

## Supporting Information

### Sustainable synthesis of a potent and selective 5-HT<sub>7</sub> receptor antagonist using a mechanochemical approach

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## I. Calculation of green metrics

Different parameters can be used to evaluate the environmental impact of the different synthetic pathways. This part details the calculation of the E Factor (EF) [1] and Ecoscale score [2].

### A. Synthetic methods

Following methods were used for the calculation of the E Factor, and the Ecoscale score. Procedures in solution are described according to our previously reported publication [3].

#### Alkylation of 2-phenylphenol in solution

2-Phenylphenol (1 g, 5.87 mmol, 1 eq) and previously grinded  $K_2CO_3$  (2.03 g 14.68 mmol, 2.5 eq), KOH (0.16 g, 2.94 mmol, 0.5 eq) were suspended in 20 ml of acetone, Next, epichlorohydrin (0.57 mL, 11.74 mmol, 2 eq) was slowly dropped to the mixture which was stirred at 60°C for 24 hours. Then, the inorganic residues were filtered off and the mixture concentrated under vacuum followed by re-solubilization in  $CH_2Cl_2$  (20 ml) and washed with 2N NaOH aqueous solution (3 x 7 mL) and saturated NaCl solution (1 x 7 mL). The organic phase was dried over  $Na_2SO_4$ , filtered and concentrated under reduced pressure. The obtained crude product was purified on silica gel with ethyl acetate/hexane (1/9 v/v) as eluting system yielding intermediate **1a** as white powder (820 mg, yield 62%).

#### Alkylation of 2-phenylphenol in ball-mill

2-Phenylphenol (80.7 mg, 0.474 mmol, 1 eq) and previously grinded  $K_2CO_3$  (196.6 mg, 1.42 mmol, 3 eq) were introduced in a 35 mL PTFE jar with one stainless steel ball ( $\varnothing_{ball}$  = 1.5 cm) followed by addition of epichlorohydrin (44.6  $\mu$ L, 0.569 mmol, 1.2 eq) and 2-propanol (50  $\mu$ L,  $\eta$  = 0.15  $\mu$ L·mg<sup>-1</sup>). The reaction was carried out for 140 minutes at rt. The mixture was then solubilized in  $CH_2Cl_2$  (15 mL) and the organic phase was washed with 2N NaOH aqueous solution (3 x 5 mL), saturated NaCl solution (1 x 5 mL), dried over  $Na_2SO_4$  and finally filtered and concentrated under reduced pressure yielding intermediate **1a** as white powder (91 mg, yield 85%).

#### Alkylation of 4-Boc-N-aminopiperidine in solution

4-Boc-N-aminopiperidine (0.85 g, 4.24 mmol, 1.2 eq) was dissolved in EtOH (15 mL), followed by addition of intermediate **1a** (0.8 g, 3.54 mmol, 1 eq) and the mixture was stirred under reflux for 4 hours. Inorganic residues were filtered off and organic mixture was concentrated under reduced pressure. The obtained crude product was purified using silica gel column chromatography with  $CH_2Cl_2/MeOH$  (9/1 v/v) as an eluting system yielding intermediate **3a** as white powder (1.17 g, yield 78%).

#### Alkylation of 4-Boc-N-aminopiperidine in ball-mill

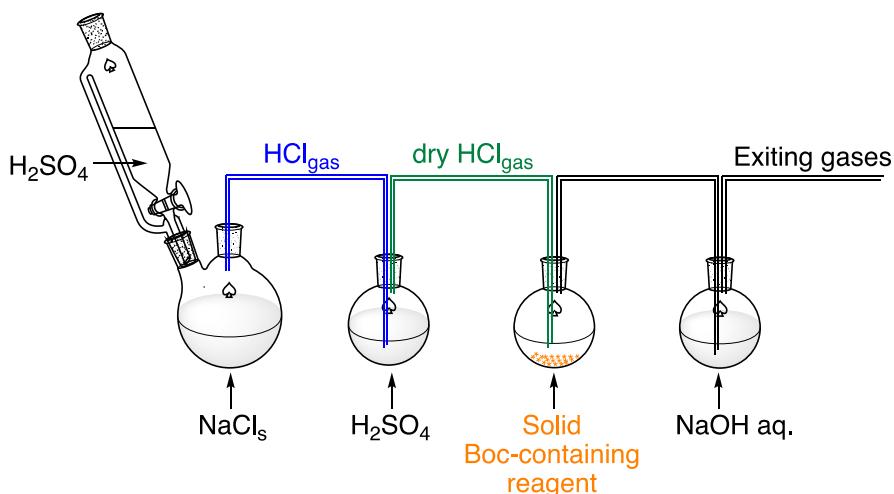
Intermediate **1a** (175.1 mg, 0.774 mmol, 1 eq) and 4-Boc-amino-piperidine (154.9 mg, 0.774 mmol, 1 eq) were introduced in a 35 mL PTFE jar with one stainless steel ball ( $\varnothing_{ball}$  = 1.5 cm) followed by addition of EtOH (33  $\mu$ L,  $\eta$  = 0.1  $\mu$ L mg<sup>-1</sup>) as liquid assistant. The reaction was carried out for 70 minutes at rt. Then, the product was solubilized in ethyl acetate (20 mL) and the organic phase was washed with  $KHSO_4$  aqueous solution at pH = 3.5 (3 x 7 mL), saturated NaCl solution (1 x 7 mL), dried over  $Na_2SO_4$  and finally filtered and concentrated under reduced pressure yielding intermediate **3a** as white powder (297 mg, yield 90%).

### General procedure for deprotection of Boc function in solution

To a solution of the intermediate **3a** (1 g, 2.35 mmol) in EtOH (35 mL), 12 mL of HCl 2N in EtOH were added. The mixture was left stirring for 6 hours at rt. The obtained white powder **4** was filtrated under vacuum and left drying overnight (832 mg, yield 98%).

### General procedure for deprotection of Boc function in solid state

Intermediate **3a** (0.5 g, 1.16 mmol) was submitted to HCl<sub>gas</sub> for 2 h at r.t., according to Scheme S1, to afford the primary amine **4** as a white hydrochloride salt (0.4 g, yield 98%).



**Scheme S1.** Experimental set-up for HCl<sub>gas</sub> production and use in solid-state reaction

### Sulfonylation of primary amine in solution

A mixture of the obtained primary amine **4** (300 mg, 0.92 mmol, 1 eq) in CH<sub>2</sub>Cl<sub>2</sub> (4 mL), and TEA (0.39 mL, 2.75 mmol, 3 eq) was cooled down (ice bath), and 3-chlorobenzenesulfonyl chloride (154 µL, 1.1 mmol, 1.2 eq) was added in one portion. The reaction mixture was stirred for 2 hours under cooling. Then, the solvent was evaporated, and the crude mixture was purified using silica gel column chromatography with CH<sub>2</sub>Cl<sub>2</sub>/MeOH (9/0.5 v/v) as an eluting system yielding final compound **5a** as white powder (330 mg, yield 72%).

### Sulfonylation of primary amine in ball-mill

Intermediate **4** (168.2 mg, 0.463 mmol, 1 eq), 3-chlorobenzenesulfonyl chloride (65.24 µL, 0.463 mmol, 1 eq), and previously grinded K<sub>2</sub>CO<sub>3</sub> (64.0 mg, 0.463, 1 eq) were introduced in a 35 mL PTFE jar with one stainless steel ball ( $\phi_{\text{ball}} = 1.5 \text{ cm}$ ). The reaction was carried out for 1 min at rt. Then, the crude mixture was solubilized in ethyl acetate (20 mL) and the organic phase was washed with KHSO<sub>4</sub> aqueous solution at pH = 3.5 (3 x 7 mL), saturated NaCl solution (1 x 7 mL), dried over Na<sub>2</sub>SO<sub>4</sub> and finally filtered and concentrated under vacuum yielding final compound **5a** (**PZ-1361**) as white powder (200.1 mg, yield 86%).

## B. Calculation of E Factor

The E Factor is the ratio of the weight of generated waste to the total weight of the end product. It is a useful tool for rapid evaluation of processes based on generated waste.

$$EF = \frac{\sum m(\text{Raw materials}) + \sum m(\text{solvent}) + \sum m(\text{water}) - m(\text{products})}{m(\text{products})}$$

E factor values are reported in the following tables for the synthesis of **PZ-1361** (final compound **5a**) in the ball-mill and magnetic stirrer conditions. A summary is given in Table S1.

**Table S1.** Summary of E factor for the synthesis of **PZ-1361**

	E factor			
	Step 1 Alkylation of 2-Phenylphenol	Step 2 Alkylation of 4-Boc-N- aminopiperidine	Step 3 Boc deprotection	Step 4 Sulfonylation
<b>Solution</b>	509	554	46	804
<b>Solid-state</b>	548	<b>181</b>	<b>2</b>	<b>230</b>

The E factor values were also calculated for the whole synthesis of PZ-1361, starting from 1 g of 2-Phenylphenol. All the quantities were then calculated according to the equivalents and yield obtained in each step. The following values were calculated:

E factor	
<b>Solution</b>	1932
<b>Solid-state</b>	<b>715</b>

### Alkylation of 2-phenylphenol in solution

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
2-phenylphenol	90-43-7	C12H10O	170.21	0.999		5.87		1
acetone	67-64-1	C3H6O	58.08	15.820	20.000		0.791	
potassium carbonate	584-08-7	K2CO3	138.21	2.03		14.68		2.5
potassium hydroxide	1310-58-3	KOH	56.11	0.165		2.935		0.5
epichlorhydrin	106-89-8	C3H5ClO	92.52	1.086	0.918	11.74	1.183	2
<i>Extraction</i>								
dichloromethane	75-09-2	CH2Cl2	84.93	26.500	20.000		1.325	
sodium hydroxide	1310-73-2	NaOH	40	1.680		42		
sodium chloride	7647-14-5	NaCl	58.44	1.82				
water	7732-18-5	H2O	18	26.18		1454.44		
<i>Purification</i>								
ethyl acetate	141-78-6	C4H8O2	88.11	45.10	50	511.86	0.902	
hexane	110-54-3	C6H14	86.18	296.55	450	3441.05	0.659	

product	m (g)	M (g/mol)	yield
alkylated 2-phenylphenol <b>1a</b>	0.82	226.28	0.62

E-factor			m (g)
	Product	alkylated 2-phenylphenol <b>1a</b>	0.820
E-factors	Reagents	2-phenylphenol acetone potassium carbonate potassium hydroxide epichlorhydrin dichloromethane sodium hydroxide sodium chloride water ethyl acetate hexane	0.999 15.820 2.028 0.165 1.086 26.500 1.680 1.820 26.180 45.100 296.55
E-factor =		417.11	/ 0.820 = 509

### Alkylation of 2-phenylphenol in ball-mill

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
2-phenylphenol	90-43-7	C12H10O	170.21	0.081		0.474		1
potassium carbonate	584-08-7	K2CO3	138.21	0.20		1.42		3
epichlorhydrin	106-89-8	C3H5ClO	92.52	0.053	0.044	0.5688	1.183	1.2
<i>iso</i> propanol	67-63-0	C3H8O	60.1	0.039	0.050		0.785	
<i>Extraction</i>								
dichloromethane	75-09-2	CH2Cl2	84.93	19.875	15.000		1.325	
sodium hydroxide	1310-73-2	NaOH	40	1.680		42		
sodium chloride	7647-14-5	NaCl	58.44	1.82				
water	7732-18-5	H2O	18	26.18		1454.444		

product	m (g)	M (g/mol)	yield
alkylated 2-phenylphenol	0.091	226.28	0.85

E-factor			m (g)
	Product	alkylated 2-phenylphenol <b>1a</b>	0.091
E-factors	Reagents	2-phenylphenol potassium carbonate epichlorhydrin <i>iso</i> propanol dichloromethane sodium hydroxide sodium chloride water	0.081 0.197 0.053 0.039 19.875 1.680 1.820 26.180
E-factor =		49.83	/ 0.091 = 548

### Alkylation of 4-Boc-N-aminopiperidine in solution

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
alkylated 2-phenylphenol <b>1a</b>			226.28	0.801		3.54		1
ethanol	64-17-5	C2H6O	46.07	11.835	15.0		0.789	
4-boc-N-aminopiperidine	73874-95-0	C10H20N2O2	200.28	0.85		4.25		1.2
Purification								
dichloromethane	75-09-2	CH2Cl2	84.93	596.250	450.0	7020.5	1.325	
methanol	67-56-1	CH4O	32.04	39.550	50.0	1234.4	0.791	

product	m (g)	M (g/mol)	yield
amination product <b>3a</b>	1.17	426.56	0.77

E-factor	Product	m (g)	
		amination product <b>3a</b>	1.170
E-factors	Reagents	alkylated 2-phenylphenol <b>1a</b>	0.801
		ethanol	11.835
		4-boc-N-aminopiperidine	0.851
		dichloromethane	596.250
		methanol	39.550
E-factor =		648.12	/ 1.170 = 554

### Alkylation of 4-Boc-N-aminopiperidine in ball-mill

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
alkylated 2-phenylphenol <b>1a</b>			226.28	0.175		0.774		1
ethanol	64-17-5	C2H6O	46.07	0.026	0.033		0.789	
4-boc-N-aminopiperidine	73874-95-0	C10H20N2O2	200.28	0.155		0.774		1
Extraction								
ethyl acetate	141-78-6	C4H8O2	88.11	18.04	20	204.74	0.902	
potassium bisulfate	7646-93-7	KHSO4	136.17	0.0002860		0.00210		
sodium chloride	7647-14-5	NaCl	58.44	1.82				
water	7732-18-5	H2O	18	26.18		1454.444		

product	m (g)	M (g/mol)	yield
amination product <b>3a</b>	0.297	426.56	0.90

E-factor	Product	m (g)	
		amination product <b>3a</b>	0.297
E-factors	Reagents	alkylated 2-phenylphenol <b>1a</b>	0.053
		ethanol	7.902
		4-boc-N-aminopiperidine	0.047
		ethyl acetate	18.040
		potassium bisulfate	0.000286
		sodium chloride	1.820
		water	26.180
E-factor =		53.74	/ 0.297 = 181

### Deprotection of Boc function in solution

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
Intermediate <b>3a</b>			426.56	1.002		2.35		1
ethanol	64-17-5	C2H6O	46.07	37.083	47.0		0.789	
Hydrochloric acid 2N in EtOH	7647-01-0	HCl	36.46	0.88		24.00		10.2

product	m (g)	M (g/mol)	yield
Intermediate <b>4</b>	0.832	362.9	0.98

E-factor	Product	m (g)	
		Amine <b>4</b>	0.832
E-factors	Reagents	Intermediate <b>3a</b>	1.002
		ethanol	37.083
		Hydrochloric acid 2N in EtOH	0.875
E-factor =		38.13	/ 0.832 = 46

*Deprotection of Boc function in solid state*

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
Intermediate 3a			426.56	0.495		1.16		1
Hydrochloric acid gas	7647-01-0	HCl	36.46	0.43		11.83		10.2

product	m (g)	M (g/mol)	yield
Intermediate 4	0.412	362.9	0.98

E-factor			m (g)			
	Product	Amine 4	0.412			
	Reagents	Intermediate 3a	1.002			
	E-factor =	1.02	/	0.412	=	2

*Sulfonylation of primary amine in solution*

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
amine 4			362.9	0.334		0.92		1
dichloromethane	75-09-2	CH <sub>2</sub> Cl <sub>2</sub>	84.93	5.300	4.000		1.325	
triethylamine	121-44-8	(C <sub>2</sub> H <sub>5</sub> ) <sub>3</sub> N	101.19	0.28	0.385	2.76	0.726	3
3-chlorobenzenesulfonyl chloride	#2888-06-4	C <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> Cl <sub>2</sub>	211.07	0.233	0.155	1.104	1.499	1.2
Purification								
CH <sub>2</sub> Cl <sub>2</sub>	75-09-2	CH <sub>2</sub> Cl <sub>2</sub>	84.93	251.750	190.000		1.325	
MeOH	67-56-1	CH <sub>4</sub> O	32.04	7.920	10		0.792	

product	m (g)	M (g/mol)	yield
final compound 5a (PZ-1361)	0.33	501.04	0.72

E-factor			m (g)			
	Product	final compound 5a	0.330			
	Reagents	amine 4 dichloromethane triethylamine 3-chlorobenzenesulfonyl chloride CH <sub>2</sub> Cl <sub>2</sub> MeOH	0.300 5.300 0.279 0.233 251.750 7.920			
	E-factor =	265.45	/	0.330	=	804

*Sulfonylation of primary amine in ball-mill*

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
amine <b>4</b>			362.9	0.168		0.463		1
potassium carbonate	584-08-7	K <sub>2</sub> CO <sub>3</sub>	138.21	0.064		0.46		1
3-chlorobenzenesulfonyl chloride	#2888-06-4	C <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> Cl <sub>2</sub>	211.07	0.098	0.065	0.463	1.499	1
<i>Extraction</i>								
ethyl acetate	141-78-6	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	88.11	18.04	20	204.74	0.902	
potassium bisulfate	7646-93-7	KHSO <sub>4</sub>	136.17	0.0002860		0.00210		
sodium chloride	7647-14-5	NaCl	58.44	1.82				
water	7732-18-5	H <sub>2</sub> O	18	26.18		1454.444		

product	m (g)	M (g/mol)	yield
final compound <b>5a</b> (PZ-1361)	0.2001	501.04	0.86

E-factor	Product	m (g)	
		final compound <b>5a</b>	0.200
	Reagents	amine <b>4</b>	0.046
		potassium carbonate	0.019
		3-chlorobenzenesulfonyl chlo	0.030
		ethyl acetate	18.040
		potassium bisulfate	0.000286
		sodium chloride	1.820
		water	26.180
E-factor =		45.93	/ 0.200 = 230

## Total synthesis of PZ-1361(5a) in solution

### Total synthesis of PZ-1361

Step 1 Alkylation								
Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
2-phenylphenol	90-43-7	C12H11O	170.21	0.999		5.87		1
acetone	67-64-1	C3H6O	58.08	15.819	19.999	272.37	0.791	46.4
potassium carbonate	584-08-7	K2CO3	138.21	2.03		14.68		2.5
potassium hydroxide	1310-58-3	KOH	56.11	0.165		2.935		0.5
epichlorohydrin	106-89-8	C3H5ClO	92.52	1.086	0.918	11.74	1.183	2
<i>Extraction</i>								
dichloromethane	75-09-2	CH2Cl2	84.93	26.522	20.000	312.28	1.325	53.2
sodium hydroxide	1310-73-2	NaOH	40	1.691		42.26		7.2
sodium chloride	7647-14-5	NaCl	58.44	1.82		31.11		5.3
water	7732-18-5	H2O	18	26.18		1454.59		247.8
<i>Purification</i>								
ethyl acetate	141-78-6	C4H8O2	88.11	45.10	50	511.86	0.902	87.2
hexane	110-54-3	C6H14	86.18	296.54	450	3440.99	0.659	586.2
Step 2 Amination								
Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
alkylated 2-phenylphenol 1a	64-17-5	C2H6O	226.28	0.824		3.64		1
ethanol	73874-95-0	C10H20N2O2	46.07	12.168	15.0	264.11	0.789	72.57
4-boc-N-aminopiperidine			200.28	0.87		4.37		1.2
<i>Purification</i>								
dichloromethane	75-09-2	CH2Cl2	84.93	612.996	450.0	7217.7	1.325	1983.2
methanol	67-56-1	CH4O	32.04	40.661	50.0	1269.1	0.791	348.7
Step 3 Boc removal								
Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
Intermediate 3a			426.56	1.211		2.84		1
ethanol	64-17-5	C2H6O	46.07	44.795	47.0	972.32	0.789	342.52
Hydrochloric acid 2N in EtOH	7647-01-0	HCl	36.46	1.06		28.96		10.2
Step 4 Sulfonylation								
Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
amine 4			362.9	1.010		2.78		1
dichloromethane	75-09-2	CH2Cl2	84.93	16.019	4.000	188.617	1.325	67.8
triethylamine	121-44-8	(C2H5)3N	101.19	0.84	1.163	8.35	0.726	3
3-chlorobenzenesulfonyl chloride	#2888-06-4	C6H4SO2Cl2	211.07	0.705	0.470	3.33843883	1.499	1.2
<i>Purification</i>								
CH2Cl2	75-09-2	CH2Cl2	84.93	761.267	190.000	8963.467	1.325	3222.0
MeOH	67-56-1	CH4O	32.04	23.950	10	747.512	0.792	268.7
product m (g) M (g/mol) yield								
final compound 5a (PZ-1361)	1.000	501.04	0.34					
E-factor								
Product	final compound 5a (PZ-1361)	1.000						
Reagents	2-phenylphenol	0.999						
	acetone	15.819						
	potassium carbonate	2.028						
	potassium hydroxide	0.165						
	epichlorohydrin	1.086						
	dichloromethane	26.522						
	sodium hydroxide	1.691						
	sodium chloride	1.818						
	water	26.183						
	ethyl acetate	45.100						
	hexane	296.54						
	ethanol	12.168						
	4-boc-N-aminopiperidine	0.875						
	dichloromethane	612.996						
	methanol	40.661						
	ethanol	44.795						
	Hydrochloric acid 2N in EtOH	1.056						
	dichloromethane	16.019						
	triethylamine	0.845						
	3-chlorobenzenesulfonyl chlo	0.705						
	CH2Cl2	761.267						
	MeOH	23.950						
E-factor =	1932.29	/	1.000	=	1932			

product	m (g)	M (g/mol)	yield
alkylated 2-phenylphenol 1a	0.82	226.28	0.62

product	m (g)	M (g/mol)	yield
amination product 3a	1.21	426.56	0.78

product	m (g)	M (g/mol)	yield
Intermediate 4	1.01	362.9	0.98

product	m (g)	M (g/mol)	yield
final compound 5a (PZ-1361)	1.00	501.04	0.72

## Total synthesis of PZ-1361(5a) in solid state

### Total synthesis of PZ-1361

#### Step 1

##### Alkylation

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
2-phenylphenol	90-43-7	C12H10O	170.21	0.081		0.474		1
potassium carbonate	584-08-7	K2CO3	138.21	0.20		1.4220		3
epichlorohydrin	106-89-8	C3H5ClO	92.52	0.053	0.044	0.5688	1.183	1.2
iso propanol	67-63-0	C3H8O	60.1	0.039	0.050		0.785	
<i>Extraction</i>								
dichloromethane	75-09-2	CH2Cl2	84.93	19.875	15.000	234.0138	1.325	493.7
sodium hydroxide	1310-73-2	NaOH	40	1.680		41.9964		88.6
sodium chloride	7647-14-5	NaCl	58.44	1.82		31.1418		65.7
water	7732-18-5	H2O	18	26.18		1454.4690		3068.5

product	m (g)	M (g/mol)	yield
alkylated 2-phenylphenol 1a	0.091	226.28	0.85

#### Step 2

##### Amination

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
alkylated 2-phenylphenol 1a			226.28	0.091		0.402		1
ethanol	64-17-5	C2H6O	46.07	0.014		0.294	0.789	0.73
4-boc-N-aminopiperidine	73874-95-0	C10H20N2O2	200.28	0.081		0.402		1
<i>Extraction</i>								
ethyl acetate	141-78-6	C4H8O2	88.11	9.37	10.391	106.37	0.902	264.5
potassium bisulfate	7646-93-7	KHSO4	136.17	0.0001486		0.0011		0.002713
sodium chloride	7647-14-5	NaCl	58.44	0.945		16.17		40.2
water	7732-18-5	H2O	18	13.602		755.69		1879.1

product	m (g)	M (g/mol)	yield
amination product 3a	0.154	426.56	0.90

#### Step 3

##### Boc removal

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
Intermediate 3a			426.56	0.154		0.362		1
Hydrochloric acid gas	7647-01-0	HCl	36.46	0.135		3.69		10.2

product	m (g)	M (g/mol)	yield
Intermediate 4a	0.129	362.9	0.98

#### Step 4

##### Sulfonylation

Reagents	CAS	Molecular Formula	M(g/mol)	m(g)	V (mL)	n (mmoles)	d	eq
amine 4			362.9	0.129		0.355		1
potassium carbonate	584-08-7	K2CO3	138.21	0.049		0.355		1
3-chlorobenzenesulfonyl chloride	#2888-06-4	C6H4ClSO2Cl	211.07	0.075	0.050	0.355	1.499	1
<i>Extraction</i>								
ethyl acetate	141-78-6	C4H8O2	88.11	13.82	15.3	156.85	0.902	442.2
potassium bisulfate	7646-93-7	KHSO4	136.17	0.0002191		0.00161		0.004536
sodium chloride	7647-14-5	NaCl	58.44	1.40		23.871		67.3
water	7732-18-5	H2O	18	20.057		1114.261		3141.4

product	m (g)	M (g/mol)	yield
final compound 5a (PZ-1361)	0.153	501.04	0.64

E-factor	Product	m (g)
	alkylated 2-phenylphenol 1a	0.153
Reagents	2-phenylphenol	0.081
	potassium carbonate	0.197
	epichlorohydrin	0.053
	iso propanol	0.039
	dichloromethane	19.875
	sodium hydroxide	1.680
	sodium chloride	1.820
	water	26.180
	ethanol	0.014
	4-boc-N-aminopiperidine	0.081
	ethyl acetate	9.372
	potassium bisulfate	0.000149
	sodium chloride	0.945
	water	13.602
	HCl	0.135
	potassium carbonate	0.049
	3-chlorobenzenesulfonyl chlo	0.075
	ethyl acetate	13.820
	potassium bisulfate	0.000219
	sodium chloride	1.395
	water	20.057
E-factor =		109.32
		/ 0.153
		= 715

### C. Calculation of Ecoscale score

According to Van Aken *et al.*: "The Ecoscale tool uses a scale from 0 to 100 with 0 representing a totally failed reaction (0% yield) and 100 representing the ideal reaction which is defined as follows: Compound **A** (substrate) undergoes a reaction with (or in the presence of) inexpensive compound(s) **B** to give the desired compound **C** in 100% yield at room temperature with a minimal risk for the operator and a minimal impact on the environment. Six general parameters which influence the quality of reaction conditions are analyzed. Within each of these parameters, individual penalty points of various relative weights are assigned that take into account all possible situations when setting up an organic chemistry experiment. The penalty points are cumulative for all components of the preparation. In order to simplify the EcoScale design, the usual differentiation between solvents, reagents, auxiliary or co-reagents and catalysts is not made."

Ecoscale can be summarized by this simple equation:  $\text{EcoScale} = 100 - \text{sum of individual penalties.}$ "[4]

Ranking of reaction conditions is defined as follow: On a scale from 0 to 100 using the following scores: > 75 is excellent; > 50 is acceptable and < 50 is inadequate.

Below, Ecoscale score tables using magnetic stirrer or ball-mill are displayed, using the web version <http://ecoscale.cheminfo.org/calculator> of this green metric. Quantities used were obtained from the protocols reported above. As suggested on the web-calculator, penalties due to hazard should be limited to a maximum score of 20. Thus, even if this penalty number was found higher on the calculator, we corrected it to 20 (corrections in red in the following pictures).

**Table S2.** Summary of Ecoscale score calculations for the synthesis of **PZ-1361**

ECOSCALE SCORE				
	Step 1	Step 2	Step 3	Step 4
	Alkylation of 2-Phenylphenol	Alkylation of 4-Boc-N- aminopiperidine	Boc deprotection	Sulfonylation
<b>Solution</b>	44	58	70	49
<b>Solid-state</b>	71	71	81	78

## Alkylation of 2-phenylphenol in solution

Reagents											
<input checked="" type="checkbox"/> Link identifier* name MF* MW density purity* ml g mmoles equiv. 1 + - 2-Phenylphenol C12H10O 170.2108 1.21 100% 0.825733 0.999137 5.87 1  2 + - Potassium carbonate CK2O3 138.2058 100% 0 2.02817 14.675 2.5  3 + - Potassium hydroxide HKO 56.10564 100% 0 0.16467 2.935 0.5  4 + - Epichlorohydrin C3H5ClO 92.5251 1.18 100% 0.920547 1.086245 11.74 2  5 + - Acetone C3H6O 58.08004 0.79 100% 20 15.8 272.03831 46.34384 											
Products											
identifier*: name: MF*: MW: g: mmoles: g theor: yield: alkylated 2-phenylphenol 226.28 0.82 3.623828 1.328264 61.7347											
Conditions											
Reagents	Name	mmoles	eq.	Bp	Hazard	Price					
	2-Phenylphenol	7.15	1	282							
	Potassium carbonate	17.89	2.5								
	Potassium hydroxide	3.57	0.5		 						
	Epichlorohydrin	14.31	2	115	 						
	Acetone	331.75	46.34	56	 						
Yield	62				-19						
Price / availability					0						
Safety					-20						
Technical setup	Possible items Any additional special glassware (Inert) gas atmosphere Glove box			Selected items Instruments for controlled addition of chemicals		-1					
Temperature / time	Possible items Room temperature, < 1h Room temperature, < 24h Heating, < 1h			Selected items Heating, > 1h		-3					
Workup and purification	Possible items Sublimation Liquid - liquid extraction or washing Classical chromatography			Selected items Adding solvent Liquid - liquid extraction or washing Classical chromatography		-13					
EcoScale					44						

## Alkylation of 2-phenylphenol in ball-mill

Reagents											
<input checked="" type="checkbox"/> Link	identifier*	name	MF*	MW	density	purity*	ml	g	mmoles	equiv.	
1 <input checked="" type="checkbox"/>	+ -	2-Phenylphenol	C12H10O	170.2108	1.21	100%	0.066678	0.08068	0.474	1	 
2 <input checked="" type="checkbox"/>	+ -	Potassium carbonate	CK2O3	138.2058		100%	0	0.196529	1.422	3	 
3 <input checked="" type="checkbox"/>	+ -	Epichlorohydrin	C3H5ClO	92.5251	1.18	100%	0.0446	0.052628	0.5688	1.2	 
4 <input checked="" type="checkbox"/>	+ -	Isopropanol	C3H8O	60.09592	0.785	100%	0.05	0.03925	0.6531225	1.3778956	  
Products											
	identifier*: name:		MF*:	MW:	g:	mmoles:	g theor:	yield:			
		alkylated 2-phenylphenol			226.28	0.091	0.4021566	0.107257	84.8429		
Conditions											
Reagents	Name	mmoles	eq.	Bp	Hazard	Price					
2-Phenylphenol	5.2	1	282								
Potassium carbonate	15.62	3									
Epichlorohydrin	6.25	1.2	115								
Isopropanol	7.17	1.37	81								
Yield	85					-8					
Price / availability						0					
Safety						-15					
Technical setup	Possible items INSTRUMENTS FOR CONTROLLED ADDITION OF CHEMICALS Unconventional activation technique Pressure equipment, > 1 atm Adv. additional general glassware			Selected items Unconventional activation technique		-2					
Temperature / time	Possible items Room temperature, < 1h Room temperature, < 24h Heating, < 1h			Selected items Room temperature, < 24h		-1					
Workup and purification	Possible items Cooling to room temperature Adding solvent Simple filtration			Selected items Liquid - liquid extraction or washing Adding solvent		-3					
EcoScale						71					

## Alkylation of 4-Boc-N-aminopiperidine in solution

Reagents										
Link										
1	[+]	[+]	Name, MF or identifier*	alkylated 2-phenylphenol	MF*	226.28	density	100%	ml	g
2	[+]	[+]	name:	4-N-Boc-Aminopiperidine	MW	200.281	purity*	0	0.801031	3.540000C
3	[+]	[+]	name:	Ethanol	C10H20N2O	46.06904	0.79	100%	15	11.85
										1
Products										
			identifier*:	name:	MF*:	MW:	g:	mmoles:	g theor:	yield:
				amination product 3			426.56	1.17	2.7428732	1.510022 77.48230C
Conditions										
Reagents		Name	mmoles	eq.	Bp	Hazard	Price			
		alkylated 2-phenylphenol	3.02	1						
		4-N-Boc-Aminopiperidine	3.63	1.2						
		Ethanol	219.84	72.66						
Yield		77							-11	
Price / availability									-8	
Safety									-10	
Technical setup		Possible items	Selected items				Score			
		Any additional special glassware (Inert) gas atmosphere Glove box	Common set-up				0			
Temperature / time		Possible items	Selected items				Score			
		Heating, > 1h Cooling to 0°C Cooling, < 0°C	Heating, > 1h				-3			
Workup and purification		Possible items	Selected items				Score			
		Sublimation Liquid - liquid extraction or washing Classical chromatography	Classical chromatography Simple filtration Cooling to room temperature				-10			
EcoScale							58			

## Alkylation of 4-Boc-N-aminopiperidine in ball-mill

Reagents												
Link												
1	[+]	[+]	identifier*	name	MF*	MW	density	purity*	ml	g	mmoles	equiv.
1	[+]	[+]	Name, MF or alkylated 2-phenylphenol 1a		226.28		100%	0	0.175141	0.774	1	 
2	[+]	[+]	4-N-Boc-Aminopiperidine	C10H20N2O	200.281		100%	0	0.155017	0.774	1	 
3	[+]	[+]	Ethanol	C2H6O	46.06904	0.79	100%	0.033	0.02607	0.565889	0.7311237	  
Products												
			identifier*:	name:	MF*:	MW:	g:	mmoles:	g theor:	yield:		
				amination product 3a		426.56	0.297	0.696267	0.330157	89.9572		
Conditions												
Reagents		Name	mmoles	eq.	Bp	Hazard	Price					
		alkylated 2-phenylphenol 1a	2.6	1			 					
		4-N-Boc-Aminopiperidine	2.6	1			 					
		Ethanol	1.9	0.73		 						
Yield		90						-5				
Price / availability								-8				
Safety								-10				
Technical setup		Possible items	Selected items									
		Common set-up	Unconventional activation technique				-2					
		Instruments for controlled addition of chemicals										
		Unconventional activation technique										
Temperature / time		Possible items	Selected items									
		Room temperature, < 1h	Room temperature, < 24h				-1					
		Room temperature, < 24h										
		Heating, < 1h										
Workup and purification		Possible items	Selected items									
		Sublimation	Liquid - liquid extraction or washing				-3					
		Liquid - liquid extraction or washing										
		Classical chromatography										
EcoScale							71					

## General procedure for deprotection of Boc function in solution

Reagents																																																											
<input checked="" type="checkbox"/> Link <table border="1"> <thead> <tr> <th></th> <th>identifier*</th> <th>name</th> <th>MF*</th> <th>MW</th> <th>density</th> <th>purity*</th> <th>mL</th> <th>g</th> <th>mmoles</th> <th>equiv.</th> <th></th> </tr> </thead> <tbody> <tr> <td>1</td> <td></td> <td>Intermediate 3a</td> <td></td> <td>426.56</td> <td></td> <td>100%</td> <td>0</td> <td>1.002416</td> <td>2.35</td> <td>1</td> <td></td> </tr> <tr> <td>2</td> <td></td> <td>Hydrochloric acid</td> <td>HCl</td> <td>36.46094</td> <td></td> <td>100%</td> <td>0</td> <td>0.873969</td> <td>23.97</td> <td>10.2</td> <td></td> </tr> <tr> <td>3</td> <td></td> <td>Ethanol</td> <td>C2H6O</td> <td>46.06904</td> <td>0.79</td> <td>100%</td> <td>47</td> <td>37.13</td> <td>805.96426</td> <td>342.96351</td> <td></td> </tr> </tbody> </table>													identifier*	name	MF*	MW	density	purity*	mL	g	mmoles	equiv.		1		Intermediate 3a		426.56		100%	0	1.002416	2.35	1		2		Hydrochloric acid	HCl	36.46094		100%	0	0.873969	23.97	10.2		3		Ethanol	C2H6O	46.06904	0.79	100%	47	37.13	805.96426	342.96351	
	identifier*	name	MF*	MW	density	purity*	mL	g	mmoles	equiv.																																																	
1		Intermediate 3a		426.56		100%	0	1.002416	2.35	1																																																	
2		Hydrochloric acid	HCl	36.46094		100%	0	0.873969	23.97	10.2																																																	
3		Ethanol	C2H6O	46.06904	0.79	100%	47	37.13	805.96426	342.96351																																																	
<b>Products</b> <table border="1"> <thead> <tr> <th>identifier*</th> <th>name:</th> <th>MF*:</th> <th>MW:</th> <th>g:</th> <th>mmoles:</th> <th>g theor:</th> <th>yield:</th> </tr> </thead> <tbody> <tr> <td></td> <td>Intermediate 4</td> <td></td> <td>362.9</td> <td>0.832</td> <td>2.2926426</td> <td>0.852815</td> <td>97.559300</td> </tr> </tbody> </table>												identifier*	name:	MF*:	MW:	g:	mmoles:	g theor:	yield:		Intermediate 4		362.9	0.832	2.2926426	0.852815	97.559300																																
identifier*	name:	MF*:	MW:	g:	mmoles:	g theor:	yield:																																																				
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Conditions																																																											
Reagents	Name	mmoles	eq.	Bp	Hazard	Price																																																					
	Intermediate 3a	2.82	1																																																								
	Hydrochloric acid	28.81	10.19	57																																																							
	Ethanol	968.7	342.96																																																								
Yield	98					-1																																																					
Price / availability						-8																																																					
Safety						-20																																																					
Technical setup	Possible items	Selected items																																																									
	Common set-up	Common set-up						0																																																			
	Instruments for controlled addition of chemicals																																																										
	Unconventional activation technique																																																										
Temperature / time	Possible items	Selected items																																																									
	Room temperature, < 1h	Room temperature, < 24h						-1																																																			
	Room temperature, < 24h																																																										
	Heating, < 1h																																																										
Workup and purification	Possible items	Selected items																																																									
	Removal of solvent using solvent	Simple filtration						0																																																			
	Simple filtration																																																										
	Removal of solvent with bp < 150°C																																																										
	Crystallization and filtration																																																										
EcoScale								70																																																			

## General procedure for deprotection of Boc function in solid state

Reagents																																																				
<input checked="" type="checkbox"/> Link identifier*: <input type="text"/> name: <input type="text"/> MF*: <input type="text"/> MW: <input type="text"/> density: <input type="text"/> purity*: <input type="text"/> ml: <input type="text"/> g: <input type="text"/> mmoles: <input type="text"/> equiv: <input type="text"/> 1 <input type="checkbox"/> <input type="checkbox"/> Name, MF or <input type="text"/> Intermediate 3a <input type="text"/> 426.56 <input type="text"/> 100% <input type="text"/> 0 <input type="text"/> 0.49481 <input type="text"/> 1.16 <input type="text"/> 1  2 <input type="checkbox"/> <input type="checkbox"/> <input type="text"/> Hydrochloric acid <input type="text"/> HCl <input type="text"/> 36.46094 <input type="text"/> 100% <input type="text"/> 0 <input type="text"/> 0.431406 <input type="text"/> 11.831999 <input type="text"/> 10.2 																																																				
Products																																																				
identifier*: <input type="text"/> name: <input type="text"/> MF*: <input type="text"/> MW: <input type="text"/> g: <input type="text"/> mmoles: <input type="text"/> g theor: <input type="text"/> yield: <input type="text"/> Intermediate 4 <input type="text"/> 362.9 <input type="text"/> 0.412 <input type="text"/> 1.1352989 <input type="text"/> 0.420964 <input type="text"/> 97.8706																																																				
Conditions																																																				
<table border="1"> <thead> <tr> <th>Reagents</th> <th>Name</th> <th>mmoles</th> <th>eq.</th> <th>Bp</th> <th>Hazard</th> <th>Price</th> </tr> </thead> <tbody> <tr> <td></td> <td>Intermediate 3a</td> <td>2.81</td> <td>1</td> <td></td> <td></td> <td></td> </tr> <tr> <td></td> <td>Hydrochloric acid</td> <td>28.71</td> <td>10.19</td> <td>57</td> <td></td> <td></td> </tr> <tr> <td>Yield</td> <td>98</td> <td></td> <td></td> <td></td> <td></td> <td>-1</td> </tr> <tr> <td>Price / availability</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td>-5</td> </tr> <tr> <td>Safety</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td>-10</td> </tr> </tbody> </table>											Reagents	Name	mmoles	eq.	Bp	Hazard	Price		Intermediate 3a	2.81	1					Hydrochloric acid	28.71	10.19	57			Yield	98					-1	Price / availability						-5	Safety						-10
Reagents	Name	mmoles	eq.	Bp	Hazard	Price																																														
	Intermediate 3a	2.81	1																																																	
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Price / availability						-5																																														
Safety						-10																																														
Technical setup	Possible items	Selected items																																																		
	Common set-up Instruments for controlled addition of chemicals Unconventional activation technique	Unconventional activation technique					-2																																													
Temperature / time	Possible items	Selected items																																																		
	Room temperature, < 1h Room temperature, < 24h Heating, < 1h	Room temperature, < 24h					-1																																													
Workup and purification	Possible items	Selected items																																																		
	None Cooling to room temperature Adding solvent	None					0																																													
EcoScale							81																																													

## Sulfonylation of primary amine in solution

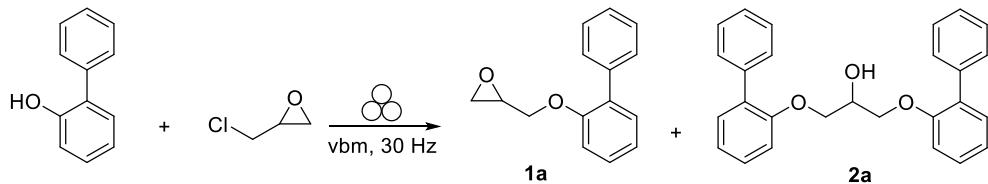
Reagents										
<input checked="" type="checkbox"/> Link										
	identifier*	name	MF*	MW	density	purity*	ml	g	mmoles	equiv.
1	<input type="checkbox"/> Name, MF or <input type="checkbox"/> Amine 4	Amine 4		362.9		100%	0	0.333868	0.92	1
2	<input type="checkbox"/> + <input type="checkbox"/>	3-Chlorobenesulfonyl chloride	C6H4Cl2O2S	211.06256		100%	0	0.233013	1.104	1.2
3	<input type="checkbox"/> + <input type="checkbox"/>	Triethylamine	C6H15N	101.1918	0.728	100%	0.383639	0.279289	2.760000	3
4	<input type="checkbox"/> + <input type="checkbox"/>	Dichloromethane	CH2Cl2	84.93288	1.325	100%	4	5.3	62.402216	67.828496
Products										
	identifier*: <input type="text"/>	name: Final compound PZ-1361	MF*: <input type="text"/>	MW: <input type="text"/> 501.04	g: <input type="text"/> 0.33	mmoles: <input type="text"/> 0.658630	g theor: <input type="text"/> 0.460957	yield: <input type="text"/> 71.590200		
Conditions										
Reagents	Name	mmoles	eq.	Bp	Hazard	Price				
	Amine 4	2.78	1							
	3-Chlorobenesulfonyl chloride	3.34	1.2	102						
	Triethylamine	8.36	3	90						
	Dichloromethane	189.09	67.82	39						
Yield	<input type="text"/> 72					-14				
Price / availability						-13				
Safety						-10				
Technical setup	Possible items Common set-up Instruments for controlled addition of chemicals Unconventional activation technique			Selected items Common set-up		0				
Temperature / time	Possible items Heating, > 1h Cooling to 0°C Cooling, < 0°C			Selected items Cooling to 0°C		-4				
Workup and purification	Possible items Sublimation Liquid - liquid extraction or washing Classical chromatography			Selected items Removal of solvent with bp < 150°C Classical chromatography		-10				
EcoScale						49				

## Sulfonylation of primary amine in ball-mill

Reagents											
<input checked="" type="checkbox"/> Link											
	identifier*	name	MF*	MW	density	purity*	ml	g	mmoles	equiv.	
1	<input type="checkbox"/> + <input type="checkbox"/> -	Name, MF or Amine 4		362.9		100%	0	0.168023	0.463	1	
2	<input type="checkbox"/> + <input type="checkbox"/> -	3-Chlorobenzenesulfonyl chloride	C6H4Cl2O2S	211.06256		100%	0	0.097722	0.463	1	
3	<input type="checkbox"/> + <input type="checkbox"/> -	Potassium carbonate	CK2O3	138.2058		100%	0	0.063989	0.463	1	
Products											
	identifier*: name:	MF*:	MW:	g:	mmoles:	g theor:	yield:				
				501.04	0.2001	0.399369	0.231982	86.2567			
Conditions											
Reagents	Name	mmoles	eq.	Bp	Hazard	Price					
	Amine 4	2.31	1								
	3-Chlorobenzenesulfonyl chloride	2.31	1	102							
	Potassium carbonate	2.31	1								
Yield	86					-7					
Price / availability						-10					
Safety						0					
Technical setup	Possible items Common set-up Instruments for controlled addition of chemicals Unconventional activation technique	Selected items Unconventional activation technique				-2					
Temperature / time	Possible items Room temperature, < 1h Room temperature, < 24h Heating, < 1h	Selected items Room temperature, < 1h				0					
Workup and purification	Possible items Sublimation Liquid - liquid extraction or washing Classical chromatography	Selected items Liquid - liquid extraction or washing				-3					
EcoScale						78					

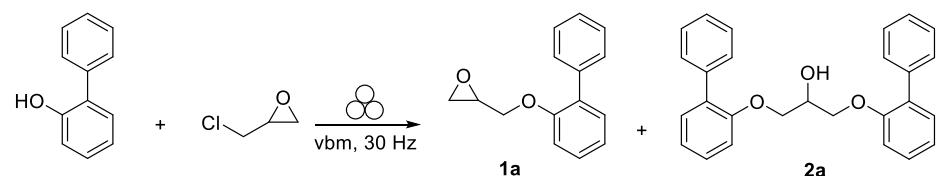
## II. Additional experimental data for alkylation of 2-phenylphenol in ball-mill

**Table S3.** Optimization of the kind of base for the alkylation of 2-phenylphenol<sup>a</sup>



Entry	Base (eq)	Conv. % <sup>b</sup>	
		1a	2a
1	NaHCO <sub>3</sub> (3)	2	0
2	Na <sub>2</sub> CO <sub>3</sub> (3)	1	0
3	K <sub>2</sub> CO <sub>3</sub> (3)	35	0
4	K <sub>2</sub> CO <sub>3</sub> /KOH (3/0.5)	26	0
5	K <sub>2</sub> CO <sub>3</sub> /KOH (2/1.5)	35	3
6	Cs <sub>2</sub> CO <sub>3</sub> (3)	29	10
7	NaOH (3)	16	0
8	KOH (3)	36	10
9	NaOtBut (3)	32	3
10	DIEA (3)	2	0
11	TEA (3)	5	0

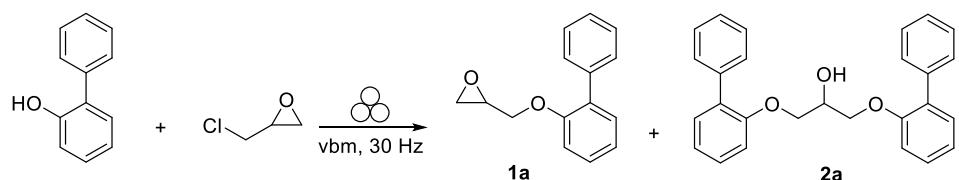
<sup>a</sup> Reaction conditions: 2-phenylphenol (1 eq), epichlorohydrin (1 eq), vbm 30 Hz, 10 mL PTFE jar,  $\varnothing = 1$  cm, 80 min, total mass of reagents: 100 mg; <sup>b</sup> Conversions were determined by UPLC/MS analysis.

**Table S4.** Optimization of milling time for the alkylation of 2-phenylphenol<sup>a</sup>

Entry	Base (eq)	Time (min)	Conv.% <sup>b</sup>	
			1a	2a
1		40	15	0
2		80	35	0
3		120	40	0.5
4	K <sub>2</sub> CO <sub>3</sub>	140	43	2
5	(3)	160	45	3.5
6		180	47.5	5
7		200	48.5	7.5
8		210	50	8.5
9		220	52	10

<sup>a</sup> Reaction conditions: 2-phenylphenol (1 eq), epichlorohydrin (1 eq), vbm 30 Hz, 10 mL PTFE jar,  $\phi = 1$  cm, total mass of reagents: 100 mg; <sup>b</sup> Conversions were determined by UPLC/MS analysis.

**Table S5.** Optimization of the milling conditions for the alkylation of 2-phenylphenol<sup>a</sup>



Entry	Base (eq)	Epich.	ε	Time (min)	Conv. % <sup>b</sup>	
					1a	2a
1		1 eq	1 cm		42.5	2.5
2	K <sub>2</sub> CO <sub>3</sub>	1 eq	1.5 cm		61	2.5
3	(3)	1.2 eq	1.5 cm	140	90	2
4		1.5 eq	1.5 cm		86	1

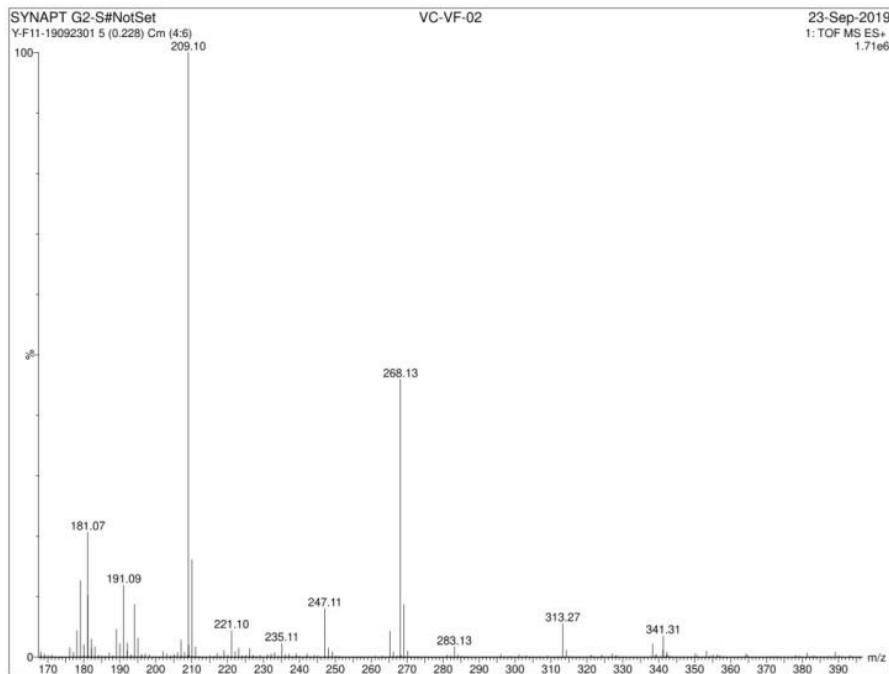
<sup>a</sup> Reaction conditions: 2-phenylphenol (1 eq), vbm 30 Hz, total mass of reagents: 100 mg; 10 mL PTFE jar;

<sup>b</sup> Conversions were determined by UPLC/MS analysis.

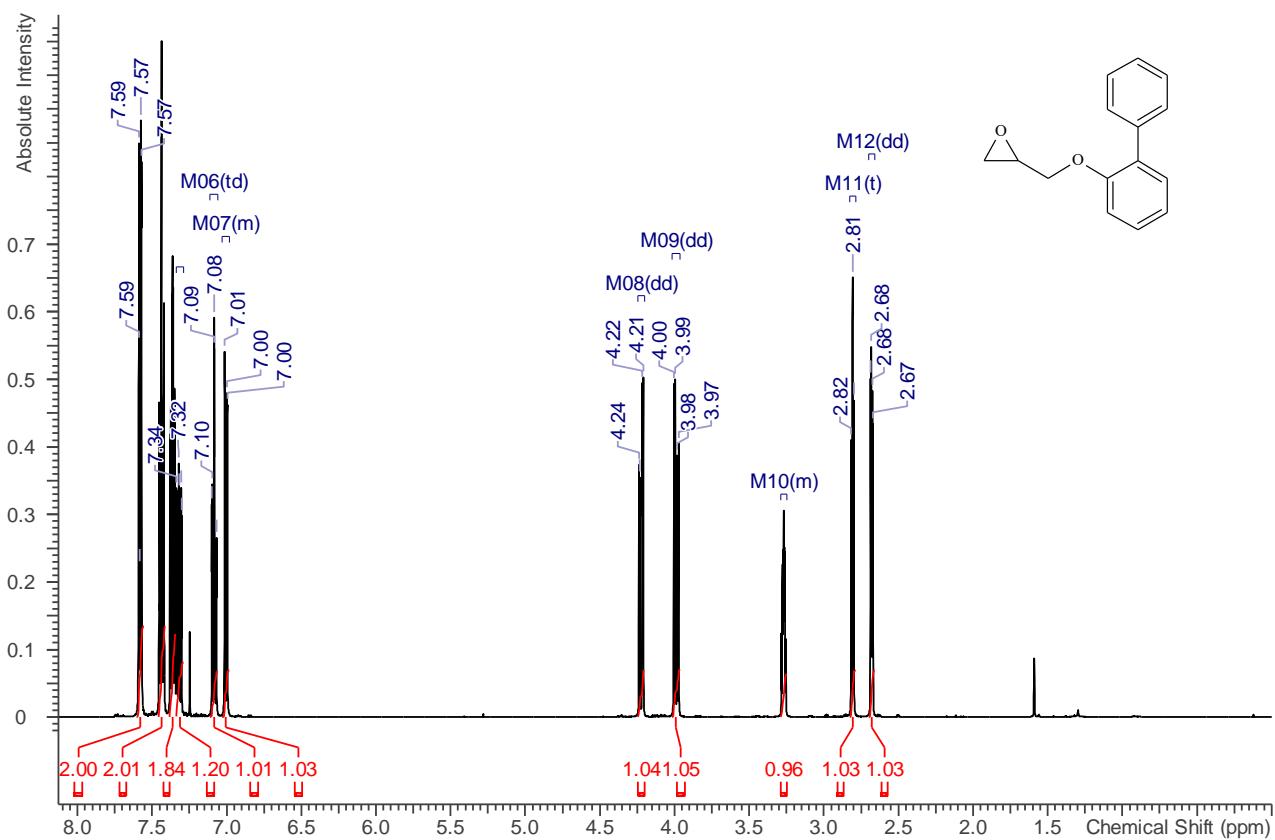
### III. HRMS, $^1\text{H}$ - and $^{13}\text{C}$ -NMR spectra

#### *2-(([1,1'-Biphenyl]-2-yloxy)methyl)oxirane (**1a**)*

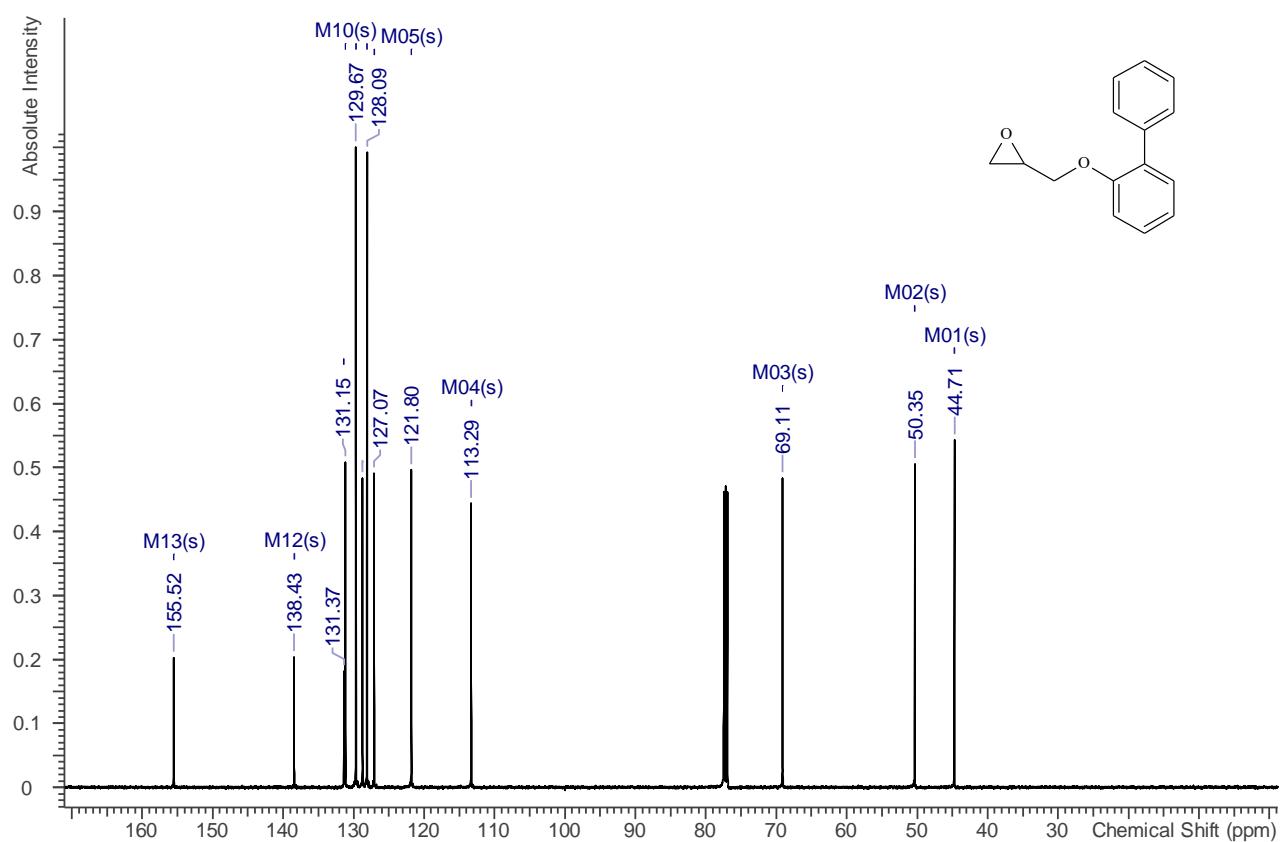
HRMS



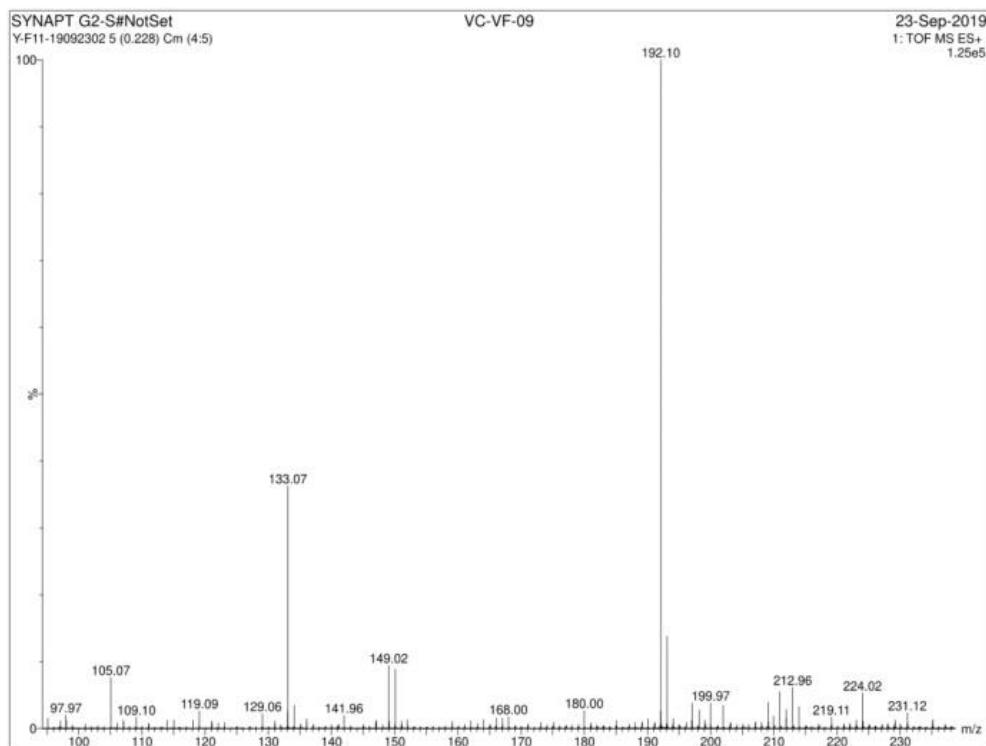
#### $^1\text{H}$ -NMR ( $\text{CDCl}_3$ , 500 MHz)



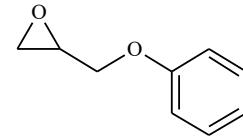
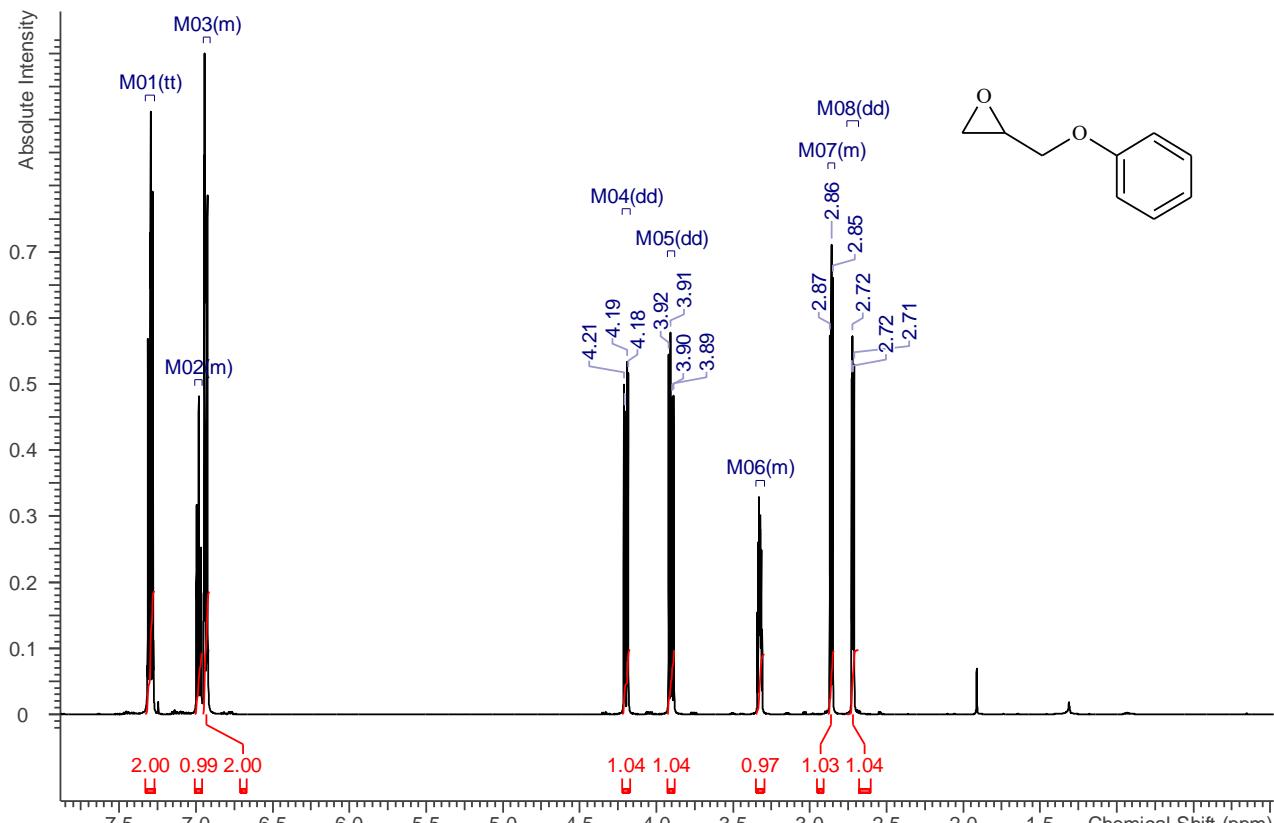
<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)



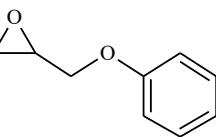
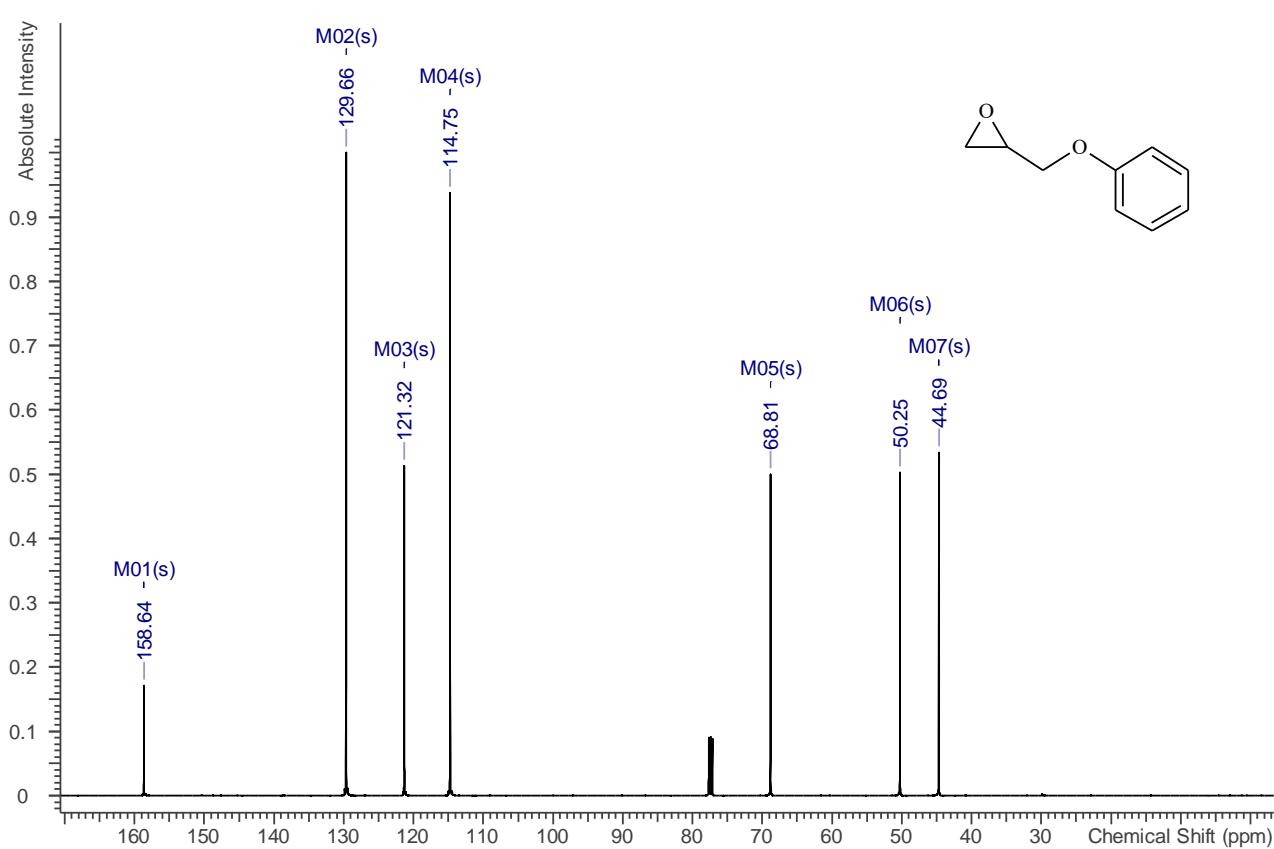
*2-(Phenoxy)methyl)oxirane (**1b**)*  
HRMS



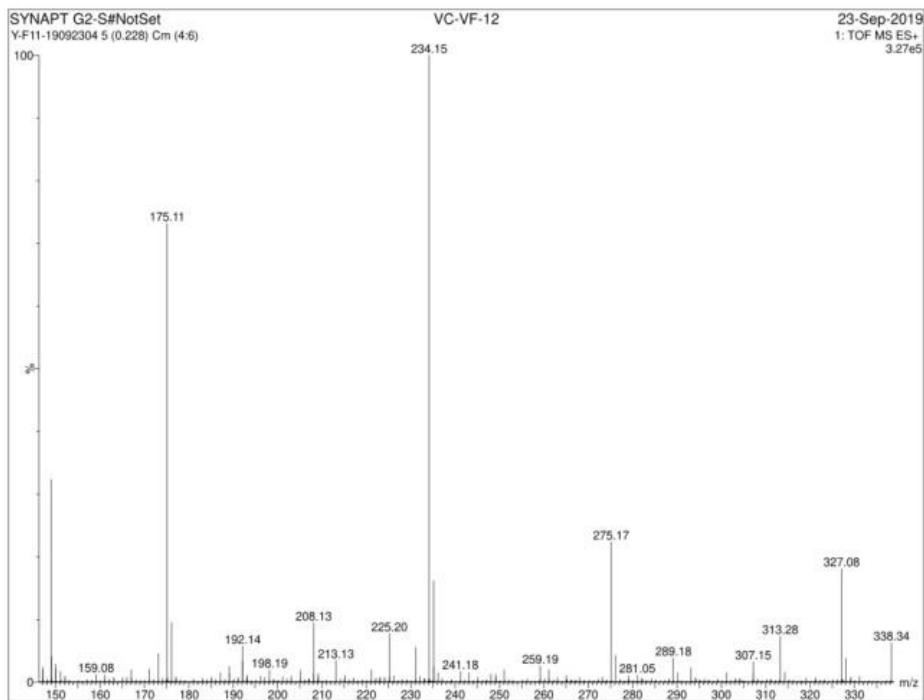
<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)



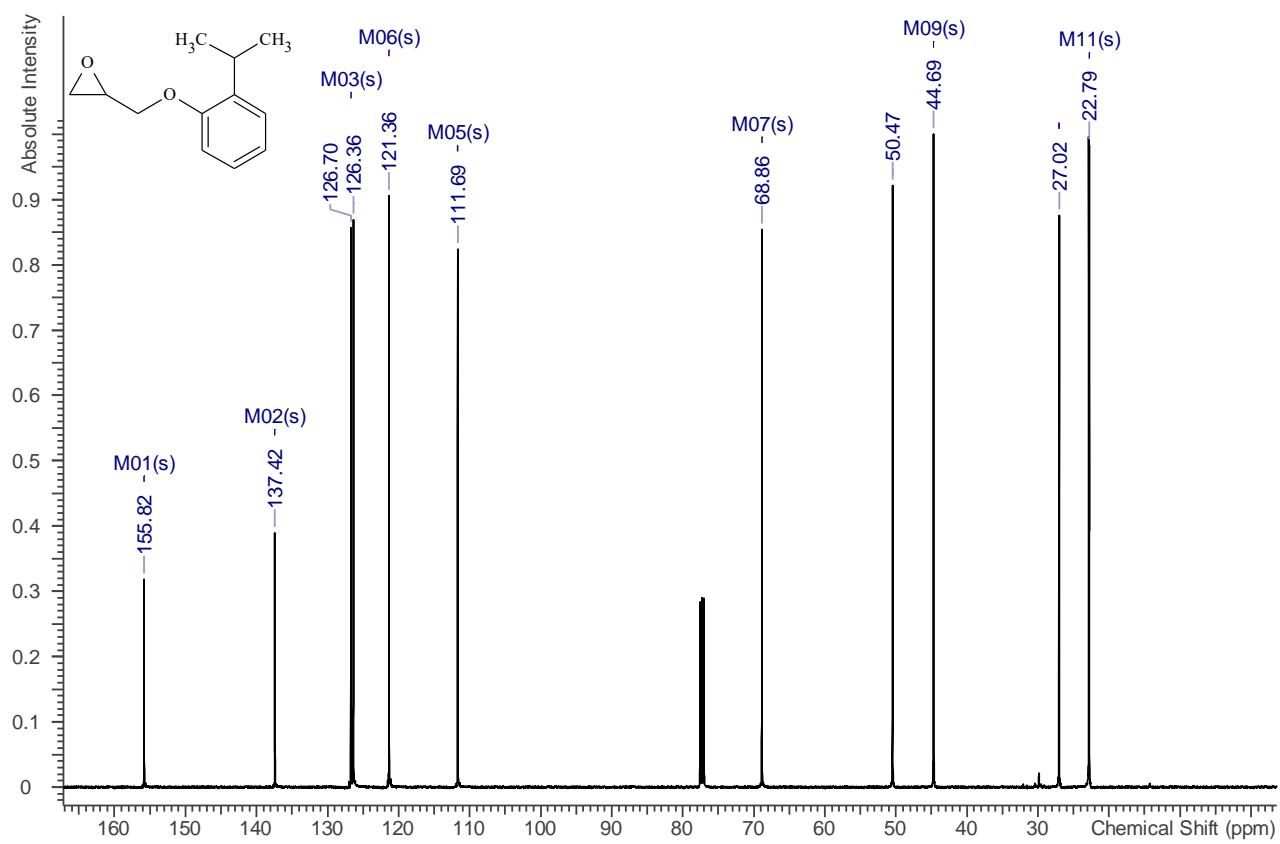
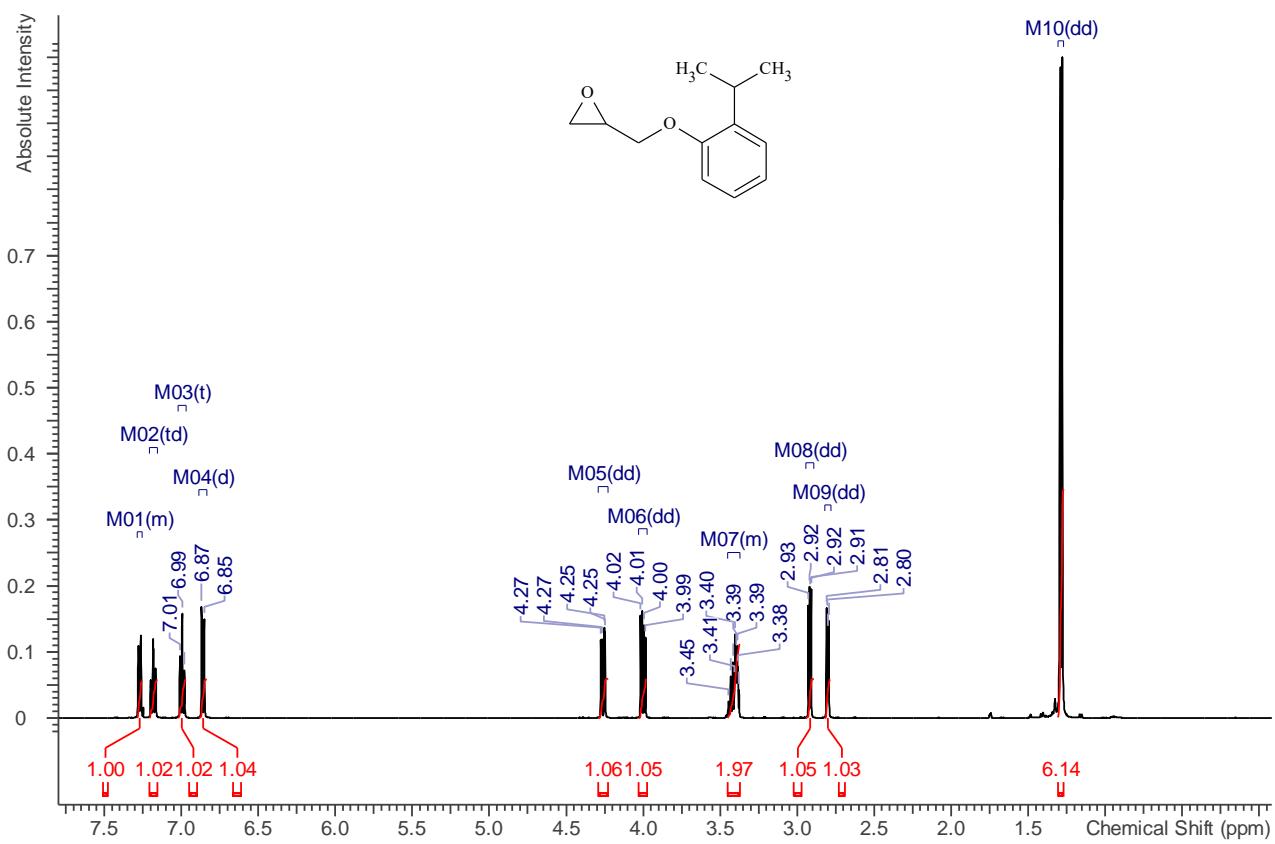
<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)



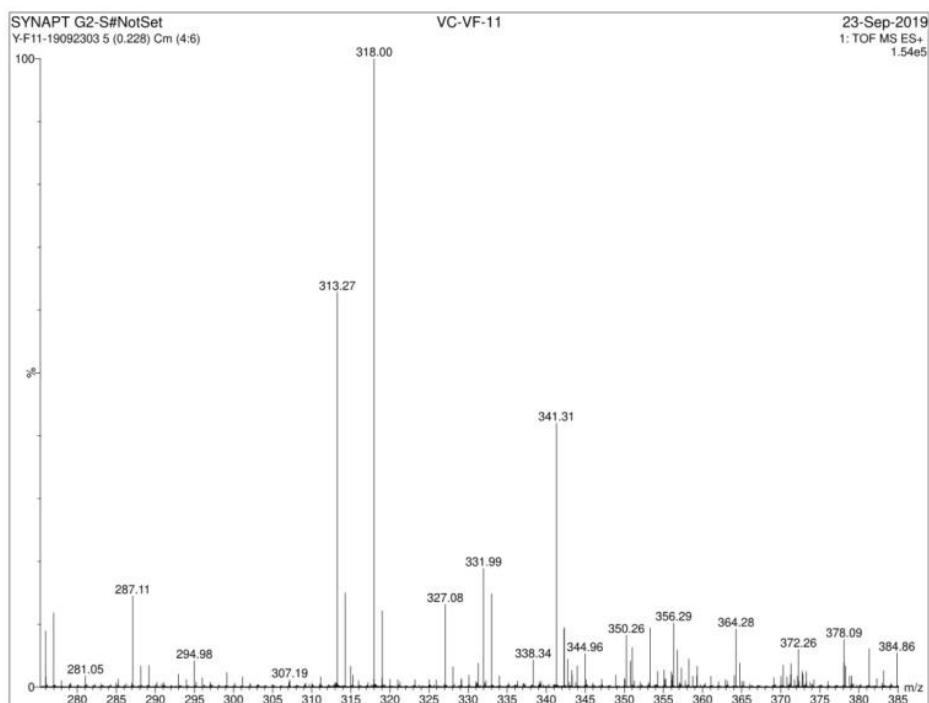
**2-((2-Isopropylphenoxy)methyl)oxirane (**1c**)**  
HRMS



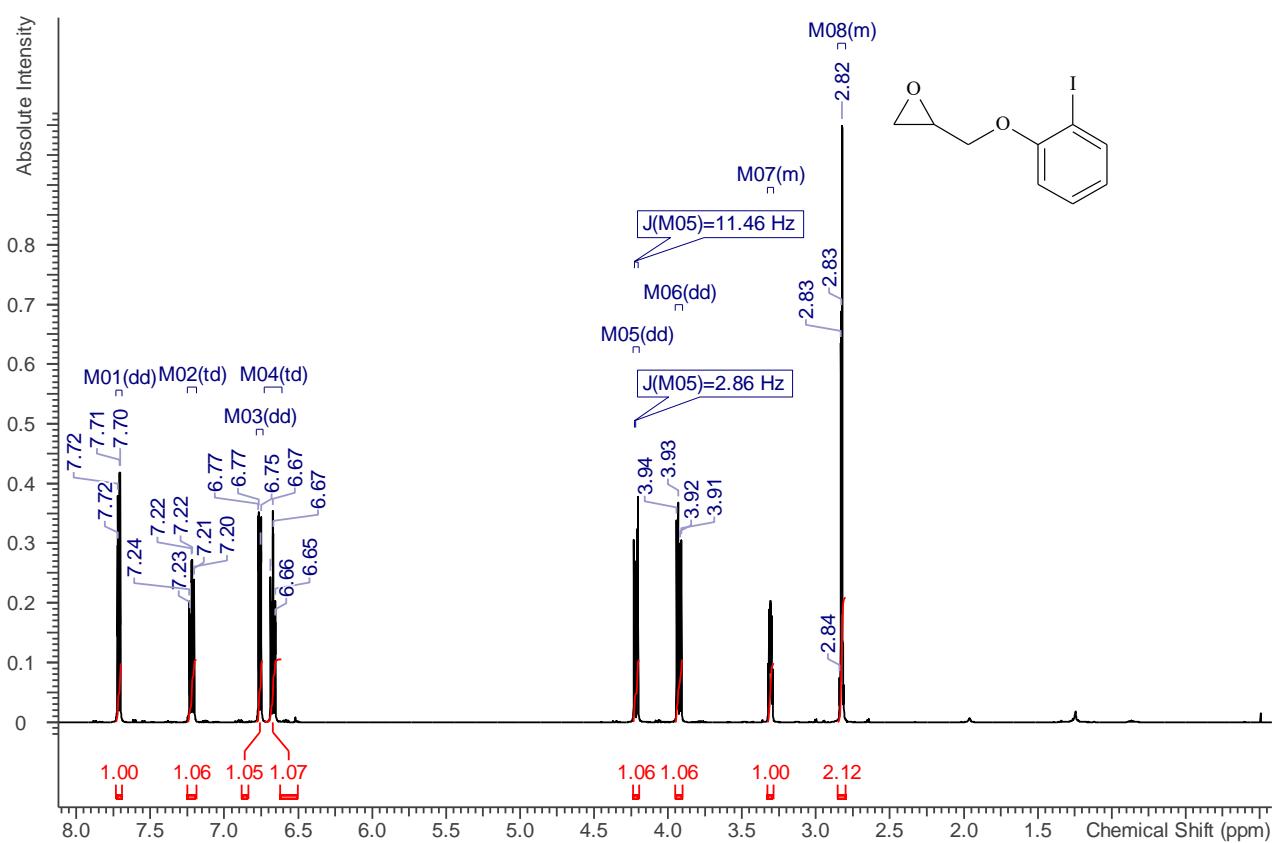
<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)



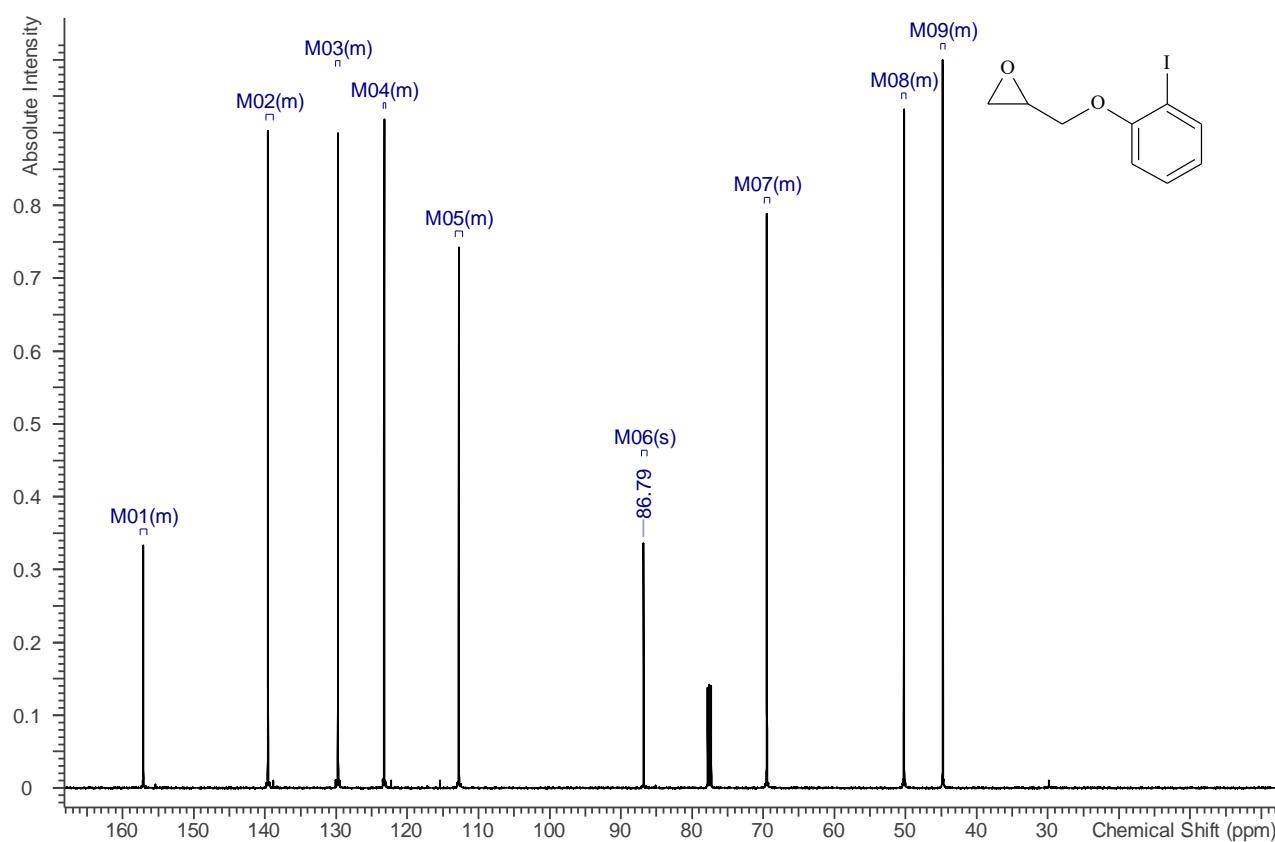
***2-((2-Iodophenoxy)methyl)oxirane (1d)***  
**HRMS**



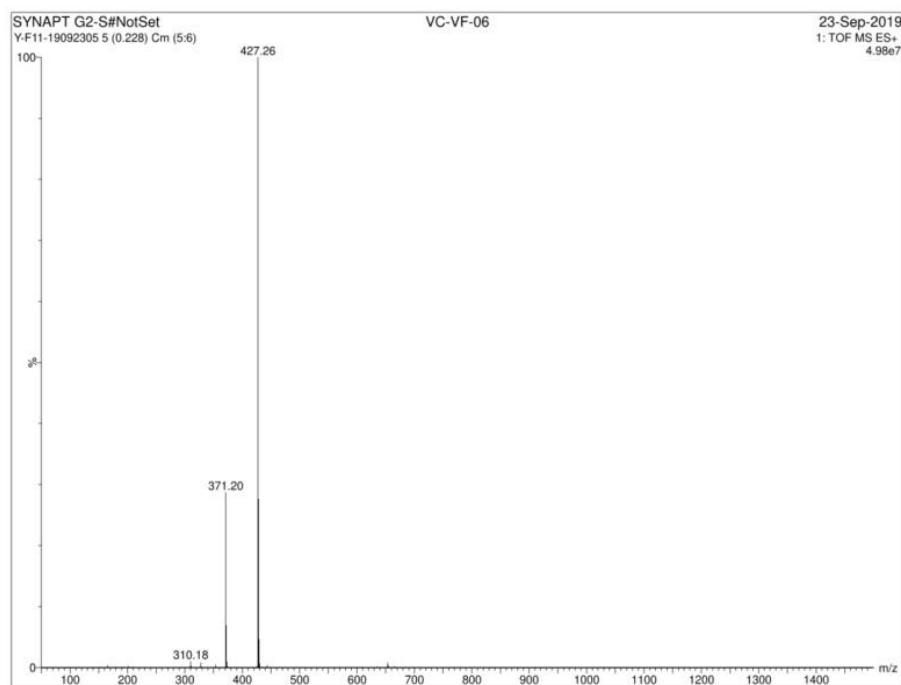
<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)



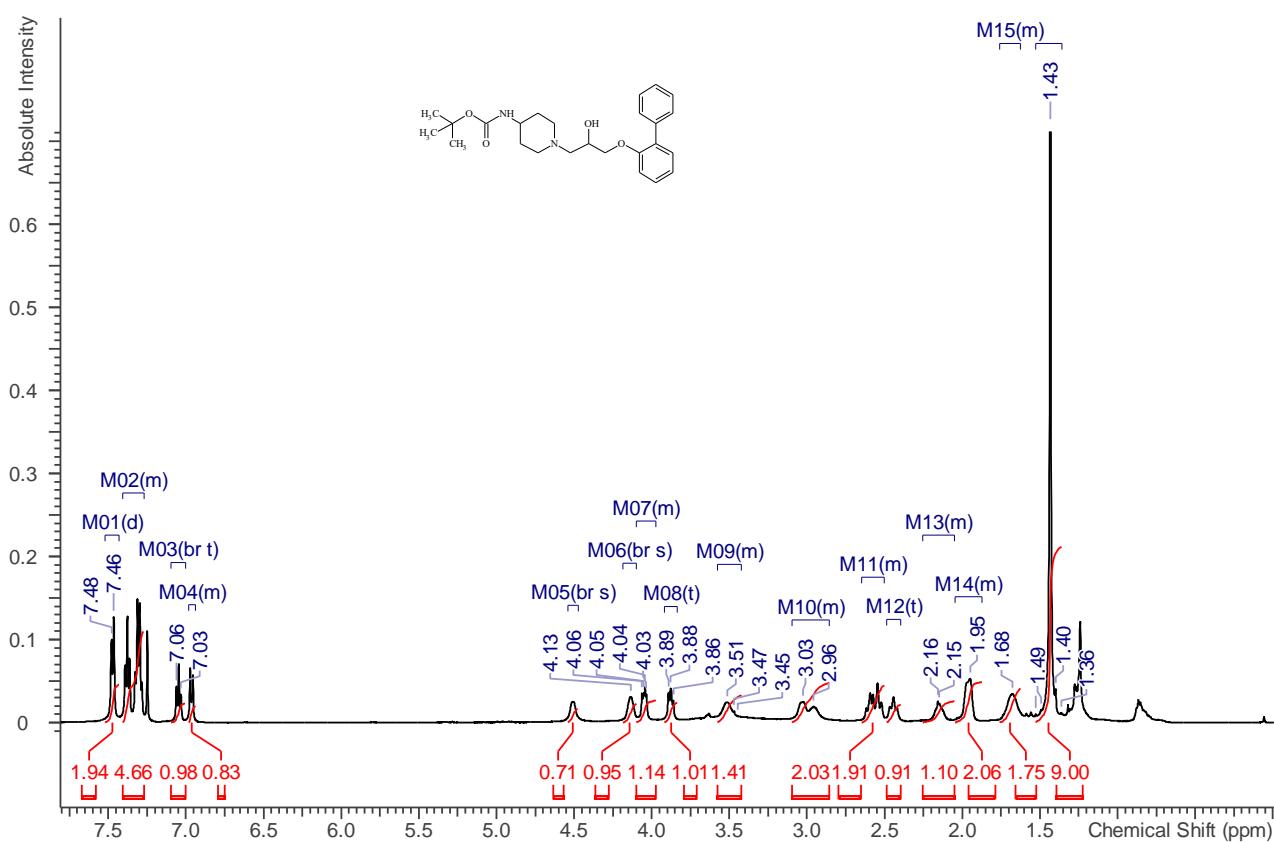
<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)



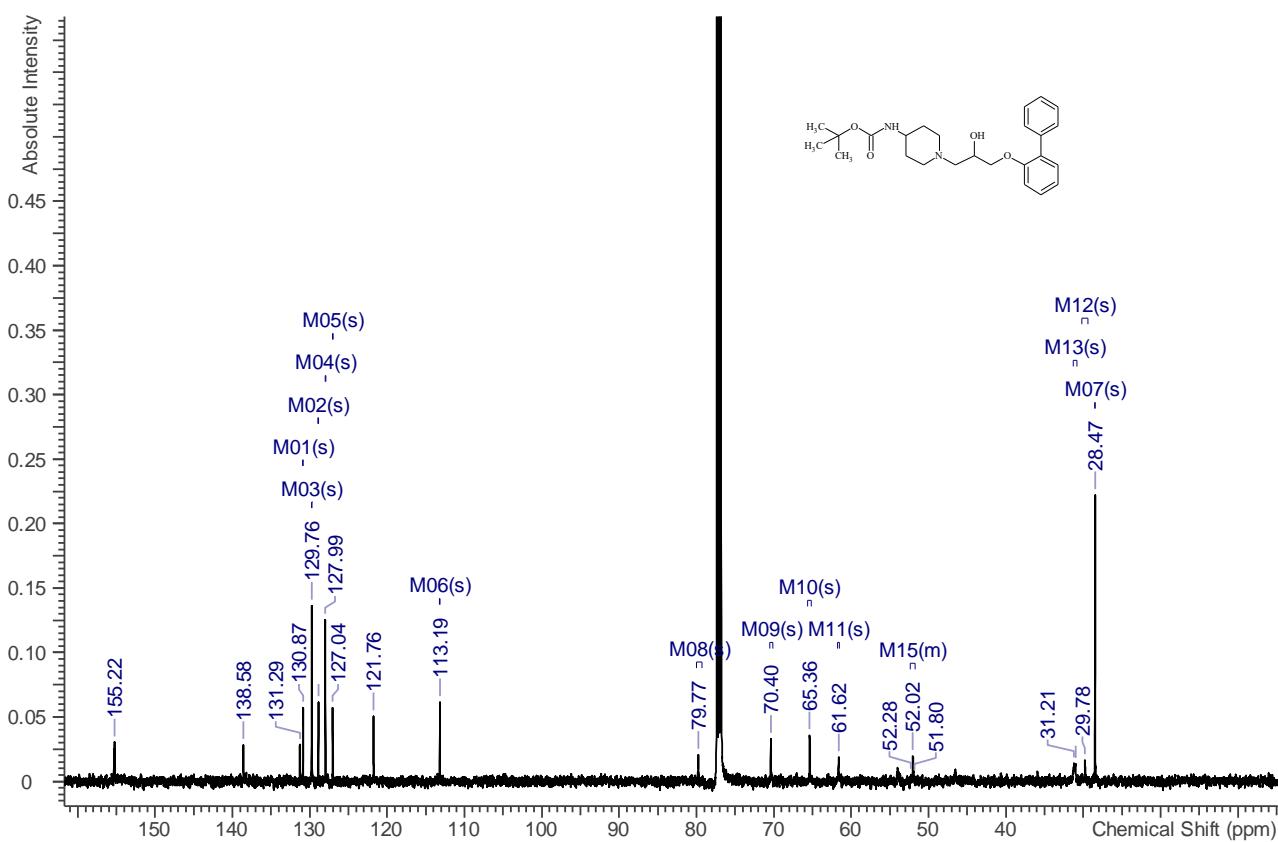
*Tert-butyl (1-(3-((1,1'-biphenyl)-2-yloxy)-2-hydroxypropyl)piperidin-4-yl)carbamate (3a)*  
HRMS



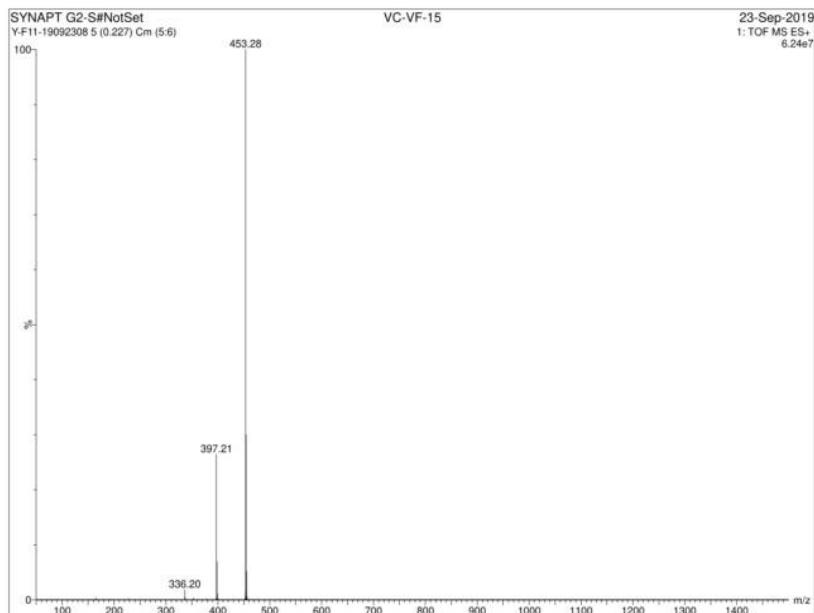
<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)



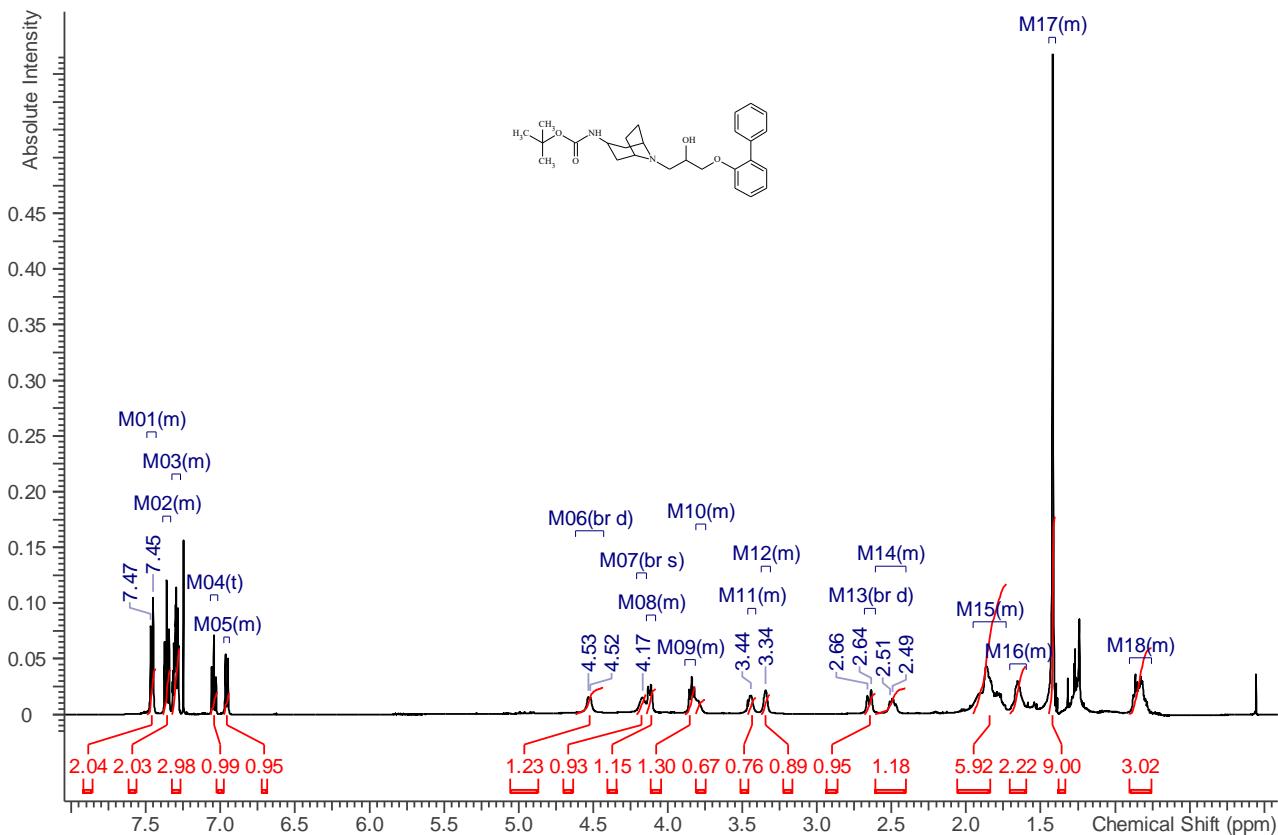
<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)



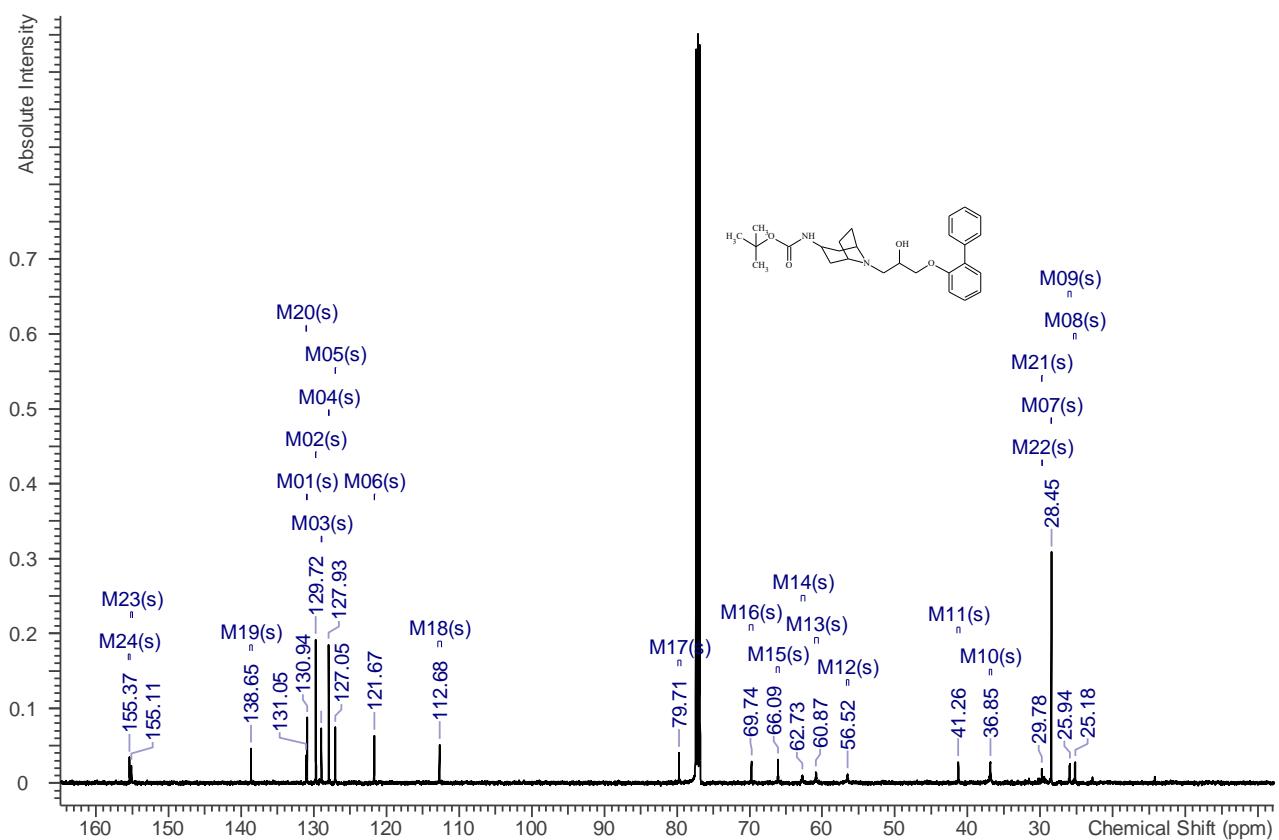
*Tert-butyl ((1*R*,5*S*)-8-(3-([1,1'-biphenyl]-2-yloxy)-2-hydroxypropyl)-8-azabicyclo[3.2.1]octan-3-yl)carbamate (**3b**)*  
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

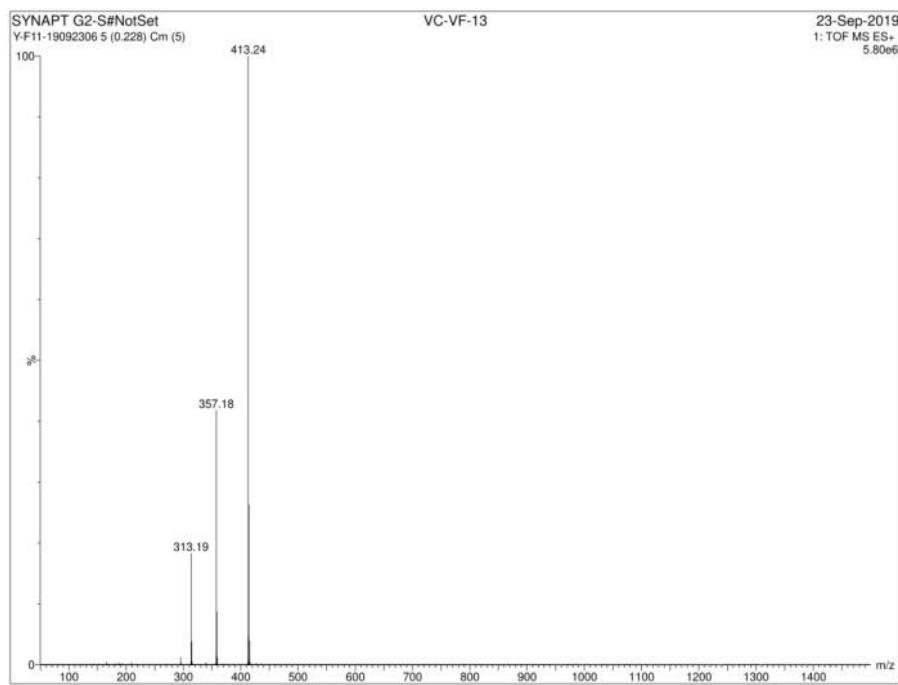


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

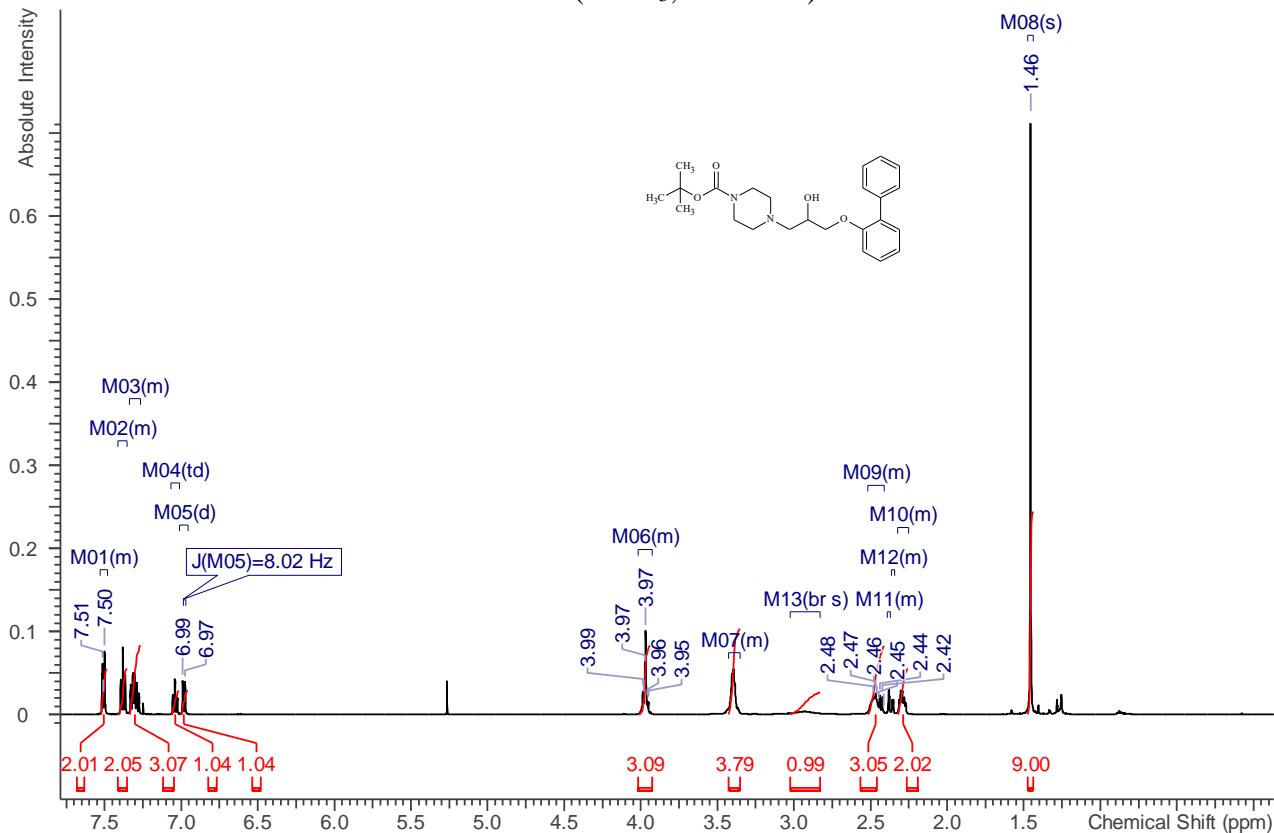


*Tert-butyl 4-(3-([1,1'-biphenyl]-2-yloxy)-2-hydroxypropyl)piperazine-1-carboxylate (3c)*

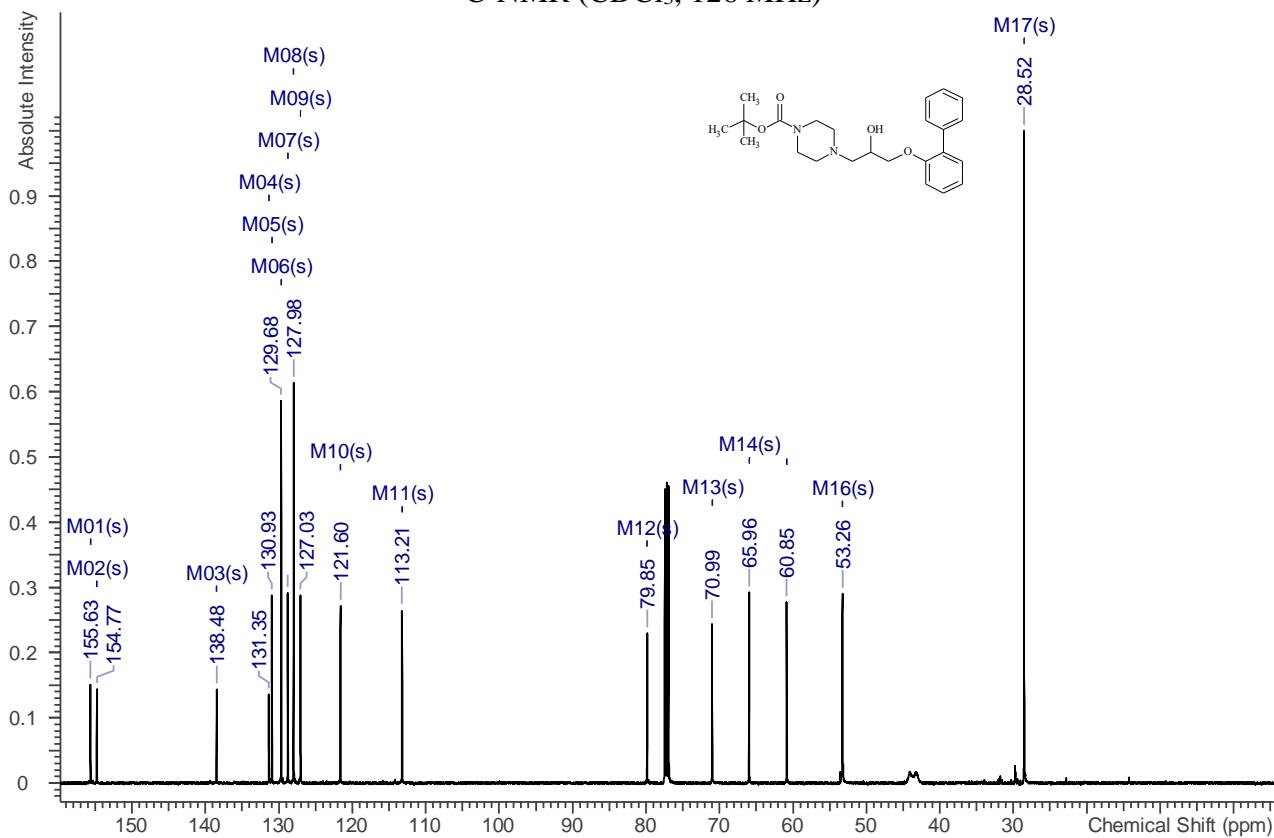
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

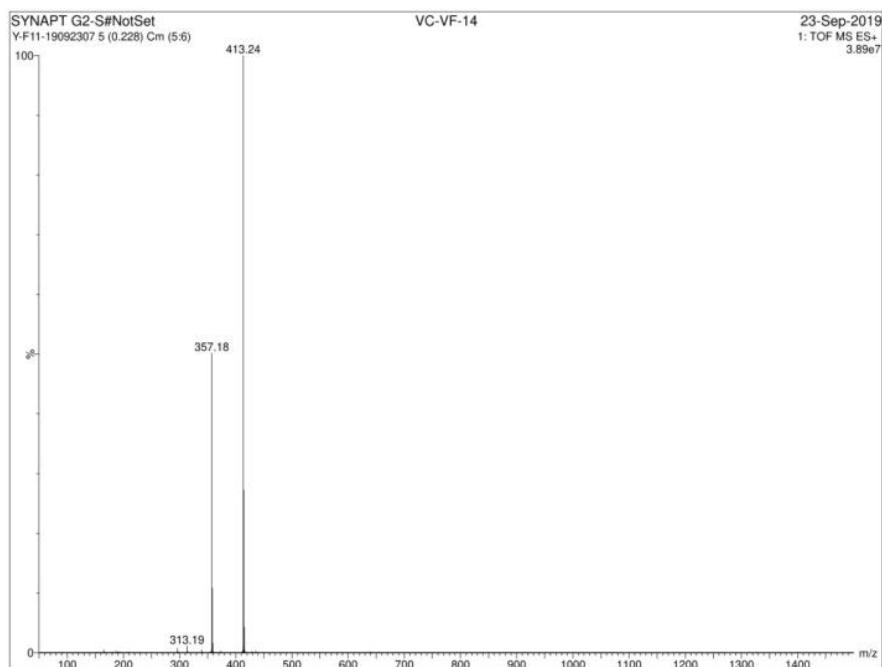


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

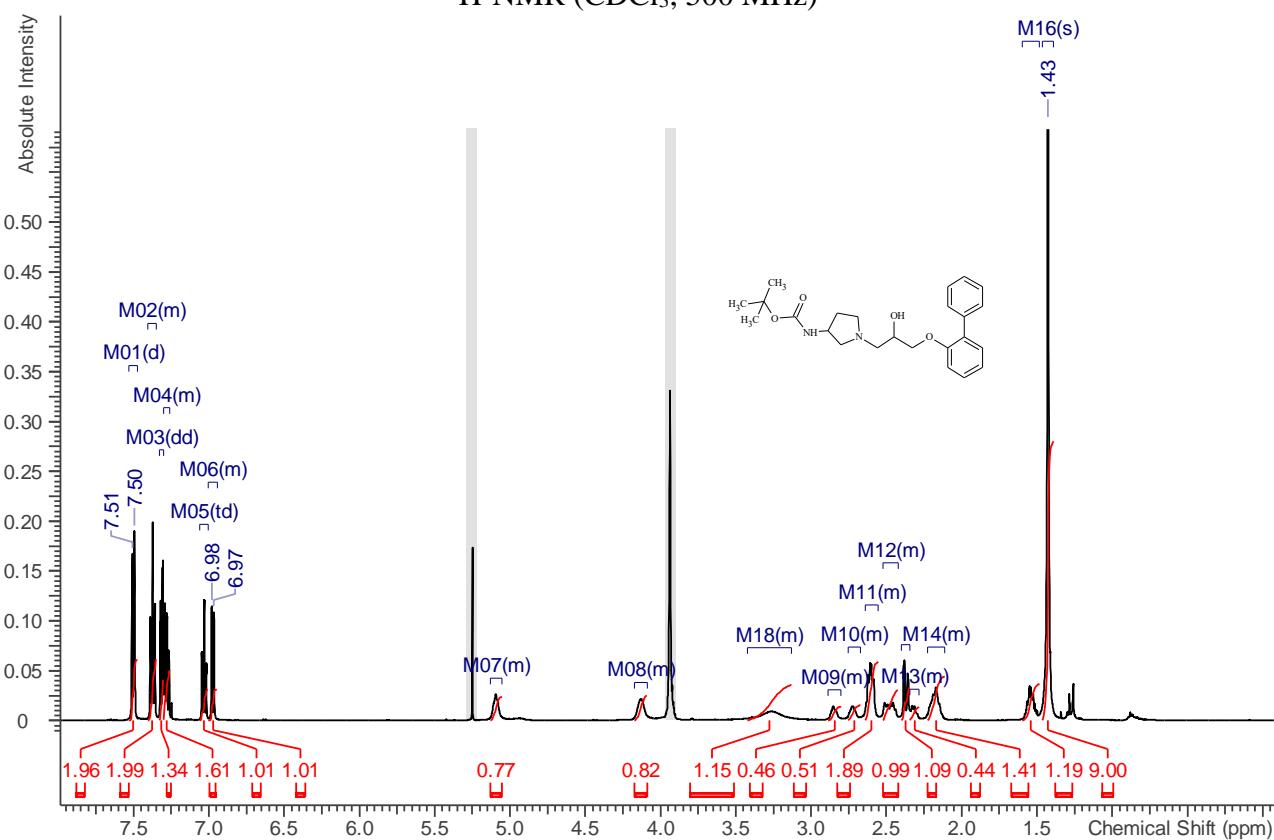


*Tert-butyl (1-(3-([1,1'-biphenyl]-2-yloxy)-2-hydroxypropyl)pyrrolidin-3-yl)carbamate (3d)*

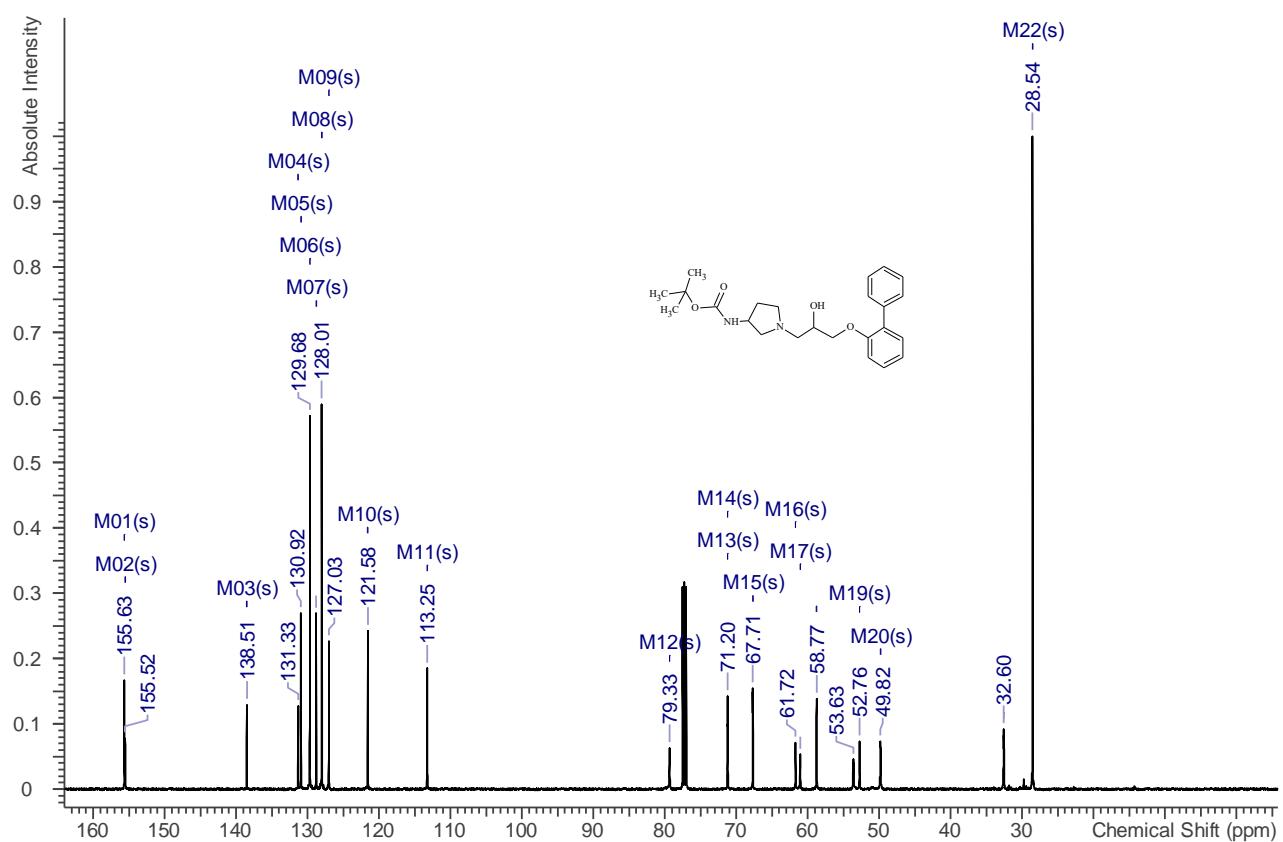
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

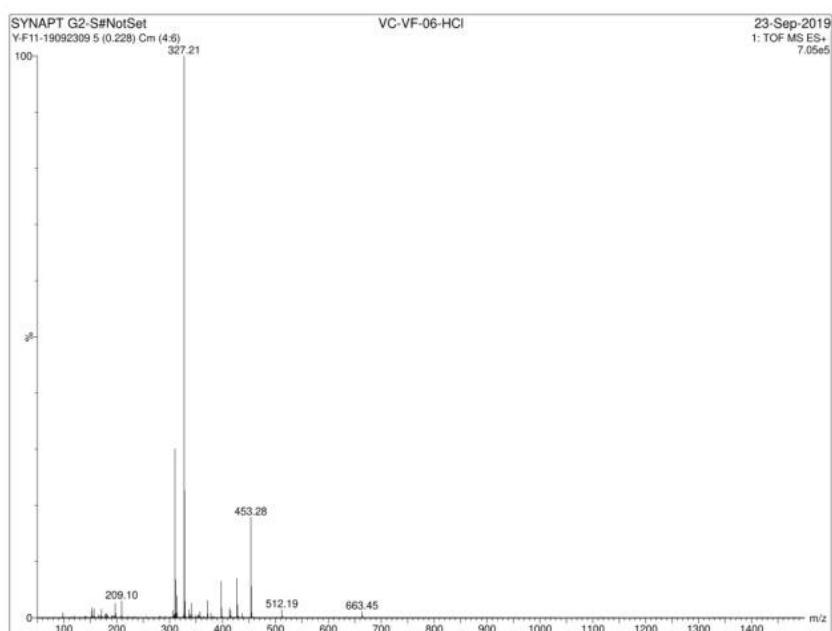


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

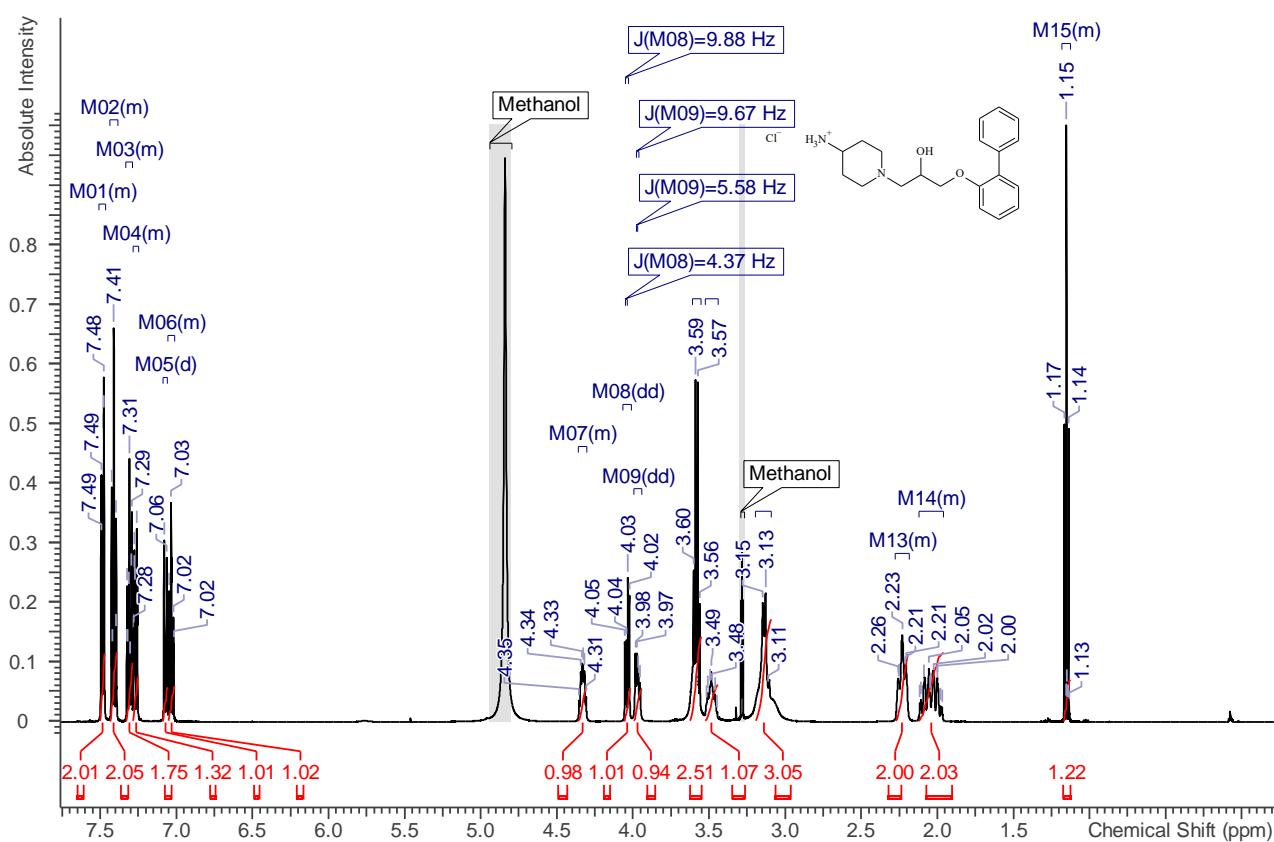


*I-([1,1'-Biphenyl]-2-yloxy)-3-(4-aminopiperidin-1-yl)propan-2-ol (4)*

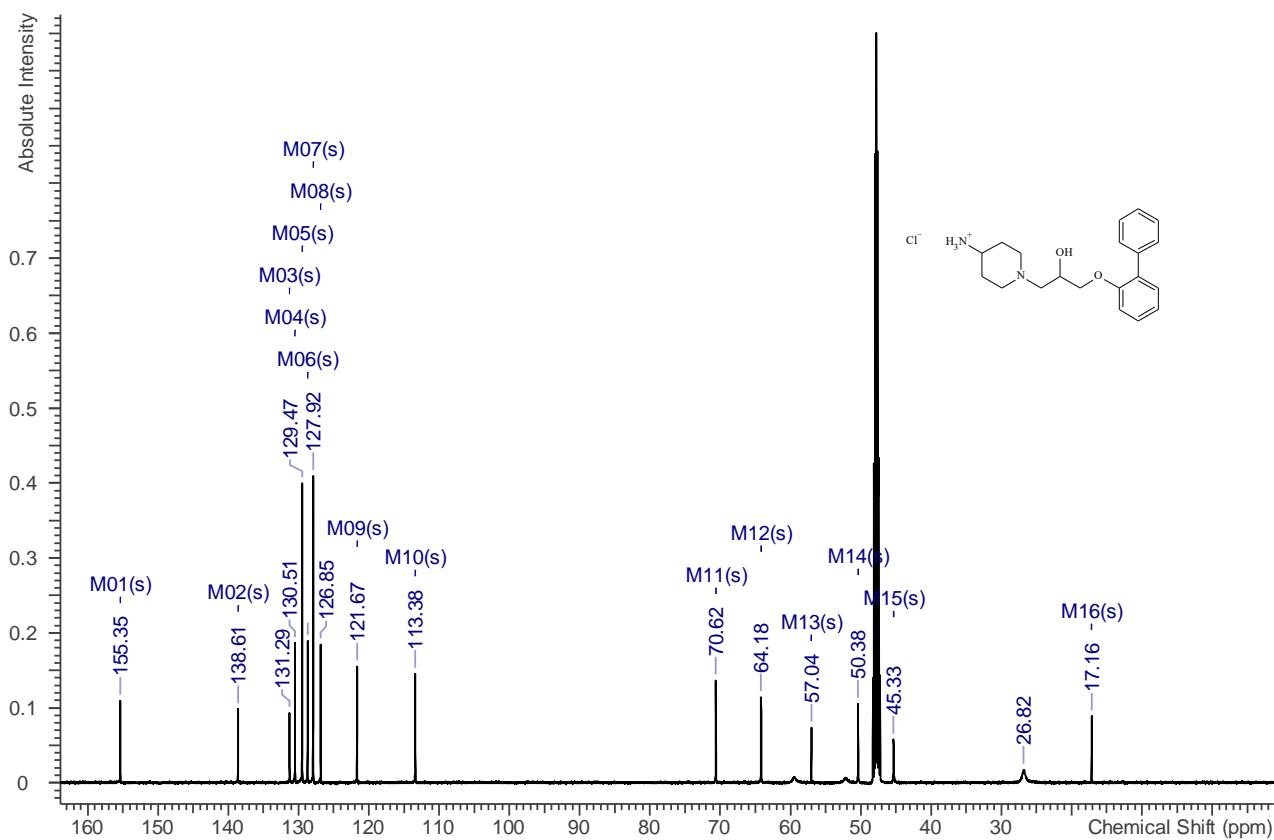
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

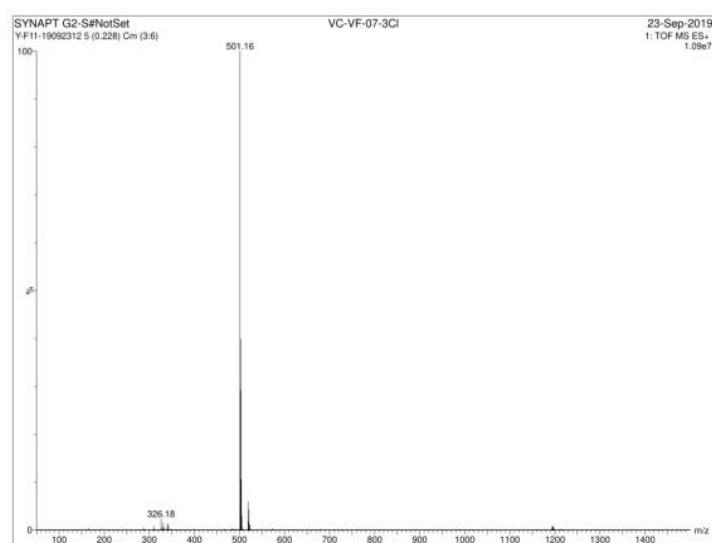


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

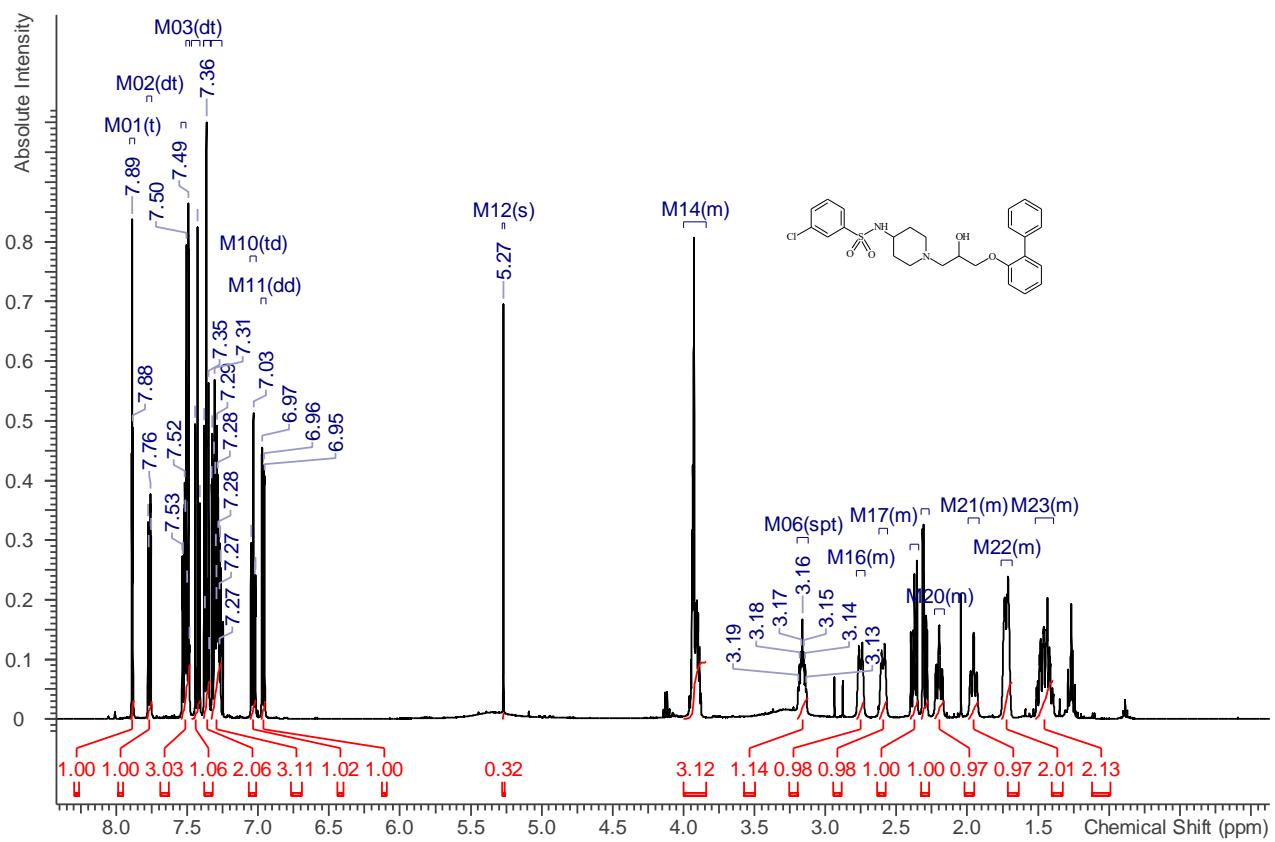


*3-Chloro-N-(1-(3-([1,1'-biphenyl]-2-yloxy)-2-hydroxypropyl)piperidin-4-yl)-benzene sulfonamide*  
**(5a)**

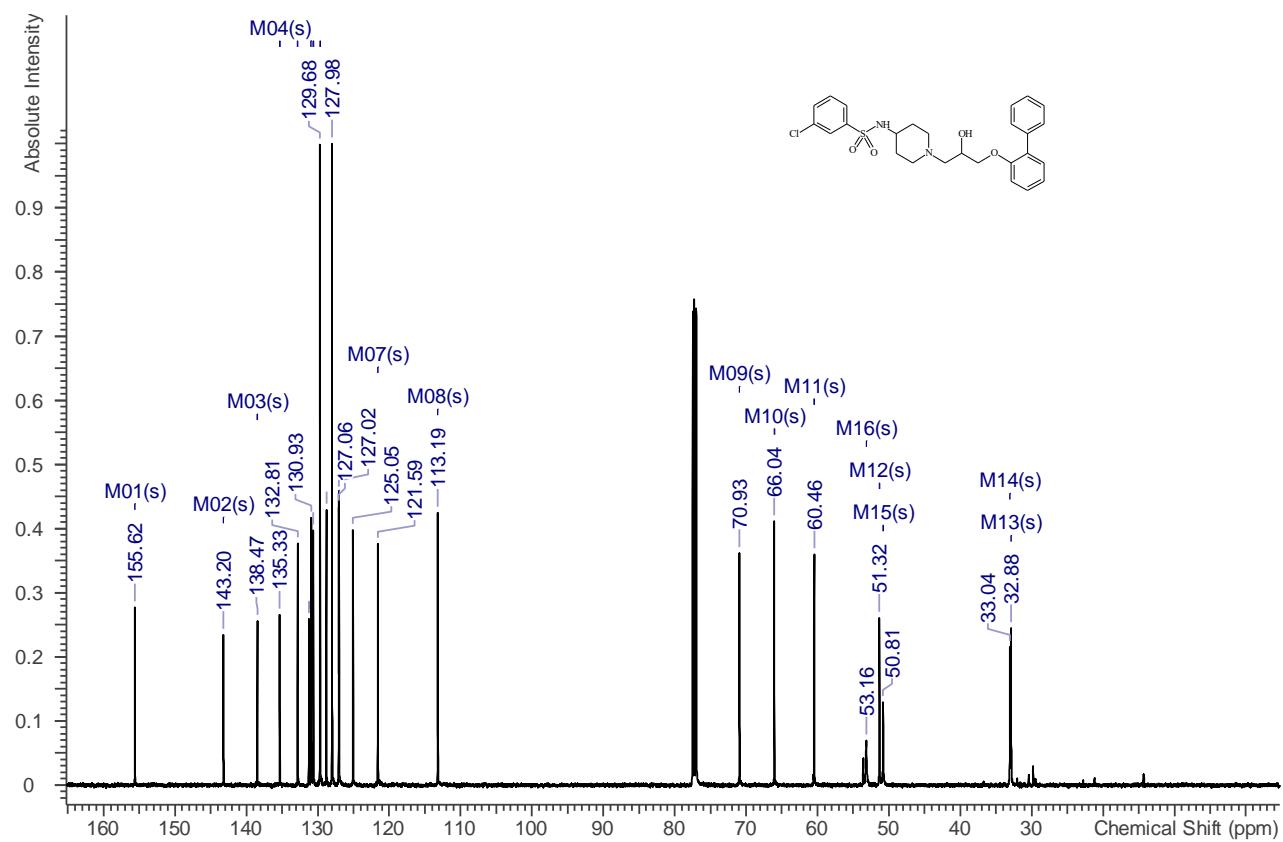
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

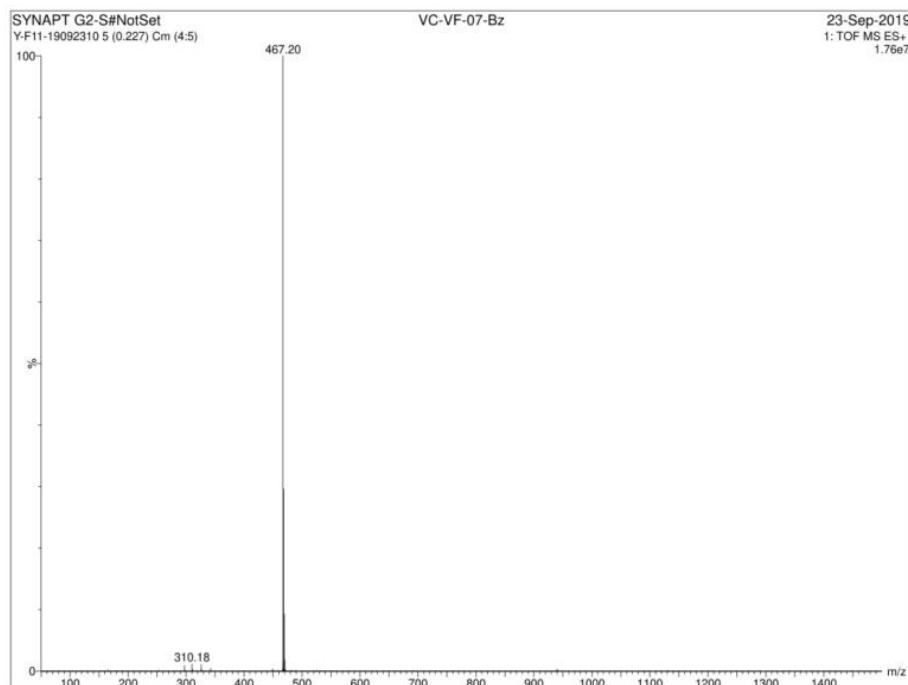


### <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

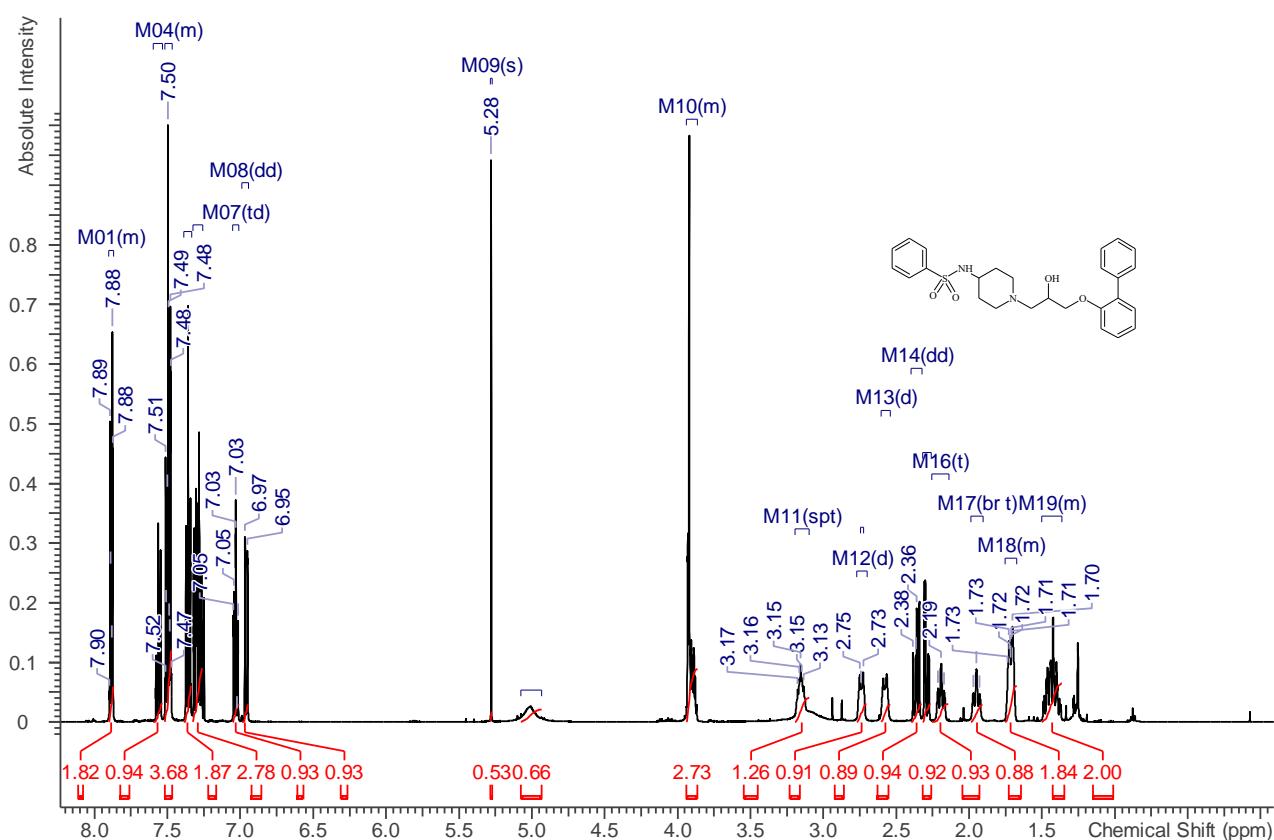


*N-(1-(3-([1,1'-Biphenyl]-2-yloxy)-2-hydroxypropyl)piperidin-4-yl)benzenesulfonamide (**5b**)*

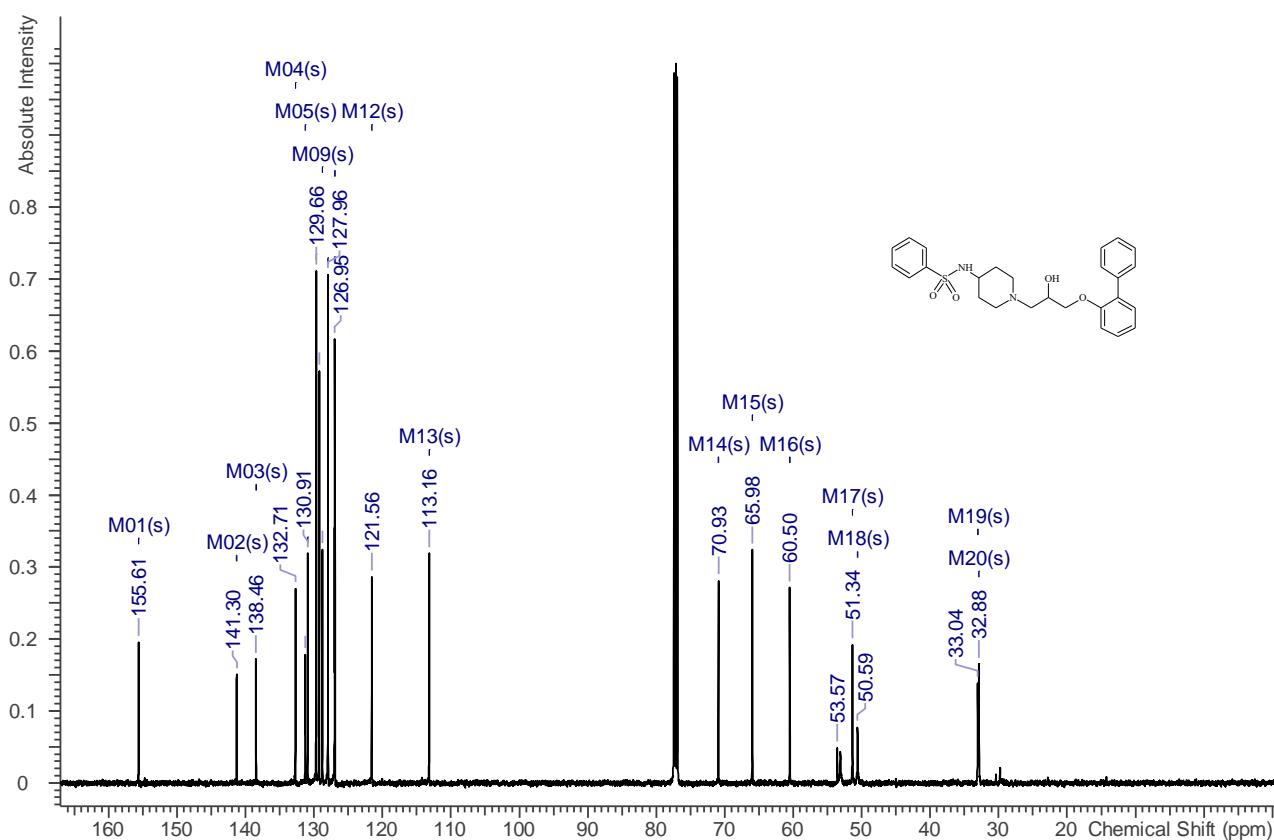
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

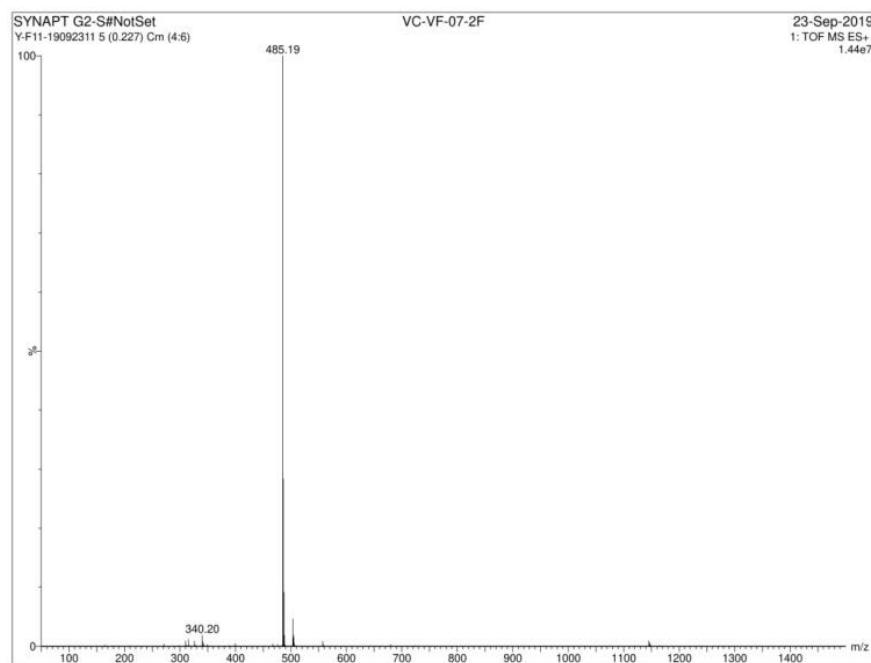


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

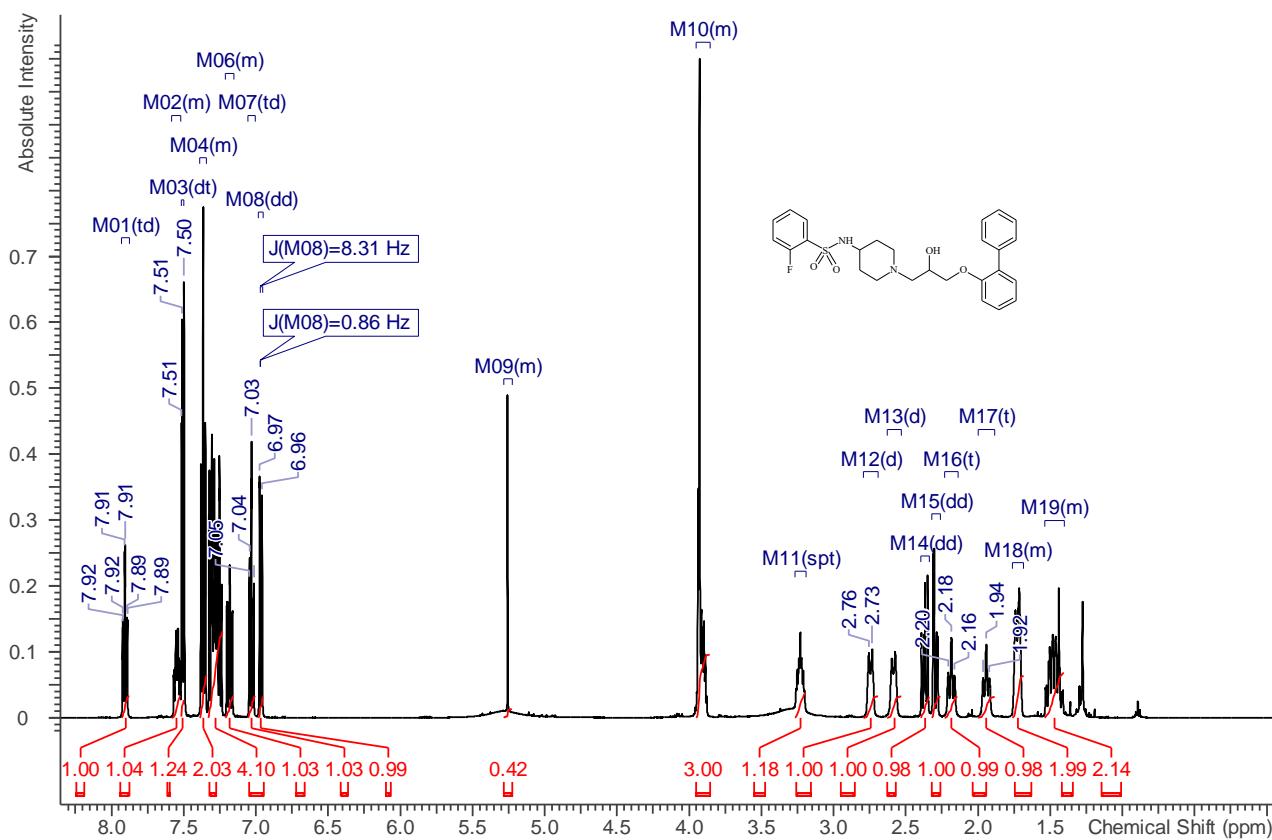


2-Fluoro-N-(1-(3-([1,1'-biphenyl]-2-yloxy)-2-hydroxypropyl)piperidin-4-yl)-benzene sulfonamide  
**(5c)**

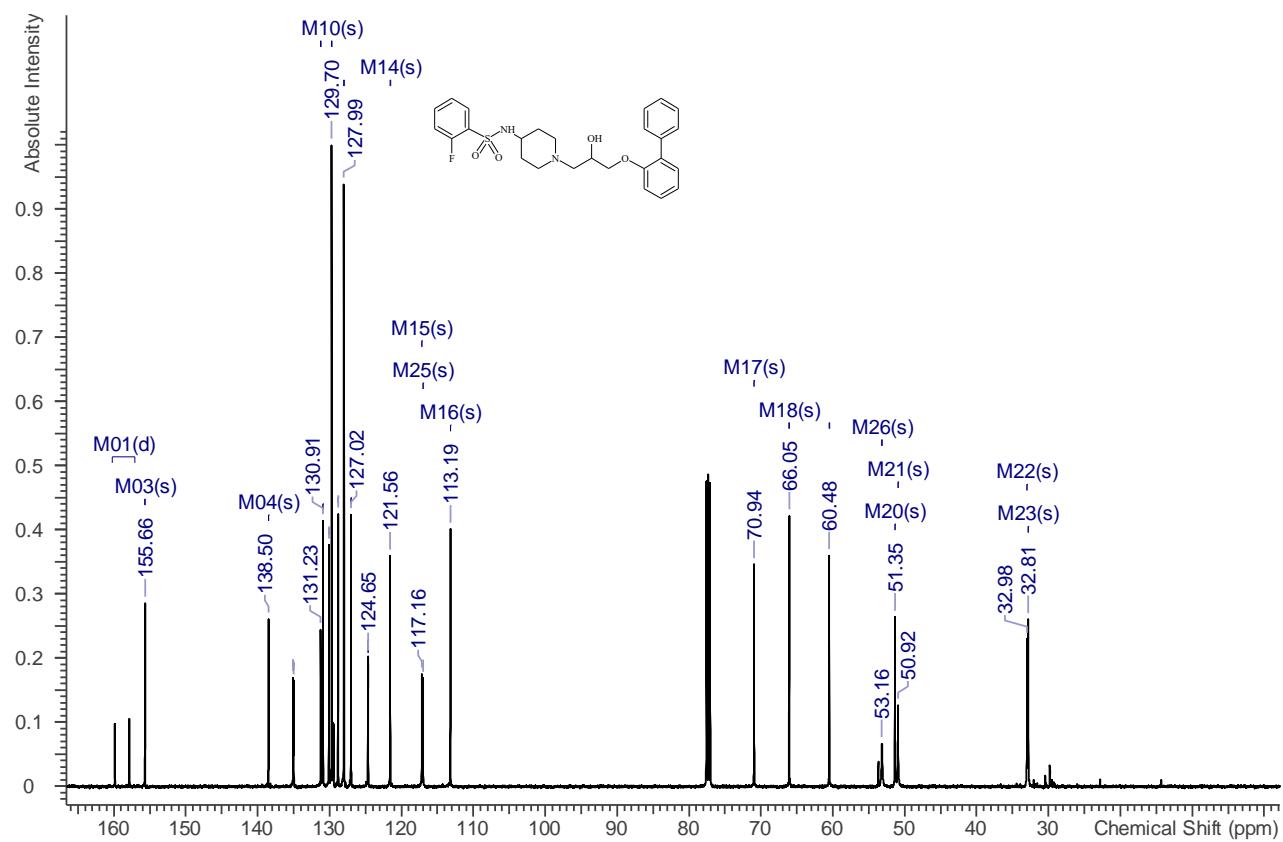
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

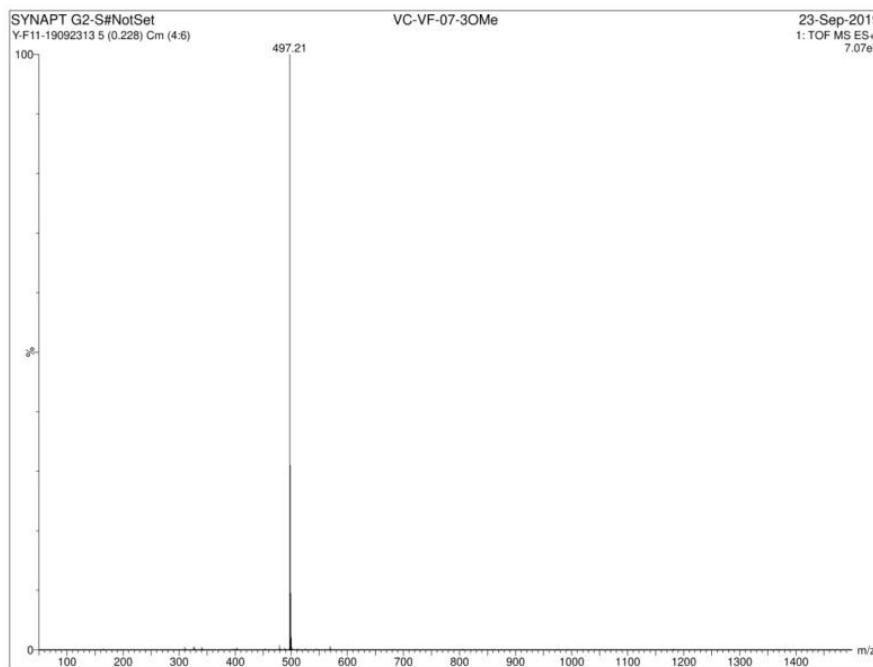


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

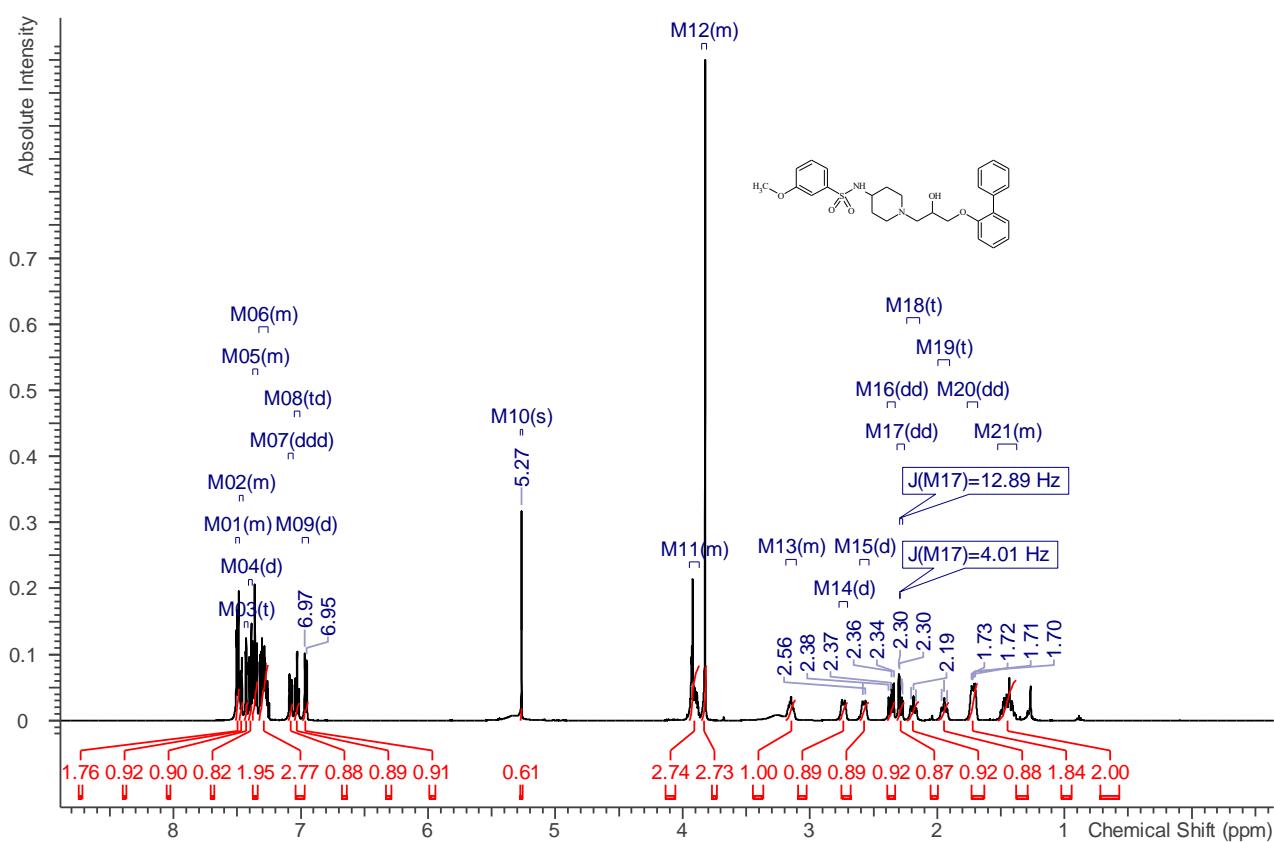


*3-Methoxy-N-(1-(3-((1,1'-biphenyl)-2-yloxy)-2-hydroxypropyl)piperidin-4-yl)-benzene sulfonamide*  
**(5d)**

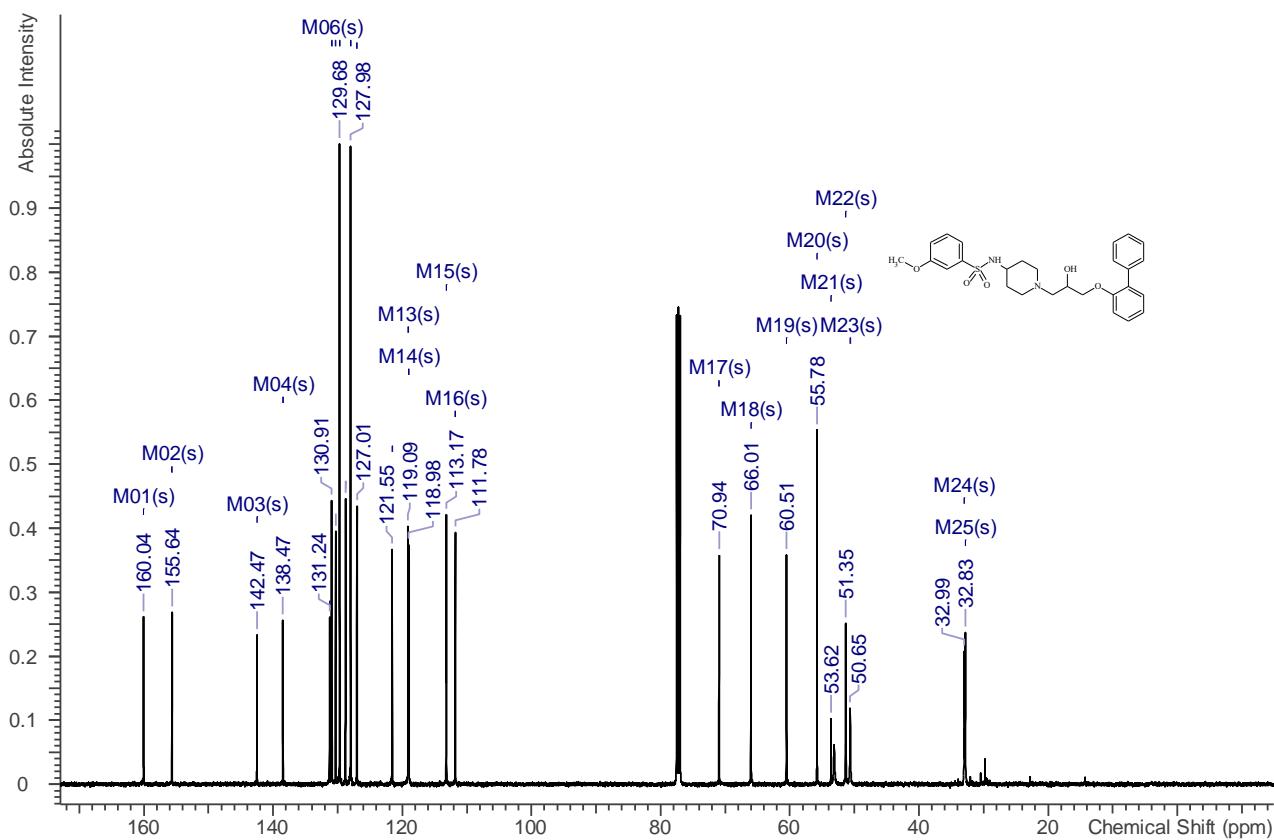
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)

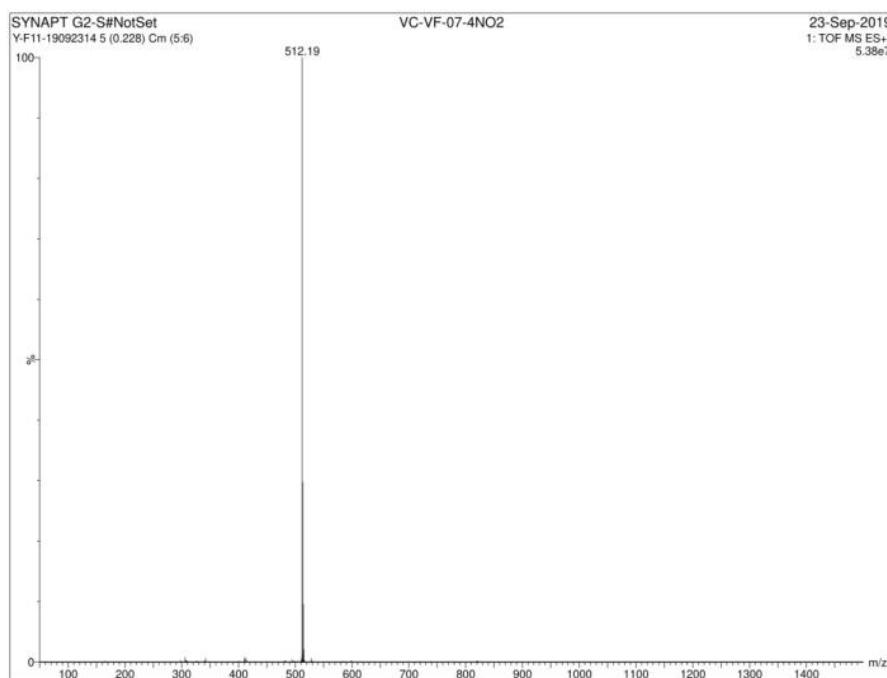


<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)

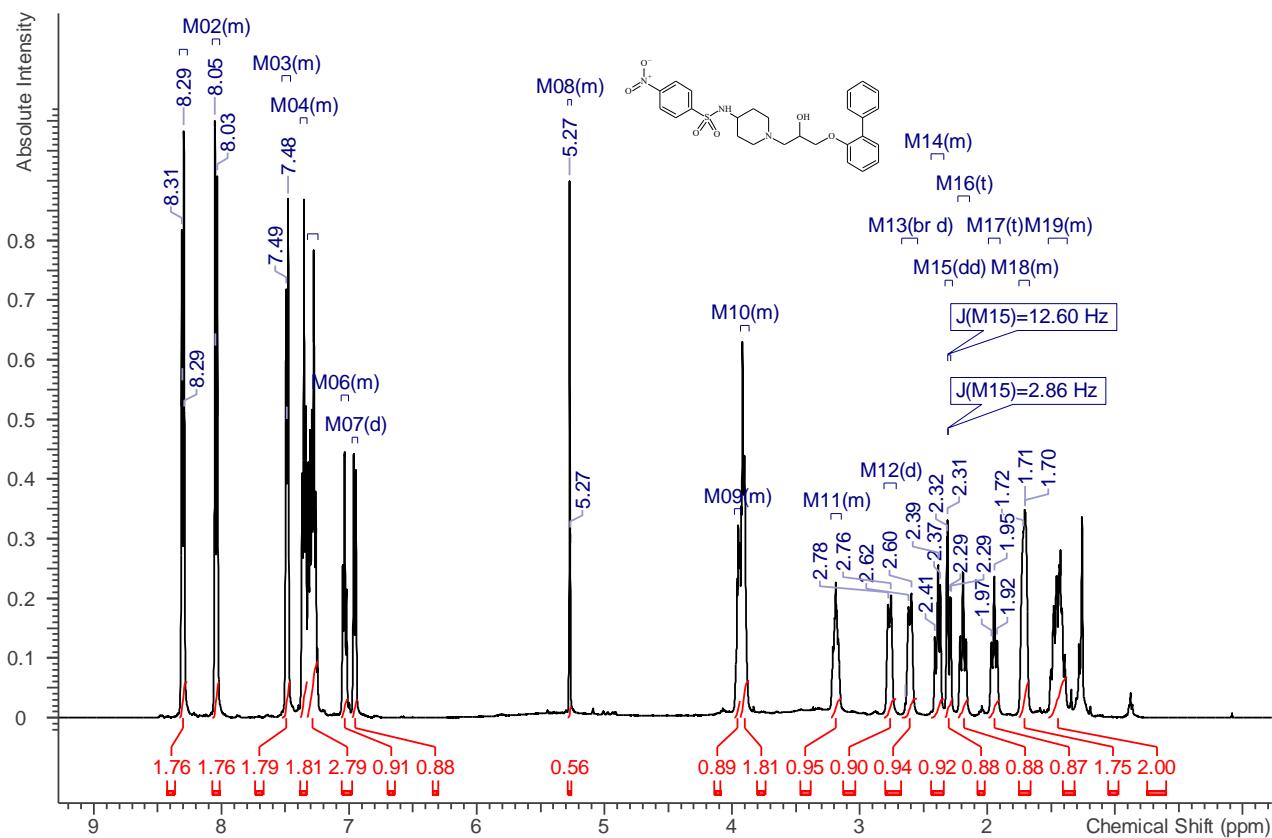


*4-Nitro-N-(1-(3-([1,1'-biphenyl]-2-yloxy)-2-hydroxypropyl)piperidin-4-yl)-benzene sulfonamide*  
**(5e)**

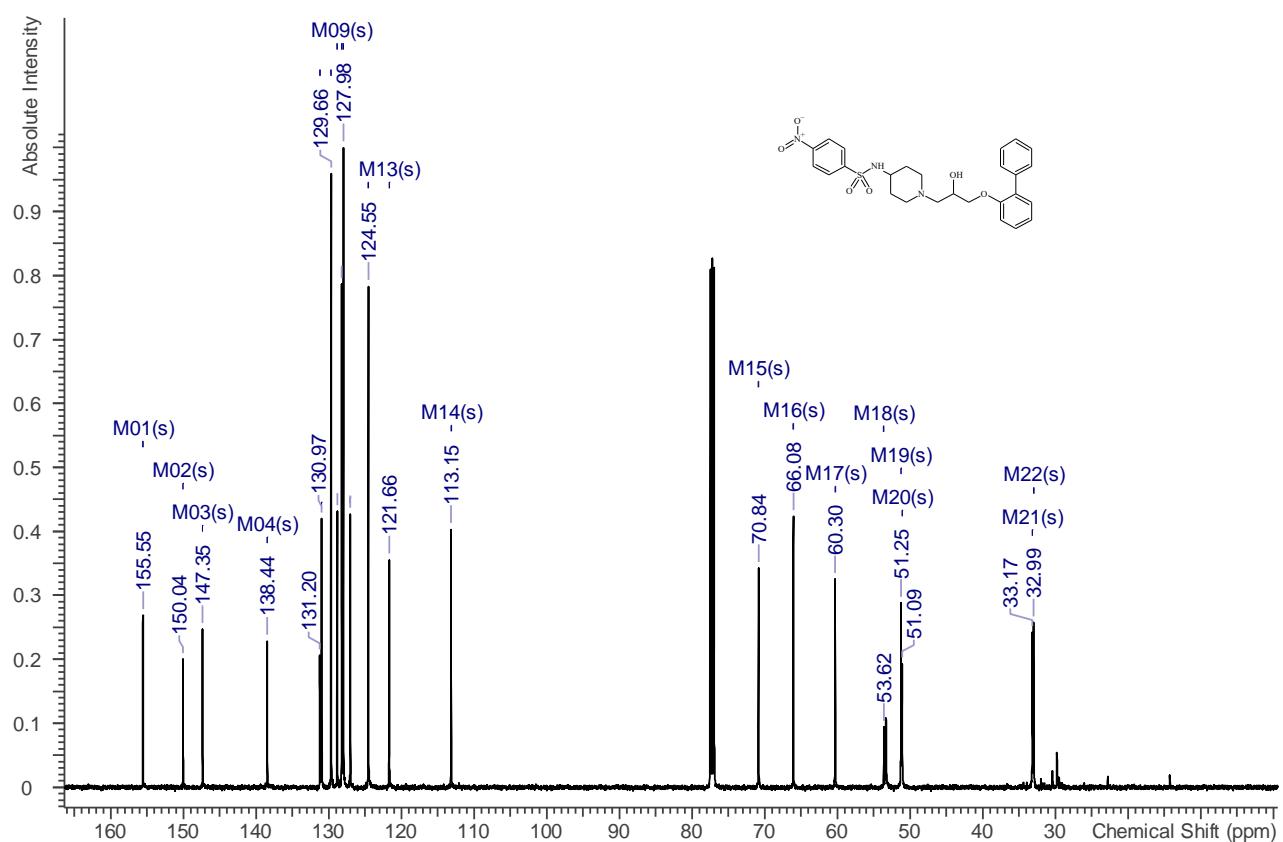
HRMS



<sup>1</sup>H-NMR (CDCl<sub>3</sub>, 500 MHz)



<sup>13</sup>C-NMR (CDCl<sub>3</sub>, 126 MHz)



#### **IV. References**

- [1] Roschangar, F.; Sheldon, R.A.; Senanayake, C.H. Overcoming barriers to green chemistry in the pharmaceutical industry – the Green Aspiration Level™ concept, *Green Chem.*, **2015**, *17*, 752–768.
- [2] Van Aken, K.; Strekowski, L.; Patiny, L. EcoScale, a semi-quantitative tool to select an organic preparation based on economical and ecological parameters. *Beilstein J. Org. Chem.*, **2006**, *2*, No 3.
- [3] Canale, V.; Partyka, A.; Kurczab, R.; Krawczyk, M.; Kos, T.; Satała, G.; Kubica, B.; Jastrzębska-Więsek, M.; Wesołowska, A.; Bojarski, A.J.; Popik, P.; Zajdel, P. Novel 5-HT<sub>7</sub>R antagonists, arylsulfonamide derivatives of (aryloxy) propyl piperidines: Add-on effect to the antidepressant activity of SSRI and DRI, and pro-cognitive profile. *Bioorg. Med. Chem.* **2017**, *25*, 2789–2799.
- [4] Hofer, A.; Cremosnik, G.S.; Müller, A.C.; Giambruno, R.; Trefzer, C.; Superti-Furga, G.; Bennett, K.L.; Jessen, H.J. A Modular Synthesis of Modified Phosphoanhydrides. *Chem. Eur. J.* **2015**, *21*, 10116–10122.