

Article

# Evaluation of the Environmental Impact and Efficiency of N-Doping Strategies in the Synthesis of Carbon Dots

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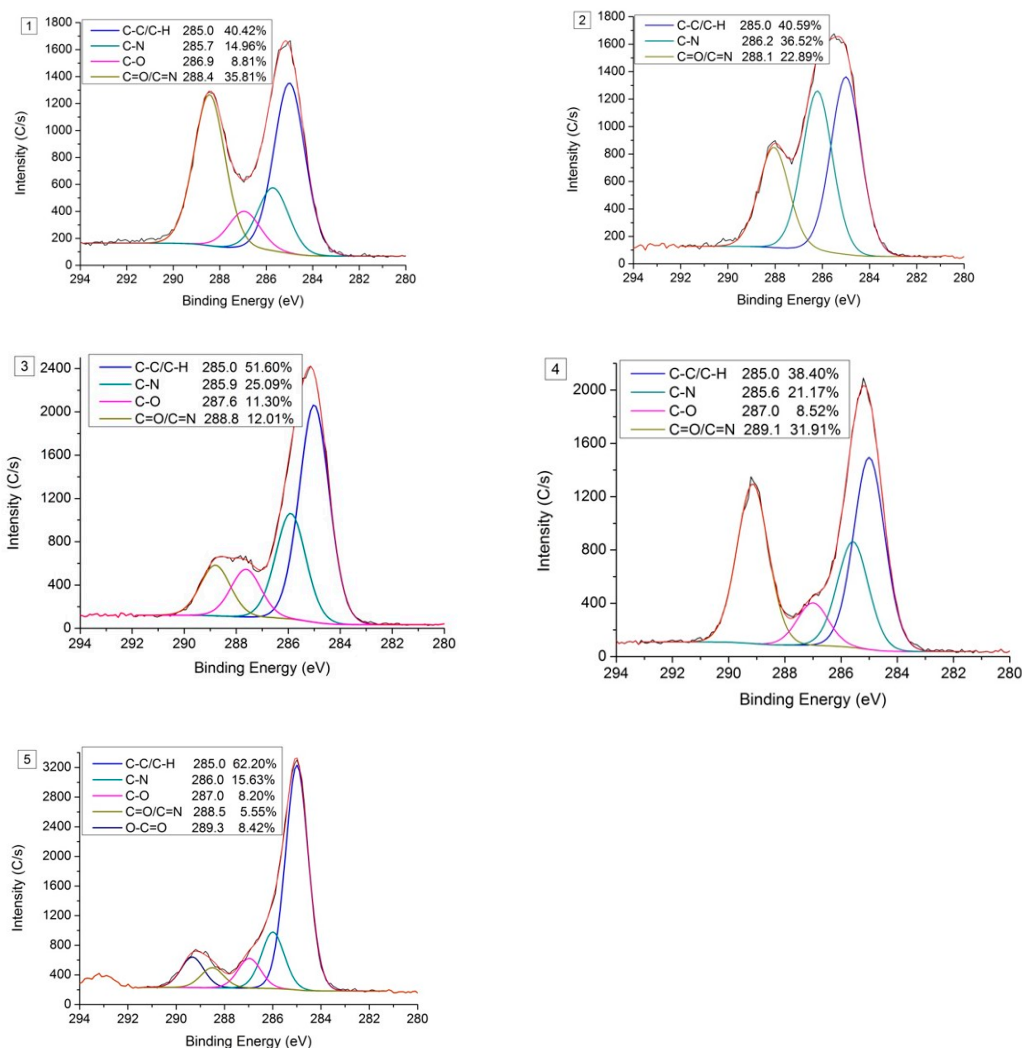


Figure S1. Detailed XPS spectra of C 1s (Carbon dots 1 to 5).

Table S1. Contributions (%) to C 1s peaks after deconvolution (CasaXPS software).

C 1s bond (BE)	1	2	3	4	5
C-C/C-H (~285 eV)	40	41	52	38	62

C-N (~286 eV)	15	37	25	21	16
C-O (~287 eV)	9	-	11	9	8
C=O/C=N (~288 eV)	36	23	12	32	6
O-C=O (~289 eV)	-	-	-	-	8

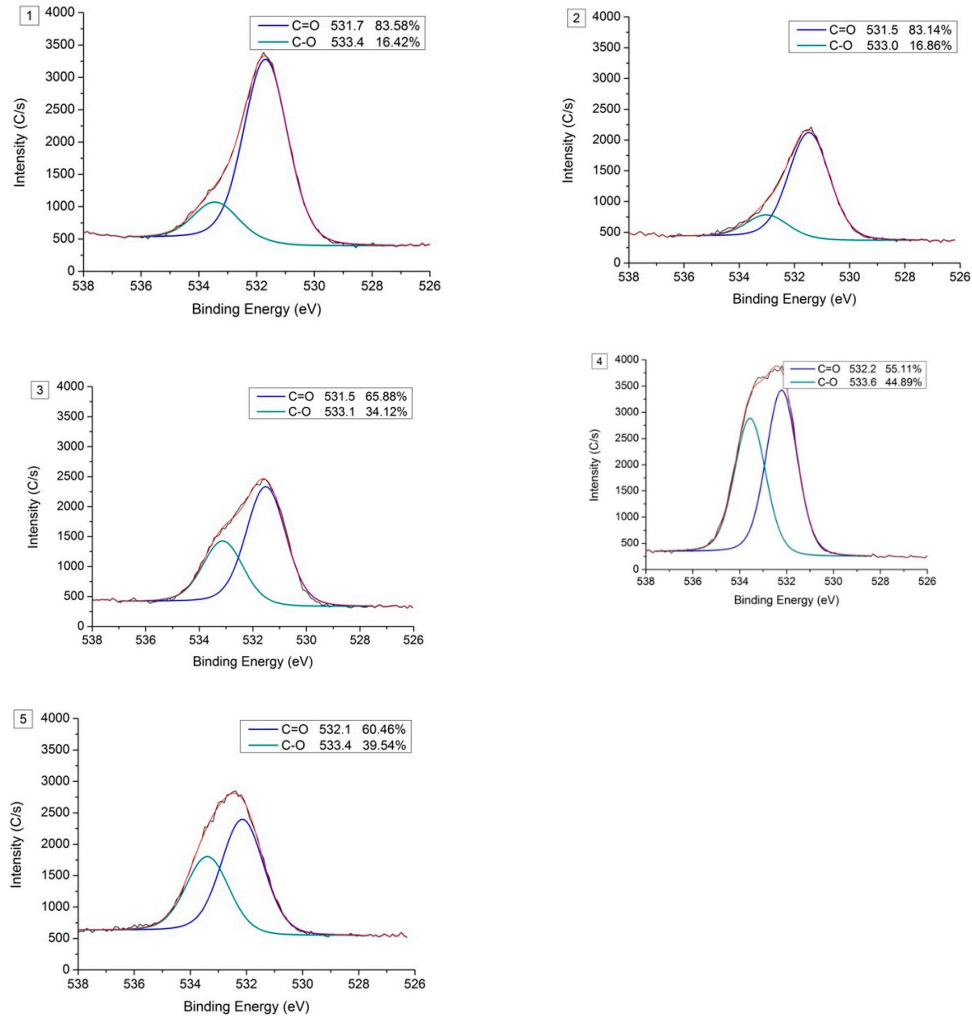
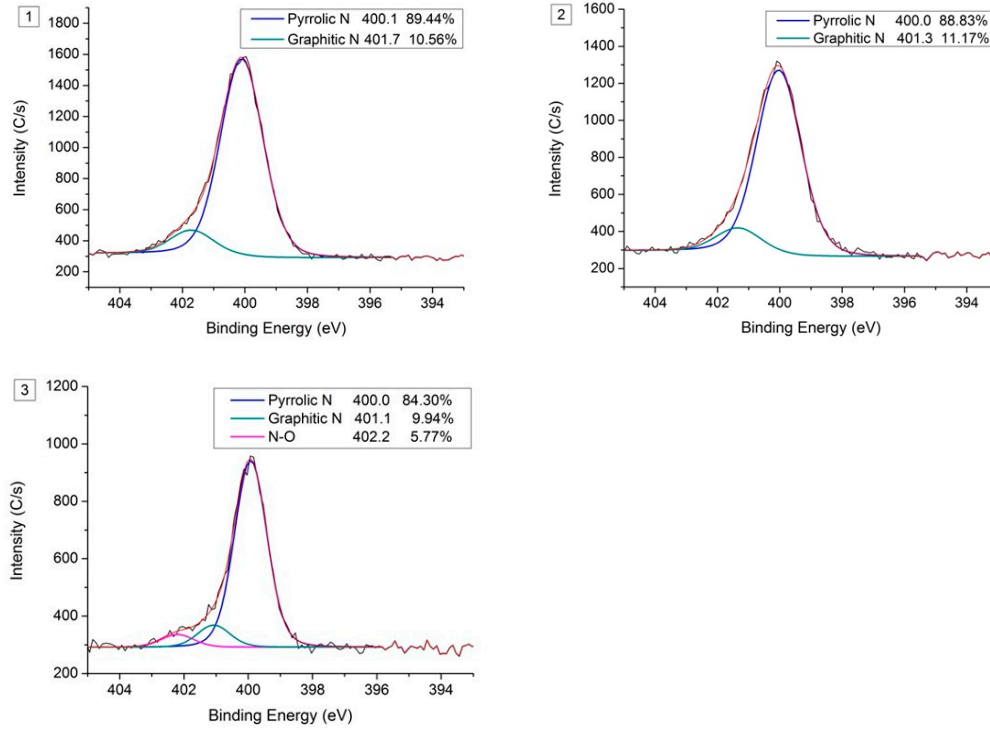


Figure S2. Detailed XPS spectra of O 1s (Carbon dots 1 to 5).

Table S2. Contributions (%) to O 1s peaks after deconvolution (CasaXPS software).

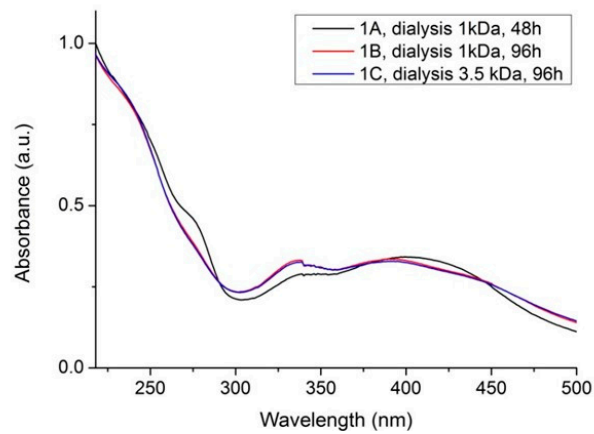
O 1s bond (BE)	1	2	3	4	5
C=O (~532 eV)	84	83	66	55	60
C-O (~533 eV)	16	17	34	45	40



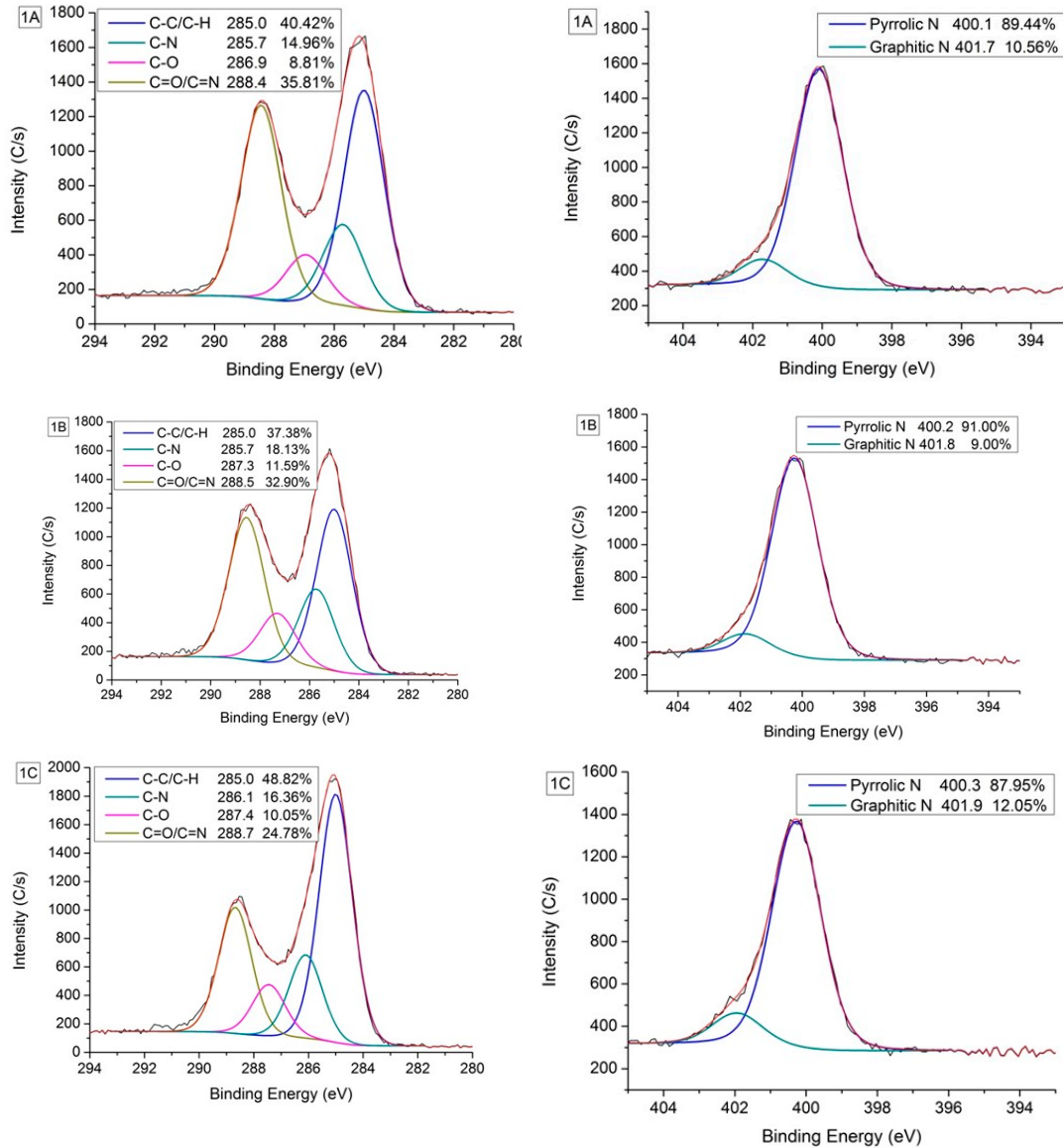
**Figure S3.** Detailed XPS spectra of N 1s (Carbon dots 1 to 3).

**Table S3.** Contributions (%) to N 1s peaks after deconvolution (CasaXPS software).

N 1s bond (BE)	1	2	3
Pyrrolic N (~400 eV)	89	89	84
Graphitic N (~401.5 eV)	11	11	10
N-O (~402 eV)	-	-	6



**Figure S4.** Absorbance spectra of carbon dots 1A, 1B and 1C.



**Figure S5.** XPS spectra C 1s (left) and N 1s (right) for carbon dots 1A, 1B and 1C.

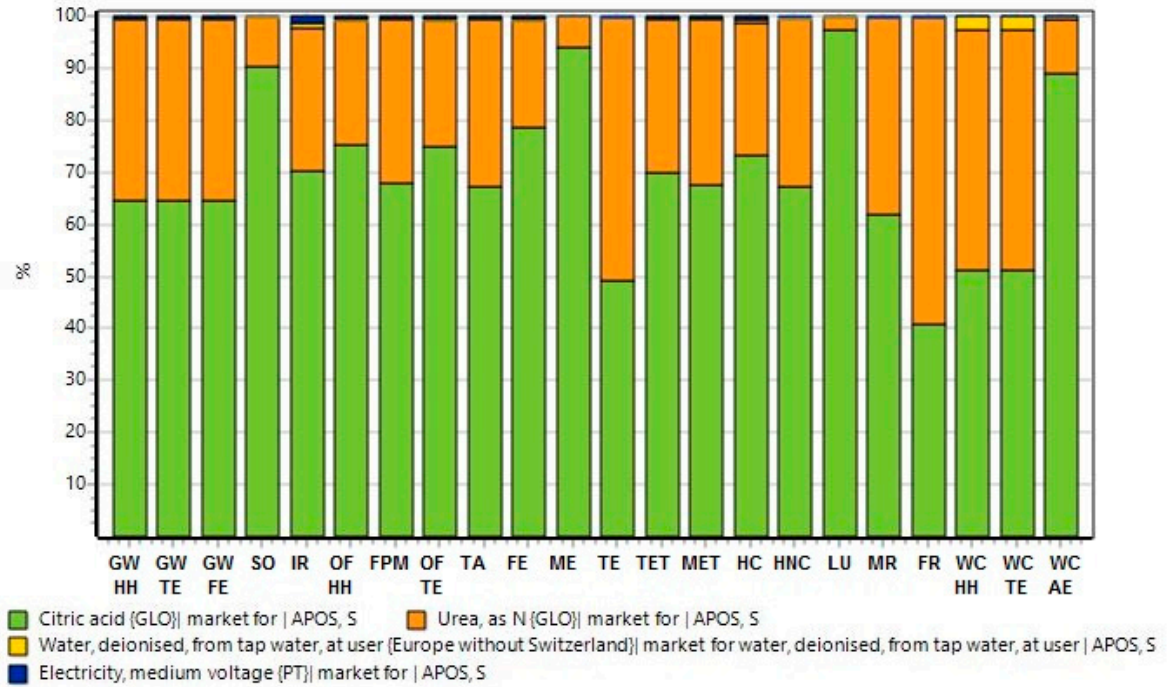
**Table S4.** Quantity of products required for the production of 1 kg of CDs (SimaPro software).

Carbon dots	Citric acid (kg)	Nitrogen source (kg)	Deionized water (kg)	Electricity (MJ)
1B	35	35	357	174
2	7.2	7.2	77.9	87.1
3	22.9	217	5.5	87.1
4	13.7	108	5.5	87.1
5	278	2720	5.5	87.1

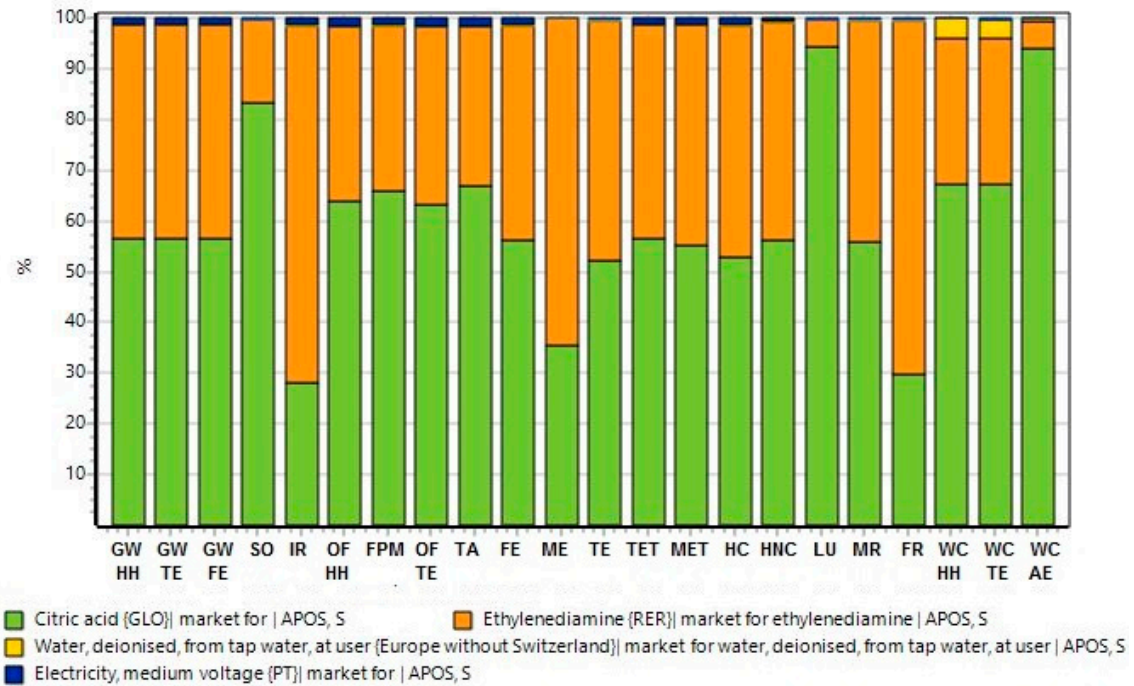
**Table S5.** Fluorescence quantum yields ( $Q_{Y_{FL}}$ ) and  $Q_{Y_{FL}}$ -based functional units.

Carbon dots	$Q_{Y_{FL}}$ (%)	$Q_{Y_{FL}}$ -based functional unit (using 1B as reference)
1B	3.5	1.0

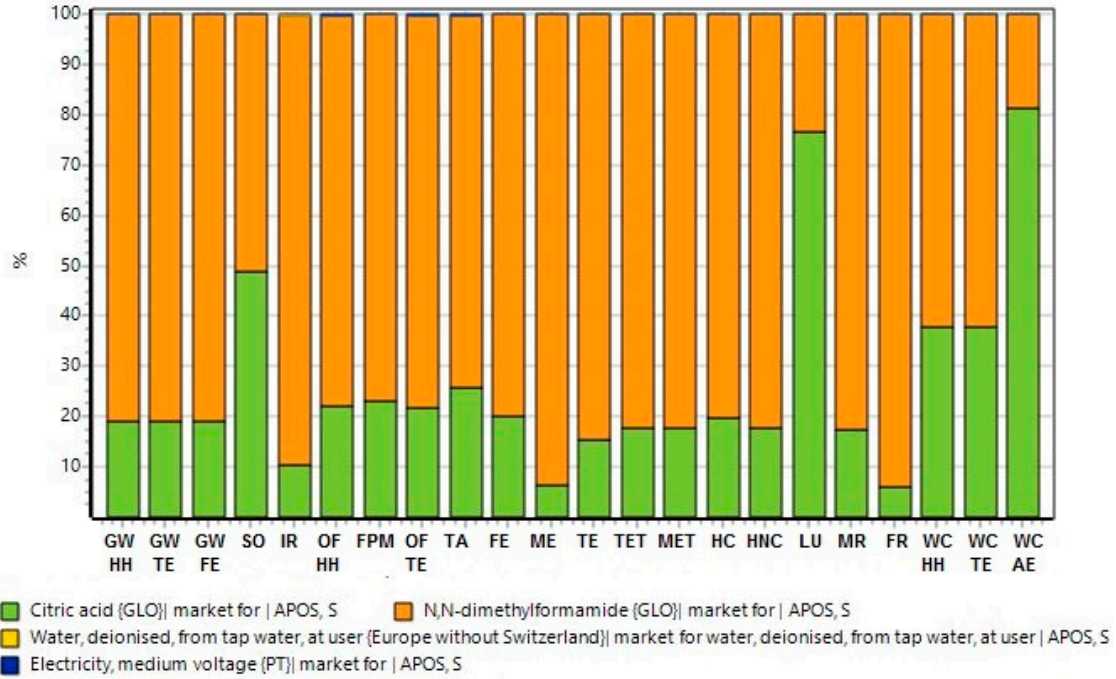
2	3.1	1.1
3	3.5	1.0
4	2.4	1.5
5	2.3	1.5



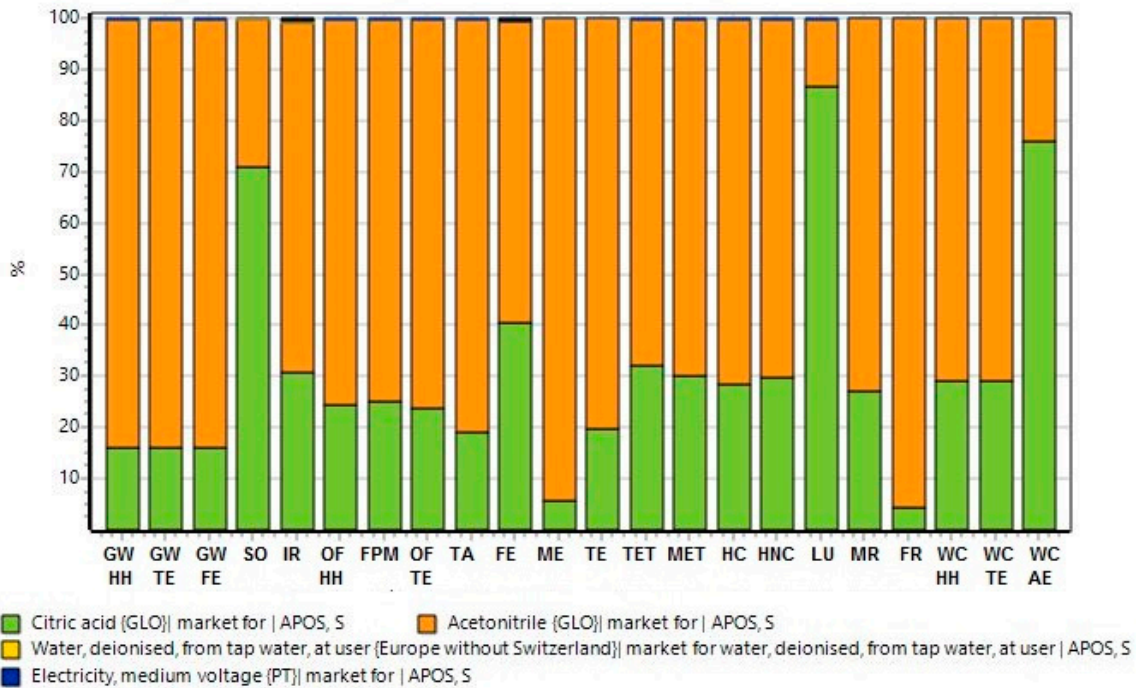
(A)



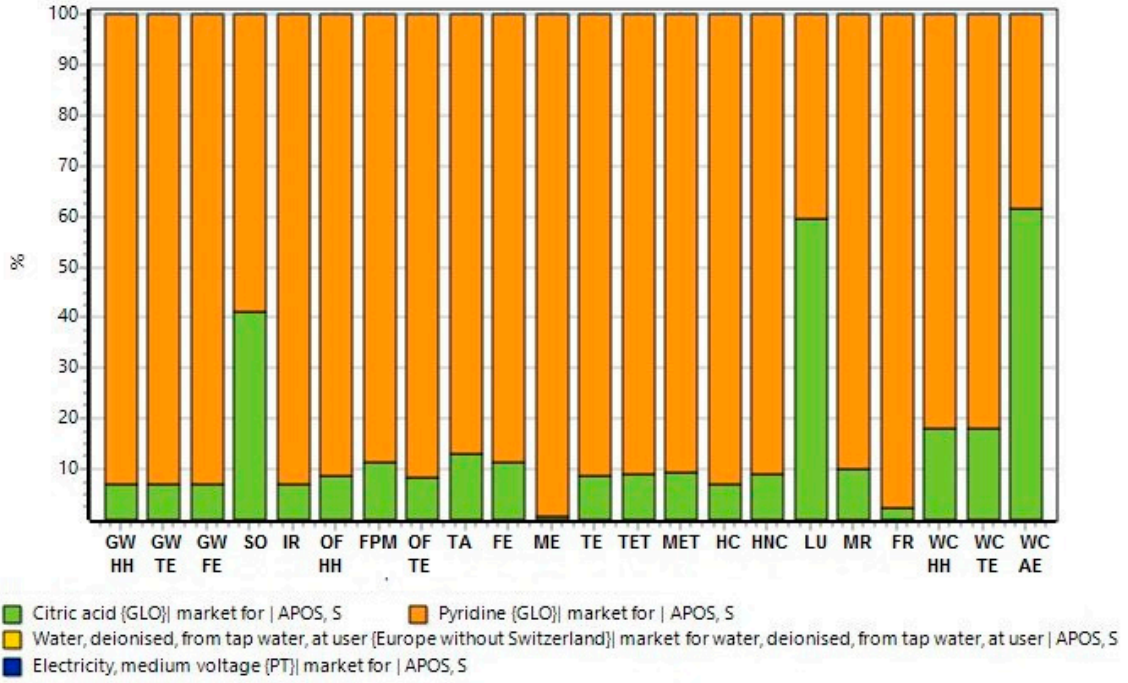
(B)



(C)



(D)



(E)

**Figure S6:** Relative environmental impacts of each carbon dots solution with the ReCiPe2016 LCIA for: (A) CD<sub>s</sub> 1B, (B) CD<sub>s</sub> 2, (C) CD<sub>s</sub> 3, (D) CD<sub>s</sub> 4 and (E) CD<sub>s</sub> 5.