Supporting information

Rendering redox reactions of cathodes in Li-ion capacitors enabled by lanthanides

Kaiqiang Zhang^{†,‡}, Tae Hyung Lee[†], Mohammad A. Khalilzadeh[§], Rajender S. Varma^{*, II}, Ji-Won Choi^{*,‡}, Ho Won Jang^{*,†}, Mohammadreza Shokouhimehr^{*,†}

[†]Department of Materials Science and Engineering, Research Institute of Advanced Materials, Seoul National University, Seoul 08826, Republic of Korea

[‡]Electronic Materials Center, Korea Institute of Science and Technology (KIST), Seoul 136-791, Republic of Korea

[§]Department of Forest Biomaterials, College of Natural Resources, North Carolina State University, Raleigh, North Carolina 27607, United States

"Regional Center of Advanced Technologies and Materials, Department of Physical Chemistry, Faculty of Science, Palacky University, Š lechtitelů 27, 783 71 Olomouc, Czech Republic

*Corresponding authors: varma.rajender@epa.gov (R.S.V.), jwchoi@kist.re.kr (J.-W. Choi), hwjang@snu.ac.kr (H. W. Jang), mrsh2@snu.ac.kr (M. Shokouhimehr)



Figure S1. Voltage profiles during the repeated cycling measurement of bare MC demonstrating the capacitive Li-ion-storage process.



Figure S2. (a) Voltage profiles and (b) magnified part of the bare MC during the consecutive cycling measurement. It depicts a stable charge–discharge behavior.



Figure S3. (a) XRD pattern, (b) Raman spectra, and (c) TGA curve of MC.



Figure S4. EIS curves of the (a) MC/Gd and (b) MC/La samples. (c) Equivalent circuit of the EIS curves with diverse components in the electrochemical system.



Figure S5. Repeated charge–discharge operations for (a) MC/Gd and (b) MC/La at a current density of 100 mA g^{-1} .



Figure S6. (a) SEM and (b) TEM images of the MC/Gd sample after long-term repeated charge–discharge measurements.



Figure S7. (a) STEM and (b,c) EDX mapping images of the MC/La sample after long-term repeated charge–discharge measurements.