

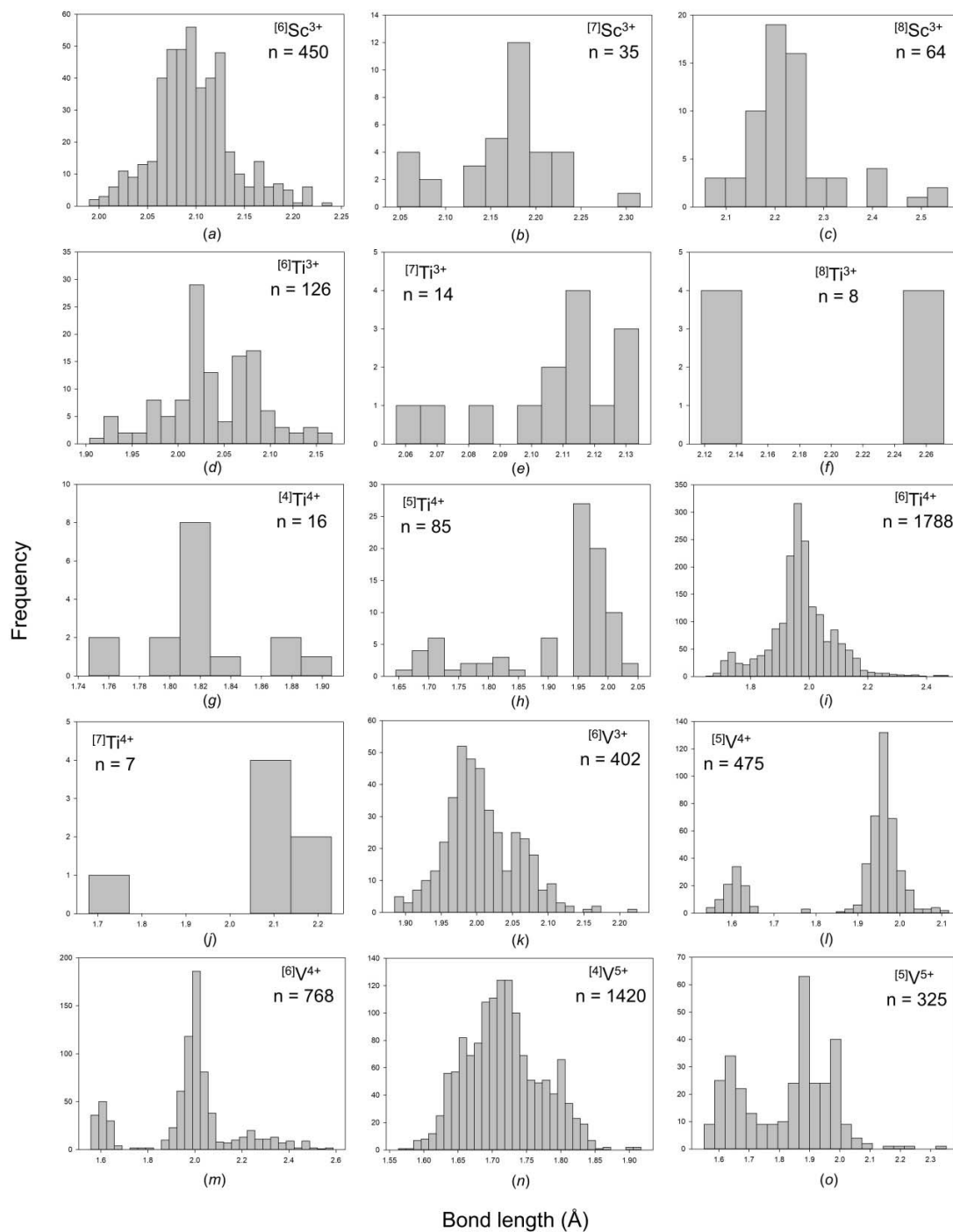
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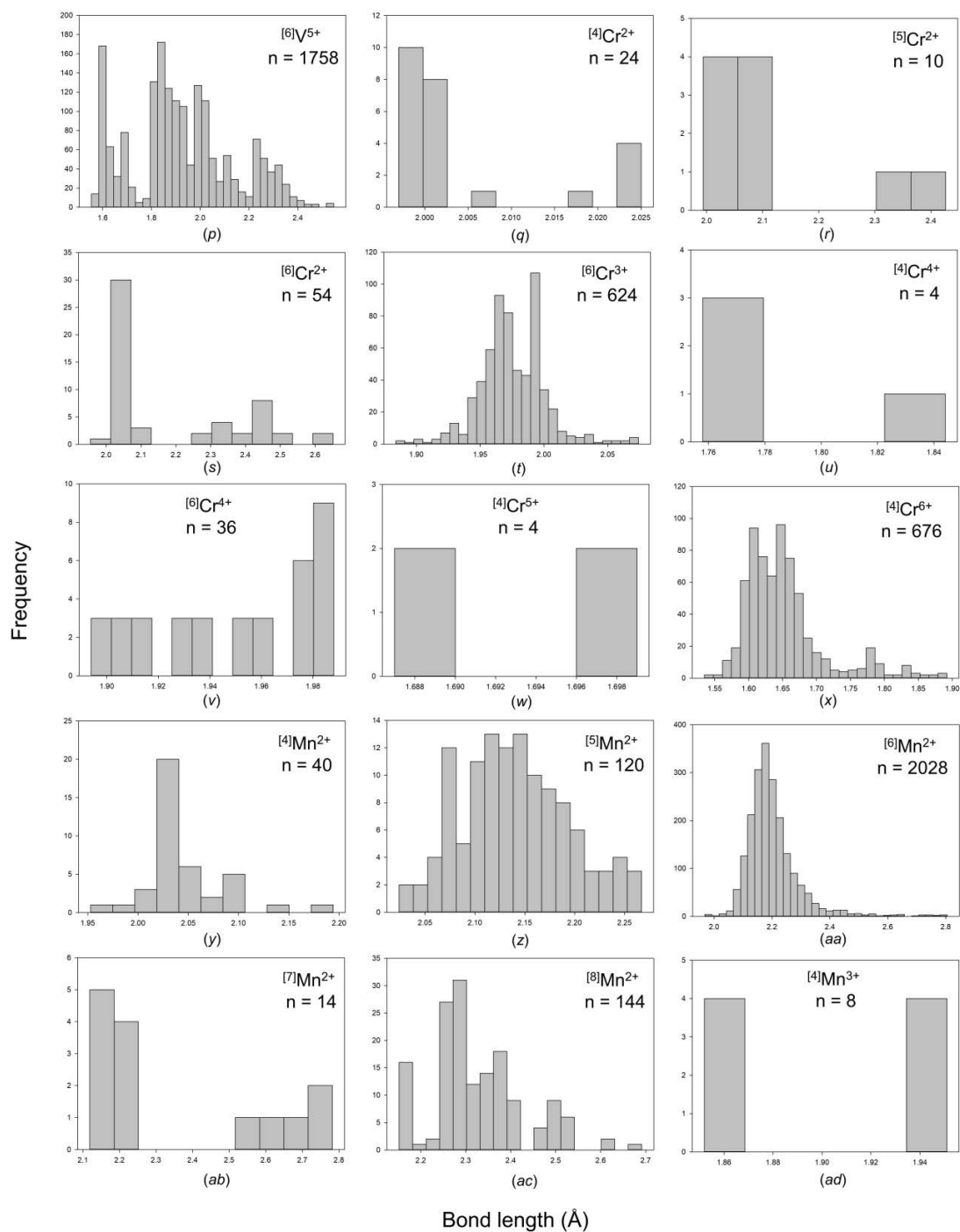
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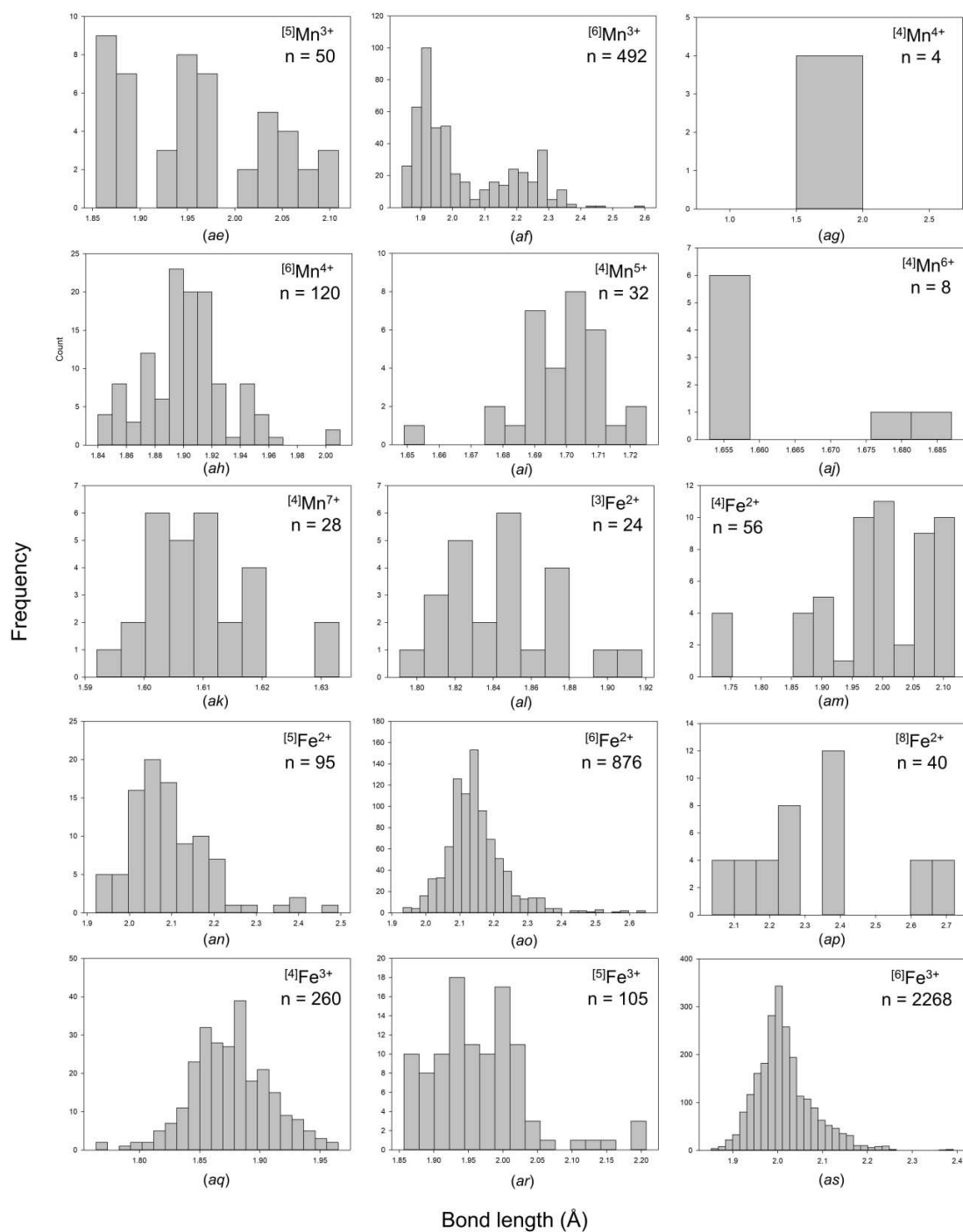
Supporting information for article:

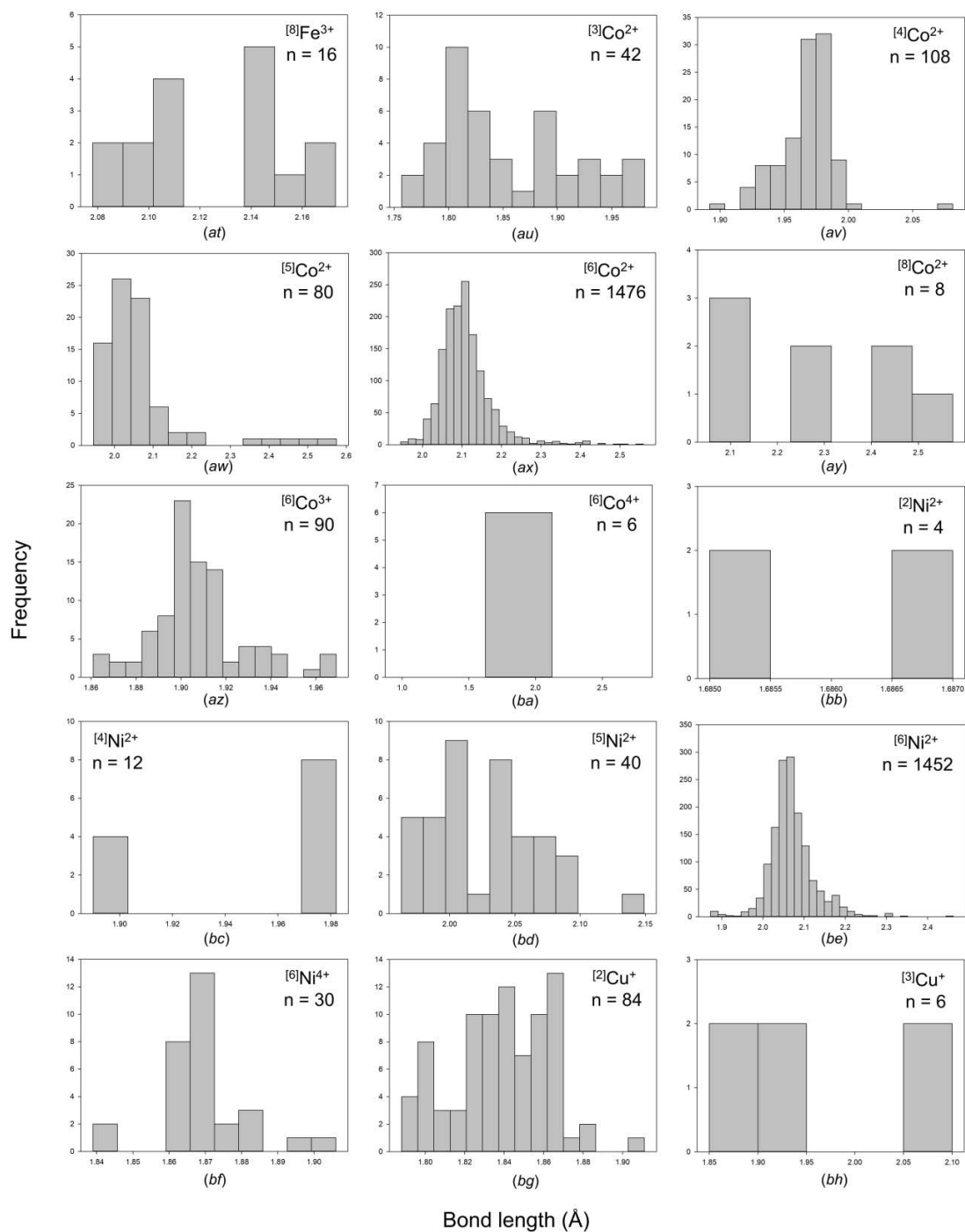
Bond-length distributions for ions bonded to oxygen: results for the transition metals and quantification of the factors underlying bond-length variation in inorganic solids

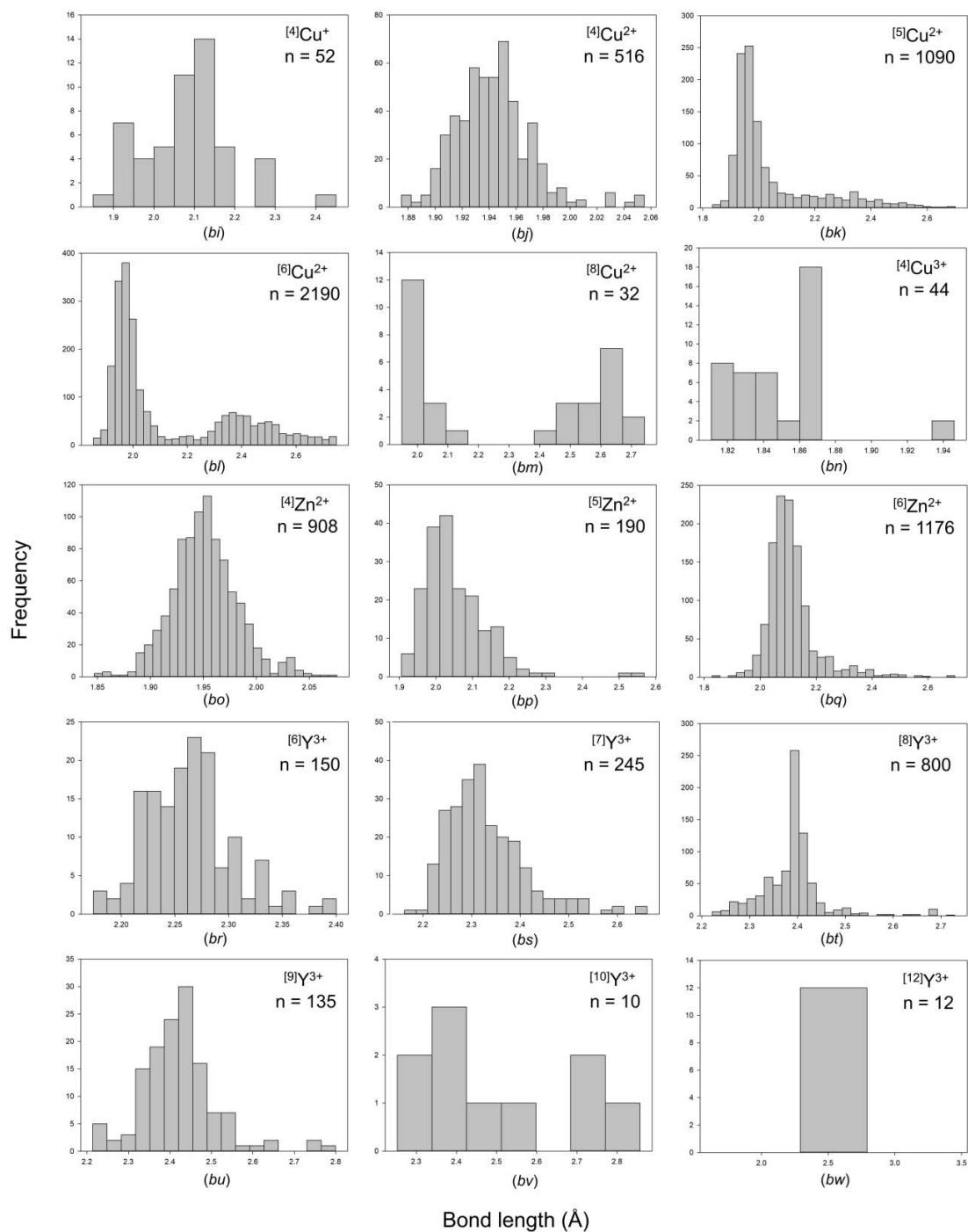
Olivier Charles Gagné and Frank Christopher Hawthorne

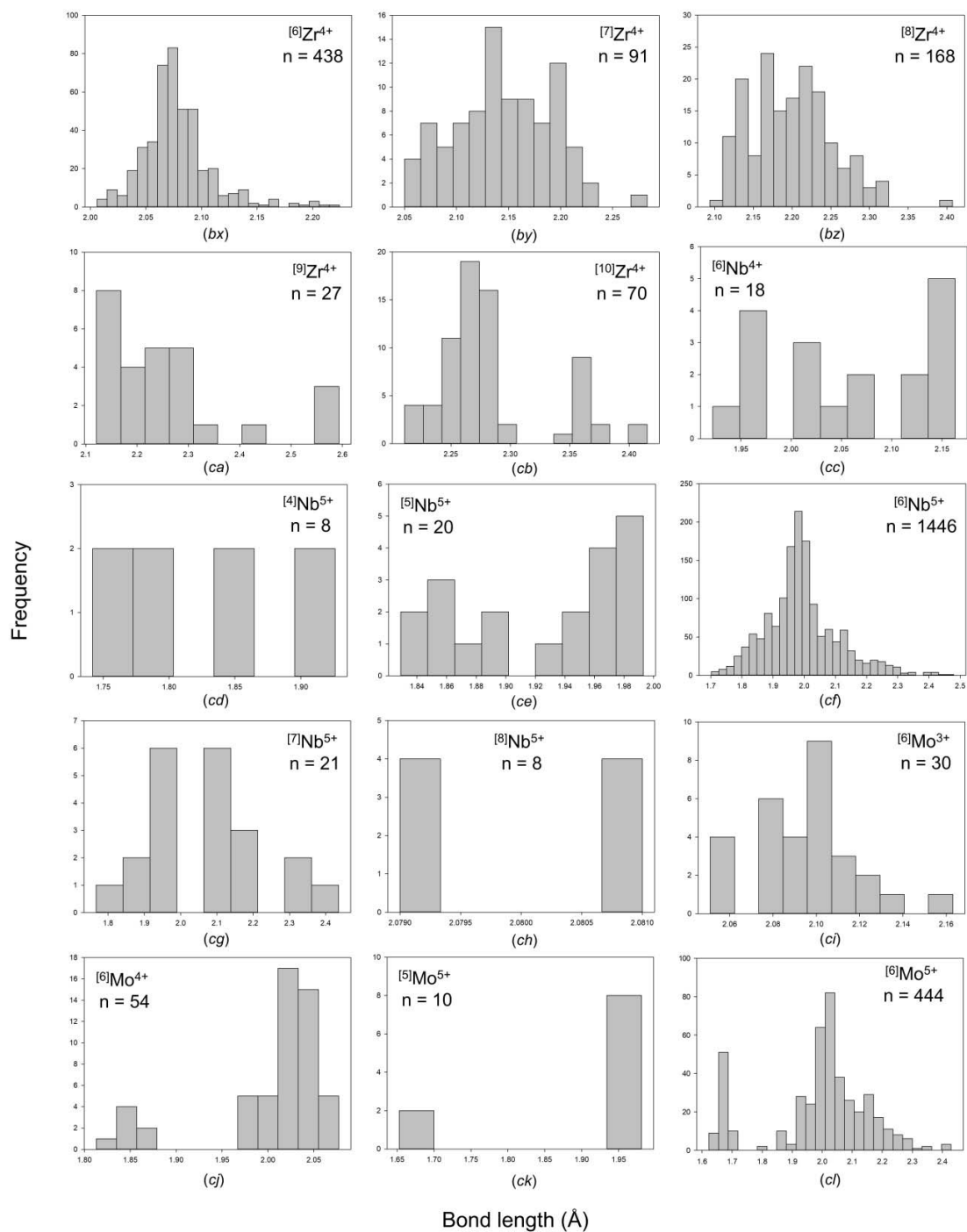


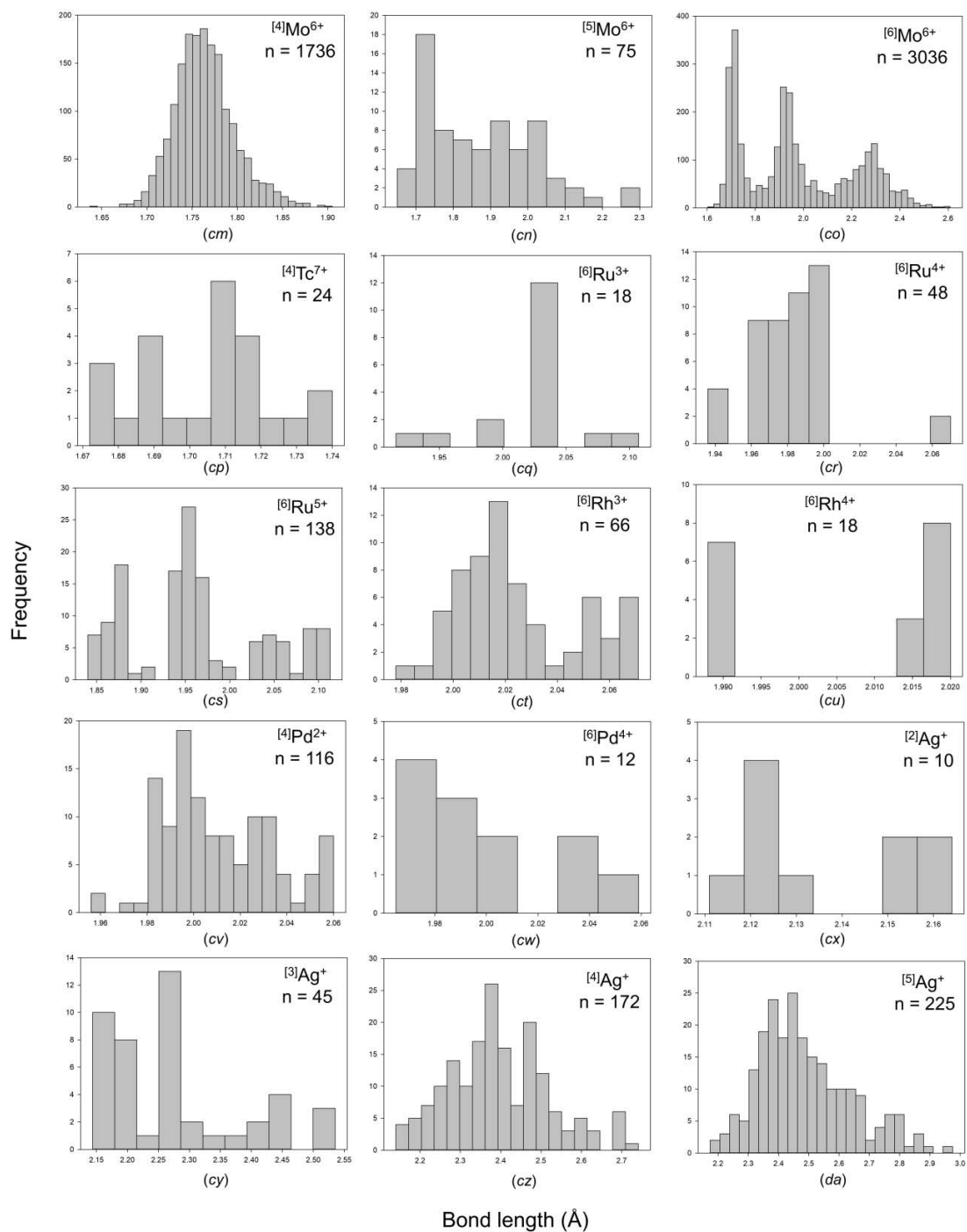


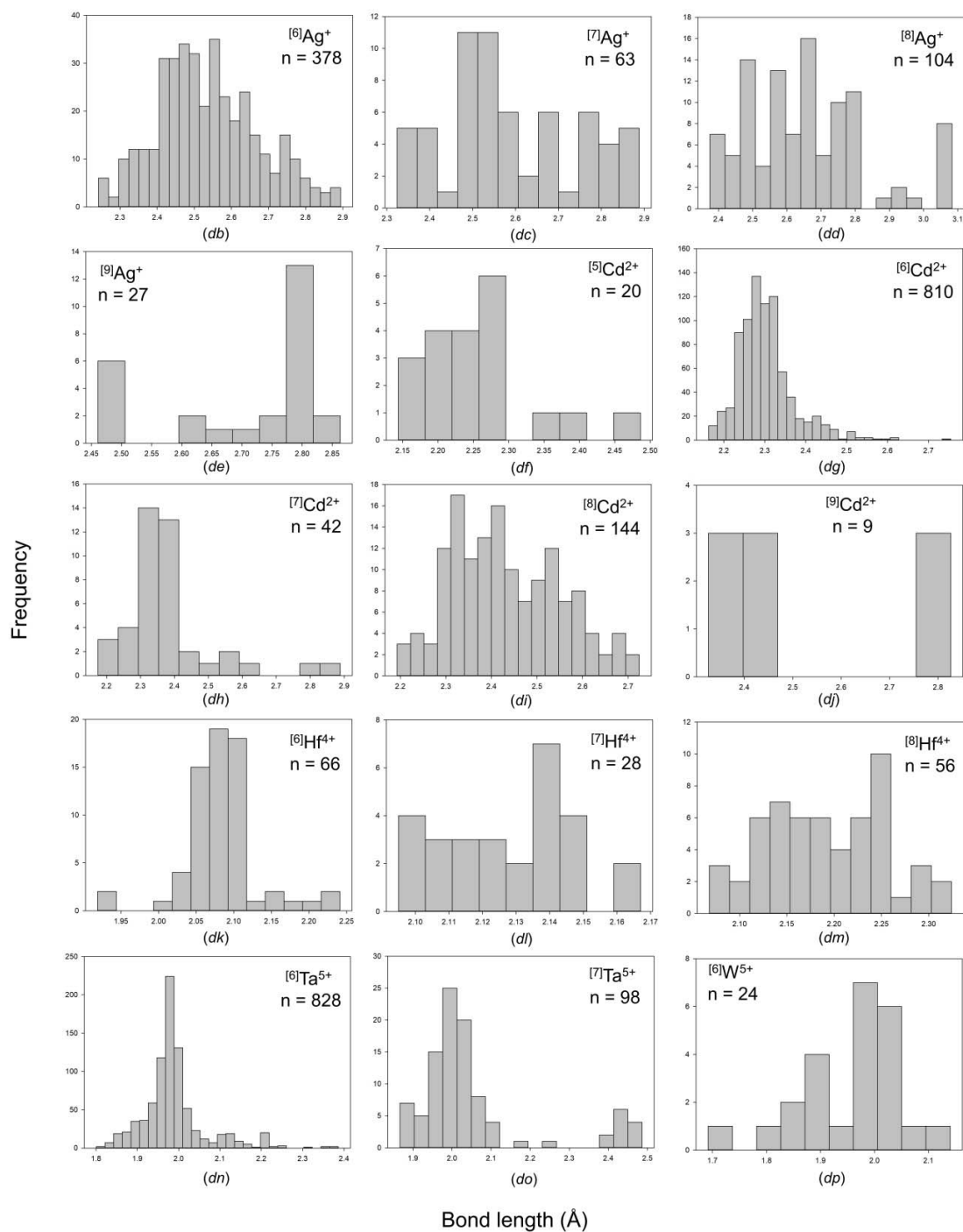


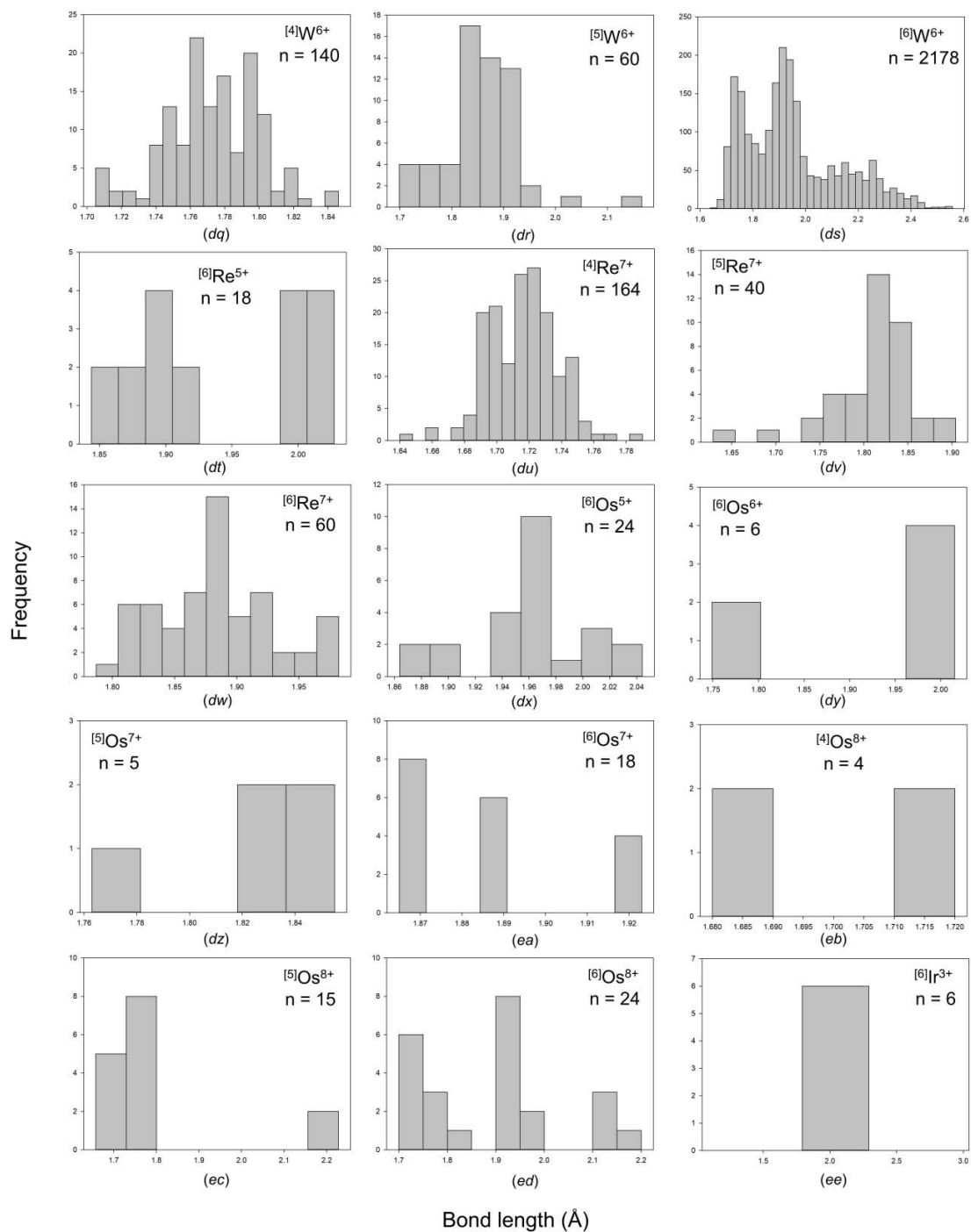












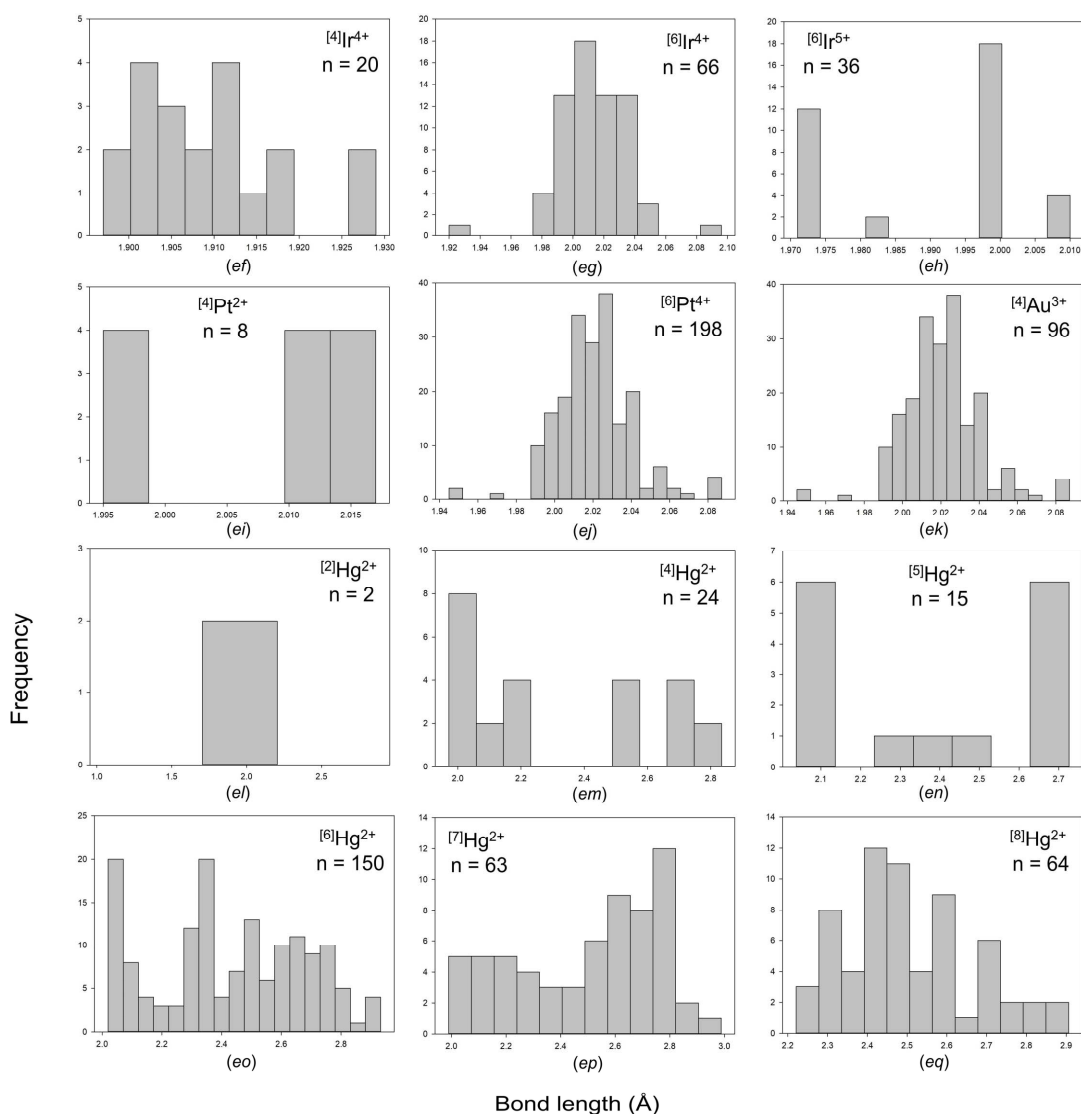
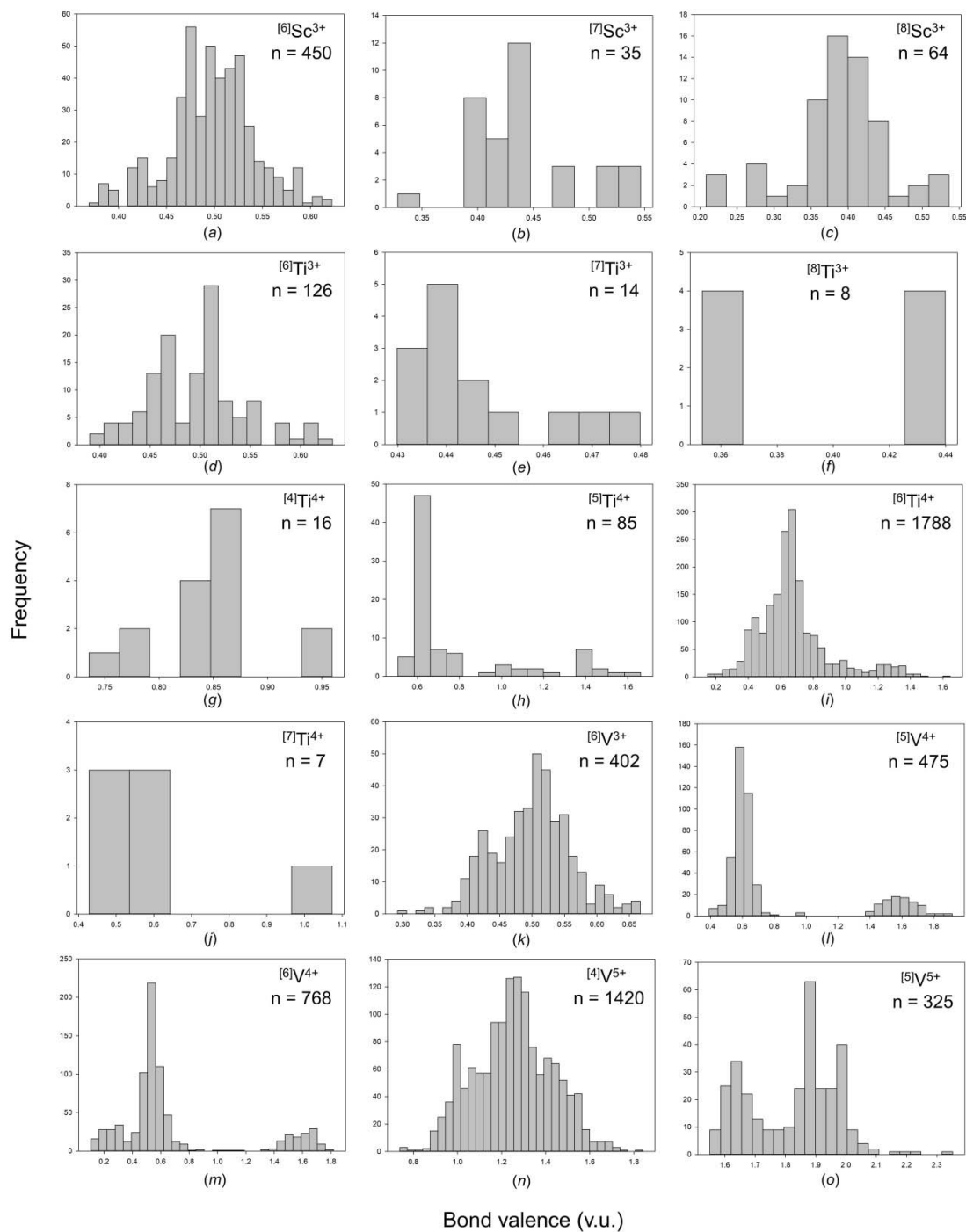
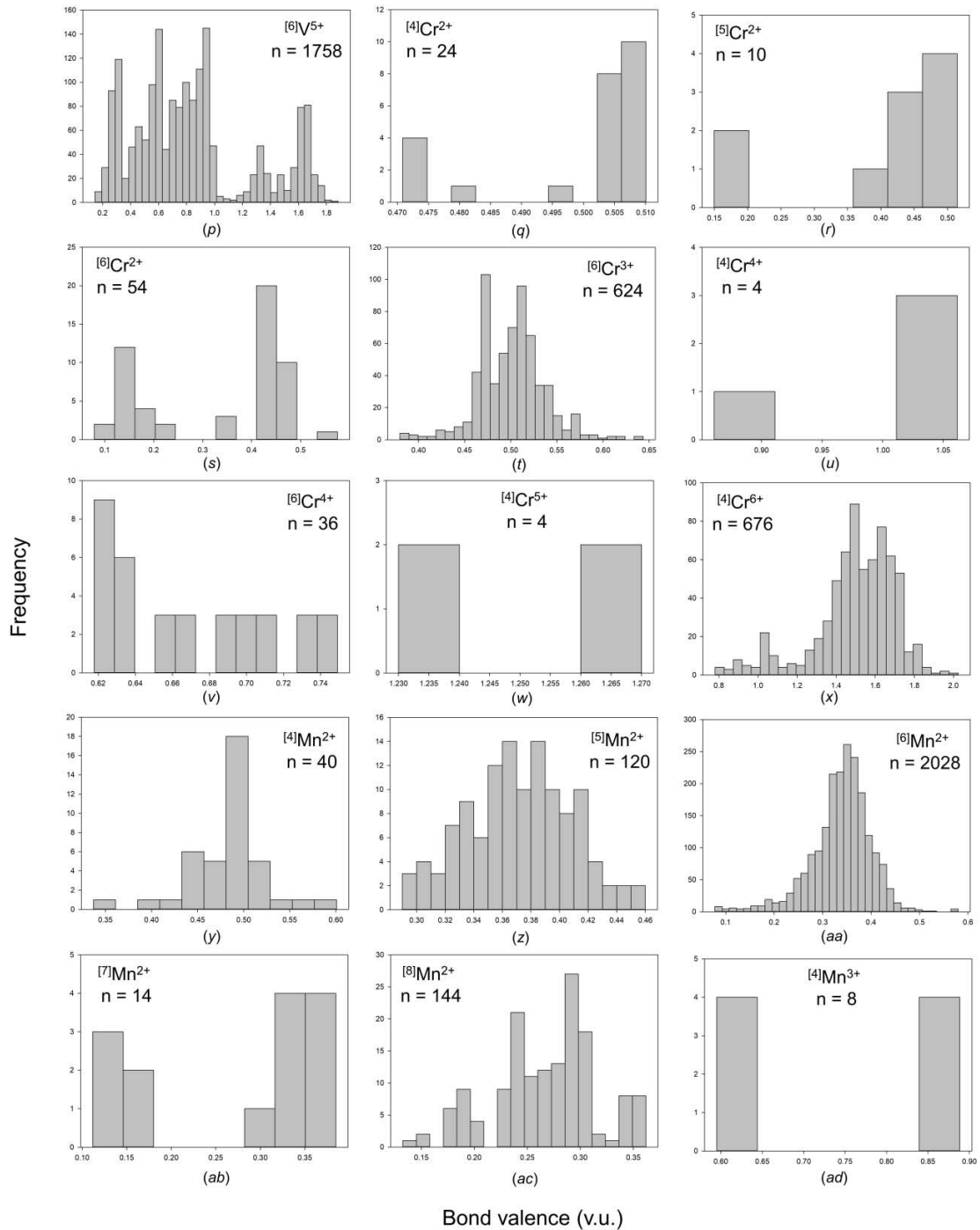
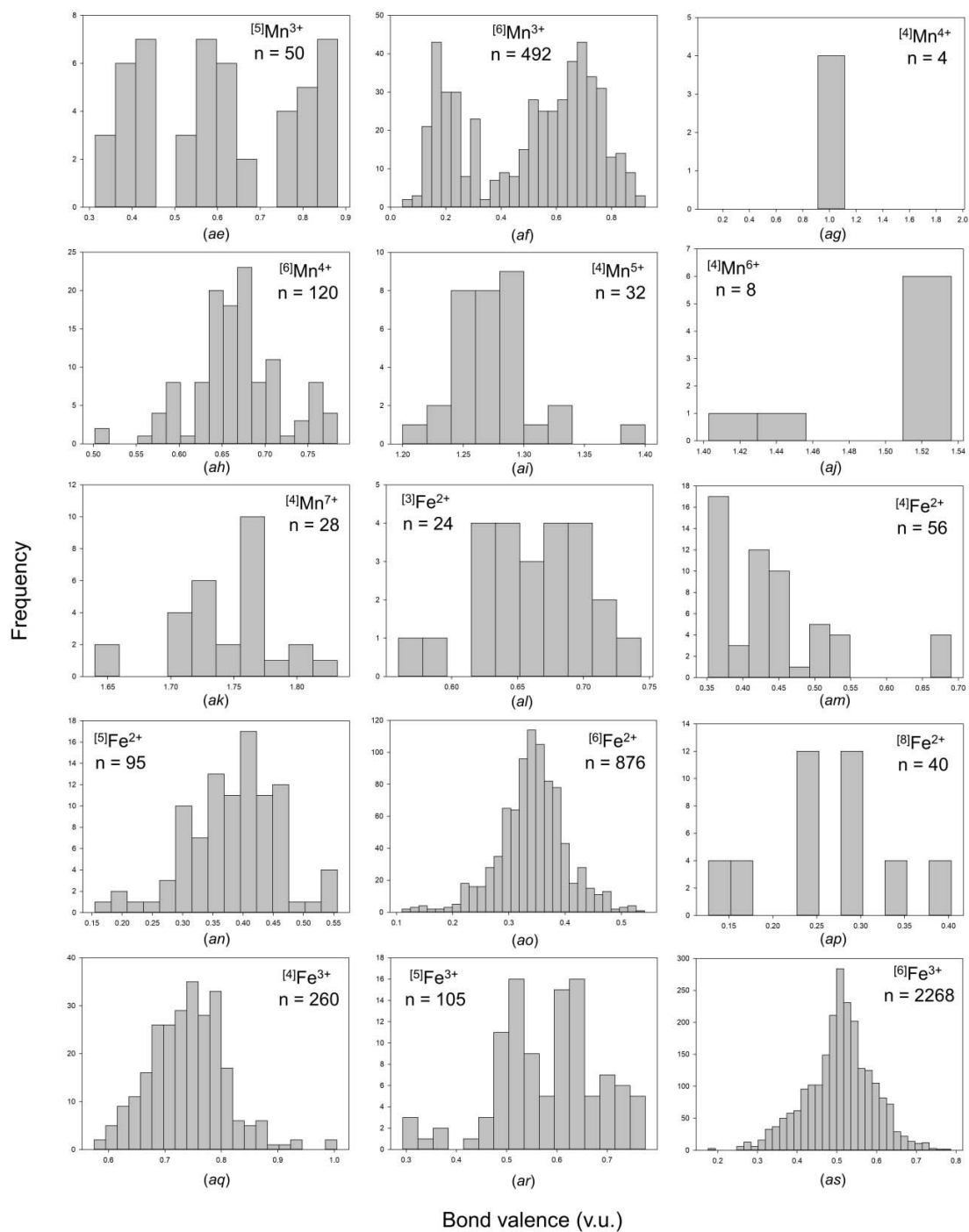


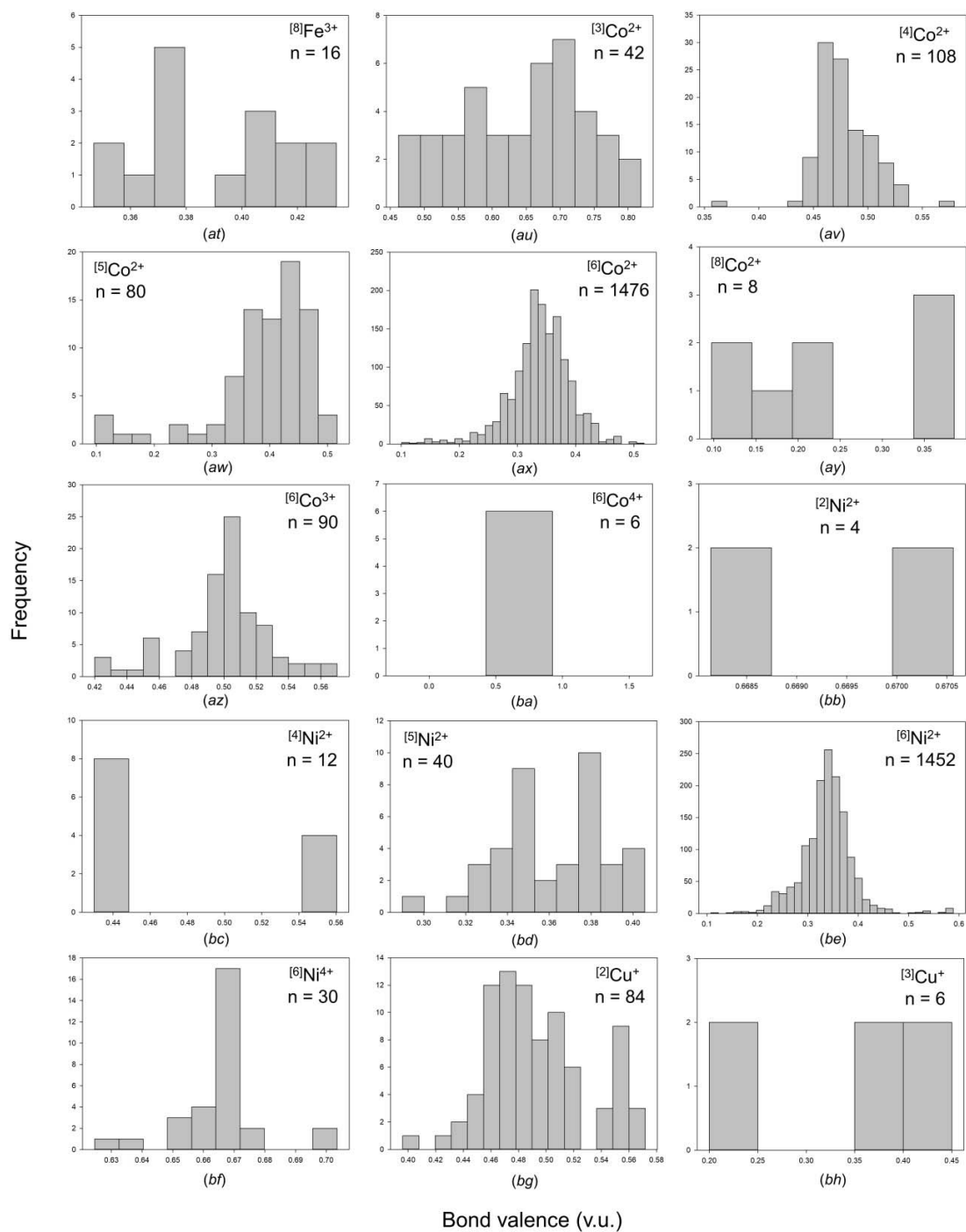
Figure S1 Bond-length distributions for all configurations of the transition metal ions bonded to O^{2-} : (a) $[6]Sc^{3+}$, (b) $[7]Sc^{3+}$, (c) $[8]Sc^{3+}$, (d) $[6]Ti^{3+}$, (e) $[7]Ti^{3+}$, (f) $[8]Ti^{3+}$, (g) $[4]Ti^{4+}$, (h) $[5]Ti^{4+}$, (i) $[6]Ti^{4+}$, (j) $[7]Ti^{4+}$, (k) $[6]V^{3+}$, (l) $[5]V^{4+}$, (m) $[6]V^{4+}$, (n) $[4]V^{5+}$, (o) $[5]V^{5+}$, (p) $[6]V^{5+}$, (q) $[4]Cr^{2+}$, (r) $[5]Cr^{2+}$, (s) $[6]Cr^{2+}$, (t) $[6]Cr^{3+}$, (u) $[4]Cr^{4+}$, (v) $[6]Cr^{4+}$, (w) $[4]Cr^{5+}$, (x) $[4]Cr^{6+}$, (y) $[4]Mn^{2+}$, (z) $[5]Mn^{2+}$, (aa) $[6]Mn^{2+}$, (ab) $[7]Mn^{2+}$, (ac) $[8]Mn^{2+}$, (ad) $[4]Mn^{3+}$, (ae) $[5]Mn^{3+}$, (af) $[6]Mn^{3+}$, (ag) $[4]Mn^{4+}$, (ah) $[6]Mn^{4+}$, (ai) $[4]Mn^{5+}$, (aj) $[4]Mn^{6+}$, (ak) $[4]Mn^{7+}$, (al) $[3]Fe^{2+}$, (am) $[4]Fe^{2+}$, (an) $[5]Fe^{2+}$, (ao) $[6]Fe^{2+}$, (ap) $[8]Fe^{2+}$, (aq) $[4]Fe^{3+}$, (ar) $[5]Fe^{3+}$, (as) $[6]Fe^{3+}$, (at) $[8]Fe^{3+}$, (au) $[3]Co^{2+}$, (av) $[4]Co^{2+}$, (aw) $[5]Co^{2+}$, (ax) $[6]Co^{2+}$, (ay) $[8]Co^{2+}$, (az) $[6]Co^{3+}$, (ba) $[6]Co^{4+}$, (bb) $[2]Ni^{2+}$, (bc) $[4]Ni^{2+}$, (bd) $[5]Ni^{2+}$, (be) $[6]Ni^{2+}$, (bf) $[6]Ni^{4+}$, (bg) $[2]Cu^{+}$, (bh) $[3]Cu^{+}$, (bi) $[4]Cu^{+}$, (bj) $[4]Cu^{2+}$, (bk) $[5]Cu^{2+}$, (bl) $[6]Cu^{2+}$, (bm) $[8]Cu^{2+}$, (bn) $[4]Cu^{3+}$, (bo) $[4]Zn^{2+}$, (bp) $[5]Zn^{2+}$, (bq) $[6]Zn^{2+}$, (br) $[6]Y^{3+}$, (bs) $[7]Y^{3+}$, (bt) $[8]Y^{3+}$,

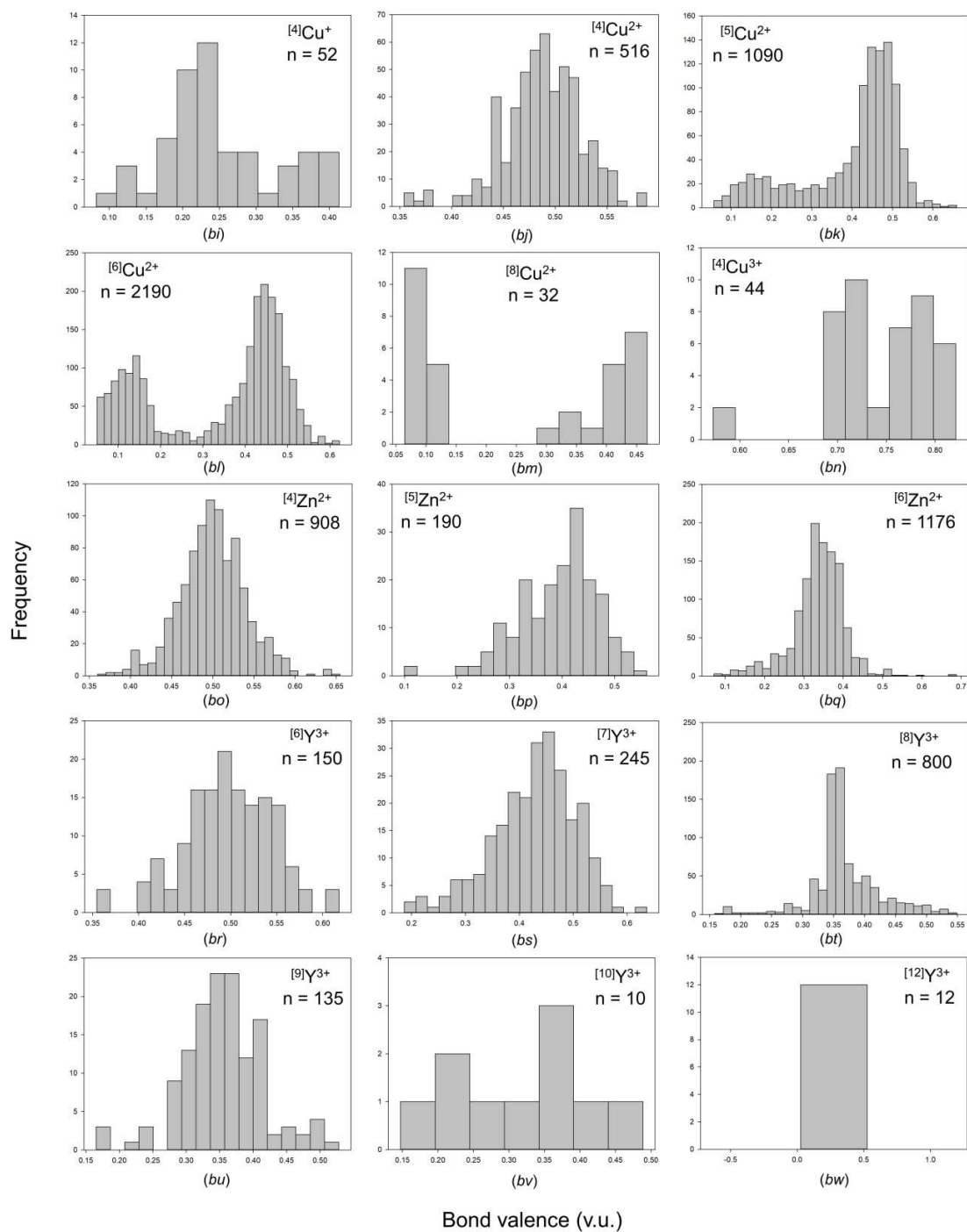
(bu) $^{[9]}\text{Y}^{3+}$, (bv) $^{[10]}\text{Y}^{3+}$, (bw) $^{[12]}\text{Y}^{3+}$, (bx) $^{[6]}\text{Zr}^{4+}$, (by) $^{[7]}\text{Zr}^{4+}$, (bz) $^{[8]}\text{Zr}^{4+}$, (ca) $^{[9]}\text{Zr}^{4+}$, (cb) $^{[10]}\text{Zr}^{4+}$, (cc) $^{[6]}\text{Nb}^{4+}$, (cd) $^{[4]}\text{Nb}^{5+}$, (ce) $^{[5]}\text{Nb}^{5+}$, (cf) $^{[6]}\text{Nb}^{5+}$, (cg) $^{[7]}\text{Nb}^{5+}$, (ch) $^{[8]}\text{Nb}^{5+}$, (ci) $^{[6]}\text{Mo}^{3+}$, (cj) $^{[6]}\text{Mo}^{4+}$, (ck) $^{[5]}\text{Mo}^{5+}$, (cl) $^{[6]}\text{Mo}^{5+}$, (cm) $^{[4]}\text{Mo}^{6+}$, (cn) $^{[5]}\text{Mo}^{6+}$, (co) $^{[6]}\text{Mo}^{6+}$, (cp) $^{[4]}\text{Tc}^{7+}$, (cq) $^{[6]}\text{Ru}^{3+}$, (cr) $^{[6]}\text{Ru}^{4+}$, (cs) $^{[6]}\text{Ru}^{5+}$, (ct) $^{[6]}\text{Rh}^{3+}$, (cu) $^{[6]}\text{Rh}^{4+}$, (cv) $^{[4]}\text{Pd}^{2+}$, (cw) $^{[6]}\text{Pd}^{4+}$, (cx) $^{[2]}\text{Ag}^+$, (cy) $^{[3]}\text{Ag}^+$, (cz) $^{[4]}\text{Ag}^+$, (da) $^{[5]}\text{Ag}^+$, (db) $^{[6]}\text{Ag}^+$, (dc) $^{[7]}\text{Ag}^+$, (dd) $^{[8]}\text{Ag}^+$, (de) $^{[9]}\text{Ag}^+$, (df) $^{[5]}\text{Cd}^{2+}$, (dg) $^{[6]}\text{Cd}^{2+}$, (dh) $^{[7]}\text{Cd}^{2+}$, (di) $^{[8]}\text{Cd}^{2+}$, (dj) $^{[9]}\text{Cd}^{2+}$, (dk) $^{[6]}\text{Hf}^{4+}$, (dl) $^{[7]}\text{Hf}^{4+}$, (dm) $^{[8]}\text{Hf}^{4+}$, (dn) $^{[6]}\text{Ta}^{5+}$, (do) $^{[7]}\text{Ta}^{5+}$, (dp) $^{[6]}\text{W}^{5+}$, (dq) $^{[4]}\text{W}^{6+}$, (dr) $^{[5]}\text{W}^{6+}$, (ds) $^{[6]}\text{W}^{6+}$, (dt) $^{[6]}\text{Re}^{5+}$, (du) $^{[4]}\text{Re}^{7+}$, (dv) $^{[5]}\text{Re}^{7+}$, (dw) $^{[6]}\text{Re}^{7+}$, (dx) $^{[6]}\text{Os}^{5+}$, (dy) $^{[6]}\text{Os}^{6+}$, (dz) $^{[5]}\text{Os}^{7+}$, (ea) $^{[6]}\text{Os}^{7+}$, (eb) $^{[4]}\text{Os}^{8+}$, (ec) $^{[5]}\text{Os}^{8+}$, (ed) $^{[6]}\text{Os}^{8+}$, (ee) $^{[6]}\text{Ir}^{3+}$, (ef) $^{[4]}\text{Ir}^{4+}$, (eg) $^{[6]}\text{Ir}^{4+}$, (eh) $^{[6]}\text{Ir}^{5+}$, (ei) $^{[4]}\text{Pt}^{2+}$, (ej) $^{[6]}\text{Pt}^{4+}$, (ek) $^{[4]}\text{Au}^{3+}$, (el) $^{[2]}\text{Hg}^{2+}$, (em) $^{[4]}\text{Hg}^{2+}$, (en) $^{[5]}\text{Hg}^{2+}$, (eo) $^{[6]}\text{Hg}^{2+}$, (ep) $^{[7]}\text{Hg}^{2+}$, (eq) $^{[8]}\text{Hg}^{2+}$.

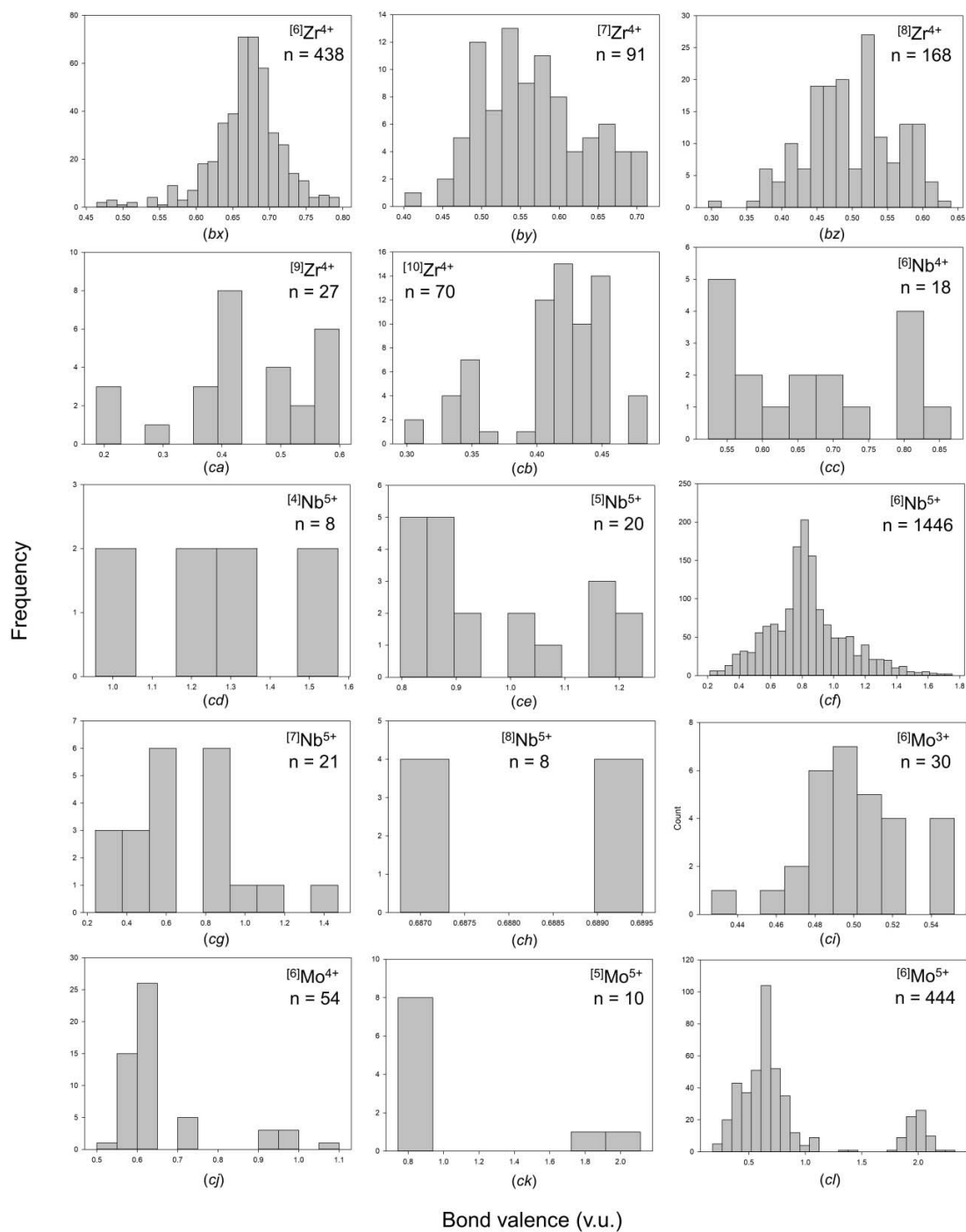


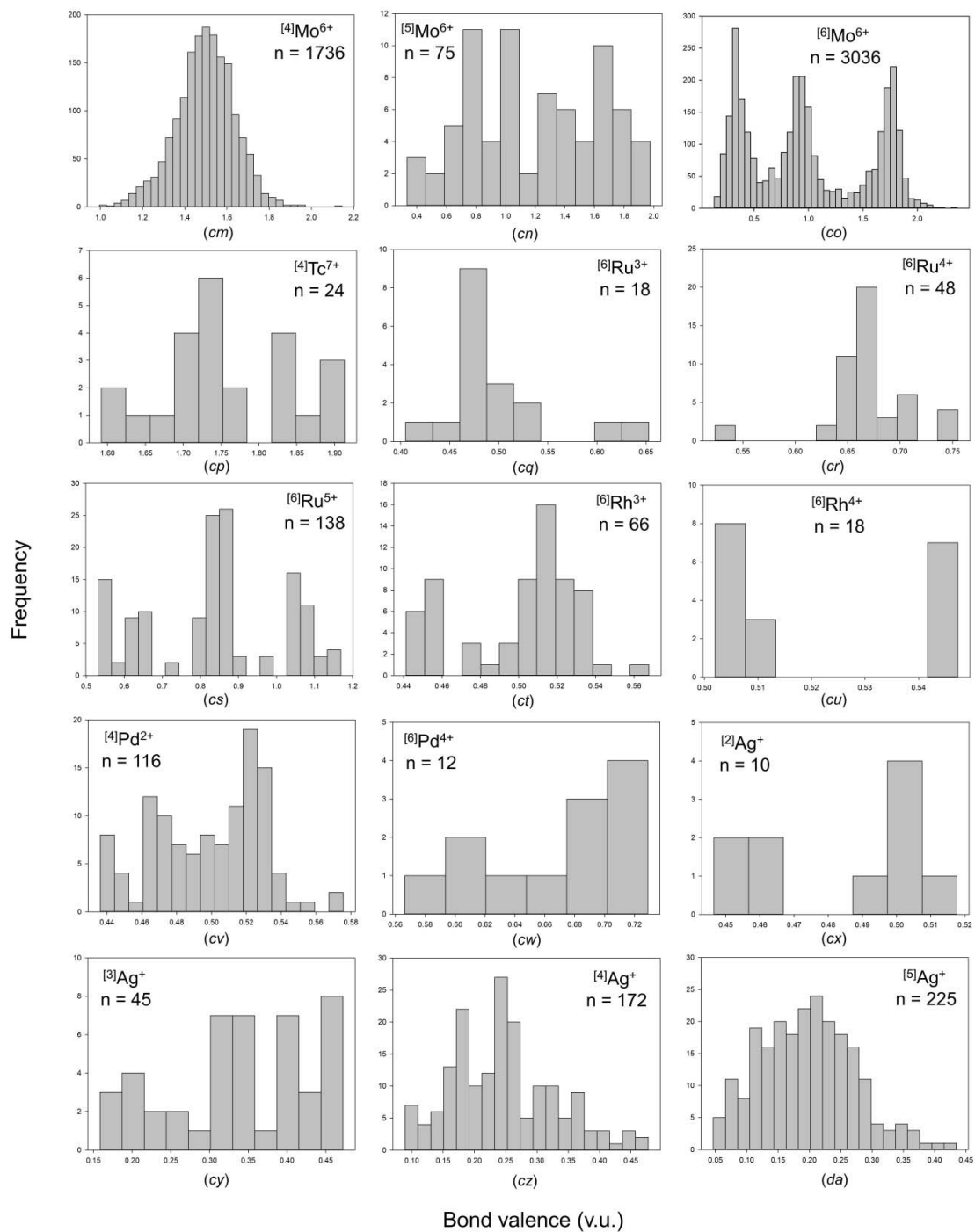


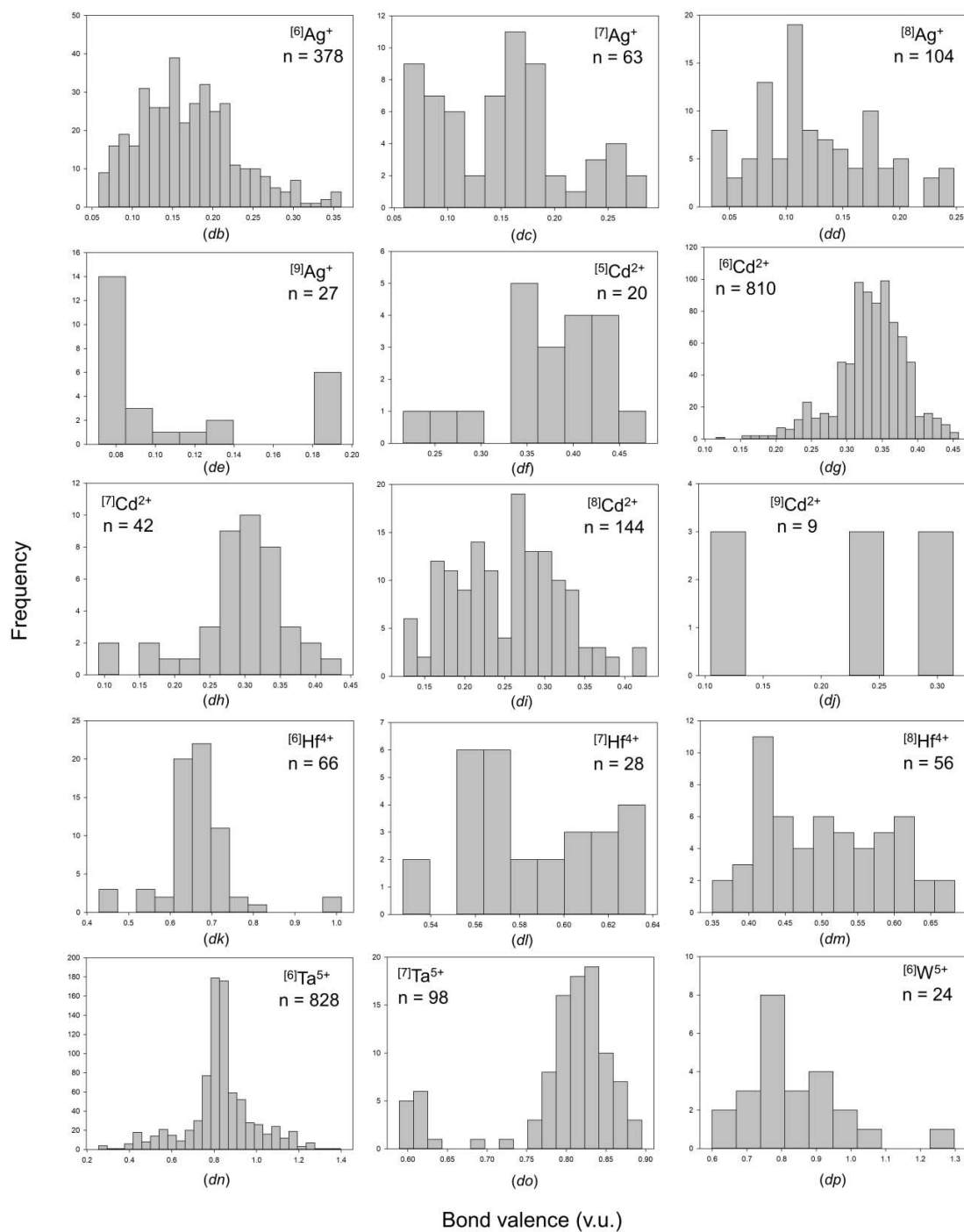


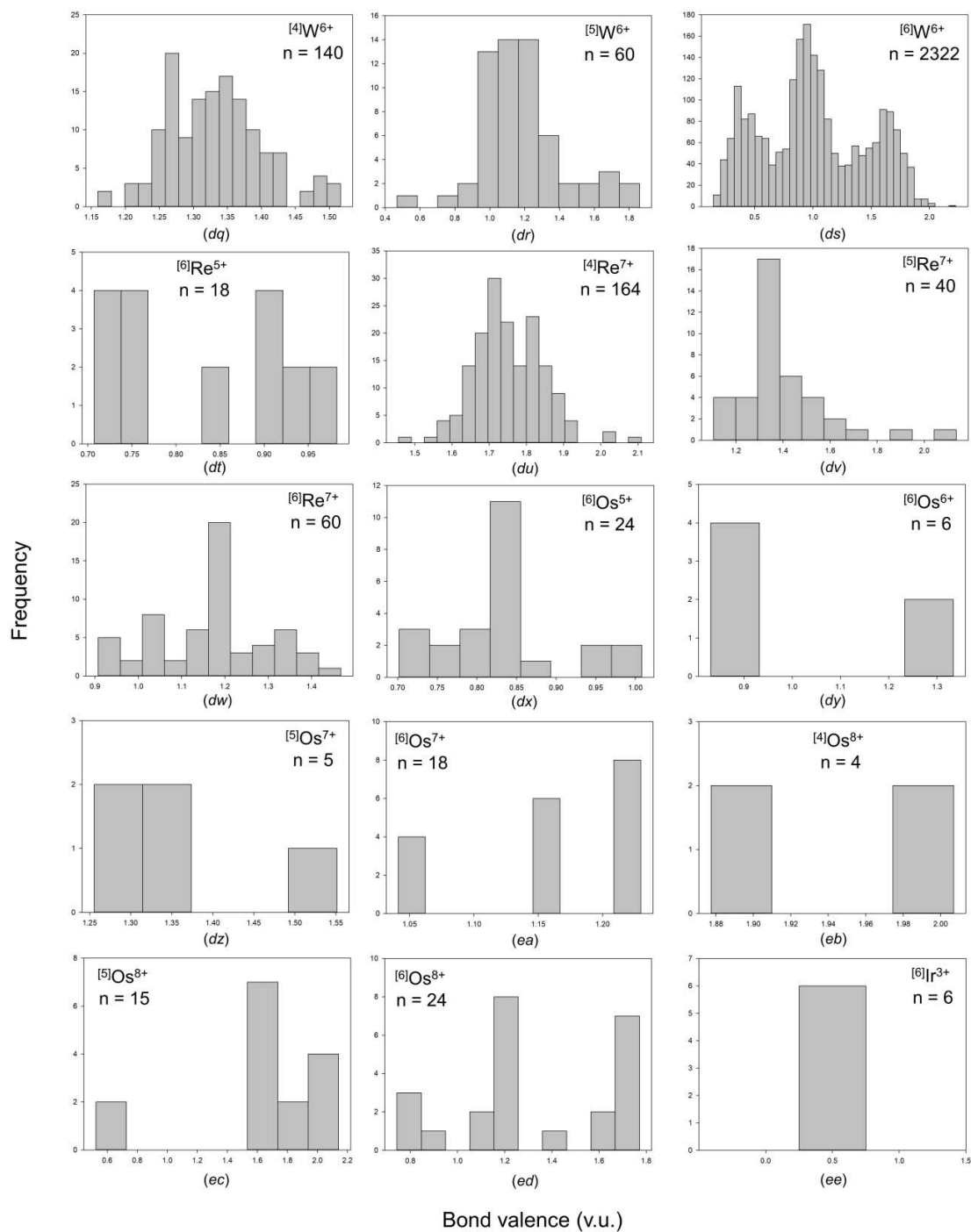












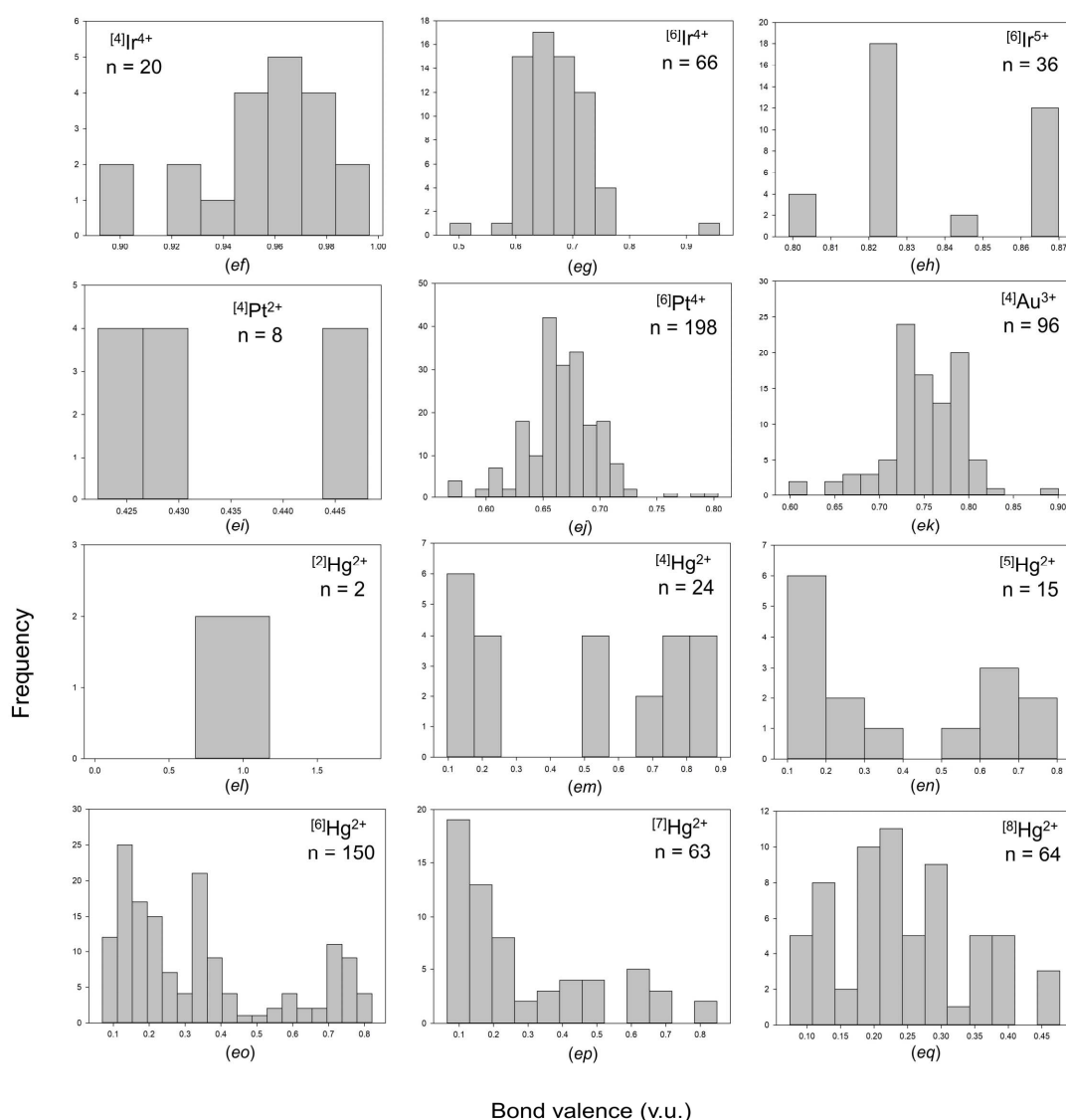
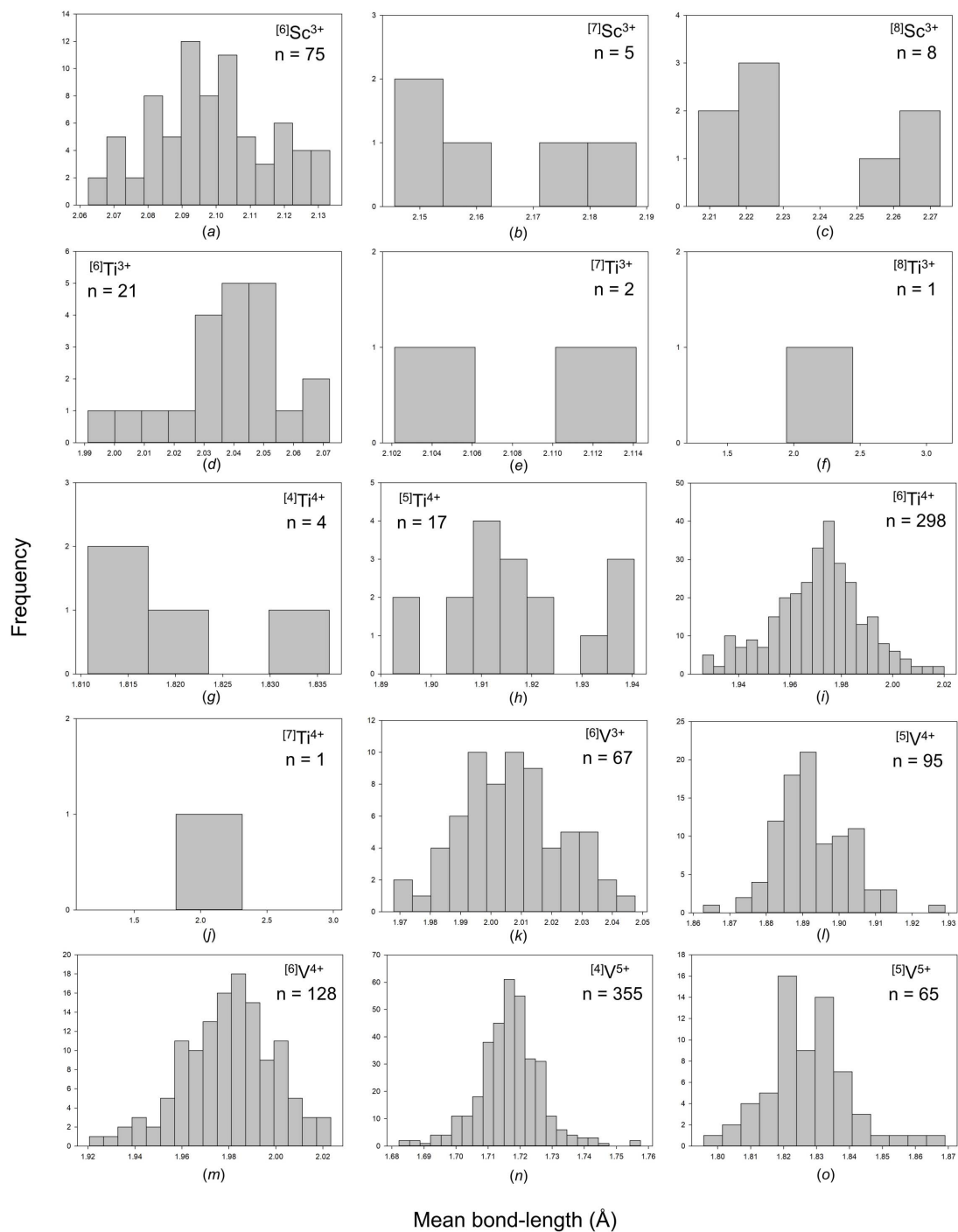
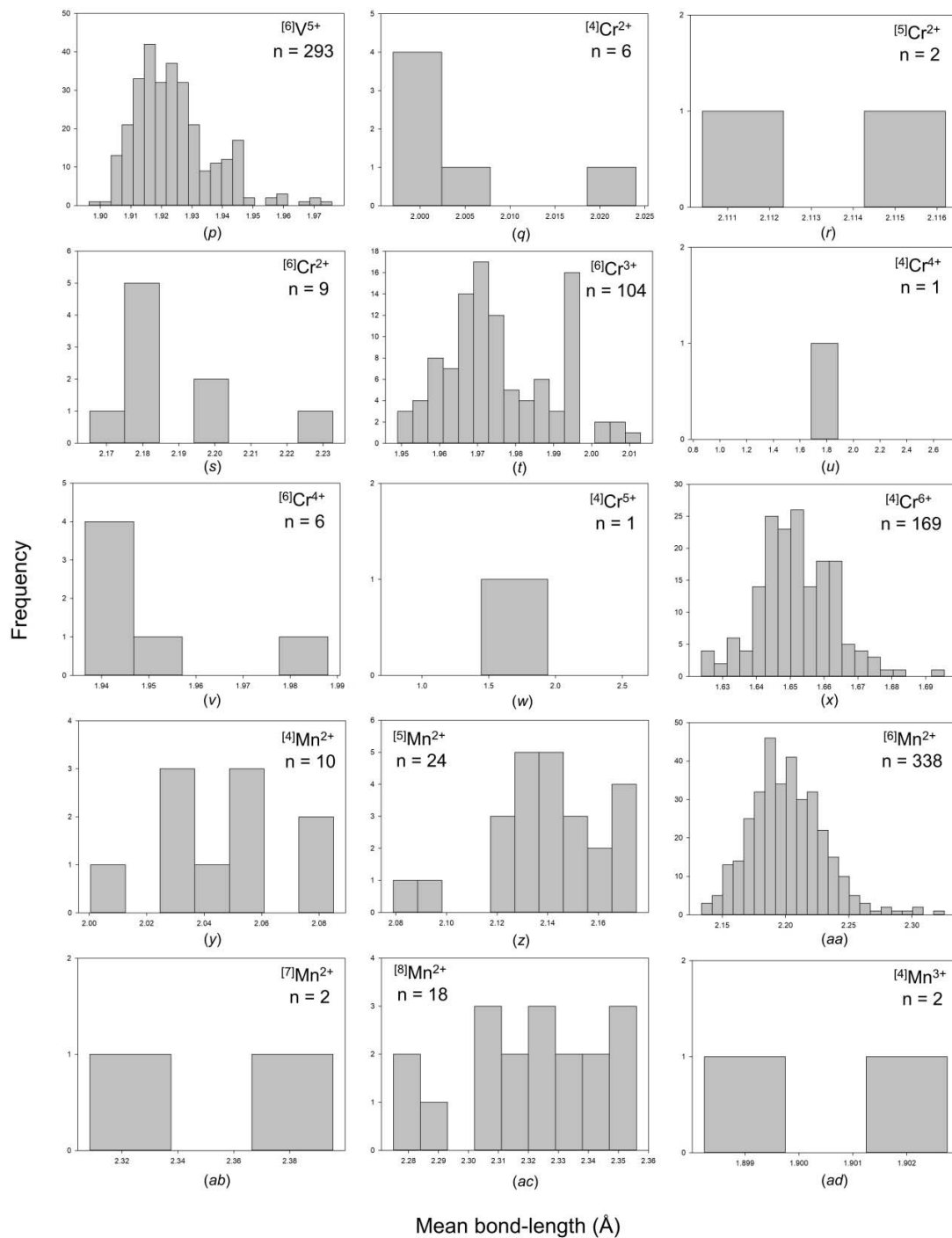


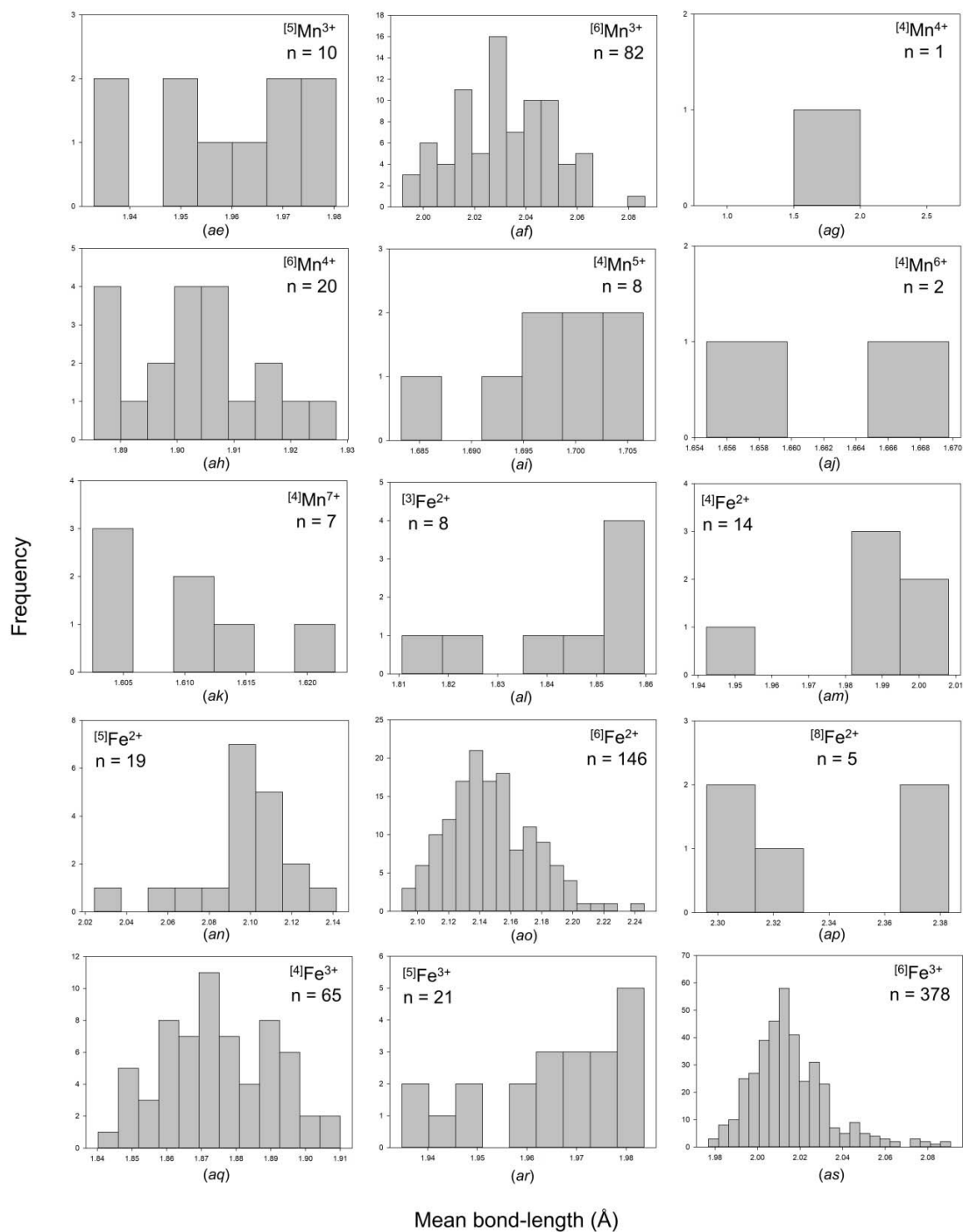
Figure S2 Bond-valence distributions for all configurations of the transition metal ions bonded to O^{2-} :

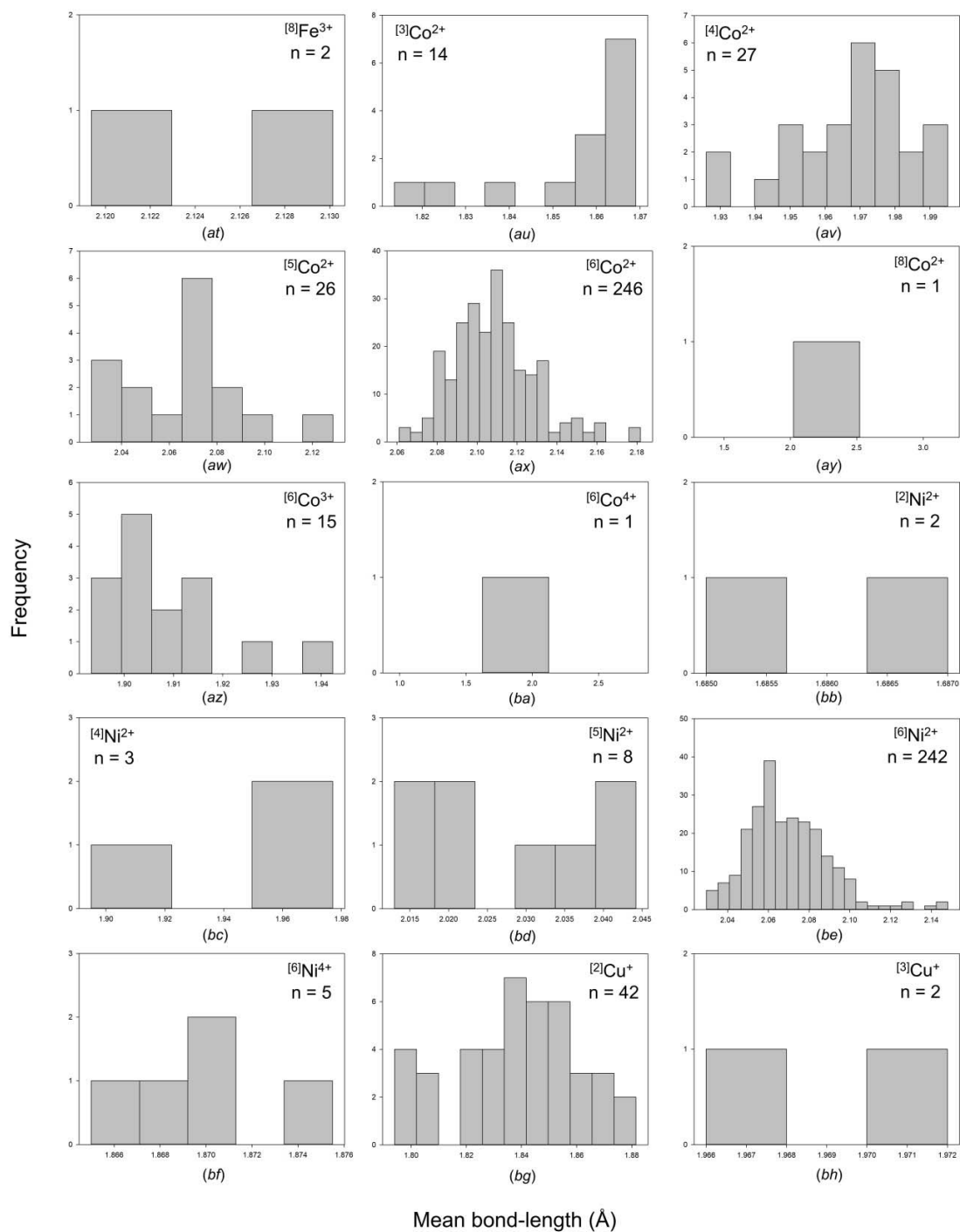
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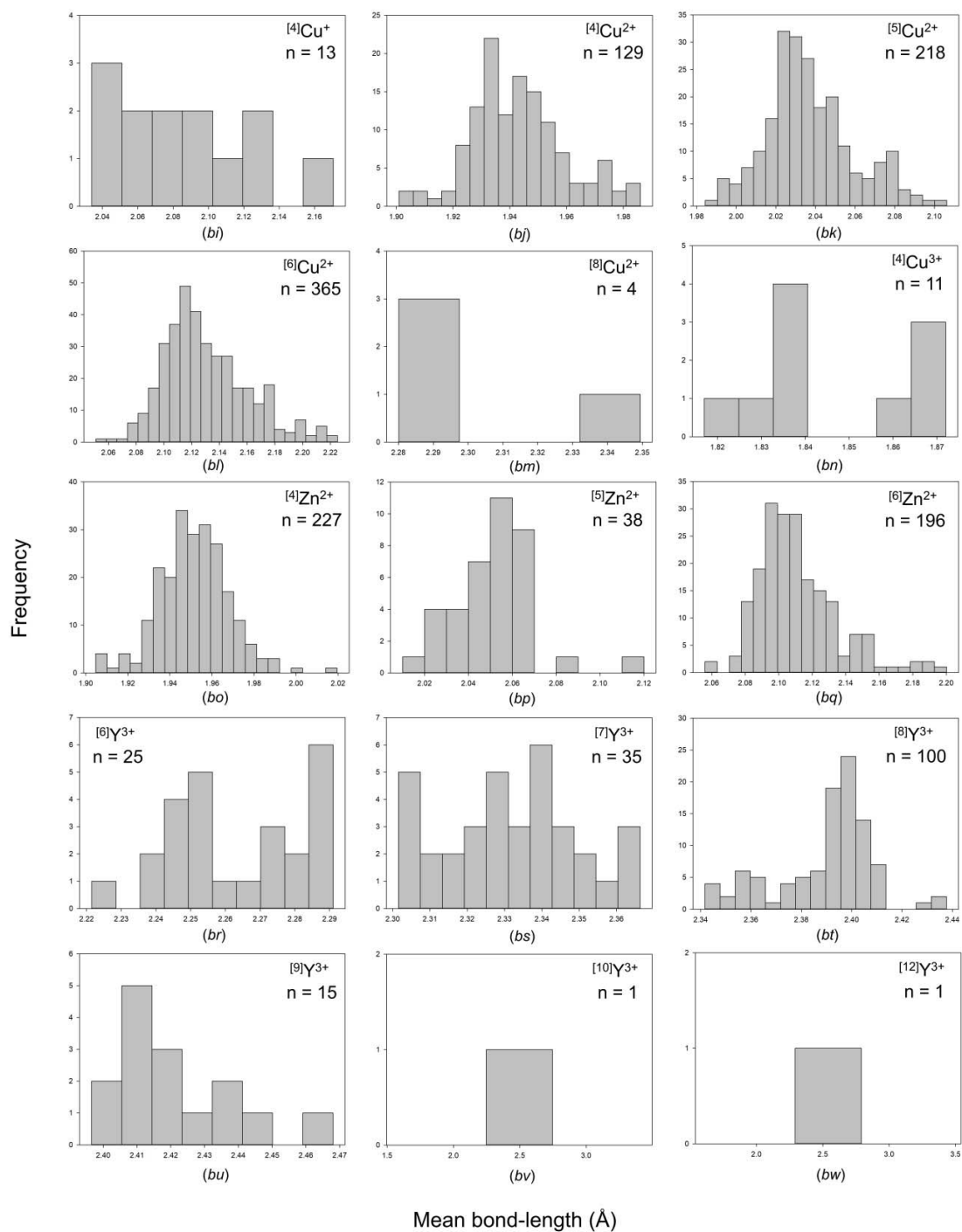
(bu) $^{[9]}\text{Y}^{3+}$, (bv) $^{[10]}\text{Y}^{3+}$, (bw) $^{[12]}\text{Y}^{3+}$, (bx) $^{[6]}\text{Zr}^{4+}$, (by) $^{[7]}\text{Zr}^{4+}$, (bz) $^{[8]}\text{Zr}^{4+}$, (ca) $^{[9]}\text{Zr}^{4+}$, (cb) $^{[10]}\text{Zr}^{4+}$, (cc) $^{[6]}\text{Nb}^{4+}$, (cd) $^{[4]}\text{Nb}^{5+}$, (ce) $^{[5]}\text{Nb}^{5+}$, (cf) $^{[6]}\text{Nb}^{5+}$, (cg) $^{[7]}\text{Nb}^{5+}$, (ch) $^{[8]}\text{Nb}^{5+}$, (ci) $^{[6]}\text{Mo}^{3+}$, (cj) $^{[6]}\text{Mo}^{4+}$, (ck) $^{[5]}\text{Mo}^{5+}$, (cl) $^{[6]}\text{Mo}^{5+}$, (cm) $^{[4]}\text{Mo}^{6+}$, (cn) $^{[5]}\text{Mo}^{6+}$, (co) $^{[6]}\text{Mo}^{6+}$, (cp) $^{[4]}\text{Tc}^{7+}$, (cq) $^{[6]}\text{Ru}^{3+}$, (cr) $^{[6]}\text{Ru}^{4+}$, (cs) $^{[6]}\text{Ru}^{5+}$, (ct) $^{[6]}\text{Rh}^{3+}$, (cu) $^{[6]}\text{Rh}^{4+}$, (cv) $^{[4]}\text{Pd}^{2+}$, (cw) $^{[6]}\text{Pd}^{4+}$, (cx) $^{[2]}\text{Ag}^+$, (cy) $^{[3]}\text{Ag}^+$, (cz) $^{[4]}\text{Ag}^+$, (da) $^{[5]}\text{Ag}^+$, (db) $^{[6]}\text{Ag}^+$, (dc) $^{[7]}\text{Ag}^+$, (dd) $^{[8]}\text{Ag}^+$, (de) $^{[9]}\text{Ag}^+$, (df) $^{[5]}\text{Cd}^{2+}$, (dg) $^{[6]}\text{Cd}^{2+}$, (dh) $^{[7]}\text{Cd}^{2+}$, (di) $^{[8]}\text{Cd}^{2+}$, (dj) $^{[9]}\text{Cd}^{2+}$, (dk) $^{[6]}\text{Hf}^{4+}$, (dl) $^{[7]}\text{Hf}^{4+}$, (dm) $^{[8]}\text{Hf}^{4+}$, (dn) $^{[6]}\text{Ta}^{5+}$, (do) $^{[7]}\text{Ta}^{5+}$, (dp) $^{[6]}\text{W}^{5+}$, (dq) $^{[4]}\text{W}^{6+}$, (dr) $^{[5]}\text{W}^{6+}$, (ds) $^{[6]}\text{W}^{6+}$, (dt) $^{[6]}\text{Re}^{5+}$, (du) $^{[4]}\text{Re}^{7+}$, (dv) $^{[5]}\text{Re}^{7+}$, (dw) $^{[6]}\text{Re}^{7+}$, (dx) $^{[6]}\text{Os}^{5+}$, (dy) $^{[6]}\text{Os}^{6+}$, (dz) $^{[5]}\text{Os}^{7+}$, (ea) $^{[6]}\text{Os}^{7+}$, (eb) $^{[4]}\text{Os}^{8+}$, (ec) $^{[5]}\text{Os}^{8+}$, (ed) $^{[6]}\text{Os}^{8+}$, (ee) $^{[6]}\text{Ir}^{3+}$, (ef) $^{[4]}\text{Ir}^{4+}$, (eg) $^{[6]}\text{Ir}^{4+}$, (eh) $^{[6]}\text{Ir}^{5+}$, (ei) $^{[4]}\text{Pt}^{2+}$, (ej) $^{[6]}\text{Pt}^{4+}$, (ek) $^{[4]}\text{Au}^{3+}$, (el) $^{[2]}\text{Hg}^{2+}$, (em) $^{[4]}\text{Hg}^{2+}$, (en) $^{[5]}\text{Hg}^{2+}$, (eo) $^{[6]}\text{Hg}^{2+}$, (ep) $^{[7]}\text{Hg}^{2+}$, (eq) $^{[8]}\text{Hg}^{2+}$.

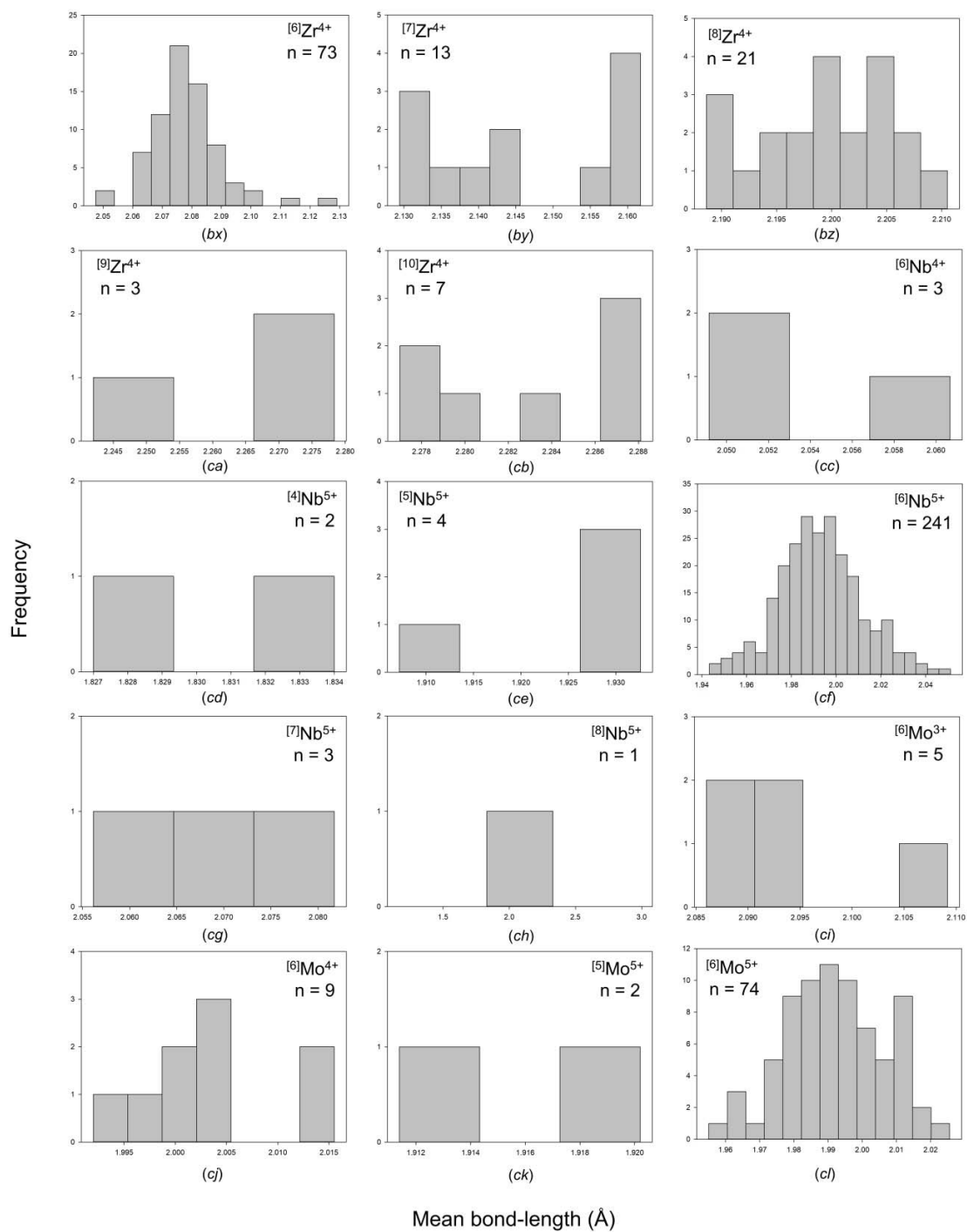


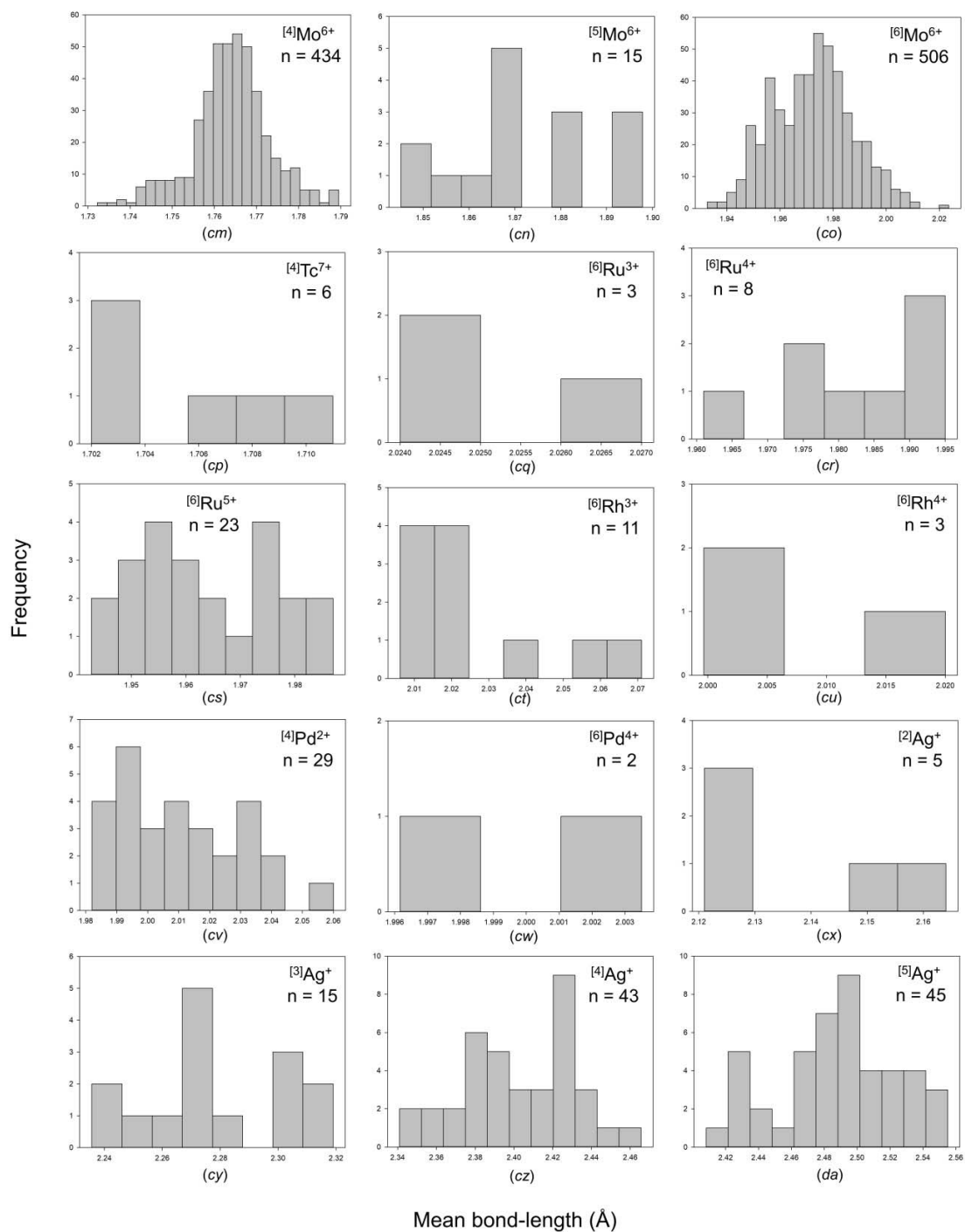


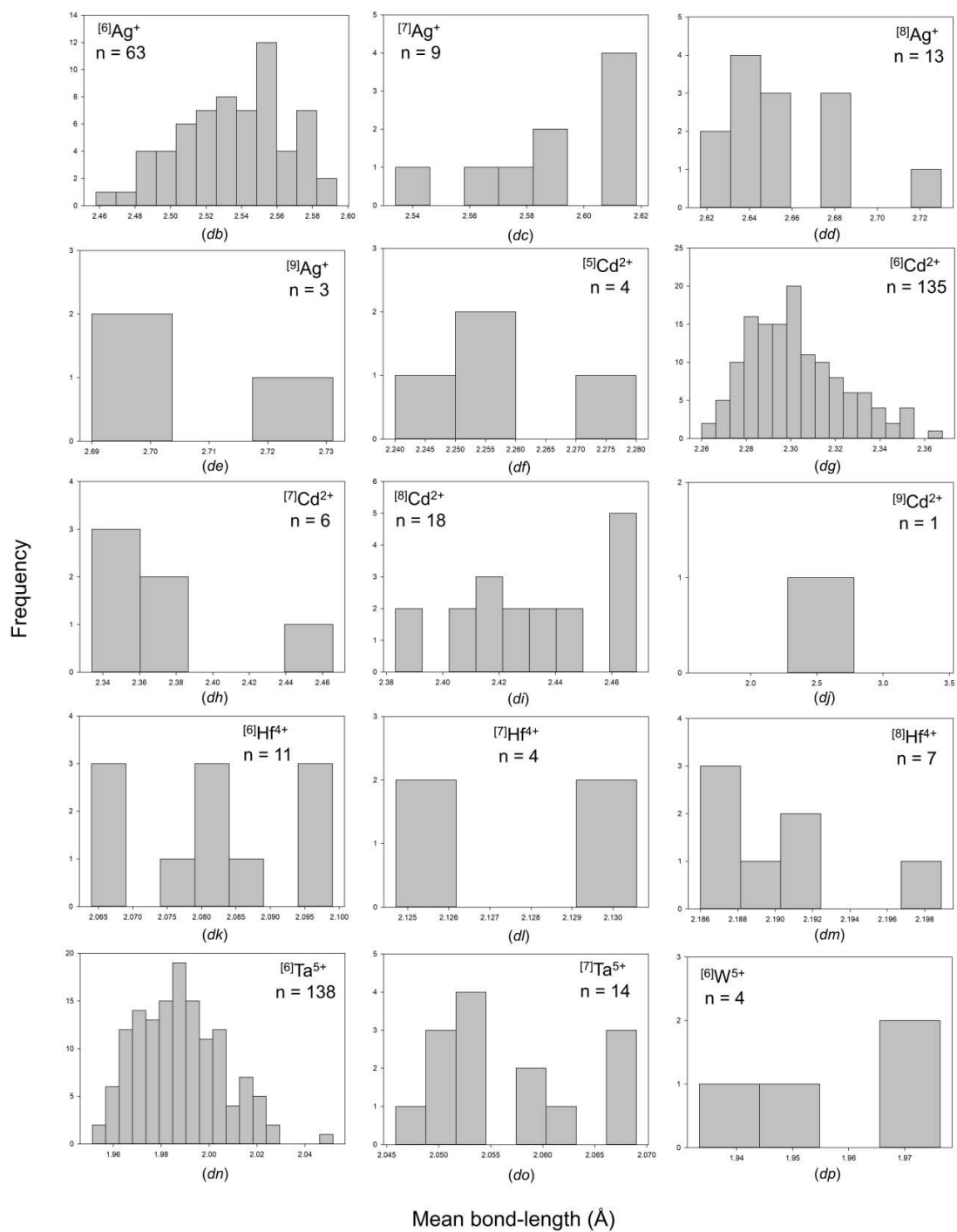


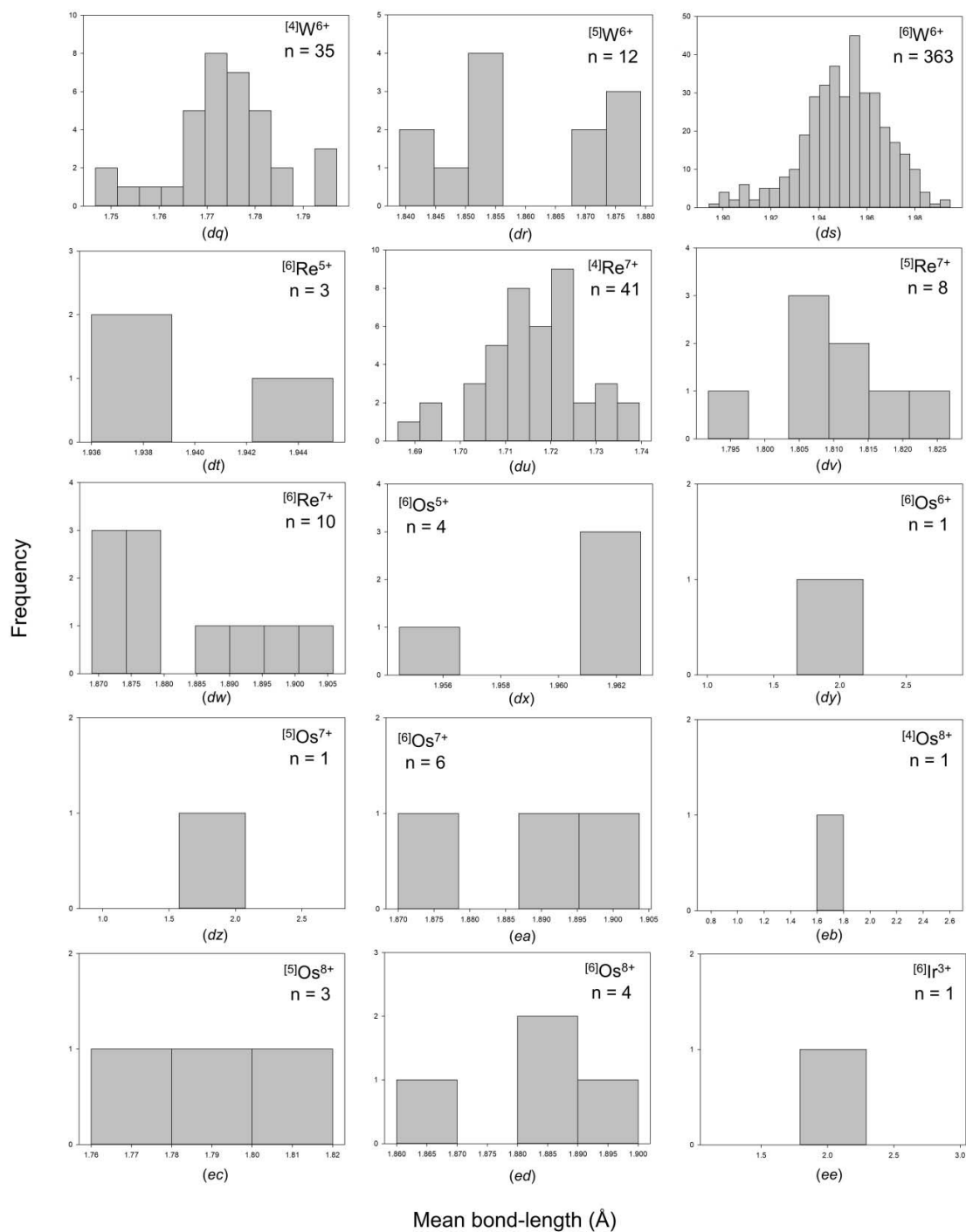












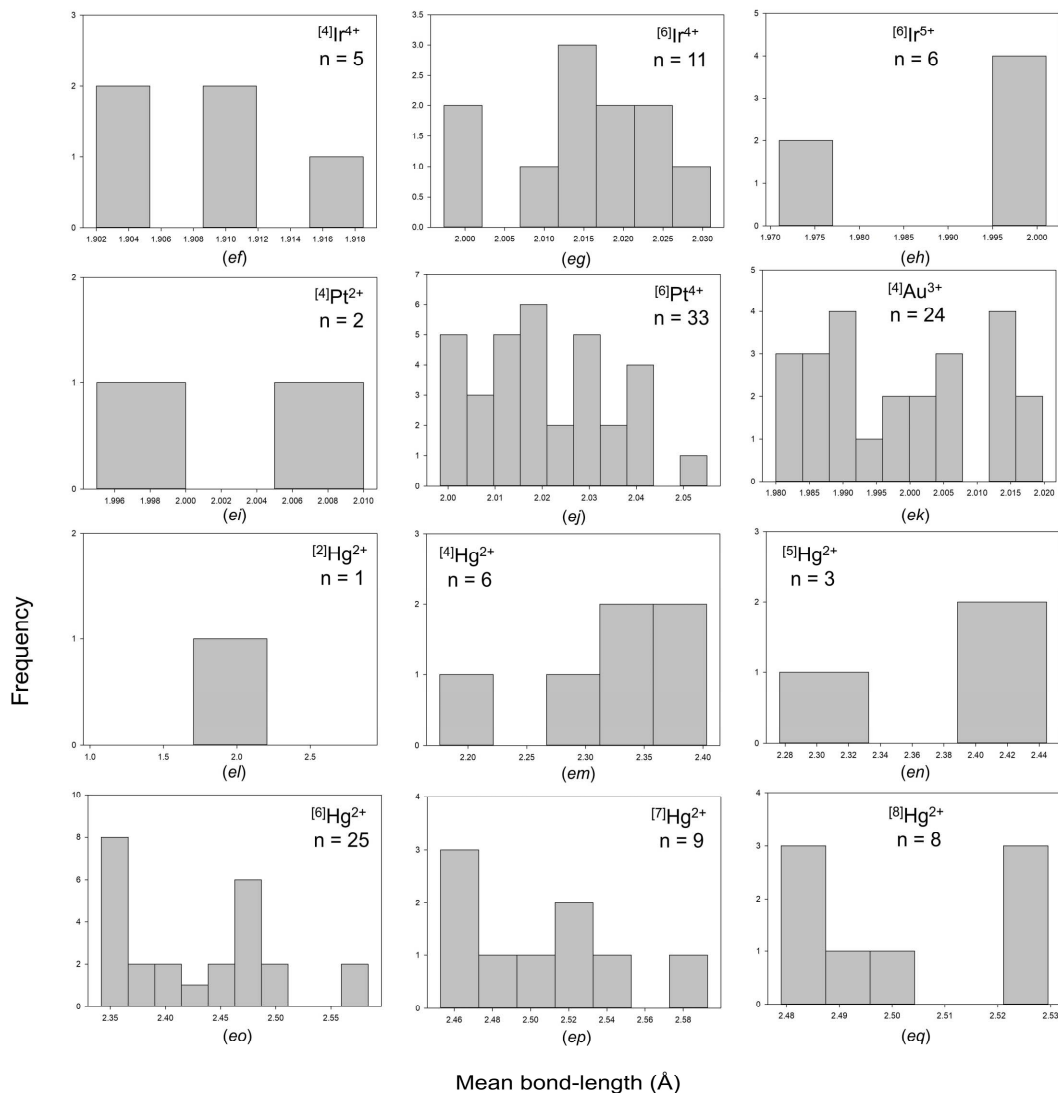
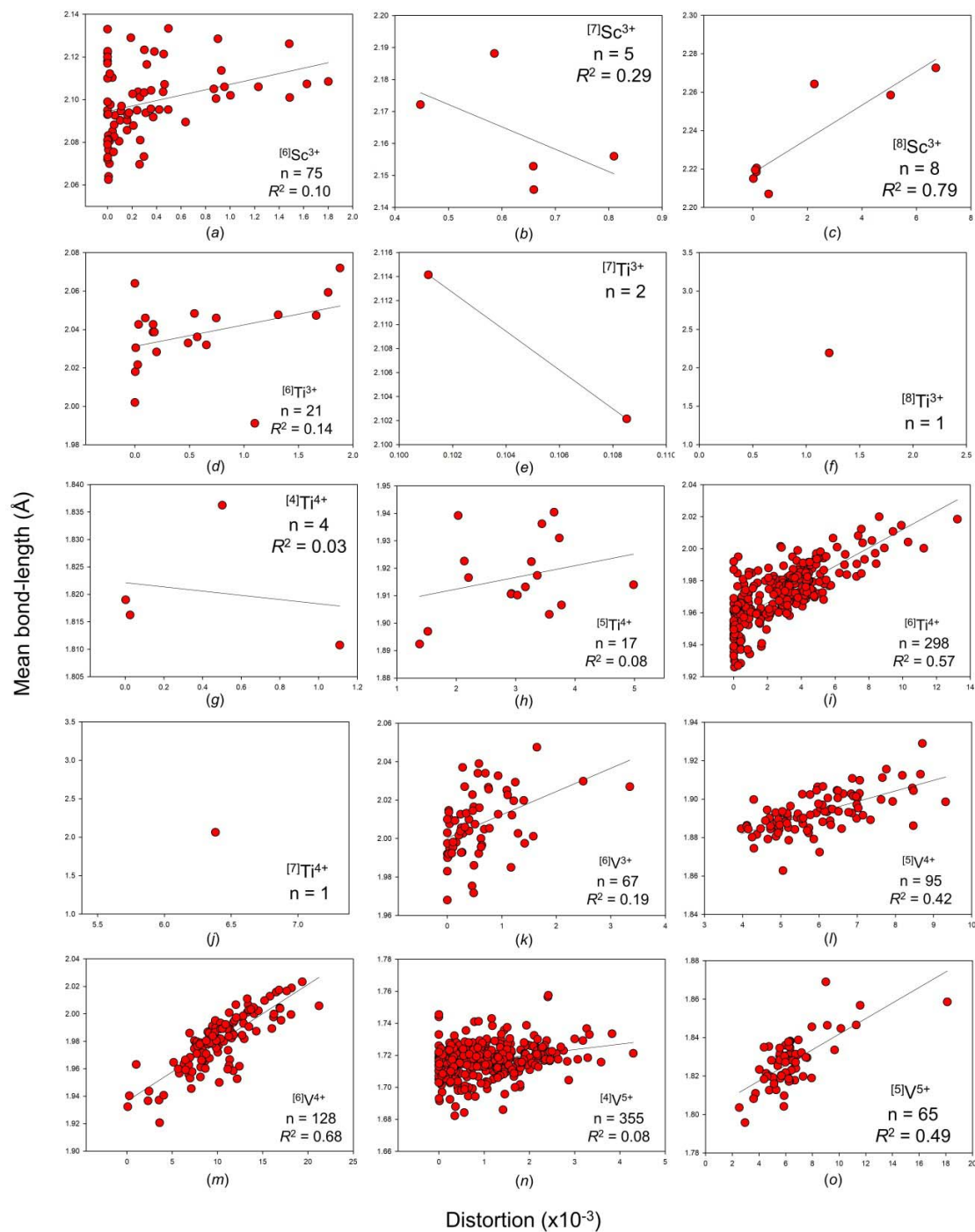
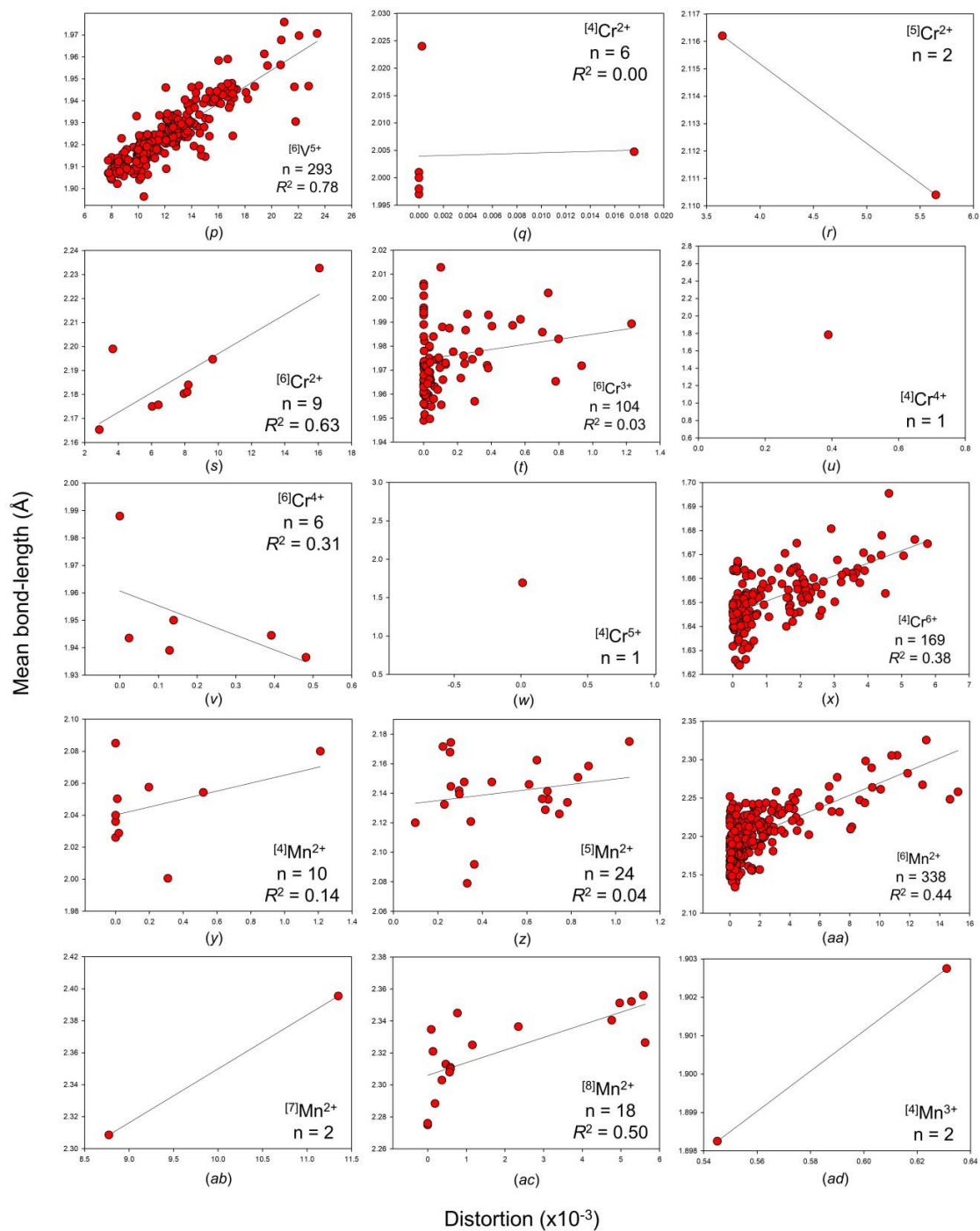
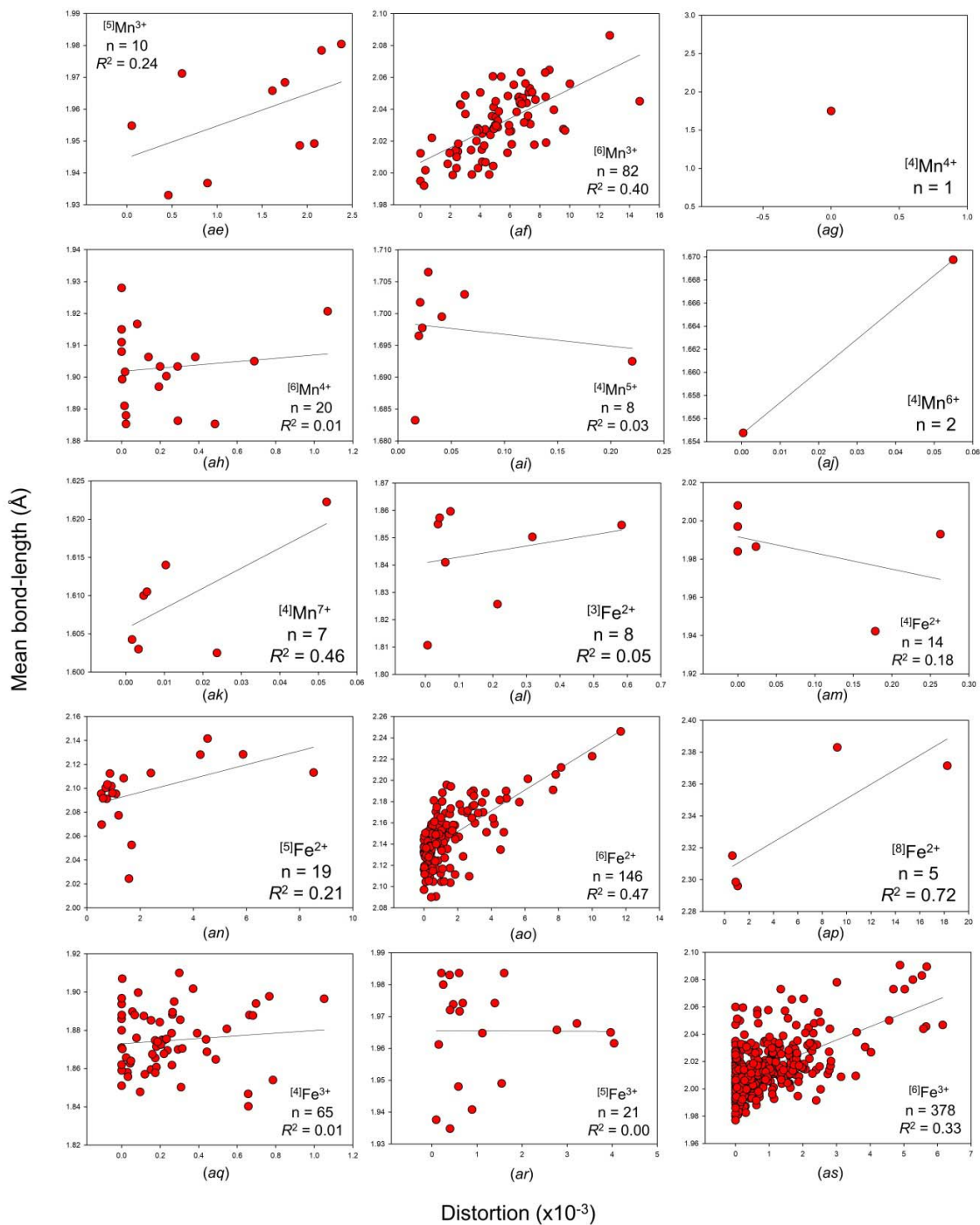


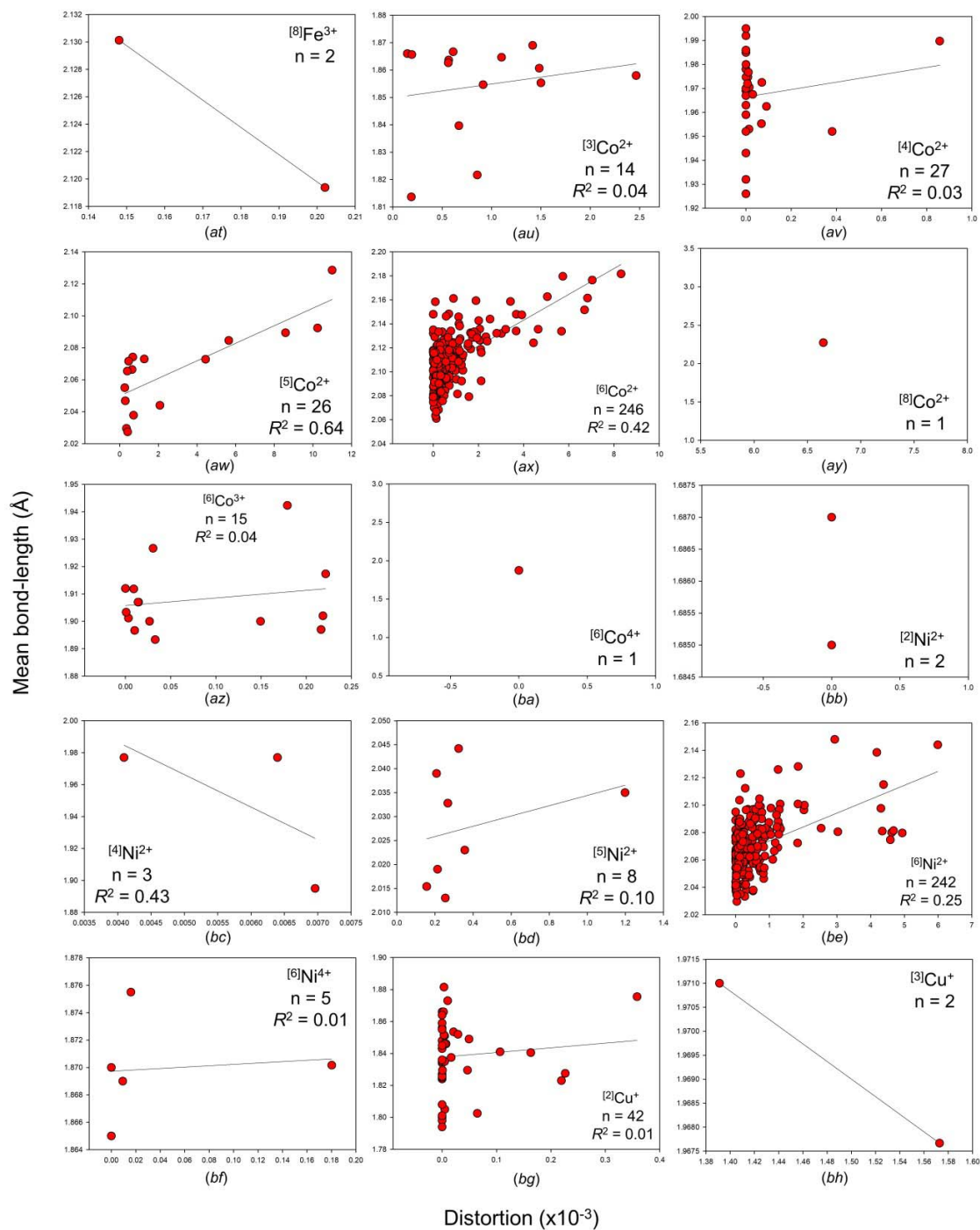
Figure S3 Mean bond-length distributions for all configurations of the transition metal ions bonded to O^{2-} : (a) $[6]Sc^{3+}$, (b) $[7]Sc^{3+}$, (c) $[8]Sc^{3+}$, (d) $[6]Ti^{3+}$, (e) $[7]Ti^{3+}$, (f) $[8]Ti^{3+}$, (g) $[4]Ti^{4+}$, (h) $[5]Ti^{4+}$, (i) $[6]Ti^{4+}$, (j) $[7]Ti^{4+}$, (k) $[6]V^{3+}$, (l) $[5]V^{4+}$, (m) $[6]V^{4+}$, (n) $[4]V^{5+}$, (o) $[5]V^{5+}$, (p) $[6]V^{5+}$, (q) $[4]Cr^{2+}$, (r) $[5]Cr^{2+}$, (s) $[6]Cr^{2+}$, (t) $[6]Cr^{3+}$, (u) $[4]Cr^{4+}$, (v) $[6]Cr^{4+}$, (w) $[4]Cr^{5+}$, (x) $[4]Cr^{6+}$, (y) $[4]Mn^{2+}$, (z) $[5]Mn^{2+}$, (aa) $[6]Mn^{2+}$, (ab) $[7]Mn^{2+}$, (ac) $[8]Mn^{2+}$, (ad) $[4]Mn^{3+}$, (ae) $[5]Mn^{3+}$, (af) $[6]Mn^{3+}$, (ag) $[4]Mn^{4+}$, (ah) $[6]Mn^{4+}$, (ai) $[4]Mn^{5+}$, (aj) $[4]Mn^{6+}$, (ak) $[4]Mn^{7+}$, (al) $[3]Fe^{2+}$, (am) $[4]Fe^{2+}$, (an) $[5]Fe^{2+}$, (ao) $[6]Fe^{2+}$, (ap) $[8]Fe^{2+}$, (aq) $[4]Fe^{3+}$, (ar) $[5]Fe^{3+}$, (as) $[6]Fe^{3+}$, (at) $[8]Fe^{3+}$, (au) $[3]Co^{2+}$, (av) $[4]Co^{2+}$, (aw) $[5]Co^{2+}$, (ax) $[6]Co^{2+}$, (ay) $[8]Co^{2+}$, (az) $[6]Co^{3+}$, (ba) $[6]Co^{4+}$, (bb) $[2]Ni^{2+}$, (bc) $[4]Ni^{2+}$, (bd) $[5]Ni^{2+}$, (be) $[6]Ni^{2+}$, (bf) $[6]Ni^{4+}$, (bg) $[2]Cu^{+}$, (bh) $[3]Cu^{+}$, (bi) $[4]Cu^{+}$, (bj) $[4]Cu^{2+}$, (bk) $[5]Cu^{2+}$, (bl) $[6]Cu^{2+}$, (bm) $[8]Cu^{2+}$, (bn) $[4]Cu^{3+}$, (bo) $[4]Zn^{2+}$, (bp) $[5]Zn^{2+}$, (bq) $[6]Zn^{2+}$, (br) $[6]Y^{3+}$, (bs) $[7]Y^{3+}$,

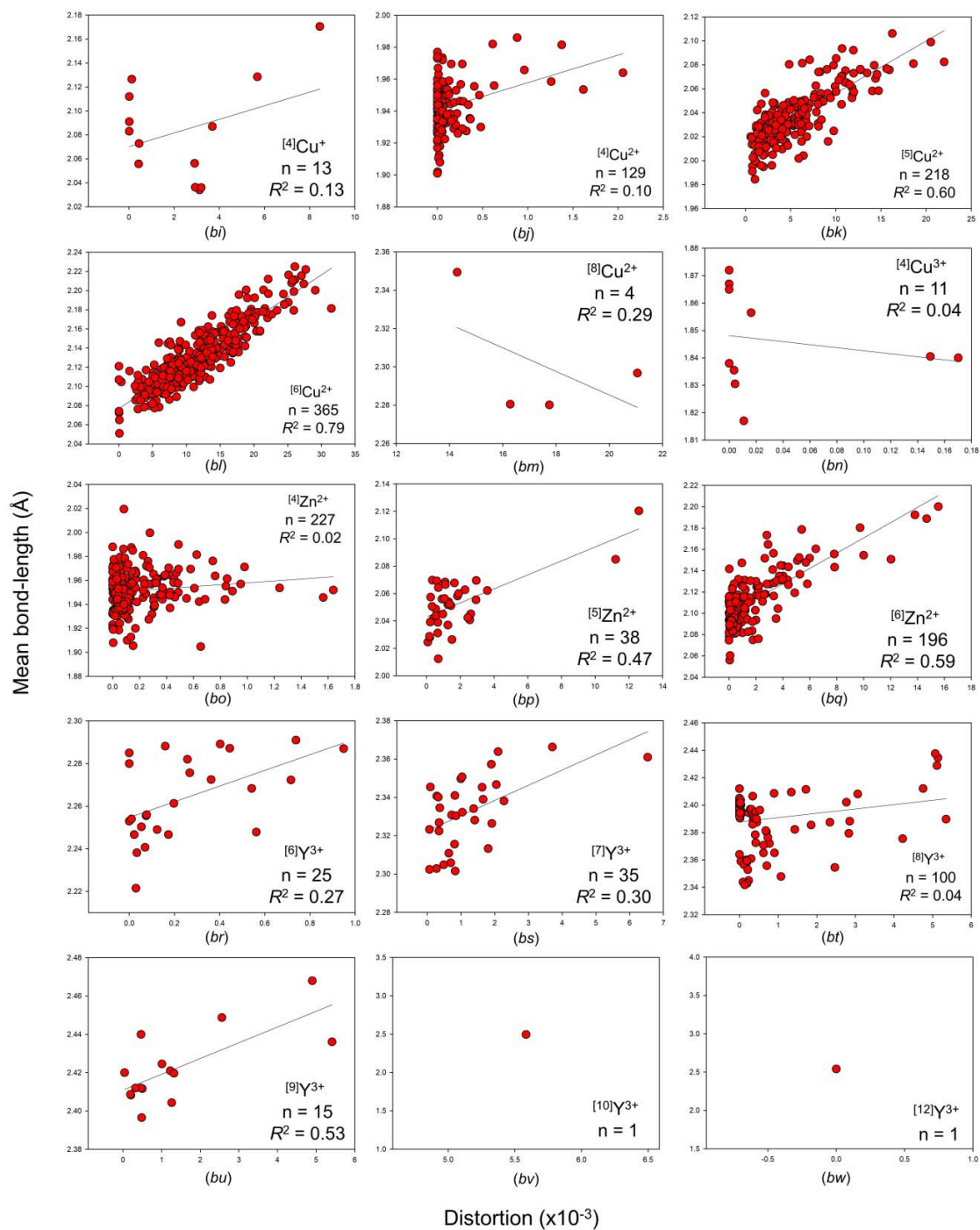
(bt) $^{[8]}\text{Y}^{3+}$, (bu) $^{[9]}\text{Y}^{3+}$, (bv) $^{[10]}\text{Y}^{3+}$, (bw) $^{[12]}\text{Y}^{3+}$, (bx) $^{[6]}\text{Zr}^{4+}$, (by) $^{[7]}\text{Zr}^{4+}$, (bz) $^{[8]}\text{Zr}^{4+}$, (ca) $^{[9]}\text{Zr}^{4+}$, (cb) $^{[10]}\text{Zr}^{4+}$, (cc) $^{[6]}\text{Nb}^{4+}$, (cd) $^{[4]}\text{Nb}^{5+}$, (ce) $^{[5]}\text{Nb}^{5+}$, (cf) $^{[6]}\text{Nb}^{5+}$, (cg) $^{[7]}\text{Nb}^{5+}$, (ch) $^{[8]}\text{Nb}^{5+}$, (ci) $^{[6]}\text{Mo}^{3+}$, (cj) $^{[6]}\text{Mo}^{4+}$, (ck) $^{[5]}\text{Mo}^{5+}$, (cl) $^{[6]}\text{Mo}^{5+}$, (cm) $^{[4]}\text{Mo}^{6+}$, (cn) $^{[5]}\text{Mo}^{6+}$, (co) $^{[6]}\text{Mo}^{6+}$, (cp) $^{[4]}\text{Tc}^{7+}$, (cq) $^{[6]}\text{Ru}^{3+}$, (cr) $^{[6]}\text{Ru}^{4+}$, (cs) $^{[6]}\text{Ru}^{5+}$, (ct) $^{[6]}\text{Rh}^{3+}$, (cu) $^{[6]}\text{Rh}^{4+}$, (cv) $^{[4]}\text{Pd}^{2+}$, (cw) $^{[6]}\text{Pd}^{4+}$, (cx) $^{[2]}\text{Ag}^+$, (cy) $^{[3]}\text{Ag}^+$, (cz) $^{[4]}\text{Ag}^+$, (da) $^{[5]}\text{Ag}^+$, (db) $^{[6]}\text{Ag}^+$, (dc) $^{[7]}\text{Ag}^+$, (dd) $^{[8]}\text{Ag}^+$, (de) $^{[9]}\text{Ag}^+$, (df) $^{[5]}\text{Cd}^{2+}$, (dg) $^{[6]}\text{Cd}^{2+}$, (dh) $^{[7]}\text{Cd}^{2+}$, (di) $^{[8]}\text{Cd}^{2+}$, (dj) $^{[9]}\text{Cd}^{2+}$, (dk) $^{[6]}\text{Hf}^{4+}$, (dl) $^{[7]}\text{Hf}^{4+}$, (dm) $^{[8]}\text{Hf}^{4+}$, (dn) $^{[6]}\text{Ta}^{5+}$, (do) $^{[7]}\text{Ta}^{5+}$, (dp) $^{[6]}\text{W}^{5+}$, (dq) $^{[4]}\text{W}^{6+}$, (dr) $^{[5]}\text{W}^{6+}$, (ds) $^{[6]}\text{W}^{6+}$, (dt) $^{[6]}\text{Re}^{5+}$, (du) $^{[4]}\text{Re}^{7+}$, (dv) $^{[5]}\text{Re}^{7+}$, (dw) $^{[6]}\text{Re}^{7+}$, (dx) $^{[6]}\text{Os}^{5+}$, (dy) $^{[6]}\text{Os}^{6+}$, (dz) $^{[5]}\text{Os}^{7+}$, (ea) $^{[6]}\text{Os}^{7+}$, (eb) $^{[4]}\text{Os}^{8+}$, (ec) $^{[5]}\text{Os}^{8+}$, (ed) $^{[6]}\text{Os}^{8+}$, (ee) $^{[6]}\text{Ir}^{3+}$, (ef) $^{[4]}\text{Ir}^{4+}$, (eg) $^{[6]}\text{Ir}^{4+}$, (eh) $^{[6]}\text{Ir}^{5+}$, (ei) $^{[4]}\text{Pt}^{2+}$, (ej) $^{[6]}\text{Pt}^{4+}$, (ek) $^{[4]}\text{Au}^{3+}$, (el) $^{[2]}\text{Hg}^{2+}$, (em) $^{[4]}\text{Hg}^{2+}$, (en) $^{[5]}\text{Hg}^{2+}$, (eo) $^{[6]}\text{Hg}^{2+}$, (ep) $^{[7]}\text{Hg}^{2+}$, (eq) $^{[8]}\text{Hg}^{2+}$.

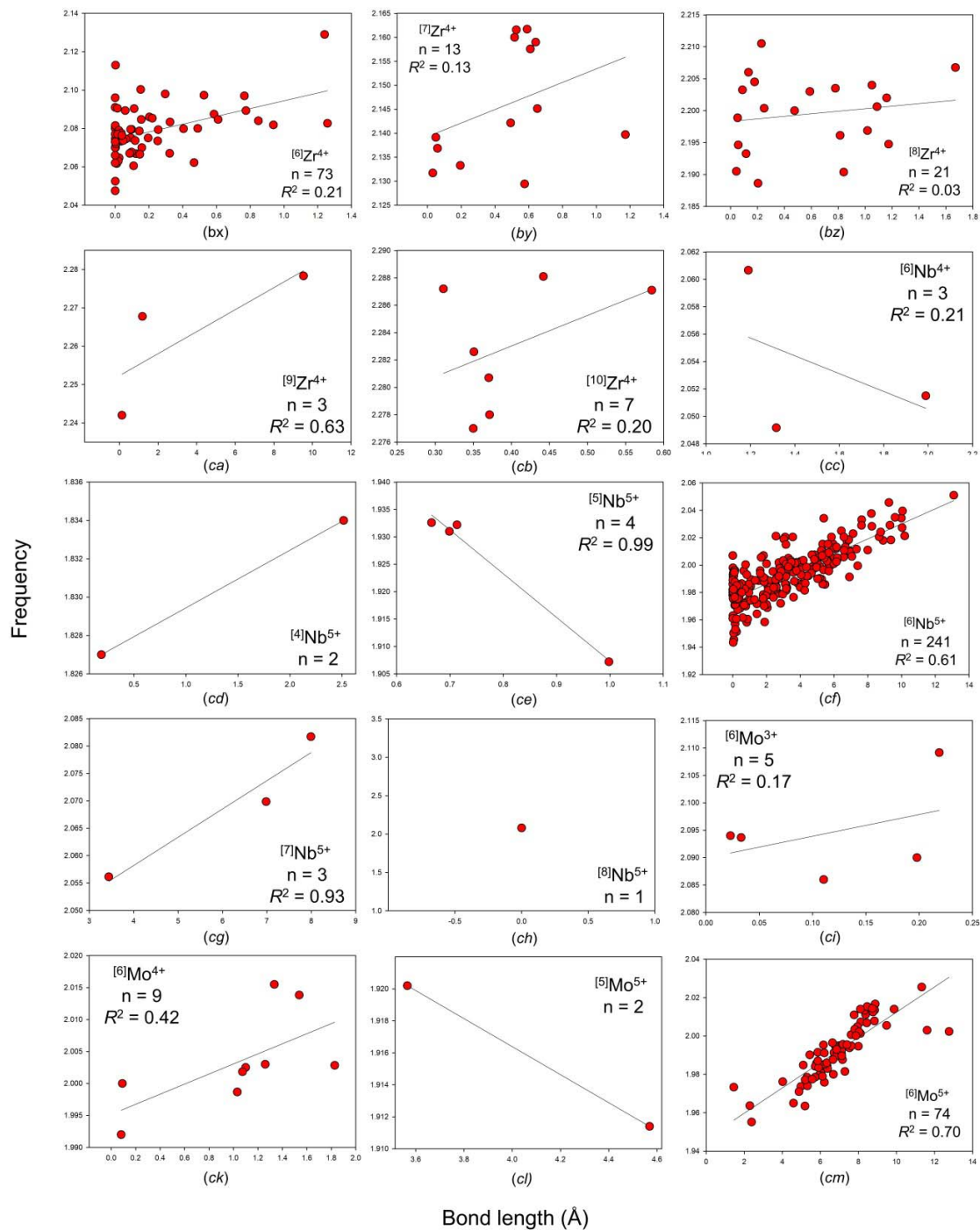


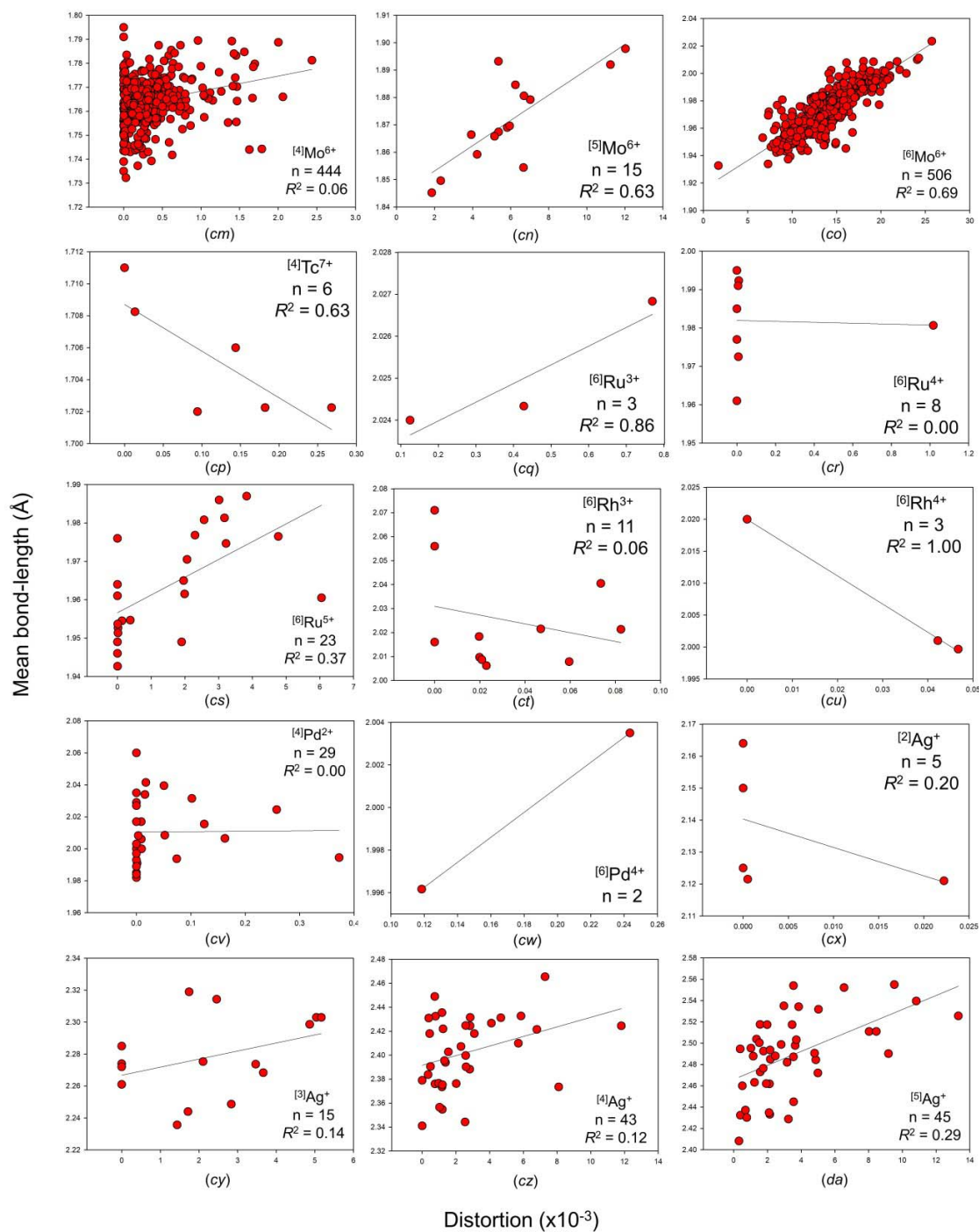


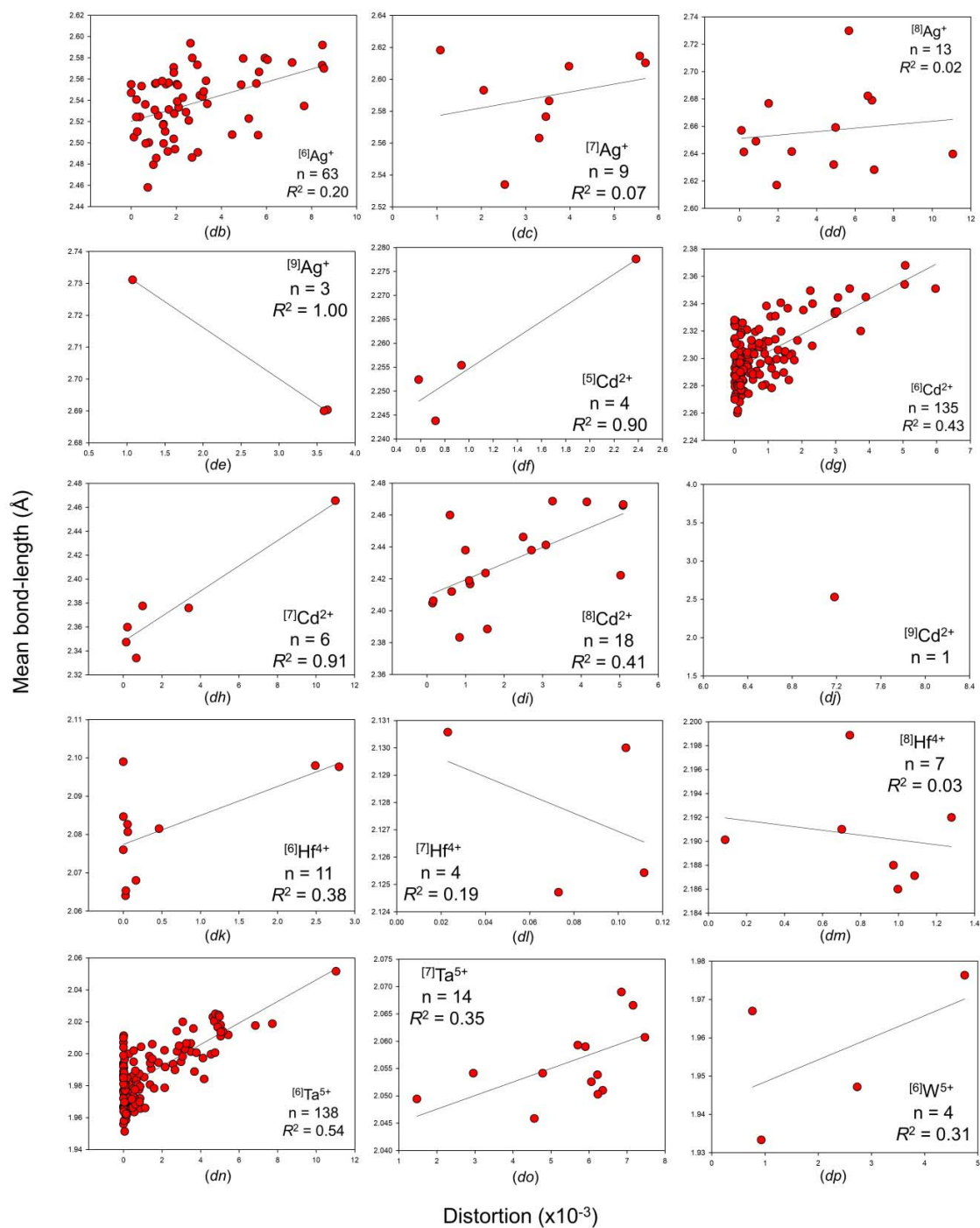


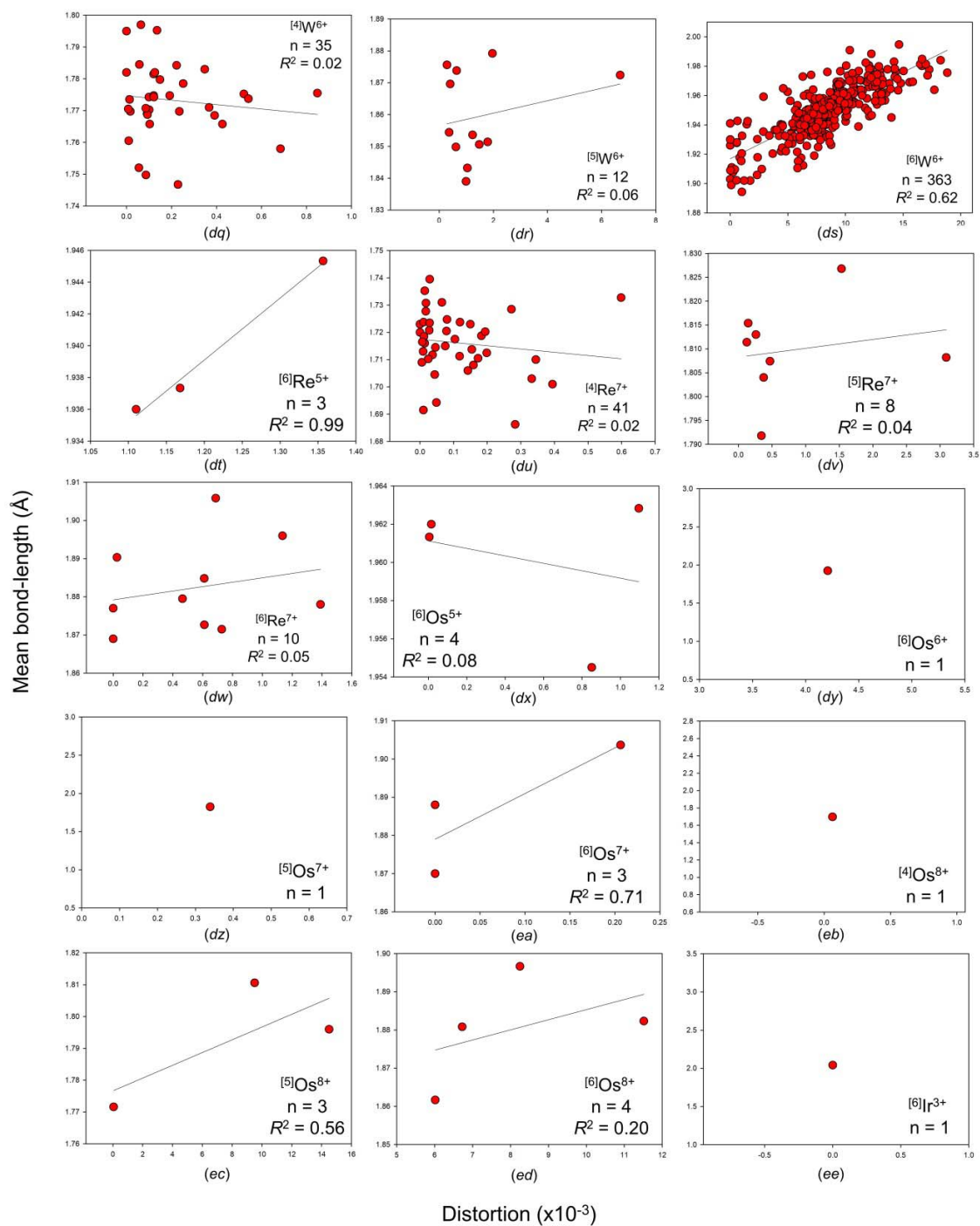












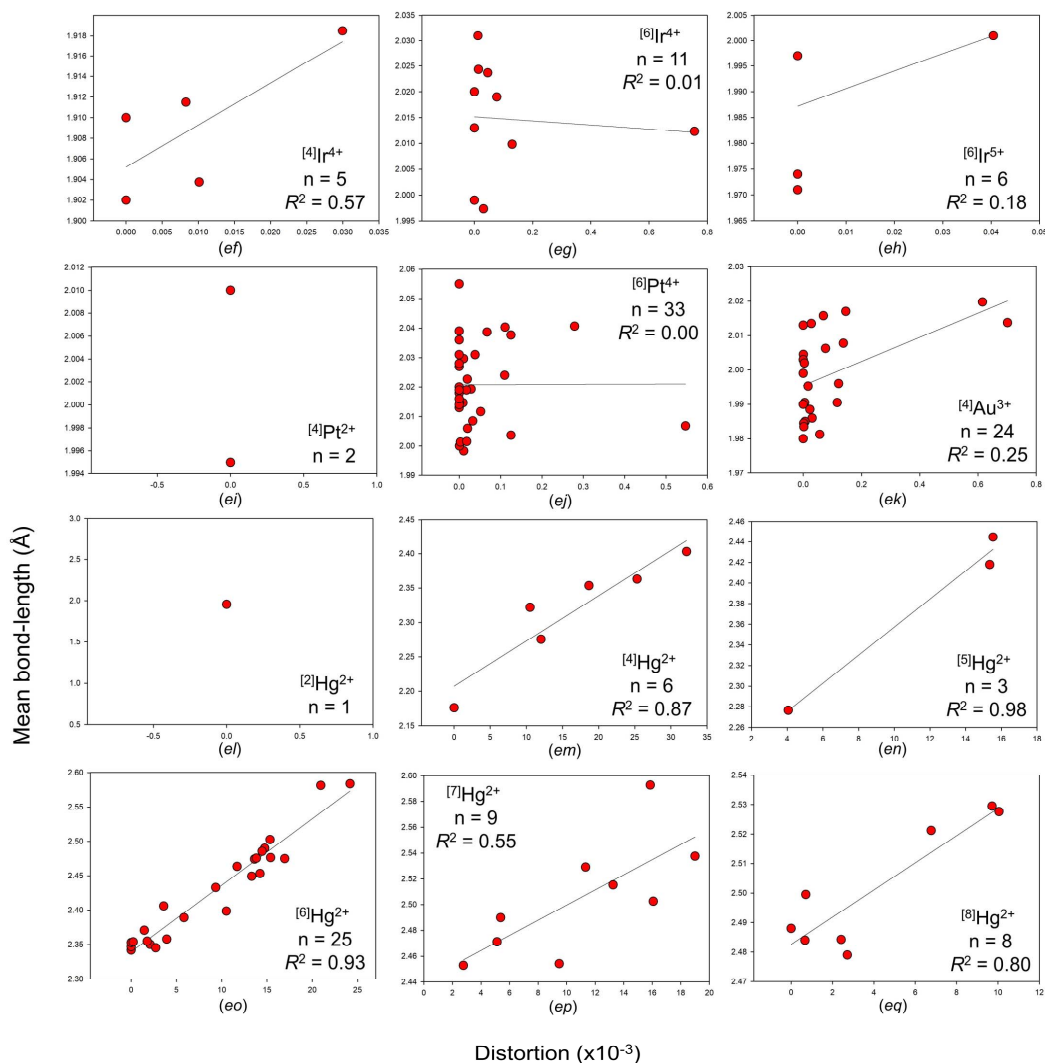


Figure S4 The effect of bond-length distortion on mean bond-length for all configurations of the transition metal ions bonded to O^{2-} : (a) $[6]Sc^{3+}$, (b) $[7]Sc^{3+}$, (c) $[8]Sc^{3+}$, (d) $[6]Ti^{3+}$, (e) $[7]Ti^{3+}$, (f) $[8]Ti^{3+}$, (g) $[4]Ti^{4+}$, (h) $[5]Ti^{4+}$, (i) $[6]Ti^{4+}$, (j) $[7]Ti^{4+}$, (k) $[6]V^{3+}$, (l) $[5]V^{4+}$, (m) $[6]V^{4+}$, (n) $[4]V^{5+}$, (o) $[5]V^{5+}$, (p) $[6]V^{5+}$, (q) $[4]Cr^{2+}$, (r) $[5]Cr^{2+}$, (s) $[6]Cr^{2+}$, (t) $[6]Cr^{3+}$, (u) $[4]Cr^{4+}$, (v) $[6]Cr^{4+}$, (w) $[4]Cr^{5+}$, (x) $[4]Cr^{6+}$, (y) $[4]Mn^{2+}$, (z) $[5]Mn^{2+}$, (aa) $[6]Mn^{2+}$, (ab) $[7]Mn^{2+}$, (ac) $[8]Mn^{2+}$, (ad) $[4]Mn^{3+}$, (ae) $[5]Mn^{3+}$, (af) $[6]Mn^{3+}$, (ag) $[4]Mn^{4+}$, (ah) $[6]Mn^{4+}$, (ai) $[4]Mn^{5+}$, (aj) $[4]Mn^{6+}$, (ak) $[4]Mn^{7+}$, (al) $[3]Fe^{2+}$, (am) $[4]Fe^{2+}$, (an) $[5]Fe^{2+}$, (ao) $[6]Fe^{2+}$, (ap) $[8]Fe^{2+}$, (aq) $[4]Fe^{3+}$, (ar) $[5]Fe^{3+}$, (as) $[6]Fe^{3+}$, (at) $[8]Fe^{3+}$, (au) $[3]Co^{2+}$, (av) $[4]Co^{2+}$, (aw) $[5]Co^{2+}$, (ax) $[6]Co^{2+}$, (ay) $[8]Co^{2+}$, (az) $[6]Co^{3+}$, (ba) $[6]Co^{4+}$, (bb) $[2]Ni^{2+}$, (bc) $[4]Ni^{2+}$, (bd) $[5]Ni^{2+}$, (be) $[6]Ni^{2+}$, (bf) $[6]Ni^{4+}$, (bg) $[2]Cu^{+}$, (bh) $[3]Cu^{+}$, (bi) $[4]Cu^{+}$, (bj) $[4]Cu^{2+}$, (bk) $[5]Cu^{2+}$, (bl) $[6]Cu^{2+}$, (bm) $[8]Cu^{2+}$, (bn) $[4]Cu^{3+}$, (bo) $[4]Zn^{2+}$, (bp) $[5]Zn^{2+}$, (bq) $[6]Zn^{2+}$, (br) $[6]Y^{3+}$, (bs) $[7]Y^{3+}$, (bt) $[8]Y^{3+}$, (bu) $[9]Y^{3+}$, (bv) $[10]Y^{3+}$, (bw) $[12]Y^{3+}$, (bx) $[6]Zr^{4+}$, (by) $[7]Zr^{4+}$,

(bz) $^{[8]}\text{Zr}^{4+}$, (ca) $^{[9]}\text{Zr}^{4+}$, (cb) $^{[10]}\text{Zr}^{4+}$, (cc) $^{[6]}\text{Nb}^{4+}$, (cd) $^{[4]}\text{Nb}^{5+}$, (ce) $^{[5]}\text{Nb}^{5+}$, (cf) $^{[6]}\text{Nb}^{5+}$, (cg) $^{[7]}\text{Nb}^{5+}$, (ch) $^{[8]}\text{Nb}^{5+}$, (ci) $^{[6]}\text{Mo}^{3+}$, (cj) $^{[6]}\text{Mo}^{4+}$, (ck) $^{[5]}\text{Mo}^{5+}$, (cl) $^{[6]}\text{Mo}^{5+}$, (cm) $^{[4]}\text{Mo}^{6+}$, (cn) $^{[5]}\text{Mo}^{6+}$, (co) $^{[6]}\text{Mo}^{6+}$, (cp) $^{[4]}\text{Tc}^{7+}$, (cq) $^{[6]}\text{Ru}^{3+}$, (cr) $^{[6]}\text{Ru}^{4+}$, (cs) $^{[6]}\text{Ru}^{5+}$, (ct) $^{[6]}\text{Rh}^{3+}$, (cu) $^{[6]}\text{Rh}^{4+}$, (cv) $^{[4]}\text{Pd}^{2+}$, (cw) $^{[6]}\text{Pd}^{4+}$, (cx) $^{[2]}\text{Ag}^+$, (cy) $^{[3]}\text{Ag}^+$, (cz) $^{[4]}\text{Ag}^+$, (da) $^{[5]}\text{Ag}^+$, (db) $^{[6]}\text{Ag}^+$, (dc) $^{[7]}\text{Ag}^+$, (dd) $^{[8]}\text{Ag}^+$, (de) $^{[9]}\text{Ag}^+$, (df) $^{[5]}\text{Cd}^{2+}$, (dg) $^{[6]}\text{Cd}^{2+}$, (dh) $^{[7]}\text{Cd}^{2+}$, (di) $^{[8]}\text{Cd}^{2+}$, (dj) $^{[9]}\text{Cd}^{2+}$, (dk) $^{[6]}\text{Hf}^{4+}$, (dl) $^{[7]}\text{Hf}^{4+}$, (dm) $^{[8]}\text{Hf}^{4+}$, (dn) $^{[6]}\text{Ta}^{5+}$, (do) $^{[7]}\text{Ta}^{5+}$, (dp) $^{[6]}\text{W}^{5+}$, (dq) $^{[4]}\text{W}^{6+}$, (dr) $^{[5]}\text{W}^{6+}$, (ds) $^{[6]}\text{W}^{6+}$, (dt) $^{[6]}\text{Re}^{5+}$, (du) $^{[4]}\text{Re}^{7+}$, (dv) $^{[5]}\text{Re}^{7+}$, (dw) $^{[6]}\text{Re}^{7+}$, (dx) $^{[6]}\text{Os}^{5+}$, (dy) $^{[6]}\text{Os}^{6+}$, (dz) $^{[5]}\text{Os}^{7+}$, (ea) $^{[6]}\text{Os}^{7+}$, (eb) $^{[4]}\text{Os}^{8+}$, (ec) $^{[5]}\text{Os}^{8+}$, (ed) $^{[6]}\text{Os}^{8+}$, (ee) $^{[6]}\text{Ir}^{3+}$, (ef) $^{[4]}\text{Ir}^{4+}$, (eg) $^{[6]}\text{Ir}^{4+}$, (eh) $^{[6]}\text{Ir}^{5+}$, (ei) $^{[4]}\text{Pt}^{2+}$, (ej) $^{[6]}\text{Pt}^{4+}$, (ek) $^{[4]}\text{Au}^{3+}$, (el) $^{[2]}\text{Hg}^{2+}$, (em) $^{[4]}\text{Hg}^{2+}$, (en) $^{[5]}\text{Hg}^{2+}$, (eo) $^{[6]}\text{Hg}^{2+}$, (ep) $^{[7]}\text{Hg}^{2+}$, (eq) $^{[8]}\text{Hg}^{2+}$.

Table S1 *A priori* bond valences for crystal-structure refinements used in this work

1292: $\text{Cu}^{2+}_3\text{O}_2(\text{PO}_4)_2$					
	<i>Cu1</i>	<i>Cu2</i>	<i>Cu3</i>	<i>P</i>	Σ
<i>O1</i>		$0.367 \times 2 \downarrow \times 2 \rightarrow$		1.265	2
<i>O2</i>		0.551		1.449	2
<i>O3</i>	$0.388 \times 2 \downarrow$		$0.277 \times 2 \downarrow \times 2 \rightarrow$	1.059	2
<i>O4</i>		0.329	0.444	1.227	2
<i>O5</i>	$0.612 \times 2 \downarrow$	0.385	$0.501 \times 2 \downarrow \times 2 \rightarrow$		2
Σ	2	2	2	5	

1640: $(\text{Hg}^{2+}(\text{H}_2\text{O})_6)(\text{Cl}^{1+}\text{O}_4)_2$				
	<i>Hg</i>	<i>Cl</i>	<i>H</i>	Σ
<i>O1</i>	$0.333 \times 6 \downarrow$		$0.833 \times 2 \rightarrow$	2
<i>O2</i>		$1.667 \times 3 \downarrow$	$0.167 \times 2 \rightarrow$	2
<i>O3</i>		2.000		2
Σ	2	7	1	

2279: $\text{Ba}_3\text{Cu}^{2+}(\text{Sb}^{5+}_2\text{O}_9)$							
Σ	<i>Ba1</i>	<i>Ba2</i>	<i>Ba3</i>	<i>Cu</i>	<i>Sb1</i>	<i>Sb2</i>	Σ
<i>O1</i>	$0.206 \times 3 \downarrow$	$0.175 \times 6 \downarrow \times 2 \rightarrow$	$0.152 \times 3 \downarrow$	$0.396 \times 3 \downarrow$	$0.896 \times 3 \downarrow$		2
<i>O2</i>	$0.080 \times 6 \downarrow \times 2 \rightarrow$	$0.149 \times 3 \downarrow$	$0.026 \times 3 \downarrow$	$0.270 \times 3 \downarrow$	$0.770 \times 3 \downarrow$	$0.724 \times 3 \downarrow$	2
<i>O3</i>	$0.300 \times 3 \downarrow$	$0.268 \times 3 \downarrow$	$0.245 \times 6 \downarrow \times 2 \rightarrow$			$0.943 \times 3 \downarrow$	2
Σ	2	2	2	2	5	5	

8269: Cr ³⁺ (W ⁵⁺ O ₄)			
	<i>Cr</i>	<i>W</i>	Σ
O1	0.542 ×4↓ ×2→	0.917 ×2↓	2
O2	0.417	0.792 ×2↓ ×2→	2
O3	0.417	0.792 ×2↓ ×2→	2
Σ	3	5	

14159: Na ₂ (Ni ²⁺ O ₂)				
	<i>Na1</i>	<i>Na2</i>	<i>Ni</i>	Σ
O1	0.219 ×2↓ ×2→	0.266 ×2↓ ×2→	0.516 ×2↓ ×2→	2
O2	0.188 ×3↓ ×3→	0.234 ×2↓ ×2→	0.484 ×2↓ ×2→	2
Σ	1	1	2	

15505: Y ₃ Re ⁷⁺ O ₈					
	<i>Y1</i>	<i>Y2</i>	<i>Y3</i>	<i>Re</i>	Σ
O1	0.495	0.486 ×2↓ ×2→	0.533		2
O2	0.495	0.486 ×2↓ ×2→	0.533		2
O3		0.254	0.302 ×2↓ ×2→	1.143	2
O4	0.282 ×2↓ ×2→	0.274		1.162	2
O5	0.361		0.399	1.240	2
O6	0.273	0.264	0.311	1.152	2
O7	0.273	0.264	0.311	1.152	2
O8	0.270 ×2↓ ×2→		0.309	1.150	2
Σ	3	3	3	7	

15545: Na ₄ Zr ₂ (SiO ₄) ₃					
	<i>Na1</i>	<i>Na2</i>	<i>Zr</i>	<i>Si</i>	Σ
<i>O1</i>	0.167 ×6↓	0.104 ×4↓ ×2→	0.646 ×3↓	0.979 ×2↓	2
<i>O2</i>		0.146 ×4↓ ×2→	0.688 ×3↓	1.021 ×2↓	2
Σ	1	1	4	4	

17062: Fe ₂₊₂ (P ₂ O ₇)					
	<i>Fe1</i>	<i>Fe2</i>	<i>P1</i>	<i>P2</i>	Σ
<i>O1</i>	0.333	0.333	1.333		2
<i>O2</i>	0.333	0.333	1.333		2
<i>O3</i>	0.333	0.333	1.333		2
<i>O4</i>			1.000	1.000	2
<i>O5</i>	0.333	0.333		1.333	2
<i>O6</i>	0.333	0.333		1.333	2
<i>O7</i>	0.333	0.333		1.333	2
Σ	2	2	5	5	

20540: Li ₂ [Os ⁸⁺ O ₄ (OH) ₂]				
	<i>Li1</i>	<i>Li2</i>	<i>Os</i>	Σ
<i>O1</i>	0.217 ×2↓	0.400 ×2↓	1.383 ×2↓	2
<i>O2</i>	0.417 ×2↓		1.583 ×2↓	2
<i>O3</i>	-0.133 ×2↓	0.050 ×4↓ ×2→	1.033 ×2↓	2
Σ	1	1	8	

20611: Rb(Os ⁸⁺ ₂ O ₈ (OH))				
	<i>Rb</i>	<i>Os1</i>	<i>Os2</i>	Σ
<i>O1</i>	0.000 ×2↓ ×2→		2.000	2
<i>O2</i>	0.000 ×2↓ ×2→		2.000	2
<i>O3</i>	0.176 ×2↓ ×2→	1.647		2
<i>O4</i>	0.265	1.735		2
<i>O5</i>	0.000		2.000	2
<i>O6</i>	0.176 ×2↓ ×2→	1.647		2
<i>O7</i>	0.000 ×2↓ ×2→		2.000	2
<i>O8</i>	0.265	1.735		2
<i>OH</i>	-0.235	1.235		1
Σ	1	8	8	

20670: YCo ²⁺ (BO ₂) ₅								
	<i>Co</i>	<i>Y</i>	<i>B1</i>	<i>B2</i>	<i>B3</i>	<i>B4</i>	<i>B5</i>	Σ
<i>O1</i>	0.300	0.246	0.715	0.738				2
<i>O2</i>	0.323		0.738			0.939		2
<i>O3</i>		0.273	0.742				0.985	2
<i>O4</i>	0.390		0.805		0.806			2
<i>O5</i>	0.300	0.246		0.738	0.716			2
<i>O6</i>		0.259 ×2↓ ×2→		0.752	0.729			2
<i>O7</i>		0.280		0.772		0.949		2
<i>O8</i>		0.259			0.749		0.992	2
<i>O9</i>	0.344 ×2↓ ×2→	0.290					1.023	2
<i>O10</i>		0.444 ×2↓ ×2→				1.113		2
Σ	2	3	3	3	3	3	3	

24819: Na ₅ Nb ⁵⁺ O ₅					
	<i>Na1</i>	<i>Na2</i>	<i>Na3</i>	<i>Nb</i>	Σ
<i>O1</i>	0.181 ×2↓	0.181 ×2↓ ×2→	0.238 ×2↓ ×2→	0.981 ×2↓	2
<i>O2</i>	0.229 ×2↓	0.229 ×2↓ ×2→	0.286	1.029 ×2↓	2
<i>O3</i>	0.181	0.181 ×2→	0.238 ×2→	0.981	2
Σ	1	1	1	5	

24973: Ca ₄ Mn ³⁺ ₃ B ₃ O ₁₂ CO ₃						
	<i>Ca1</i>	<i>Ca2</i>	<i>Mn</i>	<i>B</i>	<i>C</i>	Σ
<i>O1</i>	0.177 ×3↓	0.318	0.480	1.025		2
<i>O2</i>	0.177 ×3↓	0.318	0.480	1.025		2
<i>O3</i>		0.242	0.404 ×2↓ ×2→	0.949		2
<i>O4</i>	0.313 ×3↓	0.455	0.616 ×2↓ ×2→			2
<i>O5</i>		0.222 ×3↓ ×3→			1.333 ×3↓	2
Σ	2	2	3	3	4	

33194: Pb ²⁺ HfO ₃					
	<i>Pb1</i>	<i>Pb2</i>	<i>Hf1</i>	<i>Hf2</i>	Σ
<i>O1</i>	0.174 ×4↓ ×4→		0.645	0.661	2
<i>O2</i>		0.166 ×4↓ ×4→	0.661	0.676	2
<i>O3</i>	0.221	0.197 ×2↓ ×2→	0.692 ×2↓ ×2→		2
<i>O4</i>	0.179 ×2↓ ×2→	0.155 ×2↓ ×2→		0.666 ×2↓ ×2→	2
<i>O5</i>	0.184 ×2→	0.161 ×2→	0.655 ×2→		2
<i>O6</i>	0.179 ×2→	0.155 ×2→		0.666 ×2→	2
<i>O7</i>	0.184 ×2→	0.161 ×2→	0.655 ×2→		2
<i>O8</i>	0.179 ×2→	0.155 ×2→		0.666 ×2→	2
Σ	2	2	4	4	

33783: La(Nb ⁵⁺ ₅ O ₁₄)					
	<i>La</i>	<i>Nb1</i>	<i>Nb2</i>	<i>Nb3</i>	Σ
<i>O1</i>	0.211 ×2↓		0.894 ×2↓ ×2→		2
<i>O2</i>	0.210 ×4↓ ×2→	0.688	0.893		2
<i>O3</i>	0.348 ×2↓	0.826 ×2↓ ×2→			2
<i>O4</i>		0.591	0.796	0.614 ×2↓	2
<i>O5</i>		0.591	0.796	0.614 ×2↓	2
<i>O6</i>	0.261 ×4↓ ×2→	0.739 ×2↓ ×2→			2
<i>O7</i>				1.000 ×2↓ ×2→	2
<i>O8</i>			0.727 ×2→	0.545	2
Σ	3	5	5	5	

33800: La ₃ Ti ⁴⁺ O ₄ Cl ₅					
	<i>La1</i>	<i>La2</i>	<i>La3</i>	<i>Ti</i>	Σ
<i>O1</i>			0.542 ×2↓ ×2→	0.915	2
<i>O2</i>		0.540 ×2↓ ×2→		0.920	2
<i>O3</i>	0.253	0.245	0.252	0.625 ×2↓ ×2→	2
<i>O4</i>	0.543 ×2↓ ×2→			0.915	2
<i>Cl1</i>	0.253	0.244	0.252 ×2↓ ×2→		1
<i>Cl2</i>	0.254	0.246 ×2↓ ×2→	0.253		1
<i>Cl3</i>		0.248 ×3↓ ×3→	0.255		1
<i>Cl4</i>	0.250 ×3↓ ×3→		0.249		1
<i>Cl5</i>	0.202 ×2↓ ×2→	0.194	0.201 ×2↓ ×2→		1
Σ	3	3	3	4	

33802: Sr ₂ (Ru ⁴⁺ O ₄)			
	<i>Sr</i>	<i>Ru</i>	Σ
<i>O1</i>	0.179 ×4↓ ×4→	0.641 ×4↓ ×2→	2
<i>O2</i>	0.256 ×5↓ ×5→	0.718 ×2↓	2
Σ	2	4	

33863: Ba ₄ Ir ⁴⁺ ₃ O ₁₀					
	<i>Ba1</i>	<i>Ba2</i>	<i>Ir1</i>	<i>Ir2</i>	Σ
<i>O1</i>	0.182 ×3↓ ×3→	0.153	0.663 ×2↓	0.638	2
<i>O2</i>	0.187 ×2↓ ×2→	0.158 ×2↓ ×2→	0.668 ×2↓	0.643	2
<i>O3</i>	0.187 ×2↓ ×2→	0.158 ×2↓ ×2→	0.668 ×2↓	0.643	2
<i>O4</i>	0.326	0.297 ×3↓ ×3→		0.782	2
<i>O5</i>	0.191 ×2↓ ×2→	0.162 ×2↓ ×2→		0.647 ×2↓ ×2→	2
Σ	2	2	4	4	

34392: K ₆ (Mn ³⁺ ₂ O ₆)					
	<i>K1</i>	<i>K2</i>	<i>K3</i>	<i>Mn</i>	Σ
<i>O1</i>	0.135	0.135	0.155 ×2↓ ×2→	0.710 ×2↓ ×2→	2
<i>O2</i>	0.234 ×2↓ ×2→	0.234 ×2↓ ×2→	0.254	0.809	2
<i>O3</i>	0.198 ×2↓ ×2→	0.198 ×2↓ ×2→	0.218 ×2↓ ×2→	0.772	2
Σ	1	1	1	3	

35084: Cd ₃ Te ⁶⁺ O ₆				
	<i>Cd1</i>	<i>Cd2</i>	<i>Te</i>	Σ
<i>O1</i>	0.373 ×2↓	0.294 ×2↓ ×2→	1.039 ×2↓	2
<i>O2</i>	0.314 ×2↓	0.235 ×3↓ ×3→	0.980 ×2↓	2
<i>O3</i>	0.314 ×2↓	0.235 ×3↓ ×3→	0.980 ×2↓	2
Σ	2	2	6	

35407: CdPt ⁴⁺ ₃ O ₆				
	<i>Cd</i>	<i>Pt1</i>	<i>Pt2</i>	Σ
<i>O1</i>	0.286 ×4↓ ×2→		0.714 ×2↓ ×2→	2
<i>O2</i>	0.214 ×4↓	0.500 ×4↓	0.643 ×4↓ ×2→	2
Σ	2	2	4	

36608: Sm ³⁺ (Ti ⁴⁺ O ₃ Cl)			
	<i>Sm</i>	<i>Ti</i>	Σ
<i>O1</i>	0.320	0.560 ×3↓ ×3→	2
<i>O2</i>	0.587 ×2↓ ×2→	0.827	2
<i>O3</i>	0.507	0.747 ×2↓ ×2→	2
<i>Cl</i>	0.250 ×4↓ ×4→		1
Σ	3	4	

36626: Nb ⁵⁺ (PO ₄)O			
	<i>Nb</i>	<i>P</i>	Σ
O1	0.750	1.250	2
O2	1.000 ×2↓ ×2→		2
O3	0.750	1.250	2
O4	0.750	1.250	2
O5	0.750	1.250	2
Σ	5	5	

40249: KNa ₃ (W ⁶⁺ O ₅)					
	<i>Na1</i>	<i>Na2</i>	<i>K</i>	<i>W</i>	Σ
O1	0.192	0.197 ×2→	0.111 ×2↓ ×2→	1.192	2
O2	0.197 ×2↓	0.202 ×3↓ ×3→		1.197 ×2↓	2
O3	0.192	0.197 ×2→	0.111 ×2↓ ×2→	1.192	2
O4	0.221		0.139 ×4↓ ×4→	1.221	2
Σ	1	1	1	6	

40312: SrZn(V ⁵⁺ 2O ₇)					
	<i>Sr</i>	<i>Zn</i>	<i>V1</i>	<i>V2</i>	Σ
O1	0.260	0.414	1.326		2
O2	0.260	0.414	1.326		2
O3	0.260	0.414	1.326		2
O4	-0.046		1.021	1.025	2
O5	0.310 ×2↓ ×2→			1.381	2
O6	0.259	0.412		1.329	2
O7	0.194 ×2↓ ×2→	0.347		1.265	2
Σ	2	2	5	5	

40850: Mn ²⁺ (V ⁵⁺ ₂ O ₆)			
	<i>Mn</i>	<i>V</i>	Σ
O1	0.417 ×4↓ ×2→	1.167	2
O2	0.167 ×2↓	0.917 ×2↓ ×2→	2
O3		0.667 ×3↓ ×3→	2
Σ	2	5	

49746: Na ₅ (Os ⁷⁺ O ₆)					
	<i>Na1</i>	<i>Na2</i>	<i>Na3</i>	<i>Os</i>	Σ
O1	0.167 ×4↓ ×2→	0.167 ×4↓	0.167 ×4↓ ×2→	1.167 ×4↓	2
O2	0.167 ×2↓ ×2→	0.167 ×2↓	0.167 ×2↓ ×2→	1.167 ×2↓	2
Σ	1	1	1	7	

50010: Cs(V ⁵⁺ ₃ O ₈)				
	<i>Cs</i>	<i>V1</i>	<i>V2</i>	Σ
O1	0.500	1.500		2
O2	-0.375	0.625	0.875 ×2→	2
O3		0.500 ×2↓	0.750 ×2↓ ×2→	2
O4	0.188 ×6↓ ×3		1.438	2
O5	-0.062 ×4↓ ×2→	0.938 ×2↓	1.188	2
Σ	1	5	5	

50038: $K_2Fe^{2+}_2Nb^{5+}_4O_{13}$					
	<i>K</i>	<i>Fe</i>	<i>Nb1</i>	<i>Nb2</i>	Σ
01	-0.083 $\times 2\downarrow$	0.259 $\times 2\downarrow \times 2\rightarrow$	0.693	0.872	2
02	0.176 $\times 2\downarrow \times 2\rightarrow$	0.518		1.130	2
03	0.090 $\times 2\downarrow$		0.866	1.044	2
04	0.221 $\times 2\downarrow \times 2\rightarrow$	0.562	0.996		2
05		0.039		0.651 $\times 3\downarrow \times 3\rightarrow$	2
06	0.075 $\times 2\downarrow \times 4\rightarrow$		0.851 $\times 2\rightarrow$		2
07	0.021 $\times 2\downarrow \times 2\rightarrow$	0.363	0.797 $\times 2\downarrow \times 2\rightarrow$		2
Σ	1	2	5	5	

50459: $Cu^{2+}_4O(PO_4)_2$						
	<i>Cu1</i>	<i>Cu2</i>	<i>Cu3</i>	<i>P1</i>	<i>P2</i>	Σ
01	0.551			1.449		2
02		0.293	0.295 $\times 2\rightarrow$	1.118		2
03		0.391 $\times 2\downarrow$	0.393	1.216 $\times 2\downarrow$		2
04	0.335	0.408			1.257	2
05			0.384 $\times 2\rightarrow$		1.231	2
06	0.335 $\times 2\downarrow$		0.409		1.256 $\times 2\downarrow$	2
07	0.445	0.517	0.519 $\times 2\rightarrow$			2
Σ	2	2	2	5	5	

55272: Ca ₂ ZrB(Al ₉ O ₁₈)						
	<i>Ca</i>	<i>Al1</i>	<i>Al2</i>	<i>Zr</i>	<i>B</i>	Σ
<i>O1</i>		0.444 × 2↓ × 2→	0.444 × 2↓	0.667 × 3↓		2
<i>O2</i>	0.500 × 4↓	0.667 × 2↓ × 2→	0.667 × 2↓			2
<i>O3</i>		0.444 × 2→	0.444	0.667 × 3↓		2
<i>O4</i>		0.333 × 2→	0.333		1.000 × 3↓	2
Σ	2	3	3	4	3	

59111: HfSiO ₄			
	<i>Hf</i>	<i>Si</i>	Σ
<i>O</i>	0.500 × 8↓ × 2→	1.000 × 4↓	2
Σ	4	4	

59244: LiV ³⁺ (Si ₂ O ₆)				
	<i>Li</i>	<i>V</i>	<i>Si</i>	Σ
<i>O1</i>	0.125 × 2↓	0.450 × 4↓ × 2→	0.975	2
<i>O2</i>	0.275 × 2↓	0.600 × 2↓	1.125	2
<i>O3</i>	0.100 × 2↓		0.950 × 2↓ × 2→	2
Σ	1	3	4	

59819: Ce ⁴⁺ ₂ (Cr ⁶⁺ O ₄) ₄ (H ₂ O) ₂							
	<i>Ce1</i>	<i>Ce2</i>	<i>Cr1</i>	<i>Cr2</i>	<i>Cr3</i>	<i>Cr4</i>	Σ
01			2.000				2
02		0.632	1.368				2
03		0.632	1.368				2
04	0.737		1.263				2
05		0.474		1.526			2
06		0.474		1.526			2
07		0.474		1.526			2
08	0.579			1.421			2
09	0.553				1.447		2
010		0.447			1.553		2
011	0.553				1.447		2
012		0.447			1.553		2
013		0.421				1.579	2
014	0.526					1.474	2
015	0.526					1.474	2
016	0.526					1.474	2
Σ	4	4	6	6	6	6	

62577: CaNb ⁵⁺ ₂ (P ₄ O ₁₃)(P ₂ O ₇)O								
	Nb1	Nb2	Ca	P1	P2	P3	P4	Σ
O1			0.267 ×2↓	1.733				2
O2		0.782 ×2↓		1.218				2
O3				1 ×2→				2
O4				1.049	0.951			2
O5	0.805 ×2↓				1.195			2
O6			0.316 ×2↓		1.684			2
O7		0.831 ×2↓			1.169			2
O8	0.837 ×2↓					1.163 ×2↓		2
O9			0.348			1.652		2
O10						1.169	0.978	2
O11	0.859 ×2↓						1.141 ×2↓	2
O12		0.261					1.739	2
O13		1.515	0.458					2
Σ	5	5	2	5	5	5	5	

63103: Y ₂ Ba ₂ Cu ²⁺ Pt ⁴⁺ O ₈							
	Ba1	Ba2	Cu	Y1	Y2	Pt	Σ
O1	0.232	0.206 ×3↓ ×3→	0.440			0.711	2
O2	0.177 ×2↓	0.151 ×2↓	0.385 ×2↓	0.430 ×4↓ ×2→	0.427 ×2↓		2
O3	0.174 ×4↓ ×2→	0.148 ×2↓		0.427 ×2↓	0.424 ×2↓	0.653 ×2↓	2
O4	0.174 ×2↓ ×2→	0.148		0.427	0.424	0.653	2
O5	0.186 ×2↓	0.160 ×4↓ ×2→	0.394 ×2↓		0.436 ×2↓	0.665 ×2↓	2
Σ	2	2	2	3	3	4	

64634: V ³⁺ ₂ (V ⁴⁺ O(P ₂ O ₇) ₂) ₂					
	V1	V2	P1	P2	Σ
01		0.583		1.417	2
02	0.500 ×2↓	0.333	1.167		2
03	0.500 ×2↓	0.333		1.167	2
04		0.583	1.417		2
05		0.583		1.417	2
06		0.583	1.417		2
07			1.000 ×2→		2
08	2.000				2
09				1.000 ×2→	2
Σ	4	3	5	5	

65237: Na ₆ (O ₂ (Cu ³⁺ O ₂) ₂)										
	<i>Na1</i>	<i>Na2</i>	<i>Na3</i>	<i>Na4</i>	<i>Na5</i>	<i>Na6</i>	<i>Cu1</i>	<i>Cu2</i>	<i>Cu3</i>	Σ
<i>O1</i>	0.167 ×2↓	0.250 ×2↓	0.167 ×2↓	0.194	0.194	0.250	0.778 ×2↓			2
<i>O2</i>				0.139 ×2↓ ×2→	0.139 ×2↓ ×2→		0.722 ×2↓	0.722 ×2↓		2
<i>O3</i>	0.167 ×2↓		0.167 ×2↓	0.194	0.194	0.250 ×2↓ ×2→		0.778 ×2↓		2
<i>O4</i>				0.139 ×2→	0.139 ×2→				0.722 ×2↓ ×2→	2
<i>O5</i>	0.167 ×2↓	0.250 ×2↓	0.167 ×2↓	0.194	0.194	0.250			0.778 ×2↓	2
Σ	1	1	1	1	1	1	3	3	3	

65407: Na ₃ Sc ₂ (PO ₄) ₃					
	<i>Na1</i>	<i>Na2</i>	<i>Sc</i>	<i>P</i>	Σ
<i>O1</i>		0.106 ×4↓ ×2→	0.519 ×3↓	1.269 ×2↓	2
<i>O2</i>	0.153 ×6↓	0.067 ×4↓ ×2→	0.481 ×3↓	1.231 ×2↓	2
Σ	0.92	0.693	3	5	

65476: Ni ²⁺ ₅ HfB ₂ O ₁₀								
	<i>Ni1</i>	<i>Ni2</i>	<i>Ni3</i>	<i>Ni4</i>	<i>Hf</i>	<i>B1</i>	<i>B2</i>	Σ
<i>O1</i>	0.392 ×2↓		0.425 ×2↓	0.427	0.756 ×2↓			2
<i>O2</i>	0.308 ×2↓	0.339 ×4↓ ×2→		0.343	0.672 ×2↓			2
<i>O3</i>		0.322		0.326 ×2→		1.025		2
<i>O4</i>		0.322		0.326 ×2→			1.025	2
<i>O5</i>	0.299			0.334 ×2→		1.033		2
<i>O6</i>				0.243 ×2→	0.572		0.942	2
<i>O7</i>			0.242 ×2↓ ×2→		0.573	0.943		2
<i>O8</i>	0.300		0.333 ×2↓ ×2→				1.034	2
Σ	2	2	2	2	4	3	3	

65512: Zr(Mo ⁶⁺ O ₄) ₂				
	<i>Zr1</i>	<i>Zr2</i>	<i>Mo</i>	Σ
<i>O1</i>			2.000	2
<i>O2</i>		0.667 ×3↓	1.333	2
<i>O3</i>		0.667 ×3↓	1.333	2
<i>O4</i>	0.667 ×6↓		1.333	2
Σ	4	4	6	

65614: $Y_2Ba_2Cu^{2+}Pt^{4+}O_8$						
	<i>Ba1</i>	<i>Ba2</i>	<i>Cu</i>	<i>Y</i>	<i>Pt</i>	Σ
<i>O1</i>	0.228 $\times 4\downarrow$	0.163 $\times 4\downarrow \times 2\rightarrow$	0.380 $\times 2\downarrow$	0.402 $\times 2\downarrow$	0.663 $\times 4\downarrow$	2
<i>O2</i>		0.174 $\times 3\downarrow \times 3\rightarrow$	0.391	0.413	0.674 $\times 2\downarrow$	2
<i>O3</i>	0.272 $\times 4\downarrow$	0.207 $\times 4\downarrow \times 2\rightarrow$	0.424 $\times 2\downarrow$	0.446 $\times 4\downarrow \times 2\rightarrow$		2
Σ	2	2	2	3	4	

66994: $Cs(Np^{5+}O_2)(Mo^{6+}O_4)$				
	<i>Cs</i>	<i>Np</i>	<i>Mo</i>	Σ
<i>O1</i>	0.384 $\times 2\downarrow \times 2\rightarrow$	1.232		2
<i>O2</i>	0.288 $\times 3\downarrow \times 3\rightarrow$	1.137		2
<i>O3</i>	-0.446	0.402	1.022 $\times 2\downarrow \times 2\rightarrow$	2
<i>O4</i>	-0.292	0.557 $\times 2\downarrow \times 2\rightarrow$	1.177	2
<i>O5</i>	-0.292	0.557 $\times 2\downarrow \times 2\rightarrow$	1.177	2
<i>O6</i>	0.133 $\times 3\downarrow \times 3\rightarrow$		1.601	2
Σ	1	5	6	

67726: $Cu^{2+}_3Ba(V^{5+}O_4)_2(OH)_2$						
	<i>Ba</i>	<i>Cu1</i>	<i>Cu2</i>	<i>V</i>	<i>H</i>	Σ
<i>O1</i>		0.459 $\times 2\downarrow$	0.459 $\times 2\downarrow \times 2\rightarrow$		0.623	2
<i>O2</i>	0.195 $\times 4\downarrow$	0.270 $\times 4\downarrow$	0.270 $\times 2\downarrow$	1.264 $\times 2\downarrow$		2
<i>O3</i>	0.195 $\times 2\downarrow$		0.270 $\times 2\downarrow \times 2\rightarrow$	1.264		2
<i>O4</i>	0.138 $\times 6\downarrow \times 3\rightarrow$			1.208	0.377	2
Σ	2	2	2	5	1	

68279: Pr ³⁺ ₂ (Mo ⁶⁺ ₄ O ₁₅)							
	<i>Pr</i> 1	<i>Pr</i> 2	<i>Mo</i> 1	<i>Mo</i> 2	<i>Mo</i> 3	<i>Mo</i> 4	Σ
01		0.433	1.567				2
02		0.433	1.567				2
03	0.507		1.493				2
04	0.388	0.239	1.373				2
05	0.490			1.510			2
06	0.490			1.510			2
07		0.416		1.584			2
08	0.376	0.228		1.396			2
09					1.413	0.587	2
010		0.333			1.667		2
011	0.321	0.172			1.507		2
012					1.413	0.587	2
013						2.000	2
014		0.746				1.254	2
015	0.427					0.786 $\times 2 \downarrow \times 2 \rightarrow$	2
Σ	3	3	6	6	6	6	

68614: Cu ⁺ La(W ⁶⁺ ₂ O ₈)					
	<i>Cu</i>	<i>La</i>	<i>W1</i>	<i>W2</i>	Σ
01		0.148	0.959	0.892	2
02	-0.136		0.734 ×2↓ ×2→	0.668	2
03		0.148	0.959	0.892	2
04		0.595	1.405		2
05		0.396 ×2↓ ×2→	1.207		2
06	0.379	0.439		1.183	2
07	0.379	0.439		1.183	2
08	0.379	0.439		1.183	2
Σ	1	3	6	6	

69088: Ba(Mo ⁵⁺ ₂ P ₄ O ₁₆)					
	<i>Ba</i>	<i>Mo</i>	<i>P1</i>	<i>P2</i>	Σ
01		2.000			2
02	0.124 ×2↓	0.577	1.299		2
03	0.150 ×2↓	0.604		1.246	2
04		0.639	1.361		2
05	0.150 ×2↓	0.604		1.246	2
06	0.124 ×2↓	0.577	1.299		2
07			1.040	0.960	2
08	0.452 ×2↓			1.548	2
Σ	2	5	5	5	

71450: Sr ₂ (V ⁴⁺ O ₄)			
	<i>Sr</i>	<i>V</i>	Σ
<i>O1</i>	0.179 ×4↓ ×4→	0.641 ×4↓ ×2→	2
<i>O2</i>	0.256 ×5↓ ×5→	0.718 ×2↓	2
Σ	2	4	

71562: Ca ₂ Y(As ⁵⁺ O ₄)(W ⁶⁺ O ₄) ₂					
	<i>Ca</i>	<i>Y</i>	<i>W</i>	<i>As</i>	Σ
<i>O1</i>	0.240 ×4↓ ×2→		1.519 ×2↓		2
<i>O2</i>	0.202 ×2↓	0.317 ×4↓	1.481 ×2↓		2
<i>O3</i>	0.317 ×2↓	0.433 ×4↓		1.250 ×4↓	2
Σ	2	3	6	5	

72302: Cr ²⁺ ₃ Cr ³⁺ ₄ (PO ₄) ₆									
	Cr1	Cr2	Cr3	Cr4	P1	P2	P3	P4	Σ
01				0.509			1.491		2
02					1.020		0.980		2
03						1.000		1.000	2
04	0.459 ×2↓			0.249		1.293			2
05		0.425 ×2↓	0.383		1.192				2
06	0.583 ×2↓					1.417			2
07	0.459 ×2↓			0.249				1.293	2
08			0.467	0.255	1.277				2
09		0.438 ×2↓	0.396				1.166		2
010		0.636 ×2↓					1.364		2
011			0.710			1.290			2
012			0.584					1.416	2
013			0.460	0.248				1.292	2
014				0.489	1.511				2
Σ	3	3	3	2	5	5	5	5	

72312: Ba ₂ (Hg ₃ Pd ⁴⁺ ₅ Pd ²⁺ ₂ O ₁₄)								
	Ba	Hg1	Hg2	Pd1	Pd2	Pd3	Pd4	Σ
01	0.250			0.750		0.500 ×2↓ ×2→		2
02	-0.021 ×2↓ ×2→	0.563	1.000 ×2↓	0.479				2
03	0.378 ×2↓ ×2→						0.622 ×2↓ ×2→	2
04	0.250 ×2↓			0.750 ×2↓	0.500 ×4↓	0.500 ×2↓		2
05	0.134 ×4↓ ×2→	0.718 ×2↓		0.635 ×2↓			0.378 ×2↓	2
Σ	2	2	2	4	2	2	2	

72682: Cs ₂ (Ti ⁴⁺ O)(P ₂ O ₇)						
	Cs1	Cs2	Ti	P1	P2	Σ
01	0.009	0.017	0.739	1.235		2
02	0.191 ×2↓ ×2→	0.199		1.418		2
03	0.009	0.017	0.739	1.235		2
04	-0.115	-0.107		1.111	1.111	2
05	0.191 ×2↓ ×2→	0.199			1.418	2
06	0.314	0.321 ×2↓ ×2→	1.044			2
07	0.009	0.017	0.739		1.235	2
08	0.009	0.017	0.739		1.235	2
Σ	1	1	4	5	5	

72714: Ti ³⁺ PO ₄							
	Ti1	Ti2	P1	P2	P3	P4	Σ
01	0.429 ×2→			1.143			2
02	0.429 ×2→		1.143				2
03	0.429 ×2→					1.143	2
04	0.429 ×2→					1.143	2
05	0.643					1.357 ×2↓	2
06	0.643		1.357 ×2↓				2
07		0.643		1.357 ×2↓			2
08		0.643			1.357 ×2↓		2
09		0.429 ×2→			1.143		2
010		0.429 ×2→			1.143		2
011		0.429 ×2→		1.143			2
012		0.429 ×2→	1.143				2
Σ	3	3	5	5	5	5	

72872: Co ³⁺ Re ⁵⁺ O ₄			
	<i>Co</i>	<i>Re</i>	Σ
O1	0.417 ×2↓	0.792 ×4↓ ×2→	2
O2	0.542 ×4↓ ×2→	0.917 ×2↓	2
Σ	3	5	

72886: Ca(V ⁴⁺ O) ₂ (PO ₄) ₂				
	<i>Ca</i>	<i>V</i>	<i>P</i>	Σ
O1	0.250 ×2↓	0.500	1.250	2
O2	0.250 ×2↓	0.500	1.250	2
O3		1.000 ×2↓ ×2→		2
O4	0.250 ×2↓	0.500	1.250	2
O5	0.250 ×2↓	0.500	1.250	2
Σ	2	4	5	

73183: Ba ₃ Ca(Ru ⁵⁺ ₂ O ₉)					
	<i>Ba1</i>	<i>Ba2</i>	<i>Ca</i>	<i>Ru</i>	Σ
O1	0.119 ×6↓ ×2→	0.095 ×3↓ ×2→		0.786 ×3↓ ×2→	2
O2	0.214 ×6↓	0.190 ×9↓ ×3→	0.333 ×6↓	0.881 ×3↓	2
Σ	2	2	2	5	

73261: Cr ²⁺ ₃ Cr ³⁺ ₄ (PO ₄) ₆								
	Cr1	Cr2	Cr3	Cr4	P1	P2	P3	Σ
01			0.415 ×2↓ ×2→				1.170	2
02			0.460	0.358		1.182		2
03			0.443	0.341			1.216	2
04	0.511			0.409	1.080			2
05	0.460			0.358		1.182		2
06	0.639					1.361		2
07	0.409 ×2↓ ×2→						1.182	2
08		0.568 ×2↓					1.432	2
09		0.174 ×2↓	0.552			1.274		2
010			0.715		1.285			2
011	0.572				1.428			2
012		0.259 ×2↓		0.535	1.207			2
Σ	3	2	3	2	5	5	5	

73686: Th(V ⁵⁺ 2O ₇)							
	<i>Th1</i>	<i>Th2</i>	<i>V1</i>	<i>V2</i>	<i>V3</i>	<i>V4</i>	Σ
01	0.425 × 2↓ × 2→		1.151				2
02	0.637		1.363				2
03		0.665	1.335				2
04	0.425 × 2↓ × 2→		1.151				2
05		0.649		1.351			2
06	0.621			1.379			2
07		0.433 × 2↓ × 2→		1.135			2
08		0.433 × 2↓ × 2→		1.135			2
09	0.291 × 2↓ × 2→				1.417		2
010		0.466			1.534		2
011					1.024	0.976	2
012					1.024	0.976	2
013	0.462					1.538	2
014		0.490				1.510	2
Σ	4	4	5	5	5	5	

74212: CaTi ⁴⁺ O ₃			
	<i>Ca</i>	<i>Ti</i>	
01	0.167 × 4↓ × 4→	0.667 × 2↓ × 2→	2
02	0.167 × 8↓ × 4→	0.667 × 4↓ × 2→	2
	2	4	

75264: $\text{Li}_3\text{Nb}^{5+}\text{O}_4$			
	<i>Li</i>	<i>Nb</i>	Σ
<i>O1_{br}</i>	-0.042 $\times 3 \rightarrow$	0.708 $\times 3 \downarrow \times 3 \rightarrow$	2
<i>O2</i>	0.208 $\times 5 \downarrow \times 5 \rightarrow$	0.958 $\times 3 \downarrow$	2
Σ	1	5	

75583: $\text{La}_2(\text{Ti}^{4+}_2\text{SiO}_9)$						
	<i>La1</i>	<i>La2</i>	<i>Ti1</i>	<i>Ti2</i>	<i>Si</i>	Σ
<i>O1</i>	0.280 $\times 2 \downarrow$	0.186 $\times 2 \downarrow$	0.619 $\times 2 \downarrow$		0.914 $\times 2 \downarrow$	2
<i>O2</i>		0.668			1.332	2
<i>O3</i>	0.268	0.174 $\times 2 \downarrow \times 2 \rightarrow$		0.544	0.839	2
<i>O4</i>	0.454 $\times 4 \downarrow \times 2 \rightarrow$	0.361 $\times 2$		0.731 $\times 2 \downarrow$		2
<i>O5</i>	0.354	0.260	0.693 $\times 2 \downarrow \times 2 \rightarrow$			2
<i>O6</i>		0.315 $\times 2 \downarrow \times 2 \rightarrow$		0.685 $\times 2 \downarrow \times 2 \rightarrow$		2
<i>O7</i>			0.688 $\times 2 \downarrow \times 2 \rightarrow$	0.625		2
Σ	3	3	4	4	4	

78180: $\text{La}_2(\text{W}^{6+}\text{O}_4)_3$				
	<i>La</i>	<i>W1</i>	<i>W2</i>	Σ
<i>O1</i>	0.095	0.952 $\times 2 \downarrow \times 2 \rightarrow$		2
<i>O2</i>	0.381 $\times 2 \downarrow \times 2 \rightarrow$	1.238		2
<i>O3</i>	0.571	1.429		2
<i>O4</i>	0.429		1.571 $\times 2 \downarrow$	2
<i>O5</i>	0.571	1.429		2
<i>O6</i>	0.286 $\times 2 \downarrow \times 2 \rightarrow$		1.429 $\times 2 \downarrow$	2
Σ	3	6	6	

78842: Rb ₂ (Ti ⁴⁺ O ₃)				
	<i>Rb1</i>	<i>Rb2</i>	<i>Ti</i>	Σ
01	0.185 ×2↓ ×2→	0.1885 ×3↓ ×3→	1.074	2
02	0.185 ×3↓ ×3→	0.185 ×2↓ ×2→	1.074	2
03	0.037 ×2↓ ×2→	0.037 ×2↓ ×2→	0.926 ×2↓ ×2→	2
Σ	1	1	4	

79517: Cs(Mo ⁶⁺ ₂ O ₃ (PO ₄) ₂)						
	<i>Cs</i>	<i>Mo1</i>	<i>Mo2</i>	<i>P1</i>	<i>P2</i>	Σ
01	0.382 ×2↓ ×2→	1.237				2
02	-0.087	0.768		1.319		2
03		0.724		1.276		2
04	-0.083	0.772			1.311	2
05	-0.087	0.768		1.319		2
06		0.731			1.269	2
07	0.255 ×2↓ ×2→		1.491			2
08	0.255 ×2↓ ×2→		1.491			2
09			0.915	1.085		2
010	-0.210		1.026		1.184	2
011	-0.157 ×2↓ ×2→		1.078		1.236	2
Σ	1	5	6	5	5	

79702: $K_2Ni^{2+}(W^{6+}O_2(PO_4)_2)$					
	<i>K</i>	<i>Ni</i>	<i>W</i>	<i>P</i>	Σ
O1	-0.059		0.908 ×2↓	1.151	2
O2	-0.059		0.908 ×2↓	1.151	2
O3	0.125 ×3↓ ×3→	0.292 ×2↓		1.334	2
O4	0.156 ×2↓ ×2→	0.324 ×2↓		1.365	2
O5	0.216 ×2↓ ×2→	0.384 ×2↓	1.183 ×2↓		2
Σ	1	2	6	5	

79734: $K_3(Nb^{5+}_3O_6)(Si_2O_7)$				
	<i>K</i>	<i>Nb</i>	<i>Si</i>	Σ
O1	0.017 ×3→		0.975 ×2→	2
O2	0.105 ×3↓ ×3→	0.843 ×2↓ ×2→		2
O3	0.131 ×2↓ ×2→	0.869 ×2↓ ×2→		2
O4	0.051 ×8↓ ×4→	0.788 ×2↓	1.008 ×3↓	2
Σ	1	5	4	

80423: $CsTa^{5+}(B_2O_5)$				
	<i>Cs</i>	<i>Ta</i>	<i>B</i>	Σ
O1	0.114 ×4↓ ×2→	0.750 ×2↓	1.023	2
O2	0.114 ×4↓ ×2→	0.750 ×2↓	1.023	2
O3		1.000 ×2↓ ×2→		2
O4	0.045 ×2↓ ×2→		0.955 ×2→	2
Σ	1	5	3	

80430: $\text{KMn}^{3+}(\text{Se}^{6+}\text{O}_4)_2$				
	<i>K</i>	<i>Mn</i>	<i>Se</i>	Σ
<i>O1</i>		$0.512 \times 2\downarrow$	1.488	2
<i>O2</i>	$0.142 \times 6\downarrow \times 3$		1.574	2
<i>O3</i>	$0.037 \times 2\downarrow$	$0.494 \times 2\downarrow$	1.469	2
<i>O4</i>	$0.037 \times 2\downarrow$	$0.494 \times 2\downarrow$	1.469	2
Σ	1	3	6	

81473: $\text{BaCo}^{2+}_2(\text{Si}_2\text{O}_7)$							
	<i>Ba</i>	<i>Co1</i>	<i>Co2</i>	<i>Co3</i>	<i>Si1</i>	<i>Si2</i>	Σ
<i>O1</i>			$0.526 \times 2\downarrow$	0.449	1.025		2
<i>O2</i>	$0.246 \times 2\downarrow \times 2\rightarrow$		$0.474 \times 2\downarrow$			1.035	2
<i>O3</i>	0.227			$0.379 \times 2\downarrow \times 2\rightarrow$		1.016	2
<i>O4</i>	0.210	$0.429 \times 2\downarrow$		0.362		0.999	2
<i>O5</i>	0.161				0.889	0.950	2
<i>O6</i>	0.351	$0.571 \times 2\downarrow$			1.079		2
<i>O7</i>	$0.280 \times 2\downarrow \times 2\rightarrow$			0.432	1.008		2
Σ	2	2	2	2	4	4	

82403: $\text{Co}^{2+}_2\text{Si}(\text{P}_2\text{O}_7)_2$					
	<i>Co</i>	<i>Si</i>	<i>P1</i>	<i>P2</i>	Σ
01		1.000 ×2↓	1.000		2
02	0.429		1.571		2
03	0.286 ×2↓ ×2→		1.429		2
04			1.000	1.000	2
05	0.429			1.571	2
06	0.286 ×2↓ ×2→			1.429	2
07		1.000 ×2↓		1.000	2
Σ	2	4	5	5	

82488: $\text{Na}_4\text{Zr}_2\text{Ti}^{4+}\text{O}_4(\text{CO}_3)_4$							
	<i>Na1</i>	<i>Na2</i>	<i>Zr</i>	<i>Ti</i>	<i>C1</i>	<i>C2</i>	Σ
01	0.064	0.064		0.595 ×2↓	1.277		2
02	0.091	0.091	0.484			1.333	2
03	0.171		0.563 ×2↓ ×2→	0.702 ×2↓			2
04		0.171	0.563 ×2↓ ×2→	0.702 ×2↓			2
05	0.197 ×2↓ ×2→	0.197			1.410		2
06	-0.007		0.386 ×2↓ ×2→			1.235	2
07	0.098	0.098	0.491		1.312		2
08	0.189	0.190 ×2↓ ×2→				1.431	2
Σ	1	1	4	4	4	4	

82685: SrV ³⁺ ₂ O(PO ₄) ₂						
	Sr	V1	V2	P1	P2	Σ
O1	0.273 ×2↓	0.472 ×2↓			1.255 ×2↓	2
O2	0.269	0.468		1.263		2
O3		0.609			1.391	2
O4	0.269	0.468		1.263		2
O5	0.312	0.511	0.588 ×2↓ ×2→			2
O6	0.117		0.392 ×2↓ ×2→		1.099	2
O7	0.243 ×2↓		0.519 ×2↓	1.237 ×2↓		2
Σ	2	3	3	5	5	

83285: K(Fe ³⁺ ₁₁ O ₁₇)						
	K	Fe1	Fe2	Fe3	Fe4	Σ
O1		0.379 ×2↓ ×2→	0.742 ×3↓		0.500 ×6↓	2
O2	0.167 ×6↓	0.583 ×2↓ ×2→		0.667 ×3↓		2
O3		0.409 ×3→	0.773			2
O4		0.667 ×3→				2
O5	0.000 ×3↓ ×3→			1.000 ×2→		2
Σ	1	3	3	3	3	

85042: Mn ²⁺ Zn ₂ Ta ⁵⁺ ₂ O ₈				
	Zn	Mn	Ta	Σ
O1	0.410 ×3↓ ×3→		0.769	2
O2	0.256 ×3↓ ×3→		0.615 ×2↓ ×2→	2
O3		0.417 ×4↓ ×2→	1.167	2
O4		0.167 ×2↓	0.917 ×2↓ ×2→	2
Σ	2	2	5	

85497: NaY(GeO ₄)				
	<i>Na</i>	<i>Y</i>	<i>Ge</i>	Σ
O1	0.197 $\times 2\downarrow \times 2 \rightarrow$	0.561	1.045	2
O2	0.197 $\times 2\downarrow \times 2 \rightarrow$	0.561	1.045	2
O3	0.106 $\times 2\downarrow$	0.470 $\times 4\downarrow \times 2 \rightarrow$	0.955 $\times 2\downarrow$	2
Σ	1	3	4	

85735: ZnLiNb ⁵⁺ O ₄				
	<i>Li</i>	<i>Zn</i>	<i>Nb</i>	Σ
O1	0.212 $\times 4\downarrow \times 2 \rightarrow$	0.818 $\times 2\downarrow$	0.758 $\times 2\downarrow$	2
O2	0.076 $\times 2\downarrow$	0.682 $\times 2\downarrow$	0.621 $\times 4\downarrow \times 2 \rightarrow$	2
Σ	1	2	5	

86144: W ⁶⁺ O ₃		
	<i>W</i>	Σ
O1	1.000 $\times 2\downarrow \times 2 \rightarrow$	2
O2	1.000 $\times 4\downarrow \times 2 \rightarrow$	2
Σ	6	

88879: Nb ⁴⁺ O ₂			
	<i>Nb1</i>	<i>Nb2</i>	Σ
O1	0.667	0.667 $\times 2\downarrow \times 2 \rightarrow$	2
O2	0.667 $\times 2\downarrow \times 2 \rightarrow$	0.667	2
O3	0.667	0.667 $\times 2\downarrow \times 2 \rightarrow$	2
O4	0.667 $\times 2\downarrow \times 2 \rightarrow$	0.667	2
Σ	4	4	

89466: $V^{5+}_2Se^{4+}_2O_9$					
	<i>Se1</i>	<i>Se2</i>	<i>V1</i>	<i>V2</i>	Σ
01			2.000		2
02	1.200		0.400	0.400	2
03		1.400	0.600		2
04	1.400		0.600		2
05			1.000	1.000	2
06		1.200	0.400	0.400	2
07	1.400			0.600	2
08		1.400		0.600	2
09				2.000	2
Σ	4	4	5	5	

89506: $K(Mn^{7+}O_4)$			
	<i>K</i>	<i>Mn</i>	Σ
01	$0.083 \times 3 \downarrow \times 3 \rightarrow$	1.750	2
02	$0.083 \times 3 \downarrow \times 3 \rightarrow$	1.750	2
03	$0.083 \times 6 \downarrow \times 3$	$1.750 \times 2 \downarrow$	2
Σ	1	7	

90110: $\text{Tl}_2(\text{Te}^{4+}\text{Mo}^{6+}_2\text{O}_6(\text{PO}_4)_2)$					
	<i>Tl</i>	<i>Mo</i>	<i>P</i>	<i>Te</i>	Σ
O1	0.005	0.924		1.071 ×2↓	2
O2	-0.066 ×2↓ ×2→	0.853	1.279		2
O3	0.360 ×2↓ ×2→	1.280			2
O4	0.360 ×2↓ ×2→	1.280			2
O5	-0.088	0.831	1.257		2
O6	-0.088	0.831	1.257		2
O7	-0.137		1.208	0.929 ×2↓	2
Σ	1	6	5	4	

91748: $\text{Mg}_4\text{Nb}^{5+}_2\text{O}_9$				
	<i>Mg1</i>	<i>Mg2</i>	<i>Nb</i>	Σ
O1		0.250 ×3↓ ×2→	0.750 ×3↓ ×2→	2
O2	0.333 ×6↓ ×2→	0.417 ×3↓	0.917 ×3↓	2
Σ	2	2	5	

92317: $(\text{V}^{4+}\text{O})(\text{Re}^{7+}\text{O}_4)_2$				
	<i>V</i>	<i>Re1</i>	<i>Re2</i>	Σ
O1	0.333 ×2↓	1.667 ×2↓		2
O2	0.333	1.667		2
O3		2.000		2
O4	0.500 ×2↓		1.500 ×2↓	2
O5			2.000	2
O6			2.000	2
O7	2.000			2
Σ	4	7	7	

92489: Eu ³⁺ ₄ (Au ³⁺ ₂ O ₉)				
	<i>Au</i>	<i>Eu1</i>	<i>Eu2</i>	Σ
<i>O1</i>		0.500 ×2→	0.500 ×2→	2
<i>O2</i>	0.750	0.417 ×2↓ ×2→	0.417	2
<i>O3</i>	0.750	0.417	0.417 ×2↓ ×2→	2
<i>O4</i>	0.750	0.417 ×2↓ ×2→	0.417	2
<i>O5</i>	0.750	0.417	0.417 ×2↓ ×2→	2
Σ	3	3	3	

92508: Pr ₃ (Re ⁷⁺ O ₈)					
	<i>Pr1</i>	<i>Pr2</i>	<i>Pr3</i>	<i>Re</i>	Σ
<i>O1</i>	0.484 ×2↓ ×2→	0.518	0.513		2
<i>O2</i>	0.484 ×2↓ ×2→	0.518	0.513		2
<i>O3</i>	0.276 ×2→		0.305 ×2↓	1.144	2
<i>O4</i>	0.266	0.300 ×2↓ ×2→		1.134	2
<i>O5</i>		0.390	0.385	1.225	2
<i>O6</i>	0.267	0.301	0.296	1.136	2
<i>O7</i>	0.255	0.289	0.294	1.133	2
<i>O8</i>		0.384	0.388	1.228	2
Σ	3	3	3	7	

94743: NaKLaNb ⁵⁺ O ₅					
	<i>La</i>	<i>Nb</i>	<i>Na</i>	<i>K</i>	Σ
<i>O1</i>	0.375 ×8↓ ×2→	0.978 ×4↓	0.178 ×4↓	0.047 ×8↓ ×2→	2
<i>O2</i>		1.088	0.288	0.156 ×4↓ ×4→	2
Σ	3	5	1	1	

95493: Gd(Mn ³⁺ O ₃)			
	<i>Gd</i>	<i>Mn</i>	Σ
01	0.441 $\times 2\downarrow \times 2\rightarrow$	0.559 $\times 2\downarrow \times 2\rightarrow$	2
02	0.353 $\times 6\downarrow \times 3$	0.471 $\times 4\downarrow \times 2\rightarrow$	2
Σ	3	3	

		3		5		0.258																	
011				0.87 6	0.91 0					0.21 4													2
012				1.45 1		0.139							0.41 0										2
013	0.836	0.56 4	0.60 0																				2
014					1.67 3	0.327																	2
015 (011)							0.30 1	0.83 6	0.86 3														2
016 (012)						0.135	0.21 7			0.86 3	0.78 6												2
017 (013)						- 0.063					0.76 4	0.78 4		0.30 9								0.20 7	2
018 (014)						0.320							0.59 0	0.69 1								0.39 8	2
019 (015)							0.43 2								0.92 9	0.63 9							2
020 (016)						0.166	0.24 9										1.027	0.55 9					2
021						0.114	0.19												0.84	0.84			2

96359: SrFe ³⁺ ₃ (PO ₄) ₃ O								
	Sr1	Fe1	Fe2	Fe3	P1	P2	P3	Σ
01		0.671 ×2↓ ×2→	0.658					2
02	0.239 ×2↓	0.467 ×2↓			1.295 ×2↓			2
03	0.077 ×2↓ ×2→		0.292	0.434	1.120			2
04	0.247		0.462		1.290			2
05	0.259 ×2↓		0.474 ×2↓			1.267 ×2↓		2
06	0.134	0.362 ×2↓ ×2→				1.141		2
07				0.675		1.325		2
08	0.235 ×2↓			0.591 ×2↓			1.174 ×2↓	2
09			0.638				1.362	2
010				0.709			1.291	2
Σ	2	3	3	3	5	5	5	

96454: $\text{Pb}^{2+}_2(\text{Mo}^{4+}_2\text{O}(\text{PO}_4)_2(\text{P}_2\text{O}_7))$					
	<i>Pb2</i>	<i>Mo</i>	<i>P1</i>	<i>P2</i>	Σ
01		1.000 $\times 2 \rightarrow$			2
02	0.206	0.612		1.182	2
03	0.206	0.612		1.182	2
04		0.629	1.371		2
05		0.629	1.371		2
06	0.112 $\times 2 \downarrow \times 2 \rightarrow$	0.518	1.259		2
07			1.000 $\times 2 \rightarrow$		2
08	0.341 $\times 2 \downarrow \times 2 \rightarrow$			1.318	2
09	0.341 $\times 2 \downarrow \times 2 \rightarrow$			1.318	2
Σ	2	4	5	5	

97525: $\text{Ba}_6\text{Ru}^{5+}_2\text{Na}_2\text{Mn}^{5+}_2\text{O}_{17}$								
	<i>Na</i>	<i>Ba1</i>	<i>Ba2</i>	<i>Ba3</i>	<i>Ba4</i>	<i>Mn</i>	<i>Ru</i>	Σ
0 1	0.167 $\times 3 \downarrow$		0.167 $\times 6 \downarrow$ $\times 2 \rightarrow$	0.333 $\times 6 \downarrow$		1.167 $\times 3 \downarrow$		2
0 2	0.167 $\times 3 \downarrow$	0.267 $\times 6 \downarrow$ $\times 2 \rightarrow$	0.167 $\times 3 \downarrow$		0.233 $\times 6 \downarrow$		0.900 $\times 3 \downarrow$	2
0 3		0.133 $\times 3 \downarrow$ $\times 2 \rightarrow$			0.100 $\times 6 \downarrow$ $\times 2 \rightarrow$		0.767 $\times 3 \downarrow$ $\times 2 \rightarrow$	2
0 4			0.500			1.500		2
Σ	1	2	2	2	2	5	5	

99594: Mn ²⁺ V ⁵⁺ Sb ⁵⁺ O ₆				
	<i>Mn</i>	<i>V</i>	<i>Sb</i>	Σ
01	0.133 ×2↓	0.833 ×2↓	1.033 ×2↓	2
02	0.433 ×4↓ ×2→	1.133 ×2↓		2
03		0.533 ×2↓	0.733 ×4↓ ×2→	2
Σ	2	5	5	

100082: CaMn ²⁺ ₂ (BeSiO ₄) ₃										
	<i>Ca</i>	<i>Mn1</i>	<i>Mn2</i>	<i>Be1</i>	<i>Be2</i>	<i>Be3</i>	<i>Si1</i>	<i>Si2</i>	<i>Si3</i>	Σ
01	0.194 ×2↓ ×2→					0.556	1.056			2
02	0.194 ×2↓ ×2→			0.556				1.056		2
03	0.194 ×2↓ ×2→				0.556				1.056	2
04		0.306	0.306	0.444			0.944			2
05		0.306	0.306		0.444			0.944		2
06		0.306	0.306			0.444			0.944	2
07	0.139		0.361		0.500		1.000			2
08	0.139	0.361				0.500		1.000		2
09	0.139		0.361	0.500					1.000	2
010	0.139	0.361		0.500			1.000			2
011	0.139		0.361		0.500			1.000		2
012	0.139	0.361				0.500			1.000	2
Σ	2	2	2	2	2	2	4	4	4	

100158: Na ₂ Ca ₄ ZrNb ⁵⁺ (Si ₂ O ₇) ₂ FO ₃													
	Na1	Na2	Ca1	Ca2	Ca3	Ca4	Zr	Nb	Si1	Si2	Si3	Si4	Σ
01				0.35 5		0.34 9		0.76 2	0.53 4				2
02	- 0.019			0.43 5		0.42 8				1.15 6			2
03	- 0.252			0.20 2				0.75 8	1.29 2				2
04	- 0.233					0.36 4		0.77 7		1.09 2			2
05			0.12 4				0.75 0		1.12 6				2
06	0.170		0.14 0				0.76 7			0.92 3			2
07		0.445	0.81 9		0.578						0.15 8		2
08		0.738			0.871		0.00 4					0.38 6	2
09		- 0.206	0.16 8				0.79 5				1.24 2		2
01 0			0.15 1		- 0.090		0.77 8					1.16 0	2
01 1		- 0.003		0.28 2		0.27 6					1.44 5		2
01 2				0.26 5		0.25 9						1.36 3	2
01 3	0.309					0.038	0.90 6	0.74 7					2
01 4	0.499	0.095	0.46 9					0.93 7					2
01		0.176		0.46	0.309			1.01					2

5				1				8					
O1			0.04						1.04	0.82			
6	0.076		6						8	9			2
O1		-	0.08								1.15	1.09	
7		0.292	2								5	1	2
F	0.450	0.046			0.179	0.32							1
Σ	1	1	2	2	2	2	4	5	4	4	4	4	

100278: ScAlO ₃			
	Sc	Al	Σ
O1	0.441 × 2↓ × 2→	0.559 × 2↓ × 2→	2
O2	0.353 × 6↓ × 3→	0.471 × 4↓ × 2→	2
Σ	3	3	

100571: Ba ₁₀ (Re ⁷⁺ O ₅) ₆ Br ₂				
	Ba1	Ba2	Re	Σ
O1		0.195 × 3↓ × 3→	1.414	2
O2	0.218 × 6↓ × 2→	0.172 × 2↓	1.391 × 2↓	2
O3	0.230 × 3↓	0.184 × 4↓ × 2→	1.402 × 2↓	2
Br		0.167 × 2↓ × 6→		1
Σ	2	2	7	

156736: Ca ₃ Mn ²⁺ (Sb ⁵⁺ ₄ O ₁₄)							
	<i>Mn</i>	<i>Ca1</i>	<i>Ca2</i>	<i>Sb1</i>	<i>Sb2</i>	<i>Sb3</i>	Σ
<i>O1</i>	0.280 ×2↓	0.154 ×2↓			0.771 ×2↓	0.795	2
<i>O2</i>			0.244			0.878 ×2↓ ×2→	2
<i>O3</i>	0.278 ×2↓		0.159		0.769 ×2↓	0.793	2
<i>O4</i>		0.343 ×2↓	0.349 ×2↓ ×2→		0.960 ×2↓		2
<i>O5</i>	0.442 ×2↓	0.316 ×2↓	0.322	0.920 ×2↓			2
<i>O6</i>		0.187 ×2↓	0.194	0.791 ×2↓		0.828	2
<i>O7</i>			0.192 ×2↓ ×2→	0.789 ×2↓		0.827	2
Σ	2	2	2	5	5	5	

157733: NaFe ³⁺ Si ₂ O ₆				
	<i>Na</i>	<i>Fe</i>	<i>Si</i>	Σ
<i>O1</i>	0.113 ×2↓	0.45 ×4↓ ×2→	0.988	2
<i>O2</i>	0.263 ×2↓	0.6 ×2↓	1.138	2
<i>O3</i>	0.062 ×4↓ ×2→		0.938 ×2↓ ×2→	2
Σ	1	3	4	

170119: K ₂ Mo ⁶⁺ O ₂ (I ⁵⁺ O ₃) ₄					
	<i>K</i>	<i>I1</i>	<i>I2</i>	<i>Mo</i>	Σ
<i>O1</i>	-0.143	0.502	0.593	1.048 ×2↓	2
<i>O2</i>	0.178 ×2↓ ×2→	0.822 ×2↓ ×2→			2
<i>O3</i>	0.155 ×2↓ ×2→	0.800	0.891		2
<i>O4</i>		0.454	0.545	1.001 ×2↓	2
<i>O5</i>	0.155 ×2↓ ×2→	0.800	0.891		2
<i>O6</i>	0.155 ×2↓ ×2→	0.800	0.891		2
<i>O7</i>	-0.141		0.595 ×2↓ ×2→	0.951 ×2↓	2
Σ	1	5	5	6	

171028: Cu ²⁺ ₂ V ⁵⁺ ₂ O ₇					
	<i>Cu1</i>	<i>Cu2</i>	<i>V1</i>	<i>V2</i>	Σ
<i>O1</i>	0.356 ×2↓ ×2→			1.288	2
<i>O2</i>	0.355 ×2↓ ×2→		1.291		2
<i>O3</i>		0.363 ×2↓ ×2→	1.274		2
<i>O4</i>		0.365 ×2↓ ×2→		1.271	2
<i>O5</i>	0.044		0.980	0.976	2
<i>O6</i>	0.534			1.466	2
<i>O7</i>		0.545	1.455		2
Σ	2	2	5	5	

171758: Na ₂ Mo ⁶⁺ ₃ Te ⁴⁺ ₃ O ₁₆							
	<i>Na1</i>	<i>Na2</i>	<i>Mo1</i>	<i>Mo2</i>	<i>Te1</i>	<i>Te2</i>	Σ
<i>O1</i>				0.975	1.025 ×2↓		2
<i>O2</i>	-0.087 ×2↓				0.975 ×2↓	1.112	2
<i>O3</i>			0.772 ×2↓	0.521		0.707	2
<i>O4</i>	-0.058 ×2↓	0.323 ×2↓		0.956		1.142	2
<i>O5</i>			1.125 ×2↓	0.875			2
<i>O6</i>		-0.142 ×2↓	1.103 ×2↓			1.039	2
<i>O7</i>	0.323 ×2↓	0.341 ×2↓		1.336			2
<i>O8</i>	0.323 ×2↓	0.341 ×2↓		1.336			2
Σ	1	1	6	6	4	4	

200128: LiSc(SiO ₃) ₂				
	<i>Li</i>	<i>Sc</i>	<i>Si</i>	Σ
<i>O1</i>	0.125 ×2↓	0.450 ×4↓ ×2→	0.975	2
<i>O2</i>	0.275 ×2↓	0.600 ×2↓	1.125	2
<i>O3</i>	0.100 ×2↓		0.950 ×2↓ ×2→	2
Σ	1	3	4	

200743: Cd ₂ P ₆ O ₁₇					
	<i>Cd1</i>	<i>P1</i>	<i>P2</i>	<i>P3</i>	Σ
<i>O1</i>		0.898	1.102		2
<i>O2</i>	0.398	1.602			2
<i>O3</i>	0.398	1.602			2
<i>O4</i>		0.898	1.102		2
<i>O5</i>	0.296		1.704		2
<i>O6</i>			1.093	0.907	2
<i>O7</i>				1.000 ×2→	2
<i>O8</i>	0.389			1.611	2
<i>O9</i>	0.259 ×2↓ ×2→			1.481	2
Σ	2	5	5	5	

200854: RbNb ⁵⁺ O ₃						
	<i>Rb1</i>	<i>Rb2</i>	<i>Rb3</i>	<i>Nb1</i>	<i>Nb2</i>	Σ
<i>O1</i>		0.094 ×2↓	0.033 ×2↓ ×2→	0.917	0.923	2
<i>O2</i>	-0.240 ×2↓			0.557 ×2↓ ×2→	0.563 ×2↓ ×2→	2
<i>O3</i>		0.102 ×2↓	0.041	0.925	0.931	2
<i>O4</i>	0.121 ×2↓		0.035	0.919	0.925	2
<i>O5</i>	0.291 ×2↓		0.205 ×3↓ ×3→		1.095	2
<i>O6</i>	0.328 ×2↓	0.303 ×2↓	0.242	1.126		2
Σ	1	1	1	5	5	

201658: $V^{4+}O(HPO_4)(H_2O)_{0.5}$				
	<i>V</i>	<i>P</i>	<i>H</i>	Σ
<i>OH</i>		0.739	0.261	1
<i>O2</i>	0.522 $\times 2\downarrow$	1.478 $\times 2$		2
<i>O3</i>	0.348 $\times 2\downarrow \times 2\rightarrow$	1.304		2
<i>O4</i>	2.000			2
<i>O5</i>	0.261 $\times 2\rightarrow$		0.739 $\times 2\rightarrow$	2
Σ	4	5	1	

201733: $Cu^{2+}_3(As^{5+}O_4)_2$				
	<i>Cu1</i>	<i>Cu2</i>	<i>As</i>	Σ
<i>O1</i>	0.500 $\times 2\downarrow$	0.327	1.173	2
<i>O2</i>		0.577	1.423	2
<i>O3</i>		0.385 $\times 2\downarrow \times 2\rightarrow$	1.231	2
<i>O4</i>	0.500 $\times 2\downarrow$	0.327	1.173	2
Σ	2	2	5	

201793: $Tl^{3+}_2(Cr^{6+}O_4)_3$				
	<i>Tl</i>	<i>Cr1</i>	<i>Cr2</i>	Σ
<i>O1</i>	0.400 $\times 2\downarrow \times 2\rightarrow$	1.200		2
<i>O2</i>		2.000		2
<i>O3</i>	0.600	1.400		2
<i>O4</i>	0.500		1.500 $\times 2\downarrow$	2
<i>O5</i>	0.600	1.400		2
<i>O6</i>	0.500		1.500 $\times 2\downarrow$	2
Σ	3	6	6	

202414: W ⁵⁺ P ₈ O ₄₀			
	<i>W</i>	<i>P</i>	Σ
O1	1.000 $\times 2\downarrow \times 2\rightarrow$		2
O2	0.500	1.500	2
O3	0.750	1.250	2
O4	1.000 $\times 2\rightarrow$		2
O5	0.750	1.250	2
O6		1.000 $\times 2\rightarrow$	2
Σ	5	5	

203048: PW ⁵⁺ O ₅			
	<i>W</i>	<i>P</i>	Σ
O1	0.75	1.25	2
O2	0.75	1.25	2
O3	0.75	1.25	2
O4	0.75	1.25	2
O5	1 $\times 2\downarrow \times 2\rightarrow$		2
Σ	5	5	

203232: DyTa ⁵⁺ ₇ O ₁₉				
	<i>Dy</i>	<i>Ta1</i>	<i>Ta2</i>	Σ
O1			1.000 $\times 2\rightarrow$	2
O2	0.417 $\times 6\downarrow$	0.833 $\times 6\downarrow$	0.750	2
O3			0.667 $\times 3\downarrow \times 3\rightarrow$	2
O4			0.667 $\times 3\rightarrow$	2
O5	0.250 $\times 2\downarrow$		0.583 $\times 3\rightarrow$	2
Σ	3	5	5	

247056: LiCr ³⁺ (Mo ⁶⁺) ₂					
	<i>Li</i>	<i>Cr</i>	<i>Mo1</i>	<i>Mo2</i>	Σ
<i>O1</i>		0.385 ×2↓ ×2→		1.230	2
<i>O2</i>	0.317			1.683	2
<i>O3</i>		0.562	1.438		2
<i>O4</i>		0.562	1.438		2
<i>O5</i>	0.038	0.559		1.404	2
<i>O6</i>	0.027	0.548	1.425		2
<i>O7</i>	0.301		1.699		2
<i>O8</i>	0.317			1.683	2
Σ	1	3	6	6	

248227: K ₃ V ⁵⁺ O ₁₄				
	<i>K</i>	<i>V1</i>	<i>V2</i>	
<i>O1</i>	0.190 ×2↓ ×3→	1.430		2
<i>O2</i>	0.260 ×3↓ ×3→		1.220	2
<i>O3</i>	0.02 ×2↓ ×2→		0.980 ×2↓ ×2→	2
<i>O4</i>	-0.050 ×4↓ ×2→	1.190 ×3↓	0.910 ×2↓	2
	1	5	5	

249142: $\text{Pb}^{2+}_2(\text{V}^{4+}\text{O}(\text{PO}_4)_2)$						
	Pb1	Pb2	V	P1	P2	Σ
O1	0.123		0.648		1.229	2
O2		0.531		1.469		2
O3	0.329 $\times 2 \downarrow \times 2 \rightarrow$	0.202		1.140		2
O4	0.329 $\times 2 \downarrow \times 2 \rightarrow$	0.202		1.140		2
O5	0.123		0.648		1.229	2
O6		0.039	0.690		1.271	2
O7		0.039	0.690		1.271	2
O8	0.438	0.312		1.250		2
O9		0.675	1.325			2
Σ	2	2	4	5	5	

250466: $\text{Na}_2(\text{Co}^{2+}(\text{NO}_3)_4)$								
	Na1	Na2	Co	N1	N2	N3	N4	Σ
O1	0.101		0.271	1.628				2
O2	0.236			1.764				2
O3		0.140	0.252	1.609				2
O4	0.101		0.271		1.628			2
O5		0.140	0.252		1.609			2
O6	0.236				1.764			2
O7	0.119	0.177				1.704		2
O8		0.120	0.232			1.648		2
O9		0.120	0.232			1.648		2
O10	0.085		0.255				1.661	2
O11		0.123	0.235				1.641	2
O12	0.122	0.180					1.698	2
Σ	1	1	2	5	5	5	5	

280066: Cs ₂ (Mo ⁶⁺ ₃ O ₁₀)				
	<i>Cs</i>	<i>Mo1</i>	<i>Mo2</i>	Σ
<i>O1</i>	0.192 ×3↓ ×3→	1.424 ×2↓		2
<i>O2</i>	-0.172 ×2↓ ×2→	1.060 ×2↓	1.285	2
<i>O3</i>		0.516 ×2↓	0.742 ×2↓ ×2→	2
<i>O4</i>	0.181 ×2↓ ×2→		1.638	2
<i>O5</i>	0.136 ×3↓ ×3→		1.593	2
Σ	1	6	6	

280154: Na ₂ Ca ₃ Ta ⁵⁺ ₂ O ₉					
	<i>Na</i>	<i>Ca1</i>	<i>Ca2</i>	<i>Ta</i>	Σ
<i>O1</i>	0.167 ×6↓ ×2→	0.500 ×3↓	0.333 ×6↓	0.833 ×3↓	2
<i>O2</i>		0.167 ×3↓ ×2→		0.833 ×3↓ ×2→	2
Σ	1	2	2	5	

280292: Hg ²⁺ (PO ₃) ₂				
	<i>Hg</i>	<i>P1</i>	<i>P2</i>	Σ
<i>O1</i>	0.429	1.571		2
<i>O2</i>	0.286 ×2↓ ×2→	1.429		2
<i>O3</i>	0.286 ×2↓ ×2→		1.429	2
<i>O4</i>	0.429		1.571	2
<i>O5</i>		1.000	1.000	2
<i>O6</i>		1.000	1.000	2
Σ	2	5	5	

280309: SrCr ²⁺ (P ₂ O ₇)					
	<i>Sr</i>	<i>Cr</i>	<i>P1</i>	<i>P2</i>	Σ
<i>O1</i>	0.252	0.412	1.336		2
<i>O2</i>	0.253	0.413		1.334	2
<i>O3</i>	0.189 ×2↓ ×2→	0.349	1.273		2
<i>O4</i>			1.002	0.998	2
<i>O5</i>	0.305 ×2↓ ×2→		1.389		2
<i>O6</i>	0.253	0.413		1.334	2
<i>O7</i>	0.253	0.413		1.334	2
Σ	2	2	5	5	

280501: Rb(Ti ⁴⁺ O)(As ⁵⁺ O ₄)							
	<i>Rb1</i>	<i>Rb2</i>	<i>Ti1</i>	<i>Ti2</i>	<i>As1</i>	<i>As2</i>	Σ
<i>O1</i>	0.226	0.187	0.363		1.224		2
<i>O2</i>	0.226	0.187	0.363		1.224		2
<i>O3</i>	0.209	0.171		0.379	1.241		2
<i>O4</i>		0.240		0.449	1.310		2
<i>O5</i>	0.433	0.394	0.570	0.603			2
<i>O6</i>	0.039	0.000	0.964	0.997			2
<i>O7</i>	-0.042	-0.081	0.883			1.241	2
<i>O8</i>	-0.070		0.856			1.213	2
<i>O9</i>	-0.010	-0.049		0.786		1.273	2
<i>O10</i>	-0.010	-0.049		0.786		1.273	2
Σ	1	1	4	4	5	1.241	

280589: YMn ³⁺ O ₃				
	Y1	Y2	Mn	Σ
O1	0.444 ×3↓	0.444 ×3↓ ×2→	0.667	2
O2	0.444 ×3↓	0.444 ×3↓ ×2→	0.667	2
O3	0.333		0.556 ×3→	2
O4		0.333	0.556 ×2↓ ×3→	2
Σ	3	3	3	

280775: V ⁵⁺ AlMo ⁶⁺ O ₇				
	Al	V	Mo	Σ
O1	0.500 ×2↓		1.500 ×2↓	2
O2	0.500		1.500	2
O3	0.500		1.500	2
O4	0.500 ×2↓	0.750 ×4↓ ×2→		2
O5		2.000		2
Σ	3	5	6	

280902: NaZnFe ³⁺ ₂ (PO ₄) ₃								
	Na	Zn	Fe1	Fe2	P1	P2	P3	Σ
O1	0.252	0.416			1.332			2
O2				0.775	1.225			2
O3	0.186		0.549		1.265			2
O4	0.098	0.263	0.462		1.178			2
O5			0.621			1.379		2
O6	0.084			0.712		1.204		2
O7	0.088	0.252	0.451			1.209		2
O8	0.088	0.252	0.451			1.209		2
O9		0.550					1.450	2
O10	0.103			0.731			1.166	2
O11				0.782			1.218	2
O12	0.102	0.266	0.465				1.166	2
Σ	1	2	3	3	5	5	5	

281197: Pr ₃ Mo ⁵⁺ O ₇					
	Pr1	Pr2	Pr3	Mo	Σ
O1	0.354	0.355	0.414	0.877	2
O2		0.355 × 2↓ × 2→	0.414	0.877	2
O3	0.470 × 2↓ × 2→		0.530 × 2↓ × 2→		2
O4	0.354	0.355	0.414	0.877	2
O5		0.224	0.283	0.746 × 2↓ × 2→	2
O6	0.354	0.355	0.414	0.877	2
O7	0.499 × 2↓ × 2→	0.501 × 2↓ × 2→			2
Σ	3	3	3	5	

281210: NaAl(Mo ⁶⁺ O ₄) ₂				
	<i>Na</i>	<i>Al</i>	<i>Mo</i>	Σ
<i>O1</i>	0.062 ×2↓	0.500 ×2↓	1.438	2
<i>O2</i>	0.062 ×2↓	0.500 ×2↓	1.438	2
<i>O3</i>	0.063 ×2↓	0.500 ×2↓	1.438	2
<i>O4</i>	0.313 ×2↓		1.688	2
Σ	1	3	6	

281503: BaTe ⁴⁺ Mo ⁶⁺ ₂ O ₉					
	<i>Ba</i>	<i>Te</i>	<i>Mo1</i>	<i>Mo2</i>	Σ
<i>O1</i>		0.829	0.580	0.592	2
<i>O2</i>	0.095	1.071		0.834	2
<i>O3</i>		0.829	0.580	0.592	2
<i>O4</i>	-0.147 ×2↓ ×2→	1.272	1.023		2
<i>O5</i>	0.277 ×2↓ ×2→		1.447		2
<i>O6</i>	0.208 ×3↓ ×3→		1.377		2
<i>O7</i>			0.994	1.006	2
<i>O8</i>	0.409			1.591	2
<i>O9</i>	0.205 ×3↓ ×3→			1.386	2
Σ	2	4	6	6	

400438: $\text{Cu}^{2+}_2\text{Co}^{2+}\text{O}(\text{B}_2\text{O}_5)$						
	<i>Cu1</i>	<i>Cu2</i>	<i>Co</i>	<i>B1</i>	<i>B2</i>	Σ
<i>O1</i>		$0.373 \times 2\downarrow \times 2 \rightarrow$	0.231	1.023		2
<i>O2</i>	0.470	0.604	$0.463 \times 2\downarrow \times 2 \rightarrow$			2
<i>O3</i>	$0.304 \times 3\downarrow \times 3 \rightarrow$			1.089		2
<i>O4</i>		0.412	$0.270 \times 2\downarrow \times 2 \rightarrow$		1.047	2
<i>O5</i>	$0.309 \times 2\downarrow \times 2 \rightarrow$		0.302		1.079	2
<i>O6</i>		0.238		0.888	0.874	2
Σ	2	2	2	3	3	

400764: $\text{SrCo}^{2+}_2(\text{As}^{5+}\text{O}_4)_2$						
	<i>Sr</i>	<i>Co1</i>	<i>Co2</i>	<i>As1</i>	<i>As2</i>	Σ
<i>O1</i>		0.384	0.381		1.235	2
<i>O2</i>		0.273	$0.269 \times 2\downarrow \times 2 \rightarrow$	1.189		2
<i>O3</i>	0.169	$0.305 \times 2\downarrow \times 2 \rightarrow$		1.221		2
<i>O4</i>	0.170	0.306	0.302	1.222		2
<i>O5</i>	0.293	0.428			1.279	2
<i>O6</i>	0.294		0.426		1.280	2
<i>O7</i>	$0.316 \times 2\downarrow \times 2 \rightarrow$			1.368		2
<i>O8</i>	$0.221 \times 2\downarrow \times 2 \rightarrow$		0.352		1.206	2
Σ	2	2	2	5	5	

400802: $\text{KCu}^{2+}_5\text{V}^{5+}_3\text{O}_{13}$										
	<i>K</i>	<i>Cu</i> 1	<i>Cu</i> 2	<i>Cu</i> 3	<i>Cu</i> 4	<i>Cu</i> 5	<i>V</i> 1	<i>V</i> 2	<i>V</i> 3	Σ
01			0.303		0.303	0.242			1.151	2
02		0.450	0.494	0.562	0.494					2
03	0.094	0.475					1.431			2
04	-0.013	0.367			0.412			1.234		2
05		0.341		0.452				1.207		2
06			0.251	0.318	0.251		1.179			2
07	0.078 $\times 3\downarrow \times 3\rightarrow$					0.441		1.325		2
08	0.334		0.273		0.273				1.121	2
09			0.267	0.334		0.205	1.195			2
010	0.091 $\times 2\downarrow \times 2\rightarrow$					0.454			1.364	2
011				0.334	0.267	0.205	1.195			2
012	0.091 $\times 2\downarrow \times 2\rightarrow$					0.454			1.364	2
013	-0.013	0.367	0.412					1.234		2
Σ	1	2	2	2	2	2	5	5	5	

401042: $\text{K}_4(\text{Cu}^{2+}\text{V}^{5+}_5\text{O}_{15}\text{Cl})$					
	<i>K</i>	<i>Cu</i>	<i>V</i> 1	<i>V</i> 2	Σ
01	0.206 $\times 3\downarrow \times 3\rightarrow$		1.381		2
02	0.158 $\times 2\downarrow \times 2\rightarrow$	0.352 $\times 4\downarrow$	1.332		2
03	0.025 $\times 2\downarrow \times 2\rightarrow$		1.200	0.750 $\times 4\downarrow$	2
04	-0.087 $\times 2\rightarrow$		1.087 $\times 2\rightarrow$		2
05				2.000	2
<i>Cl</i>	0.102 $\times 4\rightarrow$	0.296 $\times 2\downarrow \times 2\rightarrow$			1
Σ	1	2	5	5	

401951: Zn ₅ Mn ⁴⁺ (BO ₃) ₂ O ₄						
	Zn1	Zn2	Zn3	Zn4/Mn	B	Σ
01		0.288 ×4↓ ×2→		0.427	0.988	2
02	0.427	0.425 ×2↓		0.574 ×2↓ ×2→		2
03	0.326 ×2↓ ×2→		0.322 ×2↓		1.025	2
04	0.343		0.339 ×4↓ ×2→	0.490 ×2↓ ×2→		2
05	0.289 ×2↓ ×2→			0.436	0.987	2
Σ	2	2	2	3	3	

405153: Ba ₂ (Nb ⁵⁺ ₂ Te ⁶⁺ O ₁₀)				
	Ba	Nb	Te	Σ
01	0.190 ×2↓ ×2→	0.810 ×2↓ ×2→		2
02	0.175	0.796	1.029 ×2↓	2
03	0.131 ×2↓ ×2→	0.752	0.985 ×2↓	2
04	0.131 ×2↓ ×2→	0.752	0.985 ×2↓	2
05	0.460 ×2↓ ×2→	1.080		2
Σ	2	5	6	

405329: PbCu ²⁺ (Cu ²⁺ Te ⁶⁺ O ₇)					
	Pb1	Cu1	Cu2	Te	Σ
01		0.321 ×2→	0.372	0.986	2
02	0.405	0.515 ×2→	0.565		2
03		0.214 ×2↓ ×4→	0.265	0.879	2
04	0.239 ×2↓	0.348	0.399 ×2↓	1.014 ×2↓	2
05	0.279 ×4↓ ×2→	0.388		1.054 ×2↓	2
Σ	2	2	2	6	

409521: Hg ²⁺ (V ⁵⁺ 2O ₆)			
	Hg	V	Σ
O1	0.167 ×2↓	0.917 ×2↓ ×2→	2
O2	0.417 ×4↓ ×2→	1.167	2
O3		0.667 ×3↓ ×3→	2
Σ	2	5	

409745: Rb ₂ (Cr ⁶⁺ 2O ₇)					
	Ag1	Ag2	Cr1	Cr2	Σ
O1	-0.225		1.131	1.094	2
O2		0.162 ×2↓ ×2→	1.676		2
O3	0.267	0.110	1.623		2
O4	0.215 ×2↓ ×2→		1.571		2
O5	0.249	0.091 ×2↓ ×2→		1.568	2
O6	0.279	0.122		1.599	2
O7		0.261		1.739	2
Σ	1	1	6	6	

411285: KY(W ⁶⁺ O ₄) ₂				
	K	Y	W	Σ
O1	0.191 ×4↓ ×2→	0.470 ×2↓	1.148	2
O2	-0.048 ×2↓	0.230 ×2↓	0.909 ×2↓ ×2→	2
O3	0.122 ×2↓	0.400 ×4↓ ×2→	1.078	2
O4	0.022 ×4↓ ×2→		0.978 ×2↓ ×2→	2
Σ	1	3	6	

413000: Rb ₂ Se ⁴⁺ Mo ⁶⁺ O ₆					
	<i>Rb1</i>	<i>Rb2</i>	<i>Mo</i>	<i>Se</i>	Σ
<i>O1</i>	0.038 ×2↓ ×2→		0.962 ×2↓ ×2→		2
<i>O2</i>	-0.039 ×4↓ ×2→	-0.074 ×2↓	0.885 ×2↓	1.267 ×2↓	2
<i>O3</i>	0.230 ×4↓ ×2→	0.194 ×4↓ ×2→	1.153 ×2↓		2
<i>O4</i>	0.160	0.125 ×3↓ ×3→		1.466	2
Σ	1	1	6	4	

415239: Pd ²⁺ (P ₂ O ₇)				
	<i>Pd1</i>	<i>Pd2</i>	<i>P</i>	Σ
<i>O1</i>	0.400 ×2↓	0.400 ×2↓	1.200	2
<i>O2</i>	0.600 ×2↓		1.400	2
<i>O3</i>			1.000 ×2→	2
<i>O4</i>		0.600 ×2↓	1.400	2
Σ	2	2	5	

415427: Ce ³⁺ Ta ⁵⁺ O ₄			
	<i>Ce</i>	<i>Ta</i>	Σ
<i>O1</i>	0.353	0.824 ×2↓ ×2→	2
<i>O2</i>	0.353	0.824 ×2↓ ×2→	2
<i>O3</i>	0.382 ×3↓ ×3→	0.853	2
<i>O4</i>	0.382 ×3↓ ×3→	0.853	2
Σ	3	5	

415460: Yb(Ta ⁵⁺ O ₄)			
	<i>Yb</i>	<i>Ta</i>	Σ
<i>O1</i>	0.500 ×4↓ ×2→	1.000 ×2↓	2
<i>O2</i>	0.250 ×4↓ ×2→	0.750 ×4↓ ×2→	2
Σ	3	5	

416590: Li(Nb ⁵⁺ U ⁶⁺ O ₆)				
	<i>Li</i>	<i>Nb</i>	<i>U</i>	Σ
<i>O1</i>	0.240 ×2↓ ×2→		1.520	2
<i>O2</i>		0.971	1.029	2
<i>O3</i>		0.647 ×2↓ ×2→	0.706	2
<i>O4</i>		0.627	0.686 ×2↓ ×2→	2
<i>O5</i>		0.627	0.686 ×2↓ ×2→	2
<i>O6</i>	0.260 ×2↓ ×2→	1.480		2
Σ	1	5	6	

417072: RbW ⁵⁺ O(P ₂ O ₇)					
	<i>Rb</i>	<i>W</i>	<i>P1</i>	<i>P2</i>	Σ
<i>O1</i>	0.394 ×2↓ ×2→	1.212			2
<i>O2</i>	-0.030	0.788		1.242	2
<i>O3</i>	-0.030	0.788		1.242	2
<i>O4</i>		0.727	1.273		2
<i>O5</i>		0.727	1.273		2
<i>O6</i>	-0.061	0.758	1.303		2
<i>O7</i>	-0.212		1.152	1.061	2
<i>O8</i>	0.182 ×3↓ ×3→			1.455	2
Σ	1	5	5	5	

Table S2 Values of Δ_{topol} and Δ_{cryst} for 266 transition metal coordination polyhedra taken from 140 crystal structures

	Oxidation state	Coordination number	Δ_{topol}	Δ_{cryst}	ICSD code
Sc	3	6	0.019	0.020	65407
		6	0.067	0.038	200128
		8	0.033	0.102	100278
Ti	3	6	0.095	0.030	72714
		6	0.095	0.042	""
	4	4	0.074	0.107	78842
		5	0.140	0.109	33800
		5	0.098	0.217	72682
		6	0.107	0.102	36608
		6	0.032	0.080	75583
		6	0.055	0.162	""
		6	0.048	0.090	82488
		6	0.235	0.250	280501
		6	0.190	0.295	""
V	3	6	0.067	0.018	59244
		6	0.111	0.030	64634
		6	0.040	0.071	82685
		6	0.072	0.052	""
	4	5	0.480	0.065	64634
		5	0.210	0.186	249142
		6	0.034	0.074	71450
		6	0.222	0.219	72886
		6	0.444	0.123	92317
		6	0.444	0.117	201658
	5	4	0.115	0.033	40312
		4	0.112	0.070	""
		4	0.021	0.156	67726

		4	0.099	0.129	73686
		4	0.115	0.093	""
		4	0.226	0.071	""
		4	0.274	0.154	""
		4	0.135	0.039	171028
		4	0.137	0.069	""
		4	0.091	0.111	400802
		4	0.038	0.040	""
		4	0.114	0.217	""
		4	0.107	0.104	401042
		5	0.250	0.112	50010
		5	0.400	0.179	280775
		5	0.400	0.073	401042
		6	0.167	0.336	40850
		6	0.292	0.314	50010
		6	0.444	0.171	89466
		6	0.444	0.160	""
		6	0.187	0.143	95929
		6	0.361	0.097	""
		6	0.325	0.027	""
		6	0.294	0.129	""
		6	0.305	0.060	""
		6	0.200	0.158	99594
		6	0.167	0.306	409521
Cr	2	5	0.057	0.076	73261
		5	0.020	0.077	280309
		6	0.170	0.118	72302
		6	0.156	0.041	73261
	3	6	0.056	0.065	8269
		6	0.055	0.067	72302
		6	0.091	0.052	""

		6	0.098	0.037	""
		6	0.074	0.038	73261
		6	0.089	0.061	""
		6	0.077	0.068	247056
	6	4	0.250	0.144	59819
		4	0.039	0.045	""
		4	0.053	0.080	""
		4	0.039	0.014	""
		4	0.250	0.164	201793
		4	0.000	0.081	""
		4	0.185	0.074	409745
		4	0.203	0.081	""
Mn	2	6	0.111	0.160	40850
		6	0.111	0.099	85042
		6	0.133	0.183	99594
		6	0.028	0.061	100082
		6	0.028	0.046	""
		6	0.072	0.110	156736
	3	4	0.040	0.092	34392
		5	0.053	0.176	280589
		6	0.077	0.211	24973
		6	0.008	0.211	80430
		6	0.039	0.186	95493
	5	4	0.250	0.097	97525
	7	4	0.000	0.028	89506
Fe	2	6	0.000	0.067	17062
		6	0.000	0.071	""
		6	0.148	0.148	50038
	3	4	0.011	0.075	83285
		4	0.125	0.039	""
		4	0.028	0.029	280902

		5	0.073	0.050	96359
		6	0.111	0.026	83285
		6	0.000	0.003	""
		6	0.114	0.070	96359
		6	0.099	0.066	""
		6	0.067	0.038	157733
		6	0.167	0.090	280902
Co	2	4	0.071	0.072	81473
		4	0.026	0.082	""
		5	0.032	0.070	81473
		6	0.026	0.044	20670
		6	0.063	0.028	82403
		6	0.086	0.063	400328
		6	0.049	0.053	400764
		6	0.053	0.089	""
		8	0.013	0.091	250466
	3	6	0.056	0.126	72872
Ni	2	4	0.016	0.052	14159
		6	0.039	0.024	65476
		6	0.007	0.017	""
		6	0.061	0.031	""
		6	0.035	0.035	""
		6	0.034	0.044	79702
Cu	2	4	0.112	0.055	1292
		4	0.000	0.028	201733
		5	0.060	0.100	1292
		5	0.099	0.033	""
		5	0.078	0.076	50459
		5	0.050	0.133	""
		5	0.051	0.063	""
		5	0.016	0.059	63103

		5	0.019	0.101	65614
		5	0.058	0.122	171028
		5	0.071	0.093	201733
		5	0.087	0.088	400438
		5	0.050	0.063	400802
		5	0.086	0.068	""
		5	0.066	0.091	405329
		6	0.063	0.070	2279
		6	0.084	0.039	67726
		6	0.084	0.041	""
		6	0.096	0.127	171028
		6	0.046	0.161	400438
		6	0.080	0.089	400802
		6	0.080	0.087	""
		6	0.116	0.115	""
		6	0.025	0.143	401042
		6	0.084	0.083	405329
	3	4	0.028	0.021	65237
		4	0.028	0.023	""
		4	0.028	0.018	""
Zn	2	4	0.250	0.277	85735
		5	0.021	0.075	40312
		6	0.077	0.129	85042
		6	0.100	0.055	280902
		6	0.034	0.038	401951
		6	0.061	0.022	""
		6	0.007	0.007	""
Y	3	6	0.040	0.046	85497
		7	0.001	0.044	63103
		7	0.004	0.056	""
		7	0.020	0.077	65614

		8	0.111	0.083	15505
		8	0.085	0.071	""
		8	0.058	0.088	71562
		8	0.072	0.131	411285
		9	0.078	0.110	15505
		10	0.086	0.079	20670
Zr	4	6	0.021	0.054	15545
		6	0.000	0.274	55272
		6	0.000	0.030	65512
		6	0.000	0.041	""
		6	0.221	0.186	100158
		8	0.063	0.040	82488
Nb	4	6	0.000	0.085	88879
		6	0.000	0.085	""
	5	5	0.023	0.094	24819
		5	0.035	0.112	94743
		6	0.061	0.210	33783
		6	0.111	0.095	36626
		6	0.071	0.189	50038
		6	0.182	0.218	""
		6	0.019	0.071	62577
		6	0.227	0.098	""
		6	0.125	0.174	75264
		6	0.030	0.056	79734
		6	0.167	0.204	85735
		6	0.083	0.161	91748
		6	0.096	0.198	100158
		6	0.184	0.178	200854
		6	0.180	0.179	""
		6	0.082	0.212	405153
		6	0.261	0.266	416590

		7	0.078	0.192	33783
		7	0.163	0.133	""
Mo	4	6	0.111	0.023	96454
	5	6	0.389	0.075	69088
		6	0.058	0.159	281197
	6	4	0.250	0.109	65512
		4	0.067	0.040	68279
		4	0.052	0.079	""
		4	0.087	0.134	""
		4	0.099	0.056	247056
		4	0.183	0.178	""
		4	0.000	0.055	280775
		4	0.094	0.047	281210
		5	0.161	0.176	66994
		5	0.232	0.256	79517
		5	0.367	0.159	280066
		6	0.418	0.235	68279
		6	0.134	0.204	79517
		6	0.186	0.329	90110
		6	0.033	0.531	170119
		6	0.152	0.340	171758
		6	0.224	0.261	""
		6	0.322	0.215	280066
		6	0.282	0.214	281503
		6	0.328	0.161	""
		6	0.102	0.496	413000
Ru	4	6	0.034	0.141	33802
	5	6	0.048	0.156	73183
		6	0.067	0.140	97525
Pd	2	4	0.100	0.058	415239
		4	0.100	0.065	""

		4	0.000	0.033	72312
		4	0.000	0.042	""
		4	0.122	0.103	""
		6	0.084	0.085	72312
Cd	2	6	0.026	0.061	35084
		6	0.062	0.020	200743
		8	0.022	0.108	35084
		8	0.036	0.027	35407
Hf	4	6	0.017	0.178	33194
		6	0.003	0.162	""
		6	0.063	0.026	65476
		8	0.000	0.092	59111
Ta	5	6	0.111	0.092	80423
		6	0.183	0.147	85042
		6	0.000	0.060	203232
		6	0.000	0.245	280154
		6	0.013	0.099	415427
		6	0.111	0.195	415460
		7	0.092	0.185	203232
W	5	6	0.056	0.026	8269
		6	0.111	0.046	203048
		6	0.167	0.098	202414
		6	0.126	0.067	417072
	6	4	0.019	0.054	71562
		4	0.071	0.141	78180
		5	0.008	0.086	40249
		5	0.149	0.169	78180
		6	0.204	0.170	68614
		6	0.183	0.239	""
		6	0.122	0.262	79702
		6	0.000	0.319	86144

		6	0.075	0.357	411285
Re	5	6	0.056	0.051	72872
	7	4	0.125	0.084	92317
		4	0.250	0.149	""
		5	0.007	0.235	100571
		6	0.025	0.128	15505
		6	0.040	0.077	92508
Os	7	6	0.000	0.105	49746
	8	4	0.000	0.077	20611
		5	0.146	0.248	20611
		6	0.200	0.320	20540
Ir	4	6	0.002	0.046	33863
		6	0.077	0.193	""
Pt	2	4	0.000	0.010	35407
	4	6	0.032	0.017	35407
		6	0.015	0.030	63103
		6	0.005	0.075	65614
Au	3	4	0.000	0.093	92489
Hg	2	2	0.000	0.203	72312
		3	0.069	0.204	72312
		6	0.000	0.010	1640
		6	0.063	0.066	280292
Average:			0.102	0.113	