

Supplementary Information for

Strain-driven growth of ultra-long two-dimensional nano-channels

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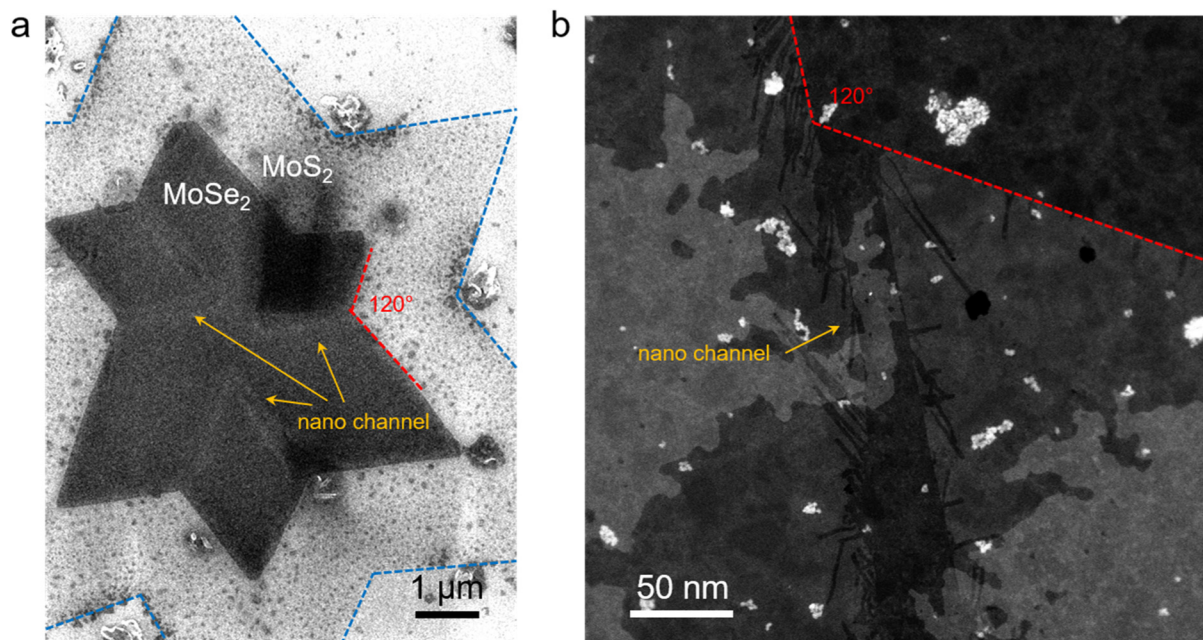
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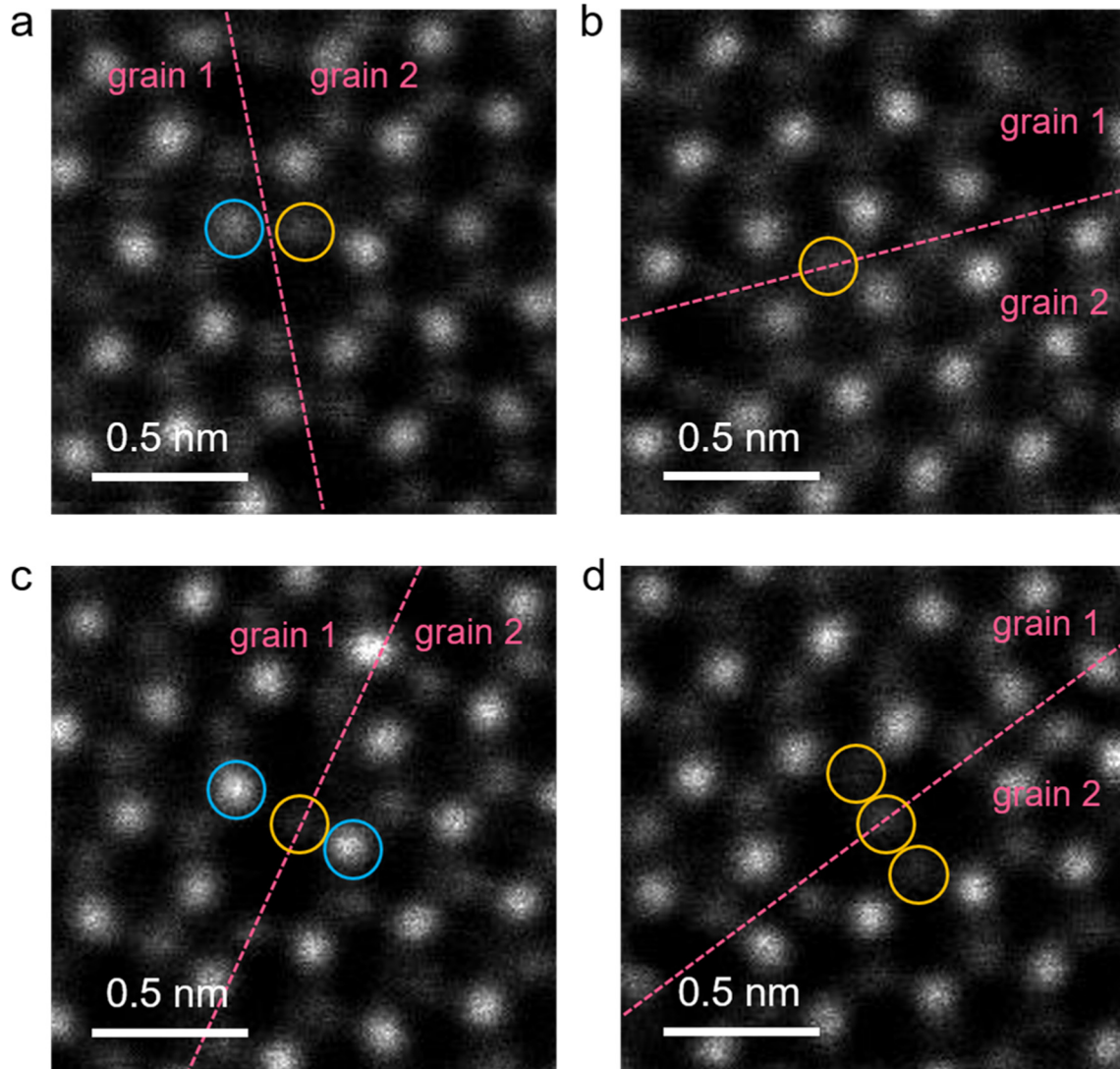
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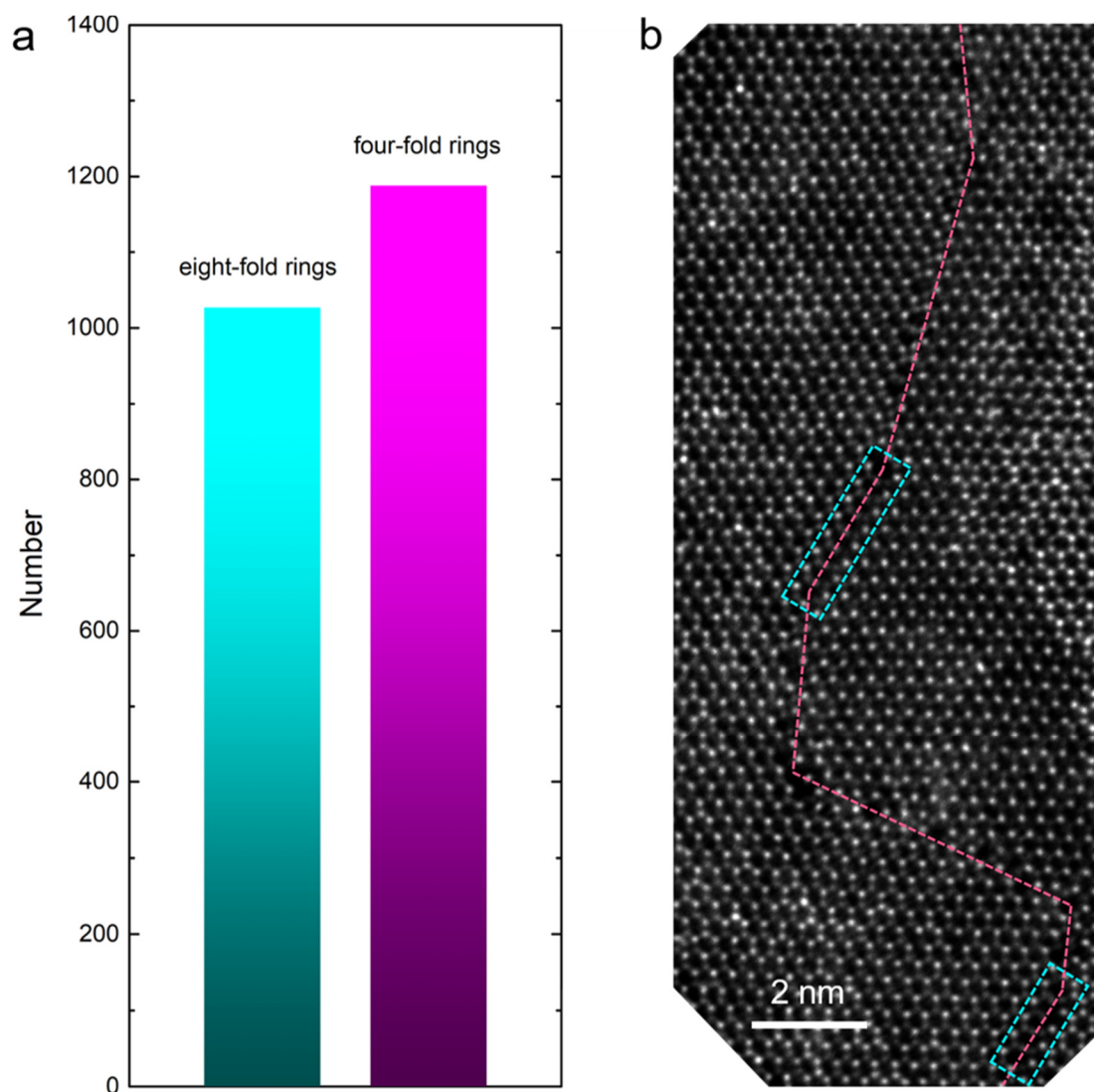
Supplementary Figures



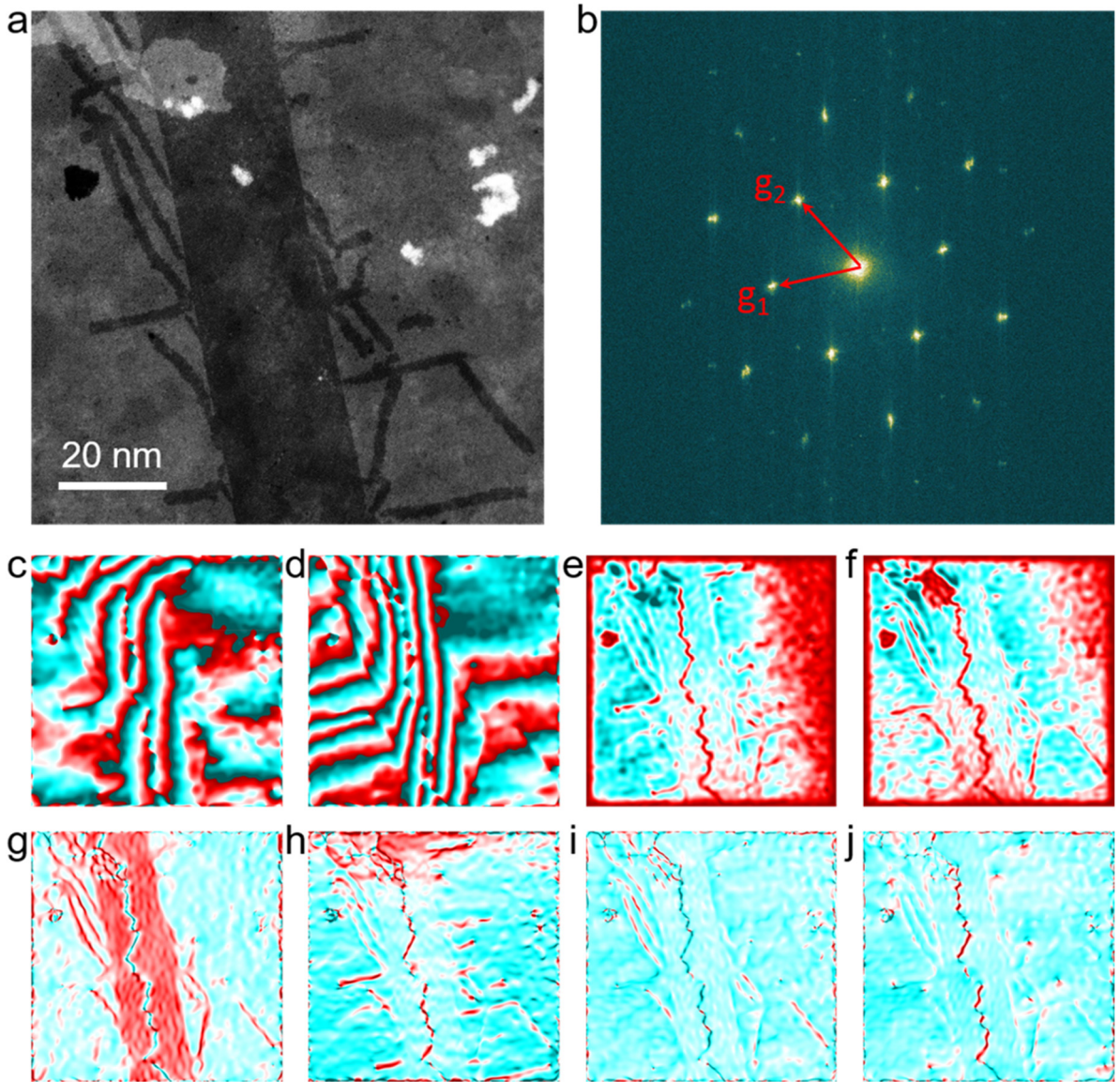
Supplementary Figure 1. Locations of MoS₂ nano-channels in MoSe₂ matrix. a, SEM image of a MoS₂-MoSe₂ heterostructure flake. The epitaxial MoS₂ has the same 6-point star shape (edges are labeled by the dashed blue lines) with the inner MoSe₂ monolayer. MoS₂ nano-channels, brighter contrast lines as indicated by the arrows, locate between neighbored corners and divide the MoSe₂ matrix into several parallelogram pieces. **b**, Large scale ADF-STEM image of the heterojunction area showing that the nano-channel origins exactly from the vertexes of obtuse angles (120°) between neighbored corners.



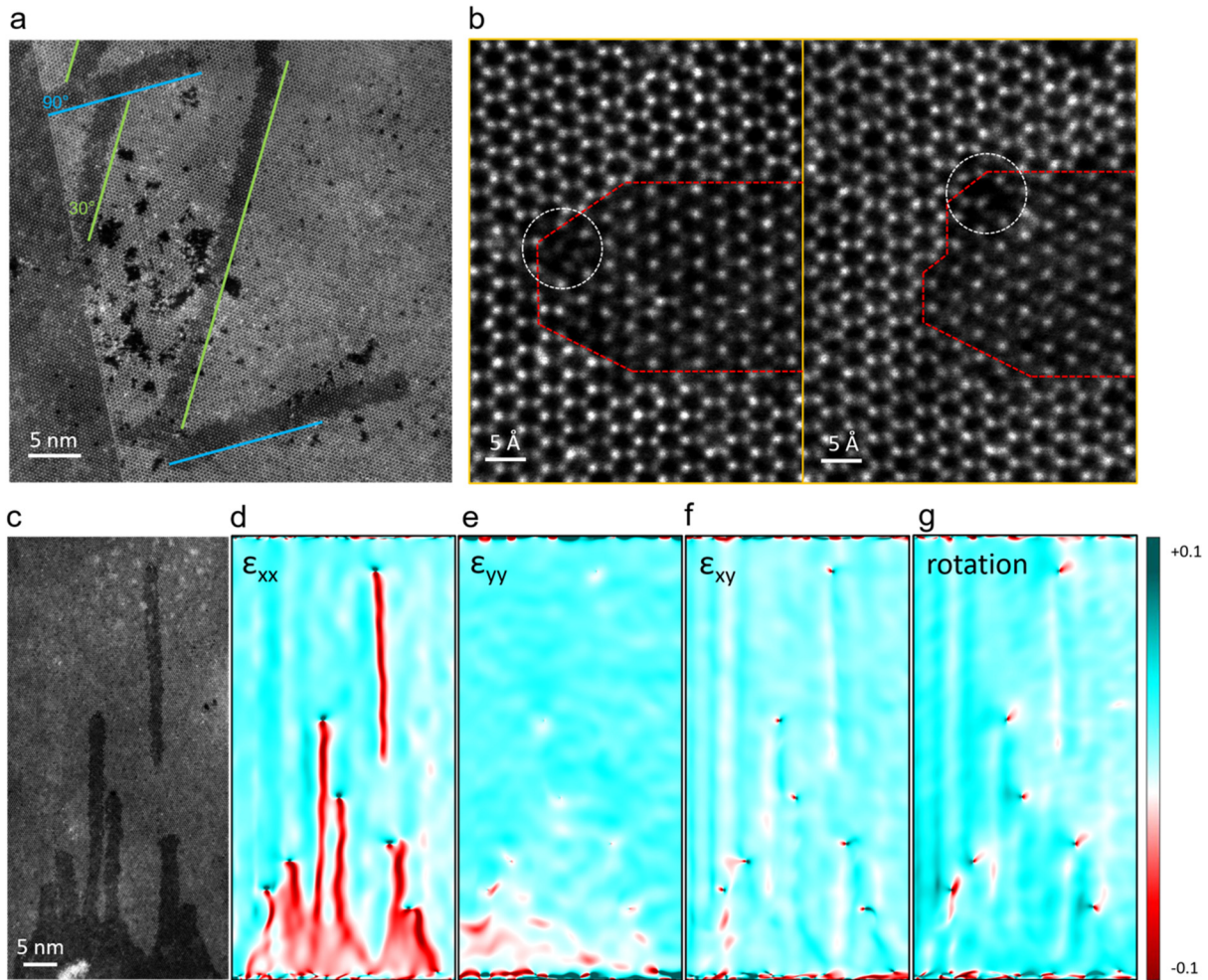
Supplementary Figure 2. Four joint types of 60° GBs. **a**, Type I: Mo atoms from one grain bond with S atoms from the other. In this type, Mo and S atoms at the GB maintain 6-fold and 3-fold coordination respectively, the same as those in defectless MoS₂ hexagonal lattice. **b**, Type II: two grains share the same S atoms with 4-fold coordination. **c**, Type III: two grains share the same 2-fold coordinative S atoms that bond with Mo atoms in each grain. **d**, Type IV: two grains share the same 2-fold coordinative S atoms that bond with S atoms in each grain. Here, blue and yellow circles represent Mo and S atoms. In our observation, 60° GBs are mainly composed of the former three joint ways, leading to the successive 4|8 rings. The density of type IV is very low, probably due to the energetically unfavorable arrangement of S atoms in this case.



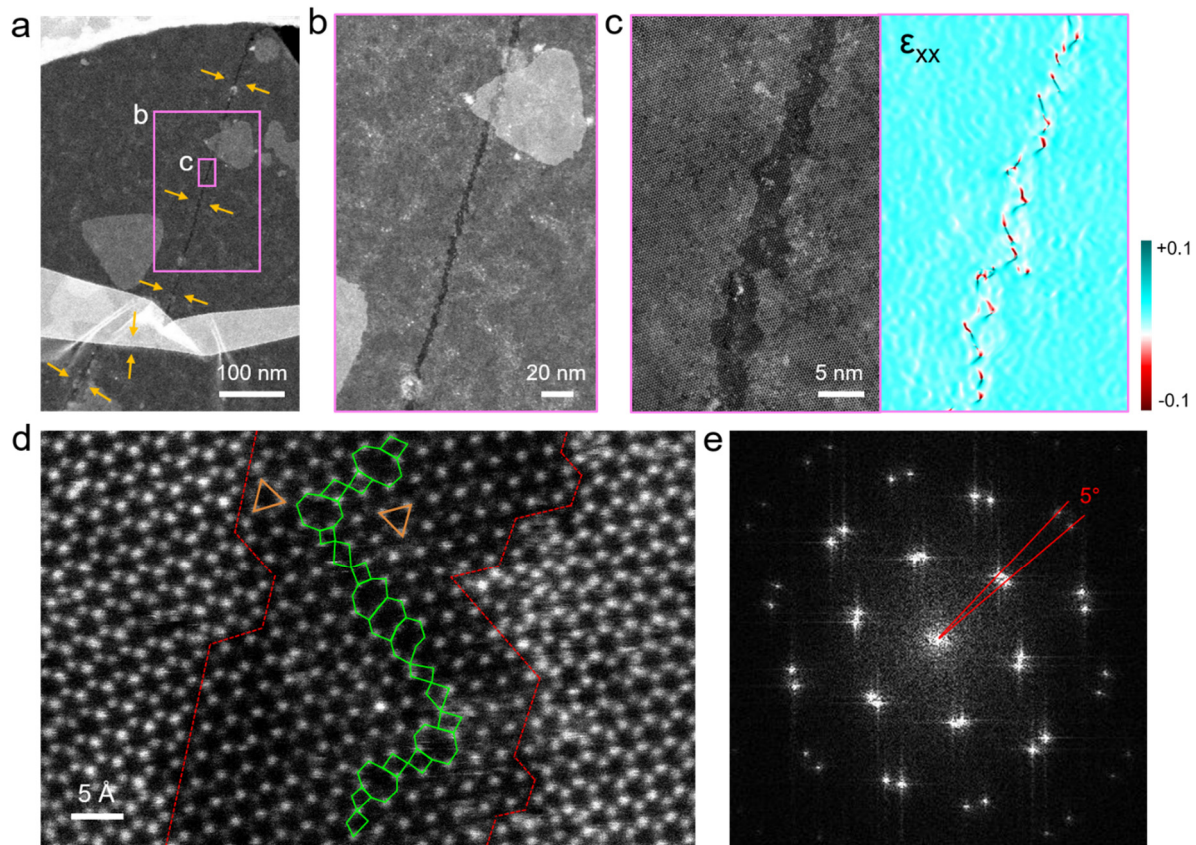
Supplementary Figure 3. Statistics of four- and eight-fold rings. **a**, Histogram of the number of four- and eight-fold rings along 1 μm -length 60° GBs, yielding a ratio of 1:1.16. **b**, ADF-STEM image shows the short chains of parallel eight-fold rings (highlighted by the rectangles).



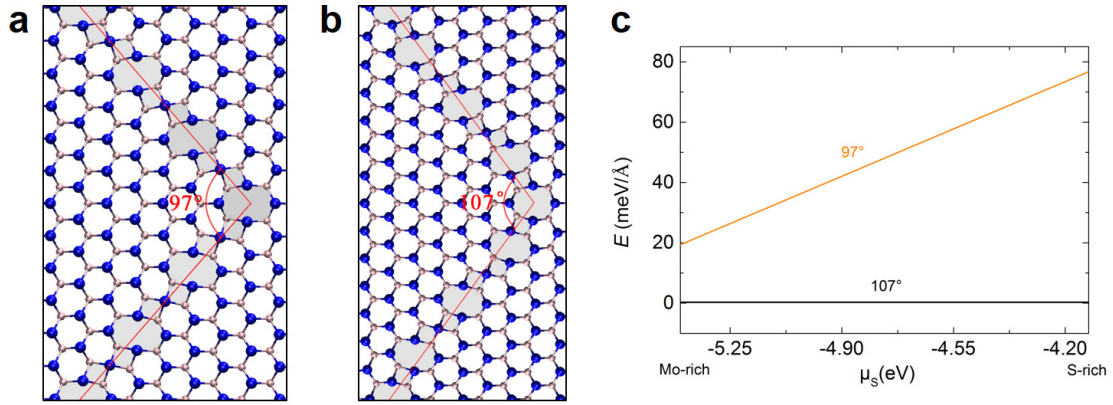
Supplementary Figure 4. Geometric phase analysis. **a**, Low-magnification ADF-STEM image of a MoS₂ channel embedded in MoSe₂ matrix. **b**, FFT image of **a**, where **g**₁ and **g**₂ are selected as the Bragg reflections for phase and strain calculation. **c** and **d**, Phase images corresponding to **g**₁ and **g**₂. **e** and **f**, Scale of reciprocal lattice **g**₁ and **g**₂. **g-i**, Strain maps (ϵ_{xx} , ϵ_{yy} , ϵ_{xy}) of **a**. **j**, Rotation map of **a**.



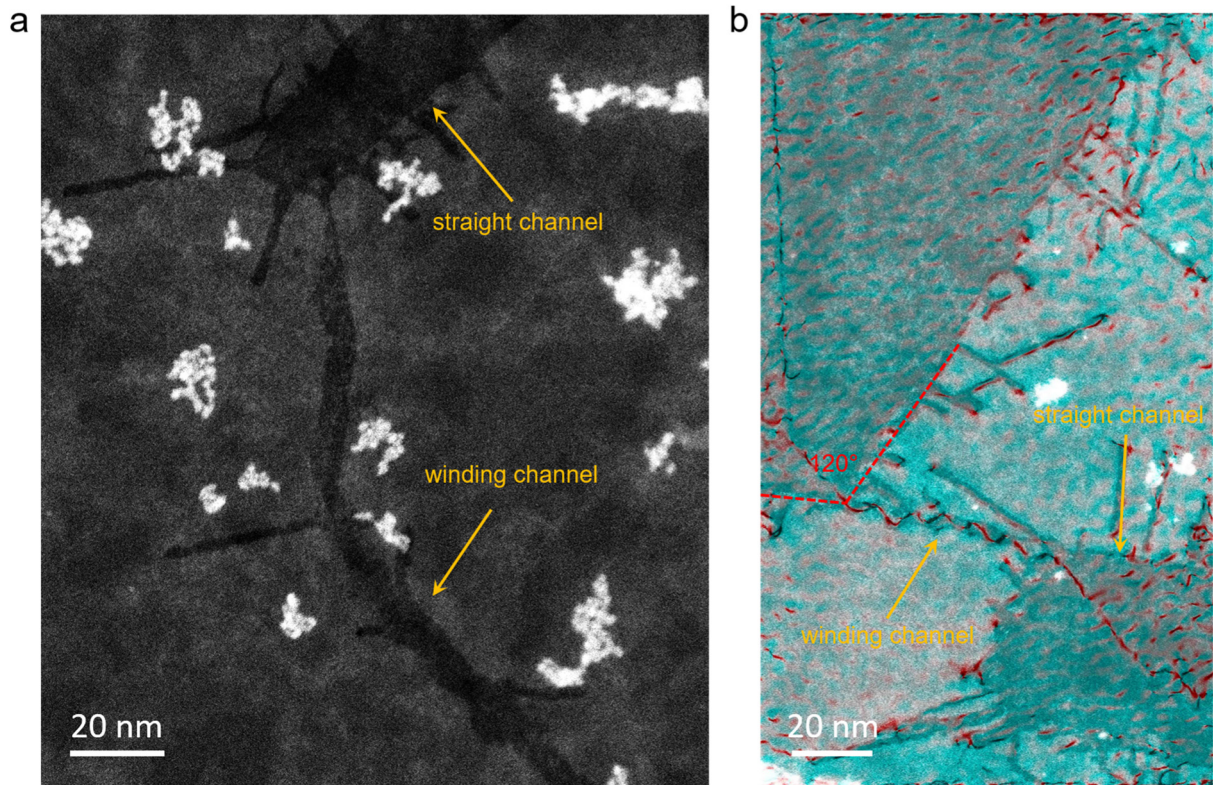
Supplementary Figure 5. Structure and GPA analysis of 2-nm-wide quantum wells. **a**, ADF-STEM image showing the growth direction of branched quantum wells: 30° (indicated by green lines) and 90° (indicated by blue lines) with respect to MoS₂ channels, consistent with our observation that straight channels grow along zigzag direction while quantum wells grow along the armchair direction of the crystal lattice. **b**, Atomic resolution ADF-STEM images showing the structure of quantum well tips. Point defects (dislocations) can be found at the tip of each quantum well (highlighted by the circles). In our observation, these defects could be missing atoms, replaced atoms or multi-fold rings such as 4- and 8- fold rings, not just 5|7 rings in previous literature. **c-g**, ADF-STEM image of quantum wells (**c**) and corresponding strain (**d-f**) and rotation maps (**g**). There are some similarities between quantum wells and nano-channels: they both have compressive ϵ_{xx} strain but uniform ϵ_{yy} strain compared with MoSe₂ matrix (see discussion in the main text). However, in shear strain and rotation maps, the point defect nature of quantum wells can be clearly distinguished from the line defect nature of nano-channels.



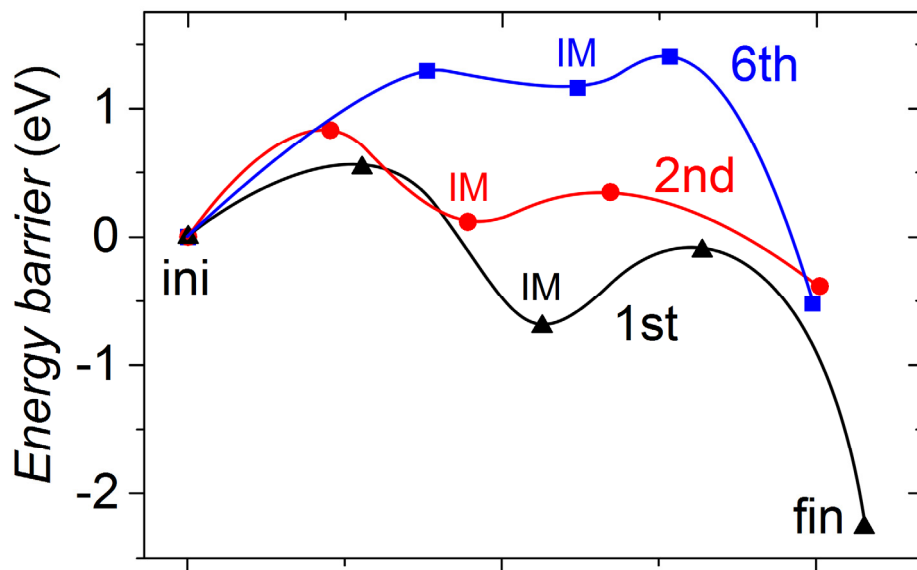
Supplementary Figure 6. MoS₂ nano-channel without quantum wells obtained at high temperature. **a**, Large field of view ADF-STEM image of a nano-channel without quantum wells. **b**, Enlarged ADF-STEM image of large rectangular region in **a**. **c**, Enlarged ADF-STEM image and corresponding strain map of the small rectangular region in **a**. **d**, High-resolution ADF-STEM image of this nano-channel showing its atomic structure. **e**, Fast Fourier transformation of **c**.



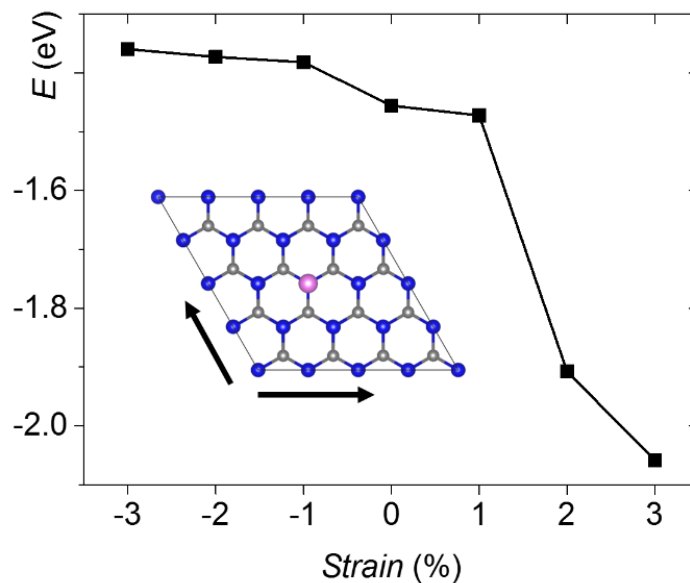
Supplementary Figure 7. Atomic structures of folded GBs. **a**, The one composed of solely of the $8|4|4|8$ dislocations. **b**, The GB composed of mixed $8|4 + 8|4|4|8$ dislocations. **c**, Energies of 97° GB relative to 107° GB as functions of the chemical potential of sulfur in the range $-5.4 \text{ eV} < \mu_s < -4.1 \text{ eV}$.



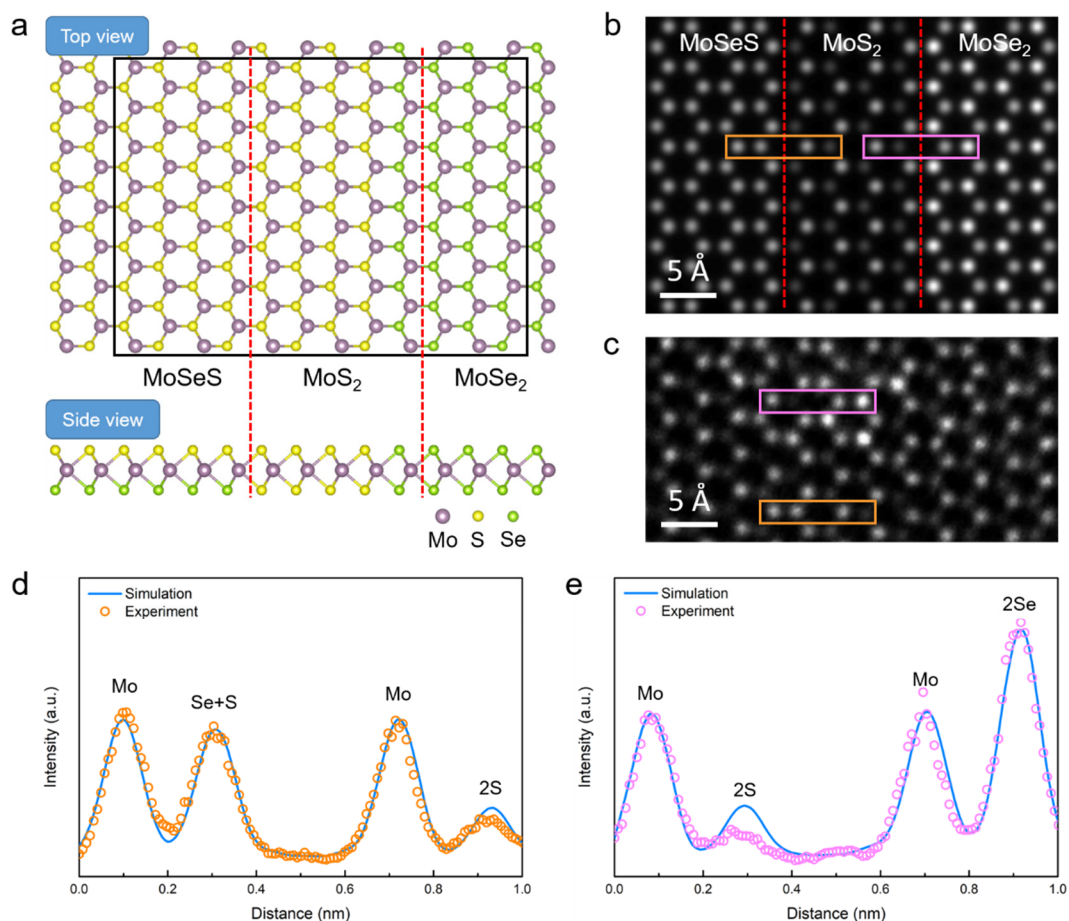
Supplementary Figure 8. Connected straight and winding channels. **a**, ADF-STEM image presenting the connected straight channels (20 nm in width) and winding channels (2-10 nm in width). **b**, Superimposed image of the connected channels and the corresponding rotation map. The inner zigzag 60° GB seamlessly extends from the winding channel to straight one, suggesting the consistency of them.



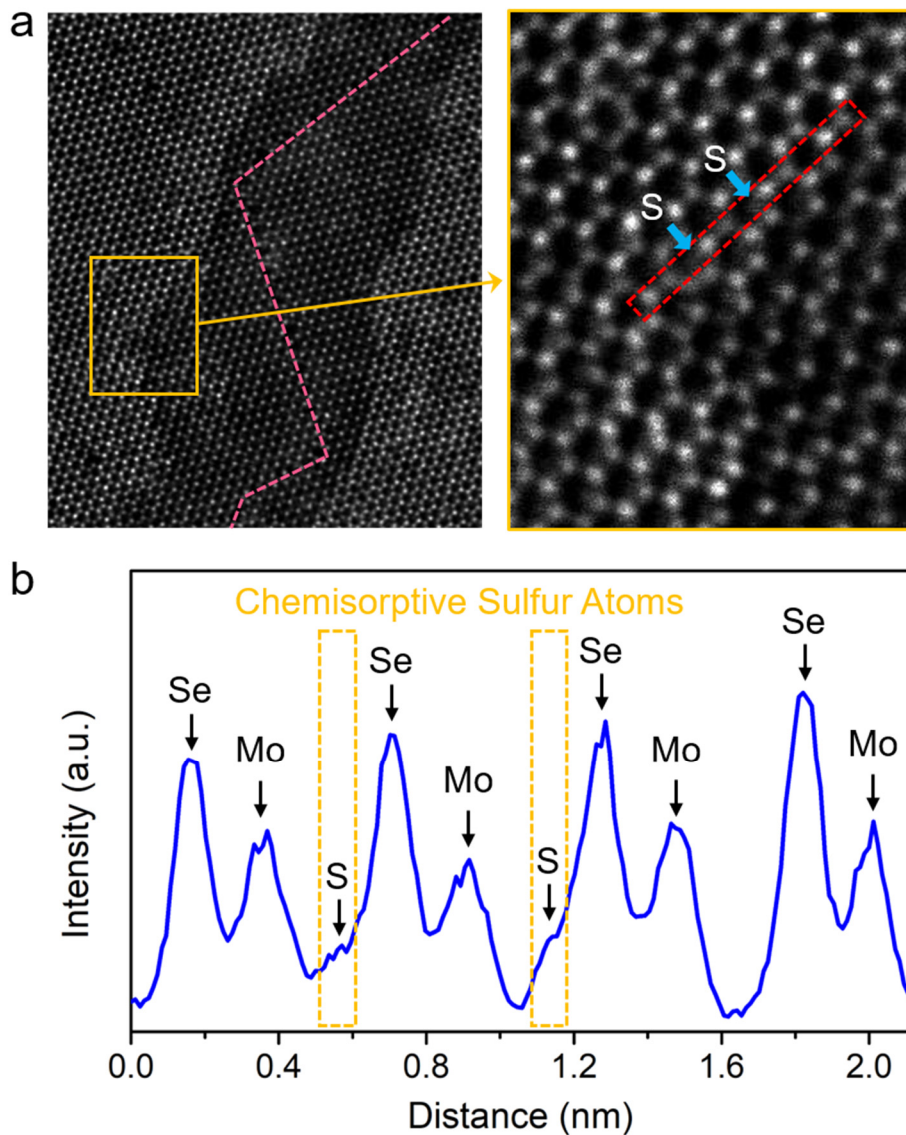
Supplementary Figure 9. Calculated minimum energy pathways energy for taking the 1st, 2nd and 6th Se atom as the substitution sites for the first S atom (see Fig. 5b). The intermediate state is marked by IM in each energy pathway.



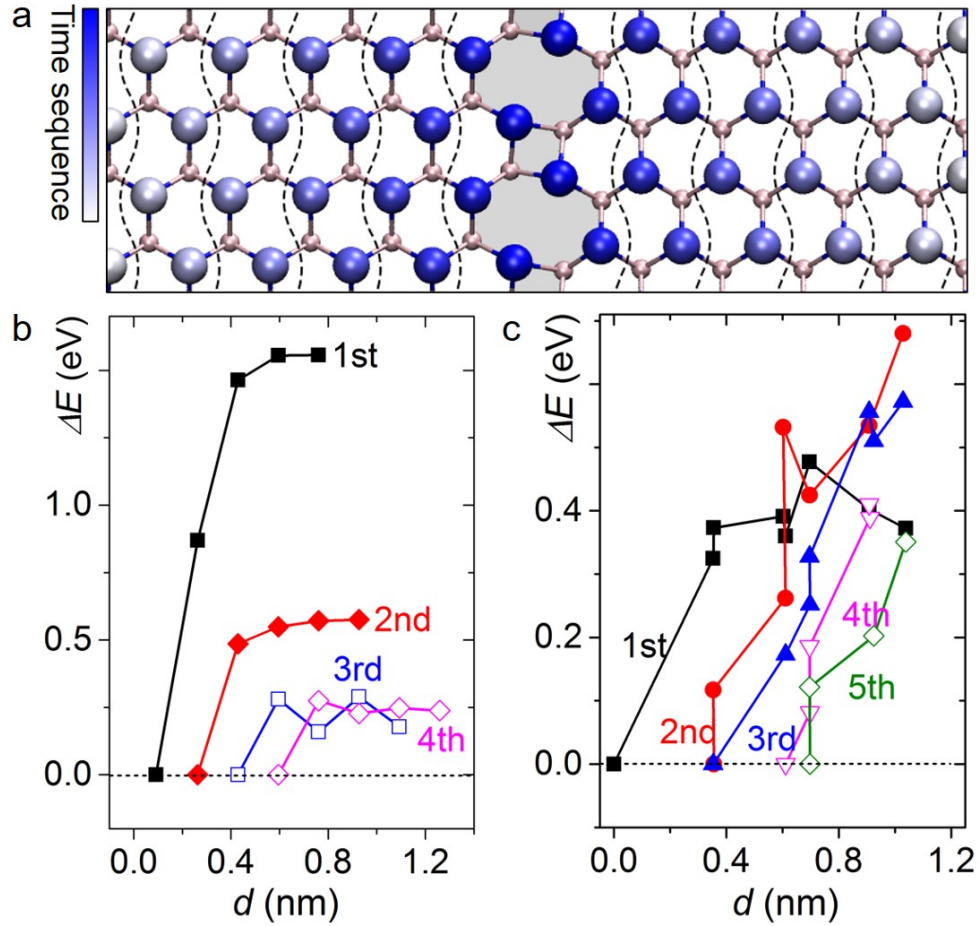
Supplementary Figure 10. Evolution of absorption energy of H atoms on a MoSe₂ sheet against applied lattice strain. Here we use the absorption energy of H atoms to reflect the correlation between the chemical reactivity and strain while excluding the effect of defects. We calculate the variation of absorption energy of H atoms on a perfect MoSe₂ sheet. The absorption energy of H atoms drops by ~0.6 eV when a moderate 3% tensile strain is applied. In comparison, the lattice strain near the GB can be up to 4% and therefore is sufficient to regulate the chemical activity thereof. We thus conclude that the chemical reactivity around the GBs is dominated by the strain. The intrinsic chemical activity of 4|8 dislocation plays a secondary role. The S substitution reaction is mainly attributed to the strain-driven mechanism as the reaction sequences also perfectly imprinted the strain patterns derived from the GBs. The inset presents the atomic structure (blue, grey and pink circles: Se, Mo and H atoms). The black arrow denotes the direction of strain.



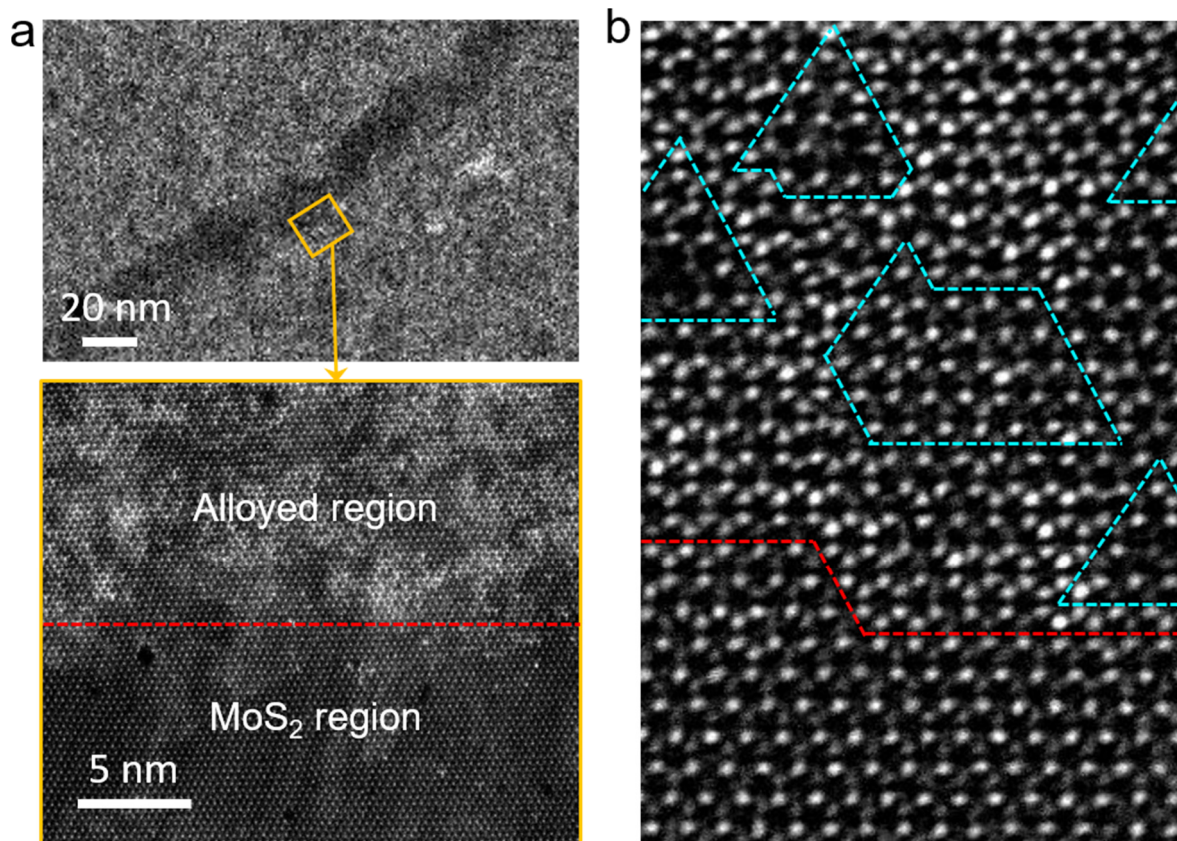
Supplementary Figure 11. Determination of SeS, SS and SeSe atom column through ADF-STEM images. **a**, Atomic model of lateral hetero-structure of monolayer MoSeS-MoS₂-MoSe₂. **b**, Simulated ADF-STEM image corresponding to the rectangular region of **a**. **c**, Experimental ADF-STEM image of some doped region in MoS₂ channel. **d**, Line intensity profile of simulated (orange rectangle in **b**) and experimental (orange rectangle in **c**) atom chains of Mo-SeS-Mo-2S. **e**, Line intensity profile of simulated (pink rectangle in **b**) and experimental (pink rectangle in **c**) atom chains of Mo-2S-Mo-2Se.



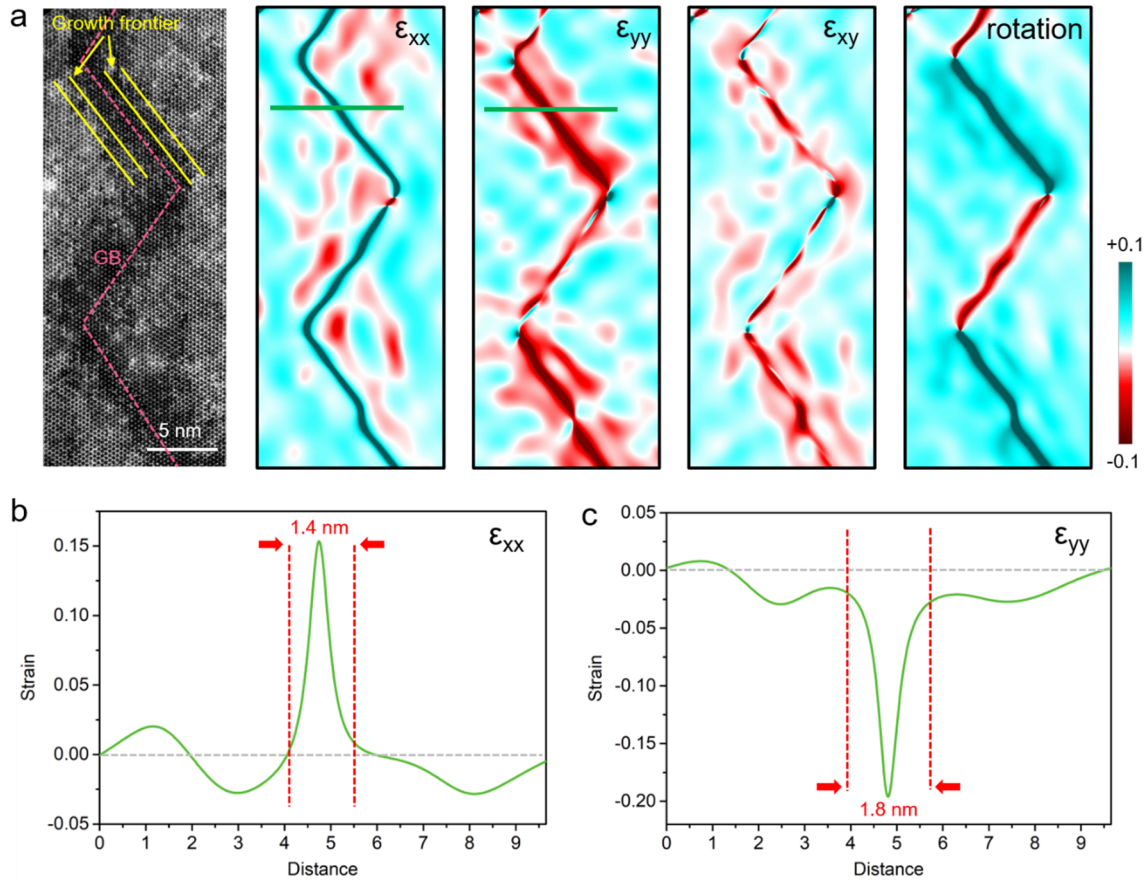
Supplementary Figure 12. Chemisorptive sulfur atoms at the interface. **a**, ADF-STEM image of the MoS₂ channel and the enlarged image of the MoSe₂-MoS₂ interface. **b**, Line intensity profile of the rectangular region in **a**, from which some chemisorptive sulfur atoms can be found at the interface, indicating the consistency of theory and experiments.



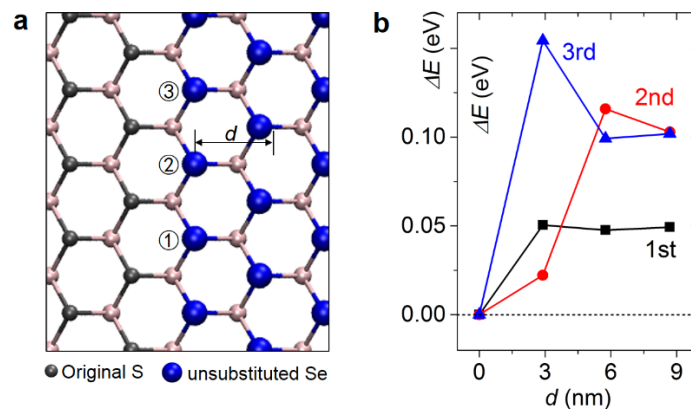
Supplementary Figure 13. **a**, Contour plot of substitution sequence (from blue to light grey) around the 8|4|8|4 GB. **b** and **c**, Calculated relative energies ΔE for the intermediate state as a function of d (the distance from the 8|4|8|4 GB or 5|7 dislocation, respectively) along the optimal substitution pathway.



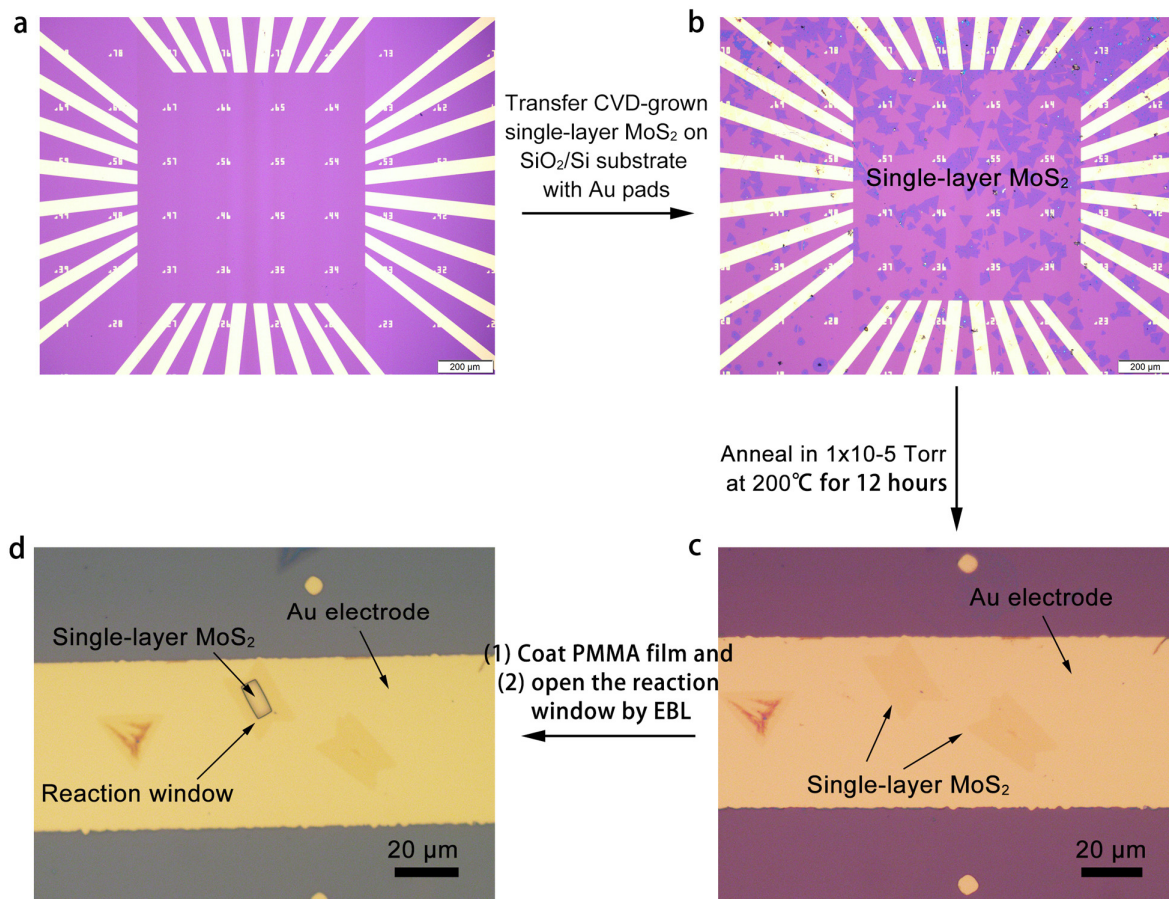
Supplementary Figure 14. Strain effect during substitution growth. In order to prove the strain-driven growth mechanism, we create some point defects (Mo and Se vacancies) with low strain through 5w Ar plasma treatment for 10 s. After that, MoS₂ growth was carried out. **a**, ADF-STEM images showing that the GBs based MoS₂ channel still can be obtained, whereas the original MoSe₂ region has become MoSe₂-MoS₂ alloy. **b**, Atomic-resolution ADF-STEM image of the interface area. The alloyed region actually consists of many MoS₂ islands (highlighted by the cyan dashed lines) with the size smaller than 2 nm, which are generated from the introduced point defects. In comparison, the channel can grow to 20 nm wide under the same condition, implying that the substitution growth occurs more easily around GBs with high strain than around vacancies with low strain.



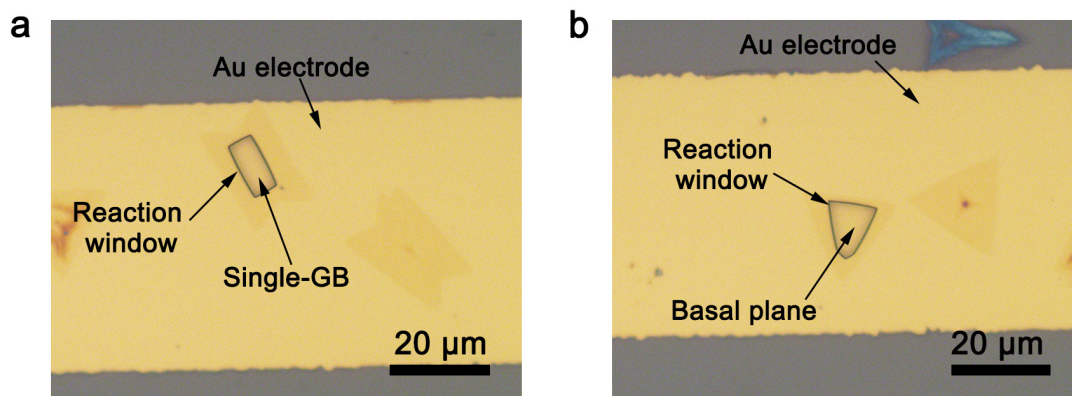
Supplementary Figure 15. Strain analysis of the growth intermediate state of MoS₂ nano-channels. **a**, ADF-STEM image of a partly-formed MoS₂ nano-channel embedded in MoSe₂, and the corresponding strain maps. The GB is noted with dashed lines and the growth frontier is highlighted with yellow solid lines. **b**, The line profile from the green line region in ϵ_{xx} strain map. **c**, The line profile from the green line region in ϵ_{yy} strain map. It can be confirmed from this figure that the driving force diminishes as the growth frontier moves away from the GB, and more specifically the driving force can be completely ignored when the growth frontier is 1 nm away from the GB.



Supplementary Figure 16. **a**, Atomic structure of the perfect MoS₂-MoSe₂ interface. The numbers 1~3 denote the sequence of S substitution. d denotes the distance of reaction sites with respect to the interface line. **b**, Calculated relative energies ΔE for the intermediate state in **a** as a function of d for the 1~3 S atoms along the optimal substitution pathway.

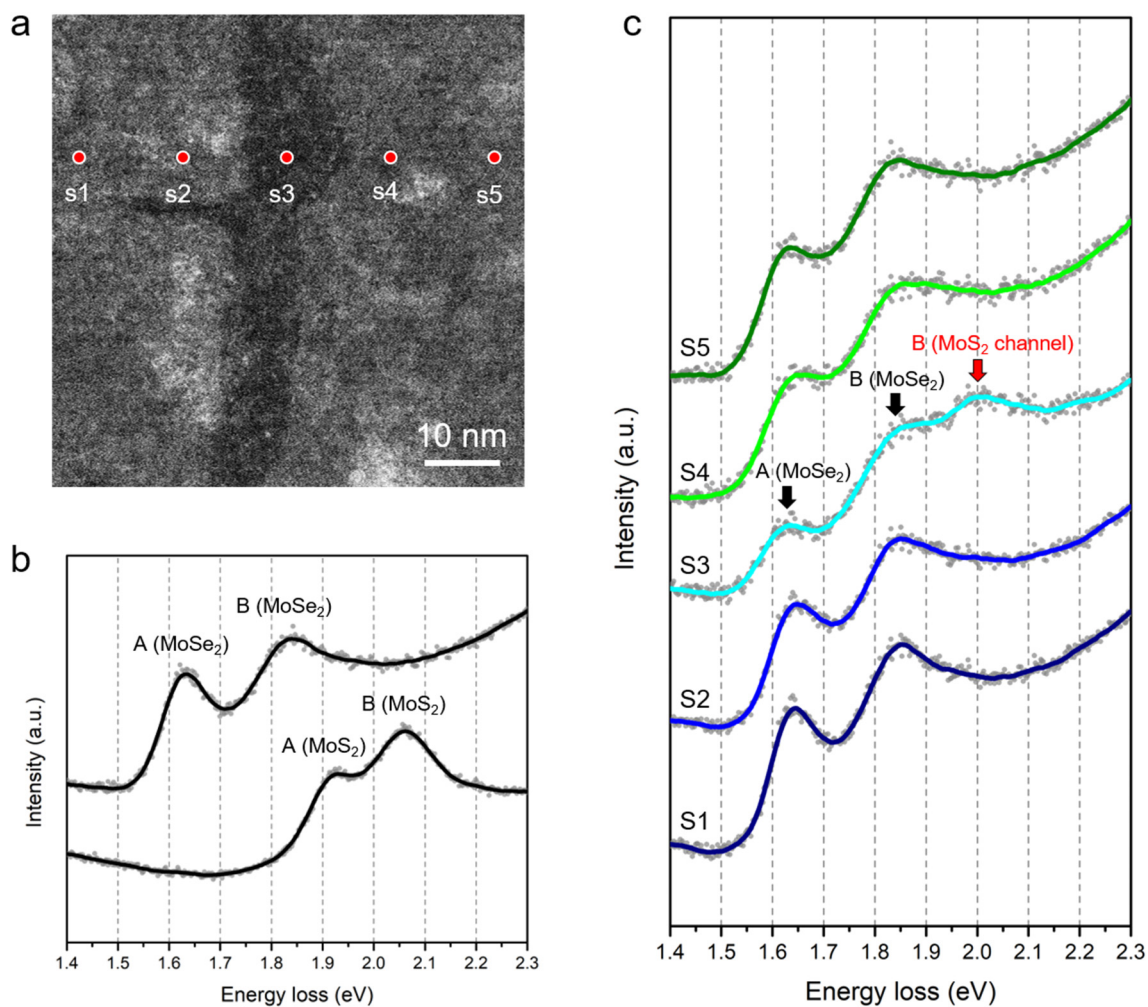


Supplementary Figure 17. Fabrication procedure for the single-layer MoS₂ microelectrode on Au substrate. **a**, SiO₂/Si substrates coated with Au pads. **b**, MoS₂/MoS₂-MoSe₂ film was transferred onto the substrate. **c**, 1-μm-thick PMMA was coated on the chip. **d**, A reaction window was opened to expose the region of interest on the film by using e-beam lithography.



Supplementary Figure 18. Optical images of pure MoS₂ micro-electrochemical devices.

The reaction windows are opened on single-GB (a) and basal plane (b) for HER measurement.



Supplementary Figure 19. Excitonic absorption spectroscopy of MoS₂ nano-channels. **a**, ADF-STEM image of a MoS₂ nano-channel. **b**, The low-loss electron energy loss spectroscopy (LL-EELS) spectra of pure MoSe₂ and MoS₂. **c**, Five LL-EELS spectra integrated at different positions highlighted in **a**. In this figure, the B exciton at 1.99 eV can be obtained of MoS₂ nano-channel (highlighted in **c**), which has 70 meV redshift compared with that of pure MoS₂ (**b**), indicating the strain-induced bandgap engineering within 10 nm scale.

Supplementary Tables

| Parameter | Symbol | Value |
|------------------------------|------------------|----------|
| GB energy (C_6 term) | σ_1 | 1 eV/Å |
| GB energy (C_3 term) | σ_2 | 0.5 eV/Å |
| Characteristic length | 1 | 50 nm |
| Interface thickness | η | 71 |
| Coupling coefficient | λ | 20 |
| Desorption time | $\tilde{\tau}_s$ | 2.5 |
| Diffusivity | \tilde{D} | 150 |
| Mobility | μ_0 | 2 |
| Interface tension anisotropy | δ | 0.1 |
| Equilibrated ζ | ξ_{eq2} | 0.1 |

Supplementary Table 1. List of parameters for the phase-field simulation. Tildes are omitted for dimensionless parameters.