# SUPPLEMENTARY INFORMATION

## An Activatable NIR Fluorescent Rosol for Selectively Imaging Nitroreductase Activity

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## <sup>1</sup>H- & <sup>13</sup>C-NMR Spectra





NO<sub>2</sub>-Rosol





NH<sub>2</sub>-Rosol





## HRMS Spectra of NO<sub>2</sub>- and NH<sub>2</sub>-Rosol



#### NO<sub>2</sub>-Rosol

#### NH<sub>2</sub>-Rosol



### <u>Spectroscopy</u>

B)

#### Solubility



[NO <sub>2</sub> -Rosol] (µM)	Absorbance (Abs) at 550 nm		
	Trial 1	Trial 2	Trial 3
1	0.034927	0.037524	0.030453
5	0.185423	0.175304	0.169592
10	0.322433	0.349062	0.364299
15	0.545783	0.508381	0.475677
20	0.542123	0.602778	0.552794
25	0.820866	0.717638	0.833552
30	0.886938	0.868470	0.815435
$r^2 = 0.98$			

**Figure S1.** Linear regression analysis and statistical measure between a select set of parameters of NO<sub>2</sub>-Rosol. A) Correlation plot interrelating the resultant absorbance value of NO<sub>2</sub>-Rosol measured at 550 nm to the various concentrations of NO<sub>2</sub>-Rosol that were utilized, which ranged between 1-30  $\mu$ M. B) The accompanying coefficient of determination (r<sup>2</sup>) of the aforementioned analyzed select set of parameters of NO<sub>2</sub>-Rosol. An r<sup>2</sup> value  $\geq$  0.70 corresponds to a strong relationship. The calculated r<sup>2</sup> value (0.98) signifies a very strong relationship exists between the resultant absorbance value measured at 550 nm and the corresponding concentration of NO<sub>2</sub>-Rosol, which suggests that NO<sub>2</sub>-Rosol demonstrates excellent water solubility throughout the range of NO<sub>2</sub>-Rosol concentrations that were utilized.

#### Photostability



**Figure S2.** Photostability of the unactivated (**NO**<sub>2</sub>**-Rosol**, blue) and activated (NH<sub>2</sub>-Rosol, red) form of the NTR-selective activatable molecular probe over 30 min before and after complete reduction of the nitro group of the pendant nitroaromatic moiety of **NO**<sub>2</sub>**-Rosol**, respectively.

### **Kinetic Analysis**



**Figure S3.** Determination of initial kinetic velocities. (a) Various concentrations of **NO<sub>2</sub>-Rosol** were activated by nitroreductase, wherein their fluorescence intensity at the maximum emission wavelength of 705 nm was monitored over time. A) The slope of the curve from the first 90 seconds was converted to velocity using B) a standard curve of the activated form of **NO<sub>2</sub>-Rosol** (i.e., NH<sub>2</sub>-Rosol) fluorescence.  $r^2 = 0.99$ .



Figure S4. Lineweaver-Burk plot for evaluating the kinetic parameters of nitroreductase activation of  $NO_2$ -Rosol. [S] = concentration of the  $NO_2$ -Rosol. V = velocity as determined from above.

## LC-HRMS Chromatograms & Spectra of Reaction between NTR with NO2-Rosol



**Figure S5.** A) LC-HRMS extracted ion count chromatograms and mass spectra of synthesized **NO<sub>2</sub>-Rosol** and NH<sub>2</sub>-Rosol used as standards. B) LC-HRMS extracted ion count chromatograms and mass spectra of the reaction between NTR and **NO<sub>2</sub>-Rosol** after 30 minutes.



**Figure S6.** Cell viability assay results from applying **NO**<sub>2</sub>**-Rosol** at relevant concentrations (200  $\mu$ M - 10  $\mu$ M) to U251 cells. Untreated cells were used as a negative control and cells treated with 15% DMSO as a positive control. Cells were stained with a live cell stain, Calcein-AM, wherein the fluorescence intensity was measured at 516 nm ( $\lambda_{ex}$  = 494 nm). N = 6, \*\*\*\**P* < 0.0001