

Supporting Information  
**Potential for Release of Pulmonary Toxic Ketene from Vaping Pyrolysis of  
Vitamin E Acetate**

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## General Methods.

Atmospheric-pressure chemical ionization (APCI) and electrospray ionization (ESI) mass spectra were acquired using Advion Expression Mass Spectrometer in positive and negative modes as required.  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were recorded at room temperature at 400 MHz and 100 MHz respectively, and calibrated using residual non-deuterated solvent as an internal reference. HPLC analysis used an Atlantis dC18 reverse phase column (5 $\mu\text{M}$ , 4.6 x 250 mm with UV-Vis detector at 254 and 220 nm). GC-MS analysis used a Waters GCT Premier Micromass system with an Optima 5 MS (length 15 m, inner diameter 0.25 mm, outer diameter 0.40 mm) GC column and ESI positive mode MS detector.

GC conditions: initial temperature: 50 °C; maximum temp 360 °C, equilibrium time: 0.20 min, inject. Inlet: initial temp: 260 °C, split ratio: 10.0; purge flow: 40.0 mL/min. Temp. ramp:

Rate (°C/min)	Final temp (° C)	Hold time (min)	Total time (min)
35.00	120	0.0	3.1
10.00	250	0.0	16.1
35.00	300	4.0	21.5

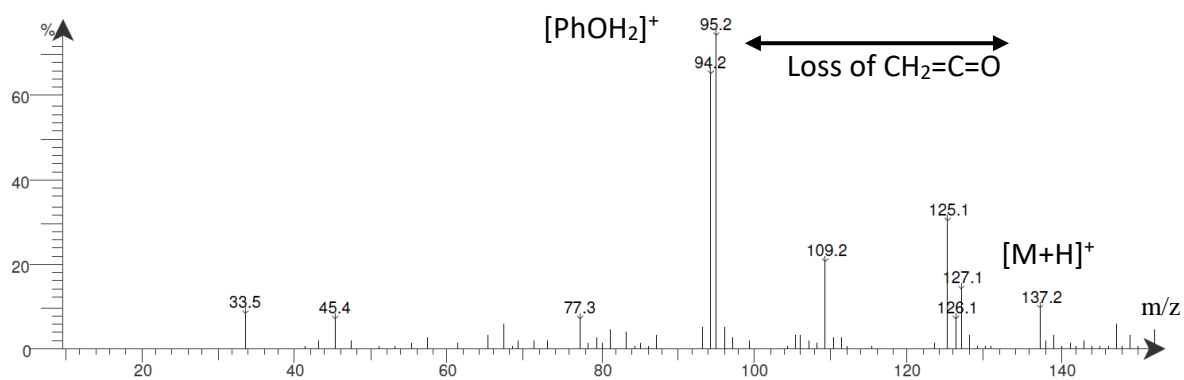
All vaping experiments were carried out in a ventilated fumehood.

## Computational Details.

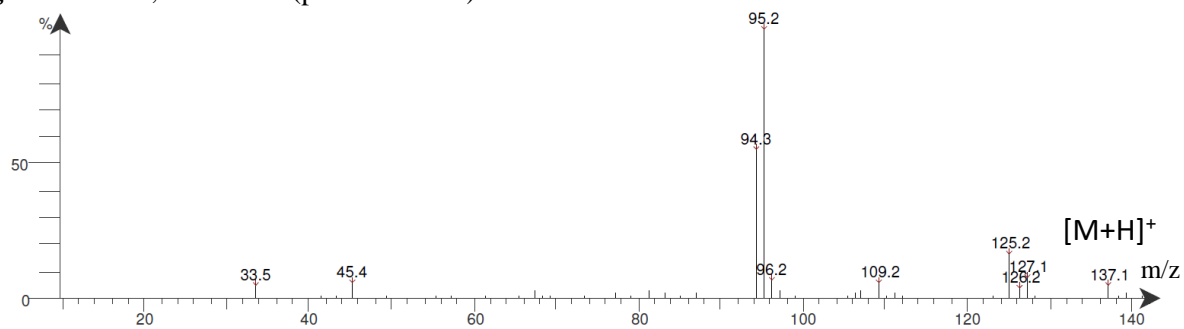
All of the calculations reported in this work were carried out using the Gaussian 09e01 package from Irish Centre for High-End Computing (ICHEC). The M06-2X density-functional method in conjunction with the 6-311G(d,p) basis set was selected for all the geometry optimizations and frequency analysis. All geometries were optimized in gas phase. Frequency calculations at 298.15 K and 1 atm pressure on all the stationary points were carried out at the same level of theory as the geometry optimizations to ascertain the nature of the stationary points. All of the presented relative energies are free energies at 298.15 K with respect to the reactants, unless otherwise stated. Each TS was confirmed by IRC calculations and vibrational analysis showing a single imaginary frequency. A six-membered TS in which proton transfer to an *ortho*-aryl ring position followed by enolization to phenol has also been proposed for the loss of ketene from phenyl acetate (**1**). As the *ortho* aryl positions of **1** are methyl substituted loss of ketene is unlikely to proceed through six-membered proton transfer to the aryl ring so the four membered TS was selected for comparative calculations.

1. E. Marquez, *et al.*, Electrocyclic [1,5] hydrogen shift in the thermal elimination kinetics of phenyl acetate and *p*-tolyl-acetate in the gas phase: a density functional theory study. *Mol. Phys.*, **112**, 462-473 (2014).

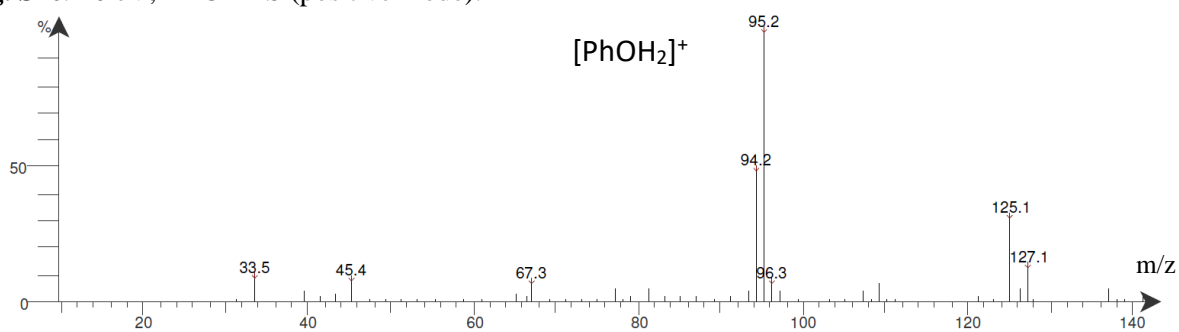
**Figure S1.** APCI-MS cone-voltage study for phenyl acetate **3**.



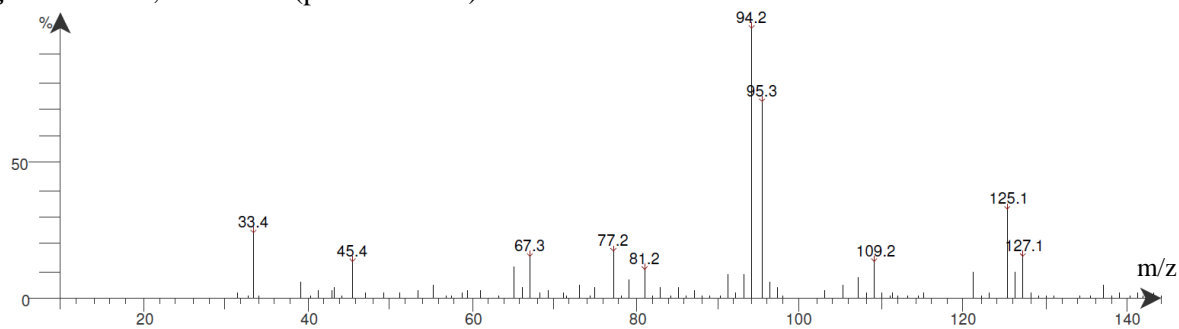
**Fig. S1a:** 10 eV, APCI-MS (positive mode).



**Fig. S1c:** 20 eV, APCI-MS (positive mode).

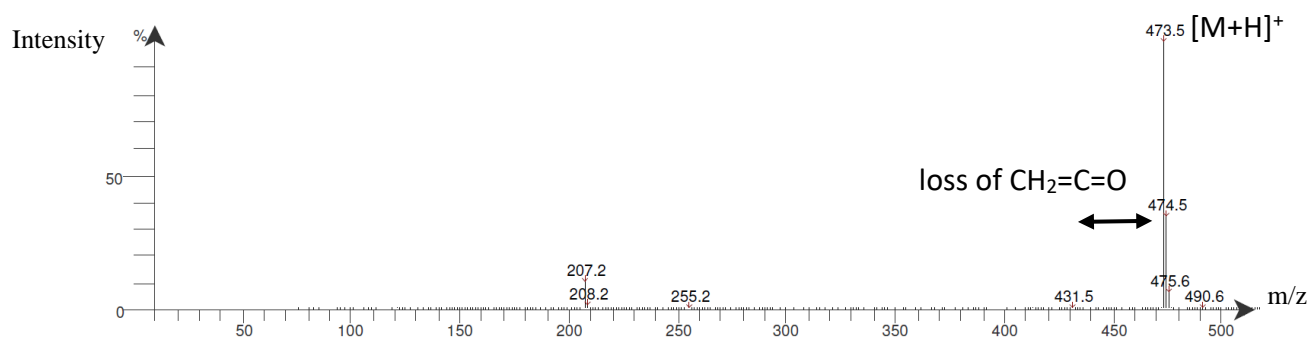


**Fig. S1d:** 25 eV, APCI-MS (positive mode).

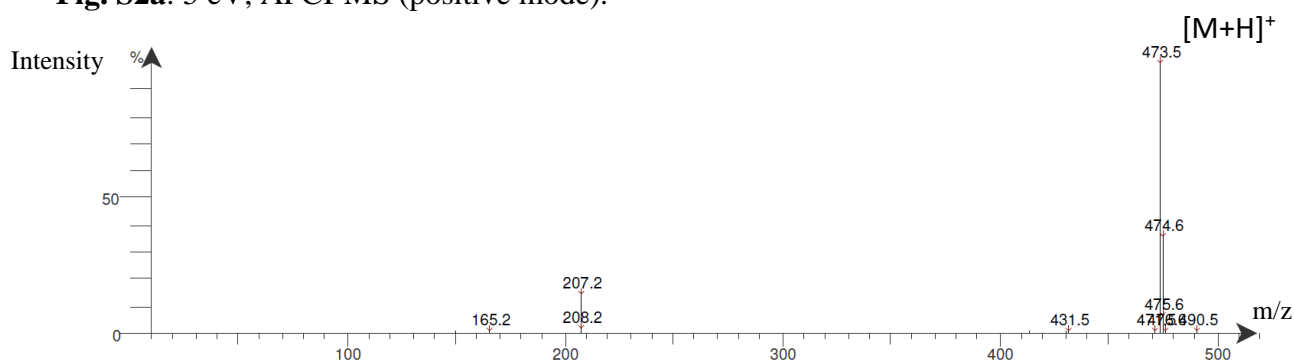


**Fig. S1e:** 30 eV, APCI-MS (positive mode).

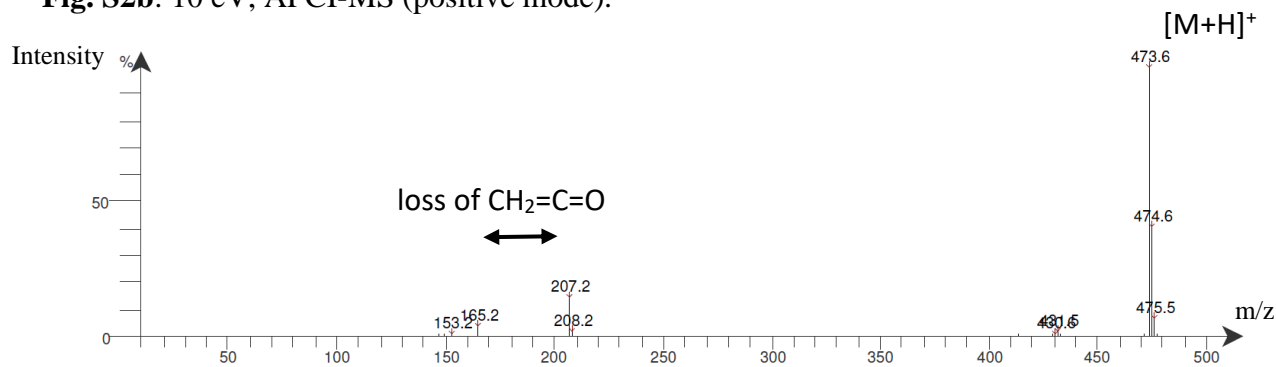
**Figure S2.** APCI-MS cone-voltage studies for vitamin E acetate **1**



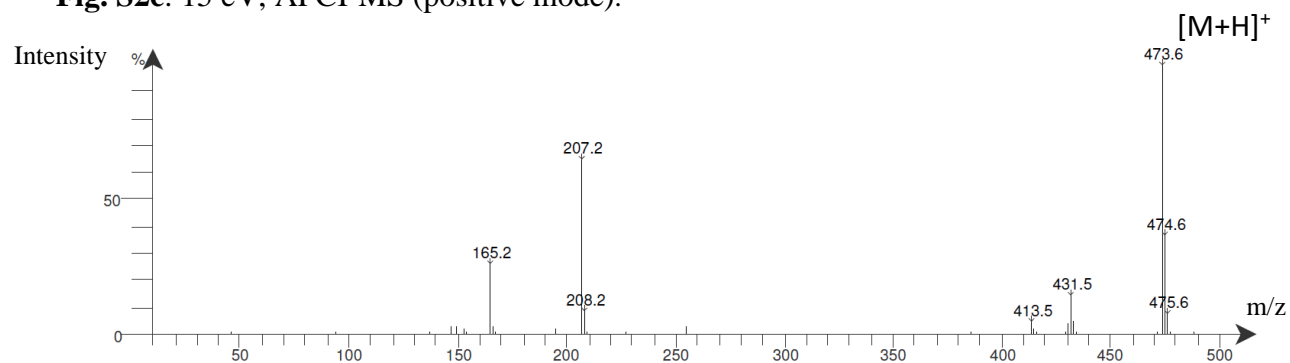
**Fig. S2a:** 5 eV, APCI-MS (positive mode).



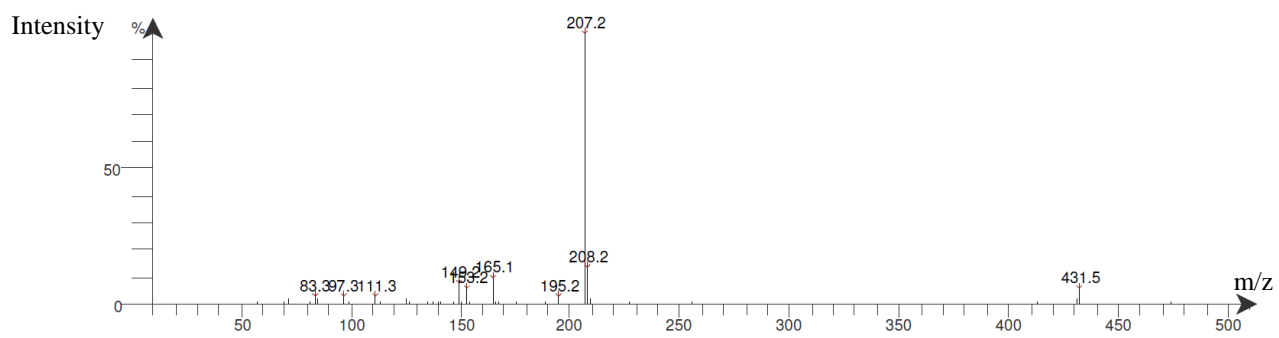
**Fig. S2b:** 10 eV, APCI-MS (positive mode).



**Fig. S2c:** 15 eV, APCI-MS (positive mode).

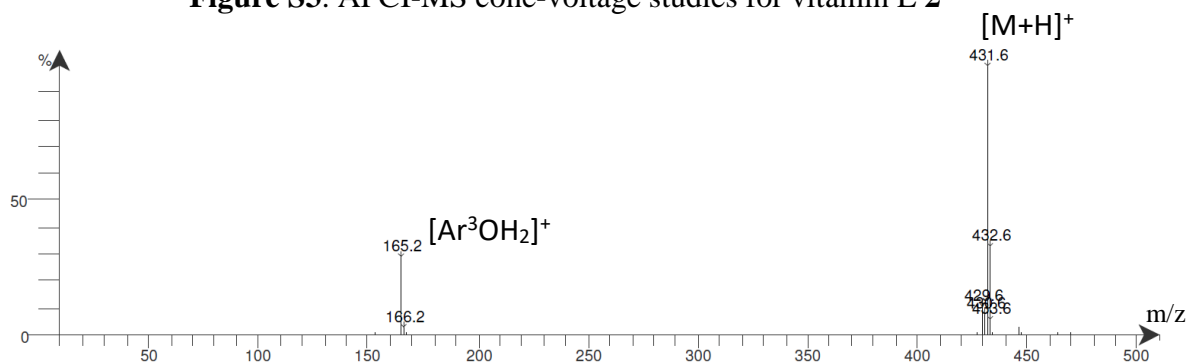


**Fig. S2d:** 20 eV, APCI-MS (positive mode).

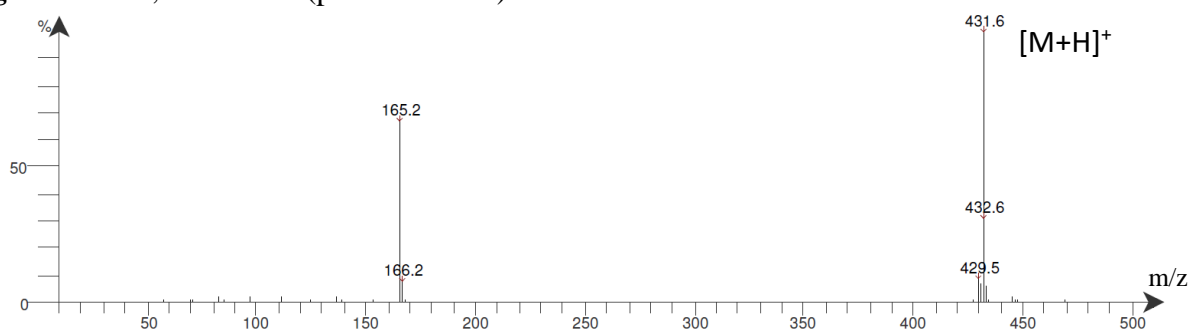


**Fig. S2f:** 30 eV, APCI-MS (positive mode).

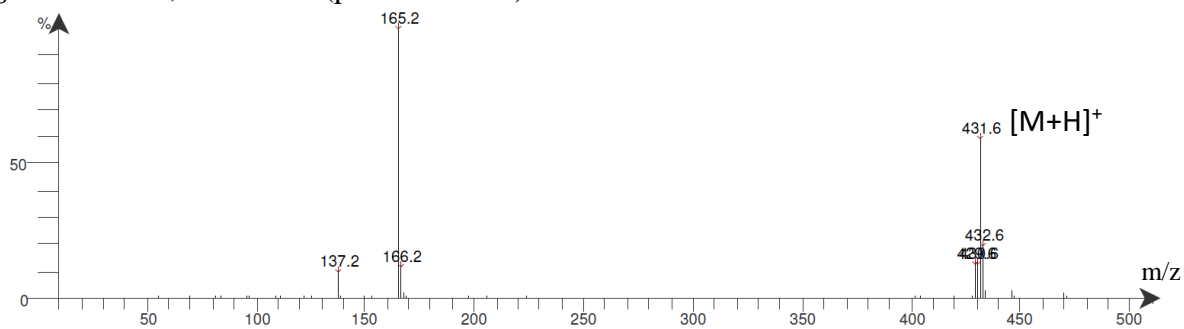
**Figure S3.** APCI-MS cone-voltage studies for vitamin E 2



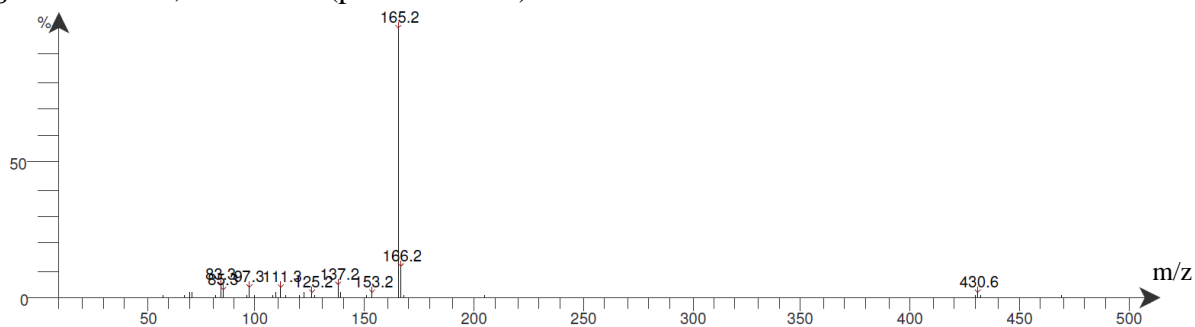
**Fig. S3a:** 5 eV, APCI-MS (positive mode).



**Fig. S3b:** 10 eV, APCI-MS (positive mode).



**Fig. S3c:** 20 eV, APCI-MS (positive mode).



**Fig. S3d:** 30 eV, APCI-MS (positive mode).

**Figure S4.** DFT calculation data for **3** and transition state

Ground state structure of phenyl acetate **3**

E(RM062X) = -460.052680107  
Zero-point correction= 0.143297 (Hartree/Particle)  
Thermal correction to Energy = 0.152147  
Thermal correction to Enthalpy = 0.153091  
Thermal correction to Gibbs Free Energy = 0.108399  
Sum of electronic and zero-point Energies = -459.909383  
Sum of electronic and thermal Energies = -459.900533  
Sum of electronic and thermal Enthalpies = -459.899589  
Sum of electronic and thermal Free Energies = -459.944281

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-2.461087	-1.071596	0.274742
2	6	0	-2.960549	0.213076	0.092541
3	6	0	-2.093537	1.249227	-0.236957
4	6	0	-0.731044	1.017001	-0.385217
5	6	0	-0.249634	-0.271794	-0.194087
6	6	0	-1.101048	-1.317314	0.133526
7	1	0	-3.128229	-1.886254	0.528666
8	1	0	-4.020091	0.406237	0.205620
9	1	0	-2.477563	2.252054	-0.379570
10	1	0	-0.053316	1.821804	-0.630791
11	1	0	-0.684038	-2.307996	0.265038
12	8	0	1.084015	-0.610537	-0.389011
13	6	0	2.077816	0.157872	0.139254
14	6	0	3.412590	-0.446371	-0.192692
15	1	0	3.500814	-0.567763	-1.272771
16	1	0	4.200639	0.198517	0.185922
17	1	0	3.484469	-1.436762	0.258890
18	8	0	1.893018	1.152982	0.770553

Ground state structure of ketene **4**

E(RM062X) = -152.578714584  
Zero-point correction = 0.032023 (Hartree/Particle)  
Thermal correction to Energy = 0.035501  
Thermal correction to Enthalpy = 0.036445  
Thermal correction to Gibbs Free Energy = 0.009110  
Sum of electronic and zero-point Energies = -152.546692  
Sum of electronic and thermal Energies = -152.543214  
Sum of electronic and thermal Enthalpies = -152.542270  
Sum of electronic and thermal Free Energies = -152.569605

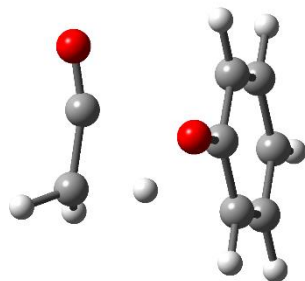
Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.000000	-0.000000	0.103973
2	8	0	-0.000000	-0.000000	1.258820
3	6	0	0.000000	0.000000	-1.204928
4	1	0	-0.000000	0.940447	-1.732415
5	1	0	-0.000000	-0.940447	-1.732415

## Ground state structure of phenol **5**

E(RM062X) = -307.41735370  
Zero-point correction = 0.105600 (Hartree/Particle)  
Thermal correction to Energy = 0.111050  
Thermal correction to Enthalpy = 0.111994  
Thermal correction to Gibbs Free Energy = 0.076670  
Sum of electronic and zero-point Energies = -307.311755  
Sum of electronic and thermal Energies = -307.306305  
Sum of electronic and thermal Enthalpies = -307.305361  
Sum of electronic and thermal Free Energies = -307.340685  
Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-1.127194	1.214015	-0.000000
2	6	0	0.263486	1.194447	0.000000
3	6	0	0.937438	-0.023909	0.000000
4	6	0	0.220142	-1.218571	0.000000
5	6	0	-1.166337	-1.185985	-0.000000
6	6	0	-1.849530	0.027643	-0.000000
7	1	0	-1.644688	2.165939	-0.000000
8	1	0	0.825228	2.123497	0.000000
9	1	0	0.769113	-2.151795	0.000000
10	1	0	-1.718883	-2.118096	-0.000000
11	1	0	-2.931768	0.046321	-0.000000
12	8	0	2.295374	-0.110676	0.000000
13	1	0	2.669969	0.773706	0.000001



Transition state image of **3** → **4** + **5**

## Transition state structure for **3** → **4** + **5**

Imaginary Frequency: -1637.0775.  
E(RM062X) = -459.939805174  
Zero-point correction = 0.135319 (Hartree/Particle)  
Thermal correction to Energy = 0.144461  
Thermal correction to Enthalpy = 0.145405  
Thermal correction to Gibbs Free Energy = 0.099782  
Sum of electronic and zero-point Energies = -459.804486  
Sum of electronic and thermal Energies = -459.795344  
Sum of electronic and thermal Enthalpies = -459.794400  
Sum of electronic and thermal Free Energies = -459.840023

Standard orientation:



Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.337300	-1.084505	-0.206101
2	6	0	1.049393	-1.220256	0.300383
3	6	0	0.267495	-0.085306	0.550677
4	6	0	0.804453	1.182431	0.287040
5	6	0	2.090228	1.303622	-0.219286
6	6	0	2.864656	0.173587	-0.472868
7	1	0	2.932915	-1.971434	-0.390245
8	1	0	0.642878	-2.198695	0.532814
9	1	0	0.197416	2.054695	0.501023
10	1	0	2.494613	2.290495	-0.414095
11	1	0	3.868617	0.275123	-0.865369
12	8	0	-0.988722	-0.201784	1.009236
13	1	0	-1.719082	-1.211474	0.552670
14	6	0	-2.467855	0.177694	-0.321309
15	8	0	-2.834906	1.252840	-0.264673
16	6	0	-2.424359	-1.207636	-0.553457
17	1	0	-3.375963	-1.628899	-0.864352
18	1	0	-1.580240	-1.456043	-1.199423

**Figure S5.** DFT calculation data for  $1^\dagger$  and its transition state

Ground state structure of  $1^\dagger$

E(RM062X) = -848.518936312  
 Zero-point correction = 0.351643 (Hartree/Particle)  
 Thermal correction to Energy = 0.371700  
 Thermal correction to Enthalpy = 0.372644  
 Thermal correction to Gibbs Free Energy = 0.303642  
 Sum of electronic and zero-point Energies = -848.167293  
 Sum of electronic and thermal Energies = -848.147237  
 Sum of electronic and thermal Enthalpies = -848.146292  
 Sum of electronic and thermal Free Energies = -848.215295

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.060561	0.492672	0.067751
2	6	0	0.631667	-0.829997	-0.074317
3	6	0	-0.717081	-1.083801	-0.356313
4	6	0	-1.591167	-0.008158	-0.456422
5	6	0	-1.173743	1.316091	-0.312559
6	6	0	0.170290	1.570826	-0.037893
7	6	0	-3.841196	-0.431864	0.237451
8	8	0	-2.926565	-0.259312	-0.763344
9	8	0	-4.977089	-0.639519	-0.057574
10	6	0	-1.212676	-2.496005	-0.529877
11	1	0	-1.276123	-3.015600	0.432525
12	1	0	-2.199160	-2.509559	-0.989559
13	1	0	-0.534251	-3.072886	-1.161404
14	6	0	0.667799	2.983101	0.126890
15	1	0	0.734610	3.487120	-0.842678
16	1	0	-0.011723	3.567133	0.750820
17	1	0	1.655685	2.994653	0.580022
18	6	0	1.608893	-1.971593	0.082420
19	1	0	1.161610	-2.758595	0.695360
20	1	0	1.808347	-2.425671	-0.895727
21	6	0	2.909801	-1.494865	0.719213
22	1	0	2.756098	-1.318641	1.788263
23	1	0	3.694425	-2.248287	0.614721
24	8	0	2.361133	0.817908	0.334202
25	6	0	3.370095	-0.178788	0.094750
26	6	0	3.596956	-0.298166	-1.411208
27	1	0	2.669595	-0.534189	-1.936335
28	1	0	3.975541	0.648081	-1.801796
29	1	0	4.327300	-1.082793	-1.624213
30	6	0	4.611956	0.352784	0.790182
31	1	0	5.453428	-0.324650	0.630370
32	1	0	4.870593	1.335502	0.391766
33	1	0	4.431878	0.449023	1.862314
34	6	0	-2.156619	2.449727	-0.450995
35	1	0	-3.098373	2.105780	-0.874702
36	1	0	-2.362903	2.914451	0.519055
37	1	0	-1.754887	3.230472	-1.099939
38	6	0	-3.328548	-0.343620	1.654308
39	1	0	-4.168591	-0.509485	2.322752
40	1	0	-2.551533	-1.089777	1.829909
41	1	0	-2.883333	0.635239	1.841931

$^\dagger$  Simplified structure used for calculation without  $C_{15}H_{32}$  alkyl group.

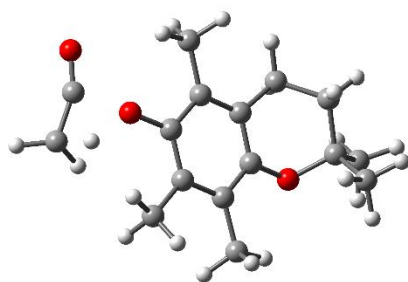
## Ground state structure of 2<sup>†</sup>

E(RM062X) = -695.883672263  
 Zero-point correction = 0.314338 (Hartree/Particle)  
 Thermal correction to Energy = 0.331069  
 Thermal correction to Enthalpy = 0.332013  
 Thermal correction to Gibbs Free Energy = 0.271751  
 Sum of electronic and zero-point Energies = -695.569334  
 Sum of electronic and thermal Energies = -695.552603  
 Sum of electronic and thermal Enthalpies = -695.551659  
 Sum of electronic and thermal Free Energies = -695.611922

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.283099	0.458090	-0.145796
2	6	0	-0.027932	-0.914811	-0.137202
3	6	0	1.296994	-1.358301	-0.019870
4	6	0	2.328073	-0.422371	0.021265
5	6	0	2.075821	0.951546	-0.010932
6	6	0	0.755360	1.395518	-0.072893
7	8	0	3.641709	-0.815216	0.118103
8	6	0	1.637259	-2.827034	0.065591
9	1	0	0.771090	-3.429260	0.333071
10	1	0	2.022775	-3.216236	-0.884432
11	1	0	2.394934	-3.003976	0.834660
12	6	0	0.445369	2.869465	-0.087428
13	1	0	1.058146	3.403307	0.641581
14	1	0	0.655999	3.302600	-1.070692
15	1	0	-0.604079	3.044553	0.136085
16	6	0	-1.171140	-1.899852	-0.255704
17	1	0	-0.915041	-2.679023	-0.978114
18	1	0	-1.327252	-2.407563	0.703823
19	6	0	-2.456883	-1.204678	-0.692782
20	1	0	-2.405404	-0.966069	-1.759461
21	1	0	-3.321702	-1.854886	-0.537521
22	8	0	-1.551114	0.974663	-0.249052
23	6	0	-2.650527	0.108418	0.062989
24	6	0	-2.712547	-0.096973	1.576096
25	1	0	-1.769613	-0.489287	1.961746
26	1	0	-2.908208	0.857488	2.068278
27	1	0	-3.513087	-0.795022	1.833673
28	6	0	-3.881754	0.847681	-0.434304
29	1	0	-4.784640	0.276323	-0.207791
30	1	0	-3.952567	1.823263	0.050349
31	1	0	-3.818285	1.000968	-1.513083
32	6	0	3.218788	1.931512	0.034787
33	1	0	4.165969	1.426897	-0.140643
34	1	0	3.091373	2.713146	-0.717334
35	1	0	3.274258	2.423842	1.010799
36	1	0	3.717877	-1.735903	-0.140304

<sup>†</sup> Simplified structure used for calculation without C<sub>15</sub>H<sub>32</sub> alkyl group.



Transition state image of  $1 \rightarrow 4 + 2$

Transition state structure of  $1^\ddagger \rightarrow 4 + 2^\ddagger$

Imaginary Frequency: -1669.7490.

E(RM062X) = -848.401721573

Zero-point correction =	0.344957 (Hartree/Particle)
Thermal correction to Energy =	0.364819
Thermal correction to Enthalpy =	0.365763
Thermal correction to Gibbs Free Energy =	0.297155
Sum of electronic and zero-point Energies =	-848.056765
Sum of electronic and thermal Energies =	-848.036903
Sum of electronic and thermal Enthalpies =	-848.035958
Sum of electronic and thermal Free Energies =	-848.104567

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.311464	1.545022	0.080560
2	6	0	-1.028275	1.409248	-0.312785
3	6	0	-1.571673	0.127753	-0.495900
4	6	0	-0.782920	-1.029314	-0.362773
5	6	0	0.540078	-0.882043	0.043397
6	6	0	1.063625	0.390805	0.272670
7	8	0	-2.889845	-0.012276	-0.813806
8	1	0	-3.720235	0.824572	-0.277777
9	6	0	-4.081070	-0.680882	0.564528
10	8	0	-4.314516	-1.796517	0.467180
11	6	0	-4.288407	0.661315	0.926738
12	1	0	-5.243116	0.854665	1.405848
13	1	0	-3.418745	1.065087	1.446098
14	6	0	-1.376921	-2.383282	-0.662730
15	6	0	1.530699	-1.999922	0.228214
16	6	0	-1.857670	2.643468	-0.575608
17	6	0	0.930281	2.901113	0.298344
18	6	0	2.817194	-1.592510	-0.499638
19	8	0	2.345900	0.484498	0.753788
20	6	0	3.373221	-0.267151	0.063798
21	1	0	-1.257287	3.411118	-1.066566
22	1	0	-2.703409	2.420487	-1.225910
23	1	0	-2.247019	3.085583	0.347915
24	1	0	1.055206	3.431464	-0.651164
25	1	0	0.302427	3.525743	0.938483
26	1	0	1.909108	2.798887	0.762022
27	1	0	-0.710181	-2.968524	-1.299251
28	1	0	-1.552927	-2.961843	0.249092
29	1	0	-2.329140	-2.272406	-1.178173
30	1	0	1.739272	-2.135579	1.296105
31	1	0	1.153051	-2.948149	-0.153324
32	1	0	2.585878	-1.478712	-1.562607

33	1	0	3.586055	-2.364809	-0.412374
34	6	0	4.422055	-0.530287	1.134092
35	6	0	3.919916	0.616378	-1.051342
36	1	0	4.697634	0.407010	1.620471
37	1	0	4.025768	-1.209198	1.892595
38	1	0	4.680334	0.084653	-1.629319
39	1	0	3.107781	0.900219	-1.726240
40	1	0	4.362887	1.523109	-0.633805
41	1	0	5.314759	-0.977273	0.691203

---

† Simplified structure used for calculation without C<sub>15</sub>H<sub>32</sub> alkyl group.

**Figure S6.** DFT calculation data of **6** and its transition state

Ground state structure of **6**

E(RM062X) = -691.255415027  
Zero-point correction= 0.236042 (Hartree/Particle)  
Thermal correction to Energy= 0.251804  
Thermal correction to Enthalpy= 0.252748  
Thermal correction to Gibbs Free Energy= 0.192424  
Sum of electronic and zero-point Energies= -691.019373  
Sum of electronic and thermal Energies= -691.003611  
Sum of electronic and thermal Enthalpies= -691.002667  
Sum of electronic and thermal Free Energies= -691.062991

Standard orientation:

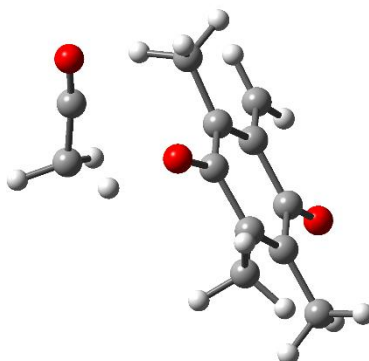
Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.638646	1.290650	0.009154
2	6	0	2.311849	-0.056226	0.100299
3	6	0	1.464417	-1.259614	-0.054193
4	6	0	0.140542	-1.119873	-0.290424
5	6	0	-0.472026	0.210974	-0.360972
6	6	0	0.196316	1.367247	-0.228511
7	8	0	-1.838881	0.244807	-0.638032
8	6	0	-2.687053	-0.047579	0.389520
9	6	0	-4.109735	-0.001232	-0.086873
10	1	0	-4.259363	-0.775618	-0.841369
11	1	0	-4.776616	-0.164359	0.755049
12	1	0	-4.311663	0.961558	-0.557195
13	8	0	-2.312280	-0.309450	1.491708
14	6	0	-0.762682	-2.304502	-0.495771
15	1	0	-1.339017	-2.496020	0.414800
16	1	0	-1.470832	-2.112999	-1.303915
17	1	0	-0.193832	-3.198923	-0.738834
18	6	0	2.157817	-2.589701	0.052739
19	1	0	1.622675	-3.264977	0.722654
20	1	0	2.235782	-3.075581	-0.924502
21	1	0	3.166427	-2.440717	0.432871
22	6	0	-0.472784	2.707483	-0.306267
23	1	0	-1.529150	2.604651	-0.542655
24	1	0	-0.376474	3.238013	0.644893
25	1	0	-0.006753	3.324124	-1.079072
26	6	0	2.410406	2.373025	0.170801
27	8	0	3.508860	-0.140831	0.288547
28	1	0	2.012901	3.379318	0.128974
29	1	0	3.470042	2.241400	0.353496

Ground state structure of **8**

E(RM062X) = -538.618811938  
Zero-point correction = 0.197824 (Hartree/Particle)  
Thermal correction to Energy = 0.209617  
Thermal correction to Enthalpy = 0.210562  
Thermal correction to Gibbs Free Energy = 0.160403  
Sum of electronic and zero-point Energies = -538.420988  
Sum of electronic and thermal Energies = -538.409194  
Sum of electronic and thermal Enthalpies = -538.408250  
Sum of electronic and thermal Free Energies = -538.458409

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	1.030883	-1.112045	-0.003198
2	6	0	-0.434815	-1.468541	0.004811
3	6	0	-1.414396	-0.356234	0.001811
4	6	0	-0.973073	0.919557	-0.001491
5	6	0	0.465755	1.231678	-0.001329
6	6	0	1.434185	0.291780	0.009974
7	8	0	0.694505	2.574115	-0.011066
8	6	0	-1.899904	2.101948	-0.004249
9	1	0	-1.706769	2.733565	0.865887
10	1	0	-1.713534	2.723230	-0.883321
11	1	0	-2.942863	1.795414	0.001688
12	6	0	-2.869714	-0.738090	0.004128
13	1	0	-3.380318	-0.348429	0.887953
14	1	0	-3.382859	-0.347917	-0.877998
15	1	0	-2.958021	-1.822317	0.004200
16	6	0	2.904772	0.606326	0.025230
17	1	0	3.128447	1.670079	0.121317
18	1	0	3.392136	0.109957	0.868468
19	1	0	3.389721	0.254386	-0.889869
20	6	0	1.898767	-2.133475	-0.029933
21	8	0	-0.795567	-2.628221	0.011721
22	1	0	2.971170	-1.986515	-0.043418
23	1	0	1.518043	-3.146968	-0.041667
24	1	0	1.638581	2.740930	-0.053011



Transitions state image of **6** → **4** + **8**

### Transition state structure of **6** → **4** + **8**

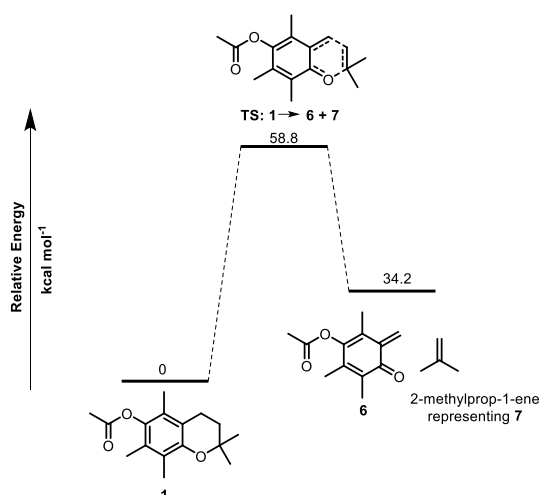
Imaginary Frequency: -886.8120.  
E(RM062X) = -691.142913169  
Zero-point correction = 0.228994 (Hartree/Particle)  
Thermal correction to Energy = 0.245075  
Thermal correction to Enthalpy = 0.246020  
Thermal correction to Gibbs Free Energy = 0.184954  
Sum of electronic and zero-point Energies = -690.913920  
Sum of electronic and thermal Energies = -690.897838  
Sum of electronic and thermal Enthalpies = -690.896894  
Sum of electronic and thermal Free Energies = -690.957959

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	2.062301	-0.740116	-0.032863
2	6	0	0.859372	-1.274797	0.264801
3	6	0	-0.366336	-0.443774	0.444835
4	6	0	-0.288484	0.926184	0.383844
5	6	0	0.973130	1.575833	0.065969
6	6	0	2.204983	0.729829	-0.153238
7	8	0	-1.488696	-1.092041	0.601300
8	1	0	-2.038285	-1.315044	-0.871127
9	6	0	-3.506101	-0.176497	-0.494674
10	8	0	-4.302903	0.016581	0.271477
11	6	0	-2.629112	-0.536344	-1.525041
12	1	0	-3.160008	-0.794197	-2.439480
13	1	0	-1.847951	0.230481	-1.633316
14	6	0	-1.494985	1.756501	0.718996
15	1	0	-2.049687	2.116074	-0.164681
16	1	0	-2.175196	1.181244	1.348327
17	1	0	-1.210968	2.650108	1.278814
18	6	0	1.130728	2.902027	-0.084000
19	6	0	0.661687	-2.755281	0.430000
20	1	0	-0.072844	-2.943424	1.212115
21	1	0	0.257794	-3.191774	-0.490514
22	1	0	1.595532	-3.264044	0.663359
23	6	0	3.308563	-1.557109	-0.244060
24	1	0	3.682371	-1.958423	0.702952
25	1	0	3.124325	-2.400277	-0.912712
26	1	0	4.087302	-0.927995	-0.670823
27	8	0	3.278201	1.236100	-0.418512
28	1	0	0.311746	3.600869	0.030149
29	1	0	2.108581	3.292539	-0.334587



**Figure S7.** DFT calculation data for chroman retro Diels-Alder transition state for **1**<sup>†</sup>



Calculated (gas phase at the M06-2X/6-311G (d,p) level) energy profile for the retro Diels-Alder pyrolysis of vitamin E acetate (calculation version). Relative free energies (kcal mol<sup>-1</sup>) for starting materials, transition state and products formed for the pyrolysis of **1**<sup>†</sup> leading to the formation of **6** and **7**<sup>††</sup>.

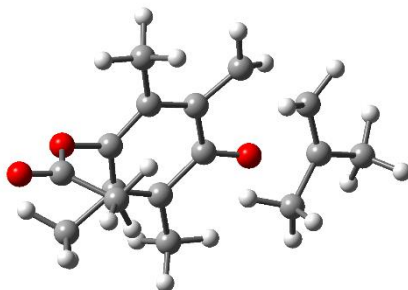
<sup>†</sup> Simplified structure used for calculation without C<sub>15</sub>H<sub>32</sub> alkyl group. <sup>††</sup> 2-Methylprop-1-ene representing **7**.

Ground state structure of 2-methylprop-1-ene (representing **7**).

```
E(RM062X) = -157.179383224
Zero-point correction= 0.108660 (Hartree/Particle)
Thermal correction to Energy = 0.113879
Thermal correction to Enthalpy = 0.114824
Thermal correction to Gibbs Free Energy = 0.081493
Sum of electronic and zero-point Energies = -157.070724
Sum of electronic and thermal Energies = -157.065504
Sum of electronic and thermal Enthalpies = -157.064560
Sum of electronic and thermal Free Energies = -157.097890
```

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.001250	-1.455560	-0.000009
2	6	0	0.000028	-0.126264	-0.000013
3	6	0	1.270638	0.679225	0.000011
4	1	0	2.153773	0.040007	-0.000100
5	1	0	1.311438	1.330002	-0.879051
6	1	0	1.311589	1.329940	0.879096
7	6	0	-1.271800	0.677270	0.000016
8	1	0	-1.313397	1.328320	0.878849
9	1	0	-1.313684	1.327697	-0.879257
10	1	0	-2.154080	0.036912	0.000323
11	1	0	0.927383	-2.019514	0.000036
12	1	0	-0.923721	-2.021392	0.000071



Transitions state image representing  $1 \rightarrow 6 + 7$

Transition state structure of  $1^\ddagger \rightarrow 6 + 7^\ddagger$

Imaginary Frequency: -398.8034.

E(RM062X) = -848.418968657

Zero-point correction=	0.346404 (Hartree/Particle)
Thermal correction to Energy =	0.367276
Thermal correction to Enthalpy =	0.368220
Thermal correction to Gibbs Free Energy =	0.297471
Sum of electronic and zero-point Energies =	-848.072565
Sum of electronic and thermal Energies =	-848.051692
Sum of electronic and thermal Enthalpies =	-848.050748
Sum of electronic and thermal Free Energies =	-848.121498

Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-2.770997	-1.863904	0.410401
2	6	0	-1.774072	-1.444261	-1.344134
3	6	0	-0.727366	-0.559637	-1.082001
4	6	0	-1.150724	0.798246	-0.782195
5	6	0	-3.163314	-0.648196	0.926375
6	6	0	0.641940	-0.953758	-0.946331
7	6	0	-0.127152	1.780199	-0.439513
8	6	0	1.182064	1.389294	-0.387710
9	6	0	1.540050	0.014960	-0.614014
10	8	0	-2.370488	1.051556	-0.732664
11	1	0	-1.931147	-2.376050	0.868544
12	1	0	-3.514242	-2.497048	-0.062753
13	1	0	-2.669044	-1.018852	-1.779611
14	1	0	-1.557305	-2.476747	-1.596290
15	6	0	-2.266920	0.040433	1.903486
16	1	0	-2.509502	1.098247	2.001620
17	1	0	-2.342116	-0.438890	2.885773
18	1	0	-1.221647	-0.053643	1.579735
19	6	0	-4.525795	-0.093204	0.661680
20	1	0	-5.184249	-0.324416	1.507709
21	1	0	-4.484006	0.987881	0.537994
22	1	0	-4.961277	-0.534481	-0.236232
23	6	0	1.059393	-2.378626	-1.186417
24	1	0	0.599494	-3.053117	-0.456285
25	1	0	0.743825	-2.709358	-2.178835
26	1	0	2.140582	-2.489116	-1.125606
27	6	0	-0.582845	3.188068	-0.166489
28	1	0	-1.670977	3.218171	-0.172243
29	1	0	-0.224931	3.547431	0.801490

30	1	0	-0.218737	3.880399	-0.931000
31	6	0	2.302132	2.349150	-0.082728
32	1	0	1.962648	3.381991	-0.116886
33	1	0	2.725482	2.166945	0.911333
34	1	0	3.116193	2.217840	-0.797375
35	8	0	2.896123	-0.291536	-0.528037
36	6	0	3.438048	-0.662086	0.670843
37	8	0	4.611250	-0.859986	0.732630
38	6	0	2.484595	-0.815673	1.832797
39	1	0	1.920939	0.103202	2.005253
40	1	0	1.759926	-1.605216	1.622884
41	1	0	3.070788	-1.071470	2.711044

-----

† Simplified structure used for calculation without C<sub>15</sub>H<sub>32</sub> alkyl group. ††2-Methylprop-1-ene representing **7**.

Figure S8. APCI-MS cone-voltage study of vaped vitamin E acetate **1**

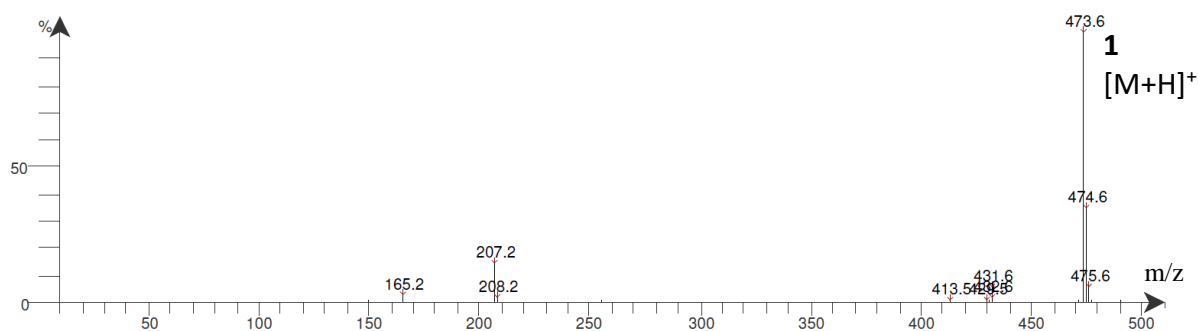


Fig. S8a: 5 eV, APCI (positive mode).

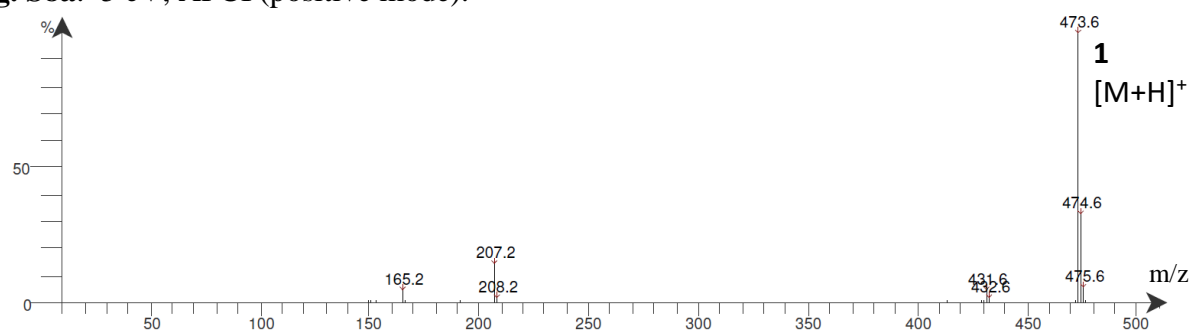


Fig. S8b: 10 eV, APCI (positive mode).

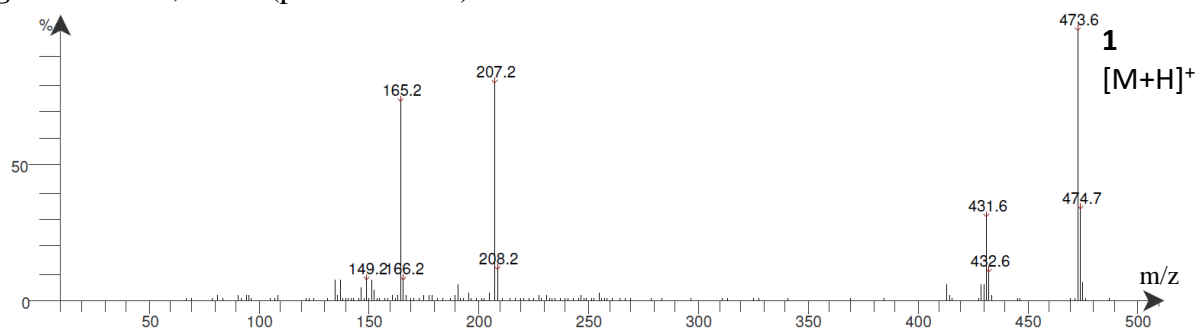


Fig. S8c: 20 eV, APCI (positive mode).

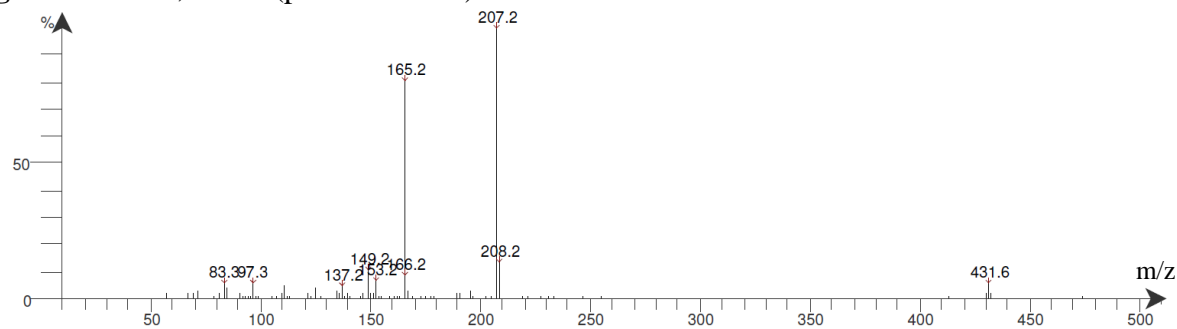
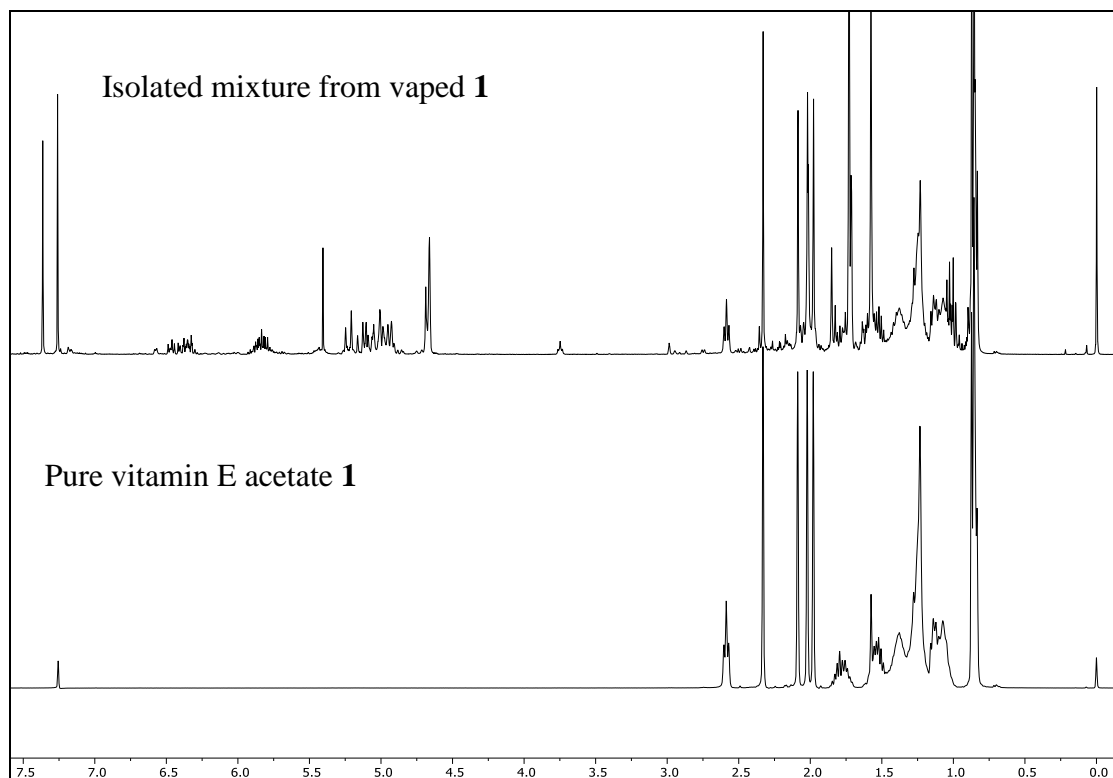


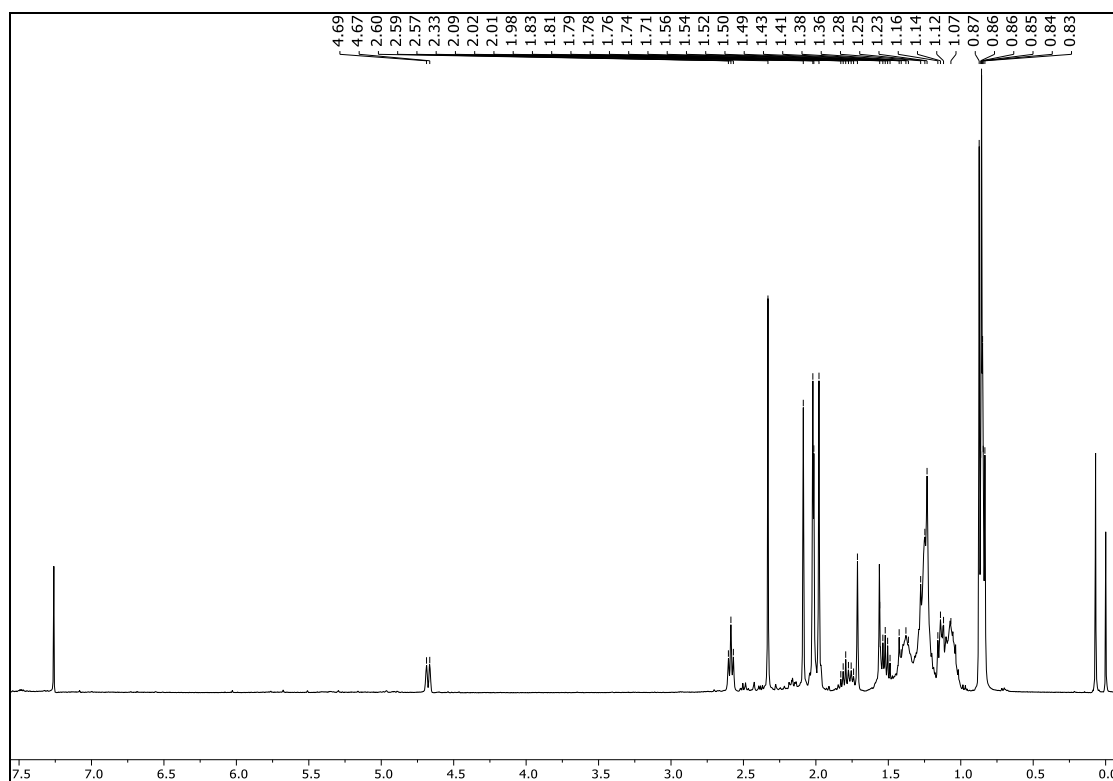
Fig. S8d: 30 eV, APCI (positive mode).

Figure S9.  $^1\text{H}$  NMR spectrum of isolated mixture from vaped **1** and pure vitamin E acetate **1**

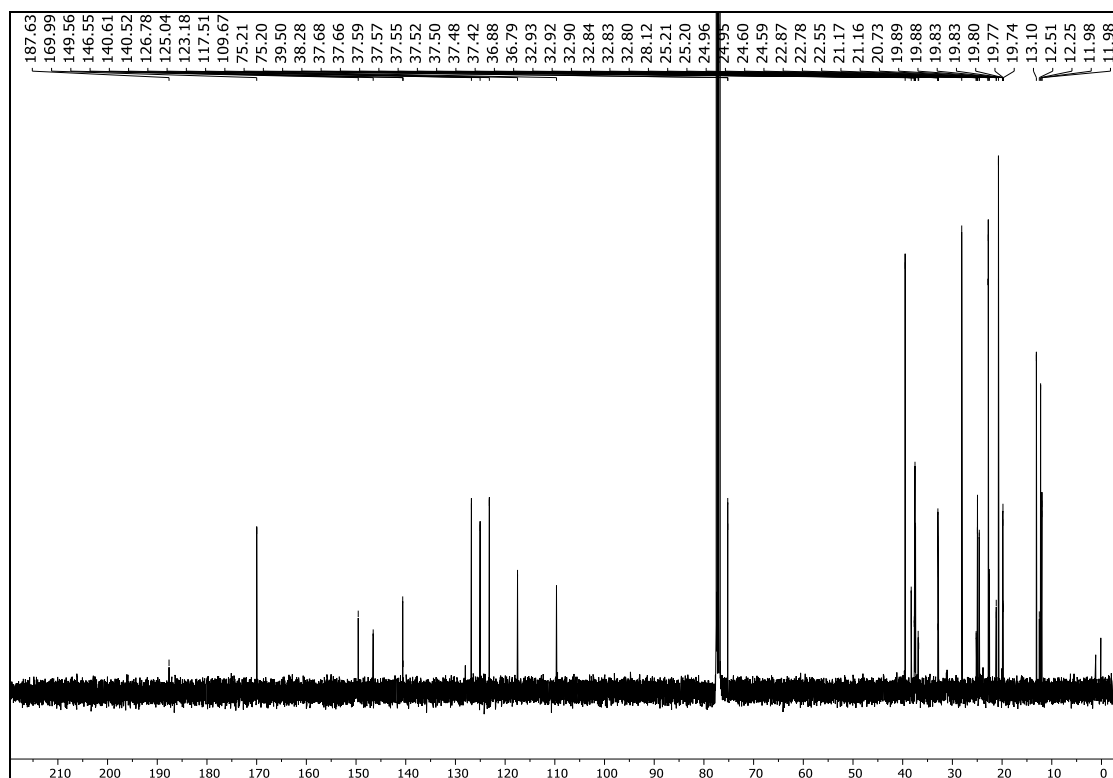


**Figure S10**  $^1\text{H}$  NMR spectrum of the non-volatile component (NVC) of isolated mixture from vaped **1**

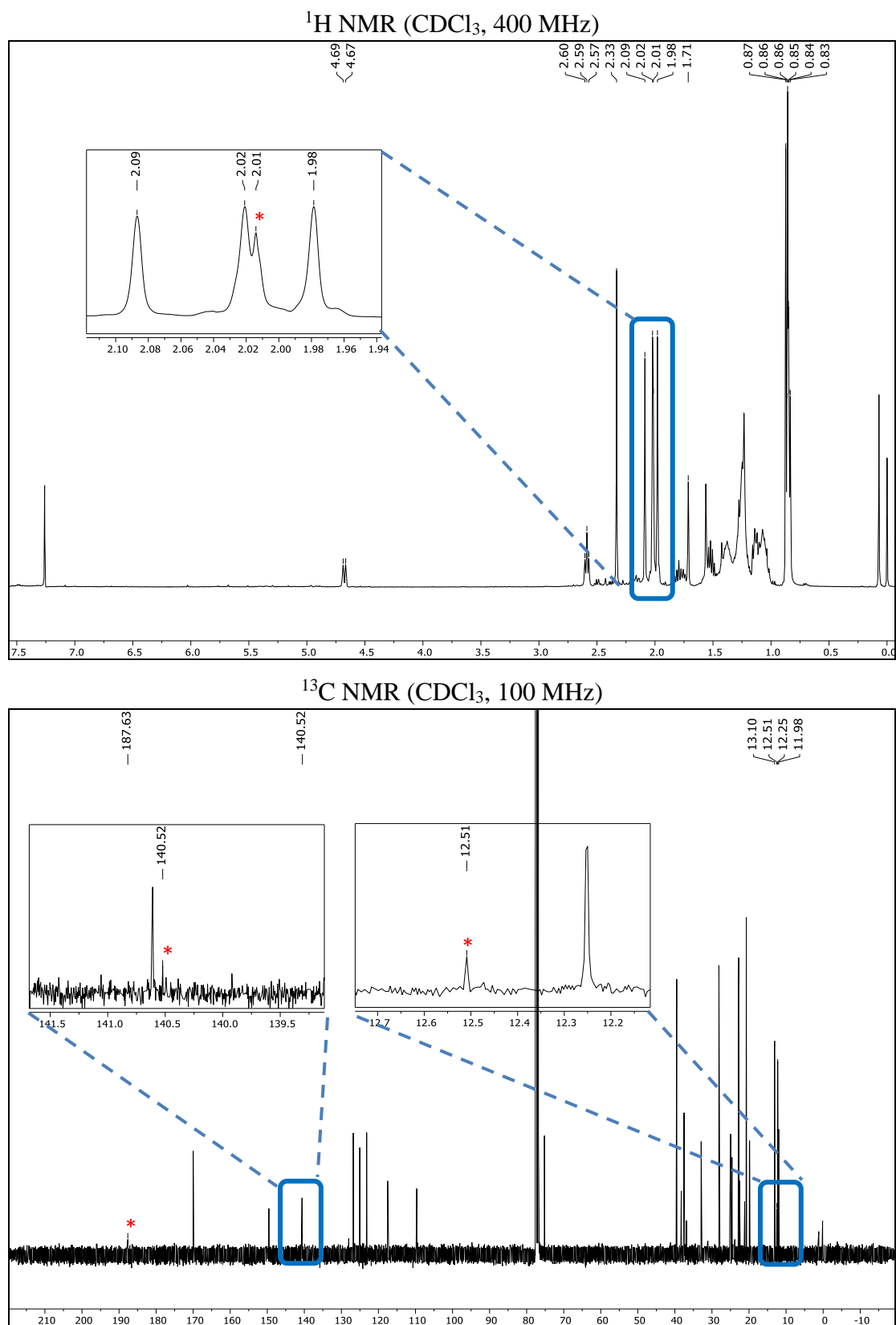
$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)



$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)

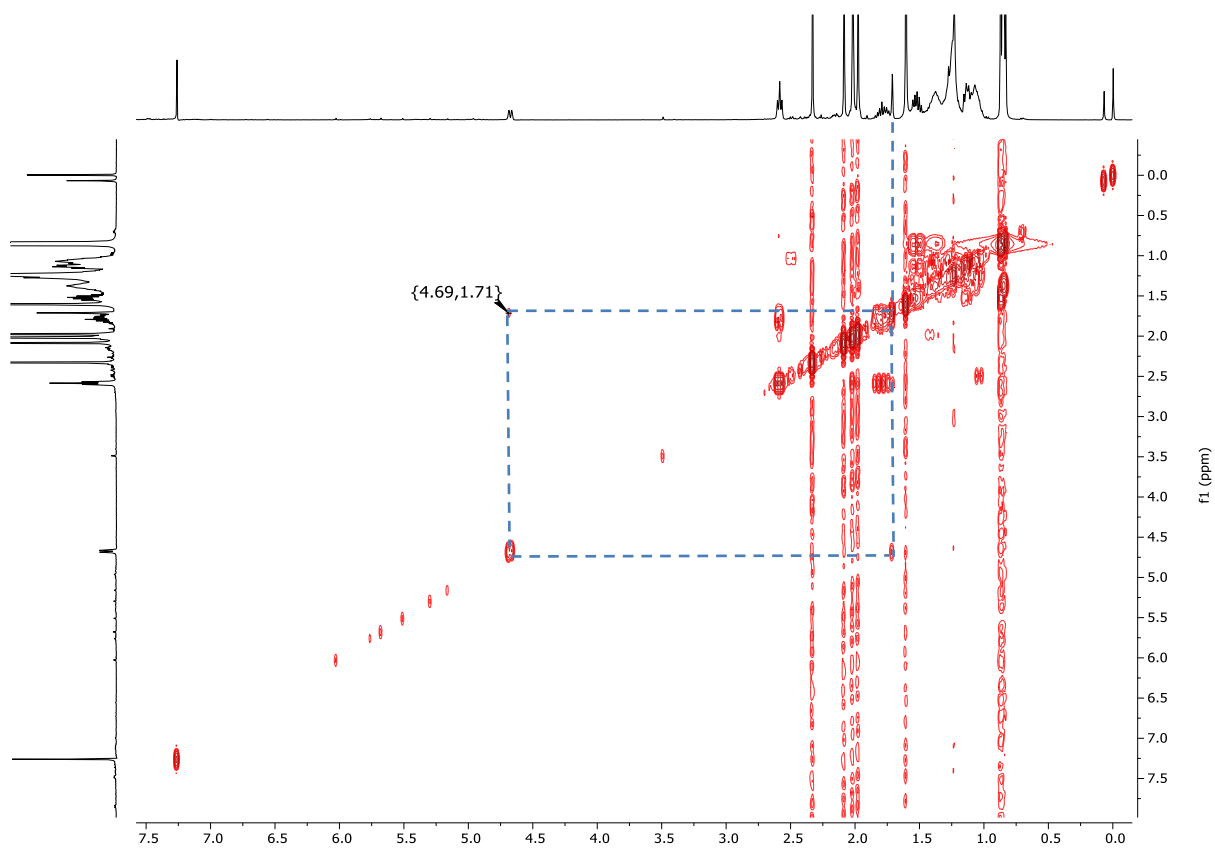


**Figure S11.** Identification of duroquinone **8** and prist-1-ene **7** in non-volatile component (NVC) of isolated mixture from vaped **1**.

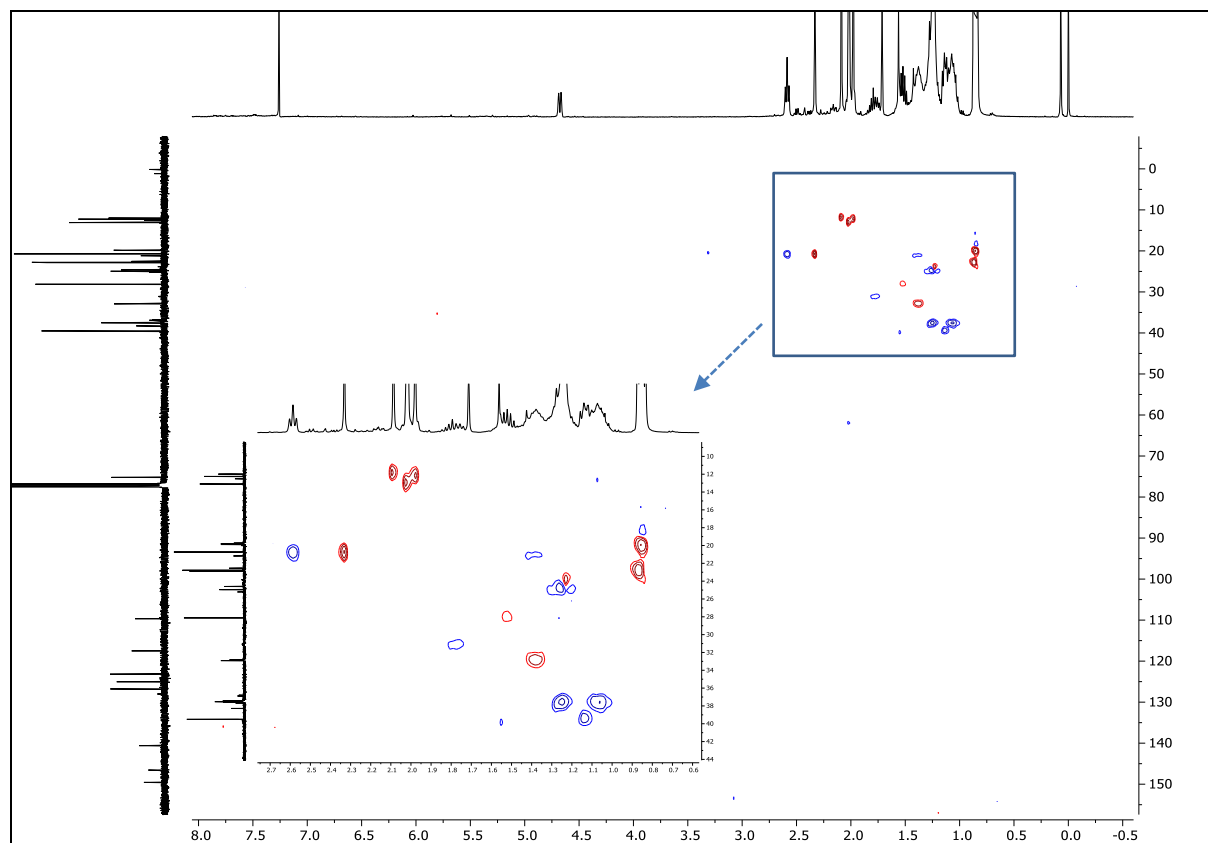


\*: Duroquinone **8**.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$ :2.01 (s, 12H) ppm.  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ )  $\delta$ :187.6, 140.5, 12.5 ppm.

# $^1\text{H}$ COSY

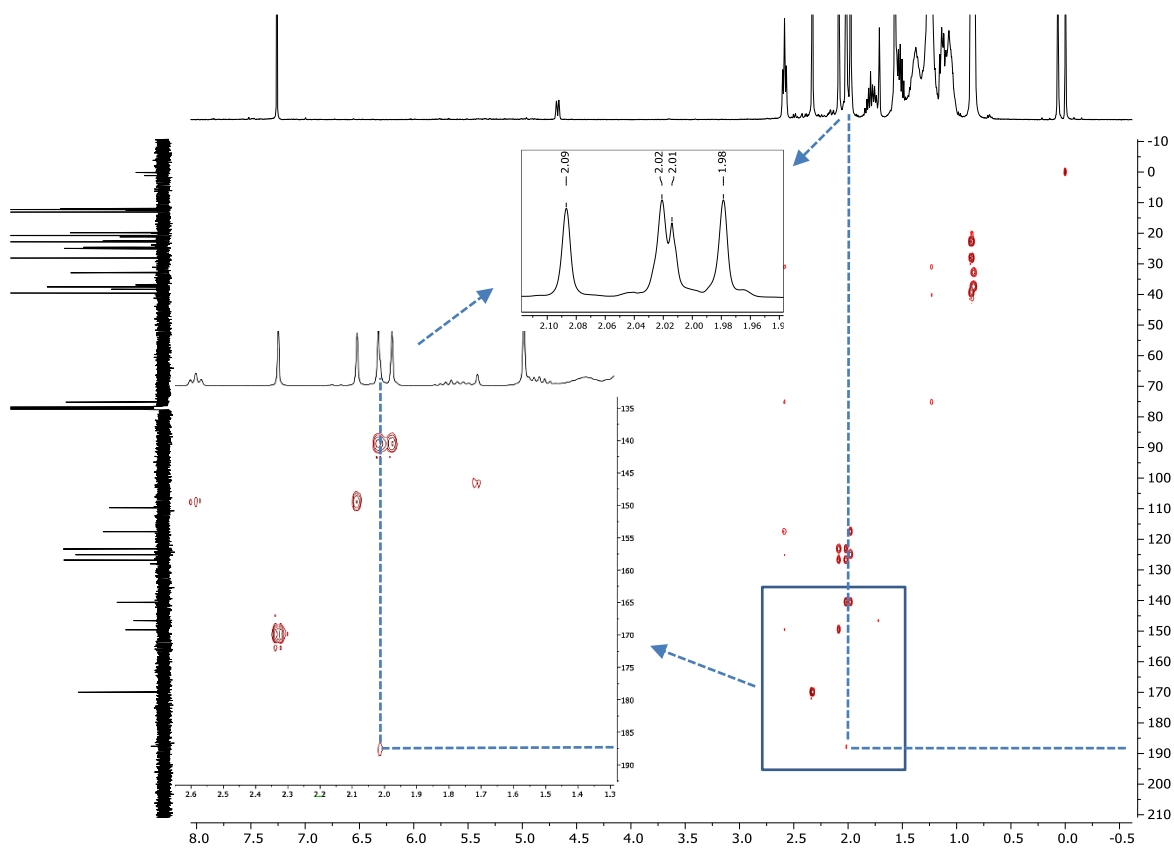


# HSQC

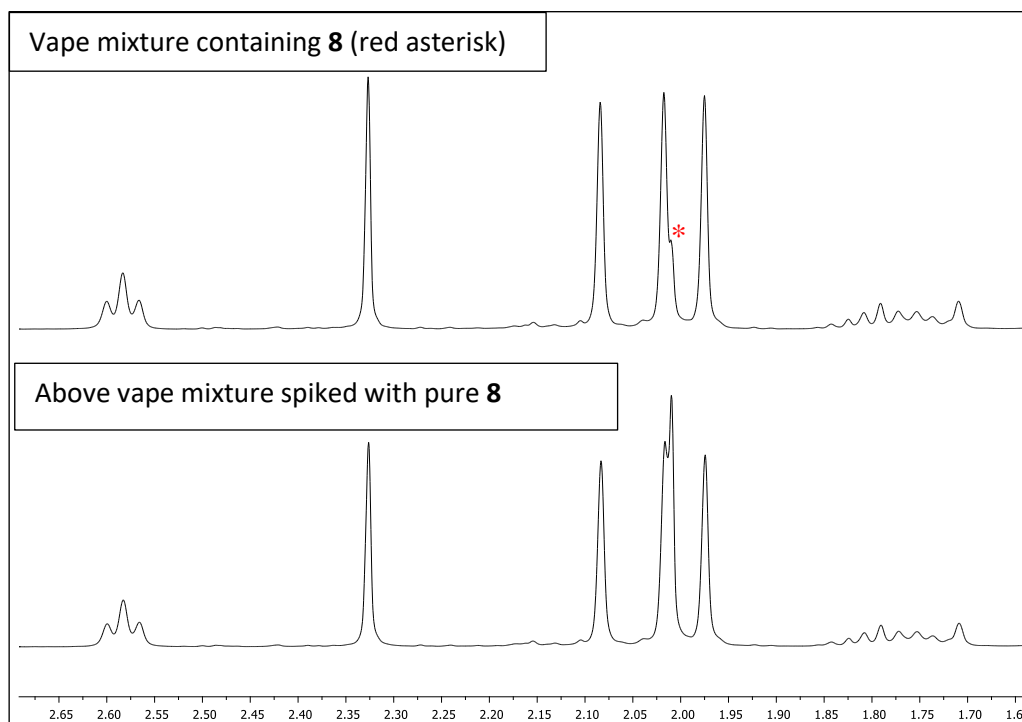




# HMBC

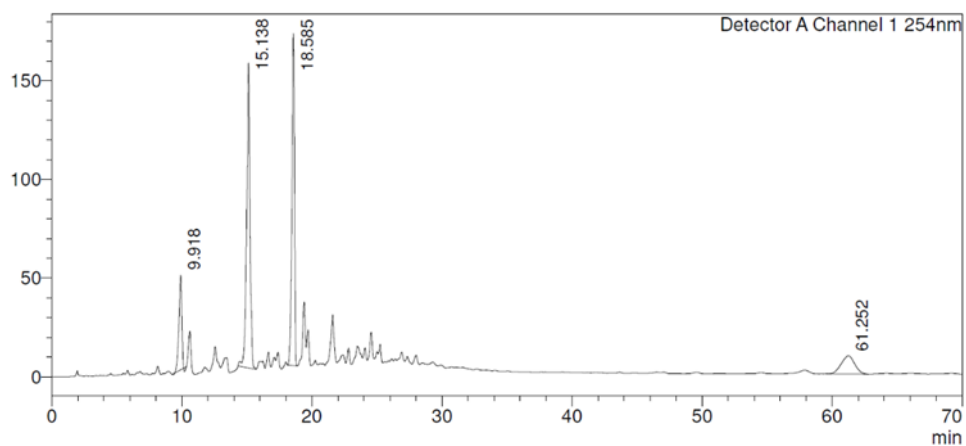


Confirmatory spiking NMR spectra. Top spectrum: vape mixture containing duroquinone **8** (red asterisk). Bottom spectrum: vaped mixture containing duroquinone following spiking with pure duroquinone **8**.

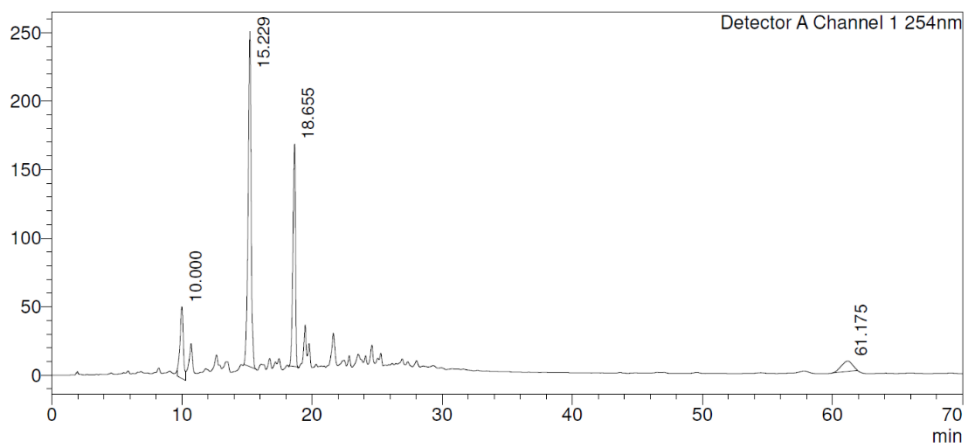


**Figure S12.** HPLC identification of duroquinone **8** in non-volatile component (NVC) of isolated mixture from vaped **1**

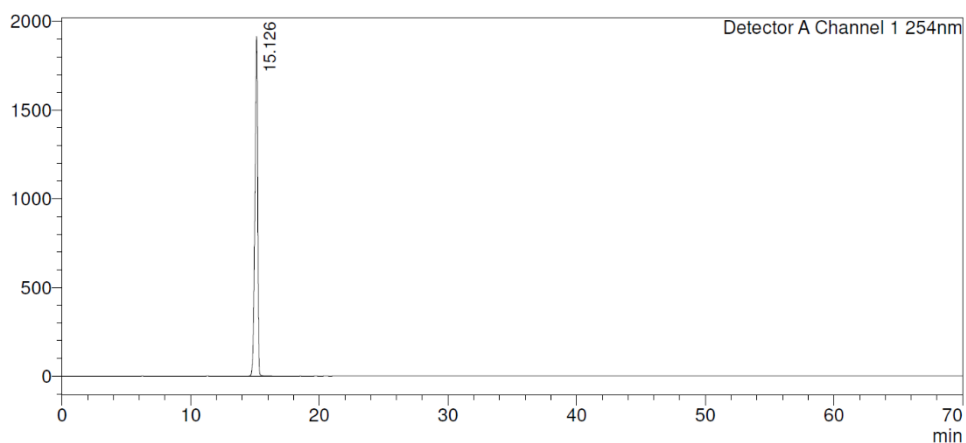
HPLC of NVC from vaped **1** with duroquinone **8** identified at RT of 15.1 min



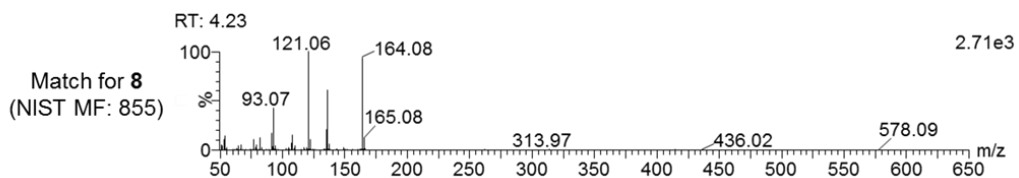
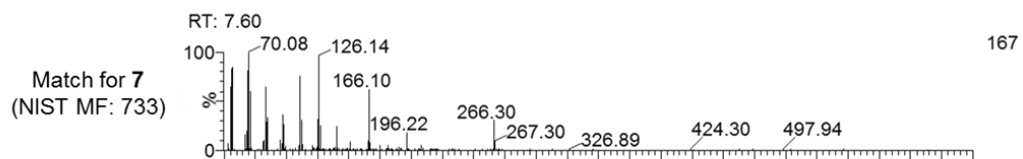
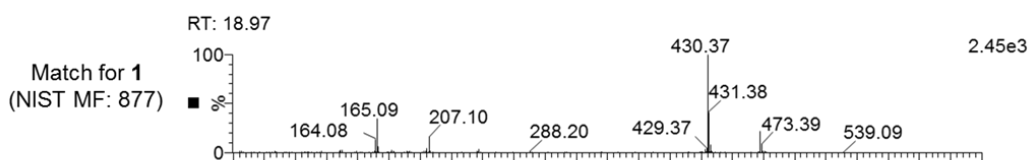
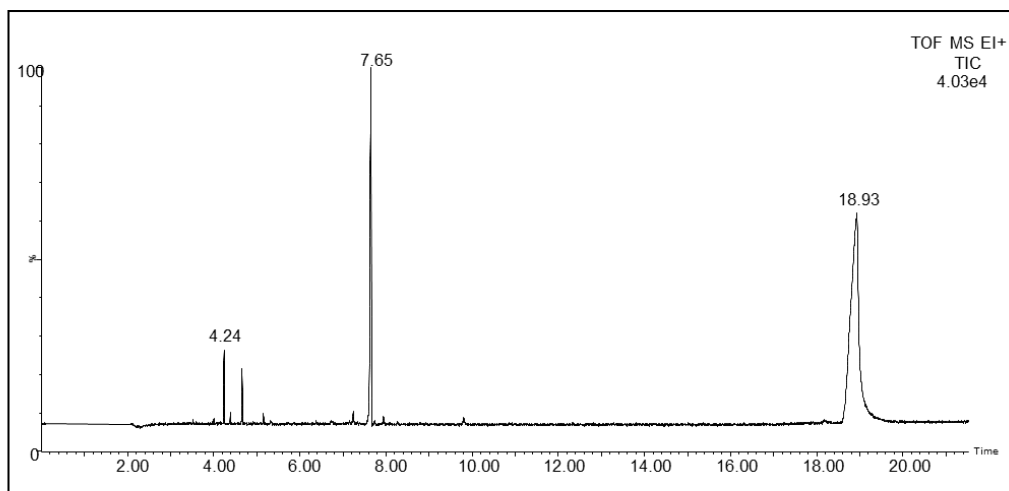
HPLC of NVC from vaped **1** spiked with pure **8**



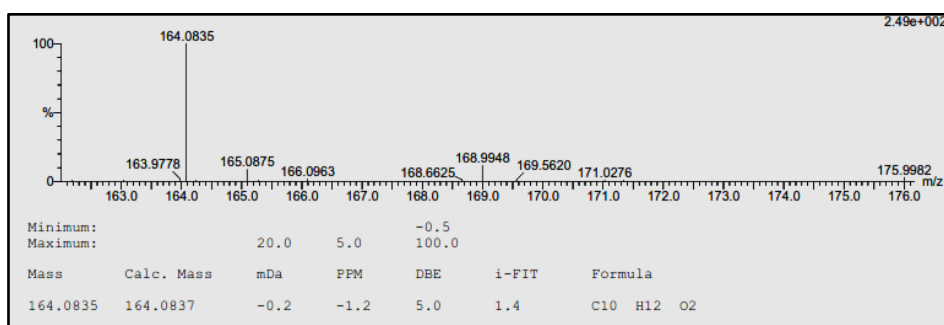
HPLC of pure duroquinone **8** (RT 15.1 min)



**Figure S13.** GC-MS identification of duroquinone **8** in non-volatile component (NVC) of isolated mixture from vaped **1**

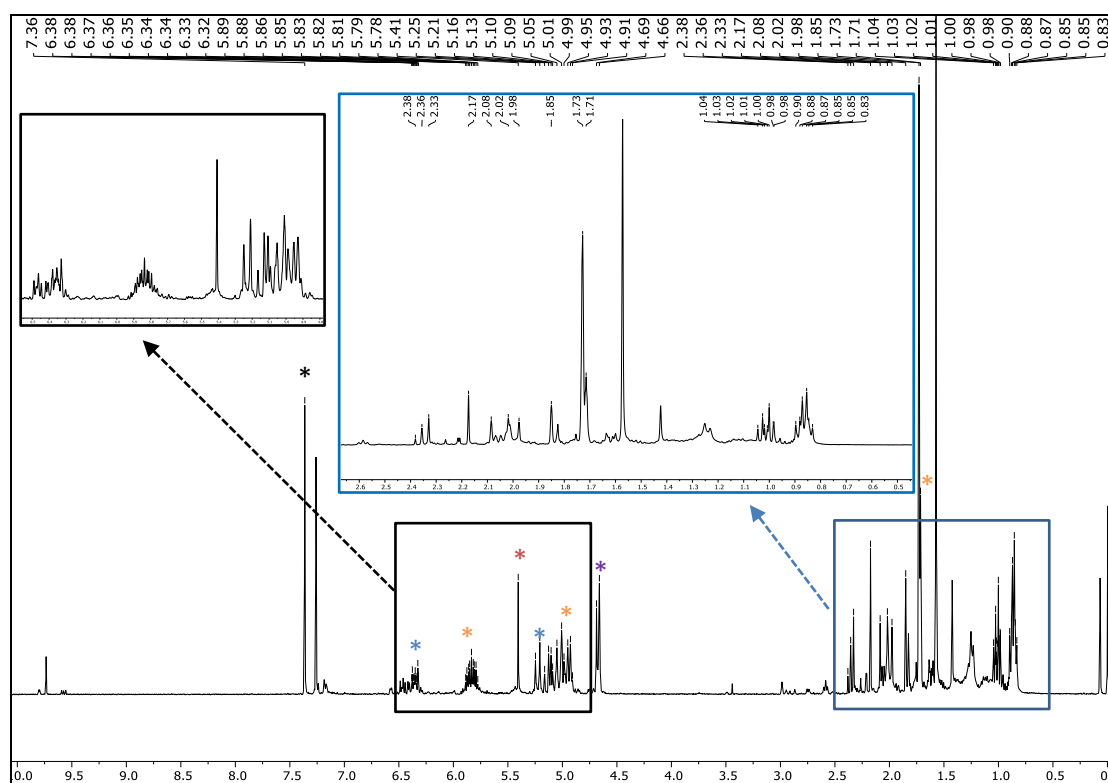


Accurate mass report for duroquinone **8**

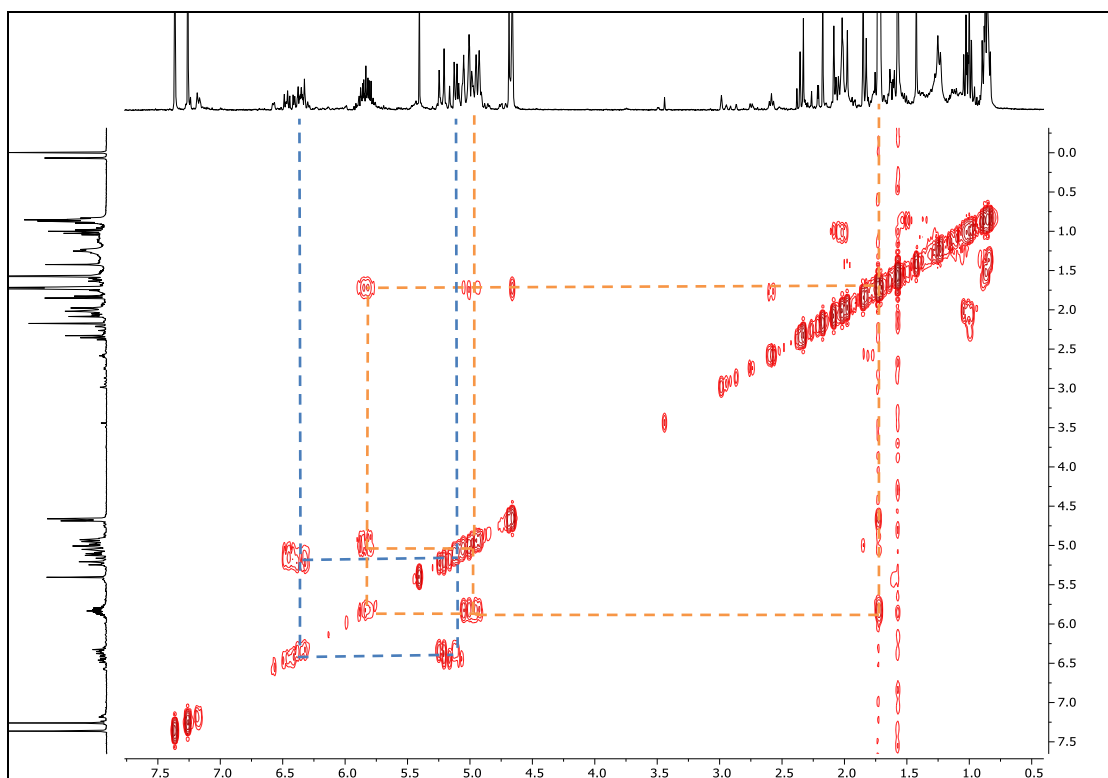


**Figure S14.** Identification of benzene, ethene, propene, butadiene, THF, 2-methylprop-1-ene in volatile component (VC) of isolated mixture from vaped **1**.

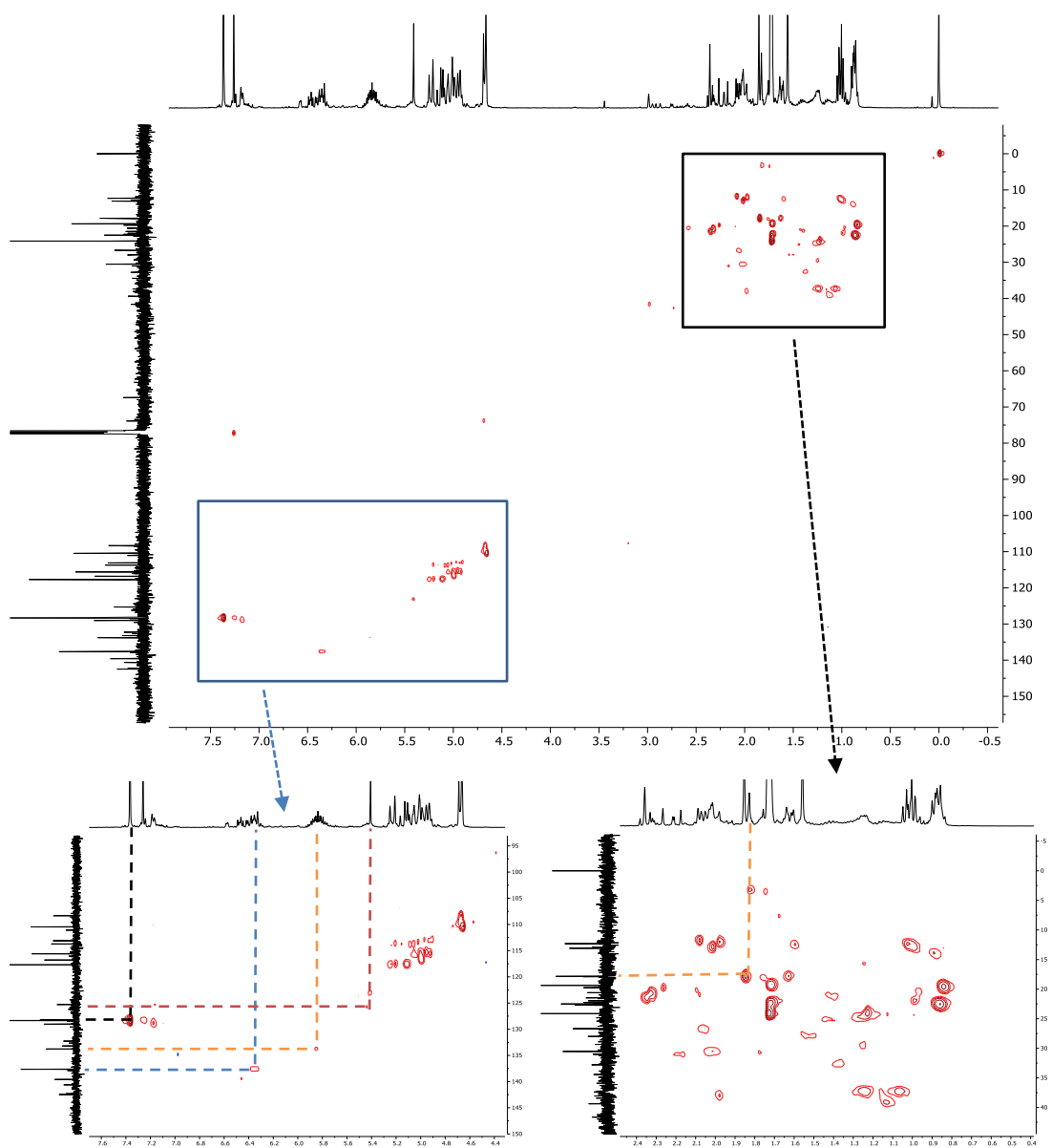
$^1\text{H NMR}$  ( $\text{CDCl}_3$ , 400 MHz)

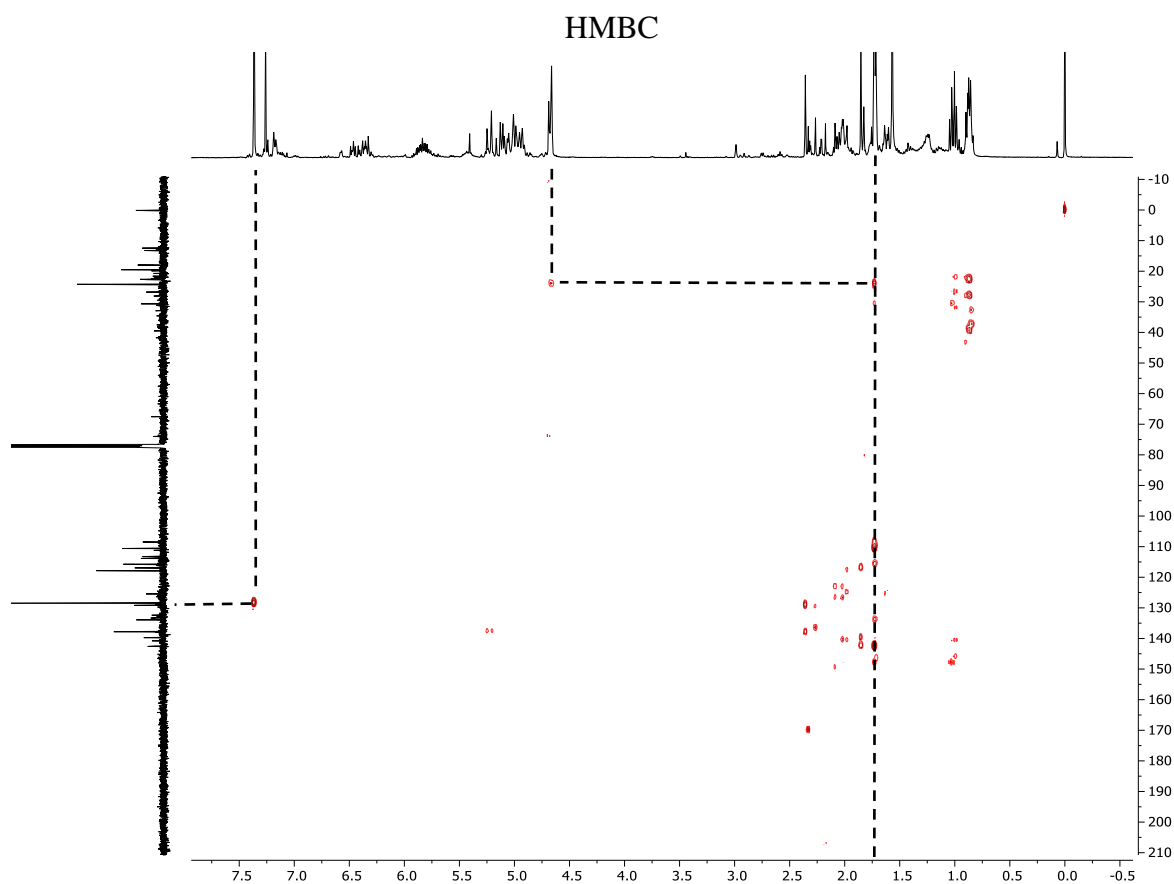


$^1\text{H COSY}$



# HSQC





\* **Benzene:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 7.36 (s) ppm.

\* **Ethene:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 5.41 (s) ppm.

\* **Propene:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 5.93-5.72 (ddq =CH, 1H), 4.97 (dd,  $J = 21.9$  Hz,  $J = 9.8$  Hz,  $\text{CH}_2=$ , 2H), 2.01 (d,  $J = 5.5$  Hz,  $\text{CH}_3$ , 3H) ppm.

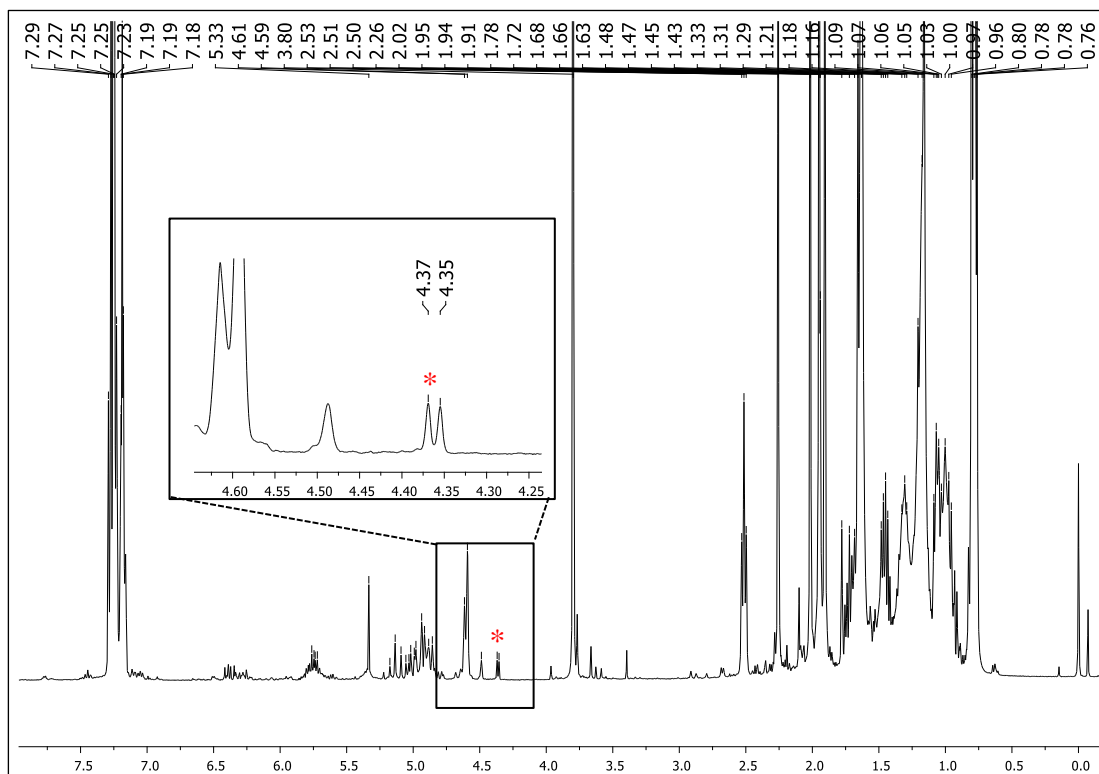
\* **Butadiene:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 6.41-6.29 (m, 2H) ppm, 5.23 (d,  $J = 16.2$  Hz, 2H), 5.12 (d,  $J = 9.5$  Hz, 2H) ppm.

**THF:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 3.76 (m, 4H), 1.85 (m, 4H) ppm.

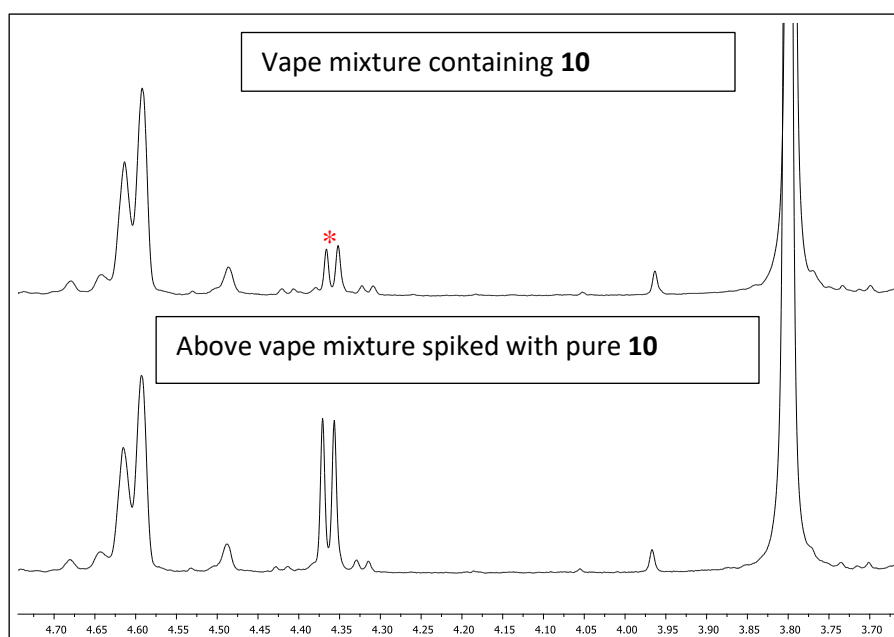
**2-Methylprop-1-ene:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 4.66 (s,  $\text{CH}_2=$ , 2H), 1.71 (s,  $\text{CH}_3$ , 6H) ppm.

**Figure S15.**  $^1\text{H}$  NMR spectra of isolated vaped **1** with benzylamine **9** as trap for ketene.

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz) of vape mixture containing benzylamide **10** (red asterisk)

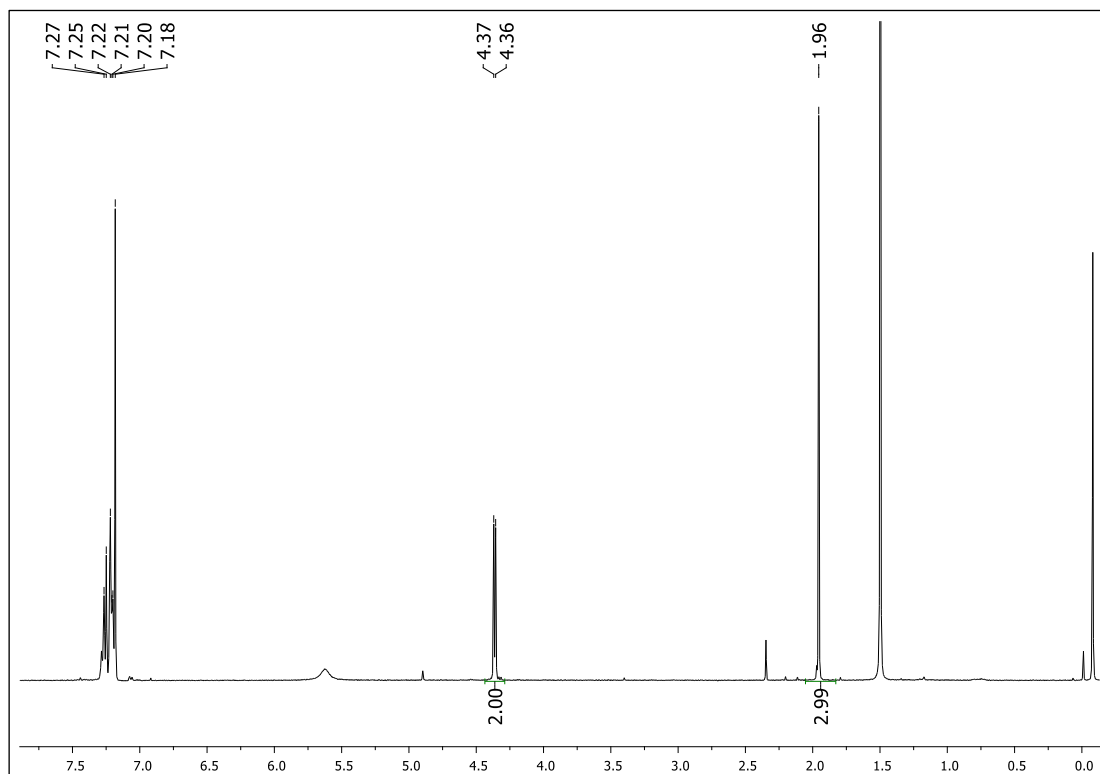


Confirmatory spiking NMR spectra. Top spectrum: vape mixture containing benzylamide **10** (red asterisk). Bottom spectrum: vaped mixture containing benzylamide **10** following spiking with pure benzylamide.



**Figure S16.**  $^1\text{H}$  NMR spectrum of synthesized benzylamide **10**.

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

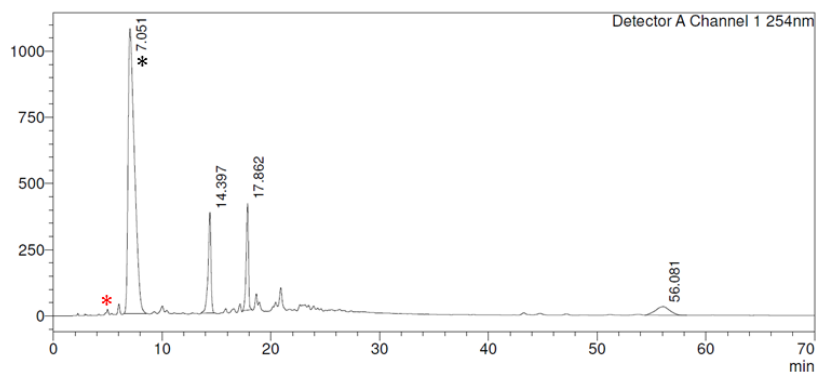


### Procedure for synthesis of benzylamide **10**

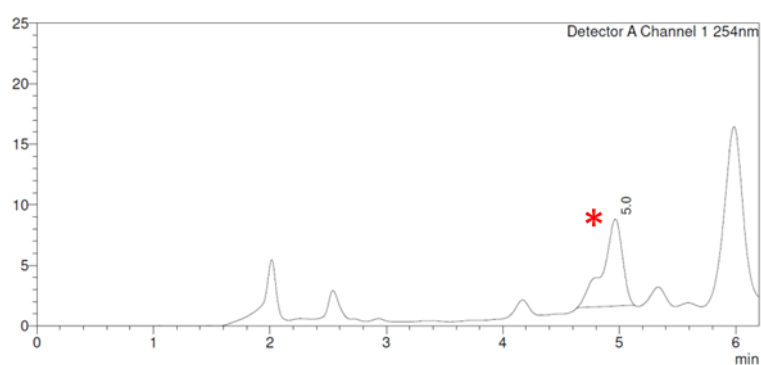
Triethylamine (7.66 mL, 54.9 mmol) was added to a solution of acetylchloride (3.71 mL, 52 mmol) in THF (5 mL), at 0 °C with stirring to in situ form ketene.<sup>†</sup> Benzylamine **9** (5.46 mL, 50 mmol) in THF (15 mL) was added and the reaction mixture stirred for 3 h, warmed to rt, and quenched with saturated  $\text{Na}_2\text{CO}_3$  solution (30 mL). The phases were separated and the aqueous layer was washed with  $\text{CH}_2\text{Cl}_2$  ( $3 \times 20$  mL). The combined organic layers were dried over  $\text{Na}_2\text{SO}_4$  and concentrated. The crude product was recrystallized in  $\text{CH}_2\text{Cl}_2$  / cyclohexane (1:1) to give *N*-benzylacetamide **10** as a solid (4.7 g, 63%, mp 61 °C).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$ : 7.27-7.18 (m, 5H), 5.65 (s, br, 1H), 4.37 (d,  $J = 5.6$  Hz, 2H), 1.96 (s, 3H) ppm.  
<sup>†</sup>Tidwell, T.T, Ketene. *Sci. Synth.* **23**, 19-21 (2006).



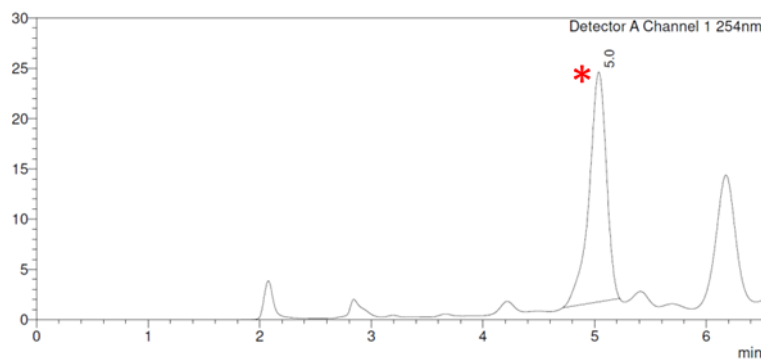
**Figure S17.** HPLC of isolated vaped **1** mixture with benzylamine **9** (black asterisk) as trap for ketene forming benzylamide **10** (red asterisk)



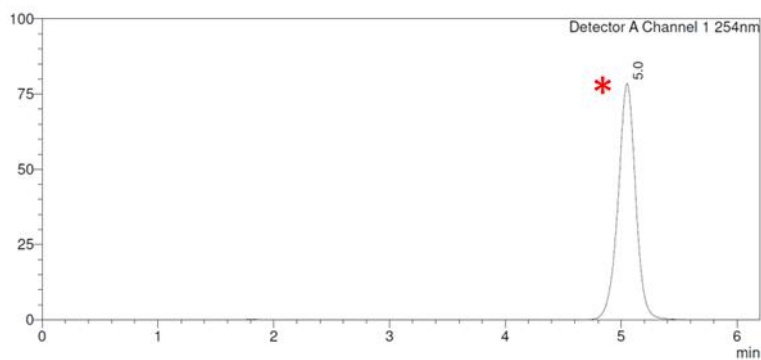
Expanded view of above HPLC trace showing benzylamide **10** (red asterisk)



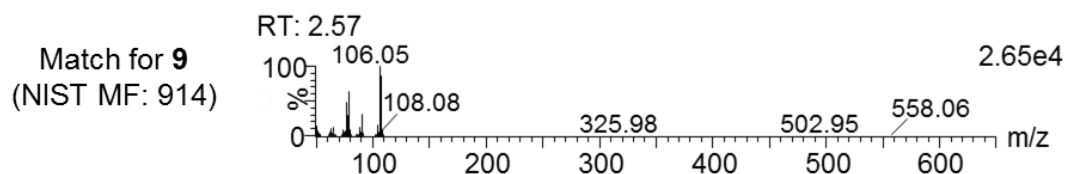
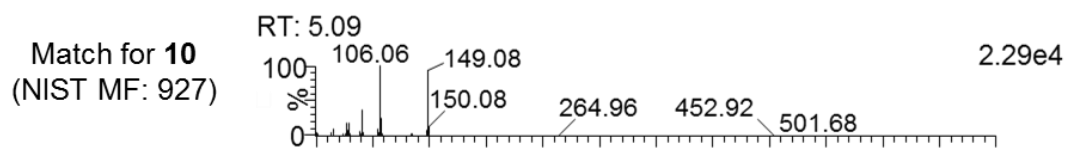
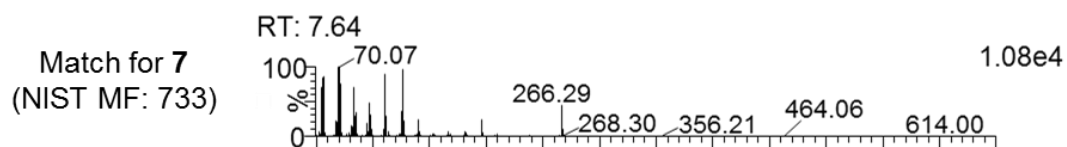
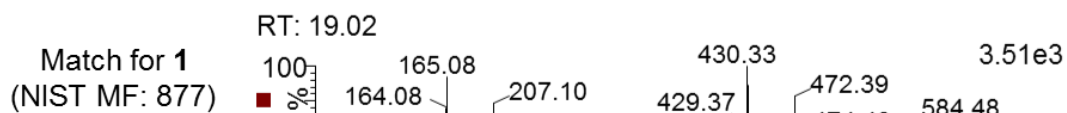
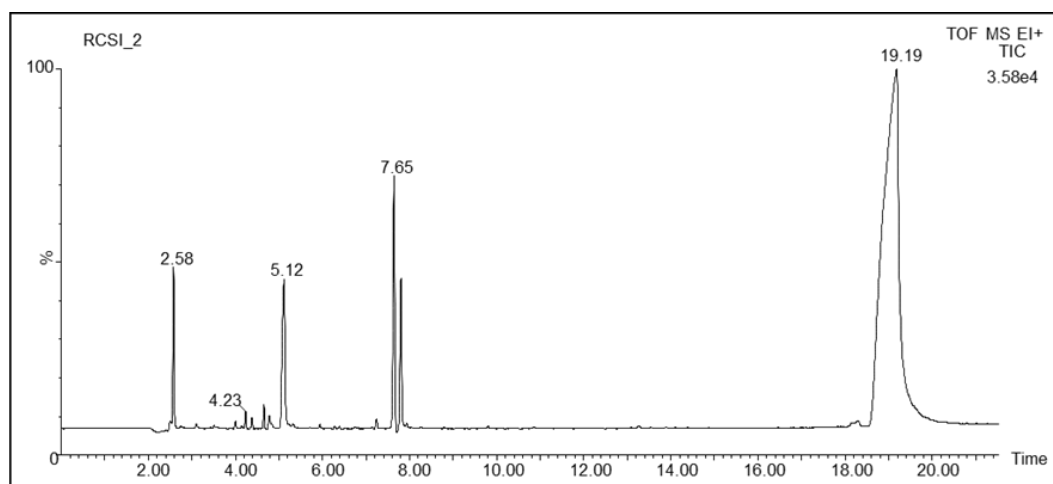
Above HPLC following spiking with pure **10** (red asterisk)



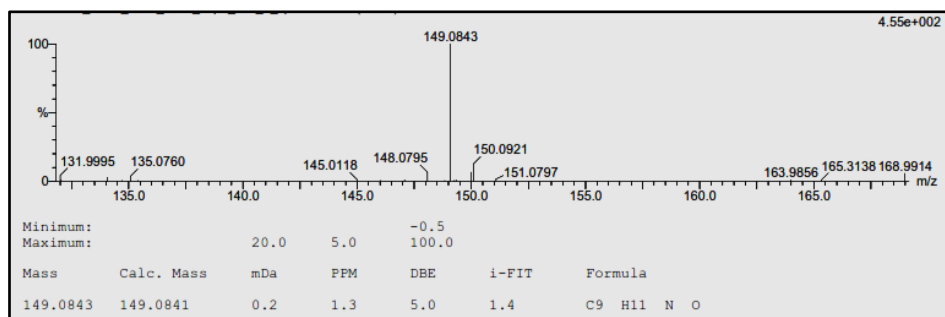
HPLC of pure benzylamide **10**



**Figure S18.** GC-MS spectra of isolated vaped **1** mixture with **9** as trap for ketene

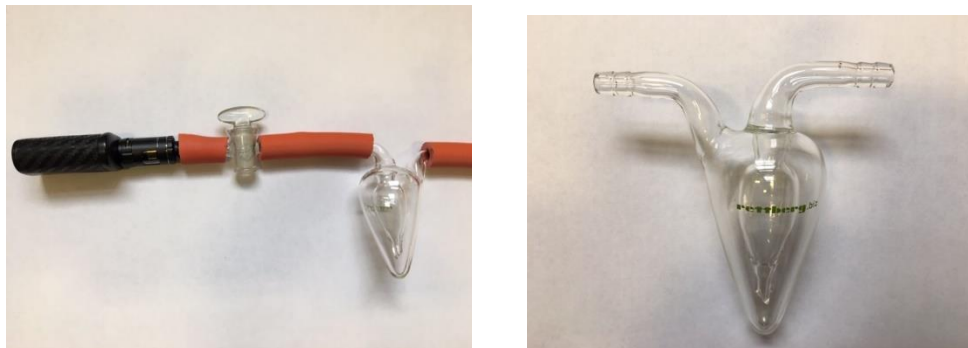


Accurate mass report for benzylamide **10**



**Figure S19.** Photographs of experimental setup for Fig. 11A-C

Apparatus set up as shown in Fig. 11A and C



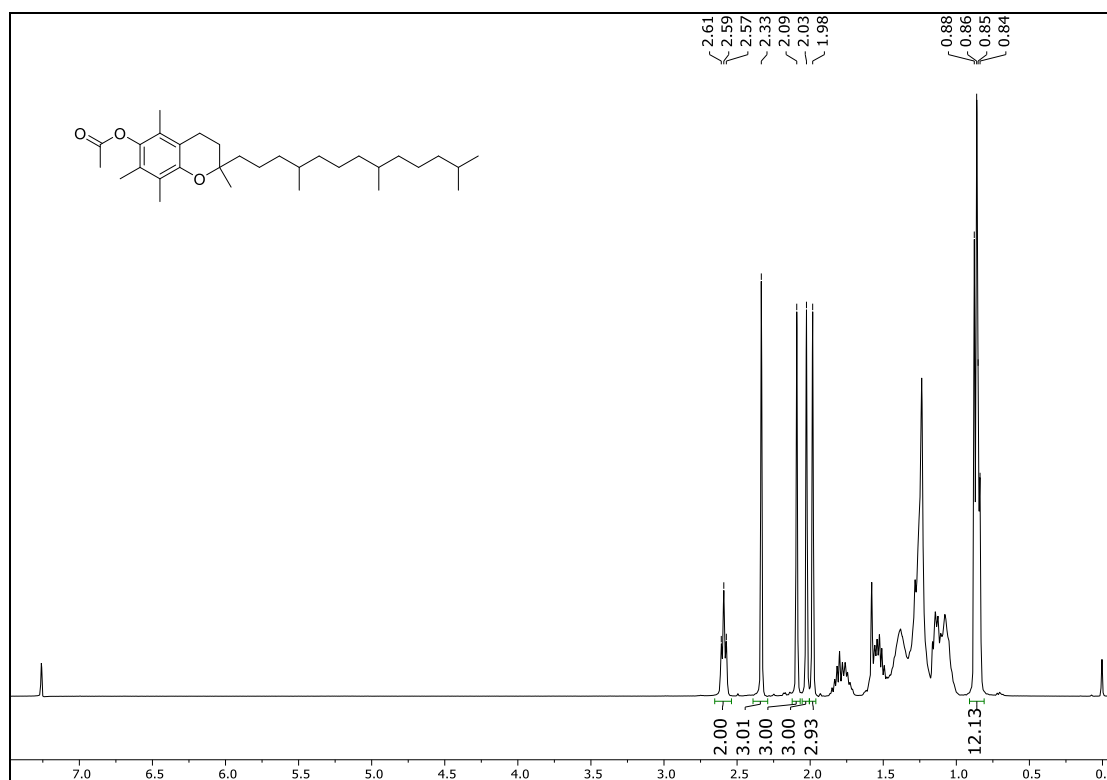
Apparatus set up as shown Fig. 11B



# Appendix NMR spectra of pure 1, 2, 8 and 9

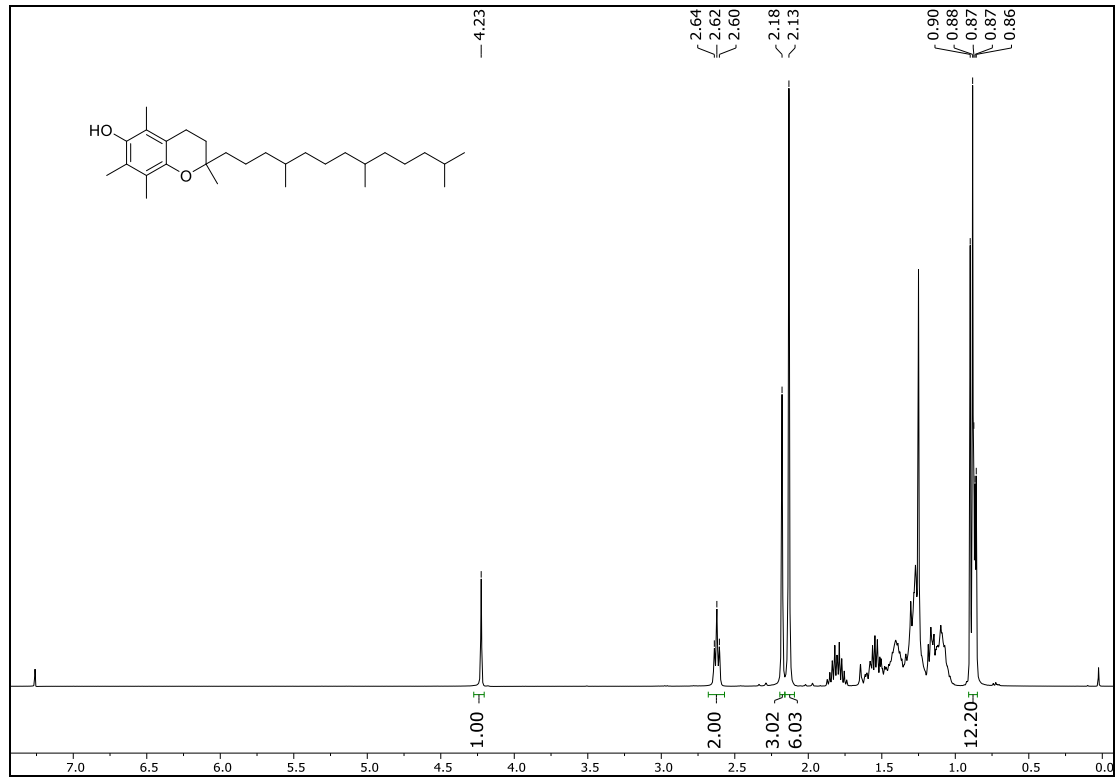
## Vitamin E acetate 1

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)



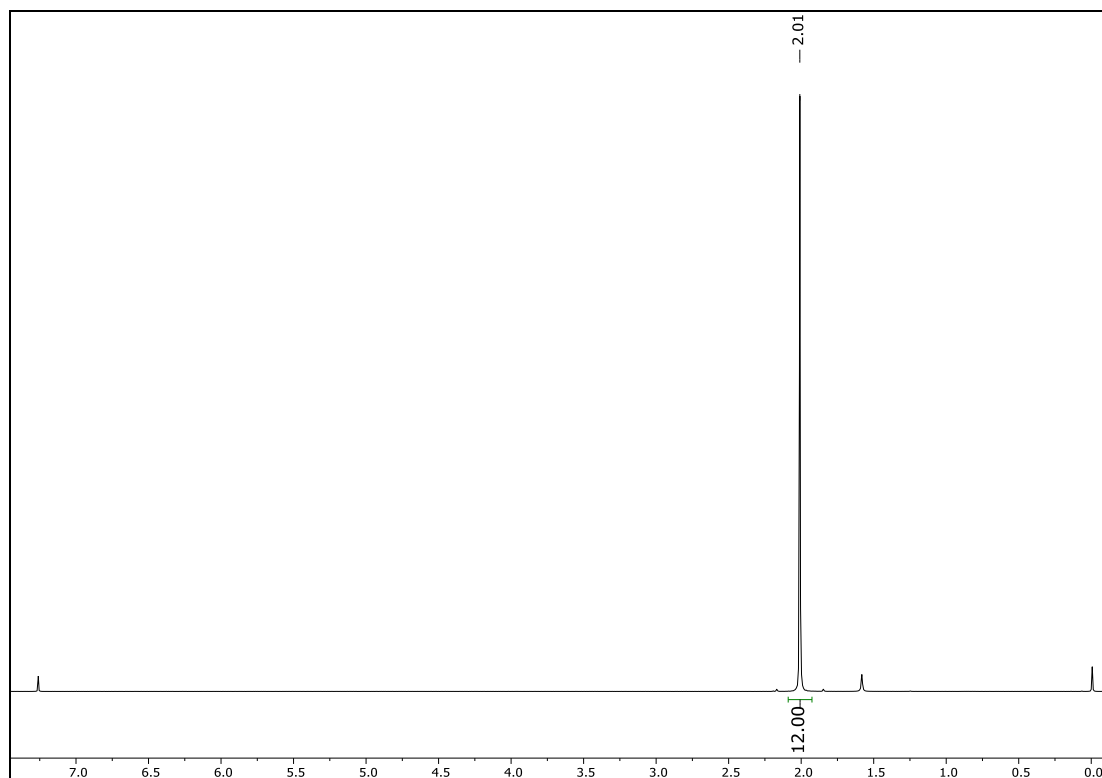
# Vitamin E 2

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

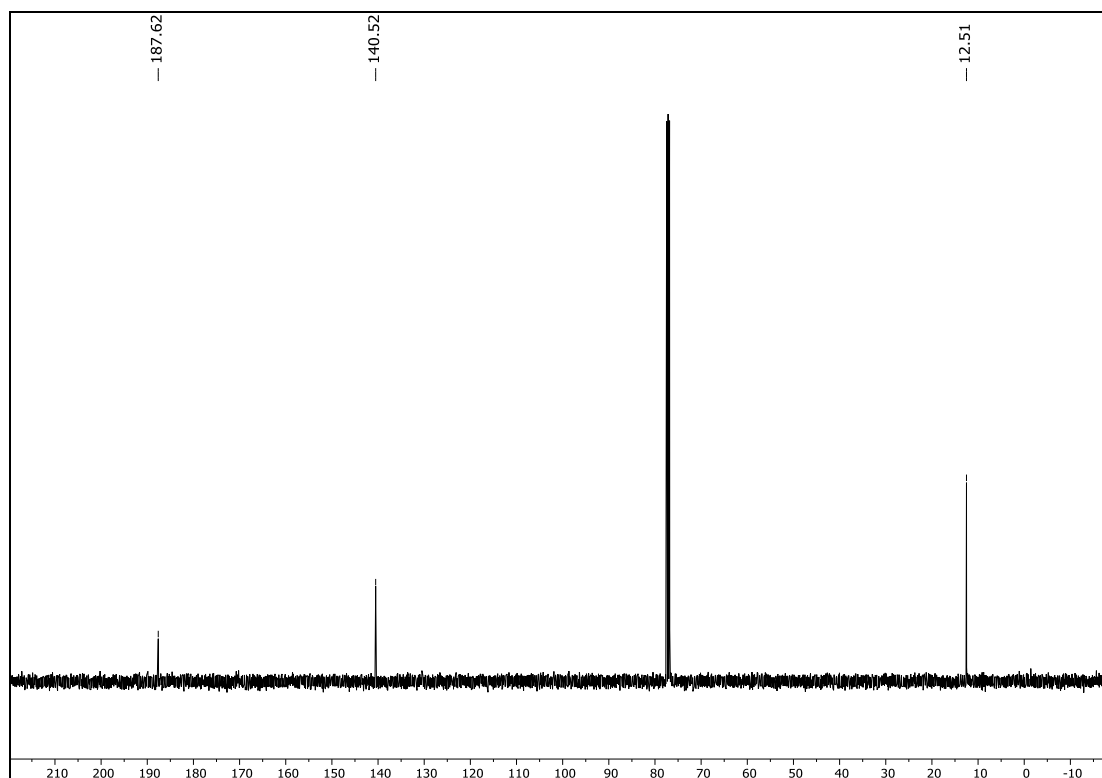


# Duroquinone **8**

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)



$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



# Benzylamine **9**

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

