

Enhanced rate capabilities in a glass-ceramic-derived sodium all-solid-state battery

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Table S1. Properties of all-solid-state batteries (ASSBs) examined in this study: specific surface areas of Na₂FeP₂O₇ (NFP) and β'' -alumina, average particle size, calcination temperature and time, composition of the cathode layer, weight of cathode materials, thickness of the cathode layer, and cell capacity.

Cell	Powder		Cathode				Cell capacity (as NFP-80mAh/g) (mAh/cm ²)			
	Calcination condition		Composition (wt%)		Cathode materials (mg)	Thickness (μm)				
	NFP	β'' -Alumina	NFP	β'' -Alumina						
A	10(0.8*)	5(2*)	550°C-1hour	72	25	3	5.41	30	3.9	0.31
B	27(0.2**)	45(0.1**)*	525°C-0.5hour	83	13.5	3.5	4.2	25	3.49	0.28
C	27(0.2**)	45(0.1**)*	525°C-0.5hour	83.4	12.4	4.2	3.79	22	3.16	0.25
D	27(0.2**)	45(0.1**)*	525°C-0.5hour	83.4	12.4	4.2	5.41	40	4.51	0.36
E	27(0.2**)	45(0.1**)*	525°C-0.5hour	83.4	12.4	4.2	4.41	30	3.68	0.36***
Ref. 1M NaPF ₆ in EC/DEC	N/A(1.7*)	—	—	90	5(PVDF)	5	6.88	40	6.19	0.6

*Laser diffraction particle size diffraction analysis, **Analysis of SEM images, *** Cell capacity as 97mAh/g of NFP

Table S2. Analysis of the Nyquist plot of Cell A.

Battery components	Materials	Composites	Symmetric cells	All-solid-state Na ion battery
Cathode (t35μm)	Na ₂ FeP ₂ O ₇ glass ceramic (calc. as thickness 35μm) Acetylene black (calc. as thickness 35μm) β"-Alumina powder (calc. as thickness 2μm)	R _{total} 35000Ω R _{bulk} N/A R _{grain boundary} N/A R _{total} 0.1Ω R _{NIP} R _{NIP-SE powder} 220000Ω R _{total} 31000Ω R _{9000Ω} R _{cathode} 27000Ω		
Interface			R _{interface} 2350Ω	R _{total} 3873Ω
Solid electrolyte (t1mm)	β"-Alumina	R _{total} 72Ω R _{bulk} 27Ω R _{grain boundary} 45Ω	R _{total} 73Ω R _{bulk} 25Ω R _{gb} 48Ω	R _{total} 72Ω R _{bulk} 27Ω R _{gb} 45Ω
Interface			R _{interface} 125Ω	R _{total} 2571Ω
Anode (t35μm)	Na metal	R 1.0Ω	R _{anode} 20Ω	R _{anode} 65Ω

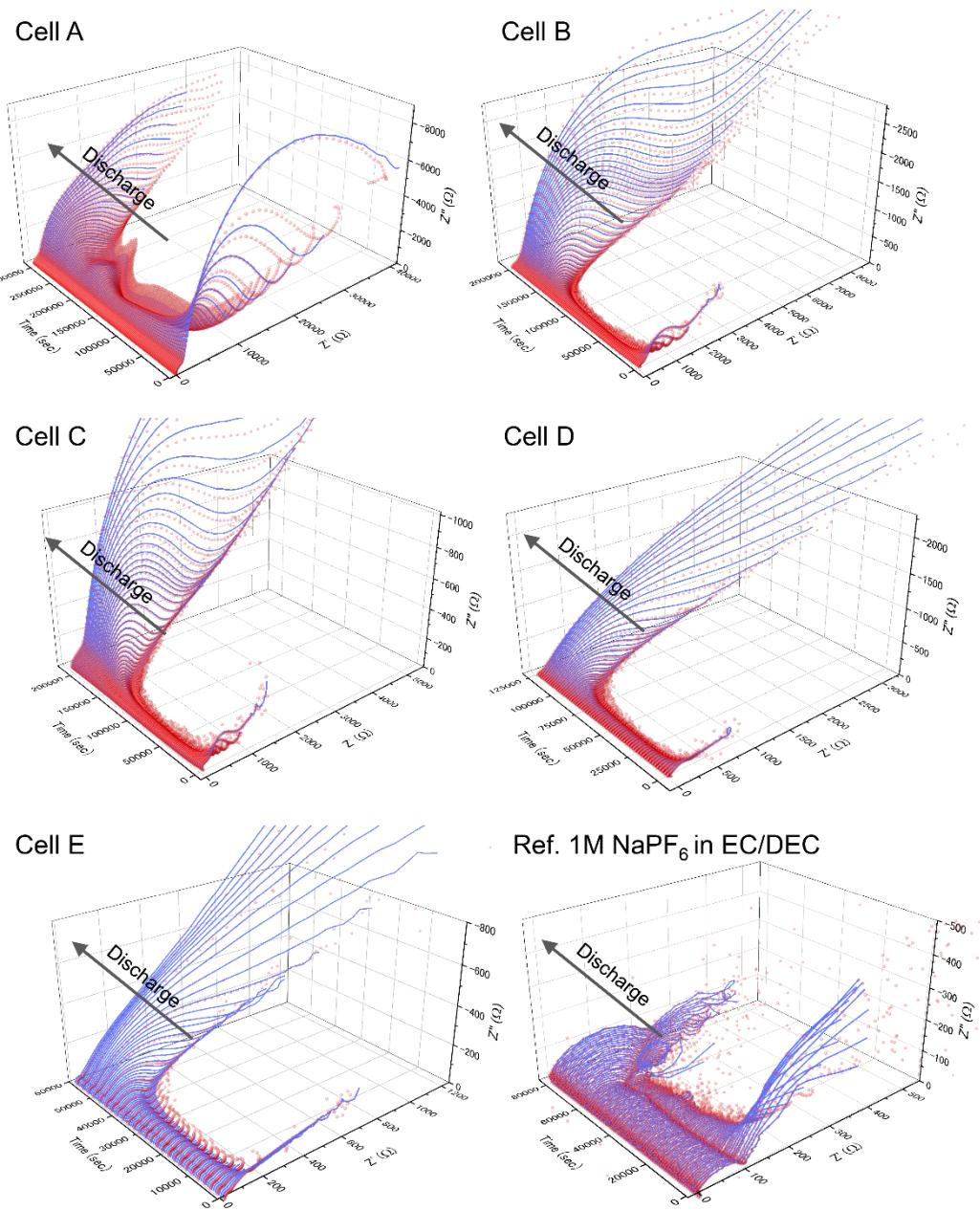


Figure S1. Three-dimensional (3D) impedance spectra of Cells A–E and 1 M NaPF₆ in EC/DEC acquired during the discharge process.

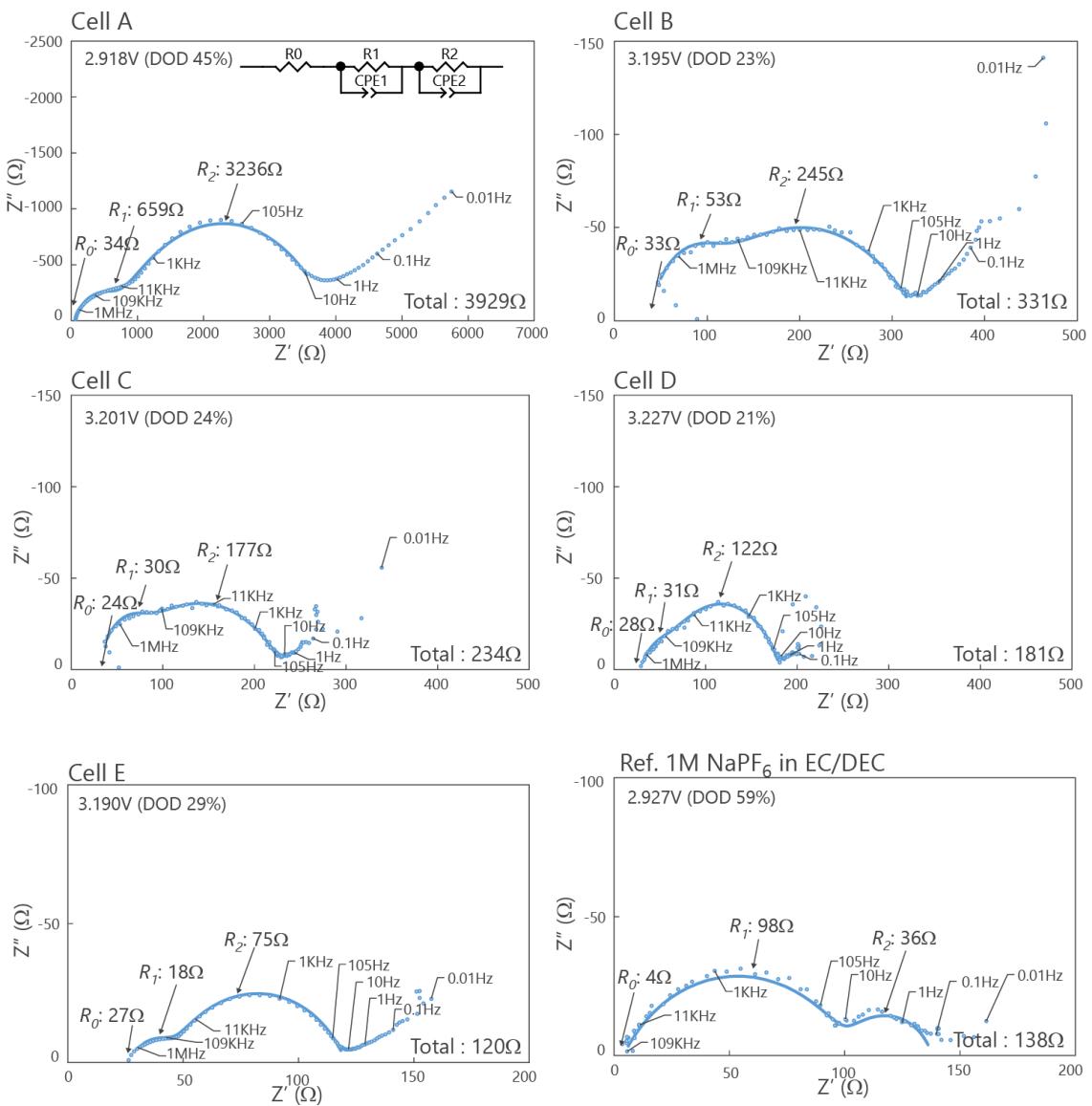


Figure S2. Nyquist plots of Cells A–E and 1 M NaPF₆ in EC/DEC obtained from 3D

impedance spectra (Fig. S1).

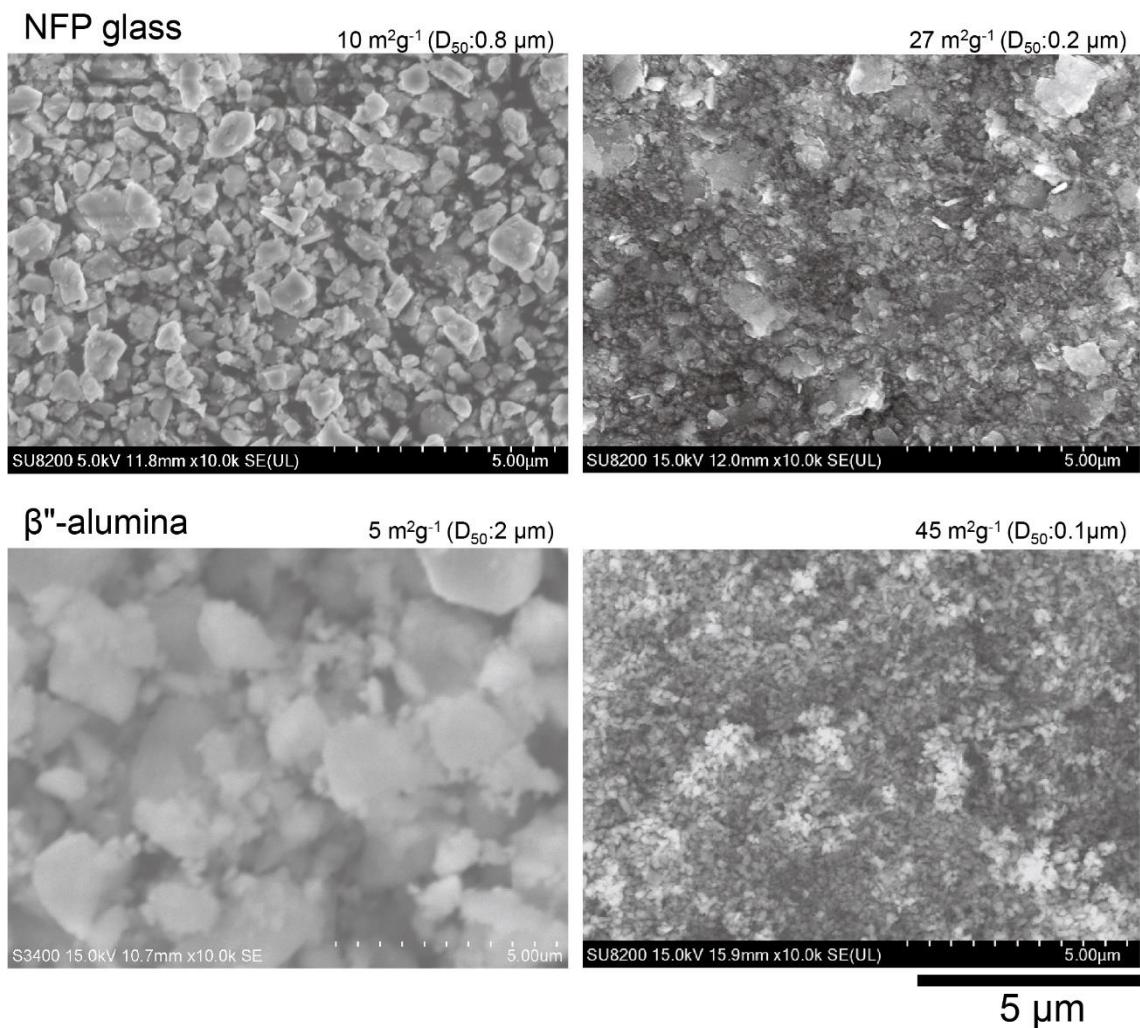


Figure S3. SEM images of NFP glass powder and β"-alumina solid solution.

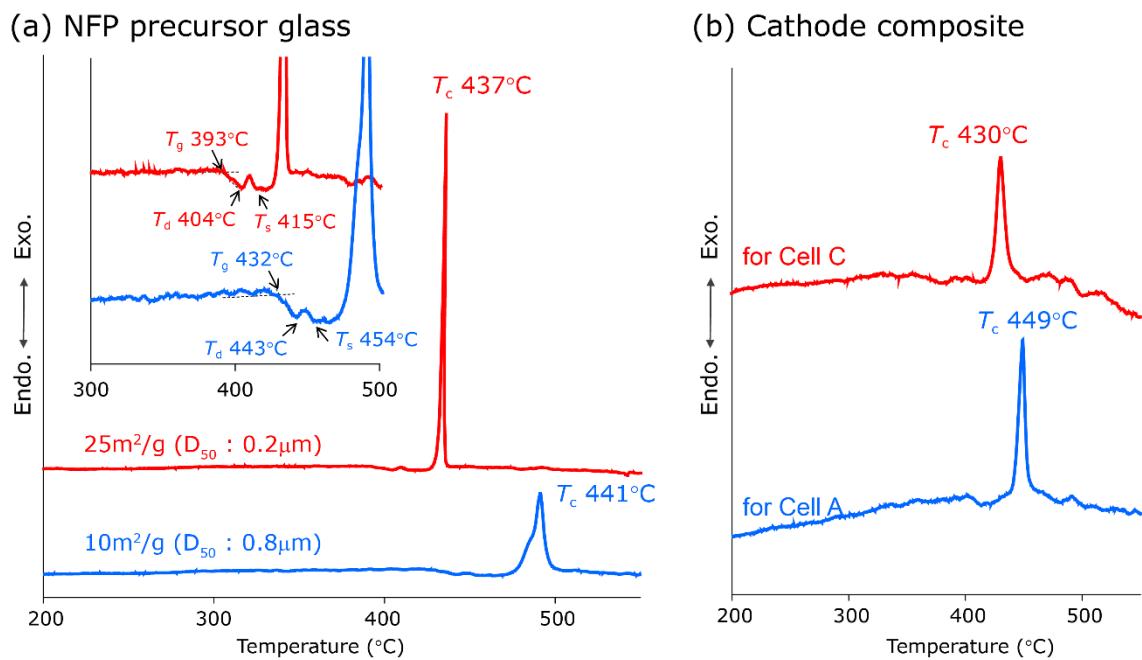


Figure S4. Differential thermal analysis curves of NFP glasses with different specific surface areas. T_d represents the yield point and T_s represents the softening point.

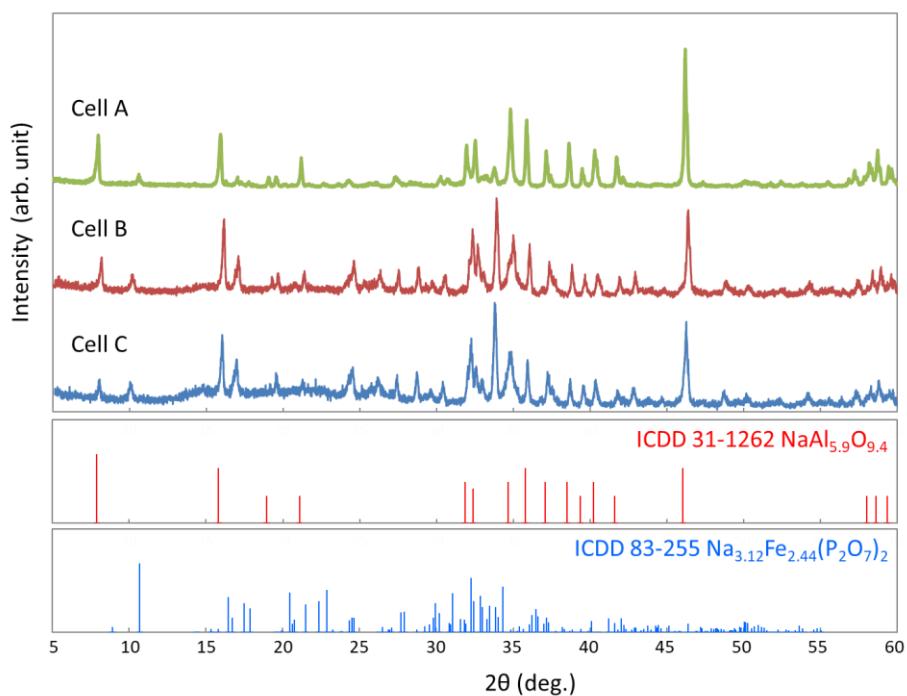


Figure S5. X-ray diffraction patterns of the NFP glass-ceramic cathode on β'' -alumina solid electrolyte substrate calcined at 550 °C for 60 min (Cell A) and 525 °C for 30 min (Cells B and C) in H₂/N₂ (4/96 v/v%). Cathode composition, NFP glass-ceramic : β'' -alumina : AB = 72 : 25 : 3 wt% (Cell A), 83 : 13.5 : 3.5 wt% (Cell B), 83.4 : 12.4 : 4.2 wt% (Cell C).

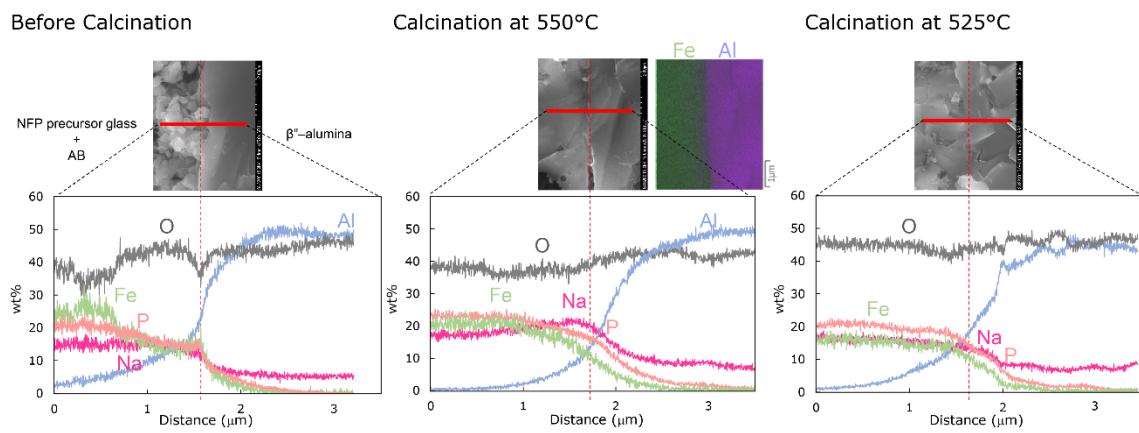


Figure S6. Scanning electron microscopy images of the NFP– β'' -alumina interface and linear analysis profiles of components. The observations were made at an acceleration voltage of 15 kV and 25000 \times magnification.

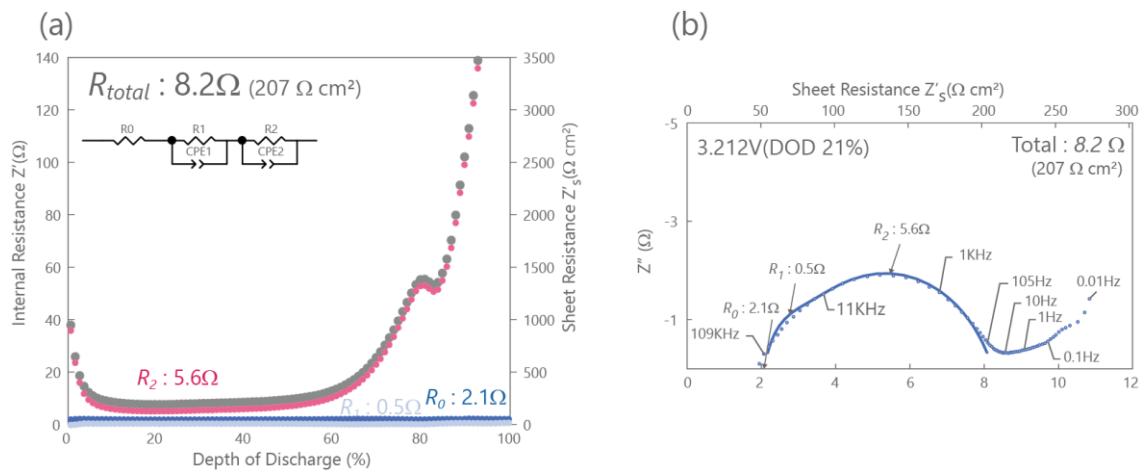


Figure S7. (a) Dependence of the internal resistance Z' (Ω) and sheet resistance Z'_s ($\Omega \text{ cm}^2$) of the pouch cell based on Cell C on the depth of discharge, (b) Nyquist plot of the pouch cell obtained from 3D impedance spectra.

Supplementary video

Electric fan powered by the $\text{Na}_2\text{FeP}_2\text{O}_7$ glass ceramic cathode| β'' -alumina|Na pouch cell sandwiched between two ice blocks.