Supporting Information

Conformational Heterogeneity and Self-Assembly of α,β,γ-Hybrid Peptides Containing Fenamic Acid: Multi-Stimuli Responsive Phase Selective Gelation

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(A) For Peptide 1 and Peptide 2



Figure S1 (A): Synthetic route of Peptide **1** and Peptide **2.** (a)Dry DCM,H-Aib-OMe, DCC, 48h, rt; (b) 2(N) NaOH, MeOH, 12h, rt; (c) Dry DCM, DCC, HOBt, H-Phe-OMe, 48h, rt (for peptide 1), H-Tyr-OMe (for peptide 2). **(B):** Synthetic route of Peptide **3** (B). (a)Dry DCM,H-Aib-OMe, DCC, 48h, rt; (b) 2(N) NaOH, MeOH, 12h, rt; (c) Dry DCM, DCC, 48h, rt, Maba-OMe (for peptide 3).



Figure S2 : (A) Concentration dependent UV-Vis spectra and (B) Concentration dependent Fluorescence spectra [excitation at 346 nm] in MeOH of Peptide 1.



Figure S3 : (A) Concentration dependent UV-Vis spectra and (B) Concentration dependent Fluorescence spectra [excitation at 338 nm] in MeOH of Peptide **2**.



Figure S4 : (A) Concentration dependent UV-Vis spectra and (B) Concentration dependent Fluorescence spectra [excitation at 342 nm] in MeOH of Peptide **3**.



Figure S5: POM image of peptide **1** (a) without polarizer and (b) with polarizer; POM image of peptide **2** (c) without polarizer and (d) with polarizer, POM image of peptide **3** (e) without polarizer and (f).with polarizer.



Figure S6: FT-IR Spectra of Peptide 3 and its xerogel.



Figure S7: The plausible effect of H_2SO_4 on peptide **3** gel to sol transition and phase transfer.

 Table S1. Crystal data and structure refinement for Peptide 1.

Identification code	NPAP
Empirical formula	$C_{27} H_{29} N_3 O_4$
Formula weight	459.53
Temperature/K	298
Crystal system	monoclinic
Space group	P 1 21 1
a/Å	6.9113
b/Å	18.1972
c/Å	10.1370
α/°	90

β/°	105.648
γ/°	90
Volume/Å ³	1227.64
Ζ	2
$\rho_{calc}g/cm^3$	1.243
μ/mm^{-1}	0.084
F(000)	488.0
Crystal size/mm ³	0.2458 imes 0.2358 imes 0.1587
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/°	4.172 to 50.044
Index ranges	$-8 \le h \le 8, -21 \le k \le 21, -12 \le l \le 12$
Reflections collected	10027
Independent reflections	4315 [$R_{int} = 0.0331$, $R_{sigma} = 0.0427$]
Data/restraints/parameters	4315/1/310
Goodness-of-fit on F ²	1.088
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0470, wR_2 = 0.1506$
Final R indexes [all data]	$R_1 = 0.0534, wR_2 = 0.1574$
Largest diff. peak/hole / e Å ⁻³	0.21/-0.17
Flack parameter	0.25(3)

 Table S2. Crystal data and structure refinement for Peptide 2.

Identification code	NPAT
Empirical formula	$C_{27} H_{29} N_3 O_5$
Formula weight	475.53
Temperature/K	100.01
Crystal system	monoclinic
Space group	P 1 21 1
a/Å	6.9049
b/Å	17.9673
c/Å	10.1305
$\alpha/^{\circ}$	90
β/°	106.637
γ/°	90
Volume/Å ³	1204.20
Ζ	2
$\rho_{calc}g/cm^3$	1.311
µ/mm ⁻¹	0.091
F(000)	506.0
Crystal size/mm ³	$0.248 \times 0.187 \times 0.107$
Radiation	MoKα (λ = 0.71073)
20 range for data collection/°	4.196 to 50.04

Index ranges	$-8 \le h \le 7, -21 \le k \le 10, -4 \le l \le 12$
Reflections collected	3435
Independent reflections	2739 [$R_{int} = 0.0337$, $R_{sigma} = 0.0502$]
Data/restraints/parameters	2739/1/320
Goodness-of-fit on F ²	0.971
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0423, wR_2 = 0.1229$
Final R indexes [all data]	$R_1 = 0.0455, wR_2 = 0.1275$
Largest diff. peak/hole / e Å ⁻³	0.49/-0.37
Flack parameter	-1.0(10)

Table S3. Crystal data and structure refinement for Peptide 3.

Identification code	NPAM
Empirical formula	$C_{25} H_{25} N_3 O_4$
Formula weight	431.48
Temperature/K	273
Crystal system	monoclinic
Space group	P 1 21/c 1
a/Å	13.2844
b/Å	9.4272
c/Å	36.075
$\alpha/^{\circ}$	90
β/°	93.302
γ/°	90
Volume/Å ³	4510.4
Ζ	8
$\rho_{calc}g/cm^3$	1.271
μ/mm^{-1}	0.709
F(000)	1824
Crystal size/mm ³	-
Radiation	$CuKa (\lambda = 1.54178)$
Theta Min-Max /°	2.5, 68.0
Dataset	-15: 15 ; -10: 11 ; -43: 43
Tot., Uniq. Data, R(int)	49064, 8166, 0.153
Observed data $[I > 2.0 \text{ sigma}(I)]$	5419
Nref, Npar	8166, 584
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.1315, wR_2 = 0.3302$
Final R indexes [all data]	$R_1 = 0.1735, wR_2 = 0.3429$
Min. and Max. Resd. Dens. [e/Ang^3]	-0.64, 0.83



Figure S8. ¹H NMR (400 MHz, CDCl₃) spectrum of Fenamic acid- Aib OMe (NPA-Aib-OMe) 4.



Figure S9. ¹³C NMR (100 MHz, CDCl₃) spectrum of Fenamic acid- Aib OMe (NPA-Aib-OMe) 4.



Figure S10. Mass spectrum of Fenamic acid- Aib OMe (NPA-Aib-OMe) 4.



Figure S11. FT-IR spectrum of Fenamic acid- Aib OMe (NPA-Aib-OMe) 4.



Figure S12. ¹H NMR (500 MHz, DMSO-d₆) spectrum of Fenamic acid- Aib OH (NPA-Aib-OH) 5.



Figure S13. ¹³C NMR (125 MHz, DMSO-d₆) spectrum of Fenamic acid- Aib OH (NPA-Aib-OH) 5.



Figure S14. Mass spectrum of Fenamic acid- Aib OH (NPA-Aib-OH) 5.



Figure S15. FT-IR spectrum of Fenamic acid- Aib OH (NPA-Aib-OH) 5.



Figure S16. ¹H NMR (400 MHz, CDCl₃) spectrum of Fenamic acid- Aib-Phe OMe 1.



Figure S17. ¹³C NMR (100 MHz, CDCl₃) spectrum of Fenamic acid- Aib-Phe OMe 1.



Figure S18. Mass spectrum of Fenamic acid- Aib-Phe OMe 1.



Figure S19. ¹H NMR (400 MHz, DMSP-d₆) spectrum of Fenamic acid- Aib-Tyr OMe 2.



Figure S20. ¹³C NMR (100 MHz, DMSO-d₆) spectrum of Fenamic acid- Aib-Tyr OMe 2.



Figure S21. Mass spectrum of Fenamic acid- Aib-Tyr OMe 2.



Figure S22. ¹H NMR (400 MHz, CDCl₃) spectrum of Fenamic acid- Aib-Maba OMe 3.



Figure S23. ¹³C NMR (100 MHz, CDCl₃) spectrum of Fenamic acid- Aib-Maba OMe 3.



Figure S24. Mass spectrum of Fenamic acid- Aib-Maba OMe 3.