

**Base-Mediated Meerwein–Ponndorf–Verley Reduction
of Aromatic and Heterocyclic Ketones**

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Supporting Information – Table of Contents

Materials and Methods	S2
Experimental Procedures	S4
A. Syntheses of Ketone Substrates.....	S4
B. Synthesis of Alcohol Reductant 5.....	S5
C. Synthesis of Enantioenriched Alcohol Reductant (<i>R</i>)-5.....	S6
D. Initial Survey of Reaction Conditions and Relevant Control Experiments.....	S7
E. Trace Metal Analysis of K₃PO₄.....	S9
F. Scope of Methodology	S12
G. Gram-Scale Reduction	S19
H. Stereospecific Reduction	S20
I. Verification of Enantiopurity.....	S21
References.....	S24
¹H NMR Spectra	S26
¹³C NMR Spectra... ..	S45

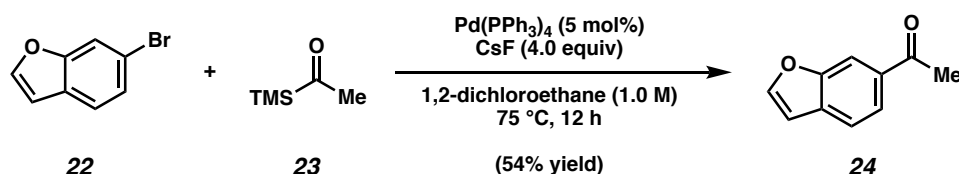
Materials and Methods. Unless stated otherwise, reactions were conducted in flame-dried glassware under an atmosphere of nitrogen or argon and commercially obtained reagents were used as received. Not-commercially available ketone substrates were synthesized following protocols specified in Section A in the Experimental Procedures. Alcohols **5** and (*R*)-**5** were synthesized following protocols specified in Section B and C in the Experimental Procedures, respectively. 1,2-Dichloroethane, 1,4-dioxane, and isopropanol were obtained from Fischer Scientific and purified by distillation. 3-Pentanol (**3**) and 1-phenylethanol (**4**) were obtained from Sigma-Aldrich and purified by distillation. Prior to use, 1,4-dioxane, isopropanol, 3-pentanol (**3**) and 1-phenylethanol (**4**) were degassed by sparging with N₂ for 1 h. Ketone **1**¹ and **36**² were prepared according to literature procedures. **22**, **28**, **29**, **30**, **31**, and **34** were obtained from Sigma-Aldrich. **26**, **21**, **27**, **32**, **35**, and **37** was obtained from Combi-Blocks. Ketone **26** was obtained from Oxchem. Ketone **33** was obtained from Alfa Aesar. Pd(PPh₃)₄ (99%) was obtained from Strem Chemicals. Acetyltrimethylsilane (**23**) (97%) was obtained from Sigma-Aldrich. Cesium fluoride (99%+) was obtained from Strem Chemicals. Potassium phosphate (K₃PO₄) was obtained from Acros. Reaction temperatures were controlled using an IKAmag temperature modulator, and unless stated otherwise, reactions were performed at room temperature (approximately 23 °C). Thin-layer chromatography (TLC) was conducted with EMD gel 60 F254 pre-coated plates (0.25 mm for analytical chromatography and 0.50 mm for preparative chromatography) and visualized using a combination of UV, anisaldehyde, iodine, and potassium permanganate staining techniques. Silicycle Siliaflash P60 (particle size 0.040–0.063 mm) was used for flash column chromatography. ¹H NMR spectra were recorded on Bruker spectrometers (at 300, 400, 500, and 600 MHz) and are reported relative to residual solvent signals. Data for ¹H NMR spectra are reported as follows: chemical shift (δ ppm), multiplicity, coupling constant (Hz), integration. Data for ¹³C NMR are reported in terms of chemical shift (at 75 and 125 MHz). IR spectra were recorded on a Perkin-Elmer UATR Two FT-IR spectrometer and are reported in terms of frequency absorption (cm⁻¹). DART-MS spectra were collected on a Thermo Exactive Plus MSD (Thermo Scientific) equipped with an ID-CUBE ion source and a Vapor Interface (IonSense Inc.). Both the source and MSD were controlled by Excalibur software v. 3.0. The analyte was spotted onto OpenSpot sampling cards (IonSense Inc.) using CHCl₃ or CH₂Cl₂ as the solvent. Ionization was accomplished using UHP He plasma with no additional ionization agents. The mass calibration was carried out using Pierce LTQ Velos ESI (+) and (-) Ion

calibration solutions (Thermo Fisher Scientific). Optical rotations were measured with a Rudolf Autopol III Automatic Polarimeter. Trace metal analysis was determined by inductively coupled plasma mass spectrometry on an Agilent 8800 Triple Quadrupole ICP-MS instrument. The level of all analytes of interest was determined in MS/MS mode, measured using He in the collision/reaction cell using an environmental calibration standard (elements not included in this standard: B, Ti, Rb, Ru, Rh, Pd, Ir, and Pt). The quantification was done using the ICP-MS MassHunter WorkStation v4.3, through the QuickScan acquisition. Nitric acid was obtained from Fisher Scientific (A467500). Determination of enantiopurity was carried out on a Mettler Toledo SFC (supercritical fluid chromatography) or Agilent HPLC (high performance liquid chromatography) using Daicel ChiralPak IC-3 and Daicel ChiralPak OD-H columns. Data for SFC and HPLC spectra are reported in enantiomeric excess (ee). For SFC and HPLC chromatograms see Section I of Experimental Procedures.

Experimental Procedures

A. Syntheses of Ketone Substrates

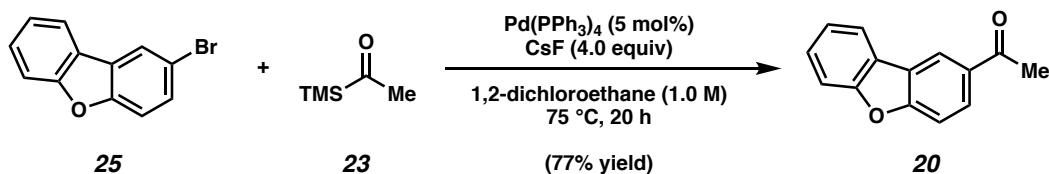
Representative Procedure for the Synthesis of Ketone Substrates (synthesis of ketone **24 is used as an example).**



A flame-dried 1-dram vial was charged 6-bromobenzofuran (**22**) (130 mg, 0.660 mmol, 1.00 equiv) and a magnetic stir bar. In the glove box, CsF (401 mg, 2.64 mmol, 4.00 equiv) and Pd(PPh₃)₄ (38.1 mg, 0.0330 mmol, 0.0500 equiv) were added to the vial. The vessel was removed from the glove box and placed under an atmosphere of N₂ on the bench. Distilled 1,2-dichloroethane (0.700 mL, 1.00 M) and silane **23** (189 μL, 1.32 mmol, 2.0 equiv) were added and the vial was sealed with a Teflon-lined screw cap. The heterogeneous mixture was heated to 75 °C for 12 h. After cooling to 23 °C, the mixture was diluted with hexanes (0.5 mL), filtered over a plug of silica gel (1.00 cm OD x 5.00 cm, 10 mL EtOAc eluent), and the volatiles were removed under reduced pressure. The crude residue was purified by flash chromatography (19:1 Hexanes:EtOAc → 14:1 Hexanes:EtOAc) to yield ketone **24** (57.0 mg, 54% yield) as a yellow oil. Ketone **24**: R_f 0.42 (5:1 Hexanes:EtOAc); ¹H NMR (600 MHz, CDCl₃): δ 8.13 (s, 1H), 7.89 (dd, *J* = 8.30, 1.38 Hz, 1H), 7.79 (d, *J* = 2.17 Hz, 1H), 7.66 (d, *J* = 8.30 Hz, 1H), 6.87–6.89 (m, 1H), 2.67 (s, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 197.8, 154.8, 148.4, 134.0, 132.0, 123.3, 121.2, 112.0, 107.0, 27.0; IR (film): 3118, 3003, 1673, 1425, 1271 cm⁻¹; HRMS-APCI (*m/z*) [M+NH₄]⁺ calcd for C₁₀H₁₂O₂N⁺, 178.08626; found 178.08536.

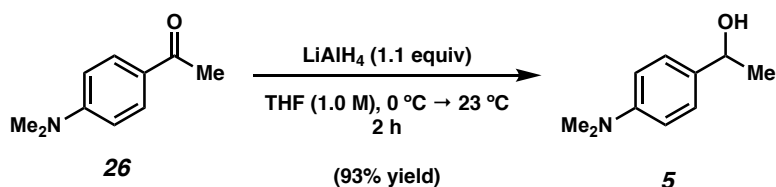
Note: Supporting information for the synthesis of ketone **36** has previously been reported.² The synthesis of the remaining substrate, **20**, is as follows:

Any modifications of the conditions shown in the representative procedure above are specified in the following scheme.

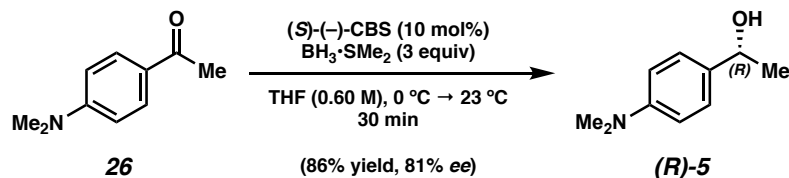


Ketone 20. Purification by flash chromatography (49:1 Hexanes:EtOAc) generated ketone **20** (263 mg, 77% yield) as a white solid. Ketone **20**: R_f 0.33 (9:1 Hexanes:EtOAc). Spectral data match those previously reported.³

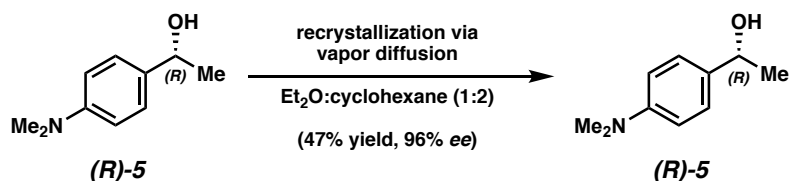
B. Synthesis of Alcohol Reductant 5



To a flame-dried flask equipped with a magnetic stir bar was added LiAlH_4 (2.56 g, 67.4 mmol, 1.10 equiv) in a glovebox. The flask was removed from the glovebox, THF (61.0 mL) was added, and the solution was cooled to 0 °C. To the solution was then added ketone **26** (2.00 g, 61.5 mmol each, 0.200 equiv) in 5 aliquots over 25 min. The reaction was then warmed to 23 °C. After stirring for 2 h, the reaction was cooled to 0 °C and quenched by the sequential addition of deionized water (5 mL), 10% aq. NaOH (7 mL), MeOH (20 mL), and deionized water (10 mL). The mixture was then warmed to 23 °C and stirred for 30 min. The mixture was then filtered over a pad of celite (100 mL EtOAc eluent). The resulting organic layer was dried over Na_2SO_4 , filtered, and the volatiles were removed under reduced pressure. The crude residue was purified by flash chromatography (5:1 Hexanes:EtOAc → 3:1 Hexanes:EtOAc) to yield alcohol **5** (9.42 g, 93% yield) as a white solid. Alcohol **5**: R_f 0.33 (3:1 Hexanes:EtOAc). Spectral data match those previously reported.⁴

C. Synthesis of Enantioenriched Alcohol Reductant (*R*)-5

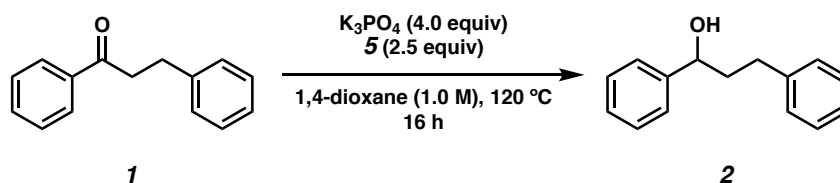
To a flame-dried flask equipped with a magnetic stir bar was added (*S*)-(-)-CBS catalyst (170 mg, 0.613 mmol, 0.100 equiv). The flask was removed from the glovebox and THF (6.13 mL) was added. Next, **26** (1.00 g, 6.13 mmol, 1.00 equiv) in a solution of THF (2.50 mL) was added to the reaction flask, which was then stirred to give a clear homogeneous solution and cooled to 0 °C. Subsequently, $\text{BH}_3\cdot\text{SMe}_2$ (1.70 mL, 18.4 mmol, 3.00 equiv) was added (1 drop/2 sec) over 7.50 min. The reaction was stirred at 0 °C for 2 min, then warmed to 23 °C. After stirring for 30 min, the reaction was cooled to 0 °C and quenched by the dropwise addition of methanol (20 mL) and water (20 mL) and diluted with Et_2O (50 mL). The layers were separated and the aqueous layer was extracted with Et_2O (3 x 50 mL). The combined organic layers were washed with sat. aq. NH_4Cl (80 mL), sat. aq. NaHCO_3 (80 mL), and brine (80 mL). The organic layer was then dried over Na_2SO_4 , filtered, and the volatiles were removed under reduced pressure. The crude residue was purified by flash chromatography (10:1 Hexanes:EtOAc \rightarrow 3:1 Hexanes:EtOAc) to yield alcohol (*R*)-5 (871 mg, 86% yield, 81% ee) as a white solid. Alcohol (*R*)-5: R_f 0.33 (3:1 Hexanes:EtOAc). The spectral data match those previously reported in the literature for *rac*-5.⁴ The SFC data match those reported in the Experimental Procedures Section G.



A solution of (*R*)-5 (25.0 mg, 0.151 mmol, 1.00 equiv) in Et_2O (1.00 mL) was filtered through a 0.45 μm Millipore Millex PTFE filter into a 1-dram vial. The vial was then placed within a 20 mL scintillation vial containing cyclohexane (2.00 mL). The scintillation vial was sealed and allowed to stand at 23 °C for 24 h, which led to the formation of white crystals. This vapor

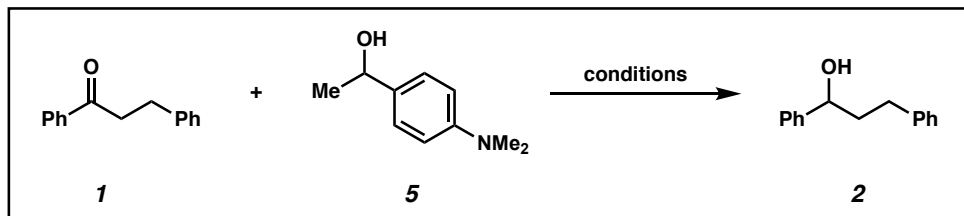
diffusion crystallization process was repeated three times to lead to the recovery of alcohol (**R**)-**5** (11.8 mg, 47% yield, 96% ee) as a white crystalline solid. $[\alpha]_D^{23.1} = +51.6$ ($c = 1.00$, CHCl_3). The spectral data match those previously reported in the literature for *rac*-**5**.⁴ The major enantiomer product was assigned by comparison to published $[\alpha]_D$ values for (**R**)-**5**.⁵

D. Survey of Reaction Conditions and Relevant Control Experiments



Representative Procedure for Base-Mediated MPV Reduction from Table S1 (reduction of ketone 1 with alcohol 5 is used as an example). A 1-dram vial was charged with anhydrous powdered K_3PO_4 (85.0 mg, 0.400 mmol, 4.00 equiv) and a magnetic stir bar. The vial and its contents were flame-dried under reduced pressure, then allowed to cool under N_2 . Ketone substrate **1** (21.0 mg, 0.100 mmol, 1.00 equiv) and alcohol reductant **5** (41.3 mg, 0.250 mmol, 2.50 equiv) were added. The vial was flushed with N_2 , and then 1,4-dioxane (0.100 mL, 1.00 M) was added. Under a stream of N_2 , the vial septum cap was quickly switched for a Teflon-lined screw cap, sealed, then further sealed with electrical tape. The reaction was stirred vigorously (800 rpm) at 120 °C for 16 h. After cooling to 23 °C, the mixture was diluted with hexanes (0.5 mL) and filtered over a plug of silica gel (1 cm OD x 5 cm, 10 mL EtOAc eluent). The volatiles were removed under reduced pressure and the yield of alcohol **2** was determined by ^1H NMR analysis with 1,3,5-trimethoxybenzene as an external standard.

Any modifications of the conditions shown in the representative procedure above are specified below in Table S1.



Reaction Conditions	Experimental Results ^a	
	1	2
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), dioxane (1.0 M), 120 °C, 16 h	0%	99%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), dioxane (1.0 M), 80 °C, 16 h	0%	99%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), dioxane (1.0 M), 120 °C, 3 h	<5%	98%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), 2-Me THF (1.0 M), 80 °C, 16 h	0%	99%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), <i>t</i> -amyl alcohol (1.0 M), 80 °C, 16 h	<5%	98%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), <i>n</i> -heptane (1.0 M), 80 °C, 16 h	11%	89%
Control Experiments:		
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), dioxane (1.0 M), 120 °C, 16 h <i>Ran in the dark</i>	0%	99%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), H ₂ O (2.0 equiv), dioxane (1.0 M), 120 °C, 16 h	<5%	95%
5 (2.5 equiv), K ₃ PO ₄ (4.0 equiv), dioxane (1.0 M), 120 °C, 16 h <i>Ran under an atmosphere of air</i>	12%	88%

^a Yields were determined by ¹H NMR analysis using 1,3,5-trimethoxybenzene as an external standard.

Table S1. Survey of Reaction Conditions and Relevant Control Experiments

E. Trace Metal Analysis

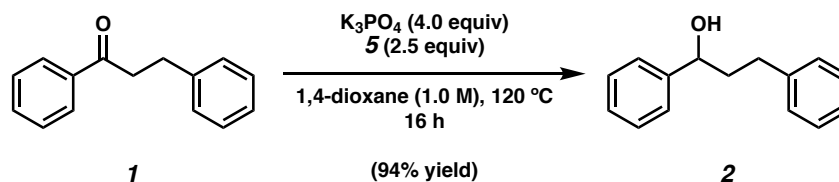
Representative Procedure for Trace Metal Analysis (preparation of K_3PO_4 is used as an example). A 15-mL conical tube was charged with K_3PO_4 (95.2 mg, 1.00 equiv) and the sample was diluted with milli-Q water (6.8 mL) to a final concentration of 1.4% (w/w). Subsequently, ICP-MS-grade 70% nitric acid (200 μ l) was added to each sample (2% final nitric acid concentration).

Sample: K_3PO_4	
Metal	Concentration (ppm) (average of two samples)
Fe	0.00809
Al	0.000
Co	0.000
B	0.0240
Ti	0.0420
Mg	0.00189
Mn	7.84×10^{-5}
Sc	0.000
Rb	0.203
Ni	0.0303
Cu	0.000
Zn	0.000
Ru	9.46×10^{-6}
Rh	0.000
Pd	1.53×10^{-5}
Ag	0.000
Ir	1.17×10^{-6}
Pt	0.000

Sample: 1,4-dioxane	
Metal	Concentration (ppm) (average of two samples)
Fe	0.00369
Al	0.00384
Co	0.000130
B	0.00570
Ti	0.00250
Mg	0.01780
Mn	0.000151
Sc	0.00190
Rb	3.41×10^{-5}
Ni	0.354
Cu	0.000
Zn	0.000
Ru	0.000
Rh	0.000
Pd	1.06×10^{-5}
Ag	2.85×10^{-5}
Ir	0.000
Pt	0.000

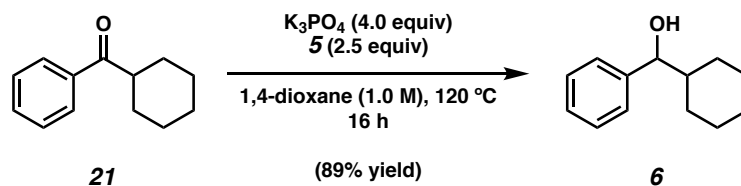
Sample: Alcohol Reductant 5	
Metal	Concentration (ppm) (average of two samples)
Fe	0.00352
Al	0.00121
Co	3.20×10^{-5}
B	0.0211
Ti	0.000642
Mg	0.0125
Mn	1.81×10^{-5}
Sc	0.000962
Rb	5.15×10^{-6}
Ni	0.113
Cu	0.000
Zn	0.000
Ru	0.000
Rh	9.31×10^{-7}
Pd	3.00×10^{-6}
Ag	7.91×10^{-6}
Ir	0.000
Pt	1.76×10^{-6}

F. Scope of Methodology

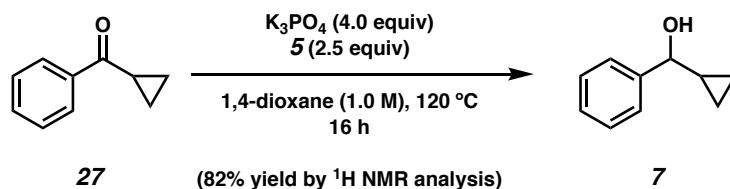


Representative Procedure for Base-Mediated MPV Reduction from Figure 2 (reduction of ketone **1 with alcohol **5** is used as an example). Alcohol **2**.** A 1-dram vial was charged with anhydrous powdered K_3PO_4 (85.0 mg, 0.400 mmol, 4.00 equiv) and a magnetic stir bar. The vial and its contents were flame-dried under reduced pressure, then allowed to cool under N_2 . Ketone substrate **1** (21.0 mg, 0.100 mmol, 1.00 equiv) and alcohol reductant **5** (41.3 mg, 0.250 mmol, 2.50 equiv) were added. The vial was purged with N_2 , and then 1,4-dioxane (0.100 mL, 1.00 M) was added. Under a stream of N_2 , the vial septum cap was quickly switched for a Teflon-lined screw cap, sealed, then further sealed with electrical tape. The reaction was stirred vigorously (800 rpm) at 120 °C for 16 h. After cooling to 23 °C, the reaction was quenched by the addition of sat. aq. NH_4Cl (1.00 mL) and diluted with EtOAc (2.00 mL) and the layers were separated. The aqueous layer was extracted with EtOAc (3 x 2.00 mL) and the combined organic layers were passed through a plug (1.00 cm OD) of silica gel (3.00 cm tall) and Na_2SO_4 (3.00 cm tall) using EtOAc (10.0 mL) as eluent. The volatiles were removed under reduced pressure. The crude residue was purified by flash chromatography (99:1 Hexanes:EtOAc \rightarrow 19:1 Hexanes:EtOAc) to yield alcohol **2** (20 mg, 94% yield, average of two experiments) as a clear oil. Alcohol **2**: R_f 0.32 (5:1 Hexanes:EtOAc). ^1H NMR (500 MHz, CDCl_3): δ 7.41–7.33 (m, 4H), 7.32–7.26 (m, 3H), 7.23–7.16 (m, 3H), 4.78–4.61 (m, 1H), 2.85–2.61 (m, 2H), 2.23–1.97 (m, 2H), 1.87 (d, $J = 3.5$ Hz, 1H). Spectral data match those previously reported.⁶

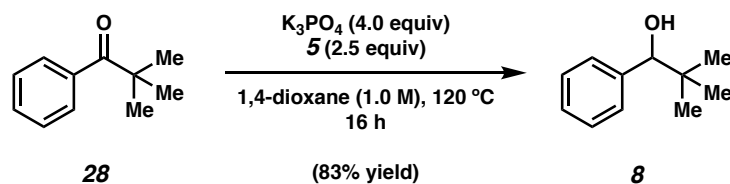
Any modifications of the conditions shown in the representative procedure above are specified in the following schemes, which depict all of the results shown in Figures 2 and 3.



Alcohol 6. Purification by flash chromatography (99:1 Hexanes:EtOAc \rightarrow 19:1 Hexanes:EtOAc) generated alcohol **6** (17 mg, 89% yield, average of two experiments) as a crystalline white solid. Alcohol **6**: R_f 0.39 (5:1 Hexanes:EtOAc). ^1H NMR (500 MHz, CDCl_3): δ 7.38–7.25 (m, 5H), 4.37 (dd, $J = 7.2$ Hz, 3.3 Hz, 1H), 2.03–1.92 (m, 1H), 1.84–1.72 (m, 2H), 1.71–1.57 (m, 3H), 1.41–1.33 (m, 1H), 1.29–1.00 (m, 4H), 0.99–0.87 (m, 1H). Spectral data match those previously reported.⁷

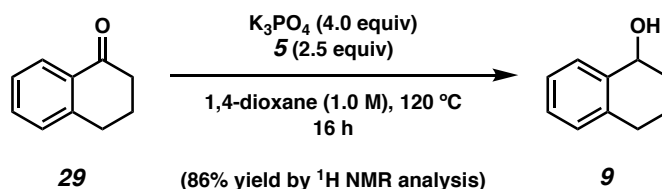


Alcohol 7. ^1H NMR analysis of the crude reaction mixture indicated an 82% yield of alcohol **7** relative to hexamethylbenzene external standard (average of two experiments). Purification by preparative thin-layer chromatography (3:1 Hexanes:EtOAc) provided an analytical sample of alcohol **7** as a clear oil. Alcohol **7**: R_f 0.30 (5:1 Hexanes:EtOAc). ^1H NMR (500 MHz, CDCl_3): δ 7.46–7.40 (m, 2H), 7.39–7.33 (m, 2H), 7.32–7.27 (m, 1H), 4.02 (dd, $J = 8.3, 3.0$, 1H), 1.90 (d, $J = 3.0$, 1H), 1.28–1.18 (m, 1H), 0.69–0.61 (m, 1H), 0.60–0.52 (m, 1H), 0.52–0.44 (m, 1H), 0.42–0.34 (m, 1H). Spectral data match those previously reported.⁸

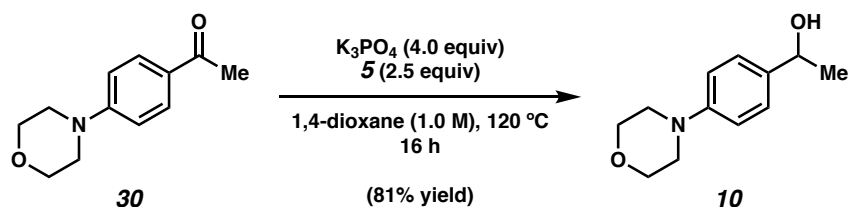


Alcohol 8. Purification by flash chromatography (24:1 Hexanes:EtOAc) generated alcohol **8** (14 mg, 83% yield, average of two experiments) as a white crystalline solid. Alcohol **8**: R_f 0.52 (5:1

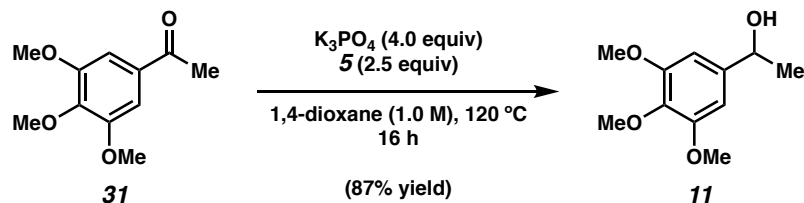
Hexanes:EtOAc). $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.35–7.30 (m, 4H), 7.29–7.26 (m, 1H), 4.40 (d, $J = 2.8$, 1H), 1.84 (d, $J = 2.8$, 1H), 0.93 (s, 9H). Spectral data match those previously reported.⁹



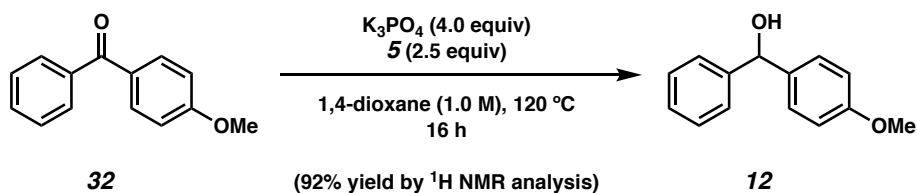
Alcohol 9. $^1\text{H NMR}$ analysis of the crude reaction mixture indicated an 86% yield of alcohol **9** relative to hexamethylbenzene external standard (average of two experiments). Purification by preparative thin-layer chromatography (9:1 PhH:Acetone) provided an analytical sample of alcohol **9** as a clear oil. Alcohol **9**: R_f 0.50 (9:1 PhH:Acetone). $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.48–7.39 (m, 1H), 7.24–7.17 (m, 2H), 7.14–7.05 (m, 1H), 4.84–4.71 (m, 1H), 2.88–2.78 (m, 1H), 2.87–2.67 (m, 1H), 2.05–1.95 (m, 2H), 1.95–1.87 (m, 1H), 1.85–1.73 (m, 1H), 1.71–1.60 (m, 1H). Spectral data match those previously reported.¹⁰



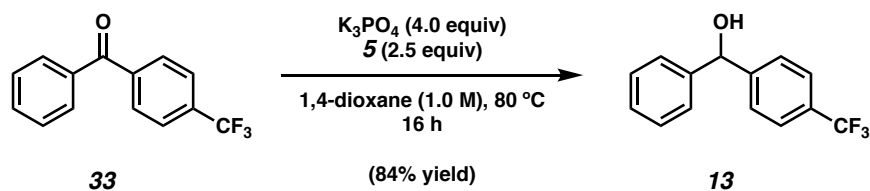
Alcohol 10. Purification by flash chromatography (2:2:1 CH_2Cl_2 : Et_2O :Hexanes) generated alcohol **10** (17 mg, 81% yield, average of two experiments) as a pale yellow solid. Alcohol **10**: R_f 0.24 (1:1:1 CH_2Cl_2 : Et_2O :Hexanes). $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.33–7.28 (m, 2H), 6.93–6.87 (m, 2H), 4.85 (dq, $J = 6.4, 3.5$, 1H), 3.90–3.83 (m, 4H), 3.19–3.12 (m, 4H), 1.67 (d, $J = 3.5$, 1H), 1.48 (d, $J = 6.4$, 3H). Spectral data match those previously reported.¹¹



Alcohol 11. Purification by flash chromatography (9:1 Hexanes:EtOAc \rightarrow 2:1 Hexanes:EtOAc) generated alcohol **11** (18 mg, 87% yield, average of two experiments) as a clear oil. Alcohol **11**: R_f 0.33 (1:1 Hexanes:EtOAc). $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 6.60 (s, 2H), 4.89–4.77 (m, 1H), 3.86 (s, 6H), 3.83 (s, 3H), 1.99–1.80 (m, 1H), 1.48 (d, $J = 6.4$, 3H). Spectral data match those previously reported.¹¹

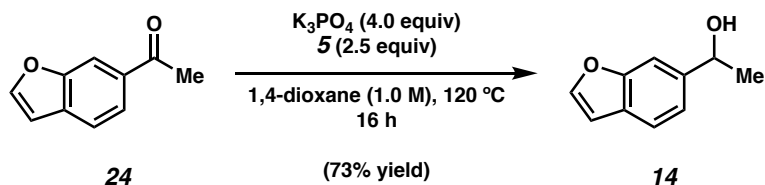


Alcohol 12. $^1\text{H NMR}$ analysis of the crude reaction mixture indicated a 92% yield of alcohol **12** relative to hexamethylbenzene external standard (average of two experiments). Purification by preparative thin-layer chromatography (3:3:2 CH_2Cl_2 : Et_2O :Hexanes) provided an analytical sample of alcohol **12** as a pale yellow solid. Alcohol **12**: R_f 0.70 (1:1:1 CH_2Cl_2 : Et_2O :Hexanes). $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.40–7.31 (m, 4H), 7.31–7.23 (m, 3H), 6.93–6.81 (m, 2H), 5.82 (d, $J = 3.0$, 1H), 3.79 (s, 3H), 2.15 (d, $J = 3.4$, 1H). Spectral data match those previously reported.⁷

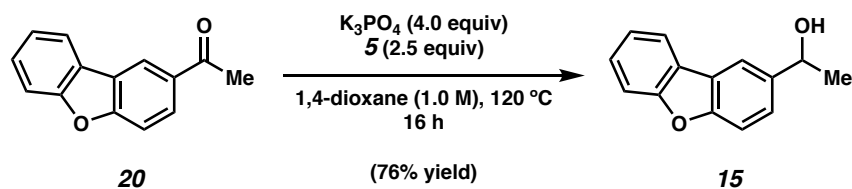


Alcohol 13. Purification by flash chromatography (9:1 Hexanes: Et_2O \rightarrow 3:1 Hexanes: Et_2O) generated alcohol **13** (21 mg, 84% yield, average of two experiments) as a clear oil. Alcohol **13**: R_f 0.30 (5:1 Hexanes: Et_2O). $^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.60 (d, $J = 8.3$, 2H), 7.51 (d, $J =$

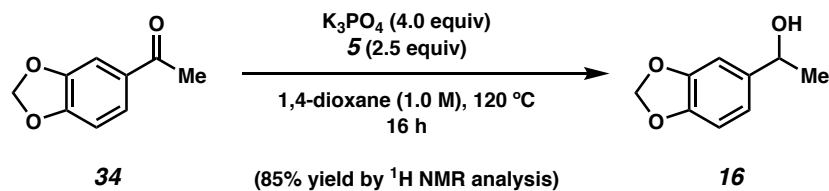
8.3, 2H), 7.42–7.34 (m, 4H), 7.34–7.28 (m, 1H), 5.88 (s, 1H), 2.46–2.32 (m, 1H). Spectral data match those previously reported.⁷



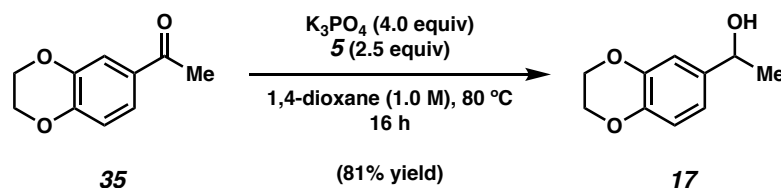
Alcohol 14. Purification by flash chromatography (90:9:1 → 15:9:1 Hexanes:PhH:Acetone) generated alcohol **14** (12 mg, 73% yield, average of two experiments) as a yellow oil. Alcohol **14**: R_f 0.39 (9:1 PhH:Acetone); $^1\text{H NMR}$ (500 MHz, C_6D_6): δ 7.50–7.46 (m, 1H), 7.35 (d, $J = 8.1$ Hz, 1H), 7.12 (ddd, $J = 8.1, 1.4, 0.5$ Hz, 1H), 6.35 (dd, $J = 2.2, 1.1$ Hz, 1H), 4.59 (q, $J = 6.5$ Hz, 1H), 1.29 (d, $J = 6.5$ Hz, 3H), 1.14, (br s, 1H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3): δ 155.3, 145.4, 142.8, 126.9, 121.2, 120.6, 108.4, 106.5, 70.7, 25.6; IR (film): 3350, 2972, 2926, 1430, 1265 cm^{-1} ; HRMS-APCI (m/z) $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{10}\text{H}_{11}\text{O}_2^+$, 163.0754; found 163.0746.



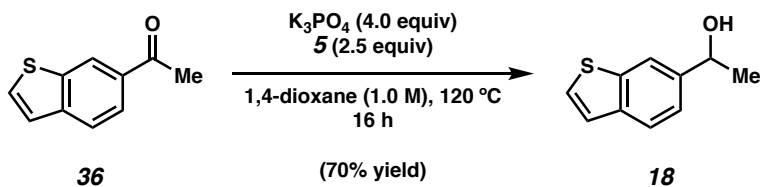
Alcohol 15. Purification by flash chromatography (90:9:1 → 25:9:1 Hexanes:PhH:Acetone) generated alcohol **15** (16 mg, 76% yield, average of two experiments) as a yellow solid. Alcohol **15**: R_f 0.35 (9:1 PhH:Acetone). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.03–7.91 (m, 2H), 7.60–7.51 (m, 2H), 7.50–7.43 (m, 2H), 7.35 (td, $J = 7.5, 1.0$, 1H), 5.09 (dq, $J = 6.4, 3.3$, 1H), 1.90 (d, $J = 3.3$, 1H), 1.60 (d, $J = 6.4$, 3H). Spectral data match those previously reported.¹²



Alcohol 16. ^1H NMR analysis of the crude reaction mixture indicated an 85% yield of alcohol **16** relative to hexamethylbenzene external standard. Purification by preparative thin-layer chromatography (13:1:1 PhH:Et₂O:CH₃CN) provided an analytical sample of alcohol **16** as a yellow oil. Alcohol **16**: R_f 0.30 (13:1:1 PhH:Et₂O:CH₃CN); ^1H NMR (500 MHz, CDCl₃): δ 6.90 (s, 1H), 6.82 (dd, $J = 8.1, 1.7$ Hz, 1H), 6.77 (d, $J = 8.1$ Hz, 1H), 5.95 (s, 2H), 4.86–4.79 (m, 1H), 1.70 (d, $J = 3.4$ Hz, 1H), 1.46 (d, $J = 6.4$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl₃): δ 147.9, 147.0, 140.1, 118.8, 108.2, 106.2, 101.1, 70.4, 25.3; IR (film): 3361, 2972, 2890, 1487, 1240 cm⁻¹; HRMS-APCI (m/z) [$M + \text{H}$]⁺ calcd for C₉H₁₁O₃⁺, 167.0703; found 167.0699.

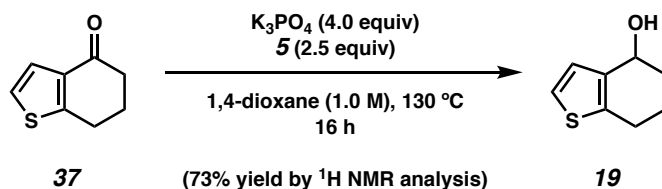


Alcohol 17. Purification by flash chromatography (98:1:1 \rightarrow 28:1:1 PhH:Et₂O:CH₃CN) generated alcohol **17** (15 mg, 81% yield, average of two experiments) as a yellow oil. Alcohol **17**: R_f 0.32 (13:1:1 PhH:Et₂O:CH₃CN). ^1H NMR (500 MHz, CDCl₃): δ 6.92–6.88 (m, 1H), 6.86–6.81 (m, 2H), 4.80 (dq, $J = 6.4, 3.6$, 1H), 4.25 (s, 4H), 1.69 (d, $J = 3.6$, 1H), 1.46 (d, $J = 6.4$, 3H). Spectral data match those previously reported.¹³



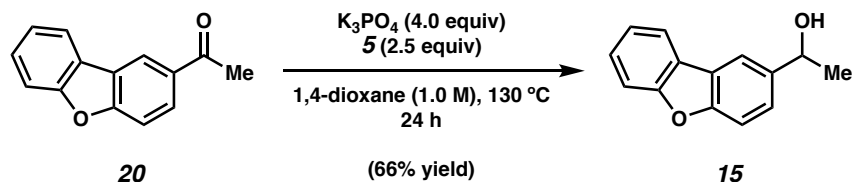
Alcohol 18. Purification by flash chromatography (90:9:1 \rightarrow 40:9:1 Hexanes:PhH:Acetone) generated alcohol **8** (12 mg, 70% yield, average of two experiments) as a yellow oil. Alcohol **18**: R_f 0.38 (9:1 PhH:Acetone); ^1H NMR (500 MHz, CDCl₃): δ 7.91 (s, 1H), 7.80 (d, $J = 8.2$ Hz,

1H), 7.43 (d, $J = 5.5$ Hz, 1H), 7.38 (dd, $J = 8.3, 1.3$ Hz, 1H), 7.32 (d, $J = 5.5$ Hz, 1H), 5.10–5.0 (m, 1H), 1.85 (d, $J = 3.4$ Hz, 1H), 1.56 (d, $J = 6.5$ Hz, 3H); ^{13}C NMR (125 MHz, CDCl_3): δ 142.3, 140.1, 139.2, 126.6, 123.8, 123.7, 122.3, 119.2, 70.7, 25.5; IR (film): 3351, 2971, 1398, 1197, 1074 cm^{-1} ; HRMS-APCI (m/z) [M] $^+$ calcd for $\text{C}_{10}\text{H}_{10}\text{OS}^+$, 178.0447; found 178.0437.



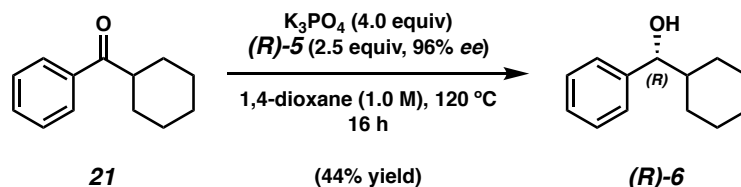
Alcohol 19. ^1H NMR analysis of the crude reaction mixture indicated a 73% yield of alcohol **19** relative to hexamethylbenzene external standard. Purification by preparative thin-layer chromatography (1:1:1 Hexanes: Et_2O : CH_2Cl_2) provided an analytical sample of alcohol **19** as a white crystalline solid. Alcohol **19**: R_f 0.48 (1:1:1 Hexanes: Et_2O : CH_2Cl_2); ^1H NMR (500 MHz, CDCl_3): δ 7.10 (dt, $J = 5.2, 0.7$ Hz, 1H), 7.03 (d, $J = 5.2$ Hz, 1H), 4.82–4.75 (m, 1H), 2.89–2.79 (m, 1H), 2.77–2.67 (m, 1H), 2.06–1.94 (m, 2H), 1.93–1.79 (m, 2H), 1.64 (d, $J = 7.0$ Hz, 1H); ^{13}C NMR (125 MHz, CDCl_3): δ 139.0, 138.1, 126.7, 122.9, 65.6, 32.5, 25.2, 20.1; IR (film): 3235, 2936, 2921, 1431, 982 cm^{-1} ; HRMS-APCI (m/z) [$\text{M} + \text{H}$] $^+$ calcd for $\text{C}_8\text{H}_{11}\text{OS}^+$, 155.0525; found 155.0521.

G. Gram-Scale Base-Mediated MPV Reduction



Alcohol 15. An 8-dram vial was charged with anhydrous powdered K_3PO_4 (4.04 g, 19.0 mmol, 4.00 equiv) and a magnetic stir bar. The vial and its contents were flame-dried under reduced pressure, then allowed to cool under N_2 . Ketone substrate **20** (1.00 g, 4.76 mmol, 1.00 equiv) and alcohol reductant **5** (1.96 g, 11.9 mmol, 2.50 equiv) were then added. The vial was flushed with N_2 and subsequently 1,4-dioxane (4.76 mL, 1.00 M) was added. Under a stream of N_2 , the vial septum cap was quickly switched for a Teflon-lined screw cap, sealed, then further sealed with electrical tape. The reaction was then stirred vigorously (800 rpm) at 130 °C for 24 h. After cooling to 23 °C, the reaction was quenched by the addition of sat. aq. NH_4Cl (8.00 mL) and diluted with EtOAc (6.00 mL) and the layers were separated. The aqueous layer was extracted with EtOAc (3 x 6.00 mL) and the combined organic layers were passed through a plug (1.00 cm OD) of silica gel (3.00 cm tall) and Na_2SO_4 (3.00 cm tall) using EtOAc (10.0 mL) as eluent. The volatiles were removed under reduced pressure. The crude residue was purified by flash chromatography (60:9:1 Hexanes:PhH:Acetone \rightarrow 5:9:1 Hexanes:PhH:Acetone) to yield alcohol **15** (664 mg, 66% yield) as a yellow solid. Alcohol **15**: R_f 0.32 (5:1 Hexanes:EtOAc). Spectral data match those previously reported.¹²

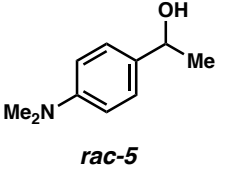
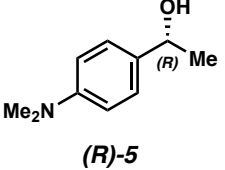
H. Stereospecific Base-Mediated MPV Reduction

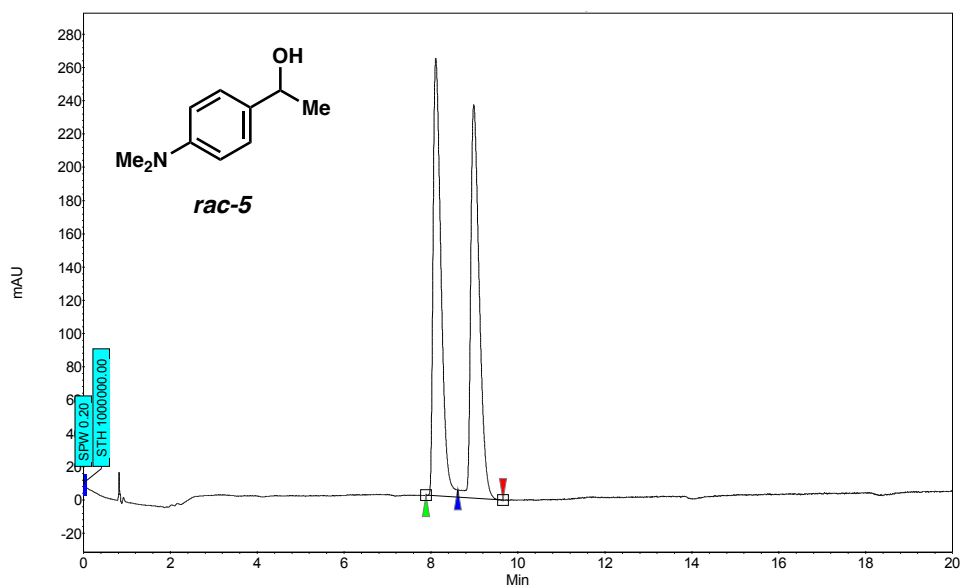


Alcohol (R)-6. A 1-dram vial was charged with anhydrous powdered K_3PO_4 (85.0 mg, 0.400 mmol, 4.00 equiv) and a magnetic stir bar. The vial and its contents were flame-dried under reduced pressure, then allowed to cool under N_2 . Ketone substrate **21** (18.8 mg, 0.100 mmol, 1.00 equiv) and alcohol reductant **(R)-5** (41.3 mg, 0.250 mmol, 2.50 equiv) were added. The vial was purged with N_2 and subsequently, 1,4-dioxane (0.100 mL, 1.00 M) was added. Under a stream of N_2 , the vial septum cap was quickly switched for a Teflon-lined screw cap, sealed, then further sealed with electrical tape. The reaction was stirred vigorously (800 rpm) at 120 °C for 16 h. After cooling to 23 °C, the reaction was quenched by the addition of sat. aq. NH_4Cl (1.00 mL) and diluted with EtOAc (2.00 mL) and the layers were separated. The aqueous layer was extracted with EtOAc (3 x 2.00 mL) and the combined organic layers were passed through a plug (1.00 cm OD) of silica gel (3.00 cm tall) and Na_2SO_4 (3.00 cm tall) using EtOAc (10.0 mL) as eluent. The volatiles were removed under reduced pressure. The crude residue was purified by flash chromatography (99:1 Hexanes:EtOAc \rightarrow 19:1 Hexanes:EtOAc) to yield alcohol **(R)-6** (8.4 mg, 44% yield, 48% ee) as a white crystalline solid. Alcohol **(R)-6**: R_f 0.39 (5:1 Hexanes:EtOAc). $[\alpha]_D^{21.1} = +20.8$ ($c = 0.50$, CHCl_3). The spectral data match those previously reported in the literature for *rac*-**6**.⁷ The major enantiomer product was assigned by comparison to published $[\alpha]_D$ values for **(R)-6**.¹⁴

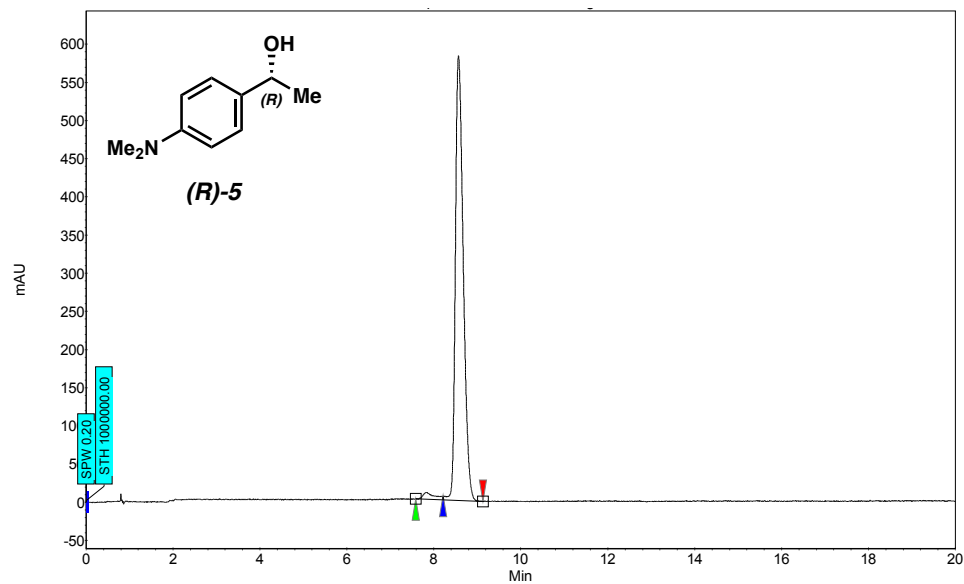
G. Verification of Enantiopurity

Chiral SFC & HPLC Assays

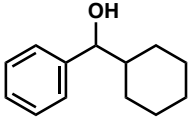
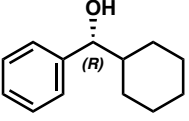
Compound	SFC Method Column/Temp. Abs. Wavelength	Solvent	Method Flow Rate	Retention Times (min)	Enantiomeric Ratio (er)
 <i>rac-5</i>	Daicel ChiralPak IC-3/35 °C $\lambda_{\text{abs}} = 210 \text{ nm}$	5% isopropanol in CO ₂	3.5 mL/min	7.88/8.61	50:50
 <i>(R)-5</i>	Daicel ChiralPak IC-3/35 °C $\lambda_{\text{abs}} = 210 \text{ nm}$	5% isopropanol in CO ₂	3.5 mL/min	7.58/8.22	98:2

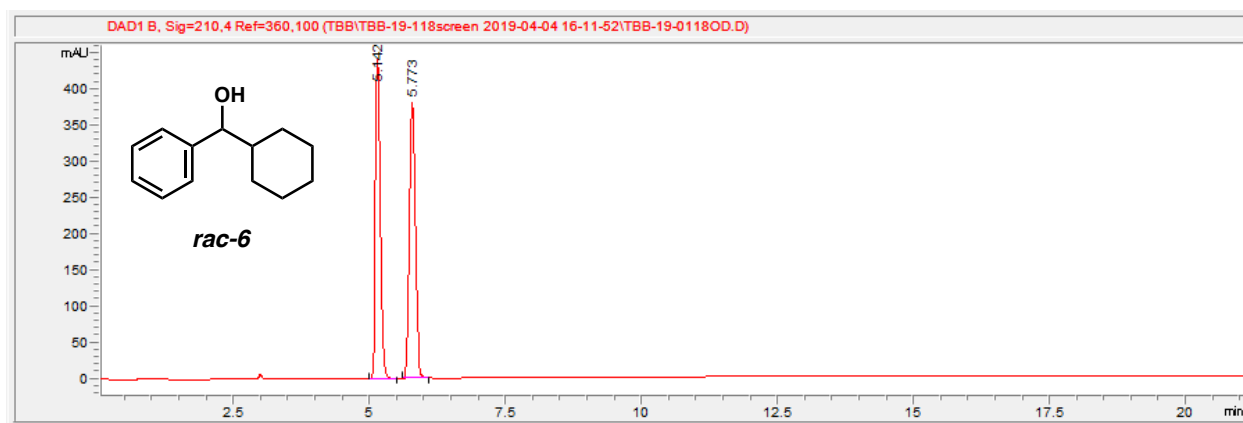


Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[μV]	[$\mu\text{V} \cdot \text{Min}$]	[%]
1	UNKNOWN	7.88	8.11	8.61	0.00	50.43	263.1	55.4	50.434
2	UNKNOWN	8.61	8.99	9.66	0.00	49.57	236.6	54.5	49.566
Total						100.00	499.7	109.9	100.000

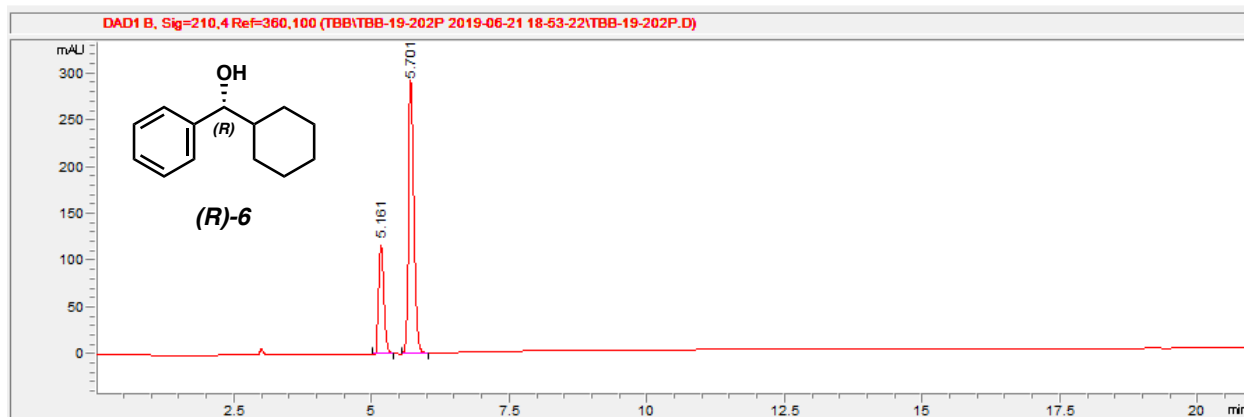


Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [μ V]	Area [μ V.Min]	Area [%]
1	UNKNOWN	7.58	7.82	8.22	0.00	2.18	8.7	2.7	2.176
2	UNKNOWN	8.22	8.57	9.13	0.00	97.82	582.5	122.1	97.824
Total						100.00	591.1	124.8	100.000

Compound	HPLC Method Column/Temp. Abs. Wavelength	Solvent	Method Flow Rate	Retention Times (min)	Enantiomeric Ratio (er)
 rac-6	Daicel ChiralPak OD-H/23 °C $\lambda_{\text{abs}} = 210 \text{ nm}$	10% isopropanol in Hexanes	1 mL/min	5.14/5.77	50:50
 (R)-6	Daicel ChiralPak OD-H/23 °C $\lambda_{\text{abs}} = 210 \text{ nm}$	10% isopropanol in Hexanes	1 mL/min	5.16/5.70	74:26



#	Time	Area	Height	Width	Area%	Symmetry
1	5.142	2774.2	442.1	0.0972	50.021	0.833
2	5.773	2771.9	380.2	0.113	49.979	0.836



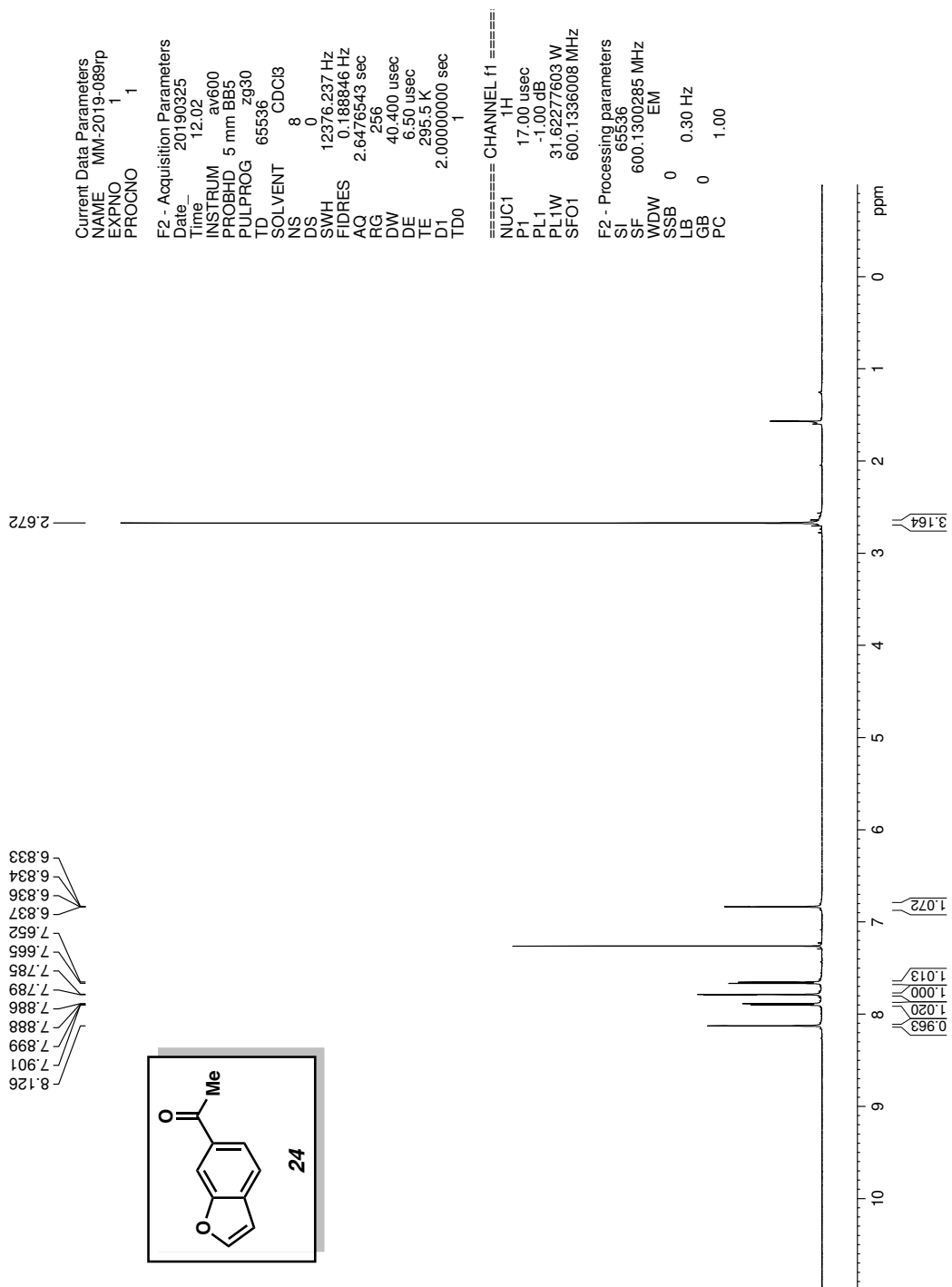
#	Time	Area	Height	Width	Area%	Symmetry
1	5.161	732.7	116.3	0.0975	25.919	0.834
2	5.701	2094.1	291.8	0.1107	74.081	0.837

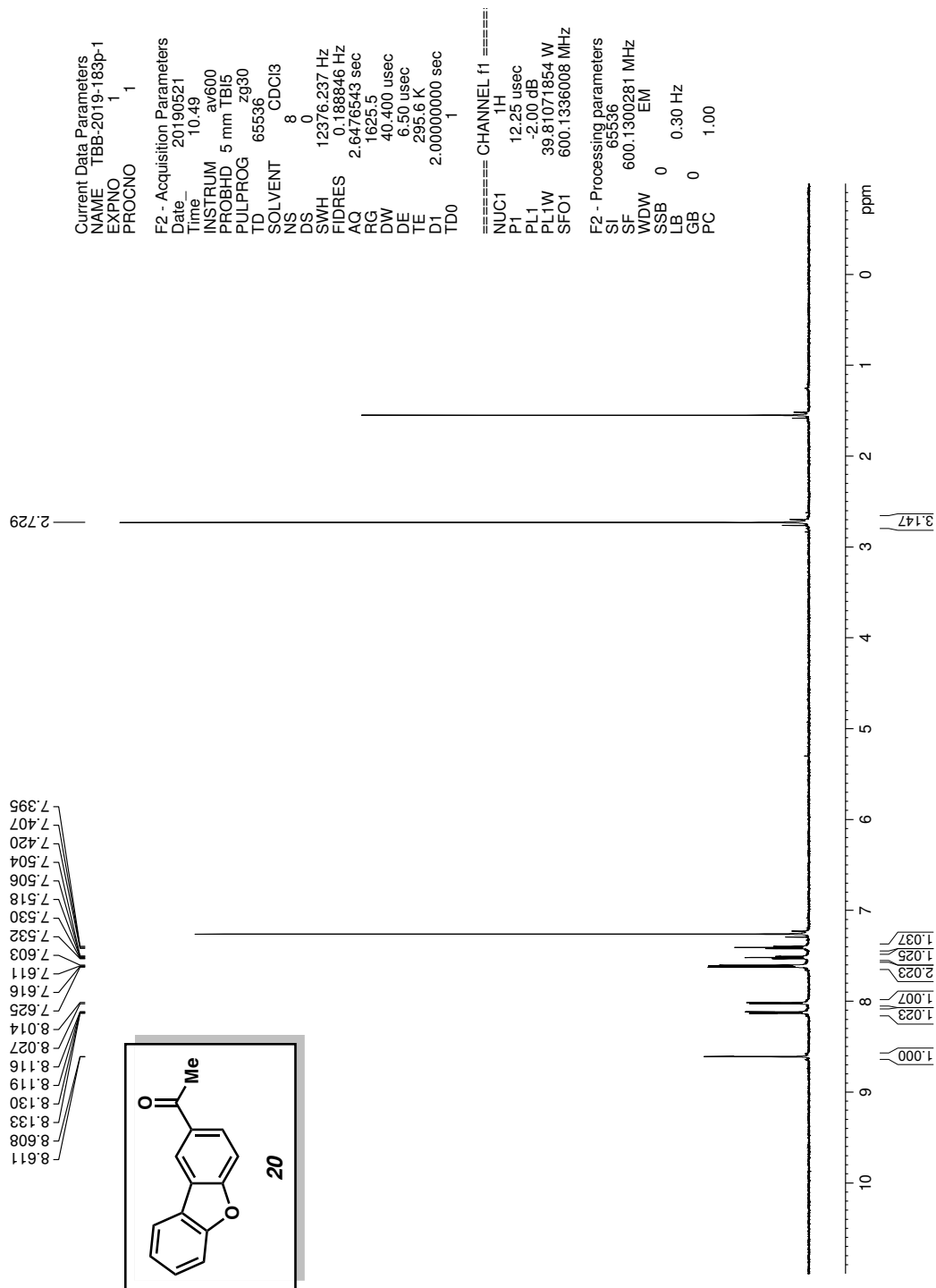
References

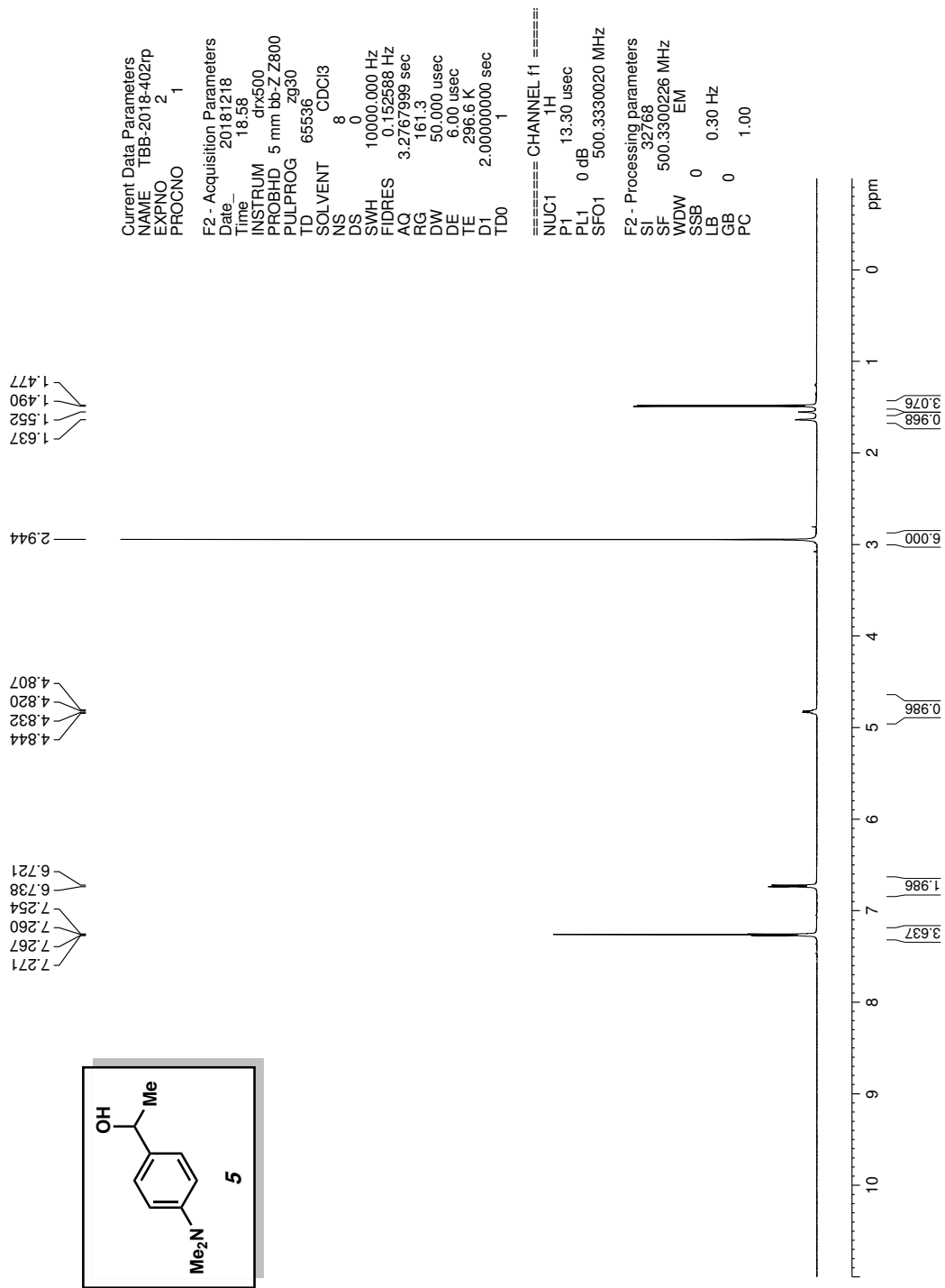
- ¹ Mori, A.; Miyakawa, Y.; Ohashi, E.; Haga, T.; Maegawa, T.; Sajiki, H. Pd/C-catalyzed chemoselective hydrogenation in the presence of diphenylsulfide. *Org. Lett.* **2006**, *8*, 3279–3281.
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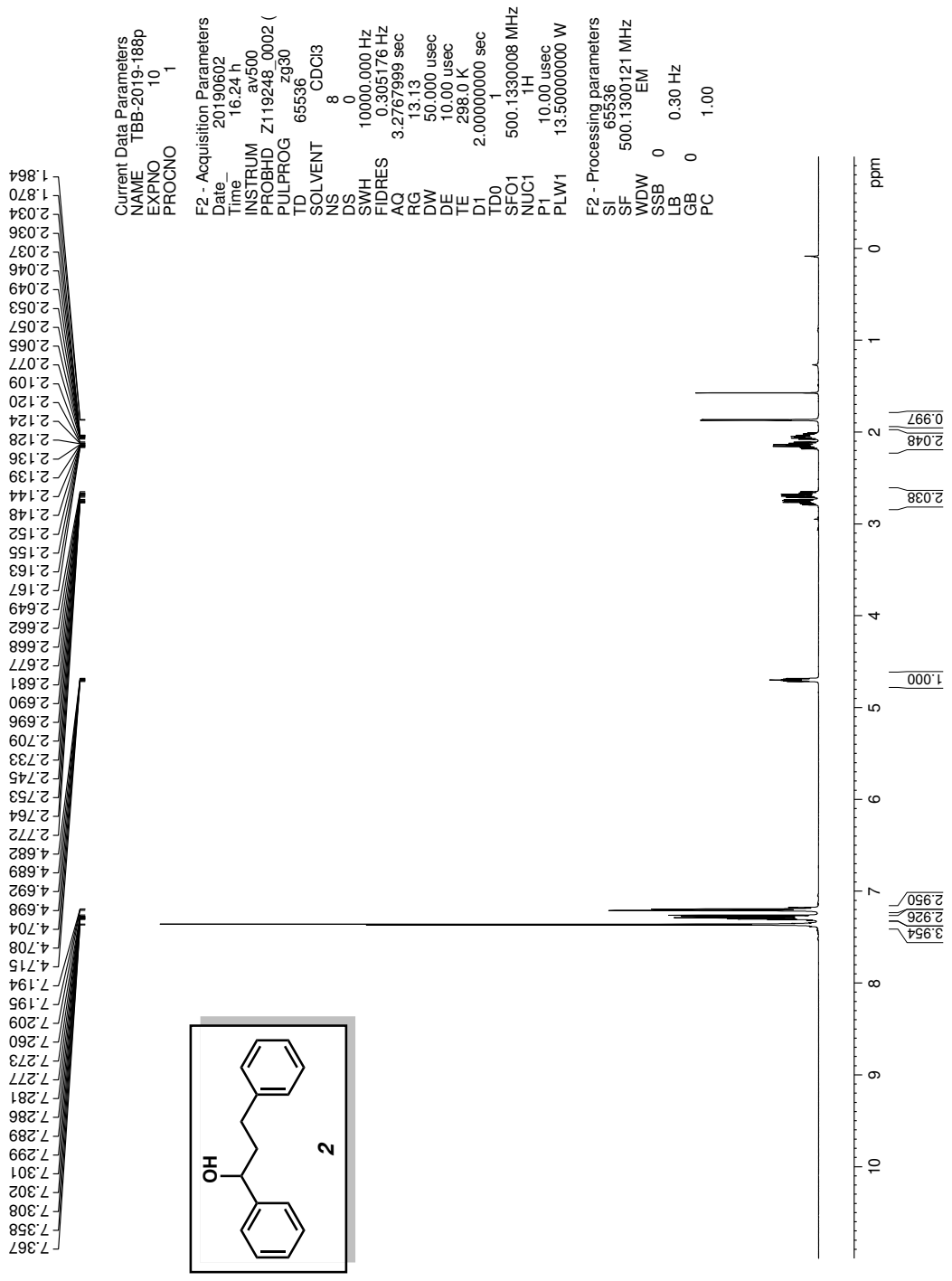
- ¹³ Moine, E.; Dimier-Poisson, I.; Enguihard-Gueiffier, C.; Logé, C.; Pénichon, M.; Moiré, N.; Delebouzé, C.; Foll-Josselin, B.; Ruchaud, S.; Bach, S.; Gueiffier, A.; Debierre-Grockiego, F.; Denevault-Sabourin, C. Development of new highly potent imidazo[1,2-b]pyridazines targeting *Toxoplasma gondii* calcium-dependent protein kinase 1. *Eur. J. Med. Chem.* **2015**, *105*, 80–105.
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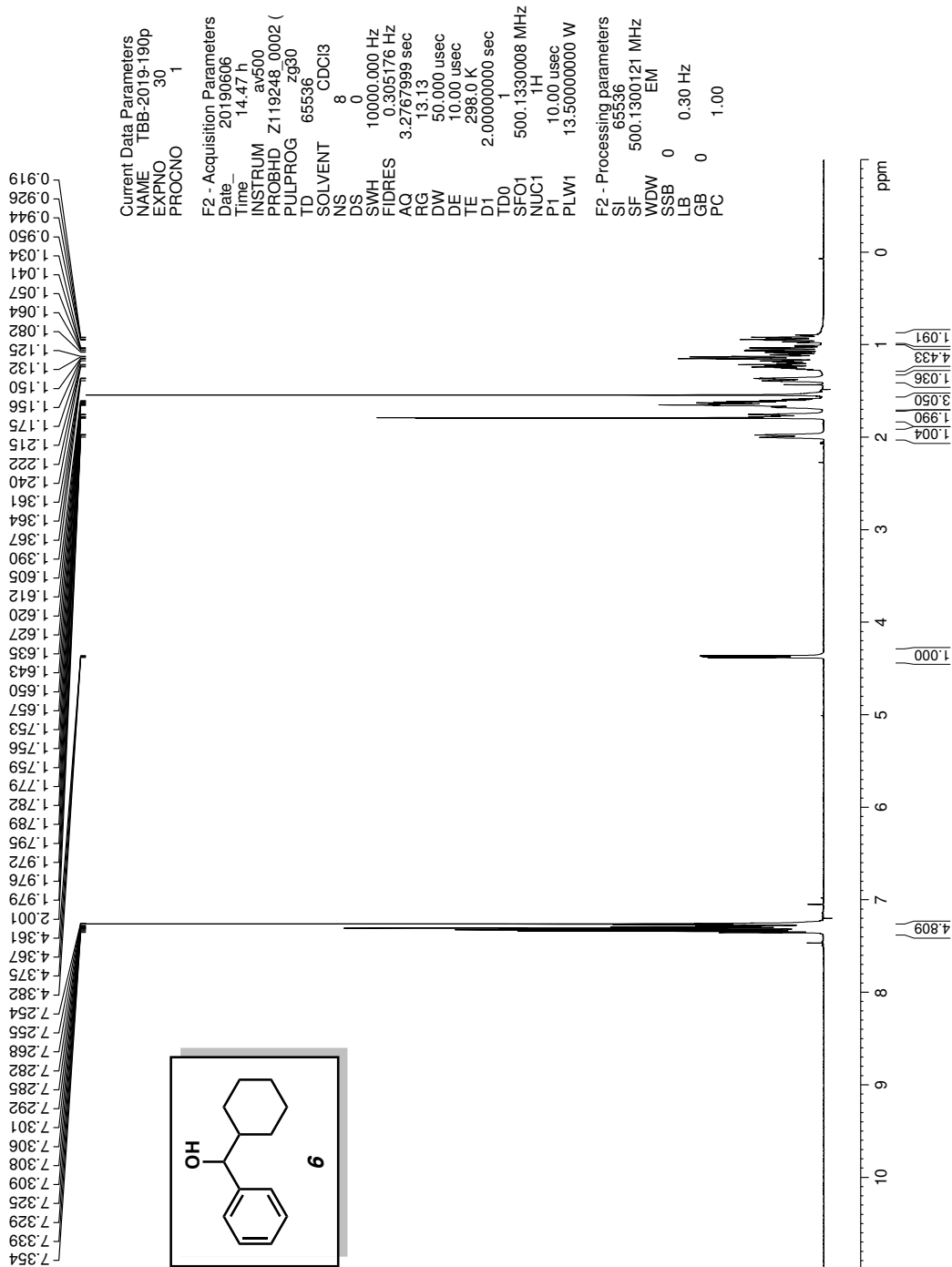
¹H NMR Spectra

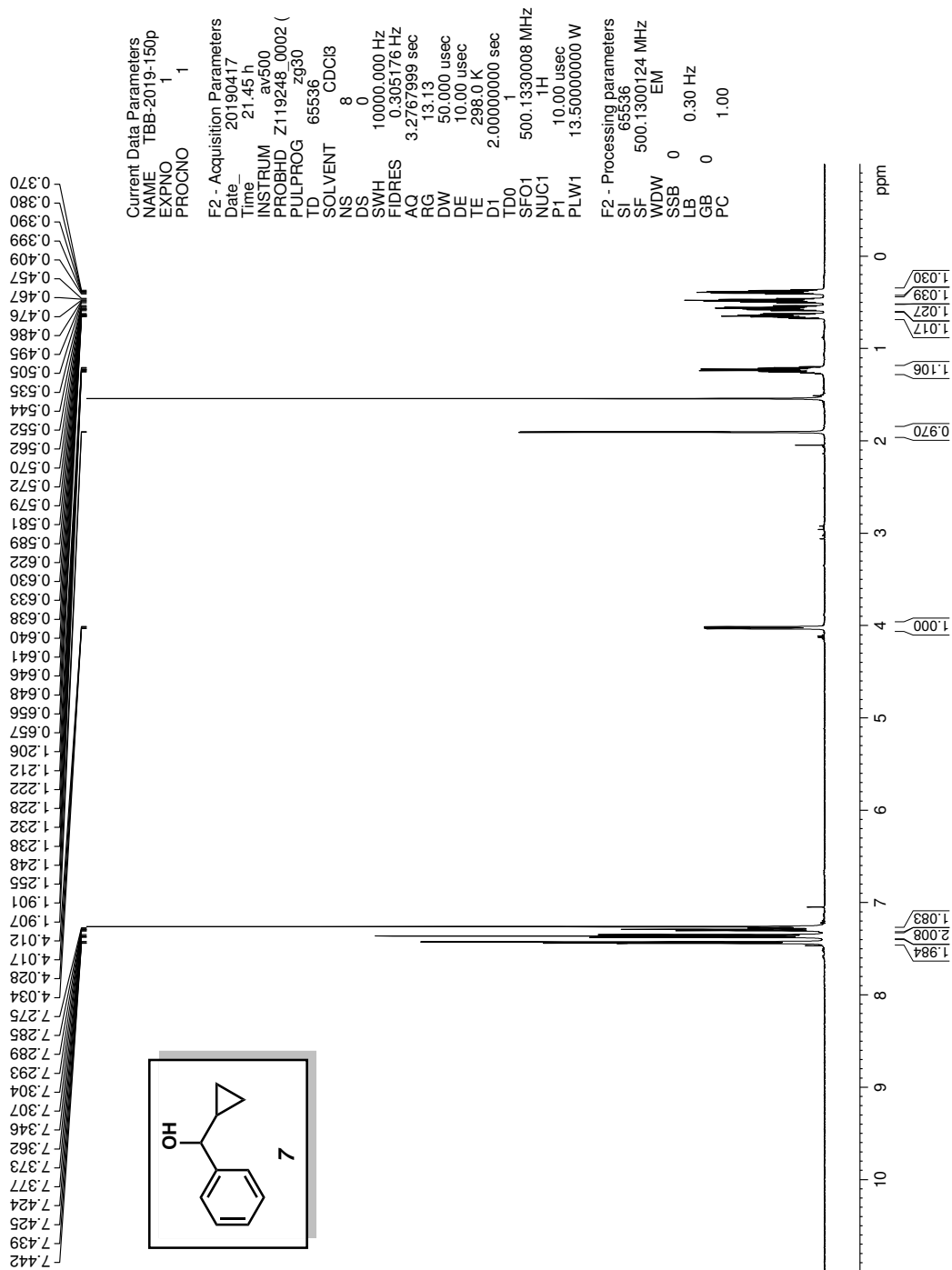


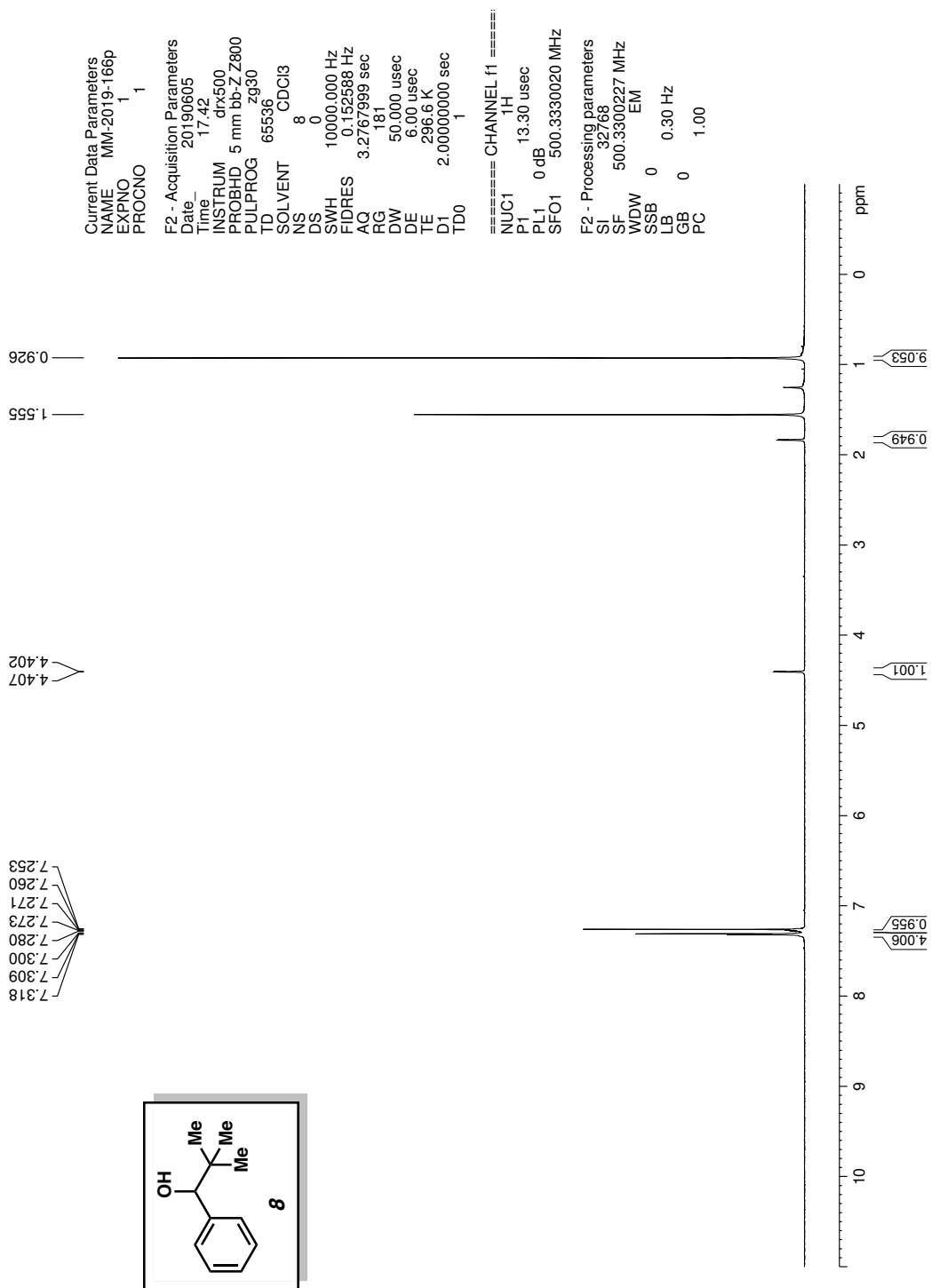


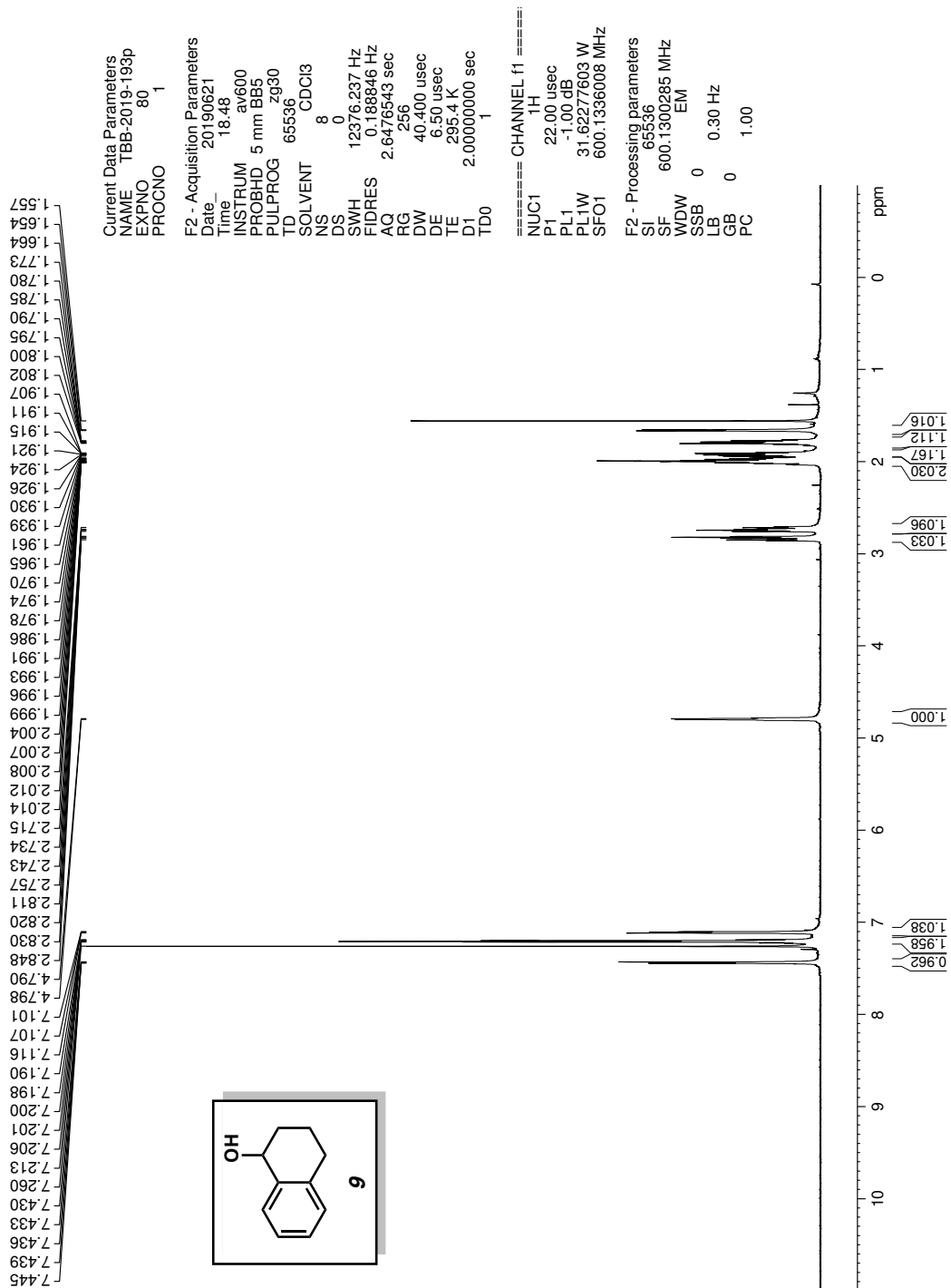


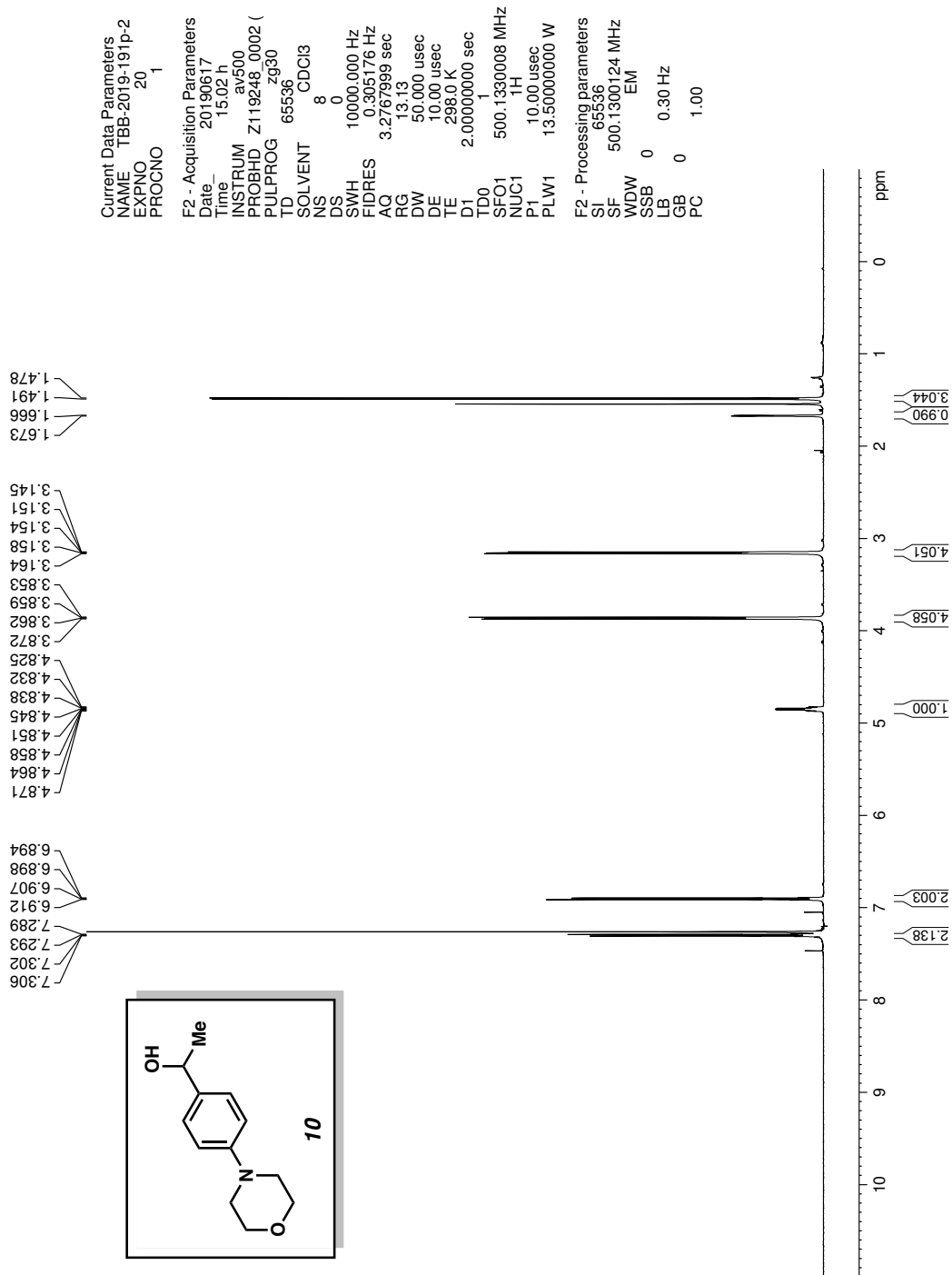


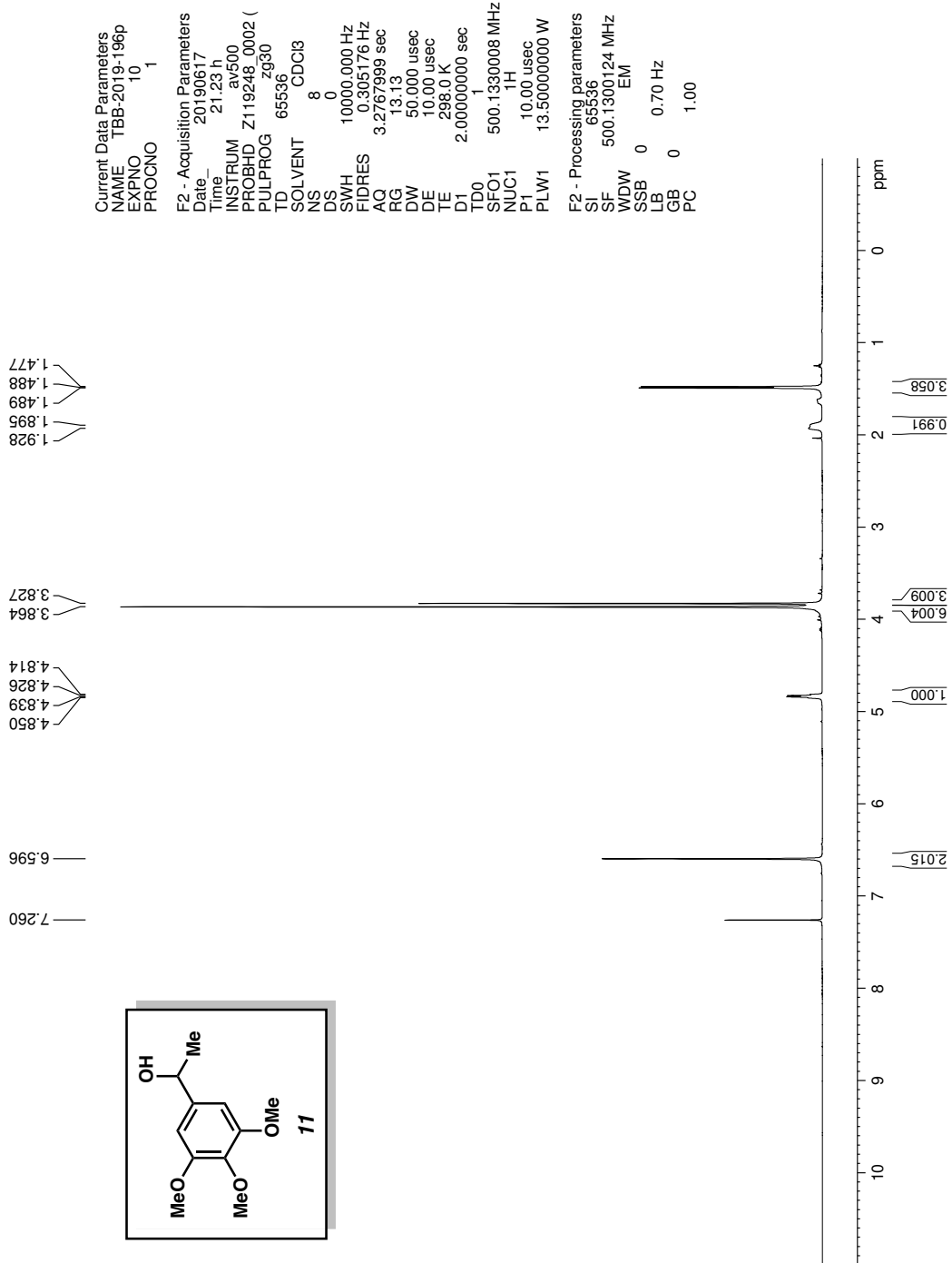


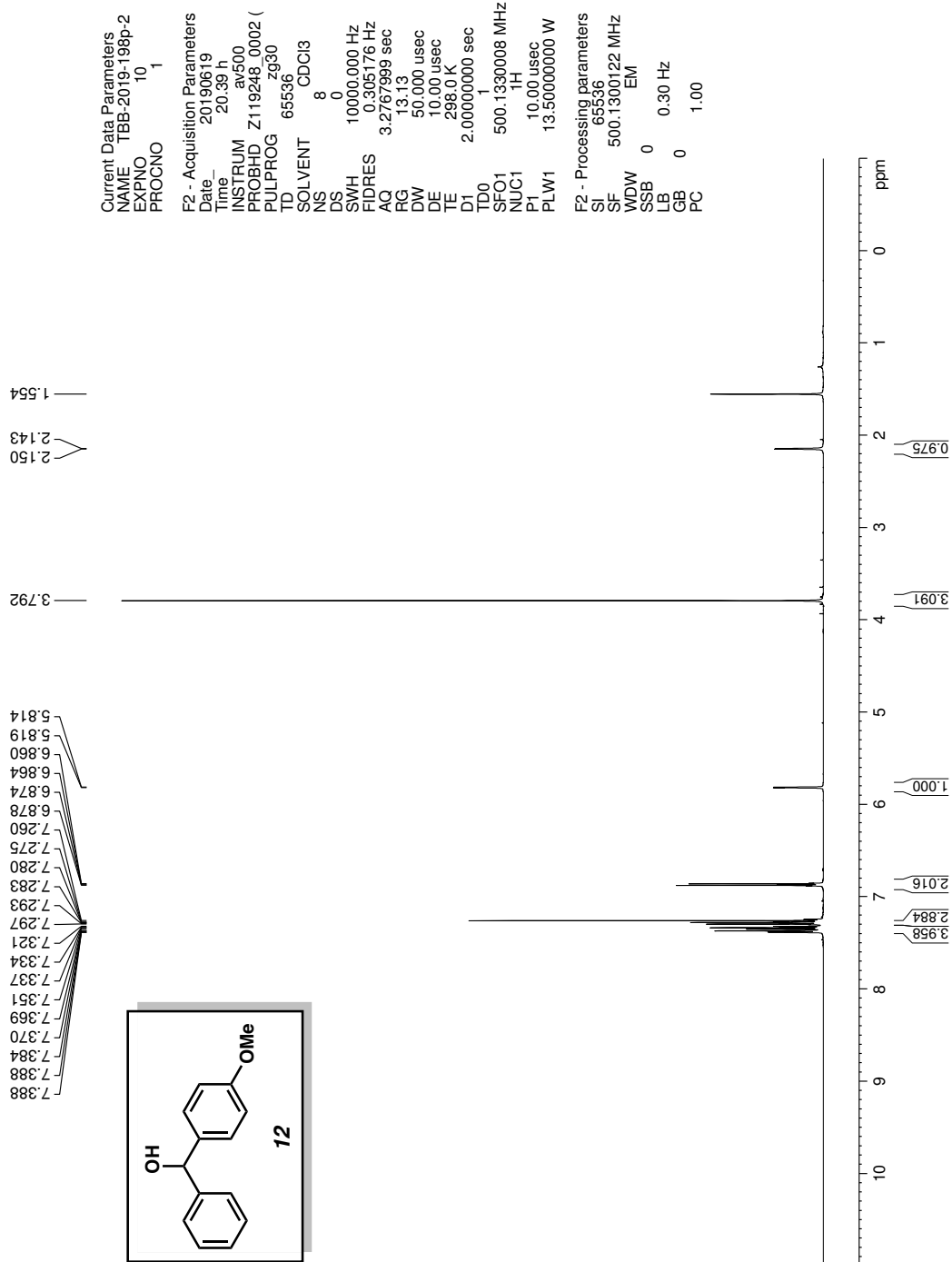


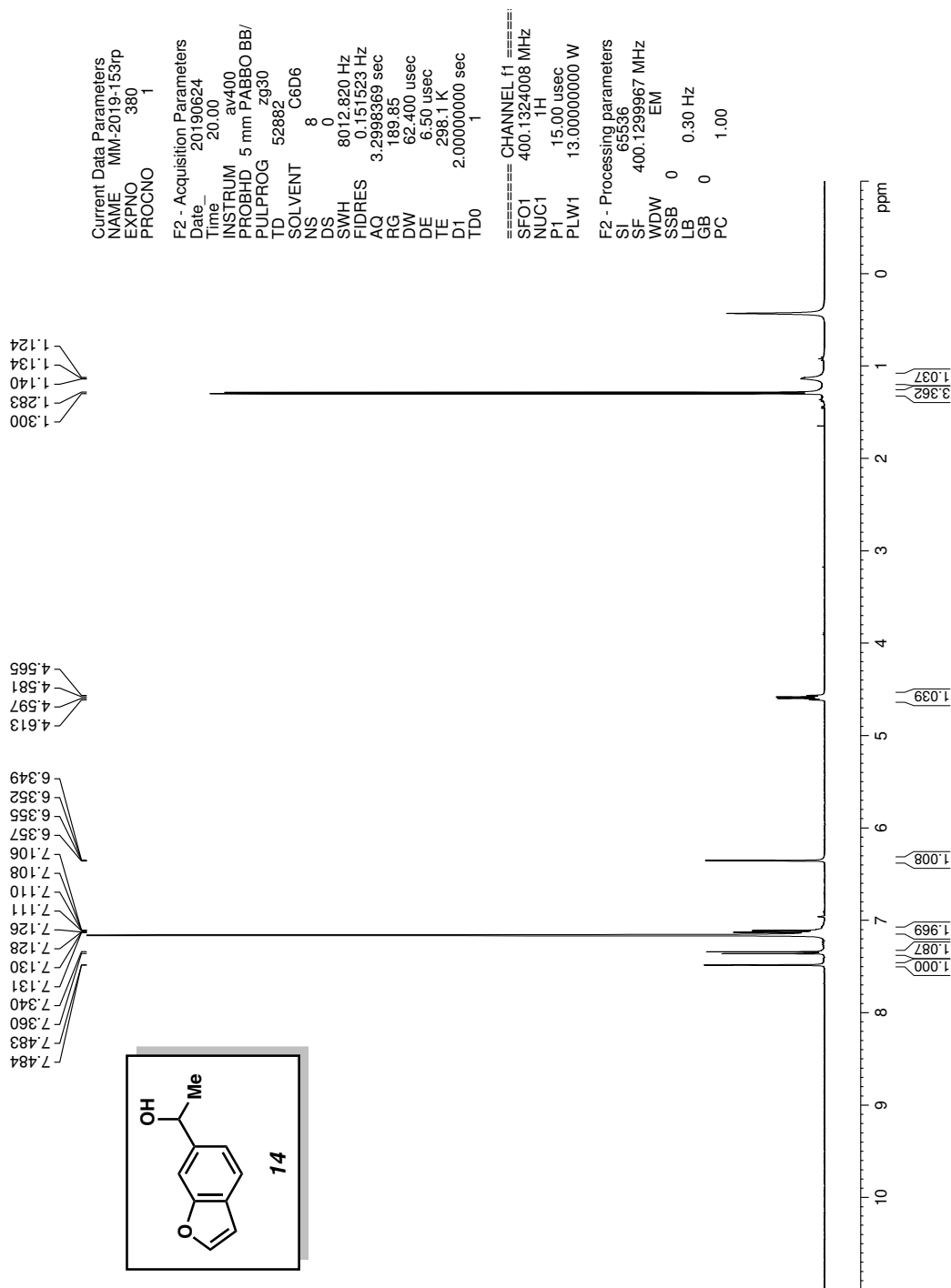


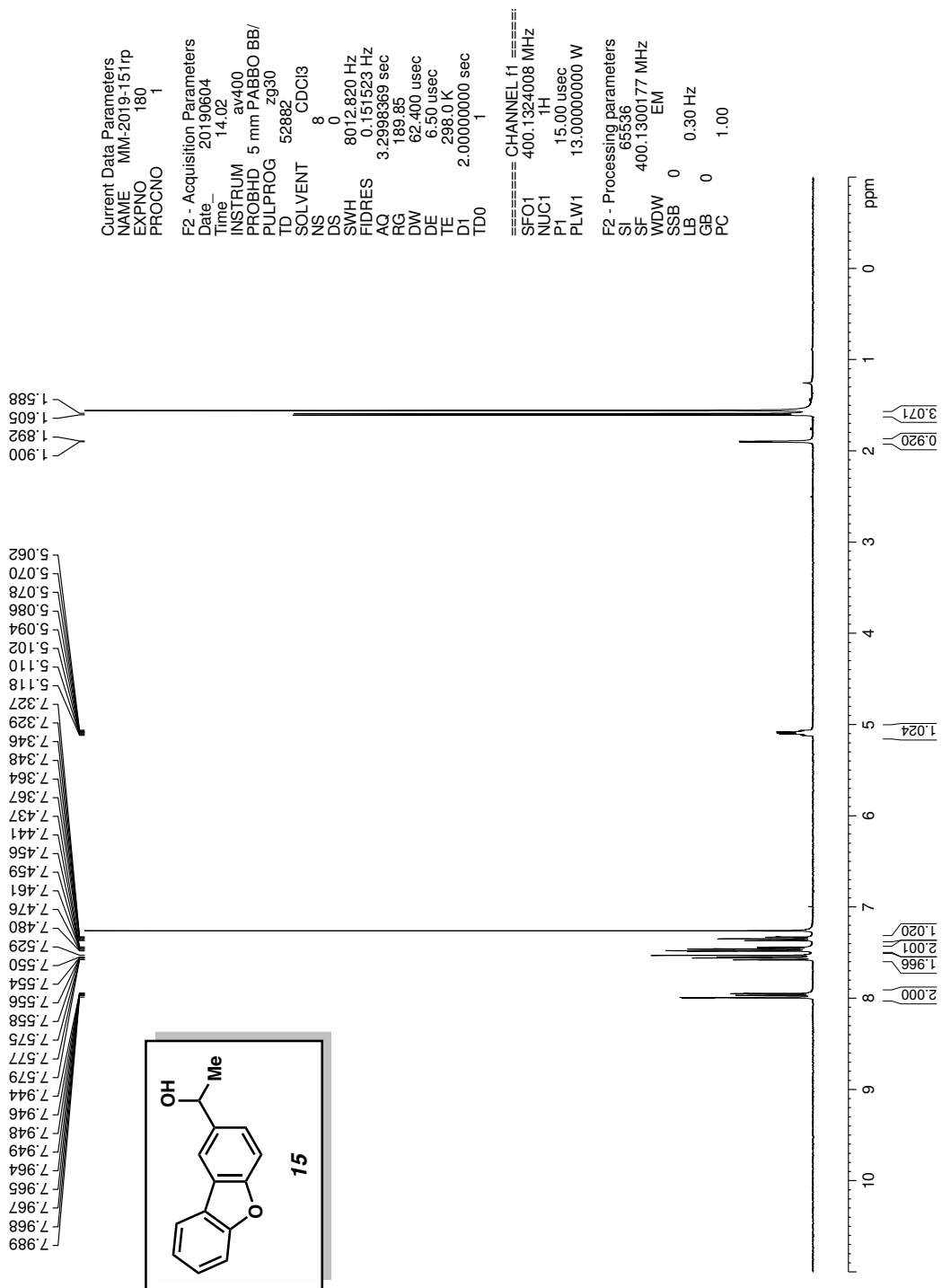


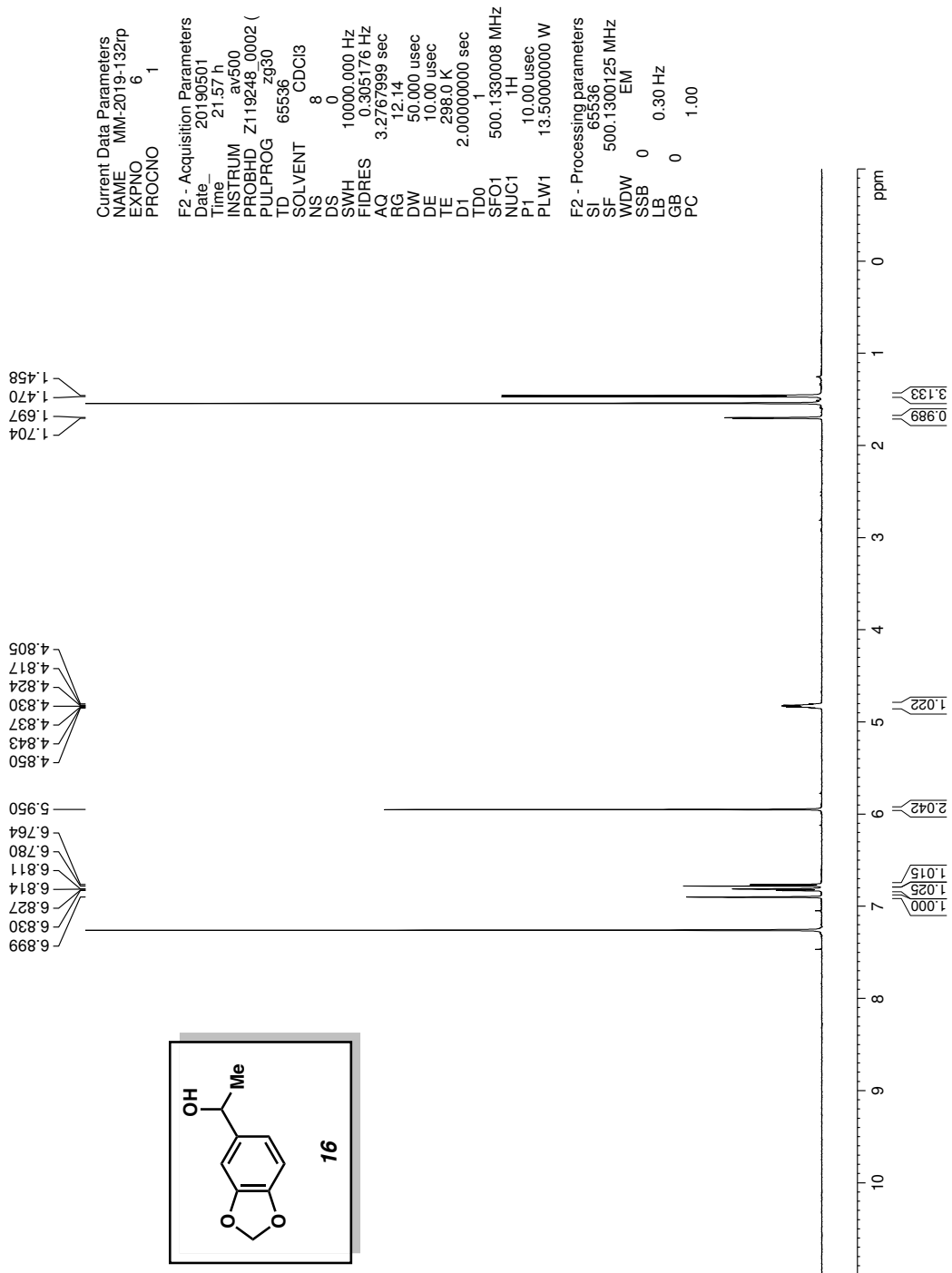


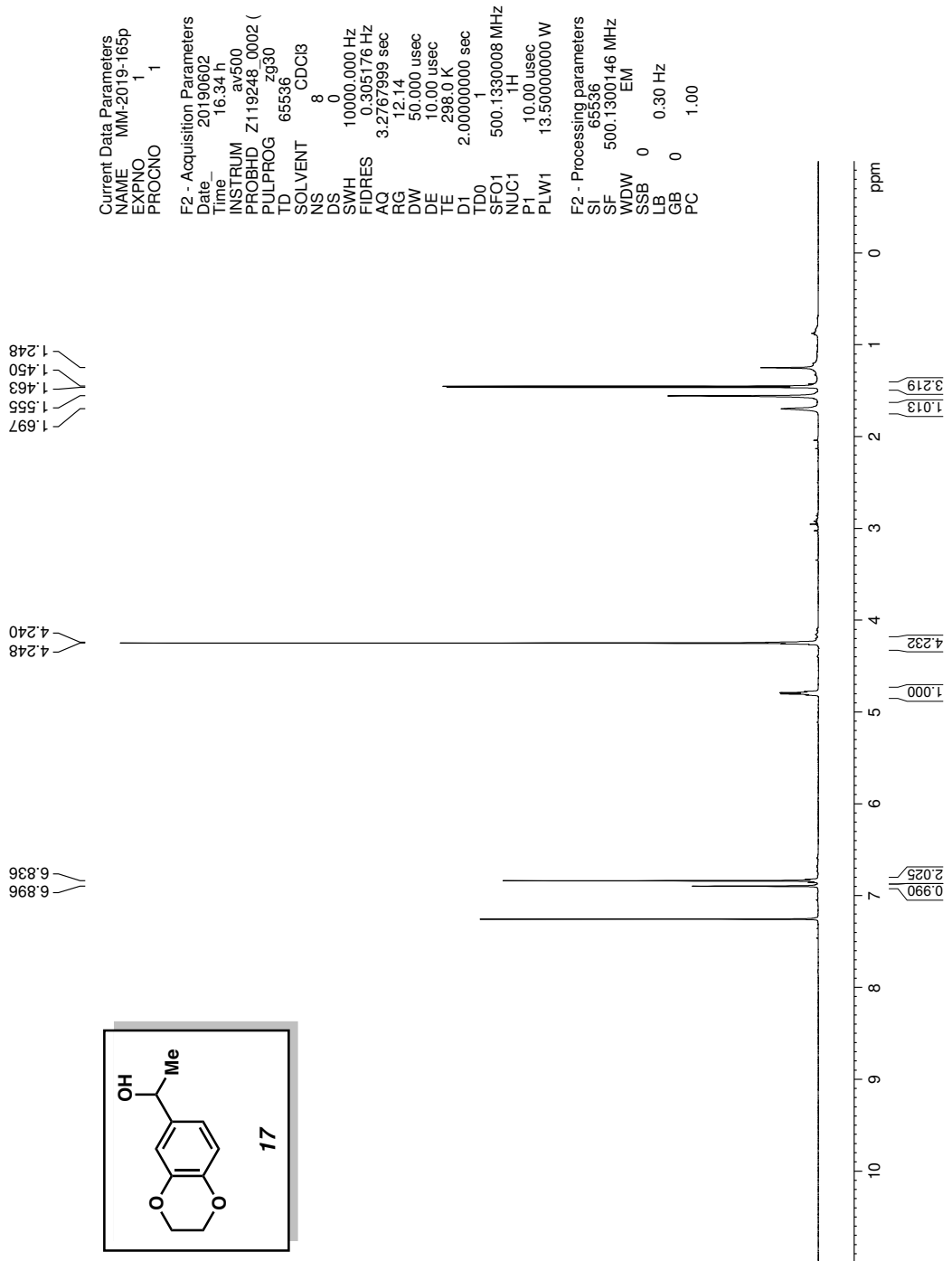


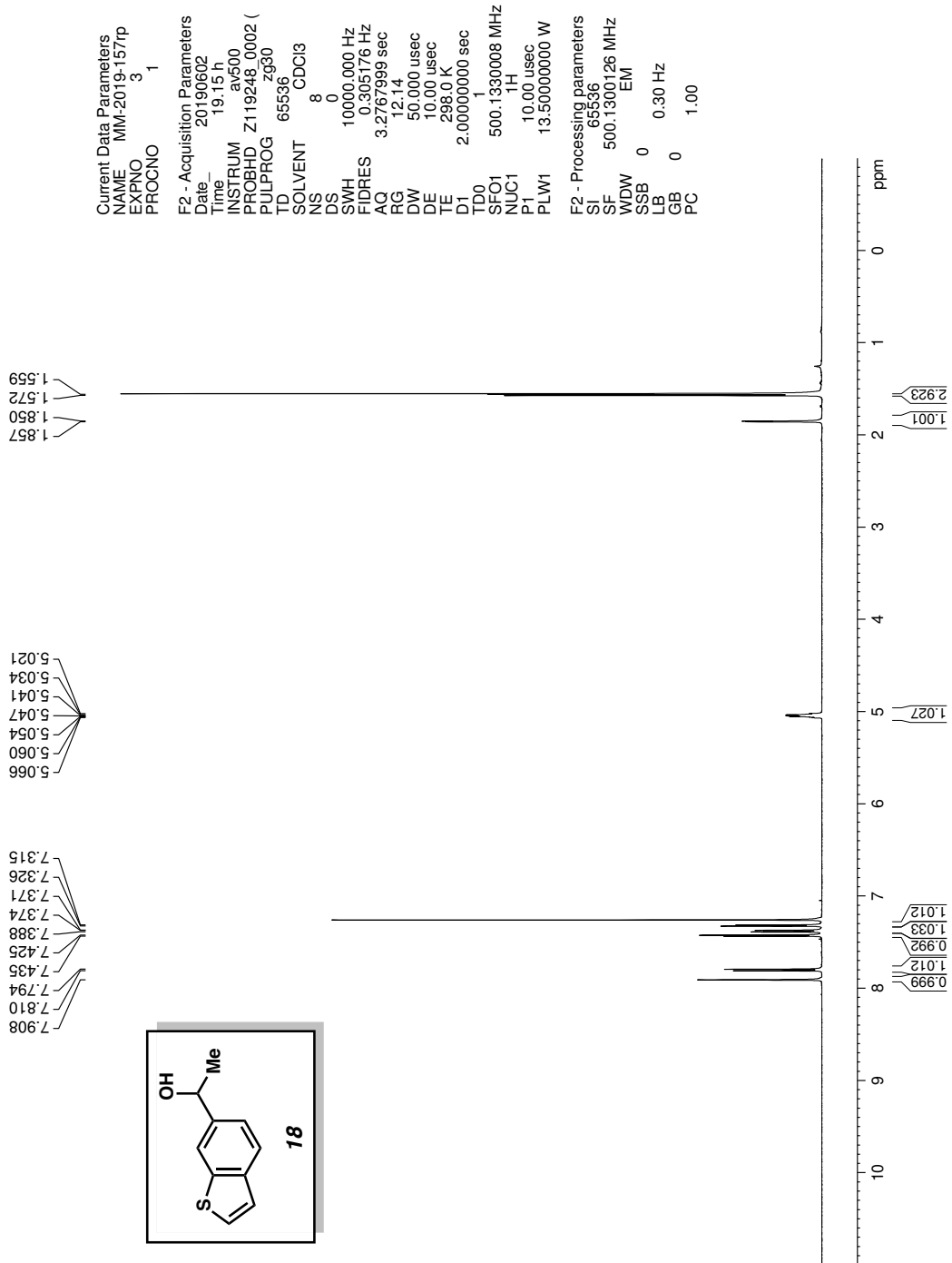


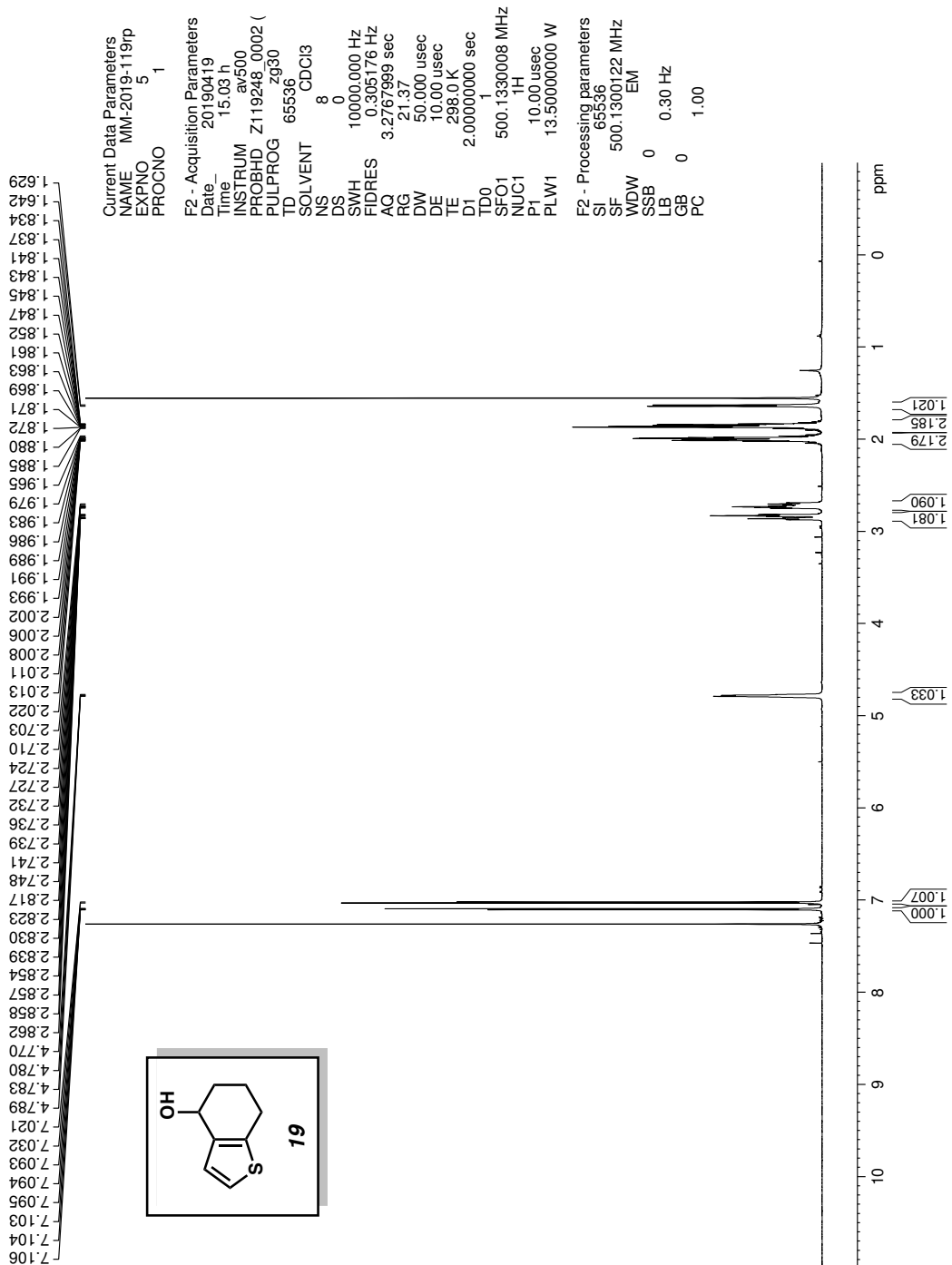










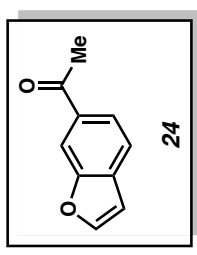
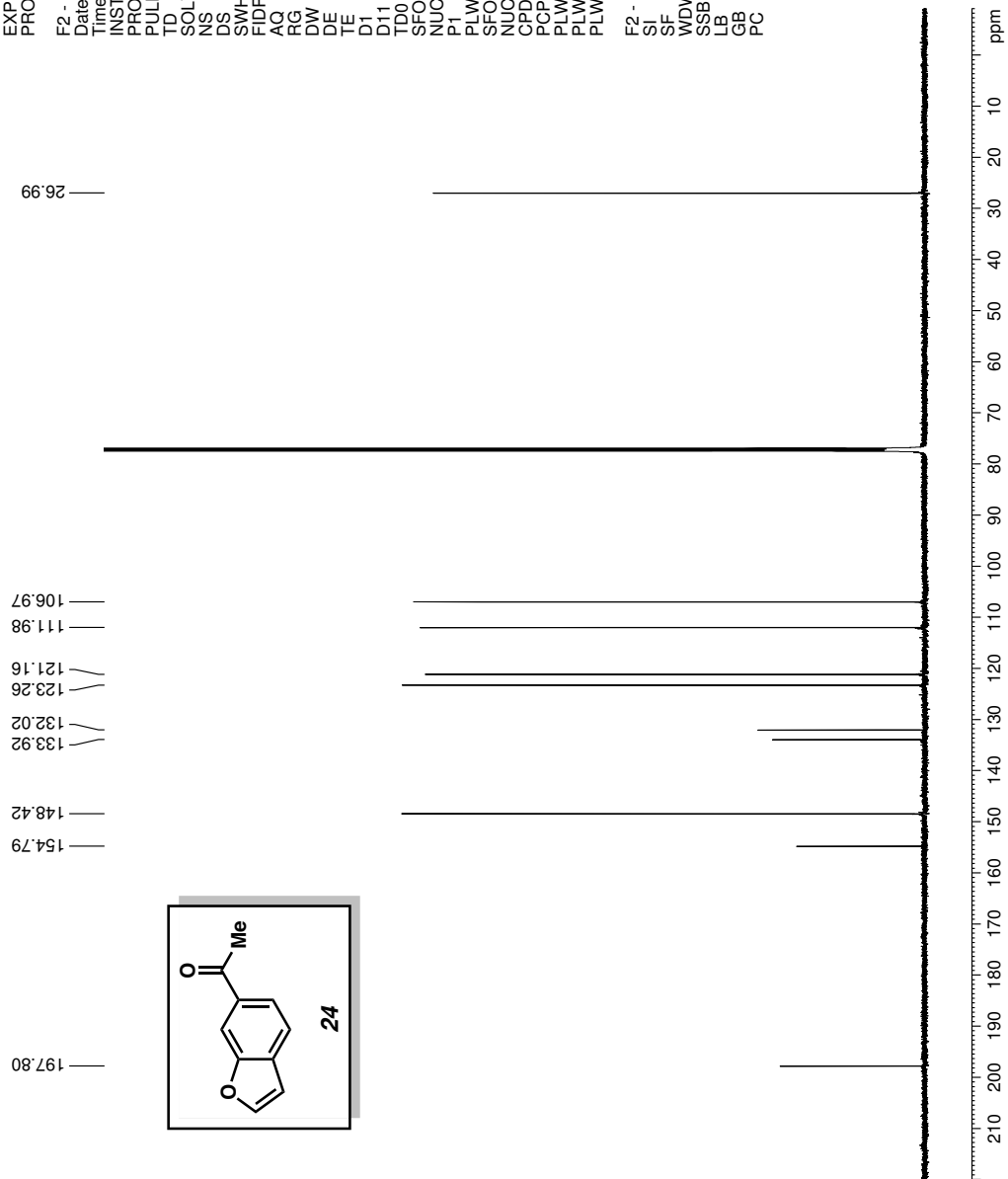


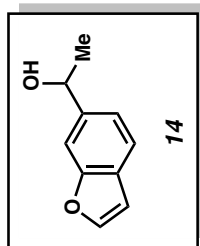
¹³C NMR Spectra

Current Data Parameters
 NAME MM-2019-159p
 EXPNO 6
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20190711
 Time_ 11.28 h
 INSTRUM av500
 PROBHD Z119248_0002 (zpgp930)
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 304
 DS 2
 SWH 31250.000 Hz
 FIDRES 0.953674 Hz
 AQ 1.0485760 sec
 RG 204.54
 DW 16.000 usec
 DE 28.00 usec
 TE 298.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1
 SFO1 125.7722511 MHz
 NUC1 13C
 P1 10.50 usec
 PLW1 23.00000000 W
 SFO2 500.1330008 MHz
 NUC2 1H
 CPDPRG2 waltz16
 PCPD2 80.00 usec
 PLW2 13.50000000 W
 PLW12 0.21094000 W
 PLW13 0.10610000 W

F2 - Processing parameters
 SI 131072
 SF 125.7577723 MHz
 WDW EM
 SSB 0
 LB 0 1.00 Hz
 GB 0
 PC 1.40

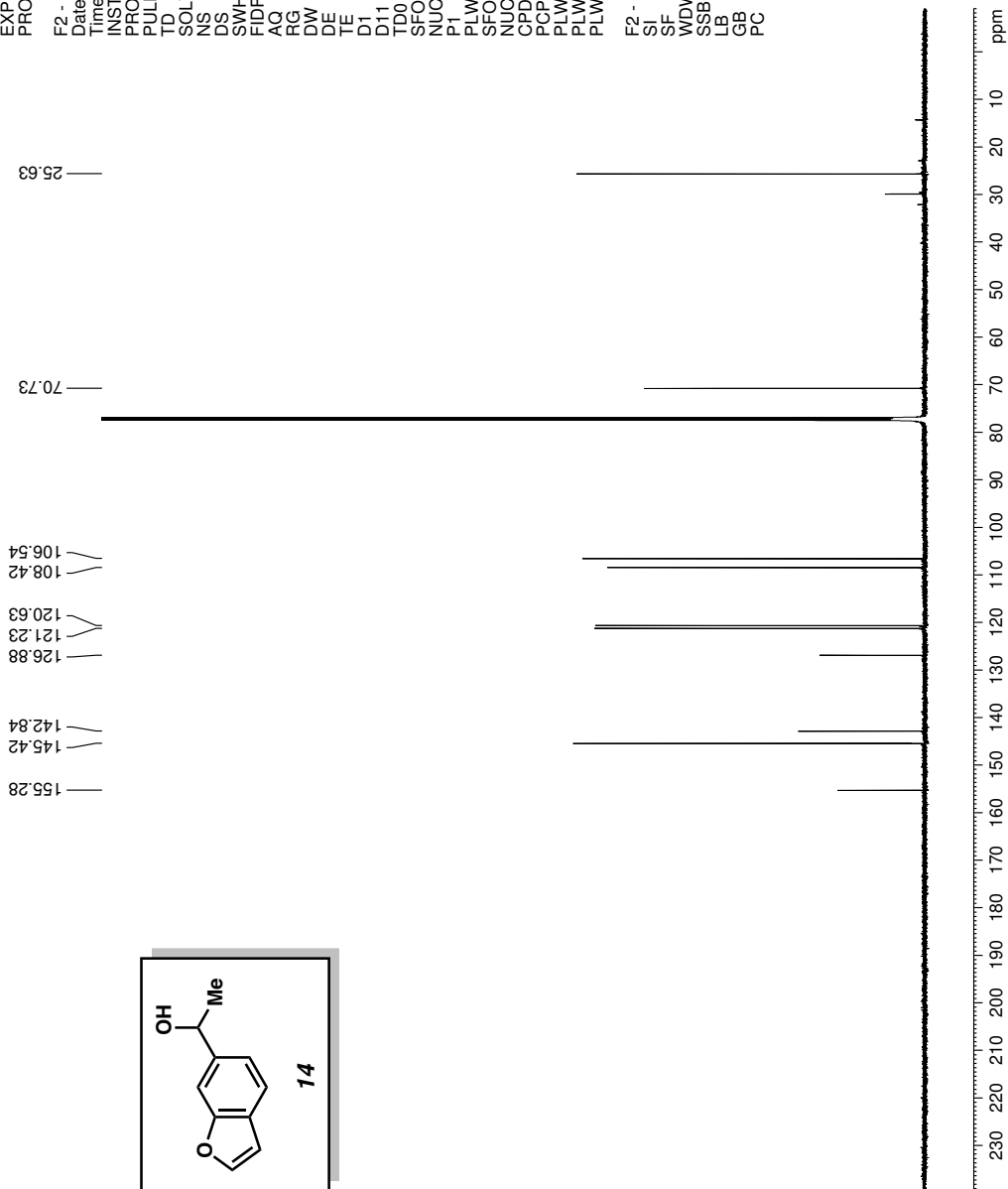


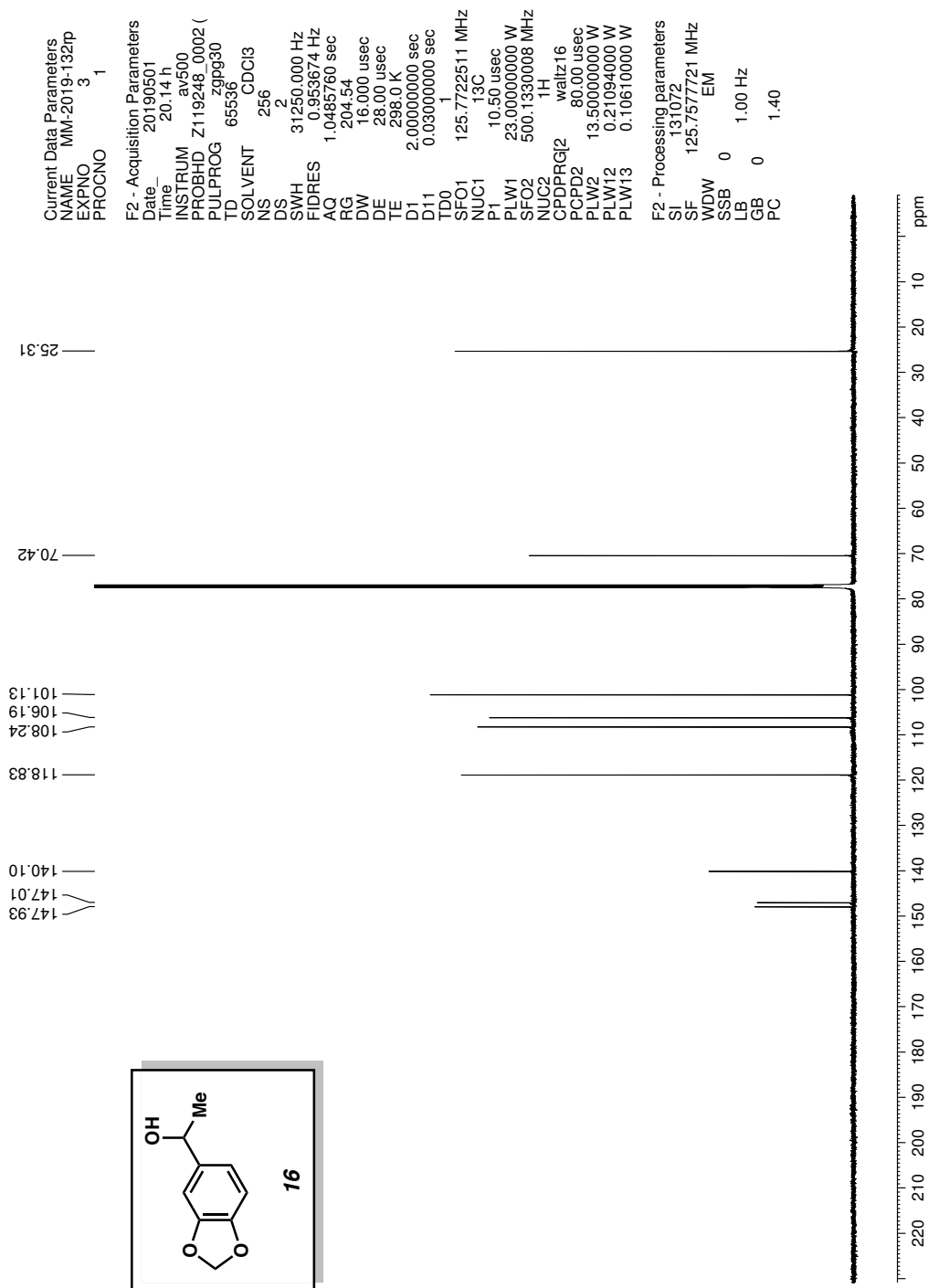


Current Data Parameters
 NAME MM-2019-146rrp
 EXPNO 6
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20190711
 Time 11:48 h
 INSTRUM av500
 PROBHD Z119248_0002 (zpgg30)
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 312
 DS 2
 SWH 31250.000 Hz
 FIDRES 0.953674 Hz
 AQ 1.0485760 sec
 RG 204.54
 DW 16.000 usec
 DE 28.000 usec
 TE 298.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1
 SFO1 125.772511 MHz
 NUC1 13C
 P1 10.50 usec
 PLW1 23.00000000 W
 SFO2 500.1330008 MHz
 NUC2 1H
 CPDPRG2 waltz16
 PCPD2 80.00 usec
 PLW2 13.50000000 W
 PLW12 0.21094000 W
 PLW13 0.10610000 W

F2 - Processing parameters
 SI 131072
 SF 125.757723 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40





Current Data Parameters
 NAME MM-2019-157p
 EXPNO 50
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20190604
 Time 9.12
 INSTRUM av400
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 512
 DS 0
 SWH 25252.525 Hz
 FIDRES 0.385323 Hz
 AQ 1.2976128 sec
 RG 189.85
 DW 19.800 usec
 DE 6.50 usec
 TE 298.7 K
 D1 2.0000000 sec
 D11 0.0300000 sec
 TD0 1

==== CHANNEL f1 =====
 SFO1 100.6243395 MHz
 NUC1 13C
 P1 10.00 usec
 PLW1 52.00000000 W

==== CHANNEL f2 =====
 SFO2 400.1324008 MHz
 NUC2 1H
 CPDPRG2 waltz16
 PCPD2 90.00 usec
 PLW2 13.00000000 W
 PLW12 0.36110000 W
 PLW13 0.29249999 W

F2 - Processing parameters
 SI 65536
 SF 100.6127572 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

