



Supplementary Information

Exploring $\text{Sn}_x\text{Ti}_{1-x}\text{O}_2$ Solid Solutions Grown onto Graphene Oxide (GO) as Selective Toluene Gas Sensors

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Table S1. Binding Energies (B. E.) relative to different Sn and Ti oxidation states, and ratios between counts of each peak and the total counts, for 32:1 SnO_2/GO , 32:1 $\text{Sn}_{0.71}\text{Ti}_{0.29}\text{O}_2/\text{GO}$, 32:1 $\text{Sn}_{0.55}\text{Ti}_{0.45}\text{O}_2/\text{GO}$, 32:1 $\text{Sn}_{0.21}\text{Ti}_{0.79}\text{O}_2/\text{GO}$ and 32:1 TiO_2/GO representative samples.

Atom	32:1 SnO_2/GO		32:1 $\text{Sn}_{0.71}\text{Ti}_{0.29}\text{O}_2/\text{G}$		32:1 $\text{Sn}_{0.55}\text{Ti}_{0.45}\text{O}_2/\text{G}$		32:1 $\text{Sn}_{0.21}\text{Ti}_{0.79}\text{O}_2/\text{G}$		32:1 TiO_2/GO	
	B. E. (eV)	Rati o	B. E. (eV)	Rati o	B. E. (eV)	Rati o	B. E. (eV)	Rati o	B. E. (eV)	Rati o
Sn(II)	–	–	485.3	0.38	484.9	0.12	484.2	0.33	–	–
Sn(III)	–	–	486.8	0.42	486.1	0.56	486.1	0.57	–	–
Sn(IV)	486.9	1.00	488.5	0.30	487.5	0.32	487.5	0.10	–	–
Ti(III)	–	–	457.5	0.40	457.3	0.36	456.4	0.32	457.3	0.33
Ti(IV)	–	–	459.4	0.36	458.6	0.44	458.0	0.32	458.3	0.51
Ti(IV+ δ) +	–	–	460.9	0.24	460.2	0.20	459.1	0.36	459.6	0.16

Table S2. Response (t_{res}) and recovery (t_{rec}) times relative to 1 ppm of (a) toluene and (b) acetone molecules, obtained at 350 °C, without UV light, in simulated air (20% O₂ – 80% N₂).

(a)

Sensor	t_{res} (s)	t_{rec} (s)
32:1 SnO ₂ /GO	20	30
32:1 Sn _{0.71} Ti _{0.29} O ₂ /GO	50	55
32:1 Sn _{0.55} Ti _{0.45} O ₂ /GO	36	65
32:1 Sn _{0.44} Ti _{0.56} O ₂ /GO	–	–
32:1 Sn _{0.35} Ti _{0.65} O ₂ /GO	55	60
32:1 Sn _{0.21} Ti _{0.79} O ₂ /GO	100	50
32:1 TiO ₂ /GO	25	30

(b)

Sensor	t_{res} (s)	t_{rec} (s)
32:1 SnO ₂ /GO	65	70
32:1 Sn _{0.55} Ti _{0.45} O ₂ /GO	30	50
32:1 TiO ₂ /GO	20	30

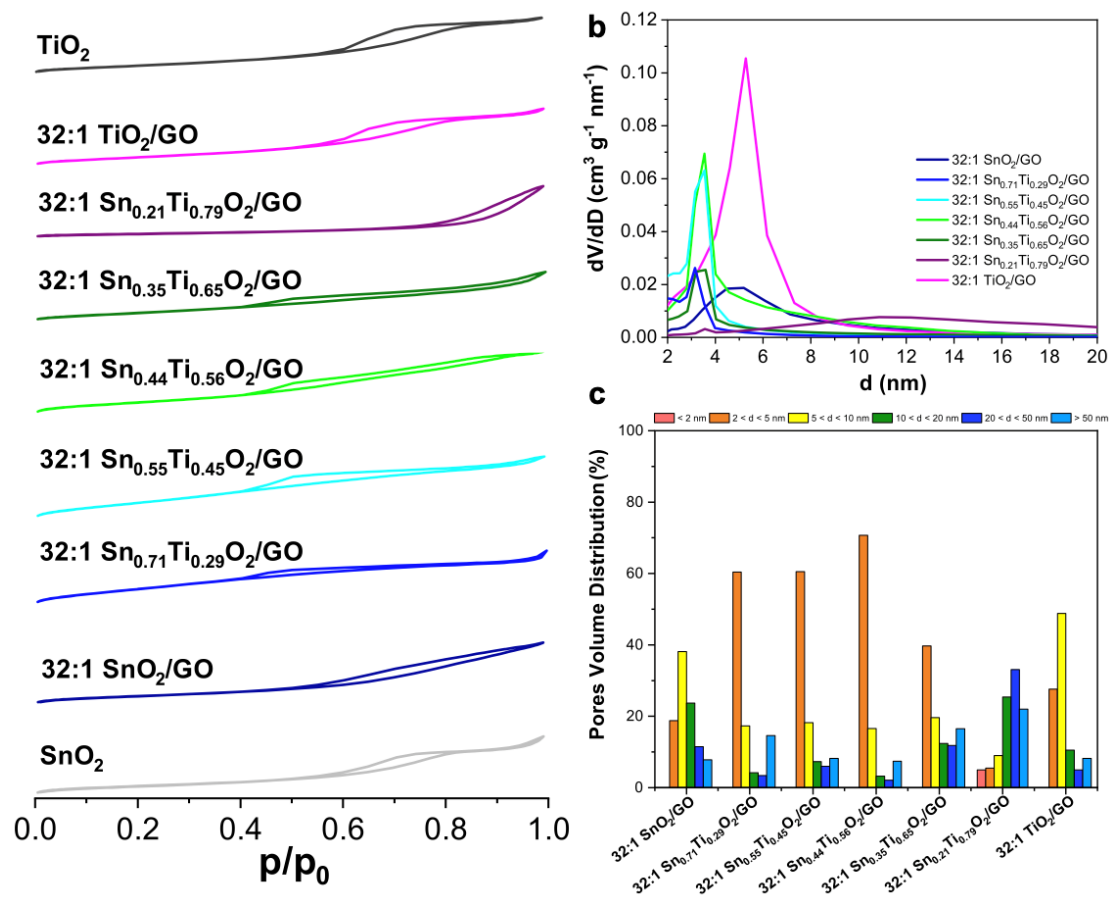


Figure S1. (a) Comparison of BET isotherms from pure cassiterite SnO_2 , 32:1 SnO_2/GO to 32:1 TiO_2/GO , pure anatase TiO_2 . (b,c) Pores size distribution by BET-BJH analysis for all the solid solutions.

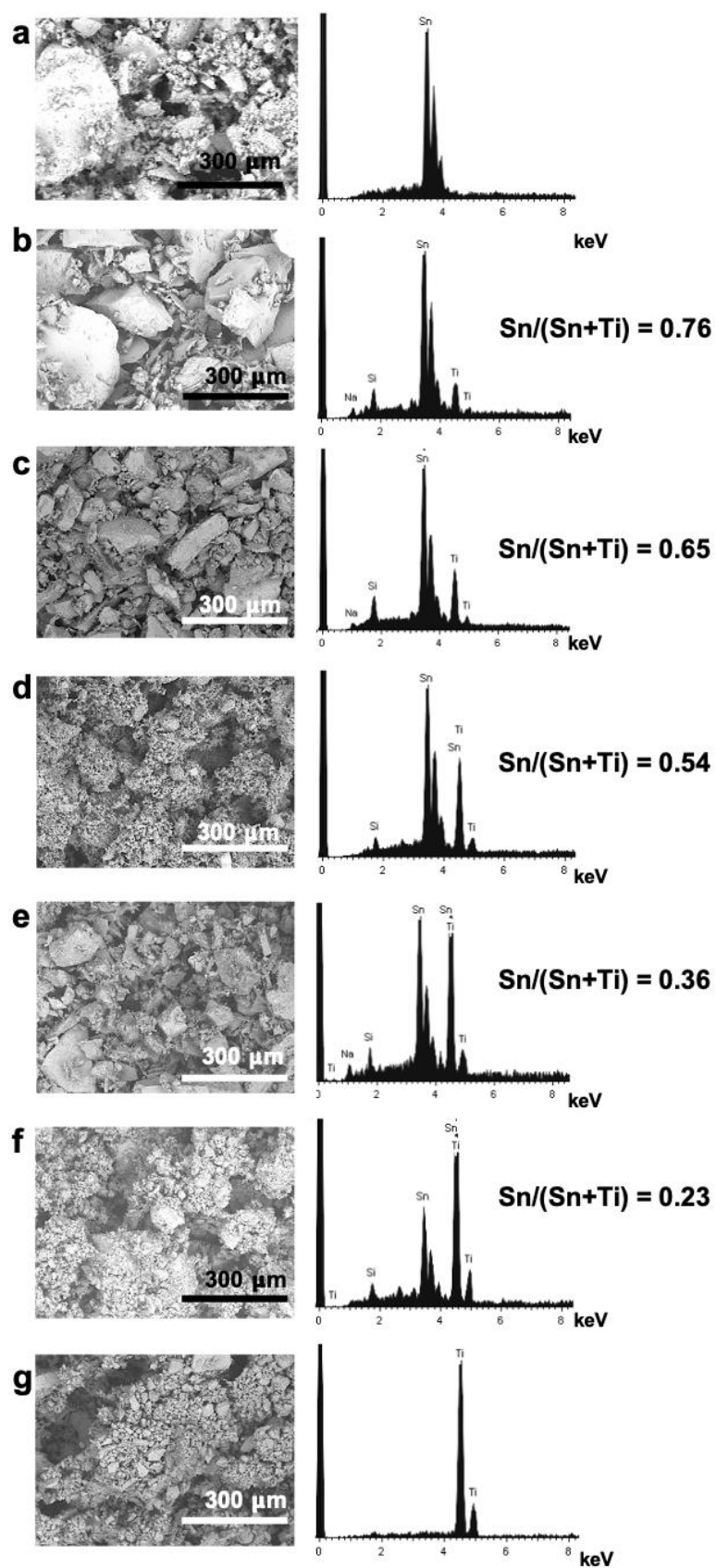


Figure S2. SEM images alongside with the relative EDX spectra of (a) 32:1 SnO₂/GO, (b) 32:1 Sn_{0.71}Ti_{0.29}O₂/GO, (c) 32:1 Sn_{0.55}Ti_{0.45}O₂/GO, (d) 32:1 Sn_{0.44}Ti_{0.56}O₂/GO, (e) 32:1 Sn_{0.35}Ti_{0.65}O₂/GO, (f) 32:1 Sn_{0.21}Ti_{0.79}O₂/GO, (g) 32:1 TiO₂/GO. The molar ratios Sn/(Sn+Ti) by EDX analysis have been reported, accordingly.

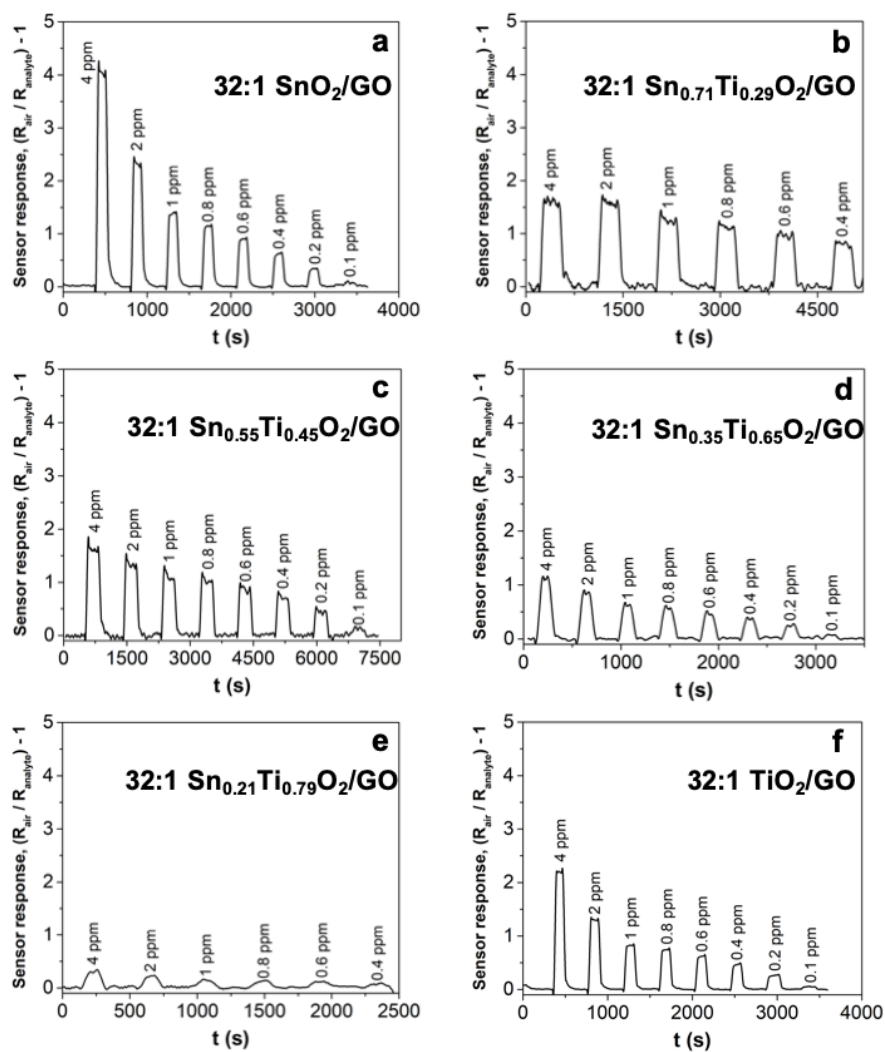


Figure S3. Toluene sensing by (a) 32:1 SnO₂/GO, (b–e) mixed oxides (with the exception of 32:1 Sn_{0.44}Ti_{0.56}O₂/GO, since it did not show any signal), and (f) 32:1 TiO₂/GO compounds. Tests were performed at 350 °C, without UV light, in simulated air (20% O₂ – 80% N₂).