1	Supplementary material to: Enrichment and characterisation of ethanol chain elongating
2	communities from natural and engineered environments
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23 S.1. Net growth of enrichments over different transfers



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25 Figure S.1. Net increase of OD at the end of each transfer over time. Colours indicate original 26 inoculum: blue comes from Lindemans brewery full-scale UASB, gray from Van Steenberghe 27 brewery full-scale UASB, orange from Ossemeersen full-scale CSTR AD, green from goat 28 faeces, purple from sheep faeces and golden from thin-stillage lab-scale pilot. Shape of symbols 29 indicates pH of enrichment: squares were enriched at pH 7, circles at pH 5.5. Lines indicate 30 enrichment replicate: full line was replicate 1, dashed line replicate 2. After the third transfer 31 (day 31), all enrichments with an OD below 0.01 were dropped. Sample names were 32 constructed of inoculum origin, enrichment pH and replicate number, e.g. Li 7 1 represents 33 replicate 1 of Lindemans inoculum enriched at pH 7.

34 S.2. Aggregated growth in enrichment series TS 5.5 1 during growth curves





Figure S.2. Pictures of growth in Balch tubes after 7 days of incubation for growth curve experiment. Panel A shows aggregated growth in enrichment TS 5.5 1 (replicate A), Panel B shows suspended growth in Li 5.5 1 (replicate C), and Panel C shows an uninoculated control

39 after 7 days of incubation at 34°C.









Figure S.4. Community composition determined by high throughput amplicon sequencing of enrichment communities after 12 transfers for all communities Sample names were constructed of inoculum origin, enrichment pH and replicate number, e.g. Li 7 1 represents replicate 1 of

- 54 or moculum origin, enrichment pH and replicate number, e.g. L1 / 1 represents replicate 1 0
- 55 Lindemans inoculum enriched at pH 7.



Figure S.5. Beta-diversity analysis of high-throughput amplicon sequencing data (Illumina) of 57 58 enriched communities after 12 transfers. pH and inoculum are indicated by color and shape of 59 the points respectively. Black symbols represent communities enriched at pH 5.5, gold points those enriched at pH 7. Symbol shapes refer to the initial inoculum the community was 60 61 enriched from; full square for Lindemans brewery full-scale UASB (Li, ■), full circle for Van 62 Steenberghe brewery full-scale UASB (VS, •) full triangle for Ossemeersen full-scale waste activated sludge digester (Os, \blacktriangle), empty square for goat feces (Go, \Box), empty circle for sheep 63 64 feces (Sh, \circ), and plus sign for + semi-pilot-scale thin stillage fermenter producing caproic 65 acid.

66 S.4. Community composition of inocula



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Figure S.6. Krona plot¹ of community presesnt in the full-scale granular sludge digester at
 Lindemans brewery (Li)







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Figure S.8. Krona plot¹ of community present in the full-scale waste activated sludge digester $\frac{1}{2}$

75 at the Ossemeersen wastewater treatment plant (Os)



Figure S.9. Krona plot¹ of community present in goat faeces used for this enrichment study
 (Go)



Figure S.10. Krona plot¹ of community present in sheep faeces used for this enrichment study
 (Sh)







Figure S.12. Growth curves of each enrichment used for further growth rate calculations. In
experiments where growth was completed after 120h, measurements were stopped. Sample
names were constructed of inoculum origin, enrichment pH and replicate number, e.g. Li 7 1
represents replicate 1 of Lindemans inoculum enriched at pH 7.

91 S.6. Impact of *Desulfovibrio* on product profile

The impact of the presence of *Desulfovibrio* on the product spectrum was investigated by first looking at the average product profile in communities at pH 7 and pH 5.5 with or without *Desulfovibrio*. The cut-off for presence was arbitrarily put at 5%, to select communities that were highly enriched in *Desulfovibrio*. Looking at the average product profile, both at pH 5.5 and 7 an increase can be seen in butyric acid when *Desulfovibrio* is present, although the spread at pH 7 is quite large (Figure S.13)





102 To confirm these observations, an ANOVA test was performed, studying the impact of the 103 presence of *Desulfovibrio* on butyric acid concentrations. Data at pH 5.5 were analysed 104 separately from data at pH 7. While both pH showed a significant p-value for the impact of 105 *Desulfovibrio*-presence (p=0.008 for pH 7; p<0.001 for pH 5.5), the Shapiro-Wilk normality 106 test showed residuals at pH 7 were not normally distributed (p=0.001), making it impossible to 107 make hard claims on this observation at pH 7. We can however conclude that at pH 5.5, the presence of *Desulfovibrio* significantly shifted the product profile towards butyric acid, as 108 109 residuals were normally distributed (p=0.19).



110 S.7. Electron balance over growth curve experiment

Figure S.14. Electron balance for each enrichment for the growth curve experiment. Error bars indicate standard deviation over all replicates (n=3-5). Sample names were constructed of inoculum origin, enrichment pH and replicate number, e.g. Li 7 1 represents replicate 1 of Lindemans inoculum enriched at pH 7.

116 S.8. Methods

- 117 S.8.1. Medium Composition
- 118 Medium for enrichments was modified from the DSM52 medium used for C. kluyveri, with a
- medium composition (in 1L basal medium): 0.31 g K₂HPO₄, 0.23 g KH₂PO₄, 0.25 g NH₄Cl,
- 120 0.05 g MgSO₄.7H₂O, 0.06 g MgCl₂, 2.50 g NaHCO₃, 4.88 g 2-(N-morpholino)ethanesulfonic
- 121 acid (MES), 5.28 g of 2-bromoethanesulfonate (BES), 4.28 g K-Ac and 15 mL ethanol· L^{-1} .
- 122 Vitamins and trace elements were added: 1 mL of trace element solution SL-10, 1 mL of
- 123 selenite-tungstate solution and 0.1 mL of 7-vitamin solution. The SL-10 trace element solution
- 124 consisted of (in 1L stock solution): 10 mL 7.7M HCl, 1.5 g FeCl₂.4H₂O, 0.07 g ZnCl₂, 0.15 g
- 125 MnCl₂.4H₂O, 0.006 g H₃BO₃, 0.19 g CoCl₂.6H₂O, 0.002 g CuCl₂.2H₂O, 0.024 g NiCl₂.6H₂O

126 and 0.036 g Na₂MoO₄.4H₂O. The selenite-tungstate solution contained (in 1L stock solution): 127 0.5 g NaOH, 0.003g Na₂SeO₃.5H₂O and 0.004 g Na₂WO₄.2H₂O. The 7-vitamin solution consisted of (in 100 mL stock solution): 0.1 g vitamin B12, 0.08 g p-aminobenzoic acid, 0.02 128 129 g D(+)-Biotin, 0.2 g nicotinic acid, 0.1 g Ca-pantothenate, 0.3 g pyridoxine hydrochloride, 0.2 g thiamine-HCl.2H₂O. Medium without ethanol was sparged for 15 min with N₂/CO₂ (90:10), 130 131 after which ethanol was added, and pH corrected to 7 using HCl or NaOH. Medium was 132 distributed in 40 mL aliquots in 120 mL serum flasks. Medium was then corrected to pH 5.5 133 using HCl and again distributed in 40 mL aliquots in 120 mL serum flasks. The headspace was 134 then replaced by N₂ using a gas exchange apparatus. For revival of the communities and 135 subsequent growth curve tests, medium without ethanol, NaHCO₃, vitamins and trace elements 136 was first distributed in Balch tubes, and autoclaved for 20 min at 121°C. Subsequently, the 137 necessary additions were supplemented from sterile anaerobic stocks and pH corrected with 138 sterile anaerobic 1M HCl or 1M NaOH. Final working volume was 9 mL per Balch tube (Vtot 139 = 26 mL), to maintain the same headspace-liquid ratio.

140 S.8.2. Enrichment strategy

Inocula	UA Linde (L	UASB Lindemans (Li)		UASB Van Steenberghe (VS)		AD Ossemeersen (Os)			at es))	Sheep feces (Sh))	Thin stillage pilot (TS)	
] × ×						
Conditions	Li 7 1	VS 7 1	Os 7 1	Go 7 1	Sh 7 1	TS 7 1		Li 5.5 1	VS 5.5 1	Os 5.5 1	Go 5.5 1	Sh 5.5 1	TS 5.5 1
	Li 7 2	VS 7 2	Os 7 2	Go 7 2	Sh 7 2	TS 7 2		Li 5.5 2	VS 5.5 2	Os 5.5 2	Go 5.5 2	Sh 5.5 2	TS 5.5 2

Figure S.15. Schematic overview of enrichment strategy from six inocula, at two pH in
 duplicate

144 S.8.3. Chemical analyses

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C2 to C8 carboxylic acids (including isoforms C4 to C6) were determined by gas 145 chromatography (GC; GC-2014, Shimadzu®, The Netherlands), with a DB-FFAP 123-3232 146 147 column (30 m \times 0.32 mm \times 0.25 µm; Agilent, Belgium) and a flame ionisation detector (FID). Liquid samples were conditioned with 2 mL sulfuric acid, 200 mg sodium chloride, and 2-148 149 methyl hexanoic acid as an internal standard for quantification before further extraction with 150 diethyl ether (1:1 volume sample:ether). The sample (1 µL) was injected at 250°C with a split ratio of 50 and a purge flow of 3 mL·min⁻¹. The oven temperature increased by 10° C·min⁻¹ 151 from 110 to 250°C where it was maintained for 5 min. The FID had a temperature of 300°C. 152 Nitrogen carrier gas was maintained at a flow rate of 2.49 mL \cdot min⁻¹. The GC was externally 153 calibrated with a minimum detection limit of 60 mg \cdot L⁻¹ for acetic acid, and 10 mg \cdot L⁻¹ for all 154 155 higher carboxylic acids.

Alcohols (acetone, methanol, ethanol, 1-propanol, 2-propanol, 1-butanol, 1-pentanol and
1-hexanol) were measured by GC (GC-2010-Plus, Shimadzu®, Belgium) with a StabilwaxDE-S column (30m x 0.32mm x 1.0 μm, Shimadzu®, Belgium) and an FID. 1 μL of sample

was injected at 230°C with a split ratio of 50 and purge flow of 3.0 mL·min⁻¹. Oven temperature was put at 35°C for one minute, after which it increased from 35°C to 230°C at a rate of 10°C min⁻¹ where it was kept for 2 minutes. FID temperature was set at 250°C, carrier gas was nitrogen gas at a flow rate of 1.09 mL·min⁻¹. The GC was externally calibrated with a minimum detection limit of 25 mg·L⁻¹.

164 The gas-phase composition was analyzed with a Compact GC (Global Analyser Solutions, 165 Breda, The Netherlands). CH_4 , O_2 and N_2 were analysed with a Molsieve 5A precolumn and 166 Porabond column using He as a carrier gas, H_2 was analysed with the same precolumn and 167 column, but using N_2 as a carrier gas. CO_2 , N_2O , and H_2S were analysed using an Rt-Q-bond 168 precolumn and column with He as a carrier gas. A thermal conductivity detector was used to 169 determine gases in the headspace.





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Figure S.16. Quality control of Illumina sequencing by comparison of a mock community with
 known composition to actual sequencing results, performed in triplicate

174 **8. References**

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