Supporting Information

Photocatalytic Deoxygenation of Sulfoxides Using Visible Light: Mechanistic Investigations and Synthetic Applications

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General Information

Except where stated, all reagents were purchased from commercial sources and used without further purification. Anhydrous CH₂Cl₂ and toluene were obtained from an Innovative Technology Inc. PureSolv® solvent purification system. Anhydrous THF was obtained by distillation over sodium benzophenone ketyl immediately before use. ¹H NMR and ¹³C NMR spectra were recorded on a JEOL ECX400 or JEOL ECS400 spectrometer, operating at 400 MHz and 100 MHz. All spectral data was acquired at 295 K unless stated otherwise. Chemical shifts (δ) are quoted in parts per million (ppm). The residual solvent peaks, δ_H 7.26 and δ_c 77.16 for CDCl₃ were used as a reference. Coupling constants (J) are reported in Hertz (Hz) to the nearest 0.1 Hz. The multiplicity abbreviations used are: br s broad singlet, s singlet, d doublet, t triplet, q quartet, dt doublet of triplets, m multiplet. Signal assignment was achieved by analysis of DEPT, COSY, HMBC and HSQC experiments where required. Infrared (IR) spectra were recorded on a PerkinElmer UATR 2 spectrometer as a thin film dispersed from either CH₂Cl₂ or CDCl₃. Mass spectra (high-resolution) were obtained by the University of York Mass Spectrometry Service, using Electrospray Ionisation (ESI) or Atmospheric Pressure Chemical Ionisation (APCI) on a Bruker Daltonics, Micro-tof spectrometer. Melting points were determined using Gallenkamp apparatus. Thin layer chromatography was carried out on Merck silica gel 60F₂₅₄ pre-coated aluminium foil sheets and were visualised using UV light (254 nm) and stained with basic aqueous potassium permanganate or vanillin. Flash column chromatography was carried out using slurry packed Fluka silica gel (SiO₂), 35–70 µm, 60 Å, under a light positive pressure, eluting with the specified solvent system.

List of starting materials

All the starting materials used in this publication are listed below. Commercially available starting materials (denoted with a *) were used as supplied, those with a reference number are known compounds prepared via the cited literature method, while for all others, preparative details and spectroscopic characterization data are provided.



General procedures

General procedure A – Sulfoxide preparation

$$R^{1}S_{R^{2}} \xrightarrow{mCPBA} O_{\parallel}$$

 $CH_{2}CI_{2}, 0 \ ^{\circ}C \text{ to } RT R^{1}S_{R^{2}}^{\circ}$

To a solution of sulfide (1.00 mmol) in anhydrous CH_2Cl_2 (3 mL) at 0 °C was added 3-chloroperbenzoic acid (1.20 mmol) portionwise. The reaction mixture was left to gradually warm to RT and stirred overnight. The reaction was quenched by the addition of sat. aq. NaHCO₃ (1 mL) and stirred for 5 min. Following this, the reaction mixture was concentrated *in vacuo* and purified by column chromatography to afford the sulfoxide product.

General procedure B – Sulfoxide reduction

$$\mathbb{R}^{1^{S}}\mathbb{R}^{2} \xrightarrow{ [Ir\{dF(CF_{3})ppy\}_{2}(dtbpy)]PF_{6} \ (1 \text{ mol}\%) \ PPh_{3} (1.2 \text{ equiv.})}{ CH_{2}Cl_{2}, RT, hv} \mathbb{R}^{1^{S}}\mathbb{R}^{2}$$

To an 8 mL vial equipped with a PTFE septum and a magnetic stirrer bar was added [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (0.003 mmol, 0.01 equiv.), PPh₃ (0.36 mmol, 1.2 equiv.) and sulfoxide (if solid, 0.30 mmol, 1.0 equiv.). The reaction vessel was purged by alternating vacuum and argon three times before anhydrous and degassed CH₂Cl₂ (1.5 mL) was added. Sulfoxide (if liquid, 0.30 mmol, 1.0 equiv.) was added and the septum additionally sealed with paraffin film. The reaction was irradiated with a 60 W blue LED floodlight for 24 h, with rapid stirring and cooling from a small fan to maintain an ambient temperature. The reaction mixture was then directly poured on to silica and purified by column chromatography affording the desired sulfide product.

Compound characterisation data and procedures

(4-Bromophenyl)(methyl)sulfane (8b)



Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1-bromo-4-(methylsulfinyl)benzene **8a** (65.7 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane, then 9:1 pentane:Et₂O) afforded the title compound **8b** as a white solid (57.0 mg, 94% yield); mp 34–36 °C; R_f 0.45 (9:1 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.39 (d, *J* = 8.7 Hz, 2H), 7.12 (d, *J* = 8.7 Hz, 2H), 2.46 (s, 3H); δ_{C} (100 MHz, CDCl₃) 137.8, 131.9, 128.2, 118.7, 16.1. Spectroscopic data is consistent with those reported in the literature.⁶

Diphenylsulfane (9b)



Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and sulfinyldibenzene **9a** (60.7 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (99:1 pentane:Et₂O) afforded the title compound **9b** as a clear and colourless oil (44.6 mg, 80% yield); R_f 0.53 (99:1 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.39–7.28 (m, 8H), 7.28–7.23 (m, 2H); δ_{C} (100 MHz, CDCl₃) 135.9, 131.2, 129.3, 127.2; HRMS (APCl⁺): Found: 187.057105; C₁₂H₁₁S⁺ (MH⁺) Requires 187.057598 (2.6 ppm error). Spectroscopic data is consistent with those reported in the literature.⁷

Di-p-tolylsulfane (10b)

Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 4,4'-sulfinylbis(methylbenzene) **10a** (69.1 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane, then 99:1 pentane:Et₂O) afforded the title compound **10b** as a white solid (62.3 mg, 97% yield); mp 53–55 °C; R_f 0.22 (100% pentane); δ_{H} (400 MHz, CDCl₃) 7.24 (d, *J* = 8.0 Hz, 4H), 7.11 (d, *J* = 8.0 Hz, 4H), 2.34 (s, 6H); δ_{C} (100 MHz, CDCl₃) 137.0, 132.8, 131.2, 130.0, 21.2; HRMS (APCI⁺): Found: 215.08827; C₁₄H₁₅S⁺ (MH⁺) Requires 215.088898 (3.1 ppm error). Spectroscopic data is consistent with those reported in the literature.⁸

tert-Butyl 10H-phenothiazine-10-carboxylate 5-oxide (11a)



Prepared according to general procedure A using *tert*-butyl 10*H*-phenothiazine-10-carboxylate **11b*** (1.20 g, 4.00 mmol) and *m*CPBA (830 mg, 4.80 mmol) in anhydrous CH₂Cl₂ (12 mL). Purification by flash chromatography on silica gel (6:4 hexane:Et₂O with 2% AcOH added to eluent, then 6:4 Et₂O:hexane) afforded the title compound **11a** as a white solid (1.00 g, 79% yield); mp 146–148 °C; R_f 0.30 (6:4 Et₂O:hexane); v_{max} (thin film)/cm⁻¹ 2978, 1717, 1475, 1458, 1322, 1295, 1249, 1232, 1153, 1098, 1021, 843, 764, 728; δ_{H} (400 MHz, CDCl₃) 7.84 (dd, *J* = 7.7, 1.6 Hz, 2H), 7.73 (dd, *J* = 8.0, 1.3 Hz, 2H), 7.49 (ddd, *J* = 7.6, 7.6, 1.7 Hz, 2H), 7.43 (ddd, *J* = 7.6, 7.6, 1.3 Hz, 2H), 1.53 (s, 9H); δ_{C} (100 MHz, CDCl₃) 151.5, 138.7, 133.5, 130.2, 126.9, 126.3, 124.2, 83.6, 28.3; HRMS (ESI⁺): Found: 316.1000; C₁₇H₁₈NO₃S⁺ (MH⁺) Requires 316.1002 (0.7 ppm error). *This material is commercially available.

tert-Butyl 10H-phenothiazine-10-carboxylate (11b)



Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and *tert*-butyl 10*H*-phenothiazine-10-carboxylate 5-oxide **11a** (94.6 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (95:5 pentane:Et₂O) afforded the title compound **11b** as a white solid (88.8 mg, 99% yield); mp 110–112 °C; R_f 0.42 (8:2 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.53 (dd, *J* = 8.1, 1.4 Hz, 2H), 7.35 (dd, *J* = 7.6, 1.4 Hz, 2H), 7.30–7.24 (m, 2H), 7.15 (ddd, *J* = 7.6, 1.4, 1.4 Hz, 2H), 1.49 (s, 9H); δ_{C} (100 MHz, CDCl₃) 152.6, 138.8, 132.3, 127.6, 127.3, 126.7, 126.2, 82.2, 28.3. Spectroscopic data is consistent with those reported in the literature.⁹

Methyl(p-tolyl)sulfane (12b)



Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1-methyl-4-(methylsulfinyl)benzene **12a** (46.3 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane, then 9:1 pentane:Et₂O) afforded the title compound **12b** as a pale yellow oil (29.5 mg, 71% yield); R_f 0.57 (9:1 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.21–7.17 (m, 2H), 7.11 (d, *J* = 8.0 Hz, 2H), 2.47 (s, 3H), 2.32 (s, 3H); δ_{C} (100 MHz, CDCl₃) 135.2, 134.8, 129.7, 127.4,

21.1, 16.7; HRMS (APCI⁺): Found: 139.057984; $C_8H_{11}S^+$ (MH⁺) Requires 139.057598 (2.8 ppm error). Spectroscopic data is consistent with those reported in the literature.¹⁰

4-(Methylthio)phenol (13b)

To an 8 mL vial equipped with a PTFE septum and a magnetic stirrer bar was added $Ir[(dF(CF_3)ppy)_2(d(CF_3)bpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 4-(methylsulfinyl)phenol **13a** (46.9 mg, 0.30 mmol). The reaction vessel was purged by alternating vacuum and argon three times before anhydrous and degassed CH₂Cl₂ (1.5 mL) was added and the septum additionally sealed with paraffin film. The reaction was irradiated with a 60 W blue LED floodlight for 4 days, with rapid stirring and cooling from a small fan to maintain an ambient temperature. The reaction mixture was then directly poured on to silica and purified by column chromatography to afford the title compound **13b** as a white solid (33 mg, 78% yield); mp 77–79 °C; R_f 0.32 (8:2 pentane:Et₂O); δ_H (400 MHz, CDCl₃) 7.23 (d, *J* = 8.7 Hz, 2H), 6.79 (d, *J* = 8.7 Hz, 2H), 4.70 (br s, 1H), 2.44 (s, 3H); δ_C (100 MHz, CDCl₃) 154.2, 130.5, 129.0, 116.2, 18.2. Spectroscopic data is consistent with those reported in the literature.¹¹

Compound **13b** was also prepared according to general procedure B using $[Ir{dF(CF_3)ppy}_2(dtbpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 4-(methylsulfinyl)phenol **13a** (46.9 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (8:2 pentane:Et₂O) afforded the title compound **13b** as a white solid (24.9 mg, 59% yield).

tert-Butyl((4-(methylsulfinyl)benzyl)oxy)diphenylsilane (14a)



To a solution of (4-(methylsulfinyl)phenyl)methanol (511 mg, 3.00 mmol) and imidazole (245 mg, 3.60 mmol) in anhydrous CH₂Cl₂ (3 mL) at 0 °C was added *tert*-butyl(chloro)diphenylsilane (0.9 mL, 3.30 mmol) dropwise. The reaction mixture was warmed to RT and stirred overnight. The reaction mixture was then concentrated *in vacuo* and purified by flash column chromatography on silica gel (7:3 EtOAc:hexane, then 8:2 EtOAc:hexane, then 100% EtOAc) to afford the title compound **14a** as a clear and colourless oil (906 mg, 74% yield); R_f 0.30 (8:2 EtOAc:hexane); v_{max} (thin film)/cm⁻¹ 2931, 2857, 1428, 1111, 1085, 1054, 1014, 824, 702; δ_H (400 MHz, CDCl₃) 7.71–7.65 (m, 4H), 7.64–7.60 (m, 2H), 7.53–7.48 (m, 2H), 7.47–7.35 (m, 6H), 4.82 (s, 2H), 2.73 (s, 3H), 1.11 (s,

9H); δ_C (100 MHz, CDCl₃) 144.7, 144.1, 135.7, 133.2, 130.0, 127.9, 126.9, 123.7, 65.1, 44.1, 27.0, 19.4; HRMS (ESI⁺): Found: 409.1650; C₂₄H₂₉O₂SSi⁺ (MH⁺) Requires 409.1652 (0.5 ppm error).

tert-Butyl((4-(methylthio)benzyl)oxy)diphenylsilane (14b)

Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and *tert*-butyl((4-(methylsulfinyl)benzyl)oxy)diphenylsilane **14a** (123 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (9:1 hexane:CH₂Cl₂, then 8:2 hexane:CH₂Cl₂) afforded the title compound **14b** as a clear and colourless oil (106 mg, 90% yield); R_f 0.38 (8:2 hexane:CH₂Cl₂); v_{max} (thin film)/cm⁻¹ 2930, 2856, 1427, 1106, 1082, 823, 798, 740, 699, 611; δ_{H} (400 MHz, CDCl₃) 7.76–7.68 (m, 4H), 7.49–7.38 (m, 6H), 7.32–7.22 (m, 4H), 4.76 (s, 2H), 2.51 (s, 3H), 1.12 (s, 9H); δ_{C} (100 MHz, CDCl₃) 138.3, 136.7, 135.7, 133.6, 129.8, 127.9, 126.9, 126.8, 65.3, 27.0, 19.4, 16.3; HRMS (ESI⁺): Found: 415.1515; C₂₄H₂₈NaOSSi⁺ (MH⁺) Requires 415.1522 (1.7 ppm error).

(4-Chlorobenzyl)(4-chlorophenyl)sulfane (15b)

S.

Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1-chloro-4-((4-chlorobenzyl)sulfinyl)benzene **15a** (85.6 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane) afforded the title compound **15b** as a white solid (76.6 mg, 95% yield); mp 62–64 °C; R_f 0.26 (100% pentane); δ_{H} (400 MHz, CDCl₃) 7.27–7.15 (m, 8H), 4.02 (s, 2H); δ_{C} (100 MHz, CDCl₃) 135.9, 134.1, 133.2, 133.0, 132.0, 130.2, 129.2, 128.8, 38.9. Spectroscopic data is consistent with those reported in the literature.¹²

Methyl 2-(phenylthio)acetate (16b)



Prepared according to general procedure B using $[Ir{dF(CF_3)ppy}_2(dtbpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and methyl 2-(phenylsulfinyl)acetate **16a** (59.5 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (9:1 pentane:Et₂O) afforded the title compound **16b** as a pale

yellow oil (52.3 mg, 96% yield); R₁0.38 (9:1 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.43–7.38 (m, 2H), 7.34–7.28 (m, 2H), 7.26–7.21 (m, 1H), 3.72 (s, 3H), 3.66 (s, 2H); δ_{C} (100 MHz, CDCl₃) 170.3, 135.0, 130.1, 129.2, 127.2, 52.7, 36.6; HRMS (APCl⁺): Found: 183.046798; C₉H₁₁O₂S⁺ (MH⁺) Requires 183.047427 (-3.4 ppm error). Spectroscopic data is consistent with those reported in the literature.¹³

2-(Methylsulfinyl)benzo[d]thiazole (17a)



Prepared according to general procedure A using 2-(methylthio)benzo[*d*]thiazole **17b*** (1.87 g, 10.0 mmol) and *m*CPBA (2.07 g, 12.0 mmol) in anhydrous CH₂Cl₂ (30 mL). Purification by flash chromatography on silica gel (7:3 Et₂O:hexane with 1% AcOH added to eluent) afforded the title compound **17a** as a white solid (1.37 g, 70% yield); mp 59–61 °C; R_f 0.35 (8:2 Et₂O:hexane); v_{max} (thin film)/cm⁻¹ 3062, 3004, 2915, 1474, 1426, 1313, 1235, 1085, 1060, 1001, 953, 758, 729, 678; δ_{H} (400 MHz, CDCl₃) 8.08-8.04 (m, 1H), 8.02-7.98 (m, 1H), 7.56 (ddd, *J* = 8.3, 7.2, 1.3 Hz, 1H), 7.51-7.46 (m, 1H), 3.07 (s, 3H); δ_{C} (100 MHz, CDCl₃) 178.5, 153.9, 136.1, 127.1, 126.4, 124.1, 122.5, 43.3; HRMS (ESI⁺): Found: 219.9862; C₈H₇NNaOS₂⁺ (MNa⁺) Requires 219.9861 (-0.4 ppm error). *This material is commercially available.

2-(Methylthio)benzo[d]thiazole (17b)

To an 8 mL vial equipped with a PTFE septum and a magnetic stirrer bar was added *fac*-lr(ppy)₃ (2.0 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 2-(methylsulfinyl)benzo[*d*]thiazole **17a** (59.2 mg, 0.30 mmol). The reaction vessel was purged by alternating vacuum and argon three times before anhydrous and degassed CH₂Cl₂ (1.5 mL) was added and the septum additionally sealed with paraffin film. The reaction was irradiated with a 60 W blue LED floodlight for 48 h, with rapid stirring and cooling from a small fan to maintain an ambient temperature. The reaction mixture was then directly poured on to silica and purified by column chromatography to afford the title compound **17b** as a white solid (52.5 mg, 97% yield); mp 43–45 °C; R₁0.41 (97:3 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.87 (d, *J* = 8.11 Hz, 1H), 7.76 (d, *J* = 7.5 Hz, 1H), 7.45-7.39 (m, 1H), 7.32-7.26 (m, 1H), 2.80 (s, 3H); δ_{C} (100 MHz, CDCl₃) 168.2, 153.5, 135.3, 126.2, 124.2, 121.5, 121.1, 16.1; HRMS (ESI⁺): Found: 182.0091; C₈H₈NS₂⁺ (MH⁺) Requires 182.0093 (0.9 ppm error). Spectroscopic data is consistent with those reported in the literature.¹⁴

Dimethylsulfane (18b)

Prepared according to general procedure B using $[Ir{dF(CF_3)ppy}_2(dtbpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and (methylsulfinyl)methane **18a** (23.4 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). The crude reaction mixture was analysed by ¹H NMR spectroscopy using a trimethoxybenzene internal standard and a 98% yield of title compound **18b** was calculated. Due to the extremely volatile nature of this compound, purification was not performed.

∕^s∖

Tetrahydrothiophene (19b)



Prepared according to general procedure B using $[Ir{dF(CF_3)ppy}_2(dtbpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and tetrahydrothiophene 1-oxide **19a** (31.3 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). The crude reaction mixture was analysed by ¹H NMR spectroscopy using a trimethoxybenzene internal standard and a 99% yield of title compound **19b** was calculated. Due to the extremely volatile nature of this compound, purification was not performed.

DibutyIsulfane (20b)



Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1-(butylsulfinyl)butane **20a** (48.7 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane) afforded the title compound **20b** as a pale yellow oil (33.7 mg, 78% yield); R_f 0.19 (100% pentane); δ_{H} (400 MHz, CDCl₃) 2.50 (t, *J* = 7.4 Hz, 4H), 1.61–1.49 (m, 4H), 1.45–1.34 (m, 4H), 0.91 (t, *J* = 7.3 Hz, 6H); δ_{C} (100 MHz, CDCl₃) 32.0 (2 x C), 22.2, 13.9. Spectroscopic data is consistent with those reported in the literature.¹⁵

Dioctylsulfane (21b)



Prepared according to general procedure B using $[Ir{dF(CF_3)ppy}_2(dtbpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1-(octylsulfinyl)octane **21a** (82.4 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane, then 99:1 pentane:Et₂O) afforded the title compound **21b** as a clear and colourless oil (73.7 mg, 95% yield); R_f 0.40 (100% pentane); δ_H (400 MHz, CDCl₃) 2.50 (t, *J* = 7.4 Hz,

4H), 1.62–1.52 (m, 4H), 1.40–1.23 (m, 20H), 0.92–0.84 (m, 6H); δ_{C} (100 MHz, CDCl₃) 32.3, 32.0, 29.9, 29.37, 29.35, 29.1, 22.8, 14.2; HRMS (APCI⁺): Found: 259.244383; $C_{16}H_{35}S^+$ (MH⁺) Requires 259.245399 (-3.9 ppm error). Spectroscopic data is consistent with those reported in the literature.¹⁶

Dibenzylsulfane (22b)

Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and (sulfinylbis(methylene))dibenzene **22a** (69.1 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane, then 98:2 pentane:Et₂O) afforded the title compound **22b** as a clear and colourless oil (62.5 mg, 97% yield); R_f 0.30 (98:2 pentane:Et₂O); δ_{H} (400 MHz, CDCl₃) 7.40–7.20 (m, 10H), 3.61 (s, 4H); δ_{C} (100 MHz, CDCl₃) 138.3, 129.1, 128.6, 127.1, 35.7. Spectroscopic data is consistent with those reported in the literature.¹⁵

Bis(methylthio)methane (23b)

∕^s∕_s∖

Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and methyl((methylsulfinyl)methyl)sulfane **23a** (37.3 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (100% pentane, then 99:1 pentane:Et₂O) afforded the title compound **23b** as a clear and colourless oil (22.7 mg, 70% yield); R_f 0.25 (95:5 hexane:Et₂O); δ_{H} (400 MHz, CDCl₃) 3.62 (s, 2H), 2.15 (s, 6H); δ_{C} (100 MHz, CDCl₃) 40.2, 14.5. Spectroscopic data is consistent with those reported in the literature.¹⁷

1,4-Dioxa-8-thiaspiro[4.5]decane (24b)



Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1,4-dioxa-8-thiaspiro[4.5]decane 8-oxide **24a** (52.9 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (9:1 pentane:Et₂O) afforded the title compound **24b** as a pale yellow oil (43.4 mg, 90% yield); v_{max} (thin film)/cm⁻¹ 2948, 2916, 2880, 1427, 1269, 1248, 1102, 1055, 1026, 885; R_f 0.21 (9:1 hexane:Et₂O); δ_{H} (400 MHz, CDCl₃) 3.94 (s, 4H), 2.77–2.67 (m, 4H), 1.92–1.85 (m, 4H); δ_{C} (100

MHz, CDCl₃) 107.3, 64.5, 36.9, 27.1; HRMS (APCl⁺): Found: 161.063557; $C_7H_{13}O_2S^+$ (MH⁺) Requires 161.063077 (-3.0 ppm error). Spectroscopic data is consistent with those reported in the literature.¹⁸

tert-Butyl (tert-butoxycarbonyl)-L-methioninate (25b)

BocHN

Prepared according to general procedure B using [Ir{dF(CF₃)ppy}₂(dtbpy)]PF₆ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and 1,4-dioxa-8-thiaspiro[4.5]decane 8-oxide **25a** (96.4 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (8:2 pentane:Et₂O) afforded the title compound **25b** as a pale yellow oil as a 7:1 mixture of rotamers (91.0 mg, 99% yield); R_f 0.34 (8:2 hexane:Et₂O); δ_{H} (400 MHz, CDCl₃) 5.10 (br s, 1H, major rotamer), 4.84 (br s, 1H, minor rotamer), 4.25 (br s, 1H, major), 4.12 (br s, 1H, minor), 2.59–2.43 (m, both, 4H), 2.16–2.01 (m, both, 8H), 1.95–1.80 (m, 2H, both), 1.45 (s, 18H, both), 1.42 (s, 18H, both); δ_{C} (100 MHz, CDCl₃) 171.5, 155.4, 82.2, 79.9, 53.5, 32.7, 30.0, 28.4, 28.1, 15.6; HRMS (ESI⁺): Found: 328.1552; C₁₄H₂₇NNaO₄S⁺ (MNa⁺) Requires 328.1553 (0.2 ppm error). Spectroscopic data is consistent with those reported in the literature.²

O,O-Diethyl O-(4-(methylthio)phenyl) phosphorothioate (26b)



To an 8 mL vial equipped with a PTFE septum and a magnetic stirrer bar was added *fac*-Ir(ppy)₃ (1.96 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and *O*,*O*-diethyl *O*-(4-(methylsulfinyl)phenyl) phosphorothioate **33a** (92.5 mg, 0.30 mmol). The reaction vessel was purged by alternating vacuum and argon three times before anhydrous and degassed CH₂Cl₂ (1.5 mL) was added and the septum additionally sealed with paraffin film. The reaction was irradiated with a 60 W blue LED floodlight for 48 h, with rapid stirring and cooling from a small fan to maintain an ambient temperature. The reaction mixture was then directly poured on to silica and purified by column chromatography to afford the title compound **33b** as a pale yellow oil (84.4 mg, 96% yield); R_f 0.35 (6:4 CHCl₃:hexane); δ_{H} (400 MHz, CDCl₃) 7.23 (d, *J* = 8.7 Hz, 2H), 7.11 (d, *J* = 8.7 Hz, 2H), 4.28–4.16 (m, 4H), 2.46 (s, 3H), 1.36 (t, *J* = 7.1 Hz, 6H); δ_{C} (100 MHz, CDCl₃) 148.6 (d, ²*J*_{C-P} = 7.7 Hz), 135.0, 128.2, 121.6 (d, ³*J*_{C-P} = 4.8 Hz), 65.2 (d, ²*J*_{C-P} = 5.6 Hz), 16.6, 16.0 (d, ³*J*_{C-P} = 7.5 Hz); HRMS (ESI⁺): Found: 293.0429; C₁₁H₁₈O₃PS_{2⁺} (MH⁺) Requires 293.0429 (1.7 ppm error). Spectroscopic data is consistent with those reported in the literature.¹⁹

Compound **33b** was also prepared according to general procedure B using $[Ir{dF(CF_3)ppy}_2(dtbpy)]PF_6$ (3.4 mg, 0.003 mmol), PPh₃ (94.6 mg, 0.36 mmol) and *O*, *O*-diethyl *O*-(4-(methylsulfinyl)phenyl) phosphorothioate **33a** (92.5 mg, 0.30 mmol) in anhydrous CH₂Cl₂ (1.5 mL). Purification by flash chromatography on silica gel (6:4 CHCl₃:hexane) afforded the title compound **33b** as a pale yellow oil (49.4 mg, 56% yield).

2-(2-Methoxy-4-(methylthio)phenyl)-1 H-imidazo[4,5-b]pyridine (34b)



To an 8 mL vial equipped with a PTFE septum and a magnetic stirrer bar was added *fac*-Ir(ppy)₃ (0.85 mg, 1.3 µmol), PPh₃ (42.0 mg, 0.16 mmol) and 2-(2-Methoxy-4-(methylsulfinyl)phenyl)-1*H*-imidazo[4,5-*b*]pyridine **34a** (37.5 mg, 0.13 mmol). The reaction vessel was purged by alternating vacuum and argon three times before anhydrous and degassed CH₂Cl₂ (0.7 mL) was added and the septum additionally sealed with paraffin film. The reaction was irradiated with a 60 W blue LED floodlight for 24 h, with rapid stirring and cooling from a small fan to maintain an ambient temperature. The reaction mixture was then directly poured on to silica and purified by column chromatography to afford the title compound **34b** as a pale yellow oil (22.0 mg, 62% yield); Rr 0.36 (100% EtOAc); v_{max} (thin film)/cm⁻¹ 3238, 3060, 2926, 1599, 1560, 1529, 1462, 1405, 1278, 1241, 1115, 1072, 1027, 879, 779; δ_{H} (400 MHz, CDCl₃) 8.49 (d, *J* = 8.3 Hz, 1H), 8.42-8.29 (m, 2H), 8.01 (d, *J* = 7.9 Hz, 1H), 7.25–7.19 (m, 1H), 6.99 (dd, *J* = 8.4, 1.3 Hz, 1H), 6.92 (s, 1H), 4.06 (s, 3H), 2.55 (s, 3H); δ_{C} (100 MHz, CD₃OD) 159.2, 153.7, 153.4, 147.1, 144.3, 131.6, 130.8, 123.6, 119.3, 119.0, 114.2, 109.6, 56.5, 14.8; HRMS (ESI⁺): Found: 272.0853; C₁₄H₁₄N₃OS⁺ (MH⁺) Requires 272.0852 (-0.3 ppm error).

Photocatalyst Comparative Studies^a



Entry	Sulfoxide	PC2 yield / %	PC8 yield / %
1	8a	99	99
2	11a	99	43
3	12a	96	60
4	14a	93	28
5	17a	34	>99 ^b
6	22a	97	60
7	2 6a	56	>99 ^c

^{*a*1}H NMR yields reported based on a trimethoxybenzene internal standard. ^{*b*}Reaction performed for 4 days. ^{*c*}Reaction performed for 48 h.

Cyclic Voltammetry Information

The cyclic voltammograms were performed using a 5 mL electrochemical cell vial containing a glassy carbon disk working electrode, platinum counter electrode and Ag/AgCl reference electrode, all from the IKA ElectraSyn range. The cell lid was modified in-house to permit connection to an EmStat potentiostat and the data was collected using the complementary PSTrace software.

The same procedure was followed for experiments conducted on both PPh₃ and sulfoxide, **8a**. First, 77 mg of tetrabutylammonium hexafluorophosphate (0.2 mmol Bu₄NPF₆, Acros Organics, 98%) was added to the cell vial, the lid was attached and then the vial was purged with N₂ for approximately 5 min via the access port. After this, 2 mL anhydrous CH_2Cl_2 was added then three control "background" cyclic voltammograms were recorded over a range of -2 V to +2 V at a scan rate of 100 mVs⁻¹ and under an atmosphere of N₂ (achieved by attaching a N₂-filled balloon to the cell lid). A 0.2 mL aliquot of 0.1 M analyte in CH_2Cl_2 was then added to the cell (final concentration 4.8 mM) and three cyclic voltammograms were recorded under the same conditions. Voltammetric data from the second potential sweep is shown (Scheme 3, C in manuscript). In between experiments the electrode surface was cleaned by abrasion using alumina.

For experiments on the mixture of sulfoxide + PPh_3 , first 0.2 mL of 0.1 M **8a** was added to the cell solution, followed by 0.3 mL of PPh_3 solution.

UV-Vis Spectroscopy Studies

UV-vis spectra were recorded using Shimadzu UV-Vis Spectrophotometer UV-2600 system. A quartz cuvette with 10mm path length (Hellma Macro, AS4C-QS/QG,) was used. Substrates were dissolved in anhydrous CH₂Cl₂ (0.005 M) unless otherwise stated.

Samples were prepared for analysis as described below:

Sulfoxide **8a** (3.3 mg, 0.015 mmol) was dissolved in CH_2Cl_2 (3 mL) and were analysed between 200–400 nm (blue line).

 PPh_3 (3.9 mg, 0.015 mmol) was dissolved in CH_2Cl_2 (3 mL) and were analysed between 200–400 nm (grey line). Equimolar quantities of sulfoxide **8a** (3.3 mg, 0.015 mmol) and PPh_3 (3.9 mg, 0.015 mmol) were dissolved in CH_2Cl_2 (3 mL) and were analysed between 200–400 nm (orange line).



Figure S1. UV-vis spectroscopy studies of sulfoxide 8a, PPh₃ and an equimolar solution of both.

¹⁹F NMR Spectra

Examination of the reaction mixture for the conversion of **8a** into **8b** using **PC2** revealed that **PC2** remained unchanged. All ¹⁹F NMR signals corresponding to pure **PC2** (*bottom, red*) are observed in the ¹⁹F NMR spectrum of the reaction mixture for the conversion of **8a** into **8b** (*top, blue*), suggesting that the catalyst does not degrade significantly or form aggregates during the course of the reaction.



Figure S2. Overlaid ¹⁹F NMR spectra of the reaction mixture for the conversion of 8a into 8b using PC2 (top, blue) and pure PC2 (bottom, red).





















































References

1. Rehaud, P. Reactions of sulfinylated radicals. Stereoselectivity in six-membered rings. *Helv. Chim. Acta* **1991**, 74, 1305–1313

2. Masuda, Y.; Maruyama, C.; Kawabata, K.; Hamano, Y.; Doi, T. Synthesis of (2S,3R,4R)-3,4-dihydroxyarginine and its inhibitory activity against nitric oxide synthase. *Tetrahedron* **2016**, *7*2, 5602–5611.

3. Motsch, S.; Schütz, C.; Huy, P. H. Systematic Evaluation of Sulfoxides as Catalysts in Nucleophilic Substitutions of Alcohols. *Eur. J. Org. Chem.* **2018**, 4541–4547.

4. Zenzola, M.; Doran, R.; Degennaro, L.; Luisi, R.; Bull, J. A. Transfer of Electrophilic NH Using Convenient Sources of Ammonia: Direct Synthesis of NH Sulfoximines from Sulfoxides. *Angew. Chem. Int. Ed.* **2016**, *55*, 7203–7207.

5. Satoh, T.; Miura, M.; Sakai, K.; Yokoyama, Y. Reaction of magnesium cyclopropylidene with N-lithio arylamines: a method for generation of α -amino-substituted cyclopropylmagnesiums and a study for their reactivity with electrophiles. *Tetrahedron* **2006**, *62*, 4253–4261.

 Amal Joseph, P. J.; Priyadarshini, S.; Lakshmi Kantam, M.; Sreedhar, B. Investigation of the scope and mechanism of copper catalyzed regioselective methylthiolation of aryl halides. *Tetrahedron* 2013, *69*, 8276–8283.
 Cheng, J.-H.; Ramesh, C.; Kao, H.-L.; Wang, Y.-J.; Chan, C.-C.; Lee, C.-F. Synthesis of aryl thioethers through the *N*-chlorosuccinimide-promoted cross-coupling reaction of thiols with grignard reagents. *J. Org. Chem.* 2012, *77*, 10369–10374.

8 Wang, M.; Wei, J.; Fan, Q.; Jiang, X. Cu(II)-catalyzed sulfide construction: both aryl groups utilization of intermolecular and intramolecular diaryliodonium salt. *Chem. Commun.* **2017**, *53*, 2918–2921.]

9. [Darvesh, S.; Darvesh, K. v.; McDonald, R. S.; Mataija, D.; Walsh, R.; Mothana, S.; Lockridge, O.; Martin, M. Carbamates with differential mechanism of inhibition toward acetylcholinesterase and butyrylcholinesterase. *Eur. J. Med. Chem.* **2008**, *51*, 4200–4212.

10. Zhang, H.; Wang, G. PyHBr₃/TBN/H₂O as catalytic system for the oxidation of sulfides to sulfoxides with air as the oxidant. *Tetrahedron Lett.* **2008**, *51*, 4200–4212.

11. Zhu, C.; Wang, R.; Falck, J. R. Mild and rapid hydroxylation of aryl/heteroaryl boronic acids and boronate esters with *N*-oxides. *Org. Lett.* **2012**, *14*, 3494–3497.

12. Ding, Q.; Cao, B.; Yuan, J.; Liu, X.; Peng, Y. Synthesis of thioethers via metal-free reductive coupling of tosylhydrazones with thiols. *Org. Biomol. Chem.* **2011**, *9*, 748–751.

Sharma, P.; Singh, R. R.; Giri, S. S.; Chen, L.-Y.; Cheng, M.-J.; Liu, R.-S. Gold-catalyzed oxidation of thioalkynes to form phenylthio ketene derivatives via a noncarbene route. *Org. Lett.* **2019**, *21*, 5475–5479.
 Kamps, J. J. A. G.; Belle, R.; Mecinović, J. Hydroxylamine as an oxygen nucleophile: substitution of sulfonamide by a hydroxyl group in benzothiazole-2-sulfonamides. *Org. Biomol. Chem.* **2013**, *11*, 1103-1108.
 Zhao, X.; Zheng, X.; Yang, B.; Sheng, J.; Lu, K. Deoxygenation of sulphoxides to sulphides with trichlorophosphane. *Org. Biomol.* Chem. **2018**, *16*, 1200–1204.

16. Ma, X.; Yu, L.; Yang, Y.; Li, H.; Xu, Q. Efficient generation of C–S bonds via a by-product-promoted selective coupling of alcohols, organic halides, and thiourea. *Adv. Synth. Catal.* **2018**, *359*, 1649–1655.

17. Levanova, E. P.; Vshivtsev, V. Y.; Sukhomazova, E. N.; Grabel'nykh, V. A.; Russavskaya, N. V.; Zhanchipova, E. R.; Klyba, L. V.; Albanov, A. I.; Korchevin, N. A. Reactions of dichloromethane with chalcogens and dimethyl chalcogenides in the hydrazine hydrate-alkali system. *Russian J. Gen. Chem.* **2008**, *78*, 1734–1741.

18. Davies, J.; Jones, J. B. Enzymes in organic synthesis. 16. Heterocyclic ketones as substrates of horse liver alcohol dehydrogenase. Stereospecific reductions of 2-substituted tetrahydrothiopyran-4-ones. *J. Am. Chem. Soc.* **1979**, *101*, 5405–5410.

19. Zhang, B.; Fan, Z.; Guo, Z.; Xi, C. Reduction of CO₂ with NaBH₄/I₂ for the Conversion of Thiophenols to Aryl Methyl Sulfides. *J. Org. Chem.* **2019**, *84*, 8661–8667.