

## **Supporting Information**

for

# Five-component, one-pot synthesis of an electroactive rotaxane comprising a bisferrocene macrocycle

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Further experimental details and NMR spectra of new compounds

#### **Electrochemistry**

Cyclic voltammetry was carried out in a Metrohm Autolab PGSTAT302N potentiostat using a three-electrode cell, with a glassy carbon working electrode, silver wire counter electrode and Ag/AgCl (3M KCl) reference electrode. The glassy carbon working electrode was polished in a figure of eight fashion using 1, 0.3 and 0.05 gamma alumina on a microcloth pad (Buehler, UK) (for 3.3 and 5 min, respectively). The electrodes were then sonicated in 1:1 ethanol/water for 5 min before being finally washed in stream of dry tetrahydrofuran. The counter electrode was a silver wire which was cleaned by annealing with a Bunsen burner. A Ag/AgCl reference electrode, stored in 3 M KCl, and washed thoroughly before use was used initially, before changing to a Ag wire pseudo-reference electrode, heat-annealed with a Bunsen burner and quenched with dry DCM.

The supporting electrolyte was a solution of 0.1M TBAPF<sub>6</sub> containing 1 mM ferrocene (to act as an internal reference), dissolved in dry CH<sub>3</sub>CN, and degassed with argon for 5 min. The rotaxane **1a** (4.0 mg) was dissolved in dry dichloromethane (0.8 mL), sonicated for 5 min, then diluted in acetonitrile (4 mL) to give a final concentration of 0.67 mM. Tetrabutylammonium hexafluorophosphate (0.10 M) was used as supporting electrolyte. Samples where degassed with argon for 5 min prior to measurement. In analogous conditions ferrocene was used as internal reference.

### NMR spectra

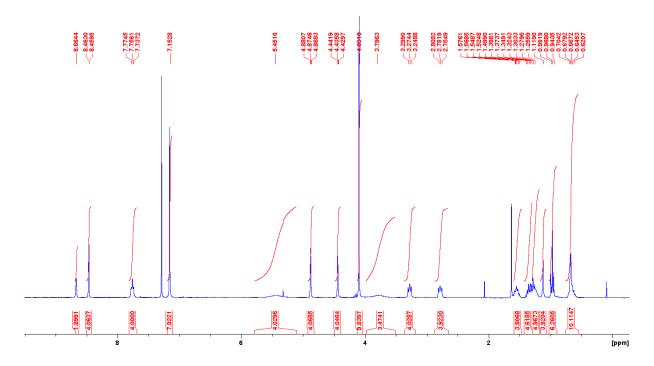
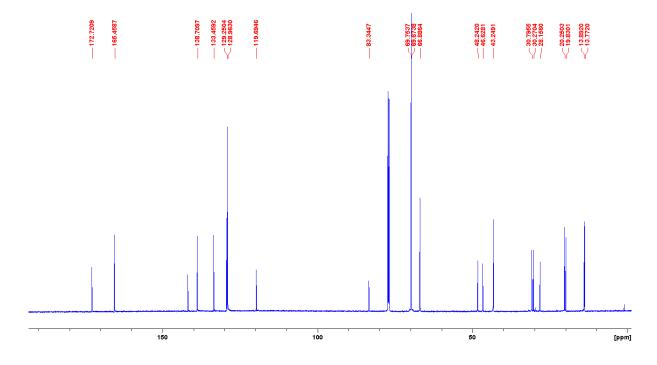


Figure S1: <sup>1</sup>H-NMR of ferrocene-containing clipped [2]rotaxane 1a (300 MHz in CDCl<sub>3</sub>).



**Figure S2:** <sup>13</sup>C-NMR of ferrocene-containing clipped [2]rotaxane **1a** (150 MHz in CDCl<sub>3</sub>).

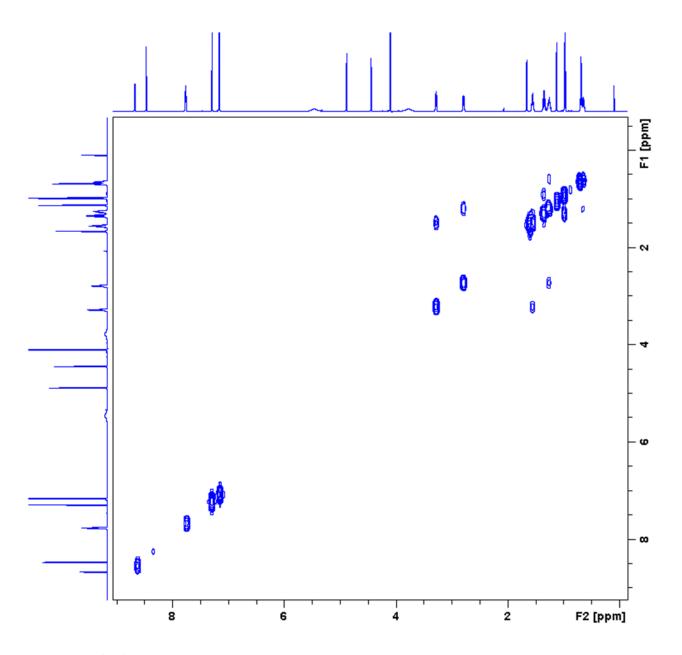


Figure S3: <sup>1</sup>H-<sup>1</sup>H COSY NMR of ferrocene-containing clipped [2]rotaxane 1a (600 MHz in CDCl<sub>3</sub>).

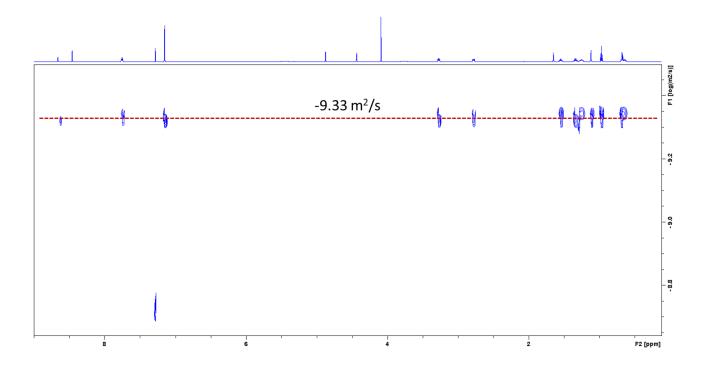


Figure S4: <sup>1</sup>H DOSY NMR of ferrocene-containing clipped [2]rotaxane 1a (600 MHz in CDCl<sub>3</sub>).

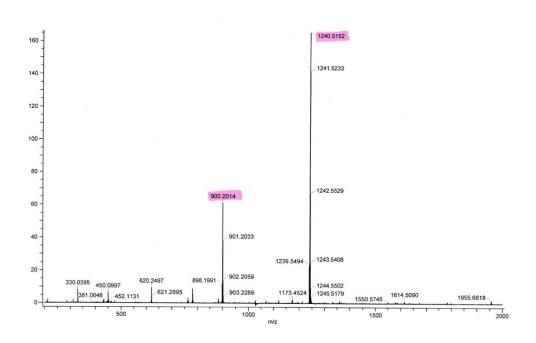


Figure S5: Mass spectrum (FD) of ferrocene-containing clipped [2]rotaxane 1a.

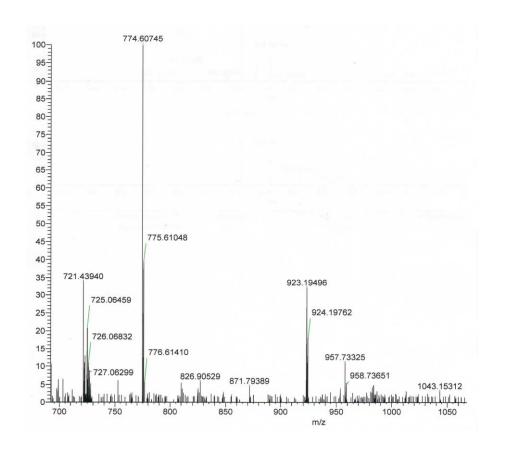


Figure S6: Mass spectrum (FD) of ferrocene macrocyle 2.

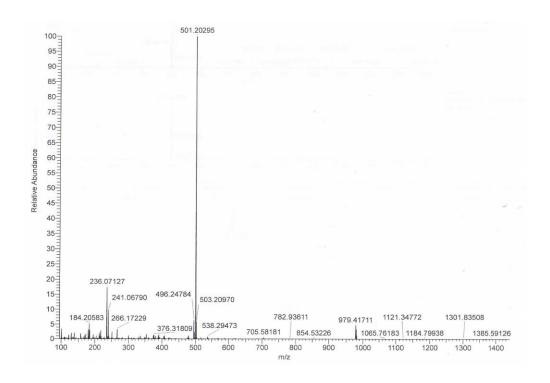


Figure S7: Mass spectrum (ESI) of double ester thread 4c.

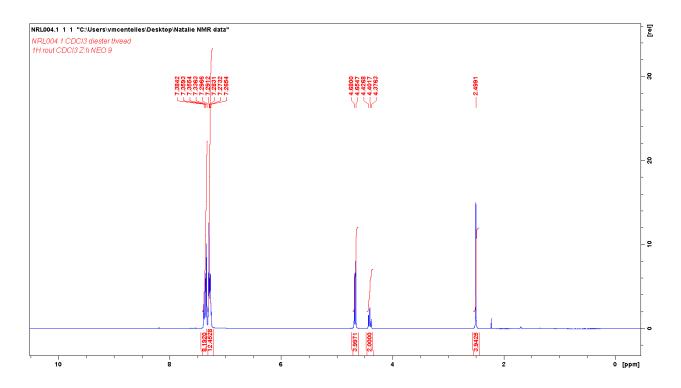


Figure S8: <sup>1</sup>H NMR of double ester thread 4c (300 MHz in CDCl<sub>3</sub>).

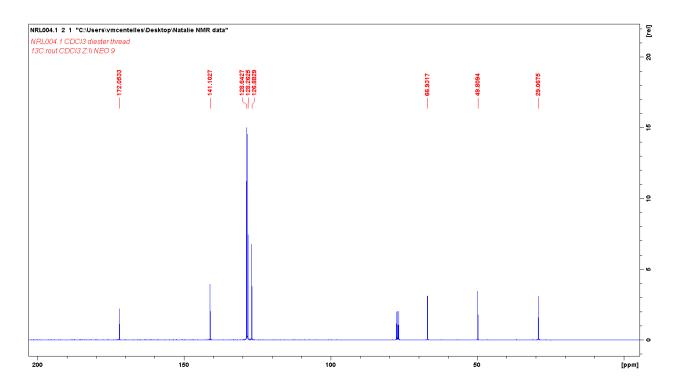
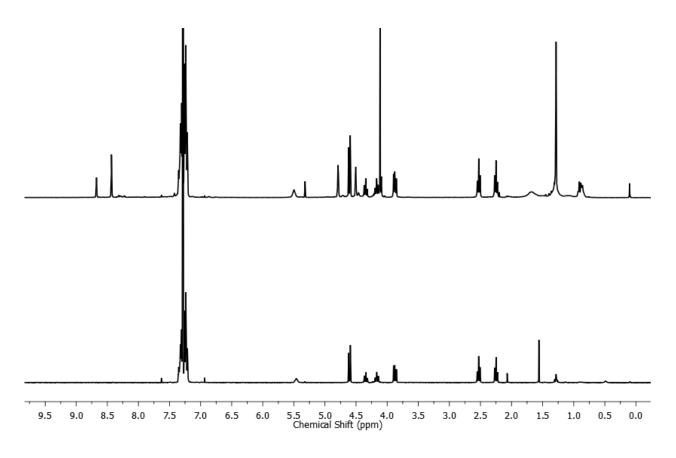


Figure S9: <sup>13</sup>C NMR of double ester thread 4c (75 MHz in CDCl<sub>3</sub>).

#### Synthesis attempt of rotaxane 1b

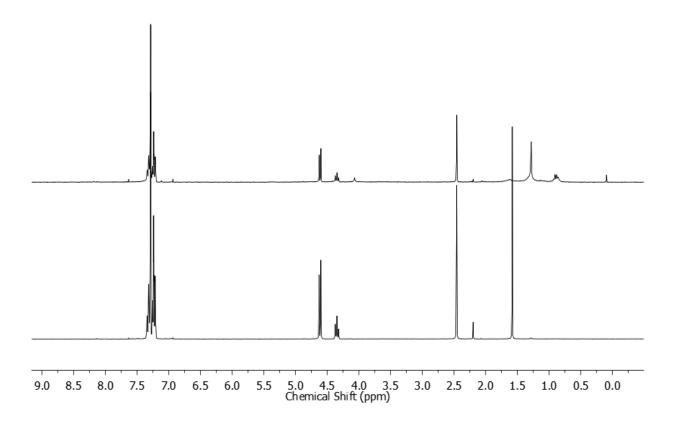
A solution of p-xylylenediamine (**6**, 68 mg, 0.5 mmol) in chloroform (20 mL) and a solution of 5-ferrocenylisophthaloyl chloride (**5**, 200 mg, 0.5 mmol) in chloroform (20 mL) were simultaneously added with a syringe pump over 5 hours to a solution of unsymmetrical ester amide thread **4b** (122 mg, 0.25 mmol) and dry triethylamine (0.35 mL, 2.6 mmol) in chloroform (20 mL), and the solution was stirred at room temperature overnight. After this time, DCM (25 mL) was added to the reaction mixture followed by filtration through Celite. The filtrate was washed with sodium carbonate solution 10% (2 × 50 mL). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and the solvent was removed in vacuo. Analysis of the crude  $^{1}$ H NMR allowed estimation of a < 5% yield of the rotaxane.



**Figure S10:** <sup>1</sup>H NMR of ester amide thread **4b** (bottom) and crude reaction mixture of the rotaxane formation reaction (top) (300 MHz in CDCl<sub>3</sub>).

#### Synthesis attempt of rotaxane 1c

A solution of *p*-xylylenediamine (**6**, 17 mg, 0.12 mmol) in chloroform (10 mL) and a solution of 5-ferrocenylisophthaloyl chloride (**5**, 50 mg, 0.12 mmol) in chloroform (10 mL) were simultaneously added with a syringe pump during 4 hours to a solution of thread **4c** (30 mg, 0.06 mmol) and dry triethylamine (86 μL, 0.6 mmol) in chloroform (10 mL). The reaction mixture was stirred at room temperature overnight. After this time, DCM (25 mL) was added to the suspension followed by filtration through Celite. The filtrate was washed with sodium carbonate solution 10% (2 × 25 mL). The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and the solvent removed in vacuo. It was not possible to identify any proton signals corresponding to the rotaxane in the crude <sup>1</sup>H NMR.



**Figure S11:** <sup>1</sup>H NMR of double ester thread **4c** (bottom) and crude reaction mixture of the rotaxane formation reaction (top) (300 MHz in CDCl<sub>3</sub>).

#### Synthesis attempts of macrocycle 2

#### Templated macrocyclization

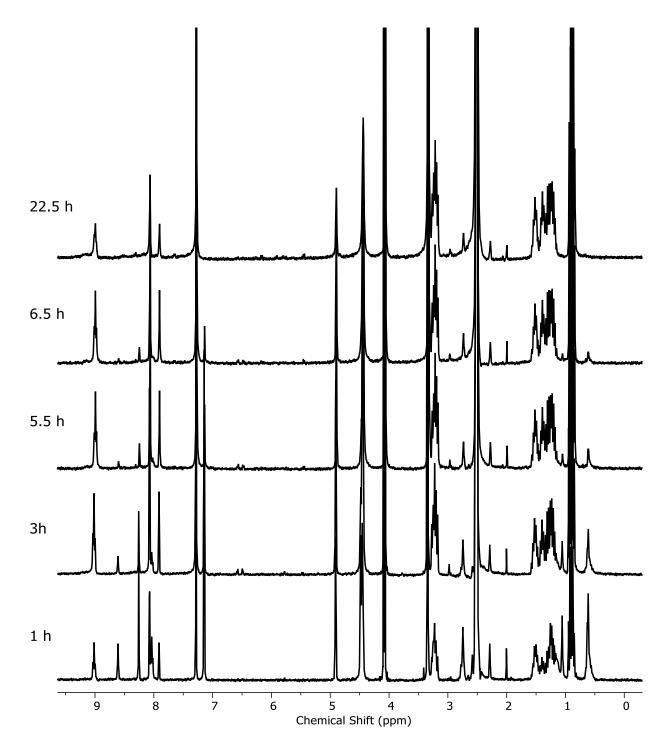
A solution of *p*-xylylenediamine (**6**, 17.6 mg, 0.129  $\mu$ mol) in chloroform (10 mL) and a solution of 5-ferrocenylisophthaloyl chloride (**5**, 50.0 mg, 0.129  $\mu$ mol) in chloroform (15 mL) were simultaneously added with a syringe pump during 3 hours to a solution of the *N*,*N*'-dihexyl-1,4-butanediamide thread **3** (36.7 mg, 0.129  $\mu$ mol) and dry triethylamine (49  $\mu$ L, 0.323  $\mu$ mol) in chloroform (10 mL). The reaction mixture was stirred at room temperature overnight. The reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> (25 mL) and filtered through Celite. The filtrate was washed with saturated NaHCO<sub>3</sub> (10 mL) and the solvent removed in vacuo. Purification by silica gel column chromatography (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 0% to 10%, v/v) afforded impure macrocycle **2** as yellow solid (17.0 mg, < 29% yield).

#### Macrocycle dethreading

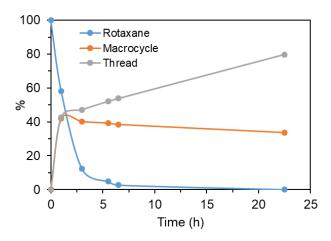
Rotaxane **1a** (30 mg, 0.024 mmol) was suspended in DMSO- $d_6$  (2.5 mL) and heated in a sand bath (160 °C, 25 hours) with stirring. As the temperature of the suspension exceeded 150 °C, the solid began to dissolve. After 4 hours it was fully dissolved and remained dissolved upon cooling to rt. After 6 hours, the color of the solution began to darken and after 25 hours the solution was dark brown and black material precipitated upon cooling. This reaction was monitored by <sup>1</sup>H NMR until completion (Figure S12). Concentration of species was estimated from the integrals of the <sup>1</sup>H NMR spectra (Figure S13). The analysis of the <sup>1</sup>H NMR at 22.5 h of reaction showed the presence of free macrocycle (Figure S14). In this reaction mixture the macrocycle could be identified by HRMS–FD (m/z): [M + Na]<sup>+</sup> calcd for C<sub>52</sub>H<sub>44</sub>N<sub>4</sub>O<sub>4</sub>Fe<sub>2</sub>Na, 923.19536; found, 923.19496.

A similar experiment was performed, where rotaxane 1a (25.7 mg, 0.023 mmol) was suspended in DMSO- $d_6$  (0.75 mL) and heated for a shorter time (6 h). Ice was added to the resulting orange solution and the orange precipitate was filtered via gravity filtration, washed with water (10 mL), and left to dry. The solid

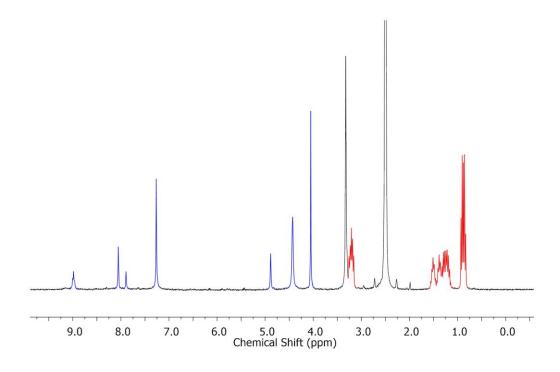
was then dissolved in dichloromethane, dried over  $Na_2SO_4$ , filtered and the solvent was removed in vacuo. The resulting solid was analyzed by  $^1H$  NMR showing only starting material.



**Figure S12:**  $^{1}$ H NMR of the dethreading reaction of rotaxane **1a** (300 MHz in DMSO- $d_{6}$ ) at 1 h, 3 h, 5.5 h, 6.5 h, and 22.5 h.

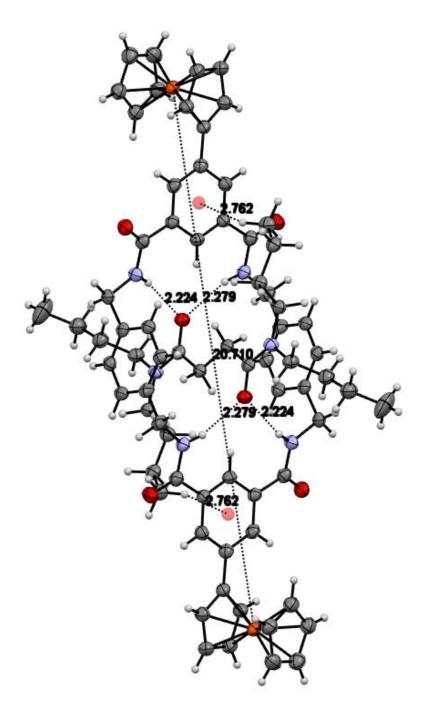


**Figure S13:** Relative distribution of species in the dethreading reaction of rotaxane **1a** estimated from the crude <sup>1</sup>H NMR integrals.



**Figure S14:** <sup>1</sup>H NMR of the dethreading reaction of rotaxane **1a** (300 MHz in DMSO-*d*<sub>6</sub>) after 22.5 h of heating 160 °C showing NMR signals of free macrocycle **2** (blue) and free thread **4a** (red).

## X-Ray data



**Figure S15:** Selected distances in the solid-state structure of rotaxane **1a**: macrocycle-thread hydrogen bond distances, Fe–Fe distance, and thread CH<sub>3</sub> to isophthalamide centroid.

 Table S1: Crystal data and structure refinement for rotaxane 1a.

CCDC Deposition Number	1968472
Empirical formula	$C_{36}H_{42}FeN_3O_3$
Formula weight	620.57
Temperature/K	120
Crystal system	triclinic
Space group	P-1
a/Å	9.5366(5)
b/Å	11.6279(7)
c/Å	14.9396(9)
α/°	99.002(5)
β/°	108.001(5)
γ/°	90.007(4)
Volume/Å <sup>3</sup>	1554.09(16)
Z	2
$\rho_{\rm calc} g/{\rm cm}^3$	1.326
$\mu/\mathrm{mm}^{-1}$	4.208
F(000)	658.0
Crystal size/mm <sup>3</sup>	$0.05\times0.01\times0.01$
Radiation	$CuK\alpha (\lambda = 1.54184)$
2Θ range for data collection/	° 9.124 to 111.358
Index ranges	$-10 \le h \le 9$ , $-12 \le k \le 12$ , $-15 \le l \le 15$
Reflections collected	14633
Independent reflections	$3928 [R_{int} = 0.0450, R_{sigma} = 0.0402]$
Data/restraints/parameters	3928/0/391
Goodness-of-fit on F <sup>2</sup>	1.041
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0454$ , $wR_2 = 0.1116$
Final R indexes [all data]	$R_1 = 0.0556$ , $wR_2 = 0.1180$
Largest diff. peak/hole / e Å	3 0.28/-0.41

## XYZ coordinates of the optimized structures

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Н	-1.516512	-2.039524	4.919922
Н	-0.154469	1.959003	4.211855
Н	2.520458	-1.341950	3.556194
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Н	-3.214154	1.761465	-6.015258
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