

1 *Supplementary information*

2 **Phage display-based homing peptide - daunomycin** 3 **conjugates for selective drug targeting to PANC-1** 4 **Pancreatic Cancer**

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16 **Table S1.** *In vitro* cytotoxic effect of the prepared conjugates on various cell lines

Code	Compounds	Viability ^a (%) at 10 ⁻⁵ M concentration, after 72 h incubation		
		Colo-205	A2058	EBC-1
1	<i>Dau=Aoa-SKAAKN-OH</i>	50.7 ± 1.6	61.5 ± 2.7	77.8 ± 1.0
2	<i>Dau=Aoa-KSKAAKN-OH</i>	51.0 ± 2.5	63.5 ± 4.6	90.2 ± 3.0
3	<i>Dau=Aoa-GFLG-KSKAAKN-OH</i>	59.3 ± 3.6	79.4 ± 17.6	76.2 ± 13.7
4	<i>Dau=Aoa-GFLG-K(Dau=Aoa)SKAAKN-OH</i>	19.8 ± 0.5	29.8 ± 4.5	32.2 ± 4.2
5	<i>Dau=Aoa-GFLG-K(Dau=Aoa-GFLG)SKAAKN-OH</i>	45.8 ± 2.6	61.9 ± 11.6	57.6 ± 11.1

17 ^a Fluorescence intensity data of the treated cells are normalized to the fluorescence intensity data values of the
18 control wells and expressed as percentages. Data are given as mean values ± standard deviation (SD), (n=3).

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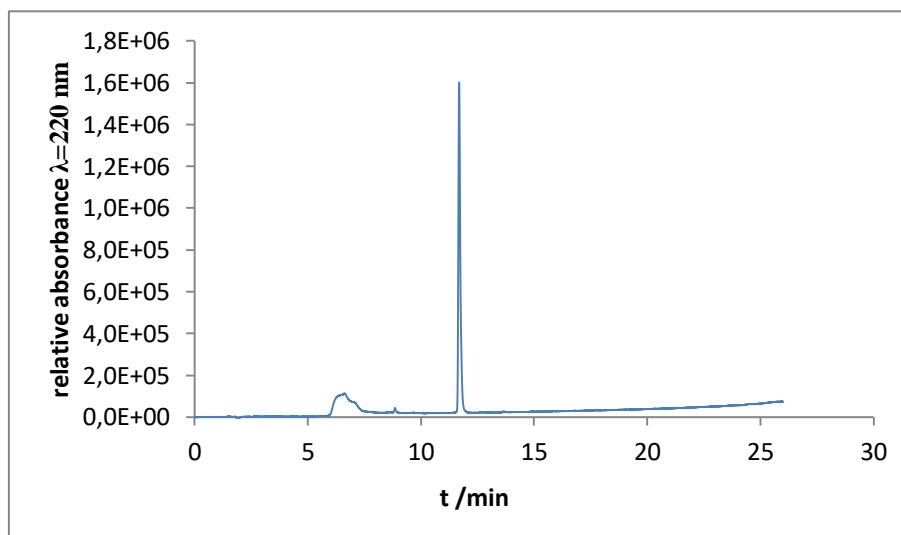
20 **Table S2.** Relative fluorescence intensity results of binding and cellular uptake measurements on
 21 PANC-1 cells

Code	Compounds	Relative fluorescence intensity (GeoMean values) ^a	
		Binding	Cellular uptake
	<i>Daunomycin</i>	72.057 ± 3.907	702.048 ± 84.340
1	<i>Dau=Aoa-SKAAKN-OH</i>	0.086 ± 0.173	1.630 ± 0.356
2	<i>Dau=Aoa-KSKAAKN-OH</i>	0.830 ± 0.157	5.228 ± 0.257
3	<i>Dau=Aoa-GFLG-KSKAAKN-OH</i>	0.012 ± 0.011	0.459 ± 0.189
4	<i>Dau=Aoa-GFLG-K(Dau=Aoa)SKAAKN-OH</i>	0.463 ± 0.147	7.666 ± 1.147
5	<i>Dau=Aoa-GFLG-K(Dau=Aoa-GFLG)SKAAKN-OH</i>	0.071 ± 0.082	2.779 ± 0.575

22 ^a Binding and cellular uptake of the compounds were studied at 10⁻⁵ M concentration. GeoMean value
 23 (geometric mean channel value) is a dimensionless value and refers to the relative fluorescence intensity. Data
 24 shown are mean of 2 parallels ± SD.

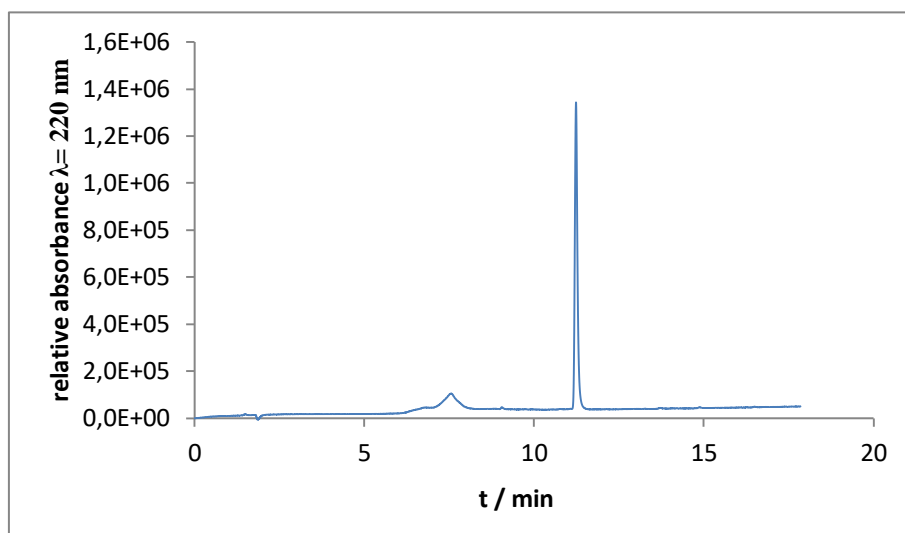
25 Analytical chromatograms of compounds:

26 Analytical RP-HPLC was performed on a Waters Symmetry (WAT 045905) C18 column
 27 (150x4.6mm I.D.) with 5 μm silica (100 Å pore size) as a stationary phase. A linear gradient elution
 28 was developed: 0 min 0% B; 2 min 0% B; 22 min 90% B with eluent A (0.1% TFA in water) and eluent
 29 B (0.1% TFA in acetonitrile-water (80: 20, v/v)). A flow rate of 1 mL/min was used at ambient
 30 temperature. Samples were applied dissolved in eluent A and 20 μl was injected. Peaks were
 31 detected at λ = 220 nm.
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 34 **Figure S1.** Analytical chromatogram of compound 1 (*Dau=Aoa-SKAAKN-OH*)

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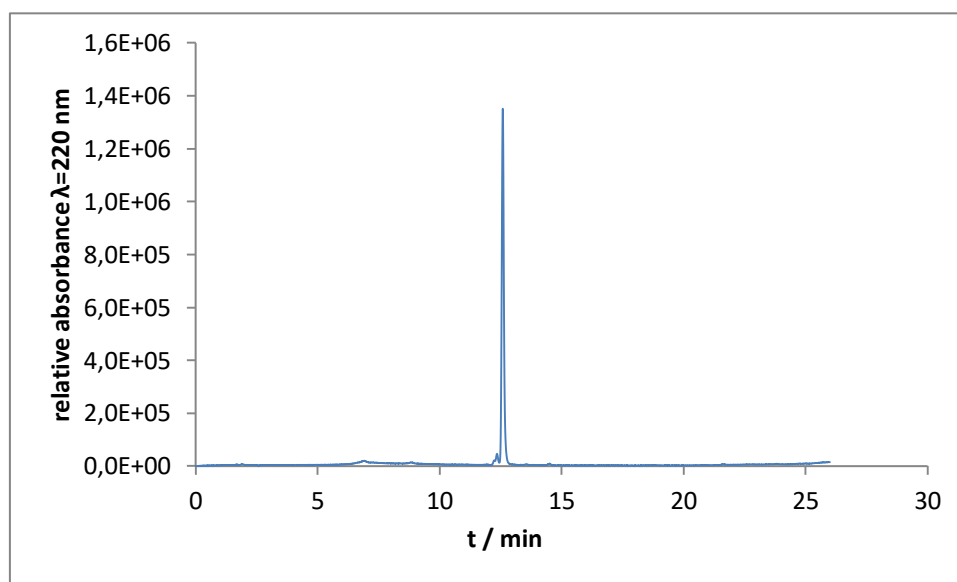


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Figure S2. Analytical chromatogram of compound **2** (*Dau=Aoa-KSKAAKN-OH*)

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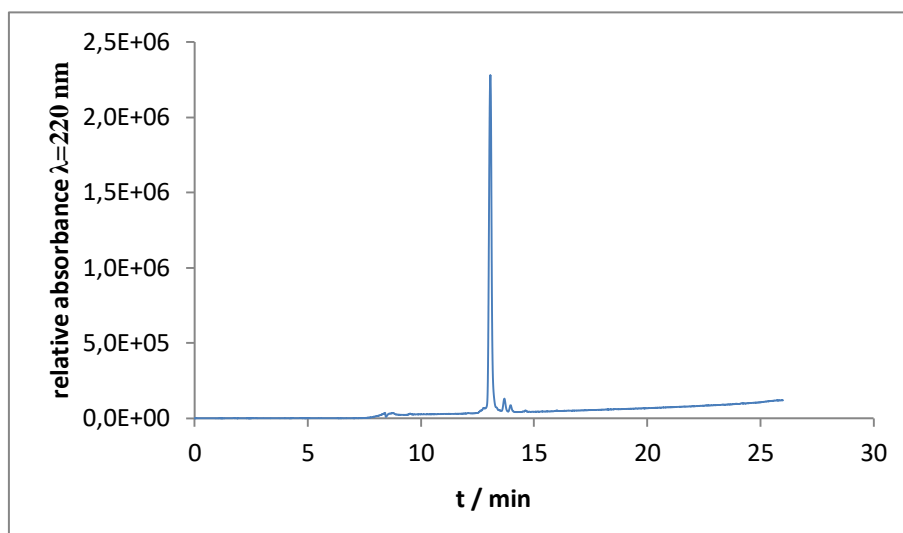


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Figure S3. Analytical chromatogram of compound **3** (*Dau=Aoa-GFLGKSKAAKN-OH*)

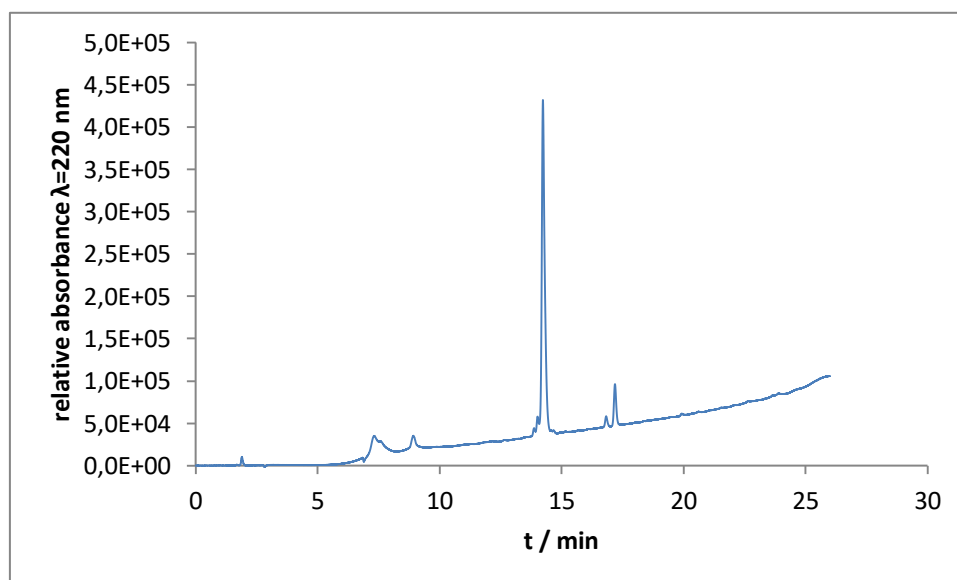
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43 **Figure S4.** Analytical chromatogram of compound **4** (*Dau=Aoa-GFLGK(Dau=Aoa)SKAAKN-OH*)

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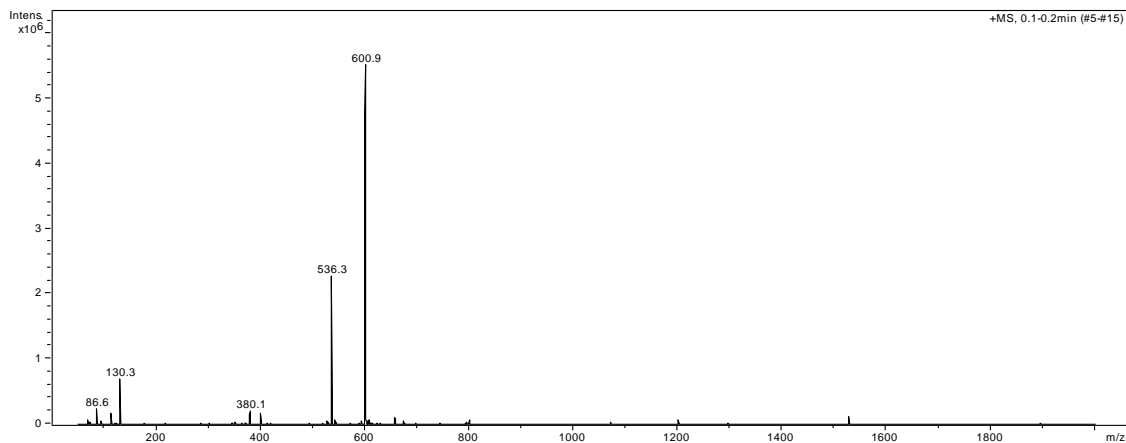
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46 **Figure S5.** Analytical chromatogram of compound
 47 (*Dau=Aoa-GFLG*)K(*Dau=Aoa-GFLG*)SKAAKN-OH)

48 **Mass spectrometry**

49 The identification of the peptide analogues and conjugates was achieved by electrospray
 50 ionization mass spectrometry (ESI-MS) on a Bruker Daltonics Esquire 3000 Plus (Bremen, Germany)
 51 ion trap mass spectrometer, operating in continuous sample injection at 4 $\mu\text{L}/\text{min}$ flow rate. Samples
 52 were dissolved in ACN-water (50:50 v/v%) mixture containing 0.1 v/v% AcOH. Mass spectra were
 53 recorded in positive ion mode in the m/z 50-2000 range.

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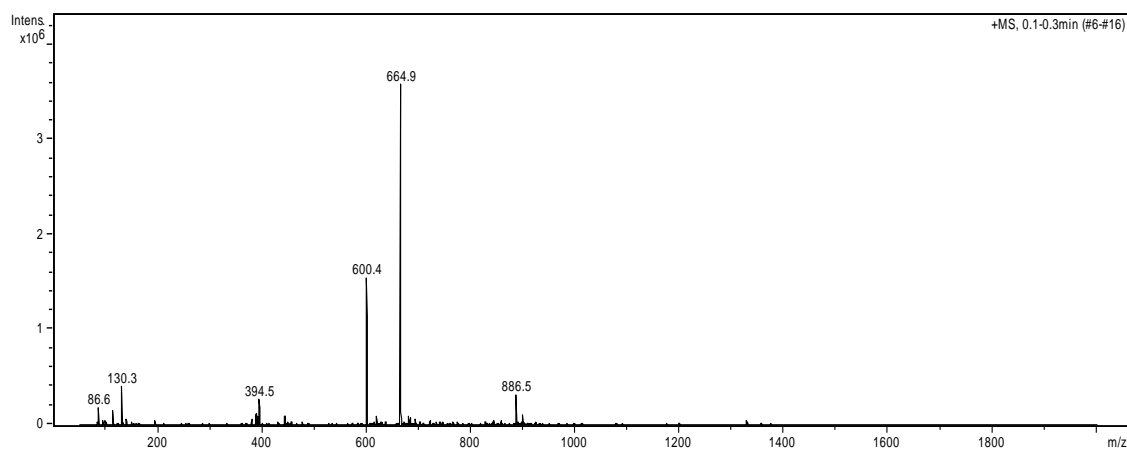


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Figure S6. Mass spectra of compound 1

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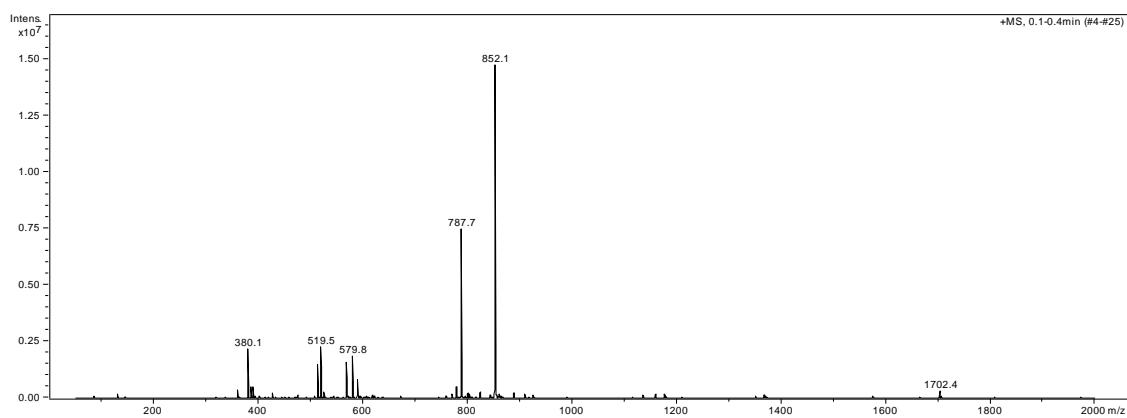


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Figure S7. Mass spectra of compound 2

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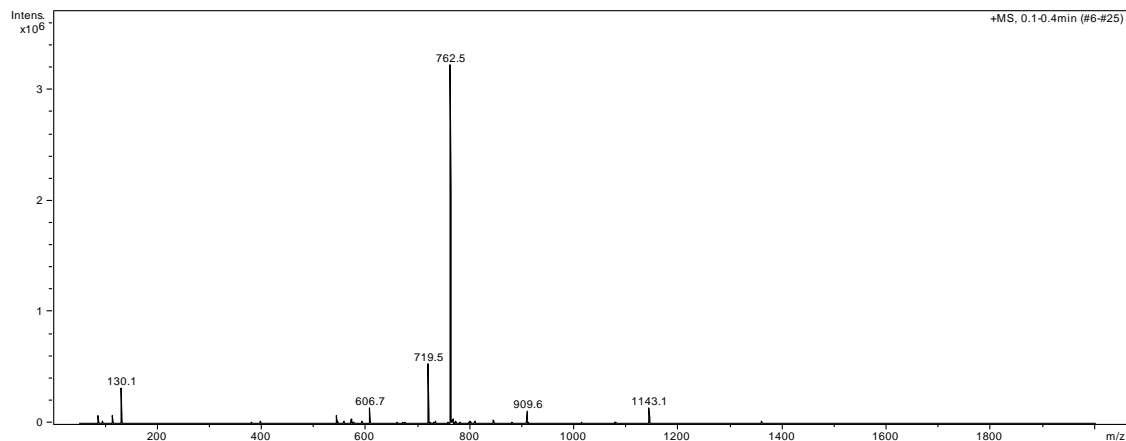


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Figure S8. Mass spectra of compound 3

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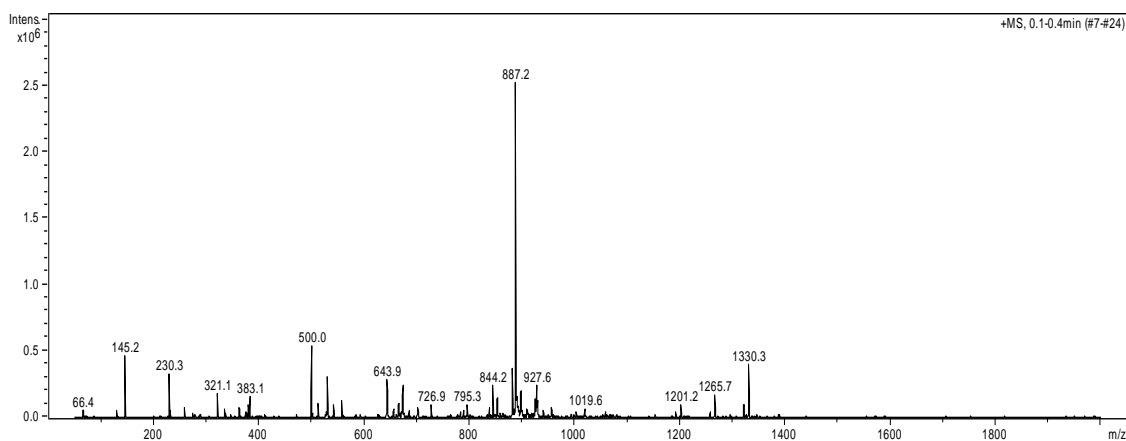


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65 **Figure S9.** Mass spectra of compound 4

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69 **Figure S10:** Mass spectra of compound 5

70 **Chemical characterization of identified fragments of the various conjugates under the enzymatic**
 71 **degradation**

72 **Table S3.** Chemical characterization of identified fragments of conjugate 1

Product of the degradation	R _t (min.)	M _(cal.)	M _(meas.)
<i>Dau=Aoa-S-OH</i>	11.17	687.2276	687.2274
<i>Dau=Aoa-SK-OH</i>	9.99	815.3225	815.3217

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74 **Table S4.** Chemical characterization of identified fragments of conjugate 2

Product of the degradation	R _t (min.)	M _(cal.)	M _(meas.)
<i>Dau=Aoa-K-OH</i>	11.4	728.2905	728.2900
<i>Dau=Aoa-KS-OH</i>	12.5	815.3225	815.3223
<i>Dau=Aoa-KSK-OH</i>	13.1	943.4175	943.4171

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Table S5. Chemical characterization of identified fragments of conjugate 3

Product of the degradation	R _t (min.)	M _(cal.)	M _(meas.)
<i>Dau=Aoa-G-OH</i>	11.4	657.2170	657.2168
<i>Dau=Aoa-GF-OH</i>	12.5	804.2854	804.2855
<i>Dau=Aoa-GFL-OH</i>	13.1	917.3695	917.3690
<i>Dau=Aoa-GFLGK-OH</i>	11.9	1102.4859	1102.4851
<i>Dau=Aoa-GFLGKSK-OH</i>	10.5	1317.6129	1317.6122
<i>Dau=Aoa-GFLGKSKAA-OH</i>	10.5	1459.6871	1459.6864
<i>Dau=Aoa-GFLGKSKAAKN-OH</i>	9.9	1701.8250	1701.8237

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Table S6. Chemical characterization of identified fragments of conjugate 4

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Table S7. Chemical characterization of identified fragments of conjugate 5

Product of the degradation	R _t (min.)	M _(cal.)	M _(meas.)
<i>Dau=Aoa-G-OH</i>	11.4	657.2170	657.2168
<i>Dau=Aoa-GF-OH</i>	12.5	804.2854	804.2855
<i>Dau=Aoa-GFL-OH</i>	13.1	917.3695	917.3690
<i>H-K(Dau=Aoa-GFLG)-OH</i>	11.9	1102.4859	1102.4851
<i>H-K(Dau=Aoa-GFLG)SK-OH</i>	10.5	1317.6129	1317.6122
<i>H-K(Dau=Aoa-GFLG)SKAA-OH</i>	10.5	1459.6871	1459.6864
<i>H-K(Dau=Aoa-GFLG)SKAAKN-OH</i>	9.9	1701.8200	1701.8239
<i>Dau=Aoa-GFLGK(Dau=Aoa-GFLG)SKAAKN-OH</i>	11.1	2658.2054	2658.2037