

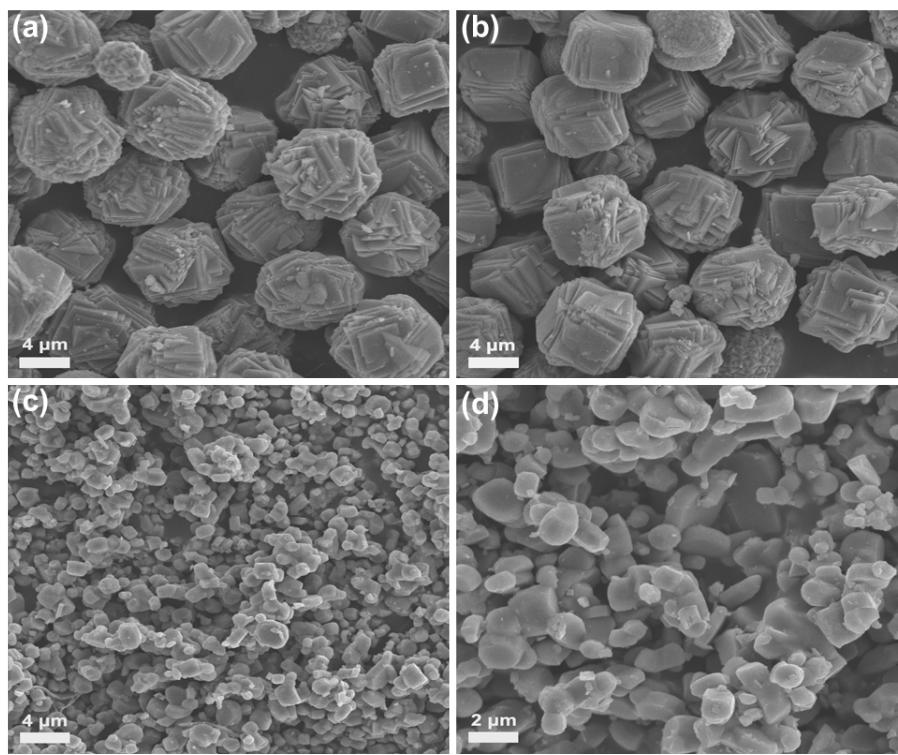
## ***Supporting Information***

# **Rational Design of a P2-Type Spherical Layered Oxide Cathode for High Performance Sodium Ion Batteries**

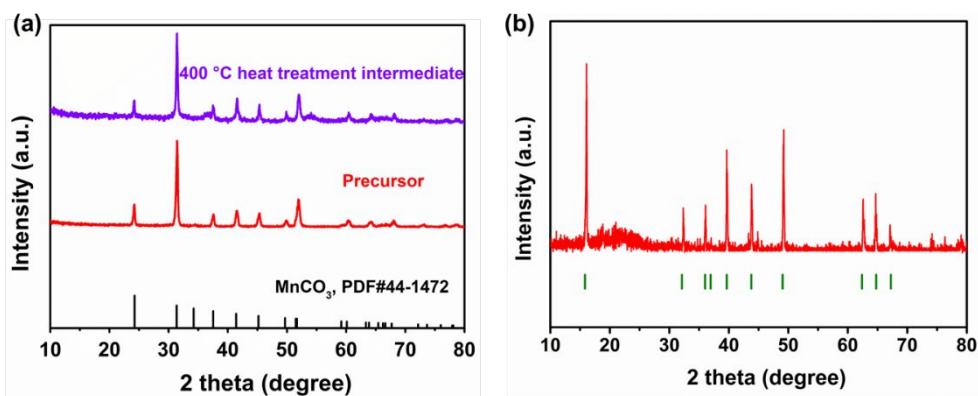
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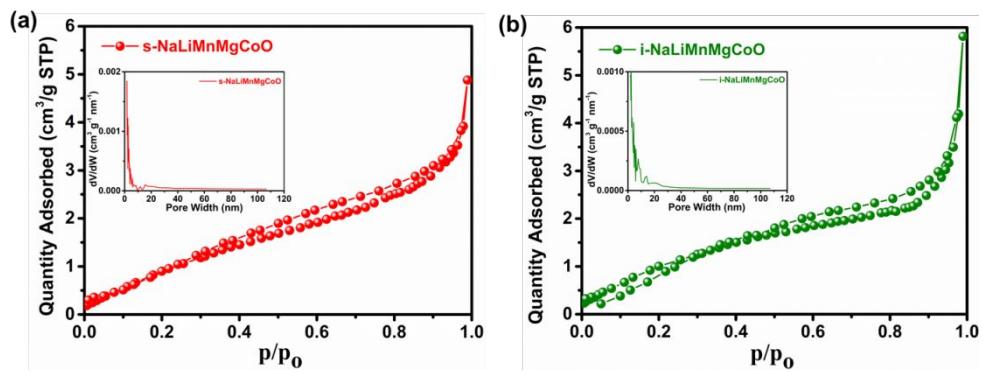
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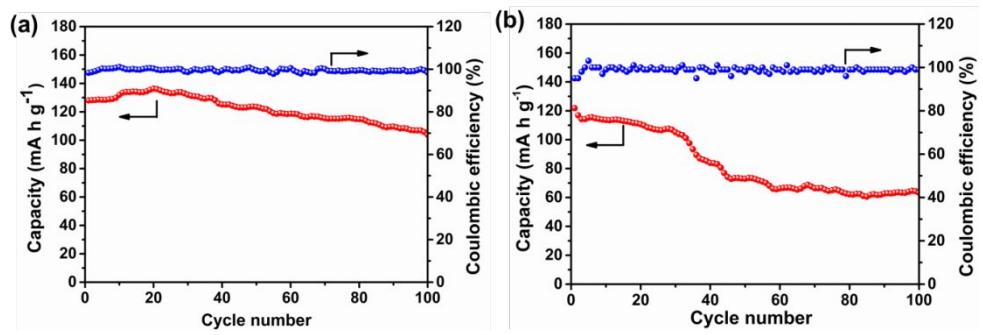
**Figure S1.** (a) SEM image of as-prepared  $\text{Mn}_{0.71}\text{Mg}_{0.21}\text{Co}_{0.08}\text{CO}_3$  precursor microspheres. (b) SEM image of 400 °C heat treatment intermediate. (c, d) SEM images of P2- $\text{Na}_{0.66}\text{Li}_{0.18}\text{Mn}_{0.71}\text{Mg}_{0.21}\text{-Co}_{0.08}\text{O}_2$  (i-NaLiMMCO) synthesized by co-precipitation method.



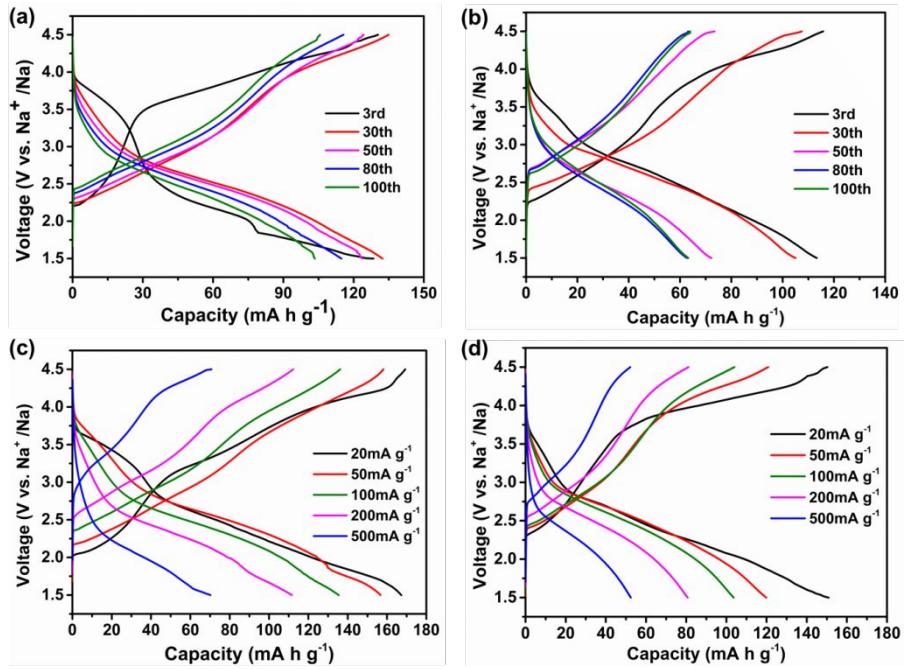
**Figure S2.** (a) XRD patterns of the Mn<sub>0.71</sub>Mg<sub>0.21</sub>Co<sub>0.08</sub>CO<sub>3</sub> precursor microspheres (red line) and 400 °C heat treatment intermediate (purple line). (b) XRD pattern of P2-Na<sub>0.66</sub>Li<sub>0.18</sub>Mn<sub>0.71</sub>Mg<sub>0.21</sub>Co<sub>0.08</sub>O<sub>2</sub> (i-NaLiMMCO) synthesized by co-precipitation method.



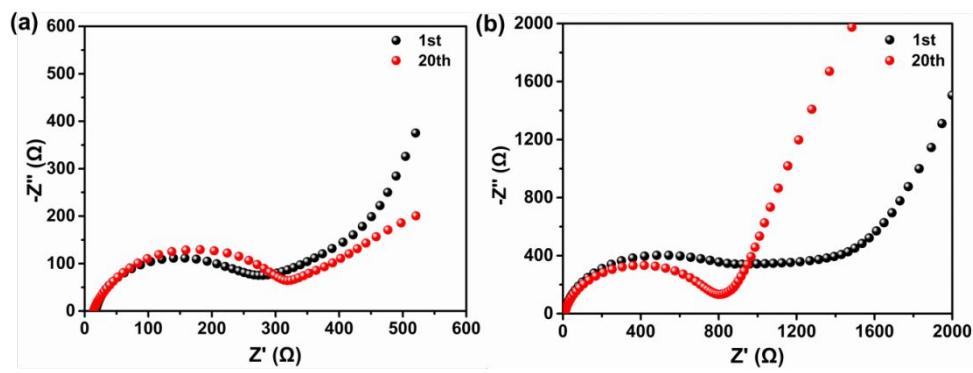
**Figure S3.** (a)  $N_2$  adsorption/desorption isotherm of s-NaLiMMCO. (b)  $N_2$  adsorption/desorption isotherm of i-NaLiMMCO.



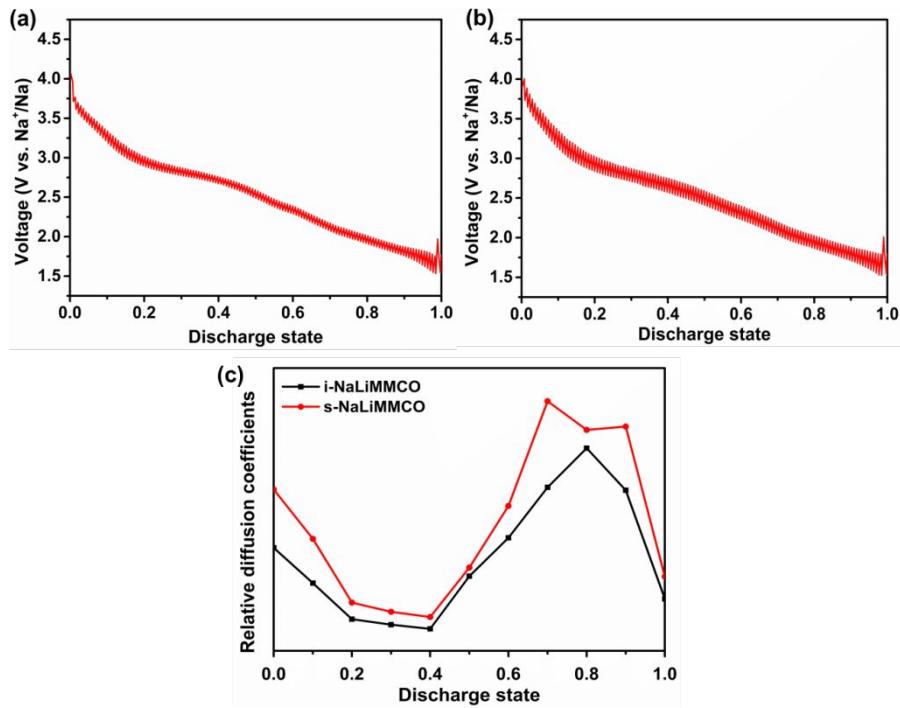
**Figure S4.** (a) The cycle performance of the s-NaLiMMCO at a rate of  $100 \text{ mA g}^{-1}$ . (b) The cycle performance of the i-NaLiMMCO at a rate of  $100 \text{ mA g}^{-1}$ .



**Figure S5.** (a) The selected charge and discharge curves for s-NaLiMMCO at a rate of 100 mA g<sup>-1</sup>. (b) The selected charge and discharge curves for i-NaLiMMCO at a rate of 100 mA g<sup>-1</sup>. (c) Charge and discharge profiles at different rates for s-NaLiMMCO. (d) Charge and discharge profiles at different rates for i-NaLiMMCO.



**Figure S6.** (a) The EIS spectra of the s-NaLiMMCO material for selected cycles at a rate of  $20 \text{ mA g}^{-1}$ . (b) The EIS spectra of the i-NaLiMMCO material for selected cycles at a rate of  $20 \text{ mA g}^{-1}$ .



**Figure S7.** (a) GITT curves of s-NaLiMMCO. (b) GITT curves of i-NaLiMMCO. (c) The calculated Na chemical diffusion coefficients calculated from GITT measurement.

**Table S1.** ICP-OES results of solvothermal precursor

Theoretical chemical formula	Measured atomic ratio		
	Mn	Mg	Co
$\text{Mn}_{0.71}\text{Mg}_{0.21}\text{Co}_{0.08}\text{CO}_3$	0.71	0.20	0.08

**Table S2.** ICP-OES results of P2- $\text{Na}_{0.66}\text{Li}_{0.18}\text{Mn}_{0.71}\text{Mg}_{0.21}\text{Co}_{0.08}\text{O}_2$ 

Theoretical chemical formula	Measured atomic ratio				
	Na	Li	Mn	Mg	Co
$\text{Na}_{0.66}\text{Li}_{0.18}\text{Mn}_{0.71}\text{Mg}_{0.21}\text{Co}_{0.08}\text{O}_2$	0.65	0.18	0.69	0.20	0.07

**Table S3.** Electrochemical performance comparison for P2-type Mn-based cathode material.

Cathode materials	Voltage range	Initial capacity (mA h g <sup>-1</sup> )	Capacity retention	Ref.
Na <sub>0.66</sub> Li <sub>0.18</sub> Mn <sub>0.71</sub> Mg <sub>0.21</sub> Co <sub>0.08</sub> O <sub>2</sub>	1.5-4.5V	166/20 mA g <sup>-1</sup>	82% (100 cycle)	This work
Na <sub>0.78</sub> Ni <sub>0.23</sub> Mn <sub>0.69</sub> O <sub>2</sub>	2.0-4.5V	138/0.1C	90% (20 cycle)	1
Na <sub>2/3</sub> Fe <sub>1/2</sub> Mn <sub>1/2</sub> O <sub>2</sub>	1.5-4.3V	126/13 mA g <sup>-1</sup>	83% (30 cycle)	2
Na <sub>0.67</sub> Ni <sub>0.29</sub> Co <sub>0.13</sub> Mn <sub>0.58</sub> O <sub>2</sub>	2.0-4.3V	164.2/16 mA g <sup>-1</sup>	77.4% (100 cycle)	3
Na <sub>2/3</sub> Ni <sub>1/3</sub> Mn <sub>5/9</sub> Al <sub>1/9</sub> O <sub>2</sub>	1.6-4.0V	118/0.1C	77.5% (100 cycle)	4
Na <sub>0.7</sub> [Cu <sub>x</sub> Fe <sub>y</sub> Mn <sub>1-x-y</sub> ]O <sub>2</sub>	2.5-4.2V	97.8/0.1C	82% (80 cycle)	5
Na <sub>2/3</sub> Fe <sub>1/2</sub> Mn <sub>1/2</sub> O <sub>2</sub>	1.5-4.2V	180 / 0.1C	55% (80 cycle)	6
Na <sub>0.66</sub> Co <sub>x</sub> Mn <sub>0.66-x</sub> Ti <sub>0.34</sub> O <sub>2</sub>	2.0-4.3V	130/0.2C	81% (100 cycle)	7
Na <sub>0.5</sub> [Ni <sub>0.23</sub> Fe <sub>0.13</sub> Mn <sub>0.63</sub> ]O <sub>2</sub>	1.5-4.6V	200/15 mA g <sup>-1</sup>	75% (70 cycle)	8
Na <sub>0.67</sub> Mn <sub>0.65</sub> Ni <sub>0.2</sub> Co <sub>0.15</sub> O <sub>2</sub>	1.5-4.2V	155/12 mA g <sup>-1</sup>	85% (100 cycle)	9
Na <sub>0.67</sub> Mn <sub>0.65</sub> Fe <sub>0.2</sub> Ni <sub>0.15</sub> O <sub>2</sub>	1.5-4.5V	204/0.05C	71% (50 cycle)	10
Na <sub>2/3</sub> Ni <sub>1/3-x</sub> Mg <sub>x</sub> Mn <sub>2/3</sub> O <sub>2</sub>	2.0-4.5V	150/10 mA g <sup>-1</sup>	84% (50 cycle)	11
Na <sub>2/3</sub> (Mn <sub>1/2</sub> Fe <sub>1/4</sub> Co <sub>1/4</sub> )O <sub>2</sub>	1.5-4.5V	195/0.1C	71% (30 cycle)	12
Na <sub>0.67</sub> Mn <sub>0.67</sub> Ni <sub>0.33-x</sub> Mg <sub>x</sub> O <sub>2</sub>	2.5-4.35V	123/17 mA g <sup>-1</sup>	85% (50 cycle)	13
Na <sub>0.66</sub> Ni <sub>0.26</sub> Zn <sub>0.07</sub> Mn <sub>0.67</sub> O <sub>2</sub>	2.0-4.4V	132/12 mA g <sup>-1</sup>	89% (30 cycle)	14
Na <sub>0.67</sub> Ni <sub>0.2</sub> Cu <sub>0.1</sub> Mn <sub>0.7</sub> O <sub>2</sub>	2.0-4.5V	125.3/17 mA g <sup>-1</sup>	72% (100 cycle)	15
Na <sub>0.44</sub> Mn <sub>0.6</sub> Ni <sub>0.4-x</sub> Cu <sub>x</sub> O <sub>2</sub>	1.5-4.0V	149.2/17 mA g <sup>-1</sup>	80% (50 cycle)	16
Na <sub>0.67</sub> Mn <sub>0.8</sub> Ni <sub>0.1</sub> Mg <sub>0.1</sub> O <sub>2</sub>	1.5-4.2V	171/0.05C	79% (50 cycle)	17
Na <sub>0.67</sub> Li <sub>0.2</sub> (Ni <sub>0.2</sub> Fe <sub>0.15</sub> Mn <sub>0.65</sub> ) <sub>0.8</sub> O <sub>2</sub>	1.5-4.3V	151/0.1C	78% (50 cycle)	18

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