-4Cl	1:0.7	0.776	18.27	67.96	9.64
ZR1:IDIC-4Cl	1:0.8	0.785	17.53	66.03	9.20

Supplementary Table 3. Device optimization of thermal annealing temperature for ZR1: Y6 with

Donor/Acceptor	ТА	<i>V</i> _{OC} [V]	$J_{\rm SC}$ [mA cm ⁻²]	FF [%]	PCE [%]
ZR1:Y6	no	0.876	14.23	40.50	5.05
ZR1:Y6	110℃	0.870	24.02	66.20	13.62
ZR1:Y6	120°C	0.861	24.34	68.44	14.34
ZR1:Y6	140℃	0.837	24.49	67.90	13.91

Donor/Acceptor	D:A	V _{oc} [V]	$J_{ m SC}$ [mA cm ⁻²]	FF [%]	PCE [%]
ZR1:Y6	1:0.4	0.855	20.56	67.07	11.80
ZR1:Y6	1:0.5	0.861	24.34	68.44	14.34
ZR1:Y6	1:0.6	0.869	21.18	62.24	11.46

Supplementary Table 4. Device optimization of D:A ratios for ZR1: Y6 at TA 120°C

Supplementary Table 5. Device optimization of solvent annealing with THF for ZR1: Y6 at TA

120℃

Donor/ Acceptor	D:A	V _{OC} [V]	J _{SC} [mA cm ⁻²]	FF [%]	TA (°C)	SVA	PCE [%]
ZR1:Y6	1:0.5	0.861	24.34	68.44	120	no	14.34
ZR1:Y6	1:0.6	0.851	22.46	62.58	120	THF (40s)	11.97

Supplementary Table 6. Device comparison of ZR1:Y6, ZR1-T:Y6, ZR1-3T:Y6 blends at TA 120℃

Donor/Acceptor	D:A	$V_{\rm OC}$ [V]	$J_{\rm SC}~[{ m mA~cm^{-2}}]$	FF [%]	PCE [%]
ZR1:Y6	1:0.5	0.861	24.34	68.44	14.34
ZR1-T:Y6	1:0.6	0.498	0.0019	25.29	0.025
ZR1-3T:Y6	1:0.5	0.754	22.51	45.47	7.67

System	284.2eV circ.int	Thickness (nm)	SQRT (TSI/thickness)	Domain purity	Domain size (nm)
ZR1:Y6 no TA	6.65E-09	120	7.4434E-06	0.8752598	73.404
ZR1:Y6 110 °C	7.16E-09	120	7.72647E-06	0.9085459	72.359
ZR1:Y6 120 °C	7.99E-09	120	8.16236E-06	0.9598021	73.913
ZR1:Y6 140 °C	8.68E-09	120	8.50421E-06	1	83.412

Supplementary Table 7. Detailed RSoXS of the ZR1: Y6 blend films under different condition

Supplementary Table 8. Hole mobility and electron mobility of blend films and pure films.

	ZR1: Y6	ZR1: IDIC-4Cl	ZR1	¥6	IDIC-4Cl
Hole mobility	1.32×10^{-4}	3.30×10 ⁻⁴	3.53×10 ⁻³	-	-
Electron mobility	3.92×10 ⁻⁴	4.55×10 ⁻⁴	-	2.29×10 ⁻⁴	1.08×10 ⁻⁴

Supplementary Methods

Materials and synthesis Compounds 5 were purchased from Hyper Chemical Company, Compounds 4 and 8 were synthesized through the method described in the literature^{1,2}, And the detailed synthetic processes of the other monomers and target small molecules were illustrated as follows.

Compound 3: A solution of Compound 1 (4.30 g, 13.34 mmol), Compound 2 (2.12 g, 11.12 mmol) and Cs₂CO₃ (13.04g, 40.02mmol) in toluene (50 mL) was degassed by bubbling N₂ through the solution for 10 min. Pd(PPh₃)₄ (0.77 g, 0.68 mmol) was added and the resulting solution was vigourously refluxed overnight under inert atmosphere. The solution was cooled to room temperature and diluted with n-hexane, filtered and concentrated in vacuo. The crude product was purified by flash chromatography (SiO2, n-hexane, 5% EtOAc) to give Compound 3 (2.20 g, 7.19 mmol, 54% yield) as a yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 9.88 (s, 1H), 7.69 (d, J = 3.7 Hz, 1H), 7.27 (s, 1H), 7.21 (d, J = 3.7 Hz, 1H), 6.97 (d, J = 5.1 Hz, 1H), 2.85-2.75(m, 2H), 1.70-1.60 (m, 2H), 1.41 – 1.20 (m, 10H), 0.87 (t, J = 6.4 Hz, 3H).

Compound 4: To a stirred solution of **3** (2.28 g, 7.44 mmol) in CHCl₃ (20 mL) and acetic acid (10 mL) kept in the dark at 0 °C was added *N*-bromosuccinimide (1.32 g, 7.44 mmol). The solution was stirred before being stirred for 1.5 h, diluted with n-hexane and washed with water, NaHCO₃ (sat.) and brine. The organic layer was dried (MgSO₄) and concentrated in vacuo. The crude product was purified via flash chromatography (SiO₂, n-hexane, 0.5 - 6% EtOAc) to give Compound 4 (2.0 g, 5.2mmol, 70% yield) as a yellow oil. ¹H NMR (400 MHz, CDCl₃) δ 9.86 (s, 1H), 7.67 (d, J = 3.7 Hz, 1H), 7.12 (d, J = 3.6 Hz, 1H), 6.91 (s, 1H), 2.72 (t, J = 7.7 Hz, 2H), 1.63-1.56 (m, J = 15.1, 7.5 Hz, 2H), 1.39 – 1.19 (m, 10H), 0.87 (t, J = 6.3 Hz, 3H).

Compound 7: Compound 5 (500mg, 0.44 mmol) and compound 6 (511 mg, 1.32mol) were dissolved into anhydrous toluene (40 mL) in a three-neck flask. The solution was flushed with nitrogen for 20 min, and Pd(PPh₃)₄ (25.56 mg) was added into the solution quickly. Then, the solution was reacted at 110 °C for 72h under nitrogen protection. After cooling to room temperature, the toluene was removed under reduced pressure. The crude product was subsequently purified by column chromatography on silica gel (DCM) to afford compound c1 as red solid.¹H NMR (400 MHz, CDCl3) δ 9.90 (s, 2H), 7.71 (d, J = 3.9 Hz, 2H), 7.34 (s, 2H), 7.22 (t, J = 3.3 Hz, 4H), 7.05 (d, J = 3.3 Hz, 2H), 7.02 (s, 2H), 2.99 (d, J = 6.6 Hz, 4H), 2.83–2.77 (m, 4H), 1.82 (dd, J = 11.9, 6.0 Hz, 2H), 1.68 (dd, J = 15.3, 7.5 Hz, 4H), 1.52–1.24 (m, 52H), 0.90 (dt, J = 20.6, 6.7 Hz, 18H). MALDI-TOF MS (m/z): 1411.1

Compound ZR1: Compound 7 (200 mg, 0.142 mmol) and Compound 8 (307 mg, 1.4 mmol) was dissolved in dry CHCl₃ (30 ml). After 3 drops of piperidine were added, the resulting solution was refluxed and stirred for 12 h under nitrogen. After cooling to room temperature, the reaction mixture was poured into water and extracted several times with chloroform. The organic phases were combined, washed with brine, and dried over anhydrous magnesium sulfate. Then the solvent was removed under reduced pressure, and the crude product was purified by column chromatography on silica gel (CHCl₃) to yield Compound ZR1 as black solid. ¹H NMR (400 MHz, CDCl₃) δ 7.65 (s, 2H), 7.20 (s, 2H), 7.15 (d, *J* = 3.3 Hz, 2H), 7.06 (s, 2H), 6.99 (d, *J* = 6.2 Hz, 4H), 6.73 (s, 2H), 3.94 (t, *J* = 7.3 Hz, 4H), 3.02 (s, 4H), 2.72 – 2.61 (m, 4H), 1.92 – 1.80 (m, 2H), 1.62 (d, *J* = 5.3 Hz, 8H), 1.35 (t, *J* = 34.6 Hz, 64H), 1.03 – 0.81 (m, 24H).MALDITOF MS (m/z): 1810.6

Compound IDIC-4Cl: Compound 9 (300 mg, 0.0.39mmol) and compound 10 (1022mg, 7.77 mmol) were dissolved in dry CHCl₃ (60 ml). After 1 ml of pyridine was added, the resulting solution was

refluxed and stirred for 12 h under nitrogen. After cooling to room temperature, the reaction mixture was poured into 200 ml methanol and filtered. The crude product was purified by column chromatography on silica gel using petroleum ether/dichloromethane (1:1) as eluent to yield IDIC-4Cl as powder. ¹H NMR (400 MHz, CDCl₃) δ 9.00 (s, 2H), 8.79 (s, 2H), 7.96 (s, 2H), 7.75 (s, 2H), 7.63 (s, 2H), 2.08 (td, *J* = 13.0, 4.1 Hz, 4H), 1.96 (td, *J* = 13.0, 4.1 Hz, 4H), 1.20 (d, *J* = 7.2 Hz, 48H), 0.80 (t, *J* = 6.8 Hz, 12H). MALDI-TOF MS (m/z): 1260.9

Supplementary References

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