### **Supplemental Information**

#### **Materials and Methods**

Redox titrations. A 4 ml solution of 40 µM FldA in 50 mM Tris-Cl pH 8.0 buffer was purged with nitrogen. In the anaerobic chamber, the following solutions of mediators were added to the final concentration of 10 µM: Phenazine methosulfate (80mV), thionine acetate (64mV), indigo carmine (-130mV), 2-hydroxy-1,4-naphthoquinone (-145mV), anthroquinone disulfonate (-170mV), sodium anthraquinone-2-sulfonate (-255mV), benzyl viologen (-358mV), methyl viologen (-440mV) and 1,1'trimethylene-2,2'-dipyridinium dibromide (-521mV). Redox titrations were performed in a custom-made quartz cuvette transformed into a spectroelectrochemical cell containing a gold flag working electrode (Gauze electrode), a platinum electrode counter and Ag/AgCl<sub>2</sub> reference electrode. The protein was reduced by electrochemical potentials applied through EZStat Pro (Nuvant systems) potentiostat. The degree of FldA reduction was monitored spectrophotometrically with CARY WinUV-Vis spectrophotometer. In order to allow for the protein to stabilize, a total of 15 spectra were taken at oneminute interval each after every 30 mV decrease in the applied potential. The background absorbance due to the mediators was recorded by dissolving mediators to a final concentration of 10 µM in 50 mM MOPS buffer pH 7.0, in order to correct for baseline, before the spectra of the sample were recorded. Absorbance changes at 360 nm, 460 nm and 385 nm were followed relative to the reference wavelength at 700 nm. The potentials were corrected to those for standard hydrogen electrode (+186 mV) and plotted against the changes in absorbances at 385nm (sq/hq) and at 460 nm (ox/hq) by applying the Nernst equation incorporated in the Igor software.

#### **Results and Discussion**

F<sub>420</sub>H<sub>2</sub>-dependent electron bifurcation by HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub> and electron transfer from HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub> to the membrane-bound electron transport chain dependent on FldA or Fdx. Preparation of materials and assay protocols were as described previously (1). The initial rate of CoMS-SCoB reduction to the thiols HSCoM and HSCoB was nearly identical in reaction mixtures containing the heterodisulfide,  $HdrA_2B_2C_2$ , reduced coenzyme F<sub>420</sub> (F<sub>420</sub>H<sub>2</sub>), and either FldA or Fdx (Fig. S6). Although at a reduced rate and final yield of thiols, the results establish that FldA is capable of replacing Fdx in the  $F_{420}H_2$ -dependent electron bifurcation by HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub>. Addition of everted membrane vesicles to the reaction mixtures enhanced the rate of heterodisulfide reduction and final yield of the corresponding thiols dependent on either FldA or Fdx. The results show that FldA is capable of mediating electron transfer from  $HdrA_2B_2C_2$  to the membrane-bound electron transport chain culminating in reduction of CoMS-SCoB, although at a reduced rate and final yield of thiols compared to Fdx. The rate of CoMS-SCoB reduction and final yield of thiols was enhanced by the addition of NaCl to reactions containing either FldA or Fdx. This result is consistent with electron transport-coupled translocation of Na<sup>+</sup> from outside to inside vesicles supporting that FldA donates electrons to the Na<sup>+</sup>-translocating Rnf complex. The rates shown in Figure 6 for FldAand Fdx-dependent membrane-bound electron transport include minor contributions from  $F_{420}H_2$ dehydrogenase-dependent membrane-bound electron transport to CoMS-SCoB, evidenced by results obtained with control reaction mixtures minus either HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub> or FldA and Fdx. Nonetheless, the conclusions regarding the ability of FldA to replace Fdx are unaffected.

**FIGURE S1. SDS-PAGE of purified recombinant FldA.** Left lane, molecular mass markers. Right lane, loaded with 1µg of purified FldA. The gel was stained with Coomassie R-250.

**FIGURE S2.** Potentiometric titration of FldA. The changes in sq absorbance at 385 nm were plotted against the actual potentials experienced by the sample (mV). (A) The  $E_m$  for the ox/sq couple is -301 ± 5.06 mV. (B) The  $E_m$  for the sq/hq couple is -464 ± 2.61 mV. Conditions were pH 7.0 in 50 mM MOPS. The data in panels A and B were fitted separately using the modified Nernst equation (igor software) (2).

**FIGURE S3. The three FldA protomers found in the asymmetric unit.** The FldA proteins and FMN molecules are shown in ribbon and stick representations (cyan, protomer A; pink, protomer B; green, protomer C). Protomers A and B form a dimer and the promoter C also forms a dimer with a symmetry related protomer C shown as white ribbon and stick models.

**FIGURE S4. Schematic of protein interactions with FMN of the FldA from** *M. acetivorans.* Negative and positive charged atoms are shown in red and blue respectively. Hydrogen bonds are shown as purple lines with arrows. Lines in teal and lime green indicate polar and hydrophobic regions respectively. Green lines ending in dots indicate Pi-Pi stacking.

**FIGURE S5. Pathway for conversion of acetate to methane in** *M. acetivorans.* Ack, acetate kinase; Pta, phosphotransacetylase; CoA-SH, coenzyme A; H<sub>4</sub>SPT, tetrahydrosarcinapterin; Fd<sub>r</sub>, reduced ferredoxin; Fd<sub>o</sub>, oxidized ferredoxin; Cdh, CO dehydrogenase/acetyl-CoA synthase; CoM-SH, coenzyme M; Mtr, methyl-H<sub>4</sub>SPT:CoM-SH methyltransferase; CoB-SH, coenzyme B; MP, methanophenazine; Hdr-DE, heterodisulfide reductase; Rnf, Rnf complex; Mrp, Mrp complex; Atp, ATP synthase. Reproduced by permission (3).

FIGURE S6.  $F_{420}H_2$ -dependent electron bifurcation by HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub> and electron transfer from HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub> to the membrane-bound electron transport chain dependent on FldA or Fdx. The atmosphere was 100% N<sub>2</sub> and the temperature was 21°C for all reactions that were also initiated by addition of  $F_{420}H_2$ . The complete reaction mixture (300 µl) contained closed everted membrane vesicles (100 µg protein), 1.5 µM HdrA<sub>2</sub>, 2 µM HdrB<sub>2</sub>C<sub>2</sub>, 200 µM CoMS-SCoB, 18 µM  $F_{420}H_2$ , 33 mM NaCl and either 5 µM Fdx or Fdx in 50 mM potassium phosphate buffer (pH 7.0). Complete: with Fdx ( $\Box$ ); with FldA ( $\blacksquare$ ), with Fdx, minus NaCl ( $\Delta$ ); with FldA, minus NaCl ( $\Delta$ ); with FldA, minus vesicles ( $\nabla$ ); with Fdx, minus vesicles ( $\nabla$ ); minus FldA or Fdx ( $\blacklozenge$ ); minus HdrA<sub>2</sub>B<sub>2</sub>C<sub>2</sub> ( $\diamondsuit$ ). There was no detectable increase in thiol content over the background level (1.2 µM) in the complete reaction mixture minus CoMS-SCoB or  $F_{420}H_2$ . Initial rates of thiol formation (µM/min) are indicated next to each reaction. Data points are the mean of three replicates for which the standard deviation is shown by error bars.

FIGURE S7. Distance tree of FldA and related proteins. The tree was constructed using the neighborjoining method with significant alignments recovered from the NCBI database queried with FldA (MA1799) from *M. acetivorans*. The scale bar indicates the average number of amino acid substitutions per site. Flavodoxin sequences: WP\_011021802.1, *Methanosarcina acetivorans*; WP\_048183459.1, *Methanosarcina siciliae*; WP\_048138299.1, *Methanosarcina horonobensis*; WP\_048125136.1, *Methanosarcina lacustris*; WP\_048118313.1, *Methanosarcina vacuolata*; WP\_011307048.1, *Methanosarcina barkeri*; WP\_048166050.1, *Methanosarcina thermophila*; WP\_048088456.1 Candidatus *Methanoperedens nitroreducens*; WP\_013194462.1, *Methanohalobium evestigatum*; WP\_048198898.1, *Methanocella arvoryzae*; WP\_013898822.1, *Methanosalsum zhilinae*; WP\_013719833.1, *Methanosaeta concilii*; WP\_015323860.1, *Methanomethylovorans hollandica*; WP\_023845719.1, *Methanolobus tindarius*; WP\_014406362.1, *Methanocella conradii*; WP\_048146851.1, *Methanolobus psychrophilus*; WP\_011022659.1, *Methanosarcina acetivorans*. WP\_032279031.1, *Escherichia coli*; WP\_027338967.1, *Halonatronium saccharophilum*; WP\_011791702.1, *Desulfovibrio vulgaris*; WP\_021760293.1, *Desulfovibrio gigas*; WP\_015326202.1, *Halobacteroides halobius*; WP\_002160966.1, *Bacillus cereus*; WP\_012058624.1, *Clostridium beijerinckii*. Lactamase sequences: WP\_028323018.1, *Desulfatiglans anilini*; WP\_0117 0051.1, *Syntrophobacter fumaroxidans*; WP\_027177280.1, *Desulfovibrio hydrothermalis*.

#### FIGURE S8. Alignment of sequences from species in clades I and II, Methanoperedens

*nitroreducens* and *Methanosaeta concilii*. Species correlating with the accession numbers are as shown in Fig. S7. Additional sequences are the generic flavodoxins from *M. acetivorans* (WP\_011022659), *D. gigas* (WP\_021760293) and *C. beijerinckii* (WP\_012058624). The boxed sequences show sequence identity to the  $S_{56}$ TFYY<sub>60</sub> FMN binding motif of *M. acetivorans* FldA (WP\_011021802).

Figure S1.



Figure S2.







Figure S4



## Figure S5







Figure S7



# Figure S8.

WP_011021802	1	MKAIVVYLSTSGNTKAMAEAIGNGIES-KNVDVQVISFYDVKLD-ELKEAEAIAVGSSTFYYKMLLPMEKFMD	71		
WP_048183459	1	MKAIVVYLSTSGNTKAMAEAIGNGIES-KNVDTKVVSFYDVKLE-ELKEAEAIAVGSSTFYYKMLLPMEKFMD	71		
WP_048138299	1	MKAIVVYLSTSGNTKVMAEAIGNGIES-KNVDVKVVSFYDVKPE-ELNEAEAIAIGSSTFYYKMLLPMEKFMN	71		
WP_048125136	1	MKAIVVYLSTSGNTKDMAEAIGKGIES-KNVDVKVISFYDVKPE-ELNEAEAIAVGSSTFYYKMLLPMEKFMN	71		
WP_048118313	1	MKAIVVYLSTSGNTKAMAEAIASGIES-KHVDAKAISFYDVKLE-DLYEADAIAVGSSTFYYRMLQPMEKFMD	71		
WP_011307048	1	MKAIVVYLSTSGNTKAMAEAIASGIES-KHVDAKAISFYDVKLE-DLNEADAIAVGSSTFYYRMLQPMEKFMD	71		
WP_048166050	1	MKAIVVYLSTSGNTKIMAEAIASGIES-KHVDAKAVSFYDVKVE-DLNEADAIAVGSSTFYYKMLQPMEKFMD	71		
WP_048088456	1	Mpkvvivylstsgntkamadaiaegvrs-rnvealavnfheadie-dlrsadaiaigsstfyykmlppmekfi	71		
WP_013194462	1	MaklavvylstQgntkTmaesiangaks-RhvdvDvssfydttvD-evaaadaiaigsstfyygmlapiekfvd	72		
WP_048198898	1	MVVYLSTSGNTKRMADAIAAGADS-RGINSEVVSFYDVDLD-KLRSADAIAIGSSTFWYKMHDAMEKFLDr-	69		
WP_013898822	1	MpKLAIVYLSTQGNTMLMAEGIAEGAIS-RNIDVEVRSFYEWNPM-DAASADGIAVGSSTFNYAMHPPIQKFL	71		
WP_013719833	1	MpKIAIIYASQSGKTKRMAEAIANGAKSvEGVDVLLKNVFQAKPD-DVLDADAVVLGGSTYNSKLIKTMDPFLA	73		
WP_015323860	1	MvKLAIVYLSTQGSTRMMAEAIAEGARQ-KHIEVRVDSFYEWEPK-DAASYDGIAVGSSTFYYKMLEPISKFL	71		
WP_023845719	1	Mpklaivylstqgstkmmaeaiaegare-khvdvdidnfyewdpa-evakydaicigsstfyytmlepiakfl	71		
WP_014406362	1	MvkpKLVVVYLSTQGNTKKMADAIAHGAED-RGMEARSVSFYEADMN-EIRDADAIALGSSTFWYKMHDAMERFIErm	76		
₩P_048146851	1	MpKLAIVYLSTQGNTQMMAEAIAEGARE-RHVEVKVNNFHEWNPA-EIADYDAIAVGSSTFYYTMLDPIAKFI	71		
WP_011022659	1	MeKTIIVYGSSTGNTEILAEEIKGTLEK-YDGKVKLMEVTDLHPT-DLANYDTILLGCSTWGDGELQDDFIPFE	72		
WP_021760293	1	MpKALIVYGSTTGNTEGVAEAIAKTLNS-EGMETTVVNVADVTAPgLAEGYDVVLLGCSTWGDDEIELQEDFVPl-	74		
WP_012058624	1	MKIVYWSGTGNTEKMAELIAKGIIE-SGKDVNTINVSDVNID-ELLNEDILILGCSAMGDEVLEESefEPFIE	71		
WP_011021802	72	-ETLVASNPQGKIGAAFGSYGWSGEA PILIAEKMREMGMTVMDPVLRILHKPTDKDLQECKRLGIDIAEKVKHKGTK	147		
WP_048183459	72	-ETLVASNPQGKIGAAFGSYGWSGEA PILIAEKMREMGMTVMDPVLRILHKPTDKDLQECKRLGIDIAEKVKHKSTK	147		
WP_048138299	72	-ETLVSTNPQGKLGAAFGSYGWSGEA PIMIAEKMRELGMTVMDPVLRILHRPTDKDLQECKRLGVDIAEKMKQRSKK	147		
₩₽_048125136	72	-ETLISANPQGKIGAAFGSYGWSGEA PILIAEKMREMGMSVVDPVLRILHKPTDKDLQECKRLGIDIAEKLKHKSKK	147		
WP_048118313	72	-ETLVSTNPKGKLGAAFGSYGWSGEA PILIAEKMRDMGMTVIDPVLRILHKPTDKDIQECKRLGVDIAEKLKHKSSK	147		
WP_011307048	72	-EILASTNPKGKLGAAFGSYGWSGEA PILIAEKMRDMGMTVIDPVLRILHKPTDKDLQECKRLGVDIAEKLKHRSSK	147		
WP_048166050	72	-ETLASTNPKGKLGAAFGSYGWSGEA PILIAEKMREMGMIVIDPVLRILHKPTDKDVQECKRLGVDIAEKMKQ-ISK	146		
WP_048088456	72	-ENLAREDVKEKIGAAFGSYGWSGEA PVMIAEKMREIGMKVIDPVLRIQYTPTDKDLEECKRLGKDIAIRLKAK	144		
WP_013194462	73	ELVQKDVDGKIGAAFGSYGWSGEA PHMIADKMRKAGMNVVDPVLRIQHIPNDKDISECERLGKDLAENIKKR	144		
WP_048198898	70	LKPLKLEGKVGAAFGSYGWSGEA PVQIAQKMRELGLDVIDPVLRVQWDPEEKDLKECERLGKDIAKALKKEPVP	143		
WP_013898822	72	-DQMLETEVKGKIGAAFGSYGWSGEA PVMIADKMRKAGMNVIDPVLRIQYRPTEKDIAECVRLGKDLAEKIKHNK	145		
WP_013719833	74	QLEKLDLKGKVGVAFGSYGWSGEG VPILIDRMKSFGMSVIEPGTTAVQLPTKEDLERCTDLGKAVANGLLN	144		
WP_015323860	72	-DALIAEGVEGKVGSAFGSYGWSGEA PVIIAEKMREAGMKVIDPVLRIQYIPEEKDIKECQRLGKDIAIKLKKLVTE	147		
WP_023845719	72	-DKLIEIGIEGKLGAAFGSYGWSGEA PVLIAEKLRKAGVEVIDPVLRIQYVPNEKDLAECIRLGKDMAKKMKKKE	145		
WP_014406362	77	gKELGQTAIEGKVGAAFGSYGWSGEA PVDIARMMRSLGIKVIDPVLRVQYEPNERDLEECKRLGKDIANAVKKTVRP	153		
WP_048146851	72	-DRIVEVGVDGKIGAAFGSYGWSGEA PVKIAEKLRNAGVEVIDPVLRIQYTPTEKDLFECNRLGKDLAERMKNR	144		
WP_011022659	73	-EEMEGVDLKGKKAACFGPGDSTFPL[4]VDLLEEKLKSCGAQIIVEGLKIDGD-VDYQLDKAEEWANEVAQAVSV	147		
WP_021760293	75	YEDLDRAGLKDKKVGVFGCGDSSYTY[4]VDVIEKKAEELGATLVASSLKIDGEPDSAEVLDWAREVLARV 1			
WP_012058624	72	EIST-KVSGKKVALFGSYGWGDGK[1]MRDFEERMNAYGCVVVETPLIVQNEPDEAE-QDCIDFGKKIANI	138		

<sup>a</sup> Data collection		Refinement	
Wavelength (Å)	0.9181	Resolution (Å)	27.09-1.68
Resolution (Å)	50-1.68	$R_{\text{work}}/R_{\text{free}}$ (%)	18.0/ 20.7
No. of Reflections	1,514,263/73,408	B-factors	
(tot./uni.)		Protein	23.110
Completeness (%)	98.1 (88.7) <sup>c</sup>	Ligand	16.968
$I / \Box (I)$	30.3 (2.6) <sup>c</sup>	Water	30.716
<sup>b</sup> R <sub>sym</sub> (%)	6.9 (57.4) <sup>c</sup>	R.m.s deviations	
		Bond lengths (Å)	0.035
		Bond angles (°)	2.612

TABLE S1: Data collection and refinement statistics of the *M. acetivorans* FldA.

<sup>a</sup>Data set was collected by synchrotron radiation (F1 line) at the MacCHESS (Cornell University, Ithaca, NY).

 ${}^{b}R_{sym} = \sigma |I - \langle I \rangle | / \sigma I$ , where *I* is observed intensity and  $\langle I \rangle$  is average intensity obtained from multiple observations of symmetry related reflections.

<sup>c</sup>Highest resolution shell (1.71-1.68 Å) is shown in parenthesis.

#### Bibliography

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- 3. Suharti S, Wang M, de Vries S, & Ferry JG (2014) Characterization of the RnfB and RnfG subunits of the Rnf complex from the archaeon *Methanosarcina acetivorans*. *PLoS One* 9:e97966.