Supporting Information: Lipid dynamics and phase transition within α -synuclein amyloid fibrils

Céline Galvagnion,^{*,†,‡,¶} Daniel Topgaard,[§] Katarzyna Makasewicz,[§] Alexander K. Buell,[∥] Sara Linse,[⊥] Emma Sparr,[§] and Christopher M. Dobson^{†,#}

[†]Centre for Misfolding Diseases, Department of Chemistry, University of Cambridge, Lensfield Road, Cambridge CB2 1EW, United Kingdom

‡German Center for Neurodegenerative Diseases, Sigmund-Freud-Str. 27, 53127, Bonn, Germany

¶Present address: Department of Drug Design and Pharmacology, Faculty of Health and Medical Sciences, University of Copenhagen, 2100 Copenhagen, Denmark

§Division of Physical Chemistry, Center for Chemistry and Chemical Engineering, Lund University, P.O. Box 124, SE-22100, Lund, Sweden

||Department of Biotechnology and Biomedicine, DTU Bioengineering, Technical University of Denmark, Soltofts Plads 227, DK-2800 Kgs. Lyngby, Denmark

 $\perp Department of Biochemistry and Structural Biology, Lund University , SE22100 Lund,$

Sweden

#Deceased September 8th, 2019

E-mail: celine.galvagnion@sund.ku.dk

Materials and methods

Materials

1,2-Dimyristoyl-*sn*-glycero-3-phospho-L-serine (sodium salt; DMPS) and 1,2-dilauroyl-*sn*-glycero-3-phospho-L-serine (sodium salt: DLPS) were purchased from Avanti Polar Lipids (Alabama, USA). Sodium phosphate monobasic (NaH₂PO₄, BioPerformance Certified, > 99.0%), sodium phosphate dibasic (Na₂HPO₄, ReagentPlus, > 99.0%) and sodium azide (NaN₃, ReagentPlus, > 99.5%) were purchased from Sigma Aldrich (Poole, UK).

Protein and lipid preparation

 α -synuclein was expressed and purified using the same protocol as that reported previously.^{S1–S3} Lipid vesicles used for the aggregation experiment and differential scanning calorimetry measurement were prepared via extrusion and sonication, respectively. Briefly, lipids (DMPS or DLPS) in powder form were suspended in phosphate buffer and the solution was stirred for 4 h at 45°C (a temperature above the melting temperature of both DLPS and DMPS). Then the lipid suspensions were sonicated on ice or extruded through 100nm pore membranes (Avanti Polar Lipids, Alabama, USA) at 45°C.

Cryo-electron Microscopy (Cryo-EM)

50 μ M α -synuclein was incubated in phosphate buffer in the presence of lipid vesicles made with DMPS or DLPS (100 μ M) under quiescent conditions and at 30°C for 4 d. The grids were then prepared as described previously.^{S4} Briefly, reaction mixtures containing α -synuclein proto-fibrils were deposited on lacey carbon filmed copper grids, which were then plunged into liquid ethane at -180°C. The resulting grids were stored under liquid nitrogen until imaged using an electron microscope (Philips CM120 BioTWIN Cryo) equipped with a post-column energy filter (Gatan GIF100) (acceleration voltage: 120 kV). The images were recorded digitally with a CCD camera under low electron dose conditions.

MAS NMR

Sample preparation

The proto-fibrils were prepared by incubating 100 μ M α -synuclein and 2 mM DMPS or DLPS solubilised as vesicles in phosphate buffer (20 mM NaH₂PO₄/Na₂HPO₄, 0.01 % NaN₃, pH 6.5) in Corning[®] 96-well half area black with clear flat bottom made of polystyrene treated with NBSTM (#3881, Corning Ltd, Corning, USA) under quiescent conditions at 30°C for 4 d. Reaction mixtures were then collected from each well, gathered and centrifuged at 90 krpm for 1 h. The pellet was then inserted into NMR rotor inserts using the following procedure: the insert was attached to a 10 μ L tip using parafilm and the sample was deposited at the larger end of the tip. The insert attached to the tip was then centrifuged for 30 s at 1,000 g to let the sample slide into the rotor.

Samples of protein-free lipid lamellar phase, referred to as "pure lipid system" throughout the text, were prepared by suspending 10 mg DMPS or DLPS powder in phosphate buffer for 2 h at maximum stirring and 50°C. The samples were then centrifuged at 90 krpm for 1 h and the pellets transferred into the rotor, as described above. All experiments were performed with fully hydrated samples to avoid dehydration-induced changes in lipid self-assembly.

Data acquisition

A set of four spectra was acquired for each sample: ¹³C MAS direct polarisation (DP), ¹³C MAS cross-polarisation (CP), ¹³C MAS Insensitive Nuclei Enhanced by Polarization Transfer (INEPT) and ³¹P MAS NMR spectra. Measurements were performed using a Bruker Avance-II 500 spectrometer (Bruker, Karlsruhe, Germany) equipped with a 4 mm CP/MAS HX probe at a field of 11.7T, resulting in ¹H, ¹³C and ³¹P resonance frequencies of 500, 125 and 200 MHz, respectively. For temperatures below 20°C, we used a BVT-2000 temperature control and a BCU-05 cooling box unit, with sample heating induced by MAS and the radiofrequency pulses being taken into account. When temperatures were set to values ranging from 25 to 60 °C, the same procedure without the cooling box unit was used. Polarisation Transfer (PT) ¹³C MAS NMR spectra were acquired using a spectral width of 200 ppm and an acquisition time of 50 ms, under 67 kHz TPPM ¹H decoupling.^{S5} For each ¹³C spectrum, 3200 scans were accumulated with a recycle delay of 4 s. ¹³C chemical shifts were externally referenced to solid α -glycine at 43.67 ppm.^{S6 1}H and ¹³C hard pulses were applied at a $\frac{\omega_1^{H/C}}{2\pi} = 80$ kHz, CP was performed with $t_{CP} = 1$ ms, $\frac{\omega_1^C}{2\pi} = 80$ kHz and $\frac{\omega_1^{H/C}}{2\pi}$ linearly ramped from 72 to 88 kHz, covering the $\pm \omega_R$ matching conditions, and INEPT spectra were recorded with the delay times $\tau = 1.8$ ms and $\tau' = 1.2$ ms. The experimental time-domain data were processed as described previously^{S4} with a Matlab in-house code partially derived from matNMR,^{S7} using line broadening of 20 Hz, and zero-filling of 8192 time-domain points. DMPS and DLPS peak assignments were made based on DMPC spectra, as described in previous studies.^{S8}

 31 P spectra were acquired using a spectral width of 200 ppm, an acquisition time of 50 ms, 2048 scans and a recycle delay of 4s. The rate of spinning in the MAS experiments was 5000Hz for the 13 C spectra and 1250Hz for 31 P spectra.

³¹P MAS NMR - Data analysis

Isotropic chemical shift values were set using the resonance of the phosphate buffer ($\delta = 0$ ppm). ³¹P chemical shift anisotropy values ($\Delta \sigma$) were determined using Herzfeld-Berger sideband analysis^{S9} as implemented in matNMR.^{S7} The anisotropic part of the $\Delta \sigma$ is defined as:

$$\Delta \sigma = \sigma_{33} - \frac{1}{2}(\sigma_{22} - \sigma_{11}) \tag{1}$$

where σ_{ii} are the principal tensor components. When the lipid bilayer is in the liquid crystalline phase, the chemical shielding tensor is averaged to an effective tensor that is axially

symmetric.^{S10} The anisotropic part of this time-average tensor has been defined as:

$$\Delta \sigma = \sigma_{\parallel} - \sigma_{\perp} = \frac{3}{2} (\sigma_{\parallel} - \sigma_{\rm i}) \tag{2}$$

where σ_i is the isotropic chemical shift, σ_{\parallel} is the low intensity shoulder ($\sigma_{\parallel} = \sigma_{33}$), and σ_{\perp} is the high intensity shoulder ($\sigma_{\perp} = \sigma_{11} = \sigma_{22}$) of the axially symmetric powder pattern.^{S10}

Differential Scanning Calorimetry

Sample preparation

DMPS-induced α -synuclein proto-fibrils were prepared as described above in the "MAS NMR - Sample preparation" section and DMPS vesicles were prepared as described above in the "Protein and lipid preparation" section.

Data acquisition

All the samples were degassed before the acquisition of the DSC thermograms. The thermograms were acquired using a Microcal VP-DSC calorimeter (Malvern Instruments) with a scanning rate of 1°C.min⁻¹ from 5 to 65°C. All of the DSC thermograms reported in this article were corrected by subtracting the thermogram of the phosphate buffer and correspond to the first scan, unless otherwise stated.

Proteinase-K treatment

The proteinase-K digested fibrils were prepared by incubating DMPS-induced α -synuclein proto-fibrils (prepared as described in "MAS NMR - Sample preparation" section) in the presence of 3 μ M proteinase-K (Ambion, Germany) under quiescent conditions for 3 h. The mixture was then placed in the calorimeter for measurements.

Supplementary Figures

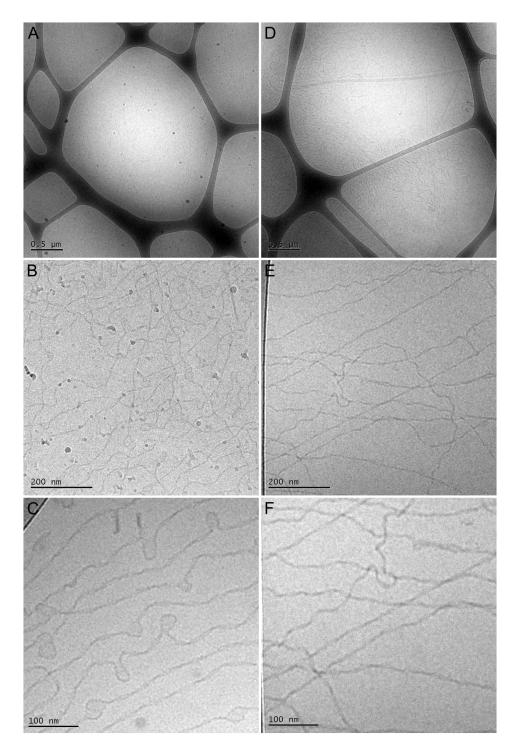


Figure S1: Cryo-EM images of proto-fibrils formed by α -synuclein in the presence of DLPS and DMPS vesicles. The proto-fibrils were formed after mixing 50 μ M monomeric α -synuclein with 100 μ M DMPS (A-C) or 200 μ M DLPS (D-F) solubilised as small unilamellar vesicles in phosphate buffer at pH 6.5 and 30°C, and incubating this mixture for 4 d under quiescent conditions (see Methods for further details).

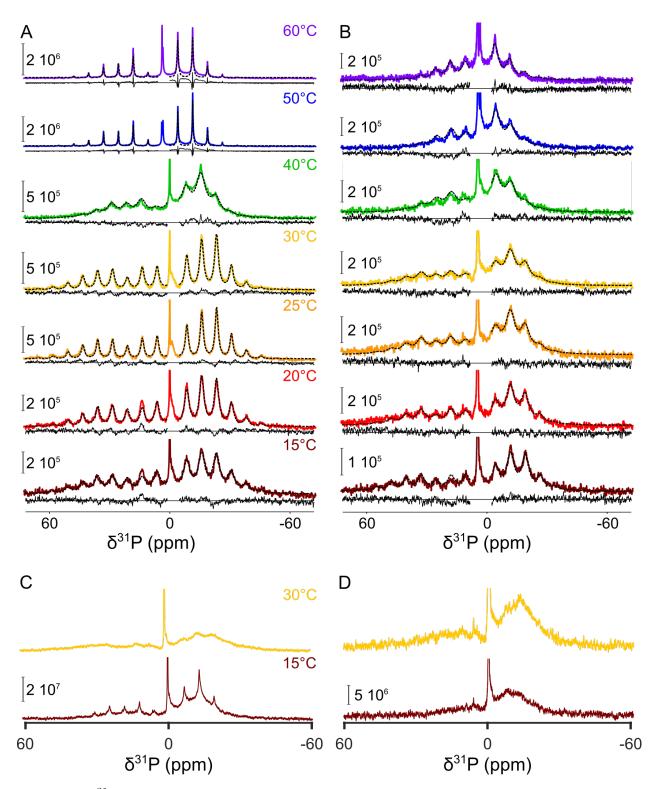


Figure S2: ³¹P MAS NMR spectra of pure lipid systems (A, DMPS, C, DLPS) and lipidinduced α -synuclein proto-fibrils (B, DMPS, D, DLPS) measured at different temperatures. The proto-fibrils were formed by incubating 100 μ M α -synuclein in the presence of 2mM DMPS or DLPS dispersed as vesicles in phosphate buffer at pH 6.5 and 30°C, and incubating this mixture for 4 d under quiescent conditions.

References

- (S1) Galvagnion, C.; Buell, A. K.; Meisl, G.; Michaels, T. C. T.; Vendruscolo, M.; Knowles, T. P. J.; Dobson, C. M. Lipid vesicles trigger alpha-synuclein aggregation by stimulating primary nucleation. *Nature chemical biology* **2015**, *11*, 229–34.
- (S2) Buell, A. K.; Galvagnion, C.; Gaspar, R.; Sparr, E.; Vendruscolo, M.; Knowles, T. P. J.; Linse, S.; Dobson, C. M. Solution conditions determine the relative importance of nucleation and growth processes in alpha-synuclein aggregation. *Proceedings of the National Academy of Sciences of the United States of America* 2014, 111, 7671–6.
- (S3) van der Wateren, I. M.; Knowles, T. P. J.; Buell, A. K.; Dobson, C. M.; Galvagnion, C. C-terminal truncation of α-synuclein promotes amyloid fibril amplification at physio-logical pH. *Chem. Sci.* **2018**, *9*, 5506–5516.
- (S4) Hellstrand, E.; Nowacka, A.; Topgaard, D.; Linse, S.; Sparr, E. Membrane lipid coaggregation with alpha-synuclein fibrils. *PloS one* **2013**, *8*, e77235.
- (S5) Bennett, A. E.; Rienstra, C. M.; Auger, M.; Lakshmi, K. V.; Griffin, R. G. Heteronuclear decoupling in rotating solids. *The Journal of Chemical Physics* 1995, 103, 6951–6958.
- (S6) Hayashi, S.; Hayamizu, K. Chemical Shift Standards in High-Resolution Solid-State NMR (1) 13C, 29Si, and 1H Nuclei. Bulletin of the Chemical Society of Japan 1991, 64, 685–687.
- (S7) van Beek, J. D. matNMR: a flexible toolbox for processing, analyzing and visualizing magnetic resonance data in Matlab. *Journal of magnetic resonance (San Diego, Calif.* : 1997) 2007, 187, 19–26.
- (S8) Nowacka, A.; Douezan, S.; Wadsö, L.; Topgaard, D.; Sparr, E. Small polar molecules

like glycerol and urea can preserve the fluidity of lipid bilayers under dry conditions. Soft Matter 2012, 8, 1482–1491.

- (S9) Herzfeld, J.; Berger, A. E. Sideband intensities in NMR spectra of samples spinning at the magic angle. J. Chem. Phys 1980, 73, 6021–6030.
- (S10) Seelig, J. 31P nuclear magnetic resonance and the head group structure of phospholipids in membranes. *Biochim. Biophys. Acta* 1978, 515, 105–40.