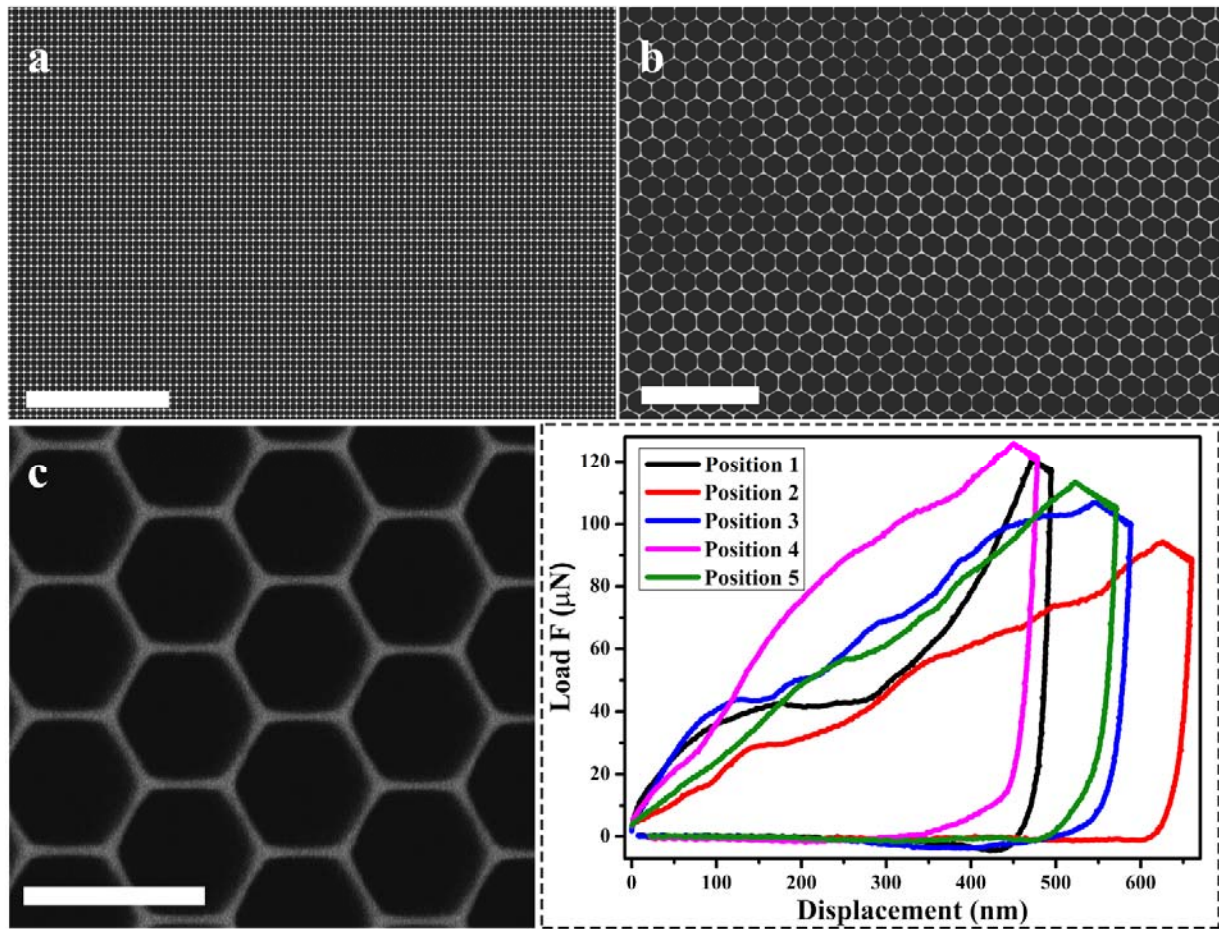


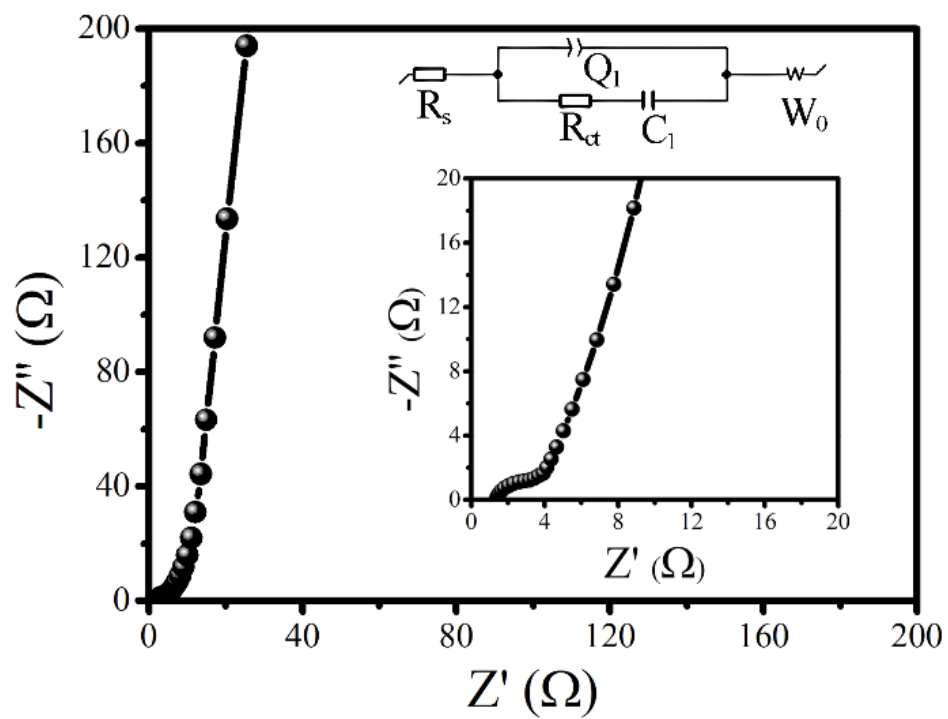
## **Supplementary Information for**

**Nanoelectrode design from microminiaturized honeycomb monolith with ultrathin and stiff nanoscaffold for high-energy micro-supercapacitors**

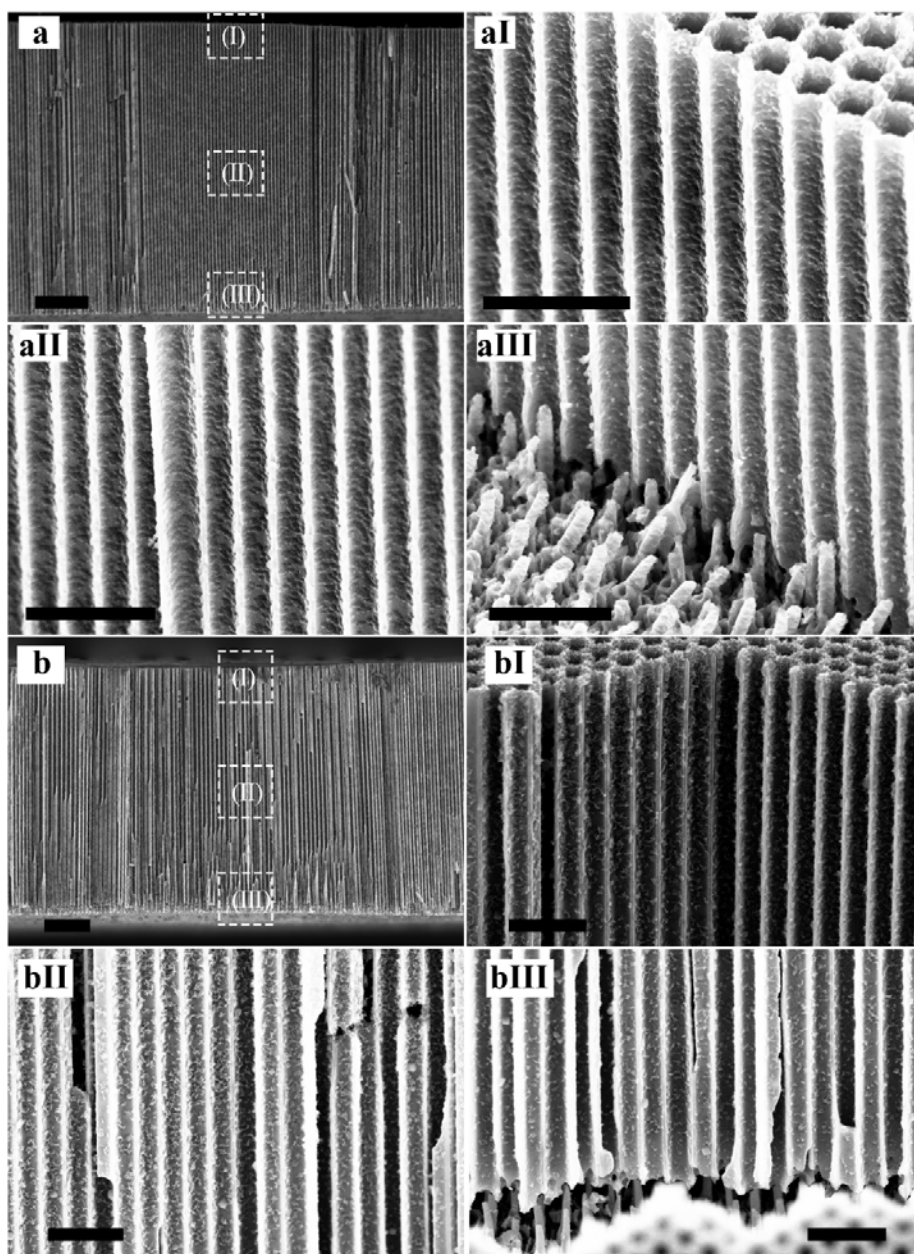
*Lei et al.*



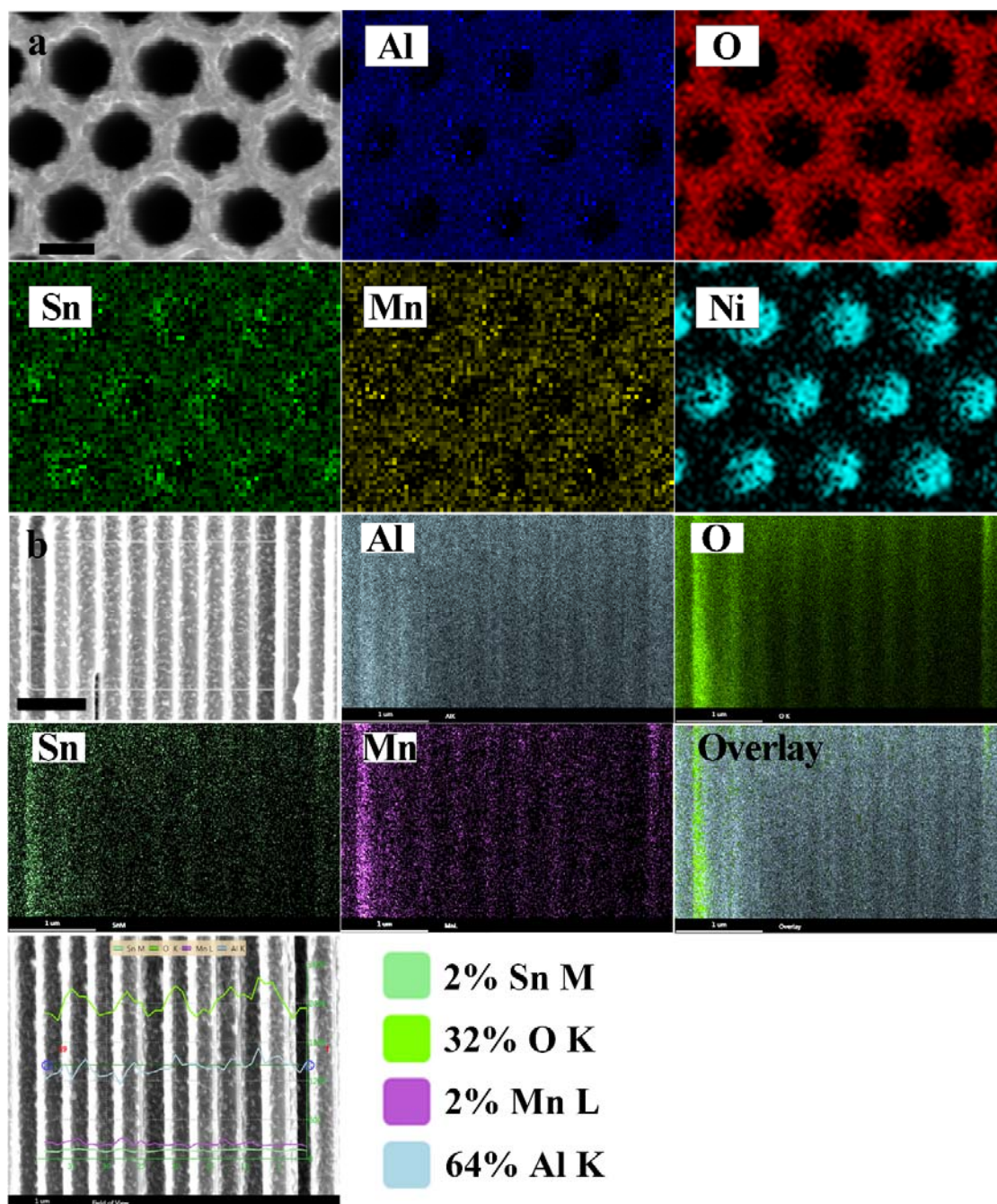
**Supplementary Figure 1 | Morphology and mechanical properties of HAN.** **a** Large-scale SEM image of HAN with square cell arrangement and 400 nm inter-cell spacing. **b** Large-scale SEM image of HAN with hexagonal cell arrangement and 800 nm inter-cell spacing. **c** The load-displacement (L-D) curves at the different positions of the HAN (hexagonal cell arrangement and 400 nm inter-cell spacing) with an indentation load of 150  $\mu\text{N}$  at room temperature. The Young's modulus of the HAN obtained from L-D curves are 3.22, 1.46, 1.50, 2.05 and 1.41 GPa at the positions 1 – 5, respectively. Scale bar: 8  $\mu\text{m}$  (**a**); 4  $\mu\text{m}$  (**b**); 500 nm (**c**)



**Supplementary Figure 2 | Electrochemical impedance of HAN@SnO<sub>2</sub>.** The Nyquist plots of HAN@SnO<sub>2</sub>/HAN@SnO<sub>2</sub> device.

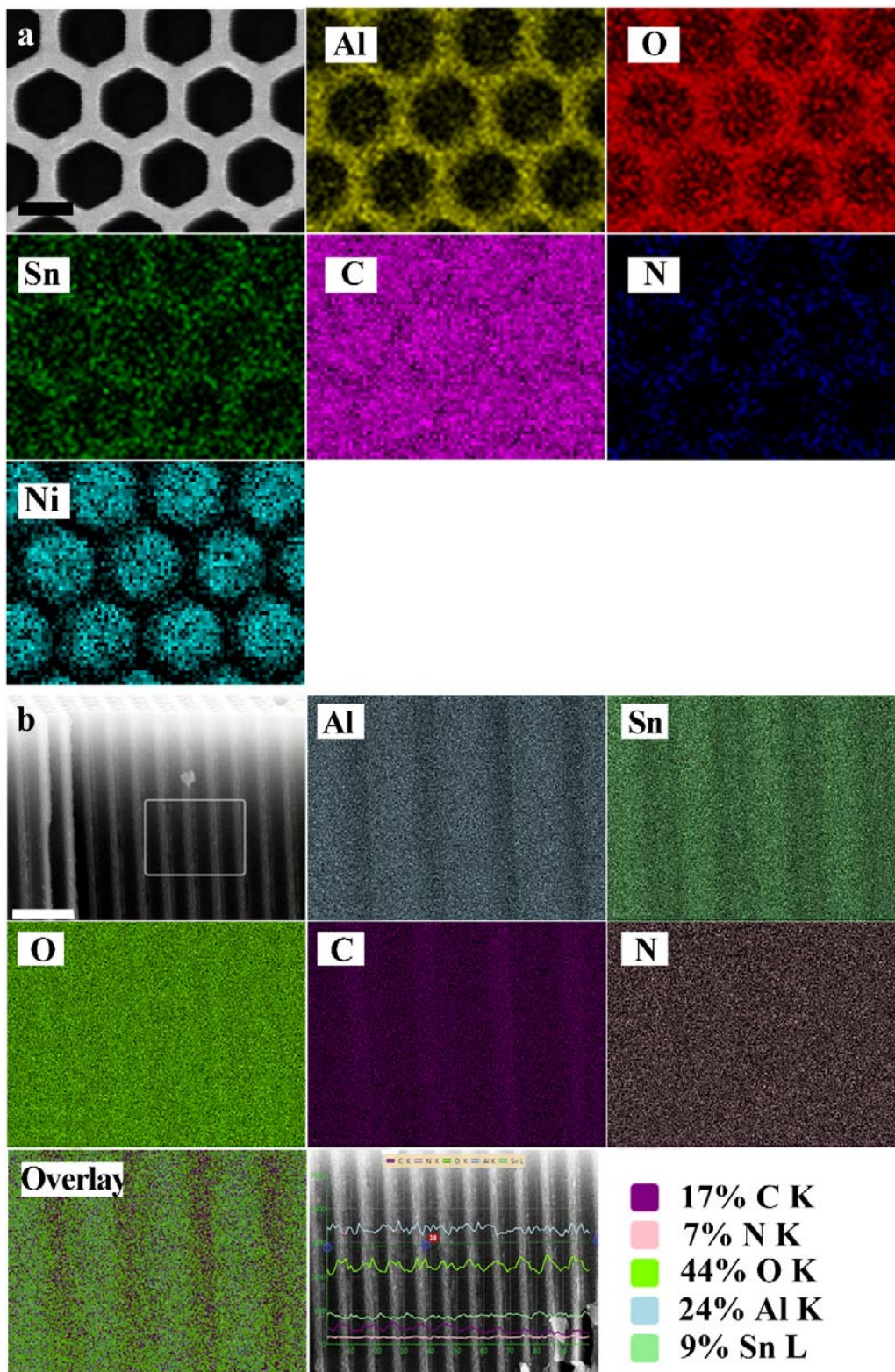


**Supplementary Figure 3 | Cross-sectional SEM images of HAN@SnO<sub>2</sub>@PPy and HAN@SnO<sub>2</sub>@MnO<sub>2</sub> electrodes with a HAN cell depth of 25 μm. a Overall view of HAN@SnO<sub>2</sub>@PPy electrode and corresponding high-resolution SEM image recorded from the region marked in aI, aII and aIII. b Overall view of HAN@SnO<sub>2</sub>@MnO<sub>2</sub> electrode and corresponding high-resolution SEM image recorded from the region marked in bI, bII and bIII. Scale bar: 5 μm (a); 1 μm (aI); 1 μm (aII); 1 μm (aIII); 5 μm (b); 1 μm (bI); 1 μm (bII); 1 μm (bIII)**



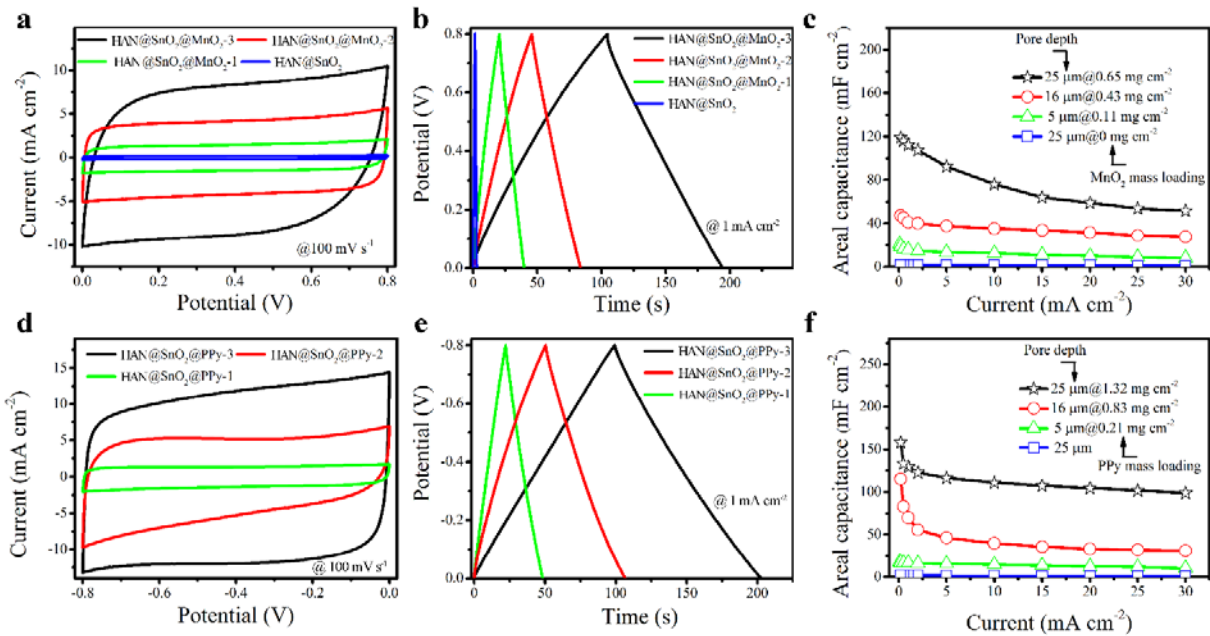
Supplementary Figure 4 | Top-view and cross-sectional EDX mapping of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>. a

Top-view and b cross-sectional view. Scale bar: 250 nm (a); 1 μm (b)

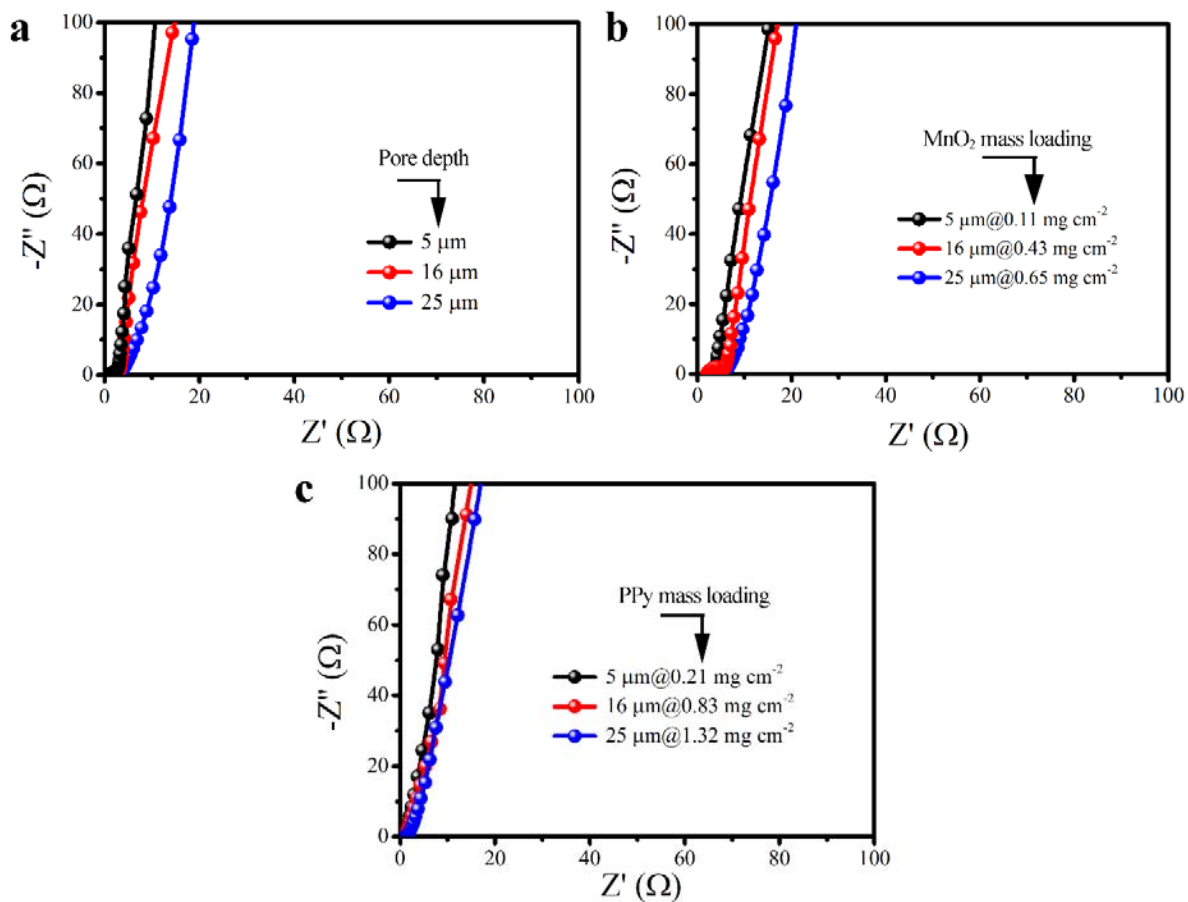


**Supplementary Figure 5 | EDX mapping of HAN@SnO<sub>2</sub>@PPy. a** Top-view and **b** cross-sectional view.

Scale bar: 250 nm (a); 500 nm (b)

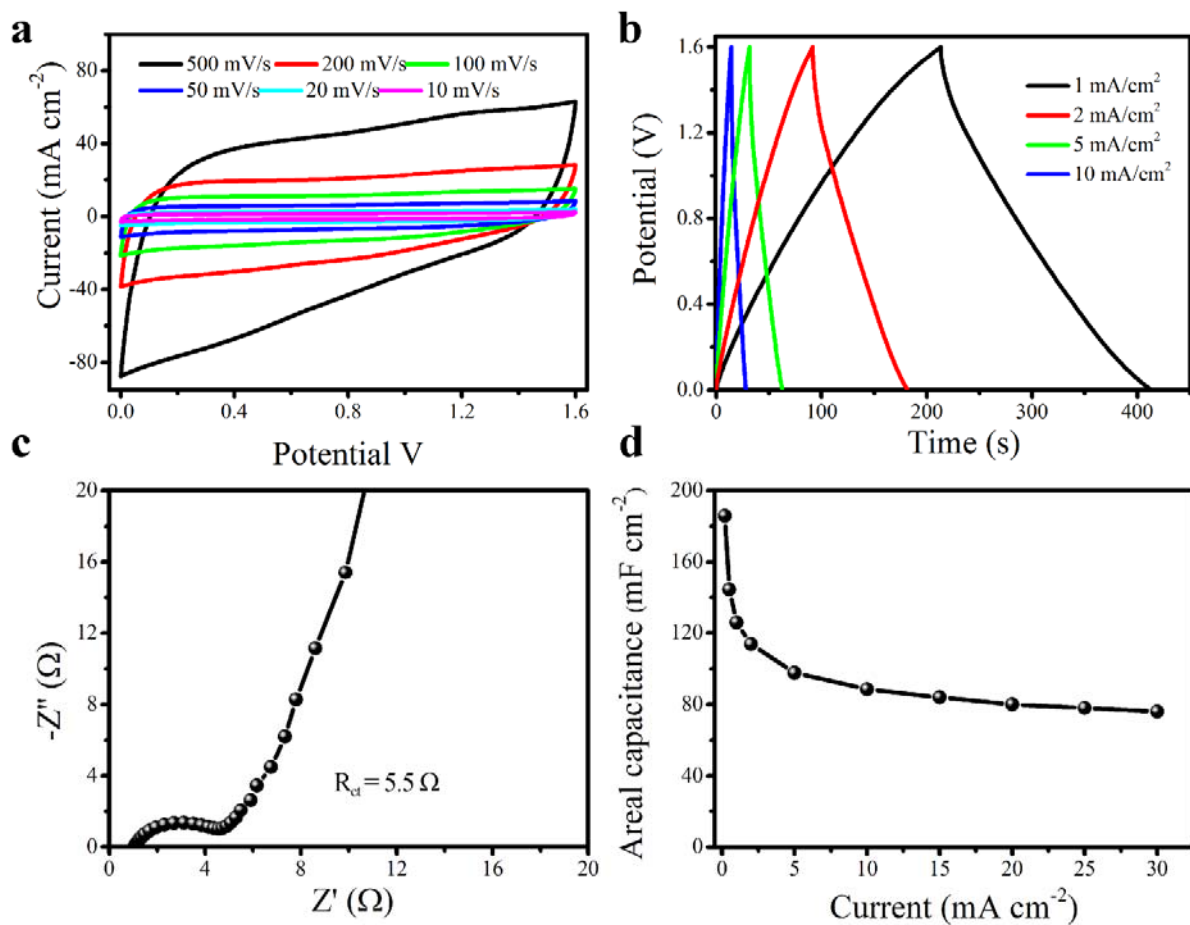


**Supplementary Figure 6 | Electrochemical performance of the symmetric stacked MSCs based on HAN-reinforced pseudocapacitive nanoelectrodes.** **a** The CV curves, **b** GCD profiles and **c** device areal capacitance as a function of current densities of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>/HAN@SnO<sub>2</sub>@MnO<sub>2</sub> MSCs with different pore thickness. **d** The CV curves, **e** GCD profiles and **f** device areal capacitance as a function of current densities of HAN@SnO<sub>2</sub>@PPy/HAN@SnO<sub>2</sub>@PPy MSCs with different HAN cell depth. The electrolyte is 1.0 M Na<sub>2</sub>SO<sub>4</sub> aqueous solution.

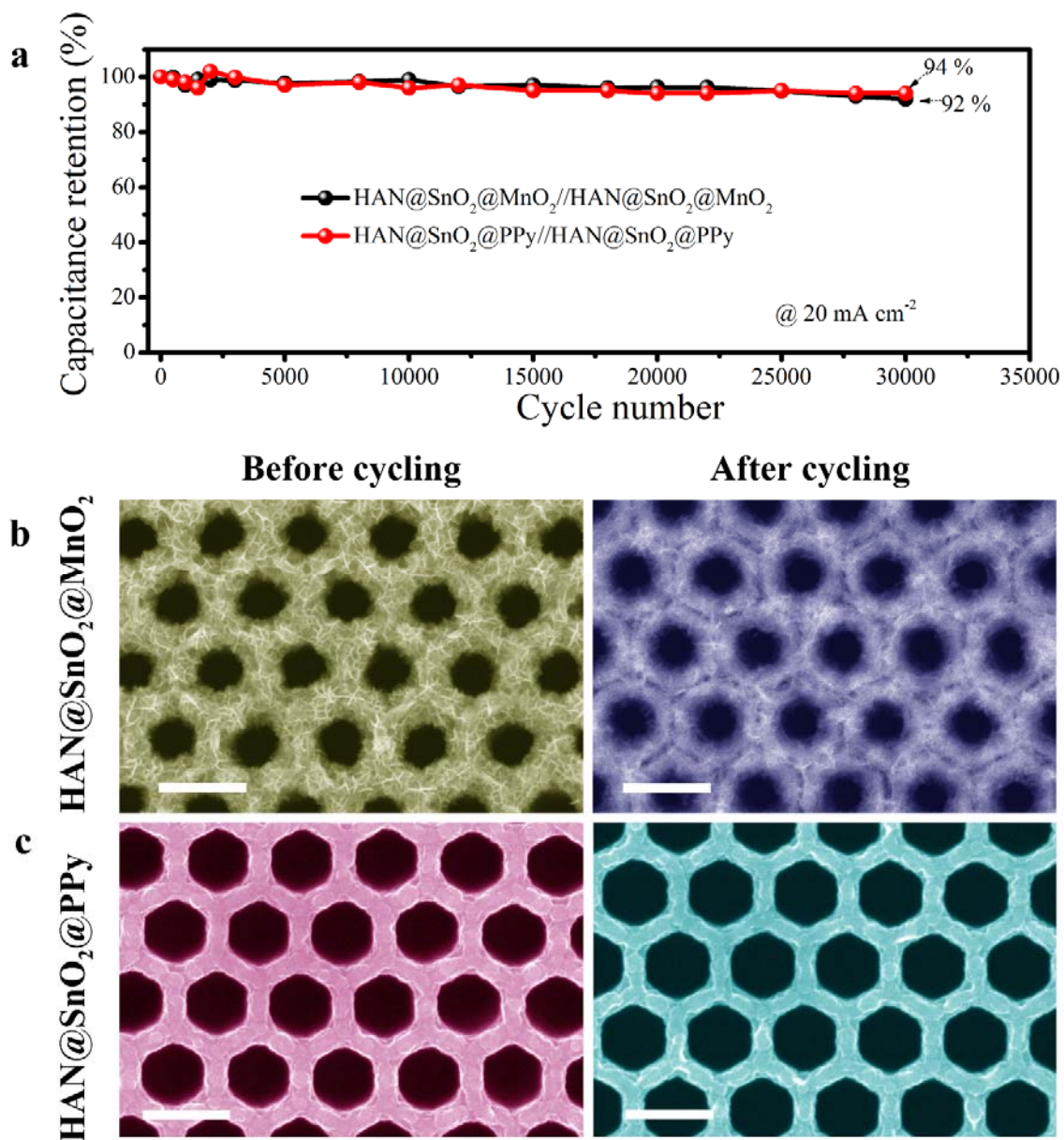


**Supplementary Figure 7 | Electrochemical impedance properties comparison of symmetric MSCs.** Warburg region of **a** HAN@SnO<sub>2</sub>, **b** HAN@SnO<sub>2</sub>@MnO<sub>2</sub>, and **c** HAN@SnO<sub>2</sub>@PPy electrodes with different pore depth of HAN. The electrolyte is 1.0 M Na<sub>2</sub>SO<sub>4</sub> aqueous solution.

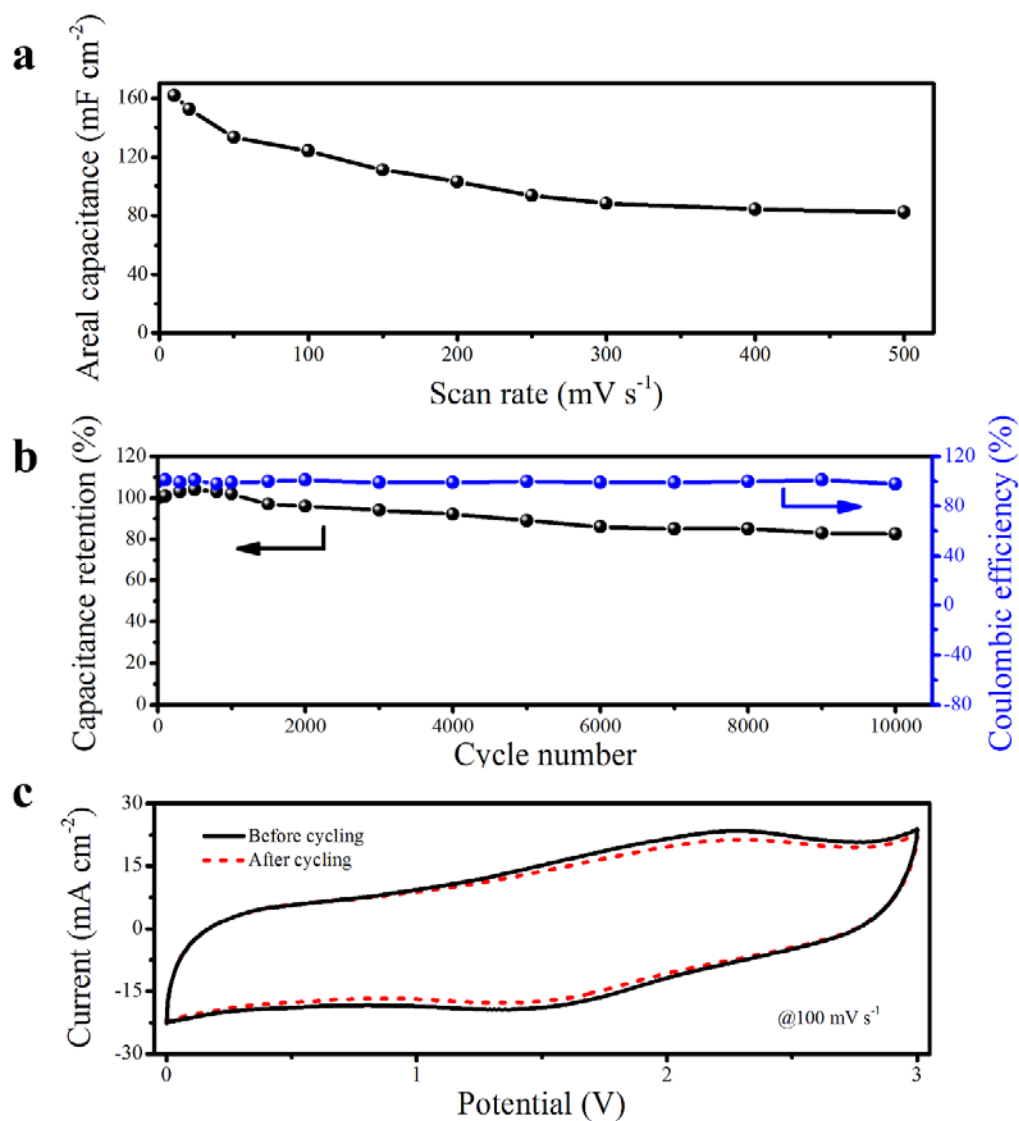




**Supplementary Figure 8 | Electrochemical performance of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>//HAN@SnO<sub>2</sub>@PPy asymmetric stacked MSCs. a** The CV curves, **b** GCD profiles, **c** Nyquist plots and **d** device areal capacitance as a function of current densities of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>//HAN@SnO<sub>2</sub>@PPy MSCs (For asymmetric MSCs based on positive and negative electrodes are both with 25- $\mu\text{m}$ -deep cell). The electrolyte is 1.0 M Na<sub>2</sub>SO<sub>4</sub> aqueous solution.



**Supplementary Figure 9 | Cyclic performance of symmetric MSCs and morphological stability of HAN-based nanoelectrodes.** **a** Cycling ability of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>//HAN@SnO<sub>2</sub>@MnO<sub>2</sub> symmetric MSCs and HAN@SnO<sub>2</sub>@PPy//HAN@SnO<sub>2</sub>@PPy symmetric MSCs, respectively, measured at a current density of 20 mA cm<sup>-2</sup> for 30,000 continued charge-discharge cycles. SEM images of **b** HAN@SnO<sub>2</sub>@MnO<sub>2</sub> and **c** HAN@SnO<sub>2</sub>@PPy electrodes before (left) and after (right) 30,000 cycles. Scale bar: 400 nm (**b, c**)



**Supplementary Figure 10 | Device capacity of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>//HAN@SnO<sub>2</sub>@PPy asymmetric stacked MSCs with EMIM-TFSI electrolyte. a** Device areal capacitance as a function of scan rates of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>//HAN@SnO<sub>2</sub>@PPy MSCs. **b** Cycling stability test at 20 mA cm<sup>-2</sup> for 10000 times and **c** corresponding CV curves at 100 mV s<sup>-1</sup> before and after cycling. The positive and negative electrodes are both with 25- $\mu\text{m}$ -deep HAN.

**Supplementary Table 1 | Performances of state-of-the-art MSCs.** Device performance of HAN@SnO<sub>2</sub>@MnO<sub>2</sub>//HAN@SnO<sub>2</sub>@PPy (abbreviated as MnO<sub>2</sub>//PPy) in comparison with various MSCs reported recently.

Electrode material	Electrode thickness (μm)	Electrolyte	Potential window (V)	Device capacitance (mF cm <sup>-2</sup> )	Device energy (μWh cm <sup>-2</sup> )	Device power (mW cm <sup>-2</sup> )	Cycling stability	C <sub>1000</sub> /C <sub>0</sub> (%)	Ref.
Graphene	0.7	PSSH	1.0	<1	<1	<1	11,000	~100	1
PG	2.0	BMIMPF <sub>6</sub>	3.0	<1	<1	<1	2,000	~100	2
CNTCs	20	BMIM-BF <sub>4</sub>	3.0	6	3.6	69	8,000	~100	3
ACF	1.5	1 M H <sub>2</sub> SO <sub>4</sub>	0.6	<1	<1	-	10,000	~100	4
OLC	7	1 M Et <sub>4</sub> NBF <sub>4</sub> in PC	3.0	<1	<1	177	10,000	~100	5
AC	5	0.5 M H <sub>2</sub> SO <sub>4</sub>	3.0	<1	<1	20	10,000	~100	5
LWG	20	1.0 M TEABF <sub>4</sub>	1.0	2.5	<1	-	10,000	~100	6
Graphene	25	1 M H <sub>2</sub> SO <sub>4</sub>	1.0	4	<1	9	9,000	~98	7
PEDOT	2.4	1 M H <sub>2</sub> SO <sub>4</sub>	0.8	18.7	2	0.2	1,000	~90	8
CDC	4.1	2 M EMIBF <sub>4</sub> in AN	3.0	41.8	40	30	11,000	~100	9
LSG/MnO <sub>2</sub>	15	1 M Na <sub>2</sub> SO <sub>4</sub>	0.9	385	60	-	10,000	100	10
MnO <sub>2</sub> //Bi <sub>2</sub> O <sub>3</sub>	-	1 M Na <sub>2</sub> SO <sub>4</sub>	1.8	97	43.4	12.9	4,000	90	11
MnO <sub>2</sub> //PPY	1.8	1 M Na <sub>2</sub> SO <sub>4</sub>	1.7	~30	~10	0.5	2,000	90	12

NN@MnO <sub>2</sub>	5	1 M Na <sub>2</sub> SO <sub>4</sub>	0.8	~100	20	10	5,000	~90	13
Zn NSs	50	2 M ZnSO <sub>4</sub>	1.5	~324	115	0.16	10,000	~100	14
rGO/IL/CP	0.3	PVA-H <sub>3</sub> PO <sub>4</sub>	3.0	27.4	32	1.05	-	-	15
MXene	-	PVA-H <sub>2</sub> SO <sub>4</sub>	0.5	43	0.32	0.15	14,000	~97	16
MHCF	~0.18	PVA-LiCl	1.8	19.8	9.49	-	5,000	96.8	17
MXene	3.2	EMIMBF <sub>4</sub> /PVDF-HFP	3.0	42	13.9	4.5	10,000	~100	18
Y-Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	~0.1	PVA/H <sub>2</sub> SO <sub>4</sub>	0.6	61	0.63	0.33	10,000	93.7	19
WJM-G-SWCNTs	~23	PVA-H <sub>3</sub> PO <sub>4</sub>	1.8	1.32	0.064	20	10,000	~100	20
H-SiC	~16	PVA-KCl	0.8	23.6	5.2	11.2	10,000	94	21
Cu(OH) <sub>2</sub> @FeOOH	14	EMIMBF <sub>4</sub>	1.5	58	18	-	10,000	80	22
3D RuO <sub>2</sub>	46	0.5 M H <sub>2</sub> SO <sub>4</sub>	0.9	-	91.4	-	2,000	~100	23
PDMS/GF	260	0.5 M Na <sub>2</sub> SO <sub>4</sub>	1.6	592	106	16	12,000	~100	24
Ni-CAT MOF	~200	PVA-LiCl	1.4	15.2	4.1	7	5,000	87	25
MnO <sub>2</sub> -ITO NWs	28	1 M Na <sub>2</sub> SO <sub>4</sub>	1.0	194	27	15	20,000	61	26
<b>MnO<sub>2</sub>//PPy</b>	<b>25 μm</b>	<b>1 M Na<sub>2</sub>SO<sub>4</sub></b>	<b>1.6</b>	<b>186</b>	<b>66</b>	<b>24</b>	<b>30,000</b>	<b>~100</b>	<b>This</b>
<b>MnO<sub>2</sub>//PPy</b>	<b>25 μm</b>	<b>EMIM-TFSI</b>	<b>3.0</b>	<b>128</b>	<b>160</b>	<b>40</b>	<b>10,000</b>	<b>~100</b>	<b>work</b>

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